

ACRST-1994

ORIGINAL

OFFICIAL TRANSCRIPT OF PROCEEDINGS

**Agency:** Nuclear Regulatory Commission  
Advisory Committee on Reactor Safeguards

**Title:** Subcommittee Meeting on Thermal  
Hydraulic Phenomena

**Docket No.**

**LOCATION:** Bethesda, Maryland

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UNITED STATES NUCLEAR REGULATORY COMMISSION  
ADVISORY COMMITTEE ON REACTOR SAFEGUARDS

DATE: January 5, 1994

The contents of this transcript of the proceedings of the United States Nuclear Regulatory Commission's Advisory Committee on Reactor Safeguards, (date) January 5, 1994, as Reported herein, are a record of the discussions recorded at the meeting held on the above date.

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1 UNITED STATES OF AMERICA  
2 NUCLEAR REGULATORY COMMISSION

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4 ADVISORY COMMITTEE ON REACTOR SAFEGUARDS

5 \*\*\*

6  
7 SUBCOMMITTEE MEETING ON THERMAL HYDRAULIC PHENOMENA

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12  
13 Nuclear Regulatory Commission  
14 Conference Room P-110  
15 7920 Norfolk Avenue  
16 Bethesda, Maryland

17  
18 Wednesday, January 5, 1994

19  
20 The above-entitled proceedings commenced at 8:30  
21 a.m., pursuant to notice, Ivan Catton, ACRS Subcommittee  
22 Chairman, presiding.

23  
24  
25  
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## 1 ACRS MEMBERS PRESENT:

2

3

I. CATTON, ACRS Subcommittee Chairman

4

E. WILKINS, JR., ACRS Member

5

P. DAVIS, ACRS Member

6

T. KRESS, ACRS Member

7

R. SEALE, ACRS Member

8

V.J. DHIR, ACRS Consultant

9

V. SCHROCK, ACRS Consultant

10

W. WULFF, ACRS Consultant

11

N. ZUBER, ACRS Consultant

12

P. BOEHNERT, Cognizant ACRS Staff Member

13

B. SHERON, NRC/RES

14

N. LAUBEN, NRC/RES

15

T. KING, NRC/RES

16

L. SHOTKIN, NRC/RES

17

J. KELLY, NRC/RES

18

R. BEELMAN, INEL

19

D. FLETCHER, INEL

20

G. JOHNSEN, INEL

21

W. WEAVER, INEL

22

L. HOCHREITER, Westinghouse

23

24

25

## P R O C E E D I N G S

[8:30 a.m.]

1 MR. CATTON: The meeting will convene.

2 Gary, your topic is level tracking model?

3 MR. JOHNSEN: That's correct.

4 [Slide.]

5 MR. JOHNSEN: This morning I would like to talk  
6 about the level tracking model that we have just recently  
7 implemented in RELAP5.

8 [Slide.]

9 MR. JOHNSEN: I am going to cover these topics.  
10 First I am going to talk about level tracking, which is the  
11 model that we recently installed and contrasted with what we  
12 used to have in the code with is vertical stratification  
13 model, a vertical stratification model.

14 I am going to skip right ahead to some preliminary  
15 results to show you how the model behaves before I go into a  
16 detailed description of how it works, and then I will  
17 describe it and show you the assessment plans that we have.

18 [Slide.]

19 MR. JOHNSEN: The comments provided to us by the  
20 consultants, the ACRS consultants in part dealt with the  
21 model described in Volume 4 which is called a vertical  
22 stratification model. The discussion of this model  
23 characterized it as a level tracking model and it is not.

1 It is a model that was designed to sharpen the interface but  
2 not track the level per se.

3 We recognized that we needed to have a true level  
4 tracking model after we --

5 MR. WULFF: I would like to make a comment to this  
6 first statement that you made. I have here page number 359  
7 from the last meeting in Idaho where I asked you whether the  
8 essential physics of the level tracking in response to a  
9 bulletin that you have in your viewgraphs were in Volume 4.  
10 Mr. Johnsen: Yes. Catton: Is that the report we got some  
11 month ago? Right. Wulff: But it is not physics. Johnsen:  
12 That is a broad statement, Wolfgang. Wulff: And I say it  
13 deliberately, I will repeat it and sign it if you would like  
14 me to. Catton: You are going to communicate that. Wulff:  
15 Yes, I will. Mr. Shotkin: Now what we would like is  
16 specific details.

17 Nowhere was it said at that time, don't consider  
18 that to be a level tracking model, so we were misled.

19 MR. JOHNSEN: I think that was your terminology  
20 and not mine. I think if you look in Volume 4 you won't  
21 find that terminology. That was your characterization of  
22 the model.

23 MR. WULFF: You had on your viewgraph level  
24 tracking that prompted this question.

25 MR. JOHNSEN: I don't recall exactly the viewgraph

1 you are talking about.

2 MR. WULFF: But I do, and I do have the viewgraphs  
3 and I do have the record here. So I think it was always  
4 called a stratification model. That was your jargon and  
5 there is nothing wrong with that, but to say now that  
6 because there is a word different from another word that we  
7 reviewed the wrong is just rejectable. So I don't like to  
8 now be told that we reviewed the wrong thing.

9 MR. JOHNSEN: I didn't say you reviewed the wrong  
10 thing. What I am trying to say here is that the model that  
11 you reviewed in Volume 4 which, of course, was issued in  
12 1990, is not a level tracking model. It is a vertical  
13 stratification model, and that in March of 1993 when we had  
14 the last meeting in Idaho Falls, we had in fact planned to  
15 implement a true level tracking model.

16 MR. WULFF: Then you should have told us that it  
17 is not in Volume 4.

18 MR. CATTON: I don't think this is leading  
19 anywhere, but Wolfgang is correct. Why don't you continue.

20 [Slide.]

21 MR. JOHNSEN: Let me just show some results first  
22 before we get into the description, and this has just  
23 recently been completed, so we are just beginning our  
24 testing of the model and we use this little test problem to  
25 verify the functionality of the model as it was coded or was



1 intended to be coded.

2 This little test model consists of a 15-foot pipe,  
3 if you will, area of three square feet consisting of 12  
4 control volumes, and initially it is filled part way into  
5 the second control volume, and the pressure you can see is  
6 1,000 psi and saturated liquid. Then what we do is, we  
7 introduce saturated vapor at the bottom using a time  
8 dependent volume, a time dependent junction, and we  
9 gradually increase the flow of steam through the bottom to  
10 watch the behavior of the mixture level.

11 MR. CATTON: So how do you define level, some  
12 percent void fraction, 99 percent or something?

13 MR. JOHNSEN: I think you are going to see that  
14 when I describe the model itself, but basically it is a  
15 sharp discontinuity in the void fraction from a liquid  
16 continuous to a vapor continuous flow.

17 MR. CATTON: But in real life if you bubbled steam  
18 in the bottom that doesn't happen.

19 MR. JOHNSEN: What doesn't happen.?

20 MR. CATTON: If your flow rate is anything  
21 meaningful with steam you don't have a sharp surface, you  
22 are throwing little droplets up in the air and all sorts of  
23 things.

24 MR. JOHNSEN: You mean if you begin to entrain the  
25 liquid.

1 MR. CATTON: Yes, so you usually pick something  
2 like .999 or something for void fraction. Go ahead.

3 MR. JOHNSEN: Okay.

4 MR. WULFF: Could you explain what the staircase  
5 vapor flow is supposed to represent?

6 MR. JOHNSEN: Basically what we are doing is  
7 bubbling steam through a stagnant column of water and then  
8 examining how the model predicts how the level swells as a  
9 function of the steam flow rate.

10 MR. WULFF: Where is the steam flow rate on the  
11 right shown on the left, is it TDV?

12 MR. JOHNSEN: Yes, it is entering right here.

13 MR. WULFF: And you are injecting it in this jump  
14 step pattern?

15 MR. JOHNSEN: We are injecting it in the bottom  
16 cell of this column of cells.

17 MR. WULFF: Okay, I understand.

18 MR. JOHNSEN: And it is flowing upward through the  
19 liquid.

20 MR. WULFF: I know that. The question is, why in  
21 this staircase pattern?

22 MR. JOHNSEN: Just to watch the behavior of the  
23 mix level as we gradually increase the steam flow.

24 MR. CATTON: And you run it long enough for it to  
25 settle out?

1 MR. JOHNSEN: YES. We are letting it settle out  
2 at each steam flow, right, and then we are increasing the  
3 flow rate to the next level, watching it stabilize out, and  
4 so on as we continue to increase the steam flow.

5 MR. WILKINS: Do you expect to get different  
6 results that way than if you had done it at one level and  
7 then walked away from the whole experiment for two weeks and  
8 come back and then started at the higher level?

9 MR. JOHNSEN: Yes, we expect to get the same  
10 results. As long as we are not losing any liquid, it will  
11 stabilize.

12 MR. SCHROCK: What are the numbers on this flow  
13 rate scale?

14 MR. JOHNSEN: I am sorry, this is pounds per  
15 second, I left it off the viewgraph.

16 MR. SCHROCK: You keep the steam flow below the  
17 flooding limit, I presume?

18 MR. JOHNSEN: Actually what we did was, eventually  
19 we let this thing go until we drove all the water out.  
20 Eventually it blows all the water out.

21 [Slide.]

22 MR. JOHNSEN: At the steam flow rates that I  
23 showed you on the prior viewgraph, there is insufficient  
24 flow to entrain the liquid out, so the amount of liquid in a  
25 column remains constant at each of those levels I showed you

1 and the code then predicts a level in the control volumes  
2 where the level is detected to be, and the level moves up  
3 the column. It starts out in the lowest most volume. Let  
4 me go back and put that up.

5 [Slide.]

6 MR. JOHNSEN: The level starts out in this control  
7 volume where the liquid starts out, and then it moves up  
8 through this control volume and into this control volume.  
9 You can see that those nodes are three feet in height each,  
10 which would be typical of what we might use in like the CMT,  
11 for example.

12 So you can see what happens is that below the  
13 entrainment limit, the mixture level swells up as we  
14 increase the steam flow. This is what we call the collapse  
15 level, that is the level that we start out with with no  
16 steam flow at all.

17 [Slide.]

18 MR. JOHNSEN: What the model does is, it  
19 calculates a void fraction above and below the level, and  
20 this is the plot that shows the calculated void fraction  
21 beneath the level in the cell in which the level is  
22 calculated to occur and it calculates a void fraction above.  
23 As you can see here for these flow rates that we chose here,  
24 the void fraction above the level remains at unity, which  
25 meaning that we are not entraining any liquid out at these

1 lower flow rates. So that is why the liquid mass remains  
2 constant during this particular set of experiments, if you  
3 will.

4 MR. WILKINS: The difference between those values  
5 of alpha-plus and alpha-minus is what you referred to  
6 earlier as "sharp discontinuity" when you were talking to  
7 Ivan a few minutes ago?

8 MR. JOHNSEN: Right, exactly.

9 So let me then, after showing this little  
10 experiment, move on to the --

11 MR. WULFF: Could I ask a question, is there a  
12 time plot of the level versus time?

13 MR. JOHNSEN: Yes. No, that was versus steam  
14 flow.

15 MR. WULFF: This means you had steady-state  
16 asymptotic values plotted, and I think four values, one each  
17 at these levels, and you have these points.

18 MR. JOHNSEN: Right.

19 MR. WULFF: How was it reached?

20 MR. JOHNSEN: I can show you. I have a back-up  
21 slide here to give you a better feeling for that.

22 [Slide.]

23 MR. JOHNSEN: This is actually a depiction of how  
24 we ran the flow up and then back down again to see if the  
25 mixture level would swell up and then come back down

1 smoothly. So this is what the flow rate looked like as a  
2 function of time starting with no steam flow and then one  
3 pound a second, five, ten, up to 30, and then back down the  
4 same way.

5 MR. CATTON: I think the question that Wolfgang  
6 was asking you is the behavior as a function of time for one  
7 of the steps?

8 MR. JOHNSEN: Yes, that is what I was going to  
9 show you next. I was just showing you the steam flow that  
10 we used.

11 [Slide.]

12 MR. JOHNSEN: This is a plot of the level as a  
13 function of time for that prior steam flow profile I showed  
14 you. So we start out with the level, and these are the  
15 different volumes. This solid line is the volume that the  
16 liquid started out in, Volume 2; then there Volume 3 above  
17 that the dashed line; and then the almost dotted line is  
18 the fourth volume. So as you can see the levels --

19 MR. WULFF: What is on the left side, alpha?

20 MR. JOHNSEN: No, this is level feet. No, this is  
21 meters.

22 MR. WULFF: But then it goes through all the  
23 volumes, why do we have three different plots?

24 MR. JOHNSEN: Once the level moves out of the  
25 lower most cell, it shows up as being full, and then the

1 level goes to the next cell up. So this is the bottom cell,  
2 and this is the level in meters. That is where it starts  
3 out. Then as you increase the floor --

4 MR. WULFF: No, that should be the level at the  
5 lowest flow rate, not at the cell. It had nothing to do  
6 with cell boundary.

7 MR. JOHNSEN: Yes, that is the lowest flow.

8 MR. WULFF: That belongs to the first flow rate,  
9 and the next rate --

10 MR. JOHNSEN: Hold on, Wolfgang, that is the zero  
11 flow rate. We start out at zero and we ended up at zero.  
12 So that is a zero flow.

13 MR. WULFF: So all these plateaus belong to a  
14 steady flow?

15 MR. JOHNSEN: That's right. It steadies out at a  
16 new level each time you increase the flow rate.

17 MR. WULFF: And it goes through all the cells?

18 MR. JOHNSEN: It goes through three cells, it  
19 passes through three cells. So you can see that we have a  
20 level moving up through the cells, and that we also have  
21 symmetry in the going up and coming back down case.

22 MR. WILKINS: And it starts above the bottom cell  
23 and never gets to the top cell?

24 MR. JOHNSEN: It never gets to the top.

25 MR. WILKINS: That is why it goes through three

1 cells.

2 MR. JOHNSEN: Yes.

3 MR. WULFF: So what is the dashed curve and the  
4 dotted curve?

5 MR. JOHNSEN: This is the level in the cell above  
6 where the liquid originally started. So, in other words --  
7 let me get that other diagram out.

8 MR. SCHROCK: Gary, the RELAP5 is supposed to be a  
9 transient two-phased code and there is an interesting  
10 density wave propagation problem here, and it is not  
11 possible to have a correct level coming out of a calculation  
12 that doesn't adequately treat these density waves  
13 propagating vertically in this channel. So I think you have  
14 to look at the development of the two-phased through the  
15 system the changes in the two-phase as you change the flow  
16 rate, as a function of time in order to test the fidelity of  
17 RELAP5 to give you the correct level. I mean you want more  
18 than just the correct level, you want the distribution of  
19 the void.

20 MR. JOHNSEN: Right.

21 MR. SCHROCK: But you are not going to get the  
22 right level if you, in a transient problem, if you do not  
23 have a reasonable representation of the transient void  
24 distribution. What you are showing us is not addressing  
25 transient void distribution.



1           MR. JOHNSEN: And I don't mean to imply that this  
2 particular test problem is a sufficient validation of the  
3 modelling, it is really just a first step to show how it  
4 behaves and that, in a steady-state condition we should be  
5 able to verify we get the correct results for the simple  
6 steady-state case.

7           MR. SCHROCK: But won't the conclusion be  
8 misleading if it only looks at steady-state conditions, and  
9 you are using very, very coarse noding which I don't think  
10 will ever follow the density waves. Then to say that the  
11 level tracking is okay, it isn't going to be okay in  
12 transient circumstances.

13          MR. JOHNSEN: But, Virgil, all I am saying right  
14 now is, this is an initial test of the model in a steady-  
15 state situation. Understanding the fact that it has to work  
16 in a transient fashion and that we have a lot more  
17 validation to do with this model, so I am really only  
18 showing you what our initial results are for simple case  
19 that we can use to verify that it is functioning the way we  
20 intended it to. So your point is well taken, and I am not  
21 saying that by virtue of running this test problem we are  
22 home free, we don't have anything else to do.

23          MR. SCHROCK: The heart of the problem is still to  
24 be addressed then, is that what you are saying?

25          MR. JOHNSEN: Yes.

1 MR. WILKINS: Gary has said that the objective  
2 here is to verify functionality.

3 MR. JOHNSEN: That's right.

4 MR. WILKINS: That is not the same as to validate.

5 MR. JOHNSEN: That remains to be accomplished. I  
6 forgot what I was going to say.

7 MR. CATTON: But you could compare this to some of  
8 the simple solutions like out of Graham Wallis'. You could  
9 compare the analytic result with your computed result?

10 MR. JOHNSEN: Yes, I could.

11 MR. CATTON: Have you done anything like that yet?

12 MR. JOHNSEN: Not yet, we just basically have this  
13 functioning in the code.

14 MR. ZUBER: When did you start on this  
15 development?

16 MR. JOHNSEN: I think about three months ago,  
17 something like that, three or four months ago. I am not  
18 positive about that.

19 MR. CATTON: I am still not clear how you define  
20 your surface.

21 MR. JOHNSEN: How I define what?

22 MR. ZUBER: He treats it as a shock.

23 MR. JOHNSEN: Basically it came out of shock.

24 MR. ZUBER: That expression really is in Ishii's  
25 book, I believe. I am almost sure it is.

1 MR. WULFF: We will come to that.

2 MR. WILKINS: There is another slide about to come  
3 in which he answers that question

4 [Slide.]

5 MR. JOHNSEN: So let's go to the description of  
6 the model now. Basically the purpose of this model is to  
7 recognize the existence and then account for the effects of  
8 a sharp void gradient representing a mixture level and a  
9 vertical component, and the principal features of the model  
10 are that there is detection logic to detect the presence of  
11 a level, and then there is the calculation of initial level  
12 parameters, and then the alteration of the convectant terms  
13 based on the fact that there is a sharp interface and  
14 alteration of the heat transfer, and I will discuss each one  
15 of these in turn.

16 [Slide.]

17 MR. JOHNSEN: Incidentally, I should preface this  
18 by saying that this model was taken from the TRAC EWR code  
19 where it was initially developed.

20 So first we have to determine what is an  
21 acceptable criteria for a mixture level and this is probably  
22 one of the most difficult things to do since there is always  
23 going to be some degree of question as to whether the  
24 detection logic is adequate.

25 What we have in the code right now is the

1 following criteria for the detection of level. First of  
2 all, if we are looking at Volume J and testing Volume J as  
3 potentially having a level in it, we first confirm that  
4 there is no level above or below Volume J.

5 Secondly, we require that there be at least a void  
6 fraction differential between Volume J and the volume above  
7 it of 0.2, or that there be the same gradient between J and  
8 the volume below it.

9 Thirdly, we require that the volume above have a  
10 void fraction of .7 or greater.

11 MR. WILKINS: Let me just see if I understand  
12 that. The first one says, in order to determine whether  
13 there is a level in Volume J, I look first to J-plus-1 and  
14 then at J-minus-1 and discover that there isn't one?

15 MR. JOHNSEN: Right.

16 MR. WILKINS: How do I do that?

17 MR. CATTON: The void fraction in J-plus-1 is  
18 what?

19 MR. JOHNSEN: In other words, from the prior time  
20 step there was no level in either the volume above or the  
21 volume below.

22 MR. WILKINS: So this is a criteria for mixture  
23 level at time T1.

24 MR. JOHNSEN: Some T.

25 MR. WILKINS: And you have already gone through it

1 at T1 minus delta T or whatever.

2 MR. JOHNSEN: Yes, right.

3 MR. WILKINS: And you got started at time zero and  
4 you knew exactly where the level was, it was in Volume 2,  
5 that is the collapse level.

6 MR. JOHNSEN: In that little test problem, that  
7 would have been the case, yes.

8 MR. KRESS: I don't understand why you use the J-  
9 minus-1 in that case to exclude the level in J because you  
10 have changed the time step, it might have moved from J-  
11 minus-1 to J.

12 MR. JOHNSEN: If it had, then there would be no  
13 level in J-minus-1 and we would satisfy the criteria.

14 MR. KRESS: Then I have the same question that  
15 Ernest had, how do you check to see if there was a level in  
16 there?

17 MR. JOHNSEN: I am not understanding your  
18 question.

19 MR. CATTON: Do you flag the cell that the level  
20 is in?

21 MR. JOHNSEN: Yes.

22 MR. CATTON: So your step number one is to look  
23 for the flag?

24 MR. JOHNSEN: Right, look for the flag, and each  
25 volume of the stack for which this model is turned on, we

1 will be checking their logic.

2 MR. WILKINS: Let me try again, let's say you are  
3 at time T that you are looking at, at time T minus delta T,  
4 which is the previous step, you look to see if there is a  
5 level in J-plus-1 and J-minus-1?

6 MR. JOHNSEN: Right.

7 MR. WILKINS: And there had better not be any,  
8 otherwise there is no chance for there to be a level in  
9 Volume J at time T.

10 MR. JOHNSEN: Right.

11 MR. WILKINS: That is what I interpret this.

12 MR. JOHNSEN: Right.

13 MR. WILKINS: Now what Tom has said is, it might  
14 have moved from J-minus-1 to J during this time delta T, and  
15 so the level might well have been in J-minus-1 at time T  
16 minus delta T and it moved up to level J at time T. That is  
17 the problem we are having in understanding the first  
18 criteria.

19 MR. CATTON: Gary, you are about to get some help.

20 MR. WEAVER: Walt Weaver from INEL.

21 This criteria is only used if there is no level in  
22 any of the cells. You can start a problem out in a stack of  
23 cells and there is no level any place. We are trying to  
24 find out if during the evolution of the transient a level  
25 appears. Once a level appears, we move it from cell to cell

1 in a deterministic way.

2 So these criteria are only for the appearance in a  
3 stack of cells for a level when previously none appeared any  
4 place.

5 MR. JOHNSEN: But it can disappear and reappear.

6 MR. WEAVER: It can disappear, and once the level  
7 disappears out of the stack, then you have to check to see  
8 whether it reappears in the stack. So these criteria are  
9 used only for the appearance of a level when none previously  
10 existed. If a level exists at the beginning of a time step,  
11 we move it with a level velocity from cell to cell or within  
12 a cell.

13 MR. CATTON: So what is your criterion for a level  
14 being in a cell?

15 MR. WEAVER: This is the criteria. We say the  
16 level is in cell J and we exclude levels in cells above and  
17 below because we don't want to have the problem that RELAP4  
18 had which was levels in adjacent cells which is unphysical.

19 MR. CATTON: I guess I am still perplexed. I  
20 would have thought maybe you would have had a void fraction  
21 criterion of some kind.

22 MR. WILKINS: They do, that is numbers 2 and 3.

23 MR. CATTON: Yes, but .2 is awful big. I mean  
24 awful small.

25 MR. JOHNSEN: That is one of the difficulties.

1 is detecting the level involves a void criteria. The  
2 question is what number do you pick.

3 MR. CATTON: That's right. For example, if you  
4 were trying to develop a mixture level for reflood just for  
5 example, not that it is important or anything else, but, for  
6 example, if you don't carry it to 99 percent, you won't get  
7 the right results. So I don't know what you would use for  
8 that difference.

9 MR. SEALE: Aren't you really saying that a level  
10 exists in the cell where you have a change in void fraction  
11 greater than two-tenths with the proviso that above that  
12 level the void fraction has to be greater than seven-tenths?

13 MR. JOHNSEN: Right.

14 MR. SEALE: That is really all you are saying?

15 MR. JOHNSEN: That's right.

16 MR. WULFF: He makes a difference, he looks at  
17 differences in void fractions.

18 MR. SEALE: A change in void fraction of two-  
19 tenths.

20 MR. CATTON: I wasn't looking at the third one up  
21 there. He is saying that if the void fraction is greater  
22 than .7, the level is below. I can be pretty wet.

23 MR. JOHNSEN: This is the cell above.

24 MR. CATTON: That is what I was asking you  
25 earlier, what your cut-off value was, and that is it.



1 MR. JOHNSEN: .7, and the transition to, let's  
2 say, annular flow from bubbly slug flow is going to be in  
3 that range. If you have to pick a number, it is

4 MR. CATTON: You are probably going to have to  
5 fool around with these criterion a little bit.

6 MR. JOHNSEN: The validation is going to have to  
7 indicate whether or not this is adequate or not. You are  
8 right, we are going to have to test it and see if it is  
9 adequate, but I suspect that for example for the CMT that  
10 most of the time we will have a fairly sharp interface.

11 MR. CATTON: Well, I guess if the ADS valve is  
12 open like Larry said, 120 seconds or something, you are  
13 right.

14 MR. DHIR: This will be a good test also to check  
15 for interfacial drag correlations, see if you go on the one  
16 route and compare it with whatever is the level 2 --

17 MR. JOHNSEN: I can tell you that we do blow all  
18 the water out eventually.

19 MR. DHIR: But to compare at what velocity you  
20 blow it out.

21 MR. JOHNSEN: Compared to the flooding velocity,  
22 yes.

23 [Slide.]

24 MR. JOHNSEN: Once the level has been detected,  
25 then the model calculates several different parameters. It

1 calculates void fractions above and below the level, and  
2 those are denoted as alpha-J-plus above the level and alpha-  
3 J-minus below the level. This would be the length of the  
4 cell delta Z, and a Z-sub-L is the position of the level,  
5 and we also calculate the velocity of the level which  
6 enables us to predict when the level will cross a cell  
7 boundary above or below.

8 [Slide.]

9 MR. JOHNSEN: The model assumes that below the  
10 mixture level alpha-J-minus is equal to the void fraction in  
11 the cell below, alpha-J-minus-1. For the void fraction  
12 above the level, the model checks to see if an entrainment  
13 criteria has been exceeded. This is a model from Rosen. If  
14 it has not, then the void fraction above alpha-J-plus is set  
15 equal to the void fraction in the cell above alpha-J-plus-  
16 1.

17 If the entrainment criteria has been exceeded,  
18 then it calculates the void fraction based on the amount of  
19 entrainment that is calculated to occur, and that involves a  
20 critical velocity which is basically a flooding type  
21 velocity.

22 MR. ZUBER: What is the V-sub-F, velocity of what?

23 MR. JOHNSEN: This is the velocity of the liquid.

24 MR. ZUBER: The droplets of both, or what liquid?

25 MR. JOHNSEN: The liquid flow above the level.

1 Let me see if I can get --

2 MR. CATTON: But in your stagnant tube isn't that  
3 zero?

4 MR. JOHNSEN: Yes, it is. In fact, I haven't  
5 confirmed this yet, but --

6 MR. CATTON: That will give your code a headache.

7 MR. JOHNSEN: I haven't confirmed this yet, but it  
8 appears that the inherent interfacial drag in the code will  
9 levitate the water before this criterion is achieved.

10 MR. WULFF: But G entrainment is the massed flow  
11 of the liquid above the level; is it not?

12 MR. JOHNSEN: Yes, it is the mass flux.

13 MR. WULFF: And so is  $\rho \cdot V \cdot V$ , if you say it is  
14 the velocity above. So that would be 1 and you would have  
15 zero for  $\alpha$ -plus.

16 MR. JOHNSEN: I am looking for another slide that  
17 I have on this entrainment. Let me try to come back to that  
18 because I have another slide that explains this in more and  
19 more detail.

20 This is the initial criteria, but the bounding  
21 limits are that the void fraction below the level cannot be  
22 greater than the average void fraction in the cell where the  
23 level exists, and that the void fraction above the level  
24 can't be smaller than the average void fraction in that  
25 cell. Those are bounding limits in think in a physical

1 sense.

2 [Slide.]

3 MR. JOHNSEN: The position and the velocity level  
4 is shown in this slide.

5 MR. WULFF: I have a question, why did these  
6 limits come up? Did the calculation come up with values  
7 alpha-plus and alpha-minus outside these bounds that you  
8 have to impose them?

9 MR. JOHNSEN: No. Again, this model was developed  
10 by GE under the TRAC code.

11 MR. WULFF: It seems to me that the calculation  
12 required to impose this boundary, and then something is  
13 wrong with the calculations

14 MR. JOHNSEN: I can't say precisely. Walt, do you  
15 know why those were included?

16 MR. WILKINS: Weaver from INEL.

17 Those are the physical limits. If the void  
18 fraction below is greater than the cell average void  
19 fraction, then the level had gone out the top of the cell.

20 MR. WILKINS: I don't think you have understood  
21 Mr. Wulff's question yet. He says, those limits are Mother  
22 Nature's limits and you shouldn't have to impose them. If  
23 your calculation is consistent with Mother Nature, they  
24 should follow. You should be able to look at your results  
25 and discover that the results you had are consistent with

1 those limits. I believe that is what Mr. Wulff is saying.

2 MR. WULFF: Correct.

3 MR. WILKINS: And that if you in fact have to  
4 impose them or if, in some place in the calculation you have  
5 to impose them because otherwise you would have an alpha  
6 that would be outside those limits, then there is something  
7 unphysical about the calculation.

8 MR. CATTON: What do you do if those limits are  
9 violated, is that just an error check in the code?

10 MR. SEALE: Check your arithmetic.

11 MR. WULFF: No, it bounds it. It is saturating.

12 MR. CATTON: Yes, but your original comment was  
13 that they really don't need them. So if they do have them,  
14 what do they do with them? Probably use them as error  
15 check, if you get outside stop, something is wrong.

16 MR. WULFF: No, they don't stop. They continue  
17 with the limit.

18 MR. JOHNSEN: I think the coding goes back to  
19 that.

20 MR. WILKINS: Yes, that is what I would expect.  
21 If in an actual case you do reset them, then I think Mr.  
22 Wulff's comment is appropriate.

23 MR. WULFF: It violates Mother Nature.

24 MR. WILKINS: It is something that needs to be  
25 explained.

1           MR. KRESS: You always violate Mother Nature when  
2 you use a finite difference code, and you could end up with  
3 finite difference problems that will give you this  
4 situation.

5           MR. WULFF: I asked for an explanation or for the  
6 reasons why this was imposed.

7           MR. KRESS: I understand.

8           MR. JOHNSEN: We have not seen a violation. That  
9 was not the reason we included those limits. Those limits  
10 made physical sense to us and they were taken from the TRAC  
11 model. We haven't seen any violation.

12           MR. WULFF: Maybe your recommendation is to take  
13 them out and see what happens.

14           MR. JOHNSEN: Your recommendation is to take them  
15 out and see what happens.

16           This is the calculation of the levels. It is just  
17 based on conservation principles using the void fractions I  
18 indicated earlier for above and below the level and the  
19 average void fraction in the cell and differentiating that  
20 with respect to time gives us an expression for the movement  
21 of the level, the rate of change of the position of the  
22 level.

23           MR. WULFF: I think here is the answer to Virgil's  
24 question before. The first expression is the starting  
25 expression. It has no kinematics in it, no kinematics from

1 the void propagation beneath. It has no kinematics in it  
2 from the gas velocity or the liquid velocity, and by  
3 differentiation you cannot recover the terms that were  
4 deleted in order to reach the steady-state expression at the  
5 top.

6 So I ask you, what is the justification for  
7 differentiating the steady-state expression on top in order  
8 to get the transient jump condition for the motion of the  
9 interface.

10 MR. JOHNSEN: Did you read my next slide?

11 MR. WULFF: No.

12 MR. ZUBER: Yes, I did, and that is wrong.

13 MR. JOHNSEN: And it is wrong?

14 MR. ZUBER: Yes.

15 MR. CATTON: Is this a quasi-steady approximation?

16 MR. JOHNSEN: Yes.

17 MR. CATTON: So you are ignoring accelerations?

18 MR. JOHNSEN: Right.

19 MR. CATTON: And your basis is the 120-second  
20 opening time?

21 MR. ZUBER: It has nothing to do with --

22 MR. JOHNSEN: No.

23 MR. WULFF: He ignores the kinematics of what  
24 happens.

25 MR. ZUBER: Actually, the answer is really on the

1 next page. What Virgil said was absolutely correct, what  
2 you really have is a jump condition which is the one below  
3 Equation 7, and this is the one that Ishii has.

4 MR. JOHNSEN: That's right.

5 MR. ZUBER: There is no argument with that. What  
6 is inconsistent is this Equation 5, that is in error.

7 [Slide.]

8 MR. JOHNSEN: Let me go through this so that there  
9 is some continuity involved. When we get to Equation 5,  
10 we --

11 MR. ZUBER: Actually, if you use this in your  
12 argument, this is correct based on Equation 5, you have no  
13 argument because Equation 5 is incorrect.

14 MR. JOHNSEN: What I am trying to show is that  
15 this model will devolve to the jump condition.

16 MR. WULFF: Only if Equation 5 were right.

17 MR. JOHNSEN: Let's go through it then. Again,  
18 the expression for the level, differential of the level, and  
19 now we have to make an assumption that these substantial  
20 derivatives are equal to zero, which says basically that the  
21 void fraction below the level and the void fraction above  
22 the level are time invariant which is a quasi-steady  
23 assumption. When you do that, then the expression  
24 simplifies to this expression for the differential.

25 [Slide.]



1 MR. JOHNSEN: Now we go to Equation 5. Equation 5  
2 basically says that the rate of change of void fraction in a  
3 cell is equal to the difference in the mass flux below and  
4 above the level times the area divided by the volume.

5 MR. ZUBER: Let me say what is wrong with that.  
6 What you have in two-phased flow, and actually we derived 20  
7 years ago and I think you will find it in textbooks, is the  
8 void propagation equations, and you can use it at volume  
9 average and you would obtain it.

10 The point is, first, you don't have a source for  
11 generation of vapor or condensation and  $\omega$ . If you look  
12 at the void propagation equation, it is at the  $\alpha$ , the  $T$ -  
13 plus- $C$ -sub- $K$ , and  $C$ -sub- $K$  is the velocity of the kinematic  
14 phase time the  $\alpha$ . This is equal to an  $\omega$ , plus you  
15 have flushing that can be minus, you have condensation.

16 Number one, you don't get that  $\omega$  at all, and  
17 what you are really basing here on fluxes and what you  
18 should really have is on a kinematic base, but you cannot  
19 apply this to either condensation or evaporation because you  
20 don't have a source for vapor.

21 Look, really go back and look at the void  
22 propagation equations, how it is formulated, I think GE  
23 probably has it, Wolfgang has it, we had it 25 years ago.

24 MR. JOHNSEN: I did. I went back and looked.

25 MR. ZUBER: Then you don't have the source there.

1 MR. JOHNSEN: I am only trying to show in these  
2 two slides that the model we are using can be shown to be  
3 consistent with the jump condition, that is all I am trying  
4 to show. I am not showing you that this is the model.

5 MR. WULFF: You tried, but it didn't work, and it  
6 because of Equation 5 being wrong, and I think doing  
7 swelling the alpha-minus-DT cannot possibly be zero or small  
8 enough to be neglected because you are coming up with vapor  
9 catching up with the interface and percolating through, and  
10 that is a transient. So I think you are making a number of  
11 assumptions and then show that this Equation 2 has something  
12 to do with this jump condition.

13 MR. JOHNSEN: Right.

14 MR. WULFF: You tried, it didn't work.

15 MR. JOHNSEN: I disagree.

16 MR. ZUBER: The jump condition, the last equation,  
17 you should also have a term either for vapor evaporation at  
18 the interface or condensation.

19 MR. JOHNSEN: This is in the absence.

20 MR. ZUBER: In absence. I think this is what  
21 Ishii had. I have no qualms. So you could have started  
22 with this, but the Equation 5, if you apply it to a CMT or  
23 anywhere in the reactor, it is wrong.

24 MR. JOHNSEN: But I am not applying that. The  
25 model doesn't apply it. All I tried to do was to start with

1 what the model has and show that --

2 MR. ZUBER: It should not have that, period.

3 MR. JOHNSEN: It could end up --

4 MR. ZUBER: No, it cannot.

5 MR. JOHNSEN: That is all I was trying to show. I  
6 am not saying the model incorporates that.

7 MR. CATTON: If you are bubbling the air through,  
8 this would be okay, wouldn't it?

9 MR. SCHROCK: No. It isn't right even for  
10 bubbling air through because the interface is moving, and so  
11 the alpha-plus/alpha-minus are changing with time due to the  
12 fact that the interface is moving through their mesh.

13 MR. JOHNSEN: They don't have to.

14 MR. SCHROCK: They have to if you are doing a  
15 transient calculation, and that is where you are going.

16 MR. WULFF: You open the valve.

17 MR. JOHNSEN: If I am moving with the mixture  
18 below the mixture level, the void fraction could look at  
19 time and variant, and yet the level is moving.

20 MR. SCHROCK: Not in your mesh it can't, not the  
21 way you calculated the void.

22 MR. WULFF: And certainly not in the experiment or  
23 the calculations you showed us at the beginning where you  
24 introduce step-wise steam that then arrived some time later  
25 at the interface and percolated through. So, at one time,

1 you had one value of alpha, then you have changed the flow  
2 rate of steam and you have a different alpha.

3 MR. JOHNSEN: That's correct.

4 MR. WULFF: So there must have been a transient in  
5 which that alpha changed.

6 MR. JOHNSEN: That's correct.

7 MR. WULFF: Maybe above, if there was no  
8 entrainment, it was always 1. That is okay.

9 MR. JOHNSEN: It was always 1, that's right.

10 MR. WULFF: But below it cannot be said equal to  
11 zero, just from physics.

12 MR. JOHNSEN: I did show you that it is not, it  
13 does change.

14 MR. ZUBER: Gary, let me say this problem can be  
15 easily formulated in the correct way, and we should not  
16 really spend our time arguing.

17 MR. JOHNSEN: All right, let's go on.

18 MR. ZUBER: It is not a good formulation period.

19 MR. JOHNSEN: Again, Equation 5 is not in the  
20 code. It is only there to illustrate how we got --

21 MR. ZUBER: The point that Virgil brought is, you  
22 are going to use this code in transient conditions, so  
23 please listen to what we are saying. I think the approach  
24 you want to do it is good, I mean the philosophy is good,  
25 the tools you are developing. They way you are developing

1     them are not good. It can be done easily and in a  
2     satisfactory way. I urge you really to do it because the  
3     only chance you have really to address the CMT and all these  
4     problems with moving interface for these will depend on how  
5     you model this.

6             MR. WULFF: Could you tell us, in RELAP the  
7     difficulties are in implementing what you have here as  
8     Equation 7 directly, or below Equation 7. Start with  
9     Ishii's jump condition and you have all the defenses behind  
10    you. Is there something in RELAP that --

11            MR. JOHNSEN: I don't think this is really  
12    inconsistent.

13            MR. WULFF: You were just convinced that it is. I  
14    am just asking, is there something in RELAP5. It is a  
15    different question now. Is there something in RELAP5 that  
16    prevents you? As far as I can tell, you have the alphas,  
17    you have the velocities of the vapor, you are computing that  
18    and you have been computing it for many years. Can you not  
19    just plug this in and calculate the time rate of change of  
20    the level, the level velocity?

21            MR. JOHNSEN: I can't answer your question. I  
22    don't know. I don't know off the top of my head.

23            [Slide.]

24            MR. JOHNSEN: Now I want to talk about what the  
25    model ends up affecting within the code itself. Without the

1 model, RELAP normally would convect with what we call a  
2 self-centered average volume properties, so if the flow were  
3 upward, for example, over here, we would convect the average  
4 void in this cell to the next cell up.

5 With the level tracking model, we alter the void  
6 fraction of the conservation equations and this little chart  
7 shows how the void fraction above and below the mixture  
8 level is altered to recognize the existence of the level. I  
9 won't go through the whole table, I think most of it is  
10 self-explanatory.

11 [Slide.]

12 MR. JOHNSEN: The hydrostatic head term was also  
13 modified again to recognize the sharp interface. Normally  
14 without the level tracking model, we simply do a  $\rho G \Delta Z$   
15 calculation using the different densities in the cell  
16 above and below to formulate the gravitational head term for  
17 the momentum equation. With the level tracking model, we  
18 hit a more accurate differential pressure due to the  
19 location of the level.

20 [Slide.]

21 MR. JOHNSEN: Finally, we alter the heat transfer  
22 calculation to recognize the position of the level relative  
23 to a heat structure that is adjacent to that vertical cell,  
24 and this proportionality is performed based on the position  
25 of the level, so that the heat transfer to the vapor space,

1 the continuous vapor space, is based on the location of the  
2 level in the cell, and likewise for the liquid.

3 MR. DHIR: How do you calculate these heat  
4 transfer coefficients?

5 MR. JOHNSEN: Pardon?

6 MR. DHIR: How would you calculate this heat  
7 transfer coefficients? Suppose you had a stagnant layer of  
8 vapor on top or gas and the liquid is at the bottom, a two-  
9 phased mixture?

10 MR. JOHNSEN: If it were stagnant?

11 MR. DHIR: Yes.

12 MR. JOHNSEN: First of all, we have a heat  
13 transfer package which, in essence, I could say is the  
14 boiling curve, but it also has condensation in it.

15 MR. DHIR: But the problem is, what lengths do you  
16 use in calculating the heat transfer coefficients?

17 MR. JOHNSEN: The length is input by the user for  
18 that particular heat structure. The user inputs the length,  
19 the length term that is appropriate for that particular heat  
20 structure. He doesn't assume it is necessarily --

21 MR. WULFF: If above the level there is  
22 condensation, you need to know the driving length above the  
23 level, that is from the level on up to the top of the wall?

24 MR. JOHNSEN: Yes.

25 MR. WULFF: How can the user put that in?

1           MR. JOHNSEN: That is one of the compromises you  
2 have to make. You have to pick a fixed level. We don't  
3 calculate -- we don't say, what is the distance between here  
4 and the top of the wall where the condensation might be  
5 taking place, so there is some imprecision in using some  
6 fraction of the length specified by the user.

7           MR. SCHROCK: Gary, this brings up one of the  
8 things that came out of the review of the documentation. I  
9 commented on it in my review and I saw that it was commented  
10 by others in their reviews. There are number of places in  
11 RELAP5 where there are very arbitrary choices made about the  
12 length scale for heat transfer coefficient evaluation. In  
13 some instances, the dimension of the computation itself is  
14 taken as a length scale for the evaluation of a heat  
15 transfer coefficient. That is unfounded, there is no basis  
16 on which to argue that an arbitrary choice made for a  
17 numerical computation has any influence on the physical  
18 problem of the heat transfer coefficient determination.

19           So this is an example of the kind of thing that is  
20 wrong in RELAP5, people have been telling you for years is  
21 there. You have heard it. Nothing happens a. t it, your  
22 sponsor doesn't think it is important, and so you don't do  
23 anything about it. How in the world can we evaluate a code  
24 that has this kind of nonsense in it when you won't  
25 recognize that the question has even been asked?



1 Is my point clear?

2 MR. JOHNSEN: Was it a point or a question?

3 MR. SCHROCK: It is a point and a question.

4 MR. JOHNSEN: Well, it has been changed in the  
5 code. It is no longer the diameter of the cell or the  
6 length of the cell, it is a user input number associated  
7 with the heat structure in the cell. So what is in the  
8 documentation from 1990 is obsolete at this point.

9 MR. CATTON: Let me see if I understand that, does  
10 that mean that if this cell, say, is five cells below the  
11 top I will have a condensation length scale and the user  
12 would input the total distance for this cell?

13 MR. JOHNSEN: The user would have to make --

14 MR. CATTON: I can understand where you might  
15 argue that in order to implement the code, and I don't know  
16 what this split is between the two here, what ZL is, I will  
17 just arbitrarily select the length scale to be from the top  
18 to the middle of that cell and let the air be what it may.  
19 What do you do, what does the user do?

20 MR. JOHNSEN: The user has to make a judgment as  
21 to what a reasonable length is in consideration of the --

22 MR. CATTON: That means the user has to be highly  
23 skilled in the world of transport phenomenon.

24 MR. JOHNSEN: Assuming, of course, that it makes a  
25 big difference.

1 MR. CATTON: In the case where it is, the Reynolds  
2 number, usually heat transfers are tied with a Reynolds  
3 number or something, and if I take one cell and compare that  
4 with three cells, I can be off a factor of two.

5 MR. DHIR: Also it is changing with time, so you  
6 can't prespecify.

7 MR. JOHNSEN: It is changing with time.

8 MR. DHIR: Every time it has to be calculated  
9 which is the proper length scale, it has to be checked.

10 MR. JOHNSEN: That's true.

11 MR. DHIR: So how can the user tell.

12 MR. JOHNSEN: Let me say that most of the  
13 application that we put the code to, this just is not all  
14 that important.

15 MR. CATTON: But we are in the CMT now where you,  
16 yourself, have discovered the condensation is important. So  
17 you almost need a length scale for force convection, a  
18 length scale for condensation, a length scale for this, that  
19 and the other.

20 MR. SEALE: Is the ultimate intent, recognizing  
21 that there is a lot of documentation that still has not  
22 appeared yet, but is the ultimate intent in the  
23 documentation to provide, among other things, instructions  
24 for the user to use in establishing what these various  
25 lengths should be as the user input so that the best

1 judgment or the best intent of the developer can be  
2 reflected in the abuse or lack of abuse that the user uses  
3 on the code when they apply it? Is that part of the  
4 documentation that you plan to have?

5 MR. JOHNSEN: Yes. There is a volume that is a  
6 user guidelines, that is part of the code record.

7 MR. SEALE: And this kind of cheat sheet will be  
8 in it, that is something that will tell them how to scale  
9 those lengths for those calculations?

10 MR. JOHNSEN: I can't confirm.

11 MR. WULFF: It is impossible to do this because,  
12 as we had seen before, the condensation heat transfer  
13 coefficient starts up very high, then comes down to a value  
14 of one-third or so, and it really represents the thickness  
15 of the film that we have. While the level is way up  
16 throughout the top, that film is thin, a very high  
17 condensation. How can some user account for the drainage of  
18 the tank and condensation during that transient, it is  
19 impossible. It ought to be computed and it is easy to  
20 compute.

21 MR. JOHNSEN: It is not easy to compute, Wolfgang.  
22 Don, do you want to comment on that?

23 MR. McELIGOT: Don McEligot, INEL.

24 I just wanted to reemphasize something that you  
25 said earlier, Gary, relating to the aspect that often this

1 question is not important. I think looking from the  
2 outside, your experience has been in most applications you  
3 have had thick walls and if you had liquid film or liquid on  
4 the inside, typically you have high heat transfer  
5 coefficients. Let's say low thermal resistances compared to  
6 the thermal resistance of the wall. So in a case like that,  
7 these questions of length scale details become less  
8 important because it is really only an order of magnitude  
9 that one needs to have in order to show that it is something  
10 else that is driving the problem.

11 That aspect was pointed out by Gary yesterday when  
12 he used order of magnitude arguments in order to come up  
13 with an order of magnitude of the BO number, which is what  
14 you need here.

15 Now let me make one other comment, in the case  
16 where you happen to have turbulent natural convection,  
17 typically the length scale drops out of the relationship  
18 because you have Nusselt number versus Rayleigh number to  
19 the one third power, and the length scale you can pick  
20 anything you want, and there you could pick a -- it wouldn't  
21 make physical sense in the way that you say, but you could  
22 pick a unit cell length and it is still going to cancel out.

23 MR. DHIR: This code is also going to be used to  
24 validate it from the experimental data and the experiments  
25 may not have thick walls. So there your evaluation of

1 condensation heat transfer coefficient becomes important.  
2 Also natural convection numbers he showed, I think they are  
3 in error by a factor of three or four at least.

4 MR. McELIGOT: Three or four is a drop in the  
5 bucket, I think. We are looking at order of magnitudes  
6 here, a factor of 10 I would begin to worry about.

7 MR. ZUBER: Let me say something, I have no qualms  
8 with order of magnitudes. I think this is good thinking. I  
9 am quite distressed when they bring something in a meeting,  
10 for example, two years ago, we brought the questions of  
11 condensation, we brought the questions of level tracking, it  
12 was pooh-poohed. Now we can see it is important. In the  
13 meantime, two years have passed and the things we bring up  
14 are in good faith, our concern is in order so that NRR has  
15 something to audit with that we can defend outside. I think  
16 take our comments with that sense, but don't try to dismiss  
17 it a priori and then the things faces up two years later, it  
18 is too late then to do anything.

19 MR. SCHROCK: There is another point, Gary, which  
20 is very simple, that is that you can argue that the heat  
21 transfer coefficient accuracy is not a very important fact  
22 in the accuracy of the overall code prediction for many  
23 situations, for many situations. The truth is, if you have  
24 garbage in the code to get the heat transfer coefficients,  
25 you are going to get garbage when it does become important.

1 You will never know from this code whether the heat transfer  
2 coefficient is important or unimportant if you continue to  
3 use garbage in the determination of the heat transfer  
4 coefficient.

5 I am not arguing, I don't think anybody here is  
6 arguing that you cannot show that there is an insensitivity  
7 to the accuracy of the heat transfer coefficient in a lot of  
8 situations. Maybe predominantly so, but that does not prove  
9 that there are not situations in which the reverse will be  
10 true and there is where you were in serious trouble.

11 MR. SHERON: I need to ask a question here,  
12 Virgil, if I could, because I have been listening to this,  
13 and I remember some time ago we were sort of admonished by  
14 the committee here for not using textbook correlations,  
15 literature-based correlations, and so forth. We tried to do  
16 that, and I am getting the impression most of these  
17 literature-based correlations have length scales in them,  
18 and now we are being admonished because we can't come up  
19 with some characteristic length when we apply it to these  
20 strange geometries that we didn't design but we have to  
21 evaluate.

22 What are you suggesting that we do, are we  
23 supposed to go back and build a CMT and run specific tests  
24 to get a heat transfer correlation for that particular  
25 geometry or what?

1 MR. SCHROCK: What I am suggesting is that there  
2 are more intelligent ways to estimate the heat transfer  
3 coefficient than to say, let's chose an arbitrary length  
4 scale which is based upon the totally unrelated issue of  
5 what the discretization of this system that we are analyzing  
6 has chosen to be.

7 MR. SHERON: He just said we are not doing that.

8 MR. JOHNSEN: I just told you we are not doing  
9 that.

10 MR. SCHROCK: I don't believe that. The code is  
11 filled with it.

12 MR. JOHNSEN: Then don't believe it.

13 MR. SCHROCK: Have you taken it all out now?

14 MR. JOHNSEN: I just told you that it is not that  
15 way in the code.

16 MR. WULFF: It is user input is what you --

17 MR. JOHNSEN: It is a user input, it doesn't  
18 assume the cell length, it doesn't assume the cell diameter.  
19 It is a user inputting number.

20 MR. SCHROCK: You are talking about the specific  
21 model that you are addressing here today, or are you talking  
22 about the code in general?

23 MR. JOHNSEN: The code in general.

24 MR. WULFF: But here you are taking the height,  
25 the full height of the tank, which is a simple geometry even

1     though the level may be close to the top and you should use  
2     a small height.

3             If it is not important, why are we splitting  
4     within one cell the distance and then make this assignment?  
5     If it doesn't matter, you can just use your heat transfer  
6     coefficient, either one, either the lower one or the upper  
7     one for the whole cell and it wouldn't matter.

8             MR. JOHNSEN: If I recall correctly from the March  
9     meeting, this was a specific question as to whether or not  
10    we recognized the level in the cell with regard to the  
11    calculation of the heat transfer.

12            MR. WULFF: Yes.

13            MR. JOHNSEN: Now you are telling me you don't  
14    think it is important; is that right?

15            MR. WULFF: No, I am not saying it is unimportant.  
16    I am saying you are inconsistent by saying it is not  
17    important on the overall result, and yet you put great  
18    importance on splitting this. Somewhere it is either  
19    important or not important, but not both at the same time.

20            MR. CATTON: If it would be possible for us to get  
21    some documentation on this, we would appreciate it.

22            MR. JOHNSEN: Of what, documentation of this  
23    model?

24            MR. CATTON: Yes. The other thing is, if it is  
25    user input, in the CMT the level is moving.



1 MR. JOHNSEN: Yes.

2 MR. CATTON: If the length scale is user input,  
3 how do you do that, or what do you do, does he input the  
4 length scale relative to a level or something?

5 MR. JOHNSEN: In those cases where the length  
6 scale plays a role in the computation of the heat transfer  
7 coefficient and, of course, that depends on what heat  
8 transfer mode you are in, but to the extent that it plays a  
9 role, then the code would use the user input value for that  
10 particular heat structure, and each heat structure is  
11 independent.

12 MR. CATTON: So it might have a heat structure  
13 that is at the mid-level of the tank.

14 MR. JOHNSEN: Right. The user has to specify a  
15 length.

16 MR. DHIR: But it can't be specified, it is  
17 changing. So either you have to give some fixed evaluations  
18 and let it track the level and then calculate what the  
19 height should be.

20 MR. JOHNSEN: Take for example the condensation.  
21 I think I showed you yesterday that beyond a half a meter  
22 length, it is relatively insensitive to length for the  
23 preconvection.

24 MR. DHIR: Let's say it is three inches, you want  
25 to calculate three inches of your surface which is three

1 inches tall and it is condensing. How did that user know if  
2 it is going to be three inches or six inches?

3 MR. JOHNSEN: Here again the user has to make a  
4 judgment as to what modes are going to occur and based on  
5 that pick a length scale that he thinks is going to give him  
6 a reasonable answer.

7 MR. SHOTKIN: Gary, can I just say something. We  
8 appreciate the ACRS' comments and we can take an action to  
9 go review our users' guidance document, update it as needed  
10 to include better guidance on the choosing of the length  
11 scales, and we will have that ready for a committee review  
12 when the documentation is ready. Is that okay?

13 MR. CATTON: That's fair enough. I think that is  
14 good.

15 MR. ZUBER: Can I just have a question on the  
16 previous slide, I didn't get a chance. H1, let's assume  
17 that we have a vapor above it.

18 MR. JOHNSEN: Right, vapor continuous.

19 MR. ZUBER: H2, we have liquid and bubbles?

20 MR. JOHNSEN: Right.

21 MR. ZUBER: Suppose I have 50 percent vapor.

22 MR. JOHNSEN: Below.

23 MR. ZUBER: Below. Can it be according to your --

24 MR. JOHNSEN: Sure.

25 MR. ZUBER: How do you calculate H2? I mean how

1 do you calculate the heat transfer coefficient, H2?

2 MR. JOHNSEN: It would be based on what the wall  
3 temperature was relative to the two-phased mixture. It  
4 could be boiling or it could be condensation.

5 MR. ZUBER: Yes, but --

6 MR. JOHNSEN: The heat transfer subroutine  
7 checks --

8 MR. ZUBER: In your subroutines you split the heat  
9 transfer coefficient according to the void fraction and you  
10 say, for example, 50 percent would be from the wall to the  
11 vapor, and 50 percent from the wall to the liquid.

12 MR. JOHNSEN: It depends on what heat transfer  
13 mode it is in. If it is in boiling, it all goes to the  
14 liquid.

15 MR. ZUBER: For example, this is not boiling, but  
16 you are heating it.

17 MR. JOHNSEN: Let's say you are heating it, for  
18 example?

19 MR. ZUBER: Yes.

20 MR. JOHNSEN: Then it would go to the liquid. A  
21 50 percent void fraction?

22 MR. ZUBER: Yes, 50 percent liquid, 50 percent  
23 vapor and it is a bubbly flow?

24 MR. JOHNSEN: Yes, it would all go to the liquid.

25 MR. ZUBER: Are you sure?

1 MR. JOHNSEN: I am going to cover that topic later  
2 on, by the way.

3 MR. ZUBER: Okay, good.

4 MR. JOHNSEN: There is a separate talk on that  
5 issue.

6 MR. CATTON: I think if you have concerns about  
7 any of this, the thing to do is to write me --

8 MR. ZUBER: I will, but since he was --

9 MR. CATTON: I understand. Because some people  
10 are going to miss their flights unless we pick up the pace.

11 [Slide.]

12 MR. JOHNSEN: This is my last slide, and I am  
13 indicating on here that as far as the level tracking model  
14 is concerned, we are going to look at several existing cases  
15 we have, GE level swell, THTF boiloff. We are in the  
16 process of doing those now. Of course, perhaps more  
17 importantly, we are going to be testing it using the AP600  
18 related experiments that are indicated here.

19 MR. SCHROCK: Gary, I would like to suggest that  
20 you add to that a test against the analytical solution from  
21 simple drift flux modelling for the void propagation. In  
22 the bubbling, your simple thought experiment problem that  
23 you started with, bubbling the air through solved that as a  
24 transient problem, and compared it with predictions from the  
25 analytical solution that you will find in textbook

1 references.

2 MR. JOHNSEN: That is a good idea.

3 MR. CATTON: I think Graham Wallis has that in his  
4 book, doesn't he, he even gives it as a homework problem,  
5 where is the level in the beer glass.

6 MR. ZUBER: They should also compare it when you  
7 have a aberration, a transient, because they should then  
8 tell you how much vapor it is generating under flashing.

9 MR. JOHNSEN: This was just a simple steady-state.

10 MR. WULFF: It could be a homework problem with an  
11 analytical solution with flashing.

12 MR. CATTON: The next speaker is Mr. Kelly on film  
13 condensation, and we are going to have to pick up the pace,  
14 so detailed questions maybe can be communicated to me and  
15 then I will forward them to RES. We are just now at 4:45  
16 p.m. yesterday.

17 [Slide.]

18 MR. KELLY: What you have is Part 1 of the  
19 handouts, there are actually three parts. I am Joe Kelly  
20 from the NRC, and I will be talking about film condensation  
21 modelling.

22 [Slide.]

23 MR. KELLY: First, what is the objective of this  
24 work? It is to develop a modelling approach for film  
25 condensation within the framework of a two-fluid model, and

1 it has a focus specifically to address concerns of the NRC  
2 consultants and also the ACRS with respect to these three  
3 issues which I will define a little bit better in a minute.  
4 Heat flux apportioning, that is how the heat flux was split  
5 between the liquid and the vapor, surface partitioning which  
6 was used in RELAP/MOD3, and that was determining how much of  
7 the surface was wet and dry, and also the effect of  
8 noncondensable gases upon condensation.

9 [Slide.]

10 MR. KELLY: Now I will give a little background on  
11 how the two-fluid model treats condensation. I give the  
12 concept of the model I am working on, and I do want to  
13 stress that this is a model that is under development so the  
14 work is not finished. Then I will go into some of the  
15 details of that. The first is film thickness, it is very  
16 important to correctly calculate the film thickness for  
17 condensation heat transfer. Then I will talk about the film  
18 side heat transfer resistances, then the effect of  
19 noncondensable gases, which would be the resistance on the  
20 vapor side, and an assessment plan.

21 [Slide.]

22 MR. KELLY: The first thing you have to do is ask  
23 yourself where is condensation potentially important in  
24 these new plans. So in the AP600, we have talked  
25 extensively about the CMT walls during the drain period, and

1 also possibly the PRHR tubes during two-phases -- two  
2 different transients, during the ATWS and during the small  
3 break LOCA during first-stage ADS actuation. Condensation  
4 could be important there.

5 The steam generator tubes, reflux cooling, this is  
6 something that has been important in small break LOCAs in  
7 traditional plants, probably is not important here because  
8 of ADS, but it is something that I will have to look at  
9 later.

10 Horizontal pipes, the condensation exists but how  
11 important it is, I don't know yet. Containment response is  
12 very important for the AP600, and that is being modelled  
13 with the CONTAIN code, so we are not handling that within  
14 the context of RELAP5.

15 For the SBWR, obviously condensation is important  
16 in the isolation condensers and also the passive containment  
17 cooling system, that is what they are designed for. Also in  
18 the containment drywell, and of course we are using CONTAIN  
19 for that.

20 [Slide.]

21 MR. KELLY: Now that we have an idea of what  
22 components condensation might be important in, what is the  
23 regime of interest for those components. I did kind of a  
24 little miniature PIRT myself to give me an idea of what  
25 regimes I have to consider. I have a list of the

1 components, the pressure in bar over the time period of  
2 which condensation may be important, vapor Reynolds number,  
3 film Reynolds number, and the noncondensable gas fraction.

4 The numbers for the SPWR I am fairly confident  
5 about because those come from the GE slide presented to the  
6 ACRS in an open meeting. These numbers --

7 MR. CATTON: But with the GE, we also felt that  
8 the bounds they put on the nitrogen concentration were too  
9 low.

10 MR. KELLY: Well, this would be in audit.

11 MR. CATTON: It should probably go all the way up  
12 to 80 or 90 percent, even one should be able to calculate  
13 starting from pure nitrogen in the tubes.

14 MR. KELLY: I agree. For the AP600, these numbers  
15 I calculated very crude hand calculations, and so note they  
16 are extremely approximate. But I wanted some idea of  
17 whether or not the film would be entirely laminar or whether  
18 some of it might be turbulent.

19 MR. SCHROCK: Excuse me, Joe. Are you using four  
20 gamma over nu or gamma over nu?

21 MR. KELLY: Four gamma.

22 MR. SCHROCK: Four gamma?

23 MR. KELLY: Yes.

24 MR. SCHROCK: Thank you.

25 MR. KELLY: That will be on one of my slides, but



1 that is always a source of confusion.

2 [Slide.]

3 MR. KELLY: That finished the introduction, so now  
4 on the background part, it is broken up into what does this  
5 physically look like, what does it look like in a two-fluid  
6 code, and in the three areas, the heat flux apportioning,  
7 surface partitioning and the noncondensable gas effects.

8 [Slide.]

9 MR. KELLY: There are two pictures here, the top  
10 one is for a pure saturated vapor, the bottom one is the  
11 vapor with noncondensable gases, and these are reprinted  
12 from Butterworth & Hewitt without permission.

13 So in the saturated vapor case, all the vapor is  
14 at saturation. There is a small temperature drop at the  
15 liquid vapor interface, and the primary resistance heat  
16 transfer is across the liquid film. In the case of  
17 noncondensable gases, there is a temperature drop across the  
18 vapor gas mixture that can be comparable or even larger than  
19 the temperature drop across the liquid film.

20 [Slide.]

21 MR. KELLY: Now we will talk about the two-fluid  
22 model and how it treats heat transfer especially with  
23 respect to condensation. The wall heat transfer rate shown  
24 here can have up to three components. As you know, it  
25 normally only has one of these, but this would be the heat

1 transfer from the wall to the liquid, wall to vapor, or  
2 something that I call the wall to interface heat transfer.

3 MR. ZUBER: Can you relate those to the pictures  
4 you had on the preceding slide?

5 MR. KELLY: Yes, and I will do that on the next  
6 slide, I will give you an example of what it looks like for  
7 film condensation.

8 MR. ZUBER: I just want to see where are these  
9 fluxes on the curve.

10 MR. KELLY: On this one, this is film condensation  
11 in a saturated steam environment. The only one of these  
12 parameters that is not zero is wall to liquid. The wall is  
13 not in contact with the vapor. The  $Q$  wall vapor should be  
14 zero, likewise  $Q$  wall interface.

15 MR. SCHROCK: What does  $Q$  wall interface mean?

16 MR. KELLY: I knew I was going to get what  
17 question.

18 MR. ZUBER: On that graph, can you show it again?

19 MR. KELLY: Yes.

20 MR. CATTON: Maybe just leave it up there.

21 MR. KELLY: I was going to get another back-up  
22 slide. I can only talk out of two sides of my mouth.

23 [Slide.]

24 MR. KELLY: The reason that QWI from wall to  
25 interface exists is because in the two-fluid model we only

1 have -- this is an example of subcooled boiling which has  
2 nothing to do with condensation, but it explains why this  
3 term exists. The two-fluid model only has one liquid  
4 temperature per computational cell, and our computational  
5 cell volumes are large with respect to things like boundary  
6 layers. In the case of subcooled nuclear boiling, in the  
7 near wall region, there is a superheated layer and the  
8 nucleation occurs at the heater surface as, of course, you  
9 know.

10           The bulk liquid is subcooled. The bubbles in this  
11 region are condensing. So our bulk liquid temperature is  
12 subcooled, T liquid. So any heat from the wall to the  
13 liquid goes into sensible heat and not generating vapor.  
14 The only way we can generate vapor is if we force it by  
15 having a heat flux from the wall to the interface, and that  
16 is this term, and that goes directly into a mass transfer  
17 term.

18           MR. SCHROCK: But the code contains the  
19 interfacial heat and mass transfer, the physics are that the  
20 heat is transferred from the wall to the liquid, and then  
21 from the liquid to the interface driving heat and mass  
22 transfer at the interface.

23           So if the interfacial area treatment is correct  
24 and the interfacial heat transfer coefficient is correct,  
25 then you ought to get the heat flux at the interface

1 correctly, the evaporation condensation effects averaged  
2 over the cells admittedly include averaging over zones in  
3 which different things are happening at different parts of  
4 the interface.

5 MR. KELLY: To do what you are suggesting, I would  
6 have to take one computational cell which may be out to  
7 here, break it into a layer next to the wall where the  
8 liquid is superheated and a layer outside the wall where the  
9 liquid is subcooled.

10 So what in effect I am doing is, I am using an  
11 empirical model to tell me what fraction of the heat flux  
12 from the wall is generating vapor in that near wall region.  
13 So I am not modelling this region exclusively, but I am  
14 forcing a vapor generation rate at the wall.

15 Then in the subcooled bulk, that vapor can  
16 condense, as you say, due to the interfacial heat transfer.

17 MR. SCHROCK: But the vapor generation is not at  
18 the wall, the vapor generation is at the interface and  
19 interface is distributed throughout the superheated boundary  
20 layer.

21 MR. WULFF: It is not at the wall.

22 MR. SCHROCK: It is not at the wall.

23 MR. KELLY: That is true, but we are talking in a  
24 very microcosm which we can't resolve within the context of  
25 the computational framework.

1 MR. DHIR: But the correlations also give you some  
2 kind of an average heat transfer coefficient at the wall.

3 MR. KELLY: Yes, and we have to take that heat  
4 flux and split it into what goes into sensible heat  
5 transport and what causes vapor generation.

6 MR. DHIR: How do you split it, you have to either  
7 know the details or it will be just a factor.

8 MR. KELLY: I agree. Actually at this point I am  
9 going to defer to Gary Johnsen because I am not talking  
10 about subcool and nuclear boiling, but if you want to wait  
11 until your talk on heat transfers, it is your choice.

12 MR. SCHROCK: The reason I questioned it is that  
13 this is something that has been questioned now for many  
14 years by consultants to the ACRS, there has been an  
15 insistence from Idaho that there is a physical basis on  
16 which this misconception makes some sense, but we never see  
17 a rational explanation of that physical basis.

18 Now you want to perpetuate the utilization of this  
19 fuzzy concept, but you want to defer to the originators of  
20 it to support its significance in the evaluation of these  
21 transient two-phased flows.

22 MR. KELLY: I will defend it on that standpoint,  
23 but what I don't know is the details of how they split the  
24 heat flux in RELAP5.

25 MR. JOHNSEN: For subcooled boiling, we use

1 Lahey's model for determining the point of net vapor  
2 generation.

3 MR. ZUBER: What did you use for bubbly?

4 MR. JOHNSEN: You mean for saturated?

5 MR. ZUBER: Yes.

6 MR. JOHNSEN: Then it all goes into vapor  
7 generation.

8 MR. KELLY: Or does it all go into the liquid,  
9 Gary?

10 MR. ZUBER: No, you just have liquid. You don't  
11 nucleate, you just have bubbly flow.

12 MR. JOHNSEN: Just saturated liquid boiling flow?

13 MR. ZUBER: Bubbly flow.

14 MR. JOHNSEN: Without heat transfer or with heat  
15 transfer?

16 MR. ZUBER: It can be heat transfer, but you are  
17 not boiling.

18 MR. JOHNSEN: I guess I don't understand.

19 MR. DHIR: Let's say he has a gas/liquid mixture  
20 flowing over the wall which is heated but no boiling on the  
21 wall.

22 MR. JOHNSEN: Okay.

23 MR. ZUBER: But you have void fraction, let's say  
24 .3-.4.

25 MR. JOHNSEN: In all likelihood, that would be

1 using Dittus Boelter to the liquid, force convection to the  
2 liquid.

3 MR. ZUBER: In all likelihood. They way I look --

4 MR. JOHNSEN: The reason I said it is that you  
5 have specified a situation without regard to mentioning flow  
6 rates, degree of superheat on the wall, or anything, so I  
7 have to give you an answer that is indefinite.

8 MR. ZUBER: The thing is, if you partition that  
9 heat flux, the way you partition the heat flux is according  
10 to the void fraction in the cell and you say 40 percent goes  
11 to the vapor and 40 percent goes to the liquid. Although  
12 you may have no vapor in contact with the wall.

13 MR. JOHNSEN: In bubbly flow it all goes to the  
14 liquid. Let me just make a point about this, I am going to  
15 talk about this later, Virgil, but there was a problem, I  
16 think, with semantics here where we are getting a little  
17 hung up.

18 MR. CATTON: If you are going to talk about it  
19 later, let's let Joe continue. I am worried about time.

20 MR. KELLY: At any rate, for condensation this  
21 term is not used, although it is in the context of  
22 COBRA/TRAC or COBRA/NC which eventually became GOTHIC, and  
23 that is because there is analog between drop-wise  
24 condensation and subcooled nuclear boiling, and it is used  
25 to start the condensation when there is no liquid present in

1 the cell. So there is no interfacial area initially, but  
2 that is not really important here.

3 For condensation -- actually, I didn't finish this  
4 slide. This is the wall heat transfer and this is what goes  
5 to the wall conduction solution. These two terms are the  
6 phasic heat transfer rates as the heat transfer coefficient  
7 and the driving the potential is wall temperature and the  
8 phasic temperature. This is what is input to the liquid  
9 energy equation, and this is what is input to the liquid  
10 vapor equation. The vapor generation due to this does have  
11 an input in both of the energy equations and, of course, in  
12 the mass equation.

13 [Slide.]

14 MR. KELLY: The condensation rate in the code is  
15 calculated as the sum of the interfacial heat transfer  
16 rates. That is liquid interface, vapor interface, over the  
17 light and heat. In RELAP5 this term does not exist for  
18 condensation.

19 The interfacial ones are calculated as a product  
20 of the heat transfer coefficient from the liquid interface,  
21 interfacial area and the driving potential is the bulk  
22 liquid temperature to the interface temperature or the bulk  
23 vapor temperature to the interface temperature.

24 MR. ZUBER: What is gamma sub-W?

25 MR. KELLY: That would be vapor generation due to



1 the heat flux from the wall to the interface, the one that I  
2 had talked about for subcool and nuclear boiling. It is not  
3 used in condensation. It was included here for  
4 completeness, that's all.

5 MR. SCHROCK: How does the code know what is  
6 happening when there is condensation in part of the  
7 computational cell and the evaporation in other parts of the  
8 computational cell, how does it identify that this is a  
9 condensation problem and not some other problem in which  
10 gamma sub-W ought to be there based on total wall heat flux  
11 or what?

12 MR. KELLY: You are talking about splitting a  
13 computational cell into two pieces and calculating different  
14 things for both halves of the computational cell, correct?

15 MR. SCHROCK: No. I am pointing out simply that  
16 there will exist in the physical world within a cross-  
17 section included in a computational cell evaporation in part  
18 of that volume and condensation in another part of the  
19 volume simultaneously in a given time period.

20 MR. KELLY: Is this the picture that you are  
21 talking about?

22 MR. SCHROCK: That is one of them.

23 MR. KELLY: In this one, the implicit solution in  
24 the code for the mass energy transfer would be condensation  
25 because the bulk liquid is subcooled.

1 MR. SCHROCK: I don't think you have gotten the  
2 gist of my question.

3 MR. KELLY: Probably not.

4 MR. SCHROCK: The gamma sub-W you have put your  
5 hand over and said, this isn't used in condensation, it is  
6 not in RELAP5 when there is condensation. What I have asked  
7 is, how does RELAP5 make a decision. I am now working on  
8 condensation instead of evaporation?

9 MR. KELLY: It depends on the temperature  
10 difference between the wall and the local saturation  
11 temperature. If a wall is subcooled with respect to that  
12 saturation temperature, it goes through and there is a logic  
13 for the heat transfer package but it depends on things like  
14 void fraction and wall temperature. So, for example, if  
15 wall temperature is above T-sat, it goes to look for a  
16 nuclear boiling type heat transfer correlations. If it is  
17 below T-sat, it looks for condensation type correlations.

18 Did you want to say something, Gary?

19 MR. JOHNSEN: I just wanted to indicate that, in  
20 fact, the code does allow you to have multiple situations  
21 within one control volume. You can have, for example, one  
22 heat structure that is cool with respect to the coolant, and  
23 another that is much warmer. So you conceivably could have  
24 boiling and condensation on two different heat structures  
25 connected to the same cell, and that would result in a

1 summation of mass transfer terms that would result in net  
2 mass transfer based on which one of those mechanisms was  
3 stronger.

4 MR. SCHROCK: So there is a logic that is based on  
5 total wall heat flux?

6 MR. JOHNSEN: No, total vapor generation or vapor  
7 generation which could be positive or negative based on  
8 summing up those terms for each heat transfer surface. So  
9 that gamma W that he is showing you is the mass transfer  
10 associated with a particular heat structure within a control  
11 volume.

12 This capability then enables you to have boiling  
13 in volume that is subcooled where there is boiling near the  
14 wall and condensing in the bulk. Depending upon the  
15 magnitudes of those terms, the condensing versus the  
16 boiling, you either have net vapor generation or you don't.

17 MR. ZUBER: I don't understand it but anyway apply  
18 it now to condensation. You would not have this term in  
19 film condensation, is that correct?

20 MR. KELLY: That's correct. This is an example of  
21 film condensation in a pure steam environment. The  
22 interface temperature and the vapor temperature at T-sat,  
23 the heat transfer terms from the vapor to the interface,  
24 wall to vapor or wall to interface are all zero.

25 The wall heat flux is simply equal to the heat

1 flux you take out of the liquid and that in turn is equal to  
2 the heat flux that you take from the saturated interface and  
3 that produces the condensation rate, just as you would  
4 expect.

5 MR. ZUBER: Okay, now can you move to the first  
6 equation for gamma? I have no problem with the first term.  
7 I think that is correct. I really don't understand the  
8 second and maybe we can talk later about it.

9 MR. KELLY: Okay, well, I'll hit it real quick.  
10 Let's go back to subcooled --

11 MR. ZUBER: Why?

12 MR. KELLY: Because that is the place where it is  
13 most important in the code. In that case the liquid  
14 temperature is less than the saturation temperature so this  
15  $Q_{li}$  is negative because we are condensing vapor in the bulk.  
16 This term we'll say is zero, so this first term is negative.  
17 We are condensing vapor but we want to generate -- this is  
18 subcooled nuclear boiling so we are generating vapor at the  
19 wall and allowing it to condense in the bulk. That is what  
20 Gary was saying.

21 This term is the vapor generation at the wall. It  
22 is the sum of these two that determines whether we have net  
23 vapor generation or net condensation but because we only  
24 have one liquid temperature in the code, the only  
25 interfacial heat transfer we can get when the liquid is

1 subcooled is condensation so this term would be condensation  
2 for that case.

3 [Slide.]

4 MR. KELLY: One of the things that was criticized  
5 by both this committee and the NRC consultants was what I  
6 call heat flux apportioning. Now this is what was done in  
7 RELAP5, Mod 3, and this was about three years ago when it  
8 was released and I am only talking about during film  
9 condensation now, okay? The wall heat flux took a heat  
10 transfer coefficient to the condensation and the driving  
11 potential was  $T_{\text{wall}} - T_{\text{sat}}$  or the wall subcooling.

12 This heat flux was taken out of the liquid phase  
13 as it should be. The other heat fluxes were set equal to  
14 zero, again as they should be, following the physical  
15 representation.

16 The condensation rate should then be that wall  
17 heat transfer divided by the latent heat, and that has to  
18 equal the liquid interfacial heat transfer and the  
19 difference between the bulk film temperature and saturation.

20 Now that is all fine, and it is as you expect but  
21 it caused a numerical problem and the reason is that the  
22 heat transfer from the wall to the liquid was using as a  
23 driving potential  $T_{\text{wall}} - T_{\text{sat}}$  and not the liquid  
24 phasic temperature, so it decoupled the heat transfer from  
25 the liquid temperature. Consequently the liquid could be

1 colder than the wall but the code would continue to remove  
2 heat from that liquid film, which obviously is not physical  
3 but it also caused a numerical problem.

4 It in effect caused the liquid to freeze and the  
5 code to bomb, have to back up and take small time steps.

6 [Slide.]

7 MR. KELLY: So this was fixed as part of the so-  
8 called "get well program" and this is how it was fixed.  
9 This is RELAP5, Mod 3.1.

10 Again calculating the wall heat transfer this same  
11 way, the condensation coefficient and the wall subcooling is  
12 the driving potential, and you do that because all of the  
13 coefficients in the literature are defined with respect to  
14 that but -- and here is what none of us were real happy  
15 about -- the wall heat flux was split in two parts, the wall  
16 to the liquid used the same heat transfer coefficient but  
17 now it took as the driving potential  $T_{\text{wall}} - T_{\text{liq}}$ ,  
18 which is good for the numerics, okay?

19 MR. DHIR: But not physically correct.

20 MR. KELLY: Right, because this number is less  
21 than this, so the extra heat transfer, because you wanted to  
22 get the right total, was taken out of the vapor phase and  
23 that does not, it is not physically correct because the  
24 vapor is not in touch with the wall, but anyway that is just  
25 to be complete about how the condensation rate was

1 calculated.

2 MR. ZUBER: Pardon me. This is the first time I  
3 really hear a good criticism of the model which you were  
4 bringing it up and I think this is a good explanation why  
5 this was introduced. The first time I hear it after five  
6 years.

7 MR. KELLY: Okay, so this solved the numerical  
8 problem and got them past that point so they could compute,  
9 but as we have just discussed, it is not physically based  
10 because the vapor is not in contact with the wall.

11 Also it causes, it induces vapor subcooling in  
12 order to get that condensation right because we are taking  
13 heat out of the vapor.

14 Secondly, the liquid temperature, the film  
15 temperature if you will, now depends on this ratio of the  
16 heat transfer coefficient due to condensation here and what  
17 the interfacial heat transfer coefficient is.

18 These were derived in separate models. They are  
19 not consistent so what the film temperature is is not really  
20 being based on a physical model.

21 [Slide.]

22 MR. KELLY: This is one of the motivations for me  
23 doing the work that I am hoping that I'll get to present.

24 MR. ZUBER: When did you start?

25 MR. KELLY: I started working at the NRC in June.

1 I started working on the condensation model about the  
2 beginning of August and I get to spend roughly 40 percent of  
3 my time doing technical work.

4 MR. CATTON: Gee, that's pretty good.

5 MR. ZUBER: It's very good.

6 MR. CATTON: I hope you can maintain that.

7 MR. SHOTKIN: We try to protect him.

8 MR. KELLY: Exactly. My management has been very  
9 good about that.

10 MR. SCHROCK: You ought to go for 80 percent.

11 [Laughter.]

12 MR. WILKINS: You can't do that at a university.

13 MR. KELLY: The surface partitioning was done in  
14 RELAP5, Mod 3 and this is where they wanted to have some  
15 idea of how much of the surface was wet or dry when there  
16 was only a very small amount of liquid in the cell, and this  
17 is what they did.

18 They took a minimum film thickness, calculated a  
19 fraction of the surface wet, took film condensation on the  
20 wet part, single phase convection to the vapor on the dry.

21 In RELAP5, Mod 3.1.1, which is the new version  
22 released I think last November, there is no surface  
23 partitioning. If the wall is subcooled, then you use  
24 condensation heat transfer and then there is a ramp to  
25 single phase liquid as the void fraction gets low.



1 MR. ZUBER: You are correct.

2 [Slide.]

3 MR. KELLY: The next item was the noncondensable  
4 gas effects. RELAP5, Mod 3.0 there was a degradation factor  
5 used. However, as was stated in Volume 4, this model was of  
6 unknown origin and also has very limited assessment and so  
7 this is the model that you reviewed, so that was recognized  
8 as a deficiency in the code.

9 They went to correct it and this is what is in  
10 RELAP5, Mod 3.1.1.

11 They put in what is now called the Berkeley-  
12 Vierow-Schrock model.

13 On the film side, it uses a Nusselt formula with  
14 an empirical flow factor to take accounts of when the film  
15 is sheered.

16 On the gas side, there is an empirical degradation  
17 factor.

18 This worked reasonably well at low pressure, for  
19 example for the PCCS of an SBWR. However, it was found to  
20 significantly over-predict condensation rates at high  
21 pressure, for example the ICS conditions, so for this what  
22 was put in was on the film side the maximum of a Nusselt  
23 formula and a Shaw correlation and on the gas side, a  
24 Colburn-Hougen type model where there was no sensible heat  
25 transfer, and the Guilliland correlation was used for the

1 mass transfer.

2 MR. DHIR: On the gas side, the degradation factor  
3 does it depend upon the composition?

4 MR. KELLY: Yes.

5 MR. DHIR: How does it relate to the compositions?

6 MR. KELLY: Pardon?

7 MR. DHIR: How does it relate to the composition  
8 of the mixture?

9 MR. KELLY: You are asking what is the form?

10 MR. DHIR: Yes.

11 MR. KELLY: I don't know.

12 MR. DHIR: There's a new version. I have not seen  
13 the write-up of RELAP.

14 MR. KELLY: Okay. On the Colburn-Hougen -- okay,  
15 now, this one you can actually ask Professor Schrock what  
16 the form is because he would know it much better than I do,  
17 and I don't have it on the top of my head.

18 MR. DHIR: Gas side --

19 MR. KELLY: But it is an empirical degradation  
20 factor based upon the bulk vapor partial pressure or bulk  
21 gas partial pressure, if you will. Okay?

22 MR. SCHROCK: Are you going to go into further  
23 detail in describing this --

24 MR. KELLY: Yes.

25 MR. SCHROCK: -- or is this it?

1 MR. KELLY: Well, that's all that I am going to  
2 give you on what is in RELAP today because that is not my  
3 mission.

4 What I am doing is trying to develop a new model  
5 that I think would be more applicable. That is what I am  
6 going to present today is the beginnings of that new model.

7 If that model is acceptable, it will be  
8 incorporated into RELAP5.

9 MR. SCHROCK: See, the reason I ask is that it  
10 isn't clear to me what is in Mod 3.1.1 from this  
11 description. I would like to know that.

12 MR. KELLY: Okay.

13 MR. SCHROCK: Am I going to learn that from a  
14 later presentation here today?

15 MR. KELLY: Probably not. What can be sent to  
16 you, they have software design implementation documents that  
17 they put out any time there is a change in the code. It's  
18 kind of a Q-A. That will describe the model and that can be  
19 sent to you.

20 MR. CATTON: That would be very helpful.

21 MR. LAUBEN: Could I make a point about this? I'm  
22 Norm Lauben.

23 Please understand that this is interim and from  
24 the schedules I showed yesterday we would like to begin  
25 3.1.2 validation in the summer and what you would see in

1 3.1.1 would be -- it's not likely to be there for very long,  
2 but I'm sure we can give it to you anyway.

3 MR. SCHROCK: Well, whether it's there long or  
4 not, I am interested in knowing what is in the code  
5 attributed to my work.

6 MR. LAUBEN: Okay.

7 MR. SCHROCK: And one of the problems that we have  
8 in looking at the relationship between what is in these  
9 computer codes and what researchers have developed in their  
10 research efforts is that there is frequently a  
11 miscommunication.

12 The code contains something which is attributed,  
13 the Chen correlation is a shining example, when John Chen  
14 has told me he's horrified by what is being represented as  
15 his correlation. I know from my own experience that my  
16 correlation on forced convection boiling was misrepresented  
17 in the RELAP codes for years. It's misrepresented in  
18 textbooks as a consequence of that misrepresentation in  
19 RELAP codes over the years so I would like to simply make  
20 the point that there is a responsibility in the professional  
21 world to reveal what you are doing with somebody else's work  
22 where you are making reference to it.

23 MR. SHOTKIN: Mr. Chairman, I just have one  
24 comment that maybe you ought to look into.

25 There could be a conflict of interest of having a

1 reviewer reviewing his own work on the subcommittee. You  
2 might want to consider that.

3 MR. CATTON: I will, but I think letting him take  
4 a look at what you did is a reasonable thing also.

5 MR. KELLY: I agree that when correlations are put  
6 in the code they need to be put into the code the way they  
7 were developed and used that way and that they should be  
8 well-documented to show that.

9 I cannot do that with respect to this because I  
10 did not do this work and so I am not familiar with the  
11 details.

12 What I can present to you is the work that I am  
13 doing now and I will give you details on that.

14 MR. CATTON: Okay.

15 MR. WILKINS: Lou, with respect to the remark you  
16 made, I don't see a conflict here. Virgil is not being  
17 asked to review his work. He's being asked to look at work  
18 other people did which they relied on some research work  
19 that he's published in the literature, and I think we're  
20 privileged to have him look at that because he knows more  
21 about that than anybody else.

22 I don't think there is a real conflict here, at  
23 least not on the basis of what has been said here this  
24 morning.

25 [Slide.]

1           MR. KELLY: The model that I am starting to  
2 develop has these objectives. I want to be able to  
3 correctly or at least as well as possible calculate the  
4 condensation rate but also the film subcooling. I also want  
5 this model to have a very large data base so it is  
6 applicable with as wide a range as possible.

7           In particular, it has to work for falling films  
8 and sheered films, and also laminar and turbulent films.

9           The constraints on this model -- it has to be  
10 physically based. It has to be compatible with the two-  
11 fluid framework, and I cannot introduce any new numerical  
12 problems.

13           The difficulties in doing this are, first, that  
14 condensation heat transfer coefficients are based on the  
15 saturation temperature as a subtemperature. I need to  
16 specify a wall heat transfer rate that uses the phasic  
17 liquid temperature as the sink temperature and then an  
18 interfacial heat transfer rate between that and the  
19 saturated interface.

20           The next part is how do you specify that liquid to  
21 interface heat transfer coefficient, because it is extremely  
22 difficult to measure.

23           MR. ZUBER: Let me ask you, maybe you could  
24 simplify it. The way you try to approach it you want to  
25 really kill all the condensation problems in one stroke,

1 subcooled boiling and condensation on walls, film, flow.  
2 The question is really what are the conditions where  
3 subcooled boiling is important compared for example to  
4 condensation with films, because if you have condensation in  
5 bubbles collapsing it's not simple boiling, so I think you  
6 could probably simplify it by just looking at films.

7 MR. KELLY: That is all I am doing, yes. I am not  
8 handling direct contact condensation here.

9 MR. ZUBER: Okay.

10 MR. KELLY: The subcooled nuclear boiling only  
11 came up to try to explain what that Qwi term was.

12 MR. ZUBER: And it only comes for subcooled  
13 boiling.

14 MR. KELLY: Right.

15 MR. ZUBER: Okay.

16 MR. KELLY: That is a completely separate issue.  
17 I am not planning on mentioning it again.

18 MR. ZUBER: Okay, good, because I am trying to  
19 understand.

20 MR. KELLY: It gets confusing. That's why I gave  
21 the background was so we could try to discuss that and make  
22 sure that we were talking about the same things.

23 The last difficulty in doing this is of course the  
24 non-condensable gas effects. We need, we believe we need a  
25 better model, and it's one in which we need to calculate the

1 interface temperature, which is a function of the partial  
2 pressure of the vapor at the interface and not the bulk  
3 partial pressure.

4 MR. DHIR: What are you trying to imply by this  
5 actual line? I don't follow.

6 MR. KELLY: Okay. Give me two slides and I'll  
7 show you where it is, okay?

8 [Slide.]

9 MR. KELLY: The idea is to calculate the wall film  
10 condensation as a series of heat transfer resistances. What  
11 I have shown here -- it is not a very good drawing -- but  
12 you see the wall, the liquid film, and the vapor region. I  
13 will identify four temperatures for you. The wall  
14 temperature. This is the bulk liquid temperature in the  
15 film, the temperature at the liquid vapor interface, and the  
16 temperature in the bulk.

17 Here I have shown the saturation at the bulk vapor  
18 partial pressure.

19 MR. ZUBER: Question. Question.

20 MR. KELLY: Yes?

21 MR. ZUBER: Why do you differentiate temperature  
22 of the interface? The interface is always in thermal-  
23 dynamically at saturation, so why do you have the I?

24 MR. KELLY: It is saturation, but it is saturation  
25 at the partial pressure of the vapor at the interface.



1 MR. ZUBER: Okay. So, you are considering --

2 MR. KELLY: Noncondensables --

3 MR. ZUBER: Okay. Go ahead.

4 MR. CATTON: You are going to use a Plesset-type  
5 heat transfer coefficient, based on the vapor pressure  
6 differences?

7 MR. KELLY: Basically, yes. Actually, I am going  
8 to follow -- that is the third part of this presentation. I  
9 am going to follow the development that was done at Berkeley  
10 by Peterson and Kajilami.

11 MR. DHIR: I don't follow. What are you trying to  
12 get? How can you get the RLI? There is no correlation  
13 which tell you RLI.

14 MR. KELLY: That is what I am going to tell you  
15 today. Okay?

16 MR. CATTON: There soon will be one?

17 MR. KELLY: I am going to infer one, yes. That is  
18 the point of this model.

19 [Slide.]

20 MR. KELLY: Okay. I am going to talk about the  
21 film side resistance now, and that has two parts: The  
22 resistance from the wall to the bulk liquid temperature, and  
23 at bulk liquid temperature to that interface -- the two  
24 resistances.

25 MR. SCHROCK: I would take exception to this

1 concept. The Nusselt analysis treats the temperature  
2 distribution in the liquid film. There is a resistance to  
3 heat transfer across the liquid film and the temperature  
4 potential driving that is the difference between the  
5 interface temperature, which is the saturation temperature  
6 for the prevailing pressure, and the wall temperature.  
7 There is only an artificial thing that is created here by  
8 dealing with the bulk temperature of the liquid film.

9 MR. KELLY: I agree.

10 MR. SCHROCK: So, what is the -- what is the  
11 physics of the problem that gives rise to a resistance  
12 between the wall temperature and the bulk temperature and  
13 then another resistance between the bulk temperature and the  
14 interface temperature? That is artificial?

15 MR. KELLY: In a way, yes. And I am splitting it  
16 this way to make it fit within the construct of the two-  
17 fluid model. Okay.

18 [Slide.]

19 MR. KELLY: Let me put this back up. So, what you  
20 are saying --

21 MR. CATTON: What you are trying to do is to match  
22 the physics against the numerical algorithms.

23 MR. KELLY: The numerics. Exactly.

24 MR. CATTON: You are going to have to divide  
25 things up a little bit differently, but you want to maintain

1 the physics; is that correct?

2 MR. KELLY: Exactly. So, what we know, if we have  
3 a condensation heat transfer coefficient, is the sum of  
4 these two resistances. We know the total resistance across  
5 the film, just as Professor Schrock said. This is the  
6 driving potential, T-Wall to the interface temperature.  
7 But, in the code numerically that doesn't work, as we  
8 discussed earlier. We need to use the phase liquid  
9 temperature for the heat flux from the wall to the liquid,  
10 otherwise, we get the numerical problem. So, my task is to  
11 find a way to approximate with some reasonability these two  
12 resistances so that the sum of them is correct and the ratio  
13 between the two is reasonable so that the film temperature  
14 that I get is reasonable.

15 MR. WULFF: Could I ask a question? The heat  
16 fluxes are the same in all of the points?

17 MR. KELLY: Yes. And, if it is steady state, yes.

18 MR. WULFF: Well, when you use Nusselt you imply  
19 that?

20 MR. KELLY: Yes, exactly. That's correct.

21 MR. WULFF: All right. Why do you need this --  
22 this breaking up if you already know the fluxes? You could  
23 match your two-fluid concept, since you know that already.  
24 I am not clear why you have to introduce this artificial  
25 resistance.

1           MR. KELLY: Okay. The way it works numerically is  
2 you specify a heat transfer coefficient in an area between  
3 the wall and the liquid phase, and then you use the liquid  
4 phase temperature as the heat sink. Okay? That is what is  
5 stable numerically. So, that is why I have to split this  
6 up. But, I only get one wall heat flux.

7           MR. WULFF: But, in Nusselt, that is one-half  
8 times the sum of the wall temperature and the interface  
9 temperature because he has the linear temperature  
10 distribution already implied.

11          MR. KELLY: Yes. And, if you give me a chance, I  
12 will go through that. Actually, it is three-eighths times  
13 the wall temperature and five-eighths times the interface  
14 temperature, because you have to take into account the  
15 velocity distribution through the film.

16          MR. WULFF: That's correct.

17          MR. KELLY: That handles it for a laminar film  
18 with a smooth interface, so no rippling, and also the film  
19 isn't turbulent. But, that is what you start from. You are  
20 exactly right. That is where you start from and that is  
21 where I will go.

22                 So, the idea is that, if I know a condensation  
23 heat transfer correlation, I know the sum of those two  
24 resistances, okay? For say a turbulent film, there are  
25 correlations for film heating and cooling where you only

1 have sensible heat transport and not latent heat. So, then  
2 the driving potential is the wall temperature to the liquid  
3 phase temperature.

4 Based on that, I can approximate that resistance  
5 from the wall to the liquid film only as this -- where this  
6 is a heat transfer coefficient from a wall to a film with  
7 sensible heat transfer only. From that, I can then infer  
8 that liquid interface -- interfacial resistance simply by  
9 taking the difference of the two. So, what what I have done  
10 is I say I know this total. I have a pretty good guess at  
11 this part. I subtract out this part, and this gives me the  
12 interfacial part. I am doing that because this is so  
13 difficult to measure.

14 MR. WULFF: And what is H sub-heat?

15 MR. KELLY: This is based upon a heating or  
16 cooling correlation. So, you just have a liquid film  
17 falling down a surface with heat transfer to the film,  
18 changing the film's temperature, but not causing a phase  
19 change.

20 MR. WULFF: Yes, but you take a correlation. Can  
21 you tell us, would that be on the flat plate?

22 MR. KELLY: Pardon? Well, if it is laminar.

23 MR. WULFF: If you show, go on.

24 MR. KELLY: Yes. I am going to go through all of  
25 the gory details.

1 MR. SCHROCK: Wait. Wait, Joey.

2 MR. KELLY: Yes?

3 MR. SCHROCK: You have got a problem with this  
4 concept. It is the following. The temperature distribution  
5 for the film cooling, in the absence of phase change, or  
6 heat transfer at the surface of the film will be --

7 MR. KELLY: They are different.

8 MR. SCHROCK: -- will be quite different from the  
9 temperature profile in the case of condensation in which the  
10 heat flux is essentially uniform through the film thickness.  
11 And the kind of problem leading to the correlation that you  
12 want to use for this heat transfer coefficient, that is not  
13 true.

14 MR. KELLY: You are correct.

15 MR. SCHROCK: Okay. So, you won't get --

16 MR. KELLY: What I am trying to do --

17 MR. SCHROCK: -- the right heat transfer  
18 coefficient for your fictitious resistance from wall to bulk  
19 --

20 MR. KELLY: You are correct, but I --

21 MR. SCHROCK: -- for the condensation problem from  
22 a film --

23 MR. KELLY: -- will get a better guess at it than  
24 what we have now. That is the point of this.

25 MR. SCHROCK: I wouldn't agree at all. You can

1 analyze the problem and get it without recourse to a  
2 correlation.

3 MR. KELLY: That is exactly what I am going to do  
4 for a laminar film is use the analysis.

5 [Slide.]

6 MR. KELLY: Going to the other side, which is  
7 supposed to be the difficult part, the vapor side  
8 resistance. In a pure steam environment it is negligible.  
9 And this just shows you what is in RELAP5. Basically, it is  
10 a very large number. It is large enough to drive the vapor  
11 towards saturation, but not so large as to cause numerical  
12 problems.

13 Now, when you have vapor with non-condensable  
14 gasses, this is different. What I am going to propose is to  
15 use the approach of a turbulent mass transfer coefficient,  
16 and then find the interface temperature.

17 MR. DHIR: What do you do if you have super-heated  
18 steam?

19 MR. KELLY: You superimpose the two -- the  
20 convective flux and you superimpose the two, so you  
21 calculate both.

22 MR. DHIR: You subtract? You are putting in heat  
23 one place and taking out heat.

24 MR. KELLY: Right. You superimpose the two, since  
25 one is positive and one is negative. You are right. You

1 actually work within the confines of the two-fluid code.

2 [Slide.]

3 MR. KELLY: Okay. Now I am going to get to start  
4 getting to some of the details of the work I am doing.  
5 Basically all I have shown you up till now is the  
6 introduction. The first thing I am going to talk about is  
7 the film thickness. The reason for this is it is very  
8 important because it determines the resistances across the  
9 film. All of the resistances scale with respect to the film  
10 thickness. I am going to break it into two parts. I am  
11 going to talk about falling films, which we might expect in  
12 something like the CMT walls, and sheared films, which you  
13 see like in the ICS condenser and in the SBWR. It is going  
14 to be further broken down to a laminar smooth, where I am  
15 going to start with Nusselt, Laminar Wavy, turbulent, et  
16 cetera. And then for a sheared film, you have to talk about  
17 interfacial shear and entrainment.

18 [Slide.]

19 MR. KELLY: As you have suggested, you can do this  
20 analytically for a smooth laminar film. This is a film  
21 thickness given by Nusselt. It is just a parabolic velocity  
22 profile integrated.  $M$  is the film thickness.  $M^*$  is the  
23 film thickness made non-dimensional with respect to the  
24 Nusselt film thickness parameter shown here.

25 The film Reynolds number is defined this way. It



1 is four times the mass flow rate per unit wetted perimeter,  
2 divided by the viscosity. I will talk about this just  
3 briefly later. This is a correlation. Actually it is the  
4 result of the calculation by Kapitsa for a laminar film with  
5 surface ripples. It is only valid over a very small range  
6 of Reynolds number.

7 MR. ZUBER: I have a question. Where do you find  
8 it?

9 MR. KELLY: In Advances in Chemical Engineering  
10 there is an article by Fulford.

11 MR. ZUBER: The interesting thing about this -- he  
12 wrote two very good papers on film flow in '48. At that  
13 time he was black-listed by Stalin, and he was not allowed  
14 to work on anything else, so he did very beautiful work on  
15 films. After 50 years or 40 years, it is still useful. It  
16 is nice that you really found it.

17 MR. DHIR: There are two points here. One is what  
18 is the transition number where you go to laminar of every  
19 film? Second, I would say that there are better  
20 correlations available in the literature now. If you look  
21 at recent textbooks, they have a correlation which includes  
22 laminar, wavy interpreted. There is one correlation which  
23 will cover all of it.

24 MR. KELLY: Can you give me a reference on that?

25 MR. DHIR: Yes.

1 MR. KELLY: Okay. Because I have not found that  
2 one.

3 MR. DHIR: Yes, there are.

4 [Slide.]

5 MR. KELLY: But, what I did was I constructed a  
6 database of all of the film thickness data I was able to  
7 find. So, these are falling films, and it is 411 points  
8 from a wide variety of sources. Most -- let's see, all of  
9 this data -- most of this data is isothermal, so most of it  
10 is just water or a different fluid running down in either  
11 the inside or outside of the tube. I plotted the non-  
12 dimensional film thickness, M-star, versus the film Reynolds  
13 number. So, there are 411 points. Here this line is a  
14 Nusselt formula. You notice there is some -- a lot of the  
15 data falls a little bit below this. This is primarily the  
16 results of Chien & Ibele. I have that as a back-up slide;  
17 but I had too much detail, so I had to take it out.

18 In the turbulent regime, I looked at I think nine  
19 different film thickness correlations. None of them matched  
20 the high Reynolds number dependents correctly. So, I did a  
21 simple curve fit over the high Reynolds number region and  
22 came up with this value. What you see here are the two.  
23 This is the predicted over measured, so it give you an idea  
24 of the error, if you just take the maximum of those two  
25 formulas. You can improve that a little, if you wish, by

1 putting in a transition region which you see here. And you  
2 will notice the transition region is very long. That is  
3 characteristic of liquid films because sublayers are large  
4 with respect to the film thickness. So, they interfere with  
5 each other, and it is a much longer transition region.

6 At the moment I am not proposing adding this extra  
7 complexity because you don't gain very much in accuracy.  
8 The data is just too widely spread.

9 [Slide.]

10 MR. KELLY: So, for falling films, what I am using  
11 at present is the following. This is the Nusselt for  
12 laminar. For turbulent, the curve fit that I showed, taking  
13 the maximum of the two. For a falling film, you can then  
14 take these non-dimensional film thicknesses and turn them  
15 into wall friction factors. That is what is shown here.  
16 Yes?

17 MR. DHIR: Where is your lens scale coming in and  
18 how do you relate this to the heat transfer coefficient?

19 MR. KELLY: Okay. The link scale and the Reynolds  
20 number?

21 MR. DHIR: Yes?

22 MR. KELLY: Okay. As I showed earlier, the  
23 Reynolds number is the standard one for films, so it is four  
24 times the flow rate, divided by the weight of perimeter.

25 MR. DHIR: I don't know that one.

1 MR. KELLY: Okay. When we talk about heat  
2 transfer I will say something about that. But, what I  
3 propose doing is using the local values calculated by the  
4 code. So, the local liquid velocity, the local liquid mass  
5 flux, calculated from the previous time step.

6 MR. DHIR: But, you are adding mass as  
7 condensation is going on. How do you know how much you are  
8 adding? Because that is driven by the temperature  
9 difference and how far you are from the leading edge of your  
10 condenser.

11 MR. KELLY: Well, I will let RELAP5 do the spacial  
12 and temporal integration. So, what you have done -- say you  
13 have a wall, okay, in a steam environment, and the wall is  
14 subcooled with respect to the steam. It will begin to start  
15 a film initially uniformly and then the film will start  
16 falling and get thicker as it goes down, okay? If the  
17 noding degradation is fine enough and you correctly specify  
18 the heat transfer resistance cross the film, then the code  
19 will do the spacial and temporal integration.

20 MR. DHIR: Heat transfer itself depends on  $M \dot{}$ .  
21 It is an iterative process. You cannot do it. You have got  
22 to go back and recheck it. Okay?

23 MR. KELLY: Well, it depends on  $M \dot{}$  because it  
24 affects the film thickness.

25 MR. DHIR: Yes. It affects the heat transfer.

1 MR. KELLY: Yes. Now, remember, we are talking  
2 about a computer code that is at the current time Currant  
3 time step-limited. So, the film thickness will change over  
4 time.

5 MR. DHIR: No, no, no. Even steady-state I am  
6 saying. Let's say steady state. Forget about the transient  
7 of how the film develops. Steady state you cannot calculate  
8 heat transfer coefficient at a given location just from this  
9 expression you have. You have to tie it to the old energy  
10 balance -- that how much energy I have taken out that gives  
11 me a condensate layer thickness. That, in turn, tells me  
12 what the heat transfer is at that location.

13 MR. KELLY: Okay.

14 MR. DHIR: The history affects it.

15 MR. KELLY: Yes. I agree. I am saying the  
16 history effects will be taken into account by the temporal  
17 integration of the partial differential equations in the  
18 code. So, will you give me that if at one point in time --  
19 if at one instant in time, at one point in space I know the  
20 film thickness.

21 MR. DHIR: You have to do it. Okay.

22 MR. WULFF: No. He takes the last time he lags -  
23 - has some damping from -- that is the way I understand it.

24 MR. CATTON: So, essentially he is using the heat  
25 transfer coefficient derived from the previous step.

1 MR. KELLY: Exactly.

2 MR. CATTON: So, you are just slightly behind in  
3 the heat transfer coefficient?

4 MR. KELLY: Exactly. All of the heat transfer  
5 coefficients --

6 MR. CATTON: That is such an improvement over what  
7 is going on now. I think that we ought to cheer.

8 SCHROCK: What about the interfacial shear  
9 effect? Your database here is simple falling film.

10 MR. KELLY: I will get to that.

11 MR. SCHROCK: Oh, you are coming to it?

12 MR. KELLY: Oh, yes.

13 MR. SCHROCK: Thank you.

14 [Slide.]

15 MR. KELLY: In fact, sheared films is the next  
16 topic. In RELAP by solving the partial differential  
17 equations, the momentum and mass equations, we calculate a  
18 void fraction or, in effect, the liquid fraction, and hence,  
19 the film thickness as a function of space and time. So, we  
20 get that directly from the solution of the field equations  
21 with a big proviso -- you have to have the correct  
22 constitutive relations for that regime. In particular, the  
23 wall friction factor, which I have just showed, interfacial  
24 friction, entrainment, if the vapor velocity is high enough,  
25 and then the momentum transfer due to phase change. Now,

1 this is normally quite small. It is only important for very  
2 low velocity vapor flows where the interfacial friction is  
3 not very high.

4 [Slide.]

5 MR. KELLY: Now, it is always best to start  
6 simple, where you know the answer. That is what I am going  
7 to do. I am going to go to a laminar smooth film and see  
8 what the effect the shear stress has. The linear shear  
9 stress distribution within the film, standard annular flow-  
10 type model, and you go through this.

11 Down at the bottom you see the film Reynolds  
12 number is expressed as a function of the film thickness and  
13 its quantity in brackets. This is the interfacial shear  
14 stress. This is two-thirds times the gravity -- the shear  
15 stress due to gravity on the film. Okay? I am going to  
16 introduce this as a characteristic shear stress on the next  
17 slide, and I am not the first person to do this. It was  
18 done by Professor Hanratty at the University of Illinois.  
19 As far as I know, that was the very first use of it.

20 [Slide.]

21 MR. KELLY: So, now I have taken -- from the  
22 equation I showed on the last slide, which I will put up  
23 here, I turned it around and I now have the film thickness,  
24 non-dimensionalized as a function of the film Reynolds  
25 number. So, all I did was take this and solve for M, and

1 you get the expression over there. So, this is following  
2 Professor Hanratty's development. So, now I have a  
3 characteristic shear stress which is defined here. The  
4 point of this is this is only a function of the film  
5 Reynolds number. So, for a laminar smooth film, M-plus is  
6 the same, whether the film is sheared or free-falling -- or,  
7 excuse me, falling on the wall.

8 MR. SCHROCK: Wait, wait. You have got tell I in  
9 there -- it is a function of tell I.

10 MR. KELLY: Yes. You are exactly right. If --

11 MR. SCHROCK: I thought you just said it is only a  
12 function of the film Reynolds number.

13 MR. KELLY: M plus. M plus is only a function of  
14 film Reynolds number. The film thickness will now depend on  
15 that. But, this right-hand side is only the film Reynolds  
16 number. So, if I say the film Reynolds number is 500, then  
17 M plus non-dimensionalized using this characteristic shear  
18 stress is the same whether it is a sheared film or a falling  
19 film. Okay? So, for a laminar smooth film you can do it  
20 analytically.

21 Now, for a falling film, there is a direct  
22 relationship between this M plus and the M star that we used  
23 earlier. This is simply it. So, if I take the relation I  
24 had for M plus -- excuse me, for M star, I can get an M plus  
25 for a turbulent film. And the question is -- we showed M



1 plus as a unique function of the film Reynolds number for a  
2 laminar film -- is it true for a turbulent film?

3 MR. SCHROCK: Could you back up just to the top of  
4 that and tell me once again how you established that first  
5 equation as M plus as -- the definition of M-plus is in the  
6 second line. The equation comes from what?

7 MR. KELLY: Okay. This equation came from  
8 integrating the shear stress distribution across the film,  
9 getting a local velocity distribution, integrating that to  
10 get the average velocity, et cetera.

11 MR. SCHROCK: I think you just answered the  
12 question. You said that it is only a function of Film  
13 Reynolds Number. But, your definition of Film Reynolds  
14 Number includes dependence on  $\mu$ . So,  $\mu$  is in that  
15 first line equation on the next page.

16 MR. KELLY: Now, my definition of film Reynolds  
17 number is the four gamma over  $\mu$ .

18 MR. SCHROCK: Well, what is the line that you just  
19 showed at the top?

20 MR. KELLY: This is the result of the equations  
21 coming down to this point. So, what you could write this as  
22 four gamma over  $\mu$  is equal to this.

23 MR. SCHROCK: Okay. Then my question is what is  
24 the basis of this equation M plus is equal to point seven  
25 Reynolds film two the one-half power?

1 MR. KELLY: From this equation, solve for M, and  
2 use this definition of M plus.

3 MR. SCHROCK: That is not my question.

4 MR. KELLY: Okay. Sorry.

5 MR. SCHROCK: The line right above that says M  
6 plus equals point seven film Reynolds to the one-half power.  
7 What is the basis of that equation?

8 MR. KELLY: I derived that from this, using this  
9 definition of M plus. It is just simple algebra.

10 MR. SCHROCK: Then what is the basis of the  
11 assertion that M plus depends only on film Reynolds number  
12 where film Reynolds number is defined simply as four gamma  
13 over MU? You see, that says there is no interfacial shear  
14 dependence.

15 MR. KELLY: On M plus. So that -- so the -- if  
16 you will, the influence of the interfacial friction is  
17 within M plus.

18 MR. SEALE: That is M in that equation.

19 MR. KELLY: Okay.

20 Let me show you the results. And I can give you  
21 the references by Professor Hanratty, if you would like to  
22 see that.

23 MR. KELLY: This is data by Andreussi and Zanelli,  
24 so it's co-current downflow which is exactly the situation  
25 we're looking at. And it gets confusing with the two

1 different nondimensional definitions of film thickness but  
2 I'll try to keep them straight.

3 This is the M plus, nondimensionalized with  
4 respect to the Nusselt film thickness parameter. This is  
5 the M plus which is the one nondimensionalized with the  
6 characteristic shear stress which involves the interfacial  
7 friction.

8 What you see is the film thickness versus Gas  
9 Reynolds Number with a parameterization on Film Reynolds  
10 Number.

11 Now basically these are different liquid flow  
12 rates. You see a range of Film Reynolds Number because out  
13 in this part entrainment begins to occur so the Film  
14 Reynolds Number goes down. So the high numbers are what the  
15 Film Reynolds Number was at the start.

16 Andreussi and Zanelli measured the film thickness  
17 and the local pressure gradient. Consequently you can  
18 calculate the interfacial shear stress.

19 Doing that, if you know the interfacial shear  
20 stress and the film thickness you can calculate by  
21 characteristic shear, then you can nondimensionalize the  
22 film thickness to get M plus.

23 When you do that, the points fall pretty close to  
24 this line. This is the Nusselt formula as I showed before  
25 and this is the one that I got for a turbulent falling film.

1 And so this is only sheared film data but it falls almost on  
2 the falling film, very close to it.

3 MR. DHIR: In between you have every film, don't  
4 you?

5 MR. KELLY: Yes. All bets are off. The  
6 uncertainties are just huge, you know, but you do the best  
7 you can.

8 MR. DHIR: And let's say you assume tau is zero?

9 MR. KELLY: No. The formula -- remember, M plus  
10 is only -- it's defined as the film thickness times the  
11 characteristic shear stress, but what I've shown is that's  
12 only a function of the Film Reynolds Number.

13 So this Nusselt relation is the same one for the  
14 falling film but on the data points were reduced using the  
15 measured interfacial shear stress to nondimensionalize it.

16 MR. CATTON: You've re-scaled it?

17 MR. KELLY: Right. And what you see here is a  
18 fairly large collection of falling film. It's all the data  
19 I showed before plus sheared film from Andreussi and  
20 Zanelli and two or three other data sources.

21 So all of it on one plot is M plus versus the Film  
22 Reynolds Number and the Nusselt relation and the turbulent  
23 relation. They're not perfect but given the uncertainties,  
24 it's not bad.

25 MR. SCHROCK: It looks excellent.

1 MR. KELLY: Thanks.

2 Okay. But I haven't said yet how I'm going get  
3 the interfacial friction, and that's one of the correlations  
4 that I need.

5 So the first thing you do is you look at the ones  
6 that are out in the literature and see if they're applicable  
7 to co-current downflow in the regime we're looking at.

8 While they're not terribly good, I've compared  
9 three: the traditional Wallis model; a model by Asali and  
10 Hanratty which included some downflow data but was primarily  
11 for co-current upflow; and model by Henstock and Hanratty  
12 which was primarily co-current upflow but had a little bit  
13 of downflow data.

14 And what I plotted is the predicted interfacial  
15 friction factor versus the measure. And what you're seeing  
16 here is only the data of Andreussi and Zanelli.

17 There are three other data sources that I've been  
18 able to find for co-current downflow and I've requested them  
19 from the library. I haven't gotten them yet.

20 And what you see is Wallis pretty consistently  
21 underpredicts interfacial friction. Asali-Hanratty, the  
22 trend's pretty good except for a couple of points, but it's  
23 a little low. Henstock-Hanratty, the magnitude's pretty  
24 good but there's a lot of points that are way off the curve.

25 MR. ZUBER: Let me ask you. Do they both have an

1 interfacial shear? I mean the tau sub i in REIAP?

2 MR. KELLY: Yes. That's how you end up  
3 calculating the void factor.

4 MR. ZUBER: No, but I mean are you using the same  
5 one or if you use their tau sub i, how would you compare the  
6 data?

7 MR. KELLY: Okay. The model that's in RELAP is  
8 very close. It's based on the Wallis model, okay? And so  
9 it's very close to what you see here. And what I'm going to  
10 say is they need to make the interfacial friction package  
11 better.

12 For the case of co-current downflow they're going  
13 to need a correlation that is based on co-current downflow  
14 data, just as you should have a different model for co-  
15 current upflow and one for counterflow.

16 MR. ZUBER: Good.

17 MR. KELLY: I have not yet finished this because I  
18 don't have the data sources yet. I've requested them and am  
19 waiting on them, but this is what it's going to look like.

20 This is the interfacial shear stress, the  
21 interfacial friction factor and what I use for annular flow  
22 is the bulk vapor velocity minus 1-1/2 times the bulk liquid  
23 velocity.

24 You know it's somewhere between 1 and 2 and 1-1/2  
25 fits for laminar flow and for turbulent flow it's not.

1           This would be the surface, velocity at the  
2 surface.

3           MR. SCHROCK: I don't follow that statement. It  
4 should be the interface liquid velocity.

5           MR. KELLY: Right. I'm approximating that  
6 interface velocity by 1-1/2 times the bulk velocity.

7           MR. SCHROCK: What's the basis of that?

8           MR. KELLY: For a laminar free falling film it is  
9 1-1/2 times the bulk at the interface. For a sheared film  
10 it's a function of the shear stress and film thickness, so  
11 it varies. And of course, when you have a film that has  
12 waves on it, the waves travel at different velocities and  
13 you don't really know what it is.

14          MR. SCHROCK: So it's basically laminar film?

15          MR. KELLY: Yes.

16          MR. SCHROCK: No interfacial shear. See, the  
17 relationship between the surface and the velocity, that is,  
18 the velocity profile and the film is influenced by the  
19 interfacial shear. So that the most simple solution of  
20 Nusselt's hydrodynamic model is of course not really the  
21 best thing where there is shear present.

22          MR. KELLY: I agree completely. However, 1-1/2 is  
23 a better approximation than 1 and the uncertainties here --

24          MR. SCHROCK: but for the laminar part of the  
25 calculation which is the most important thing, these

1 condensers rarely get into turbulent film flow and so it's  
2 indeed possible to get a good determination of the  
3 interfacial velocity from the hydrodynamic model including  
4 shear in the absence of interfacial disturbances.

5 MR. KELLY: Exactly.

6 MR. SCHROCK: But that would be much better than  
7 choosing the number 1-1/2 that comes from the theoretical  
8 profile based on no interfacial shear.

9 MR. KELLY: But what I will then say is that the  
10 interfacial disturbances are always going to exist. So a  
11 more accurate analytical result for that -- and what you'll  
12 see is all the uncertainty is going to be here. What you  
13 put here almost doesn't matter.

14 MR. SCHROCK: But my point is that you don't need  
15 to do something so simplistic as 1-1/2 times the average  
16 velocity of the film which comes from the zero interfacial  
17 shear hydrodynamic model. There is a very simple  
18 hydrodynamic model with the interfacial shear dependence.  
19 Why not use that instead? It will be a much better  
20 approximation than the one that you've proposed here.

21 MR. KELLY: Well, basically there's very large  
22 uncertainty aside from that.

23 MR. ZUBER: Why don't you give it to him.

24 MR. KELLY: I will. If you go back a couple of  
25 slides you'll see a relation for velocity profile and a film



1 with interfacial shear stress. That's it. If I know the  
2 interfacial shear stress and the film thickness I know that  
3 velocity, just as Professor Schrock is saying.

4 MR. DHIR: But you could use  $V_i$  to calculate that.

5 MR. KELLY: Exactly. So I could, but it just  
6 makes this more complicated and all the uncertainty is here.  
7 Most of the models that you see in the literature for an  
8 interfacial friction either use 1 or 2 or 0. A lot of the  
9 co-current upflow ones ignore the liquid velocity, saying  
10 it's small relative to the vapor velocity.

11 Most of Professor Hanratty's work is done that  
12 way.

13 So all I'm doing is making it a little bit better  
14 than what's been used in the past.

15 For an interfacial friction factor, this  $F_{sub\ i}$ ,  
16 I'm going to base it upon an interfacial roughness and I'm  
17 going to use an explicit approximation to the Colebrook-  
18 White equation, an empirical formula to try and fit  
19 Colebrook-White in exquisite manner so you don't have to do  
20 it recursively.

21 So here you see a roughness factor. This is the  
22 interfacial roughness. From everything that I've looked at  
23 so far for co-current downflow, it appears to be a function  
24 of this nondimensional film thickness and the Vapor Reynolds  
25 Number. And this is what has to be determined.

1 I could give you something for Andreussi and  
2 Zanelli but the more I've looked at things the less I feel  
3 comfortable about only using one data set to predict  
4 anything.

5 Okay. That finishes the first part of the  
6 presentation.

7 MR. CATTON: I thought you were through with the  
8 presentation.

9 MR. KELLY: Unfortunately, not. You're going to  
10 have to sit through a little bit more unless you want to  
11 kick me off the stage.

12 MR. CATTON: No, no. What I would like to do  
13 would be to take a break and then maybe I could talk to Lou  
14 and Brian about the rest of the day and what we're going to  
15 do.

16 (Brief recess.)

17 MR. CATTON: Before you start, Jim, one second.

18 What I'm going to do is -- we're just going to  
19 march right through this stuff until we hit 3:30 and bingo,  
20 it's over. So what I'd like you to do is take a look at the  
21 Friday agenda and if you have any preferences for the five  
22 topics -- four topics, we should rearrange it right now.

23 MR. SHERON: I would just -- I've talked with  
24 Idaho and we've sort of decided that we would like to -- any  
25 presentations anyone would want to hear, we would like to

1 give the full presentation rather than there shouldn't be  
2 any -- in other words, no abbreviation or do it fast.

3 MR. CATTON: I understand and had sort of come to  
4 that conclusion with Lou already. What I'm trying to figure  
5 out is the order so that whatever drops off the end is the  
6 least interest. And I think pretty much the order that it's  
7 in looks go, doesn't it? Interfacial heat transfer,  
8 critical flow and the momentum equation and closure and then  
9 RELAP 5 field equations.

10 And when we reach a point we'll just stop. How is  
11 that?

12 Okay. Let's do it.

13 Joe, it's yours again.

14 MR. KELLY: Okay. We've finished with the film  
15 thickness and now we're going to talk about the heat  
16 transfer resistance on the film side. So this is the heat  
17 transfer resistance across the film and the one that I have  
18 within the film.

19 Again, I'm going to break it down the same way  
20 into falling films and sheared films where the sheared films  
21 here is specific to co-current downflow.

22 And again, the same type of structure, starting  
23 with what we can calculate, which is the laminar smooth  
24 film, and proceeding on. And I'll give a summary.

25 So, again, you always start with a Nusselt

1 analysis. The film thickness which we've shown before. And  
2 this is the local Nusselt number for condensation heat  
3 transfer. It's simply the inverse of that nondimensional  
4 film thickness. And it's shown here.

5           What people normally talk about when they talk  
6 about the Nusselt model is the integral form where it's been  
7 averaged over a heat transfer surface. And the only reason  
8 I'm introducing this is because a lot of -- almost all the  
9 laminar condensation data has been taken this way. And that  
10 will be denoted by these brackets.

11           I do not advocate using this in the code because  
12 of having to specify that distance. I much prefer using the  
13 local formulation using the calculate liquid Reynolds  
14 number.

15           Now, what I'm going to do -- remember, I want the  
16 resistance across the film to be correct but I need to  
17 distribute it between those two resistances, the wall to  
18 liquid and liquid to interface.

19           I'm going to start by doing it analytically with a  
20 Nusselt type analysis. So this is the heat transfer  
21 coefficient, conductivity over the film thickness where the  
22 film thickness is what became on the previous slide from  
23 Nusselt.

24           You can now calculate for this one specific case  
25 of a laminar smooth film, you can calculate the bulk fluid

1 temperature using the parabolic velocity profile and the  
2 assumption that Nusselt made of a linear temperature  
3 distribution within the film.

4 This is it. So this is the bulk liquid  
5 temperature in the film. It's  $5/8$  times the interface  
6 temperature;  $3/8$  times the wall temperature.

7 From that, given that the heat flux is constant  
8 through the film, I can then calculate the resistance from  
9 the wall to the liquid. And it's  $5/8$  for the total  
10 resistance, which comes out to a Nusselt number from the  
11 wall to the liquid of 1.8.

12 On the other side of that going from the bulk film  
13 temperature to the interface, you do the same thing and so,  
14 of course, since the two resistances have to add up to 1  
15 times the film resistance, it's  $3/8$ . This is now the  
16 interfacial heat transfer coefficient. It's  $8/3$  times this.  
17 The Nusselt number is simply  $8/3$ .

18 MR. WULFF: And when you take the local values,  
19 should there not be an integral from the beginning of the  
20 cell to the end of the cell or how do you approximate that?

21 MR. KELLY: That's a decision -- you're correct.  
22 That's a decision that we're going to have to make when we  
23 think about implementing it in RELAP 5. And you're correct  
24 if one wants to consider that because the discretization in  
25 RELAP 5 is relatively coarse.

1           Now, what RELAP 5 does calculate is the cell  
2 averaged volume fraction which you can convert that to a  
3 liquid fraction and hence a thickness.

4           MR. WULFF: If there are no dispersed droplets?

5           MR. KELLY: Correct. If there are dispersed  
6 droplets then you have to calculate what fraction of the  
7 liquid is in the entrained form.

8           MR. SCHROCK: That requires a great degree of  
9 accuracy on void fraction prediction. It's not really a  
10 very sound way to get at the film thickness and therefore  
11 the film resistance. That is, the computed void fraction in  
12 a cell done by RELAP 5. What I'm saying, it's a poor way to  
13 come to a determination of liquid film thickness. It will  
14 be very inaccurate.

15           MR. KELLY: Well, that's what I'll find out. What  
16 I'm doing is building a set of correlations of constituent  
17 relations for the specific circumstances for interfacial  
18 thickness, the wall shear stress, et cetera. If all those  
19 work correctly, then you have a good prediction of the film  
20 thickness within the constraint that the discretization is  
21 relatively course.

22           MR. SCHROCK: But I think you need to realize up  
23 front that that approach demands a very great precision on  
24 void fraction determination. And given that your  
25 determination is over a large computational cell, it's

1 doomed to be inaccurate for the determination of the film  
2 thickness.

3 MR. KELLY: Well, I agree with what you're saying  
4 and it's going to have to be something I'm going to have to  
5 demonstrate that we can do. If we cannot do it, then I'll  
6 have to modify the approach.

7 MR. CATTON: How does --

8 MR. KELLY: Well, I've shown the analytical result  
9 and the first question is how does it compare with the data?  
10 What I'm showing here is some points. This is  
11 falling film, so it's not sheared. It's on the outside of  
12 tubes. It's from a paper by Kutateladze and Gogonin. The  
13 first plot is water data. This Prandtle number should say  
14 approximately because all of this is averaged, so the panel  
15 number changes as the film temperature changes. And so this  
16 is a rough estimate of the Prandtle number.

17 So this is water data that Kutateladze compiled  
18 from a wide variety of data sources. This was from freon  
19 and these are the data sources. The solid line is the  
20 Nusselt prediction.

21 And what I'm showing is the Nusselt prediction  
22 averaged over the total heat surface transfer length,  
23 because this is how the older condensation data was taken.  
24 As I'm sure Professor Schrock can tell you, taking good  
25 quality condensation data is very difficult. So these are

1 based on difficult energy balances.

2 MR. SCHROCK: But Joe, I think the older data  
3 where the focus was on getting local values was simply not  
4 there means that that older data isn't really a very good  
5 basis on which to build a new model. You ought to be  
6 looking at newer data where local values have been  
7 determined.

8 MR. KELLY: Yes. I have not yet been able to find  
9 any for a falling film. For sheared films, yes. For a  
10 simple falling film, no. I've only been able to find fairly  
11 old data, and all of it averaged, with the exception of if I  
12 look at the evaporation data or film heating data, then I  
13 have some local data and I'll show that to you.

14 MR. SCHROCK: See, part of the problem of the  
15 average is that each of these experimenters has done  
16 different vertical heights. Not even comparing things on a  
17 common basis very well.

18 MR. KELLY: I agree completely. And the point  
19 here is all of this data lies -- or almost all of it lies  
20 about Nusselt by a fairly significant fraction.

21 This is attributed to rippling or waviness of the  
22 liquid film.

23 So the question is what kind of behavior do we  
24 expect on a rippled film as opposed to a smooth laminar  
25 film? So I'm going to call this laminar wavy. You expect



1 to see at least two types of waves. I'm going to break them  
2 just into two categories; ripple waves and disturbance  
3 waves.

4           These are roughly the film Reynolds numbers ranges  
5 over which you see those waves, but remember these are  
6 rough. Every experimenter gives different numbers. And  
7 also, those are for water.

8           For a ripple wave, and there are a fairly low  
9 amplitude periodic type waves, they reduce the film  
10 thickness because some of the ripples were able to be  
11 transported with the wave instead of the laminar substrate.

12           They also induce velocities normal to the  
13 interface. That's extremely important in mass transfer  
14 data. It's not terribly important for us.

15           There's also the possibility of recirculation  
16 under the wave crest, but in general, these waves are small  
17 enough amplitude that they don't recirculate.

18           Disturbance waves which occur at Films Reynolds  
19 Numbers approximately greater than 350, they're definitely  
20 kinematic. A very significant fraction of the liquid flow  
21 can be transported by the waves, which makes the substrate  
22 between the wave crest much thinner.

23           Also, these waves -- well, they're called  
24 disturbance waves so you can imagine they mix the film in  
25 the region of the wave crest. So it's like you have a

1 developing length for each wavelength. So these are the  
2 ways the disturbance waves can enhance the heat transfer.

3 Looking through there's a lot of data on waves in  
4 mass transfer literature, but actually what I found here had  
5 to do with countercurrent flow. There's an experiment by  
6 Professor Dukler and we are looking at -- this is a  
7 probability density distribution of film thickness and I  
8 have it in two type waves, a ripple wave and a disturbance  
9 wave.

10 The point for me here is we're going to calculate  
11 our heat transfer coefficient based upon an average film  
12 thickness. That's shown here; the inverse of the average  
13 value. But what one should actually use is the average of  
14 the inverse film thickness. And that's shown here.

15 If you ratio these two, you get a relative  
16 enhancement of 8 percent for a ripple wave and 11 percent  
17 for a disturbance wave.

18 Now these probability distributions were built up  
19 over about 2,000 waves for each one. But again, these are  
20 rough numbers. What I want to do is get some kind of idea  
21 of how large the phenomena should be that we're looking at.

22 So what do I expect for this enhance due to  
23 rippling? Well, if I take the film thickness by Kapitza and  
24 ratio it to the one from Nusselt, it's about 93 percent. So  
25 already the average film thickness is going to be a little

1 bit lower, a little bit thinner.

2 You add to that the effect of averaging the  
3 inverse versus the inverse of the average and this is what  
4 you get, approximately a 16 percent enhancement with respect  
5 to Nusselt.

6 Now, it's not any hard fast great correlation, but  
7 it gives us an idea of what to expect.

8 And oddly enough, when you look in the low  
9 Reynolds Number region where you expect this to be  
10 applicable, the enhancement is roughly 15 to 20 percent.

11 There are a number of correlations that I found in  
12 the literature that I have in a backup slide. There's one  
13 by Kutateladze and some others, but when you get to higher  
14 film Reynolds Numbers, the enhancement grows from 15 percent  
15 up to 30, maybe even to 50. But once you get up to this  
16 part, you're beginning the transition to turbulent.

17 And I compiled a list. I think -- yes. It's 745  
18 data points. And again, this is all averaged Nusselt number  
19 and I did a simple curve fit, saying that the wavy was equal  
20 to Nusselt times an enhancement factor. It's not perfect,  
21 but it represents the data reasonably well.

22 Well, I did that for condensation data that was  
23 average. I then converted that to a local correlation. It's  
24 the same form, just a slightly different coefficient which  
25 you will see later. And I'm going to compare it to

1 evaporation data. And the reason is evaporation data was  
2 easier to take and there's more recent data where they have  
3 the local values.

4 And I've got the results of three different  
5 experimenters: Chun and Seban, Faghri and Seban and Fujita  
6 and Ueda. And I'm plotting the nondimensional Nusselt  
7 number, nondimensional as with respect to Nusselt film  
8 thickness versus Films Reynolds number.

9 So this is the turbulent regime and you see that  
10 it's a function of Prandtle number. There's not much laminar  
11 data, but you see it significantly enhanced relative to  
12 Nusselt and it's pretty close to the curve fit that I've  
13 proposed.

14 MR. CATTON: That is a really strong dependence on  
15 Prandtle number.

16 MR. KELLY: You're exactly right. You've got very  
17 good eyes.

18 MR. CATTON: The fixed Reynolds number of 5 times  
19 10 to the 3 or so, right under the V of wavy laminar. And  
20 then you go from the bottom one which is at 5.7 up to 1.7.

21 MR. KELLY: It's Prandtle number to the .65 and I  
22 agree it's extremely high. And that's something we'll talk  
23 about when we get to the turbulent ones.

24 MR. CATTON: That look a little suspicious.

25 MR. KELLY: I was, too. It surprised me. And

1 we'll talk about that, too. And also, maybe Professor  
2 Schrock can comment on it.

3 MR. SCHROCK: Right. I do think you ought to have  
4 a look at the Blangetti data also. You have that, don't you?

5 MR. KELLY: I have two examples of it.

6 MR. SCHROCK: Okay.

7 MR. KELLY: I don't have the original reference  
8 yet. I've requested it from the library but the NRC Library  
9 works somewhat slowly.

10 MR. DHIR: It's also how Prandtle number is  
11 defined. These are heated surfaces.

12 MR. KELLY: These Prandtle numbers are based on -  
13 - I believe they're based upon the saturation temperature.

14 MR. SCHROCK: There is the additional problem that  
15 is very hard to deal with, I think, that the Prandtle number  
16 variation within the liquid film in the domain of  
17 application for low pressure condensation is very important.  
18 There's a strong property dependence within the liquid film  
19 dynamics that is very difficult to deal with in terms of  
20 your simple correlations that you've put together.

21 MR. KELLY: You're correct. Fortunately, the  
22 Prandtle numbers that we'll be seeing most of the time lie  
23 between the high point 9's and 2.

24 So, it's -- I mean, it is if you're trying to get  
25 something within a couple percent, but when you look at all

1 this data it's so scattered that if we're anywhere within  
2 plus or minus 30 percent, I'm going to be quite happy.

3 Okay. This was the good news. The one that  
4 wasn't quite as good is the one shown in the bottom by  
5 Fujita and Ueda. And so again, the same coordinates. This  
6 was a curve fit that I proposed but the dashed line here is  
7 a Nusselt number assuming laminar flow but heating to a  
8 film.

9 So it's like two parallel plates but you split it  
10 in between and say it's the film. And this is the  
11 asymptotic Nusselt number in that situation.

12 And this data falls in between those two. And my  
13 attempt at an explanation is that the inlet temperature of  
14 those data were subcooled, so part of this was subcooled  
15 versus evaporation.

16 MR. SCHROCK: Are these temperatures F or C?

17 MR. KELLY: Those are C. Excuse me. Sorry about  
18 that. I'm trying to use SI units wherever.

19 Okay. I'm going to switch gears slightly. Now  
20 that's all the data that I would have for evaporation and I  
21 just wanted to see if what I was proposing was reasonable.

22 Now what I'm going to switch and talk about is the  
23 heat transfer correlation to a film that's being heated as  
24 it falls down a wall.

25 I have two data sources. The classical one is by

1 Wilke, and I also have some data by Ueda and Tanaka. The  
2 Prandtle numbers are shown here. Again, they're  
3 approximate. Nusselt number versus Films Reynolds number,  
4 so here you see the turbulent effects and the effect of  
5 Prandtle number. And here you see the laminar behavior.

6 This solid line is a Nusselt number for a laminar  
7 film in heating with a constant temperature wall battery  
8 condition. It's basically the parallel plate solution  
9 looking at only one half with a line of symmetry down the  
10 center.

11 What this shows is that there's very little effect  
12 of surface rippling upon the heat transfer from the wall to  
13 the film when you're only doing sensible heat transfer.

14 So putting this together for the laminar wavy  
15 regime, that part is a simple local value of a Nusselt  
16 formula. this part is what I've proposed as a correct if  
17 you will for laminar wavy effects. So this is the heat  
18 transfer coefficient for condensation.

19 For wall to fluid heat transfer -- and remember  
20 this is what I need to specify in the code. And what I'm  
21 going to subtract from the condensation will give me the  
22 interfacial value.

23 I saw no effect of waves on the heating data.  
24 Consequently, I'm going to use the asymptotic Nusselt number  
25 for film heating. Actually, this should read 1.88.

1 MR. CATTON: What happened to the Prandtle number?

2 MR. KELLY: This is laminar.

3 MR. CATTON: Oh.

4 MR. KELLY: We're getting to turbulent.

5 MR. CATTON: The way those -- the Nusselt number  
6 relationship deviates from laminar looks like the turbulent  
7 characteristics or the initiation of the turbulent wave is a  
8 strong function of Prandtle number.

9 MR. KELLY: Yes. And these are extremely high  
10 Prandtle numbers for heat transfer data.

11 MR. CATTON: The bottom one is almost infinite.

12 MR. KELLY: Yes. It's starting to look more like  
13 mass transfer data.

14 MR. CATTON: That's right. Those Schmidt numbers  
15 are usually a few thousand.

16 MR. KELLY: So your thermal boundary number is  
17 usually very small compared to the fluid boundary layer.

18 MR. CATTON: Well, normally the way you get a high  
19 Prandtle number is on low thermal conductivity.

20 MR. KELLY: Right.

21 MR. CATTON: And low thermal conductivity, the  
22 turbulence has a much bigger impact on the changes on the  
23 Nusselt number. What I don't understand is why the  
24 initiation is so dependent on the Prandtle number anyway.  
25 It doesn't matter for RELAP 5.



1 MR. KELLY: Yes. I won't claim to understand  
2 everything.

3 MR. CATTON: If you have any thoughts on that I'd  
4 like to hear them.

5 MR. KELLY: Well, I've wondered about it some and  
6 haven't -- I mean, for me it seems strange to say that a  
7 film at a Reynolds number of 200 would be turbulent, but it  
8 departs from the laminar behavior there. Now, it may not  
9 really be turbulent.

10 MR. CATTON: But you see when you look at the  
11 Nusselt number/Reynolds number behavior, if you had roll  
12 waves, for example, which is really not turbulence, that  
13 would cause what you're seeing there because the convective  
14 contribution for a high Prandtle number fluid would be  
15 significant.

16 MR. KELLY: Okay.

17 MR. CATTON: I suspect that that's what the  
18 process is.

19 MR. KELLY: And also if you look at the Reynolds  
20 number dependence here, they're different than the wall's  
21 Prandtle number.

22 MR. CATTON: Yes. That implies maybe --

23 MR. KELLY: A transition regime.

24 Okay. So what I showed you was the condensation  
25 heat transfer coefficient and using the asymptotic Nusselt

1 number for the heating part or from the wall to the liquid.  
2 So for my interfacial I subtract a 2 and there's no point in  
3 going through the algebra.

4 Now we're going to talk about turbulent films and  
5 what is the expected behavior or at least the behavior as I  
6 expected it.

7 You expect a transition region to begin somewhere  
8 between a Reynolds number of 1,000 to 2,000. Every  
9 investigator picks a different transition point. There's a  
10 very wide spread.

11 MR. SCHROCK: But Rohsenow's proposal was that it  
12 depends on interfacial shear rather strongly. Are you doing  
13 this first without the interfacial shear and then --

14 MR. KELLY: Yes. We're only talking about falling  
15 films then we're going to transition to sheared films.

16 I've read Professor Rohsenow's paper and I assumed  
17 that was the case. But I haven't seen it in the data, I'll  
18 be frank with you. And I'll show that later. I have not  
19 seen an early transition to turbulent behavior due to  
20 interfacial shear.

21 MR. CATTON: I think you're going to have trouble  
22 unless you can get a whole batch of data at the same  
23 Prandtl number because at the very high Prandtl numbers  
24 any kind of a disturbance in the film is going to  
25 significantly enhance the heat transfer.

1 MR. KELLY: I agree with you.

2 MR. CATTON: At the low Prandtle number, the  
3 initial disturbances aren't going to show up hardly at all.  
4 So I think what you're seeing is two processes occurring.

5 MR. KELLY: Well, you're completely correct. On  
6 one of my earlier slides when I introduced the concept there  
7 was a bullet I didn't go through and that was where after I  
8 introduced the process I said waht I'd like to do is use  
9 mass transfer data as a back check on what that interfacial  
10 heat transfer coefficient should be.

11 And exactly what you said. The Schmidt numbers  
12 are very high so the penetration depth is very small. And  
13 any surface disturbance overcomes everything. And it simply  
14 doesn't apply in our case because our Prandtle numbers are  
15 more close to 1. And everything we'll be looking at,  
16 they're based between .9 and 2.

17 So the point is very well taken and I'm trying now  
18 to develop a larger data base, especially for shear films  
19 where data exist of water steam data so that it will be  
20 directly applicable.

21 So I expect to see a transition somewhere in that  
22 regime and we're talking about falling films. Also, I  
23 expect that transition region to persist over a much wider  
24 range of Reynolds number than we're used to seeing for pipe  
25 flow. And that's because what I'll loosely call sublayers

1 or buffer layers are large with respect to the film  
2 thickness.

3 So the film Reynolds number has to be get so the  
4 film can be thick enough that the buffer layer can be  
5 contained within the film before the film can act truly  
6 turbulent.

7 Also, I expect to see the primary resistance to  
8 heat transfer is going to be across what we'll call these  
9 laminar sublayers and there'll be one at the wall and one at  
10 the free surface because the free surface tends to damp out  
11 the turbulent eddies.

12 At high Reynolds number, just based on experience,  
13 you expect the Reynolds number dependence to go to .8.  
14 Simple Dittus Boelter type thing.

15 Now what about the Prandtle number dependence? On  
16 heating data it's tradition to use  $1/3$  or roughly  $1/3$ . For  
17 cooling data,  $0.4$ . But for interface mass transfer it's  
18 normally  $1/2$  and most of the mass transfer data uses Schmidt  
19 number to the  $1/2$ . So I expected to see something between  
20  $.4$  and  $.5$ .

21 MR. DHIR: What is the boundary condition on this  
22 falling film? Is this a water film, a liquid film, falling  
23 off this heated surface? Is the interface supposed to steam  
24 or what?

25 MR. KELLY: If it's condensation or evaporation,

1 it would be steam.

2 MR. DHIR: Right. You're talking about falling  
3 film which are heated?

4 MR. KELLY: Yes.

5 MR. DHIR: What is the boundary condition?

6 MR. KELLY: What is the boundary condition?

7 MR. DHIR: Yes.

8 MR. KELLY: You mean for shear?

9 MR. DHIR: No. If you shear or for massive  
10 transfer across or heat transfer across. You see  
11 condensation you have vapor that's condensing on the  
12 interface and then you have film evaporation that's taking  
13 place. You're talking about these heated films, the falling  
14 film heated case. What is that?

15 MR. KELLY: Okay. In that case, it an ambient  
16 air. It's an air on the outside. So it's rough -- it's  
17 approximately adiabatic. Pardon?

18 MR. DHIR: There will be evaporation while it's  
19 transfer.

20 MR. KELLY: No. Well, okay.

21 MR. SCHROCK: It depends on the volatility of the  
22 liquid and the experiments when it --

23 MR. KELLY: Correct. But that's the way the  
24 experiments were constructed, was to minimize transport due  
25 to latent heat.

1 [Slide.]

2 MR. KELLY: I looked at condensation heat transfer  
3 coefficient for falling films and there's a lot of them in  
4 the literature and they all contradict each other, in  
5 particular, if you look at the Reynolds number dependence or  
6 the Prandtle number dependence. Let's start with Prandtle  
7 number.

8 A number of the older correlations have no  
9 Prandtle number dependence in them, which we know is wrong.  
10 But then if you look at other ones, Colburn to one-third  
11 power, Soliman -- and this is actually for sheared films  
12 where you sat Tau i to zero. It degenerates to falling film  
13 with Prandtle number to the .65.

14 One by Blangetti. I don't know how he got four  
15 decimal places, but that was the value. And then the  
16 evaporation heat transfer data that you saw was fit by Chun  
17 and Seban to Prandtle number .65 and that greatly surprised  
18 me, but it really fits the evaporation data.

19 MR. SCHROCK: Have you looked at the range of  
20 Prandtle numbers and the database for each of these  
21 experimenters?

22 MR. KELLY: On the older experiments, they don't  
23 even quote the physical properties. It's extremely  
24 difficult to get anything out of those. The only ones that  
25 made any kind of systemic effort -- most people never even

1 mentioned the Prandtle number dependence. It just all of a  
2 sudden appears.

3           However, that wasn't the case with Chun and Seban,  
4 and if you go back to the view graph, you'll see what the  
5 Prandtle number dependence was. I don't remember off the  
6 top of my head. But also Soliman. They had a large  
7 database for sheared films and then if you believe their  
8 model it had to have -- then they plotted it versus Prandtle  
9 number and saw what dependence would have to be for it to  
10 work and it came out to .65.

11           Basically, I'm going to have to end up doing the  
12 same thing when I get to the end of all of this and I  
13 haven't gotten that far yet. But again, fortunately, our  
14 Prandtle numbers are in the range of .9 to 2, so whether  
15 it's .5 or .65 is not critical. But, of course, like you I  
16 would like to know what is it really. But as you said, I  
17 suspect it's in this range.

18           What you see over here, which I don't remember if  
19 I put that on the handout or not, this is a Nusselt number  
20 with respect to the film thickness. So it's heat transfer  
21 coefficient times film thickness over conductivity. So  
22 these are the Reynolds number dependents. Remember, I told  
23 you high Reynolds numbers are expected to be the .8 power.  
24 Well, you can get any power you want here basically. It  
25 depends --

1 MR. CATTON: Well, between .7 and what?

2 MR. KELLY: And it basically depends on where the  
3 person took their data and how careful they were about  
4 fitting it, and I'll show you that in a minute.

5 [Slide.]

6 MR. KELLY: But what I -- the next thing I did was  
7 I plotted these up as a function of Reynolds number at two  
8 different Prandtle numbers. This is a Prandtle number more  
9 where we expect to be and this is for fairly cold water --  
10 room temperature if I remember correctly. There is a wide  
11 spread in the data. Excuse me -- in the correlations.

12 [Slide.]

13 MR. KELLY: So then failing to find a great  
14 agreement between the correlations, I started looking at  
15 some of the data, and I showed you earlier some of the  
16 falling film data. This is some of that same data, but now  
17 I'm showing you the turbulent part and I'm plotting the  
18 Nusselt with respect to the film thickness.

19 So this is the data from Kutateladze for water to  
20 freon in a turbulent range, it appears, at something like  
21 Reynolds number to the two-thirds.

22 This is data from Grover's textbook for three  
23 different fluids -- water, diphenyl and diphenyl oxide.  
24 Prandtle numbers were really approximate in this data. But  
25 the water was roughly one and half, diphenyl was five, and



1 for diphenyl oxide it wasn't even quoted. But this looks  
2 more like Reynolds number to the one power.

3 So in other words, I didn't learn very much, and  
4 this data is so scattered I can't get a Prandtle number  
5 dependence out of it either.

6 [Slide.]

7 MR. KELLY: So I fall back on the evaporation  
8 data, and this is from all three different investigators --  
9 Chun and Seban, Faghri and Seban and Fujita and Uedei. I  
10 broke the data points into three different Prandtle number  
11 ranges. So the open triangles are 1.77 to 2. So that's  
12 here. This is a Prandtle number of about 2.9 and this is  
13 5.1 to 5.7.

14 Now, I then nondimensionalized it with respect to  
15 film thickness and divided by Prandtle number to .65.  
16 Actually, I tried other Prandtle number dependencies and  
17 they didn't work. I had to go to the .65 as suggested by  
18 Chun and Seban. When you do that, they pretty much all fall  
19 in one line. This is the equation of that line, where this  
20 Reynolds number dependents I picked it and then plotted it  
21 and it seems to work. This corresponds to one of the  
22 correlations by Wilke for heating.

23 [Slide.]

24 MR. KELLY: I introduced Wilke's. I'll show you  
25 what those correlations are. So this is falling films.

1 Wilke presented an ensemble of correlations to cover a wide  
2 variety of Reynolds number ranges, going from laminar to  
3 fully turbulent with two different transition regions in  
4 between.

5 The number two only appears through very high  
6 Prandtle number fluids. Well, actually, it appears at the  
7 Prandtle numbers greater than two. Otherwise, you don't see  
8 it. If you put -- and this is the way he presented the  
9 correlations. If you put them in a nondimensional form with  
10 respect to the Nusselt film thickness parameter, and using  
11 the values of film thickness that I have, they come out this  
12 way.

13 [Slide.]

14 MR. KELLY: There are other correlations for film  
15 heating. Here are some of the ones I found. So you see,  
16 the Prandtle number dependence for heating is not a great  
17 controversy. Almost all of them are one-third or .34, and  
18 one by Gimbutis has a correction due to a hot wall effect.

19 This is the Reynolds number dependence for Nusselt  
20 number using the film thickness. So again, there's the same  
21 kind of spread that you saw in laminar. Excuse me -- that  
22 you saw with the condensation data.

23 [Slide.]

24 MR. KELLY: Now, if I plot these correlations  
25 against each other, there's not as wide a spread. First,

1 I'll back up. In the laminar regime, this is the asymptotic  
2 Nusselt number for a constant wall temperature boundary  
3 condition and a constant heat flux boundary condition.

4 I've shown two different Prandtle numbers as I did  
5 for condensation. Because there the Prandtle number  
6 dependencies were about the same, there is not a big  
7 difference in these, except for one thing that I'll show  
8 you.

9 This is the transition region in the Wilke  
10 correlation for the higher Prandtle number fluids. And in  
11 this -- you'll see it both here and here -- is a short  
12 regime between a Reynolds number of 1,600 and 3,200, where  
13 the Reynolds number dependence is greater than one.

14 [Slide.]

15 MR. KELLY: I had a fairly good database for film  
16 heating. So I put all the points together,  
17 nondimensionalized them with respect to film thickness and  
18 Prandtle number to the one-third power, because this is  
19 heating data, in plotted them versus film Reynolds number.

20 In this regime I'm using the -- the correlation  
21 Number 2 is by Wilke. I think it's his Number 4. I'm sorry  
22 about the confusion in the numbers. So it's Reynolds number  
23 to the .93 and it fits the data quite well. But if I  
24 remember, it was only valid up to Reynolds number of about  
25 11,500 -- was his database.

1           So at the high Reynolds numbers I simply assumed  
2           the Reynolds number dependence would be .8 and correlated it  
3           and just did a curve fit getting this coefficient, and they  
4           intersected at Reynolds number of about 13,000.

5           I also did a curve fit at what is a transition  
6           region for high Prandtle number fluids. I assumed that the  
7           Reynolds number dependence would be .5, which you normally  
8           have for a transition region between laminar and turbulent,  
9           and got this. But, fortunately, we don't have to worry  
10          about that because our Prandtle numbers are always going to  
11          be two or less. So I only needed this because there's a lot  
12          of high Prandtle number data on this slide.

13          This shows it as a predicted over measured so you  
14          can get an idea of what the error is. This band is plus or  
15          minus 15 percent. So that's what I'm going to use for  
16          heating.

17          [Slide.]

18          MR. KELLY: So that finishes falling films. But I  
19          have not yet found what I consider to be a satisfactory  
20          correlation for condensation in falling films.

21          MR. DHIR: I have one comment.

22          MR. KELLY: Yes.

23          MR. DHIR: There's a correlation by Chen and Tien  
24          and that is published in recent textbooks on heat transfer.

25          MR. KELLY: Yes. I have the paper now.

1 MR. DHIR: Then there's a review article by I  
2 think Fujita in recent conference on condensation. It was a  
3 foundation conference. He lists all kinds of sources for  
4 condensation heat transfer data.

5 MR. KELLY: Okay. What was the name again? I  
6 missed it.

7 MR. DHIR: I think it's Fujita or Chita. I  
8 forget. One of the two.

9 MR. KELLY: Okay, because I'll check that. I did  
10 just recently get the paper that you mentioned, and I will  
11 plot that one against these.

12 MR. SCHROCK: But it deals with the heat transfer  
13 coefficient for the whole film, that is the whole  
14 resistance, not this divided resistance that you're trying  
15 to create here for --

16 MR. KELLY: Right. But I need the whole  
17 resistance to make sure I get the right answer.

18 [Slide.]

19 MR. KELLY: So now I'm going to switch and talk  
20 about sheared films, and again we're going to start over  
21 with laminar smooth and go from something that we can  
22 calculate.

23 This was a nondimensional film thickness that I  
24 introduced before,  $M$  plus. Then you can just take that as  
25 thickness as a function of interfacial friction and get a

1 Nusselt number from that that's a function of interfacial  
2 friction. You notice that if the interfacial friction is  
3 very large it becomes proportional to the interfacial shear  
4 to the one-half, which is what's been seen in flat plate  
5 data.

6 For laminar wave -- now we're going to go to  
7 laminar wavy. If I look just at an enhancement due to  
8 interfacial shear as the interfacial shear is thinning the  
9 film. So this would be relative to Nusselts because the  
10 film is thinner than the Nusselt thickness.

11 Then the question I want to pose is, I have two  
12 effects -- the film being thinned by interfacial shear and  
13 ripples on the film surface. Can I simply add the two  
14 effects or do they interact in some way?

15 I expected them to interact because as you start  
16 to shear the film you're going to change the characteristics  
17 of the waves on the surface. Tip for co-current down flow  
18 typically the amplitudes go down and the wavelengths become  
19 shorter and you'd expect that to affect the heat transfer.  
20 Don't read the conclusion yet.

21 [Slide.]

22 MR. KELLY: This is the data mentioned earlier by  
23 Blangetti and Schlunder, nondimensional Nusselt number  
24 versus film Reynolds number for two different values of a  
25 nondimensional shear stress. So this is the interfacial

1 shear stress nondimensionalized with the liquid density  
2 times gravity times the Nusselt film thickness parameter,  
3 which is only a function of properties, okay?

4 So values of 1.18 and 2.82. So these are slightly  
5 to moderately sheared films. Okay, the data is the symbols.  
6 This finely dashed line is a Nusselt formula for the local  
7 value. The dashed line is what you get if you calculate the  
8 film thickness, based on this value of the shear stress.  
9 You can calculate what the thickness would be, assuming  
10 laminar flow and the film, and assuming a linear temperature  
11 profile across that film.

12 The solid line is what I get if I take the  
13 enhancement due to waves that I introduced earlier and use  
14 this film thickness with it. So, in effect, it looks like  
15 the two effects are additive. This surprised me.

16 MR. CATTON: Why should that -- it shouldn't  
17 surprise you that they're additive. The question is, what  
18 are the characteristics as a function of the shearing? I  
19 think the additive part is clear.

20 MR. KELLY: Yes, you're correct. What I'm  
21 surprised is that I did not have to change my correlation  
22 for rippling to account for this.

23 MR. CATTON: Well, that's right. You would think  
24 there would be some affect of shear on that.

25 MR. KELLY: That's what I expected. Now, I admit

1 it. This is a very small database.

2 MR. CATTON: On the other hand, if you're talking  
3 about if these are roll waves and the axis of the roll wave  
4 is along the wall, it uncouples from the shearing.

5 MR. DHIR: But the ripples are contributing only  
6 10 to 15 percent. So it's not much. Basically, it depends  
7 on shear.

8 MR. CATTON: It depends on Prandtle number. If I  
9 take a high Prandtle number, the contribution is big.

10 MR. KELLY: This is water data, so the Prandtle  
11 number was on the order of two. I don't have that on the  
12 slide. Sorry. But what you see is the enhancement due to  
13 rippling is this much, and the enhancement due to the film  
14 thinning is only this much. But I have to superpose to get  
15 close to the right answer, and I thought that was pretty  
16 good.

17 MR. DHIR: I thought you calculated it to 16  
18 percent or something?

19 MR. KELLY: Well, it's 16 percent at the low  
20 Reynolds numbers and gets close to 50 percent just before  
21 you transition the turbulence. At 50 percent, it's  
22 considerable.

23 [Slide.]

24 MR. KELLY: So that was the conclusion, that you  
25 do superpose the two, and so what I will have is a Nusselt



1 number. It's a Nusselt number due to standard Nusselt  
2 analysis times the enhancement that I proposed to the  
3 rippling and then taking into account the actual film  
4 thickness. So that's what shown here, conductivity over  
5 film thickness times the enhancement due to rippling.

6 MR. SCHROCK: There is an aspect of the Blangetti  
7 experiment that might be responsible for these higher values  
8 and I'm not yet convinced that it is the explanation but  
9 their data are high. The nature of the experiment is that  
10 it's a short section. The film flow is introduced. It's  
11 not produced by condensation. And the change in the film  
12 thickness over the length of the experiment is small  
13 compared to the film thickness entering --

14 MR. KELLY: A lot of it's in the developing  
15 length.

16 MR. SCHROCK: So there is a development length  
17 here which is sort of a part of the experimental data and  
18 it's not been really quantitatively assessed. So it's  
19 possible that the higher values that they're showing come  
20 from an entrance effect which is a part of the experimental  
21 database.

22 MR. KELLY: Okay. Thank you. Those are the kind  
23 of things I need to know. Unfortunately, I do not yet have  
24 the original reference. I've requested it, but again, it  
25 takes a long time to get things.

1 MR. SCHROCK: It's a good experiment but it's only  
2 like a foot in height.

3 MR. KELLY: Yes. Okay, thanks.

4 [Slide.]

5 MR. KELLY: My conclusion was that I could simply  
6 superpose the two. Excuse me. Now, we're going to talk  
7 about turbulent sheared films. I'm just going to do a back  
8 of the envelope calculation to see what I expect, okay?

9 So I'm going to assume that the total heat  
10 transfer resistance is due to what we'll call laminar  
11 sublayers, one at the wall and one at the interface. For  
12 now, I'm not going to specify how thick these layers are.  
13 Nusselt number, I'm going to define with  $t$  film thickness.  
14 So, in effect, it's going to be the film thickness divided  
15 by the thickness of the sublayers, which is going to be  
16  $y(s1)$ , and it's going to be both of the sublayers combined.

17 Nondimensionalize it as a standard using a  
18 friction velocity, and for the characteristic shear stress  
19 I'm going to use the one that I used before, which is  
20 interfacial plus two-thirds times  $\rho g$  film thickness.

21 So this is the definition of the nondimensional  
22 one, and you go through and you get a Nusselt number being a  
23 constant times Reynolds number to .85. Now, this was using  
24 my correlation for film thickness, the nondimensional one.

25 But the point is that if you use that

1 characteristic shear stress, you get only one function of  
2 Reynolds number. There's nothing on this right hand side  
3 that is a function of interfacial shear. The interfacial  
4 shear affects are captured through the film thickness.

5 [Slide.]

6 MR. KELLY: Well, does that work? Okay, it's a  
7 supposition.

8 So I'm going to first look at some condensation  
9 data, and this is condensation data where the interfacial  
10 friction is known. They measured the pressure drop and the  
11 film thickness so they could estimate the interfacial  
12 friction. All of these equal signs should be approximate,  
13 and there's a fair amount of error in this.

14 So there's a wide range of interfacial friction  
15 from 10 to 300. This is at nondimensional Nusselt number  
16 versus film Reynolds number. And you just see a wide  
17 scatter basically. But, in effect, as the shear stress gets  
18 higher, the Nusselt number gets higher, and there appears to  
19 be some affect with Reynolds number. But the scatter is so  
20 bad it's hard to see.

21 If I change and use a Nusselt number with respect  
22 to film thickness, if I look in the turbulent regime, most  
23 of this data has now collapsed to one line, you know, given  
24 that the data is very scattered. The slope I have is the  
25 .93 that I showed earlier.

1 MR. DHIR: How was the interfacial shear  
2 determined?

3 MR. KELLY: They measured pressure drop and also  
4 the film thickness.

5 MR. DHIR: I see. So you used their correlation?

6 MR. KELLY: Well, I used their measured values of  
7  $\tau_i$ . I didn't have to calculate it for myself. But these  
8 values are approximate and that's part of the reason this  
9 looks so scattered because it's not a parameter that you can  
10 control. So what, basically, all of their data that was any  
11 where near ten, they said it was equal to ten and that's  
12 part of the reason this is so scattered.

13 It's nice to see this but it's not very satisfying  
14 because the data is so scattered.

15 MR. ZUBER: I have a question. Your film Reynolds  
16 number is based on film thickness --

17 MR. KELLY: Well, it's --

18 MR. ZUBER: On the bottom.

19 MR. KELLY: Yes, I understand. It's the film --  
20 okay. The definition I used does not explicitly have the  
21 film thickness in it. It's the film flow rate divided by  
22 the wetted perimeter.

23 MR. ZUBER: The same for the Nusselt number?

24 MR. KELLY: Yes.

25 MR. ZUBER: How did you determine that one? The

1 same way?

2 MR. KELLY: Actually, to go from here to here I  
3 calculate the film thickness using these values of the  
4 interfacial shear, the Reynolds number that was reported in  
5 the data, and the correlations that I have that I presented  
6 earlier.

7 MR. ZUBER: Okay.

8 MR. KELLY: So the film thicknesses were not  
9 reported, so I calculated them. Thanks for asking that. I  
10 forgot to say it.

11 MR. ZUBER: I was afraid that you were plotting N  
12 versus M.

13 MR. KELLY: Yes.

14 [Laughter.]

15 MR. KELLY: It's a good trick, but I knew you  
16 wouldn't let me get away with it.

17 So then I went back and looked at Blangetti and  
18 Schlunder again, now with the emphasis on the turbulent  
19 regime. I do the same thing. So this is Nusselt number  
20 using the film thickness. Again, I calculated the film  
21 thickness because I don't have the measured values. Plotted  
22 it versus film Reynolds number, and pretty much the two data  
23 sets overlay each other when you do that.

24 This was the correlation for laminar wavy that I  
25 introduced and this is -- the solid line here is the

1 Reynolds number to the .9 and Prandtle to the .65. The  
2 coefficient I got just by fitting the data here. It's  
3 slightly different than the coefficients you've seen on the  
4 earlier slides. So there's not a complete consistency.

5 But what there is, is a consistency in that if you  
6 use the real film thickness for a sheared film, it looks  
7 like a falling film. What I've done here is just simply use  
8 these two correlations and combine them in a way that's  
9 somewhat standard for doing a transition. It's just an  
10 interpolation, basically, between them from anywhere to  
11 square to fifth power.

12 MR. CATTON: He has exercised this to a fairly --  
13 well, this sort of approach.

14 MR. KELLY: Yes, I actually got it from Professor  
15 Hanrarty, but you know, it's the same type thing.

16 [Slide.]

17 MR. KELLY: Well, that was condensation data. I  
18 then went and looked at the heating data for sheared films  
19 and I have some by Ueda and Tanaka. So it's the same type  
20 thing where they measured the film thickness and the  
21 pressure gradient and calculated the interfacial shear  
22 stress. But I only have two values, 10 and 30, and again,  
23 they're spread in this and these should say approximate. I  
24 plot it as the nondimensional Nusselt number -- the Nusselt  
25 number star here, and what you see is as you go from a

1 falling film to a sheared film the Nusselt number increases,  
2 as you expect.

3 If you then define the Nusselt number using film  
4 thickness, it collapses to one line, and what I plotted here  
5 are the correlations of Wilke for falling films. It's  
6 correlations Number 3 and 4, and they fall very well right  
7 on top of each other. I'm actually surprised it works out  
8 this well.

9 So that allows me to get to the summary on film  
10 side heat transfer.

11 [Slide.]

12 MR. KELLY: For laminar falling film, we've seen  
13 in condensation heat transfer is significantly enhanced by  
14 waves on the surface. That the curve fit that I've proposed  
15 for that. Wall to fluid heat transfer, conversely, is  
16 relatively unaffected by these surface waves, and I'm going  
17 to use the asymptotic Nusselt number for that.

18 For turbulent falling film I don't have so good a  
19 conclusion at the moment. The condensation data that I have  
20 are simply too scattered to with any degree of confidence  
21 determine the Reynolds number and Prandtle number  
22 dependents. So I'm going to have to use something like the  
23 evaporation data or infer it from sheared film data.

24 This is what the evaporation heat transfer data in  
25 the turbulent regime was fit by.

1 [Slide.]

2 MR. KELLY: For a turbulent falling film, pretty  
3 much is standard of the correlations by Wilke. So that's  
4 the laminar. Number 2 is the one that applies for higher  
5 Prandtle numbers so we don't really need it. So we won't  
6 have to use that one.

7 This is only for this relatively small Reynolds  
8 number region. Unfortunately, it's the Reynolds number  
9 regime that's fairly important to us because for most of our  
10 applications -- for example, in the SBWR the tubes are  
11 simply not long enough to get to high film Reynolds numbers.  
12 So we're just moving into the turbulent regime and quite a  
13 bit of it is over the laminar regime. So I will be using  
14 these two.

15 But then at high Reynolds number -- and you can in  
16 something like a PRHR, if you assume you have single phase  
17 steam coming into it because the tubes are much longer, you  
18 can go to high film Reynolds numbers, and so for that case,  
19 this was the curve fit I had, assuming that the Reynolds  
20 number dependence was Point A.

21 [Slide.]

22 MR. KELLY: Now for sheared films. I made the  
23 supposition that most of the important effects were captured  
24 in the film thickness and I believe that I have sufficiently  
25 demonstrated that. I need, of course, to assess it against



1 the much wider data set before I'll even believe it, but  
2 that's what I'm assuming for the moment.

3           So what that says is that falling film heat  
4 transfer correlations are applicable to the sheared film  
5 case if you use a Nusselt number based upon film thickness.  
6 And that's what I show here -- Nusselt number and it's equal  
7 to the film thickness divided by conductivity times a  
8 function of Reynolds and Prandtl number, where this  
9 function will now be the same whether the film is sheared or  
10 falling.

11           That ends Part II.

12           MR. SCHROCK: I'd like to say that you're to be  
13 congratulated on a very fine piece of work, Joe.

14           MR. KELLY: Thank you.

15           MR. SCHROCK: I don't agree with every aspect of  
16 it as I've already indicated. I wish I could persuade you  
17 that the breaking up of the film resistance into two parts  
18 is a sort of unnecessary and not very desirable exercise,  
19 but maybe you can convince me that it is that.

20           I think that you've brought forward a tremendous  
21 amount of experimental data to make your points, but in  
22 relationship to this one point of whether there is a solid  
23 relevance of the film heating data in the absence of mass  
24 transfer, I don't think you've really thought through real  
25 enough yet the fact that those temperature profiles are very

1 unlike the temperature profiles that exist in the  
2 condensation problem.

3 In the condensation problem, there is very little  
4 deviation from linearity of temperature distribution across  
5 the film thickness and, for that reason, the determination  
6 of the bulk temperature which you're getting is very  
7 different than the determination of bulk temperature in  
8 these heated film empirical correlations.

9 So you have to look at what does bulk temperature  
10 mean in those heated film experiments. It means something  
11 quite different in your condensation application because the  
12 temperature profiles are very different.

13 MR. KELLY: Yes, I agree with what you're saying  
14 and I'm doing the best I can within the constraints that I  
15 need to use. Now, in one response to that, if I go to  
16 laminar film where the temperature distribution is linear,  
17 almost linear, the Nusselt number for condensation was  
18 obviously one, if you make it based on film thickness.

19 If I did that, if I get my resistance of a Nusselt  
20 number between the wall and the bulk liquid temperature, if  
21 you remember, it was 1.8, okay? Just simply integrating it  
22 with Nusselt but only going to the bulk temperature. The  
23 heating data is well correlated by a Nusselt number of 1.9.  
24 So at least for a laminar film it's not that big a  
25 difference.

1                   But you're exactly correct that the temperature  
2 profiles are different.

3                   MR. DHIR: And they will depend on the distance  
4 from the leading edge. The shorter the distance, the  
5 farther you will be from 1.8.

6                   MR. SCHROCK: What they're correlating is a long  
7 enough section, so it's similar to things like Dittus  
8 Boelter where the correlation represents an average over the  
9 length, including an entry region. You must specify a  
10 minimum  $l$  over  $d$  for the correlation. There has to be a  
11 minimum  $l$  over  $d$  in Wilke's correlations because the film  
12 thickness is so small compared to the distances of  
13 interests.

14                   MR. KELLY: Wilke's values were local. So they  
15 were fully developed.

16                   MR. SCHROCK: But fully developed. That's the  
17 point, fully developed profile.

18                   MR. KELLY: So that they expect him to under  
19 predict the values near an entrance. That's correct. Now,  
20 I'll go on to Part III. That's noncondensable gases.

21                   MR. SEALE: I have another question. You can  
22 answer it perhaps while you're just distributing that. You  
23 mentioned several times difficulties with getting materials  
24 from the library.

25                   MR. KELLY: Yes.

1 MR. SEALE: How long are you talking about?

2 MR. KELLY: It depends. A lot of the things I've  
3 had to go Xerox myself if it's something that the library  
4 has. Unfortunately, the NRC library is horribly  
5 understaffed. The people there are very responsive and  
6 they've been very helpful. You know, they've made a great  
7 effort to try to help me. But there's only three of them  
8 for the whole agency.

9 MR. CATTON: Is that the library that's in this  
10 building?

11 MR. KELLY: Yes. So if it's something that they  
12 have to get on interlibrary loan, it takes months.

13 MR. ZUBER: My experience was they're very, very  
14 helpful and very proficient and efficient but they're  
15 understaffed. But they're most, most helpful.

16 MR. CATTON: You have the National Bureau  
17 Standards library, though, don't you? It's probably about  
18 as close.

19 MR. KELLY: What I've decided is I have to go talk  
20 to the librarians here and see if they can direct somewhere  
21 else and I guess that's the one.

22 MR. ZUBER: Or the Library of Congress.

23 MR. KELLY: It's just difficult for me to take  
24 time out from the office to go get an article. I'm used to  
25 being able to send a requisition to the library and have it

1 come back.

2 MR. CATTON: And I guess you don't a student to  
3 send.

4 MR. KELLY: If you one --

5 [Laughter.]

6 MR. JOHNSEN: Joe, you can always call us to at  
7 the IML.

8 MR. KELLY: Okay, thanks, Gary.

9 MR. JOHNSEN: Don't hesitate. Pick up the phone  
10 and call.

11 MR. KELLY: Thank you. I will take you up on  
12 that. But you know that at the NRC we can't do E-mail  
13 outside of the agency.

14 MR. BOEHNERT: Yes, you can.

15 MR. KELLY: Well, if you have a hook-up to  
16 INTERNET, which I don't. I don't think there's one in my  
17 building, which is Nicholson Lane North.

18 MR. BOEHNERT: I'm surprised. I thought most of  
19 the agency was getting it.

20 MR. KELLY: Eventually, yes.

21 [Slide.]

22 MR. KELLY: Now, we're going to go to  
23 noncondensable gases. Here, I have not done the work that  
24 you saw on the other. All I'm going to do is lay out the  
25 approach that I'm intending to follow and, basically, I'm

1 going to follow the method proposed -- used by Peterson et  
2 alibi and Kageyama et alibi. So these are basically the  
3 models developed at Berkeley.

4 What they do is they take an energy balance at the  
5 interface. So this is the wall condensation. The heat flux  
6 through the wall condensation is split into two parts, a  
7 flux to the latent heat and a flux of sensible heat. I  
8 should go back and briefly remind you of the physical  
9 situation.

10 We're talking about the picture on the bottom now  
11 where there's a significant temperature drop between the  
12 bulk temperature and the interface temperature. In our  
13 brief audit we do not know that interface temperature and  
14 this is because there is a distribution of the mole  
15 fractions of the vapor and the noncondensable gases across  
16 the boundary layer. Okay?

17 But, of course, this is the number we want, is the  
18 wall condensation. Now, you can express that, as shown  
19 here, the heat transfer coefficient. The driving potential  
20 is the interface temperature, the saturated value at the  
21 local vapor partial pressure. Then the two heat transfer  
22 coefficients, one due to the light flux -- flux of light and  
23 heat, one due to sensible heat, and you see the driving  
24 potential. This is the bulk saturation of the vapor and the  
25 interface saturation temperature.

1 [Slide.]

2 MR. KELLY: Now, we're going to follow the  
3 development done at Berkeley. The first assumption we make  
4 is the one that everyone makes, and that's that you can  
5 treat the vapor gas mixture as a mixture of perfect gases.  
6 That allows you to take the mole concentrations and express  
7 them as partial pressures.

8 The next step that was made is using the Clausius-  
9 Clapeyron relation to take the partial pressures and express  
10 them as a temperature difference, and this is the  
11 temperature difference I want between the vapor bulk and the  
12 saturation value at the interface.

13 In doing this, we introduced a vapor liquid  
14 specific volume. That's approximated this way: sets the  
15 gas constant an average across the boundary layer and the  
16 gas, and this  $X$  is the mole fraction of vapor, the wall  
17 gradient mean value across that boundary layer.

18 Then the diffusion coefficient, the diffusivity,  
19 is pretty much proportional to  $T$  squared over  $P$ . The  
20 relationship we're going to derive, there is this value of  
21 the diffusivity times  $P$  over  $T$  squared. That's  
22 approximately a constant. So the assumption has been made  
23 you can evaluate as a constant at as a reference value and  
24 then you don't have to worry about the dependents.

25 [Slide.]

1 MR. KELLY: So then from this, from those  
2 assumptions, you can formulate a model. So what you have to  
3 do is assume an interface temperature and the mole fraction  
4 of noncondensable gas or vapor at that interface.

5 Then they have defined a condensation conductivity  
6 and it's shown here. Everything in these brackets is a  
7 constant. It can be evaluated once. This is the latent  
8 heat. This Phi is defined here. It's the ratio of the log  
9 mean values of the gas and the vapor mole fractions across  
10 that boundary layer, and it's shown here in terms of the  
11 bulk value and the value at the interface of the gas.

12  $T_{ave}$  is an average of the temperature at the  
13 interface and the temperature in the bulk. So you make  
14 these two assumptions. You can evaluate Phi and you can  
15 evaluate this condensation conductivity.

16 From that and the Sherwood number you can get a  
17 heat transfer coefficient due to condensation. This is on  
18 the vapor side, okay? The small case "d" is a diameter.  
19 From this you can calculate -- from this and continuity of  
20 heat flux through the film and through this boundary layer,  
21 you can calculate directly what that interface temperature  
22 is.

23 Then you have to check this against what you  
24 assumed and iterate until you converge. When you've done  
25 that, you can then calculate what the wall heat flux is,



1 which, of course, is what we wanted.

2 [Slide.]

3 MR. KELLY: Now, this model -- and again, it's not  
4 by model. But it has one very strong advantage. That's  
5 that this iteration parameter,  $\Phi$ , is much less sensitive  
6 than the partial pressures that you see in some like a  
7 Colburn Hougen model. Consequently, you can get a better  
8 initial guess and few iterations. I like this approach very  
9 much, which is why I selected it.

10 Also, what we might be able to do is make one  
11 further assumption, but this is tentative. It's something  
12 that I have to check out when I actually get the model up  
13 and running. And what I'm going to suppose is that when  
14 noncondensable gas effects are important then the bulk value  
15 of the gas mole concentration is not a small number. Okay,  
16 it's not infinitesimal.

17 Also, due to the noncondensable gas effect the  
18 heat flux is not terribly high. This means that the  
19 difference between the interface and the bulk concentration  
20 is not a large number and the denominator here is not very  
21 small.

22 So maybe this value isn't very large. If this is  
23 not large, then  $\Phi$  can simply be expressed as the ration of  
24 one minus the bulk gas concentration over the bulk  
25 concentration. If that's true, you don't need to iterate.

1 Now, this is totally supposition.

2 MR. SCHROCK: I don't think that's a good  
3 assumption to make.

4 MR. KELLY: You're probably right. I'll show you  
5 what the form of that is.

6 MR. SCHROCK: Based on experimental data that is.

7 MR. KELLY: I haven't closely looked at that yet.  
8 I will be. What I can do is use this as a first guess.

9 [Slide.]

10 MR. KELLY: At the very least, I can use it as a  
11 first guess if I don't have a better value.

12 What I've plotted is the actual minus the  
13 approximate value divided by the actual value. So that  
14 would be the error versus that ratio I just talked about.  
15 This is the difference between the interface value and the  
16 bulk divided by the bulk. If that number is less than about  
17 ten percent, the error is on the order of five or so percent  
18 and it's not terribly large.

19 For small values it's a very good approximation.  
20 But as the values become large, you see it falls apart, as  
21 expected. So it's just something I'm going to check to try  
22 to make the computational effort easier if possible.

23 [Slide.]

24 MR. KELLY: To use this model you need several  
25 constitutive relations. You need a convected heat transfer

1 coefficient,  $h$  of  $s$ . We also need to know the ratio of a  
2 Sherwood to the Nusselt number because I used the Sherwood  
3 number earlier.

4 For the convected heat transfer coefficient, there  
5 are at least four effects that we need to consider --  
6 entrance links affects, boundary layer suction. Because  
7 we're condensing, it looks like you're sucking the boundary  
8 layer closer to the liquid film and that enhances the heat  
9 transfer. Roughness due to interfacial waves and also the  
10 possibility of mist formation in the vapor gas mixture.

11 All four of those act to enhance the heat  
12 transfer, and that's what I talk about here. Entrance  
13 effects are possibly important for the PCCS and the ICS  
14 because roughly the  $l/d$  for the entire condensers on the  
15 order of 50. So definitely you see these entrance effects  
16 in the experimental data.

17 The boundary layer suction is probably small from  
18 the things I've seen in papers. The roughness effect is not  
19 small, and what I've plotted here -- I didn't plot it. This  
20 is reprinted from Rohsenow and Choi's textbook. It's a  
21 Nusselt number over a smooth tube Nusselt number versus the  
22 friction factor over the smooth tube friction factor. So  
23 this is error data turbulent force convection and you see  
24 enhancements between 1 and 3 over this roughness range.

25 For interfacial the friction factors go on out

1 close to 100. What I'm planning to approximate this as the  
2 heat due to surface roughness divided by the smooth is a  
3 square root of the ratio of interfacial friction to smooth  
4 tube friction. That's basically assuming that things like  
5 subway or scales is the square root of a shear stress.

6 There are some correlations for this, but again,  
7 there all in contradiction with each other, and for me all  
8 the uncertainty is in the value of  $f$  of  $i$ .

9 Mist formation -- I haven't researched this yet  
10 but in the paper by Peterson et alibi they quote the data of  
11 Mori and Hijikata where it was found that the heat transfer  
12 coefficient for sensible heat transport was approximately a  
13 factor of two greater than the convective, and in Peterson's  
14 work they used a value of seven, and that's something that  
15 I'll have to look at.

16 [Slide.]

17 MR. KELLY: The other thing I need is the ratio of  
18 the Sherwood to the Nusselt number and it's traditionally of  
19 Schmidt number to Prandtle to some exponent  $n$ . Again,  
20 what's the exponent? Dittus-Buelter for cooling, .4, mass  
21 transfer gas absorption, .5, the Gilliland and Sherwood --  
22 and this was evaporation of a liquid film into an air  
23 environment. They correlated their data with a ratio --  
24 with an exponent .44. In a paper by Kageyama et alibi, they  
25 use an exponent of .6, and that's probably what I'll start

1 with because that was based upon condensation data.

2 [Slide.]

3 MR. KELLY: My last slide -- I actually skipped  
4 the summary because I didn't want to tell you all this over  
5 again, but this is the assessment plan. Once I get the  
6 model finished and into a code and start checking it out,  
7 this is how I'm going to do it.

8 For forced convection, I found it hard to find  
9 pure steam data. There is a large database done by  
10 Goodykoontz and Dorsch. I believe this is NASA data, and  
11 that's what I've requested from the library. We don't have  
12 it yet. There are four cases that are in "Multiphase  
13 Science and Technology series. So at the moment I have four  
14 cases there.

15 I have two cases from Babcock and Wilcox, which is  
16 at high pressure, and from the MIT tests there are eight of  
17 them that are pure steam.

18 For the effect of noncondensables, I have the data  
19 from MIT from two different experimenters, Siddique and  
20 Husanein. Now, this is what I call developmental  
21 assessment. This is the data I can use in developing the  
22 model and adjusting these correlations to make sure they all  
23 work together.

24 Now, the other term --

25 MR. SCHROCK: You don't have our data you're

1 saying?

2 MR. KELLY: Well, if you remember, I did write to  
3 you and you were kind enough to reply and you told me that  
4 the data was propriety and that I would have to ask --

5 MR. SCHROCK: No. I didn't say that. I said that  
6 it had to be requested from General Electric and I assumed  
7 that you would do that and have it. Have you not done that?  
8 I mean the data have been supplied to NRC in the regulatory  
9 process so I don't understand why it wouldn't be available  
10 to you.

11 MR. KELLY: Okay. I misunderstood then. I  
12 thought the data was proprietary, in which case I could not  
13 use it for developmental assessment. You'll notice it's  
14 down here under applications.

15 MR. SCHROCK: Well, some of it is published  
16 already and at the International Multiphase Conference Mr.  
17 Cooper --

18 MR. KELLY: Yes, you did send me the papers. But  
19 I needed the raw data and I will check --

20 MR. SCHROCK: The thesis were requested. I mean  
21 you requested the thesis.

22 MR. KELLY: Correct.

23 MR. SCHROCK: They have been supplied to NRC in  
24 relationship to the certification program, and what I  
25 replied in my letter to you is that I should not be the

1 person that you solicit this from in view of its role in the  
2 certification program. That's the message I intended to  
3 convey. You have to get it from GE because I think they're  
4 quite -- they have already given it to the people in Regs so  
5 I don't understand why you don't have it.

6 MR. KELLY: Yes, I'll check and find out because I  
7 --

8 MR. CATTON: Well, the thesis isn't proprietary,  
9 is it Richard?

10 MR. SCHROCK: Of course not.

11 MR. CATTON: He can get the thesis.

12 MR. KELLY: Yes, that's what I had asked for and  
13 was told I needed to get it from GE and by that response I  
14 misunderstood and thought that it was proprietary, and if it  
15 was proprietary I cannot use it in developmental assessment  
16 but I can use it in applications assessment. What that  
17 means is that I can't adjust a model based on it but I can  
18 see how good the model is.

19 MR. SCHROCK: Well, GE puts no limit on us. We  
20 don not accept research contracts that do not permit us to  
21 publish our results. On the other hand, we do respect the  
22 sponsors right to deal in the regulatory arena in this way  
23 and so you have the data in the Agency or other data which  
24 have yet to be fully documented for GE which will shortly be  
25 available. But I don't think that you should represent this

1 as Berkeley data as unavailable for your assessment. It is  
2 certainly available.

3 MR. KELLY: Good. I'm glad to hear that because,  
4 frankly, I can use all the data I can get and I intend to do  
5 an extremely large amount of assessment, not using RELAP 5,  
6 but I have a small two-foot study state code that I use for  
7 correlation developmental. I'm having them modify it for  
8 condensation, but that's what I'll do. So I'll talk to NRR  
9 first and see if they have the data.

10 MR. CATTON: Well, they have the two -- were they  
11 masters or Ph.D.'s, Richard? Because we got them.

12 MR. KELLY: Good. That's good news. Thank you.

13 So then, this will be moved up to developmental  
14 assessment as well and I will then talk to you later about  
15 this, Professor Schrock, to see which ones are the best ones  
16 to run because the cleaner the boundary conditions, for me  
17 the better.

18 For what I'll call applications assessment, I will  
19 look at the CMT tests because I need to worry about falling  
20 films on the inside wall of the CMT. Now, the CMT tests --  
21 these are the ones done by Westinghouse. They'll be two  
22 sets. One is pure steam and one with steam and a  
23 noncondensable gas, and three types of tests. One is where  
24 the tank is empty and there is just pure wall condensation  
25 and they drain off the condensate at the bottom. So it is a



1 transient. But it's only wall condensation.

2 Then there will be the series of tests where they  
3 maintain a stable level in the CMT so they get both wall and  
4 pool condensation. But they have that at different levels,  
5 and the idea is that way you have different wall  
6 condensation amounts and you can subtract it to get an  
7 estimate of what the condensation on the pool surface is.

8 So admittedly the aspect ratio isn't the same as  
9 the plant but it's a whole lot better than no data.

10 MR. CATTON: Well, if it a stable level then it  
11 will behave almost like an air function. The condensation  
12 on the water interface will die out pretty quickly. It's  
13 almost like air function.

14 MR. KELLY: These tests will be done with RELAP 5.  
15 So it will be done on a transient mode.

16 MR. CATTON: The drain test will be the  
17 interesting one.

18 MR. KELLY: Right. And I will plan on doing all  
19 of those test with RELAP 5.

20 MR. ZUBER: What is scheduled when?

21 MR. KELLY: Well, for getting a model into the  
22 code, that's scheduled to be done by June. So the  
23 developmental assessment will be done by then. Then the  
24 applications assessment will be done after that. That's my  
25 goal.

1           Then finally, we have the PANTHERS experiment for  
2 SBWR. So it's full scale, typical, you know, prototypical  
3 conditions for both the ICS and the PCCS. So believe it or  
4 not, I'm finished.

5           MR. CATTON: Well, I would first like to  
6 compliment Joe. I think you've done a very good job and you  
7 carried this through from the beginning to the end in what  
8 is a very scrutable presentation. It's fairly easy to  
9 follow and I think you're going to be successful with it and  
10 I thank you for the presentation.

11           MR. KELLY: Thank you.

12           MR. CATTON: Novak?

13           MR. ZUBER: I'd like to make five comments.

14           MR. CATTON: Five?

15           MR. ZUBER: Five. Number one, I would really like  
16 to complement you. It was a very, very good technique of  
17 presentation. It was a good technique of presentation. It  
18 was candid and it was really pleasurable to listen to you.  
19 I'm sure this was one of the best I have heard.

20           MR. KELLY: Thank you.

21           MR. ZUBER: Two, I have a concern I have to adjoin  
22 here with Joe. What you really found is very interesting  
23 and very appealing but everything is dominated by the film  
24 thickness. I think this is a nice piece of work. I see,  
25 however, a problem because I don't think -- I'm afraid that

1 data won't be really accurate to predict alpha and that all  
2 these refinements which you went through -- and they're  
3 really good and sound -- will be automatically completely  
4 overshadowed by the inaccuracy of calculating the liquid  
5 under those conditions.

6 Three, I would really urge you to write a report  
7 as soon as possible. I think this will give credit to the  
8 work this agency is doing. Four, I was really pleased to  
9 see that the concerns which this Committee has voiced  
10 over the past three years have been addressed in this model.  
11 Finally, we are getting somewhere.

12 Five, it is really too bad that this was not done  
13 two or three years ago. We could have avoided many  
14 unpleasant discussions, and more than that I think the  
15 problem would have much ahead, the schedule would have been  
16 much better.

17 So, altogether, I think it was a pleasurable thing  
18 to sit and listen to you. Thank you.

19 MR. SEALE: One other thing. I think the  
20 management, that Joe's supervision, ought to be  
21 congratulated also for providing the opportunity for him to  
22 do this. This was a very nice piece of work.

23 MR. LAUBEN: Can I say something? I'm his  
24 immediate supervisor and I don't want to take very much  
25 credit.

1 MR. CATTON: Why not?

2 MR. LAUBEN: Because he's had to work a lot on his  
3 own at home with this stuff and that's a real credit to Joe  
4 and at the same time trying to satisfy his Ph.D. supervisor  
5 at Berkeley and do the things we've asked --

6 MR. KELLY: At MIT.

7 MR. LAUBEN: Sorry. At MIT.

8 MR. KELLY: Unfortunately, this is not my Ph.D.  
9 work. I wish it were.

10 MR. LAUBEN: Right. He's in essence doing  
11 virtually three jobs -- this job, the things we ask him to  
12 do, and completing his Ph.D. requirements. So I think  
13 that's quite an effort. I want to give him all the credit  
14 for it.

15 MR. SEALE: We urge you to keep this in mind  
16 though.

17 MR. WULFF: In view of what I said about the level  
18 tracking this morning, I also like to join Novak in what he  
19 had said. I think this was a good example of how we can  
20 find out what is being done and it was very good work. That  
21 was a good example of a presentation.

22 MR. ZUBER: May I add one more thing? I hope that  
23 this agency will take work like this as a standard. I think  
24 our work with you would be much easier to do and it would be  
25 much easier than to confront any concerned scientists or

1 whatever. I think this should be a standard for the work  
2 and it's an example it can be done. It depends on the  
3 quality of their people and the vision of the management.

4 MR. CATTON: I agree. I agree. I think what I'd  
5 like to do is to -- if people will accept my suggestion --  
6 is take just a 30 minute break for lunch, come back at 1:00  
7 and then we'll just work our way through these topics until  
8 we reach 3:30 and end it at that point. Okay? Let's do it.

9 [Whereupon, at 12:33 p.m, the meeting in the  
10 above-entitled matter was recessed, to reconvene at 1:00  
11 p.m. this same day.]

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## AFTERNOON SESSION

[1:00 p.m.]

1 MR. SHOTKIN: The next topic is critical flow.

2 MR. SHERON: Ivan, we were wondering if we could  
3 do the RELAP5 field equation since Professor Ransom is here.

4 MR. CATTON: All right.

5 MR. SHERON: Trying to get him and the right  
6 people in the same room at the same time is kind of tough.  
7 If we could start with Walt Weaver.

8 MR. CATTON: He's the speaker for the rest of the  
9 day.

10 MR. SHERON: Except for Gary, on the interfacial  
11 heat transfer.

12 MR. CATTON: That's right. Why don't we do that,  
13 do the field equations and then back to interfacial heat  
14 transfer.

15 [Slide]

16 MR. WEAVER: Good afternoon. My name is Walt  
17 Weaver, from Idaho National Engineering Laboratory. I will  
18 be making a presentation on the RELAP5 field equations. The  
19 presentation is divided into two pieces. I will make some  
20 general comments about the documentation of the field  
21 equations since that seems to be a concern of the Advisory  
22 Committee, and we believe that the totality of the  
23 documentation is rather complete but it was not all  
24  
25

1 contained in Volume 4 of the manual which was what the  
2 Committee reviewed. We are going to rectify that situation,  
3 and I will speak to that in detail.

4 Then, there was a specific comment -- that was a  
5 general comment -- there was a specific comment made on the  
6 form of the momentum equation; that it was in the non-  
7 conservative form, so I will discuss that at some length.  
8 First, regarding the documentation of the field equations.

9 [Slide]

10 MR. WEAVER: Most of the documentation that we  
11 prepared was directed toward the code user, so that the user  
12 could tell what was actually in the code. The documentation  
13 was not prepared with the idea of defending the particular  
14 form of the equations as they finally appeared.

15 Volume 4 which was originally prepared as a stand-  
16 alone document in support of the CSAU effort was intended to  
17 show the context in which the constitutive relations  
18 appeared in the equations, so that it would motivate the  
19 discussion of the source and applicability of the individual  
20 constitutive relations. That's repeated in the last bullet.

21 The section on the field equations was to show  
22 where the factors fit within the calculational scheme rather  
23 than to derive the equations from first principles.

24 [Slide]

25 MR. WEAVER: We are proposing to issue a new RELAP

1 volume based on Vic Ransom's internal of the work that he  
2 did and presented at this summer school in France in 1989.  
3 We are proposing to take that report and put it through tech  
4 editing and issue it as a NUREG report. This report starts  
5 from the continuum equations of two inter-penetrating media  
6 and does all the kinds of mathematical manipulations that  
7 you need to get down to the partial differential equations.

8 I must say that we had not issued this before  
9 because Vic was thinking of using this as a basis of a book.  
10 He has since decided not to do that because of the press of  
11 being the Chairman of the Department at Purdue, and has  
12 given us permission to reissue this report as a NUREG.

13 We will take the equations as derived in this new  
14 RELAP volume as partial differential equations and develop  
15 in more complete detail the difference equations that are  
16 actually coded and solved in the RELAP5 program, so that you  
17 can see the tie between the partial differential equations  
18 as derived in this new volume and the difference equations  
19 that are actually solved by the computer program.

20 Finally, we will take out of Volume 4, the  
21 material on the derivation of the equations which was  
22 admittedly sketchy, and reference the final equations in  
23 Volume 1 as the starting point for Volume 4 for showing  
24 where the constitutive relations fit within the framework of  
25 the calculational procedure.



1           To facilitate doing that, for converting our  
2           documentation into an electronic form so that we can keep  
3           the documentation much more up to date, we are converting it  
4           to a desktop publishing program called Framemaker which has  
5           a very good equation editor so that the engineers themselves  
6           can get the equations in the manual and get them correct.  
7           The procedure at the laboratory prior to the preparation of  
8           the manual in 1990 was very cumbersome, where we had to  
9           iterate back and forth between the word processing people.  
10          They did not have what you call a WYSIWYG way of looking at  
11          the equations. They had to print them out to see what they  
12          would really look like. Every time we iterated back with  
13          them they would correct one mistake in an equation and  
14          introduce another one. It was quite a cumbersome process.

15                 With this new desktop publishing system and all  
16          the engineers having it available on their own workstations  
17          we will be able to maintain living documents in that, when  
18          an engineer makes a change to the code not only does he  
19          submit changes in the FORTRAN coding into the code architect  
20          but he also must submit changes to the manual. The manual  
21          will always be up to date with the current version of the  
22          code, and it will alleviate some of the problems we have had  
23          with us saying that's not really what the code looks like  
24          now. This has become available as we have been able to get  
25          workstations and this high quality publishing type software.

1           That is all I wanted to say about the  
2 documentation of the field equations. Now, I would like to  
3 address the particular comment that was made by Dr. Schrock,  
4 that the momentum equations were written in non-conservative  
5 form. He didn't understand why it was done and where the  
6 justification was.

7           MR. ZUBER: Let me ask you, these Volumes, are  
8 they going to be available in September?

9           MR. WEAVER: Yes.

10          MR. ZUBER: All of them?

11          MR. WEAVER: Yes.

12          MR. ZUBER: Okay.

13          MR. WEAVER: They are planned to come out with the  
14 code so that everything will be documented and will all be  
15 current with the version of the code, and we won't have to  
16 send out interim internal documentation to describe the  
17 model. When the code comes out the documentation will  
18 automatically become available.

19          MR. CATTON: If you put it on a CD ROM and you put  
20 and use Framemaker's indexing program, you can do a search  
21 of whatever you want.

22          MR. WEAVER: That's right. Framemaker has a very  
23 strong what they call cross reference feature, so that you  
24 can automatically generate nomenclature and make sure that  
25 the mass transfer, kilograms per second in a cell, is always

1 gamma rather than M dot AB as is currently in some of the  
2 documentation. It ensures consistency.

3 You can also generate an index as well by doing  
4 key word searches. It's a very, very nice program. Because  
5 it's so versatile, of course, it takes some getting used to,  
6 just like any of the other ones would. It's intended for  
7 the assembly of multi-volume books as you will, where  
8 various sections are prepared by different people. There  
9 are automatically numbers from one section to the next, so  
10 the numbers are all consistent.

11 When you have an equation number you can cross  
12 reference it in the text. If you put another one in between  
13 it, it changes the cross reference in the text. It goes a  
14 long way to automating a lot of the problems that you see  
15 when you try to edit things by hand and then change. When  
16 you add a reference, for example, it slides all the ones  
17 down and changes all the reference numbers in the text.  
18 It will go a long way, to alleviating some of the problems  
19 we have had with the documentation.

20 The third thing is that, maintaining the living  
21 manuals will help alleviate losing the institutional memory.  
22 So that, when a model is put into the code the manual will  
23 be under document conversion control just like the code is.  
24 Are there any questions regarding what we are planning to do  
25 for the documentation? This is really the first section of

1 my presentation. It kind of stands by itself.

2 [No response.]

3 [Slide]

4 MR. WEAVER: Let's go on to the momentum equation.

5 The observation was made that the momentum equation was  
6 written in non-conservative form, and the manual didn't  
7 discuss why it was in non-conservative form versus  
8 conservative form.

9 What I have done is extract some of the design  
10 principles for the numerical procedure which were contained  
11 in Volume 1 of the manual. These were the principles that  
12 were used to derive the particular form of the difference  
13 equations that were solved in the code. Mass and energy  
14 conservation were considered paramount, and a greater degree  
15 of approximation to the momentum conservation was considered  
16 acceptable at that time. This is historical now. I am  
17 talking about where the equations came from and why they are  
18 the way that they are in the code.

19 Implicit evaluation was used only in the terms  
20 needed to exceed the sonic Courant limit because that was  
21 the problem at that time, trying to run at greater than the  
22 sonic Courant limit. The time level of the various pieces  
23 of the terms, whenever you had a product of void fraction,  
24 density, times of velocity for example, the time levels of  
25 those pieces were adjusted to make the implicit terms

1 linear.

2 The mass flux term in the continuity equation, the  
3 fluid property which was void fraction times density, was  
4 evaluated at the beginning of the time step. The velocity  
5 was evaluated at the end of the time step. So, the velocity  
6 was implicit in that term. The fluid properties were  
7 explicit, but that makes that term linear.

8 I would like to say that the non-conservative form  
9 has errors of the same order as the conservative form. That  
10 comes from the differencing of the spacial gradient  
11 operators. If you difference the gradients in first order  
12 differencing, whether it's conservative form or non-  
13 conservative form, the method is still first order. It may  
14 converge faster for a conservative form because the  
15 coefficient in the leading error term may be smaller, but  
16 it's still only first order.

17 [Slide]

18 MR. WEAVER: Using the design principles as stated  
19 in the first bullet on the previous slide --

20 MR. CATTON: The energy equation, is it written in  
21 conservative form?

22 MR. WEAVER: Strictly speaking, the expanded --  
23 what is the so-called expanded form of the equation -- none  
24 of them are conservative form because you have expanded the  
25 products. The definition of conservative form from the

1 textbook states that it's in conservative form if all the  
2 coefficient of the derivative terms are constants or, if  
3 they are not constant, they do not appear within the  
4 derivative terms.

5           When you expand for example the product of Alpha  
6 Rho in the time derivative of the continuity equation and  
7 use the expanded form that makes that particular part non-  
8 conservative, according to the strict definition of  
9 conservative form.

10           MR. CATTON: This definition means that the  
11 convective terms are written as Del dot, Rho V dot.

12           MR. WEAVER: Right. That, too.

13           MR. CATTON: Del dot Rho, whatever.

14           MR. WEAVER: The same thing happens when you split  
15 that Del dot of whatever product is into the coefficient  
16 times the divergence of the individual and expand it all  
17 out, that also destroys the strict definition of  
18 conservative. There is non-conservative in the strict sense  
19 in both the time derivative and in the spacial derivative.

20           The reason that the non-conservative form was used  
21 is that when you combine the non-conservative form with  
22 these design principles where we have adjusted the time  
23 levels of the properties of some of the primitive variables  
24 and not others you get a Poisson equation in pressure, which  
25 means that the system matrix that you have to solve, instead

1 of being a six N by six N where N is the number of cells for  
2 example, it comes down to an N by N and then you back  
3 substitute for the other five variables.

4 When the numerical procedure was first developed -  
5 - how many years ago, 77 -- computer capacity was not what  
6 it is today. That was a great advantage to reducing the  
7 amount of time spent in the matrix solution. Another part  
8 of this is, RELAP does not iterate. It takes a single  
9 advance, controls the truncation error based on the time  
10 step. You may be able to take larger time steps but then  
11 you are balancing off the time required to re-evaluate the  
12 matrix coefficients and solve the matrix against the series  
13 of smaller time steps without iteration.

14 I believe that when the algorithm was first put  
15 together they looked at iteration and decided that a single  
16 step method was more efficient computationally.

17 MR. CATTON: Virgil, this is your question.

18 MR. SCHROCK: I am sorry I am late.

19 MR. CATTON: You got here just in time.

20 MR. WEAVER: One of things that I think is the  
21 major concern is the modeling of the momentum flux term in  
22 the momentum equation, because those are the ones that you  
23 normally see big errors in and you can see errors in your  
24 solution as a result of the modeling of the momentum flux  
25 terms. I am going to concentrate on the momentum flux terms

1 now.

2 [Slide]

3 MR. WEAVER: What are the consequences of  
4 inaccuracies of modeling the momentum flux terms. You get  
5 artificial pressure losses in your system which result in  
6 lower flow rates for a given momentum source, if you have a  
7 pump for example, running at a certain speed and putting out  
8 a certain head. If you have numerical losses of momentum in  
9 addition to your physical losses of momentum, physical  
10 losses being wall friction, interfacial friction, you will  
11 get in a steady state a lower flow rate than you should.  
12 This also shows up as the incorrect pressure drop or  
13 recovery through area changes.

14 To look at this problem I have come up with a  
15 simple problem to look at the accuracy of the momentum flux  
16 formulation in RELAP.

17 MR. WULFF: Should there not also be among these  
18 two a third one, referring to the numerical damping when you  
19 have oscillatory flow that the amplitudes of your flow  
20 oscillations --

21 MR. WEAVER: That's right.

22 MR. WULFF: Are smaller. Is there an assessment  
23 of the magnitude of this damping, not to the friction but to  
24 the numerical damping?

25 MR. WEAVER: We have run oscillatory problems like



1 the manometer which is a well known thought problem, to  
2 investigate for oscillatory problems, whether we do get  
3 numerical damping. That's shown in Volume 3 of the manual,  
4 I believe. Yes, you are right.

5 MR. WULFF: What frequency is that?

6 MR. WEAVER: What frequency? I believe it's about  
7 2.8 hertz, the magnitude of the manometer, the frequency of  
8 the period of manometer that we set up.

9 MR. WULFF: Two oscillations per second?

10 MR. WEAVER: That's not true.

11 MR. WULFF: That's what I thought.

12 MR. WEAVER: The period is 2.8 seconds.

13 MR. WULFF: Right.

14 MR. WEAVER: I don't want to put words in your  
15 mouth, but I believe your concern might come out of  
16 stability, where we know that these finite difference  
17 methods have finite damping associated with them due to the  
18 upwind differencing in the --

19 MR. WULFF: Whatever numerical damping there is,  
20 it can perhaps be easily assessed for oscillatory flow. But  
21 there is a relation between the damping of oscillatory flow  
22 and non-oscillatory flow, and you have a distortion in the  
23 times and time constants. You have indicated that.

24 Is there a quantification of this artificial  
25 pressure loss? Do you have a ratio between --

1           MR. WEAVER: I am going to show you, based on  
2 these little simple thought problems, what the numerical  
3 loss is. The problem consists of a closed loop laying in  
4 the horizontal plane, so there's no gravity head. It has  
5 area changes in it, so that we look to see whether we are  
6 getting the correct pressure drops and losses through  
7 contractions. We turn off the wall friction. RELAP has a  
8 user flag that you can run frictionless. It was put in to  
9 do some of these kinds of things. Running single phase, so  
10 we don't have any interfacial frictions.

11           The problem has no momentum sources. If I start  
12 the loop up it should go forever, if I am conserving  
13 momentum.

14           [Slide]

15           MR. WEAVER: This is just a little diagram. I  
16 said before it's a horizontal pipe that closes back on  
17 itself. There's a 50 percent area of deduction here. We  
18 are running either single phase liquid or single phase  
19 vapor. The guy that ran this for me picked a pressure of 50  
20 bars. To be more extensive we should run more pressures.

21           The liquid velocities and the vapor velocities he  
22 picked were based on a constant mass flux. RELAP has an  
23 input option of specifying the mass flux at all the  
24 junctions, and it automatically adjusts the velocities for  
25 you. This is the conditions that were run.

1 [Slide]

2 MR. WEAVER: We ran two cases for single phase  
3 liquid, where the irreversible or physical losses at the  
4 contraction and at the expansion were both user set to zero  
5 by the user. You have the opportunity of putting in a loss  
6 coefficient to represent the irreversible loss through a  
7 fitting. Then, we ran it again, where the irreversible loss  
8 of the contraction was 0.5, and for the expansion was 1.0.  
9 We are going to look at what the velocity does in this,  
10 whether it keeps circulating infinitely or not.

11 [Slide]

12 MR. WEAVER: Now, for single phase liquid with no  
13 losses which is this case, it looks like it's constant. I  
14 have another expanded view of that to show you that I am not  
15 trying to pull the wool over your eyes.

16 MR. CATTON: How many cycles was that?

17 MR. WEAVER: That's 2,000 cycles. This is with  
18 the physical losses. For this case it looks like you have  
19 no losses at all.

20 [Slide]

21 MR. WEAVER: When you expand that and plot it on a  
22 much larger expanded scale, you will notice that now these  
23 numbers are varying in the sixth significant figure. You  
24 really do have a small numerical loss. This is because  
25 liquid is not incompressible. If you look at the equations

1 you will see that it's a velocity equation for the  
2 divergence. It has area ratios in it but it does not have  
3 density ratios.

4 Looking at it, I first thought that it would be  
5 correct for an incompressible fluid. For this case liquid  
6 is basically incompressible, and you can see that that's  
7 been borne out. Once again, that's 2,000 cycles. Zero  
8 sources, only the numerics. Since we are solving the matrix  
9 equation for the pressure solution with a numerical  
10 procedure, some of this might be round off in the matrix  
11 solution. You can't tell. It's hard to tell when it's this  
12 small.

13 Wolfgang, you look perplexed.

14 MR. WULFF: No. I don't recognize what I asked  
15 before. It's the relative importance of this. You simply  
16 picked some loss coefficients.

17 MR. WEAVER: Right.

18 MR. WULFF: I am not sure the coefficient of 0.5  
19 is for a tube going into an infinite tank, perhaps. I don't  
20 know where you picked this from and how relevant they are  
21 for the entrance in the core or for the space or grids. Are  
22 we dealing in this range?

23 MR. WEAVER: Yes. The loss coefficients for  
24 fittings and grid spacers are in the range of one to ten.

25 MR. WULFF: One to ten.

1 MR. WEAVER: Yes. You might argue that the "K"  
2 for the contractions should be .25 rather than 0.5. The  
3 guide picked 0.5 based on the 50 percent contraction.

4 MR. WULFF: When you have a two-phase multiplier  
5 with two-phase flow, what is the assessment of the numerical  
6 damping then?

7 MR. WEAVER: I have not run two-phase flow cases.  
8 I am going to show you on the next slide that the magnitude  
9 of the loss does depend on the density gradient around the  
10 loop. Again, it is small, compared to the physical losses.  
11 If you have two-phase flow with multipliers the physical  
12 loss only goes up. It doesn't go down, it goes up. It goes  
13 down, relative to single phase flow.

14 I would have liked to have been able to -- for  
15 this one, if I put in a momentum source so that it was  
16 absolutely flat to 13 significant figures, it would have  
17 been very, very, very, very small. Whether I have four  
18 varies on there or five varies on there, I can't tell you.

19 MR. WULFF: But we have ten seconds and not ten  
20 hours or ten minutes.

21 MR. WEAVER: That's right. But in a real plant  
22 you have momentum sources, you don't have sourceless flow.  
23 You have real friction which is a sink, and you have real  
24 gravity head which is a source. I am just trying to get the  
25 magnitude for frictionless here. I am trying to put it in

1 perspective, what the numerical loss is for this  
2 frictionless case.

3 [Slide]

4 MR. WEAVER: If I do the same thing for single  
5 phase vapor you can see that there is a small loss. I say  
6 small, relative to losses associated with real physical  
7 losses in a real system.

8 MR. SCHROCK: Have you compared the K --

9 MR. WEAVER: I have been trying to do that, and I  
10 was going to speak to that, Virgil.

11 MR. SCHROCK: I don't think you heard my question  
12 yet.

13 MR. WEAVER: Okay.

14 MR. SCHROCK: Have you compared the K with the FL  
15 over D for the loop that you simulated?

16 MR. WEAVER: No, I haven't. F is zero, for these  
17 cases.

18 MR. SCHROCK: The way you have calculated F is  
19 zero. You have a geometry and you know what the friction is  
20 in that geometry, have you compared the FL over D for that  
21 loop that you have calculated with the K's.

22 MR. WEAVER: I have not calculated the FL over D,  
23 if I allowed F to be greater than zero. I ran these  
24 frictionless, so by definition F is zero.

25 MR. SCHROCK: I know, but your conclusion is that

1 friction is represented by K's. I am asking you to tell us  
2 if you know, how FL over D for the loop that you have  
3 calculated compares with the K's that you have used in your  
4 computation.

5 MR. WEAVER: No, I do not know.

6 MR. SCHROCK: That ought to be compared.

7 MR. WEAVER: If you put in real friction this  
8 would only go down faster, because you are putting in more  
9 momentum sink.

10 MR. WULFF: I think Virgil's point is that you  
11 dominate with form losses which are always greater. But if  
12 you have geometries that have only the small L over D, then  
13 you have a much smaller decay due to physical reasons.

14 MR. WEAVER: Yes, I would agree with that.

15 MR. WULFF: What you have here, I think you can  
16 really -- instead of making these calculations look at the  
17 truncation error and normalize it with the friction, and you  
18 can give an expression for the ratio of these two forces of  
19 diffusion types, one through numerics and one through wall  
20 shear.

21 MR. WEAVER: In the single phase incompressible  
22 case, the ratio will be zero, incompressible. You will have  
23 no numerical loss that will be exact.

24 MR. CATTON: I really find that a little bit  
25 difficult to believe. It may be true, but I still have

1 difficulty with it. What did you use for V, Delta T over  
2 Delta X.

3 MR. WEAVER: Courant number, I did not run these  
4 calculations myself. This was run with a fixed one  
5 millisecond time step, so I suspect that you are fairly low  
6 below the Courant number of one. As I said, I did not run  
7 these myself.

8 MR. WULFF: You do have these extra terms, VISC F  
9 and VISC G. How large are they compared to the friction  
10 term?

11 MR. WEAVER: The combination of the difference of  
12 the cell centered velocity squared and the VISC term gives  
13 you a conservative form for the velocity diversions, in that  
14 the momentum cell velocity is convected by the cell centered  
15 velocity which is at its edge. That VISC term has an area  
16 ratio in it, which is why when I looked at it I suspected  
17 that I would get these nice good answers when I was  
18 incompressible.

19 [Slide]

20 MR. WEAVER: But because it did not contain a  
21 density ratio, I also suspected that I would have a  
22 numerical loss when I went to compressible flow, which is  
23 represented by this:

24 [Slide]

25 MR. WEAVER: This confirmed my suspicions. This



1 is an area where we need to look some more.

2 MR. CATTON: Is this also for 2,000 steps?

3 MR. WEAVER: No. It's 1,000 steps, I think.

4 MR. CATTON: That's still a lot.

5 MR. DHIR: Considering the time is only one  
6 second, that's large.

7 MR. WEAVER: I have to quantify the numerical K  
8 somehow. Since RELAP, if I put a pump in here and change  
9 the pump speed to make up the numerical pressure loss so I  
10 get an equivalent K, because of the heating of the fluid due  
11 to the pump I never get to steady state. The code will not  
12 allow me to put a negative K someplace to make up this loss,  
13 to get an idea of it. I have to make some changes to the  
14 code, adjust the K, so this is flat. That will give me an  
15 idea of the relative magnitude of the numerical loss versus  
16 losses which are more physical for real systems.

17 I also think I know how to correct this, by  
18 putting density ratios into this term so that this will be  
19 flat.

20 MR. WULFF: The original purpose for these VISC F  
21 and VISC G terms was to stabilize numerical oscillations;  
22 was it not? Is there a second stabilizing step? What we  
23 are really interested in is the combination of the numerical  
24 damping that comes from this source and any other source in  
25 this case on the momentum balance.

1 I think you have shown us some examples, but they  
2 are selective. I am not prepared to generalize these  
3 selective answers.

4 MR. WEAVER: The momentum flux term is written as  
5 the difference or the squares of the cell centered  
6 velocities plus the so-called VISC term, which maybe is  
7 misnamed but that's historically what it has been called.  
8 When you do the algebra you discover that the combination of  
9 the difference of the cell centered velocities and the VISC  
10 terms gives you a "conservative form in velocity" for the  
11 velocity diversions, in that the momentum cell quantity  
12 which is the junction velocity is convected to the next  
13 momentum cell with the cell centered velocity between them.  
14 What leaves one momentum cell enters the same face on the  
15 other side.

16 I don't believe that the VISC term was added after  
17 the design of the numerical procedure. It was put in that  
18 way from the beginning, but I can't confirm that. I wasn't  
19 around 15 or 20 years ago.

20 You can show algebraically, that the difference of  
21 the cell centered velocity squared and the particular form  
22 of the VISC terms does lead to a so-called conservative form  
23 in velocity, not in momentum. The areas are in there but  
24 the densities are not in there, which is why we see this.

25 MR. WULFF: Is that documented?

1 MR. WEAVER: It will be. As we improve the manual  
2 through this process we will put that in and t. . . l be  
3 part of the expansion of Volume 1, so that you will see  
4 exactly where all these terms come from.

5 MR. RANSOM: May I address the Chair?

6 MR. CATTON: Sure.

7 MR. RANSOM: I would just like to add a little  
8 clarification to this. What you are talking about is work  
9 that John and I did years ago, and maybe I can add some  
10 clarification.

11 We purposely formulated the momentum equation to  
12 be in divergence form in terms of the velocity squared, such  
13 that it would degenerate to Euler's equation in cases of  
14 variable area ducts and steady flows. When we did that the  
15 usual approach was to use upwind differencing for the  
16 momentum flux term, which in a variable area duct leads to  
17 losses. We felt that was unacceptable.

18 What we did is, we formulated the VISC term which  
19 would normally arise as a result of upwind differencing, in  
20 such a way that for incompressible flow and variable duct  
21 flow would lead to zero viscosity. Flows like you have just  
22 examined here, indeed, compressibility is the only effect  
23 which would lead to some loss, just as Walt has discovered.  
24 That's completely consistent.

25 The only type of way which is actually damped by

1 that upwind differencing or equivalent artificial viscosity  
2 are compression waves, which are propagated through the  
3 flow. There will be some damping associated with that type  
4 of thing. Oscillatory flows like a manometer for example  
5 that don't involve compressibility significantly, will not  
6 have any damping, and that's what the test problem showed  
7 for varying and significant damping.

8 I would just like to add one more thing. I have  
9 talked with these people repeatedly. This word,  
10 conservation form I think Walt has corrected pointed out,  
11 this is something that came from the literature about 10 or  
12 15 years ago having to do with hydro codes and calculational  
13 schemes. I think like it was correctly pointed out, it  
14 means divergence form, when the convective term can be put  
15 into divergence form.

16 Really, in RELAP5, both the mass and energy  
17 equation are in what you would call conservational or  
18 divergence form. Even though the time derivatives have  
19 expanded, that's not considered a non-conservational form.  
20 The thing you have to realize is, whenever you deal with  
21 partial differential equations if you want to use method of  
22 characteristics or other schemes to examine how boundary  
23 conditions and things like that should be formulated, it's  
24 always necessary to look at them in the primitive variable  
25 form.

1           This idea of conservational form came about as a  
2 result of difference equation formulations, in which  
3 presumably the things in the divergence would be preserved  
4 across discontinuous changes such as shocks. When we  
5 originally worked on these codes it was never envisioned  
6 that these would be applied to high speed flows. Indeed,  
7 there still today, is not any application where that's the  
8 case, where compressibility is of major importance.

9           Although not shown here, we have run nozzle  
10 problems with super sonic flow, where actually the normal  
11 shock that occurs in the super sonic flow is well captured  
12 by even these methods. That would be an interesting case  
13 maybe for you to look at, to get some idea of what kind of  
14 damping really is inherent in the schemes.

15           Conservational form does not mean that momentum is  
16 not conserved. The differential equations that are being  
17 used are just as valid as any form of the equations you  
18 would want to use. When you get to the finite difference  
19 form, obviously, there is going to be truncation involved  
20 there and some differences from that point of view.  
21 Hopefully, just a little clarification.

22           MR. CATTON: Meteorologists use the conservative  
23 form. I guess it was first developed by Arakawa at UCLA.  
24 The reason they give is large scale, large nodes, long  
25 times, not necessary to be in conservative form.

1 MR. RANSOM: I guess I am not familiar with that.  
2 There's nothing wrong with using conservation variables.  
3 That's Rho V, Rho V and Rho U, the energy variable. There's  
4 nothing wrong with that. That's just as good. It's just  
5 that here, when we talk about velocity, that's considered  
6 one of the primitive variables which you really would unfold  
7 anyway.

8 MR. CATTON: I understand the meaning of primitive  
9 variable.

10 MR. RANSOM: One point that I would like to make  
11 that I don't think came up yesterday, if you look at the  
12 conservation variable that should be preserved for a  
13 material interface or contact surface, it's velocity.  
14 Velocity is a conservation variable. It seems that we are  
15 dealing more with contact surfaces and material interfaces  
16 than we are with compressibility.

17 Really, velocity as a variable in divergence form,  
18 is the more appropriate variable for most of our cases.  
19 That's an interesting fact.

20 MR. SCHROCK: I guess I am a little surprised that  
21 you view the problems addressed by RELAP5 as not involving  
22 compressible flows. Surely, the flashing in the CMT is  
23 something that has to be described in terms of a  
24 compressible fluid phenomena.

25 MR. RANSOM: When I say compressibility, again

1 Virgil, what I mean is relative to the mach number that the  
2 velocity represents and generally accepted critical flow  
3 points. That's the only place we approach a mach number  
4 one.

5 Normally, we are well below mach numbers of one in  
6 most of the internal system. That's the sense that I am  
7 using compressibility here, in a sense that divergence of  
8 the velocity is basically near zero.

9 MR. SCHROCK: In terms of a two-phase sound speed,  
10 I am not all together sure that you really always are at  
11 such low mach number.

12 MR. RANSOM: Not always. That's true. We can  
13 choke internally, as the code will predict at times because  
14 of this low velocity of the sound speed. It's not 100  
15 percent, that's true. If you would like to see how it  
16 responds to compressible flows, this nozzle problem that I  
17 am talking about is a good example, and actually does  
18 surprisingly well.

19 MR. WULFF: Can I ask a question?

20 MR. WEAVER: I want to follow up Virgil. I  
21 believe you received a reading list of papers that we were  
22 going -- should have received a reading list.

23 MR. CATTON: When?

24 MR. WEAVER: We gave them to NRC quite a long time  
25 ago.

1 MR. CATTON: We haven't received any.

2 MR. WEAVER: A list of all the references we --

3 MR. SHOTKIN: We sent them to Mr. Boehnert, the  
4 list.

5 MR. WEAVER: The paper that Vic talks about, where  
6 this nozzle problem with shocks, is on that list. If you  
7 want to look at it you can.

8 MR. SHOTKIN: We could send it again.

9 MR. WULFF: The smallness of the slope here at the  
10 top curve is due to the fact that you have almost equal  
11 velocities across the cells. That is, one should look at -  
12 -

13 MR. WEAVER: I can change the velocities --

14 MR. WULFF: Let me suggest that this should be  
15 checked if you have strong dilatation in the cells. That  
16 is, if you have either large evaporation in the cell or  
17 condensation, contraction. Then, determine what the  
18 relative significance of the numerical diffusion is relative  
19 to the wall shear.

20 MR. WEAVER: I can do that.

21 MR. RANSOM: I might caution you with that. The  
22 one thing you are going to find is, it's very hard to find a  
23 steady state problem that involves that kind of problem.  
24 There are normally propagating type problems, so the  
25 information is swept out of the system and you can't really



1 find that. That's an aside.

2 I also should point out one other thing. On this  
3 diagram that you drew, Walt, you really should have shown  
4 those abrupt area changes as smooth. The way you are  
5 modeling that, I presume is, they are smooth.

6 MR. WEAVER: Smooth, with --

7 MR. RANSOM: The toughest test for this is  
8 actually a variable area duct, if you really want to see  
9 whether or not Bournolli's equation is satisfied. That's  
10 exactly the artificial viscosity -- if you want to call it  
11 that -- was formulated, so that it does affect recovering  
12 Bournolli's equation for frictionless incompressible flow  
13 under steady state.

14 That's in contrast to a lot of the other hydro  
15 codes. They will actually have sources and sinks as a  
16 result of the upwind differencing. We purposely tried to  
17 avoid that.

18 MR. ZUBER: I hope all this will be explained in  
19 your Volume 1.

20 MR. WEAVER: Yes.

21 MR. ZUBER: Good.

22 [Slide]

23 MR. WEAVER: I was trying to respond to a question  
24 about the magnitude of this loss versus what I call physical  
25 losses. To do that I will have to change the code, to put

1 an explicit momentum source as a negative K into the code.  
2 The code does not presently allow me to do this. These were  
3 done about last Thursday.

4 I am going to pursue this, to try to quantify the  
5 numerical loss as a K versus the physical losses that you  
6 have in real systems, so that we can get a handle on the  
7 relative magnitudes of this.

8 MR. CATTON: Shouldn't you also be doing these  
9 kinds of tests with a Courant number that is close to what  
10 you run them at in an actual plant calculation?

11 MR. WEAVER: In a real plant calculation the  
12 Courant number runs the gambit, from one all the way down to  
13 0.1. The velocities are all different in various places,  
14 and the lengths of the cells. The only thing I can say when  
15 I look at a code calculation is, one of the cells in the  
16 code is running close to Courant number of 0.1. I can say  
17 nothing on the printout about the Courant numbers of any of  
18 the other cells or the spectrum of Courant numbers, or  
19 things like that.

20 That's why we go to these little simple problems  
21 where you have equal length cells and where you have  
22 constant velocities. There is actually two Courant numbers  
23 in there. There's one for the big area and one for the  
24 small area. There's just two Courant numbers in that  
25 problem. When you go to real decks it gets real messy, very

1 fast.

2 MR. CATTON: I understand.

3 MR. ZUBER: Just one other point. I agree with  
4 Wolfgang, to put this literally to rest you really should  
5 consider a large change of density. You show a density  
6 right here which is relatively small. Since you are really  
7 working with compressible fluids, very much compressible, I  
8 think you should really address that problem and put it to  
9 rest, one way or the other.

10 MR. WEAVER: The code will not currently let me  
11 run frictionless in the two-phase case. I can turn off wall  
12 friction in the two-phase case but I cannot turn off  
13 interfacial friction. Interfacial friction is a damping  
14 mechanism, a real damping mechanism, that we have in the  
15 code.

16 MR. WULFF: But you can add it up for cell and get  
17 the sum, and use that as your basis.

18 MR. WEAVER: But I can't get a steady state with  
19 two-phase without a momentum source to make up the numerical  
20 loss. Somehow I have to get a steady state so that I can do  
21 all this adding up, so the transient effects don't affect  
22 the answer. I agree with you.

23 MR. WULFF: Then, you should use a simple  
24 analytical solution.

25 MR. CATTON: I would like to make a suggestion.

1 If you have any recommendations for how they should do this  
2 why don't you communicate them to me, and I would like to  
3 move on.

4 Are you about finished?

5 MR. WEAVER: Yes. All I have is a summary of  
6 those four test cases.

7 [Slide]

8 MR. WEAVER: I make a rather bold statement here  
9 on the last slide. Through a variety of thought problems  
10 and separate effects tests and integral tests we have never  
11 seen anything that we can attribute to the non-conservative  
12 form of the momentum equation. I believe your concern was  
13 for natural circulation flows where the driving heads are  
14 very small. I reference some simulations that we have done  
15 with natural circulation. These two are in the manual,  
16 Volume 3 of the manual, which I believe you have.

17 There is a compendium of natural circulation  
18 experiments and simulations as an NRC NUREG. These BETHSY  
19 tests, natural circulation tests that were done in France,  
20 compared the data and the same semiscale natural circulation  
21 tests that were done in semiscale. These appear in the same  
22 place.

23 MR. ZUBER: Who did the BETHSY calculations; you  
24 did?

25 MR. WEAVER: One of our analysts, as part of the

1 standard problem. I don't know what the standard problem  
2 number was. It was Paul Roth, at Idaho. That's the end of  
3 that one.

4 MR. CATTON: Thank you. I think we would like to  
5 hear next, the interfacial heat transfer.

6 [Slide]

7 MR. JOHNSEN: My presentation this afternoon on  
8 interfacial heat transfer modeling was specifically  
9 motivated by comments that were forwarded to us from the  
10 consultants.

11 [Slide]

12 MR. JOHNSEN: I am going to address three specific  
13 comments that were conveyed. Professor Schrock made the  
14 observation that the modeling of the interfacial area is  
15 simplistic; that the heat transfer coefficients in the  
16 interfacial heat transfer model are ad hoc; and, I think  
17 both Dr. Wulff and Professor Schrock had problems with the  
18 mass transfer at the wall terminology. I want to address  
19 each one of those areas.

20 [Slide]

21 MR. JOHNSEN: I am going to start by providing  
22 sort of a summary response, and acknowledge to you that the  
23 models for interfacial heat and mass transfer are based on  
24 idealizations of the various flow regimes and to the extent  
25 possible augmented with literature based coefficients,

1 again, where that's possible.

2 We believe the models are faithful to the  
3 philosophy of --

4 MR. ZUBER: This slide is missing.

5 MR. JOHNSEN: The one that says summary response  
6 is missing? I will try to get a copy to you before the end  
7 of the day.

8 We believe the models are faithful to the  
9 philosophy of using a simple but plausible model, and then  
10 validating it using relevant data. Ultimately, the level of  
11 sophistication in the model is based on the principle of  
12 providing an adequate simulation for the intended purpose to  
13 which the code will be placed. That's a bit of philosophy,  
14 I guess, more than anything else.

15 [Slide]

16 MR. JOHNSEN: I am listing here some of the  
17 idealizations of the flow regime interfacial area that we  
18 use in the code. Bubbly, assumes spherical bubbles and a  
19 size distribution of bubbles. I mentioned yesterday, the  
20 maximum bubble size is based on a Weber number criterion.  
21 The interfacial heat transfer coefficients are in fact based  
22 on the idea of spherical bubbles, so there's consistency  
23 between the interfacial area assumption and the coefficients  
24 that were being used.

25 So on down the line, you can see for yourself what

1 the idealization is that is being made.

2 [Slide]

3 MR. JOHNSEN: One of the consequences of adopting  
4 the two fluid concept is that it requires you to specify a  
5 lot of coefficients that you cannot derive from direct  
6 measurement, simply because there's no feasible way of doing  
7 it. In fact, the combination of possible thermodynamic  
8 states and flow regimes in RELAP requires us to specify 48  
9 phases of heat transfer coefficients. These are the liquid  
10 side and vapor side heat transfer coefficients that connect  
11 at the interface.

12 Of those 48 coefficients, 24 of them apply to  
13 metastable states. For the most part, the ones that apply  
14 to the non-metastable states -- in fact, in 18 out of 24  
15 cases -- the heat transfer coefficients are from the  
16 literature.

17 MR. DHIR: Excuse me. What do you mean by non-  
18 metastable.

19 MR. JOHNSEN: I mean in other words, conditions  
20 that are plausible, super heated liquid for example in  
21 contact with the subcooled liquid, that's a non-metastable  
22 state. Super cooled vapor is metastable. Super heated  
23 liquid, metastable. That's what I mean by metastable.

24 MR. DHIR: No, you are saying non-metastable.  
25 Suppose I have wall, that vapor is condensing on it; what is

1 it?

2 MR. JOHNSEN: You have super heated vapor in  
3 contact with a cold wall.

4 MR. DHIR: Cold wall.

5 MR. JOHNSEN: Yes, that would be a stable state,  
6 not metastable. Subcooled vapor is metastable. Superheated  
7 liquid is metastable.

8 MR. DHIR: You talk about it, that you have  
9 something sitting at a neutral equilibrium and it can go  
10 either way.

11 MR. JOHNSEN: Yes, stable. Stable, as opposed to  
12 non-stable. The coefficients that are used for the  
13 metastable states are in fact designed to ensure that these  
14 states are transitory. They are not stable, so they should  
15 be not long lasting.

16 I did point out at the bottom of the slide that  
17 there is an error in the Volume 4 relative to bubbly flow  
18 with super heated vapor -- that's what that stands for. The  
19 Volume says that the NUSSELT number is equal to ten to the  
20 fourth, and that's not correct. What it should say is, the  
21 heat transfer coefficient on the vapor side is set to ten to  
22 the fourth watts per squared meter degree Calvin, which was  
23 suggested by this reference here.

24 MR. SCHROCK: Gary, I have the statement in front  
25 of me that I wrote on this topic. What it says is, the heat



1 transfer coefficient in most cases is just arbitrated and  
2 motivated only by what the developer deems a desirable  
3 magnitude. An example is the coefficient HIG. I quote from  
4 the documentation on page 4.1-15, following as the  
5 quotation.

6 "The volumetric heat transfer coefficient HIG for  
7 bubbly SHG is not based on theoretical empirical  
8 correlation. The NUSSELT number,  $NU_{ib}$  equals ten to the  
9 four is chosen to be large in order to bring the gas  
10 temperature rapidly towards the saturation temperature.  
11 Function F6 clearly enhances this tendency, especially as  
12  $\Delta T_{SG}$  increases magnitude. Function F7 apparently  
13 improves numerical stability for low void fractions. The  
14 determination of volumetric interfacial area is discussed in  
15 Section 4.1.1.1.

16 Clearly, there is room for improving the  
17 determination of HIG for this case, although to the best of  
18 our knowledge this might require further experimental work."

19 That's what I said. I have added my own comment  
20 on that. "This statement is typical of those following 14  
21 various coefficients presented in this section. It is  
22 evident that the interfacial heat transfer calculation has  
23 little engineering basis. By their nature these ad hoc  
24 equations cannot be compared directly with experimental data  
25 to assess their value. The interfacial heat transfer

1 package is weak."

2 Those were my comments. I think you have  
3 misrepresented them in the way you have presented the  
4 comment that you are trying to address here. Further, there  
5 is a numerical discrepancy between your assessment of 18 of  
6 24 or from the literature, whereas I have stated here in my  
7 written report that I counted 14 in the RELAP5  
8 documentation.

9 MR. JOHNSEN: The following two slides in your  
10 packet should summarize all 48 coefficients. You can count  
11 them for yourself. Where they come from literature, that's  
12 so indicated. I wasn't going to go through them in this  
13 presentation but they are in your packet for completeness.

14 There is, regrettably, some mistakes in Volume 4.  
15 It went out as a draft. I would apologize to you for those  
16 areas in the documentation where there are mistakes, such as  
17 the one I have indicated here.

18 MR. SCHROCK: You are saying that the document  
19 that I reviewed was inaccurate, not that my count is  
20 inaccurate. What are you saying? Did you look at the  
21 documentation to verify that what I said is incorrect?

22 MR. JOHNSEN: I was not able to establish that  
23 there were 14 coefficients that would be considered ad hoc,  
24 that is to say, having no basis in some form from the  
25 literature.

1 MR. SCHROCK: There's a considerable difference  
2 between 14 and six. I think what you are doing is kissing  
3 this off as something that is kind of irrelevant. It's not  
4 irrelevant.

5 MR. JOHNSEN: I am not kissing it off at all,  
6 Virgil. I am trying to be forthright, and explain that in  
7 those instances where there was a way to attach a literature  
8 based model to the code we have done so. Not in every case  
9 can that be done. In those instances where you cannot do  
10 it, then you try to pick a reasonable --

11 MR. SCHROCK: Here's a very nice example of the  
12 fact that it's recognized by somebody writing this  
13 documentation, that this might require further experimental  
14 work.

15 MR. JOHNSEN: Again, that was a value judgment.

16 MR. SCHROCK: Is this on the list of things at  
17 INEL that need to be done? Have you advised your sponsor  
18 that there ought to be some experimental work done to  
19 provide a basis for the coefficient disputed in RELAP5.

20 MR. JOHNSEN: No. No, not for the bubbly flow.  
21 We see no need whatsoever at this point to require any  
22 further information.

23 MR. SCHROCK: That was the thrust of my comments  
24 here.

25 MR. JOHNSEN: I think what I want to try to convey

1 today is that the code does embody assumptions, and that  
2 they can be looked at and criticized. We could argue all  
3 day about some of these things. In the end what we try to  
4 do is, come up with a plausible approach to modeling  
5 something and then testing it against what we think is  
6 relevant data, that being data that can be associated with  
7 the ultimate application of the code.

8           What I am trying to do today is not so much defend  
9 every coefficient that is in the code -- I don't think  
10 that's very productive. What I would rather do is show you  
11 a few examples of the kinds of experiments we have used to  
12 either find out our model was adequate or was inadequate,  
13 irrespective of its simplicity or how arbitrary it might  
14 seem.

15           [Slide]

16           MR. JOHNSEN: I have given you here, a list of the  
17 experimental problems we have used to specifically test,  
18 validate and --

19           MR. WULFF: Excuse me. You are not explaining the  
20 tables, telling us what --

21           MR. JOHNSEN: I am not going to spend the time to  
22 go through the tables.

23           MR. WULFF: I don't mean to go through the tables.  
24 I think you need to know what the "F's" are, and you have to  
25 give us dimensions. This table is really not readable.

1 What is three times ten to the sixth?

2 MR. JOHNSEN: If we start into the table I am  
3 afraid we are going to be here all day long. If you have a  
4 specific question, I promise to give you an answer to it.

5 MR. CATTON: That sounded like a specific  
6 question. What is three time ten to the sixth?

7 MR. JOHNSEN: Which one are you referring to?

8 MR. CATTON: Liquid film, under hif,SHL.

9 MR. WULFF: Is that 3.6 by Theofanous, or how do I  
10 interpret this table? Where does it come from?

11 MR. JOHNSEN: You are referring to --

12 MR. WULFF: Table one.

13 MR. JOHNSEN: This number right here; is that  
14 correct?

15 MR. WULFF: That's one place, yes.

16 MR. JOHNSEN: This is the liquid side heat  
17 transfer coefficient where there is super heated liquid.  
18 What this says is that the coefficient on the liquid side is  
19 set to be very high.

20 MR. WULFF: Is it in watts per meter squared in  
21 degree?

22 MR. JOHNSEN: The specific units there are watts  
23 per square meter degree, Calvin. This is one of the  
24 metastable conditions, super heated liquid.

25 MR. ZUBER: I have a short question. In annular

1 flow you have a liquid. How do you calculate heat transfer  
2 at the wall.

3 MR. JOHNSEN: That depends on the wall condition,  
4 whether it's --

5 MR. ZUBER: Suppose you don't boil.

6 MR. JOHNSEN: Suppose it's condensation?

7 MR. ZUBER: You have annular flow through a pipe  
8 which may be heated, you are routing energy to it but not  
9 burning. How do you calculate the heat transfer.

10 MR. JOHNSEN: In that case, assuming that the  
11 velocities were high enough, it would be forced convection  
12 to the liquid phase.

13 MR. ZUBER: Only liquid.

14 MR. JOHNSEN: Yes.

15 [Slide]

16 MR. JOHNSEN: I don't want to spend time  
17 discussing each one of these cases. They are in your  
18 handout. In each case what I am describing is the  
19 experiment that we used and the specific objective in terms  
20 of assessing the code. What I am going to do is show you  
21 several that relate directly to the interface heat transfer  
22 modeling for various flow regimes.

23 [Slide]

24 MR. JOHNSEN: The first one I am going to show you  
25 is this Christensen Subcooled boiling experiment. This

1 experiment involves a heated tube in which we introduce  
2 subcooled liquid at the bottom, subcooling of about 12  
3 degrees Calvin. The experiment was run up at five and one-  
4 half mega-pascals. The skin heater power is 70 kilowatts, a  
5 little over a meter per second.

6 This experiment was used to assess the interface  
7 mass transfer and subcooled nucleate boiling models. This  
8 is the example that Joe Kelly was actually talking about  
9 this morning.

10 MR. DHIR: Here, I have question. How do you  
11 calculate first the subcooled nucleate boiling heat transfer  
12 coefficient?

13 MR. JOHNSEN: Using the Chen correlation.

14 MR. DHIR: Is it valid?

15 MR. JOHNSEN: Rex, do you want to comment on that,  
16 the applicability of Chen and subcooled boiling?

17 MR. SHUMWAY: This correlation is referred to as  
18 applicable for subcooled boiling when you use T-wall minus  
19 T-liquid on the convection part of the correlation, T-wall  
20 minus T-sat on the boiling part of the correlation, Zuber.

21 MR. JOHNSEN: From his original reference, you  
22 mean?

23 MR. SHUMWAY: No. He originally, just based upon  
24 saturated liquids. Other people later on have said it's  
25 also applicable for subcooled if you use the liquid

1 reference temperature on the Dittus Buelter part of the  
2 correlation.

3 MR. DHIR: Even if you go back and look at your  
4 documentation in Volume 4, clearly, there is a comparison of  
5 heat flux as a function of super heat. Chen's correlation  
6 and Tom's correlation, see how much difference you get with  
7 wall super heat itself. Chen's correlation gives you very  
8 weak dependence on wall super heat. In reality, there's a  
9 very strong dependence on wall super heat, Delta T to almost  
10 cubed. Chen's correlation gives you 148 or something. It's  
11 a misuse of Chen's correlation to subcooled boiling.

12 MR. JOHNSEN: As Rex said, I think other  
13 investigators have --

14 MR. DHIR: Why don't you give reference to who has  
15 used it?

16 MR. JOHNSEN: Rex, the other investigators that  
17 you mentioned, are they contained in the references we have  
18 provided for Lou Shotkin?

19 MR. SHUMWAY: No, I don't think so.

20 MR. JOHNSEN: Could we come up with those?

21 MR. SHUMWAY: Yes.

22 MR. JOHNSEN: We will provide those to you through  
23 the NRC. I am talking about the specific references that  
24 claim that the model is applicable to the subcooled boiling.

25 [Slide]



1 MR. JOHNSEN: Here is a plot of the result using  
2 MOD3. What I am showing you here is the void fraction as a  
3 function of the elevation of the tube in meters. On the  
4 left hand scale the void fraction, this is the experimental  
5 data in the RELAP calculation. In the experiment they did  
6 not measure the subcooling, so I am just showing you the  
7 RELAP prediction of what the subcooling is as a function of  
8 the elevation.

9 This is a good example of where the bubbly flow is  
10 predominant and where you have subcooled boiling going on,  
11 resulting in a combination of generation and condensing in  
12 the bulk.

13 MR. WULFF: This is a combination of heat transfer  
14 and vapor generation rates. It is not an assessment, an  
15 isolation and then an assessment of heat transfer.

16 MR. JOHNSEN: That's correct. It's not heat  
17 transfer alone. As a matter of fact, it involves also the  
18 interfacial as well. There's a combination of things going  
19 on here. I am not going to say we are isolating just one  
20 thing. You can see that the overall capacity to model this  
21 kind of situation appears to be rather good.

22 MR. ZUBER: What criterion do you have to stop  
23 that void curve?

24 MR. JOHNSEN: In the code?

25 MR. ZUBER: Yes.

1 MR. JOHNSEN: The code uses a nucleation criteria  
2 that spans one degree of superheat, such that at one degree  
3 of superheat you begin nucleation with a minimum void  
4 fraction of ten to the minus five.

5 MR. ZUBER: I am familiar with some of the  
6 Christensen data.

7 MR. JOHNSEN: I may have misunderstood your  
8 question. We are using Leahy's model for the point of net  
9 vapor generation.

10 MR. ZUBER: What would have been probably better  
11 is if you had started where he starts boiling, not from the  
12 entrance but somewhere down that pipe, then you can assess  
13 whether you really predict that point correctly and  
14 calculate correctly and then you predict the curve.

15 MR. JOHNSEN: You mean, extend this back in this  
16 direction?

17 MR. ZUBER: That's right. I don't know whether  
18 this is the entrance of the pipe or what, the zero.

19 MR. JOHNSEN: This is the end of the heated  
20 section. I believe there was a lead in section that was  
21 unheated.

22 MR. ZUBER: Then you have a portion which is  
23 subcooled at the superheated region at the wall, and then  
24 you start taking off.

25 MR. JOHNSEN: What I am showing you is, the bulk

1 fluid temperature was subcooled all the way down the  
2 channel. All the way down the channel it remains subcooled,  
3 according to the code calculation. The code says it was  
4 subcooled all the way down the channel.

5 MR. ZUBER: There are data where you start with  
6 subcooling and then you develop superheated point somewhere  
7 down the pipe, and then you develop --

8 MR. JOHNSEN: Then you start getting voids,  
9 pulling you away from the wall.

10 MR. ZUBER: The "S" shaped curve, I think that  
11 would be a better test. Here, you start immediately  
12 developing the voids from the very entrance.

13 MR. JOHNSEN: I know what you are saying. We have  
14 run those tests specifically for Savannah River because they  
15 were concerned about --

16 MR. ZUBER: Those, we didn't see. I think I see -  
17 - it's better for you to provide us with something -

18 MR. SCHROCK: I wonder if you are correctly  
19 representing Christensen's data. My recollection of it is,  
20 there is a substantial region upstream, more than a couple  
21 of centimeters as this illustrates, before you get in that  
22 void generation.

23 MR. JOHNSEN: Yes. I think this zero was the  
24 beginning of the heated section.

25 MR. SCHROCK: I don't think so, that's what I am

1 saying.

2 MR. ZUBER: It doesn't mean that you start boiling  
3 at the entrance.

4 MR. JOHNSEN: I understand that.

5 MR. ZUBER: You have to build up a superheat at  
6 the wall, and then you start boiling.

7 MR. JOHNSEN: I understand.

8 MR. ZUBER: I think that would be a better test  
9 for your capability to calculate subcooled boiling than this  
10 curve here.

11 MR. SCHROCK: Can we get a comment from the person  
12 that did that; is this really the correct representation of  
13 Christensen data? I don't think it is. I think the  
14 Christensen data shows a non-boiling temperature preceding  
15 that vapor generation.

16 MR. JOHNSEN: You are really just quibbling about  
17 what this zero really means.

18 MR. ZUBER: It's not quibbling.

19 MR. SCHROCK: I am not quibbling. I am asking you,  
20 are you plotting this data correctly.

21 MR. JOHNSEN: It's within the latitude to decide  
22 to put zero where the heating begins or put it somewhere  
23 else. I don't remember in the reference exactly where the  
24 zero point is.

25 MR. SCHROCK: In the Christensen experiment the

1 heating began at the beginning. But with a high level of  
2 subcooling at the entrance, there's a substantial non-  
3 boiling or at least zero net void generation distance before  
4 you begin this S-shaped curve that Novak is talking about.

5 MR. CATTON: Beyond the edge of the heater. If I  
6 extrapolate your black line -- there is a void fraction at  
7 zero. Somebody didn't plot it very well.

8 MR. JOHNSEN: I know what you are saying.

9 MR. ZUBER: Gary, I am not sure whether subcooled  
10 boiling is very important for AP600. This has to be thought  
11 about. The thing is, if you want to apply this code to  
12 Savannah that was a critical question, how well you predict  
13 it.

14 MR. JOHNSEN: Right.

15 MR. ZUBER: Had you done these calculations you  
16 would have seen that there is a region of subcooled flow  
17 without any voids. Then, you start developing the curve  
18 somewhere downstream. That is a test of your capability.

19 MR. JOHNSEN: Significant void and then net vapor  
20 generation that precede.

21 MR. WULFF: Your 12 degrees subcooling are  
22 representative of entrance subcooling in a boiling water  
23 reactor under normal conditions and has tremendous heat  
24 flux, more than in this experiment most likely. It takes on  
25 the order of 30 centimeters before net vapor generation

1 begins. That's where your zero should take off in the void  
2 fraction.

3 MR. ZUBER: Again, this may not be important for  
4 this reactor.

5 MR. JOHNSEN: The main reason for these to be in  
6 this presentation was to try to illustrate the philosophy of  
7 formulating a model and then testing it against relevant  
8 experiments. This is not AP600-specific.

9 MR. ZUBER: I am not convinced that you really  
10 have the capability until you show the other parts.

11 MR. JOHNSEN: Okay.

12 [Slide]

13 MR. JOHNSEN: The Oak Ridge heat transfer test  
14 facility void profile test is another one we have used. We  
15 have a bundle, eight by eight electrically heated bundle,  
16 four and one-half mega-pascals. You see the heat flux, mass  
17 flux. Again, inlet subcooling that was quite substantial.

18 MR. CATTON: Is this one of the THTF tests, where  
19 they actually measure the void fraction?

20 MR. JOHNSEN: Yes.

21 MR. CATTON: Good.

22 [Slide]

23 MR. JOHNSEN: This is the predicted of an actual  
24 void fraction in the experiment. The data is the dash line.  
25 RELAP is the solid line. The comparison is reasonably good.

1 I believe that these were the grid spacer affects in the  
2 experiment that couldn't be captured in the calculation.

3 MR. ZUBER: Do you know for these experiments  
4 where you have equilibrium conditions, cross sectional, for  
5 this set of data?

6 MR. JOHNSEN: Not off hand, I don't know where the  
7 point of equilibrium was reached. Right off hand, I don't  
8 know.

9 MR. CATTON: They get super heated steam up on the  
10 top when the void fraction is very high.

11 [Slide]

12 MR. JOHNSEN: The next one I am showing you is the  
13 Bankoff stratified flow case. Our objective here was to  
14 look at the condensation for stratified flow. This is an  
15 experiment that was run down at atmospheric pressure. It  
16 involves a rectangular duct, roughly six by 30 by 160  
17 centimeters in dimensions. You have liquid and steam flow  
18 at the entrance -- at one end of the test section --  
19 wherein, the liquid is coming in basically at close to room  
20 temperature and the steam is coming in above the boiling  
21 point at atmospheric pressure, somewhat superheated steam.

22 The idea is to examine the condensation of the  
23 steam on the liquid as it flows down the test section. This  
24 is the RELAP5 diagram, if you will of this experiment, where  
25 we are introducing both steam and liquid at one end of the

1 test section and using an atmospheric sink, if you will, on  
2 the other end. The flow of the liquid and the steam is low  
3 enough that you get a stratified condition running down this  
4 channel.

5 What Bankoff did was to measure the flow of the  
6 steam at various locations down the duct as a means of  
7 inferring the contact condensation of the steam on the  
8 liquid surface. This is the first run. There were two  
9 different runs as I indicated on there. One is considered  
10 the low flow and one is a little higher flow case.

11 MR. DHIR: Excuse me. What heat transfer  
12 coefficient do you use, what correlation?

13 MR. JOHNSEN: For stratified flow?

14 MR. DHIR: Yes, for this situation. Steam is  
15 flowing over cold water.

16 MR. JOHNSEN: Right. Can you help me with that  
17 one, Rich? It's in the Dittus table.

18 MR. DHIR: Stabilized heat, Dittus Boelter  
19 correlation. Is that valid?

20 MR. JOHNSEN: Let me go back and look at the  
21 table.

22 MR. CATTON: That's what it says.

23 MR. JOHNSEN: Yes, that's what we use. On the  
24 liquid side it's Dittus Boelter.

25 MR. DHIR: On the vapor side it's also Dittus.



1 MR. JOHNSEN: That's correct.

2 MR. DHIR: Is that correct way of doing it?

3 MR. JOHNSEN: In this case with the stratified  
4 condition, it's an assumption that we can apply a pipe-like  
5 forced convection correlation.

6 MR. DHIR: You have to justify the assumption  
7 before you can apply it.

8 MR. JOHNSEN: Don.

9 MR. MCELIGOT: Back in the late 1960's there were  
10 some studies done on non-compressed gases, so they would  
11 apply for either the liquid side or the saturated vapor side  
12 for square and for triangular tubes over the range from  
13 laminar where there is a slight difference in the  
14 coefficient you would use up through turbulent flow. I  
15 suspect maximum Reynolds numbers were around 60 to 100,000  
16 or so, based on hydraulic diameter.

17 Essentially, these results agreed with the Dittus  
18 Boelter equation as presented by McAdams for this particular  
19 case, the vapor, where it would be a coefficient  $c = 0.023$ .  
20 There is some justification for using it in the case where  
21 you have a rectangular cross section.

22 MR. DHIR: No, that's not the question. You have  
23 a developing flow from the liquid side as well as from the  
24 vapor side. You have to calculate first the number for the  
25 vapor and the liquid and see how long that length would be.

1 Then, you have to go back and correct your heat transfer  
2 coefficient on both sides. What was the number for the  
3 liquid for example?

4 MR. JOHNSEN: I don't know, off hand.

5 MR. DHIR: Is it laminate or what?

6 MR. JOHNSEN: Obviously, they are not a high flow,  
7 otherwise they wouldn't be stratified --

8 MR. DHIR: Then you have very long entry length.

9 MR. CATTON: Looking at the channel most of it was  
10 probably entry length.

11 MR. MCELIGOT: That would depend on the Reynolds  
12 number. It was not, but if it had been a circular tube a  
13 Reynolds number of say 2,000, in order to get the NUSSELT  
14 number within about five percent it would take 100  
15 diameters. That entry length varies linearly essentially,  
16 with the Reynolds number. A Reynolds number of say 100 the  
17 necessary entry length would be the order of ten diameters.

18 That's just giving you order of magnitude  
19 estimates.

20 MR. DHIR: My point is, one has to assess first  
21 before just blindly applying a correlation.

22 MR. WULFF: My problem is that Bankoff's  
23 experiments were on a flat bed of liquid with vapor on top  
24 of it. Dittus Boelter was internal duct flow, whether that  
25 is rectangular or circular, it's still internal flow. Why,

1 with this difference in geometry, Dittus Boelter apply here.

2 MR. JOHNSEN: Under a stratified condition what  
3 the vapor essentially sees is the liquid as being a wall  
4 surface and vice versa. That's the analogy.

5 MR. SCHROCK: I could contrast that to the case of  
6 flow in a rectangular cross section heated on one side only.  
7 Of course, there's a lot of data in the literature, much of  
8 analytical, to show you that the heat transfer coefficient  
9 is not the same for that case when you heat from one side  
10 only.

11 This liquid layer is being heated only at the  
12 interface.

13 MR. JOHNSEN: That's correct.

14 MR. SCHROCK: It's being heated on one side only.  
15 Consequently, Dittus Boelter equation should not be expected  
16 to apply.

17 MR. JOHNSEN: What would you say would be the  
18 quantifiable -- how would you quantify the distortion?

19 MR. SCHROCK: As rather significant.

20 MR. JOHNSEN: Can you give me a number?

21 MR. SCHROCK: It's one of the early papers, about  
22 1951, in which it shows the determination of heat transfer  
23 coefficients for flow in that kind of channel heated from  
24 one side only. The important difference here is that the  
25 temperature profile is essentially only one-half of the kind

1 of temperature profile that you are looking at in the full  
2 symmetrically heated tube. It goes to an adiabatic wall on  
3 the other side.

4           When you calculate both temperature you get a very  
5 different number there. The heat transfer coefficient has a  
6 dependency on that temperature profile. That's what it's  
7 really representing, is a lump parameter characterization of  
8 the dimensionless temperature profile. That's what it  
9 means.

10           MR. DHIR: Many textbooks have tables showing  
11 these geometries, one side heated, both side heated,  
12 rectangular, circular, whatever it is, you can get for  
13 steady state or fully developed flow. Still, you have to  
14 account for the development of the length.

15           MR. JOHNSEN: I guess this is another -- Vic, do  
16 you want to comment?

17           MR. RANSOM: Can I make a remark?

18           MR. CATTON: Sure.

19           MR. RANSOM: I have listened to this presentation,  
20 and the thing that seems to be constantly happen here as we  
21 look at the details of the physical process that's going on  
22 -- while I think that's good from an insight point of view  
23 and the developer should do that -- it seems worth  
24 remembering that this thing is a systems code. Actually  
25 from the start we have sort of assumed fully developed flow

1 everywhere. It's only a special case, if you really have to  
2 depart from that.

3 So often times the attempt has been to use a fully  
4 developed flow approximation for friction coefficients and  
5 heat transfer coefficients. In this case for example,  
6 friction has been calculated around a friction factor  
7 correlation and a relative velocity. If you simply use the  
8 Reynolds analogy for the heat transfer you would get a  
9 Dittus Boelter type of heat transfer correlation.

10 I don't think they are out to lunch in a case like  
11 this. You can argue with the details but the actual fact of  
12 the matter is, I don't think any heat transfer coefficients  
13 have been measured. They are very difficult to do between  
14 an interface and two fluids, like you are talking about  
15 here.

16 Really, some of these rectangular duct condensing  
17 flow type experiments are about the only cases where there's  
18 data that you can kind of assess are you in the ball park or  
19 not. I guess what I would look for is, are you  
20 qualitatively predicting the phenomena that go on. That's  
21 the first criteria. If you are not, clearly, you are out to  
22 lunch. I mean, you have to do something.

23 If you disagree with the data but you are  
24 qualitatively correct, then it's a matter of assessment of  
25 how important is this phenomena and perhaps in the end what

1 does it mean as far as reactor safety is concerned. If it's  
2 big, obviously there may be some concern. I think you have  
3 to make that connection.

4 MR. ZUBER: May I reply to this?

5 MR. CATTON: After I do.

6 MR. ZUBER: Since you are the Chairman.

7 MR. CATTON: I will take the Chairman's  
8 prerogative. I don't really understand the remarks that you  
9 just made. You pick a problem, where it looks like the  
10 entry length effects are probably pretty strong. Where that  
11 steam first sees the water the gradients are very steep and  
12 you compare with a Dittus Boelter.

13 I think you have to ask yourself the question as  
14 to why it looks so good. You have to ask the reverse  
15 question as well, because your correlation should not have  
16 done well for this. Why did it.

17 MR. RANSOM: I would argue that actually it did  
18 reasonably well, because the heat transfer coefficient which  
19 they calculated was approximately what is actually  
20 occurring. I think Bankoff measured the temperatures and  
21 the steam, the liquid flow rates, all the data were there.  
22 I don't think they got the right answer for the wrong  
23 reason.

24 I think the disagreement that you see -- there is  
25 some trend, but it's not a strong effect.

1 MR. CATTON: I wouldn't look at this as  
2 disagreement. I certainly would ask as to why the agreement  
3 is so good. Novak.

4 MR. ZUBER: I would like to agree with Vic in one  
5 respect. I don't think that they can predict everything  
6 perfectly. I think what we have to do in the system code is  
7 to predict reasonably well, important phenomena. In this  
8 sense, I agree with you. I don't exactly know how important  
9 subcooling boiling is for AP600.

10 This aside, what bothers me is the ROSA which has  
11 been going for years, at least for three years. Whenever  
12 this Committee brings up a question it's been dismissed, is  
13 it important, is it relevant to safety and so on and so  
14 forth. Two years later, three years later, we find that yes  
15 it's important, and they have a crash program to put better  
16 models. Which puts during that time we have a  
17 confrontational discussion, criticism and defense. When we  
18 start resolving the problem we don't have enough time and  
19 people around.

20 I think what is really needed is a change of  
21 attitudes. We are not really criticism, at least I don't  
22 think we are. On the other hand, if something is important  
23 one should have open eyes and open mind to accept it and  
24 address it. I agree with you, we cannot predict everything  
25 to the last digit. But, we have to be sure that what is

1 important is correctly addressed.

2 My experience during the last three years, we have  
3 been washing all this and all has been put under the rug. A  
4 given example is the condensation, thermal layers and now  
5 it's important. We brought this program three years ago.

6 MR. CATTON: I think that's a fair criticism.

7 MR. RANSOM: I guess I can't really address that  
8 issue since I haven't been involved in it that much. I  
9 think there are some legitimate concerns there. I would  
10 agree that today -- back in 1974 when we first developed  
11 this code we thought interfaces would be an important part  
12 of the problem. I, personally, worked on that for about a  
13 year and one-half. After getting nowhere basically because  
14 it is a tough problem, we decided it was more important to  
15 move ahead because we weren't seeing a lot of interfacial  
16 type of problems at that time. These were large break  
17 blowdowns, small breaks, et cetera.

18 Today, I think you are seeing the importance of  
19 that kind of phenomena in the kinds of systems we are  
20 dealing with today and the kinds of accidents. Clearly,  
21 there is a reason to go back and address that issue. I  
22 would agree with that and support that. I think this  
23 Committee's concerns should be important.

24 MR. CATTON: Gary, it's back to you.

25 MR. SHOTKIN: The only thing I would like to point



1 out is, I agree with Dr. Zuber. But I think we have agreed  
2 with him for three years. I think we have pointed out in  
3 1990 or 1991 that we needed improvements in condensation  
4 modeling, we needed improvements in level tracking, thermal  
5 stratification, the list that you see that's being worked  
6 on. I don't disagree.

7 I just agree, that you are not giving us any  
8 credit for having identified it.

9 MR. SCHROCK: I would like to comment on that. I  
10 made the point at the conclusion of my review of the  
11 documentation that the interfacial heat transfer package is  
12 weak. I am convinced, the interfacial heat transfer package  
13 is weak, and it would not be hard to make substantial  
14 improvements. If we go away from here with that view having  
15 been suppressed and NRC management contention that it is a  
16 phenomena in which the integral performance of the system  
17 has low sensitivity to the heat transfer coefficients, we  
18 are going to go on with the same low quality interfacial  
19 heat transfer package.

20 I don't think that's in the agency interest. I  
21 think you ought to be looking at it realistically and  
22 recognizing that it's not hard to fix most of what is weak  
23 in the interfacial heat transfer package and it ought to be  
24 done.

25 There may in fact arise situations in which there

1 is a significant dependence upon interfacial heat transfer  
2 package that you do not recognize today, that nobody in this  
3 room recognizes today. You won't be prepared to cope with  
4 it if you go on with this weak interfacial heat transfer  
5 package.

6 MR. SHERON: I feel compelled to respond. As  
7 someone who has to sit here and make decisions on how to  
8 allocate finite resources in the face of declining budgets,  
9 I just got hit with a \$3.7 million budget cut. I have to  
10 figure out where to take it. I was just told that certain  
11 other areas in my division are high priority on the agency's  
12 agenda as well as thermal hydraulics.

13 I have asked repeatedly of the Subcommittee, if  
14 they could please tell me what the safety significance is of  
15 all of these things that you desire us to do, to make this  
16 code academically pure, if you don't mind me using that  
17 term. All I have heard is that this could bite you in the  
18 rear end sometime in the future. It may be a problem, we  
19 don't know. It's your responsibility.

20 That sort of says that we have to do everything  
21 and prove it out before anybody's happy. I have said  
22 before, we have to make engineering judgments a lot of times  
23 on a lot of this stuff in order to get the job done. I get  
24 bothered by the fact that somehow engineering judgment is  
25 not acceptable.

1 I have asked the Committee, tell me what are the  
2 priorities within the Office of Research that you think we  
3 should be spending money on. Should I be putting unlimited  
4 funds to solve all of these problems on the RELAP code when  
5 in fact I can show you that the risk is not even dominated  
6 by this area. I have a hard time with that.

7 As a consequence, I have to make a decision that  
8 maybe we are not going to be putting the resources on that,  
9 and then I have to come down and continuously get criticism  
10 because we get accused of not addressing something.

11 MR. CATTON: I think you are missing a point,  
12 Brian. I think in this particular case what Virgil says is  
13 true. If you would just say look, I understand and I  
14 believe and Virgil is right, but I don't have the money to  
15 spend on it. Finished. To me, that's okay.

16 What gets perplexing is, when these things are  
17 pointed out and then huge arguments ensue as to whether or  
18 not this thing is weak or not. If you take a look at the  
19 table it's filled with ten to the four. To me, I agree with  
20 Virgil. That says that a lot of the basis for this is weak.  
21 If you decide a lot of this is weak but we don't think it's  
22 very important furthermore, I agree it's weak but I am not  
23 going to spend the money because I don't have it.

24 I think that's a position that you can take  
25 because you are the one who is spending the money. Don't

1 tell us that it's not weak when it is. That's where we get  
2 into this kind of a back and forth, is when they hold up the  
3 ten to the four and say that's fine, it's plenty robust.

4 MR. SHERON: There is just a difference of opinion  
5 then, okay, if that's the case.

6 MR. CATTON: I don't know how to deal with that.

7 MR. ZUBER: Let me add something else. At least  
8 two years ago the problem of thermal stratification, level  
9 tracking and condensation was brought up. You made the same  
10 argument, tell me what is important. Two years later we  
11 find out that yes, it is important and now trying to do a  
12 good engineering job in six months, which is very hard.

13 MR. SHOTKIN: Novak, those have been in our plans  
14 for two years.

15 MR. ZUPER: I shall go through --

16 MR. SHOTKIN: I will comment after you are done.

17 MR. ZUBER: I shall go through the minutes of the  
18 meetings and report that these comments were dismissed by  
19 the staff, and they are on the same grounds. It is being  
20 shown, and we have to address it two years later.

21 MR. SHOTKIN: I will show you our plans, Novak.  
22 Those models were in our plans for at least two years that I  
23 know of.

24 MR. CATTON: Why don't you continue, Gary.

25 MR. JOHNSEN: Okay.

1 MR. SHOTKIN: Gary, could you comment on the ten  
2 to the fourth that Professor Catton brought up. Would you  
3 explain that.

4 MR. CATTON: It's not one, it's eight of them.

5 MR. SHOTKIN: Yes. Could you explain what that  
6 is.

7 MR. JOHNSEN: You will notice that most of them  
8 cover the metastable conditions. I mentioned in the  
9 preceding slide that of the 24 that cover the stable  
10 conditions, 18 of them are at the literature and six are  
11 engineering judgment numbers.

12 MR. CATTON: Fourteen, I guess, if I include the  
13 three times ten to the six.

14 [Slide]

15 MR. JOHNSEN: This is the low flow case. This is  
16 the higher flow case. You can see that the match is pretty  
17 good, despite the fact that some would believe that the  
18 Dittus Boelter is not applicable.

19 MR. WULFF: But the fact is that this is a  
20 combination of coefficients that you present here, and we  
21 cannot decide whether Dittus Boelter is applicable or not on  
22 the basis of the agreement on these. I have said this  
23 before, and you keep repeating that this tells you that  
24 Dittus Boelter is applicable.

25 MR. JOHNSEN: What I am saying is that the

1 correlation is producing a heat transfer coefficient that is  
2 obviously leading to a reasonable result. That's what I am  
3 saying.

4 MR. WULFF: Along with other things you cannot --  
5 compensating errors.

6 MR. JOHNSEN: Is this the right answer for the  
7 wrong reason again?

8 MR. CATTON: We don't know.

9 [Slide]

10 MR. JOHNSEN: Bankoff also measured the liquid  
11 temperature exited of the rectangular duct. This is a  
12 comparison of the liquid temperature at the end, after  
13 having condensed most of the steam. The temperature  
14 comparison is not bad, it's reasonable.

15 MR. CATTON: Is this counterflow?

16 MR. JOHNSEN: Cocurrent, stratified flow.

17 [Slide]

18 MR. JOHNSEN: That's the stratified experiment.  
19 This is the completely opposite kind of experiment. This  
20 one is by Aoki, looking at steam being injected at one end  
21 of a test section and a spray of liquid in the middle of the  
22 test section, and watching the condensation of the steam on  
23 the droplets. Again, this is a rectangular/horizontal test  
24 section run at atmospheric temperature.

25 The steam coming in at 100 degrees C just

1 saturated at atmosphere, the liquid flow at the center of  
2 the test section being introduced as a spray, and the water  
3 being pretty much at room temperature. So now, we are going  
4 to have dispersed flow rather than stratified flow,  
5 droplets.

6 [Slide]

7 MR. JOHNSEN: This is the diagram of the  
8 experiment, steam coming in at one end from the time  
9 dependent volume, liquid being introduced toward the center  
10 of the test section, again, on a spray form, and atmosphere  
11 on the other end.

12 MR. SEALE: That arrow has to be wrong.

13 MR. JOHNSEN: Yes, it should be pointing the other  
14 way.

15 MR. SEALE: Okay.

16 [Slide]

17 MR. JOHNSEN: What Aoki did was to measure the  
18 pressure along the test section, and you can see that what  
19 occurs is a pressure depression in the vicinity of the  
20 injection point that the code qualitatively matches but  
21 doesn't quantitatively agree real well, let's put it that  
22 way.

23 The characteristic of this pressure suppression  
24 and recovery downstream is captured. There are some three  
25 dimensional effects going on here that the 1-D code just

1 can't capture. The basic trend is --

2 MR. SCHROCK: Is this test section horizontal or  
3 vertical?

4 MR. JOHNSEN: Horizontal. Now, I would like to  
5 turn to the last item on --

6 MR. WULFF: I have a question. You say on this  
7 previous viewgraph, assess condensation model for this flow.  
8 Would you show us the -- it shows pressure. Why is pressure  
9 the thing that you show us. I don't think it's very  
10 sensitive to vapor generation rate. How sensitive is it.  
11 What I would really like to see is the objective that you  
12 list here, plotted versus experiments.

13 MR. JOHNSEN: We would like to have done that too,  
14 but this is the measurement that was made. It is an  
15 indirect indication of condensation in this particular  
16 situation, where you have droplets being entrained in  
17 saturated steam.

18 MR. CATTON: All they measured was the pressure  
19 distribution?

20 MR. JOHNSEN: That's all they measured.

21 MR. WULFF: Then, it was inadequate experiment to  
22 use for the purpose.

23 MR. JOHNSEN: These kinds of experiments are quite  
24 difficult to find.

25 MR. SCHROCK: Why should there be a discrepancy



1 upstream of the injection point.

2 MR. JOHNSEN: The local pressure suppression in  
3 the calculation is causing the flow upstream to be greater  
4 in the code, therefore, the pressure dropped from the  
5 entrance to the over stated.

6 MR. SCHROCK: What does Aoki control in his  
7 experiment, upstream pressure or inlet flow rate.

8 MR. JOHNSEN: The inlet flow rate of both steam  
9 and the liquid.

10 MR. CATTON: So, at zero which is where the  
11 injection is, there's a lot of multi-dimensional  
12 effect?

13 MR. JOHNSEN: Yes. There are also multi-  
14 dimensional effects that we can't capture.

15 MR. CATTON: So, it's probably causing a barrier  
16 and effective constriction and that gives them the larger  
17 pressure drop and you don't calculate that. I understand.

18 MR. SEALE: How long was Aoki's section?

19 MR. JOHNSEN: I think this is 60 centimeters.

20 MR. SEALE: It must be a misprint on the --

21 MR. JOHNSEN: It says 2.86 centimeters by 60.

22 MR. SEALE: That's centimeters, again.

23 MR. JOHNSEN: Centimeters. It's a mistake on that  
24 first slide. It says meters, and it should have said  
25 centimeters.

1 MR. CATTON: Sixty meters would be pretty --

2 MR. SEALE: Yes. I guess it also depends on how  
3 directed the spray really is.

4 MR. JOHNSEN: Right.

5 [Slide]

6 MR. JOHNSEN: I want to turn to the last item I  
7 had on my outline slide, which is a discussion of the mass  
8 transfer of model, in particular an explanation of the term  
9 "gamma W" which was questioned in the comments as being  
10 confusing, and that there was no such thing as vapor  
11 generation at the wall. That observation is absolutely  
12 correct.

13 The total vapor generation model in RELAP consists  
14 of two parts, vapor generation the bulk, the first term  
15 here, and vapor generation in the vicinity of the wall. In  
16 the bulk this vapor generation term is given by this  
17 expression here. These, again, are the interfacial heat  
18 transfer coefficients. This is the jump condition which  
19 depends on the thermodynamic states of the phases.

20 These are discussed in Volume 1 and in Volume 4.  
21 I think most of the confusion arose from reading about gamma  
22 W in the section that had the conservation equations, where  
23 it was not completely described as it was elsewhere in the  
24 documentation. So, again, gamma W<sub>j</sub> then stands for the mass  
25 transfer at the liquid vapor interface that is near the wall

1 of a heat structure J, attached to the volume. For every  
2 heat structure that is attached to a control volume we have  
3 the potential for a term of vapor generation.

4 This goes back to what I said earlier, that it is  
5 conceivable to have a control volume that has a cold heat  
6 structure in it and a hot one, and you could have boiling at  
7 one and condensation on the other. That would be reflected  
8 in the sum of these terms.

9 The reason this approach was used is because as  
10 Joe Kelly pointed out earlier this morning, RELAP has only  
11 one liquid temperature. It does not calculate the fine  
12 details involving thermal gradings near the wall and some  
13 other approach must be used, and that's why we used this  
14 approach. A perfect example of where this is important is,  
15 again, subcooled boiling, where the bulk liquid can be  
16 subcooled but there is vapor generation near the wall which  
17 may or may not persist out into the bulk.

18 [Slide]

19 MR. JOHNSEN: This is the diagram that Joe showed  
20 this morning. He clarified it better than I did, by drawing  
21 a line down here to indicate a superheated layer next to the  
22 wall. The idea behind the diagram was to indicate that the  
23 overall vapor generation rate is the sum of what goes on in  
24 the bulk plus what occurs at the wall.

25 In this particular case as would be the case in

1 the Christensen experiment, we have subcooled liquid.  
2 Therefore, in the bulk there's condensation going on.  
3 Therefore, this is negative, the vapor generation near the  
4 wall. This is positive. In the case of subcooled boiling  
5 the sum of those two is greater than zero, so we do get a  
6 net generation of vapor.

7 MR. DHIR: For calculating gamma  $W_j$ , do you just  
8 use the Chen correlation and convert that into --

9 MR. JOHNSEN: No. In the case of boiling that is  
10 the so-called Gamma  $W_j$ , is where that's going.

11 MR. DHIR: Near the wall, gamma  $W_j$ .

12 MR. JOHNSEN: That's right. That's partitioned.  
13 Some of it goes into vapor generation and some of it goes  
14 into the --

15 MR. DHIR: What partition, on what basis?

16 MR. JOHNSEN: That's Leahy's subcooled boiling  
17 model, for the prediction of vapor generation and subcooled  
18 boiling.

19 MR. DHIR: That is combined with Chen's  
20 correlation then.

21 MR. JOHNSEN: That is correct.

22 MR. SCHROCK: This goes back historically to the  
23 work done at -- I am trying to remember the fellow's name -  
24 - on the subcooled boiling. The modeling was not in the  
25 context of two-phased multi-fluid models at that time. The

1 reference that you are making precedes this form of two-  
2 phased modeling.

3           What you have done is to borrow ideas out of the  
4 older modeling in which there was a vapor generation in the  
5 bubbly layer close to the wall. That was not identified as  
6 vapor issuing from the wall. There is a vapor generation  
7 within the liquid near the wall.

8           Now, you have come along and introduced the multi-  
9 fluid formulations. You have an interfacial heat transfer,  
10 you have an interfacial area description, but you choose to  
11 isolate some unidentified part of the interfacial area and  
12 say we are going to call this a wall vapor source. That's  
13 basically what you have done. You don't clearly identify  
14 what is the amount of the interfacial area per unit volume  
15 in your computational cell. Instead, you are treating it as  
16 though it is literally vapor issuing away from the wall.

17           That puzzles you, I can see. But I think this is  
18 the source of your difficulty in describing it. It's true,  
19 you have used a correlation from Leahy subcooled boiling  
20 model, but that subcooled boiling model is not structured in  
21 the form of your two fluid model used in RELAP5. You have  
22 somehow made a determination that that particular part of it  
23 can be adapted for your two fluid model. It's not clear how  
24 you accomplish that adaptation. This is the difficulty that  
25 you are having, I think.

1 MR. JOHNSEN: I think the adaptation simply says  
2 that a portion of the heat transfer is going into vapor  
3 generation, and a portion is going into heating a liquid.

4 MR. SCHROCK: But you have interface on all the  
5 bubbles that are present. There are some bubbles that are  
6 growing and there --

7 MR. JOHNSEN: Once they are generated, yes. Once  
8 they are generated, then they are --

9 MR. SCHROCK: Once they are nucleated.

10 MR. JOHNSEN: Yes.

11 MR. SCHROCK: Once they are nucleated. They grow,  
12 as a consequence of energy flow to the interface out of the  
13 liquid.

14 MR. JOHNSEN: Right.

15 MR. SCHROCK: The physics of that is not being  
16 described, at least in the description of the documentation  
17 that you are using. You have the problem of how do you  
18 average over computational cell, interface which is involved  
19 with condensation and interface which is involved in  
20 evaporation. That's what you need to be partitioning  
21 somehow. I don't think you know how to do it.

22 It isn't correctly determined simply by taking  
23 wall heat flux and heated vaporization to determine a vapor  
24 source. It ends up being per unit volume of computational  
25 cell in a totally artificial way.

1           MR. JOHNSEN: Again, this is a situation where you  
2 are trying to take the two fluid model and stretch it into a  
3 regime that is indeed more complex than the basic model to  
4 begin with. We think this is a reasonable approach to the  
5 situation of subcooled boiling. The results we have  
6 achieved seem to bear that out.

7           Are you suggesting that we somehow within the  
8 context of the two fluid model, take the position that  
9 there's a certain amount of interfacial area associated near  
10 the wall and a separate --

11           MR. SCHROCK: I said that there is interfacial  
12 area in a computational cell, some of which is associated  
13 with evaporation process simultaneously some of which is  
14 associated with the condensation process.

15           MR. JOHNSEN: I agree with that.

16           MR. SCHROCK: The change in the vapor content is a  
17 result of the net process in that computational cell.

18           MR. JOHNSEN: Yes, I agree.

19           MR. SCHROCK: It's not a source of vapor at the  
20 wall.

21           MR. RANSOM: I think Virgil's view is absolutely  
22 right, as a matter of fact. It would be better if they  
23 talked about it in that same way. The way you are trying to  
24 do it is model the vapor which is produced in the wall layer  
25 in the subcooled boiling situation. I think the only part

1 of the idealization that missing is when they create vapor  
2 it's the net vapor coming out of that layer. They are not  
3 trying to model the bubbles that exist within that layer.

4           Within the context of the structure it immediately  
5 begins trying to condense those bubbles. It interfaces then  
6 with the bulk mixture. Really, it's like talking about the  
7 vapor which is now, after net vapor generation, is being  
8 transmitted from the wall layer to the bulk. It's the heat  
9 exchange mechanism that carries the energy from the wall  
10 into the bulk.

11           Obviously, they could do better in terms of  
12 documenting that and clarifying that, and I think that needs  
13 to be done. It's that kind of view that was the basis of  
14 that model.

15           I will make one other comment with regard to these  
16 heat transfer coefficients. I think this Committee was a  
17 member -- many of the people were -- remember MOD 0 and MOD  
18 1, we only used one energy equation. In actuality I think I  
19 can take credit for that approximation. From early days I  
20 thought if you look at situations it's very hard to find a  
21 situation where both phases are away from saturation.

22           Generally, the Committee agreed that it was a good  
23 approximation. The problem came that we couldn't model  
24 subcooled boiling, we couldn't model the post-CHF heat  
25 transfer process, because those are two examples where each



1 phase exists away from the saturation point. The logic just  
2 became untenable.

3 When you look at these large coefficients they  
4 have in the table what those really are is, they are saying  
5 from an engineering approximation point of view you have  
6 assumed that that phase is essentially saturated. I don't  
7 believe the database exists actually, to do a whole lot  
8 better than that.

9 They have in a couple of cases in subcooled  
10 boiling post-CHF heat transfer. You do find mechanistic  
11 modeling of the heat transfer from both phases to the  
12 interface, but many of these other cases is really a  
13 problem. What these documents should be saying is, they  
14 have essentially from an engineering judgment point of view  
15 assume that in that regime that phase exists near the  
16 saturation point. It also exists in a metastable state, so  
17 it's tending to relax to the stable state.

18 You are always moving -- the physics are moving  
19 you in the correct direction. Whether at the correct rate  
20 or not, it's difficult to tell.

21 MR. SEALE: Could I make a comment?

22 MR. CATTON: Yes.

23 MR. SEALE: Having not been a party to those other  
24 decisions and so on, I would also observe that when one made  
25 a model which was based on a single energy equation, that

1 model clearly had limitations and they were understood  
2 reasonably well by all.

3           When one finds that one can no longer use a single  
4 energy model and you begin to have to model things in more  
5 detail, I think it's important that we still recognize that  
6 some of the models we evolve are better in quality than some  
7 of the other models. Furthermore, from a practical point of  
8 view, we should recognize that some things need to be  
9 modeled better than other things do. Indeed, that's the  
10 decision that the management people have to make, where you  
11 spend your bucks.

12           The problem is, there seems to be a tendency to  
13 want to defend the validity of all models with equal  
14 tenacity. That's clearly not the case. The problem is that  
15 when you do that and you find yourself five years down the  
16 road having to look at the situation again because now the  
17 situation is a little bit different than it was the last  
18 time you fixed it, you have damned the people that have that  
19 job five years from now to go through all of the detail  
20 again instead of giving them a list that says this is a  
21 pretty good model. This one isn't, but it's as good as it  
22 needs to be.

23           Do you see what I am suggesting?

24           MR. CATTON: I think that's absolutely right.

25           MR. RANSOM: I can't disagree with that. I think

1 the thing that tends to be missing is documentation.

2 MR. CATTON: There is that too, but the defense of  
3 poor models leads to lively discussion and ill feelings. I  
4 think it would be better just to lay it out. As long as you  
5 are mentioning history and you referred to this Committee,  
6 this Committee made strong recommendations about the need  
7 for an experimental database for two fluid modeling and that  
8 never came about. That's why you don't have the data that  
9 you are looking for.

10 Any time that you decide that you are going to  
11 take something and model it in more detail you have to ask  
12 yourself if you have the information to do it.

13 MR. RANSOM: That's the message that I basically  
14 get, there's plenty of blame to go around.

15 MR. CATTON: You bet. I am just giving a little  
16 of it back. Gary.

17 [Slide]

18 MR. JOHNSEN: I am just about done here.

19 MR. CATTON: We are going to finish at 3:30.  
20 That's when these guys said they had to run.

21 MR. JOHNSEN: I am on my last slide.

22 MR. CATTON: Maybe we are going to finish at 3:15.

23 MR. JOHNSEN: There are some other presentations  
24 that we haven't gotten to.

25 MR. CATTON: I understand that. We had decided

1 that we would go until we got to 3:30 and quit, because I  
2 lose some of the people.

3 MR. JOHNSEN: Again, I just wanted to summarize  
4 this particular presentation by acknowledging that we do  
5 have simplifications in the code, idealizations, engineering  
6 approximations, which are subject to question. I think it  
7 was correctly observed by Professor Seale that these are  
8 situational judgments, what's good yesterday might not be  
9 good tomorrow.

10 I showed you yesterday that in the CMT behavior  
11 that we saw that there were a number of instances where we  
12 have work to do. The current code is not going to do the  
13 job for that situation. Whereas, we didn't have a similar  
14 situation with the current generation reactor. These  
15 judgments are situational in nature.

16 That's why you have to go back to the code  
17 applicability question every time you have a new design or a  
18 new transient that you have never done before. The  
19 philosophy is that once you have identified the design and  
20 the transient state that you want to run and you test your  
21 code, test your assumptions, you test your approximations  
22 using relevant data, until you reach a level of  
23 acceptability in the results. That's the basic philosophy.

24 I think it goes to the question of how good is  
25 good enough. I think the answer to how good is good enough

1 is that there's a situational decision that has to be made.

2 MR. CATTON: I would agree with that.

3 MR. JOHNSEN: Condensation was not very important  
4 in the last generation of reactors.

5 MR. ZUBER: Absolutely right.

6 MR. JOHNSEN: It is very important in the current  
7 generation.

8 MR. ZUBER: Absolutely right. That was pointed  
9 two years ago or three years ago. The same thing with the  
10 certification, the same thing with the tracking. These were  
11 really not important for the large break LOCA and we didn't  
12 address them in detail, and they were successful.

13 If you can model them correctly we are scot free.  
14 If not, we are open to questioning. The point is, this was  
15 not known in August and this work on condensation started.  
16 It was known two and three years ago. Whether it went to  
17 the large code previous generation, you knew the  
18 shortcomings of this codes. They were not able to address  
19 the problems of tomorrow or of today.

20 MR. JOHNSEN: The planning on the code development  
21 was begun in 1992. We didn't begin working on AP600 or SBWR  
22 until September, 1992. At the time that we had the last  
23 meeting in Idaho Falls, we had been working on the advanced  
24 reactors for less than six months. There was more work  
25 that went back earlier than that, but not code development.

1 MR. CATTON: I have INEL reports that are 1991.

2 MR. JOHNSEN: I am talking about code development  
3 work now.

4 MR. ZUBER: Let me just say, I don't know when you  
5 started. I just ask questions on the thing that started in  
6 August and I think on the thermal stratification a month  
7 ago. Those are really the essential problems to be  
8 addressed for this reactor, and they are addressed within  
9 the last six months.

10 MR. LAUBEN: Schedule, it looks like we were  
11 pretty late. I don't think there was any intention of  
12 dragging our heels. I remember the first time I got  
13 involved and submitted a presentation to the Committee was  
14 March of 1992. That was a year ago, before the previous  
15 March. We discussed these things.

16 Not only did we know these things were a problem  
17 but we ran into them as problems when AP600 was first  
18 analyzed in early 1992 and late 1991.

19 You may criticize us for going slowly, but I don't  
20 think we were trying to sweep those things under the rug,  
21 nor did we not recognize that they were important. In fact,  
22 this is now the third iteration on condensation heat  
23 transfer. Maybe we didn't apply a sufficient enough effort  
24 but we certainly were attempting to address it within the  
25 context of the funding and resources that we had.

1 MR. CATTON: I think with that, I would like to -  
2 - we only have 15 more minutes before these other guys  
3 leave. I think we will just hold the three that we missed,  
4 critical flow -- unless critical flow can be done in ten  
5 minutes which I don't think so, not with Virgil.

6 MR. ZUBER: I think this is something that should  
7 be done carefully.

8 MR. CATTON: I understand that. What we will have  
9 to do is -- Norm, you said something about the results of  
10 PIRT, and that would be something that would be a good topic  
11 for us to start digging into.

12 MR. LAUBEN: Yes.

13 MR. CATTON: It could be, if we do that reasonably  
14 soon we could couple these other two that we skipped,  
15 critical flow and -- really, it's just critical flow.

16 MR. ZUBER: Momentum.

17 MR. CATTON: We kind of talked about momentum  
18 equations. We could bring those two in with the discussion  
19 of the PIRT results. I think you indicated that they would  
20 be available in June. I don't want to wait for the final  
21 report. I would like us to interact before the final  
22 report.

23 MR. LAUBEN: Gary, did you want to --

24 MR. WILSON: Yes. I would like to respond to  
25 that. By June, we would be willing to come in -- I think

1 BNL with the SBWR concurs with this -- we would be willing  
2 to come in for both the AP600 and the SBWR, and show you the  
3 approach and the plans to achieve the PIRT. By June, I do  
4 not believe we will be in a position to show you significant  
5 PIRT results because they will be interim results.

6 It will be more in the order of September or  
7 October when we would be able to come in and show you the  
8 results of the PIRT. I am perfectly happy to come in June  
9 or even maybe in May and show you the approach and show you  
10 the scheduling, but before September I don't think we will  
11 be in a position, either BNL or ourselves, to talk about  
12 PIRT results.

13 MR. CATTON: Norm, what do you think? There's a  
14 remote possibility that we could be helpful to you.

15 MR. LAUBEN: That's right. Listening to what Gary  
16 says, I think there's some interesting things to hear about  
17 the approach, understanding that this is work in progress  
18 and you would be looking at interim results. I would kind  
19 of leave it up to you. It may be fine to do this in the  
20 summer.

21 MR. ZUBER: I would really advise to do it as soon  
22 as possible. What is really distressing is that something  
23 which should have come at the very beginning of this program  
24 to identify what we have to do is coming almost at the same  
25 time the code is already available. I don't know who



1 approves what.

2 MR. CATTON: As long as it's iterative, I guess.

3 MR. ZUBER: The point being, the sooner we hear we  
4 can have a better dialogue. One more thing. I think what  
5 is also important is to address these two questions which  
6 remain. Also, establish a dialogue where we do not dismiss  
7 things which are not important as you prove this is  
8 important, prove this is important for safety, or us  
9 criticizing something which may not be important for the  
10 reactor. I think we have to establish a better working  
11 relationship. I think maybe this new meeting as soon as  
12 possible would help.

13 MR. CATTON: We can keep it on a technical --

14 MR. WILSON: You heard a very strong pitch  
15 yesterday from Ron Beelman about welcoming you to come on an  
16 informal basis whenever you choose, and we would love you to  
17 come in the office and see what's going on and talk to us on  
18 an informal basis. That same invitation exists for the PIRT  
19 work and for the AP600 PIRT work, and I believe Kumar would  
20 make the same offer.

21 Any of your consultants, any of the Subcommittee  
22 members that would like to come in at any time, we would  
23 love to have you. I think we could --

24 MR. ZUBER: There is a safety concern that I am  
25 concerned with here. This is the isometrical behavior. The

1 question is, are the conditions during a transient whether  
2 we lose a sink. I think it will be really worthwhile to  
3 address this questions early in the game, not to leave it  
4 for October or later on.

5 MR. CATTON: I agree with you.

6 MR. ZUBER: I think this could also be combined  
7 with PIRT, and do it at a very early stage so that we can at  
8 least discuss it and see what experiments we can use and how  
9 we can address that problem. I think this may be a safety  
10 issue. If not, we should put it --

11 MR. CATTON: We have both AP600 and SBWR. The  
12 question is already asked. This is getting to be pretty  
13 free wheeling. I think we can end the recording.

14 [Whereupon, at 3:20 p.m., the transcribed portion  
15 of the meeting concluded.]

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This is to certify that the attached proceedings  
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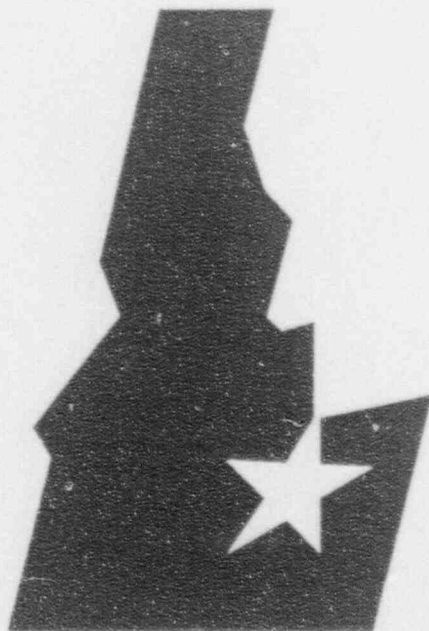
NAME OF PROCEEDING: ACRS Thermal Hydraulic Phenomena

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*Idaho  
National  
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Laboratory*

**RELAP5 Field Equations**

**Presented by**

**Walter L. Weaver III**

**ACRS Meeting**

**January 4-5, 1994**

## Summary of Presentation

- Documentation of Field Equations
  - Totality of documentation is rather complete but contained in several volumes and reports
- Momentum Equation Differencing
  - Use of the “non-conservative” form of momentum equation acceptable because the numerical momentum loss is small compared to physical losses

## Documentation Of Field Equations

- Most of RELAP5 documentation was directed towards code users.
- Intent of documentation was to state what was in code rather than show complete derivation.
- Volume IV was never intended as a complete derivation of the field equations but was intended to document the sources and details of the constitutive relations for interphase heat/mass transfer, wall heat transfer, interphase drag, and wall friction.
- Section in Volume IV on field equations was to show factors and terms that required closure to complete solution.

## Proposed Improvements To RELAP5 Documentation

- New RELAP5 volume.

An internal report, V. H. Ransom, "Course A-Numerical Modeling Of Two Phase Flows For Presentation At Ecole d'Ete d'Analyse Numerique," EGG-EAST-8546 (May 1989) develops the field equations used in Relap5 from basic conservation principles. ( available from NTIS as CONF-8906249-1. Also available at OSTI-Oak Ridge)

This report will be retitled and issued as an NUREG report.

- Volume I

Hydrodynamic section would start from equations developed in Ransom report and show more complete derivation of numerically convenient set of difference equations.

Increased discussions on numerical approximations.

- Volume IV

Material on derivation of field equations will be removed and Vol IV will repeat the final equations derived in Vol I

## Planned Improvements To RELAP5 Documentation

- Existing RELAP5 documentation is being converted to electronic form using document publishing software for technical material (Framemaker)
- Input requirements and some other text and equations have been converted.
- Intend to maintain 'living manuals'; that is manuals will be updated as code source is updated.



## Basis of Momentum Equation

- Momentum equations are written in “non-conservative” form for numerical convenience.
  1. Mass and energy conservation were considered paramount. A greater degree of approximation to momentum conservation considered acceptable
  2. Implicit evaluation used only in terms needed to exceed the sonic Courant limit
  3. Time level evaluations adjusted to make implicit terms linear
- A “conservative” form has errors of the same order as a “non-conservative” form

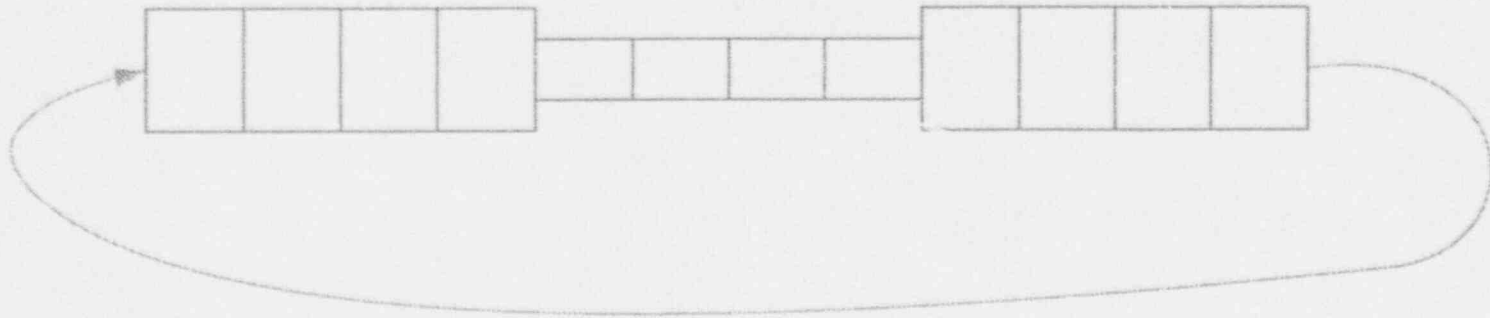
## Basis of Momentum Equation(cont.)

- Use of “non-conservative” momentum equation along with design principles reduces the system of field equations to a Poisson equation in pressure reducing the computation time spent in matrix solution.
- Major concern is the modeling of the momentum flux terms in the momentum equation

## Momentum Flux Modeling

- Consequences of inaccuracies in modeling of momentum flux term
  - artificial pressure losses in equations resulting in lower flow rates for a given momentum source (i.e. pump work)
  - incorrect pressure losses or recovery through area changes
  
- Several simple problems used to assess accuracy of momentum flux term modeling
  - frictionless loop with area changes and no momentum sources

## Loop Test Case Nodalization



Horizontal Pipe

50% Area Reduction

Single Phase Liquid or Vapor

Pressure = 50 bar

Liquid Velocity =  $\sim 3$  m/s

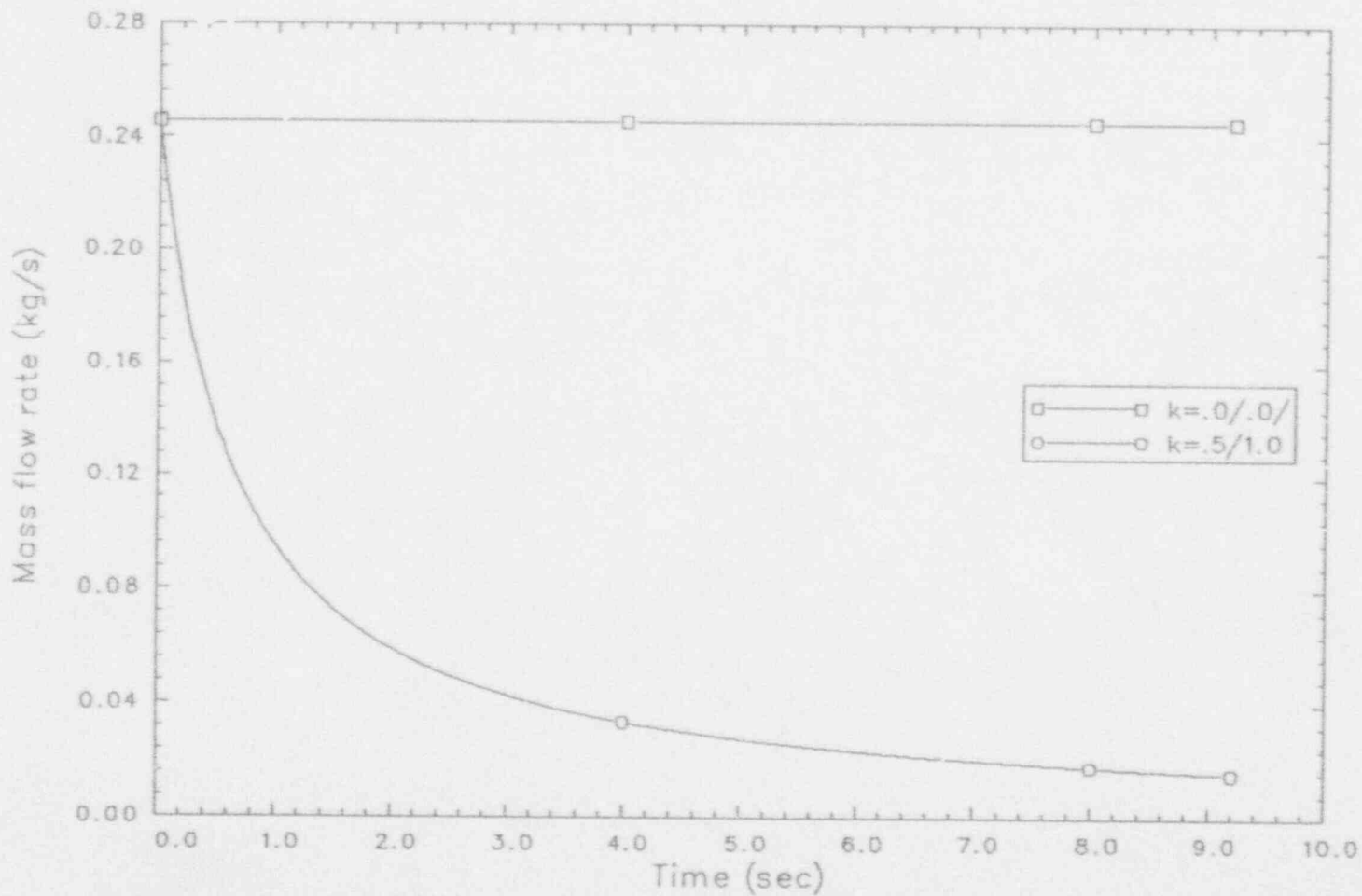
Vapor Velocity =  $\sim 100$  m/s

## Momentum Flux Test Cases

- Single phase liquid with and without irreversible pressure losses at the area changes
  1.  $K_{\text{cont}} = 0.0, K_{\text{exp}} = 0.0$
  2.  $K_{\text{cont}} = 0.5, K_{\text{exp}} = 1.0$
  
- Single phase vapor with and without irreversible pressure losses at the area changes.
  1.  $K_{\text{cont}} = 0.0, K_{\text{exp}} = 0.0$
  2.  $K_{\text{cont}} = 0.5, K_{\text{exp}} = 1.0$

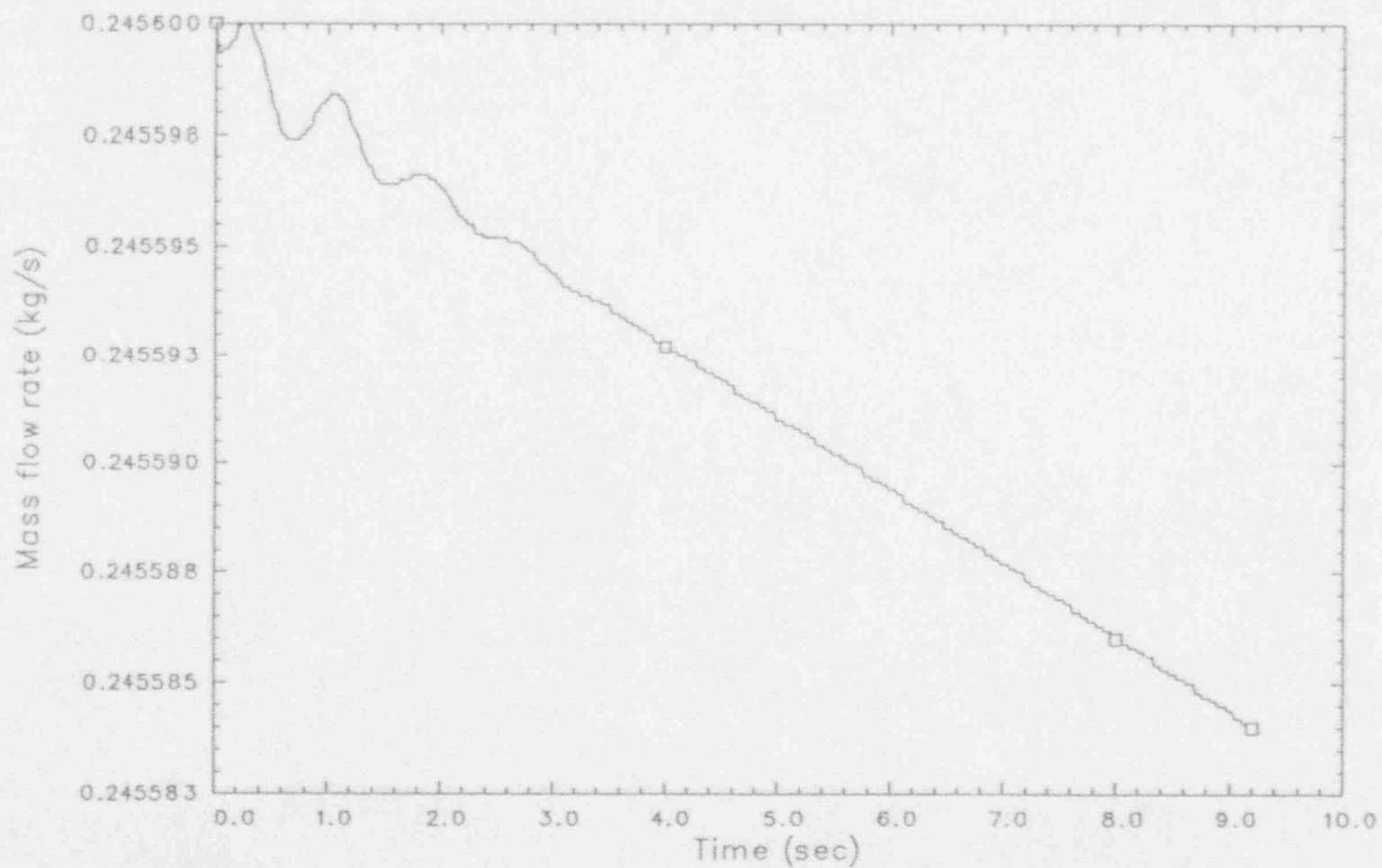
# Single Phase Liquid Test Case

Insignificant numerical losses relative to physical losses



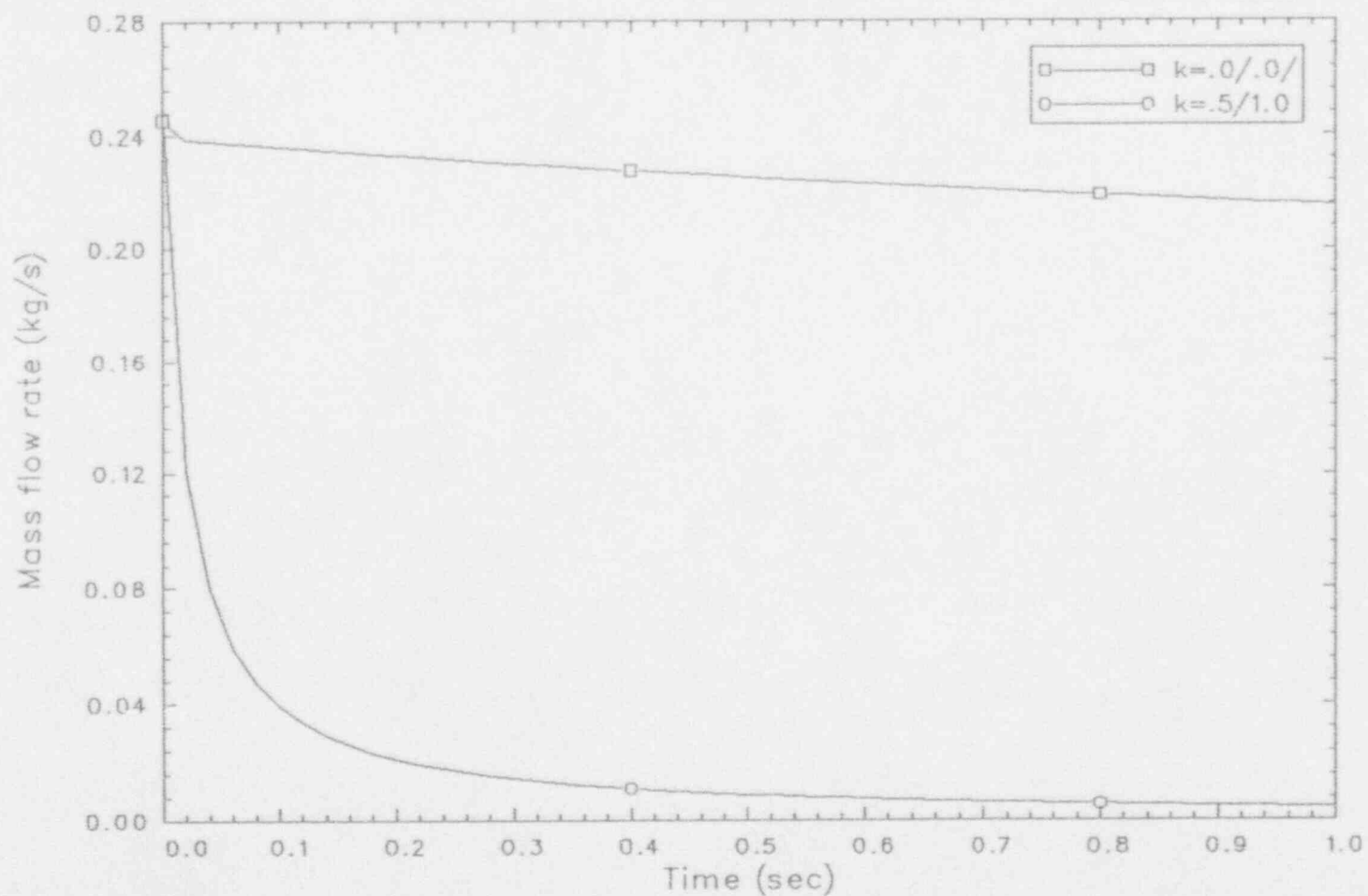
# Single Phase Liquid Test Case

Numerical losses in 6<sup>th</sup> significant place for single phase liquid



## Single Phase Vapor Test Case

Small numerical losses for single phase vapor relative to physical losses





## Summary of Momentum Flux Test Case Results

- Summary of test results
  - Loop problem
    1. Velocity remains constant to 5 significant figures during 2000 advancements for single phase liquid with no physical losses
    2. Velocity decays rapidly when physical losses introduced for single phase liquid.
    3. Small velocity loss for single phase vapor with no physical losses.
    4. The physical loss through an area change much larger than the numerical loss for single phase vapor

## Summary on Momentum Equation Modeling

- Simulation of a variety of simple “thought” problems, separate effects tests and integral tests have not uncovered any observable effects of the “non-conservative” form of the momentum equations

References for natural circulation test cases

Vol III of RELAP5/MOD3 manual reports results of LOFT L3-7 Small Break simulation with natural circulation as well as results of the simulation of the Semiscale Natural Circulation test series

P. R. McHugh and R. D. Hentzen, Natural Circulation Cooling in U.S. Pressurized Water Reactors, NUREG/CR-5769, EGG-2653, Jan 1992 shows results of simulations of BETHSY natural circulation tests as well as the Semiscale NC test series



*Idaho  
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# **RELAP5/MOD3 Interfacial Heat and Mass Transfer Modeling**

Presented by  
Gary W. Johnsen

Advisory Committee on  
Reactor Safeguards  
Bethesda, Maryland  
January 4-5, 1994

## Consultants' Comments Addressed

- Modeling of interfacial area is simplistic
- Interfacial heat transfer coefficients are ad-hoc
- Mass transfer "at the wall" confusing

**RELAP5 Interfacial Area Modeling uses idealizations that have proven to provide adequate results when tested using relevant experiments**

Regime

Bubbly

Slug

Annular mist

Mist

Stratified

Interfacial Geometry

Spherical bubbles, size distribution

Spherical bubbles & Taylor bubbles

Spherical drops & liquid film

Spherical drops

Quiescent surface

# Interfacial Heat Transfer Coefficient Modeling

- The combination of possible thermodynamic states and flow regimes requires the specification of 48 phasic heat transfer coefficients ( $h_{if}$ ,  $h_{ig}$ )
- For non-metastable thermodynamic states 18 of 24 heat transfer coefficients are from the literature
- Large coefficients are used for the metastable states (i.e., superheated liquid, subcooled vapor) to ensure they are transitory

Note: Volume 4 is in error.  $Nu \neq 10^4$  for bubbly SHG. Rather  $h_{ig} = 10^4 \text{ W/m}^2\text{K}$ , which was suggested by Bruker and Sparrow

## Summary of Interfacial Areas and Heat Transfer Coefficients

Flow Type	$a_{gf}$	$h_{if,SCL}$	$h_{if,SHL}$	$h_{ig,SHG}$	$h_{ig,SCG}$
Bubbly	$\frac{3.6 \alpha_{bub}}{d_b}$	Unal	Lee-Ryley Plesset-Zwick	$10^4 f(\Delta T_{sg})$	$10^4 f(\Delta T_{sg})$
Slug: Bubbles	$\frac{3.6 \alpha_{gs}(1-\alpha_{TB})}{d_b}$	Unal	Lee-Ryley Plesset-Zwick	$10^4 f(\Delta T_{sg})$	$10^4 f(\Delta T_{sg})$
Taylor bubble	$\frac{4.5}{D} \alpha_{TB}(2.0)$	Seider-Tate	$3 \times 10^6 f(\Delta T_{sf})$	Lee-Ryley	$10^4 f(\Delta T_{sg})$
Annular mist: Drops	$\frac{3.6 \alpha_{fd}}{d_d} (1-\alpha_{ff})$	Brown $f(\Delta T_{sf})$	$\frac{k_f}{d_d} f(\Delta T_{sf})$	Lee-Ryley	$10^4 f(\Delta T_{sg})$
Liquid Film	$\frac{4}{D} (1-\alpha_{ff})^{1/2}(2.5)$	Theofanous	$3 \times 10^6$	Dittus-Boelter	$10^4 f(\Delta T_{sg})$
Horizontal stratified	$\frac{4 \sin \theta}{\pi d}$	Dittus-Boelter	Dittus-Boelter $f(\Delta T_{sf})$	$10^4 f(\Delta T_{sg})$ Dittus-Boelter	$10^4 f(\Delta T_{sg})$
Vertical stratified	$\frac{A_c}{V}$	McAdams	$h_{if,REG}$	McAdams	$h_{ig,REG}$

SCL = subcooled liquid; SHL = superheated liquid; SHG = superheated gas; SCG = subcooled gas;

$f(\Delta T_{sg})$  = function of  $\Delta T_{sg} = T^s - T_g$ ;  $f(\Delta T_{sf})$  = function of  $\Delta T_{sf} = T^s - T_f$ ; REG = flow regime when not vertically stratified

## Summary of Interfacial Areas and Heat Transfer Coefficients (cont'd)

Flow Type	$a_{gf}$	$h_{if,SCL}$	$h_{if,SHL}$	$h_{ig,SHG}$	$h_{ig,SCG}$
Inverted annular: Bubbles	$\frac{3.6 \alpha_{bub}}{d_b} (1-\alpha_g)$	Unal	Lee-Ryley Plesset-Zwick	$10^4 f(\Delta T_{sg})$	$10^4 f(\Delta T_{sg})$
Vapor film	$\frac{4}{D} (1-\alpha_B)^{1/2} (2.5)$	Dittus-Boelter	$3 \times 10^6$	$\frac{k_g}{D} f(\Delta T_{sg})$	$\frac{k_g}{D} f(\Delta T_{sg})$
Inverted slug: Drops	$\frac{3.6 \alpha_{drp}}{d_d} (1-\alpha_B)$	Brown $f(\Delta T_{sf})$	$\frac{k_f}{D} f(\Delta T_{sf})$	Lee-Ryley	Lee-Ryley
Taylor drop	$\frac{4.5}{D} (\alpha_B) (2.5)$	Brown $f(\Delta T_{sf})$	$\frac{k_f}{D} f(\Delta T_{sf})$	$\frac{k_g}{D} f(\Delta T_{sg})$	$\frac{k_g}{D} f(\Delta T_{sg})$
Mist (dispersed)	$\frac{3.6 \alpha_{drp}}{d_d}$	Brown $f(\Delta T_{sf})$	$\frac{k_f}{D} f(\Delta T_{sf})$	Lee-Ryley $f(\Delta T_{sg})$	$10^4 f(\Delta T_{sg})$

SCL = subcooled liquid; SHL = superheated liquid; SHG = superheated gas; SCG = subcooled gas;

$f(\Delta T_{sg})$  = function of  $\Delta T_{sg} = T^s - T_g$ ;  $f(\Delta T_{sf})$  = function of  $\Delta T_{sf} = T^s - T_f$ ; REG = flow regime when not vertically stratified



# Experiments Used to Validate RELAP5 Interphase Heat and Mass Transfer Models

<u>Experiment</u>	<u>Model Assessed</u>
Edwards Pipe Blowdown	Vapor generation model in the bulk, $H_{if}$ superheated liquid in bubbly flow
Christensen Subcooled Boiling Test 15	Vapor generation model near the wall, condensation model in the bulk, $H_{if}$ for subcooled liquid in bubbly flow
MIT Pressurizer Test ST4	Wall condensation model near the wall, condensation model in the bulk, $H_{if}$ for subcooled liquid in vertical stratified flow
ORNL THTF Void Profile Test 3.09.10i	Vapor generation model in flow regimes for void fraction 0 $\rightarrow$ 1

# Experiments Used to Validate RELAP5 Interphase Heat and Mass Transfer Models (cont'd)

<u>Experiment</u>	<u>Model Assessed</u>
ORNL Bundle CHF Tests 3.07.9B, 3.07.9N, 3.07.9W	Vapor generation model in flow regimes for void fraction 0 $\rightarrow$ 1
Swedish Royal Institute of Technology Heated Tube CHF Test 261	Vapor generation model in flow regimes for void fraction 0 $\rightarrow$ 1
Bennett's Heated Tube CHF Tests 5358, 5294, and 5394	Vapor generation model in flow regimes for void fraction 0 $\rightarrow$ 1
Bankoff Cocurrent Flow Condensation Tests 253 and 279	Condensation model in the bulk, $H_{if}$ for subcooled liquid in horizontal stratified flow
Aoki Steam-Water Mixing Experiment	Condensation model in the bulk, $H_{if}$ for subcooled liquid in mist (dispersed) flow

# Experiments Used to Validate RELAP5 Interphase Heat and Mass Transfer Models (cont'd)

<u>Experiment</u>	<u>Model Assessed</u>
Chen Heated Tube Post CHF Tests 174-93 and 318-84	Mass transfer model at high void fraction
Neptunus Pressurizer Test Y05	Condensation model in the bulk, $H_{if}$ for subcooled liquid in vertical stratified flow
Semiscale Pressurizer Insurge Test	Condensation model in the bulk, $H_{if}$ for subcooled liquid in vertical stratified flow

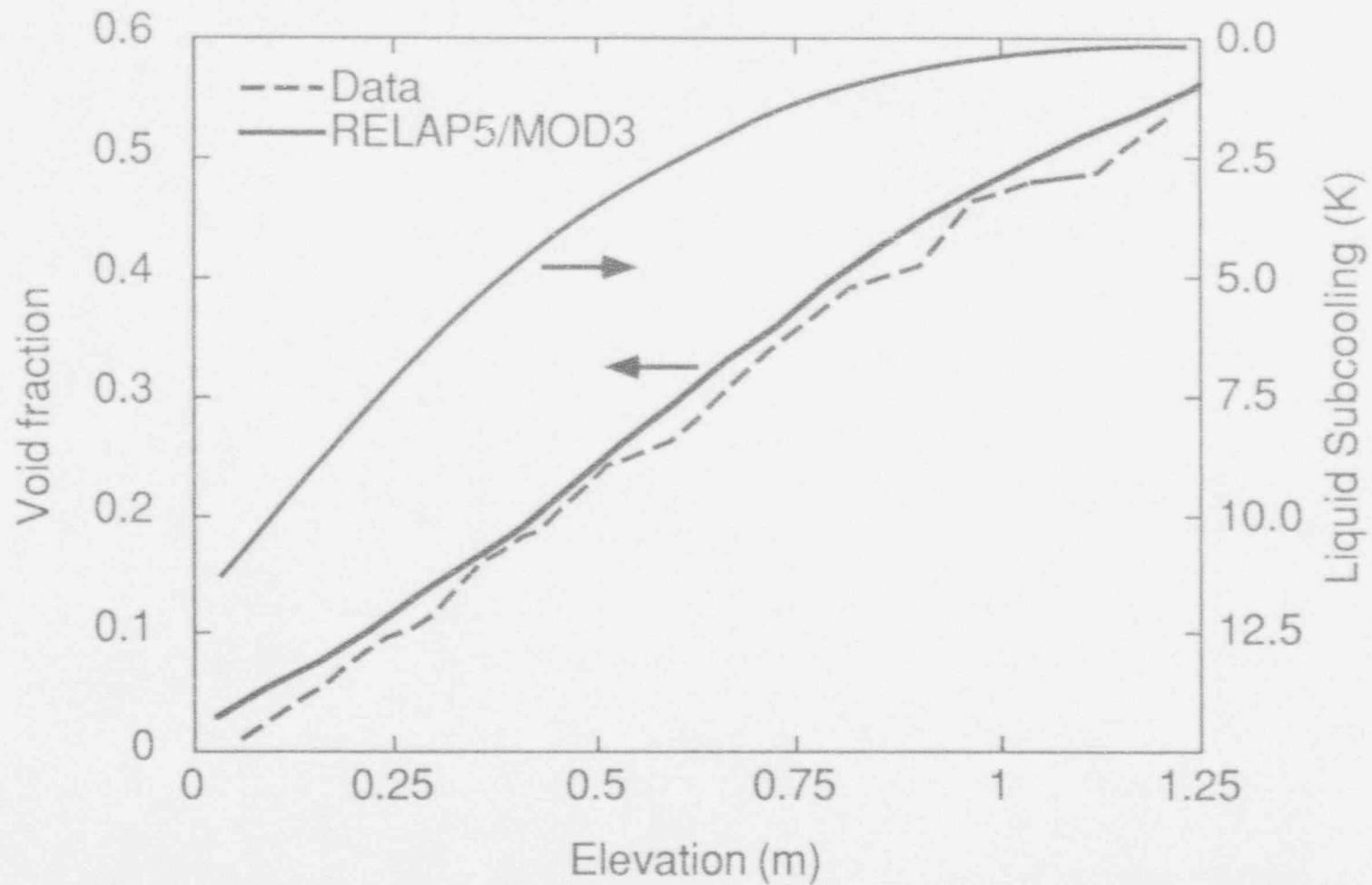
# Christensen Subcooled Boiling Experiment 15

Objective: Assess interphase mass transfer and subcooled nucleate boiling model

Description:

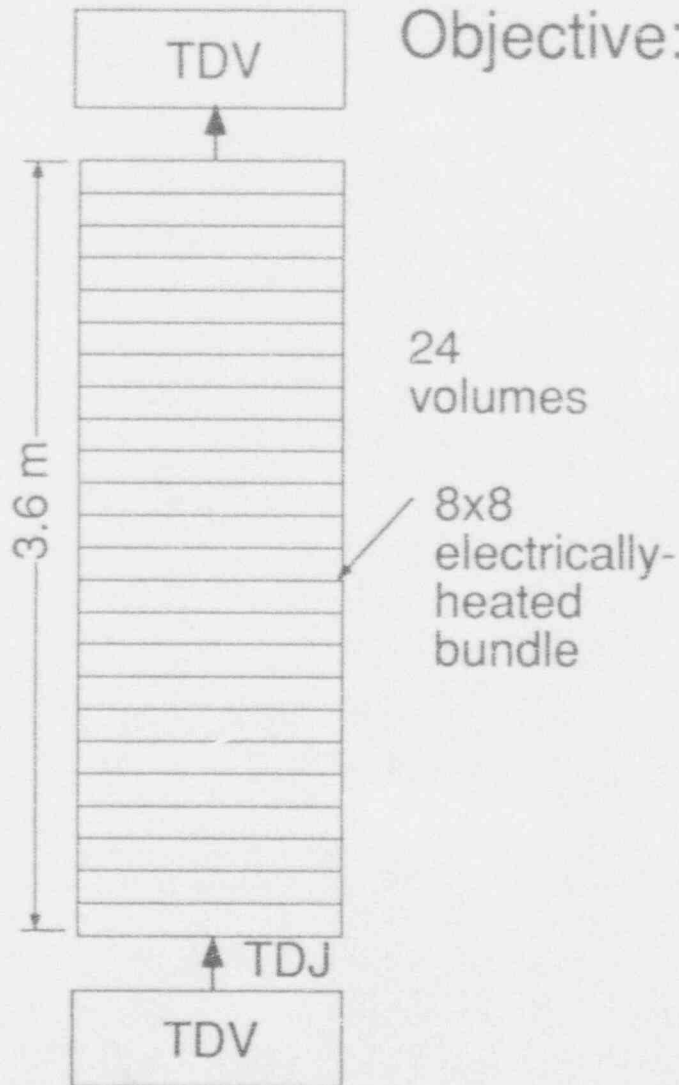
- Pressure =  $5.512 \times 10^6$  Pa
- Power = 70 KW
- Flow Rate = 1.15 m/s
- Inlet subcooling = 12.5 K

# Measured and RELAP5/MOD3 – Calculated Axial Void Fraction for the Christensen Subcooled Boiling Test 15



# ORNL THTF Void Profile Test

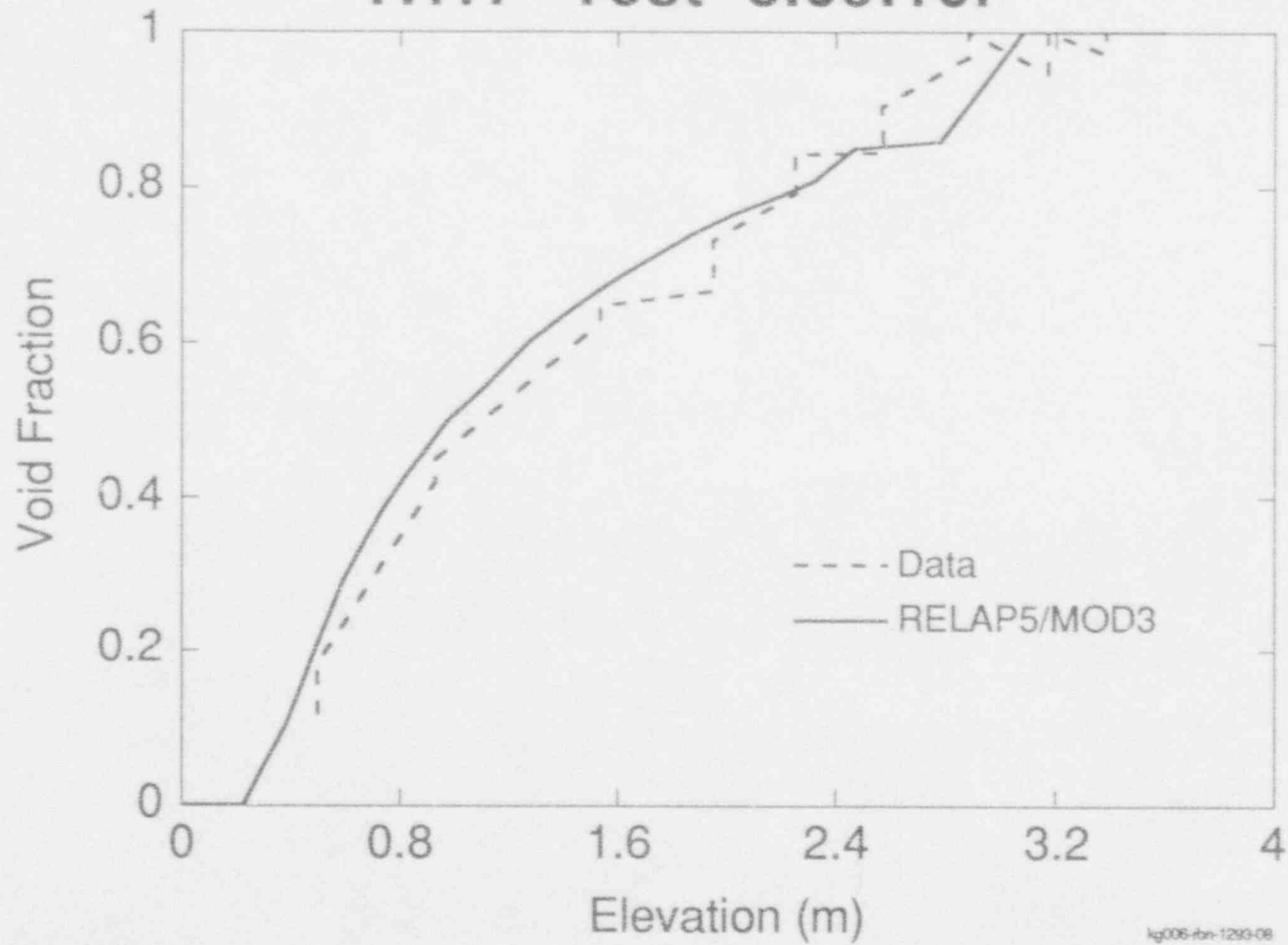
Objective: Subcooled boiling, void generation, void propagation



## Test Conditions

- Pressure = 4.5 MPa
- Heat flux = 0.38 MW/m<sup>2</sup>
- Mass flux = 29.8 Kg/s-m<sup>2</sup>
- Inlet subcooling = 57.6 K°

# Measured and Calculated Void Fraction THTF Test 3.09.10i



# Bankoff Cocurrent Steam/Water Mixing Experiment

Objective: Assess condensation for stratified flow

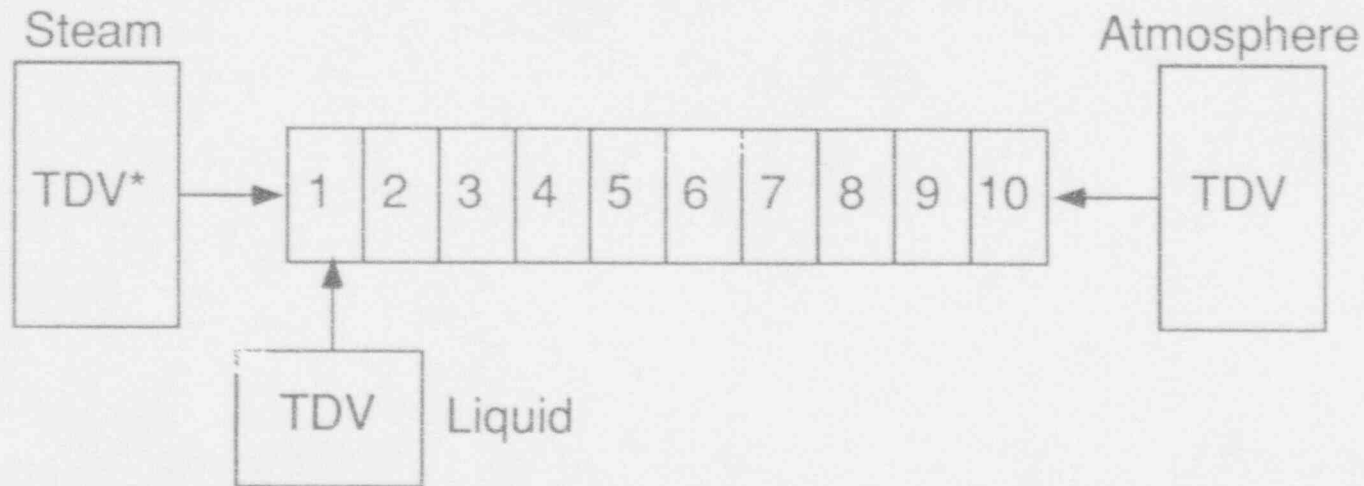
Description:

- Atmospheric pressure
- Rectangular horizontal test section (6.4 cm x 30.5 cm x 160 cm)
- Liquid and steam flow at entrance of test section

	<u>Run 253</u>	<u>Run 279</u>
Liquid		
Flow (kg/s)	0.657	1.04
Temp (K)	295	298
Steam		
Flow (kg/s)	0.0651	0.160
Temp (K)	411	415

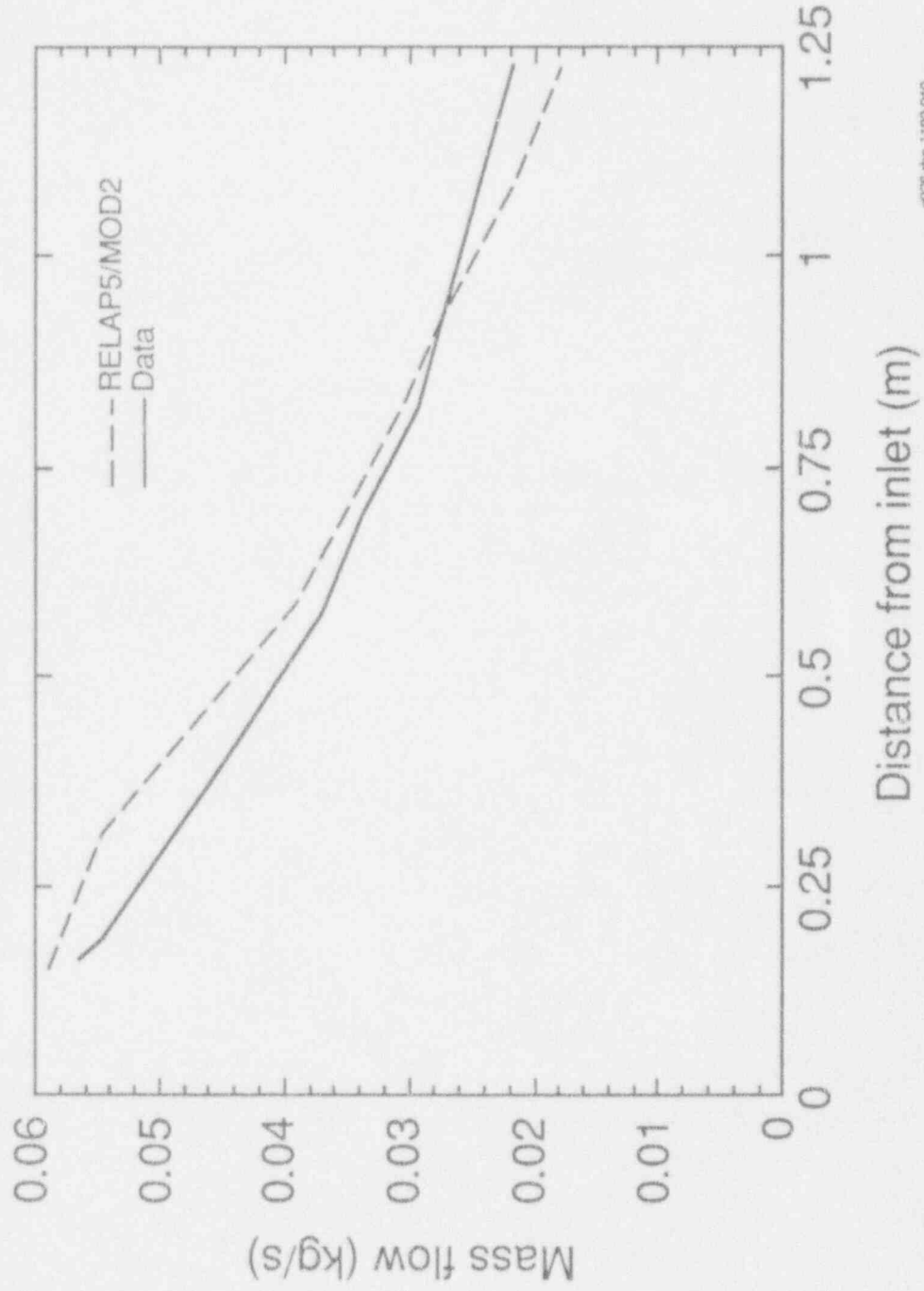


# RELAP5/MOD2 Nodalization for Bankoff Experiment



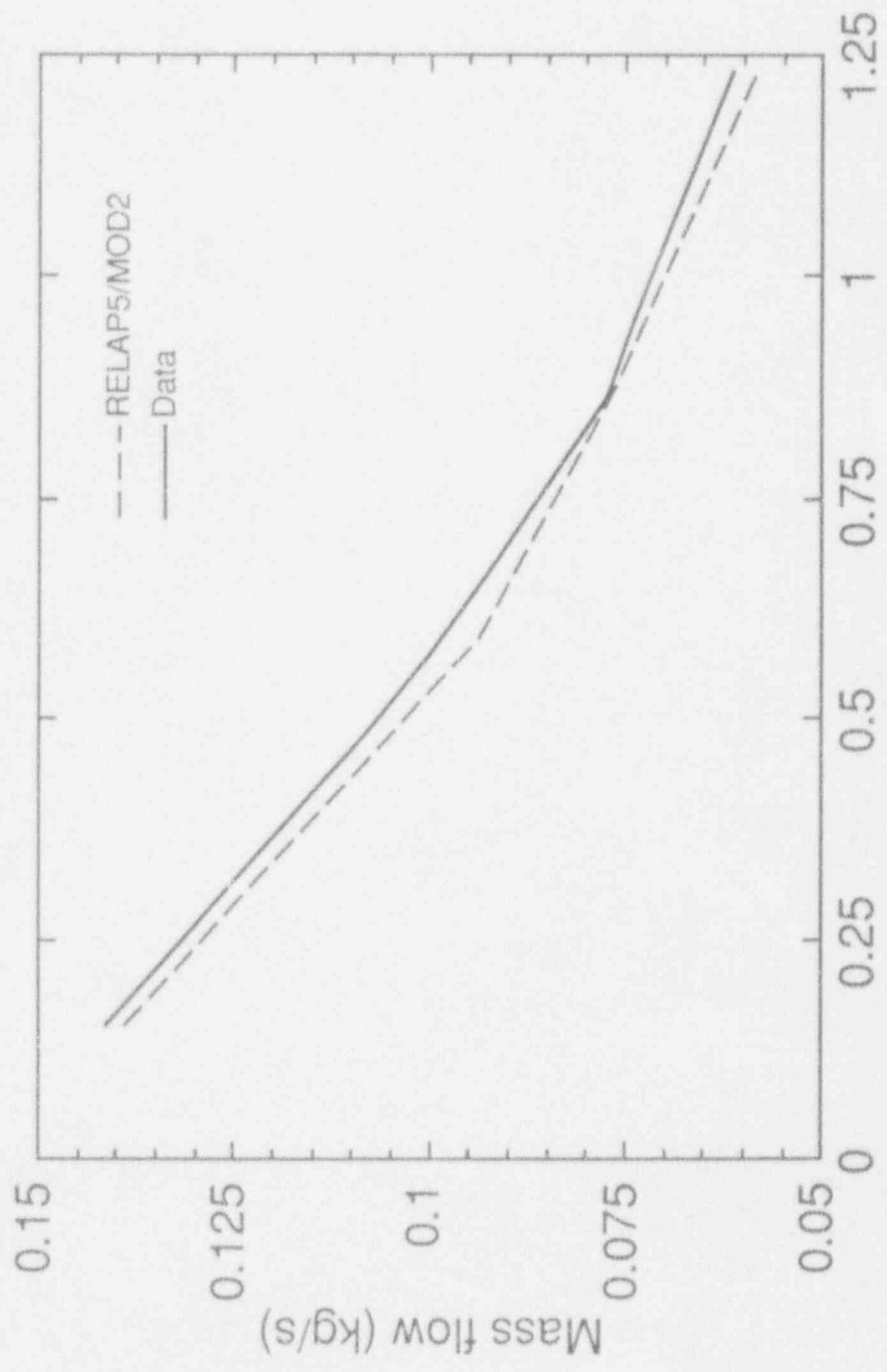
\*TDV = Time Dependent Volume

# Calculated and Measured Axial Steam Flow Profile, Bankoff Run 253



4026-dbr-1193-040

# Calculated and Measured Axial Steam Flow Profile, Bankoff Run 279



Distance from inlet (m)

# Measured and Predicted Liquid Outlet Temperatures Bankoff Cocurrent Flow Experiments

Test Run	Outlet Temperature (K)	
	Data	RELAP5/MOD2
253	342	337
279	358	357

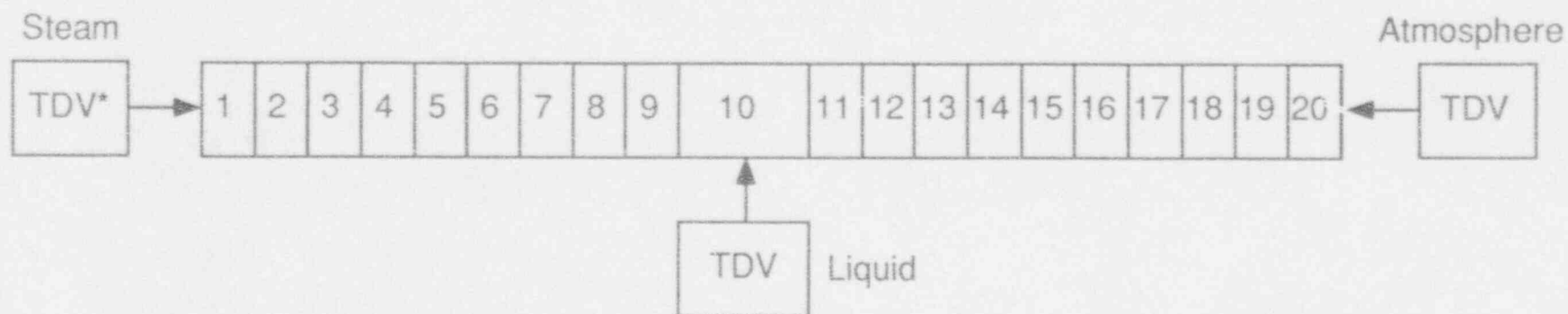
# Aoki Steam/Water Mixing Experiment

Objective: Assess condensation model for dispersed flow

Description:

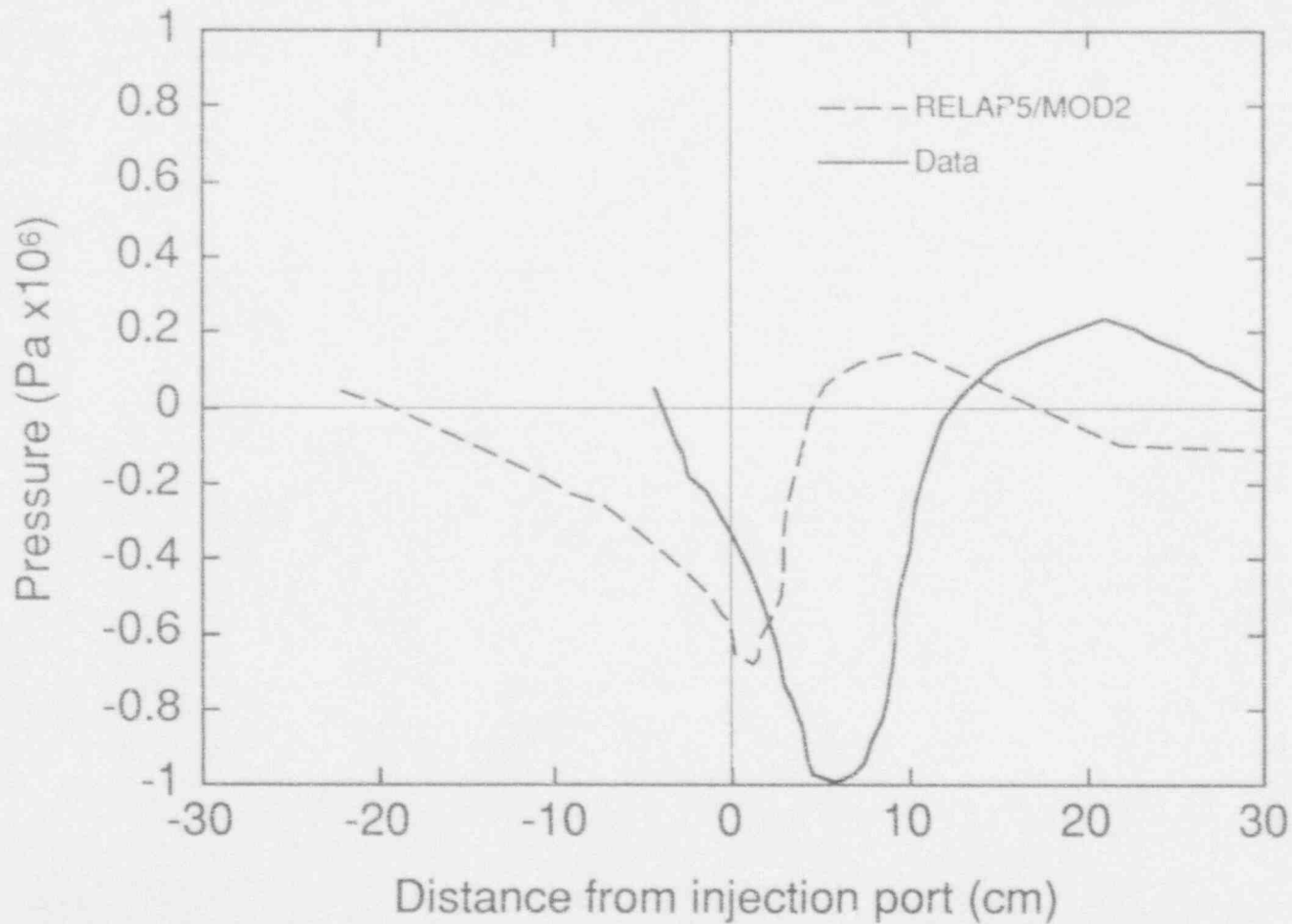
- Atmospheric pressure
- Rectangular horizontal test section (2.86 cm x 60 m)
- Steam flow at entrance to test section (60 kg/s-m<sup>2</sup> at 373 K)
- Liquid flow at center of test section (330 kg/s-m<sup>2</sup> at 293 K)

# RELAP5/MOD2 Nodalization for Aoki Experiment



\*TDV = Time Dependent Volume

# Calculated and Measured Axial Pressure Profiles for Aoki Experiment



# Vapor Generation Model

- The mass transfer model is of the form

$$\Gamma_g = \Gamma_{ig} + \sum_j \Gamma_{wj}$$

- The term  $\Gamma_{ig}$  is the mass transfer at the interface in the bulk (away from the wall) and is given by

$$\Gamma_{ig} = - \frac{H_{if}(T^s - T_f) + H_{ig}(T^s - T_g)}{h_f^* - h_g^*}$$

- where  $h_f^*$  and  $h_g^*$  are bulk or saturation enthalpies (discussed in Volume 1) and  $H_{if}$  and  $H_{ig}$  are interfacial heat transfer coefficients (discussed in Volume 4)

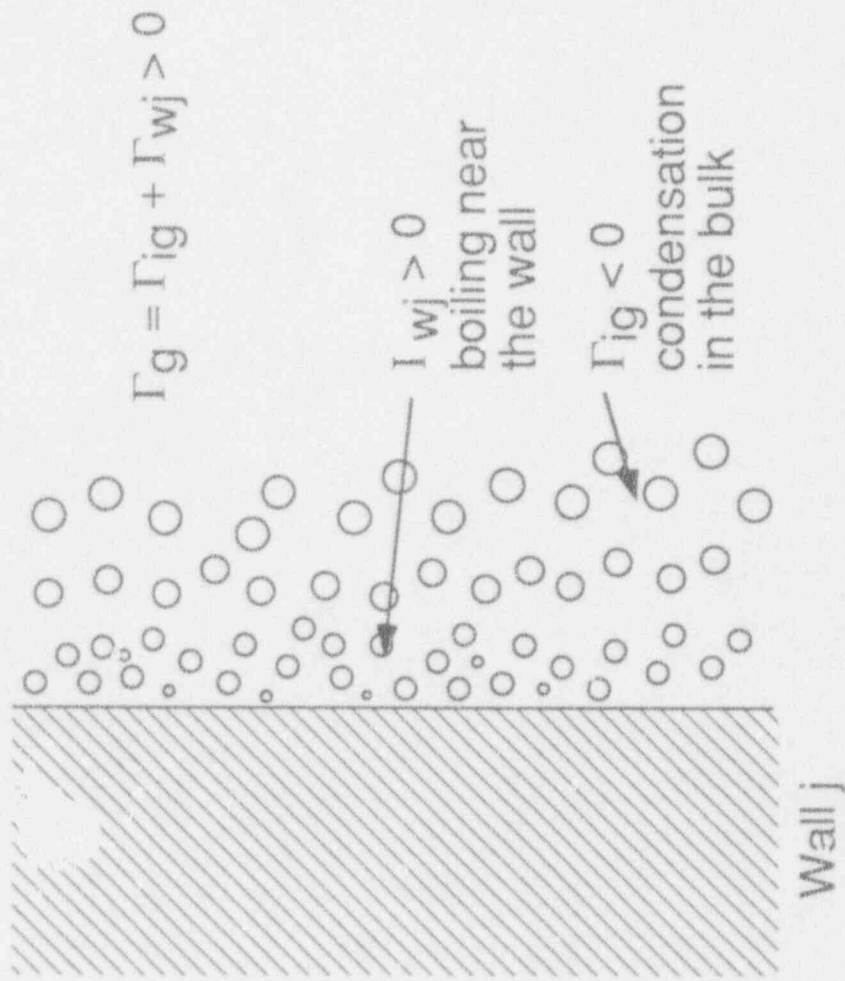


## Vapor Generation Model (cont'd)

- The term  $\Gamma_{wj}$  is the mass transfer at the liquid-vapor interface near the wall of heat structure  $j$  attached to the volume
- Because RELAP5 has just one liquid temperature in a volume and does not calculate thermal gradients in the wall boundary layer, another model must be used for  $\Gamma_{wj}$
- For subcooled boiling, the bulk liquid can be subcooled while water in the boundary layer is warmer and is flashing to steam, resulting in net vapor generation

# Subcooled Boiling Mass Transfer

(Net Vapor Generation)



## Vapor Generation Model (cont'd)

- To capture this effect, the mechanistic method proposed by Lahey is used for  $\Gamma_{wj}$
- These words are currently in Volume 1 and 4 in the constitutive relations section, and similar words will be added to the field equation section when  $\Gamma_{wj}$  is first introduced.

# Summary

- RELAP5's constitutive modeling includes idealizations and engineering approximations where further sophistication is either not possible or impractical
- Validation using relevant data is the essential element for acceptability

## FILM-SIDE HEAT TRANSFER RESISTANCES

### \* FALLING FILMS

- Laminar Smooth
- Laminar Wavy
- Turbulent

### \* SHEARED FILMS: Cocurrent Downflow

- Laminar Smooth
- Laminar Wavy
- Turbulent

### \* MODEL SUMMARY

### \* ASSESSMENT PLANS

## FALLING FILMS: LAMINAR SMOOTH

### \* NUSSELT MODEL ASSUMPTIONS:

- 1) Fully Developed, Steady, Laminar Flow: Flat plate
- 2) Constant Properties:  $\frac{1}{2}(T_w + T_{set})$
- 3) Subcooling of Liquid Negligible in Energy Balance
- 4) Inertia Effects Negligible in Momentum Balance
- 5) Interfacial Shear Stress Negligible
- 6) Liquid-Vapor Interface is Smooth (No Ripples)
- 7) Heat Transfer Across Film is by Conduction
- 8) Wall Temperature is Uniform Axially
- 9) Saturated Vapor with  $X = 1$  at Inlet
- 10) Condensation is only Limited by Liquid Resistance

# FALLING FILMS: LAMINAR SMOOTH

## \* NUSSELT MODEL :

- Film Thickness

$$m^* = \left( \frac{3}{4} \cdot \text{Re}_f \right)^{\frac{1}{3}}$$

where

$$m^* = m \cdot \left( \frac{\rho_f \cdot \Delta\rho \cdot g}{\mu_f^2} \right)^{\frac{1}{3}}$$

$$\text{Re}_f = \frac{4\Gamma'}{\mu_f}$$

- Local Nusselt No.

$$\text{Nu}^* = \left( \frac{3}{4} \cdot \text{Re}_f \right)^{\frac{1}{3}} = \frac{1}{m^*}$$

where

$$\text{Nu}^* = \frac{h}{k_f} \cdot \left( \frac{\mu_f^2}{\rho_f \cdot \Delta\rho \cdot g} \right)^{\frac{1}{3}}$$

## FALLING FILMS: LAMINAR SMOOTH

### \* NUSSELT MODEL :

- Average Nusselt No.

$$\langle Nu^* \rangle = \frac{1}{L} \cdot \int_0^L Nu^* \cdot dz$$

$$\langle Nu^* \rangle = \frac{4}{3} \cdot \left( \frac{3}{4} \cdot Re_f(L) \right)^{-\frac{1}{3}}$$

$$\langle Nu^* \rangle = 0.943 \cdot \left\{ \frac{h_{lv} \cdot \rho_f \cdot \Delta\rho \cdot g \cdot L^3}{k_f \cdot \mu_f \cdot (T_i - T_w)} \right\}^{\frac{1}{4}}$$



## FALLING FILMS: LAMINAR SMOOTH

### \* NUSSELT TYPE ANALYSIS :

- Condensation Heat Transfer Coefficient:

$$h_{cond} = \frac{k_f}{m}$$
$$q_w = \frac{k_f}{m} \cdot (T_w - T_i)$$

- Bulk Fluid Temperature:

- Parabolic velocity profile with linear temperature profile yields

$$T_f = \frac{5}{8} \cdot T_i + \frac{3}{8} \cdot T_w$$

- Wall-Fluid Heat Transfer

$$R_{wl} = R_{film} \cdot \left[ \frac{T_f - T_w}{T_i - T_w} \right] = \frac{5}{8} \cdot R_{film}$$

$$h_{wl} = \frac{8}{5} \cdot h_{cond} = \frac{8}{5} \cdot \frac{k_f}{m}$$

$$Nu_w = 1.8$$

# FALLING FILMS: LAMINAR SMOOTH

## \* NUSSELT TYPE ANALYSIS :

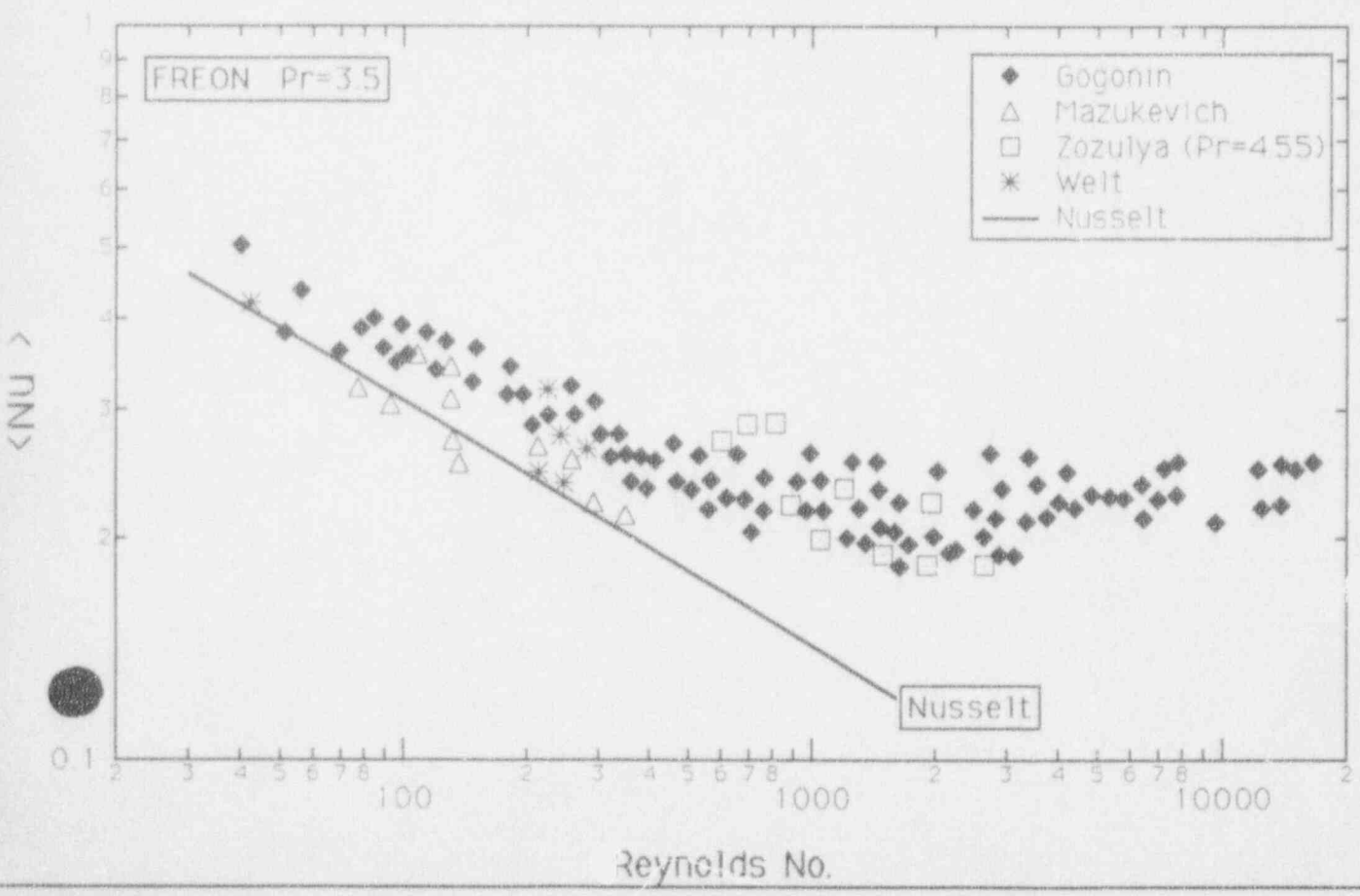
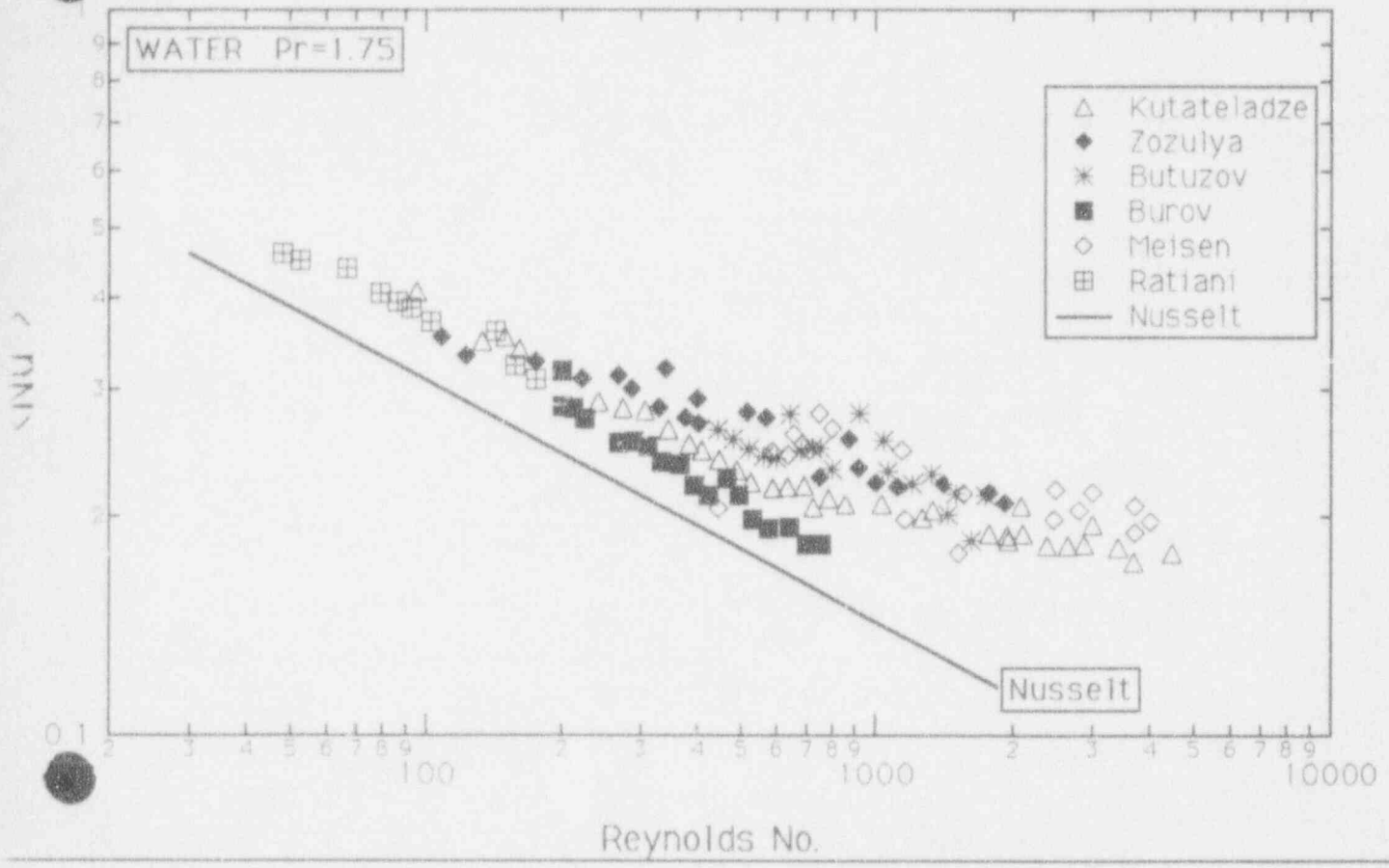
- Interfacial Heat Transfer: Liquid-Interface

$$R_{li} = \frac{3}{8} R_{film}$$

$$h_{li} = \frac{8}{3} h_{cond} = \frac{8}{3} \cdot \frac{k_l}{m}$$

$$Nu_{li} = 2\frac{2}{3}$$

# FALLING FILM: CONDENSATION



## FALLING FILMS: LAMINAR WAVY

### \* EXPECTED BEHAVIOR

#### - RIPPLE WAVES:

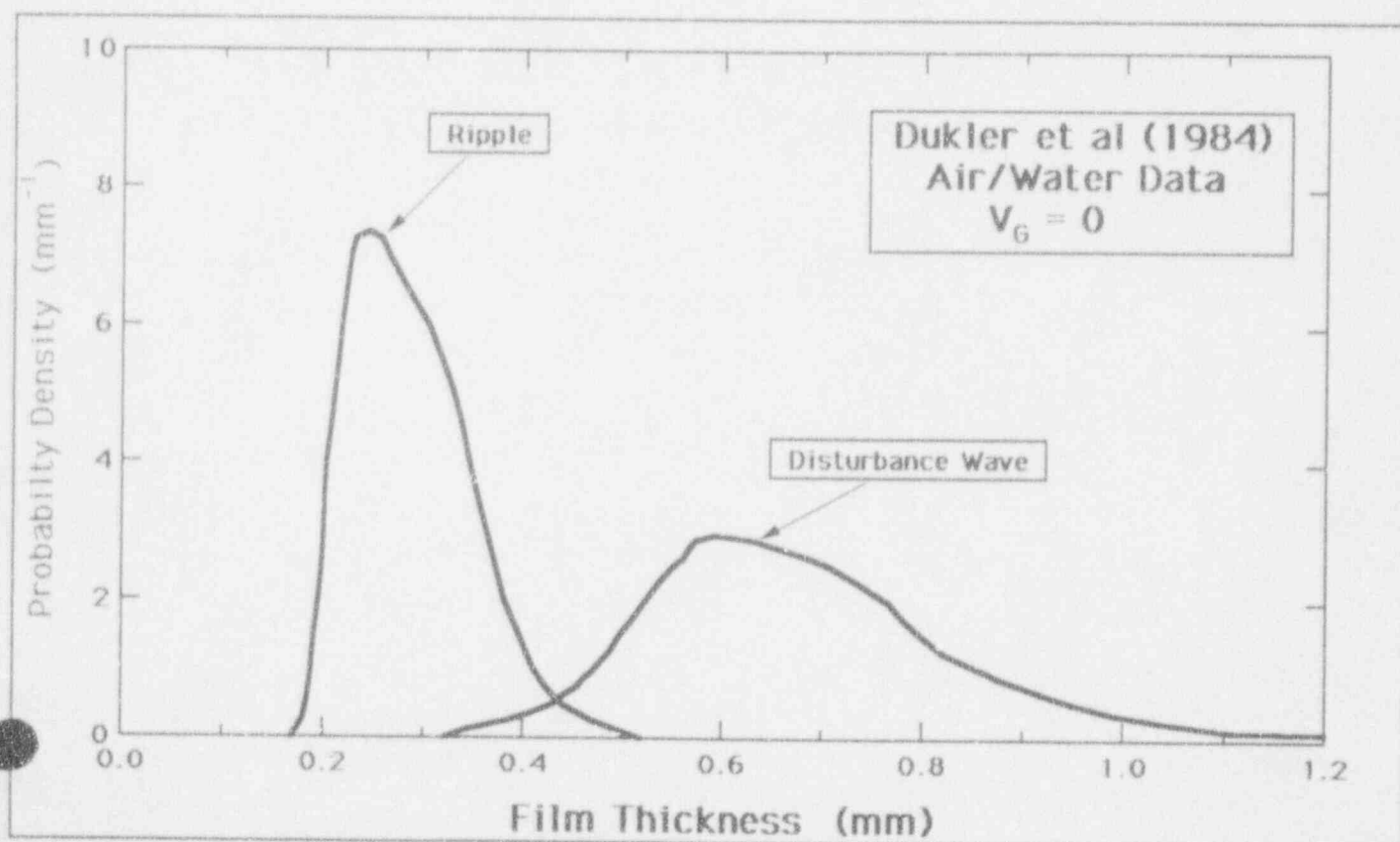
- Reduce Film Thickness
- Induce Velocities Normal to Interface
- Recirculation under Wave Crest?

#### - DISTURBANCE WAVES:

- Kinematic: significant fraction of fluid can be transported by waves, thereby reducing substrate.
- Mixes Film under Wave Crest: a 'developing length' for each wavelength.

# WAVE CHARACTERISTICS

	Ripple Wave	Disturbance Wave
Film Height (mm)	0.280	0.722
Avg. Amplitude (mm)	0.054	0.290
$1/\langle m \rangle$	3.571	1.385
$\langle 1/m \rangle$	3.858	1.535
Enhancement	1.08	1.11



# FALLING FILM: LAMINAR WAVY

## \* ENHANCEMENT DUE TO RIPPLING

- Reduced Film Thickness (Kapitsa, 1948)

$$m^* = \left[ \frac{2.4}{4} \cdot Re_f \right]^{\frac{1}{3}}$$

$$\frac{m_{Ka}}{m_{Nu}} = 0.9283$$

- Effect of Averaging Inverse of Thickness

$$\left\langle \frac{1}{m} \right\rangle_{wavy} \approx 1.08 \cdot \frac{1}{\langle m \rangle}$$

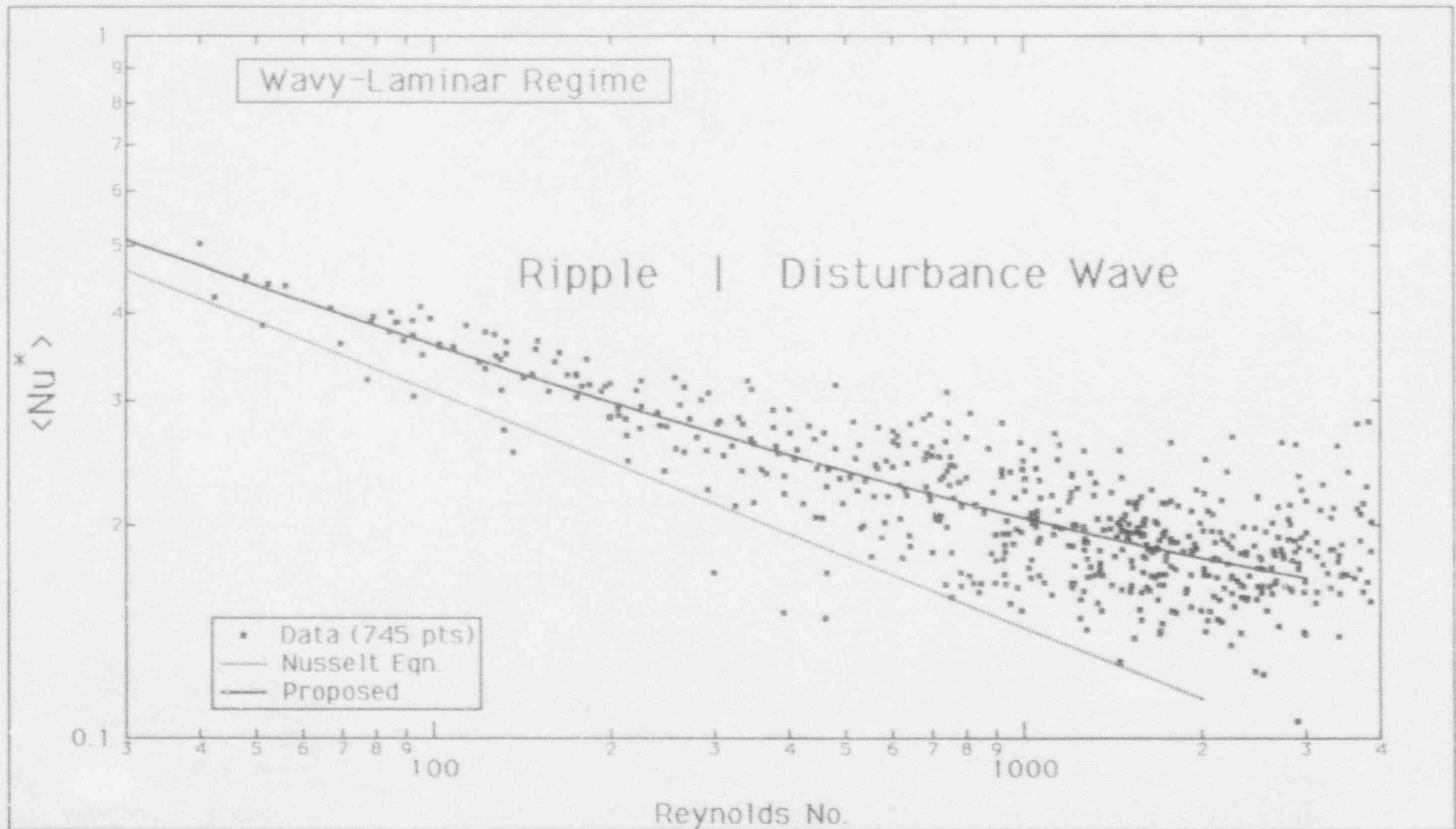
- Enhancement Relative to Nusselt

$$\left\langle \frac{1}{m} \right\rangle_{wavy} \approx 1.08 \cdot \frac{1}{\langle m \rangle_{Ka}}$$

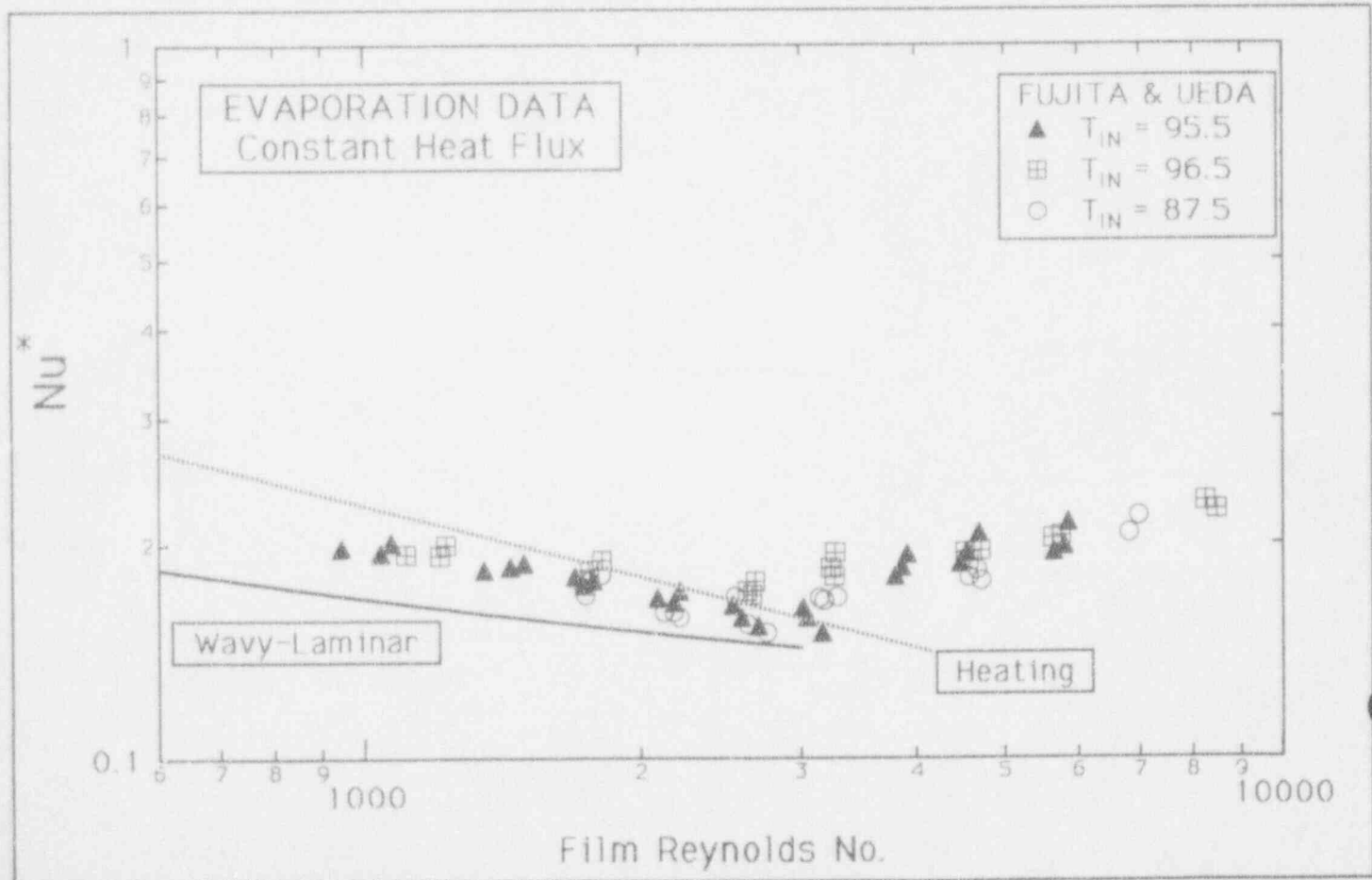
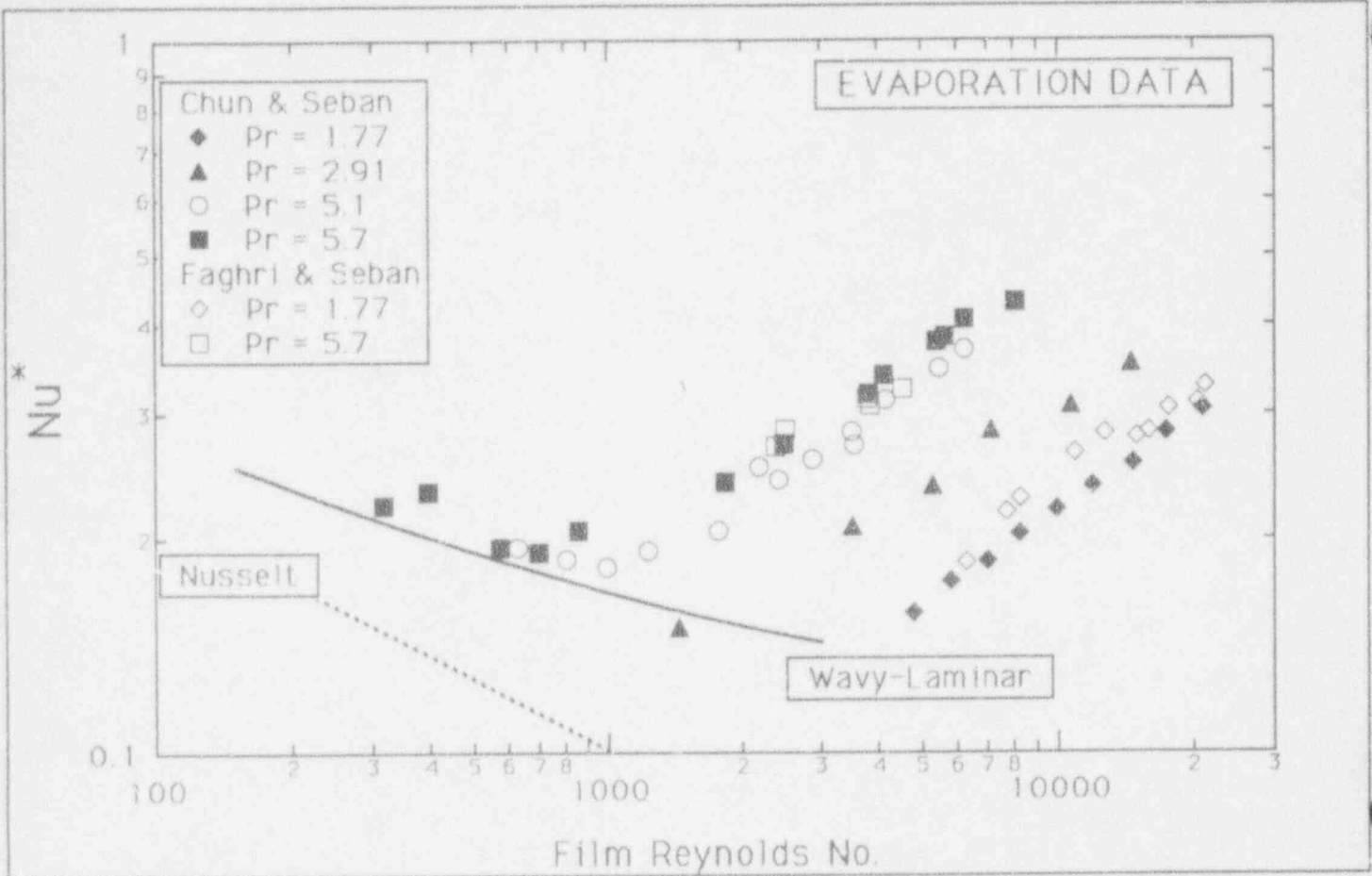
$$\left\langle \frac{1}{m} \right\rangle_{wavy} \approx 1.16 \cdot \frac{1}{\langle m \rangle_{Nu}}$$

# Wavy-Laminar: Proposed Correlation

$$\langle Nu^* \rangle_{wavy} = \langle Nu^* \rangle_{Nu} \cdot (1 + 0.0164 \cdot Re_f^{.46})$$

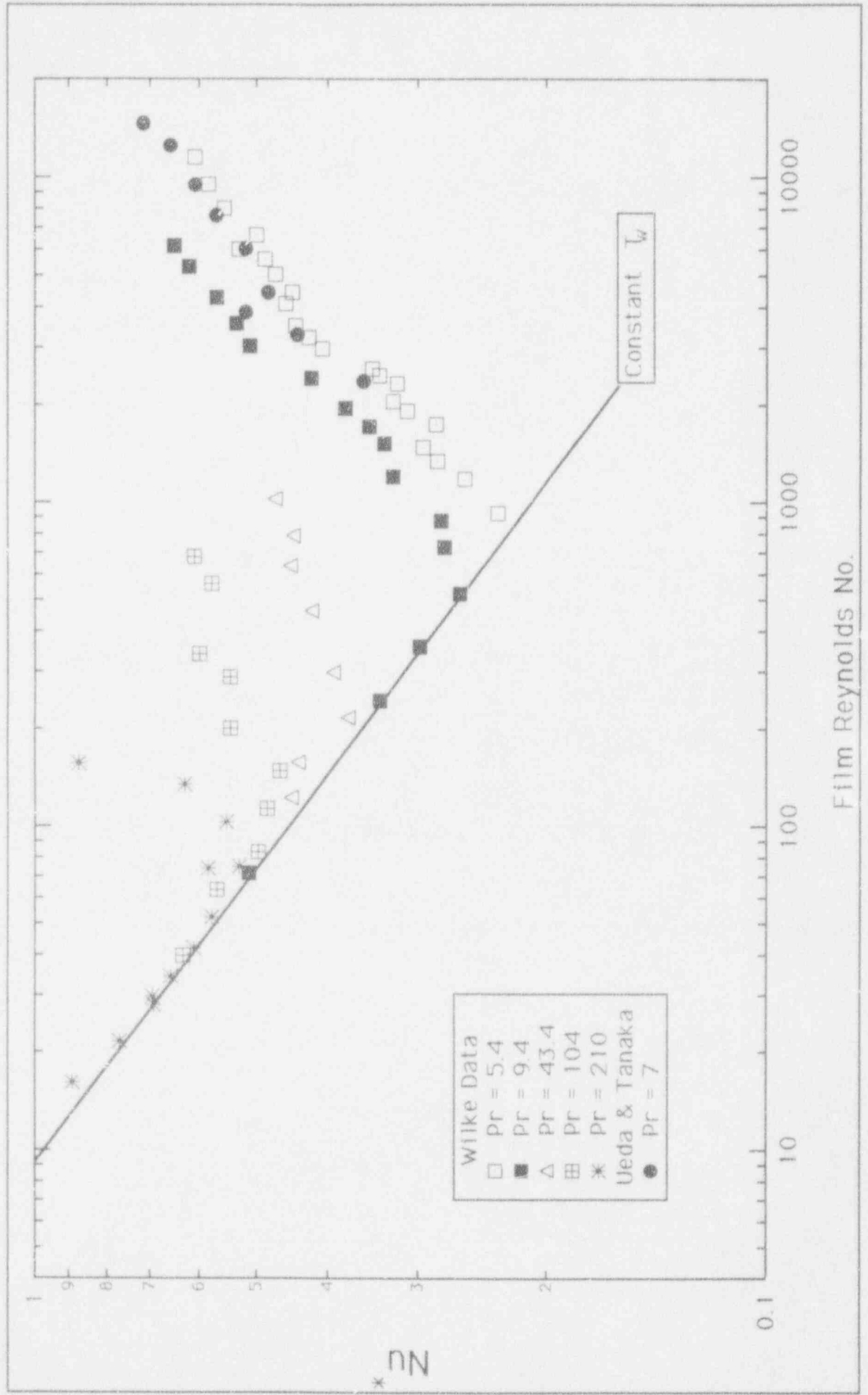


# FALLING FIL: EVAPORATION





# FALLING FILM: Heating Data



# FALLING FILMS: LAMINAR WAVY

## \* Condensation Heat Transfer Coefficient

$$Nu^* = 1.10 \cdot (Re_f)^{-\frac{1}{3}} \cdot (1 + 0.022 \cdot Re_f^{46})$$

$$h_{cond} = \frac{k_f}{m} \cdot (1 + 0.022 \cdot Re_f^{46})$$

## \* Wall-Fluid Heat Transfer

- No effect of waves evident in heating data
- Use asymptotic Nusselt No. for film heating/cooling with constant wall temperature b.c.

$$Nu = 1.9$$

$$Nu^* = 2.09 \cdot (Re_f)^{-\frac{1}{3}}$$

$$h_{wl} = 1.9 \cdot \frac{k_f}{m}$$

## FALLING FILMS: LAMINAR WAVY

### \* Interfacial Heat Transfer: Fluid-Interface

$$R_{ii} = \frac{1}{h_{cond}} - \frac{1}{h_{wt}}$$

$$R_{ii} = \frac{m}{k_f} \left( \frac{1}{Enh} - \frac{1}{1.9} \right)$$

$$h_{ii} = \frac{k_f}{m} \left( \frac{1.9 - Enh}{1.9 \cdot Enh} \right)$$

where,

$$Enh = \left( 1 + 0.022 \cdot Re_f^{.46} \right)$$

# FALLING FILMS: Turbulent

## \* EXPECTED BEHAVIOR

- Transition expected to begin:  $1000 < Re_f < 2000$
- Transition region expected to persist to much higher  $Re_f$  than for pipe flow:
  - Film thickness must be greater than that of buffer layers (wall & interface).
- Primary resistance to heat transfer is due to 'laminar' sublayers at wall & interface.
- At high  $Re_f$ , dependence on  $Re_f$  should tend to 0.8 power.
- Prandtl No. dependence should be between 0.4 (wall cooling) and 1/2 (interface mass transfer).

## FALLING FILMS: Turbulent

### \* CONDENSATION HTC'S:

- Grigull (1942):

$$\langle Nu^* \rangle = 0.0131 \cdot Re_j^{\frac{1}{2}}$$

- Colburn (1933):

$$\langle Nu^* \rangle = 0.056 \cdot Re_j^{0.2} \cdot Pr^{\frac{1}{3}}$$

- Kirkbride (1933):

$$\langle Nu^* \rangle = 0.0077 \cdot Re_j^{0.4}$$

- Soliman et al. (1968):

$$Nu^* = 0.0132 \cdot Re_j^{2.92} \cdot Pr^{0.65}$$

- Blangetti et al. (1982):

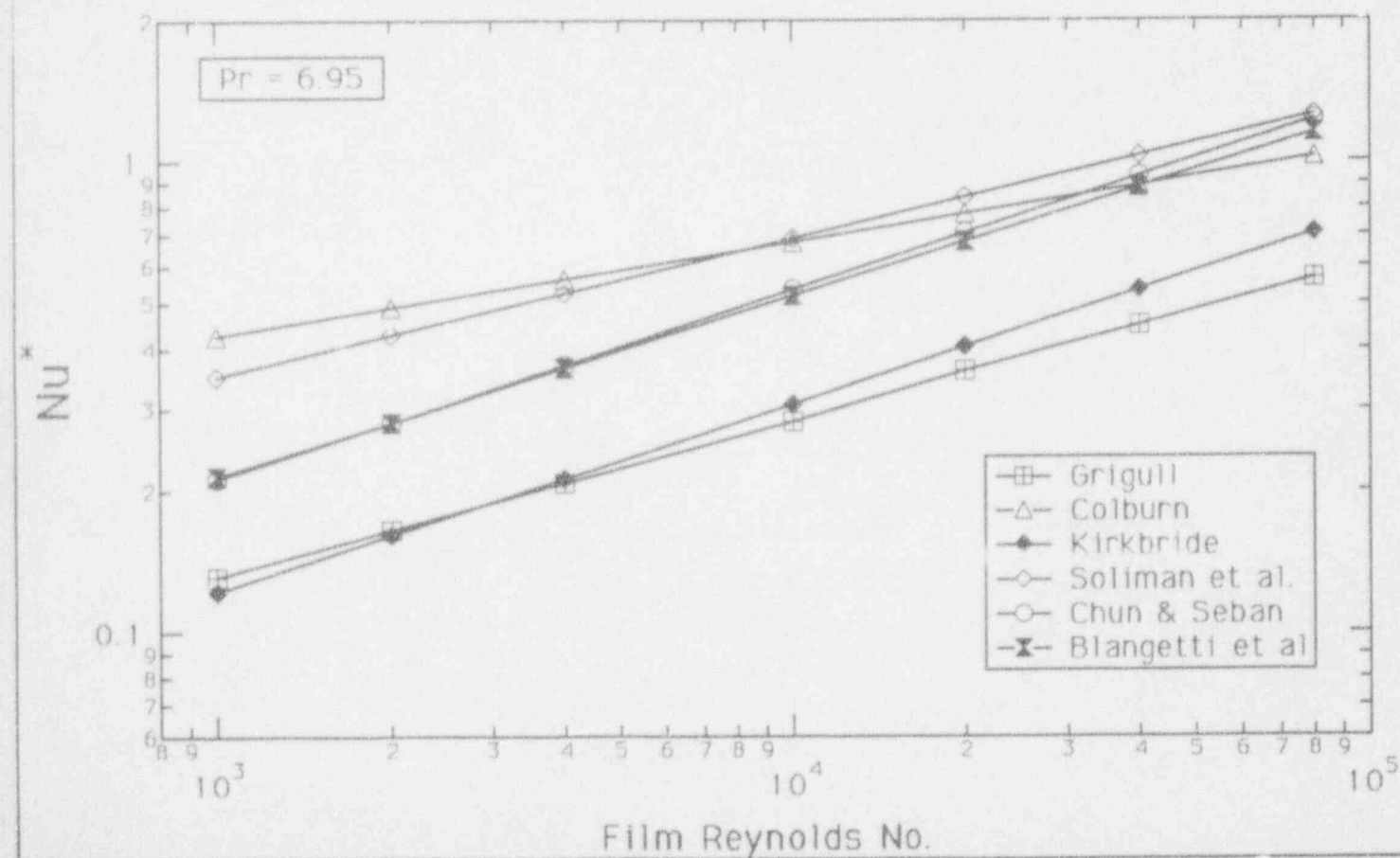
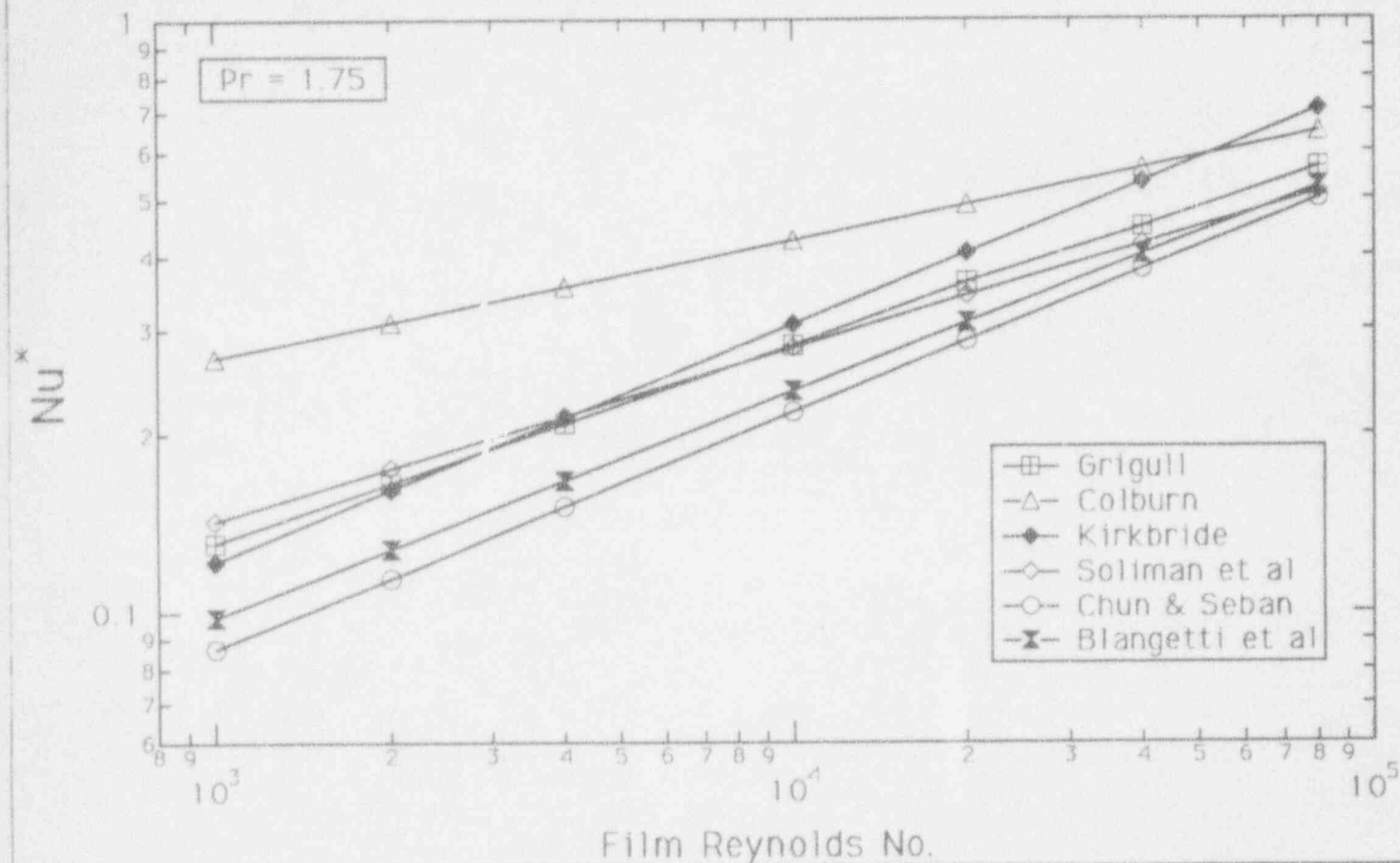
$$Nu^* = 0.0051 \cdot Re_j^{3.82} \cdot Pr^{0.5689}$$

### \* EVAPORATION HTC'S

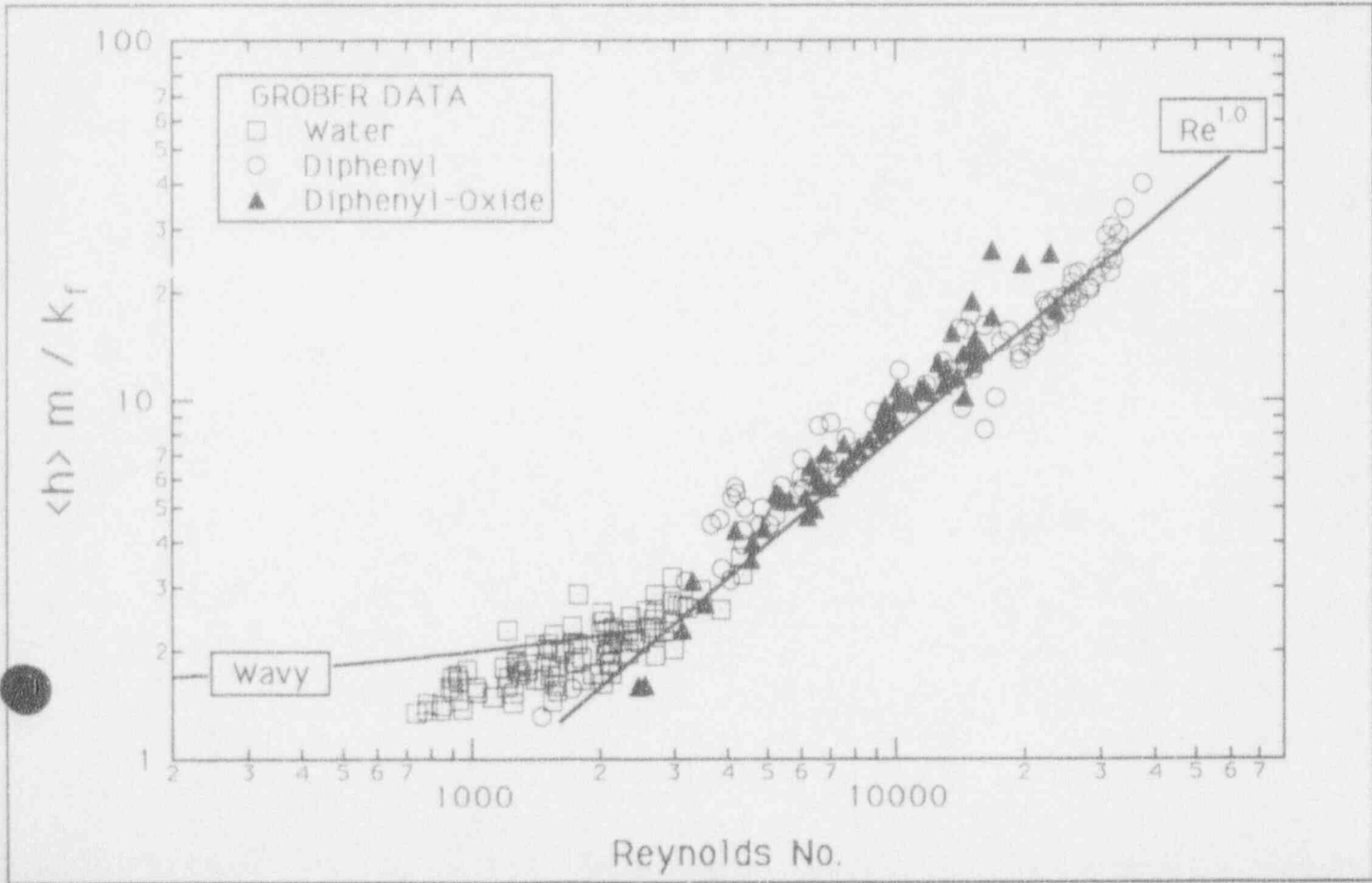
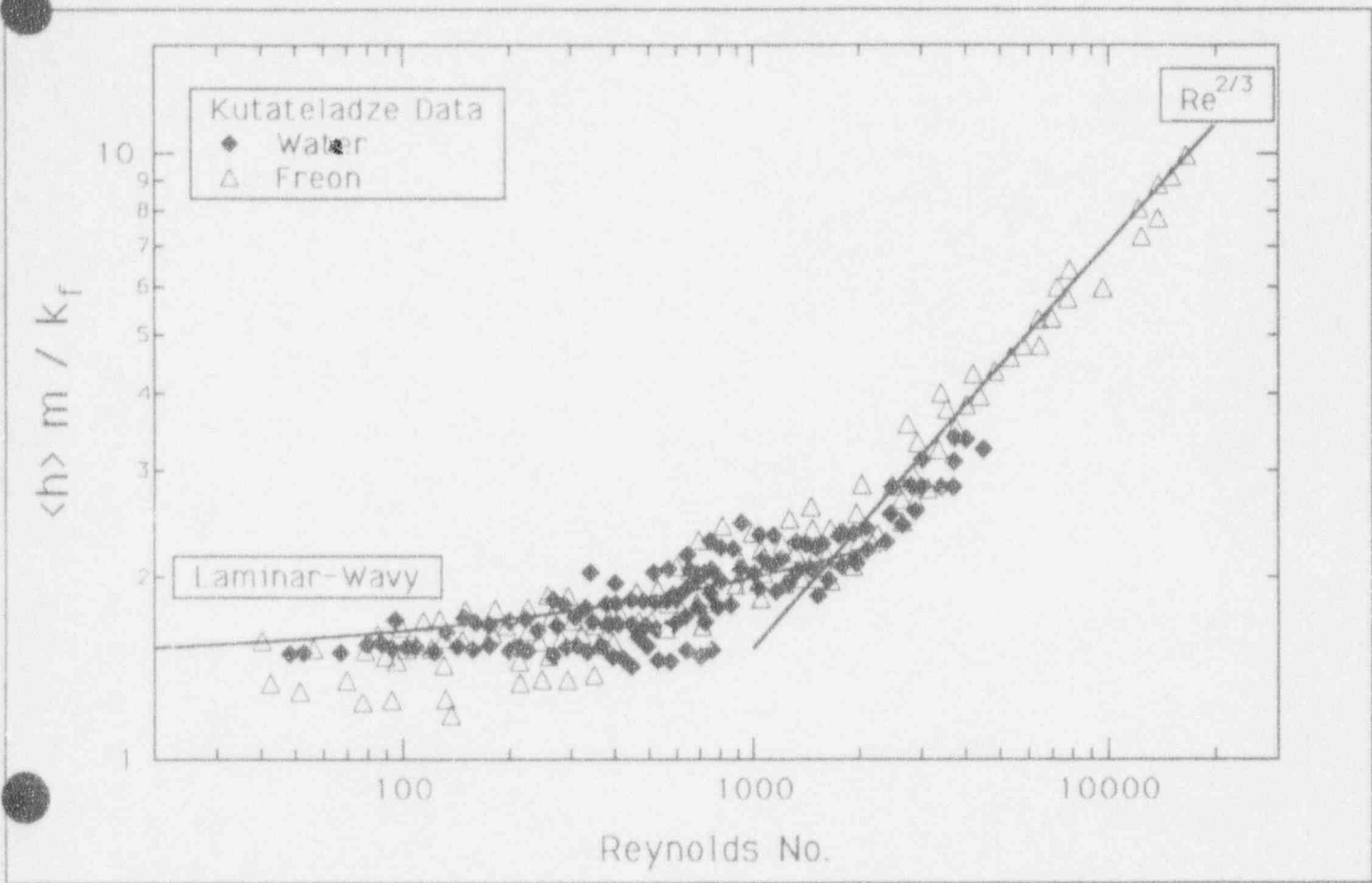
- Chun & Seban (1971):

$$Nu^* = 0.0038 \cdot Re_j^{0.4} \cdot Pr^{0.65}$$

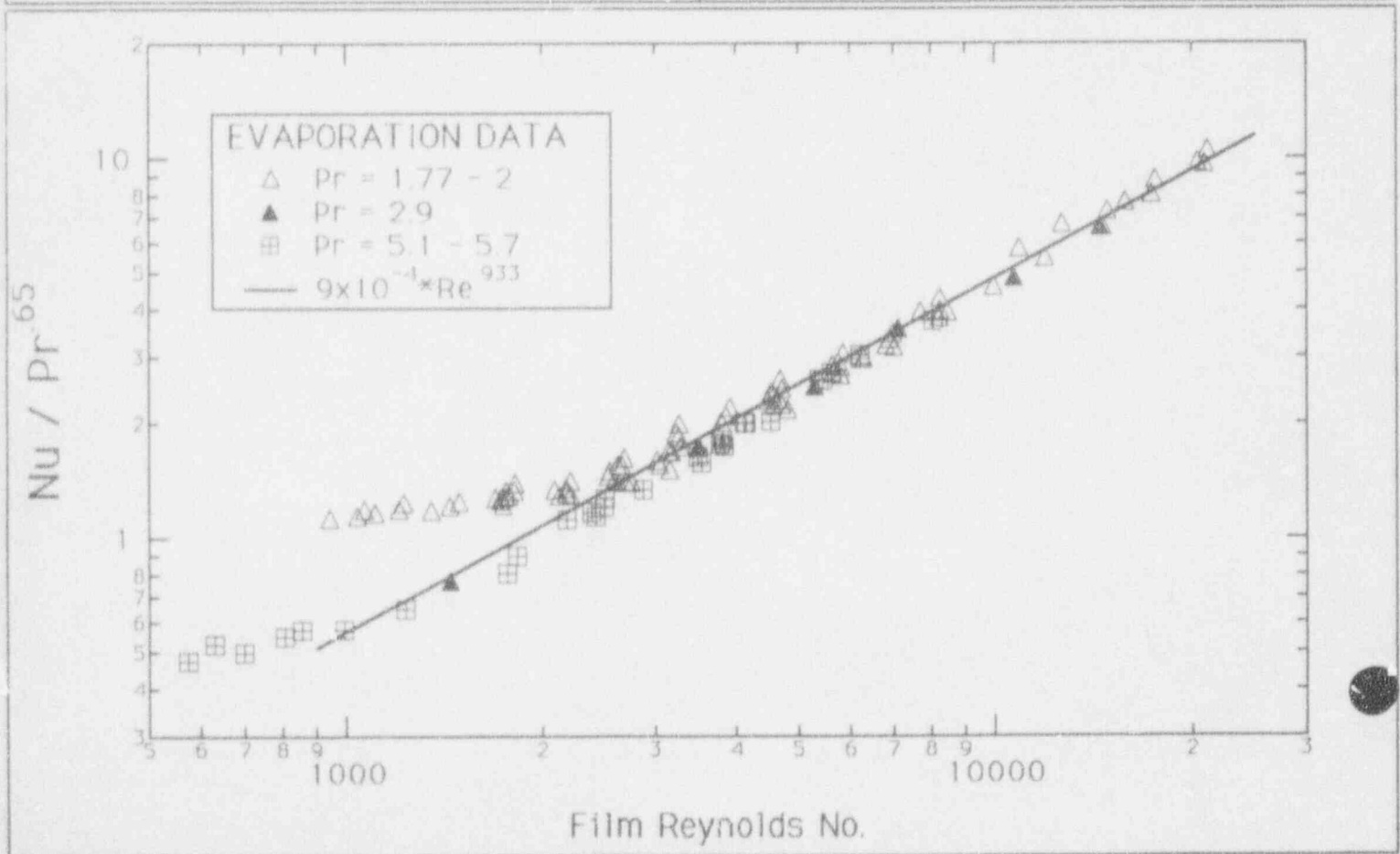
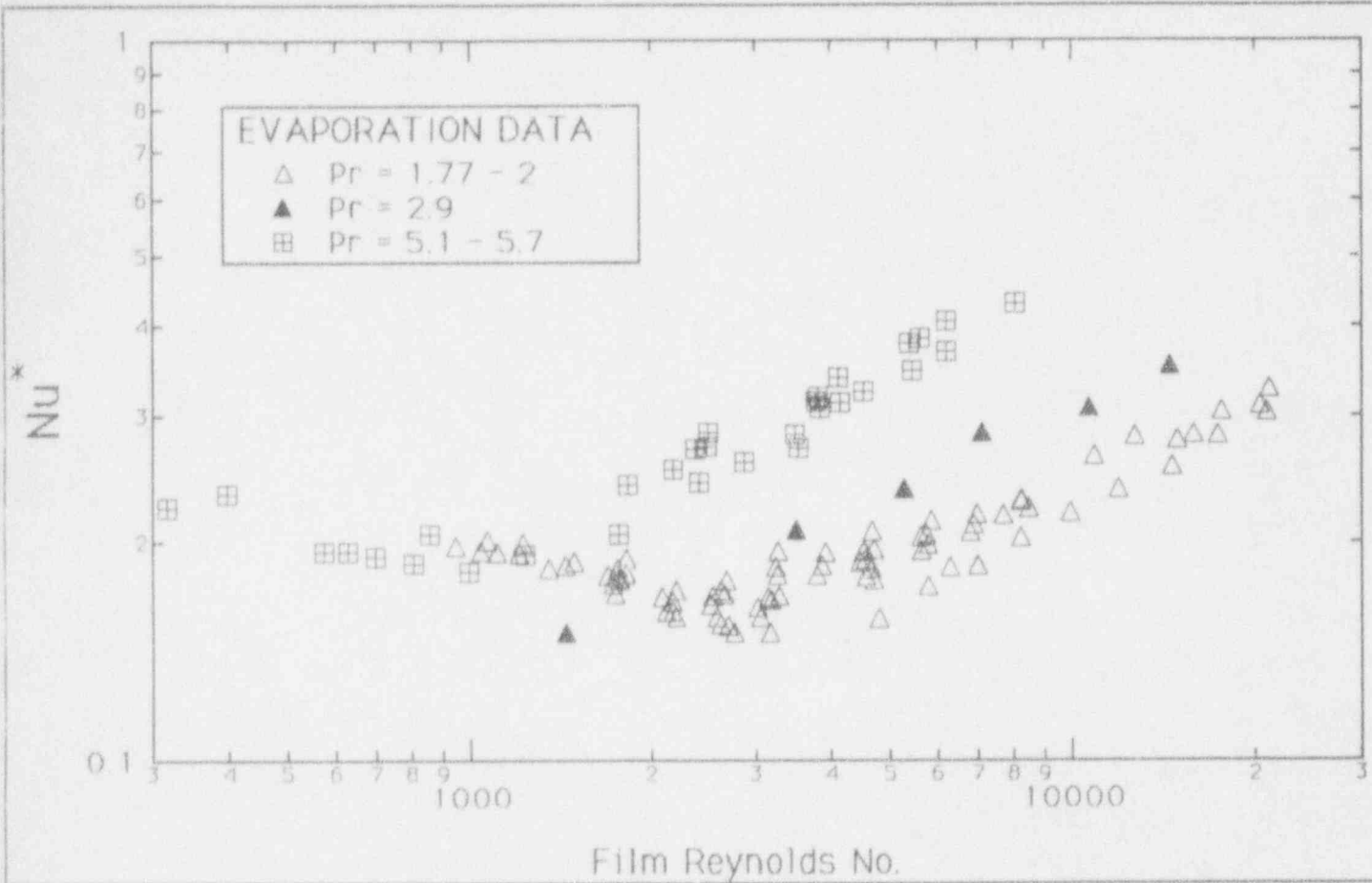
# FALLING FILM: Condensation HTC



# FALLING FILM: CONDENSATION



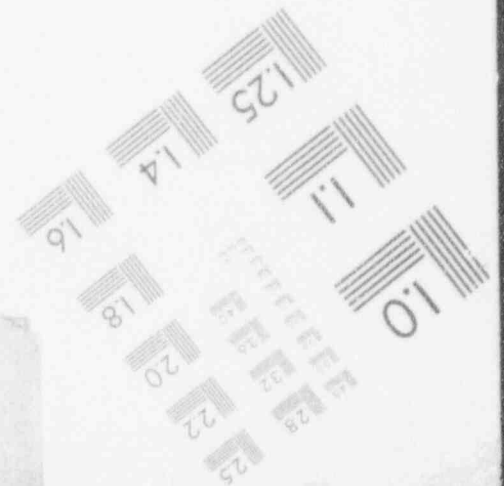
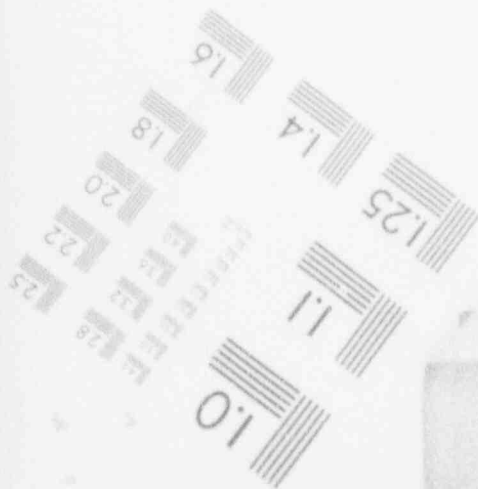
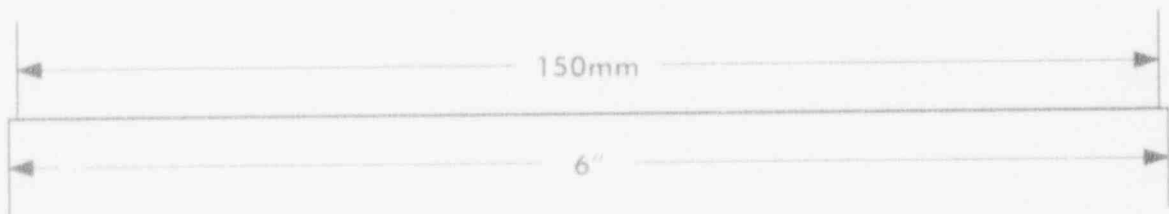
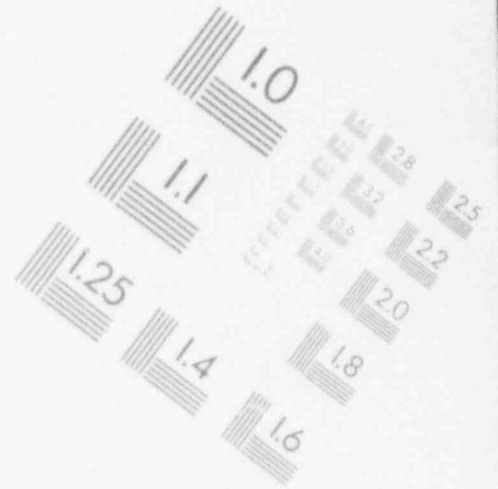
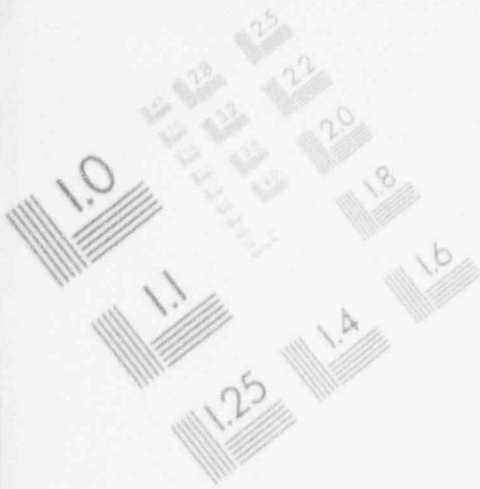
# FALLING FILM: Evaporation Data





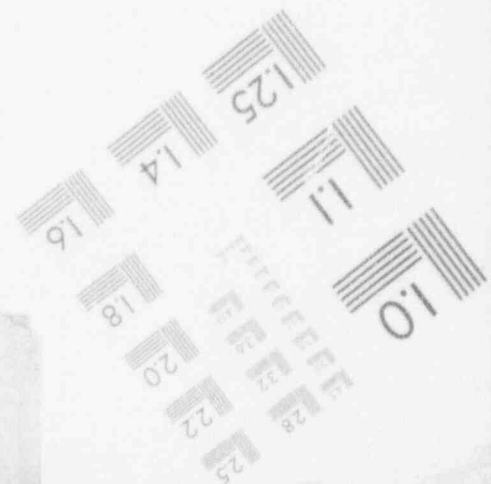
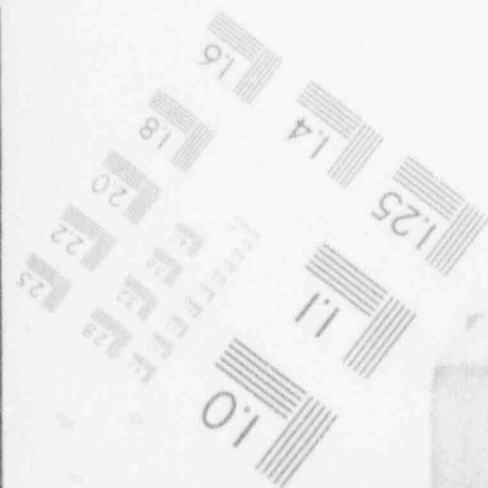
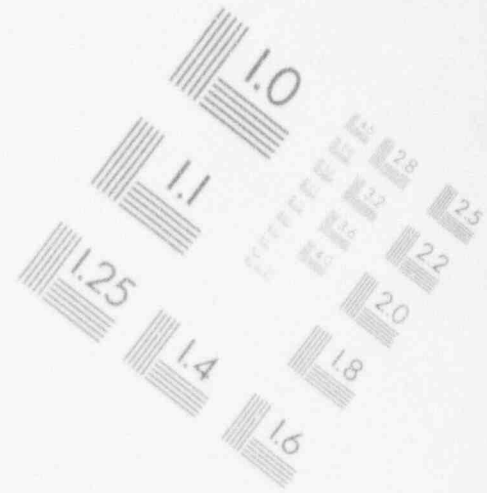
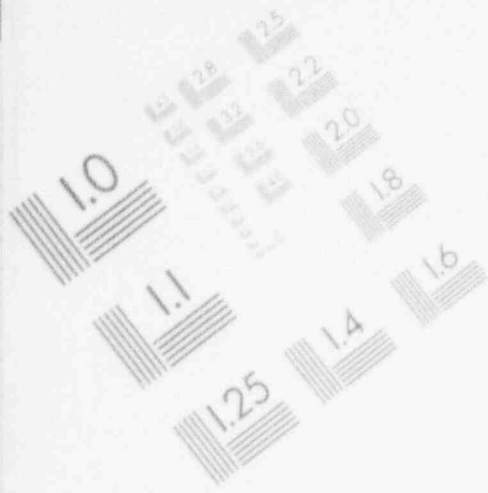
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## IMAGE EVALUATION TEST TARGET (MT-3)



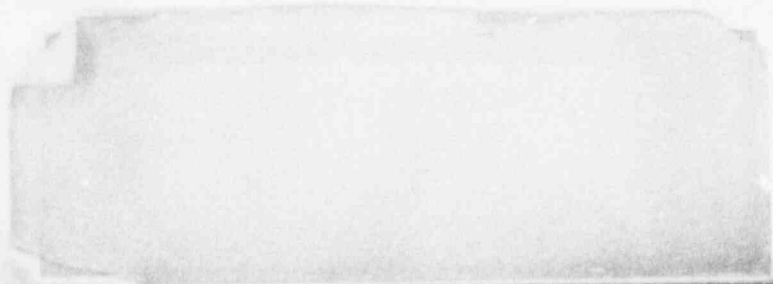
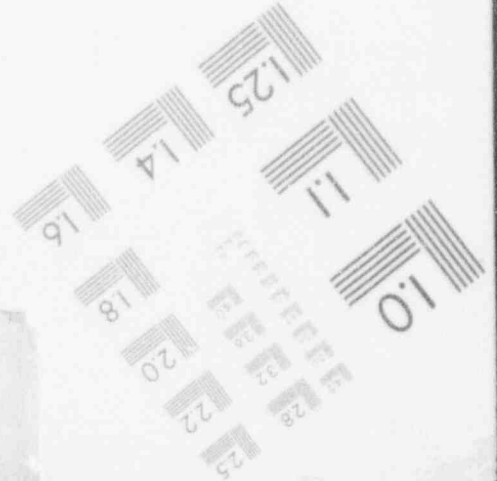
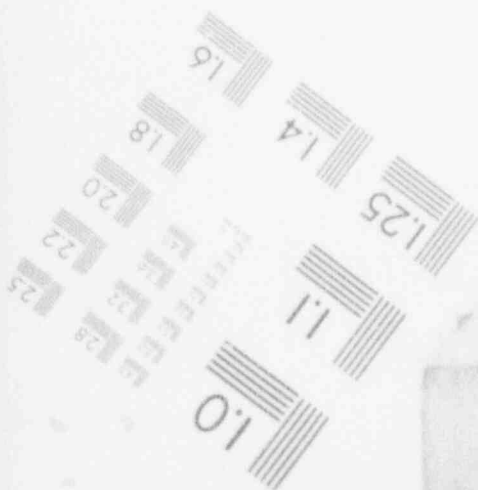
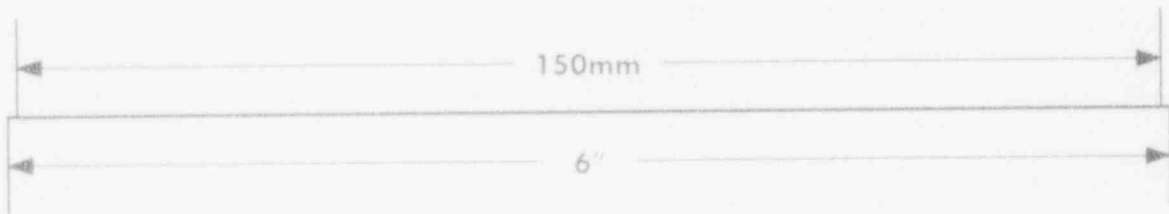
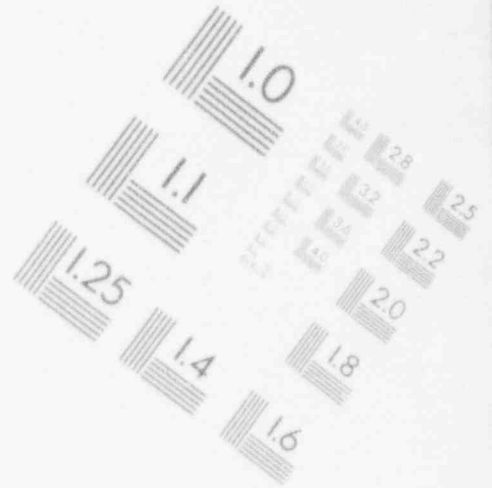
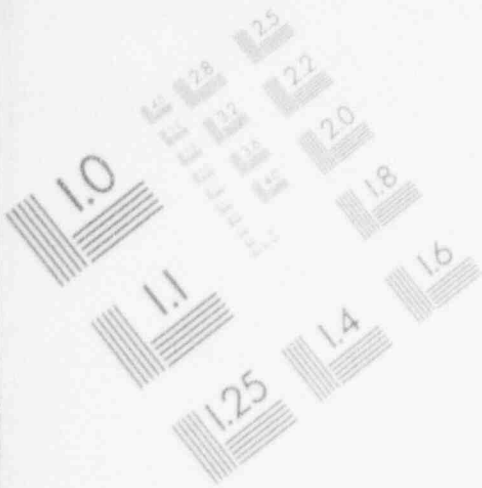
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## IMAGE EVALUATION TEST TARGET (MT-3)



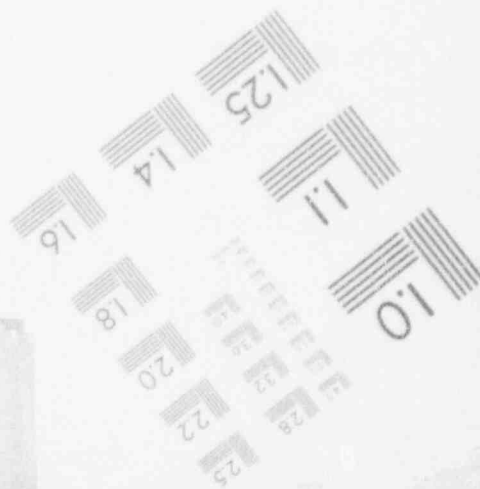
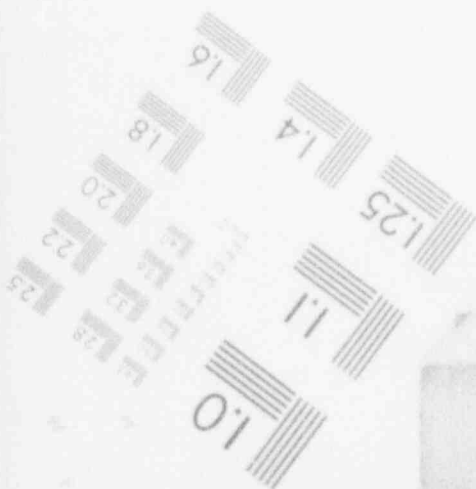
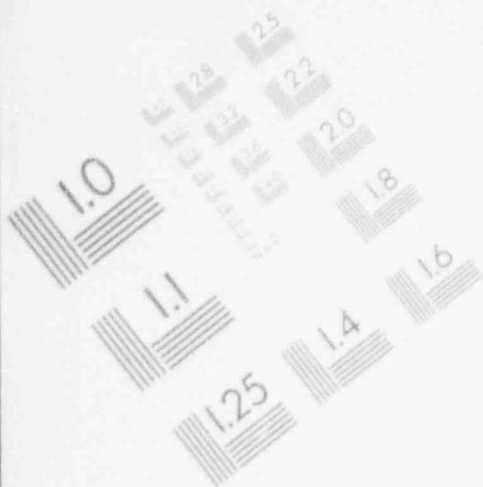
# 1

## IMAGE EVALUATION TEST TARGET (MT-3)



# 1

## IMAGE EVALUATION TEST TARGET (MT-3)



# FALLING FILMS: Turbulent

## \* HEATING HTC'S:

- Wilke (1962):

- original form:

$$1) \quad Nu = 1.9 \qquad Re_f \leq 2460 \cdot Pr^{-6.5}$$

$$2) \quad Nu = .0292 \cdot Re_f^{\frac{3}{5}} \cdot Pr^{.344} \qquad 2460 \cdot Pr^{-.046} \leq Re_f \leq 1600$$

$$3) \quad Nu = 2.12 \times 10^{-4} \cdot Re_f^{1.2} \cdot Pr^{.344} \qquad 1600 \leq Re_f \leq 3200$$

$$4) \quad Nu = 1.81 \times 10^{-3} \cdot Re_f^{\frac{14}{5}} \cdot Pr^{.344} \qquad 3200 \leq Re_f \leq 11500$$

- or

$$1) \quad Nu^* = 2.07 \cdot Re_f^{-\frac{1}{3}} \qquad Re_f \leq 2460 \cdot Pr^{-.646}$$

$$2) \quad Nu^* = .0322 \cdot Re_f^{\frac{1}{5}} \cdot Pr^{.344} \qquad 2460 \cdot Pr^{-.646} \leq Re_f \leq 1600$$

$$3) \quad Nu^* = 1.28 \times 10^{-3} \cdot Re_f^{.634} \cdot Pr^{.344} \qquad 1600 \leq Re_f \leq 3200$$

$$4) \quad Nu^* = 1.10 \times 10^{-3} \cdot Re_f^{.367} \cdot Pr^{.344} \qquad 3200 \leq Re_f \leq 11500$$

# FALLING FILMS: Turbulent

## \* HEATING HTC'S:

- Carey (1985):

$$Nu^* = .0259 \cdot Re_j^{2.92} \cdot Pr^{\frac{1}{3}}$$

- Gimbutis (1974):

$$Nu^* = (.165 \cdot Re_j^{16} - 0.4) \cdot Pr_j^{34} \cdot (Pr_j / Pr_w)^{\frac{1}{4}}$$

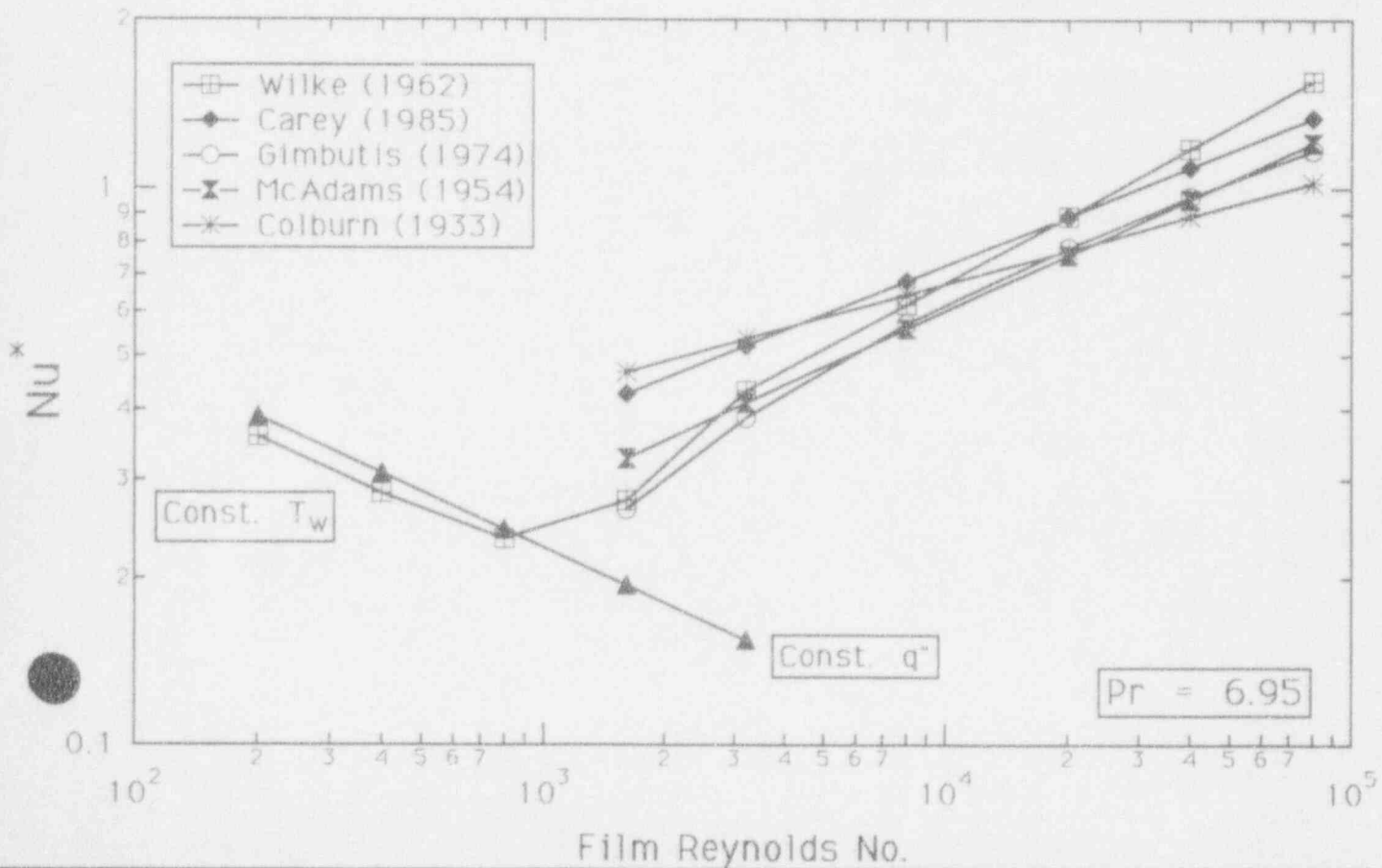
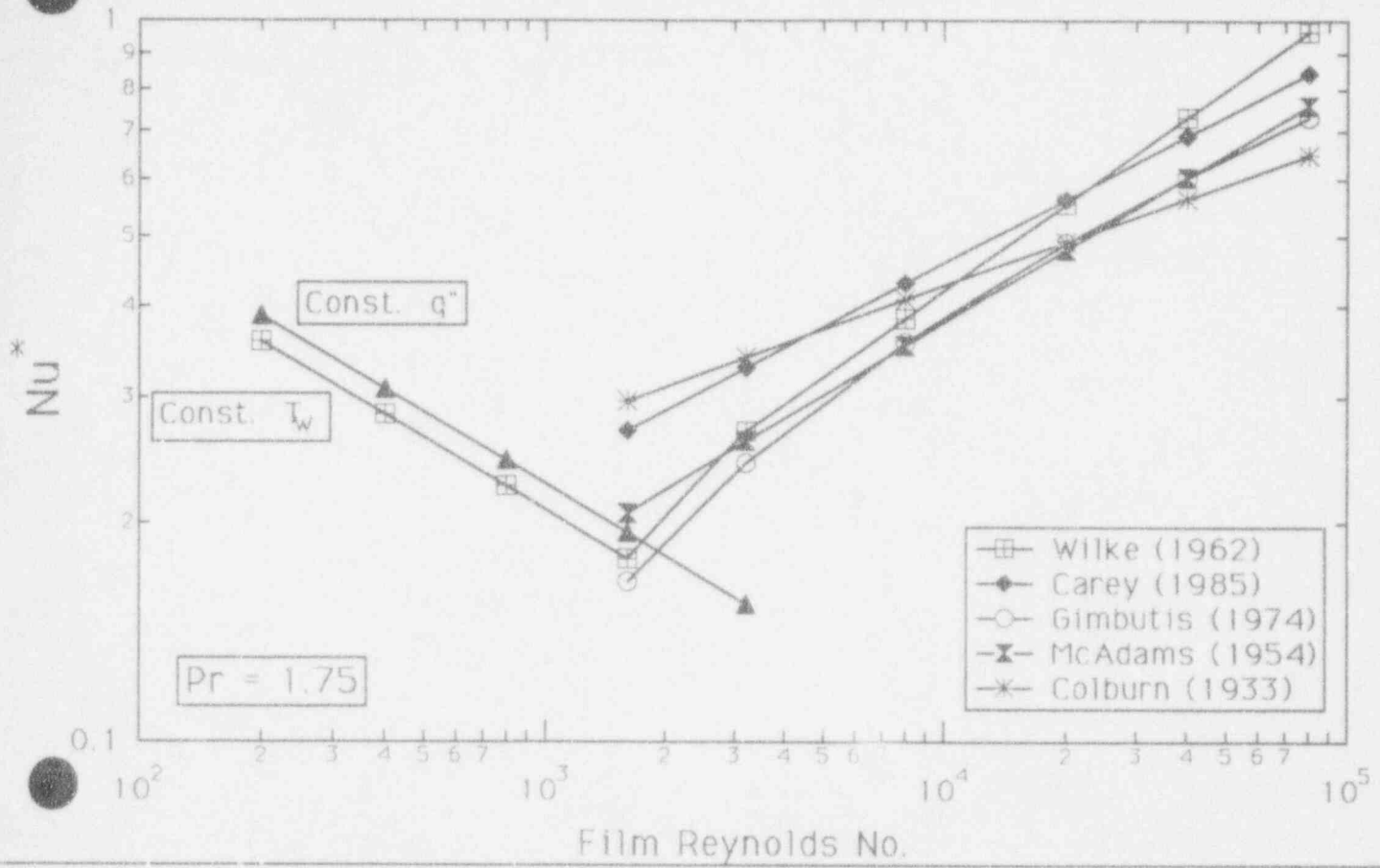
- McAdams (1954):

$$Nu^* = .01 \cdot \pi^{\frac{1}{3}} \cdot Re_j^{\frac{1}{3}} \cdot Pr^{\frac{1}{3}}$$

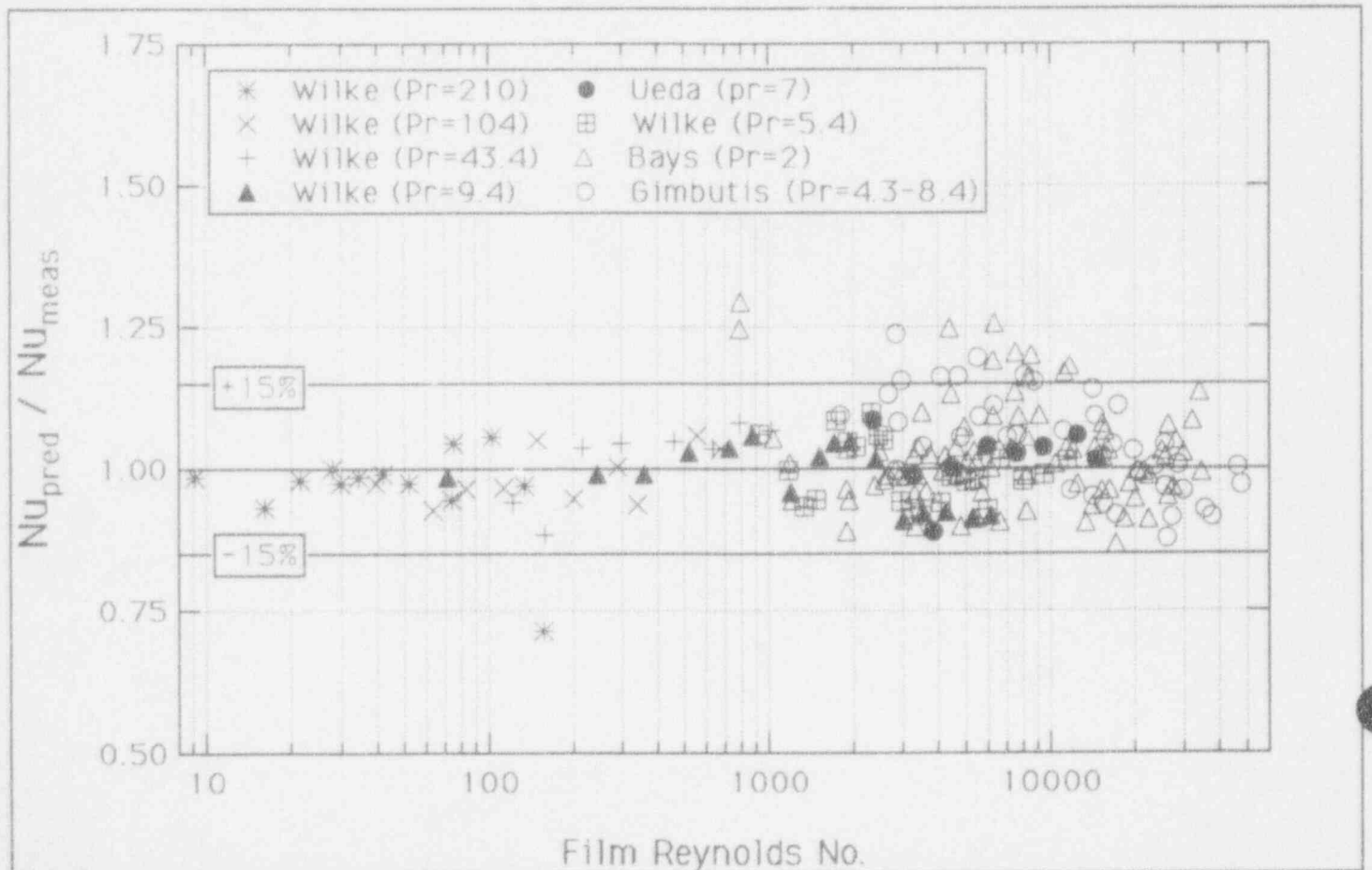
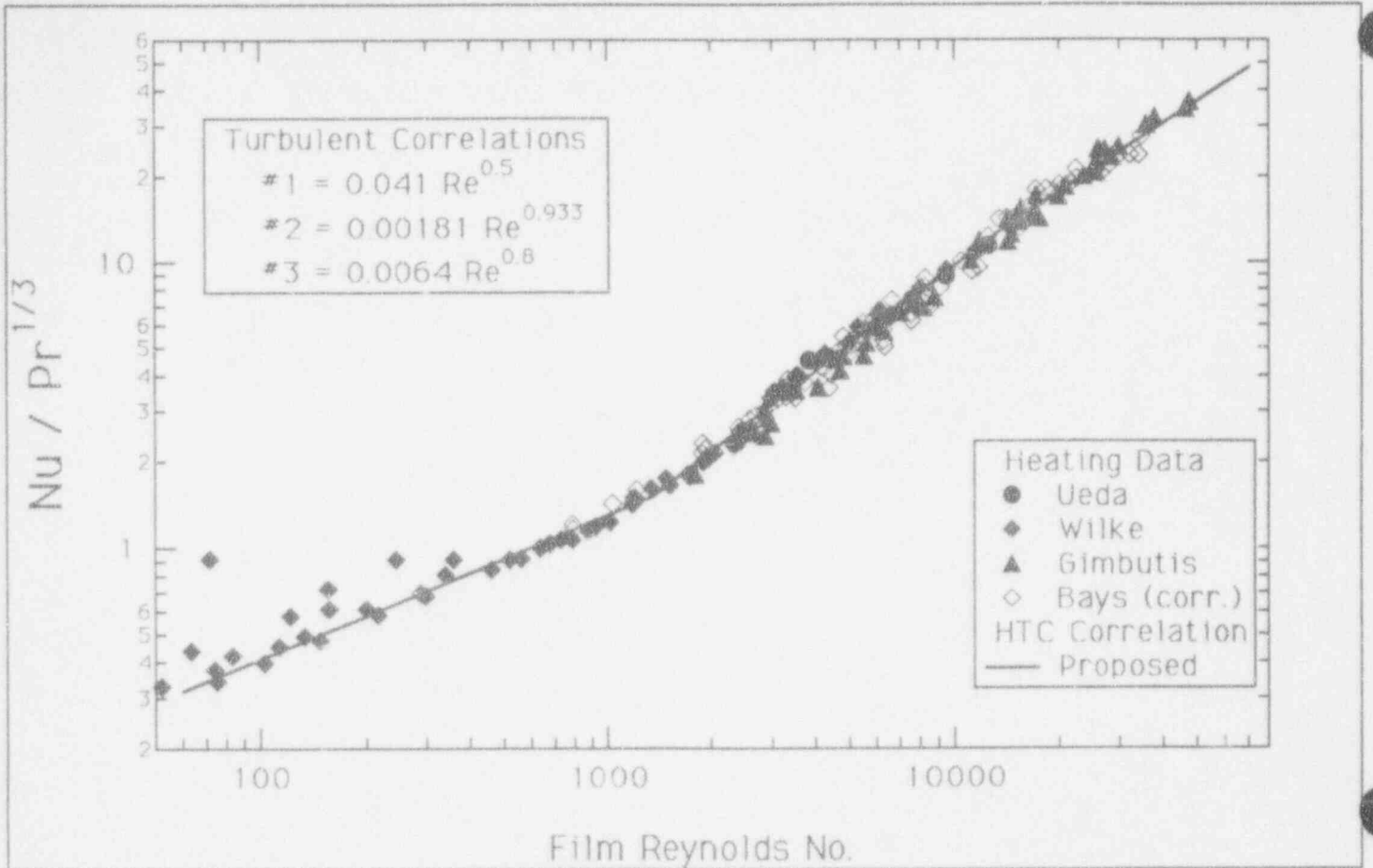
- Colburn (1933):

$$Nu^* = .056 \cdot Re_j^2 \cdot Pr^{\frac{1}{3}}$$

# FALLING FILM HEATING CORRELATIONS



# FALLING FILM: Heating/Cooling





## ● SHEARED FILMS: Laminar-Smooth

\* Dimensionless Film Thickness:

$$m^+ = 0.707 \cdot \text{Re}_f^{\frac{1}{2}}$$

\* Thickness vs. Interfacial Friction:

$$m^* = \frac{0.707 \cdot \text{Re}_f^{\frac{1}{2}}}{\left(\frac{2}{3} \cdot m^* + \tau_i^*\right)^{\frac{1}{2}}}$$

\* Nusselt No. Dependence:

$$\text{Nu}^* = \frac{1}{m^*} = \frac{\left(\frac{2}{3} \cdot m^* + \tau_i^*\right)^{\frac{1}{2}}}{0.707 \cdot \text{Re}_f^{\frac{1}{2}}}$$

## SHEARED FILMS: Laminar-Wavy

- \* Enhancement due to Interfacial Shear:

$$Enh_i = \left( 1 + \frac{3 \cdot \tau_i^*}{2 \cdot m^*} \right)^{\frac{1}{2}}$$

- \* QUESTION: Does Interfacial Shear affect the enhancement due to waves?

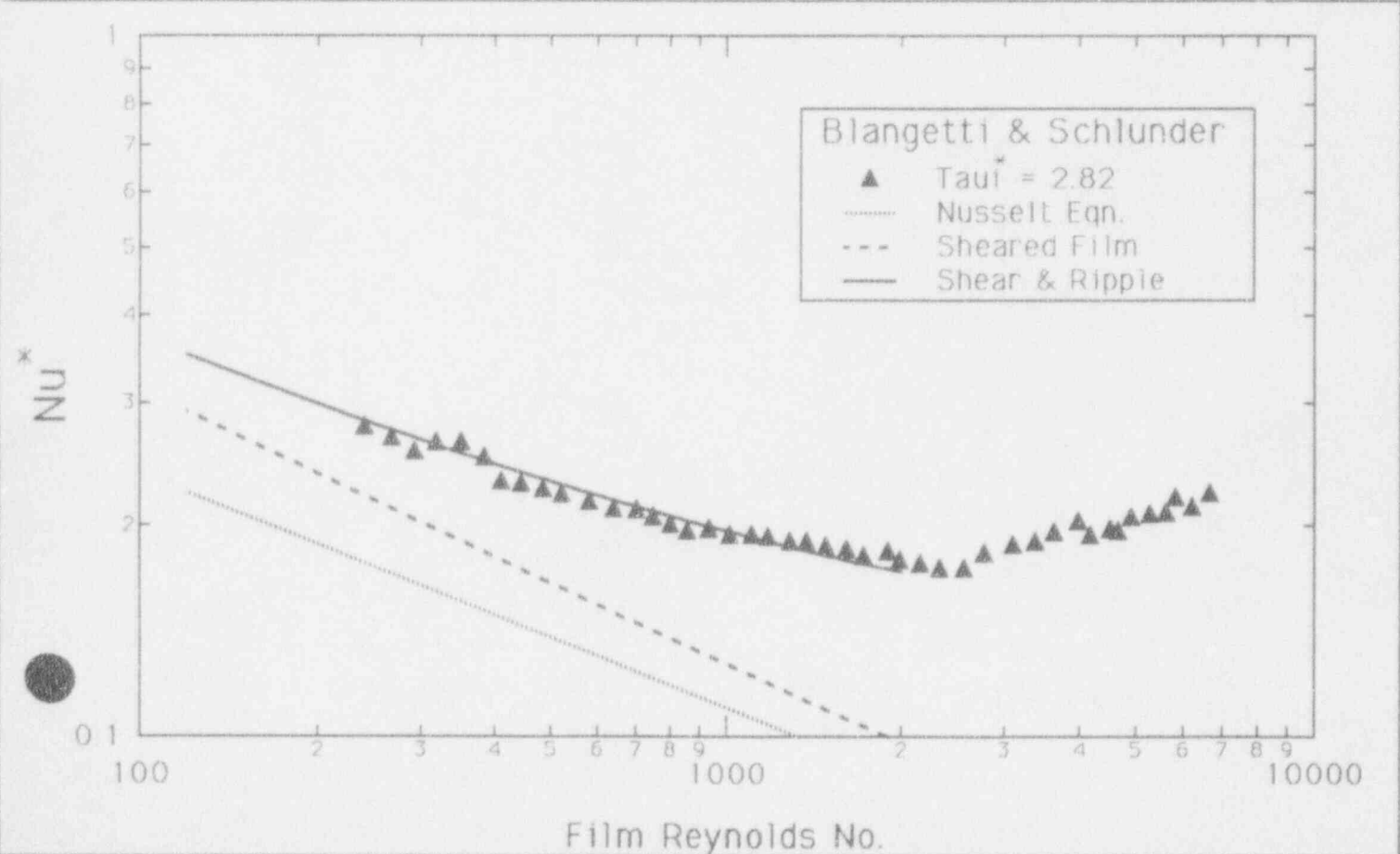
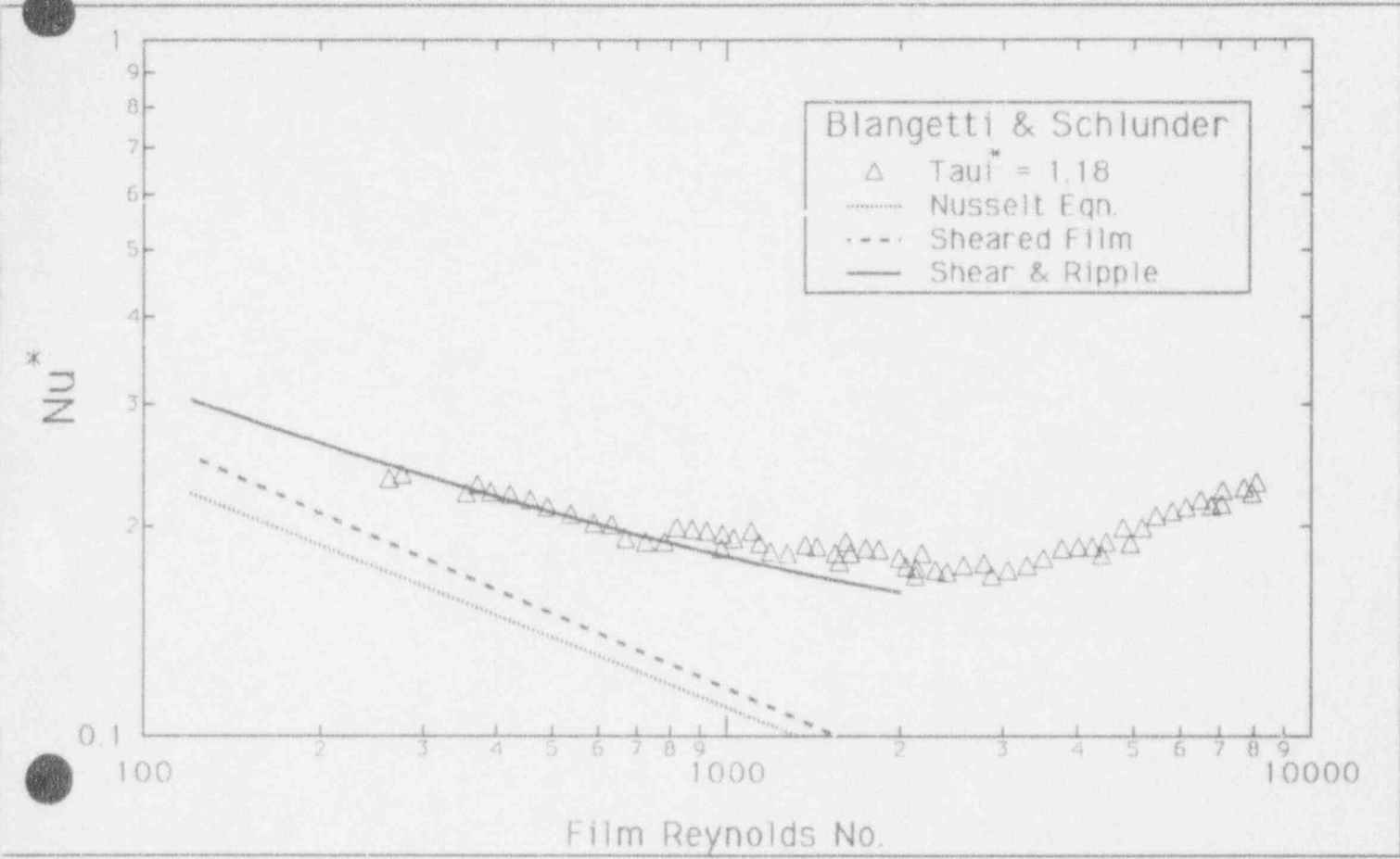
- \* CONCLUSION: for a laminar film the wave and interfacial effects appear to be additive.

$$Nu^* = Nu_0^* \cdot Enh_{wavy} \cdot Enh_i$$

thus,

$$h_w = \frac{k_f}{m} \cdot Enh_{wavy}$$

# SHEARED FILM: Laminar



## SHEARED FILMS: Turbulent

- \* Assume total HT resistance across film is due to 'laminar sublayers' (wall & interface)

$$Nu = \frac{h \cdot m}{k_f} = \frac{m}{y_{sl}}$$

where  $y_{sl}$  represents combined thickness,

$$y_{sl}^+ = \frac{\sqrt{\tau_c / \rho_f}}{v_f}$$

then,

$$y_{sl} = \text{const} \cdot \frac{m}{m^+}$$

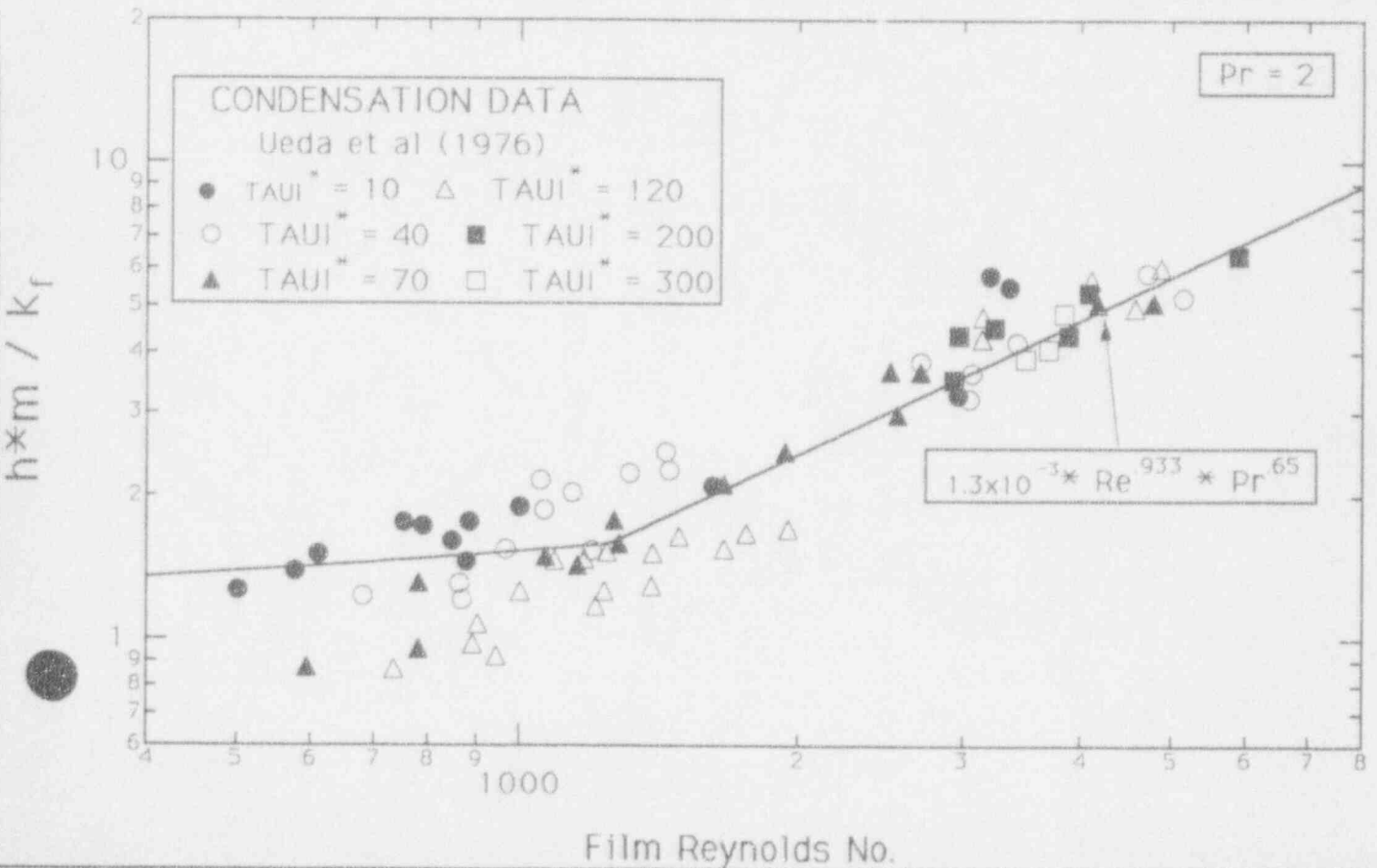
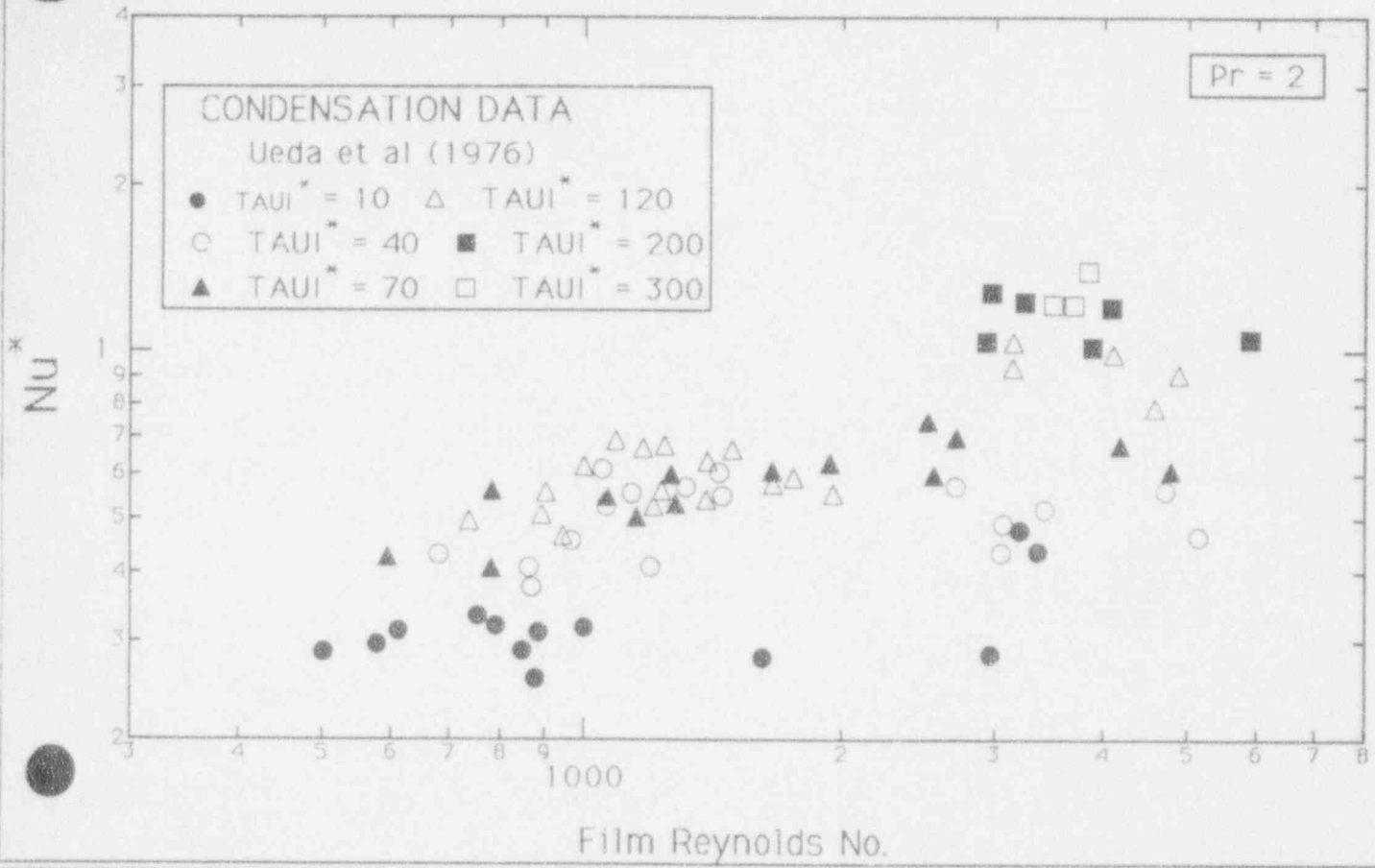
- \* Nusselt No. Behavior:

$$Nu = \text{const} \cdot m^+ = \text{const} \cdot \text{Re}_f^{0.85}$$

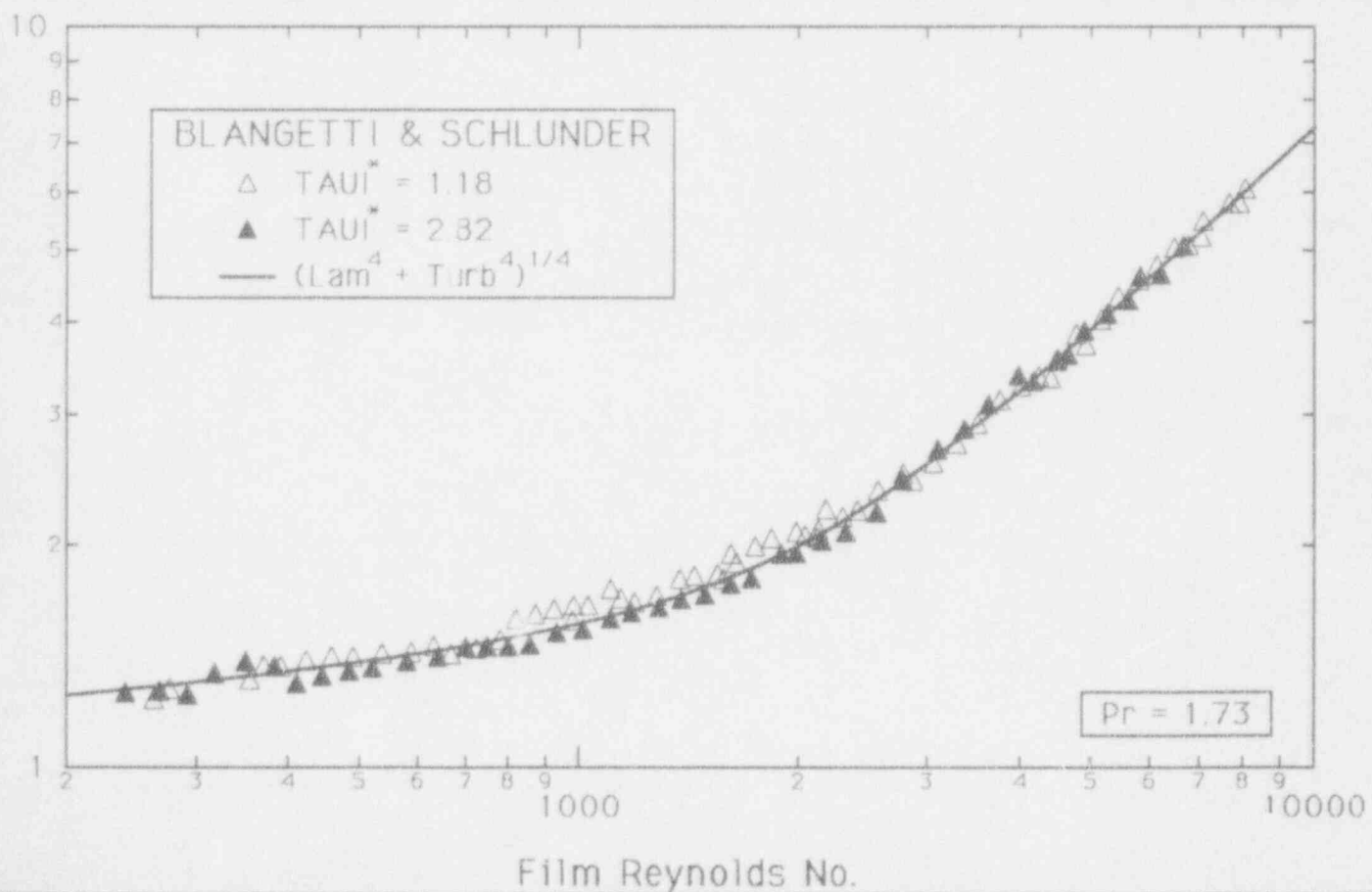
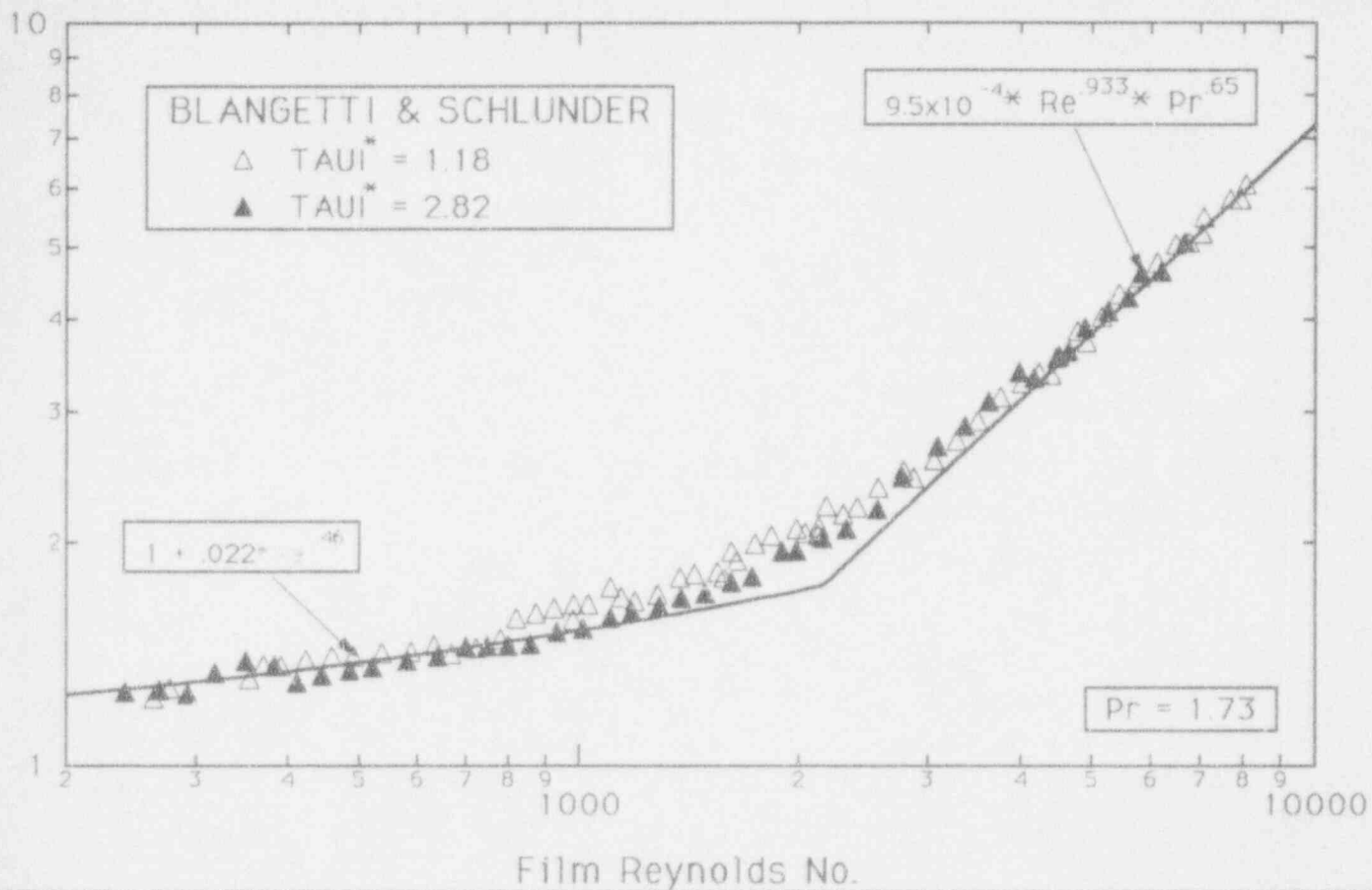
$$Nu^+ = \text{const} \cdot \frac{m^+}{m} = \text{const} \cdot \text{Re}_f^{2.84}$$

→ same for sheared & falling film.

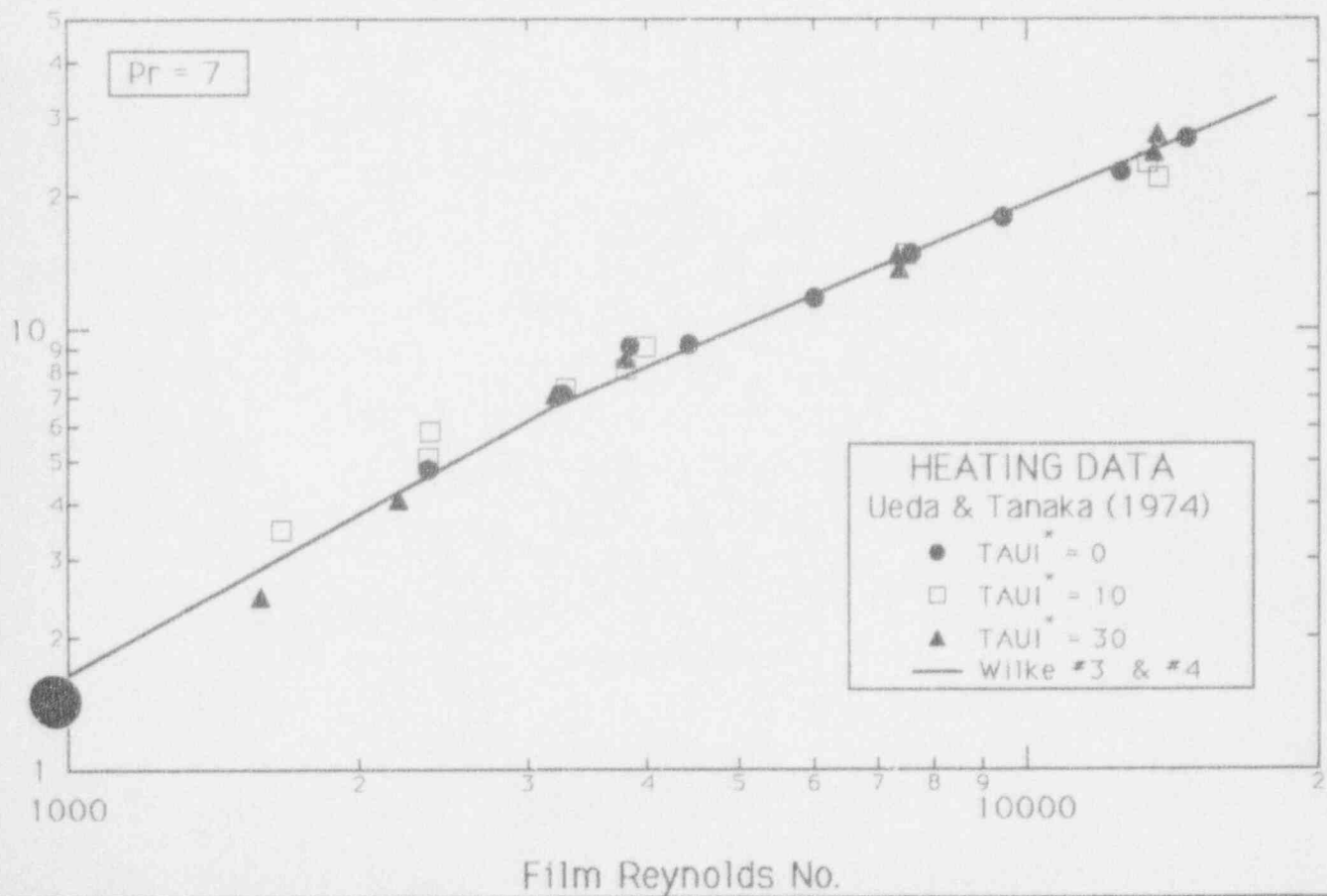
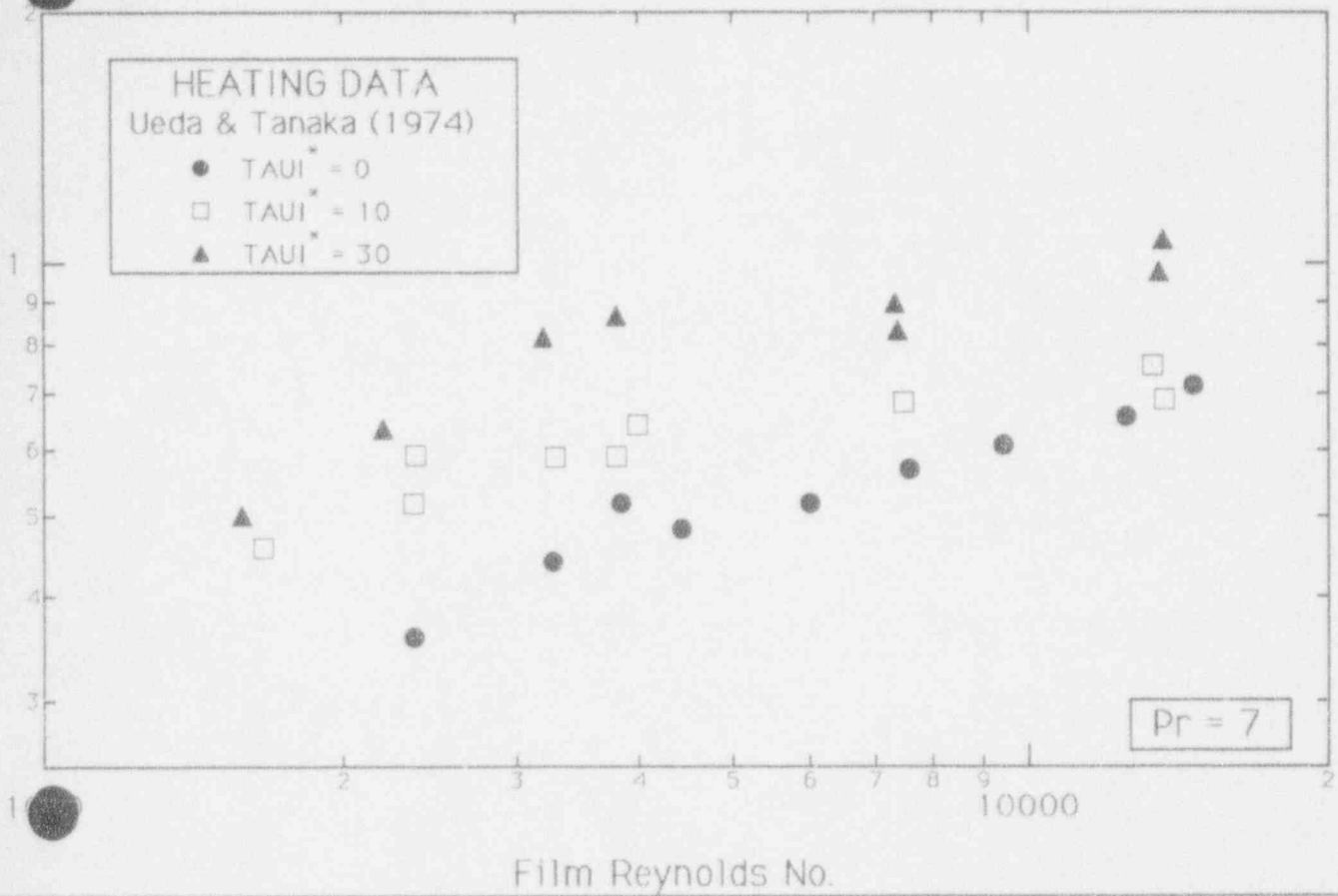
# SHEARED FILMS: Condensation Data



# SHEARED FILMS: Condensation Data



# SHEARED FILMS: Heating Data



# FILM-SIDE HEAT TRANSFER

## \* SUMMARY

### - Laminar Falling Film:

- Condensation Heat Transfer enhancement by surface waves:

$$Enh = 1 + 0.022 \cdot Re_j^{46}$$

- Wall-Fluid Heat Transfer relatively unaffected by waves:

$$Nu = 1.88$$

### - Turbulent Falling Film:

- Condensation data too scattered to determine Reynolds and Prandtl No. dependence.
- Evaporation Heat Transfer data fit by:

$$Nu^* = .0054 \cdot Re_j^{367} \cdot Pr_j^{65}$$

$$Nu = 9 \times 10^{-4} \cdot Re_j^{933} \cdot Pr_j^{65}$$



# FILM-SIDE HEAT TRANSFER

## \* SUMMARY

### - Turbulent Falling Film: (cont.)

- Heating Data well fit by Wilke:

$$1) \quad Nu = 1.9 \qquad Re_j \leq 2460 \cdot Pr^{-.646}$$

$$2) \quad Nu = .0292 \cdot Re_j^{\frac{3}{5}} \cdot Pr^{.344} \qquad 2460 \cdot Pr^{-.646} \leq Re_j \leq 1600$$

$$3) \quad Nu = 2.12 \times 10^{-4} \cdot Re_j^{1.2} \cdot Pr^{.344} \qquad 1600 \leq Re_j \leq 3200$$

$$4) \quad Nu = 1.81 \times 10^{-3} \cdot Re_j^{\frac{14}{15}} \cdot Pr^{.344} \qquad 3200 \leq Re_j \leq 11500$$

where #2 is used for  $Pr > 2$ , and at high  $Re$

$$Nu = .0064 \cdot Re_j^8 \cdot Pr_j^{\frac{1}{2}}$$

# FILM-SIDE HEAT TRANSFER

## \* SUMMARY

### - Sheared Films:

- Important effects of interfacial shear are captured through the film thickness.
- Falling Film Heat Transfer Correlations are applicable to Sheared Films when the Nusselt No. is based on the film thickness.
- Thus:

$$Nu = \frac{h \cdot m}{k_f} = \frac{m}{k_f} \cdot fn\{Re_f, Pr_f\}$$

→ same for sheared & falling film.

# NONCONDENSABLE GASES

## \* MODELING APPROACH:

- Follows Method of Peterson et al (1992) and Kageyama et al (1993).

- Energy Balance at Interface:

$$q_w'' = q_c'' + q_s''$$

where,

$$q_w'' = \text{total heat flux to wall}$$

$$q_c'' = \text{flux of latent heat}$$

$$q_s'' = \text{flux of sensible heat}$$

and,

$$h_w(T_i^s - T_w) = h_c(T_b^s - T_i^s) + h_s(T_b - T_i^s)$$

where,

$$T_i^s = \text{sat. at interface vapor pressure}$$

$$T_b^s = \text{sat. at bulk vapor pressure}$$

$$h_c = \text{HTC for condensation at interface}$$

$$h_s = \text{HTC for convection to interface}$$

# NONCONDENSABLE GASES

## \* ASSUMPTIONS:

- VAPOR/GAS MIXTURE: Perfect Gas
  - Mole concentrations expressed as partial pressures
  
- CLAUSIUS-CLAPEYRON EQN. APPLIES
  - $(P_{vb} - P_{vi}) \rightarrow (T_{vb} - T_{vi})$
  
- VAPOR/LIQUID SPECIFIC VOLUME APPROXIMATED:
  - $v_{lv} \approx \frac{R \cdot T_{ave}}{M_v \cdot P_T \cdot x_{v,ave}}$
  
- DIFFUSION COEFFICIENT APPROXIMATED:
  - $D \cdot \frac{P}{T^2} = D_0 \cdot \frac{P_0}{T_0^2}$

# NONCONDENSABLE GASES

## \* MODEL FORMULATION:

- CONDENSATION 'CONDUCTIVITY': assume  $T_i$  &  $x_i$

$$\bullet k_c = \frac{h_{lv}^2}{\Phi \cdot T_{ave}} \left\{ \frac{M_v^2 \cdot P_0 \cdot D_0}{R^2 \cdot T_0^2} \right\}$$

where,

$$\bullet \Phi = \frac{x_{g,ave}}{x_{v,ave}} = \frac{\text{Ln}[(1 - x_{gb}) / (1 - x_{gi})]}{\text{Ln}[x_{gb} / x_{gi}]}$$

- CONDENSATION HTC:

$$\bullet h_c = \frac{k_c}{d} \cdot Sh$$

- INTERFACE TEMPERATURE:

$$\bullet T_i^s = \frac{h_w \cdot T_w + h_c \cdot T_b^s + h_s \cdot T_b}{h_w + h_c + h_s}$$

- WALL HEAT FLUX:

$$\bullet q_w'' = \frac{h_c(T_b^s - T_w) + h_s(T_b - T_w)}{1 + (h_c + h_s)/h_w}$$

# NONCONDENSABLE GASES

## \* ADVANTAGE:

- Iteration Parameter,  $\Phi$ , is Less Sensitive:
  - better initial guess
  - fewer iterations

## \* FURTHER ASSUMPTION (?)

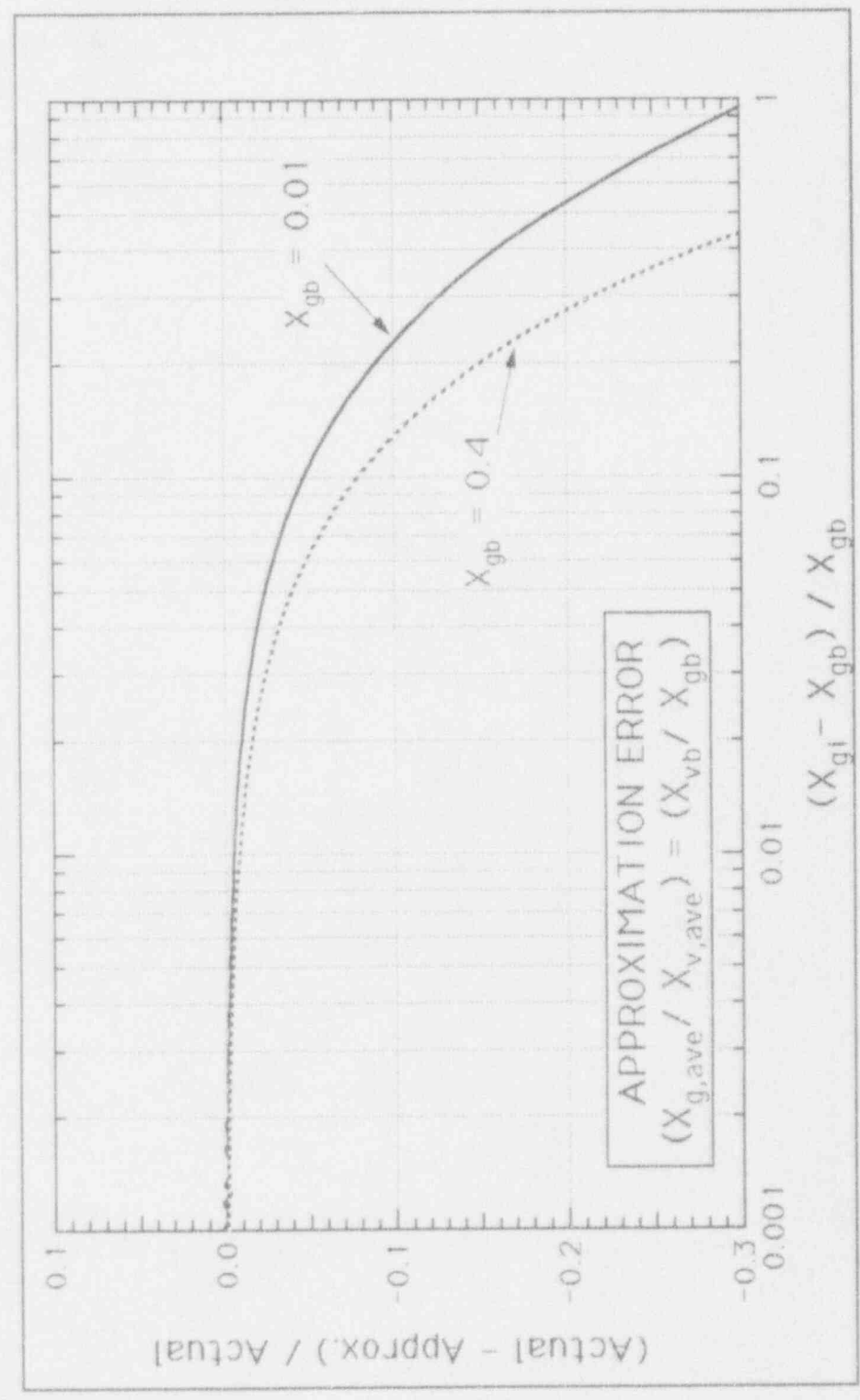
- when NC gas effects are important:
  - $x_{gb}$  is appreciable (i.e. not infinitesimal)
  - Condensation heat flux is not large
  - thus,  $(x_{gi} - x_{gb})/x_{gb}$  is not large
  - then,

$$\Phi = \frac{x_{g,ave}}{x_{v,ave}} \approx \frac{1 - x_{gb}}{x_{gb}}$$

$\Rightarrow$  No Iteration Required !

# NONCONDENSABLE GASES

\* FURTHER APPROXIMATION:



# NONCONDENSABLE GASES

## \* CONSTITUTIVE MODELS REQUIRED:

- Convective HTC:  $h_s$
- Ratio of Sherwood No. to Nusselt No.

## \* Convective HTC: Effects to consider

- Entrance Effects
- Boundary Layer Suction
- Roughness (interfacial waves)
- Mist Formation

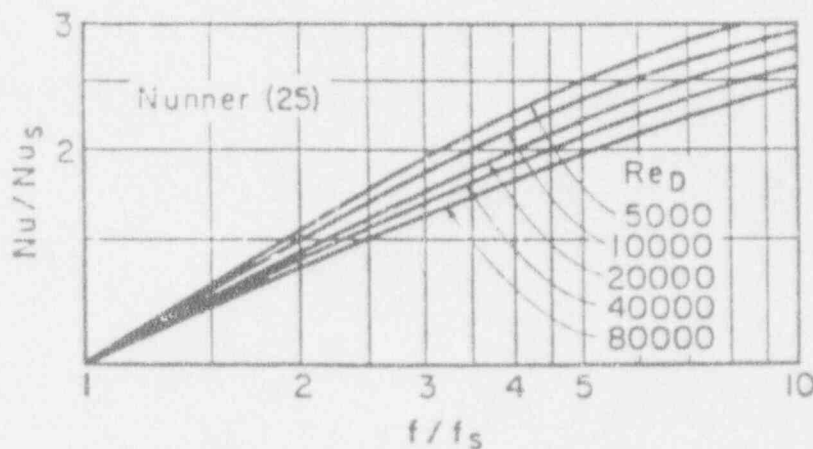


# NONCONDENSABLE GASES

## \* HEAT TRANSFER ENHANCEMENT:

- ENTRANCE EFFECTS: Possibly important for the PCCS and ICS ( $L/D = 50$ ).
- BOUNDARY LAYER SUCTION: probably small, neglect initially.
- ROUGHNESS EFFECT: approximate as

$$\bullet \frac{h_{rough}}{h_{smooth}} = \left( \frac{f_i}{f_s} \right)^{\frac{1}{2}}$$



## - MIST FORMATION:

- Mori & Hijikata (1973):  $h_s \approx 2 \cdot h_{conv}$
- Peterson et al (1992):  $h_s \approx 7 \cdot h_{conv}$

# NONCONDENSABLE GASES

## \* HEAT/MASS TRANSFER ANALOGY:

$$\bullet \quad \frac{Sh}{Nu} = \left( \frac{Sc}{Pr} \right)^n$$

### - Exponent n ?

- Dittus-Boelter (cooling):  $n = 0.4$
- Mass Transfer (gas absorption):  $n = 0.5$
- Gilliland & Sherwood (1934):  $n = 0.44$
- Kageyama et al (1993):  $n = 0.6$

# ASSESSMENT PLANS

## \* DEVELOPMENTAL ASSESSMENT

### - Forced Convection

#### - Pure Steam:

- Goodykoontz & Dorsch (4 tests)
- B & W (2 tests)
- M.I.T. (8 tests)

#### - Noncondensables:

- M.I.T. (Siddique & Husanein)

## \* APPLICATIONS ASSESSMENT

### - CMT Tests (steam, steam & NC)

- Empty (wall condensation)
- Stable Level (wall & interface)
- Drain Test

### - U.C.B. Tests (?)

### - PANTHERS (ICS & PCCS)



*Idaho  
National  
Engineering  
Laboratory*

# Level Tracking Model

Presented by  
Gary W. Johnsen

Advisory Committee on  
Reactor Safeguards  
January 4, 1994  
Bethesda, MD

# Outline

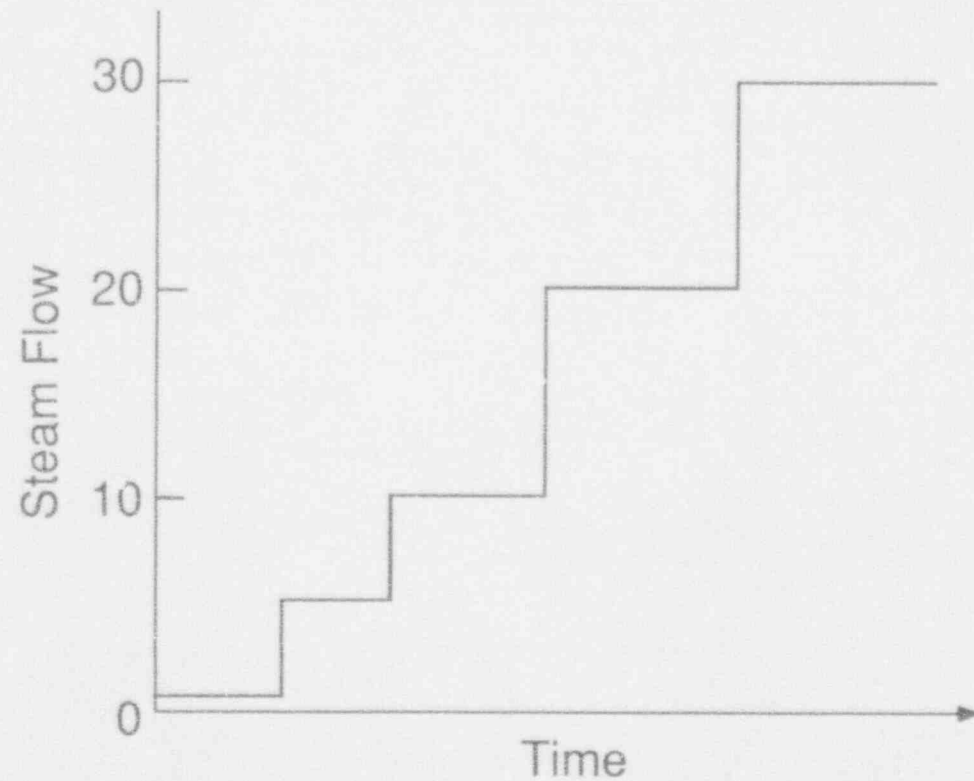
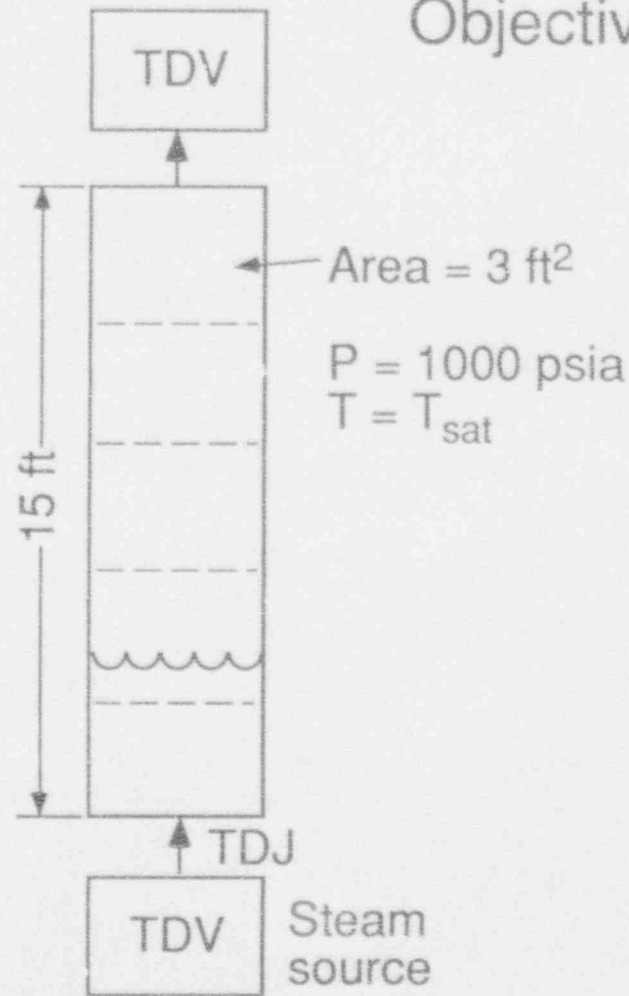
- Level Tracking versus Vertical Stratification Models
- Preliminary Verification Results
- Description of the New Level Tracking Model
- Planned Assessment

# Level Tracking versus Vertical Stratification Models

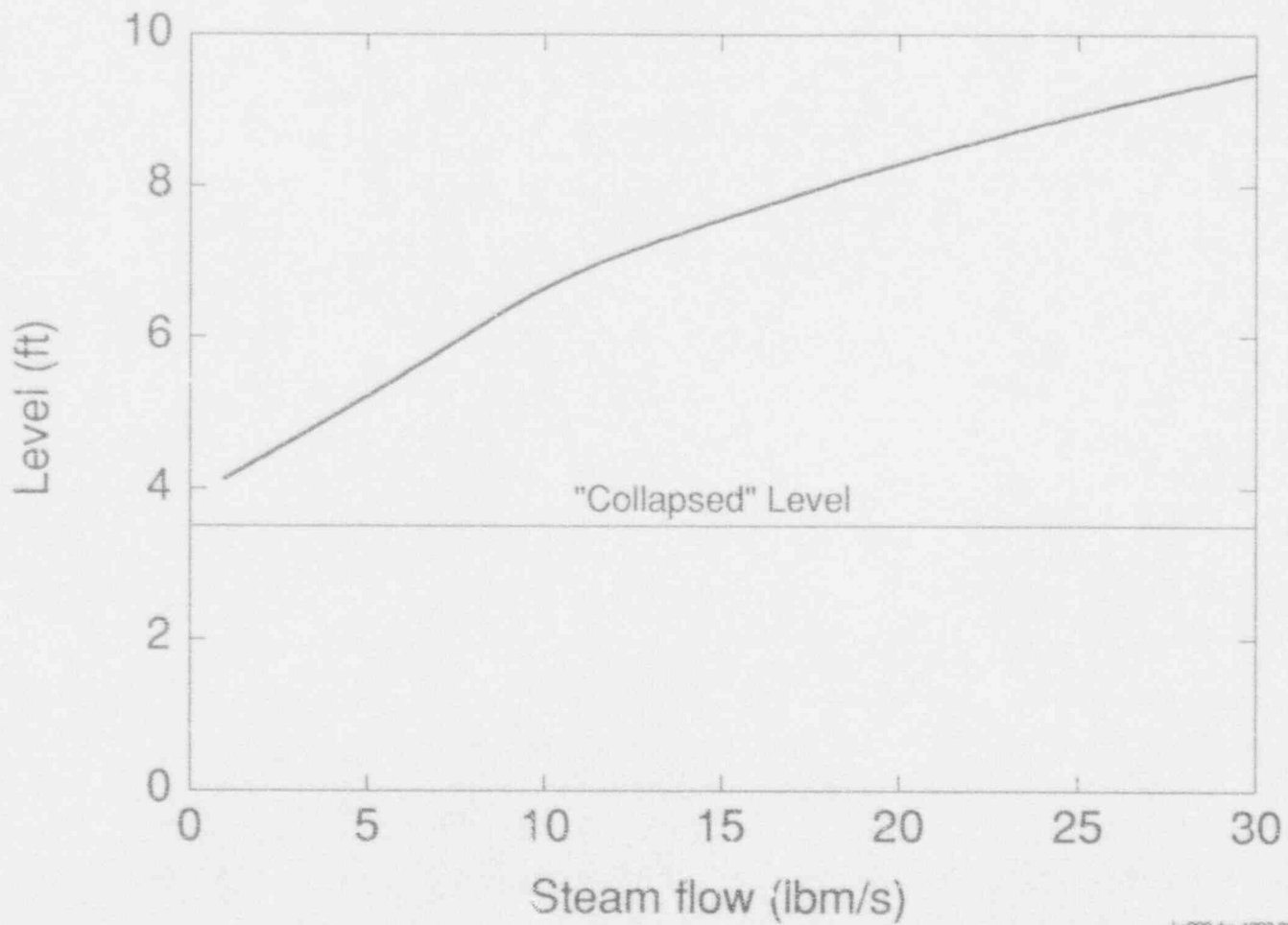
- The model described in Vol. 4 is a vertical stratification model, not a level tracking model
- The need for a true level tracking model was recognized in the identification of modeling capabilities for the advanced light water reactors, particularly for modeling the CMT
- A new model has been implemented and will be described

# Level Tracking Test Problem— Bubbling Steam Through Liquid

Objective: Verify Functionality

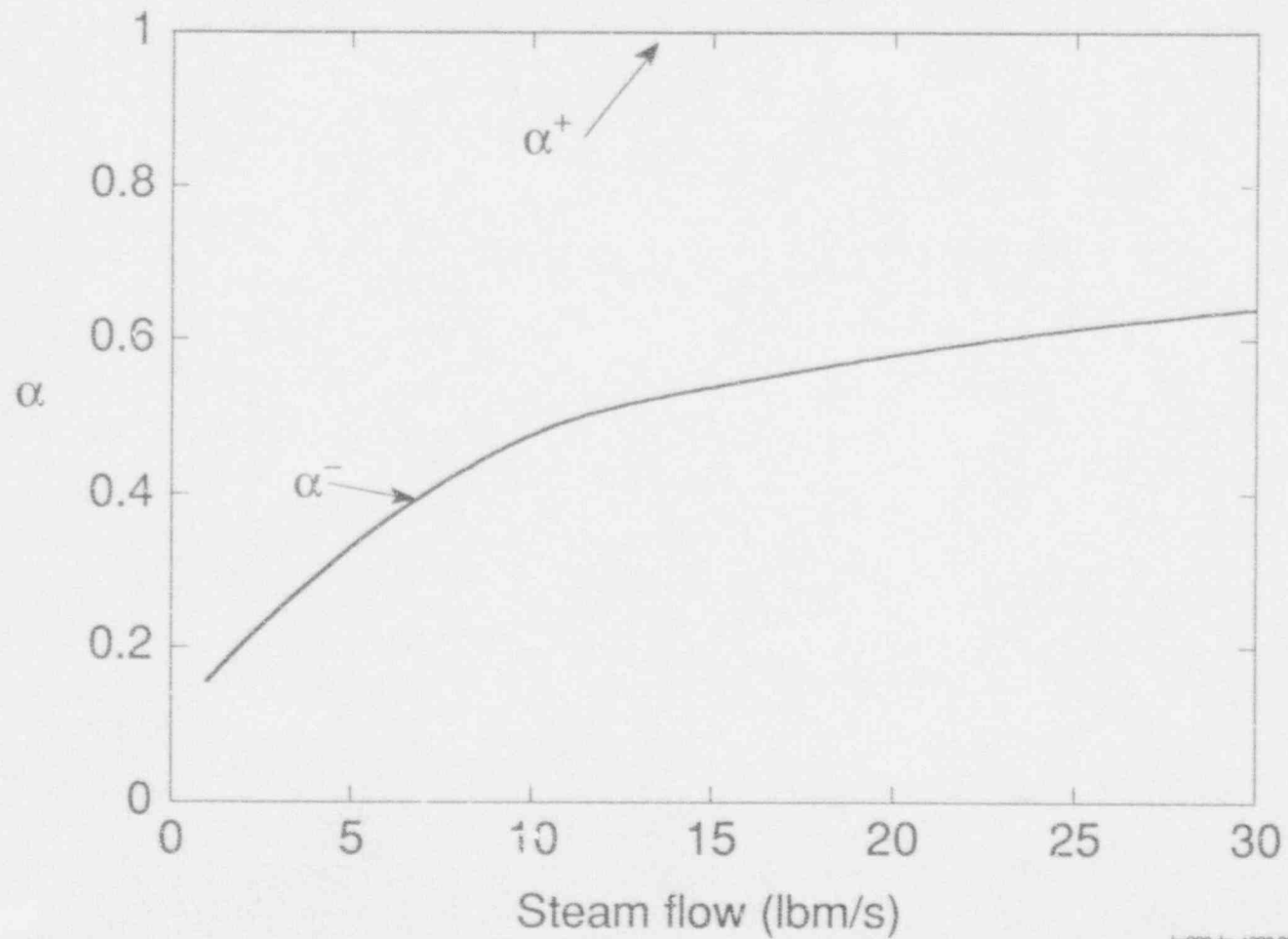


## Mixture Level vs. Steam Flow





## Void Fraction Above and Beneath Level vs. Steam Flow



# Level Tracking Model

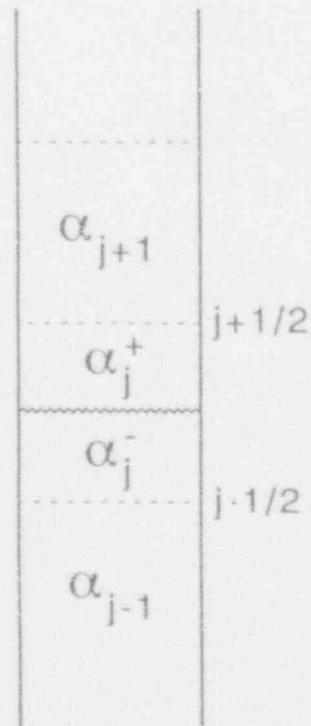
- Purpose:

Recognize and account for the effects of a sharp void gradient representing a mixture level in a vertical component

- Principal features:

- Level detection
- Calculation of mixture level parameters
- Alteration of convected terms
- Alteration of wall heat transfer

## Mixture Level Detection Logic (Focusing on volume j)



Criteria for a mixture level:

1. No level in  $j+1, j-1$

2.  $\alpha_{j+1} - \alpha_j > 0.2$

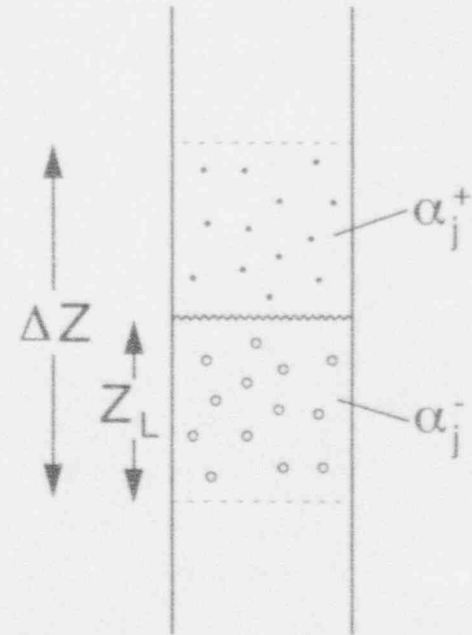
or

$\alpha_j - \alpha_{j-1} > 0.2$

3.  $\alpha_{j+1} > 0.7$

## Calculated Mixture Level Parameters

- Void fractions above and below level
- Location of level within cell
- Velocity of level



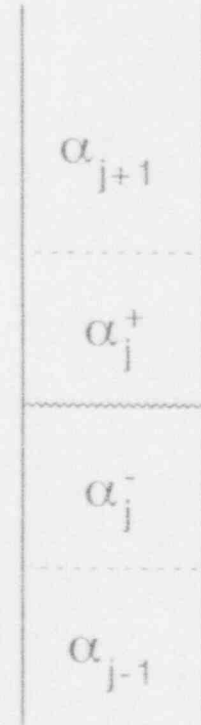
# Void Fractions Above And Below Level

- $\alpha_j^- = \alpha_{j-1}$
- $\alpha_j^+ = \alpha_{j+1}$  (no entrainment case)  
 $= 1 - \frac{G_{ent}}{\rho_f V_f}$  (entrainment case)\*

- Bounding limits

$$\alpha_j^- \leq \alpha_j$$

$$\alpha_j^+ \geq \alpha_j$$

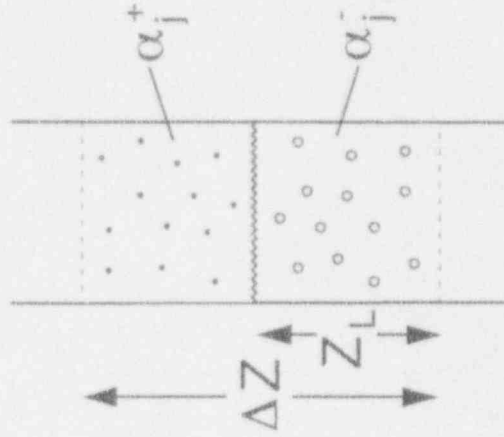


\* $G_{ent}$  is taken from A. Rosen, et al, Teploenergetika, No. 11, p. 59, 1976.

## Position and Velocity of Level

$$\bullet Z_L = \Delta Z \left( \frac{\alpha_j^+ - \alpha_j}{\alpha_j^+ - \alpha_j^-} \right)$$

$$\bullet \dot{Z}_L = \frac{\Delta Z \frac{d\alpha_j}{dt} - Z_L \frac{d\alpha_j^-}{dt} - (\Delta Z - Z_L) \frac{d\alpha_j^+}{dt}}{\alpha_j^- - \alpha_j^+}$$



$\alpha_j$  = avg. void fraction in cell j

## Consistency of the Level Model And the Jump Condition

- Position of the mixture level

$$Z_L = \Delta Z \left( \frac{\alpha_j^+ - \alpha_j}{\alpha_j^+ - \alpha_j^-} \right) \quad (1)$$

- Mixture level velocity by differentiation

$$\dot{Z}_L = \frac{\Delta Z \frac{d\alpha_j}{dt} - Z_L \frac{d\alpha_j^-}{dt} - (\Delta Z - Z_L) \frac{d\alpha_j^+}{dt}}{\alpha_j^- - \alpha_j^+} \quad (2)$$

- Assumption:  $\frac{d\alpha_j^-}{dt} = 0, \frac{d\alpha_j^+}{dt} = 0$  (3)

- Simplified form of (2)

$$\dot{Z}_L = \frac{\Delta Z \frac{d\alpha_j}{dt}}{\alpha_j^- - \alpha_j^+} \quad (4)$$

## Consistency of the Level Model And the Jump Condition (cont'd)

- Total Average Volume Void Time Derivative

$$\frac{d\alpha_j}{dt} = \frac{A}{V} (j_g^- - j_g^+) \quad (5)$$

$$V = A\Delta Z \quad (6)$$

$$\frac{d\alpha_j}{dt} = \frac{1}{\Delta Z} (j_g^- - j_g^+) \quad (7)$$

- Substitute (7) into (4)

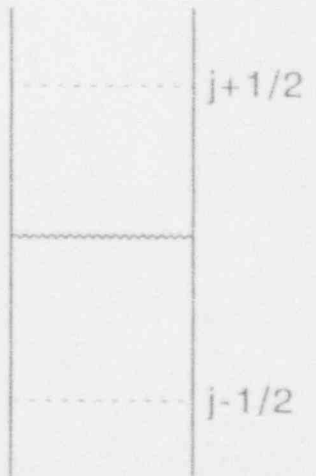
$$\dot{Z}_L = \frac{j_g^- - j_g^+}{\alpha_j^- - \alpha_j^+}$$



# Alteration of Convected Terms

- Without Level Tracking model, RELAP5 convects cell-centered, average volume properties
- Level Tracking model alters the convected void in the conservation equations:

<u>Junction</u>	<u><math>V_g</math></u>	<u><math>\alpha_g</math></u>	<u><math>V_f</math></u>	<u><math>\alpha_f</math></u>
$j+1/2$	$> 0$	$\alpha_j^+$	$> 0$	$1 - \alpha_j^+$
	$< 0$	$\alpha_{j+1}$	$< 0$	$1 - \alpha_{j+1}$
$j-1/2$	$> 0$	$\alpha_{j-1}$	$> 0$	$1 - \alpha_{j-1}$
	$< 0$	$\alpha_j^-$	$< 0$	$1 - \alpha_j^-$



## Alteration of Hydrostatic Head Terms

- Without Level Tracking model, RELAP5 presumes homogeneous flow for computing hydrostatic head term

$$\Delta P_H = \frac{\rho_{j+1} g \Delta Z_{j+1} + \rho_j g \Delta Z_j}{2}$$

- Level Tracking model produces a more accurate computation using the position of the level

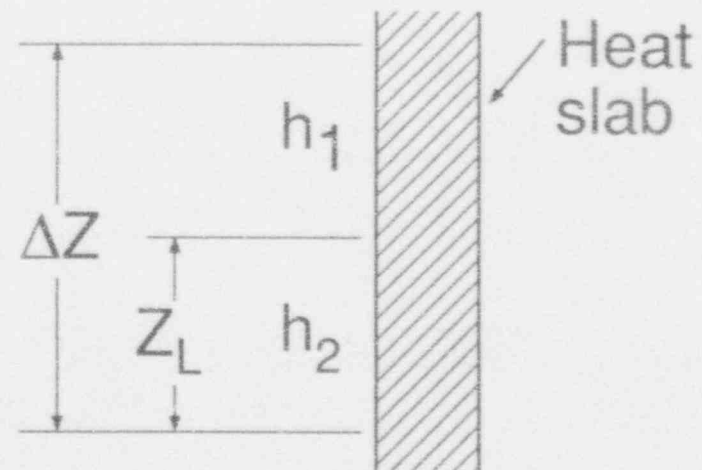
# Alteration of Heat Transfer Calculation

- Without Level Tracking model, wall heat transfer uses volume average properties and parameters
- Level Tracking model "partitions" heat transfer based on position of level

$$q_{WF} = h_2 (1 - \phi) (T_W - T_F)$$

$$q_{WG} = h_1 \phi (T_W - T_G)$$

$$\phi = \frac{\Delta Z - Z_L}{\Delta Z}$$



# Planned Assessment

- G.E. Level Swell
  - THTF Boiloff
  - CMT, ROSA, OSU, SPES Tests
- } In-Process



# FILM CONDENSATION MODELING

J.M. Kelly

January 4, 1994

# FILM CONDENSATION MODEL

## \* OBJECTIVE:

- Develop a modeling approach for film condensation heat transfer within the framework of the two-fluid model.

## \* FOCUS:

- Address concerns of NRC T/H consultants and of the ACRS, specifically
  - Heat Flux Apportioning (Liquid/Vapor)
  - Surface Partitioning (Wet/Dry)
  - Effect of Noncondensables

# CONTENTS

← INTRODUCTION

← BACKGROUND: Two-Fluid Model

← MODELING CONCEPT

← FILM THICKNESS

← FILM-SIDE HEAT TRANSFER RESISTANCES

\* NONCONDENSABLE GASES

\* ASSESSMENT PLAN

\* SUMMARY

# INTRODUCTION

## \* WHERE IS FILM CONDENSATION POTENTIALLY IMPORTANT?

### - AP600:

- CMT WALLS: Drain Period
- PRHR TUBES: ATWS & SBLOCA (1st ADS)
- SG TUBES: Reflux Cooling (?)
- HORIZONTAL PIPES (?)
- CONTAINMENT RESPONSE (CONTAIN)

### - SBWR:

- ICS TUBES:
- PCCS TUBES:
- CONTAINMENT DRYWELL (CONTAIN)



# INTRODUCTION

## What are the Regimes of Interest?

---

Component	Pressure (bar)	Vapor Re	Film Re	Noncondesable Gas Fraction (%)
CMT	5 - 75	1E+3 - 1E+4	< 5E+4	?
PRHR	70, 140	< 1E+5	< 3E+4	0
ICS	70 - 80	1E+3 - 1E+4	500 - 1000	0 - 33
PCCS	2 - 5	1E+3 - 1E+4	250 - 1700	0 - 40

NOTE: Values are extremely approximate and intended for illustration only.

## BACKGROUND

- \* PHYSICAL REPRESENTATION
- \* TWO-FLUID REPRESENTATION
- \* HEAT FLUX APPORTIONING
- \* SURFACE PARTITIONING
- \* NONCONDENSABLE GAS EFFECTS

# BACKGROUND

## \* PHYSICAL REPRESENTATION

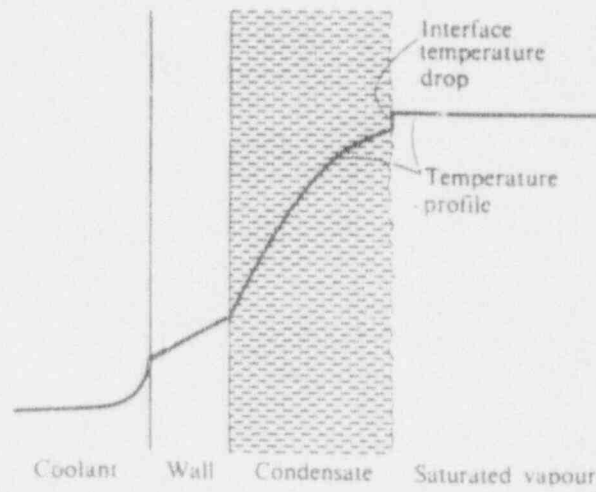


FIG. 18.2. Temperature profile for condensing a pure saturated vapour.

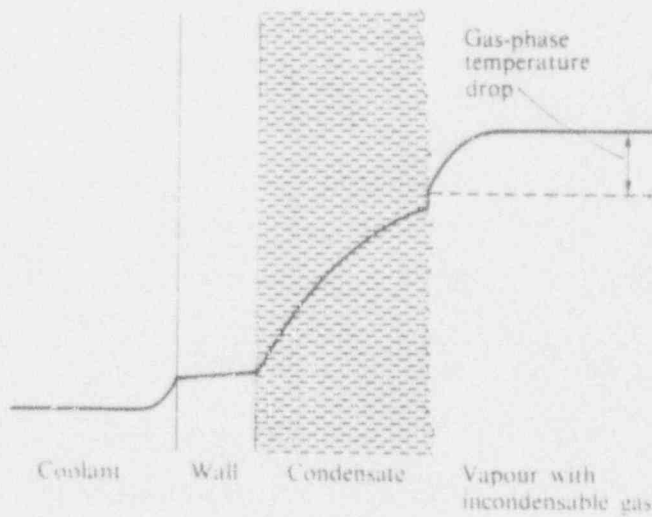


FIG. 18.3. Temperature profile for condensing a vapour which contains incondensable gas.

# BACKGROUND

## \* TWO-FLUID REPRESENTATION

- WALL HEAT TRANSFER RATE:

$$q_w = q_{wl} + q_{wv} + q_{wi} \Rightarrow \text{Wall Conduction}$$

where:

$$q_{wl} = h_{wl} \cdot A_w \cdot (T_w - T_l) \Rightarrow \text{Liquid Energy Eqn.}$$

$$q_{wv} = h_{wv} \cdot A_w \cdot (T_w - T_v) \Rightarrow \text{Vapor Energy Eqn.}$$

and,

$q_{wi}$  is the heat transfer to the interface in a near-wall fluid layer where the local fluid temperature is significantly different from the bulk temperature (e.g. subcooled nucleate boiling).

NOTE: Typically, only one of the three wall heat flux terms,  $q_{wl}$ ,  $q_{wv}$ , or  $q_{wi}$ , is non-zero.

# BACKGROUND

## \* TWO-FLUID REPRESENTATION

- CONDENSATION RATE:

$$\Gamma = \frac{q_{li} + q_{vi}}{h_{vs} - h_{ls}} + \Gamma_w$$

where:

$$q_{li} = h_{li} \cdot A_i \cdot (T_l - T_i)$$

$$q_{vi} = h_{vi} \cdot A_i \cdot (T_v - T_i)$$

and,

$\Gamma_w$  is the mass transfer due to  $q_w$

- EXAMPLE FILM CONDENSATION: (Pure Steam)

$$T_i = T_{sat} = T_v$$

$$q_{vi} = q_{wv} = q_{wi} = 0$$

$$q_w = q_{wl} = q_{li}$$

$$\Gamma = \frac{q_w}{(h_{vs} - h_{ls})}$$

# BACKGROUND

## \* HEAT FLUX APPORTIONING

### - RELAP5 MOD 3.0:

#### - Wall Heat Transfer Rate:

$$q_w = h_{cond} \cdot A_w \cdot (T_w - T_{sat})$$

$$q_{wl} = q_w$$

$$q_{wv} = q_{wi} = 0$$

#### - Condensation Rate:

$$\Gamma = \frac{q_w}{(h_{vs} - h_{ls})} = \frac{h_{li} \cdot A_i \cdot (T_l - T_{sat})}{(h_{vs} - h_{ls})}$$

#### - Numerical Problem:

- use of  $T_{sat}$  rather than  $T_l$  as the sink temperature decouples wall HT from fluid energy eqn. This results in 'freezing' liquid.

# BACKGROUND

## \* HEAT FLUX APPORTIONING

### - RELAP5 MOD 3.1:

#### - Wall Heat Transfer Rate:

$$q_w = h_{cond} \cdot A_w \cdot (T_w - T_{sat})$$

but,

$$q_{wl} = h_{cond} \cdot A_w \cdot (T_w - T_l)$$

$$q_{wv} = q_w - q_{wl}$$

#### - Condensation Rate:

$$\Gamma = \frac{q_{wl} + q_{wv}}{(h_{vs} - h_{ls})}$$

$$\Gamma = \frac{h_{li} \cdot A_l \cdot (T_l - T_{sat}) + h_{vi} \cdot A_i \cdot (T_v - T_{sat})}{(h_{vs} - h_{ls})}$$

#### - Numerical Problem (Mod 3.0) Solved

#### - Problem: Not Physically Based

- Vapor subcooling
- Liquid temperature depends on  $h_{cond}/h_{li}$ , and these are not consistent.

# BACKGROUND

## \* SURFACE PARTITIONING

### - RELAP5 MOD 3.0:

- Minimum Film Thickness: Fraction of surface wet
- Wet Surface: Film condensation
- Dry Surface: SPV convection

### - RELAP5 MOD 3.1.1:

- No surface partitioning, if  $(T_w < T_{sat})$ , then condensation heat transfer with ramp to single phase liquid.



# BACKGROUND

## \* NONCONDENSABLE GAS EFFECTS

### - RELAP5 MOD 3.0:

- Degradation factor model used, but of unknown origin and limited assessment.

### - RELAP5 MOD 3.1.1:

#### - UCB Vierow-Schrock Model:

- Film-Side: Nusselt with empirical flow factor
- Gas-Side: empirical degradation factor

#### - High Pressure Option:

- Film-Side:  $\text{Max}[\text{Nusselt}, \text{Shah}]$
- Gas-Side: Colburn-Hougen type model
  - no sensible heat transfer
  - Gilliland correlation for mass transfer.

# FILM CONDENSATION MODEL

## \* CONCEPT

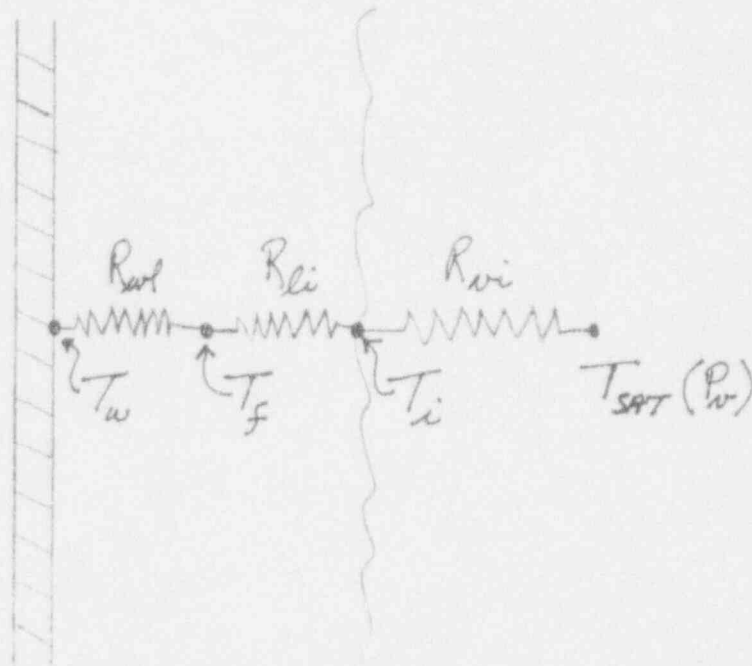
- Model Objectives:
  - Condensation Rate
  - Film Subcooling
  - Large Data Base
  
- Model Constraints:
  - Physically Based
  - Compatible with Two-Fluid
  - Avoid Numerical Problems

## \* PROBLEMS

- Condensation HTC's:
  - use  $T_{sat}$  as sink temperature
  
- Interfacial HTC's:
  - how to specify  $h_{ij}$
  
- Noncondensable Gas Effects:
  - better model needed

# FILM CONDENSATION MODEL

## \* CONCEPT



- Use Heat Transfer Resistances in Series:
  - $R_{wl}$ : wall - liquid
  - $R_{li}$ : liquid - interface
  - $R_{vi}$ : vapor - interface

# FILM CONDENSATION MODEL

## \* CONCEPT

### - Film-Side Resistance:

#### - Condensation HTC:

$$h_{cond} \cdot A_w = \frac{1}{R_{wl} + R_{li}}$$

#### - Film Heating/Cooling HTC:

$$h_{wl} = h_{heat}$$

$$R_{wl} = \frac{1}{h_{heat} \cdot A_w}$$

#### - Infer Liquid-Interface Resistance:

$$R_{li} = \frac{1}{h_{cond} \cdot A_w} - R_{wl}$$

#### - Use Mass Transfer Data as a Check?

- Data scatters widely .
- $Sc \ll Pr$ , penetration depth is much less than thermal boundary layer, different controlling phenomena ?

# FILM CONDENSATION MODEL

## \* CONCEPT

### - Vapor-Side Resistance:

- Pure Steam: Negligible

$$Nu_{vi} = 10^4 \cdot F_6 \cdot F_{10}$$

where

$F_6$  = smoothing between SCV and SHV

$F_{10}$  = smoothing as film disappears

- Vapor with Noncondensables:
  - use turbulent mass transfer coefficient
  - find interface temperature,  $T_i$

NOTE: From kinetic theory (Butterworth & Hewitt, 1977)

$$h_{vi} \approx 10^7 \text{ (W/m}^2 \cdot \text{K)}$$

for pure water vapor at 1 bar.

# FILM THICKNESS

## \* IMPORTANCE:

- Heat transfer resistances scale with film thickness.

## \* FALLING FILMS

- Laminar Smooth
- Laminar Wavy
- Turbulent
- Proposed Model

## \* SHEARED FILMS: Cocurrent Downflow

- Laminar Smooth
- 'Universal' Dimensionless Thickness
- Interfacial Shear Model
- Effect of Entrainment
- Assessment Results

# FILM THICKNESS

## \* FALLING FILMS

- Laminar Smooth: Nusselt

$$m^* = \left(\frac{3}{4} \cdot \text{Re}_f\right)^{\frac{1}{3}}$$

where

$$m^* = m \cdot \left(\frac{\rho_f \cdot \Delta\rho \cdot g}{\mu_f^2}\right)^{\frac{1}{3}}$$

$$\text{Re}_f = \frac{4\Gamma'}{\mu_f}$$

and,

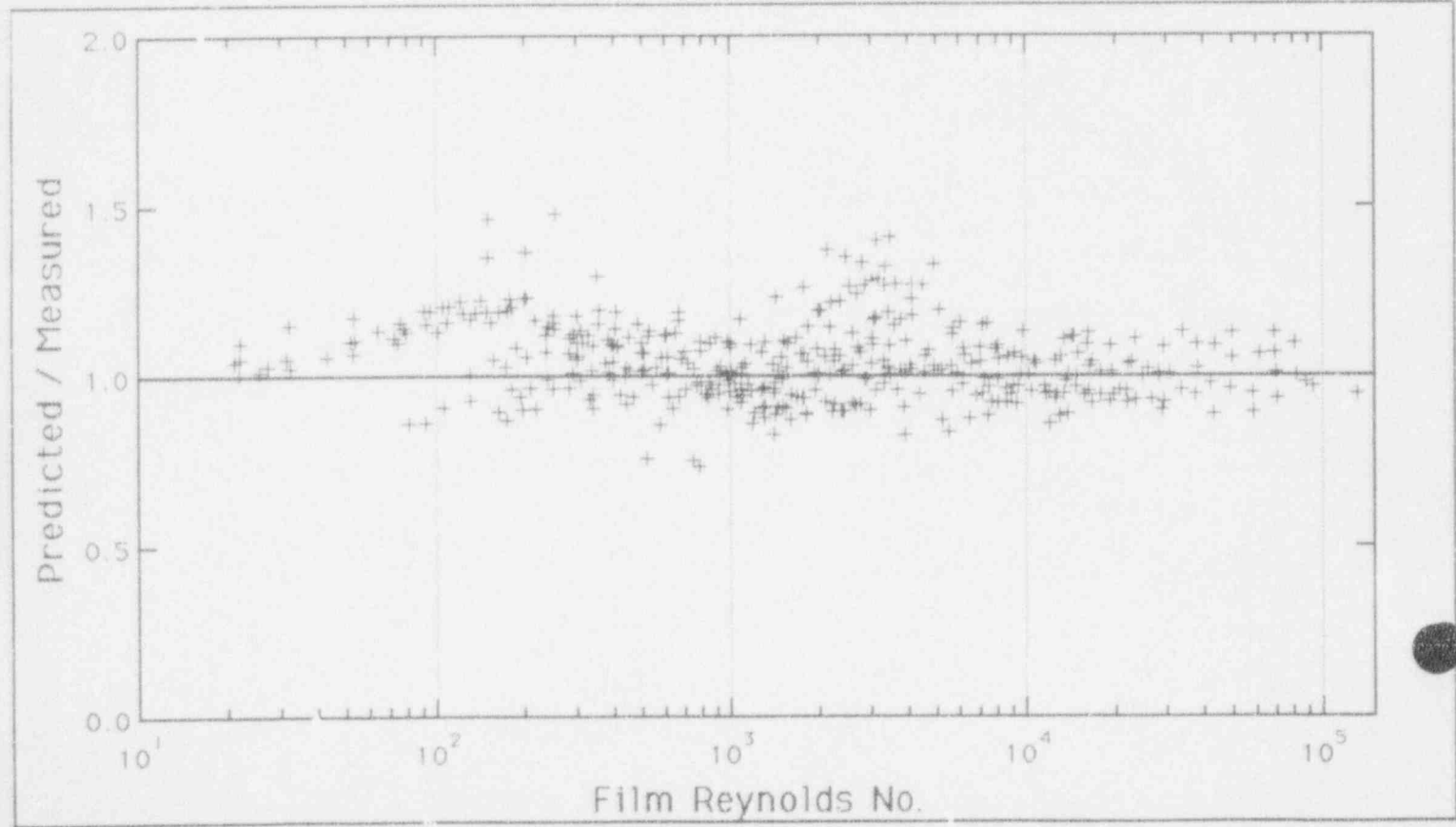
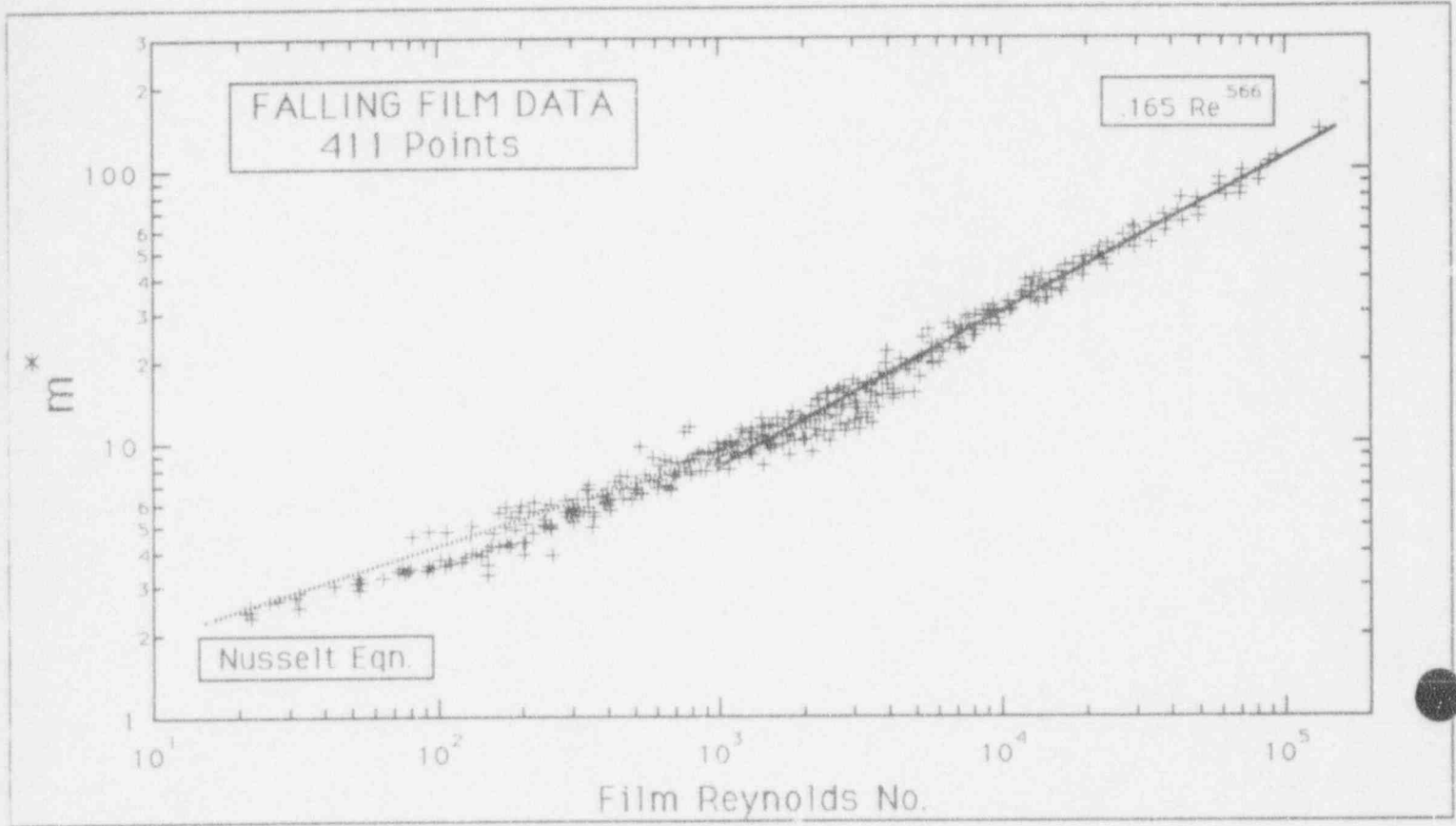
$\Gamma'$  is the flowrate per unit perimeter

- Laminar Wavy: Kapitza (1948)

$$m^* = \left(\frac{24}{4} \cdot \text{Re}_f\right)^{\frac{1}{3}}$$

# FALLING FILM: THICKNESS

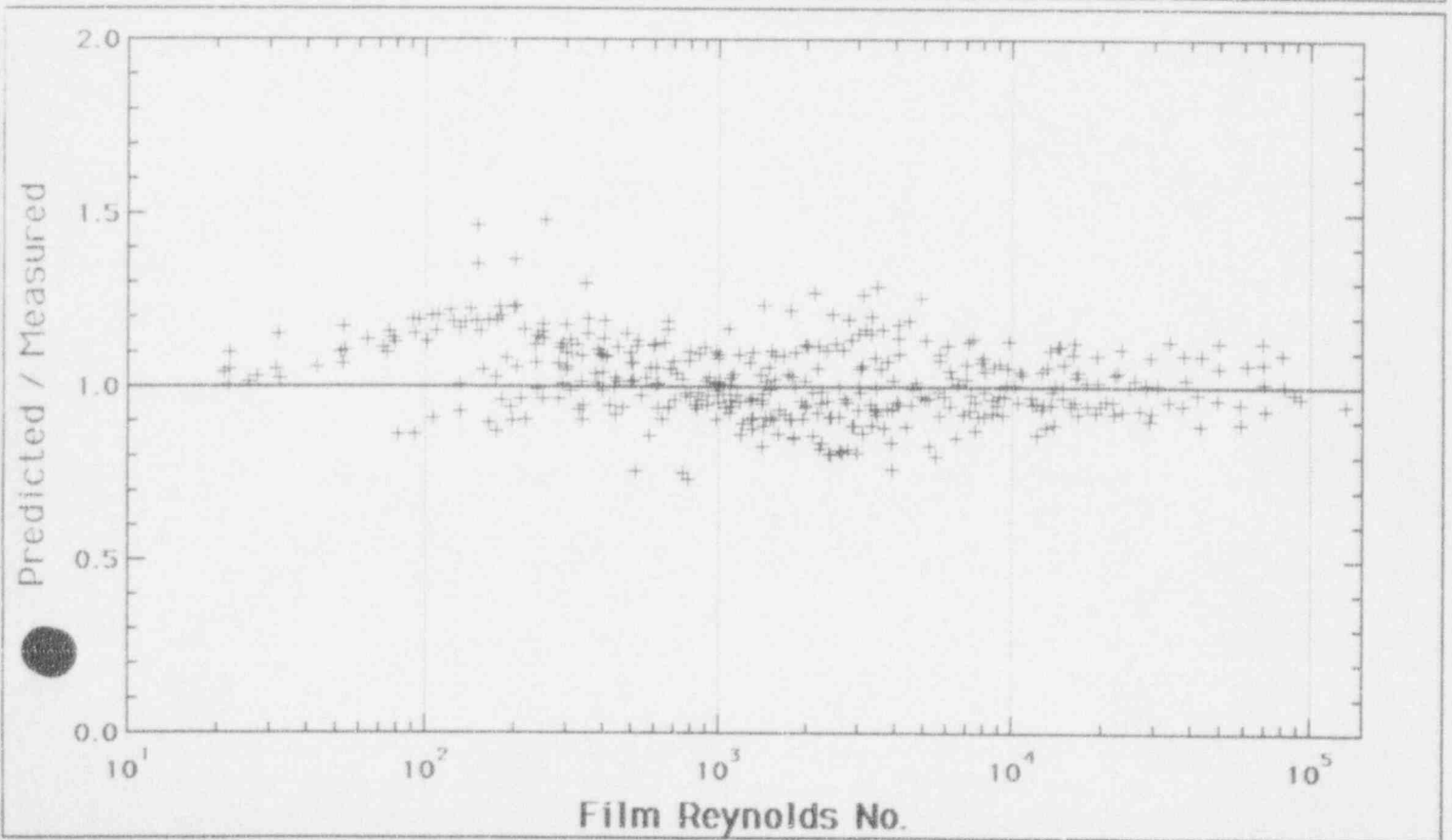
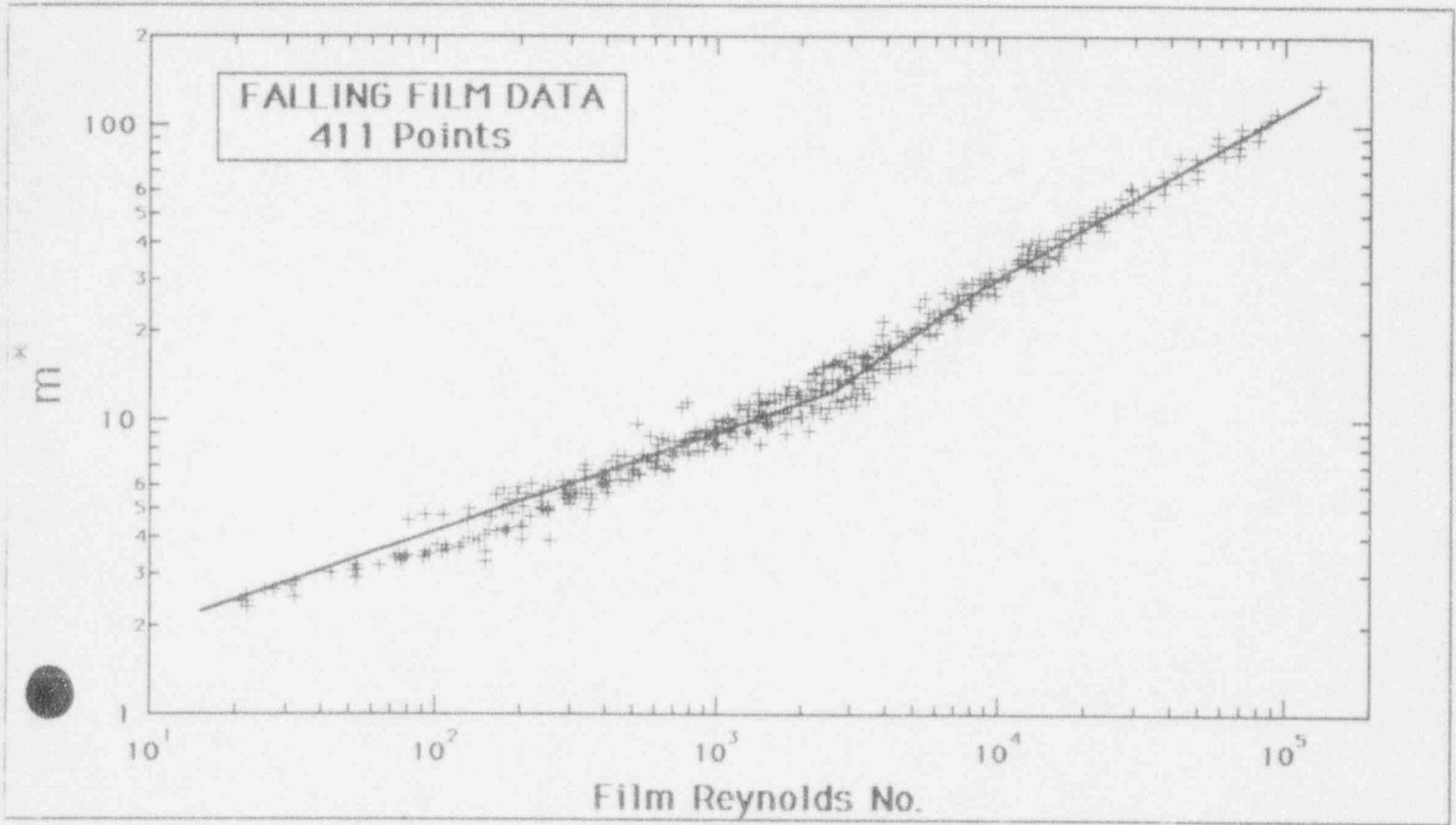
$$m^* = \max[\text{Nusselt}, .165 \cdot \text{Re}^{.566}]$$





# FALLING FILM: THICKNESS

$$m^* = \max \left\{ \text{Nusselt}, \min \left[ .0663 \cdot \text{Re}^{\frac{2}{3}}, .165 \cdot \text{Re}^{.566} \right] \right\}$$



# FILM THICKNESS

## \* FALLING FILMS

- Proposed Model:

$$m_L^* = .9085 \cdot Re_j^{\frac{1}{2}}$$

$$m_T^* = .165 \cdot Re_j^{.566}$$

- Wall Friction Factors:

$$f_L = \frac{24}{Re_j}$$

$$f_T = \frac{.144}{Re_j^{.03}}$$

# FILM THICKNESS

## \* SHEARED FILMS

- RELAP5:
  - Void fraction, and hence film thickness, results from solution of field equations.
  
- CONSTITUTIVE MODELS REQUIRED:
  - Wall friction factor
  - Interfacial friction factor
  - Entrainment fraction
  - Momentum transfer due to phase change

# FILM THICKNESS

## \* SHEARED FILMS - Laminar Smooth

- Shear Stress Distribution:

$$\tau = \tau_i + \Delta\rho \cdot g \cdot (m - y)$$

- Velocity Profile:

$$V_f = \frac{1}{\mu} \cdot \left\{ \tau_i \cdot y + \Delta\rho \cdot g \cdot \left( m \cdot y - \frac{1}{2} y^2 \right) \right\}$$

- Average Velocity:

$$\bar{V}_f = \frac{m}{\mu} \cdot \left\{ \frac{1}{2} \tau_i + \frac{1}{3} \Delta\rho \cdot g \cdot m \right\}$$

- Film Reynolds No.

$$Re_f = 2 \cdot \rho_f \cdot \left( \frac{m}{\mu} \right)^2 \cdot \left\{ \tau_i + \frac{2}{3} \Delta\rho \cdot g \cdot m \right\}$$

## FILM THICKNESS

### \* SHEARED FILMS - Laminar Smooth

- Dimensionless Film Thickness:

$$m^+ = 0.7071 \cdot \text{Re}_j^{\frac{1}{2}}$$

where

$$m^+ = m \cdot \left( \frac{\tau_c}{\rho_j} \right)^{\frac{1}{2}} \cdot \frac{1}{v_j}$$

$$\tau_c = \tau_i + \frac{2}{3} \cdot \rho_j \cdot g \cdot m$$

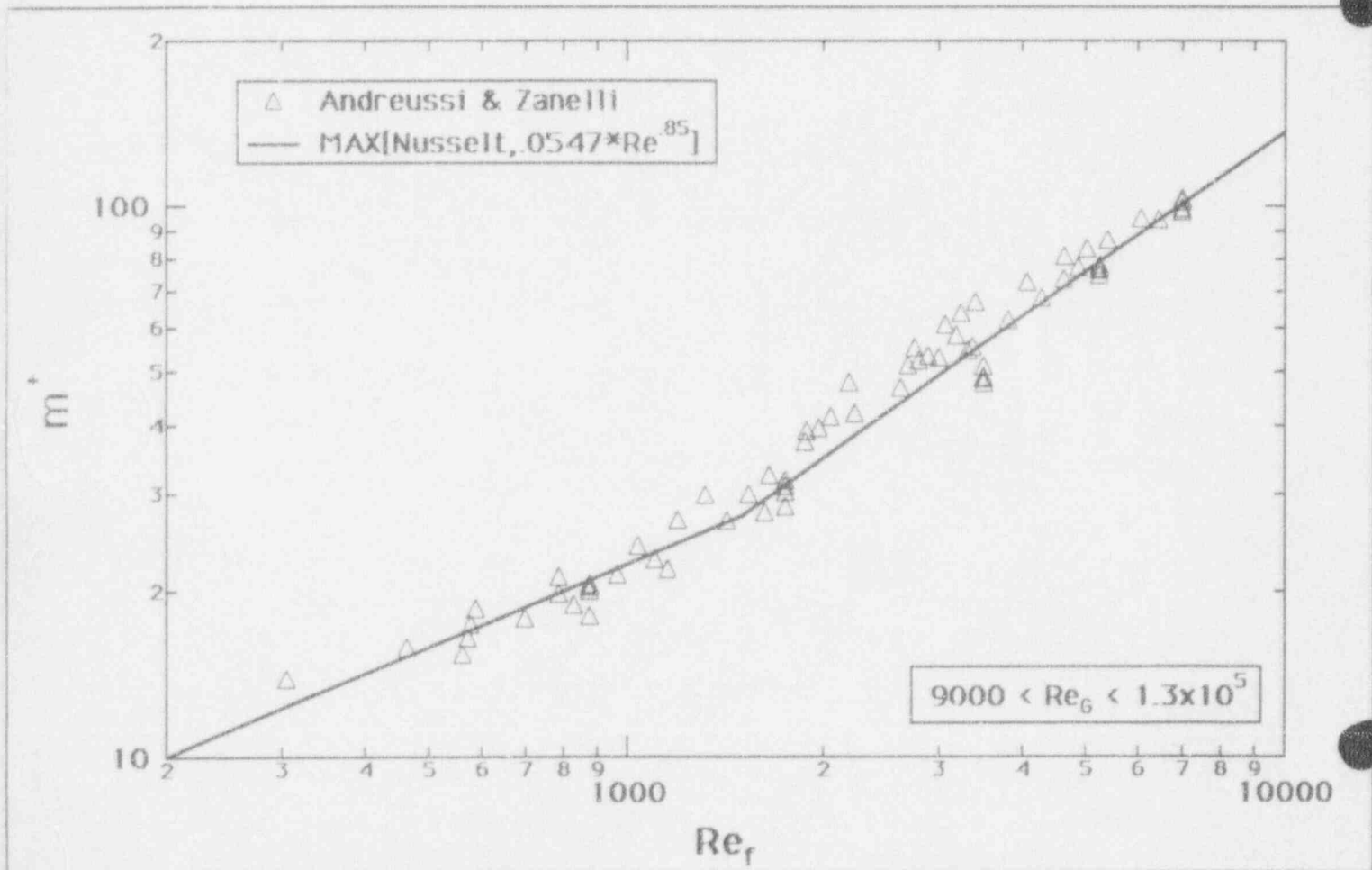
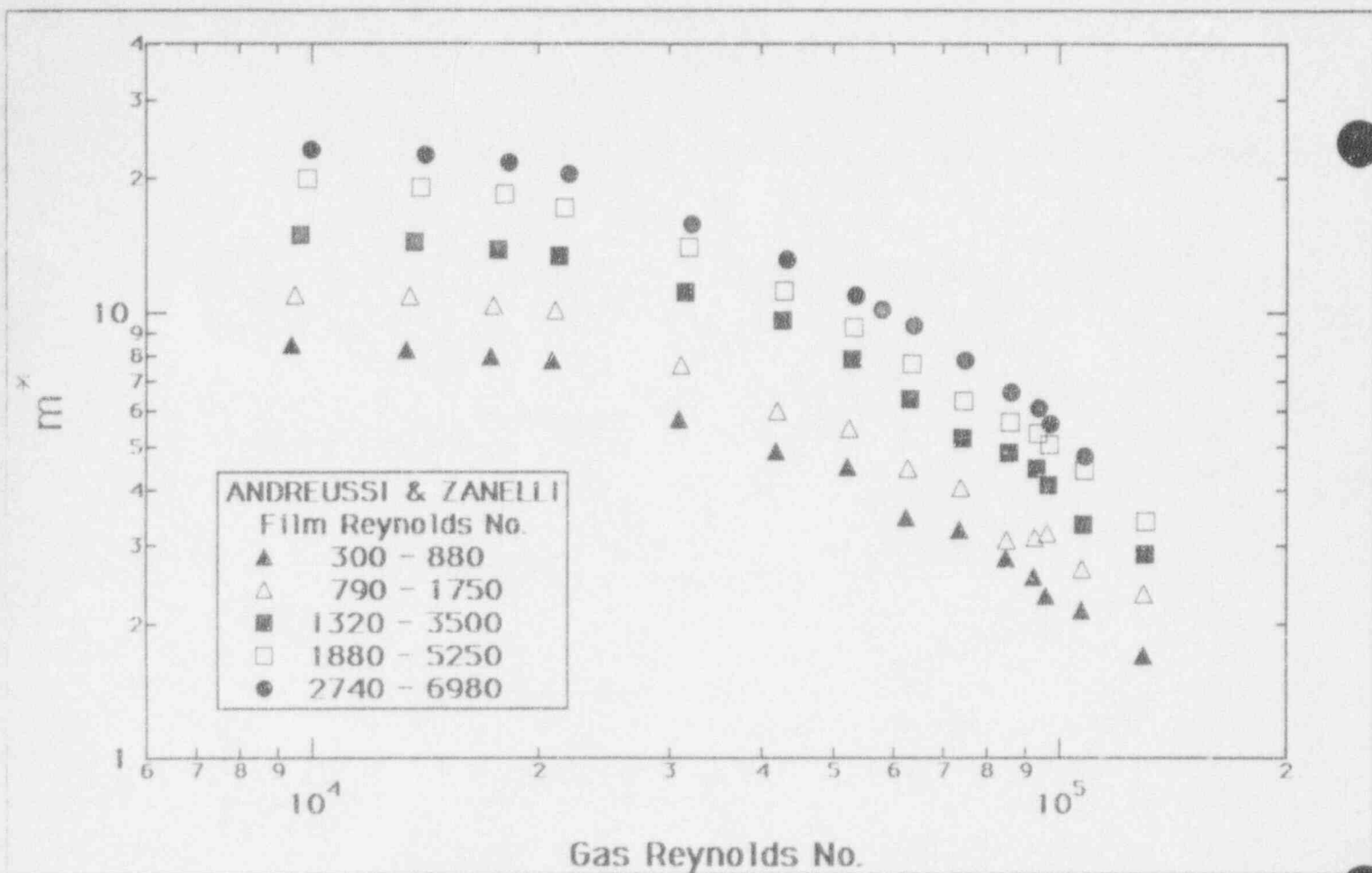
- Note: for a falling film:

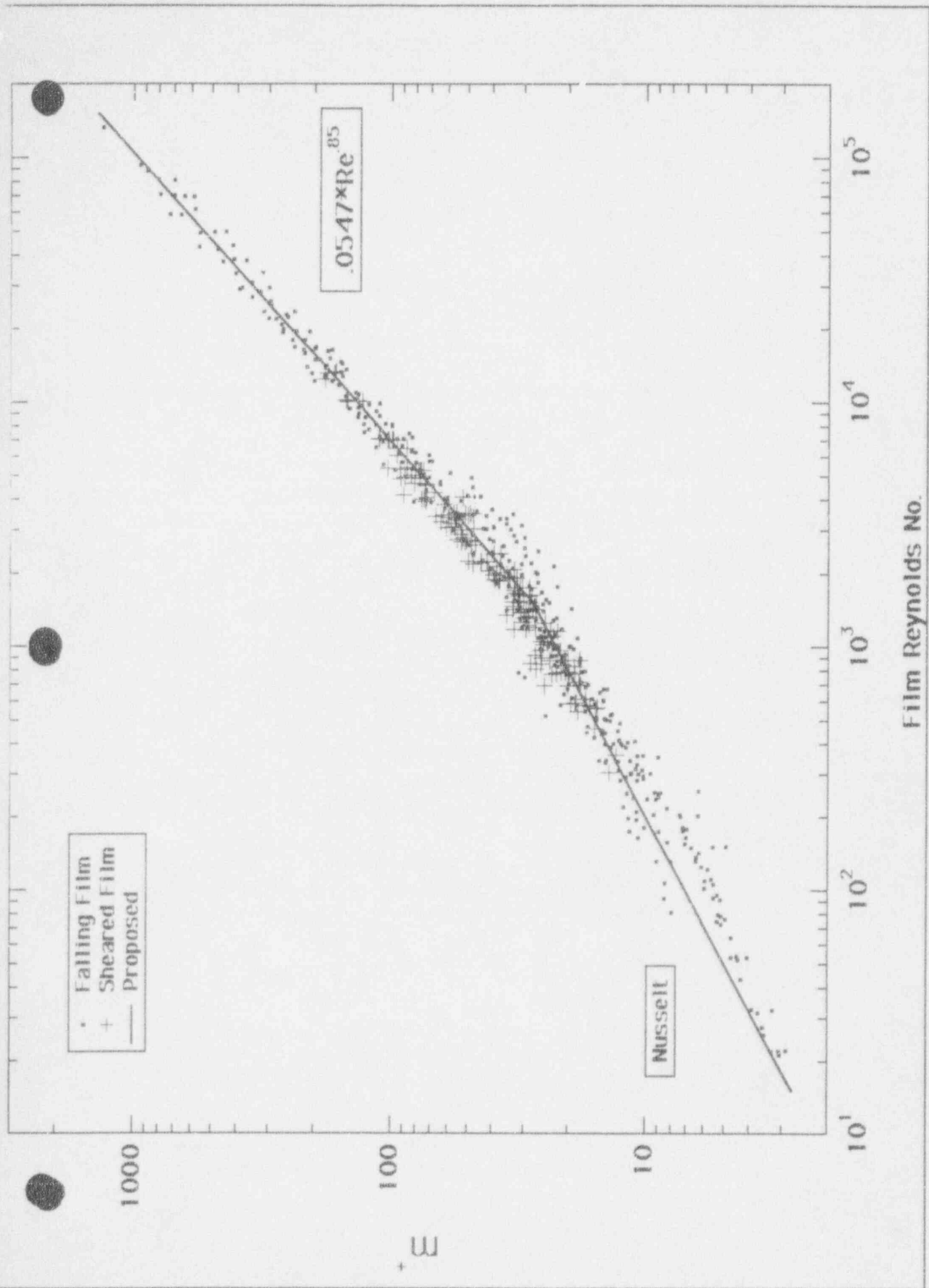
$$m^+ = \left( \frac{2}{3} \right)^{\frac{1}{2}} \cdot (m^*)^{\frac{3}{2}}$$

so, for a turbulent falling film:

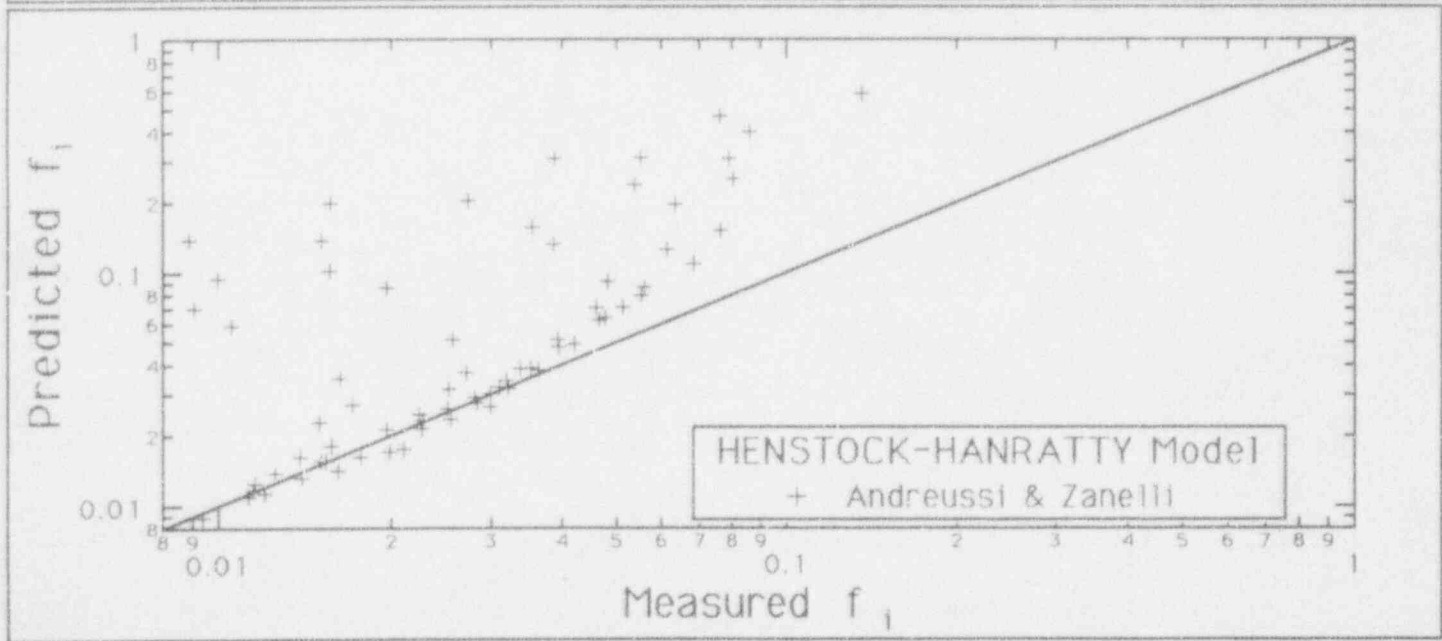
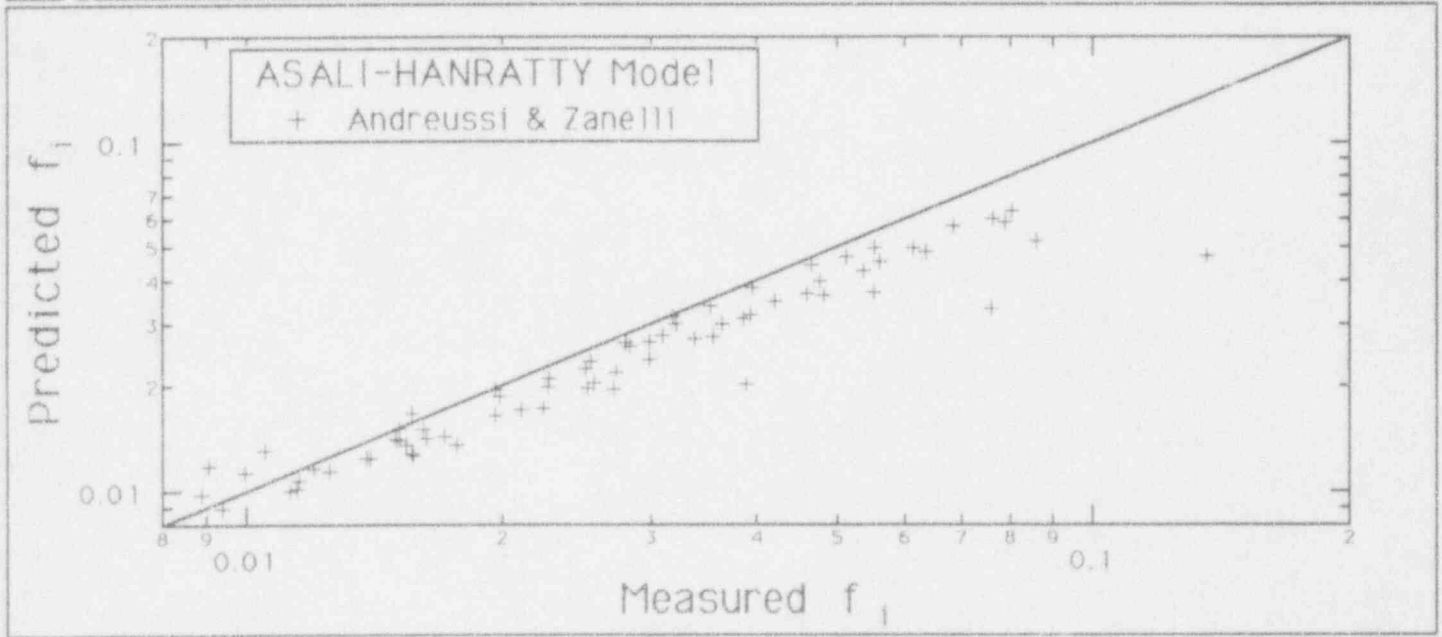
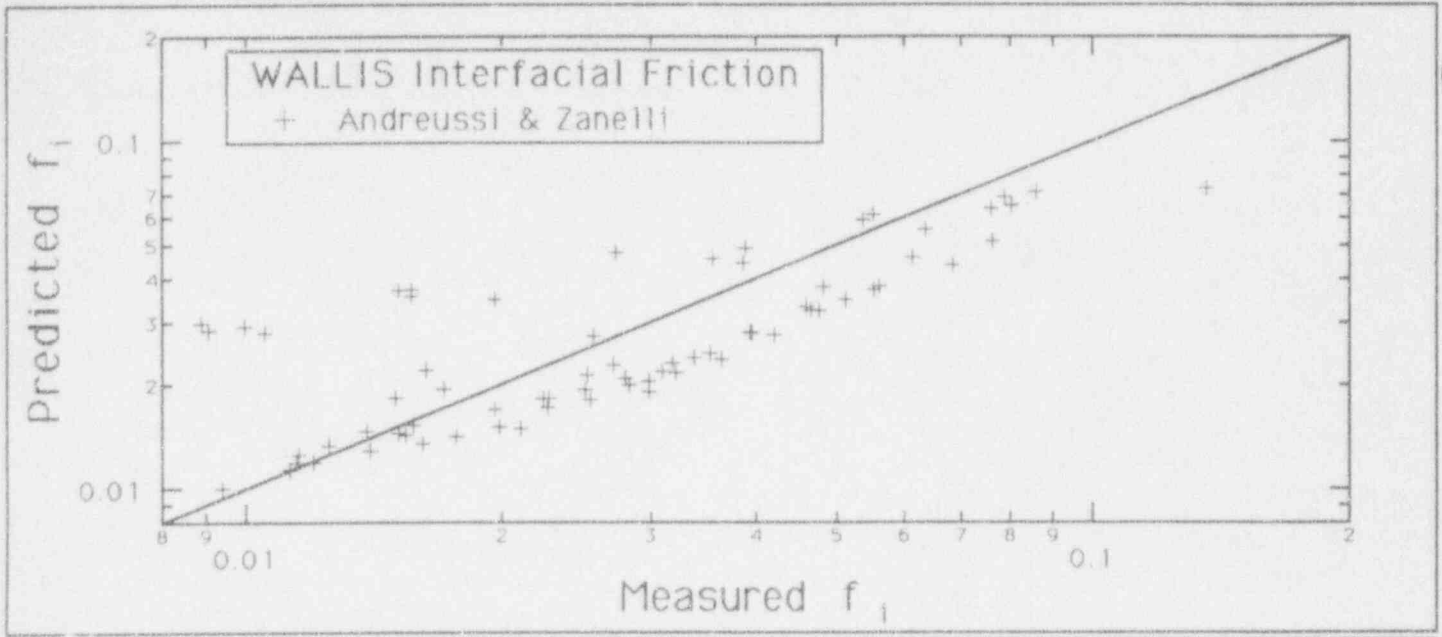
$$m_j^+ = .0547 \cdot \text{Re}_j^{.85}$$

\* QUESTION: is  $m^+$  a unique function of  $\text{Re}$  for turbulent sheared films also ?





# INTERFACIAL FRICTION





# FILM THICKNESS

## \* INTERFACIAL FRICTION MODEL

- Interfacial Shear Stress:

$$\tau_i = \frac{1}{2} \cdot f_i \cdot \rho_v \cdot (V_v - 1.5 \cdot V_l)^2$$

- Interfacial Friction Factor:

$$\frac{1}{\sqrt{f_i}} = -1.8 \cdot \log \left\{ \frac{6.9}{Re_v} + \left( \frac{\epsilon/d}{37} \right)^{1.11} \right\}$$

- Interfacial Roughness:

$$\frac{\epsilon}{d} = \text{function}(m^+, Re_v)$$

- to be determined.