

1 UNITED STATES OF AMERICA  
2 NUCLEAR REGULATORY COMMISSION

DOCKETED  
USNRC

3 ATOMIC SAFETY AND LICENSING BOARD

'90 OCT 31 P2:42

4 Before Administrative Judge  
5 Peter B. Bloch

OFFICE OF SECRETARY  
DOCKETING & SERVICE  
BRANCH

6  
7 In the Matter of )  
8 )  
9 THE CURATORS OF )  
10 THE UNIVERSITY OF MISSOURI )  
11 )  
12 (Byproduct License )  
13 No. 24-00513-32; )  
14 Special Nuclear Materials )  
15 License No. SNM-247) )  
16 )

Docket Nos. 70-00270  
30-02278-MLA

RE: TRUMP-S Project

ASLBP No. 90-613-02-MLA

17 AFFIDAVIT OF DR. J. STEVEN MORRIS  
18 REGARDING PLUTONIUM CONTENT

19 I, J. Steven Morris, being duly sworn, hereby state as  
20 follows:

21 1. I am Interim Director of the University of Missouri-  
22 Columbia Research Reactor Facility ("MURR"), a position I have  
23 held since March 1, 1989.

24 2. I received a B.S. in Chemistry from Central Missouri  
25 State University in 1966 and a Ph.D. in Chemistry from the  
26 University of Missouri-Columbia in 1973. I have been employed at  
27 the MURR since 1973, in the positions of Radiochemist (1973 to  
28 1975), Research Scientist (1975 to 1978), Sr. Research Scientist  
29 (1978 to 1983) and as Group Leader, Nuclear Analysis Program  
30 (1983 to present), a position I hold concurrently with that of  
31 Interim Director.

32 3. In the foregoing positions I have had a variety of  
33 responsibilities of progressive importance under the NRC licenses  
34 relating to the MURR held by the Curators of the University of  
35 Missouri ("Licensee"). For three years (1975 to 1978) my job  
36 responsibilities at the MURR were principally that of reactor  
37 chemist. As such, my group evaluated all experiments involving  
38 the use of the reactor. This group was responsible for all  
39 chemistry quality assurance measurements as required by Reactor  
40 License No. R-103, Material Licenses No. SNM-247 and  
41 No. 24-00513-32, and the Technical Operating Specifications.  
42 During this period of time, and afterwards, I have conducted,  
43 supervised and published research related to the reactor  
44 chemistry function. From 1978 to the present, I have continued  
45 to have responsibility for the reactor chemistry function in a  
46 supervisory capacity. In 1988, I supervised the development of

1 the quality assurance program for another MURR materials license.  
2 Since 1978, I have served as a member of the Safety Subcommittee  
3 of the Reactor Advisory Committee at MURR. This Subcommittee, by  
4 charter, must review all changes in the operation, maintenance or  
5 utilization of the reactor that have any safety significance.  
6 Hence, based on my education and experience, I am qualified to  
7 evaluate the isotopic composition, quantity and safety  
8 significance of the plutonium involved in the TRUMP-S experiments  
9 at MURR under the amendment to NRC License No. SNM-247 issued on  
10 March 19, 1990.

11 4. I have reviewed the Written Presentation of Arguments of  
12 Intervenors and Individual Intervenors ("Intervenors' Written  
13 Presentation") (October 15, 1990) including Exhibit 1 thereof,  
14 and other relevant materials, including Intervenors' Renewed  
15 Request for Stay Pending Hearing ("Renewed Stay Request")  
16 (October 15, 1990). I have also reviewed the Summary of Actions  
17 Taken During Conference Call of October 19, 1990 ("Summary of  
18 Actions"), which requires that:

19 "a. Licensee will respond to Intervenors' allegation  
20 that the plutonium possessed by Licensee exceeded 2  
21 curies and that Licensee possessed plutonium in excess  
22 of the amount authorized by the subject license  
23 amendment; . . ."

24 Summary of Actions, at 3. Finally, I have reviewed the  
25 Memorandum and Order (Grant of Temporary Stay) ("Temporary Stay  
26 Order") (issued on October 20, 1990, reissued with editorial  
27 changes on October 22, 1990).

28 5. Intervenors have alleged that the Licensee's application  
29 dated February 20, 1990 for amendment of NRC License No. SNM-247  
30 (the "Application") was deficient because it failed to identify  
31 accurately the isotopic composition and curie content of the  
32 plutonium to be used by Licensee in the TRUMP-S experiments.  
33 Intervenors' Written Presentation at 16-19. They also allege  
34 that the Licensee possesses unlicensed plutonium. Id. at 19.

35 6. These allegations are based upon numerous errors by  
36 Intervenors and their experts in Intervenors' Written  
37 Presentation, including the principal exhibit thereof, Exhibit 1,  
38 "Declaration of the Review Panel." Because of the large number  
39 of such errors, this response must necessarily be quite  
40 comprehensive and will include discussions of the following:

- 41 1.) The impurities required to be identified in an NRC  
42 license application are any significant contaminants,  
43 taking into account their dose contribution.
- 44 2.) The Licensee's Application did not exceed the 2 Ci  
45 quantity of plutonium specified under 10 CFR § 70.22(i).

- 1 3.) The plutonium that the Licensee has received is a  
2 single 5 gram lot of New Brunswick Laboratory (NBL)  
3 Certified Reference Material (CRM) 127, formally  
4 National Bureau of Standards (NBS) Standard Reference  
5 Material (SRM) 945.
- 6 4.) Laboratory analyses of the five grams of NBL CRM 127  
7 and associated documentation demonstrate that Licensee  
8 has correctly identified the isotopic composition of  
9 the plutonium on its Application.
- 10 5.) The  $^{241}\text{Pu}$  and  $^{241}\text{Am}$  present in such plutonium are not  
11 significant contaminants, in the context of the  $^{239}\text{Pu}$   
12 and  $^{240}\text{Pu}$  standard material of which they are  
13 impurities, and therefore are not required to be  
14 identified on a license application.
- 15 6.) The total activity of the five grams of NBL CRM 127  
16 that Licensee currently possesses and the ten grams  
17 that Licensee could possess under the 10 gram license  
18 limit are less than 2 Ci including the  $^{241}\text{Pu}$  impurity.
- 19 7.) The Intervenors have misrepresented the curie content  
20 of plutonium that appears in the Application which in  
21 turn potentially misled the Presiding Officer.
- 22 8.) The Intervenors' experts (TRUMP-S Review Panel) have  
23 misrepresented the availability of plutonium with low  
24 isotopic composition of  $^{241}\text{Pu}$ .
- 25 9.) The Intervenors' experts (TRUMP-S Review Panel) have  
26 selectively misused isotopic composition data.
- 27 10.) The Intervenors' experts (TRUMP-S Review Panel) have  
28 miscalculated  $^{241}\text{Pu}$  activity.
- 29 11.) The Intervenors' experts (TRUMP-S Review Panel) did not  
30 consider obvious sources such as the National Bureau of  
31 Standards and New Brunswick Laboratory for obtaining  
32 certified plutonium reference standards.
- 33 12.) The Intervenors' experts (TRUMP-S Review Panel) only  
34 considered activity and ignored the concept of  
35 effective dose equivalent.

#### 36 Background--Plutonium Isotopic Composition

37 7. In filing its Application for amendment of the SNM-247  
38 license, Licensee followed Regulatory Guide 10.3, "Guide for the  
39 Preparation of Applications for Special Nuclear Material Licenses  
40 of Less Than Critical Mass Quantities" (RG 10.3). Section 4.3,

1 "Specifications of Special Nuclear Material", provides the  
2 specific guidance pertaining to the description of the isotopic  
3 composition. The pertinent part of Section 4.3 states:

4 "The special nuclear material requested should be  
5 identified by isotope, chemical or physical form,  
6 activity in curies, millicuries, or microcuries, and  
7 mass in grams. Specification of isotope should include  
8 principal isotope and significant contaminants. Major  
9 dose-contributing contaminants present or expected to  
10 build up are of particular interest." (Emphasis added)

11 8. Licensee knew that the plutonium to be supplied for the  
12 TRUMP-S research project would contain impurities of radioactive  
13 isotopes, including  $^{241}\text{Pu}$  and its daughter  $^{241}\text{Am}$ . However, for the  
14 reasons set forth below, such impurities did not have to be  
15 identified in the Application. Moreover, even if they had been  
16 included, Applicant would not have exceeded the 2 curie level of  
17 plutonium requiring the Licensee to address the need or actually  
18 submit an emergency plan under 10 CFR § 70.22(i). 1/

19 9. Licensee's position that it listed the appropriate  
20 isotopes in its Application and that  $^{241}\text{Pu}$  did not need to be  
21 listed is based on RG 10.3. As can be noted from the emphasized  
22 language in RG 10.3 quoted above, only "significant contaminants"  
23 are to be specified in an application, and significance is based  
24 upon the dose contribution of such contaminants in the context of  
25 the principal isotopes. The key term is dose-contributing. The  
26 applicability of reporting a minor or trace constituent depends  
27 on the dose that constituent contributes relative to the dose of  
28 the major constituents--in this case  $^{239}\text{Pu}$  and  $^{240}\text{Pu}$ . To address  
29 this further requires first that the record concerning the actual  
30 plutonium content be set straight.

31 Plutonium Certified Reference Material (NBL CRM 127)

32 10. The plutonium for which the Licensee applied and  
33 received authorization under License No. SNM-247 is a high-purity  
34 (99.8%) plutonium standard. Originally the material was

---

35 1/ Nevertheless, as discussed in more detail in the Affidavit  
36 of Walter A. Meyer Regarding Emergency Plan, the University  
37 of Missouri Research Reactor (MURR) Facility Emergency Plan  
38 does apply to the TRUMP-S activities. That Plan, along with  
39 the NRC correspondence related to its acceptance, is part of  
40 the supplement to the hearing file that was provided to the  
41 Intervenor and the Presiding Officer by the NRC Staff on  
42 August 16, 1990. Moreover, the Plan and associated  
43 procedures had been voluntarily made available to  
44 Intervenor by the Licensee on June 26, 1990.

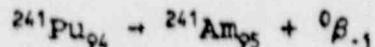
1 available from the National Bureau of Standards (NBS) 2/ as  
2 Standard Reference Material (SRM) 945. In 1987, NBS SRM 945 was  
3 transferred to the New Brunswick Laboratory (NBL) and became  
4 Certified Reference Material (CRM) 127. Copies of the  
5 "Certificate of Analysis" for NBL CRM 127 (Attachment 1), the  
6 "Certificate of Analysis" for NBS SRM 945--the previous NBS  
7 designation for this material (Attachment 1B), DOE/NRC Form 741  
8 (as completed by the NBL when the material was originally shipped  
9 to Rockwell International) (Attachment 2), and Form DOE-CH393(10-  
10 79) (also completed by NBL) (Attachment 3) are attached. The  
11 shipping papers and Form 741 (Attachment 4) for the transfer of  
12 the NBL CRM 127 from Rockwell International to the MURR at the  
13 University of Missouri-Columbia are also attached.

14 11. Forms 741 and CH393 (Attachments 2 and 3) were  
15 available to the Licensee at the time the Application was  
16 prepared. Data from these forms were used by the Licensee in  
17 completing that part of the Application regarding isotope  
18 composition. These forms include information on both NBL CRM 115  
19 (75 grams of depleted uranium) and NBL CRM 127 (5 grams of  
20 plutonium metal) because these two materials were shipped  
21 together by the NBL.

22 12. Form CH393 lists the total activity for both NBL CRM  
23 115 and NBL CRM 127 to be 390 millicuries which essentially can  
24 be attributed entirely to the plutonium standard 3/. The  
25 plutonium is listed as 4.72 grams <sup>239</sup>Pu and 0.279 grams <sup>240</sup>Pu.  
26 These masses of <sup>239</sup>Pu and <sup>240</sup>Pu translate to 94.42 and 5.58 weight  
27 percent for <sup>239</sup>Pu and <sup>240</sup>Pu, respectively. These are precisely the  
28 numbers used by the Licensee in the Application. See page  
29 1, ¶2(d) of Licensee's Application (Attachment 5).

30 13. On Form 741, NBL did not list <sup>241</sup>Pu or <sup>241</sup>Am. On Form  
31 CH393 it did not list <sup>241</sup>Am, and it checked <sup>241</sup>Pu as present but  
32 left the gram content as blank. The reason that quantities for  
33 <sup>241</sup>Pu and <sup>241</sup>Am were not listed was that they are present only at  
34 trace levels. Since Licensee knew that trace levels of these  
35 isotopes not listed on the shipping documents could not be  
36 significant contaminants in the context of the <sup>239</sup>Pu and <sup>240</sup>Pu, it  
37 did not list them on the Application.

38 14. The relationship between <sup>241</sup>Pu and <sup>241</sup>Am is shown in the  
39 following equation:



41 2/ NBS is now known as the National Institute for Standards and  
42 Technology (NIST).

43 3/ The 75 grams of depleted uranium accounts for only ~30µCi  
44 (0.03 mCi) of the total activity for that shipment.

1 Of particular note is the fact that the decay of  $^{241}\text{Pu}$  is almost  
 2 entirely via beta emission (99+ %) and only a small fraction via  
 3 alpha emission (0.00245%). The radiological risk due to  
 4 plutonium is essentially entirely due to the internal dose, and  
 5 alpha emitting isotopes are of much greater concern than beta  
 6 emitting isotopes. Therefore, the alpha emitting Pu isotopes are  
 7 of principal regulatory interest, and the beta-emitting  $^{241}\text{Pu}$  does  
 8 not contribute significantly to the overall potential dose  
 9 assessment in high-isotopic-purity samples such as NBL CRM 127.  
 10 The  $^{241}\text{Am}$  that builds up from the decay of  $^{241}\text{Pu}$  is an alpha  
 11 emitter; however, it does not contribute significantly to the  
 12 overall dose because its specific activity (curies/gram) is a  
 13 factor of 30 less than that of  $^{241}\text{Pu}$  which is already at a trace  
 14 level with respect to mass.

15 15. The certificate for NBL CRM 127 lists the  $^{241}\text{Am}$  content  
 16 from the beta decay of  $^{241}\text{Pu}$  to be approximately 146  $\mu\text{g}$  of  $^{241}\text{Am}$   
 17 per gram of plutonium as of September 1970. The certificate also  
 18 states that the build up of  $^{241}\text{Am}$  is less than 150  $\mu\text{g}$  per gram of  
 19 plutonium per year. From these two values and the half-lives of  
 20  $^{241}\text{Pu}$  and  $^{241}\text{Am}$ , reasonable estimates can be computed regarding the  
 21 present-day levels of these isotopic impurities. The  
 22 calculations have been summarized and attached (Attachment 6).  
 23 The isotopic distribution for NBL CRM 127 including the  $^{241}\text{Pu}$  and  
 24  $^{241}\text{Am}$  impurity levels are given in Table 1 below.

25 Table 1. Total mass and activity for each plutonium isotope in  
 26 CRM-127 as of September 1990 4/.

27	28	29	Isotope	wt. %	License Limit at	
					MURR is 10 grams	
					grams	curies
30			$^{239}\text{Pu}$	94.42	9.44	0.586
31			$^{240}\text{Pu}$	5.58	0.558	0.126
32			$^{241}\text{Pu}$ ( $\beta$ )	< 0.1	< 0.01	< 1
33			$^{242}\text{Pu}$	< 0.1	< 0.01	< $4 \times 10^{-5}$
34			$^{241}\text{Am}$	< 0.2	< 0.02	< 0.07
35			total plutonium alpha			0.712 Ci
36			total alpha activity (including			
37			americium)			0.782 Ci
38			total plutonium activity ( $\alpha + \beta$ )			1.71 Ci

39 4/ Data are taken or computed from Forms 741, CH:93 and the CRM  
 40 127 certificate. Less-than values (<) result from  
 41 calculations based on less-than values in the certificate.  
 42 The plutonium was received by Licensee in September 1990.

1 16. In addition to the data in Table 1 from the NBL CRM 127  
2 certificate and other documents, the isotopic composition was  
3 determined in 1975 at Los Alamos National Laboratory (LANL). A  
4 copy of a 1982 letter to Dr. Thomas Gills at NBS discussing these  
5 analysis results is attached (Attachment 7). The data in the  
6 table on page two of that letter, under the "Feb-1975" column  
7 heading are given in atom percent. These data converted to  
8 weight percent and decay corrected to September 1990 (date CRM  
9 127 standard was received at MURR) are given in Table 2.

10 **Table 2. Isotopic composition (wt%) and curies per 10 grams for**  
11 **NBL CRM 127 (NBS SRM 945) from 1975 LANL analysis decay corrected**  
12 **to September 1990.**

13	<u>Pu Isotope</u>	<u>wt. %</u>	<u>curies per 10 grams</u>
14	<sup>238</sup> Pu	0.0081	0.014
15	<sup>239</sup> Pu	94.2	0.584
16	<sup>240</sup> Pu	5.52	0.126
17	<sup>241</sup> Pu	0.116	1.21
18	<sup>242</sup> Pu	0.018	< 1 x 10 <sup>-6</sup>
19	total alpha activity		0.728 <sup>5/</sup>
20	total plutonium activity (α + β)		1.94

21 These results are in good agreement with those reported in  
22 Table 1, especially for the major isotopes, <sup>239</sup>Pu and <sup>240</sup>Pu. In  
23 addition, the atomic weight computed from the atom percent data  
24 given in the 1982 letter discussing the 1975 LANL analysis  
25 (Attachment 7), and the atomic weight of the plutonium isotopes  
26 as given in the Chart of the Nuclides is 239.11. This is in  
27 excellent agreement with the value of 239.12 as given in the 1973  
28 NBS certificate (Attachment 1B) and the NBL certificate  
29 (Attachment 1) for this plutonium standard.

30 Intervenors Have Misrepresented Licensee's Application--.07 Ci is  
31 not 0.71 Ci

32 17. Intervenors have argued in their direct case that the  
33 Licensee has failed to comply with 10 CFR § 70.22(a)(4). They  
34 begin this argument by stating:

35 "Regulatory Guide 10.3 is of considerable significance  
36 in this respect, and should be read in its entirety."  
37 (Emphasis added) Intervenors' Written Presentation at  
38 10.

39 <sup>5/</sup> Does not include <sup>241</sup>Am data.

1 Intervenor continue at pages 10 and 11 to quote a part of  
2 Section 4.3 of RG 10.3:

3 "The special nuclear material requested should be  
4 identified by isotope, chemical or physical form,  
5 activity in curies, millicuries, or microcuries, and  
6 mass in grams. Specification of isotope should include  
7 principal isotope and significant contaminants."

8 Note that even though intervenors have instructed that RG 10.3  
9 should be read in its entirety, they conveniently conclude their  
10 quotation of Section 4.3 immediately prior to the following  
11 sentence:

12 "Major dose-contributing contaminants present or  
13 expected to build up are of particular interest."  
14 (Emphasis added)

15 This sentence is critical in deciding what impurities should be  
16 listed in a materials license application. It establishes that a  
17 major factor in the decision is dose-contribution, not merely  
18 activity. This omission is relevant and will be further  
19 discussed below.

20 At page 16, intervenors state:

21 "There is realistically no way that the authorized 10  
22 grams of plutonium would contain only .07 curies, as  
23 claimed by the University at page 1, § 2(d) of the  
24 application." (Emphasis added)

25 At page 17 intervenors state:

26 "...Absent such an affidavit, the University's  
27 extraordinary representations that the plutonium  
28 represents only .07 curies, obviously based on hearsay,  
29 bordering on fantasy, cannot withstand Professor Warf's  
30 analysis." (Emphasis added)

31 At page 18 intervenors state:

32 "The definition of § 70.4(r) excepts research and  
33 development activities utilizing 'unsubstantial  
34 amounts' of plutonium. Believing that only .07 curies  
35 of plutonium were involved, the Staff might reasonably  
36 conclude that this is an "unsubstantial amount" and  
37 therefore not within the definition." (Emphasis added)

38 18. In each of these statements, intervenors state that the  
39 Licensee has represented in its application that the curie  
40 content of the 10 grams of plutonium requested is .07 curies.  
41 This is not correct. In the Application at page 1, § 2(d)

1 (Attachment 5), Licensee correctly listed the total activity of  
2 the  $^{239}\text{Pu}$  and  $^{240}\text{Pu}$  as 710 mCi, 10 times the number stated by  
3 intervenors.

4 19. Licensee notes that intervenors filed intervenors'  
5 correction on this point on October 25, 1990. Licensee does not  
6 agree that the substance of the points is not affected by the  
7 error--especially given the sarcastic rhetoric used by the  
8 intervenors. The misstated curie content could well have  
9 influenced the Presiding Officer's decision to temporarily stay  
10 the experiments. The curie content is correctly stated in  
11 intervenors' Exhibit 1, however there is no indication in the  
12 Temporary Stay Order that the Presiding Officer had observed the  
13 discrepancy and was acting on the correct version. At any rate,  
14 the error reflects on the competence of the intervenors, not the  
15 licensee. Unfortunately, it was not perceived that way by the  
16 Presiding Officer or the local and national media, which have  
17 characterized the licensee as incompetent based on language used  
18 by the Presiding Officer in the Temporary Stay Order that was put  
19 in place prior to allowing licensee to respond.

20 Intervenors Have Misrepresented the Availability of Plutonium  
21 With Low Isotopic Composition of Pu-241

22 20. Intervenors argue that all plutonium must contain at  
23 least 5.3 Ci of  $^{241}\text{Pu}$  per 10 grams of total plutonium, and perhaps  
24 as much as 120 Ci. Intervenors' Written Presentation at 16 and  
25 Exhibit 1 at page 9, ¶27 and ¶29. They then contrast this with  
26 the .07 Ci of total plutonium that they contend the licensee used  
27 in its Amendment application. As licensee has now shown, the  
28 quantitative aspects of the intervenors' argument are entirely  
29 incorrect. The .07 Ci figure for plutonium alpha activity should  
30 be 0.71 Ci and the 5.3 to 120 Ci range for  $^{241}\text{Pu}$  beta activity  
31 should be  $\approx 1$  Ci in the case of 10 grams of the NBL CRM 127  
32 plutonium standard.

33 21. Intervenors at page 17, citing Dr. Warf and the "TRUMP-  
34 S Review Panel" 6/, 7/ have argued that licensee should  
35 demonstrate the actual isotopic composition of the plutonium that  
36 it possesses. Licensee has shown by Certificate of Analysis,  
37 issued first by the United States National Bureau of Standards  
38 (Attachment 1B), that CRM 127 has a different isotopic

---

39 6/ At Exhibit 1, ¶3, the intervenors TRUMP-S Review Panel is  
40 defined to consist of: James C. Warf, Daniel Hirsch,  
41 Sheldon C. Plotkin, Miguel Pulido and Lowell Wayne.

42 7/ The Exhibit 1, Declaration of the Review Panel, paragraphs  
43 relevant to this discussion include but may not be limited  
44 to: ¶15 through ¶36 inclusive (pages 6-11), ¶54[a] (page  
45 15), and Appendix I (no page number).

1 composition than those materials cited by Warf 8/, et.al, at  
2 Exhibit 1, §17 and §18 and in Appendix I of Exhibit 1.

3 22. Dr. Warf, et al., have stated at Exhibit 1, page 7, §18:  
4 "The lowest figure we have seen reported for <sup>241</sup>Pu is 0.44 wt% for  
5 twenty-year-old plutonium, which would result in about 0.53  
6 Ci/gram WG-Pu 9/, or ten grams representing about 5.3 curies  
7 total." In point of fact, 0.44 wt% is:

$$8 \quad 0.0044 \times 103 \text{ Ci/g } 10/ = 0.45 \text{ Ci/g}$$

9 Therefore 10 grams would be 4.5 Ci not the 5.3 Ci reported by the  
10 TRUMP-S Review Panel.

11 23. The fact that the lowest percentage of <sup>241</sup>Pu found by  
12 Dr. Warf and the Review Panel in the literature was 0.44 wt% is  
13 irrelevant. Licensee has shown that a certified reference  
14 material exists that is considerably lower. This reference  
15 material was once available from the National Bureau of Standards  
16 (NBS) and now from the New Brunswick Laboratory (NBL) and is  
17 listed in catalogues from both of these agencies.

18 24. It is astounding that a scientist concerned with the  
19 development of chemical or physical methods would not consult the

---

20 8/ As a separate point Intervenor's have been critical of  
21 Licensee (specifically Dr. Morris) in his citations of  
22 release factors, accusing him of citing only low values  
23 rather than representative values (i.e., Intervenor's  
24 Written Presentation at p. 39 and Exhibit 1, p. 21, § 73).  
25 Justification for these citations will be given in  
26 Licensee's response to be filed as part of its direct case.

27 The following is of interest: Warf, et.al., at §17 of  
28 Exhibit 1, has listed data "adapted from J.C. Warf, All  
29 Things Nuclear, page 117". A copy of that page is attached  
30 (Attachment 8). The entries for <sup>239</sup>Pu and <sup>241</sup>Pu are of  
31 particular interest because in his book, Dr. Warf cites <sup>239</sup>Pu  
32 to be 92.0-92.4% and selects 92.0 to enter in Table I of  
33 §17, Exhibit 1. On the other hand, for <sup>241</sup>Pu, Dr. Warf cites  
34 the range to be 1.7-2.0% and selects 2.0% to enter in  
35 Table I. Licensee would be interested in receiving Dr.  
36 Warf's technical rationale for selecting the low end of the  
37 <sup>239</sup>Pu range and the high end of the <sup>241</sup>Pu range.

38 9/ Licensee believes that WG-Pu stands for weapons-grade  
39 plutonium.

40 10/ The 103 Ci/g factor for <sup>241</sup>Pu is taken directly from Exhibit  
41 1, Appendix I. Licensee has checked this factor and agrees  
42 that it is correct.

1 National Bureau of Standards (formally NBS, now NIST) prior to  
2 concluding that a desired material or method was not available.  
3 The NBS has both a national and international reputation in the  
4 development, characterization and distribution of standard  
5 reference materials. In fact, the NBS was established as a  
6 methods and materials resource by an act of Congress on March 3,  
7 1901. They have catalogued their materials and services since  
8 1906 (Attachment 9)11/.

9 25. Had Dr. Warf, or others on the Intervenor's TRUMP-S  
10 Review Panel checked any recent NBS catalog, up to 1987  
11 (Attachment 10)12/, they would have found:

- 12 1.) The catalogs (up to 1987) contain plutonium standards,  
13 even some technical information concerning isotopic  
14 composition for some of the standards.
- 15 2.) Addresses and telephone numbers to obtain more  
16 information specific to special nuclear materials.
- 17 3.) An open letter of invitation to users from S.D.  
18 Rasberry (Chief of the Office of Standard Reference  
19 Materials) beginning with the words: "Please do not  
20 hesitate to call us if you have any questions about the  
21 SRMs (Standard Reference Materials) described, their  
22 availability, or if you cannot find what you need."

23 26. In the most recent NBS SRM Catalog (1990-91)  
24 (Attachment 11)13/, information is included directing  
25 inquiries concerning special nuclear material standards to the  
26 New Brunswick Laboratory (NBL) even though the NBS had  
27 transferred responsibility for distribution of SNM materials to  
28 NBL on October 1, 1987.

29 27. Had Dr. Warf or other members of the TRUMP-S Review  
30 Panel pursued their interests in plutonium to the New Brunswick  
31 Laboratory, they would have discovered a number of plutonium

---

32 11/ Attachment 9 is a copy of the cover, and inside cover page,  
33 of the 1982-83 Edition of the NBS Standard Reference  
34 Materials Catalog recognizing their 75th anniversary.

35 12/ Attachment 10 consists of a copy of the cover of the 1986-87  
36 Edition of the NBS SRM Standard Reference Material Catalog,  
37 an open letter to users from Chief Rasberry, and page 78  
38 illustrating information on plutonium standards.

39 13/ Attachment 11 consists of the 1990-91 Edition NBS SRM  
40 Catalog cover page and page 10 directing inquiries on  
41 special nuclear materials standards to the New Brunswick  
42 Laboratory.

1 standards (Attachment 12)<sup>14/</sup> are available. For some of  
2 these--for example, CRM 138 listed on catalog page 5, see  
3 Attachment 12--there are data in the catalog showing isotopic  
4 compositions of <sup>241</sup>Pu less than the 0.44 wt.% that Warf, et.al.,  
5 state is "the lowest figure we have seen reported for <sup>241</sup>Pu in  
6 weapons-grade plutonium...."(Intervenors Exhibit 1, page 7, ¶18).  
7 Whether or not the material is weapons grade plutonium is  
8 irrelevant. For any of these materials, including the CRM 127  
9 standard being used in the actinide experiments at the MURR,  
10 information can be readily obtained by calling the New Brunswick  
11 Laboratory.

12 28. These steps just outlined are not difficult nor  
13 innovative. They are ones taken every day by working scientists  
14 and engineers around the world. The question remains as to why  
15 the TRUMP-S Review Panel did not check with such an obvious  
16 possibility as the National Bureau of Standards for information  
17 on the availability of a standard before they took this  
18 litigation down such a non-productive pathway that departs so  
19 greatly from any legitimate concern regarding public safety.

#### 20 Relative Radiological Hazards of the Plutonium Isotopes

21 29. Although the activity of <sup>241</sup>Pu is greater than the  
22 activity of <sup>239</sup>Pu and <sup>240</sup>Pu, the radiological hazard associated  
23 with <sup>241</sup>Pu, a beta emitter, is much less than that associated with  
24 <sup>239</sup>Pu and <sup>240</sup>Pu, both of which are alpha emitters. This difference  
25 in radiological risk is reflected in the Nuclear Regulatory  
26 Commission regulations, e.g., 10 CFR Part 20, Appendix B, and  
27 10 CFR Part 71, Table A-2. Part 20, Appendix B, permits  
28 concentration in air above background for <sup>241</sup>Pu at a level 50  
29 times greater than for the alpha-emitting <sup>239</sup>Pu and <sup>240</sup>Pu isotopes  
30 (soluble). Similarly Part 71, Table A-2 defines the activity  
31 limits not to be exceeded for shipping Type A quantities of  
32 radioactive materials. The <sup>241</sup>Pu activities listed in this table  
33 are at least 50 times greater than the corresponding ones for  
34 <sup>239</sup>Pu.

35 30. This factor of 50 also appears in NUREG-1140, "A  
36 Regulatory Analysis on Emergency Preparedness for Fuel Cycle and  
37 Other Radioactive Material Licensees - Final Report" (Jan 1988)  
38 which was prepared by the NRC in considering whether emergency  
39 preparedness requirements should be imposed on certain fuel cycle  
40 and radioactive materials licensees. Table 13 of NUREG-1140,  
41 "Quantities of Radioactive Materials Requiring Evaluation of the  
42 Need for Offsite Emergency Preparedness" (based on 1 rem

---

43 <sup>14/</sup> Attachment 12 consists of pages from the most current  
44 edition (1988) of the New Brunswick Laboratory catalog of  
45 Certified Reference Materials.

1 effective dose equivalent outside the building), identifies a  
2 dose conversion factor (in rem/ $\mu$ Ci inhaled) for many radioactive  
3 materials and, taking into account solubility class and release  
4 fraction, calculates a quantity for each material that would  
5 require evaluation for need for emergency preparedness. For  
6  $^{239}\text{Pu}$ ,  $^{240}\text{Pu}$  and  $^{241}\text{Pu}$ , these result in action levels of 2, 2 and  
7 100 curies, respectively.

8 31. These numbers point up the misleading nature of  
9 intervenors' argument. By simply adding their alleged (and  
10 incorrectly derived, as shown above) quantities of  $^{239}\text{Pu}$ ,  $^{240}\text{Pu}$  and  
11  $^{241}\text{Pu}$ , they have argued that Licensee exceeds the 2 curie level  
12 for plutonium set forth in 10 CFR  $\S$  70.22(i). As demonstrated  
13 above, even if beta activity is blindly added to alpha activity,  
14 Licensee does not exceed the 2 curie level.

15 32. But, if any attempt is made to calculate rationally the  
16 dose contribution of the  $^{241}\text{Pu}$ , it is shown to be negligible in  
17 the context of  $^{239}\text{Pu}$  and  $^{240}\text{Pu}$ . This could be done by calculating  
18 the potential impact of the additional isotopes based upon their  
19 effective dose equivalent. 15/ A common mechanism for such  
20 calculation is computing a sum of the ratios. 16/

21 33. Table 1 of my affidavit above shows the quantities of  
22  $^{239}\text{Pu}$  and  $^{240}\text{Pu}$  for a 10 gram CRM 127 sample. For those two  
23 isotopes the curie quantities listed in Table 13 of NUREG-1140  
24 (as upper limits based on 1 rem effective dose equivalents) are 2  
25 curies and 2 curies, respectively. Thus, considering only those  
26 two isotopes, the following sum of the ratios is computed:

27 sum of ratios =  $[(C_{i_{\text{Pu-239}}}/2) + (C_{i_{\text{Pu-240}}}/2)]$

28 sum of ratios =  $[(0.586/2) + (0.126/2)] = \underline{0.356}$

29 If the  $^{241}\text{Pu}$  is added (considering its 100 curies level in  
30 Table 13 of NUREG-1140), the trace quantity shown in my Table 1  
31 adds a value of  $(1 \text{ curie}/100) = 0.01$ , which brings the foregoing

---

32 15/ Even if Licensee requested authorization for more than  
33 2 curies of plutonium, an alternative under 10 CFR  
34  $\S$  70.22(i)(1)(i) would be to provide: "An evaluation  
35 showing that the maximum dose to a member of the public  
36 offsite due to a release of radioactive materials would not  
37 exceed 1 rem effective dose equivalent . . . ."

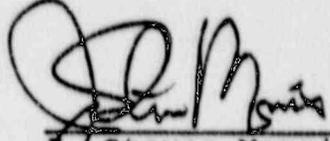
38 16/ See for example the Note appended to Part 20, Appendix B.

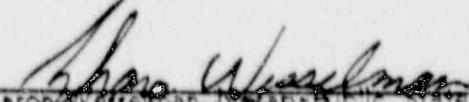
1 sum of the ratios to a total of 0.366. This is an insignificant  
2 addition. 17/

3 Summary:

4 34. The Intervenors and their experts (TRUMP-S Review  
5 Panel) have made false accusations based on their failure to take  
6 simple steps to ascertain the possibility that Licensee had  
7 properly represented the isotopic and curie content of the  
8 plutonium in the Application. Moreover, they have chosen to make  
9 simplistic arguments based on the activity of  $^{241}\text{Pu}$  rather than  
10 its effective dose equivalent. In so doing they have diverted  
11 the attention of this proceeding away from public safety and  
12 hence the public's interests. In making their arguments, the  
13 Intervenors have altered the record concerning Licensee's  
14 Application, misled the Presiding Officer regarding isotopic  
15 composition of plutonium, mistakenly accused the Licensee of  
16 being in violation of regulations, miscalculated  $^{241}\text{Pu}$  activity,  
17 inappropriately based arguments on activity rather than dose,  
18 and, in general, totally misrepresented the magnitude of the  
19 activity of the plutonium being used in the TRUMP-S experiments  
20 being done at the MURR. In summary, there is no merit to the  
21 allegations made by Intervenors that Licensee has improperly  
22 applied for a license for plutonium or is in violation of the  
23 license it received.

24 Subscribed and sworn  
25 before me in  
26 BOONE County,  
27 Missouri this 22<sup>nd</sup> day of  
28 October 1990

  
\_\_\_\_\_  
J. Steven Morris  
Interim Director

29   
30 Sharon Weikel, Notary Public, State of Missouri  
My commission expires February 21, 1991  
Boone County, Missouri

31 My Commission Expires

32 2-21-91

33 17/ Even if the trace quantity of  $^{241}\text{Am}$  is included from Table 1,  
34 the addition of a fourth term [ $C_{\text{Am-241}}/2$ ] having a value of  
35  $(0.07/2 \approx 0.04)$  brings the sum of the ratios to a total of  
36 only 0.406. This addition is still not significant, and the  
37 sum of the ratios is still well less than one.



U. S. Department of Energy  
New Brunswick Laboratory

ATTACHMENT 1

# New Brunswick Laboratory Certified Reference Materials Certificate of Analysis

## CRM 127

### Plutonium Metal - Matrix Standard

This Certified Reference Material (CRM) is for use as a source material for the preparation of emission spectroscopy  $\text{PuO}_2$  matrix standards. Each unit of CRM 127 consists of several pieces of metal having a total weight of about 5 grains, sealed in a glass tube under a reduced-pressure argon atmosphere. **NOTE:** *The tube and its containment should be handled under proper radiologically-controlled conditions at all times.*

This CRM was originally issued in 1973 by the National Bureau of Standards (NBS) as Standard Reference Material (SRM) 945. The material was prepared and analyzed by the Los Alamos National Laboratory (LANL) of the University of California, in collaboration with NBS. In 1987, the technical and administrative transfer of NBS Special Nuclear SRMs into the NBL CRM Program was coordinated by the NBS Office of Standard Reference Materials and N. M. Trahey, NBL.

The metal is known to be of a high-purity ( $>99.8\%$ ), as it was prepared from a single lot of material selected because of its low metallic impurity content. The major fixed (nonradioactive) impurities in CRM 127 are: tungsten -  $80 \mu\text{g/g}$ , silicon -  $11 \mu\text{g/g}$ , gallium -  $6 \mu\text{g/g}$ , and iron -  $3 \mu\text{g/g}$ . Total detected fixed metallic impurities are estimated to be  $105 \mu\text{g/g}$ . The americium-241 content, resulting from the decay of plutonium-241, is approximately  $146 \mu\text{g/g}$  as of September 1970, and should increase less than  $150 \mu\text{g/g}$  per year. All values given are not certified and are provided for informational purposes only. Impurity determinations indicate that the material is quite homogeneous.

The atomic weight of the plutonium is 239.12.

#### RECOMMENDED PROCEDURE FOR USING CRM 127

Each unit (tube) of CRM 127 is provided with a serial number and unique weight. Weighings are accurate to the nearest 1.0 milligram. Very small pieces of metal may be separated from the larger pieces in the tube. The contents of the tube should be washed thoroughly with dilute HCl to insure quantitative removal of the metal pieces.

October 1, 1987  
Argonne, Illinois  
(Revision of NBS Certificate dated January 29, 1973)

Carleton D. Bingham  
Director

(Over)

The matrix material should be prepared to closely approximate the chemical composition of the sample under study, since best results are obtained when the standards have the same chemical matrix as the samples. Also, the matrix material should be prepared under controlled conditions that parallel the ignition of the sample prior to its analysis.

A recommended procedure<sup>1</sup> for determining metallic impurities in plutonium metal is as follows: The entire 5 grams of CRM 127 should be ignited as one batch. Impurity elements are then added in the form of oxides and blend into the matrix using a mixer-mill. Care should be taken to guard against impurity contamination. The material should then be ignited at 800°C for 30 minutes using a Pt-10% Rh crucible, precleaned with hot concentrated HNO<sub>3</sub>, and a Pt-lined electric furnace. Addition of 4% Ga<sub>2</sub>O<sub>3</sub> as a carrier is performed prior to excitation of the material in a dc arc. Plastic ware is recommended for the blending operation and for storage of the standards.

#### REFERENCE

1. Unpublished Method, LANL.

U. S. Department of Commerce  
 Frederick B. Dent  
 Secretary  
 National Bureau of Standards  
 Richard W. Roberts, Director

# National Bureau of Standards

## Certificate of Analysis

### Standard Reference Material 945

### Plutonium Metal – Standard Matrix Material

This plutonium metal is issued to provide a source material for the preparation of  $\text{PuO}_2$  to be used as matrix material for the preparation of emission spectroscopy standards. Although it is known to be of high purity, it is not intended to replace SRM 949, as a primary chemical standard. The Assay and Atomic Weight values are supplied primarily as information in the use of this standard as a matrix material.

Assay  $99.9 \pm 0.1$

Atomic Wt 239.12

This plutonium metal has been prepared from a single lot of material selected because of its low metallic impurity content. The main impurities detected, exclusive of americium, are: tungsten, 80 ppm; silicon, 11 ppm; gallium, 6 ppm; and iron, 3 ppm--with a total detected metallic impurity content of about 105 ppm. The americium resulting from the decay of plutonium-241 is approximately 146 ppm as of September 1970, and should increase less than 150 ppm per year. Impurity determinations indicate that the material is quite homogeneous.

Each of these SRM's consists of several pieces of metal having a total weight of about 5 grams sealed in a glass tube under a reduced-pressure argon atmosphere. The serial number of the SRM and its weight are given on each tube. Weighings are accurate to the nearest 1.0 milligram. Because of the method used to fabricate the material, very small pieces of metal may be separated from the larger pieces in the tube. These small pieces may be left behind unless the tube is carefully washed out, preferably with dilute hydrochloric acid.

This material was prepared and analyzed by the Los Alamos Scientific Laboratory of the University of California, Los Alamos, New Mexico, in collaboration with the National Bureau of Standards.

Washington, D. C. 20234  
 April 6, 1971  
 (Reissued January 29, 1973)

J. Paul Cali, Chief  
 Office of Standard Reference Materials

(over)

## DIRECTIONS FOR USE

For the emission spectrographic analysis of either Pu metal, U metal, or alloys of these two, best results are obtained when standards are prepared having the same matrix, chemically, as the samples. This is because the chemical matrix significantly influences the response of impurity elements to dc arc excitation. Further, the physical properties, particularly the state of subdivision and density, also significantly affect the response of impurity elements to dc excitation. For these reasons, it is recommended that the matrix material should not only closely approximate the chemical composition of the samples, but should also be prepared in a closely controlled manner that parallels the ignition of the sample prior to its analysis.

A recommended method<sup>(1)</sup> for determining metallic impurities in plutonium metal involves ignition of the metal under controlled conditions (800 °C for 30 minutes), addition of 4% Ga<sub>2</sub>O<sub>3</sub> as a carrier, and excitation of the material in a dc arc.

To prepare matrix material for standards for this analysis, the PuO<sub>2</sub> should be produced by ignition of high-purity metal under similar conditions of temperature and time. It is recommended that 5 grams of metal be ignited as one batch. Impurity elements are then added in the form of oxides and carefully blended into the matrix using a mixer-mill. The usual precautions must be taken to guard against impurity contamination in each preparation step. Use of a Pt-10% Rh crucible precleaned with hot concentrated HNO<sub>3</sub>, and a Pt-lined electric furnace is recommended for the ignition step. Plastic ware is recommended for the blending operation and for storage of the standards.

(1) Unpublished Method, Los Alamos Scientific Laboratory.



**ATTACHMENT 3**

NBL SHIPPING FORM  
NBL RECORD COPY  
(RECORD CLERK/CENTRAL FILE)

No. 3233

DATE: \_\_\_\_\_  
CARRIER NO.: \_\_\_\_\_  
VEHICLE NO.: \_\_\_\_\_

BRUNSWICK LABORATORY  
Building 350  
8800 South Cass Avenue  
Argonne, Illinois 60439  
(312) 972-2442



SHIP TO  
Rockwell International  
Rocketdyne Div., Bldg. 064  
Santa Susana Test Site  
Top of Woolsey Canyon  
Chatsworth, CA 91311  
ATTENTION: V. J. Schaubert

DATE REQUIRED AT DESTINATION: 5/11/89  
NBL REF. NO.: 89-043

SHIPMENT AUTHORIZED BY: *Handwritten signature*  
DATE: 5/11/89  
M. J. Marks/R. A. MARSH, JR. 5/11/89  
CONSIGNEE SIGN & RETURN COPY BY: \_\_\_\_\_ DATE: \_\_\_\_\_

RADIOACTIVITY PRESENT?  YES  NO  
TYPE:  ALPHA  BETA  GAMMA  NEUTRON  
CONTAINER RETURN REQUIRED?  YES  NO  
CONSIGNEE REF. NO. R94PNFB9240804

COMPLETE ROUTING INSTRUCTIONS:  
RVNX \$.40 per lb

ITEM #	BY PAGES	PKM	DESCRIPTION	TOTAL QUANTITY	WEIGHT (Units or Cans)	RATE	CHARGES
1	1	X	Radioactive Material, N.O.S., UN2982: Plutonium and Uranium (Depleted) as Solid, Metal; 390.0290 mCi Fissile Exempt U.S.A. DOT Type B Package Two Radioactive White I Labels Applied  (one unit each NBL CRM#s 127 & 115)	80.0 g	56 lb		

FISSILE CLASS:  EXEMPT  I  II  III  NA  
TRANSPORT GROUP: N/A  I  II  III  IV  V  VI  VII  VIII  SPECIAL FORM  
RADIOACTIVE LABEL:  NOT REQUIRED  WHITE I  YELLOW II  YELLOW III  
ACTIVITY: 390.0290 mCi  
OTHER LABELS: N/A  OXIDIZER  CORROSIVE  OTHER  
 POISON  FLAMMABLE  
RADIATION SURVEYS:  SURFACE  SURFACE BY: 200 dpm/100 cm<sup>2</sup> AT 1 METER (1y) 0.02 mSv/h  
TRANSPORT INDEX: 0.02  
TRANSFERABLE ACTIVITY: 200 dpm/100 cm<sup>2</sup>  
E 6060 BY E 80 DATE: 5/11/89

D.O.T. CERTIFICATION: This is to certify that the above listed items are properly classified, described, packaged, marked and labeled, and are in proper condition for transportation according to applicable regulations of the U.S. Department of Transportation.  
D.O.T. CERTIFICATION BY: Frank P. Orlowicz DATE: 5/11/89

*Handwritten signature* MAY 12 1989

CARRIER NAME: \_\_\_\_\_ CARRIER SIGNATURE: \_\_\_\_\_ DATE: \_\_\_\_\_

Transportation hereunder is for the U.S. Government and actual transportation costs paid to carrier(s) by the shipper or receiver is to be reimbursed by the government.

MODE OF SHIPMENT:  PREPAID  U.S.  EXCLUSIVE USE  
 COLLECT  PARCEL POST  REF. MAT'L  SALE  
 AIR FREIGHT  PROJECT. SERV.  LOAN/RETURN  REPAIR  
 MOTOR FREIGHT  OTHER: \_\_\_\_\_  OTHER: \_\_\_\_\_  WASTE DISPOSAL/RECOVERY  
FORM OF MATERIAL:  SOLID  ALLOY  NITRATE  
 LIQUID  OXIDE  ENCAPSULATED  
 GAS  CARBIDE  OTHER: \_\_\_\_\_  
 METAL  SULFATE  
NUCLIDES:  U-235 0.151g  Pu-239 4.72g  OTHER Pu-240 0.279g  
 U-238 74.832g  Pu-240 0.279g  
TOTAL ELEMENTS:  U 74.983g  OTHER  
 Pu 4.999g  
FINAL INSPECTION - INNER CONTAINER:  PACKING/SHIELDING SECURE  CLOSURE DEVICES SECURE  
 GASKET SEATED SECURELY  VALVES/VENTS SECURE  
 NO SIGNIFICANT RUST, DENTS, ETC.  
FINAL INSPECTION - OUTER CONTAINER:  ADDRESS LABELS  COVER LETTER/INSTRUCTIONS  
 GASKETS SEATED SECURELY  CLOSURE DEVICES SECURE  
 PACKING/SHIELDING SECURE  VALVES/VENTS SECURE  
 OTHER REQ'D LABELS  CONTAINER CONDITION

APPROVED BY 900, 900, 900, 900  
 900, 900, 900, 900  
 INFORMATION CONTAINED HEREIN IS UNCLASSIFIED EXCEPT WHERE SHOWN OTHERWISE  
 DATE 10/15/01 BY 60322 UCBAW

NUCLEAR MATERIAL TRANSACTION REPORT

1. REPORT NUMBER: LAF  
 2. MATERIAL NUMBER: ZNK  
 3. TRANSACTION NUMBER: 002  
 4. QUANTITY: A  
 5. UNIT: A  
 6. DATE: A  
 7. MONTH: B  
 8. YEAR: -

1. NAME OF SELLER: SNM-21  
 2. ADDRESS OF SELLER: Maxwell International  
 Rockledge Division  
 6633 Canoga Ave., Canoga Park, CA  
 P. D. Rutherford 91303  
 P. O. Box 100000  
 P. O. Box 100000  
 P. O. Box 100000

1. NAME OF BUYER: SNM-247  
 2. ADDRESS OF BUYER: University of Missouri, Columbia  
 Research Reactor Facility  
 Research Park, Columbia, MO  
 Roland Hiltsch 65211  
 (314) 902-4211  
 P. O. Box 100000

1. NAME OF BUYER: LM  
 2. ADDRESS OF BUYER: U.S. Department of Energy  
 San Francisco Operations Office  
 1333 Broadway  
 (415) 946-1212

- ZNK-LAF
- ZNK
- ZNK-LM
- ZNK-OFA
- LAF
- LM
- OFA
- IRG

U. of Mo Letter 2/20/90 & Mo License Amend #12  
 PU metal for TRAMP-S program

DATE	QUANTITY	UNIT	WEIGHT	ANALYSIS
9 5 90				
9 7 90				

135 lbs

Ref. original receipts OJ-LAF-01.  
 Deliver to Dr. Roland Hiltsch

DATE	QUANTITY	UNIT	WEIGHT	ANALYSIS
01	0.127	T	50.771	5.0
				5.0
				5.50
				5

SHIPPER'S DATA: P. Rutherford  
 RECEIVING DATA: 7/1/90  
 APPROVED BY: [Signature]  
 DATE: 7/20/90  
 PIECE-COUNT VERIFIED  
 RECEIVER ACCEPTS SHIPPER'S  
 WEIGHTS AND ANALYSIS

WARNING: FALSE STATEMENTS IN THIS CERTIFICATE MAY BE SUBJECT TO CIVIL AND/OR CRIMINAL PENALTIES. NRC REGULATIONS REQUIRE THAT SUBMITTERS TO THE NRC BE COMPLETE AND ACCURATE IN ALL MATERIAL RESPECTS. 10 USC, SECTION 1001 MAKES IT A CRIMINAL OFFENSE TO MAKE A WILLFULLY FALSE STATEMENT IN REPRESENTATION TO ANY DEPARTMENT OR AGENCY OF THE UNITED STATES AS TO ANY MATTER WITHIN ITS JURISDICTION.

ATTACHMENT 4



**THIS MEMORANDUM**  
TST

is an acknowledgment that a Bill of Lading has been issued and is not the Original Bill of Lading as a copy or duplicate, covering the property named herein, and is intended solely for filing or record

(Name of Carrier)

RECEIVED, subject to the classification and tariffs in effect on the date of the issue of this Bill of Lading from ROCKWELL INTERNATIONAL CORPORATION the property described herein in express, parcel, or other as noted, contents and condition of packages unknown, marked, consigned and delivered as indicated herein, which are carried by the said carrier herein under and through this contract of carriage and pursuant to possession of the property under the control of carrier to its own place of delivery at said destination, it or its route, others to be delivered to another carrier on the route to said destination, it is mutually agreed as to each carrier of all or any of the property over all or any portion of said route to destination, and as to each party or any party interested in all or any of said property, that every article to be performed hereunder shall be subject to all the terms and conditions of the Uniform International Bill of Lading or Form 11 or Official Southern, Western and Illinois Freight Classification, in effect on the date hereof, if this is a rail or a rail-water shipment, or (2) in the applicable motor carrier classification or tariff if this is a motor carrier shipment. Shipper hereby certifies that he is familiar with all the terms and conditions of the said Bill of Lading, including those on the back thereof, in form as the classification or tariff which governs the transportation of this shipment, and the said terms and conditions are hereby agreed to by the shipper and accepted for himself and his assign.

A1 \_\_\_\_\_ 19 \_\_\_\_\_ Shipper's No. **555** **836**

CONSIGNEE TO \_\_\_\_\_ (Mail or street address of consignee - For purposes of notification only) Subject to Section 7 of conditions, if this shipment is to be delivered to the consignee with out receipt on the consignee, the consignee shall sign the following statement:

DESTINATION \_\_\_\_\_ STATE OF \_\_\_\_\_ COUNTY OF \_\_\_\_\_ The carrier shall not make delivery of this shipment or take payment of freight and all other lawful charges.

Delivery Address \_\_\_\_\_ (To be filled only when shipper desires and governing tariffs provide for delivery thereof)

ROUTE \_\_\_\_\_ ROCKWELL INTERNATIONAL CORPORATION

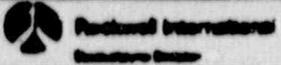
DELIVERING CARRIER \_\_\_\_\_ CAR INITIALS \_\_\_\_\_ CAR NUMBER \_\_\_\_\_ (Signature of Consignor)

No. Pkgs	Description of Articles, Special Marks and Exceptions	*Weight (Subject to Carr. Tariff)	Class. or Rate	Ch. Col.	
	Dual Driver Requested				If charges are to be prepaid or collect, write or stamp form "To Be Prepaid" or "Collect".  Received \$ _____ to apply to prepayment of the charges on the property described herein.  Agent or Cashier: per: _____ (The signature here denotes only the amount prepaid)  Charges Advanced \$ _____  This is to certify that the above-named materials or property classified, described, packaged, marked and labeled and are in proper condition for transportation according to applicable regulations of the Department of Transportation.  *If this shipment moves in two or more parts by a carrier or carriers, the law requires that the Bill of Lading shall state whether it is "carrier's" or "shipper's" weight.  NOTE—Where the law is a general one unless otherwise required to state specifically, giving the agreed or declared value of the property. The agreed or declared value of the property is hereby specifically stated in the shipping bill to be sent.  Consignee: Per: _____
	No Placards required				
	In the event of an emergency, contact				
	Rockwell International 818-700-5124.				
	If arrival is anticipated after noon Friday, contact				
	Charlie McKibben at 314-882-5204 for delivery instructions.				

ROCKWELL INTERNATIONAL CORPORATION

AGENT *[Signature]* PE  
T.S. No. 2582

Rockwell International Corporation  
10000 Canyon Blvd., Canyon Park, CA 95024  
PUSH 2-5



1. PRIME CONTRACT PURCHASE PRICE

**COMMERCIAL SHIPPING DOCUMENT**

RD4PNZ90561327

NO. 1 1  
 DATE  
 ACCEPTANCE POINT

SHIPMENT NO. LAE-ZNK-002    DATE SHIPPED 09-05-90    A. BL. 555    ORDER NO.  
 VEH. TSMT    PPD

CONTRACTOR: Rockwell International, Rocketdyne Division, 6633 Canoga Avenue, Canoga Park, CA 91303

SHIPPED FROM: Rockwell International, Rocketdyne Division, Santa Susana Field Laboratory, Top of Woolsey Canyon Road, near Chatsworth, CA 91311

DATE: \_\_\_\_\_ COMPANY REPRESENTATIVE: \_\_\_\_\_  
 THE REQUIRED POA OF LISTED ITEM HAS BEEN PERFORMED

TYPED NAME AND OFFICE: \_\_\_\_\_ DATE: \_\_\_\_\_ SIGNATURE OF AUTHORIZED GOVT. REP.: \_\_\_\_\_

SHIPPED TO: University of Missouri, Research Reactor Facility, Research Park, Columbia, MO 65211

MARKED FOR: Dr. Roland Hultsch

13. ITEM NO.	14. PART NUMBER	DESCRIPTION	17. QUANTITY SHIPPED	18. UNIT	19. UNIT PRICE	20. AMOUNT
1		Metal, Plutonium Activity: 0.39 curies 95% Pu-239 5% Pu-240  TRACE Pu-241 & Am-241	5	gm		

70830 50090

CONTROL NUMBER

OTHER INFORMATION: P. Horton, D/635 T020

GROSS WT. 135    CUBE

LENGTH 18    BIRTH 18    HEIGHT 26

APPROVAL SIGNATURE: *J. A. Kucinskas*

PRINTED/TYPED NAME: J. A. Kucinskas    DEPT. 014    MAIL CODE EA35    PHONE EX. 5159(c)

PACKAGING INFORMATION: DOT 6M - Type B Container    PREP. DATE 9/4/90    PACKING LIST NO. Z90-3336

DATE \_\_\_\_\_ CUSTOMER REPRESENTATIVE \_\_\_\_\_

## AMENDMENT TO BNN-247

February 20, 1990

## RADIOACTIVE MATERIAL (RE: RG 10.3, SUBSECTION 4.3)

1. a. Uranium (Depleted)  
b. any  
c. 500 grams  
d. 0.2 mCi
2. a. Plutonium (94.42 wt% Pu-239, 5.58 wt% Pu-240)  
b. any  
c. 10 grams  
d. 710 mCi

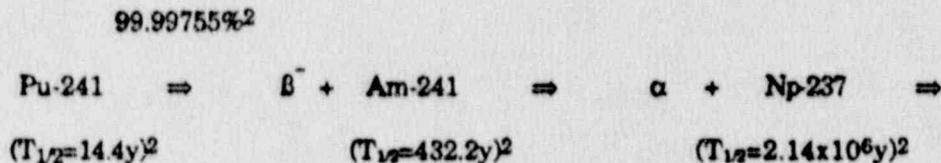
PURPOSES FOR WHICH LICENSED MATERIAL WILL BE USED  
(RE: RG 10.3, SUBSECTION 4.2)

A research project has been funded at the University of Missouri, the objective of which is to make basic scientific measurements using small amounts (one gram or less) of depleted uranium or a transuranic (TRU) to obtain thermochemical properties data. The data are needed for design of a process to separate actinide and rare-earth metals in nearly pure form from PUREX wastes, which requires computing the separation efficiencies and the purity of the actinide and rare-earth metals that are recovered. Utilizing such a process, uranium and transuranic materials can be recovered from spent reactor fuel material without generation of liquid wastes. The uranium and TRU can then be recycled for use as fuel or transformed to shorter-lived isotopes by use of "actinide burner" reactors or disposed of in much smaller monoliths. As a result, the mass of extremely long-lived, high-level radioactive waste from fuel reprocessing that requires disposal is greatly reduced, thus reducing the time needed for isolation of this waste material by more than a factor of 10,000--a major reduction in the potential environmental impact. Therefore, use of the actinide material for this experiment constitutes "research and development" as defined in 10 CFR 30.4.

The proposed process is based on the electrolytic separation of lanthanide rare earth from transuranic metals. Actinide and rare-earth fission products are dissolved in molten cadmium. The cadmium acts as an anode at the bottom of a crucible in a furnace. Above the cadmium is a liquid salt electrolyte composed of the chloride salts into which the cathode is suspended. When an electric potential is applied, the heavy metal ions flow from the cadmium anode through the salt and are collected in nearly pure form at the metal cathode.

Estimation of Pu-241 and Am-241Contaminants in CRM 127 as of September 1990<sup>1</sup>

The amount of Pu-241 and Am-241 contaminants in CRM 127 can be calculated for any point in time knowing the decay scheme, and the Am-241 concentration and production rate at a given point in time. Pu-241 decays to Am-241 as shown:



As stated in CRM 127 Certificate:

"The americium-241 content, resulting from the decay of plutonium-241, is approximately 146  $\mu\text{g/g}$  as of September 1970, and should increase less than 150  $\mu\text{g/g}$  per year."

Pu-241 Calculation

The production rate of Am-241 is equivalent to 99.99755% of the decay rate for Pu-241. Therefore, if Am-241 production rate equaled 150  $\mu\text{g/year}$  for one gram of CRM 127 material in September 1970, then the decay of Pu-241 would have been:

$$\begin{aligned}
 (\lambda_{\text{Pu-241}} \times N_{\text{Pu-241}}) &= \frac{(150 \mu\text{g/y})(6.023 \times 10^{23} \text{atoms})(10^{-6} \text{g}/\mu\text{g})}{(241\text{g})} \\
 &= 3.7 \times 10^{17} \text{atoms/y per g of CRM 127 material}
 \end{aligned}$$

Since the Pu-241 decay constant is:

$$\lambda_{\text{Pu-241}} = (\ln 2)/(14.4\text{y}) = 0.0481\text{y}^{-1}$$

---

1 Receipt of plutonium material at MURR.

2 KOCHER, D.C., Radioactive Decay Data Tables: A Handbook of Decay Data for Application to Radiation Dosimetry and Radiological Assessments, DOE/TIC-11026 (1981), p.211.

then the number of Pu-241 atoms would have been:

$$\begin{aligned} N_{\text{Pu-241}} &= (3.7 \times 10^{17} \text{atoms/y}) / (0.0481 \text{y}^{-1}) \\ &= 7.7 \times 10^{16} \text{atoms per g of CRM 127 material} \end{aligned}$$

This would correspond to a Pu-241 mass of:

$$\begin{aligned} \text{Mass}_{\text{Pu-241}} &= \frac{(7.7 \times 10^{16} \text{atoms})(241 \text{g})}{(6.023 \times 10^{23} \text{atoms})} \\ &= 0.0031 \text{g per g of CRM 127 material} \end{aligned}$$

Therefore, the mass of Pu-241 in September 1990 (20 year decay time) would be:

$$\begin{aligned} \text{Mass}_{\text{Pu-241}} &= (0.0031 \text{g}) e^{-(0.0481 \text{y}^{-1})(20 \text{y})} \\ &= 0.0012 \text{g per g of CRM 127 material} \end{aligned}$$

and the corresponding Pu-241 activity in September 1990 would be:

$$\begin{aligned} \text{Act}_{\text{Pu-241}} &= \frac{(0.0012 \text{g})(6.023 \times 10^{23} \text{atoms})(0.0481 \text{y}^{-1})(1 \text{Ci})}{(241 \text{g})(3.16 \times 10^7 \text{sec/y})(3.7 \times 10^{10} \text{dps})} \\ &= 0.12 \text{Ci per g of CRM 127 material} \end{aligned}$$

Since the actual production rate of Am-241 in September 1970 was less than 150  $\mu\text{g/g}$  per year, then the values estimated to one significant figure for Pu-241 in September 1990 are:

$$\text{Mass}_{\text{Pu-241}}(\text{Sep 90}) < 0.001 \text{g Pu-241/g of CRM 127 material}$$

and

$$\text{Act}_{\text{Pu-241}}(\text{Sep 90}) < 0.1 \text{Ci Pu-241/g of CRM 127 material}$$

# Los Alamos

Los Alamos National Laboratory  
Los Alamos, New Mexico 87545

20 January 1982

Dr. Thomas E. Gillis  
Standards Coordinator  
Office of Standard Reference Materials  
U. S. Department of Commerce  
National Bureau of Standards  
Washington, D.C. 20234

Dear Tom:

In accordance with your recent telephone request, fourteen units of SRM 945 were sent to New Brunswick Laboratory, Attention: Nancy Trahey. This is one less than the quantity stated in your letter of January 12, but the fourteen had been selected, packaged, and shipped prior to receipt of your letter.

This reference material was reanalyzed during February, 1975, for americium-241, uranium, and plutonium isotopic composition. Based upon the original analyses and the reanalyses in 1975, the calculated results for uranium and americium are as follows:

## Uranium Total, $\mu\text{g/g}$

<u>Year</u>	Jan.	Feb.	Mar.	Apr.	May	Jun.	Jul.	Aug.	Sep.	Oct.	Nov.	Dec.
1981	360.4	363.2	365.9	368.6	371.3	374.1	376.8	379.5	382.3	385.0	387.7	390.4
1982	393.2	395.9	398.6	401.4								

## Americium, $\mu\text{g/g}$

1981	1343	1350	1358	1365	1374	1380	1387	1394	1402	1409	1416	1423
1982	1431	1438	1445	1452								

Plutonium isotopic distribution, atom percent, was determined April 1970 (probably by ORNL) and February 1975 by our laboratory. The results, including the ORNL data decay-corrected to February 1975 are given below. Half life values, in years, used for the computation are 87.74(238), 24 119(239), 6 540(240), 14.4(241), and 387 000(242).

Dr. Thomas E. Gills

- 2 -

20 January 1982

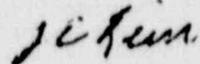
<u>Isotope</u>	<u>April 1970</u>	<u>April 1970 decay-corrected</u>	<u>Feb. 1975</u>
238	0.0096	0.0092	0.0092
239	94.130	94.194	94.198
240	5.528	5.530	5.530
241	0.314	0.249	0.245
242	0.018	0.018	0.018

Plutonium assay and metal impurities were not redetermined. The values as of April, 1970, were:

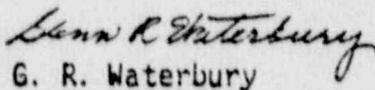
Pu Assay, %: 99.97

Metal Impurities,  $\mu\text{g/g}$ : W, 59; Fe, 3; Ga, 2; others less than detection limits ( $< \Sigma = 93 \mu\text{g/g}$ ).

Very truly yours,



James E. Rein  
CMB-1 Acting Group Leader  
Analytical & Instrumental Chemistry  
MS-740



G. R. Waterbury

JER:GRW:vmw

cc: C. D. Bingham, NBL  
Nancy Trahey, NBL  
CRM-4 (2), MS-150  
File (2)

# ALL THINGS NUCLEAR

BY

**JAMES C. WARF**



**The Southern California  
Federation of Scientists**

Pu-240 content is high. Thus the plutonium-239, which is the major component of the bomb core, is awash in neutrons, a circumstance which tends to initiate premature chain reaction ("pre-ignition"). This means that the higher the plutonium-240 content in a bomb core, the more below critical size it must be and the more surface its component parts must have in order to permit the escape of the excess neutrons to prevent a chain reaction. It also means that the implosion must be all the more powerful to cause the intended uncontrolled explosive chain reaction on detonation. The presence of plutonium-240 complicates fabrication and degrades reliability. Also, some plutonium-241 is unavoidably present, and this too requires heavier shielding.

Weapons grade plutonium generally contains 6% Pu-240 or less. Since plutonium-240 does not undergo fission by slow neutrons (as do U-235 and Pu-239), it tends to accumulate in the fuel of nuclear reactors. Typically, weapons-grade plutonium has the following isotopic analysis:

Pu-238	0.06%
Pu-239	92.0-92.4
Pu-240	5.8
Pu-241	1.7-2.0
Pu-242	0.1

The question arises whether the plutonium from commercial, power-generating reactors can be employed in bombs. The answer is definitely yes. Such plutonium generally contains 10 to 20% plutonium-240, and while troublesome, it can be made into a weapon. Test bombs using commercial plutonium have been exploded successfully in the Nevada desert as early as 1962. Such bombs require especially powerful implosion. Although plutonium-240 can be fissioned by fast neutrons, such survives in explosions, making them extraordinarily dirty. The word "dirty" here means that the fission product debris contains such plutonium-240, a particularly dangerous substance. All isotopes of plutonium undergo fission under suitable conditions.

---

Bombs have been  
made from com-  
mercial plutonium

---

Warheads containing several percent plutonium-240 emit both neutrons and gamma rays, posing a danger to personnel, especially in confined quarters such as submarines. This is one of the motivations for preparing plutonium which contains little or no plutonium-240. One way to lower the content of the unwanted isotope is to blend it with plutonium containing about 3% of the 240 variety (called supergrade, made in the Savannah River reactors) and this is actually done.

A newer, more sophisticated technique is the physical separation of Pu-240 (and Pu-241) from Pu-239. A factory for this purpose has been proposed (Idaho Nuclear Engineering Laboratory). This is by a process called atomic vapor laser isotope separation (AVLIS). The liquid plutonium metal is vaporized by electron

U.S. Department  
of Commerce

National Bureau  
of Standards



NBS Special Publication 260

# NBS Standard Reference Materials Catalog 1981-83 Edition

Four 1606 cast irons (right), and nearly  
1,000 SRM's of today



1906-1981  
75 YEARS OF SERVICE



## NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards<sup>1</sup> was established by an act of Congress on March 3, 1901. The Bureau's overall goal is to strengthen and advance the Nation's science and technology and facilitate their effective application for public benefit. To this end, the Bureau conducts research and provides: (1) a basis for the Nation's physical measurement system, (2) scientific and technological services for industry and government, (3) a technical basis for equity in trade, and (4) technical services to promote public safety. The Bureau's technical work is performed by the National Measurement Laboratory, the National Engineering Laboratory, and the Institute for Computer Sciences and Technology.

**THE NATIONAL MEASUREMENT LABORATORY** provides the national system of physical and chemical and materials measurement; coordinates the system with measurement systems of other nations and furnishes essential services leading to accurate and uniform physical and chemical measurement throughout the Nation's scientific community, industry, and commerce; conducts materials research leading to improved methods of measurement, standards, and data on the properties of materials needed by industry, commerce, educational institutions, and Government; provides advisory and research services to other Government agencies; develops, produces, and distributes Standard Reference Materials; and provides calibration services. The Laboratory consists of the following centers:

Absolute Physical Quantities<sup>2</sup> — Radiation Research — Thermodynamics and Molecular Science — Analytical Chemistry — Materials Science.

**THE NATIONAL ENGINEERING LABORATORY** provides technology and technical services to the public and private sectors to address national needs and to solve national problems; conducts research in engineering and applied science in support of these efforts; builds and maintains competence in the necessary disciplines required to carry out this research and technical service; develops engineering data and measurement capabilities; provides engineering measurement traceability services; develops test methods and proposes engineering standards and code changes; develops and proposes new engineering practices; and develops and improves mechanisms to transfer results of its research to the ultimate user. The Laboratory consists of the following centers:

Applied Mathematics — Electronics and Electrical Engineering<sup>2</sup> — Mechanical Engineering and Process Technology<sup>2</sup> — Building Technology — Fire Research — Consumer Product Technology — Field Methods.

**THE INSTITUTE FOR COMPUTER SCIENCES AND TECHNOLOGY** conducts research and provides scientific and technical services to aid Federal agencies in the selection, acquisition, application, and use of computer technology to improve effectiveness and economy in Government operations in accordance with Public Law 89-305 (40 U.S.C. 759), relevant Executive Orders, and other directives; carries out this mission by managing the Federal Information Processing Standards Program, developing Federal ADP standards guidelines, and managing Federal participation in ADP voluntary standardization activities; provides scientific and technological advisory services and assistance to Federal agencies; and provides the technical foundation for computer-related policies of the Federal Government. The Institute consists of the following centers:

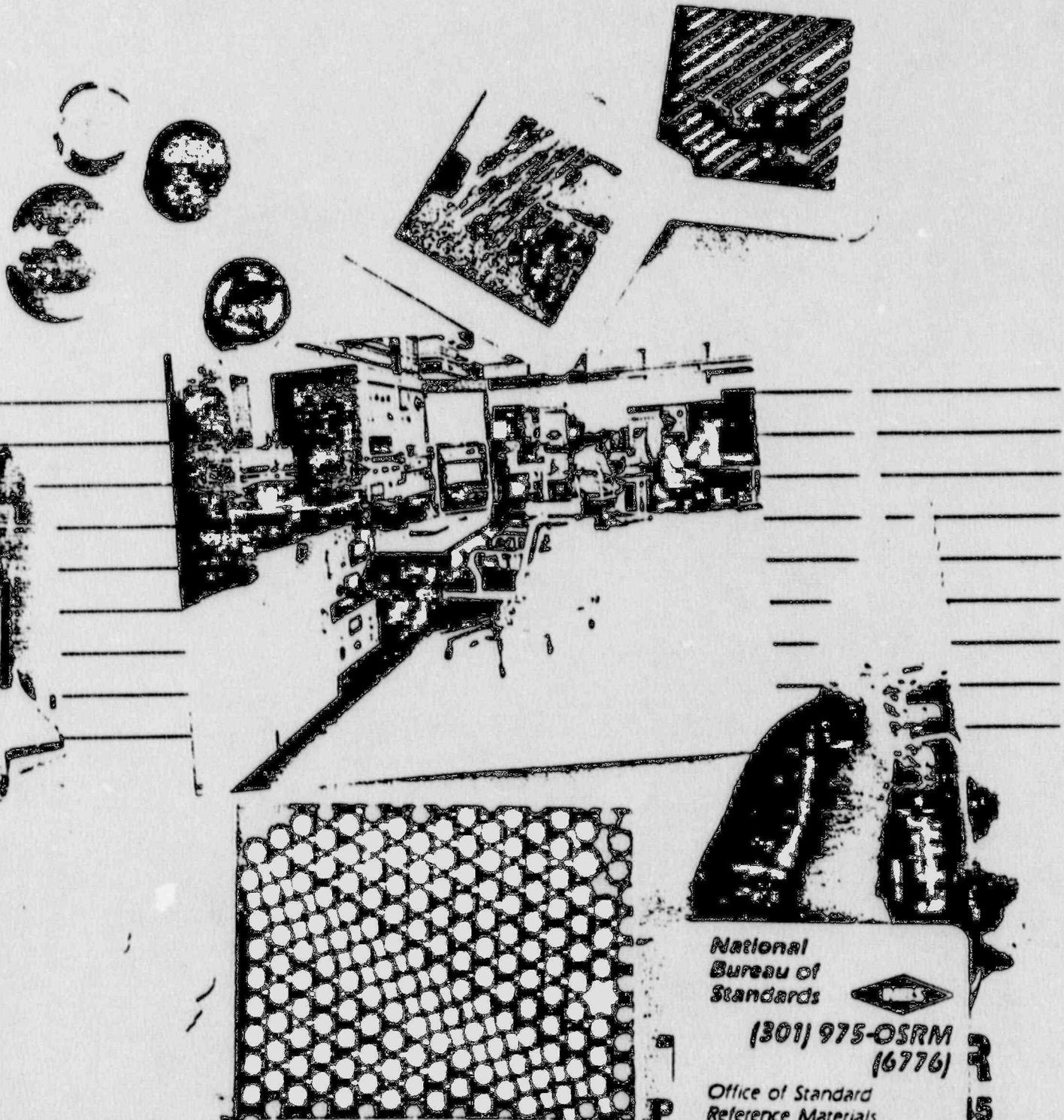
Programming Science and Technology — Computer Systems Engineering.

<sup>1</sup>Headquarters and Laboratories at Gaithersburg, MD, unless otherwise noted; mailing address Washington, DC 20234.

<sup>2</sup>Some divisions within the center are located at Boulder, CO 80303.

# NBS Standard Reference Materials Catalog 1986-87

U.S. Department of Commerce  
National Bureau of Standards



National  
Bureau of  
Standards   
(301) 975-OSRM  
(6776)

Office of Standard  
Reference Materials  
Bldg 222, Room B-311  
Gaithersburg, MD 20899

R  
15



UNITED STATES DEPARTMENT OF COMMERCE  
National Bureau of Standards  
Gaithersburg, Maryland 20899

Dear Colleague:

I hope that you find this new Standard Reference Materials Catalog to be the best edition we have issued. We have tried to make it user oriented to help you find the materials you need both quickly and easily. We think, and hope you will agree, that the revised alphabetical index is a major improvement over earlier editions.

Because finding the right SRM out of almost 1,000 can be difficult, I would like to suggest the following approach:

1. Start with the Contents,
2. Flip though the Catalog to see its organization,
3. Browse through the Alphabetical Index.

Most of the materials are classified by matrix (such as steel) or by use (such as clinical chemistry). However, with such diverse offerings, the categories are not mutually exclusive, and you may find some materials of interest to you in any part of the Catalog.

The 1984-1985 Catalog went to about 45,000 people. At least 60,000 copies of this one will be distributed. I think this indicates an increasing interest in quality measurements. We are happy to be part of this tradition and welcome you to the growing family of SRM users.

Sincerely,

Stanley D. Rasberry

P.S. Please do not hesitate to call us if you have any questions about the SRM's described, their availability, or if you cannot find what you need. We would be happy to have your suggestions for improved service and new SRM's (see Guide for Requesting the Development of New SRM's).

# Nuclear Materials

## Special Nuclear Materials

These SRM's are available to Department of Energy contractors, Nuclear Regulatory Commission, or State Licensees, and foreign governments that have entered into an agreement of cooperation with the U.S. Government regarding the use of these materials. Purchase orders and requests for information regarding ordering procedures, availability, and shipment of these SRM's should be directed to:

NBS Special Nuclear Standard Reference Materials  
 U.S. Department of Energy  
 New Brunswick Laboratory, D-350  
 9800 South Cass Avenue  
 Argonne, IL 60439  
 (312) 972-2453  
 FTS: 972-2453

Plutonium Assay				
SRM	Identification (Batch Name)	Constituent Certified	Element Weight* (g)	(Weight Percent)
945	Plutonium Metal, standard matrix	Impurities	5	99.9
949f	Plutonium Metal Assay	Plutonium Content	0.5	99.99

\*Nominal weight

Plutonium Isotopic								
SRM	Identification (Batch Name)	Element Weight (g)	Certified Isotopes (Atom Percent)					
			<sup>239</sup> Pu	<sup>240</sup> Pu	<sup>241</sup> Pu	<sup>242</sup> Pu		
946	Plutonium Sulfate Tetrahydrate	0.25	0.232	84.464	12.253	2.477	0.574	
947	Plutonium Sulfate Tetrahydrate	0.25	0.278	77.089	18.610	2.821	1.202	
948	Plutonium Sulfate Tetrahydrate	0.25	0.010	91.736	7.922	0.299	0.0330	
996	Plutonium—Spike	0.001	0.005	0.034	0.677	0.092	1.325	97.867

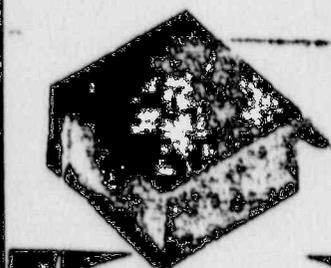
# NIST

## Standard Reference Materials Catalog 1990-91

ATTACHMENT 11  
NIST Special Publication 260  
U.S. DEPARTMENT OF COMMERCE  
National Institute of Standards and Technology

TO ORDER  
Phone: (301) 975-3089 (6776)  
Fax: (301) 948-3750

# GLASCOCK



# *Certified Reference Materials From Other Sources*

## **Special Nuclear Materials**

On October 1, 1987, the New Brunswick Laboratory began issuing special nuclear reference materials as NBL Certified Reference Materials (CRM's). These CRM's include the plutonium and uranium assay and isotopic materials previously issued by the National Institute of Standards and Technology. All orders or inquiries should be addressed to:

U.S. Department of Energy  
New Brunswick Laboratory  
Attn: Reference Materials Sales  
9800 S. Cass Avenue, Bldg. 350  
Argonne, IL 60439

708 (312) 972-2767

## **International CRM's**

Certified reference materials (CRM's) are available from many sources. The International Organization for Standardization (ISO), through its Council Committee on Reference Materials (REMCO), has prepared an international Directory of Certified Reference Materials. Inquiries may be directed to:

Dr. M. Parkany  
Secretary for REMCO  
International Organization for Standardization  
1, Rue de Varembe  
Case Postale 56  
1211 Geneva 20  
Switzerland

The International Union of Pure and Applied Chemistry (IUPAC), through its Commission on Physicochemical Measurements and Standards, issues a catalog of CRM's that are useful for the realization of physicochemical properties. It also has prepared a number of related documents. The current IUPAC edition is: "Physicochemical Measurements: Catalogue of Reference Materials from National Laboratories," Revised 1976, Pure & Appl. Chem., 48, 503-515 (1976).

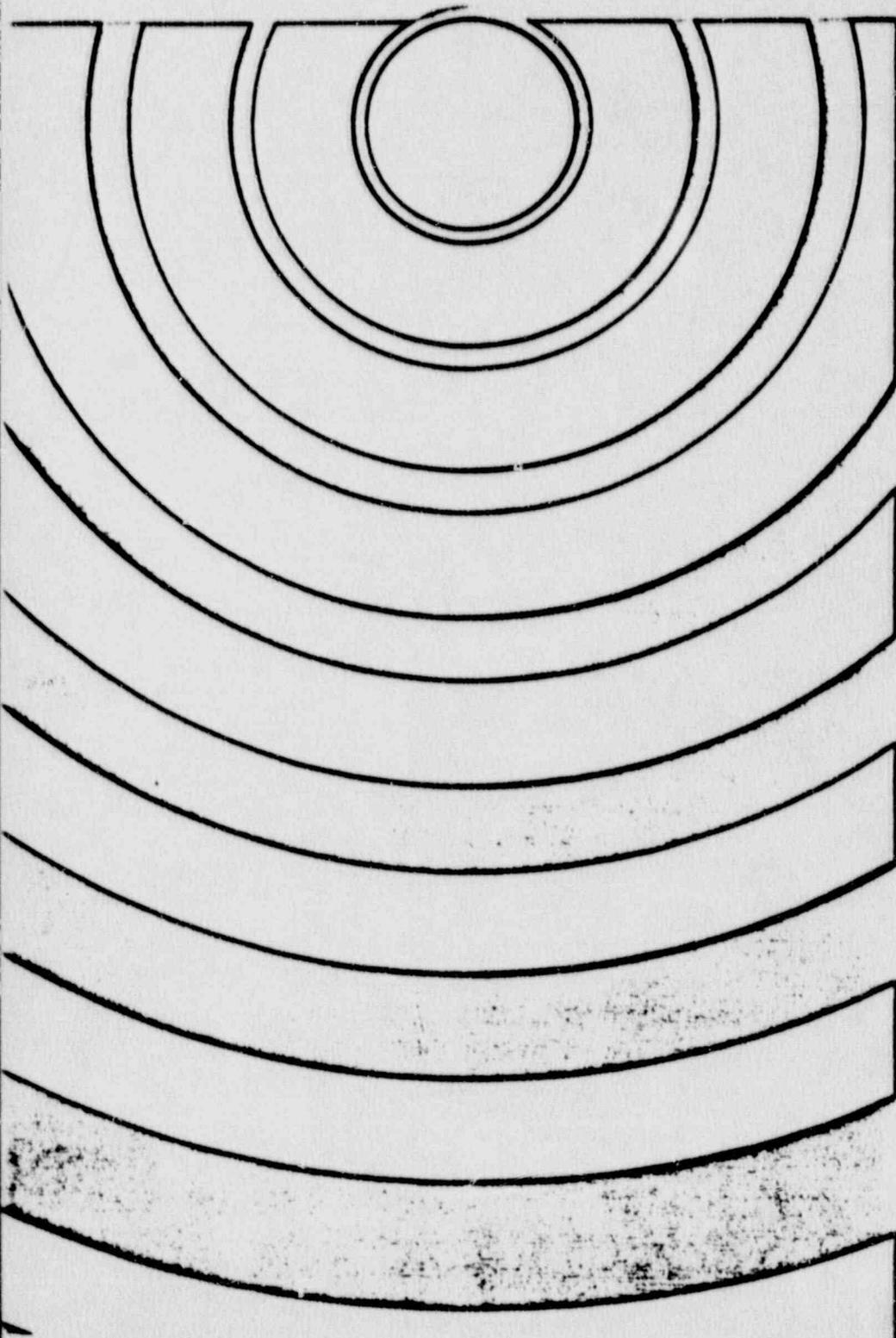
*Julie Frum makes the Division Office a pleasant and productive place by her dependability and warm spirit of hospitality.*



**NBL**  
**Certified**  
**Reference**  
**Materials**  
**Catalog**  
**1988**



United States Department of Energy



### **Certified Assay and Isotopic Composition CRMs**

These NBL CRMs are prepared for use primarily in: 1) analytical method development, calibration, and evaluation; 2) radiometric instrument and method calibration; 3) nuclear materials accountability verification; and 4) field source or working standards preparation. They are available to private companies, academic institutions, DOE contractors, and NRC or state licensees. They are also available to countries authorized to possess radioactive materials and/or to countries which have entered into an agreement for cooperation with the U. S. Government concerning the civil uses of nuclear energy.

### AMERICIUM ASSAY AND ISOTOPIC

CRM	Type	Mass/Unit (g)	Weight % Am	Atom %	
				<sup>241</sup> Am	<sup>243</sup> Am
•132	Americium-241 Spike, Nitrate	In Preparation		100	—
•133	Americium-243 Spike, Nitrate	In Preparation		—	100
•134	Americium Isotopic, Nitrate	In Preparation		<sup>241</sup> Am/ <sup>243</sup> Am = 1	

### PLUTONIUM ASSAY AND ISOTOPIC

CRM	Type	Mass/Unit (g)	Weight % Pu	Atom %					
				<sup>238</sup> Pu	<sup>239</sup> Pu	<sup>240</sup> Pu	<sup>241</sup> Pu	<sup>242</sup> Pu	<sup>244</sup> Pu
•122	Plutonium Oxide-PuO <sub>2</sub>	1	87.79	0.05	87.31	11.54	0.92	0.18	—
•126	Plutonium Metal, chunk	1	99.96	—	97.92	—	—	—	—
•128	Plutonium Isotopic, nitrate, solid	0.002	—	<sup>239</sup> Pu/ <sup>242</sup> Pu = 1					
•130	Plutonium-242 Spike, nitrate, solid	0.002	50	0.004	0.005	0.02	0.02	99.95	0.0004
•131	Plutonium-244 Spike, nitrate, solid	0.002	50	0.005	0.03	0.68	0.09	1.32	97.87
•136	Plutonium Isotopic, sulfate tetrahydrate, solid	0.5	—	0.23	84.46	12.25	2.48	0.57	—
•137	Plutonium Isotopic, sulfate tetrahydrate, solid	0.5	—	0.28	77.09	18.61	2.82	1.20	—
•138	Plutonium Isotopic, sulfate tetrahydrate, solid	0.5	—	0.01	91.74	7.92	0.30	0.03	—

### **Certified Impurity Element Composition CRMs**

These NBL CRMs are designed primarily for impurity element chemical analysis by spectrographic or spectrometric techniques. They can be used for: 1) analytical method calibration; 2) instrument calibration; and 3) nuclear materials accountability verification.

## PLUTONIUM

CRM	Type	Mass/Unit (g)	Elements Certified	Weight % Pu
*127	Plutonium Metal, standard matrix material, chunk	5	Fe, Ga, Si, W	99.9

## THORIUM

CRM	Type	Mass/Unit (g)	Elements Certified
66(1-7)	Thorium Oxide-ThO <sub>2</sub>	7 x 25	Ag, Al, B, Be, Bi, Ca, Cd, Cr, Cu, Fe, K, Mg, Mn, Mo, Na, Ni, P, Pb, Si, Sn, V, Zn

## URANIUM

CRM	Type	Mass/Unit (g)	Elements Certified
17-B*	Uranium (normal) Fluoride-Ur <sub>4</sub>	200	Fe, Mo, Ni, V
18*	Uranium (normal) Oxide-UO <sub>3</sub>	500	Cd, Cr, Cu, Fe, Mo, Ni
114*	Uranium (normal) Oxide-U <sub>3</sub> O <sub>8</sub>	50	Al, Cr, Fe, Mn, Ni, P, Si
123(1-7)	Uranium (normal) Oxide-U <sub>3</sub> O <sub>8</sub>	7 x 25	Al, B, Ca, Cd, Cr, Cu, Fe, Mg, Mn, Mo, Na, Ni, Pb, Si, Sn, V, Zn, Zr
124(1-7)	Uranium (normal) Oxide-U <sub>3</sub> O <sub>8</sub>	7 x 25	Ag, Al, B, Be, Bi, Ca, Cd, Co, Cr, Cu, Fe, Mg, Mn, Mo, Na, Ni, Pb, Si, Sn, Ti, V, W, Zn, Zr

\*Also certified for assay.

This index provides a numerical listing of current NBL CRM Certificates. For each CRM unit ordered a certificate is placed in the shipping container with the material. Additional Certificate copies are available upon request.

#### Index to Certificates

CRM	Certificate Date
1-A	March 1980
3-B	June 1969
4	September 1950
5	September 1950
17-B	July 1961
18	August 1957
42(1-4)	August 1957
66(1-7)	Undated
101-A	February 1981
102-A	February 1981
103-A	February 1981
104-A	February 1981
105-A	February 1981
106-A	February 1981
107-A	February 1981
108-A	February 1981
109-A	February 1981
110-A	February 1981
111	October 1, 1987 R
112-A	October 1, 1987 R
113	February 15, 1985 R
114	June 1978
115	June 1978
116	June 1978
117	January 1, 1986 R
118	June 1978
119	June 1978
122	May 1, 1985
123(1-7)	September 1, 1983
124(1-7)	September 1, 1983
125	October 1, 1982
126	January 1, 1986

127	October 1, 1987 R
128	October 1, 1985
129	October 1, 1987 R
130	July 1, 1987
131	October 1, 1987 R
132	In preparation
133	In preparation
134	In preparation
135	October 1, 1987 R
136	October 1, 1987 R
137	October 1, 1987 R
138	October 1, 1987 R
969	October 1, 1987 R
U0002	October 1, 1987 R
U005-A	October 1, 1987 R
U010	October 1, 1987 R
U015	October 1, 1987 R
U020-A	October 1, 1987 R
U030-A	October 1, 1987 R
U050	October 1, 1987 R
U100	October 1, 1987 R
U150	October 1, 1987 R
U200	October 1, 1987 R
U350	October 1, 1987 R
U500	October 1, 1987 R
U750	October 1, 1987 R
U800	October 1, 1987 R
U850	October 1, 1987 R
U900	October 1, 1987 R
U930	October 1, 1987 R
U970	October 1, 1987 R

R denotes Certificate revision.