

TO: KEN JENISON, DRP  
FROM: SURRY R.I. OFFICE

April, '83

SRI Log

4/4-7 - AL, Burke

4/8 - FU on HF items, U-1 on tape work, U-2 operating, etc.  
9 AM - RT/Vepu phone mtg, discuss COA Lot 6  
COA letter finalized.

84 JUL 24 P 2: 37

8502090619 840809  
PDR FOIA  
MAYBERR84-574 PDR

4/12 - HF inspection - Reviewed RT 17.1 + 6 in result to ...  
x'd ... energy ... Co. ... height ... 2.4 ... per ...  
Here ...  
...  
All meters & TLD's checked for ... (12' for 1 MeV B)  
...  $CS^{124}$ ,  $CS^{137}$   $\beta$  ~ 0.1 to 0.2 MeV  
As ... TLD ...  $\beta$  ... correlation decreases  
... window ... Energy req'd to part. TLD ribbon.

# U-1 'C' Cubicle Air Activities

	EA	MYC	RWP
4/19 - 0416, <u>15'</u>		24.3%	1
4/20 - No samples - 0045 Press'ing for Type A			
4/21 - No samples			
4/22 - 0356 - Type A test 'sat', 0600 - depress'ing 3 psi/hr			
4/22 - 1816 - 47' C.V ~ 4%, no cubes			
4/23 - 0900 - (10.3) 8.4%			
etc.			

U-1 'C' Cubicle Air Samples (~10 min. sample,  
(-3'6" level, unless noted)

Date/Time	(E.g.) % MPC	RWP
Sat. - 4/23, 0900 -	8.4%	1
0933 (18') -	96.8%	1
1125 -	18.61 X MPC	446 - $C_2^{134-6-8}$ , $C_2^{137-1.2-7}$ , $C_2^{58-35-11}$ "60 1.5-9
1246 (past decon)	11%	446
(Pre-wk survey for snubbers work) 1241 - 18' level	153%	1
1610 -	23.7%	1023
1940 (no work at sample time) -	15.9%	1
2050 -	7.3%	1023
Sunday 4/24, 0100	11.2%	1023
0345	12%	"
0440 (18')	11%	" ?
0507	7%	"
0950	13.4%	1
1127	10%	361
ANS record discarded (9.5m pipe 14'C) 1245	7%	1
(1300 - sidpipe isolated)	90%	working snubbers
1617	14%	above platform
1700	11%	gen'l area
2258	8%	

Cont	Date/Time	(Σ <sub>n</sub> ) % MPC	RWP
	4/25, 0125 -	9%	
	0745	10%	
	0925 (18')	5%	
	0930	27.6%	
	1620	26.5%	Gen/Area
	1635 (18')	18%	CS 137 ~ 1.7-8 CS 134 ~ 9.1-9 CS 58 ~ 1.4-10
	2045	61.4%	RWP 1 Before work? 102.3
	4/26, 0930 -	68.5%	102.3
	1550	17.8%	"
	4/27, 0135 -	94.8%	102.3
	221.5 -	27.8%	"
	4/28, 0013 -	37.6%	102.3

# MEMORANDUM

TO File  
FROM Supervisor-Health Physics

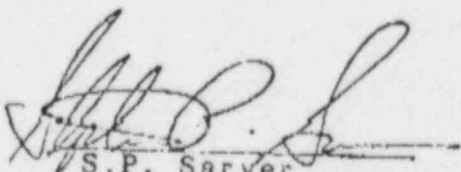
Surry Power Station

May 3, 1983

On March 11, 1983, Mr. M.L. Wells, HP Technician, found a PuBe neutron source (Source No. MRC-H-SS-W-PUS Bc 141) in the Calibration Source Room still in its transfer shield and not in the proper storage shield. Mr. Wells conducted and documented a survey of the source room to determine neutron dose rates which existed (copy attached). Upon investigation, it was determined that the source had been returned to the source room on February 26, 1983 after use by station instrument technicians in Unit No. 1 containment. Apparently, the person returning the source had not replaced it in its storage shield and, thus, it was left in a partially exposed position for the period 2-26-83 until 3-11-83 (13 days).

Mr. F. Weimer had worked in the source room decontaminating dosimeters during this time period and thus, assignment of a proper dose for this exposure is necessary. As indicated by the survey, the dose rates in Mr. Weimer's work area were approximately 1.0 to 2.5 mrem/hr at 5 inches from the floor and 1.0 to 1.5 mrem/hr at 3 feet from the floor. Mr. Weimer's estimated total occupancy time was 48 hours. Therefore, his maximum extremity exposure (ankle) was estimated to be 120 mrem and maximum whole body exposure was estimated to be 72 mrem.

The exposure estimated for this incident was added to Mr. Weimer's first quarter 1983 exposure records for an accumulated total of 1368 mrem extremity and 1920 mrem whole body. This total exposure remains below the permissible dose levels specified in 10CFR20.101(b).

  
S.P. Sarver  
Supervisor-Health Physics

*Pu-Be ~ .064 n/sec x 10<sup>-6</sup> / gm  
(α-n source) Curie  
Avg ~ 4.5 Mev neut's  
(Max ~ 10.7 Mev)*

UNCLASSIFIED

11 - Tech. Report (60) 100  
12 May 1983

10 File

request for Supervision for Home Control  
under assignment of New Line 100.

On 2-22-83 and 2-24-83 four instrument  
technicians and I were in source in the  
calibration of instruments. The source used  
was a 3 Ci PuBe source with an exposure  
level of 35.6 mREM at 18 inches.

Name	Dist. from source	Exposure Period	Dose Assigned
Chandler, H.J.	18"	12 min	0.007 REM
Crossin, A.W.	18"	12 min	0.007 REM
McGinnis, T.A.	18"	4 min	0.002 REM
Stalling, T.E.	18"	4 min	0.002 REM

SIGNED: *John W. [unclear]*

all stages of fuel-element treatment, it is necessary to make absolutely certain that a critical mass of fissile material is not attained either in the solid or in solution [24]. Second, because of the intense radioactivity of the fuel, even after cooling, special techniques and plant designs are required for carrying out the operations and for maintaining the equipment [25].

COOLING IRRADIATED FUEL ELEMENTS

The purpose of the so-called "cooling" phase is to permit the decay of radioactive nuclides present in the spent fuel. These nuclides include fission products and isotopes of elements of high mass number, e.g., 92 (uranium) and (8.113). The more significant consequences of cooling are the following: (1) the beta and gamma activity is decreased to a level at which radiocomposition of the processing reagents is not significant, (2) fission product half-life decay almost completely and the reprocessing operation is simplified by the decrease in the number of impurity elements which must be removed from the spent fuel, and (3) certain heavy isotopes, which cannot be separated chemically from the final products, decay to the point where their activity is no longer a problem.

In the extraction (or separation) phase of the reprocessing operation the use of solutions, the first two of these objectives are important and, as a general rule, so also is the third. In these circumstances an extended cooling period, on the order of 100 days, is required. For certain separation processes in which solutions are not employed and in which remote handling is quite possible, quite short cooling periods may be possible. From the economic standpoint, a decrease in cooling time is highly desirable. Cooling costs, which are charged for storage facilities, for use (or lease) of the fuel, and for the interest in the fuel inventory (§ 14.21 *et seq.*), can add appreciably to the cost of nuclear power.

According to equations (2.53) and (2.54), the gross beta and gamma activities of the fission products are roughly proportional to  $t^{-1/2}$ , where  $t$  is the time of removal from the reactor. Consequently, the gross activity after 100 days of cooling is about  $10^{-1}$  of the activity at 1 hr. Most of the decay occurs in the early stages, but the 100-day cooling period is often required for safety reasons.

The major contributions to the total activity made by individual fission products during the cooling period are shown by the curves in Fig. 8.113. These are based on the assumption that the fuel elements have been in the reactor for an extended period, so that at the time of their removal an equilibrium distribution of equilibrium has been attained. It is seen that after 100 days only about a dozen elements are responsible for nearly the whole of the activity. These are the elements from which uranium and plutonium are separated, and reprocessed, and which are the main contributors to the activity of the spent fuel.

8.113]

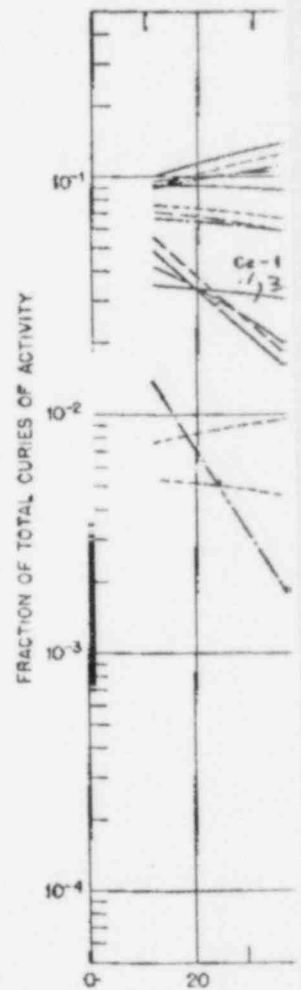


FIG. 8.113. Activity of

few elements, such as the rare earths, which are not removed because they have large half-lives. The manner in which the activity of a fuel consisting of uranium-235 decays is shown by arrows pointing to the right representing (alpha, beta) reactions with gamma decay. Of immediate interest are the (alpha, gamma) stages starting with uranium-238.

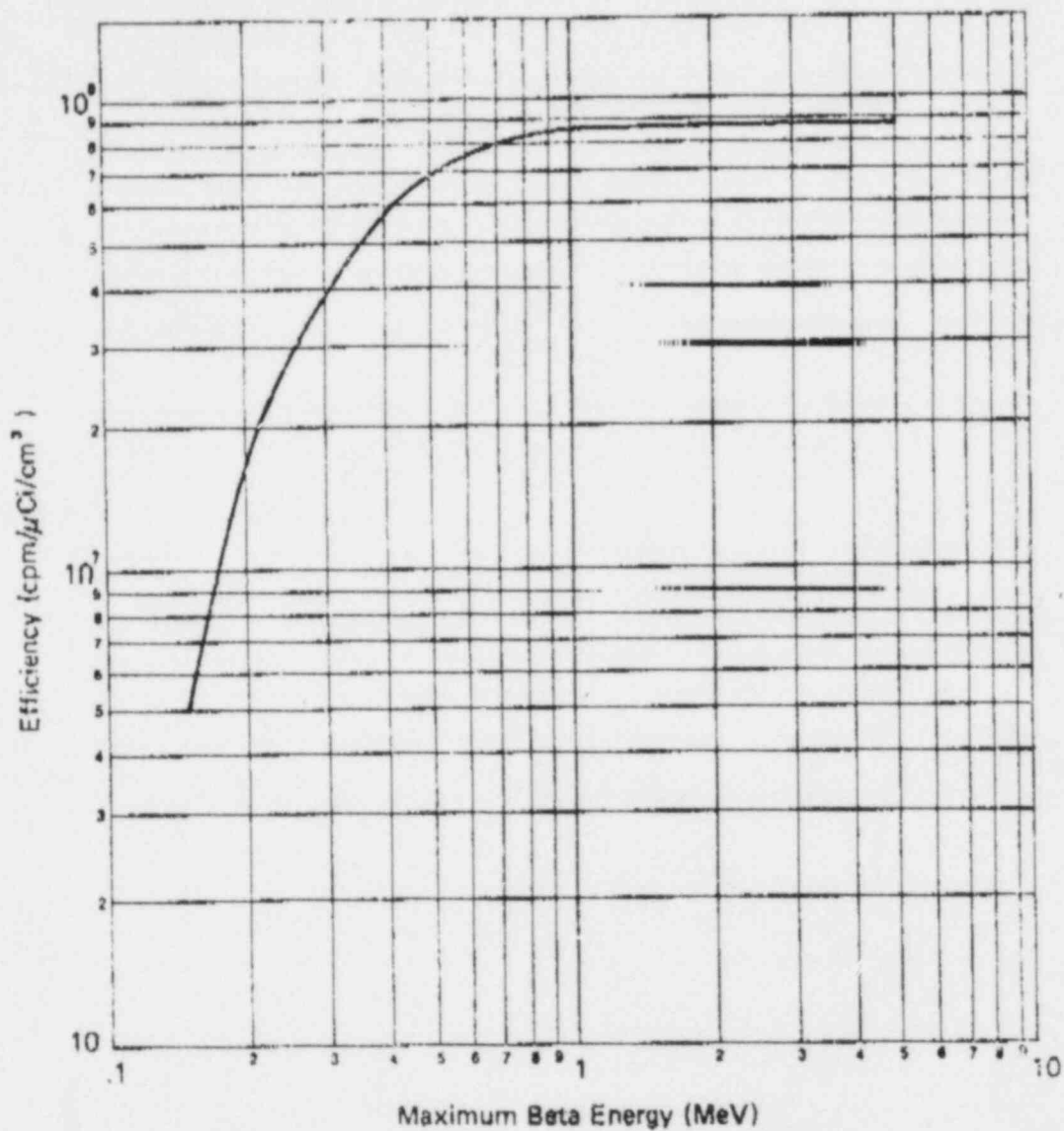


FIGURE 1

MODEL KDB BETA DETECTOR IN MODEL KSG GAS SAMPLER  
 EFFICIENCY TO RADIOACTIVE GAS VS. MAXIMUM BETA ENERGY  
 FROM POINT SOURCE ENERGY RESPONSE CORRECTED  
 TO Xe-133 & Kr-85 GAS EFFICIENCY



Alpha Particles

1. It requires an alpha particle of at least 7.5 MeV to penetrate the protective layer of the skin, 0.07 mm thick.
2. With 2π geometry, the surface of a thick source of tuballoy will give about 2,400 alpha cpm/cm<sup>2</sup>; plutonium will give about 70,000 alpha cpm/μg; 16.2 g of <sup>239</sup>Pu has an activity of 1 Ci.

Beta Particles

1. When working with <sup>198</sup>Au, experience has shown that under certain conditions, the beta dose will be five times the gamma value. Therefore, only 1/5 of the total dose will be recorded by gamma dosimeters.
2. It requires a beta particle of at least 70 keV to penetrate the protective layer of the skin, 0.07 mm thick.
3. The range (R) of beta particles in g/cm<sup>2</sup> (thickness in cm multiplied by the density in g/cm<sup>3</sup>) is approximately equal to the maximum energy (E) in MeV divided by 2 (i.e.,  $R \approx E/2$ ).
4. The range of beta particles in air is about 12 ft per MeV; for example, a 3 MeV beta has a range of about 36 ft in air.
5. A chamber wall thickness of 30 mg/cm<sup>2</sup> will reduce a flux of 1 MeV (max.) betas by 30% and a flux of 0.4 MeV betas by a factor of 4 or 5. *let's ~ 7 mg/cm<sup>2</sup> (windows)*
6. The intensity of bremsstrahlung increases approximately with the energy of the beta particle and about the square of the atomic number of the absorbing material.
7. When betas of 1 to 2 MeV pass through light materials such as water, aluminum, or glass, less than 1% of their energy is dissipated as bremsstrahlung.
8. The bremsstrahlung from 1 Ci <sup>32</sup>P aqueous solution in a glass bottle is about 1 mR/hr at 1 meter.
9. When the beta particles from a 1 Ci source of <sup>90</sup>Sr-<sup>90</sup>Y are absorbed, the bremsstrahlung hazard is approximately equal to that presented by the gamma from 12 mg of radium. The average energy of the bremsstrahlung is about 300 keV.
10. For a point source of beta radiation (neglecting self- and air-absorption) of strength Ci curies, the dose rate at 1 ft is approximately equal to 300 Ci rads/hr. The variation with energy is small over a wide range.
11. Beta-ray surface dose rates with 7 mg/cm<sup>2</sup> filter:

<u>Source</u>	<u>mrad/hr</u>
U slug. . . . .	233
UO <sub>2</sub> (brown oxide) . . . . .	207
UF <sub>4</sub> (green salt) . . . . .	179
UO <sub>2</sub> (NO <sub>3</sub> ) <sub>2</sub> ·6H <sub>2</sub> O (yellow uranyl nitrate hexahydrate) . . . . .	111
UO <sub>3</sub> (orange oxide) . . . . .	204
U <sub>3</sub> O <sub>8</sub> (black oxide) . . . . .	203
UO <sub>2</sub> F <sub>2</sub> (cliptite or uranyl fluoride) . . . . .	176
Na <sub>2</sub> U <sub>2</sub> O <sub>7</sub> (soda salt or sodium diuranate) . . . . .	167