

DUKE POWER COMPANY
PROCEDURE MAJOR CHANGE
PROCESS RECORD

(1) ID No: CP/0/A/8100/05
Change No: 2
Permanent/~~XXXXXXXXXX~~

- (2) STATION: Catawba
- (3) PROCEDURE TITLE: Chemistry Procedure for the Determination of Chloride
(Manual Method)
- (4) SECTION(S) OF PROCEDURE AFFECTED: 3.1 and 3.2
- (5) DESCRIPTION OF CHANGE: (Attach additional pages, if necessary.)

See Attachment

- (6) REASON FOR CHANGE:

Procedure Improvement

- (7) PREPARED BY: Cheryl M. Hize DATE: 12-1-82
- (8) SAFETY EVALUATION

This change:

Yes ☐ No ☒ Represents a change to the station or procedures as described
in the FSAR, or a test or experiment not described in the FSAR?
Yes ☐ No ☐ Requires a change to the station Technical Specifications?
Yes ☐ No ☒ Involves an unreviewed safety question?

If the answer to any of the above is "Yes", attach a detailed explanation.
As appropriate attach a completed "Nuclear Safety Evaluation Check List" form.

By: Cheryl M. Hize Date: 12-1-82

- (9) REVIEWED BY: R.H. Chest DATE: 12-1-82

Cross-Disciplinary Review By: (N/R) RHE

- (10) TEMPORARY APPROVAL (IF NECESSARY):

By: _____ (SRO) Date: _____
By: _____ Date: _____

- (11) APPROVED BY: H.S. Tuckman DATE: 12/3/82

- (12) MISCELLANEOUS:

Reviewed/Approved By: _____ Date: _____
Reviewed/Approved By: _____ Date: _____

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Attachment I
CP/O/A/8100/05

Change Section 3.1 to read:

Ferric, Nitrate, (10%)

Dissolve 100.0 ± 0.1 g of reagent grade Ferric Nitrate $[\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}]$ in 120 ± 1 ml of demineralized water. Add 750 ± 1 ml of concentrated Nitric Acid (HNO_3 , sp. gr. 1.42). Dilute to 1000 ± 1 ml with demineralized water. This solution is stable indefinitely.

Change Section 3.2 to read:

Mercuric Thiocyanate, (0.3%)

Dissolve 0.750 ± 0.001 g of Mercuric Thiocyanate $[\text{Hg}(\text{CNS})_2]$ in 250 ± 1 ml of reagent grade methanol (CH_3OH). Allow to stand for at least 24 hours. Filter & store in an amber reagent bottle. Do not use if more than 4 weeks old.