NUREG-0770

Glossary of Terms

Nuclear Power and Radiation

U.S. Nuclear Regulatory Commission

Office of Public Affairs Office of Inspection and Enforcement Office of Administration

Compilers: J. G. Hanchett, OPA, F. W. Hasselberg, OIE Editor: M. H. Singh, ADM



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Manuscript Completed: May 1981 Date Published: June 1981

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ABSTRACT

This "Glossary of Terms: Nuclear Power and Radiation" is a compilation of words and concepts commonly used in the nuclear power field defined to assist the news media and members of the public in understanding this often complex technology. The glossary was compiled by the NRC Office of Inspection and Enforcement and the Office of Public Affairs from a variety of internal and external sources. It is a part of the agency's response to a recommendation by the President's Commission on the Accident at Three Mile Island that government agencies, utility companies and the news media better prepare themselves to disseminate information to the public, in a form that is understandable.

CONTENTS

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37

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ABSTRACTINTRODUCTION	iii 1
DEFINITIONS	3
SIMPLIFIED SCHEMATIC OF BOILING WATER	
REACTOR (BWR) PLANT	47
SIMPLIFIED SCHEMATIC OF PRESSURIZED WATER	
REACTOR (PWR) PLANT	48
ELEMENTS AND ELEMENT ABBREVIATIONS	49
PREFIXES	50

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Introduction

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The President's Commission on the Accident at Three Mile Island (the Kemeny Commission) made several recommendations under the general heading of "The Public's Right to Information." These recommendations were mainly concerned with the responsibility held by government agencies, utility companies and the news media to do a better job of dissewinating information to the public.

Among the recommendations, the President's Commission stated that "Federal and state agencies, as well as the utility, should make adequate preparation for a systeratic public information program so that in time of a radiation-related eme gency, they can provide timely and accurate information to the news media and the public <u>in a form that is understandable</u>" [emphasis added].

The President's Commission also recommended that major media outlets and those other media, regardless of size, located near nuclear power plants, "hire and train specialists who have more than a passing familiarity with reactors and the language of radiation." Furthermore, the Commission added, reporters should discipline themselves "to place complex information in a context that is <u>understandable to the public</u> [emphasis added] and that allows members of the public to make decisions regarding their health and safety."

As part of the U.S. Nuclear Regulatory Commission response to those recommendations, the Office of Inspection and Enforcement and the Office of Public Affairs have developed this "Glossary of Terms: Nuclear Power and Radiation" as part of the course material for an introductory seminar on nuclear power plants and radiation to be presented to the news media. It is our first attempt to define in readily understandable language, a wide variety of terms and concepts commonly used in the disciplines related to nuclear power.

There is certainly no shortage of excellent scientific and technical dictionaries. However, most of them are compiled for scientists and engineers and are not sufficiently broad to cover the range of topics--nuclear physics, radiation, nuclear reactor components and operations, fuel cycle, and health physics, to name a few--involved in a nuclear power plant.

We have therefore developed our own glossary, borrowing liberally from a variety of internal and external sources. We have attempted to redefine many of the terms and concepts in a way that will be understandable to the news reporter and the layman. In some instances, we suspect that our simplification--and, perhaps, oversimplification--of certain terms may offend some of our technical colleagues. In other Č.

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• • • cases, the definitions may still be too technical. When the definitions include terms that are also defined elsewhere in the text, they are identified by underlining.

Our goal is to provide a useful start toward defining terms that are, at times, both complex and mystifying. The editors will welcome, from any source, comments and suggestions for additions, deletions, corrections and further clarification and simplification of the terms defined in this glossary.

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Definitions

absorber

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absorbers (like boron, hafnium, and cadmium) are used in <u>control rods</u> for <u>reactors</u>. Concrete and steel absorb <u>gamma rays</u> and neutrons in reactor <u>shields</u>. A thin sheet of paper or metal will absorb or weaken <u>alpha particles</u> and all except the most energetic <u>beta particles</u>. (See <u>control</u> rod; <u>shielding</u>.) The process by which the number of particles or

Any material that absorbs or lessens the

intensity of ionizing radiation. Neutron

absorption

access hatch (air lock)

activation

photons entering a body of matter is reduced or attenuated by interaction with the matter. (See neutron capture.)

An airtight door system that preserves the pressure integrity of a reactor <u>containment</u> building while allowing access to personnel and equipment.

The process of making a material <u>radioactive</u> by bombardment with <u>neutrons</u>, <u>protons</u>, or other nuclear radiation. (See induced radioactivity.)

activation products

activity

acute exposure

acute radiation sickness (syndrome)

air lock

air sampling

ALARA

s See induced radioactivity.

See radioactivity.

See exposure.

See radiation sickness (syndrome).

See access hatch.

The collection and analysis of samples of air to measure its <u>radioactivity</u> or to detect the presence of radioactive substances, particulate matter or chemical pollutants.

Acronym for "As Low as Reasonably Achievable," a basic concept of <u>radiation</u> protection that specifies that radioactive discharges from nuclear plants and radiation exposure to personnel be kept as far below regulation limits as feasible. The term was originally "As Low as Practicable." alpha particle

anion

atom

atomic energy

Atomic Energy

Commission (AEC)

A

A positively charged particle ejected spontaneously from the <u>nuclei</u> of some <u>radioactive</u> elements. It is identical to a helium <u>nucleus</u> that has a <u>mass</u> <u>number</u> of 4 and an electrostatic charge of +2. It has low-penetrating power and short range. The most energetic alpha particle will generally fail to penetrate the skin. Alphas are hazardous when an alpha-emitting <u>isotope</u> is introduced into the bedy.

Negatively charged ion. (See ionization.)

The smallest particle of an <u>element</u> that cannot be divided or broken up by chemical means. It consists of a central core called the <u>nucleus</u>, which contains <u>protons</u> and <u>neutrons</u>. <u>Electrons</u> revolve in orbits in the region surrounding the nucleus.

Energy released in <u>nuclear reactions</u>. Of particular interest is the energy released when a <u>neutron</u> initiates the breaking up or <u>fissioning</u> of an <u>atom</u>'s <u>nucleus</u> into smaller pieces (<u>fission</u>), or when two nuclei are joined together under millions of degrees of heat (<u>fusion</u>). It is more correctly called "<u>nuclear</u> energy."

Federal agency created in 1946 to manage the development, use and control of <u>nuclear energy</u> for military and civilian application. Abolished by the Energy Reorganization Act of 1974 and succeeded by the Energy Research and Development Administration (now part of the U. S. Department of Energy) and the U.S. Nuclear Regulatory Commission.

The number of positively charged protons in the nucleus of an atom.

atomic weight

atomic number

See mass number.

See absorption.

attenuation

auxiliary building

Building at a <u>nuclear power plant</u>, frequently located adjacent to the <u>reactor containment</u> building, that houses most of the reactor

auxiliary and safety systems, such as radioactive waste systems, chemical and volume control systems and emergency cooling water systems.

auxiliary feedwater

Backup feedwater supply used during nuclear plant startup and shutdown; also known as emergency feedwater. (See feedwater.) A

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background radiation The radiation in man's natural environment, including cosmic rays and radiation from the naturally radioactive elements, both outside, and inside the bodies of humans and animals. It is also called natural radiation. The usually quoted average individual exposure from background radiation is 125 millirem per year. beta particle A charged particle emitted from a nucleus during radioactive decay, with a mass equal to 1/1837 that of a proton. A negatively charged beta particle is identical to an electron. A positively charged beta particle is called a positron. Large amounts of beta radiation may cause skin burns, and beta emitters are harmful if they enter the body. Beta particles are easily stopped by a thin sheet of metal or plastic. binding energy The minimum energy required to separate a nucleus into its component neutrons and protons. bioassay The collection and analysis of human hair. tissue, nasal smcars, urine or fecal samples to determine the amount of radioactive material that might have been ingested by the body. biological halflife The time required for a biological system, such as that of a human, to eliminate by natural processes half the amount of a substance (such as a radioactive material) that has entered it. biological shield A mass of absorbing material placed around a reactor or radioactive source to reduce the radiation to a level safe for humans. body burden The amount of radioactive material present in the body of a human or an animal. boiling water reactor A reactor in which water, used as both coolant (BWR) and moderator, is allowed to boil in the core. The resulting steam can be used directly to drive a turbine and electrical generator. bone seeker A radioisotope that tends to accumulate in the bon s when it is introduced into the body. An example is strontium-90, which behaves chemically like calcium.

breeder

A reactor that produces more nuclear fuel than it consumes. A fertile material, such as uranium-238, when bombarded by neutrons, is transformed into a fissile material, such as plutonium-239, which can be used as fuel. (See fissile, fissionable and fertile material.)

Btu

A British thermal unit. The amount of heat required to change the t-mperature of one pound of water one degree Fahrenheit at sea level.

A boiling water reactor.

BWR

B

è

calibration

cask

L

cation

chain reaction

charged particle

chemical compound

chemical recombination

China syndrome

The check or correction of the accuracy of a measuring instrument to assure proper operational characteristics. (See counter.)

A heavily shielded container used to store and/or ship radioactive materials. Lead and steel are common materials used in the manufacture of casks.

A positively charged ion. (See ionization.)

A reaction that stimulates its own repetition. In a <u>fiscion</u> chain reaction, a fissionable <u>nucleus</u> absorbs a <u>neutron</u> and fissions, releasing additional neutrons. These in turn can be absorbed by other fissionable nuclei, releasing still more neutrons. A fission chain reaction is self-sustaining when the number of neutrons released in a given time equals or exceeds the number of neutrons lost by <u>absorption</u> in nonfissionable material or by escape from the system.

An ion. An elementary particle carrying a positive or negative electric charge.

See compound.

Following an ionization event, the positive and negatively charged ion pairs may or may not realign themselves to form the same chemical substance they formed before ionization. Thus, chemical recombination could change the chemical composition of the material bombarded by radiation.

The hypothetical result of a <u>power reactor</u> core met. accident in which molten fuel melts through the reactor <u>pressure vessel</u> and the bottom of the <u>containment</u> building and into the earth "all the way to China."

chronic exposure

cladding

See exposure.

The thin-walled metal tube that forms the outer jacket of a nuclear <u>fuel rod</u>. It prevents corrosion of the fuel by the <u>coolant</u> and the release of <u>fission products</u> into the coolant. Aluminum, stainless steel and zirconium alloys are common cladding materials. cleanup system

coastdown

compound

condensate

condenser

contamination

containment

A system used for continuously filtering and demineralizing the reactor coolant system to reduce contamination levels and minimize corrosion.

An action that permits the <u>reactor</u> power level to decrease gradually as the fuel in the <u>core</u> is depleted.

cold shutdown The term used to define a reactor coolant system at atmospheric pressure and at a temperature below 212°F following a reactor cooldown. (See control rod.)

> A chemical combination of two or more <u>elements</u> combined in a fixed and definite proportion by weight.

> Water that has been produced by the cooling of steam in a condenser.

A large heat exchanger designed to cool exhaust steam from a <u>turbine</u> below the boiling point so that it can be returned to the heat source as water. In a <u>pressurized water reactor</u>, the water is returned to the <u>steam generator</u>. In a <u>boiling</u> <u>water reactor</u>, it returns to the <u>reactor core</u>. The heat removed from the steam by the condenser is transferred to a circulating water system and is exhausted to the environment, either through a <u>cooling tower</u> or directly into a body of water. (See cooling tower.)

The deposition of unwanted <u>"adioactive</u> material on the surfaces of structures, areas, objects, or personnel.

The provision of a gastight shell or other enclosure around a <u>reactor</u> to confine <u>fission</u> <u>products</u> that otherwise might be released to the atmosphere in the event of an accident.

control rod A rod, plate or tube containing a material such as hafnium, boron, etc., used to control the power of a <u>nuclear reactor</u>. By absorbing <u>neutrons</u>, a control rod prevents the neutrons from causing further fission. (See <u>poison</u>.)

С

control	led	area	

control room
(building)

coolant

cooldown

cooling tower

С

A defined area in which the occupational exposure of personnel to <u>radiation</u> or <u>radioactive</u> material is under the supervision of an individual in charge of radiation protection.

The area in a nuclear power plant from which most of the plant power production and emergency safety equipment can be operated by remote control.

A substance circulated through a <u>nuclear reactor</u> to remove or transfer heat. The most commonly used coolant in the United States is water. Other coolants include <u>heavy water</u>, air, carbon dioxide, helium, liquid sodium and sodium-potassium alloy.

The gradual decrease in reactor fuel rod temperature caused by the removal of heat from the reactor coolant system.

A heat exchanger designed to aid in the cooling of water that was used to cool exhaust steam exiting the turbines of a power plant. Cooling towers transfer exhaust heat into the air instead of into a body of water.

The central portion of a <u>nuclear reactor</u> containing the <u>fuel elements</u>, <u>moderator</u>, neutron poisons and support structures.

See China syndrome

Penetrating <u>ionizing radiation</u>, both particulate and electromagnetic, originating in outer space. Secondary cosmic rays, formed by interactions in the earth's atmosphere, account for about 45 to 50 millirem of the 125 millirem <u>background</u> radiation that an average individual receives in a year.

A general designation applied to <u>radiation</u> <u>detection instruments</u> or <u>survey meters</u> that <u>detect</u> and measure radiation. The signal that announces an <u>ionization</u> event is called a count. (See Geiger-Mueller counter.)

The smallest mass of fissionable material that will support a self-sustaining chain reaction.

core

core melt accident

cosmic radiation

counter

critical mass

С

critical organ

criticali'y

crud

cumulative dose

curie (Ci) The body organ receiving a <u>radionuclide</u> or radiation <u>dose</u> that results in the greatest overall damage to the body.

A term used in <u>reactor</u> physics to describe the state when the number of neutrons released by fission is exactly balanced by the neutrons being absorbed (by the fuel and poisons) and escaping the reactor core. A reactor is said to be "critical" when it achieves a selfsustaining nuclear chain reaction.

A colloquial term for corrosion and wear products (rust particles, etc.) that become <u>radioactive</u> under a <u>radiation</u> flux. (See <u>induced radio-</u>activity.)

The total <u>dose</u> resulting from repeated exposures of <u>radiation</u> to the same region, or to the whole body, over a period of time.

The basic unit used to describe the intensity of radioactivity in a sample of material. The curie is equal to 37 billion disintegrations per second, which is approximately the rate of decay of 1 gram of radium. A curie is also a quantity of any radionuclide that decays at a rate of 37 billion disintegrations per second. Named for Marie and Pierre Curie, who discovered radium in 1898.

D

daughter products

decay heat

decay, radioactive

deconter nation

depleted uranium

design-basis accident

design-basis phenomena

detector

deuterium

<u>Isotopes</u> that are formed by the <u>radioactive decay</u> of some other isotope. In the case of <u>radium-226</u>, for example, there are 10 successive daughter products, ending in the <u>stable</u> isotope lead-206.

The heat produced by the <u>decay</u> of <u>radioactive</u> <u>fission products</u> after the <u>reactor</u> has been shut down. (See residual heat.)

The decrease in the amount of any <u>radioactive</u> material with the passage of time, due to the spontaneous emission from the atomic nuclei of either <u>alpha</u> or <u>beta</u> <u>particles</u>, often accompanied by gamma radiation. (See halflife; radioactive.)

The reduction or removal of contaminating radioactive material from a structure, area, object, or person. Decontamination may be accomplished by (1) treating the surface to remove or decret the <u>contamination</u>; (2) letting the material stan. o that the <u>radioactivity</u> is decreased as a result of natural decay; and (3) covering the contamination to shield or attenuate the radiation emitted.

Uranium having a percentage of uranium-235 smaller than the 0.7% found in <u>natural uranium</u>. It is obtained from <u>spent</u> (used) <u>fuel elements</u> or as by-product <u>tails</u>, or residues, from uranium isotope separation. (See <u>mill tailings</u>.)

A postulated accident that a nuclear facility must be designed and built to withstand without loss to the systems, structures and components necessary to assure public health and safety.

Earthquakes, tornadoes, hurricanes, floods, etc., that a nuclear facility must be designed and built to withstand without loss to the systems, structures, and components necessary to assure public health and safety. (See seismic Category I.)

A material or device that is sensitive to radiation and can produce a response signal suitable for measurement or analysis. A radiation detection instrument. (See counter.)

An <u>isotope</u> of hydrogen with one proton and one neutron in the nucleus. (See heavy water.) desteron

The nucleus of deuterium. It contains one proton and one neutron.

differential pressure (DP)

The difference in pressure between two points of a system, such as between the inlet and outlet of a pump.

disintegration

Doppler coefficient

dose

dose equivalent

dosimeter

dosimetry

dose rate

drywell

See decay, radioactive. See fuel temperature coefficient of reactivity. A quantity (total or accumulated) of ionizing radiation received. The term "dose" is often used in the sense of the exposure dose, expressed in roentgens, which is a measure of the total amount of ionization that the quantity of radiation

could produce in air. This should be distinguished from the absorbed dose, given in rads, that represents the energy absorbed from the radiation in a gram of any material. Furthermore, the biological dose, given in rem, is a measure of the biological damage to living tissue from the radiation exposure.

A term used to express the amount of effective radiation when modifying factors have been considered. The product of absorbed dose multiplied by a quality factor multiplied by a distribution factor. It is expressed numerically in rem.

A portable instrument for measuring and registering the total accumulated exposure to ionizing radiation. (See dosimetry.)

The theory and application of the principles and techniques involved in the measurement and recording of radiation doses. Its practical aspect is concerned with the use of various types of radiation instruments with which measurements are made. (See film badge; survey meter.)

The radiation dose delivered per unit of time. Measured, for example, in rem per hour.

The containment structure enclosing a boiling water reactor vessel and its recirculation system. The drywell provides both a pressure suppression system and a fission product barrier under accident conditions.

D

effective halflife

E

efficiency, plant

electrical generator

electromagnetic radiation

electron

element

emergency core
 cooling system
 [ECC(S)]

emergency feedwater

enrichment

excursion

The time required for the amount of a <u>radioactive</u> element deposited in a living organism to be diminished 50 percent as a result of the combined action of radioactive <u>decay</u> and biological elimination. (See biological halflife.)

The percentage of the total energy content of a power plant's fuel that is converted into electricity. The remaining energy is lost to the environment as heat.

An electromagnetic device that converts mechanical (rotational) energy into electrical energy. Most large electrical generators are driven by steam or water turbine systems.

A traveling wave motion resulting from changing electric or magnetic fields. Familiar electromagnetic radiations range from X-rays (and gamma rays) of short wavelength, through the ultraviolet, visible, and infrared regions, to radar and radio waves of relatively long wavelength. All electromagnetic radiations travel in a vacuum with the velocity of light. (See photon.)

An elementary particle with a unit negative charge and a mass 1/1837 that of the proton. Electrons surround the positively charged <u>nucleus</u> and determine the chemical properties of the <u>atom</u>. (See beta particle.)

One of the 103 known chemical substances that cannot be broken down further without changing its chemical properties. Some examples include hydrogen, nitrogen, gold, lead and uranium.

Reactor system components (pumps, valves, heat exchangers, tanks and piping) that are specifically designed to remove residual heat from the reactor <u>fuel rods</u> should the normal core cooling system (reactor coolant system) fail.

See auxiliary feedwater.

See isotopic enrichment.

A sudden, very rapid rise in the power level of a reactor caused by supercriticality. Excursions are usually quickly suppressed by the <u>negative</u> <u>temperature coefficient</u>, the <u>fuel temperature</u> <u>coefficient or the void coefficient</u> (depending on reactor design), and by rapid insertion of control rods.

exposure

The absorption of radiation or ingestion of a rediation of the second s

external radiation

Exposure to ionizing radiation when the radiation source is located outside the body.

extremities

The hands and forearms and, with restrictions, the head, feet, and ankles. (Permissible radiation exposures in these regions are generally greater than in the whole body because they contain less blood-forming material and have smaller volumes for energy absorption.)

E

Glossary of Terms - 16

fast fission

F

fast neutron

fast reactor

feedwater

fertile material

film badge

fissile material

fission

Fission of a heavy atom (such as uranium-238) when it absorbs a high-energy (fast) neutron. Most fissionable materials need thermal (slow) neutrons in order to fission.

A <u>neutron</u> with <u>kinetic energy</u> greater than its surroundings released during fission.

A reactor in which the fission chain reaction is sustained primarily by fast neutrons rather than by slow-moving neutrons. Fast reactors contain little or no moderator to slow down the neutrons from the speeds at which they are ejected from fissioning nuclei.

Water supplied to the reactor pressure vessel (in a <u>BWR</u>) or the <u>steam generator</u> (in a <u>PWR</u>) that removes heat from the reactor <u>fuel rods</u> by boiling and becoming steam. The steam becomes the driving force for the plant turbine generator.

A material, which is not itself fissile (fissionable by thermal neutrons), that can be converted into a fissile material by irradiation in a reactor. There are two basic fertile materials, uranium-238 and thorium-232. When these fertile materials capture neutrons, they are converted into fissile <u>plutonium-239</u> and uranium-233, respectively.

A pack of photographic film used for approximate measurement of radiation exposure for personnel monitoring purposes. The badge may contain two or three films of differing sensitivity, and it may contain a filter that shields part of the film from certain types of radiation.

Although sometimes used as a synonym for fissionable material, this term has acquired a more restricted meaning; namely, any material fissicnable by thermal (slow) neutrons. The three primarily fissile materials are uranium-233, uranium-235 and plutonium-239.

The splitting of a <u>nucleus</u> into at least two other nuclei and the release of a relatively large amount of energy. Two or three <u>neutrons</u> are usually released during this type of transformation.

17 - Glossary of Terms

Those fission products that exist in the gaseous fission gases state. Primarily the noble gases (krypton, xenon, radon, etc.). The nuclei (fission fragments) formed by the fission products fission of heavy elements, plus the nuclides formed by the fission fragments' radioactive decay. fissionable material Commonly used as a synonym for fissile material, the meaning of this term has been extended to include material that can be fissioned by fast neutrons, such as uranium-238. flux A term applied to the amount of some type of radiation crossing a certain area per unit time. The unit of flux is the number of particles, energy, etc., per square centimeter per second. A cluster of fuel rods (or plates). Also called fuel assembly a fuel element. Many fuel assemblies make up a reactor core. The series of steps involved in supplying fuel for fuel cycle nuclear power reactors. It can include mining, milling, isotopic enrichment, fabrication of fuel elements, use in a reactor, chemical reprocessing to recover the fissionable material remaining in the spent fuel, reenrichment of the fuel material, refabrication into new fuel elements, and waste disposal. See fuel assembly. fuel element The processing of reactor fuel to separate the fuel reprocessing unused fissionable material from waste material. A long, slender tube that holds fissionable fuel rod material (fuel) for nuclear reactor use. Fuel rods are assembled into bundles called fuel elements or fuel assemblies, which are loaded individually into the reactor core.

fuel temperature coefficient of reactivity The physical property of fuel <u>pellet</u> material (<u>uranium-238</u>) that causes the uranium to absorb more <u>neutrons</u> away from the <u>fission</u> process as fuel <u>pellet</u> temperature increases. This acts to stabilize <u>power reactor</u> operations. Also known as the Doppler coefficient.

F

fusion (thermonuclear reaction) A <u>nuclear reaction</u> characterized by joining together of light <u>nuclei</u> to form heavier nuclei, the energy for the reactions being provided by violent thermal agitation of particles at very high temperatures. If the colliding particles are properly chosen and the agitation is violent enough, there will be a release of energy from the reaction. The energy of the stars is derived from such reactions.

F

G

gap

gamma ray

(gamma

radiation)

The space inside a reactor <u>fuel rod</u> that exists between the fuel <u>pellet</u> and the fuel rod <u>cladding</u>.

High-energy, short wavelength electromagnetic radiation (a packet of energy) emitted from the nucleus. Gamma radiation frequently accompanies alpha and beta emissions and always accompanies fission. Gamma rays are very penetrating and are best stopped or shielded against by dense materials, such as lead or uranium. Gamma rays are similar to X-rays, but are usually more energetic.

A nuclear reactor in which a gas is the coolant.

Normally formless fluids that completely fill the

space and take the shape of their container.

gas-cooled reactor

gases

gaseous diffusion (plant)

Geiger-Mueller counter

graphite

A method of <u>isotopic separation</u> based on the fact that gas <u>atoms</u> or <u>molecules</u> with different masses will diffuse through a porous barrier (or membrane) at different rates. This method is used to separate <u>uranium-235</u> from uranium-238; it requires large gaseous diffusion plants and enormous amounts of electric power.

A <u>radiation</u> detection and measuring instrument. It consists of a gas-filled tube containing electrodes, between which there is an electrical voltage but no current flowing. When <u>ionizing</u> <u>radiation</u> passes through the tube, a short, intense pulse of current passes from the negative electrode to the positive electrode and is measured or counted. The number of pulses per second measures the intensity of radiation. It was named for Hans Geiger and W. Mueller who invented it in the 1920s. It is sometimes called simply a Geiger counter, or a G-M counter.

A form of carbon, similar to the lead used in pencils, used as a moderator in some nuclear reactors.

Glossary of Terms - 20

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halflife	The time in which half the <u>atoms</u> of a particular radioactive substance disintegrate to another nuclear form. Measured halflives vary from mill- iont's of a second to billions of years. Also called physical halflife.
halflife, biological	The time required for the body to eliminate half of the material takes in by natural biological means.
halflife, effective	The time required for a <u>radionuclide</u> contained in a biological system, such as a human or an animal, to reduce its accivity by half as a combined resul of radioactive <u>decay</u> and biological elimination.
half-thickness	The thickness of any given <u>absorber</u> that will reduce the intensity of a beam of <u>radiation</u> to one-half its initial value. (See <u>attenuation</u> ; <u>shielding</u> .)
head, reactor vessel	The removable top section of a reactor pressure vessel. It is bolted in place during power oper- ation and removed during refueling to permit access of fuel-handling equipment to the <u>core</u> .
health physics	The science concerned with recognition, evalu- ation and control of health hazards from <u>ionizing</u> <u>radiation</u> .
heat exchanger	Any device that transfers heat from one fluid (liquid or gas) to another fluid or to the environment.
heat sink	Anything that absorbs heat; usually part of the environment, such as the air, a river or outer space.
heatup	The rise in temperature of the reactor <u>fuel rods</u> resulting from an increase in the rate of fission in the <u>core</u> .
heavy water (D ₂ O)	Water containing significantly more than the natural proportions (one in 6500) of heavy hydro- gen (deuterium) atoms to ordinary hydrogen atoms. Heavy water is used as a <u>moderator</u> in some <u>reactors</u> because it slows down neutrons effec- tively and also has a low probability for <u>absorp-</u> tion of <u>neutrons</u> .

t

heavy-water-moderated reactor	A <u>reactor</u> that uses <u>heavy water</u> as its <u>moderator</u> . Heavy water is an excellent moderator and thus permits the use of inexpensive (unenriched) <u>uranium</u> as a fuel.
high radiation area	Any area in which a major portion of the body could receive a <u>radiation</u> <u>dose</u> of 100 <u>millirem</u> (0.1 <u>rem</u>) in one hour. These areas must be posted as "high radiation areas" and access into

hot

hot spot

The region in a radiation/contamination area in which the level of radiation/contamination is noticeably greater than in neighboring regions in

these areas is maintained under strict control.

A colloquial term meaning highly radioactive.

the area.

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Glossary of Terms - 22

I

induced Radioactivity that is created when stable substances are bombarded by ionizing radiation. radioactivity For example, the stable isotope cobalt-59 becomes the radiactive isotope cobalt-60 under neutron bombardment. internal radiation Nuclear radiation resulting from radioactive substances in the body. Some examples are iodine-131 found in the thyroid gland, and strontium-90 and plutonium-239 found in bone. ion An atom that has too many or too few electrons, causing it to be chemically active; an electron that is not associated (in orbit) with a nucleus. (See ionization.) ionization The process of adding one or more electrons to, or removing one or more electrons from, atoms or molecules, thereby creating ions. High temperatures, electrical discharges, or nuclear radiations can care ionization. ionization chamber An instrument that detects and measures ionizing radiation by measuring the electrical current that flows when radiation ionizes gas in a chamber, making the gas a conductor of electricity. (See counter.) ionizing radiation Any radiation capable of displacing electrons from atoms or molecules, thereby producing ions. Examples: alpha, beta, gamma, X-rays, neutrons and ultraviolet light. High doses of ionizing radiation may produce severe skin or tissue damage. irradiation Exposure to radiation. isotone One of several different nuclides having the same number of neutrons in their nuclei. isotope One of two or more atoms with the same number of protons, but different numbers of neutrons in their nuclei. Thus, carbon-12, carbon-13 and carbon-14 are isotopes of the element carbon, the numbers denoting the approximate atomic weights. Isotopes have very nearly the same chemical properties, but often different physical properties (for example, carbon-12 and -13 are stable, carbon-14 is radioactive.

isotope separation The process of separating isotopes from one another, or changing their relative abundances, as by <u>gaseous diffusion</u> or electromagnetic separation. Isotope separation is a step in the <u>isotopic enrichment</u> process.

isotopic enrichment

A process by which the relative abundances of the <u>isotopes</u> of a given <u>element</u> are altered, thus producing a form of the element that has been enriched in one particular isotope and depleted in its other isotopic forms.

K

kilo-A prefix that multiplies a basic unit by 1000.
Example: 1 kilometer = 1000 meters.kilovolt
(kV)The unit of electrical potential equal to 1000
volts.kinetic energyThe energy that a body possesses by virtue of its
mass and velocity; the energy of motion.

J

Glossary of Terms - 24

The dose of radiation expected to cause death within 30 days to 50 percent of those exposed. Generally accepted to range from 400 to 450 rem received over a short period of time [also known as lethal dose (LD)]. light water Ordinary water (H2O) as distinguished from heavy water (D_20) . light-water reactor A term used to designate reactors using ordinary water as coolant, including boiling water reactors (BWRs) and pressurized water reactors (PWRs), the most common types used in the United States. loop In a pressurized water reactor, the coolant flow path through piping from the reactor pressure

vessel to the steam generator, to the reactor coolant pump, and back to the reactor pressure vessel. Large PWRs may have as many as four separate loops.

low population zone An area of low population density often required around a nuclear installation. The number and (LPZ) density of residents is of concern in emergency planning so that certain protective measures (such as notification and instructions to residents) can be accomplished in a timely manner.

LD 50/30

mass-energy equation

The equation developed by Albert Einstein which is usually given as $E = mc^2$, slowing that, when the energy of a body changes by an amount E (no matter what form the energy takes), the mass, m, of the body will change by an amount equal to E/c^2 . The factor c^2 , the square of the speed of light in a vacuum, may be regarded as the conversion factor relating units of mass and energy. The equation predicted the possibility of releasing enormous amounts of energy by the conversion of mass to energy. It is also called the Einstein equation.

mass number The number of <u>Eucleons</u> (<u>neutrons</u> and <u>protons</u>) in the <u>nucleus</u> of an <u>atom</u>. Also known as the <u>atomic</u> weight of an atom.

A prefix that multiplies a basic unit by 1,000,000.

One million curies. (See curie.)

A prefix that divides a basic unit into one million parts.

A one-millionth part of a second.

microcurie

microsecond

mega-

micro-

megacurie

mill tailings

milli-

millirem

milliroentgen

moderator

thorium, <u>radium</u>, polonium and <u>radon</u>. A prefix that divides a basic unit by 1000.

A one-millionth part of a curie. (See curie.)

Naturally radioactive residue from the processing

of <u>urarium</u> ore into <u>yellowcake</u> in a mill. Although the milling process recovers about 93 percent of the uranium, the residues, or tailings, contain several radioactive elements, including uranium,

A one-thousandth part of a rem. (See rem.)

A one-thousandth part of a roentgen. (See roentgen.)

or A material, such as ordinary water, <u>heavy water</u>, or graphite, used in a <u>reactor</u> to slow down high-velocity <u>neutrons</u>, thus increasing the likelihood of <u>fission</u>.

moderator temperature
coefficient of
reactivityThe property of a reactor moderator to
slow down fewer neutrons as its temperature
increases. This acts to stabilize power reactor
operations.

M

molecule

monitoring

A group of <u>atoms</u> held together by chemical forces. A nolecule is the smallest unit of a compound that can exist by itself and retain all its chemical properties.

Periodic or continuous determination of the amount of ionizing radiation or radioactive contamination present in an occupied region, as a safety measure, for purposes of health protection. (See <u>radio-</u> logical survey.) nano-

nanocurie

natural radiation

natural uranium

A prefix that divides a basic unit by one billion. One billionth part of a <u>curie</u>. See <u>background radiation</u>. <u>Uranium</u> as found in nature. It contains 0.7 percent uranium-235, 99.3 percent uranium-238

and a trace of uranium-234.

captures a neutron.

negative temperature See moderator temperature coefficient.

neutron

An uncharged elementary particle with a mass slightly greater than that of the proton, and found in the <u>nucleus</u> of every <u>atom</u> heavier than hydrogen.

neutron capture

neutron chain reaction A process in which some of the neutrons released in one <u>fission</u> event cause other fissions to occur. There are three types of chain reactions:

The process in which an atomic nucleus absorbs or

- Nonsustaining chain reaction--An average of less than one <u>fission</u> is produced by the neutrons released by each previous fission (reactor subcriticality.)
- (2) Sustaining chain reaction--An average of exactly one <u>fission</u> is produced by the neutrons released by each previous fission (reactor criticality.)
- (3) Multiplying chain reaction--An average of more than one <u>fission</u> is produced by the neutrons released by previous fission (reactor <u>super-</u> cricitality.)
- neutron generation The release, <u>thermalization</u> and <u>absorption</u> of fission neutrons by a <u>fissile material</u> and the <u>fission</u> of that material producing a second generation of neutrons. In a typical reactor system, there are about 40,000 generations of neutrons every second.
- neutron leakage Neutrons that escape from the vicinity of the fissionable material in a reactor core. Neutrons that leak out of the fuel region are no longer

N

Glossary of Terms - 28

N

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NA:

* 🛒 _ *- available to cause fission and must be <u>absorbed</u> by <u>shielding</u> placed around the reactor <u>pressure</u> vessel for that purpose.

neutron, slow

neutron source

A <u>radioactive material</u> (decays by neutron emission) that can be inserted into a <u>reactor</u> to ensure that a sufficient quantity of neutrons is available to start a <u>chain reaction</u> and register on neutron detection equipment.

neutron, thermal

noble gas

A <u>neutron</u> that has (by collision with other particles) reached an energy state equal to that of its surroundings. (See <u>thermalization</u>.)

A gaseous chemical <u>element</u> that does not readily enter into chemical combination with other elements. An inert gas. (See <u>fission</u> gases.)

non-vital plant Systems at a nuclear facility that may or may not be necessary for the operation of the facility (i.e., power production), but that would have little or no effect on public health and safety should they fail. These systems are not

safety related.

See neutron, thermal.

nozzle

As used in PWRs and BWRs, the interface for fluid (inlet or outlet) between reactor plant components (pressure vessel, coolant pumps, steam generators, etc.) and their associated piping systems.

The energy liberated by a nuclear reaction (fission or fusion) or by radioactive decay.

nuclear disintegration See decay, radioactive.

nuclear energy

nuclear fission

auclear force

See fission.

A powerful short-ranged attractive force that holds together the particles inside an atomic nucleus.

nuclear fusion

See fusion.

29 - Glossary of Terms

An electrical generating facility using a nuclear power plant nuclear reactor as its power (heat) source. nuclear radiation See radiation, nuclear. nuclear reaction See reaction, nuclear. See reactor, nuclear. nuclear reactor Common name for a constituent particle of the nucleon atomic nucleus. At present, applied to protoes and neutrons but may include any other particles found to exist in the nucleus. The small, central, positively charged region of nucleus (or atomic an atom that carries essentially all the mass. nucleus); Except for the nucleus of ordinary (light) nuclei (plural) hydrogen, which has a single proton, all atomic nuclei contain both protons and neutrons. The number of protons determines the total positive charge, or atomic number; this is the same for all the atomic nuclei of a given chemical element. The total number of neutrons and protons is called

nuclide

A general term referring to all known isotopes, both stable (279) and unstable (about 5000), of the chemical elements.

the mass number. (See isotope.)

N

operating basis earthquake

oralloy

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994 N 19

r.,]

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J,

(%) *

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200 10 An earthquake that could be expected to affect the plant site, but for which the plant power production equipment is designed to remain functional without undue risk to public health and safety. (See design-basis phenomenon.)

Uranium enriched in the isotope uranium-235. This material is an excellent fission fuel and is capable of sustaining a chain reaction. parent

(pom)

parts per million

photodosimetry

photon

pico-

pig

pile

picocurie

A <u>radionuclide</u> that upon radioactive <u>decay</u> or disintegration yields a specific nuclide (the daughter).

Parts (molecules) of a substance contained in a million parts of air (or water) by volume.

pellet, fuel As used in <u>PWRs</u> and <u>BWRs</u>, a pellet is a small cylinder approximately 3/8-inch in diameter and 5/8-inch in length consisting of <u>uranium</u> fuel in a ceramic form--uranium dioxide, UO₂. Typical fuel pellet <u>enrichments</u> range from 2 to 3.5 percent uranium-235.

periodic table An arrangement of chemical <u>elements</u> in order of increasing <u>atomic number</u>. Elements of similar properties are placed one under the other, yielding groups or families of elements. Within each group, there is a variation of chemical and physical properties, but in general there is a similarity of chemical behavior within each group.

personnel monitoring The determination of the degree of radioactive contamination on individuals using survey meters, or the determination of <u>radiation</u> dosage received by means of dosimetry devices.

The determination of the cumulative dose of ionizing radiation by use of photographic film.

A quantum (or packet) of energy emitted in the form of <u>electromagnetic radiation</u>. <u>Gamma rays</u> and X-rays are examples of photons.

A prefix that divides a basic unit by one trillion.

One trillionth part of a curie.

A container (usually lead) used to ship or store <u>radioactive</u> materials. The thick walls protect the person handling the container from <u>radiation</u>. Large containers are commonly called casks.

A <u>nuclear reactor</u>; called a pile because the earliest reactors were "piles" of graphite and uranium blocks. P

Glossary of Terms - 32

plutonium (Pu)

P

pocket dosimeter

poison

pool reactor

positron

power reactor

pressure vessel

pressurized water reactor (PWR)

pressurizer

A heavy, radioactive, manmade metallic <u>element</u> with <u>atomic number</u> 94. Its most important <u>isotope</u> is <u>fissile</u> plutonium-239, which is produced by <u>neutron</u> irradiation of <u>uranium-238</u>.

A small ionization detection instrument that indicates radiation exposure directly. An auxiliary charging device is usually necessary.

In reactor physics, a material other than fissionable material in the vicinity of the reactor core that will absorb neutrons. The addition of poisons, such as control rods or boron, into the reactor is said to be an addition of negative reactivity.

A reactor in which the <u>fuel elements</u> are suspended in a pool of water that serves as the <u>reflector</u>, <u>moderator</u> and <u>coolant</u>. Popularly called a "swimming pool reactor," it is used for research and training, not for electrical generation.

Particle equal in mass, but opposite in charge, to the electron; a positive electron.

A reactor designed to produce heat for electric generation, as distinguished from reactors used for research, for producing radiation or fissionable materials, or for reactor component testing.

A strong-walled container housing the core of most types of power reactors; it usually also contains the moderator, neutron reflector, thermal shield and control rods.

A power reactor in which heat is transferred from the core to a heat exchanger by hightemperature water kept under high pressure in the primiry system. Steam is generated in a secondary circuit. Many reactors producing electric power are pressurized water reactors.

A tank or vessel that acts as a head tank (or surge volume) to control the pressure in a pressurized water reactor.

primary system

See reactor coolant syster.

p

propertional counter

An instrument in which an electronic detection system receives pulses that are proportional to the number of ions formed in a gas-filled tube by ionizing radiation.

proton

An elementary nuclear particle with a positive electric charge located in the <u>nucleus</u> of an atom. (See <u>atomic number</u>.)

PWR

A pressurized water reactor.

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quality factor

The factor by which the absorbed <u>dose</u> is to be multiplied to obtain a quantity that expresses, on a common scale for all <u>ionizing radiations</u>, the biological damage to exposed persons. It is used because some types of radiation, such as <u>alpha particles</u>, are wore biologically damaging than other types

quantum theory

The concept that energy is radiated intermittently in units of definite magnitude called quanta, and absorbed in a like manner. (See photon.)

and the second

R

rad

radiac

Acronym for radiation absorbed <u>dose</u>. The basic unit of absorbed dose of radiation. A dose of one rad means the absorption of 100 ergs (a small but measurable amount of energy) per gram of absorbing material.

An acronym derived from "radioactivity detection indication and computation." a generic term applying to radiological instruments or equipment.

Particles (alpha, beta, neutrons) or photons (gamma) emitted from the nucleus of an unstable (radioactive) atom as a result of radioactive decay.

radiation area

radiation, nuclear

Any accessible area in which the level of radiation is such that a major portion of ar individual's body could receive in any one hour a dose in excess of 5 millirem, or in any five consecutive days a dose in excess of 100 millirem.

A device that detects and records the charac-

teristics of ionizing radiation. (See counter.)

radiation detection instrument

radiation shielding

radiation monitoring See monitoring.

Reduction of <u>radiation</u> by interposing a shield of absorbing material between any radioactive source and a person, work area or radiationsensitive device.

The complex of symptoms characterizing the disease radiation sickness known as radiation injury, resulting from exces-(syndrome) sive exposure of the whole body (or large part) to ionizing radiation. The earliest of these symptoms are nausea, fatigue, vomiting, and diarrhea, which may be followed b, loss of hair (epilation), hemorrhage, inflamation of the mouth and throat, and general loss of energy. In severe cases, where the radiation exposure has been relatively large, death may occur within two to four weeks. Those who survive 6 weeks after the receipt of a single large dose of radiation may generally be expected to recover.

radiation source Usually a manmade sealed source of radiation used in teletherapy, <u>radiography</u>, as a power source for batteries, or in various types of R

industrial gauges. Machines such as accelerators and radioisotope generators and natural radionuclides may be considered sources.

radiation standards

Exposure standards, permissible concentrations, rules for safe handling, regulations for transportation, regulations for industrial control of radiation and control of <u>radioactive</u> material by legislative means.

An officially prescribed symbol (a magenta trefoil) on a yellow background that must be

doses of radiation could be received.

displayed where certain quantities of radioactive materials are present or where certain

radiation syndrome See radiation sickness (syndrome).

radiation warning symbol

radioactive

radioactive

contamination

radioactive isotope

radioactive series

radioaccive waste

radioactivity

radiography

Exhibiting <u>radioactivity</u> or pertaining to radioactivity.

Deposition of radioactive material in any place where it may harm persons or equipment.

A radiois tope.

A succession of <u>nuclides</u>, each of which transforms by radioactive <u>disintegration</u> into the next until a <u>stable</u> nuclide results. The first member is called the <u>parent</u>, the intermediate members are called <u>daughters</u>, and the final stable member is called the end product.

See waste, radioactive.

The spontaneous emission of <u>radiation</u>, generally alpha or beta particles, often accompanied by gamma rays, from the <u>nucleus</u> of an unstable isotope.

The making of shadow images on photographic film by the action of ionizing radiation.

radioisotope An unstable <u>isotope</u> of an <u>element</u> that <u>decays</u> or disintegrates spontaneously, emitting <u>radiation</u>. Approximately 5000 natural and artificial radioisotopes have been identified. radiological survey

The evaluation of the <u>radiation</u> hazards accompanying the production, use, or existence of <u>radioactive materials</u> under a specific set of conditions. Such evaluation customarily includes a physical survey of the disposition of materials and equipment, measurements or estimates of the levels of radiation that may be involved, and a sufficient knowledge of processes affecting these materials to predict hazards resulting from expected or possible changes in materials or equipment.

radiology

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radionuclide

radiosensitivity

radium (Ra)

radon (Rn)

reaction

reactivity

reactor coolant system

reactor, nuclear

The relative susceptibility of cells, tissues, organs, organisms, or other substances to the injurious action of radiation.

That branch of medicine dealing with the diag-

nostic and therapeutic applications of radiant energy, including X-rays and radioisotopes.

A radioisctope.

A radioactive metallic element with atomic number 88. As found in nature, the most common isotope has a mass number of 225. It occurs in minute quantities associated with uranium in pitchblend, carnotite and other minerals.

A <u>radioactive element</u> that is one of the heaviest gases known. Its <u>atomic number</u> is 86, and its mass number is 222. It is a daughter of radium.

Any process involving a chemical or nuclear change.

A term expressing the departure of a <u>reactor</u> system from <u>criticality</u>. A positive reactivity addition indicates a move toward <u>supercriticality</u> (power increase). A negative reactivity addition indicates a move toward <u>subcriticality</u> (power decrease).

The cooling system used to remove energy from the reactor core and transfer that energy either directly or indirectly to the steam turbine.

A device in which nuclear fission may be sustained and controlled in a self-supporting nuclear reaction. The varieties are many, but all incorporate certain features, including fissionable material or fuel, a moderating material (unless

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Glossary of Terms - 38

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recycling

reflector

rem

reprocessing

restricted area

roentgen (r)

roentgen equivalent man (or mammal) the reactor is operated on <u>fast neutrons</u>), a reflector to conserve escaping neutrons, provisions for removal of heat, measuring and controlling instruments, and protective devices.

The reuse of fissionable material after it has been recovered by chemical processing from spent o. depleted reactor fuel, reenriched and then refabricated into new fuel elements.

A layer of material immediately surrounding a reactor <u>core</u> that scatters back (or reflects) into the core many <u>neutrons</u> that would otherwise escape. The returned neutrons can then cause more <u>fissions</u> and improve the neutron economy of the reactor. Common reflector materials are graphite, beryllium, water and natural uranium.

Acronym of roentgen equivalent man. The unit of dose of any ionizing radiation that produces the same biological effect as a unit of absorbed dose of ordinary X-rays. (See quality factor.)

See recycling.

Any area to which access is controlled for the protection of individuals from exposure to radiation and radioactive materials.

A unit of exposure to <u>ionizing radiation</u>. It is that amount of <u>gamma</u> or <u>X-rays</u> required to produce <u>ions</u> carrying 1 electrostatic unit of electrical charge in 1 cubic centimeter of dry air under standard conditions. Named after Wilhelm Roentgen, German scientist who discovered X-rays in 1895.

See rem.

safeguards

safe shutdown earthquake

scintillation

counter

scram

detector or

The protection of <u>special nuclear material</u> (SNM) to prevent theft, loss or sabotage. (See <u>special</u> nuclear material.)

A design-basis earthquake. (See <u>design-basis</u> phenomenon.)

safety injection The rapid insertion of a chemically soluble neutron <u>poison</u> (such as boric acid) into the reacter coolant system to ensure reactor shut-<u>down</u>. (See <u>shutdown</u>.)

safety related The managerial controls, administrative documents, operating procedures, systems, structures and components that have been designed to mitigate the consequences of postulated accidents that could cause undue risk to public health and safety.

safety rod See control rod; scram.

scattered radiation Radiation that, during its passage through a substance, has been changed in direction. It may also have been modified by a decrease in energy. It is one form of secondary radiation.

> The combination of phosphor, photomultiplier tube, and associated electronic circuits for counting light emissions produced in the phosphor by ionizing radiation. (See counter.)

Sudden shutting down of a nuclear reactor, usually by rapid insertion of control ods, either automatically or manually by the reactor operator.

secondary radiation Radiation originating as the result of absorption of other radiation in matter. It may be either electromagnetic or particulate in nature.

secondary system The <u>steam generator</u> tubes, <u>steam turbine</u>, <u>condenser</u> and associated pipes, pumps and heaters used to convert the heat energy of the <u>reactor coolant</u> <u>system</u> into mechanical energy for electrical generation. Most commonly used in reference to <u>pres</u>-<u>surized water reactors</u>.

seismic Category I A term used to define structures, systems and components that are designed and built to withstand the maximum potential (earthquake) stresses for the particular region that a nuclear plant is sited.

S

Glossary of Terms - 40

S

S

shielding	Any material or obstruction that absorbs <u>radiation</u> and thus tends to protect personnel or materials from the effects of <u>ionizing radiation</u> .
shutdown	A decrease in the rate of <u>fission</u> (and heat pro- duction) in a reactor (usually by the insertion of <u>control rods</u> into the <u>core</u>). See <u>subcrit</u> - <u>icality</u> .)
somatic effects of radiation	Effects of radiation limited to the exposed indiv- idual, as distinguished from genetic effects, which may also affect subsequent unexposed generations.
special nuclear material	includes <u>plutonium</u> , <u>uranium</u> -233, or uranium <u>enriched</u> in the <u>isotopes</u> uranium-233 or uranium-235.
spent (čepleted) fuel	Nuclear reactor fuel that has been used to the extent that it can no longer effectively sustain a chain reaction.
spent fuel pool	An underwater storage and cooling facility for $\frac{\text{fuel elements}}{\text{reactor}}$ that have been removed from a $\frac{\text{reactor}}{\text{reactor}}$.
stable isotope	An isotope that does not undergo radioactive decay.
startup	An increase in the rate of <u>fission</u> (and heat pro- duction) in a reactor usually by the removal of control rods from the <u>core</u>). (See <u>supercriti</u> - <u>cality</u> .)
stay time	The period during which personnel may remain in a restricted area before accumulating some permis-sible dose.
steam generator	The <u>heat exchanger</u> used in some <u>reactor</u> designs to transfer heat from the primary (<u>reactor coolant</u>) system to the secondary (steam) system. This design permits heat exchange with little or no contamination of the secondary system equipment.
subcriticality	The condition of a <u>nuclear reactor</u> system when the rate of production of <u>fission neutrons</u> is lower than the rate of production in the previous generation due to increased <u>neutron leakage</u> and <u>poisons</u> .
subcritical mass	An amount of fissionable material insufficient in

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An amount of <u>fissionable material</u> insufficient in quantity or of improper geometry to sustain a fission chain reaction. supercriticality

The condition for increasing the level of operation of a reactor. The rate of fission neutron production exceeds all neutron losses, and the overall neutron population increases. (See critical mass; criticality.)

supercritical reactor

superheating

A reactor in which the power level is increasing.

The heating of a vapor, particularly steam, to a temperature much higher than the boiling point at the existing pressure. This is done in some power plants to improve efficiency and to reduce water damage to the turbine.

survey

survey meter

within an area of interest; (2) locate regions of higher-than-average intensity; i.e., <u>hot spots</u>. (See <u>personnel monitoring</u>.) Any portable radiation detection instrument

A study to (1) find the radiation or contamin-

ation level of specific objects or locations

Any portable radiation detection instrument especially adapted for inspecting an area to establish the existence and amount of radioactive material present. (See counter.) S

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tailings, tails	See mill tailings.
tenth thickness	The thickness of a given material that will decrease the amount (or <u>dose</u>) of <u>radiation</u> to one-tenth of the amount incident upon it. Two-tenth thicknesses will reduce the dose received by a factor of 10 x 10; i.e., 100, and so on. (See <u>shielding</u> .)
terrestrial radiation	The portion of <u>natural radiation</u> (background) that is emitted by naturally occurring <u>radioactive</u> materials in the earth.
thermal breeder reactor	A breeder reactor in which the fission chain reaction is sustained by thermal neutrons. (See neutron, thermal.)
thermalization	The process undergone by high-energy (fast) neutrons as they lose energy by collision. (See <u>neutron, thermal</u> .)
thermal neutron	See neutron, thermal.
thermal reactor	A reactor in which the <u>fission chain reaction</u> is sustained primarily by thermal neutrons. Most current reactors are thermal reactors.
thermal shield	A layer or layers of high-density material located within a reactor <u>pressure vessel</u> or between the vessel and the <u>biological shield</u> to reduce radia- tion heating in the vessel and the biological shield.
thermonuclear	An adjective referring to the process in which

wery high temperatures are used to bring about the fusion of light nuclei, such as those of the hydrogen isotopes, deuterium and tritium, with the accompanying liberation of energy. (See fusion.)

> A change in the reactor coolant system temperature and/or pressure due to a change in power output of the reactor. Transients can be caused by adding or removing neutron poisons, by increasing or decreasing the electrical load on the turbine generator, or by accident conditions.

trip, reactor

transient

See scram.

43 - Glossary of Terms

tritium

A radioactive isotope of hydrogen (one proton, two <u>neutrons</u>). Because it is chemically identical to natural hydrogen, tritium can easily be taken into the body by any ingestion path. <u>Decays</u> by <u>beta</u> emission. Its radioactive <u>halflife</u> is about 12-1/2 years.

turbine A rotary engine made with a series of curved vanes on a rotating shaft. Usually turned by water or steam. Turbines are considered to be the most economical means to turn large electrical generators.

turbine generator (TG)

A steam (or water) <u>turbine</u> directly connected to an electrical generator. The two devices are often referred to as one unit. T

ultraviolet	Electromagnetic radiation of a wavelength between
	the shortest visible violet and low-energy X-rays.
unrestricted area	The area outside the owner-controlled portion of a nuclear facility (usually the site boundary).
unstable isotope	A radioisotope.
uranium	A radioactive element with the atomic number 92,
(0)	and as found in natural ores, an atomic weight of approximately 238. The two principal natural
	isotopes are uranium-235 (0.7 percent of natural
	percent of natural uranium), which is fissionable
	by fast neutrons and is fertile. Natural uranium
	and an and a minute amount of aranitar 254.
uranium enrichment	See isotopic enrichment.
uranium millings	See mill tailings.
(tails)	

V

U V U

vapor	The gaseous form of substances that are normally in liquid or solid form.
vessel	See pressure vessel.
vital plant systems	See safety related.
void	An area of lower density in a moderating system (such as steam bubbles in water) that allows more <u>neutron leakage</u> than does the more dense material around it. (See <u>moderator</u> ; <u>void coeffi-</u> <u>cient</u> ; <u>neutron leakage</u> .)
void coefficient of reactivity	Property of a reactor plant moderating system where, as temperature increases in the system, <u>neutron leakage</u> increases due to an increase in the number and size of <u>voids</u> (steam bubbles) in the moderator.

waste, radioactive

Solid, liquid and gaseous materials from nuclear operations that are radioactive or become radioactive and for which there is no further use. Wastes are generally classified as high level (having radioactivity concentrations of hundreds of thousands of <u>curies</u> per gallon or cubic foot), low level (in the range of 1 <u>microcurie</u> per gallon or cubic foot), or intermediate level (between these extremes).

whole-body counter A device used to identify and measure the <u>radiation</u> in the body (<u>body burden</u>) of human beings and animals; it uses heavy <u>shielding</u> to keep out <u>background radiation</u> and ultrasensitive radiation detectors and electronic counting equipment.

whole-body exposure An exposure of the body to radiation, in which the entire body rather than an isolated part is irradiated. Where a <u>radioisotope</u> is uniformly distributed throughout the body tissues, rather than being concentrated in certain parts, the irradiation can be considered as a whole-body exposure.

wipe sample A sample made for the purpose of determining the presence of removable <u>radioactive contamination</u> on a surface. It is done by wiping, with slight pressure, a piece of soft filter paper over a representative type of surface area. It is also known as a "swipe sample." x

X-rays

Penetrating <u>electromagnetic radiation</u> (<u>photon</u>) having a wavelength that is much shorter than that of visible light. These rays are usually produced by excitation of the <u>electron</u> field around certain nuclei. In <u>nuclear reactions</u>, it is customary to refer to photons originating in the <u>nucleus</u> as <u>gamma rays</u>, and to those originating in the electron field of the atom as X-rays. These rays are sometimes called <u>roentgen</u> rays after their discoverer, W. K. Roentgen.

Y

yellowcake

A product of the <u>uranium</u> milling process, yellowcake is a solid uranium-oxygen compound (U_3O_8) that takes its name from its color and texture Yellowcake is the feed material used for fuel enrichment and fuel pellet fabrication.



SIMPLIFIED SCHEMATIC OF BOILING WATER REACTOR (BWR) PLANT



SIMPLIFIED SCHEMATIC OF PRESSURIZED WATER REACTOR (PWR) PLANT

Name	Symbol	Name	Symbol
actinium	Ac	mercury	Чg
aluminum	Al	molybdenum	Mo
amaricum	Am	neodymium	Nd
antimony	Sh	neon	Ne
anchaoly	As	neptunium	Np
arsenic	At	nickel	Ni
astatine	Ra	niobium	Nb
barian	Bk	nitrogen	N
berkellun	Be	pobelium	No
beryllium	Bi	osmium	Os
bismuch	B	oxvgen	0
boron	Br	palladium	Pd
bromine	Cd	phosphorus	Р
cadmium	Ca	platinum	Pt
calcium	Cf	plutonium	Pu
californium	C	polonium	Po
carbon	Ce	potassium	K
cerium	Ce	praseodymium	Pr
cesium	C1	promethium	Pm
chlorine	Cr	protactinium	Pa
chromium	Co	radium	Ra
cobalt	Cu	radon	Rn
copper	Cu	rhenium	Re
curium	Dar	rhodium	Rh
dysprosium	Dy	rubidium	Rb
einsteinium	ES E-	ruthenium	Ru
erbium	Er	samarium	Sm
europium	Eu	samatium	Sc
fermium	Fm	colonium	Se
fluorine	F	scilicon	Si
francium	Fr	silver	Ag
gadolinium	Gđ	silvei	Na
gallium	Ga	stroptium	Sr
germanium	Ge	sulfur	S
gold	Au	tentalum	Та
hafnium	Ht	tachatium	Tc
helium	He	tellurium	Te
holmium	Но	terium	Th
hydrogen	н	terbium	T1
indium	In	thallium	Th
iodine	I	thorium	Tm
iridium	Ir	Chullum	Sn
iron	Fe	tin	Ti
krypton	Kr	titanium	ŵ
lanthanum	La	tungsten	
lawrencium	Lr	uranium	v
lead	Pb	vanadium	Ye
lithium	Li	xenoli	Vh
lutetium	Lu	ytterbium	v
magnesium	Mg	yttrium	70
manganese	Mn	zinc	211
mendelevium	Md	zirconium	61

ELEMENTS AND ELEMENT ABBREVIATIONS

d	deci	$(= 10^{-1})$	da	deka	(= 10)
с	centi	$(= 10^{-2})$	h	kecto	$(= 10^2)$
m	milli	$(= 10^{-3})$	k	kilo	$(= 10^3)$
μ	micro	$(= 10^{-6})$	M	mega	$(= 10^6)$
n	nanc	$(= 10^{-9})$	G	giga	$(= 10^9)$
р	pico	$(= 10^{-12})$	Т	tera	(= 1012)
f	femto	$(= 10^{-15})$			(10)
a	atto	$(= 10^{-18})$			

PREFIXES

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