Bulletin Reply 82-02



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SACRAMENTO MUNICIPAL UTILITY DISTRICT 🗆 6201 S Street, Boy 15830, Sacramento, California 95813; (916) 452-3211

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August 3, 1982

REGION VIVE

R H ENGELKEN, REGIONAL ADMINISTRATOR REGION V OFFICE OF INSPECTION AND ENFORCEMENT U S NUCLEAR REGULATORY COMMISSION 1450 MARIA LANE SUITE 210 WALNUT CREEK CA 94596

DOCKET NO. 50-312 LICENSE NO. DPR-54 RESPONSE TO IE BULLETIN NO. 82-02, SUPPLEMENTAL INFORMATION

The Sacramento Municipal Utility District hereby submits the following information to supplement our report dated July 28, 1982 on the above-mentioned bulletin.

Bulletin item number 3.b requested the chemical composition of factener lubricant used on bolted closures. As stated in our report, Molykote G has been and is being used on Reactor Coolant Pump B fasteners. Initially, the District was unable to obtain analysis of Molykote G because the manufacturer, Dow Corning Corporation, had indicated that they considered the chemical composition proprietary. The District has since obtained an analysis of Molykote G. The analysis was performed in 1971 and gives the chemical composition as follows:

Chloride. < 0.01 (%) Fluoride Sulfur as SO₃ 65.4 (%) Phosphorus < .01 (%) Sodium 3.75%(% ash @900°C) Ash Other: 29.6% Mineral oil CHCl₃ Extraction Moly. as MoO₃ 40.6% as Mo 26.8%

| Spec Semi-Quant Ana | lysis (%)* | Analysis for Metals (AA) |
|--|----------------|---|
| Silicon as SiO ₂ | 3.7 | Aluminum as Al |
| Aluminum as Al ₂ 0 ₃ | 1.8 | Antimony as Sb |
| Iron as Fe ₂ 0 ₃ | 0.7 | Arsenic as As |
| Titanium as TiO2 | 0.1 | Bismuth as Bi |
| Calcium as CaO | 0.1 | Boron as B |
| Magnesium as MgO | 0.9 | Cadmium as Cd |
| Sodium as Na ₂ 0 | 0.5 | Copper as Cu |
| Nickel as NiO | <0.2 | Lead as Pb |
| Chromium as Cr ₂ 0 ₃ | <0.2 | Mercury as Hg |
| Molybdenum as MoO3 1 | Major (>10.0%) | Silver as Ag |
| Vanadium as V ₂ 0 ₅ | <0.1 | Tin as Sn |
| Cobalt as CoO | <0.06 | Zinc as Zn |
| Manganese as MnO ₂ | <0.06 | |
| Copper as CuO | <0.1 | |
| Zinc as ZnO | <0.5 | |
| Lead as Pb0 | <0.06 | *These elements are reported as the oxide because of the analysis method used. This does not neces- sarily imply their presence as such. |
| Tin as SnO ₂ | <0.06 | |
| Zirconium as ZrO2 | <0.06 | |
| | | |

John J. Mattimoe

Assistant General Manager

and Chief Engineer

Subscribed and sworn to before me this

971√ day of August, 1982.

Notary Public Lay

