

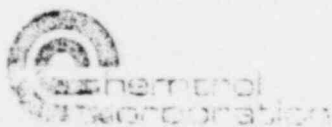
TEST REPORT ON
CHEMTROL CT-40-NS Type "B"

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TEST REPORT
ON
THE EFFECTS OF COMBINED
GAMMA AND NEUTRON RADIATION
ON
BORATED SILICONE MATRIX
CT-40-NS TYPE "B"
INSTALLED
WITHIN THE REACTOR CAVITY
OF THE
NORTH ANNA NUCLEAR STATION
UNITS 1 AND 2

STONE AND WEBSTER, P.O. NO: NA 1574

DATE ISSUED: February 23, 1981



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Attachments

- A). Specification for Supplemental Neutron Shielding
- B). University of Michigan (Irradiation and Physical Analysis)
- C). Dow Corning Corporation (Chemical Analysis)
- D). Southwest Research (Impact Test)
- E). Wyle Laboratories (LOCA Exposure Testing)

1.0 OBJECTIVE

The objective of this test program is to simulate the shield material's specified operating conditions and possible one-time abnormal condition. Also to demonstrate, under these one-time conditions, the neutron shielding material will not decompose, crumble, dissolve, or melt in any significant degree and will not evolve quantities of combustible gases or become a fire hazard. The shield material need not remain functional as a shield after this one-time event.

The specified normal operational requirements are as follows:

1. NORMAL OPERATION

- 1.1). Ambient temperatures between 86° F to 105° F.
- 1.2). Relative humidity form 0 to 100 percent.
- 1.3). Containment pressures of 9 PSIA to 11 PSIA.
- 1.4). The neutron shield material shall be capable of withstanding an integrated gamma radiation dose of 3.3×10^{10} rads over its design life of 40 years.

- 1.5). The shield material shall be capable of withstanding neutron fluence over its design life of 40 years of:
- 1.5.1). $E < .11$ Mev - 6.0×10^{18} neutrons/sq.cm.
- 1.5.2). $.11 \leq E \leq 1.1$ Mev - 3.0×10^{18} neutrons/sq.cm.
- 1.5.3). $E > 1.1$ Mev - 1.5×10^{17} neutrons/sq.cm.
- 1.6). The neutron shielding material will be in direct contact with sections of the reactor vessel and nozzle insulation. During normal reactor operations, the continuous contact surface temperature of the shield at the insulation-to-shield interface will be 247°F to 393°F.

2. ABNORMAL ONE-TIME CONDITIONS

As specified, the shielding material may be subjected to the following abnormal one-time condition:

2.1).

<u>TIME</u>	<u>TEMPERATURE</u>	<u>PRESSURE</u>	<u>PH</u>
0 to 2 min.	440°F	45 PSIG	5-11
2 to 60 min.	280°F	45 PSIG	5-11
60 min to 30 days	150°F	0 PSIG	7-9

- 2.2). 100 percent relative humidity.
- 2.3). Steam and water jet at 600°F in a localized area for a one minute period.
- 2.4). A boric acid solution soak (2,000 to 2,500 PPM) with sodium hydroxide added, ph 5.0 to 11.0 at 150 F for 30 days.

2.0 TESTING CRITERIA

In order to properly accommodate the above normal and abnormal operational requirements, some test criteria integration was required. This integration resulted in the following test requirements.

1.0 The Shield Material In Pre-Operational Mode

- a). Quantify chemical values.
 - a.1). Hydrogen
 - a.2). Carbon
 - a.3). Oxygen
 - a.4). Silicone
- b). Quantify physical values.
 - b.1). Specific Gravity
 - b.2). Durometer
 - b.3). Stress/Strain
- c). Quantify LOCA Resistance
 - c.1). Steam Impingement
 - c.2). Borated Water Soaking

2.0 The Shield Material In Operational Mode

- a). Gamma and Neutron radiation exposure to 2 levels of integrated doses.

a.1). Sample 1 - 8.25×10^9 Rads/Gamma

1.5×10^{18} N/sq.cm. $E < .11$ Mev

7.5×10^{17} N/sq.cm. $.11 \leq E \leq 1.1$ Mev

3.5×10^{16} N/sq.cm. $E > 1.1$ Mev

Above dose equivalent to 10 operational reactor years.

a.2). Sample 2 - 3.3×10^{10} Rads/Gamma

6.0×10^{18} N/sq.cm. $E < .11$ Mev

3.0×10^{18} N/sq.cm. $.11 \leq E \leq 1.1$ Mev

1.5×10^{17} N/sq.cm. $E > 1.1$ Mev

Above dose equivalent to 40 operational reactor years.

Since the shield material is to be in contact with reactor vessel and nozzle insulation, the shield material shall experience normal operating temperatures of 247°F to 393°F and therefore, shall undergo radiation exposure testing under equivalent conditions of $380^\circ\text{F} \pm 20^\circ\text{F}$.

b). Quantify Chemical values (See Item 1.0 above)

c). Quantify Physical values (See Item 1.0 above)

d). Quantify LOCA Resistance

d.1). Steam Impingement (Sample 1 and 2)

e). Quantify Impact Resistance

e.1). Accelerate Sample 1 and 2 above to 1500
F.P.S. to simulate Loss of Coolant piping
rupture. Recover accelerated material and
submit to Borated Water Soaking.

f). Quantify LOCA Resistance

f.1). Borated Water Soaking.

3.0 Chemtrol CT-40-NS Type "B"

Samples Preparation

The preparation of sample specimens was carried out at the Chemtrol facilities April 11, 1979. No elaborate procedure were involved in the proper compounding of the matrix other than those approved by the Chemtrol QA/QC. These procedures basically reflect compounding normally associated with dual component polymers.

For the sake of traceability (Quality Control), all samples produced were labeled to provide continuity of samples. A series of samples were labeled A1 thru A12 with another series B1 thru B12. The A series of samples were submitted to the University of Michigan (Phoenix Memorial Laboratory) for experimental purposes; i.e., identification of a reliable irradiation mode within the Ford Nuclear Reactor. As soon as reliability was established the B series of samples were submitted to the University of Michigan for the required irradiation process.

Due to limitations within the reactor irradiation area, the above A and B series of samples were molded into 1"x7"x3/8"

long specimens. This geometry provided symmetry of testing since the 380°F environmental heating required during the irradiation process did not allow sample specimens of greater size. Moreover, the sample geometry allowed greater ease of physical properties testing.

In conjunction with the above sample preparation, four additional sample specimens were prepared with an approximate size of 4"x4"x4". These unirradiated samples were submitted to Wyle Laboratories for the purpose of undergoing exposure testing per Loss of Coolant Accident conditions.

Note should be taken of the fact that the compounding of Chemtrol CT-40-NS Type "B" Silicone matrix involves the additional compounding of Boron Carbide (B4C) in fine powder form within the basic silicone polymer. This compounding is the 2nd step in the 2 step process. In the first step, the basic 2 part silicone components A and B are mixed in equal weights or volume. Immediately thereafter, the boron carbide (B4C) is added to mix as the second step in mixing process. The amount of B4C to be added to the basic polymer mix is sufficient to result in a net 1.5% minimum weight of boron of the matrix weight.

An additional note to be made is that ancillary samples were prepared for the purpose of assuring reliability of boron homogeneity within the silicone matrix. Said sample testing resulted in such assurance.

4.0 Chemtrol CT-40-NS Type "B"

Samples Irradiation

The purpose of the neutron shielding material irradiation study is to determine the changes to physical and chemical properties of the shielding material as a function of irradiation exposure under conditions similar to those present within the reactor cavity of the North Anna Nuclear Station.

In a radiation field, elastomeric materials are the most sensitive to radiolysis of all construction materials. Likewise, radiation progressively deteriorates the initially optimized properties of elastomers. The general effect is to increase the modulus of elasticity and hardness while decreasing ultimate elongation and tensile strength.

The general process whereby such changes occur is ionization. The transfer of energy from the radiation beam to the atoms of the polymer directly or indirectly causes the ejection of orbital electrons. The energetic electrons produce secondary ionizations. The ions formed by the ejection of orbital electrons combine with free electrons to form energetic, unstable molecules. The excited

molecules quickly dissipate their excess energies, largely by bond scission, physical transfer, and possibly molecular rearrangement. The bond scissions produce free radicals and unsaturation, and most of the subsequent overt effects result from these. Cross linking, chain scission, molecular rearrangement, and chemical reaction with environmental agents, especially oxidation and ozonization, occur and constitute the preponderant changes. Essentially all the changes in physical properties of the elastomer material ensue from these basic processes. Although the primary and secondary processes are obviously not temperature sensitive, the resultant chemical reactions are temperature dependant.

A secondary point of importance, is the effect of dose rate on the elastomer compound. Prior experimentation has shown that there is no dose-rate effect for elastomers within the broad range of 10^4 through 10^7 rads/hr. i.e.; at least over this range, any variation in dose rate does not significantly effect the change in physical properties of the elastomer compound. What does matter is the total amount of radiation energy absorbed and the uniformity of absorption.

A final point to be made is that prior experimngntation has revealed no significant post-irradiation effects in the

compound. Measurements of hardness and elasticity has shown no change over a period of three months after irradiation.

The irradiation test program was comprised of two parts. The first part involved irradiation of test specimens for a period equivalent to 10 years of reactor operation i.e., to the same level of accumulative gamma and neutron doses. The second part of the test program involved the irradiation of test specimens to 40 years equivalent reactor operation. The above irradiation process was to be conducted with material samples in a $380\text{ F} \pm 20\text{ F}$ environment.

All of the above irradiations were performed at the University of Michigan Ford Nuclear Reactor operated by the Michigan Memorial Laboratory. These facilities were chosen for these test experiments simply because the Ford Reactor flux profiles accomidated the majority of incident flux profiles required for the material testing. In addition, the University of Michigan reactor material testing personnel had considerable experimental experience with similar material testing requirements.

The respective irradiation exposure levels required along with the actual accumulated doses are as follows:

<u>10 Year Exposure Level</u>		
<u>Irradiation</u>	<u>Desired</u>	<u>Actual</u>
Integrated Gamma Exposure	8.25×10^9 Rads	8.4×10^9 Rads
Neutrons $E < .11$ Mev	1.5×10^{18} N/sq.cm.	2.0×10^{18} N/sq.cm.
Neutrons $.11 \leq E \leq 1.1$ Mev	7.5×10^{17} N/sq.cm.	8.7×10^{17} N/sq.cm.
Neutrons $E > 1.1$ Mev	3.5×10^{16} N/sq.cm.	7.1×10^{17} N/sq.cm.

<u>40 Year Exposure Level</u>		
<u>Type</u>	<u>Desired</u>	<u>Actual</u>
Integrated Gamma Exposure	3.3×10^{10} Rads	4.1×10^{10} Rads
Neutrons $E < .11$ Mev	6.0×10^{18} N/sq.cm.	9.8×10^{18} N/sq.cm.
Neutrons $.11 \leq E \leq 1.1$ Mev	3.0×10^{18} N/sq.cm.	4.3×10^{18} N/sq.cm.
Neutrons $E > 1.1$ Mev	1.5×10^{17} N/sq.cm.	3.5×10^{18} N/sq.cm.

The above exposure levels were administered at the following incident flux levels.

<u>10 Year and 40 Year Exposure Level</u>		
<u>Irradiation</u>	<u>Measured Dose Rate</u>	<u>Dose Rate -10%</u>
Gamma	8.4×10^7 rad/hr.	7.6×10^7 rad/hr.
$E > .11$ Mev	5.6×10^{12} n/cm ² /sec.	5.0×10^{12} n/cm ² /sec.
$.11 \text{ Mev} \leq E \leq 1.1 \text{ Mev}$	2.4×10^{12} n/cm ² /sec.	2.2×10^{12} n/cm ² /sec.
$E > 1.1$ Mev	2.0×10^{12} n/cm ² /sec.	1.8×10^{12} n/cm ² /sec.

The 10 year exposure level required 110 hours of exposure to reach its desired accumulated dose while the 40 year level required 545.6 hours of exposure. The actual exposure levels were in excess of the desired levels. i.e.; the 40

year specimen received 20% more gamma and over 50% more neutron exposure than specified.

An additional deviation from the norm is the above referenced measured dose rates. A 10% reduction was applied to all gamma and neutron incident flux rates to allow for shielding of the samples 1/4" wall aluminum container. This value reduction was applied per estimates generated by University of Michigan Ford Reactor Engineering personnel. These estimates are highly conservative.

From the standpoint of irradiation exposure and resulting physical and chemical data, the 40 year material sample specimens are not representative to conditions within the reactor cavity of the North Anna Nuclear Station. The material's physical and chemical properties may have undergone radical change beyond irradiation to 3.3×10^{10} Rads. A more than 50% over exposure to neutrons in the thermal energy range surely had a prominent role in end data. Each thermal neutron to hydrogen reaction (capture) results in the release of a single 2.2 mev gamma photon. The magnitude of this problem was somewhat reduced since B4C compounding reduced the potential for the H + N event. Instead, a neutron + Boron 10 event results in gamma emissions of .5 mev. However, the accompanying release of a

highly energetic alpha particle does provide a mechanism for localized ionization. The magnitude of this event is appreciated when one equates the thermal flux rate, and atoms of Boron 10 as a function of cross section.

The temperature environment sustained by the material samples during the irradiation process most assuredly contributed to Loss of Chemical and Physical properties. Cross radiation damage is accelerated by heating or stressing the elastomeric material.

On the matter of material activation, the 10 year sample reflected a gamma dose rate on contact of approximately 50 Mrem/hr. 16 days after removal from the reactor core area. The identified activation species were CO-60, ZN-65, SB-124, SC-46 and CS-134. No data was obtained for the 40 year exposure sample.

5.0 Chemtrol CT-40-NS Type "B"

Physical Analysis

In its cured state, Chemtrol CT-40-NS Type "B" has the following physical properties:

Specific Gravity-----1.36

Durometer (Shore A)-----55

(Lb) Tensile Force	(In) Sample Length	(In) Sample Elongation	(PSI) Stress	(In/In) Strain
0	Lo=2.50	0	0	0
1	2.58	.08	8.40	.032
2	2.68	.18	16.81	.072
3	2.75	.25	25.21	.100
4	2.85	.35	33.61	.140
9.5	UTS		79.83	.250

Subsequent to irradiation exposure (10 year exposure level) the shield material manifested changes in all areas of physical properties tested. These changes adhered to the general effect of increased hardness and modules of elasticity while decreasing ultimate elongation and tensile strength.

Specific Gravity-----1.57 g/cc

Durometer (Shore A)-----95

(Lb) Tensile Force	(In) Sample Length	(In) Sample Elongation	(PSI) Stress	(In/In) Strain
0	1.80	0	0	0
20	1.80	0	204.08	0
	UTS			

No physical data was obtained from the 40 year exposure level sample other than specific gravity due to difficulty in handling sample. Generally speaking said sample was fragile upon handling. As discussed in Section 4.0, the over-irradiation most likely resulted in an exaggerated representation of expected physical affects to shield material.

The impact resistance testing simulating accelerations as a result of loss of coolant accident was conducted by the Southwest Research Institute of San Antonio, Texas. The institute received the 10 year and 40 year exposure samples from Wyle Laboratories who had the initial test responsibility of subjecting the samples to steam impingement exposure.

The general results of the impact testing, i.e., accelerating samples to 1500 FPS, was to reduce the sample specimens both (10 year and 40 year) to a fine powder with a density much greater than water.

6.0 Chemtrol CT-40-NS Type "B"

Chemical Analysis

Since the primary function of the above referenced shield material system is to act as a neutron shield, careful elemental analysis is required. With this fact in mind, Chemtrol Corporation was able to retain the services of Dow Corning Corporation to perform the chemical analysis. Dow Corning Corporation is the main supplier of the base silicone polymer utilized in the formulation of the shield material and therefore, is well familiar with silicone analysis techniques.

Initially, one control sample (non-irradiated) and one of the 10 year exposure level specimens were submitted to them for analysis. Soon thereafter, the 40 year exposure level specimen was also submitted to them for the required analysis.

The following is the elemental data obtained from the control, 10 year and 40 year exposed samples:

Element/Property	Control	10 Year Sample	40 Year Sample
Carbon (C)	17.77 + .17%	14.22 + .14%	5.96 + .06%
Hydrogen (H)	4.38 + .12%	3.00 + .09%	1.07 + .03%
Silicon (Si)	40.5 + .2%	41.2 + 0.2%	40.2 + .5%
Oxygen (O)	37.35 + .2%	41.58 + .2%	58.42 + .5%
Specific Gravity	1.359 + .005%	1.571 + .005%	1.68 + .005

Oxygen Values by Difference.

A review of the elemental data reflects a significant loss in hydrogen and carbon.

It should be noted that the irradiated samples were thin strips which were given a near uniform irradiation under uniform temperature. The installed shield material's thickness and orientation allow for self shielding and a temperature gradient throughout its thickness. Accumulated dose and resultant degradation of material properties would, therefore, be attenuated as a function of shield thickness. In addition, the test specimens were given an approximate 50% excess irradiation, as discussed in Section 4.0.

In conclusion, the test specimens demonstrated a definite degradation in shielding properties. The degree of degradation or efficiency which will be experienced by the installed shield material, however, can only be accurately determined by a plant survey program of neutron dose vs. life of shield material.

7.0 Chemtrol CT-40-NS Type "B"
Pre and Post Irradiation
LOCA Exposure Testing

Qualification testing was conducted on CT-40-NS Type "B" to simulate Loss of Coolant Accident (LOCA) conditions. The test program was performed by Wyle Laboratories of Huntsville, Alabama.

The test program was differentiated into two basic programs. One program element reflected qualification testing of unirradiated shield material, while the other element involved qualification testing of the 10 year and 40 year irradiated material samples. In conjunction with the second program and subsequent to the initial steam impingement portion of the LOCA testing, the material samples were submitted to Southwest Research Institute for impact testing (refer to Section 5.0 and Exhibit E). Thereafter, Southwest Research Institute returned the impact test specimens back to Wyle Laboratories for continuation of LOCA Testing Requirements (Borated Water Submersion Test).

The initial LOCA testing involved unirradiated material samples (4 ea. - 4"x4"x4") comprising of steam impingement (600 F steam - 60 sec.) and submersion in borated water at 150 F for 30 days.

The effects of steam impingement on the sample specimens were non-existent. However, the results from the Borated Water Submersion Test merits some discussion. The area of discussion only relates to the generation hydrogen off-gassing since all four material samples appeared to be unchanged in all other physical respects after removal from the borated water test cylinder.

The LOCA test cylinder had a free volume of approximately .7 cubic foot (20 liters). Of this free volume, approximately 75% (15 liters) was occupied by borated water and 4 unirradiated material sample specimens. Gas that evolved collected in the remaining 5 liters of the cylinder at 150 F (66 C) and 1 psi (1.07 atmospheres). Gas analysis, performed at room temperature and atmospheric pressure, resulted in 32 volume percent hydrogen. Five liters of gas at 1.07 atmospheres and 150 F is equal to approximately 4.31 liters at room temperature and standard pressure (STP). The hydrogen component under the same conditions would be 32% of 4.31 liters, or approximately 1.38 liters. The evolution of gas is a function of the surface area of 2.7 sq.ft. Therefore, the evolution of hydrogen per square foot of material is .62 liters/sq.ft.

Chemtrol's CT-40-NS Type "B" Neutron Shielding Material has

been installed in the North Anna Unit 1 and 2 Nuclear Facility with an exposed surface of approximately 10,000 sq.ft. This material surface area, in an unirradiated form, would generate 6,200 liters of hydrogen gas in the event of LOCA. The area of installation (containment) has an approximate volume of 1.8×10^6 cubic feet (5.1×10^7 liters). Therefore, the hydrogen gas evolved would account for less than 1.24×10^{-4} percent of the containment volume. This volume percent does not even remotely approach the LEL for hydrogen in air (4%).

The introduction of the methane (1.1%) and ethane (.01%) components of the off-gassing to the above equation does not alter the basic conclusions.

The LOCA testing on the 10 year and 40 year irradiated specimens was somewhat different in that the test specimens for the Borated Water Submersion were in powder form as a result of impact testing. Nevertheless, the basic results with the exception of increased ethane off-gassing and much reduced hydrogen off-gassing. The overall off-gassing was minute in quantity and therefore, does not provide a potential for concern.

A comparative evaluation of CR-40-NS Type "B" (irradiated

vs. unirradiated) shield material system as a function of LOCA exposure testing reveals that as the shield system undergoes irradiation exposure, its potential for off-gassing is greatly reduced.

ATTACHMENT A

J.O. No. 13075.20
NAS 1007

June 8, 1979

Specification for
SUPPLEMENTAL NEUTRON SHIELDING

North Anna Power Station - Unit 1
Virginia Electric and Power Company
Richmond, Virginia

	Rev. 0	Rev. 1	Rev. 2
S&W Originator	<i>R.R. Cline</i> 6-7-79		
S&W Lead Engineer	<i>W. H. Cline</i> 6-8-79		
S&W Specialist	<i>R. H. Cline</i> 6/7/79		
S&W Project Engineer	<i>[Signature]</i> 6/8/79		
VEPCO Project Engineer			
VEPCO QA Supervisor			
VEPCO Approved			

I.	<u>GENERAL</u>	1.9
	The reactor neutron shielding will be used to attenuate neutrons escaping from the reactor vessel during normal plant operations. The shielding will be located between and around the reactor nozzles and will be supported on top of the neutron shield tank. The shielding will remain in place during all phases of operation except when portions of the nozzle shields must be removed for periodic inspection of welds. The shield is designed to contain both hydrogen and boron to attenuate and absorb neutrons, thus reducing exposure to operating personnel.	1.12 1.13 1.14 1.16 1.17 1.18 1.19
II.	<u>SCOPE</u>	1.22
	1. The work includes fabrication, quality control, and delivery of silicone-based neutron shielding material to the North Anna jobsite.	1.25 1.26
	The supplier will be provided with previously fabricated molds (50° and 70° shields) as shown on the attached sketches, 13075-MKS-13, sheets 1 and 2. These prefabricated molds are to be filled with the silicone neutron shielding material.	1.29 1.30 1.31
	The supplier shall also supply molds and mold the silicone material into the various shapes shown on the attached sketches, 13075-MKS-13, sheets 3 through 10. Quantities shall be as specified.	1.32 1.33 1.34
	2. The prefabricated molds require that the 10 gage, inside diameter be supported along its vertical length such that it is not distorted during injection of the liquid silicone neutron shield material by the liquid hydrostatic force. The purchaser will provide this support. The supplier shall inform the purchaser of any restrictions on the support material. Items to be considered are cure temperature that the support may experience and other supplier identified considerations. This information is required as part of the supplier's bid.	1.36 1.37 1.38 1.39 1.40 1.41 1.42
III.	<u>OPERATING CONDITIONS</u>	1.45
	The neutron shielding will be located within the containment building in close proximity to the reactor vessel. It will be subjected to containment atmospheric conditions as follows:	1.48 1.49 1.51

<u>1.</u>	Normal Operation				1.54	
	<u>a.</u>	Ambient temperatures between 86°F to 105°F			1.56	
	<u>b.</u>	Relative humidity from 0 to 100 percent			1.57	
	<u>c.</u>	Containment pressures of 9 psia to 11 psia			2.1	
	<u>d.</u>	The neutron shield material shall be capable of withstanding an integrated gamma radiation dose of 3.3×10^{10} rads over its design life of 40 yr.			2.2 2.3	
	<u>e.</u>	The materials shall be capable of withstanding neutron fluence over its design life of 40 yr of:			2.4	
		1. $E < .11$ Mev - 6.0×10^{16} neutrons/sq cm			2.6	
		2. $.11 \leq E \leq 1.1$ Mev - 3.0×10^{13} neutrons/sq cm			2.8	
		3. $E > 1.1$ Mev - 1.5×10^{17} neutrons/sq cm			2.10	
	<u>f.</u>	The neutron shielding material will be in direct contact with sections of the reactor vessel and nozzle insulation. During normal reactor operations, the continuous contact surface temperature of the shield at the insulation-to-shield interface will be 247°F to 393°F.			2.13 2.14 2.15 2.16	
<u>2.</u>	The shielding material may be subjected to the following abnormal one-time condition in addition to those listed in III.1.				2.18 2.19	
	<u>a.</u>	<u>Time</u>	<u>Temperature</u>	<u>Pressure</u>	<u>pH</u>	2.22
		0 to 2 min	440°F	45 psig	5-11	2.25
		2 to 60 min	280°F	45 psig	5-11	2.26
		60 min to 30 days	150°F	0 psig	7-9	2.27
	<u>b.</u>	100 percent relative humidity				2.32
	<u>c.</u>	Steam and water jet at 600°F in a localized area for a one-minute period				2.33
	<u>d.</u>	A boric acid solution soak (2,000 to 2,500 ppm) with sodium hydroxide added, pH 5.0 to 11.0 at 150°F for 30 days.				2.34 2.35
	<u>Under these one-time conditions, the neutron shielding material must not decompose, crumble, dissolve, or melt in any significant degree and must not evolve quantities of combustible gases or become a fire</u>					2.37 2.38

hazard. The degree and type of degradation expected 2.39
 under the one-time conditions shall be indicated in
 the supplier's bid. The shield material need not 2.40
 remain functional as a shield after this one-time
 event.

IV. DESIGN REQUIREMENTS 2.43

1. The shielding material shall be of a Dow Corning 2.46
 "Sylgard 170" base with a boron loading of 2 percent 2.47
 by weight. If there are any exceptions or changes in 2.49
 the Seller's proposed composition, list under the
 "Exception to Requirements" section of the proposal. 2.50
The following information shall be provided in the bid 2.51
 package:

Specific gravity	_____	2.54
(Minimum 1.3 g/cc)		2.55
Thermal conductivity	_____	2.57
Hydrogen density after 32EFPY	_____	3.1
(Minimum 0.055 g/cc)		3.2
Carbon (percent)	_____	3.4
Silicone (percent)	_____	3.7
Oxygen (percent)	_____	3.9
Hydrogen (percent)	_____	3.11
Boron (percent)	_____	3.13

2. Toxicity: The material shall exhibit no toxic effects 3.18
 to personnel handling it.
3. Fire resistance: The following requirements are set 3.20
 forth in the U.S. Regulatory Guide 1.120, Revision 1,
 dated November 1977. The material should be 3.22
 noncombustible or listed by a nationally recognized
 testing laboratory such as Factory Mutual Underwriters 3.23
 Laboratory, Inc., for:
1. Surface flamespread rating of 50 or less when 3.25
 tested under ASTM E-84.
2. Potential heat release of 3,500 Btu/lb or less 3.26
 when tested under ASTM D-3286 or NFPA 259.

- The supplier shall provide the test data necessary to support ANI and NRC acceptance. Any test data which is not available shall be identified in the supplier's bid. 3.21
3.22
4. The cured shielding material shall be sufficient strength to be self-supporting when subjected to the normal and one-time operating conditions specified in Section III above. The supplier shall certify that the neutron shield material will exhibit the above physical and chemical properties for 40 yr under the operating condition (Section III) specified above. 3.31
3.32
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3.35
5. The silicone neutron shield material shall be cured by the Seller to ensure that it will exhibit the physical and chemical properties listed above in Section IV when subjected to the normal and one-time operating conditions of Section III. The supplier shall ensure that void spaces in the silicone material are minimized during mold injection. The supplier shall provide his curing procedure as part of his bid. 3.36
3.37
3.38
3.39
3.40
- V. INSTRUCTIONS AND DOCUMENTATION 3.43
1. The Seller shall furnish the buyer with any available documentation required for support of licensing applications. This would include such things as fire protection, radiation resistance, quantity of hydrogen offgassing during the one-time conditions of Section III, chemical properties, physical properties, etc. 3.46
3.47
3.48
3.50
2. The Seller shall furnish with his bid the proposed quality control procedures to ensure that the supplied neutron shield material meets the above-listed (Section IV) physical and chemical properties and the method for which void spaces are minimized during manufacturing. 3.51
3.52
3.53
3. The Seller shall furnish with this bid a schedule of fabrication and shipping milestones for the material identified in Section II. 3.54
3.55

VI.	<u>IDENTIFICATION AND PREPARATION FOR SHIPMENTS</u>	3.58
1.	Packing slips or other identification papers shall accompany all shipments and must bear the following:	4.3 4.4
	Supplementary Neutron Shielding	4.6
	North Anna Power Station - Unit 1	4.7
	Purchase Order No. _____	4.8
	Item No. _____	4.9
2.	Each item shall be marked with the appropriate item number.	4.13
3.	The Seller or subcontractor shall prepare and weatherize all articles for shipment in a good commercial manner so as to protect the material from damage to which it might reasonably be subjected, both in transit and handling. Packaging shall allow for temporary outdoor storage.	4.14 4.15 4.16 4.17
VII.	<u>QUALITY ASSURANCE PROGRAM</u>	4.20
	Each bidder shall submit with his original proposal one copy of his quality assurance procedure, covering the quality control and quality assurance measures (1) imposed by him on his work and (2) imposed by him upon sub-suppliers or subcontractors.	4.23 4.24 4.25
	A procedure acceptable to the Engineers shall be a prerequisite for a bidder being chosen as Seller.	4.27
	The Seller and his sub-suppliers shall have in effect in their shops at all times an inspection, testing, and documentation program that will ensure that the equipment furnished under this technical document meets in all respects with the requirements of this technical document.	4.28 4.29 4.30
	The Seller shall implement and maintain this procedure carrying out the requirement of this technical document, and all proposed major changes shall be submitted to and approved by the Engineers prior to implementation.	4.31 4.32 4.33
	In order to attain assurance that the neutron shielding material will give satisfactory, dependable, trouble-free service, this technical document requires that a "Certificate of Conformance" be submitted by the Seller, verifying that the neutron shielding material is in conformance with the set-forth requirements.	4.34 4.35 4.36
	The Seller shall specifically ensure that a copy of this technical document with all addenda thereto, or	4.37

appropriate work instructions which include the technical document requirements are readily available at each of his fabricating or production locations where work covered by this document is in progress. 4.38
4.39

VIII. INSPECTION 4.42

The Seller shall give the Purchaser or Engineers notification 72 hours prior to initial pouring of molds for possible inspection. 4.45
4.46

Authorized shop inspectors or other representatives of the Purchaser or the Engineers shall be allowed access to the engineering offices, shops, and working area of the Seller and his subsuppliers at all reasonable times. They shall have the right to such information as is necessary to demonstrate that engineering, procurement, and production are proceeding in accordance with the established schedules. They shall also have the right to inspect the materials or equipment, or the Seller's or subsupplier's production and inspection procedures, to confirm that the requirements of this document are being complied with; the Seller or subsupplier shall provide all tools, instruments, etc, necessary to facilitate these inspections. A final inspection will be made prior to shipment of the neutron shielding material. 4.48
4.49
4.51
4.52
4.53
4.54
4.55
4.56

IX. GUARANTEE 5.1

The materials furnished shall be new and guaranteed to perform as required herein. Materials and workmanship shall be guaranteed to meet the requirements set forth above. Should any defect in material, workmanship, or performance develop during the first year following initial commercial operation of the plant, the Seller shall agree to make repairs or necessary replacements free of charge as long as the materials were not subjected to unusual operating conditions or other hazardous requirements. 5.4
5.6
5.7
5.8
5.9
5.10

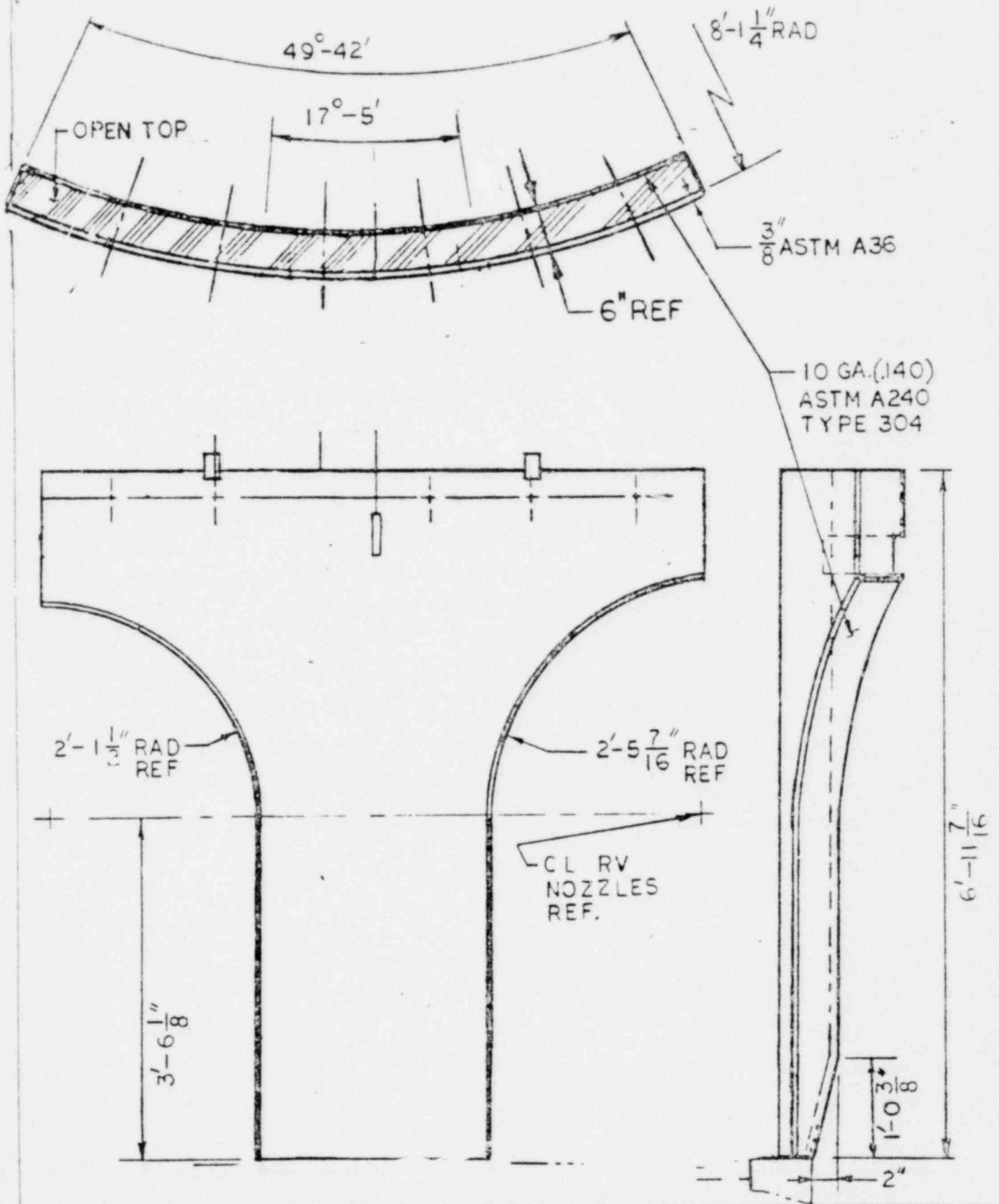
The Seller shall repair or replace any item which is torn, gouged, or broken upon receipt. The proposed repair procedure shall be submitted with the supplier's bid. 5.11
5.12

X. PROPOSAL DATA 5.15

1. The bidder shall include in his proposal sufficient data and other information to completely describe the material to facilitate a thorough evaluation. The proposal information shall include, but not be limited to, the following: 5.16
5.19
5.21

7.

- a. Completed summary of proposal 5.23
 - b. Warranty statement 5.24
 - c. Material specification and data sheets 5.25
 - d. Performance test reports 5.26
 - e. Quality Assurance and Quality Control procedures to be followed during the mixing and injecting of the neutron shielding material. 5.27
5.28
 - f. Any other data or information requested elsewhere in this document. 5.29
 - g. Any other information that may assist the Purchaser or Engineers in evaluating the bidder's proposal. 5.30
2. If the bidder takes exception to any of these requirements, these exceptions should be listed separately under the heading "Exceptions to Requirements." Any exceptions should reference the paragraph to which exception is taken. If no exceptions are taken, the bidder should so state. 5.32
5.33
5.34
5.35



POWER INDUSTRY GROUP TITLE

CHECKED *AP Clicks*

CORRECT *Stonchen*

APPROVED *[Signature]*

REVISIONS ②

③

④

⑤

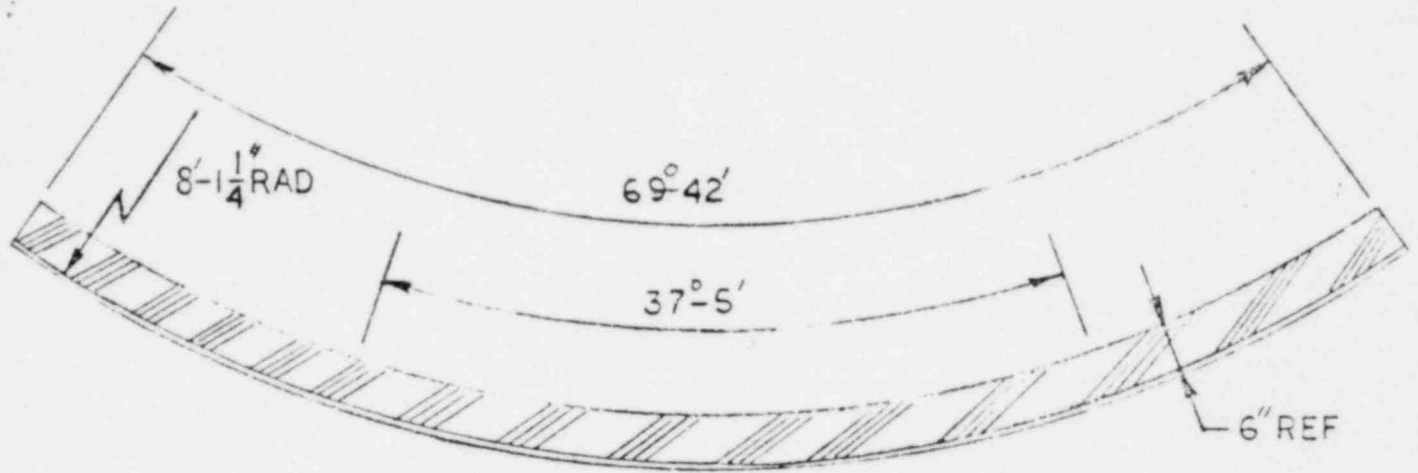
50° SHIELD
(QTY-3)

SCALE: NONE

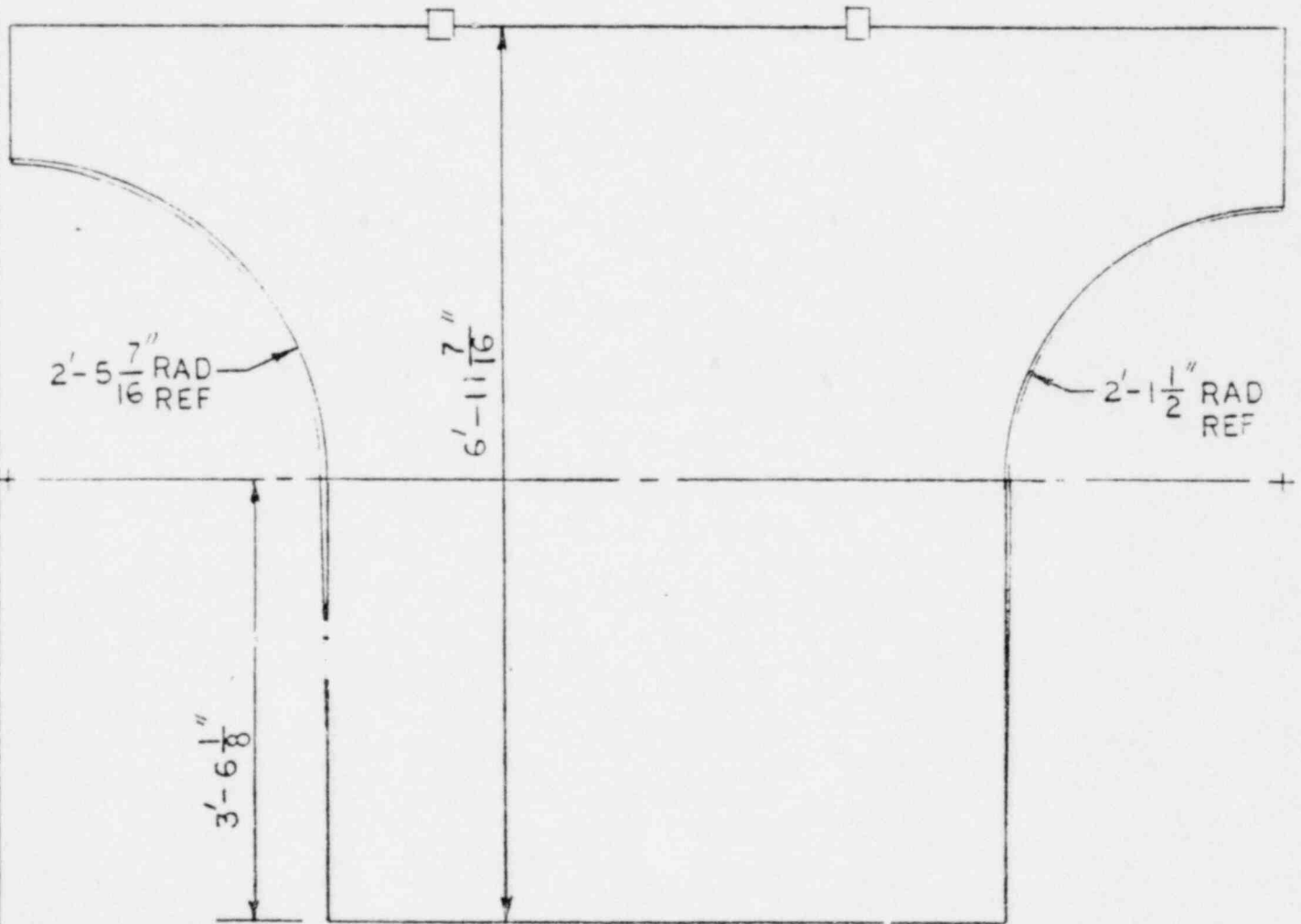
DATE: 5-24-79

SKETCH NUMBER SH. 108 IC

13075-MKS-13



NOTE: MATERIALS OF CONSTRUCTION AND SIDE VIEW DETAILS SHOWN ON 50° SHIELD



POWER INDUSTRY GROUP TITLE

CHECKED *McClure*

CORRECT *Strick*

APPROVED *[Signature]*

REVISIONS (2) (3) (4) (5)

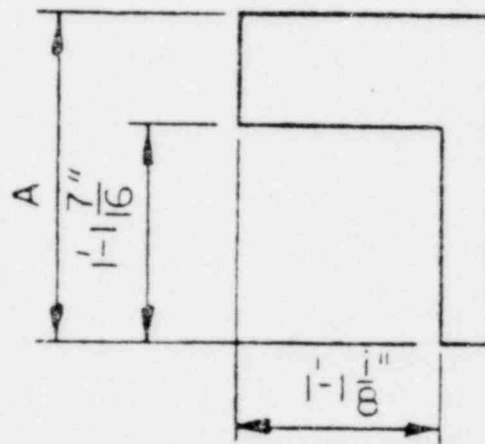
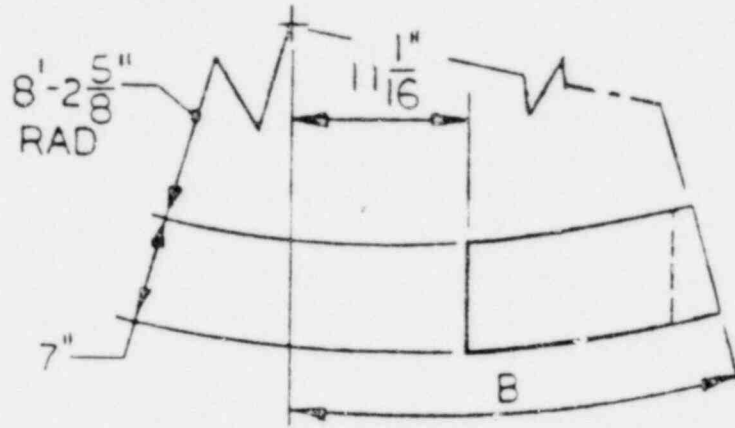
70° SHIELD
(QTY-3)

SCALE: NONE

DATE: 5-24-79

SKETCH NUMBER SH. 20+10

13075-MK 5-13



ITEM	A	B	TYPE	QTY
17-1	$1'-5 \frac{5}{16}"$	$18^{\circ}39'$	AS SHOWN	3
17-2	$1'-5 \frac{5}{16}"$	$18^{\circ}39'$	OPP HAND	3
18-1	$1'-8 \frac{7}{16}"$	$16^{\circ}10'$	AS SHOWN	3
18-2	$1'-8 \frac{7}{16}"$	$16^{\circ}10'$	OPP HAND	3

DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
$\pm \frac{1}{32}$	$\pm \frac{1}{16}$	$\pm \frac{1}{8}$	$\pm 0^{\circ}-30'$

(Handwritten initials)
VMC

POWER INDUSTRY GROUP TITLE

CHECKED *PP clake*

CORRECT *PP clake*

APPROVED *(Signature)*

REVISIONS ②

③

④

⑤

BLOCK

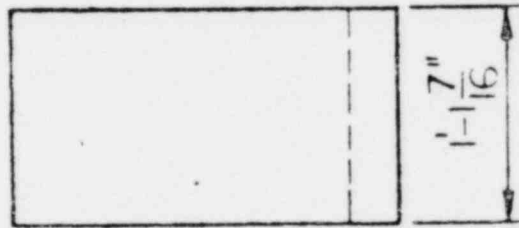
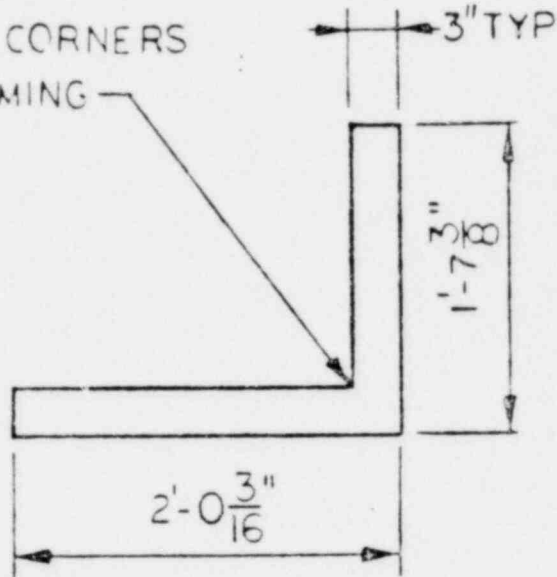
SCALE: NONE

DATE: 5-24-79

SKETCH NUMBER SH 30710

13075-MKS-13

CLEAN OUT CORNERS
AFTER FORMING



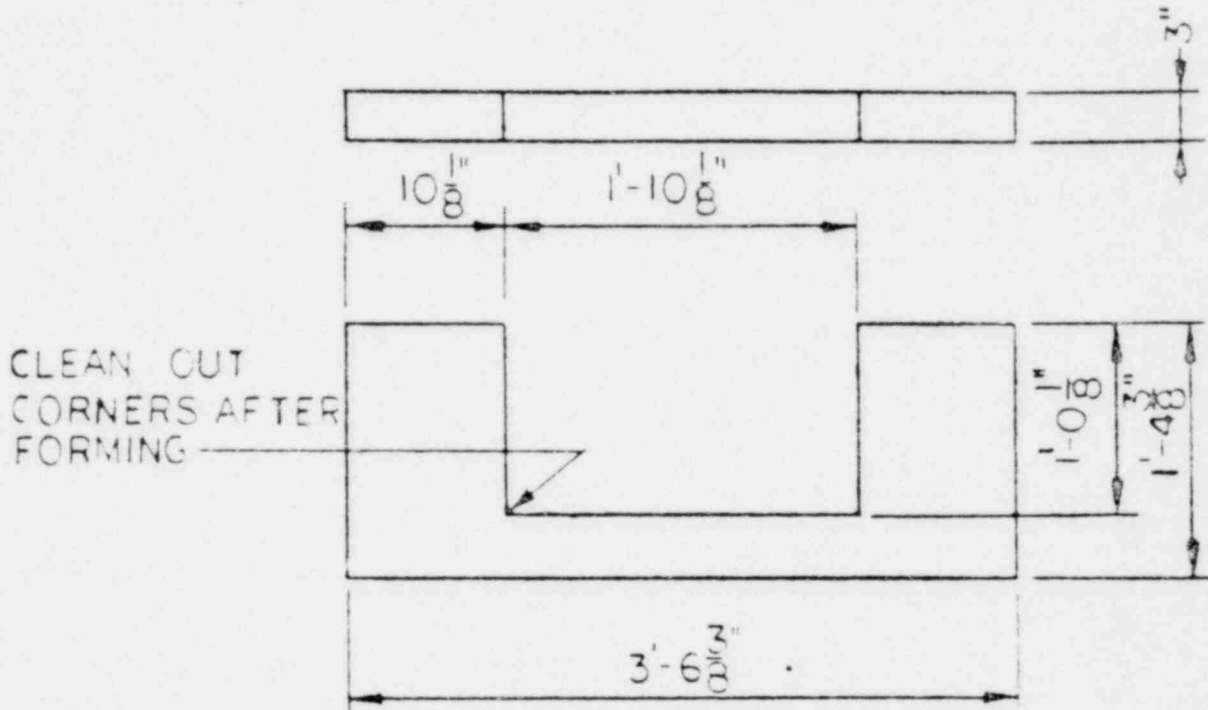
ITEM *	QTY
17-3	6
18-3	6

DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
$\pm \frac{1}{32}$	$\pm \frac{1}{16}$	$\pm \frac{1}{8}$	$\pm 0'-30'$

VMC.

* IDENTICAL PIECES GIVEN
DIFFERENT ITEM NOS. TO
FACILITATE INSTALLATION

POWER INDUSTRY GROUP	TITLE	SCALE: NONE
CHECKED <i>AR</i>	CORNER	DATE: 5-24-79
CORRECT <i>J.T.M.</i>		SKETCH NUMBER 5H. 40710
APPROVED <i>[Signature]</i>		13075-MKS-13
REVISIONS	(2) (3) (4) (5)	



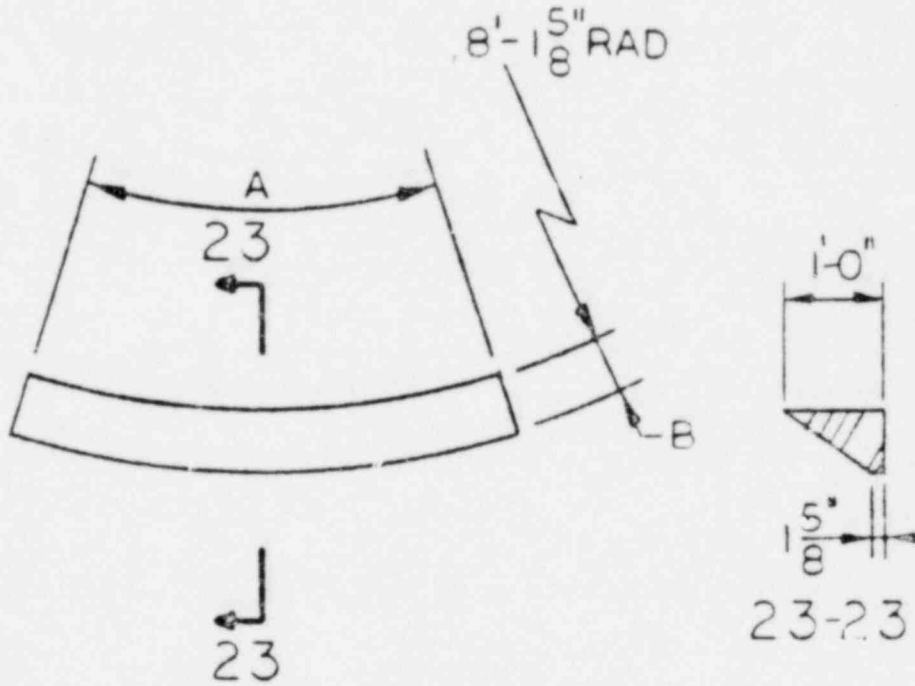
ITEM*	QTY
17-4	3
18-4	3

DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
$\pm \frac{1}{32}$	$\pm \frac{1}{16}$	$\pm \frac{1}{8}$	$\pm 0'-30'$

CD VPE

* IDENTICAL PIECES GIVEN DIFFERENT ITEM NOS. TO FACILITATE INSTALLATION

POWER INDUSTRY GROUP	TITLE	SCALE: NONE			
CHECKED <i>APChick</i>	TOP	DATE: 5-24-79			
CORRECT <i>J.M.C.S.</i>					
APPROVED <i>[Signature]</i>					
REVISIONS	②	③	④	⑤	SKETCH NUMBER SH 504 10 13075-MKS-13

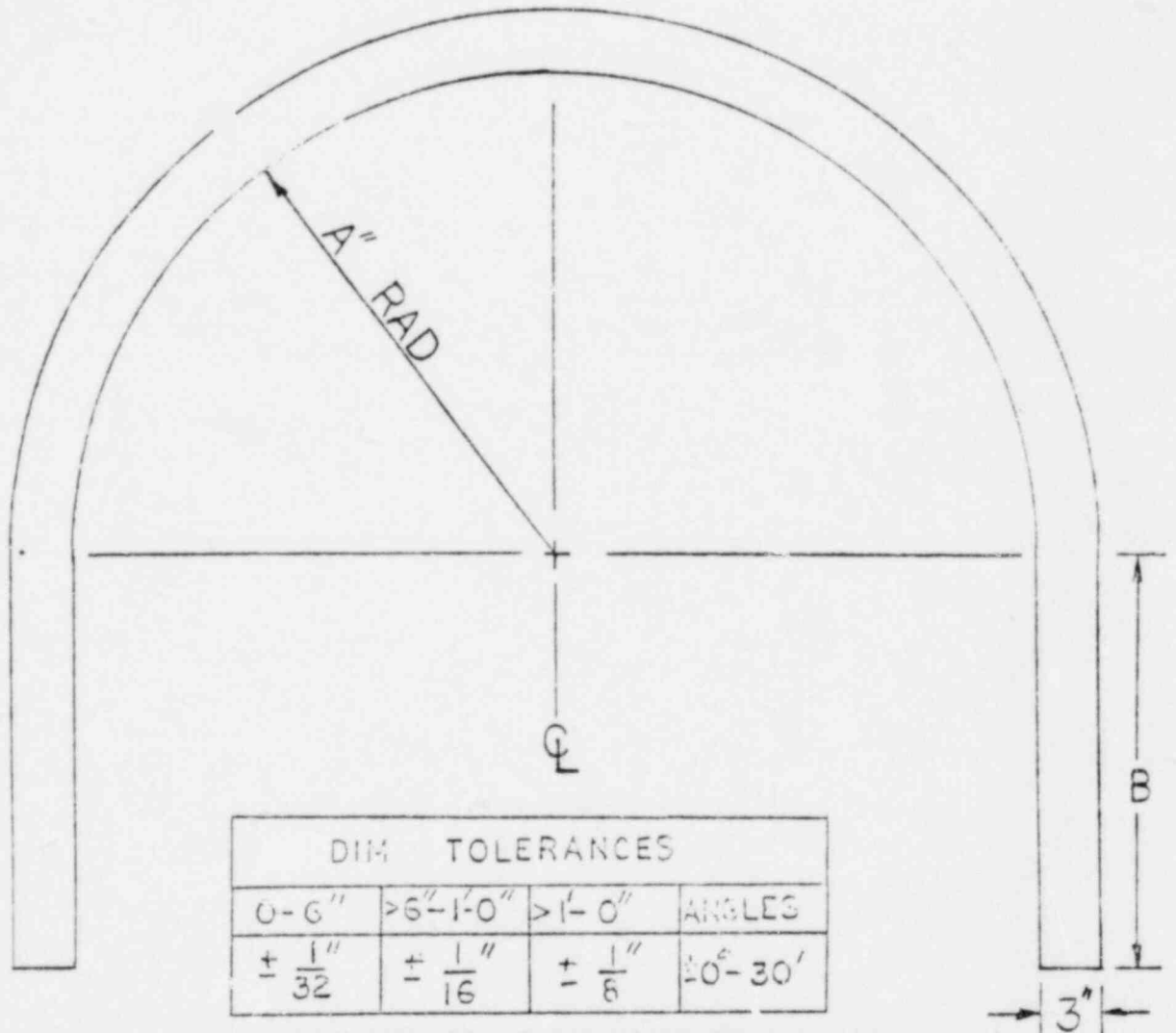
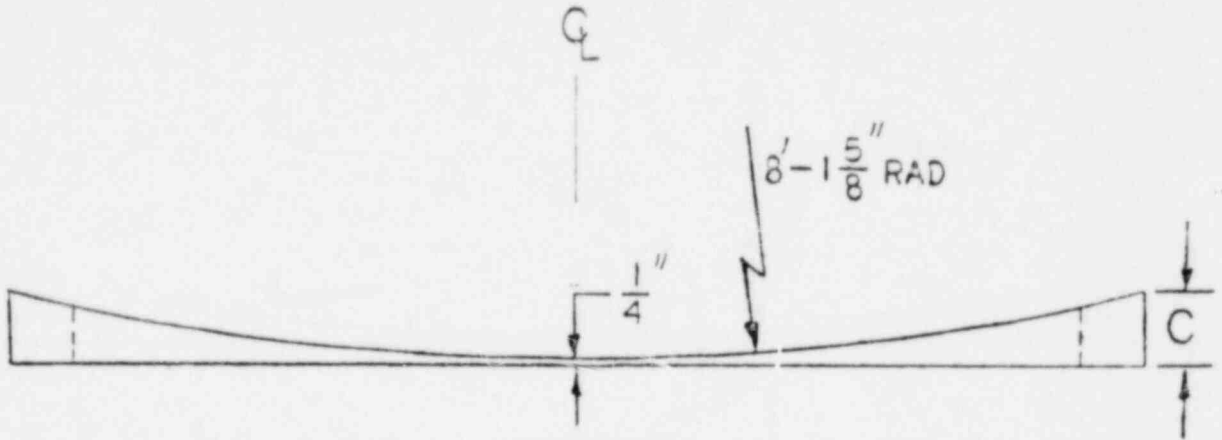


ITEM	A	B	QTY
11	35°-2'	7 3/4"	2
12	15°-2'	7 3/4"	3
13	35°-2'	6 1/2"	1

DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
± 1/32"	± 1/16"	± 1/8"	±0°-30'

(Signature)
V.M.R.

POWER INDUSTRY GROUP	TITLE	SCALE: NONE
CHECKED <i>(Signature)</i>	FILLER	DATE: 5-24-79
CORRECT <i>(Signature)</i>		SKETCH NUMBER SH. 60P10
APPROVED <i>(Signature)</i>		13075-MKS-13
REVISIONS		②
		④
		⑤



DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
$\pm \frac{1}{32}$	$\pm \frac{1}{16}$	$\pm \frac{1}{8}$	$\pm 0^{\circ}-30'$

ITEM	A	B	C	QTY
19	$29\frac{1}{16}''$	$26\frac{13}{16}''$	$5.649''$	3
24	$25\frac{1}{4}''$	$21\frac{7}{16}''$	$4.419''$	3

VMC.

POWER INDUSTRY GROUP TITLE

CHECKED *Al Cliche*

CORRECT *J.M.C.*

APPROVED *[Signature]*

REVISIONS (2) (3) (4) (5)

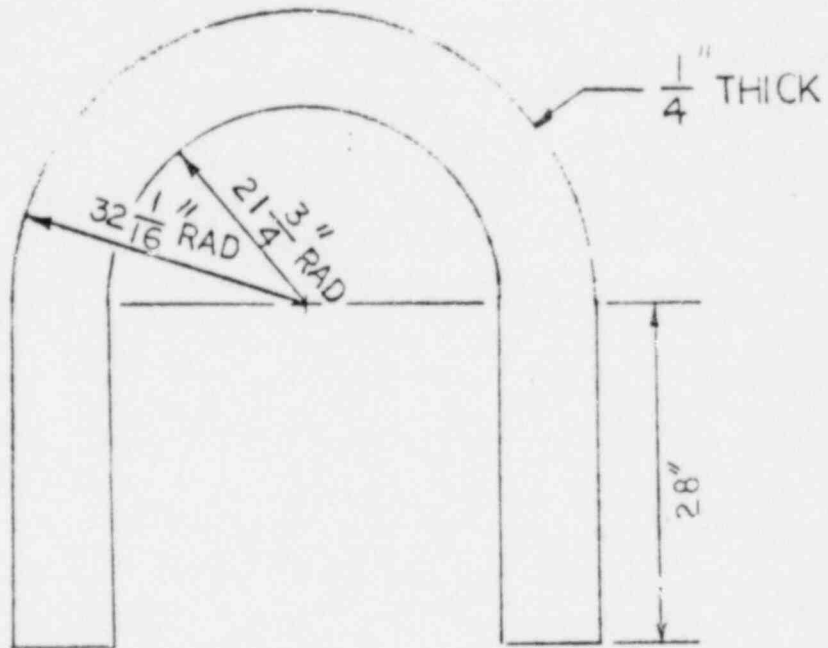
SADDLE

SCALE: NONE

DATE: 5-24-79

SKETCH NUMBER SH. 7 of 10

13075-MKS-13



DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
± 1/32"	± 1/16"	± 1/8"	±0°-30'

ITEM	QTY
21	84
26	29

POWER INDUSTRY GROUP TITLE

CHECKED *AK Clibe*

CORRECT *stmcb*

APPROVED *[Signature]*

REVISIONS ②

③

④

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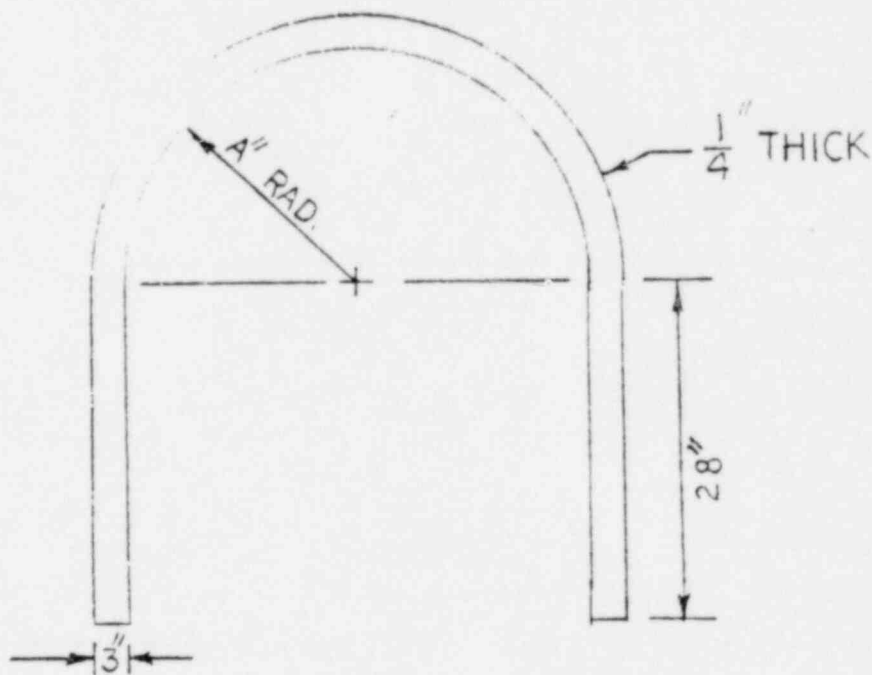
SADDLE

SCALE: NONE

DATE: 5-24-79

SKETCH NUMBER SH 86710

13075-MKS-13



DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
$\pm \frac{1}{32}$	$\pm \frac{1}{16}$	$\pm \frac{1}{8}$	$\pm 0^{\circ}-30'$

ITEM	A	QTY
20	$29 \frac{1}{16}$ "	136
22	$23 \frac{3}{4}$ "	140
23	$21 \frac{3}{4}$ "	69
25	$25 \frac{1}{4}$ "	140
27	$23 \frac{7}{16}$ "	202
28	$21 \frac{7}{16}$ "	41

POWER INDUSTRY GROUP TITLE

CHECKED *RR Clarke*

CORRECT *Franklin*

APPROVED *[Signature]*

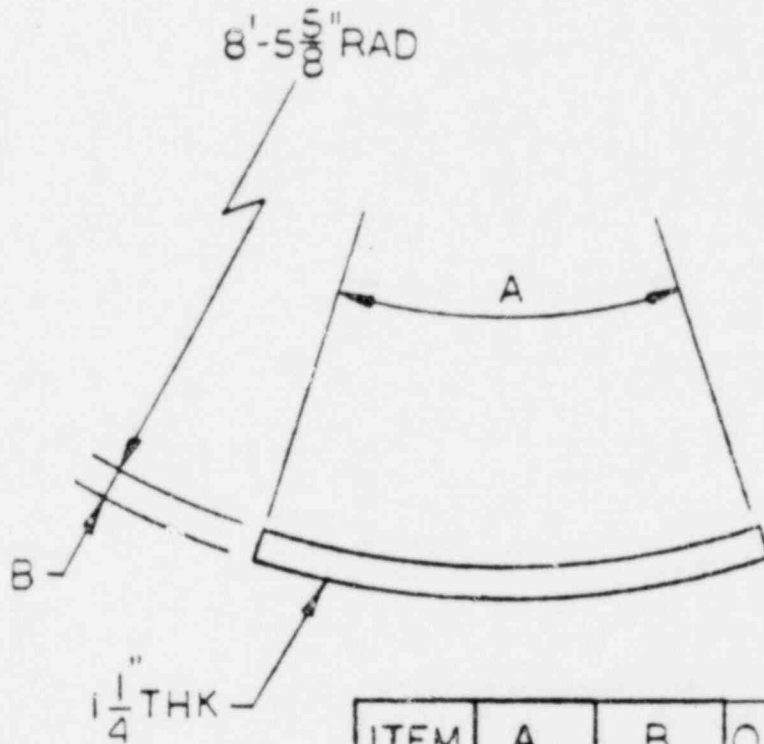
REVISIONS (2) (3) (4) (5)

SADDLE

SCALE: NONE

DATE: 5-24-79

SKETCH NUMBER SH. 90710
13075-MKS-13



ITEM	A	B	QTY
14	35°2'	4'	2
15	15°2'	4'	3
16	35°2'	3"	1

DIM TOLERANCES			
0-6"	>6"-1'-0"	>1'-0"	ANGLES
± 1/32"	± 1/16"	± 1/8"	±0°-30'

(Signature)
VME

POWER INDUSTRY GROUP		TITLE	SCALE: NONE	
CHECKED	<i>ARChick</i>	SUPPORT	DATE: 5-24-79	
CORRECT	<i>JTMS</i>		SKETCH NUMBER SH-10.7/0	
APPROVED	<i>(Signature)</i>		13075-MKS-13	
REVISIONS	②	③	④	⑤

VIRGINIA ELECTRIC AND POWER COMPANY

CERTIFICATE OF CONFORMANCE

Project Name _____

Seller _____ Address _____

Item or Service _____ Mark No. _____

Specification No. and Title _____

Purchase Order No. _____ J.O.No. _____

Seller Identifying No. _____ Drawing No. _____

Deviations from Specification Requirements: (If none, so state) attach

Copies of Deviation Approval Documents

- | | |
|----------|----------|
| 1. _____ | 4. _____ |
| 2. _____ | 5. _____ |
| 3. _____ | 6. _____ |

The Seller, including his subsuppliers, hereby certifies that the item or service, supplied on this order, complies with the above-listed specifications, drawings, applicable codes, standards, and procedures. The Seller certifies that all deviations from specification requirements are listed above and that deviation approval documents are attached.

Signature _____
Quality Assurance Manager
or Equivalent

INSTRUCTIONS FOR COMPLETING VEPCO'S
CERTIFICATE OF CONFORMANCE

The vendor shall complete the lines numbered below, as applicable, and return one copy with the shipment, and another copy mailed to the project location - ATTN: Resident QC Engineer:

<u>LINE NO.</u>	<u>INSTRUCTIONS</u>
1.	PROJECT NAME (Surry, North Anna)
2.	VENDORS NAME
3.	VENDORS ADDRESS
4.	NAME OF COMPONENT OR SERVICE PERFORMED
5.	MARK OR PART NUMBER OF COMPONENT
6.	THE SPECIFICATION TITLE, NUMBER, REVISION AND DATE, SPECIFIC INDUSTRY STANDARDS OR CODES ATTESTED TO AS REQUIRED BY THE PURCHASE ORDER.
7.	VEPCO PURCHASE ORDER NUMBER PLUS ANY CHANGE ORDERS
8.	JOB ORDER NUMBER
9.	VENDORS JOB NUMBER OR SHOP NUMBER
10.	APPROVED FABRICATION OR ENGINEERING DRAWINGS AND LATEST REVISION
11.	ALL VENDOR DEVIATIONS FROM THE PURCHASE ORDER OR SPECIFICATION WITH APPROVAL LETTERS ETC. TO VERIFY ACCEPTANCE OF DEVIATION
12.	Q.A. MANAGER OR EQUIVALENT RESPONSIBLE VENDOR REPRESENTATIVE

ATTACHMENT B

THE UNIVERSITY OF MICHIGAN
PHOENIX MEMORIAL LABORATORY
FORD NUCLEAR REACTOR
ANN ARBOR, MICHIGAN 48109
(313) 764-6220

September 17, 1979

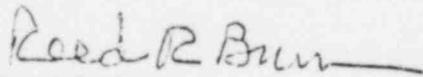
Mr. Vincent Cataldo
Chemtrol Corporation
8805 Solon
Building G5
Houston, Texas 77064

Dear Mr. Cataldo:

Enclosed are tabulations of the total gamma and neutron doses administered to samples of Chemtrol material CT-40-NS Type B. Samples were irradiated at $380 \pm 20^{\circ}\text{F}$ for 110 hours and 545.6 hours at two megawatts. The specific gravities of three separate pieces from the 110 hour samples are also enclosed.

Pieces of the 110 hour samples have been sent to Southwest Research for analysis. The 545.6 hour samples are still too radioactive to be handled. Present plans are to prepare them for analysis on September 24, 1979.

Sincerely,



Reed R. Burn
Reactor Manager

enclosure

RRB/sv

CHEMTROL MATERIAL CT-40-NS TYPE B

Short Term Irradiation

Irradiation Temperature: $380 \pm 20^\circ\text{F}$

<u>Irradiation</u>	<u>2 MW Time</u>	<u>Measured Dose Rate</u>	<u>Dose Rate -10%</u>	<u>Total Dose</u>	
				<u>Actual</u>	<u>Desired</u>
Gamma	110 hr	8.4×10^7 rad/hr	7.6×10^7	8.4×10^9	8.25×10^9
$n < .11$ MeV	110 hr	5.6×10^{12} n/cm 2/5	5.0×10^{12}	2.0×10^{18}	1.50×10^{18}
$.11$ MeV $< n < 1.1$ MeV	110 hr	2.4×10^{12}	2.2×10^{12}	8.7×10^{17}	7.5×10^{17}
$n > 1.1$ MeV	110 hr	2.0×10^{12}	1.8×10^{12}	7.1×10^{17}	3.5×10^{16}

Long Term Irradiation

Gamma	545.6 hr	8.4×10^7 rad/hr	7.6×10^7	4.1×10^{10}	3.3×10^{10}
$n < .11$ MeV	545.6 hr	5.6×10^{12} n/cm 2/5	5.0×10^{12}	9.8×10^{18}	6.0×10^{18}
$.11$ MeV $< n < 1.1$ MeV	545.6 hr	2.4×10^{12}	2.2×10^{12}	4.3×10^{18}	3.0×10^{18}
$n > 1.1$ MeV	545.6 hr	2.0×10^{12}	1.8×10^{12}	3.5×10^{18}	1.4×10^{17}

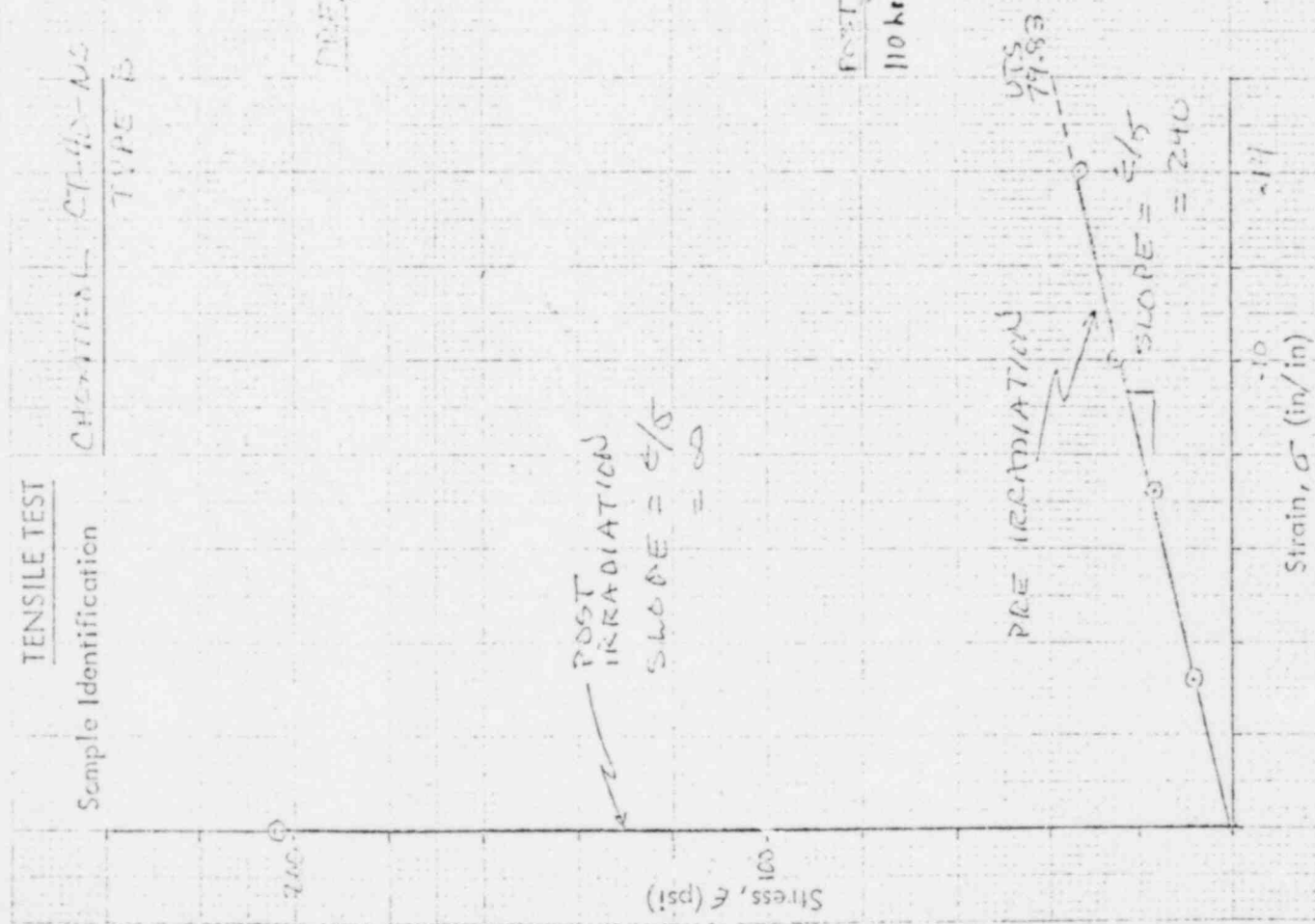
FORD NUCLEAR REACTOR
UNIVERSITY OF MICHIGAN

SPECIFIC GRAVITY

Sample Number	Dry Weight			Net Weight			Volume	Specific Gravity	Water Temperature		
	Net A	Tare B	Sample C = A - B	Net D	Tare E	Sample F = D - E			H = C - FG	C/H	°C
	(gm)	(gm)	(gm)	(gm)	(gm)	(gm)	(cm ³)	(gm/cm ³)		25	
POST-IRRADIATION (110 hr.)	11.42021	9.20000	2.22021	1.84024	1.00000	.84024	1.37745	1.61183	G = 1.0030		
	CHEMTROL MATERIAL CT-40-NS TYPE B								Correction		
									T	Value	
	11.42077	9.20000	2.22077	1.85148	1.00000	.85148	1.36674	1.62487	°C	G	
									20	1.0018	
									.2	1.0018	
									.4	1.0019	
	12.49373	9.20000	3.29373	2.25730	1.00000	1.25730	2.03266	1.62041	.6	1.0019	
									.8	1.0020	
									21	1.0020	
								.2	1.0021		
								.4	1.0021		
								.6	1.0022		
								.8	1.0022		
								22	1.0023		
								.2	1.0023		
								.4	1.0024		
								.6	1.0024		
								.8	1.0025		
								23	1.0025		
								.2	1.0026		

FORD NUCLEAR REACTOR
UNIVERSITY OF MICHIGAN

TENSILE TEST		CHEMICAL CT-40-MS		TYPE B		D. Cross Sectional Area		PRE POST	
Sample Identification								.115 .012 in	
A	B	A	B	A	B	A	B	A	B
Tensile Force (lb)	Sample Length (in)	Sample Elongation (in)	Stress ϵ (psi)	Strain σ (in/in)	Sample Elongation (in)	Stress ϵ (psi)	Strain σ (in/in)	Sample Elongation (in)	Stress ϵ (psi)
	$L_0 =$	$C = B - L_0$		C/L_0					
0	2.50	0	0	0	0	0	0	0	0
1	2.58	.08	8.40	.032	.08	8.40	.032	.08	8.40
2	2.68	.18	16.81	.072	.18	16.81	.072	.18	16.81
3	2.75	.25	25.21	.100	.25	25.21	.100	.25	25.21
4	2.85	.35	33.61	.140	.35	33.61	.140	.35	33.61
9.5	UTS		79.83	.250		79.83	.250		79.83
0	1.80	0	0	0	0	0	0	0	0
20	1.30	0	204.08	0	0	204.08	0	0	204.08
	UTS								



THE UNIVERSITY OF MICHIGAN
PHOENIX MEMORIAL LABORATORY
FORD NUCLEAR REACTOR
ANN ARBOR, MICHIGAN 48105

August 13, 1979

Mr. Vince Cataldo
Chemtrol Corporation
8805 Solon
Building G5
Houston, TE 77064

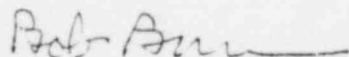
Dear Vince:

Enclosed is sample B6 of your CT-40NS, Type B material which has been irradiated as follows:

Gamma		8.4×10^9 rad
Neutron	$E < .11$ MeV	2.0×10^{18} n/cm ²
	$.11$ MeV $< E < 1.1$ MeV	8.7×10^{17}
	$E > 1.1$ MeV	7.1×10^{17}

The gamma dose rate on contact from this sample was approximately 50 mrem/hr. 16 days after removal from the reactor core.

Sincerely,



Reed R. Burn
Reactor Manager
Ford Nuclear Reactor

Enclosure



RECEIVED

AUG 2 1979

Chemtrol

THE UNIVERSITY OF MICHIGAN
PHOENIX MEMORIAL LABORATORY
FORD NUCLEAR REACTOR
ANN ARBOR, MICHIGAN 48109
(313) 764-6220

July 29, 1979

Dear Vince;

Enclosed are the residual radioactivity results from the short term sample which was irradiated at 350-400 degrees.

The long term sample completed its first cycle at temperature and is doing well.

Sincerely,

Bob Burr

DISTRIBUTION LIST:

- | | |
|-------------------------------------|---|
| <input type="checkbox"/> Young | <input type="checkbox"/> Catala- |
| <input type="checkbox"/> Alford | <input type="checkbox"/> Welch |
| <input type="checkbox"/> Block | <input type="checkbox"/> Fleming |
| <input type="checkbox"/> Chute | <input type="checkbox"/> Yeldell |
| <input type="checkbox"/> Fariss | <input type="checkbox"/> Bell <input type="checkbox"/> Pank |
| <input type="checkbox"/> Ferris | <input type="checkbox"/> West Coast C. |
| <input type="checkbox"/> Spriggs | <input type="checkbox"/> Shop |
| <input type="checkbox"/> Russell | <input type="checkbox"/> Subject |
| <input type="checkbox"/> G/A | <input type="checkbox"/> Reading |
| <input type="checkbox"/> Insurance | <input type="checkbox"/> _____ |
| <input type="checkbox"/> Accounting | <input type="checkbox"/> _____ |

 ***** 20 JUL 1979 2:39:28 PM *****

CHEMTRON EXPERIMENT

SAMPLE DATE: 19JUL79 1446:00
 SAMPLe IDENTIFICATION: 110
 TYPE OF SAMPLe: SOLID
 SAMPLe QUANTITY: 0.1510000 UNITS: GRAMS
 SAMPLe GEOMETRY: SAMPLe GEOMETRY
 EFFICIENCY FILE NAME: EFF.TABM

ACQUISITION DATE: 19JUL79 1446:00 * FWHM(1332) 2.719
 PRESET TIME(LIVE): 1800. SEC * SENSITIVITY: 16.000
 ELAPSED REAL TIME: 1840. SEC * SHAPE PARAMETER : 10.0 %
 ELAPSED LIVE TIME: 1800. SEC * NBR ITERATIONS: 5

DETECTOR: ADC DETECTOR * LIBRARY: NUCL.LIBM
 DATE CALIBRATED: 19JUL79 744:00 * ENERGY TOLERANCE: 2.500KV
 KEV/CHNL: 1.0012299 * HALF LIFE RATIO: 8.00
 OFFSET: -0.7146839 KEV * ABUNDANCE LIMIT: 50.00%
 Q. COEFF. : -2.733E-07 KEV/C**2 *

ENERGY WINDOW 99.405 TO 2000.652

PK	IT	ENERGY	AREA	BKGD	FWHM	CHANNEL	LEFT	PW	CTS/SEC	NERR	FIT
1	1	510.85	20633.	13157.	3.20	511.01	504	14	1.146E-01	1.1	4.51E-01
2	1	602.67	1422.	8638.	2.73	602.74	599	9	7.903E-01	9.6	1.29E-01
3	1	889.01	3427.	16571.	2.52	888.84	885	9	1.904E-00	5.6	4.13E-01
4	1	1115.21	182385.	4137.	2.58	1114.89	1107	13	1.013E-02	0.2	1.79E-01
5	1	1172.95	361.	236.	3.35	1172.99	1167	16	2.008E-01	8.0	3.87E-01
6	1	1291.44	436.	257.	3.09	1291.02	1284	19	2.424E-01	7.1	1.78E-01
7	1	1332.17	312.	117.	2.63	1331.73	1326	13	1.732E-01	7.5	1.09E-01
8	1	1407.33	141.	153.	3.31	1406.86	1401	18	7.819E-02	15.0	2.40E-01
9	1	1460.48	137.	75.	3.04	1459.98	1456	10	7.626E-02	12.3	1.71E-01
10	1	1690.46	229.	56.	2.63	1689.88	1683	16	1.326E-01	7.8	2.13E-01

PEAK SEARCH COMPLETED

60-01 NUCLIDE IDENTIFICATION SYSTEM
SUMMARY OF NUCLIDE ACTIVITY

PAGE 3

TOTAL LINES IN SPECTRUM 10
 LINES NOT LISTED IN LIBRARY 0
 IDENTIFIED IN SUMMARY REPORT 10 100.00%

ACTIVATION GAS

NUCLIDE	HLIFE	HLSEC	DECAY	UC/UT	ERROR	%ERR
AR-41	1.83E 00H	6.588E 03	6.111E -4	7.646E -3	5.400E -4	7.06

ACTIVATION PRODUCT

uci/gm

NUCLIDE	HLIFE	HLSEC	DECAY	UC/UT	ERROR	%ERR
CO-60	5.26E 00Y	1.660E 08	5.556E -4	5.062E -3	3.790E -4	7.49
ZN-65	2.44E 02D	2.108E 07	5.556E -4	4.971E 0	1.190E -2	0.24
SB-124	6.02E 01D	5.201E 06	5.556E -4	1.061E -2	8.329E -4	7.85
SC-46	8.39E 01D	7.249E 06	5.556E -4	3.858E -2	2.153E -3	5.58
CS-134	2.05Y	6.457E 07	5.556E -4	**KEY LINE NOT PRESENT**		

NATURAL PRODUCT

NUCLIDE	HLIFE	HLSEC	DECAY	UC/UT	ERROR	%ERR
K-40	1.00E 03Y	3.156E 10	5.556E -4	2.210E -2	2.728E -2	12.35



THE UNIVERSITY OF MICHIGAN
PHOENIX MEMORIAL LABORATORY
FORD NUCLEAR REACTOR
ANN ARBOR, MICHIGAN 48109
(313) 764-6220

RECEIVED

OCT 10 1979

Chemtrol

October 5, 1979

Mr. Vincent Cataldo
Chemtrol Corporation
8805 Solon
Building G5
Houston, Texas 77064

Dear Vince:

I sent both short term and long term samples of CT-40-NS-Type B to Ron Estaphan, Southwest Research Institute and G.L. Elam, Wyle Laboratories. G.L. Elam is Wyle's Radiation Control Officer. You might contact the actual user of the material to warn him to expect it. Enclosed are letters sent preceding the packages.

Sincerely,

Bob Brun

Reed R. Brun
Reactor Manager

Enclosures

RRB/sv

THE UNIVERSITY OF MICHIGAN
PHOENIX MEMORIAL LABORATORY
FORD NUCLEAR REACTOR
ANN ARBOR, MICHIGAN 48109
(313) 764-6220

October 5, 1979

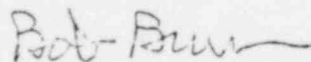
Mr. G. L. Elam
Radiation Control Officer
Wyle Laboratories
7800 Governor's Drive, West
Huntsville, Alabama 35807

Dear Mr. Elam:

I have packaged and sent to you two irradiated samples of Chemtrol silicone polymer base neutron shielding material CT-40-NS-Type B. The samples are in lead pigs within the box. Each sample is labeled on its lead pig. An irradiation data sheet is enclosed which provides irradiation details.

The samples are radioactive. Dose rates are provided on each lead pig.

Sincerely,



Reed R. Burn
Reactor Manager

Enclosure

RRB/sv

xc: Vincent Cataldo
Chemtrol Corporation

ATTACHMENT C

DOW CORNING

April 16, 1980

RECEIVED
APR 21 1980

Mr. Vincent M. Cataldo
Chemtrol Corporation
P.O. Box 38556
Houston, TX 77088

Chemtrol

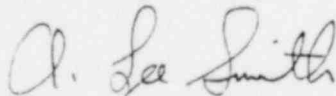
Dear Mr. Cataldo:

In response to your recent telephone inquiry about the labeling of Sample No. 16705 (CT-40NS Type B), designated as irradiated with 110 hours exposure per R. G. Niemi's letter of March 3, 1980, I have talked further with Mr. R. R. Burn at the University of Michigan and also rechecked our records. Mr. Burn sent us two samples on December 7, 1979; one irradiated 110 hours, the other 545.6 hours. In recording these samples, our sample clerk assigned both of them the same sample number (16705). Thus, we do not know which of these two samples was actually analyzed. Since the identification of the material labeled 16705 is ambiguous, I suggest that you simply delete that data from your consideration. The sequence of samples is therefore as follows:

Sample 12751 - Non-irradiated
12750 - Irradiated 110 hours
17296 - Irradiated 545.6 hours

I regret the inconvenience that this confusion has caused, and hope the data will now be useful to you.

Yours very truly,



A. Lee Smith
Manager
Analytical Services Dept.

ALS/jlm

xc: Mr. Reed R. Burn, Reactor Mngr.
Ford Nuclear Reactor
Phoenix Memorial Laboratory
Ann Arbor, MI 48109

R. G. Niemi (Dow Corning Corp.)



RECEIVED

OCT 13 1979

Chemtrol

October 8, 1979

Mr. Vincent M. Cataldo
Chemtrol Corporation
P.O. Box 38556
Houston, TX 77088

Dear Mr. Cataldo:

We received the two cured samples of SYLGARD® 170 elastomer which you submitted specially formulated with boron carbide filler and designed for neutron shielding applications. As per your communication and designation, sample B-7 has undergone neutron and gamma irradiated exposure at the University of Michigan. For comparison, your sample designated B-12 represents a non-irradiation control. Both samples were submitted to our Analytical Department (Dr. A. L. Smith) for specific element and specific gravity analysis. These results, along with our Analytical Department reference numbers, and Corporate Test Methods (CTM) are listed below:

<u>CTM</u>	<u>Element/Property</u>	<u>B-7 (12750) Irradiated</u>	<u>B-12 (12751) Non-Irradiated</u>
0030	Carbon (C)	14.22 ± .14	17.77 ± .17
0030	Hydrogen (H)	3.00 ± .09	4.38 ± .12
0522	Silicon (Si)	41.2 ± 0.2	40.5 ± 0.2
0540	Specific Gravity	1.571 ± .005	1.359 ± .005

I hope that this analytical data will satisfy your requirements. Let me know if you have any questions.

Very truly yours,

R. G. Niemi
Sr. Specialist
Elastomers, Technical Service
and Development
Phone: (517)-496-5380

sdk



March 3, 1980

RECEIVED

MAR 7 1980

Chemtrol

Mr. Vincent M. Cataldo
Chemtrol Corporation
P. O. Box 38556
Houston, TX 77088

Dear Mr. Cataldo:

We received the two additional samples of cured SYLGARD® 170 Elastomer that was specially formulated with boron carbide by Chemtrol (earlier samples were designated B-7 and B-12). These latest samples were designated by Chemtrol as CT-40 NS Type B and were respectively subjected to 110 hours and 545.6 hours of neutron and gamma irradiation exposure at the University of Michigan. These samples were submitted to our Analytical Department (Dr. A. L. Smith) for specific elemental and specific gravity analysis. These results, along with our Analytical Department reference numbers and Corporate Test Methods (CTM) are listed below:

CTM	Element/Property	(17296)	(16705)
		CT-40 NS Type B 545.6 Hours Exposure	CT-40 NS Type B 110 Hours Exposure
0030	Carbon (C)	5.96 ± 0.06%	4.97 ± 0.05%
0030	Hydrogen (H)	1.07 ± 0.03%	0.83 ± 0.02%
0522	Silicon (Si)	40.2 ± 0.5%	38.0 ± 1.0%
0540	Specific Gravity	1.681 ± 0.005	Insufficient Sample

Again, I hope that this analytical data will satisfy your requirements. Let me know if you have any questions.

Very truly yours,

R. G. Niemi
Sr. Technical Specialist
Elastomers, Technical Service
and Development
Phone: (517)-496-5380

sdk

ATTACHMENT D

SOUTHWEST RESEARCH INSTITUTE

POST OFFICE DRAWER 28510 · 6220 CULEBRA ROAD · SAN ANTONIO, TEXAS 78284 · (512) 684-5111

EVALUATION OF SILICONE POLYMERS

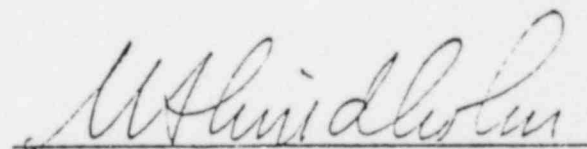
SwRI Project No. 02-5814-118
FINAL REPORT

Prepared for

Stone & Webster Engineering Corporation
P. O. Box 2325
Boston, Massachusetts 02107

June 9, 1980

APPROVED:



U. S. Lindholm, Director
Department of Materials Sciences

Prepared by:
David R. Williams



SAN ANTONIO, HOUSTON, TEXAS, AND WASHINGTON, D. C.

INTRODUCTION

Three pieces of irradiated silicone polymer were received at Southwest Research Institute (SwRI) from Wyle Laboratories on March 12, 1980. The specimens were packaged in plastic bags contained in lead pigs, which were in turn contained in a smaller box within the wooden shipping crate. The package was identified as containing radioactive material. The specimens were weighed and measured upon receipt to provide information for the design of sabots to guide them in the gun tube and for the determination of the gas pressure required to obtain the required velocity of 1500 fps. The information given on the labels affixed to the lead pigs is given in Table I, along with the initial and as-tested weights of the specimens and the muzzle velocities measured for each firing.

PROCEDURE

The specimens were received in the form of rectangular solid pieces. In order to fire them from the 1-in. bore diameter cold gas gun, they had to be sanded into smooth blocks and fitted with sabots to guide them down the gun barrel and contain the pressure behind them. The sabots were made of low-density urethane foam cast into a 1-in. inside diameter tube with a rectangular block inserted to form the cavity into which the specimen was placed for firing. Each sabot was cast as one piece, then cut into two side pieces plus a round pusher disk as shown in Figure 1. The projectile was then assembled by placing the specimen between the side pieces and inserting this assembly into the breech end of the gun, which opens at Joint A in Figure 2. The pusher disk was then inserted behind the specimen.

Occasionally, paper shims were inserted between the specimen and side pieces in order to achieve a good seal between the sabot and the gun tube to minimize the leakage of propellant gas past the projectile.

The gun used to accelerate the specimens uses compressed gas as propellant. A series of reservoirs accumulates the required pressure, and the gun is fired by venting the main reservoir into the gun tube behind the projectile. Two photodiodes are mounted near the muzzle of the gun to detect the passage of the front of the projectile. The photodiode output signals are fed to a waveform generator for direct display and to a signal conditioner which passes a step function to the waveform generator. The four waves are displayed on an oscilloscope. A camera attachment fixture on the oscilloscope allows the waves to be photographed for measurement. The first photodiode also detects the turbulent compressed gas which precedes the projectile down the barrel, so its output must be interpreted accordingly. The projectile passage is taken to be at the discontinuity in the slope of the output of the first photodiode. The muzzle velocity of the projectile is then calculated by dividing the distance between the photodiodes by the time required for the projectile to travel that distance. Figure 3 shows the oscilloscope traces obtained for the live specimens.

Figure 2 shows the gun and the specimen containment structure. Box A is open at both ends and serves to contain the fragments resulting from the stripping of the sabot from the specimen. Box B is the catch box, and contains the 34-lb concrete block target. Box B is lined with plastic sheeting and is sealed with a lid secured by four screws. Duct tape is used to seal the lid to the box. A 2-in. diameter aperture is provided for the passage of the projectile, with crossed slits in the plastic liner to allow free

entry of the specimen while containing the resulting debris. The distance from the muzzle of the gun to the target is 69.5 in. The Special Operating Procedure required by the SwRI Radiation Hazards Officer is included as an Appendix to this report.

CALIBRATION

The gun was calibrated by firing a series of dummy specimens out from a block of unirradiated material supplied by Stone & Webster Engineering Corporation. The calibration firings provided information on the relationship between reservoir pressure and muzzle velocity. The required velocity of 1500 fps could only be obtained by exceeding the 600 psi rating of the gun, and then only with relatively poor reliability. When attempting to achieve a velocity of 1500 fps, the actual velocity fell in a range of ± 100 fps of the desired value. This limitation is presumed to be caused by variations in the fit of the sabot in the gun tube, in the leakage of propellant gas past the sabot, and in the limits of resolution of photodiodes and associated electronic gear.

TESTING

The two live specimens were sanded to fit the sabots, then weighed. The sabots were also weighed to obtain the net projectile weight. Specimen A was fired first. The oscilloscope traces are given in Figure 3(a). Several minutes were allowed for the dust to settle inside the catch box before the box was opened and the remains of the specimen removed. The

concrete block showed some damage as a result of the impact of the specimen. A region approximately 1 in. in diameter was chipped out to a maximum depth of approximately .13 in. (visual estimate). The fragments of specimen, sabot, and concrete were removed from the box and returned to a plastic specimen bag. Since the specimen was reduced to a fine powder, and since other debris was also present, no attempt was made to quantify the particle size distribution.

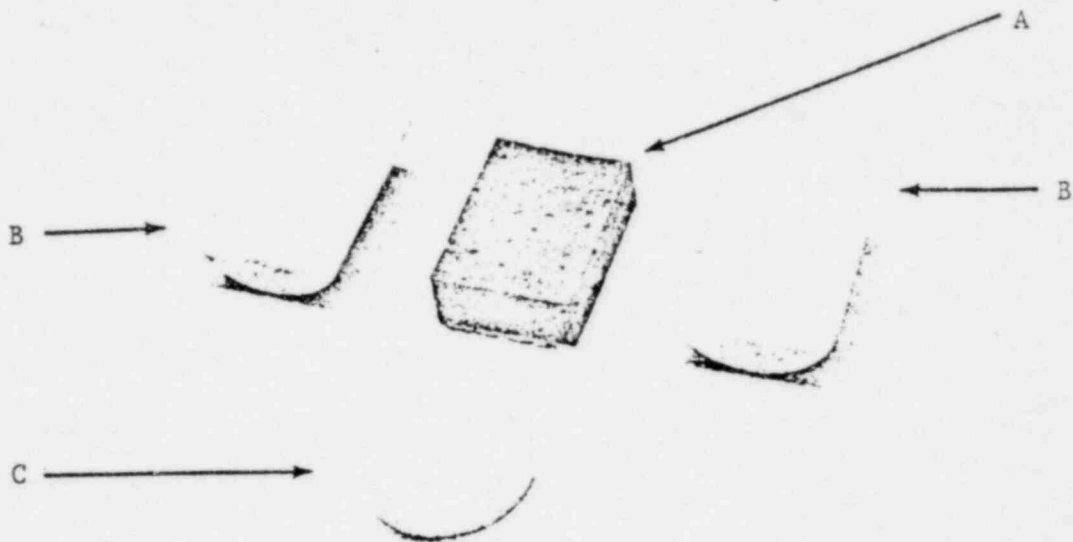
Specimen B was fired next, using a fresh face of the target block. The oscilloscope trace is given in Figure 3(b). Again, the specimen was reduced to a fine powder on impact. The damage to the target block was less severe, with a maximum penetration of only about .07 in. by visual estimate. The powder was removed from the box to a separate specimen bag, which was placed inside another specimen bag containing a chip removed from this specimen before firing.

CONCLUSION

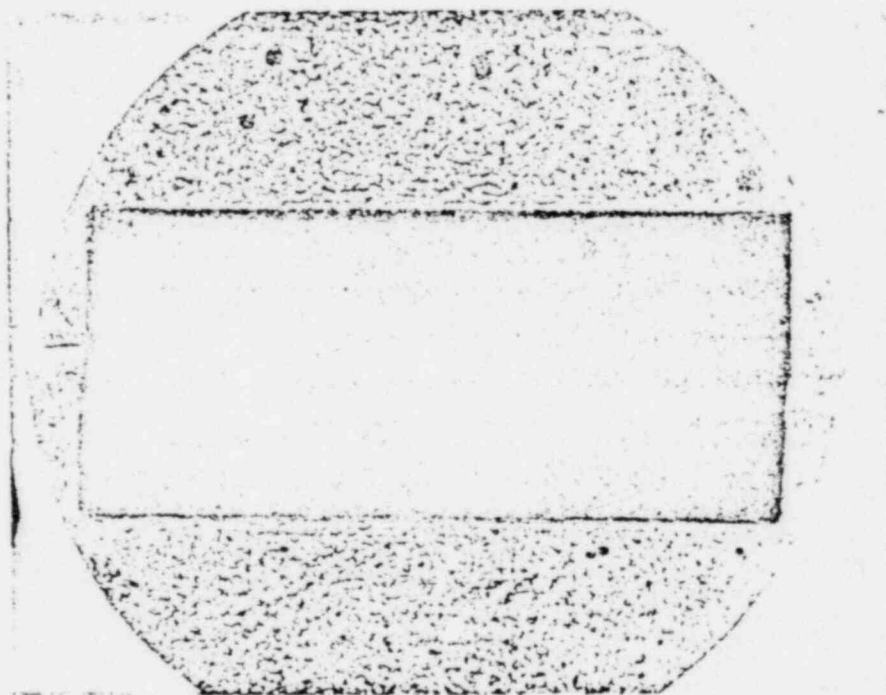
The fragments from each specimen were collected and placed in plastic specimen bags marked with the specimen designation. The bagged specimens were then placed in the lead slugs and returned to the SwRI "Hot Lab" for packing prior to shipment back to Wyle Labs. Both sets of fragments also contain bits of urethane foam from the sabot and bits of concrete from the target.

TABLE I
SPECIMEN DATA

Specimen:	A	B
Type:	Chemtrol CT-40-NS Type B	Chemtrol CT-40-NS Type B
Exposure:		
• Time, hr:	545.6	110
• Temperature, °F:	380 ± 20	380 ± 20
• Power, MW:	2	2
• Y radiation, rads:	4.1 x 10 ¹⁰	8.4 x 10 ⁹
• Isotope:	Zn ⁶⁵	Zn ⁶⁵
• B rate, mR/hr:	21	195
• Y rate, mR/hr:	18	15
Approximate Initial Weight, g:	12.4	10.7
Approximate Initial Dimensions, mm:	35 x 22 x 10	35 x 22 x 10
Weight as Tested, g:	10.00	9.42
Projectile Weight, g:	16.04	15.49
Muzzle Velocity, fps:	1450	1410



(a) Sample Specimen and Sabot Disassembled to Show Components: A - Specimen; B - Sabot Side; and C - Pusher Disk.



(b) Front View of Assembled Projectile, 4X.

FIGURE 1. SPECIMEN AND SABOT

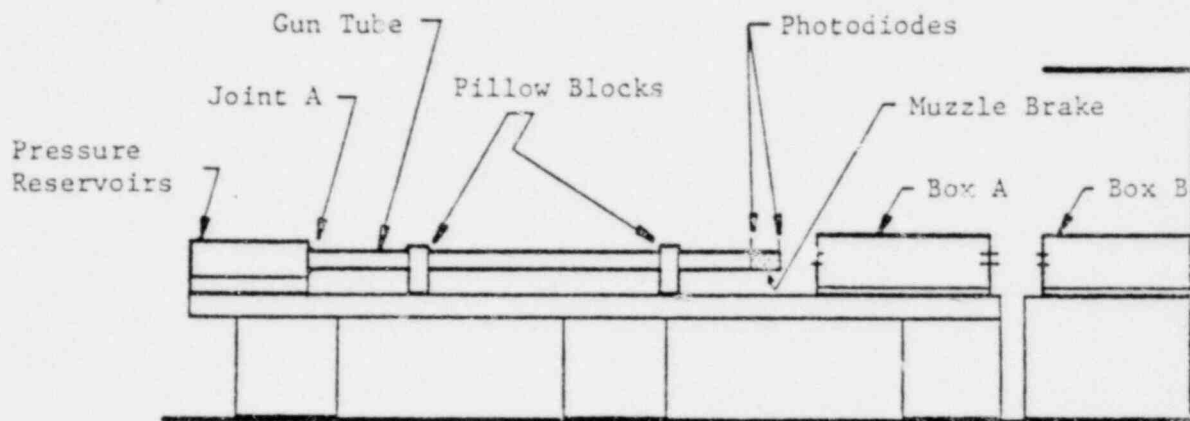
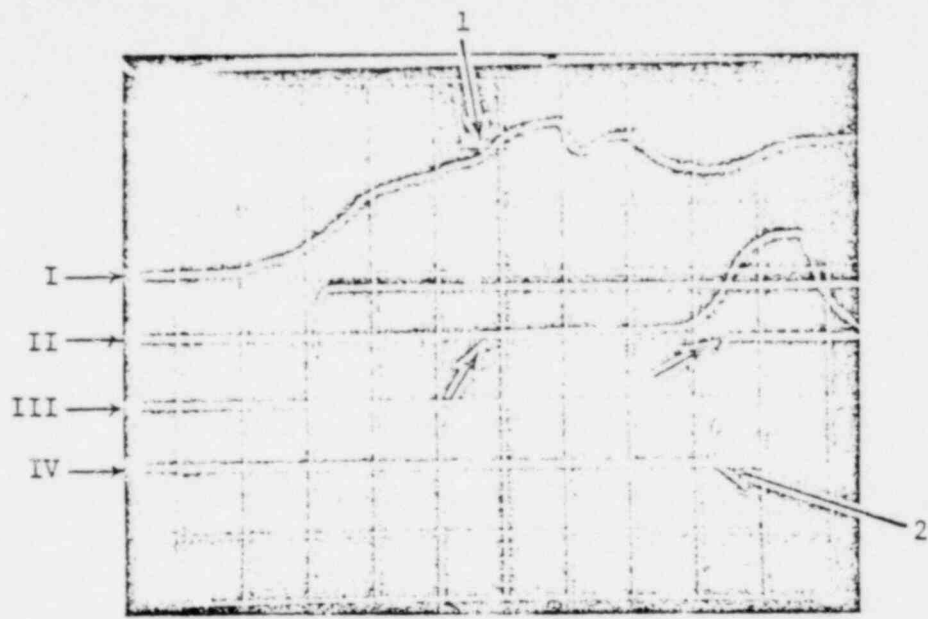
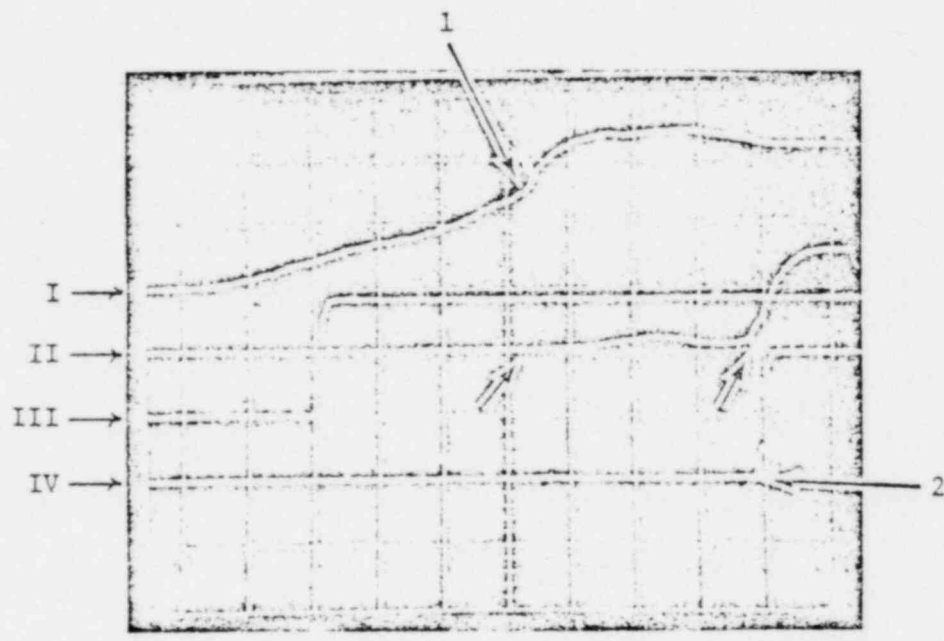


FIGURE 2. GUN AND CONTAINMENT ARRANGEMENT (SCHEMATIC)



(a) Specimen A



(b) Specimen B

FIGURE 3. OSCILLOSCOPE TRACES OF PHOTODIODE OUTPUTS

1 Division = 100 μ s

- I = First photodiode output
- II = Second Photodiode output
- III = Conditioned signal from first photodiode
- IV = Conditioned signal from second photodiode
- 1 = Projectile at first photodiode
- 2 = Projectile at second photodiode

APPENDIX

SPECIAL OPERATING PROCEDURE

PURPOSE:

Because of the hazards involved with radioactive materials, this operating procedure is provided to implement special procedures in support of Project 02-5845-118. This addendum is intended to supplement the SwRI Radiological Health and Safety Manual.

HAZARDS:

The tests will involve the firing of two active silicone polymer samples from an air gun at the Ballistics Range. The projectile will impact on a concrete target within a containment chamber. Methods of containment, protection, and monitoring will be covered in this procedure.

SPECIAL PROCEDURES:

The area to be used for testing will be divided and marked as a controlled area with ropes and radiation warning signs.

The projectile will be fired as received. However, should any sizing be necessary it will be done at the "Hot Lab."

The active projectile will be fired into a containment box through a small opening. Within the box, the projectile will impact the concrete target. Should any fragmentation of the projectile occur, it will be contained within the box. During removal of the fragments, the operator will be required to wear a respirator, surgical gloves, and a disposable lab coat. After the test, the fragments will be recovered for subsequent measurements in an area approved by the Radiation Safety Officer.

Personnel handling the projectile and fragments will be required to wear TDL badges and surgical gloves. Additionally, pocket dosimeters will be used to monitor "current" dosages.

All contaminated waste products, such as used gloves and concrete fragments, will be disposed of in plastic bags provided by the Radiation Safety Officer. These bags will then be disposed of by the "Hot Lab."

After completion of these tests, the area will be monitored carefully for contamination. If contamination is indicated, the area will be cleaned in accordance with the SwRI Radiological Health and Safety Manual.

QUALIFICATION TEST REPORT
ON
NEUTRON SHIELDING MATERIAL
FOR
CHEMTROL CORPORATION
HOUSTON, TEXAS

NEQ

Nuclear Environmental Qualification

Test Report

REPORT NO. 44925-1

WYLE JOB NO. 44925

CUSTOMER P. O. NO. 4087

PAGE 1 OF 33 PAGE REPORT

DATE September 10, 1980

SPECIFICATION (S) See Paragraph 5.0



1.0 CUSTOMER Chemtrol Corporation

ADDRESS 330 North Belt East, Houston, Texas 77060

2.0 TEST SPECIMEN Neutron Shielding Material Samples

3.0 MANUFACTURER Chemtrol Corporation

4.0 SUMMARY

Two samples of Neutron Shielding Material were subjected to a Qualification Test Program to verify functional integrity when subjected to the environmental tests specified herein.

A description of the Neutron Shielding Material (specimens) is presented in Paragraph 6.0.

STATE OF ALABAMA } ss. Alabama Professional Eng.
COUNTY OF MADISON } Reg. No. 7913

Robert A. Hall, being duly sworn,

deposes and says: The information contained in this report is the result of complete and carefully conducted tests and is to the best of his knowledge true and correct in all respects.

SUBSCRIBED and sworn to before me this 25th day of Sept, 19 80

Joyce Oliver
Notary Public in and for the County of Madison, State of Alabama

My Commission expires June 3, 19 84

TEST BY NEQ Department

PROJ. ENGINEER Joe Marshall

WYLE Q. A. Murvin Kimbrell
Murvin Kimbrell

WYLE LABORATORIES
SCIENTIFIC SERVICES AND SYSTEMS GROUP
HUNTSVILLE, ALABAMA

4.0 SUMMARY (CONTINUED)

The Qualification Test Program was performed in accordance with References 5.1 and 5.2. The Shielding Material showed no visual physical effects from the Steam Impingement Tests. Test results are presented in Section I of this report. The test results of the Borated Water Submersion Test are presented in Section III.

This Qualification Test Report contains three sections as listed below. The test program was conducted in the sequence indicated by the section number.

- o Section I Steam Impingement Test
- o Section II Impact Test
- o Section III Borated Water Submersion Test

5.0 REFERENCES

- 5.1 Wyle Quotation No. 543/1640-2/ES
- 5.2 Chemtrol Corporation Purchase Order No. 4087

6.0 TEST ITEM AND EQUIPMENT DESCRIPTION

6.1 Test Item Description

Sample A: CT-40-NS, gamma dose = 4.1×10^{10} rads--Sample A is in one piece approximately 1" x 3/4" x 3/8".

Sample B: CT-40-NS, gamma dose = 8.4×10^9 rads--Sample B is in two pieces approximately 1" x 3/4" x 3/8" and 3/4" x 3/4" x 3/8".

6.2 Test Equipment Description

The test equipment used in recording data is shown on Instrumentation Equipment Sheets located in the appendices of the appropriate sections of this report.

**STEAM IMPINGEMENT
TEST**

SECTION 1

STEAM IMPINGEMENT TEST

1.0 REQUIREMENTS

Each of the Neutron Shielding Materials will be subjected to a Steam Impingement Test.

The specimens will be exposed to a 600°F steam jet impinging directly on the free surface of the samples for 120 seconds.

2.0 PROCEDURES

A 1/4-inch fitting for an exit nozzle was located 6 inches from the Shielding Material sample surface. A 1/4-inch line from a high pressure/temperature steam source, with a valve for regulating the flow, furnished the steam supply to the exit nozzle. A bleed valve was provided upstream of the flow regulating valve to condition the steam in the line prior to starting the impingement. A thermocouple was installed in the steam line, approximately 18 inches upstream of the exit nozzle, for measuring the steam temperature. The Shielding Material sample was placed flat in a holding basket located under the exit nozzle, the bleed valve opened to condition the steam, the flow control valve opened and the 600°F steam allowed to impinge on the as-molded free surface for 120 seconds.

The specimen weights were recorded prior to and after impingement.

3.0 RESULTS

The test specimens were subjected to the requirements as specified in Paragraph 1.0 and as described in Paragraph 2.0. No visual degradation was evident. The samples were cracked and discolored to some degree when received. Small powder-like specks of material fell from the specimens whenever handled.

The weight of each specimen prior to and after test is presented in Appendix II, Table I.

Photographs are presented in Appendix II. Photographs 1-1 through 1-4 are pretest specimens. Photographs 1-5 and 1-6 show the test setup. Photographs 1-7 through 1-10 show the post-test specimens.

An Instrumentation Equipment Sheet listing the equipment used in this test is presented in Appendix III.

PAGE NO. 1-2

REPORT NO. 44925-1

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PAGE NO. 1-3

REPORT NO. 44925-1

APPENDIX I

TABLE

TABLE I

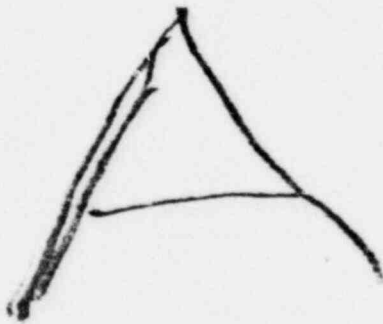
I	<u>Pretest Weight</u>
	Sample A = 10.741 ± 0.001 grams
	Sample B = 18.743 ± 0.001 grams
II	<u>Post-Test Weight</u>
	Sample A = 10.732 ± 0.001 grams
	Sample B = 18.701 ± 0.001 grams
III	<u>Weight Loss</u>
	Sample A = 0.009 ± 0.002 grams
	Sample B = 0.042 ± 0.002 grams

PAGE NO. 1-5

TEST REPORT NO. 44925-1

APPENDIX II

PHOTOGRAPHS



PHOTOGRAPH 1-i

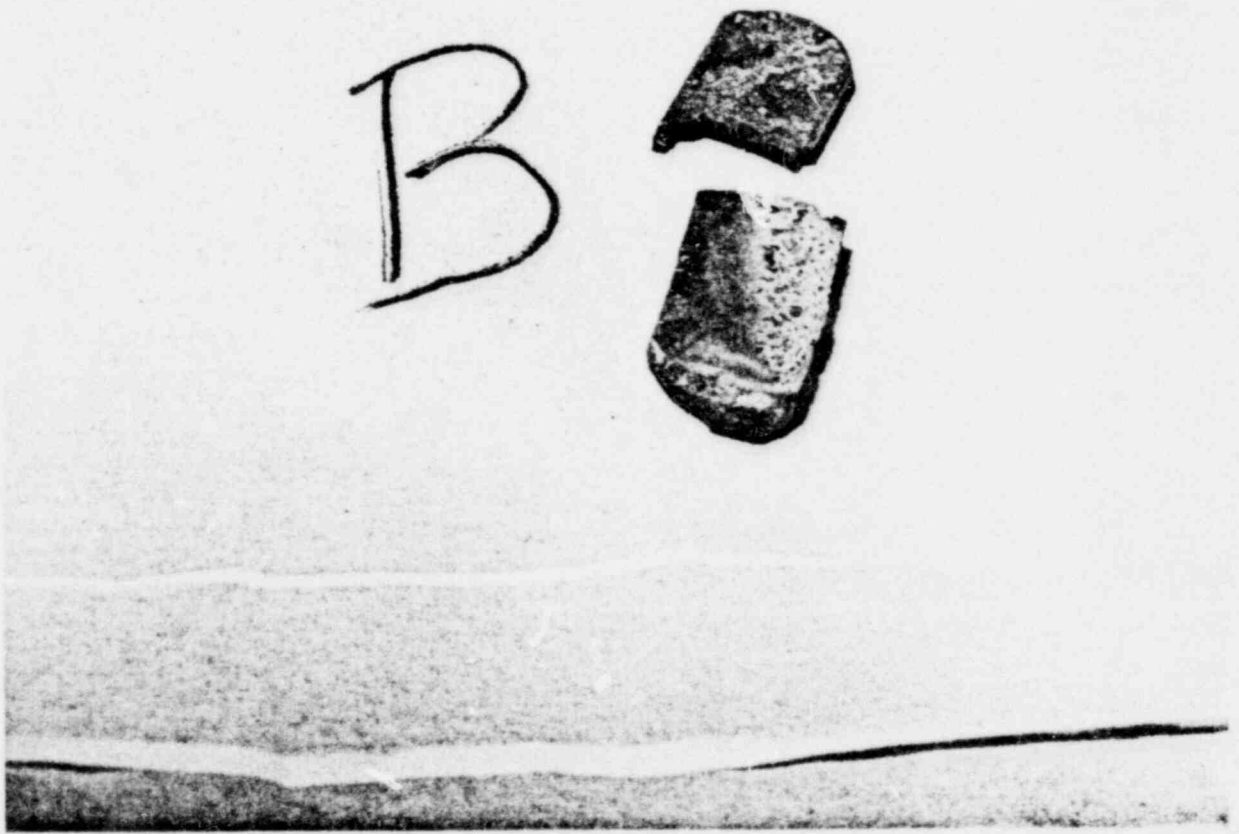
FR. 10% TO STEAM IMPINGEMENT



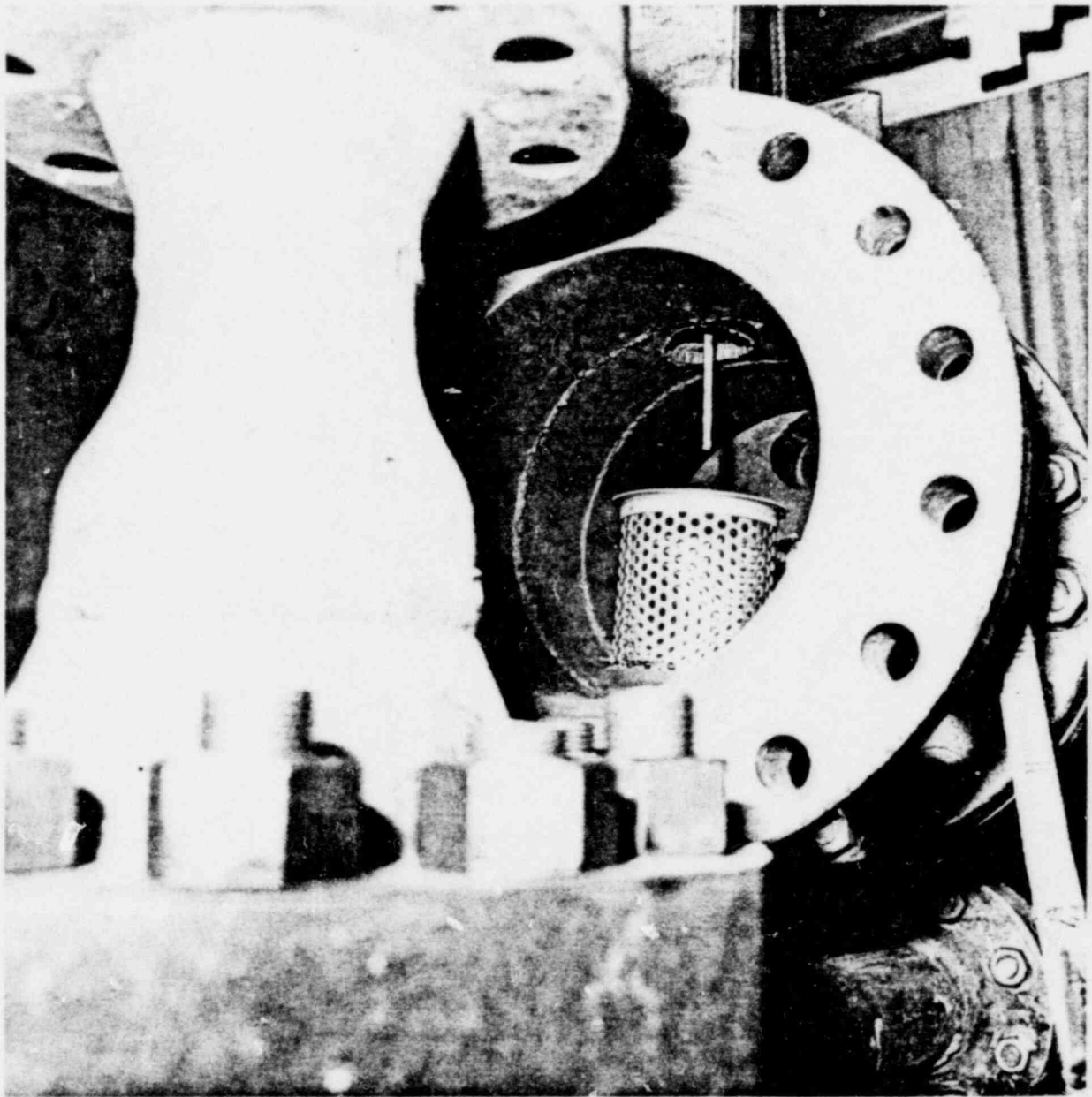
PHOTOGRAPH 1-2
PRIOR TO STEAM IMPINGEMENT



PHOTOGRAPH 1-3
PRIOR TO STEAM IMPINGEMENT

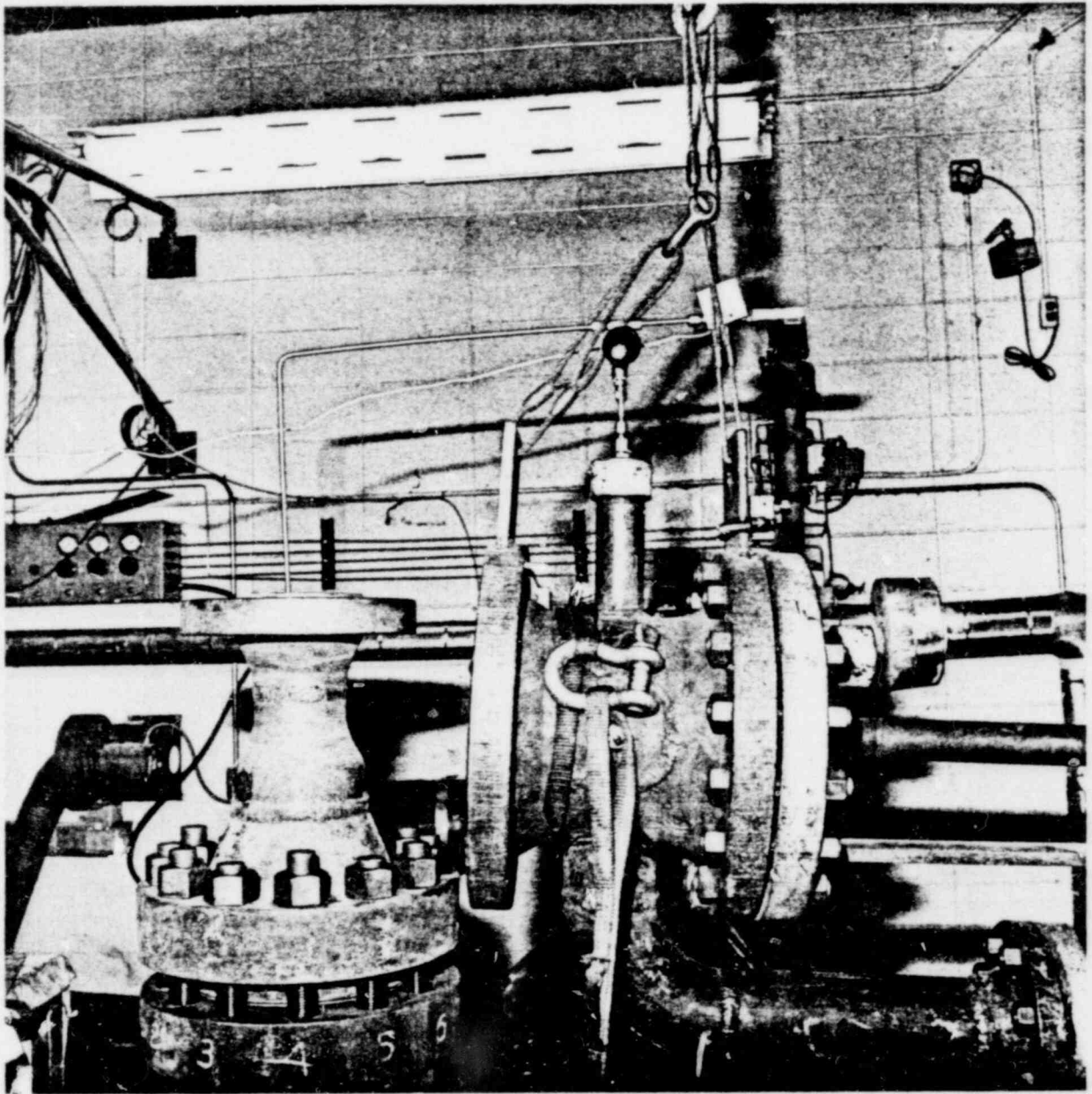


PHOTOGRAPH 1-4
PRIOR TO STEAM IMPINGEMENT



PHOTOGRAPH 1-5

TEST SETUP STEAM IMPINGEMENT



PHOTOGRAPH 1-6

TEST SETUP STEAM IMPINGEMENT



A



PHOTOGRAPH 1-7
POST-STEAM IMPINGEMENT

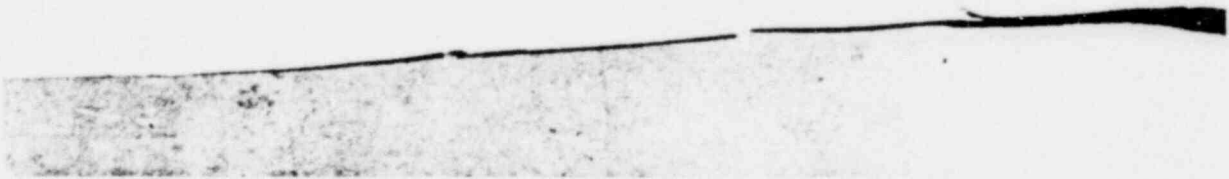


A

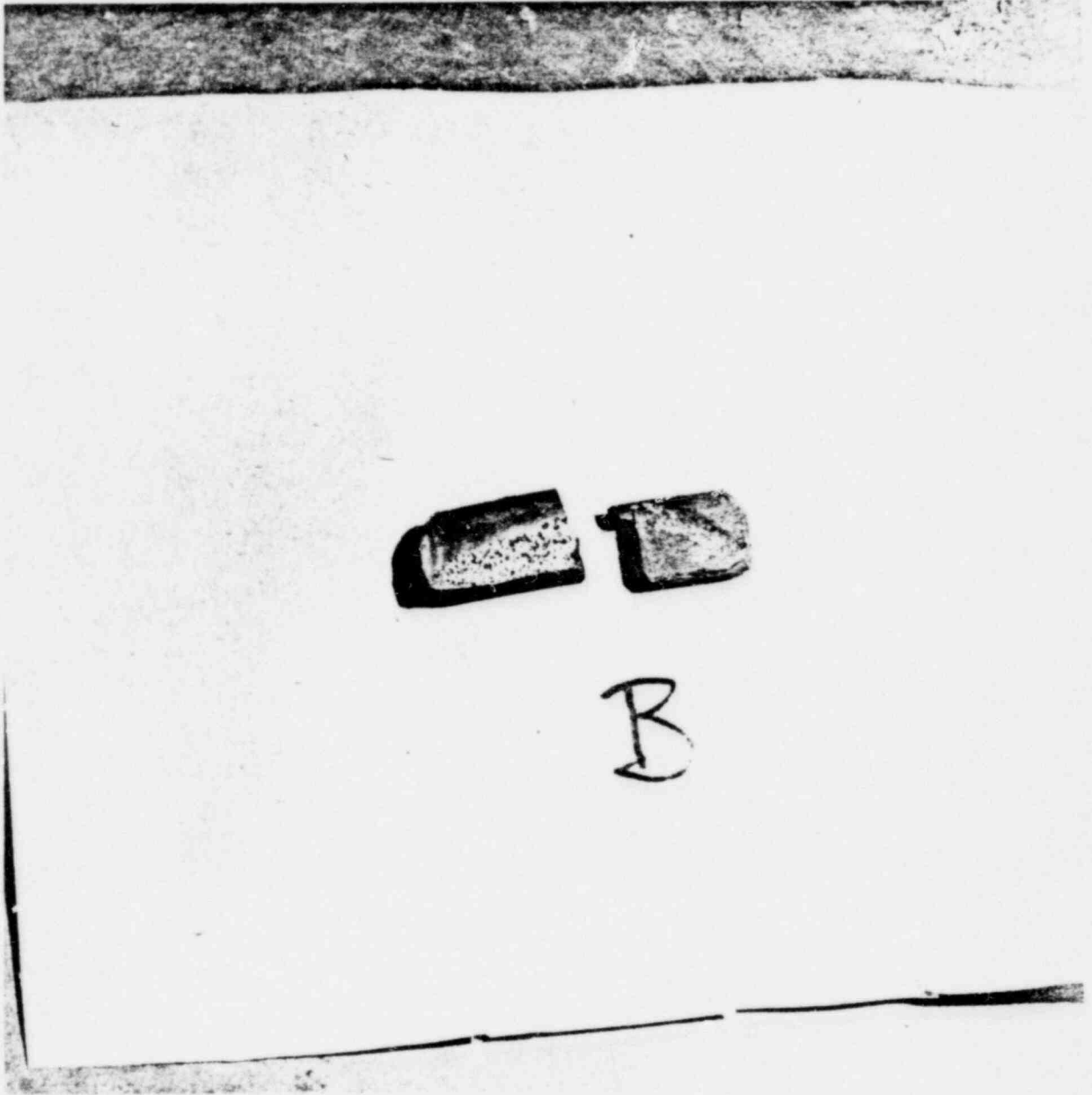
PHOTOGRAPH 1-8
POST-STEAM IMPINGEMENT



B



PHOTOGRAPH 1-9
POST-STEAM IMPINGEMENT



PHOTOGRAPH 1-10
POST-STEAM IMPINGEMENT

PAGE NO. 1-16

TEST REPORT NO. 44925-1

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PAGE NO. 1-17

TEST REPORT NO. 44925-1

APPENDIX III

INSTRUMENTATION EQUIPMENT SHEET

INSTRUMENTATION EQUIPMENT SHEET

Date 1.20.80 Job No. 44925 Test Area SRV BUILDING
 Technician G. J. ELAM Customer CHEMTRON Type Test EMPLOYMENT

No.	Instrument	Manufacturer	Model No.	Serial No.	Wyle or Gov't No.	Range	Accuracy	Calibration	
								On	Due
1	TEMP. RECORDER	MONITOR LABS	9300		11554	0-750	± 0.02%	9.19.79	3.19.80
2	MASTER WEIGHTS	CHRISTIAN BAKER	NA		81033	0-160 gm	CLASS S	10.24.78	10.24.80

Instrument Test Engineer J. M. Myhall Checked & Received By A. J. [Signature]

**IMPACT
TEST**

SECTION II

IMPACT TEST

1.0 REQUIREMENTS

Ship the Neutron Shielding Material to:

*Southern Research Institute
6220 Culebra Road
San Antonio, TX 78284

Attention: Mr. David K. Curtice

2.0 PROCEDURES

Upon receipt of *Southern Research Institute's Radioactive Material License, the specimens were properly prepared and shipped. *Southern Research complete the Impact Test and returned the specimens to Wyle Laboratories for completion of the test program.

3.0 RESULTS

The specimens were received as a powdered and granular substance.

*Southwest Research Institute

**BORATED WATER
SUBMERSION TEST**

SECTION III

BORATED WATER SUBMERSION TEST

1.0 REQUIREMENTS

The Borated Water Submersion Test shall consist of maintaining the specimens at 150°F for 30 days in a solution of 2000 to 2500 ppm of boron concentration. Sodium hydroxide shall be used to adjust the pH between 7.0 and 9.0. Gas samples shall be taken at the 24th and 30th day of exposure and a chromatograph analysis performed to determine the hydrogen, methane and ethane gas contents.

2.0 PROCEDURES

The samples of shielding material were placed in separate containers and filled with the specified borated water to approximately 75% of the container volume, the remaining 25% being available for gas evolution sampling. The borated water was mixed 2000 to 2500 ppm boron concentrate sodium hydroxide (NaOH) was added to the solution which produced a room temperature pH of 7.5. Foreign substances were observed within the solution. These substances were removed as requested by the Chemtrol Technical Representative. The containers were sealed and a gas sampling line, containing a hand valve and pressure gauge, was attached. The containers were placed in an environmental chamber and the 150°F temperature was maintained for 30 days.

Gas samples were taken on the 24th and 30th day of exposure. No pressure developed within the containers, therefore, the samples were removed by evacuating 250 cc sample bottles to 15 inches of mercury. The gas was then drawn into the sample bottles.

A gas chromatograph was performed to determine hydrogen, methane and ethane gas contents.

3.0 RESULTS

The test specimens were subjected to the requirements as specified in Paragraph 1.0 and as described in Paragraph 2.0. One equipment problem occurred. It is described in Notice of Anomaly No. 1.

The gas chromatograph was performed by Southern Research Institute, Birmingham, Alabama.

Notice of Anomaly No. 1 is presented in Appendix I.

3.0 RESULTS (CONTINUED)

The results of the gas chromatograph are shown in the certification letter presented in Appendix II.

Photographs 111-A and 111-B, showing the solution and samples, are presented in Appendix III.

An Instrumentation Equipment Sheet listing equipment used in this test sequence is presented in Appendix IV.

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APPENDIX I
NOTICE OF ANOMALY

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Test Report No. 44925-1
NOTICE OF ANOMALY

NOTICE NO. 1 P. O. NUMBER: 4087 WYLE JOB NO. 44925 PAGE NO. —
CONTRACT NUMBER: N/A REPORT NO. —
CATEGORY: SPECIMEN PROCEDURE TEST EQUIPMENT DATE: 7/7/80

TO: Chemtrol Corporation ATTN: Vince Cataldo
PART NAME: Neutron Shielding Material PART NO. N/A
TEST: Borated Water Submersion Test I. D. NO. N/A
SPECIFICATION: 543/1640-2/ES PARA. NO. N/A
NOTIFICATION MADE TO: Vince Cataldo DATE: July 7, 1980
NOTIFICATION MADE BY: J. Mayhall VIA: Telephone

REQUIREMENTS:

Samples shall be submerged in borated water in a pressure-tight container maintained at 150°F for 30 days.

At the end of the 30-day submersion test, the gas shall be analyzed for evolution of hydrogen and methane gases. The samples shall be evaluated for any physical change, i.e., dissolving, deformation, or weight change.

DESCRIPTION OF ANOMALY:

A test equipment problem occurred causing the temperature to decrease to ambient for approximately 29 hours.

DISPOSITION – COMMENTS – RECOMMENDATIONS:

At the customers request, the test duration will be extended 29 hours.

DISTRIBUTION:
Original: Dept.
(1) Copies: Customer
2 Copies: C. C.
2 Copies Project Office
1 Copy: Contracts
1 Copy: Operations Director

TEST WITNESS —
—
REPRESENTING —

ENGINEER J. Mayhall *Res*
QUALITY CONTROL J. W. Osborne
PROJECT MANAGER J. R. G. Hill

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APPENDIX II

GAS CHROMATOGRAPH RESULTS AND
CERTIFICATION LETTER

Southern Research Institute



2000 NINTH AVENUE SOUTH
BIRMINGHAM, ALABAMA 35255
TELEPHONE 205-323-6592

September 2, 1980

Wyle Laboratories
7800 Governors Drive West
Huntsville, Alabama 35807

Attention: Joe Mayhall

Reference: P. O. 4-5482-S

Gentlemen:

The results of the analyses of the four gas samples are given below:

Sample	Date received	Percent by volume					
		H ₂	O ₂	N ₂	CH ₄	C ₂ H ₆	C ₂ H ₄
A	7-30-80	<0.3	20	78	0.001	0.002	<0.001
B	7-30-80	<0.3	20	77	<0.001	<0.001	<0.001
A	8-13-80	<0.3	20	78	0.001	0.001	<0.001
B	8-13-80	<0.3	20	78	<0.001	<0.001	<0.001

The samples were analyzed on a Perkin-Elmer Sigma I gas chromatograph with a Molecular Sieve 5A column. A thermal conductivity detector was used to determine H₂, O₂, and N₂ and a flame ionization detector was used for the remainder of the gases. The detector responses to the gases were calibrated by analyses of standard gas mixtures under conditions identical to those used for the samples.

Yours very truly,

Approved by:

Lee Ann Wallace
Chemical Research Technician

Herbert C. Miller, Head
Analytical and Physical
Chemistry Division

Ruby H. James
Head, Environmental Analytical
Chemistry Section

SORI-EAS-80-641
NB No. 8921, pgs 123 and 124
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(3:20) cs

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CONTAINED THEREIN

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Report No.: Project 4587, Report 1
SORI-EAS-80-641

To: Wyle Laboratories
7800 Governors Drive West
Huntsville, Alabama 35807

Date: September 2, 1980

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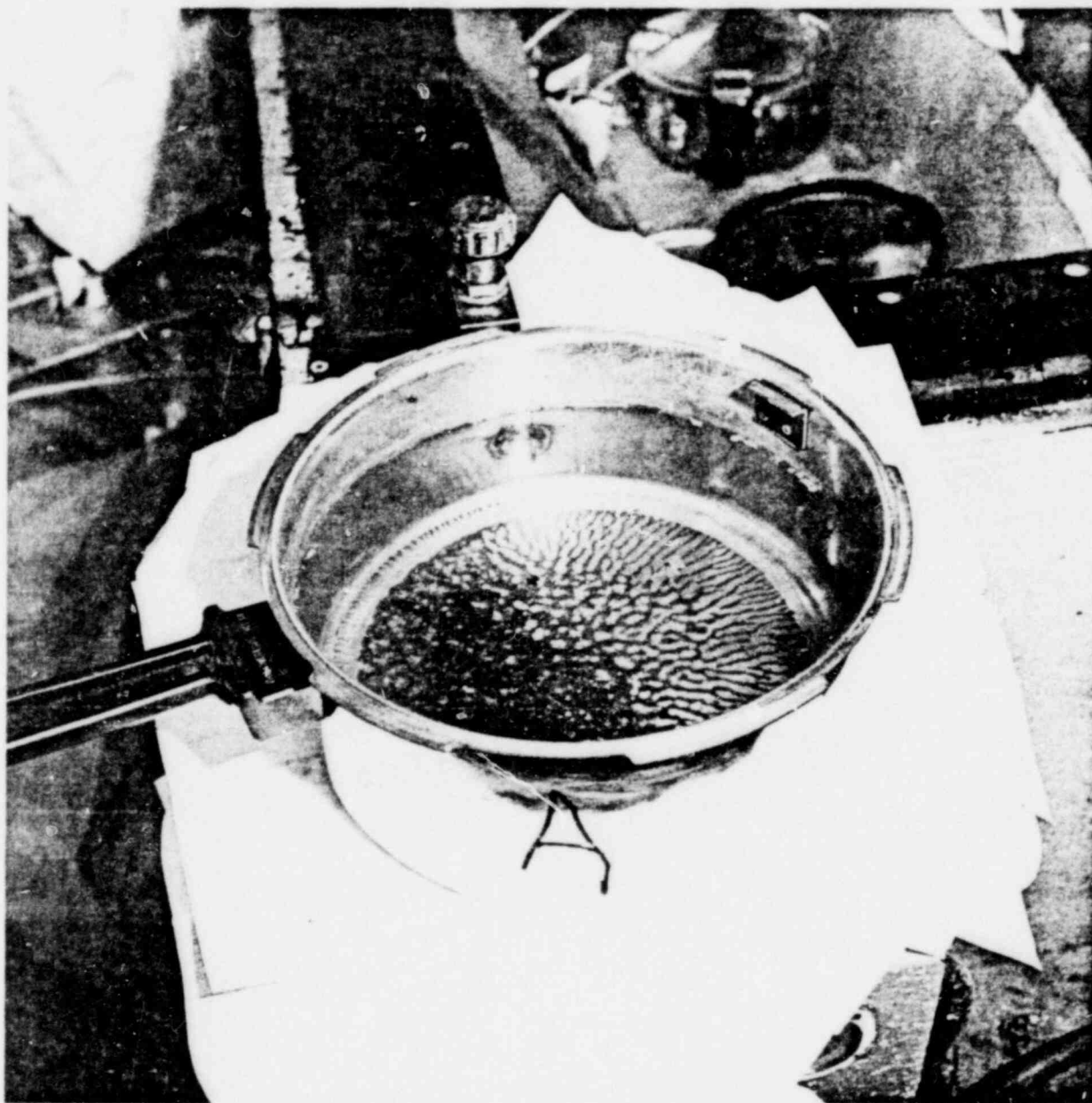
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APPENDIX III

PHOTOGRAPHS



PHOTOGRAPH III-1
SAMPLE A IN SOLUTION



PHOTOGRAPH 111-2
SAMPLE B IN SOLUTION

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APPENDIX IV
INSTRUMENTATION EQUIPMENT SHEET

INSTRUMENTATION EQUIPMENT SHEET

Date 6/30/80 Job No. 44425-00-3168 Test Area SRV
 Technician J. H. Kelly Customer Spentel Type Test High Temp

No.	Instrument	Manufacturer	Model No.	Serial No.	Wyle or Gov't No.	Range	Accuracy	Calibration	
								On	Due
1	Temp Recorder	Fluorwell	45	N/A	92815	0-200°F	±.5%	6/16/80	9/14/80
2	Recorder	R/I	639LLP	N/A	11292	0-1000°F	±.5%	5/18/80	11/18/80
3	Temp Controller	R/I	639 B	N/A	11297	0-1000°F	±.5%	5/18/80	11/18/80

Instrument Test Engineer B. Garcia

Checked & Received By J. H. Kelly