

1 UNITED STATES OF AMERICA
2 NUCLEAR REGULATORY COMMISSION

3
4 ADVISORY COMMITTEE ON REACTOR SAFEGUARDS
5 SUBCOMMITTEE ON CLASS 9 ACCIDENTS

6 Room 1046
7 1717 H Street, N.W.
8 Washington, D.C.

9 Thursday, August 28, 1980

10 The meeting of the Advisory Committee on Reactor
11 Safeguards Subcommittee on Class 9 Accidents was convened,
12 pursuant to notice, at 8:30 a.m.

13 MEMBERS PRESENT:

14 W. KERR, presiding
15 H. ETHERINGTON
16 S. LAWROSKI
17 J. C. MARK
18 C. P. SISS

19 ACRS CONSULTANTS PRESENT:

20 J. LEE
21 R. SEALE
22 S. SIEGEL
23 G. SCHOTT
24 D. GREGORY
25 R. STREHLOW

DESIGNATED FEDERAL EMPLOYEE:

G. R. QUITTSCHREIBER

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P R O C E E D I N G S

1 MR. KERR: The meeting will come to order.

2
3 This is a meeting of the Advisory Committee on
4 Reactor Safeguards, specifically the Subcommittee on Class 9
5 accidents.

6 My name is William Kerr. I am Subcommittee
7 chairman. Other ACRS members here today are Mr.
8 Etherington, Mr. Lawrowski, Mr. Mark and Mr. Siess. As
9 consultants we have Messrs. Lee, Seale and Siegel. As
10 invited experts -- I am not sure I know what the difference
11 is -- we have Messrs. Shott, Strehlow and Gregory.

12 The purpose of the meeting is to discuss hydrogen
13 generation and control. This represents an interesting
14 departure from the usual in that we spend much of our time
15 discussing the more complicated atoms, and I guess we have
16 learned our lesson and we are going to start now looking
17 more at simple atoms, and perhaps we can solve that problem.

18 We have as a continuing consideration the
19 evolution and treatment of accidents more serious than the
20 design basis accident, which I guess we refer to as the
21 Class 9 accident. As a part of that consideration, we had
22 originally planned today's meeting to look at some of the
23 information available and some of the considerations
24 associated with the production, distribution and possible
25 consequences of hydrogen generation during accidents.

1 After the meeting was planned, we also received
2 some specific questions from Commissioner Gilinsky having to
3 do with the Sequoyah case specifically. I think you perhaps
4 have those questions, but I might repeat them because part
5 of our effort today will be to get information that may help
6 the Committee to deal with those questions.

7 They are specifically: Does the Committee believe
8 additional hydrogen control measures are necessary for ice
9 condenser containments; and second, is the Committee
10 reasonably persuaded of the effectiveness of distributed
11 igniters in ice condenser containments? Can such igniters
12 be counted on to keep pressure increases caused by hydrogen
13 burns at suitably low values, which I would define as design
14 pressures, during accident sequences involving TMI-like
15 quantities of hydrogen?

16 We have received, those of us on the Subcommittee,
17 the consultants and, I presume members of the Committee,
18 eventually, rather voluminous information hydrogen, some of
19 which deals specifically with the Sequoyah situation, some
20 of which deals with the broader questions. This information
21 varies from a so-called compendium on hydrogen which has
22 been put together by Sandia, among others, to a set of SECY
23 papers prepared by the staff for consideration by the
24 Commission.

25 I would suggest that as we explore this question,

1 both the Subcommittee and the Committee, we give continuing
2 attention to an effort to understand the approach that the
3 staff plans to take in eventually trying to deal with this
4 question on a general basis.

5 I assume that will finally occur in the rulemaking
6 hearings and in the process of rulemaking, and I would guess
7 that the philosophy is now being developed. I have had some
8 problems myself in understanding what approach the staff
9 plans to take eventually.

10 I could get the impression in reading the SECY
11 papers that the present approach of the staff and, perhaps,
12 even the Commission, based on reading of transcripts of
13 meetings, might be one which will make the generation of
14 hydrogen due to x percent metal/water reaction -- where x is
15 still undefined in y minutes, and where the number of
16 minutes is also not completely specified -- on a
17 non-mechanistic but conservative basis as a design basis
18 accident and would then plan to require that a system exists
19 which will keep the containment pressure produced thereby to
20 an acceptable level where the acceptable level is still not
21 thoroughly defined.

22 It might be some multiple of design pressure
23 between perhaps 1 and 3. It is not yet clear to me on what
24 basis one would require the operation of the mitigation
25 system, whether one would talk about a single failure

1 criterion or some probabilistic basis. That, insofar as I
2 have seen, has not been made very clear, and indeed, I may
3 be misinterpreting the approach. I have been trying to
4 understand it.

5 Comments made by some of the staff and even some
6 members of the Commission in the discussions I read might
7 lead me to conclude that an event producing an amount of
8 hydrogen equal to that produced at TMI, given in some cases
9 as 200 to 300 kilograms of hydrogen, and in others to 30 to
10 50 percent of metal-water reaction, is likely with high
11 probability to occur at each operating reactor during its
12 useful life, and that those changes in equipment, procedure,
13 training and staffing which have already been introduced,
14 and some of which soon will be, have had little or not
15 effect on the probability of occurrence of a
16 hydrogen-producing accident.

17 Again, I am not sure I am interpreting the
18 discussion correctly. But that could be an impression one
19 would get.

20 Also, it seems to me, in spite of all the comments
21 that have been made since TMI-2 about early attendance to
22 concentrate on low probability accidents and to ignore
23 higher probability but presumably lower consequence
24 accidents, I don't see a clear approach yet to an attempt to
25 specify a consideration of scenarios or sets of accidents

1 which are classified according to probability.

2 In fact, it seems to me that at least in the SECY
3 paper, that the staff estimates of quantitative hydrogen
4 production appear to be based on consideration of production
5 of hydrogen by a large LOCA followed by failure of the ECCS.
6 Again, I am giving my effort to interpret what I have seen,
7 and I may be misinterpreting it.

8 What I am trying to suggest to you and to me and
9 to each of us is that we try to explore and understand the
10 ultimate approach that is going to be taken in deciding how
11 one deals not just with the Sequoyah problem but with the
12 hydrogen problem in general, and as it fits into
13 consideration of Class 9 accidents.

14 Now, in the Sequoyah situation itself, as far as I
15 can tell, a good bit of attention has been given to what at
16 least TVA and the staff interpret to be a set of scenarios
17 somewhat similar to TMI-2. It would be interesting to me,
18 although I am not sure how feasible this is, to see
19 attention given to scenarios chosen on the basis of
20 probabilities, with a goal of trying to attach some sort of
21 probability to the generation of some quantity of hydrogen.

22 It seems to me, and I am not sure this is feasible
23 at all, that one can postulate sets of scenarios that will
24 lead to generation of almost any amount of hydrogen up to
25 that that would be produced by a metal/water reaction with

1 all the zirconium and all of the steel in the system, if one
2 wants to carry things that far.

3 So, it would appear to me that at some point one
4 almost must deal with probabilities. Again, I recognize
5 that this may be early in the game and that this may be an
6 ultimate intention.

7 It also seems to me that at some point we need to
8 try to deal with whether the changes that have been
9 suggested and in many cases mandated by TMI have decreased
10 the probability of hydrogen generation, and how much?

11 For example, has the probability, once these are
12 in place, been decreased by a factor of 10 or 100 compared
13 to what it was before TMI-2? Is it the same for all kinds
14 of reactors, or does one need to give specific
15 consideration? And is there some probability below which
16 one does not worry about hydrogen generation?

17 It is also not clear to me that whether in the
18 treatments we are seeing, we are taking a conservative
19 approach or the staff plans to take a conservative approach
20 or whether a best estimate approach is the one that is
21 appropriate. Does the staff have an ultimate goal? And if
22 it does, is this goal understood by licensees who have to
23 deal with this problem?

24 That perhaps is enough of a set of comments from
25 me, and I simply indicate these are some of the questions

1 that have occurred to me as I have tried to read the
2 information. I do think that the information made available
3 to the Subcommittee represents quite a lot of work, and I
4 find it quite useful, and I look forward to learning further
5 as we go through the day.

6 You will note that we have scheduled a fairly full
7 day's work. I have a plane schedule to leave at 6:15 this
8 evening and I hope to be able to catch it, which means we
9 probably will be finished by about 5 o'clock, if anybody has
10 a tight schedule.

11 MR. MARK: Mr. Chairman.

12 MR. KERR: Yes, sir.

13 MR. MARK: One thing about your description I
14 thought gives the picture very beautifully that bothers me
15 is that those objectives which you referred to as perhaps
16 being ultimately going to become clear, it is absolutely
17 urgent that they become clear before one goes into rulemaking
18 or anything else.

19 MR. KERR: Thank you, Mr. Mark.

20 After those comments, I should remember to add
21 that the meeting today is being conducted in accordance -- I
22 have not mentioned that, have I?

23 MR. QUITTSCHREIBER: No.

24 MR. KERR: I must. -- with the the provisions of
25 the Federal Advisory Committee Act and the Government in the

1 Sunshine Act and all other applicable rules and
2 regulations.

3 Mr. Quittschreiber is the designated Federal
4 employees. Rules for participation in today's meeting have
5 been announced as part of a Federal Register notice on
6 August 6 of this year.

7 A transcript is being kept and will be available
8 as stated in the notice. Those speaking who want to be
9 recorded should try to use microphones so that the recorder
10 can understand what you say.

11 I gather that we have not had any written comments
12 nor requests for time to make oral statements.

13 I will proceed now with the meeting; but before I
14 do, I should ask, I am sure, if there are comments or
15 specific questions with which members of the Subcommittee
16 would like to deal at this point, or the consultants.

17 (No response.)

18 I see none. I will therefore call upon Mr. Butler
19 of the NRC. Mr. Butler.

20 MR. RUBENSTEIN: My name is Lester Rubinstein. I
21 am assistant director for core containment systems. As Dr.
22 Kerr has mentioned, the staff is here to answer many of the
23 questions which he has raised. I am not sure in this time
24 period we will be able to answer all of the questions;
25 however, we have prepared a comprehensive discussion on most

1 of the issues he touched upon.

2 We do provide early in the discussion a
3 perspective on the current state of affairs of hydrogen
4 management in all containments. We touch briefly on this.
5 We move on to place this in perspective with the current
6 regulations and the plans that the staff has to deal with
7 hydrogen management in the interim through the interim rule
8 and the requirements for Sequoyah, and finally, with the
9 long-term rulemaking hearing.

10 We have invited our consultants to give you a
11 detailed presentation on the status of our work on the
12 distributed ignition system, with particular attention to
13 the Sequoyah plant.

14 Dr. Butler will lead off with an introduction
15 which considers the agenda we will follow and some
16 particular remarks on general containments.

17 (Slide.)

18 MR. BUTLER: Good morning. My name is Walter
19 Butler of the NRC staff.

20 As Mr. Rubinstein described for you, we will
21 follow as close as we can the agenda given by the
22 Subcommittee. We have presenters today. Following myself
23 will be Shapaker, Tinkler, instead of Phil Di Benedetto, we
24 will just cover for his subject matter with respect to the
25 response of equipment, and Messrs. Fleishman, Madeiros,

1 Bowman and Cybulskis.

2 I thought it would be helpful to get started
3 simply by putting on the board the agenda which each of you
4 already have. The presentations we have are basically as
5 prescribed here, which allows about an equal amount of time
6 for questioning.

7 We will start off here with a background
8 discussion and setting perspective of how we got to this
9 point today. Some things have been moving very fast on the
10 subject of hydrogen.

11 (Slide)

12 The second page of your agenda is shown there. We
13 hope to conclude the NRC staff portion of presentation soon
14 after lunch, with the statement of our view of where we
15 stand on the general subject of hydrogen and on the specific
16 nature with respect to the Sequoyah station.

17 This will be followed by a presentation by R&D
18 Associates of Los Angeles on subject matter similar to what
19 the staff will have previously addressed. Similarly, TVA
20 will express its views on the same subject. The last hour
21 here will be presented by Mr. Miller of EPRI.

22 (Slide)

23 I would like to discuss now the statement of the
24 objectives as we view them for today's presentation. We
25 would like to provide summary discussions of the general

1 hydrogen issue as it affects all plants, including a
2 description of our overall program to resolve the issue of
3 hydrogen generation and control for all plants.

4 We would like to provide detailed discussions on
5 the specific hydrogen issue as it affects the Sequoyah
6 station, and also the specific attention to the interim
7 resolution for the Sequoyah station.

8 It is my understanding that there are two issues
9 that need some attention by the ACRS or on which the
10 Commission would like the ACRS's views. One issue which
11 will be dealt with today is the hydrogen issue and the
12 response to hydrogen combustion. The other issue is the
13 structural response -- the behavior of the structure to the
14 pressures as a result of hydrogen burn.

15 The second issue will be addressed by another
16 subcommittee on Tuesday the 2nd of September. We will only
17 have a brief discussion here, simply for perspective
18 reasons, a discussion of our view of the yield pressure and
19 failure pressure for the ice condenser containments.

20 (Slide)

21 A slide that describes some of the background to
22 how we got to this point today. First of all, the pre-TMI
23 requirements, pre-TMI licensing requirements are prescribed
24 in the regulations, 10 CFR, Section 50.44, and in the
25 general design criteria number 50 of Appendix A to Part 50.

1 Following the TMI accident there was prepared the
2 staff's Action Plan, NUREG-0660. Items relevant to hydrogen
3 generation and control are dealt with in items II.P.6, 7 and
4 8 of the Action Plan, as well as the so-called RSSMAP
5 studies of the Probabilistic Assessment Staff of Research.

6 Those entries in the Action Plan identify work
7 that needs to be done on prescribed schedules dealing with
8 hydrogen. The previously mentioned series of SECY papers,
9 107A and B, were prepared in response to item II.B.7 of the
10 Action Plan. SECY paper 80-283 was prepared in response to
11 the RSSMAP studies of the Probabilistic Assessment Staff.

12 There was a concluding paper prepared very
13 recently, item 1.3.3 of that slide, Hydrogen Control for
14 Sequoyah, which was considered during the Commission
15 meetings of August 14 and 21, just these past two weeks.

16 There are also some related reports available, two
17 R&D Associates reports, one dealing with structural response
18 and the other dealing with a critique of the hydrogen
19 generation and control work that was sponsored by the NRC
20 staff. There are also a series of Sandia reports, including
21 the compendium and a number of reports on selected topics
22 that are available.

23 (Slide)

24 A slide of the chronology on the recent
25 activities, starting with the February 22 issuance of SECY

1 107. That paper gave a general discussion of the
2 responsiveness or sensitivity of the various types of
3 containments to hydrogen generation and control. That paper
4 concluded with staff recommendations that MARK I and MARK II
5 containments on boiling water reactors be required to be
6 inerted.

7 An interim rule that carries out or proposes to
8 carry out that recommendation will be discussed later on in
9 today's work.

10 On March 19 there was a Commission briefing on
11 that general paper, followed by on March 28 a Commission
12 request for certain additional information relative to the
13 107 paper. The requested information was furnished in SECY
14 papers dated April 22 and June 20, the 107A and 107B papers.

15 On June 26 the Commission heard a briefing on the
16 entire series again. There was a meeting with the ACRS on
17 July 11, followed by the ACPS report on Sequoyah dated July
18 15. The last two entries there were the August 14 and
19 August 21 briefings of the Commission.

20 (Slide)

21 It is my understanding that these issues will be
22 considered again at the full committee meeting of the ACRS
23 on September 4 and that there will be a Commission meeting
24 soon thereafter. I am not sure whether it is the evening of
25 that same day or the very following day, the 5th, but we

1 should hear that very soon.

2 I thought it would be a good idea to briefly
3 summarize where the different plants with different
4 containments stack up. The MARK I BWRs we recommend and the
5 draft interim rule would require that they be inerted.
6 There are only two operating BWRs that are not currently
7 inerted. These are the Vermont Yankee and Hatch-2 plants.

8 MARK II BWRs, because of their small size and low
9 pressure - well, moderately low pressure design -- are also
10 recommended for inerting. Ice condensers. The specific
11 licensing requirements relative to hydrogen generation and
12 control, as we all know, are under consideration by the
13 Commission and are presently under discussion.

14 MARK III containments are in the same hydrogen
15 sensitivity situation as the ice condensers, and licensing
16 requirements for the MARK II relative to hydrogen generation
17 and control are still under consideration by the staff.

18 The first plant to be licensed with a MARK III is
19 the Grand Gulf station. Subatmospheric containments and dry
20 containments, because of their large size and high design
21 pressure, do not require hydrogen control requirements
22 beyond Section 50.44, 10 CFR 50.44, pending the rulemaking
23 proceeding that was discussed earlier.

24 On the subject of ice condenser plants, we have
25 listed all ten of them here at five sites.

1 (Slide)

2 Their fuel load dates are indicated on the far
3 right. Three operating plants include Cook-1 and 2 and
4 Sequoyah-1. Though Sequoyah is at this time loaded to 5
5 percent power, all the others are in various stages of
6 construction, with McGuire-1 being the very next plant
7 scheduled for fuel loading. All the others are 1981 and
8 forward.

9 (Slide)

10 A brief paper that characterizes design features
11 important to hydrogen control. The structure of all ten ice
12 condenser units are all freestanding steel, with the
13 exception of the two Cook units, which are reinforced
14 concrete. Design pressure: they are all at 15 pounds except
15 Sequoyah and Cook which are at 12 pounds.

16 The net free volume of containment: they are all
17 about 1.2 million cubic feet, about half the size of the
18 large dry containments. The provisions for containment
19 sprays: they are all in the upper compartments with the
20 exception of D.C. Cook, which has the sprays in the lower
21 compartment as well.

22 (Slide)

23
24
25

1 The staff approach toward developing resolution on
2 the subject of hydrogen control, the staff has broken the
3 program up into short-term and long-term. The object of the
4 short-term is to define and implement those requirements to
5 assure no undue risk to the health and safety of the public
6 pending the rulemaking proceeding.

7 The long-term approach would require the owners of
8 nuclear plants to conduct analyses and experiments where
9 these studies should be designed to establish the data base
10 for defining the design features that make the plant's
11 response to degraded and melted core accidents acceptable.

12 We will also establish NRC-sponsored research and
13 technical assistance programs to confirm the results
14 obtained by LWR plant owners, and to establish the
15 acceptance criteria for the design features we expect will
16 be proposed for mitigating degraded and melted core
17 accidents.

18 (Slide.)

19 A description of our short-term program for
20 Sequoyah and for other ice condenser plants. We have
21 technical assistance work underway with Livermore National
22 Laboratory where they are charged with developing data to
23 determine that the interim distributed ignition system
24 proposed for Sequoyah will not degrade the safety of the
25 plant, and that the igniters will in fact successfully lean

1 mixtures of hydrogen, air and steam.

2 At Battelle-Columbus we have a program underway to
3 assess the IDIS in terms of the extent to which that system
4 can improve hydrogen control in ice condenser plants. At
5 Sandia we have work underway to develop a data base on
6 alternative systems. The point being that should the
7 current efforts on the IDIS lead to a conclusion that they
8 are not suitable for hydrogen control, we would like to have
9 some backup system and efforts to understand potential
10 backup systems are underway at Sandia. These are the halon
11 system and the water fogging system.

12 At Ames Laboratory we have work underway to
13 understand the structural response of various containments.
14 Included in that program is a determination of the response
15 of the structures to localized detonations. TVA and the
16 other ice condenser owners have an association designed to
17 study the sensitivity of ice condensers by use of the
18 so-called CLASIX code. More details on that code will be
19 discussed later today.

20 They will have programs underway to assess the
21 accuracy and reliability of the CLASIX code. They will
22 perform -- they have completed tests at the Singleton
23 Laboratory and have just yesterday submitted the results of
24 those tests. They will have an extended program of tests at
25 the Fenwall Laboratory in Massachusetts, and that will be a

1 subject of discussion by the staff later today.

2 Finally, part of the Fenwall tests will include an
3 assessment of the response of selected vital equipment to
4 the anticipated burn environment.

5 (Slide.)

6 The final slide I have here, to give you a
7 perspective on the time in which we are doing the various
8 things and selected pieces of milestones here, starting with
9 September 2, next Tuesday, we expect to receive from IVA
10 their safety analysis report on the IDIS. There will be the
11 full committee meeting on September 4 followed by a
12 Commission meeting on Sequoyah. October 1 we expect to have
13 the testing at the Fenwall Labs completed, at least for the
14 early series of tests. The work at Livermore will be
15 completed October 30. The Sandia work on assessment of
16 alternative hydrogen control measures, October 30.

17 Similarly, the Battelle-Columbus analyses, we
18 expect to have that work completed by October 30 as well.
19 Reports on all these activities should be available to the
20 staff by November 15 with a target of a staff SER on the
21 interim distributed ignition system by September 15. We
22 expect to present the results of the staff review to the
23 ACRS in January and brief the Commission soon thereafter.

24 MR. LAWROSKI: Will there be any test results
25 available by the September 4 meeting?

1 MR. BUTLER: While there will be the results of
2 the very elementary tests at Singleton Labs, it is a very
3 small unit that just -- there will be results of the
4 durability of the selected igniter system. There will be
5 completed certain tests at Fenwall labs, but data from them
6 will not be too meaningful. They are just kind of checkout
7 tests.

8 MR. LAWROSKI: I was referring to the expectation
9 based on the Sequoyah hydrogen control plans report.

10 MR. BUTLER: I'm sorry. I did not hear the
11 question.

12 MR. LAWROSKI: My question was based on a date I
13 thought I saw in the --

14 MR. KERR: Will you use a microphone, Mr. Lawroski?

15 MR. LAWROSKI: I am now. Thank you, Bill.

16 I thought there was a date of the availability of
17 such tests by August 15 according to the report -- draft
18 report on hydrogen control for the Sequoyah nuclear plant,
19 which bears no date unfortunately.

20 MR. BUTLER: Yes. We were expecting a safety
21 analysis report from TVA by August 15. The safety analysis
22 report -- there was material furnished August 15, but we
23 felt it was not really a complete report with respect to
24 what should be in a safety analysis report. We have since
25 sent a letter to TVA identifying the topics that need to be

1 addressed, and they intend to furnish a response to that by
2 September 2, on Tuesday.

3 Nevertheless, those will not contain the results
4 of any substantial testing. They are results primarily of
5 analytical efforts.

6 MR. LAWROSKI: I want to correct myself. The
7 report does have a date, August 13, 1980.

8 MR. BUTLER: Oh, yes, that is the -- that is the
9 hydrogen control paper.

10 MR. LAWROSKI: Yes.

11 MR. BUTLER: But that submittal of August 15 was
12 expected to be a safety analysis report which was not
13 expected to contain results of experimental work.

14 MR. LAWROSKI: That is also mentioned. It is
15 expected to be available by September, as well as some test
16 results.

17 MR. KERR: Do you understand Mr. Lawroski's
18 questions? I am not sure I do, but if you do it is okay.

19 MR. BUTLER: No, I don't understand the last
20 question.

21 MR. LAWROSKI: He has answered the -- at least for
22 me enough, the extent of the tests that will be available.

23 MR. KERR: Okay.

24 MR. MARK: Could you say in a very few words, just
25 topic headings, what is in and what is not in the MARCH code?

1 MR. BUTLER: I think I will have to defer to Mr.
2 Cybulskis or Charlie Tinkler.

3 MR. RUBINSTEIN: We have a presentation on that
4 later in the afternoon.

5 MR. MARK: That will be fine.

6 MR. KERR: Does it say in a very few words what
7 Mr. Mark wants?

8 MR. RUBINSTEIN: I cannot promise that.

9 (Laughter.)

10 MR. KERR: Please continue.

11 MR. BUTLER: The very next topic on the agenda is
12 a discussion of the rulemaking proceedings, starting off
13 with the interim rule by Mr. Fleishman, and he will be
14 followed by Mr. Madeiros on the advance notice of rulemaking
15 for the final rule.

16 MR. KERR: Mr. Butler, if your presentation is
17 complete, on one of your transparencies labeled "Staff
18 Approach Toward Developing Resolutions," I find that the
19 long-term approach requires owners of nuclear plants to
20 conduct analytical and experimental studies to establish a
21 data base for defining those design features that make the
22 plant response to degraded/melted core accidents acceptable.

23 Now, it would seem to me that in order to do that
24 they would need a fairly clear definition of what degraded
25 core/melt accidents they were protecting against. Is the

1 staff in the process of developing a scenario or a design
2 basis accident against which they are to mitigate?

3 The transparency does not mention this, and it
4 would seem to me that that would be a fairly important part
5 of the total picture.

6 MR. BUTLER: I believe the answers to that
7 question will be the outcome of the rulemaking proceeding
8 itself.

9 Now, there are a series of questions in that area
10 contained in the advance notice of rulemaking, and Mr.
11 Madeiros will describe the kinds of questions we are asking
12 to adduce the necessary information.

13 MR. KERR: But you are not going into the
14 rulemaking hearing blind, are you? The staff will have a
15 position before it goes in -- a tentative position at least
16 before it goes into rulemaking, won't it?

17 MR. BUTLER: I would rather defer to Mr. Madeiros
18 to answer the question.

19 MR. KERR: Surely this is going to be an NRC
20 position and not a Mr. Madeiros position.

21 MR. BUTLER: That is true.

22 MR. KERR: Well, does the NRC have a position
23 now? Is it going into the rulemaking hearing with a
24 tentative position?

25 MR. BUTLER: Well, I believe by the class of

1 questions that we ask it indicates the direction we are
2 taking.

3 MR. KERR: So that you will have a direction but
4 nothing other than that.

5 MR. RUBINSTEIN: We do have a position going into
6 the rulemaking hearing for the small containments such as
7 the MARK I's, the MARK II's. We will have a position --

8 MR. KERR: The position that I see now is a
9 position that says given that one has hydrogen in the
10 containment, what does one do about it. But what I see
11 written here is require owners to conduct analytical
12 experimental studies to establish a data base for defining
13 design features that respond to degraded/melted core
14 accidents.

15 Now, in order to do that it seems to me one has to
16 know what kind of accidents one is designing against. And
17 the position for Mark I and Mark II does not define the
18 accident against one which is designing.

19 It is a non-mechanistic thing. It defines one has
20 a certain amount of hydrogen, and what I am trying to get a
21 picture of is that the sort of definition that you plan or
22 -- I recognize the answer may be that you simply have not
23 decided yet. And if that is the answer, then do you plan to
24 decide by the time you go into rulemaking what at least the
25 staff would consider to be a sensible approach to this?

1 MR. NORBERT: Jim Norbert from Office of Standards
2 Development.

3 In the interim rule which Mort will be talking
4 about here shortly there is a number that we are asking the
5 vendors to do design studies on for the interim -- as part
6 of the interim rule, a number on how much hydrogen release
7 they should -- to look at as a basis. There is nothing
8 fixed. It is not the staff's final position, but it gives
9 them a bound to shoot for; and that is in the interim rule.

10 MR. KERR: I guess what I am reading from is
11 labeled as a long-term approach.

12 MR. NORBERT: I was not quite sure about that with
13 Walt either.

14 MR. RUBINSTEIN: The staff has not defined a set
15 of scenarios which would lead specifically to severe core
16 damage and large amounts of hydrogen generation, and we have
17 had discussions, and TVA has identified for the Sequoyah
18 plant four scenarios which would have the highest likelihood
19 of leading to severe core damage.

20 We believe we have taken mitigation -- pardon me.
21 We believe we have addressed those features in the Action
22 Plan and through operating procedures and guidelines.

23 Our current approach that we are dealing with
24 right now is -- I think you are right on target -- is
25 mitigation consequences in dealing with Sequoyah.

1 MR. KERR: I am referring more to the long-term
2 approach on Mr. Butler's slide.

3 MR. RUBINSTEIN: The staff does not have a
4 position on sequences which would lead to severe core
5 damage, and it is likely that we will look to the rulemaking
6 hearing for guidance in this matter.

7 MR. KERR: Okay. So at this point one will not
8 require the owners to do these studies until after
9 rulemaking has been completed, is that correct?

10 MR. RUBINSTEIN: We want them to do studies and
11 help us in the identification --

12 MR. KERR: How can they do studies to mitigate
13 against something without knowing what they are trying to
14 mitigate against, and they won't know this until after
15 rulemaking. It seems to me that one has something of --
16 well, at this point since inflation has set in, it is
17 probably a "Catch-33" situation.

18 (Laughter.)

19 I really --

20 MR. BUTLER: I recognize the question, and it is a
21 valid question. In summary fashion you are asking if the
22 staff will undertake a study program prefatory to the
23 rulemaking to see if the staff can prescribe certain bounds
24 or guidelines for the rulemaking.

25 At the present time the staff does not plan that

1 kind of activity. It might be an appropriate activity for
2 the staff to undertake. I guess the staff really has not
3 decided it, okay.

4 Now, why don't we go through the discussion on the
5 rulemaking, and if the questions continue to prevail, it
6 might be an appropriate comment of the ACES.

7 MR. KERR: Okay. Thank you, Mr. Butler.

8 MR. LAWROSKI: Are we having this discussion and
9 then the rulemaking presentation?

10 MR. BUTLER: I believe I described what the
11 present activities --

12 MR. KERR: I am satisfied with your answer.

13 MR. BUTLER: Thank you.

14 MR. KERR: Mr. Lawroski, more questions?

15 MR. LAWROSKI: It seems to me we have the cart
16 before the horse, but maybe this is the way we arranged the
17 agenda. I don't know.

18 MR. KERR: Your problem is you are living back in
19 the horse-cart era.

20 (Laughter.)

21 MR. LAWROSKI: I see. Do you have an inflationary
22 term for that?

23 MR. KERR: Please continue.

24 (Slide.)

25 MR. FLEISHMAN: My name is Mort Fleishman, and I

1 am with the Office of Standards Development. I am going to
2 describe the rulemaking on the interim rule that we are
3 presently working on.

4 The staff is currently involved with two major
5 rulemaking actions, one involving a longterm related to
6 consideration of degraded or melted cores safety regulation,
7 and the other one involving an interim rule on hydrogen
8 control and certain degraded core considerations.

9 Now, the long-term rulemaking consists of four
10 parts. One is an advance notice of proposed rulemaking
11 which will be discussed by Mr. Madeiros after I am
12 finished. That has already been sent to the Commission, and
13 it is a SECY-80-357 paper. And we view this long-term
14 rulemaking as a two to four year effort depending upon what
15 questions or problems may come up during the course of the
16 rulemaking.

17 There is a very good possibility that there will
18 have to be a rulemaking hearing similar to what happened at
19 the ECCS hearing, and depending upon what comes up there,
20 that will dictate essentially how long the rulemaking will
21 last.

22 The interim rule, which is the subject of my talk,
23 has already been sent to the Commission via SECY-80-399. It
24 was sent to the Commission this past week, and we look at
25 that as going through just a proposed rule and an effective

1 rule phase. Depending upon what the Commission decides,
2 that rule could be made effective by early 1981.

3 Now, before going into the details of the rule I
4 would just like to give a little bit of background on the
5 rule. Following the TMI-2 accident there were a number of
6 studies that were initiated which culminated in several
7 documents of which most of you, I am sure, are familiar with.

8 The documents that were related to the Three Mile
9 Island accident was NUREG-0878, which was issued in July of
10 1979 and essentially was a status report and summary of the
11 short-term recommendations made by the Lessons Learned Task
12 Force; and NUREG-0885 was issued in October 1979, which was
13 their final report.

14 Based on these two reports, letters were sent to
15 licensees of operating nuclear power plants and also
16 applicants and other people that would be affected by that
17 describing what followup actions should be taken resulting
18 from the WRC reviews of the Three Mile Island accident.

19 (Slide.)

20 Clarifying letters were also sent out on October
21 30. After the September 13 letter regional meetings were
22 held the week of September 24 to clarify the recommendations
23 made. I believe the ACPS has already reviewed these
24 documents. In fact, the September 13 letter had a number of
25 recommendations in it that were attributed to the ACPS. In

1 fact, the ACRS letter concerning that was included in the
2 September 13 letter.

3 Some recommendations that the ACRS made were on
4 accident monitoring and instrumentation such as containment
5 pressure, hydrogen concentration, and things like that. In
6 May 1980 the staff and the Commission reviewed and approved
7 NUREG-0660, which was the TMI-2 Action Plan, and under Task
8 II.P.8 of that Action Plan they describe the various rules
9 that we are presently working on.

10 That is sort of the background, and it sort of
11 gives you an indication of the relationship of the long-term
12 rule and the interim rule to each other.

13 The interim rule itself consists of two major
14 parts.

15 (Slide.)

16 The first part is based on the -- is related to
17 hydrogen management in containment, and the other is related
18 to design and other requirements.

19 Now, these design and other requirements were
20 revealed from the TMI-2 accident and had been discussed in
21 the various NUREG reports, and the licensees are already
22 familiar with most of these requirements.

23 The recommendations that we have made -- that we
24 have in this paper were already transmitted to the
25 Commission. As far as our position on the NUREG, we have

1 already discussed that, and we are codifying regulations
2 which the staff feels should be put into place immediately
3 to protect the public health and safety.

4 The staff believes that if these requirements are
5 put into effect, it will improve the capability of nuclear
6 power plants to deal effectively with TMI-2 type accidents.
7 We don't have any specific quantification of just how much
8 an improvement that will be, but essentially based on
9 engineering judgment, we feel there will be an improvement.

10 (Slide.)

11 MR. KERR: Mr. Mark.

12 MR. MARK: In the interim rule it is certain
13 degraded core considerations. You also mentioned TMI-2
14 type. Is it then clear and agreed that one is limiting
15 one's attention to the sort of boiloff picture for TMI and
16 nothing else?

17 MR. FLEISHMAN: In this interim rule that is
18 correct.

19 MR. MARK: So this also involves hydrogen coming
20 at a quite slow rate, like 1 percent reaction per minute.
21 It has nothing to do with 107 which does not observe such
22 restrictions, and the only thing uncertain is how much
23 hydrogen was actually involved in TMI which no one yet knows.

24 MR. FLEISHMAN: I believe we feel the TMI-2 type
25 accident was somewhere in the 25 to 50 percent metal/water

1 reaction.

2 MR. MARK: I'm aware that you feel that, but I'm
3 not aware of any evidence that really puts the finger on it.

4 MR. FLEISHMAN: That is correct.

5 MR. KERR: Please continue.

6 MR. FLEISHMAN: Okay. Relative to hydrogen
7 managment, the interim rule has four items within the rule.
8 We are going to require that the Mark I and II reactors --
9 BWR reactors be inerted. We are not saying anything at all
10 about ice condensers or other pressurized water reactors in
11 the rule.

12 It would mean that the Vermont Yankee and Hatch-2
13 reactors, which are not presently, will have to be inerted.
14 This recommendation as made to the Commission by the staff
15 in SECY-107(d) -- the whole SECY-101 series.

16 We are also requiring in the interim rule that
17 design analysis studies be performed to look at various
18 measures to handle large amounts of hydrogen. We are not
19 specifying any specific accident scenario, but we are just
20 saying that we want them to look at measures to study what
21 they could do to handle various amounts of hydrogen.

22 We told them to look at hydrogen up to 75 percent
23 metal/water reaction. The measure we are suggesting that
24 they look at in the evaluation would include inerting,
25 hydrogen recombiners, purge systems, halon suppressant

1 systems, filtered vent systems, hydrogen combustion systems,
2 water fog spray, or combinations of these.

3 Depending upon the results of these studies
4 further requirements may turn out to be necessary in the
5 long-term.

6 MR. MARK: Is it made clear in that connection
7 anything about the suddenness of the appearance of this
8 hydrogen; that is, does it come on at TMI-type rates or not
9 come on in a delta function?

10 MR. FLEISHMAN: We say that -- in the rule that
11 the hydrogen generated within the first eight hours, and we
12 do not say anything in the rule about how fast that reaction
13 rate should be. We just say hydrogen generated within the
14 first eight hours, which would basically avoid having to
15 look at long-term radiolytic composition.

16 MR. MARK: That, I think, is a good point, but
17 there is still quite an important difference, it seems to
18 me, and it seems to me it will seem so to the designers, if
19 you assume that the hydrogen is bled out uniformly
20 throughout the eight hours or comes on in the particular ten
21 minutes during the eight hours.

22 MR. FLEISHMAN: The rule says nothing about that.
23 We did not give any specific guidance on that in our --

24 MR. KERR: Don't you think you should?

25 MR. FLEISHMAN: What do you think about that, Walt?

1 MR. BUTLER: I think with respect to the lower
2 bound -- Mort identified the upper bound of eight hours, and
3 that will indicate clearly that they are not to rely on the
4 heat removal systems.

5 With respect to the rapid rate of -- that is, the
6 lower bound for a rate of hydrogen generation, I guess what
7 we are trying to do is have them consider it to come out as
8 fast as possible. Yet, we are not prescribing --

9 MR. KERR: Mr. Butler, I think the question was
10 not what you are going to specify but whether you are going
11 to specify something, and Mr. Fleishman says you are not
12 giving any rate of hydrogen generation.

13 I think the question is shouldn't you, if you are
14 asking designers to design something, give them some idea of
15 what they are designing for? And apparently at this point
16 nothing is said about the rate of hydrogen evolution.
17 Wouldn't that have a significant effect on how one was going
18 to handle it?

19 MR. BUTLER: If the rate comes out -- if you
20 prescribe a rate that is very fast, it certainly will have
21 an effect. It is our hope, though, in not prescribing it in
22 the interim rule that that issue be considered during the
23 final rule.

24 MR. KERR: But how can one do a design to handle
25 this without knowing what one is designing for, Mr. Butler?

1 I don't understand.

2 MR. BUTLER: I think the problem is one with
3 schedule. In the interim rule we are requesting that
4 certain studies be conducted on a sensitivity basis. Those
5 results will then be available for assessment during the
6 final rule phase.

7 If after reviewing the results of those analyses
8 it looks like additional studies are needed, those
9 additional studies will have to be done.

10 MR. KERR: Then what you expect they will do is
11 consider everything from an instantaneous release to a
12 release uniformly over eight hours. Even though you have
13 not asked for that, you expect that is what they will do.

14 MR. BUTLER: Well, I am not sure what to expect in
15 the way of response. I guess questions will have to wait
16 until answers come in.

17 MR. KERR: Well, I guess there is good biblical
18 precedent for doing this sort of thing. I can remember
19 Daniel and King Nebuchadnezzar when the king called him in
20 and asked him to interpret his dream, and the only problem
21 was the king could not remember what the dream was.

22 So he asked Daniel to tell him what the dream was
23 first. But I would hope that people would produce better
24 results from a study if they knew what it was you wanted
25 them to study.

1 MR. BUTLER: I might indicate that the reason we
2 did not put the lower bound in was because we did not think
3 that the analysis results was going to be very sensitive to
4 any lower bounds, except for the terminal lower bound, and
5 we were not ready to require that terminal lower bound
6 pending receipt of the results of some of these studies.

7 So I guess any studies of any rate would be
8 satisfactory, and as we get further into the rulemaking and
9 we decide that we really want that end point, well, we want
10 to defer that decision until we have the results of some of
11 these early studies.

12 MR. ETHERINGTON: Are you assuming the same amount
13 of hydrogen with the only difference being the temperature
14 and steam content in the containment: that is, an immediate
15 release of one cubic foot of hydrogen leads to the same end
16 concentration as a release over a long period? Or are you
17 removing some of that hydrogen in the interim?

18 MR. FLEISHMAN: The rule itself did not get that
19 specific.

20 MR. ETHERINGTON: What is the concept? What is
21 your concept?

22 MR. FLEISHMAN: We are asking for sensitivity
23 analyses to be done, so we assume that they would look at
24 various rates of generation and various scenarios.

25 MR. ETHERINGTON: And various rates of removal as

1 well?

2 MR. FLEISHMAN: Depending on the systems we are
3 asking them to look at, they would have different
4 capabilities. Hydrogen combustion is one of the systems we
5 are suggesting they look at. It is actually in the
6 statement of considerations. We have given them
7 essentially guidance on some of the areas we want them to
8 look at. The hydrogen combustion system was one of the
9 areas we suggested that they look at.

10 MR. ETHERINGTON: In other words, you will get a
11 higher pressure from an immediate release than could be
12 obtained with a long-term release, assuming that you have
13 these hydrogen-relieving systems.

14 MR. FLEISHMAN: Yes. In other words, that would
15 be the things they would look at, depending upon which
16 system they would look at. It would affect the --

17 MR. ETHERINGTON: You are including immediate
18 release but you are not insisting on that being in the
19 criteria.

20 MR. FLEISHMAN: We have not specified it in the
21 rule at this point. Now, when we start getting information
22 in as a result of these studies, we may want to revise our
23 guidance, revise the studies. This is going to be an
24 ongoing thing. There is going to be interaction between the
25 staff and the licensee, so depending on what comes in, there

1 is going to be an interplay, and the studies, I am sure,
2 will be modified.

3 There will be discussions between the staff and
4 the licensees as to what action finally gets -- the rule
5 itself is quite general, actually. It just says we want the
6 studies to be done. The specifics of what is going to be
7 done will vary from plant to plant as well as from licensee
8 to licensee.

9 MR. KFRR: Well, I think we have all been faced
10 with the fact that there is a limited supply of manpower and
11 resources available for these studies. Now, if we tell
12 people to go study something which was not what we had in
13 mind, and they spent a lot of effort studying something that
14 was quite different from what we wanted them to study, it
15 seems to me that that is not very efficient.

16 I recognize that perhaps you cannot know exactly
17 what you want done until you have had some preliminary
18 results, but it certainly seems to me that one could perhaps
19 save manpower and resources if one could be more specific
20 than just sort of saying go away and study the problem. I
21 mean if you have in mind some limitations or some general
22 guidance, it certainly seems to me it might be to
23 everybody's advantage to provide it.

24 MR. BUTLER: What you say certainly is true.
25 However, we did not want to cast in concrete in the form of

1 a rule the specifics on what should be analyzed. It was our
2 hope to maintain some measure of flexibility, so that when
3 we start implementing the rule, we intend to meet quite
4 frequently with the various owners group and at that point
5 provide appropriate guidance as to scope of the studies.

6 MR. KERR: Please continue.

7 Excuse me. Mr. Etherington.

8 MR. ETHERINGTON: Regardless of the basis of
9 calculation, you are going to calculate the pressure which
10 you believe will result in combustion of a certain amount of
11 hydrogen. Now, I was told recently that you use the
12 specific heat and constant pressure in calculating the
13 pressure from the combustion. I was told this.

14 Is this true; and if so, why don't you
15 specifically use heat and constant volume?

16 MR. FLEISHMAN: I would defer to Mr. Tinkler on
17 that.

18 MR. TINKLER: The calculations that you saw in the
19 SECY paper 107 were done assuming specific heat of gases at
20 constant volume.

21 MR. ETHERINGTON: Constant volume.

22 MR. TINKLER: Constant volume.

23 MR. ETHERINGTON: Then I was misinformed.

24 MR. TINKLER: There was a calculation presented at
25 a meeting sometime ago. I do not recall the details of it.

1 Someone did present a calculation with pressure calculated
2 using heat and constant pressure. But the calculations
3 which you have seen to date in the SECY papers performed by
4 the staff have been --

5 MR. ETHERINGTON: That is about 40 percent higher.

6 MR. TINKLER: If you do the calculation
7 consistently, it may not be that much different. The
8 calculations which you see are also done assuming the gas
9 constants for the various gases as a relatively low
10 temperature, so that it does contain some additional
11 conservatism in that regard.

12 MR. ETHERINGTON: Not very much, though.

13 MR. TINKLER: Two thousand or three thousand
14 degrees. It is 10 or 15 percent, yes.

15 MR. ETHERINGTON: Yes.

16 MR. LEE: Based on the present understanding of
17 the amount of hydrogen generation, and even perhaps the rate
18 of hydrogen generation, the interim rule would require
19 inerting for MARK I and MARK II BWPs.

20 MR. FLEISHMAN: That is correct. We feel it would
21 be appropriate at this time to require the inerting at least
22 until we have more information, which will be determined
23 during the long-term rulemaking.

24 MR. LEE: But that decision would have assumed
25 certain rates of hydrogen generation, I presume, which could

1 have an instantaneous generation of a certain amount; or is
2 it different from that?

3 MR. FLEISHMAN: Well, right now the way the
4 regulations presently stand, most plants would not have to
5 be inerted based on the way 50.44 is written. The
6 regulations now call for no more than a maximum of 5 percent
7 metal/water reaction to be considered by the regulations.

8 TMI-2 showed us that that was no longer accurate.
9 So, in lieu of having any further information which is going
10 to be determined in the long-term rulemaking, we felt it
11 would be appropriate to require inerting right now. I guess
12 the feeling is that inerting will be beneficial. Just how
13 beneficial it is and what sort of reactions it can correct
14 for, I don't think we know exactly.

15 MR. LEE: I guess my question is really related
16 to one of the comments. General Electric Company has
17 somewhere -- I can place where I read it in the bulk of
18 documents I have -- but they were criticizing the staff
19 position relative to the fact that in BWR systems you cannot
20 expect hydrogen generation immediately nor a loss of coolant
21 accident or anything like that.

22 So they feel there is no need to go inerting MARK
23 I and MARK II containment systems.

24 MR. FLEISHMAN: That is correct. That is G.E.'s
25 position.

1 MR. LEE: Staff has taken the position that
2 regardless of comments, it would be necessary at this point
3 to go to inerting.

4 MR. RUBINSTEIN: Let me address that. We did not
5 do a mechanistic analysis of the generation of hydrogen in
6 dealing with the BWR MARK I's and II's. For the interim,
7 staff essentially assumed that hydrogen could be generated,
8 looked at the structural response and design pressures,
9 looked at the amounts of hydrogen which could be generated
10 from a severely damaged core, and came to the conclusion
11 that we had better prepare a mitigation scheme for this.

12 The benefit of inerting a MARK I and II was small,
13 but so were the costs, and the interim we felt it was a
14 prudent course of action.

15 MR. KERR: In that connection, Mr. Rubinstein, I
16 would be interested in your comments on a document that is
17 labeled "Decision Rationale for the Staff's Position on
18 Inerting." On page 3 of the document I find the statement:
19 "MARK I and MARK II containments should be inerted. The
20 decrease in residual risk is small based on probabilistic
21 analyses because the likelihood of this accident scenario is
22 one to two orders of magnitude smaller than the dominant
23 core melt containment failure accident scenario for BWRs."

24 Then this statement: "The persuasive argument for
25 inerting, however, is not the magnitude of risk decrease.

1 It is rather there are no significant countervailing safety
2 disincentives." I am not sure I know what that statement
3 means. I guess it means that inerting will not make things
4 any less safe. "The cost of inerting is small and there has
5 been substantial satisfactory experience with inerting MARK
6 I containments."

7 Now, except for the third statement, it seems to
8 me one can make exactly the same arguments for painting the
9 containment red, white and blue. There are no safety
10 disincentives. It does not cost much. And one has had
11 satisfactory operating experience up till now with red,
12 white and blue containments.

13 But I would feel better if one -- I just cannot
14 believe that the staff cannot come up with a better
15 justification for doing something which does cost some money
16 and does have some safety considerations than to say, well,
17 there don't seem to be any safety disincentives and it
18 doesn't cost much, so let's do it.

19 I don't think it is as persuasive. You must have
20 other reasons for proposing what you have proposed.

21 MR. BUTLER: The subject of deciding whether MARK
22 I's and II's ought to be inerted was a very difficult one to
23 arrive at because we were faced with judgmental views that
24 said they should be inerted because their sensitivity to
25 hydrogen generation was very intense; they had a very small

1 volume and it did not take but about 6 percent metal/water
2 reaction to reach the detonable mixtures of hydrogen
3 concentrations; and a prudent course of action would say: by
4 golly, you can expect operators to do the wrong things at
5 the wrong times and give you the adverse concentrations of
6 hydrogen.

7 Nevertheless, we were faced with results from the
8 Probabilistic Analysis staff that concluded with respect
9 boiling water reactors that the dominant sequences rarely
10 were the core melt sequences, and they found their analysis
11 with their reactor safety methodology which led to their
12 conclusion that inerting did very little to improve the
13 safety of the BWRs.

14 So we had to provide some kind of justification
15 there of a judgmental view that you needed to inert against
16 some objective information that said if you believe
17 probabilistic assessment approaches, then you would not
18 require inerting because it did you very little good.

19 MR. KERR: I agree. I think it would be a good
20 idea to provide some justification. But I believe if a
21 licensee came in to you with the sort of justification that
22 you have given, you would fall on the floor laughing. I
23 mean I would.

24 Now, I recognize that you have a letter from the
25 ACPS suggesting that these things be inerted also, and you

1 have been kind enough not to mention that.

2 (Laughter.)

3 Well, I was just struck by what appeared to me to
4 be a rather weak justification, and I have an idea that
5 there must be better justifications than that.

6 Mr. Fleishman, we have interrupted you
7 periodically. Why don't we get back to you?

8 MR. FLEISHMAN: I have just one other comment. I
9 think the justification is a result of committee decision.
10 That is about what everyone could agree to.

11 The next item that we have relative to hydrogen
12 control in the interim rule is we are going to require that
13 for plants that rely upon external recombiners or venting to
14 satisfy the hydrogen control requirements, that they have
15 dedicated penetrations. In other words, the penetrations
16 should be dedicated for that service only.

17 Finally, we have one other item that we are
18 adding, and that is that plants that rely upon venting have
19 to have external recombiner capability installed. Now, this
20 was a minority recommendation of the Lessons Learned Task
21 Force. However, the staff now believes that in order to
22 reduce the likelihood of release of radioactive material to
23 the environment, means other than venting should be
24 available for control of hydrogen.

25 So, we are requiring in the rule that these plants

1 have penetrations installed so that they could install
2 external hydrogen recombiners if required.

3 MR. LEE: What kind of venting capacity do you
4 have in mind when you have included these two items 3 and 4?

5 MR. FLEISHMAN: Right now the venting capacity and
6 even the external recombiners only have to meet the
7 requirements of Section 50.44, which is up to at most a 5
8 percent metal/water reaction. To actually take into account
9 something like TMI-2, just how these plants should be
10 modified, whether they should be backfit, what the new
11 design should be: that is going to be the subject of the
12 rulemaking.

13 MR. LEE: It is mostly having the recombiners on
14 line, available on-line? Is that the main difference
15 between the present requirements and these suggested rules?

16 MR. FLEISHMAN: The last one. Right now many,
17 many plants just have venting. They do not have recombiner
18 capability.

19 MR. LEE: But I thought oft-times most of them
20 have had --

21 MR. FLEISHMAN: The plant licensed prior to
22 November 5, 1970, I believe, did not have to have recombiner
23 capability. Some of them did but not all of them did, I
24 believe.

25 This is going to require that all plants now have

1 recombiner capability.

2 MR. LEE: Thank you.

3 MR. MARK: Assuming what you mean by that is that
4 there be such fixtures, nozzles, whatnot, outside that a
5 recombiner could be brought from Columbus, Ohio and hitched
6 on.

7 MR. FLEISHMAN: That is correct. Not only that;
8 they have to have proper procedures and shielding available
9 so that they can install the recombiners during and
10 following an accident.

11 MR. KEER: Mr. Seale.

12 MR. SEALE: In this 5 percent metal/water reaction
13 requirement, there still is not a rate specifically
14 indicated there, is there?

15 MR. FLEISHMAN: That is correct.

16 MR. SEALE: As I recall the analysis of the
17 capabilities of recombiners that were available at TMI, even
18 a 5 percent availability would have swamped the capacity of
19 the recombiners that were there.

20 MR. FLEISHMAN: Actually, I said up to 5 percent.
21 Most recombiners probably are not required to meet the 5
22 percent because they have to meet -- I think the way 50.44
23 is written now, they have to be able to accommodate up to
24 five times the metal/water reaction calculated that would
25 occur during a design basis ICCM, which could be less than

1 one percent, actually.

2 MR. SEALE: Well, I think this is a very good
3 example of the kind of fix which is generic, which is
4 tremendously sensitive to the input assumptions. In line
5 with Dr. Kerr's earlier remarks regarding the effect of not
6 being specific with regard to design requirements, it is
7 almost inevitable that the first presentation is going to be
8 a nonacceptable --

9 MR. RUBINSTEIN: Excuse me. The case of the
10 recombiners are really directed toward mitigating the
11 effects of a LOCA, which we do have this fix on. One of the
12 alternatives I think Mort said was the facility had the
13 option of either using a recombiner or purging.

14 The staff believed that it is possible in the
15 future we would not find purging for the mitigation of LOCA
16 hydrogen generation under 5 percent desirable, so we are
17 asking for a dedicated penetration capability with the
18 appropriate isolation and procedures for primarily dealing
19 under 50.44 with the mitigation of the hydrogen generation
20 for LOCA.

21 So it does not deal with the severe core damage
22 and it is not a mitigation feature for TMI-type accident.

23 MR. SEALE: So this is one case where the
24 rulemaking is more general than just the --

25 MR. RUBINSTEIN: This is an interim step to deal

1 with the potential for purging from a LOCA-type design basis
2 accident.

3 MR. SEALE: And then is more general than just the
4 TMI kind of accident.

5 MR. RUBINSTEIN: It is for any accident which
6 would generate hydrogen up to about 5 percent.

7 MR. KERR: Please continue, Mr. Fleishman.

8 MR. FLEISHMAN: The other aspect of this interim
9 rule would require other design requirements and
10 improvements.

11 (Slide)

12 I will not go into that very much because they are
13 not related as much to hydrogen control, except for the
14 first one, which would require high point vents in the
15 reactor coolant system and the reactor vessel head to
16 control noncondensable gas buildup. That is also a
17 requirement that has been previously mentioned in the
18 letters.

19 So we have items on protection of safety equipment
20 in vital areas, implant iodine instrumentation, sampling
21 capability, leakage integrity outside containment, accident
22 monitoring instrumentation, detection of inadequate core
23 cooling and training to mitigate degraded core accidents.
24 These are all recommendations that had already been made in
25 these letters, and we are just going to be codifying them.

1 And that is the interim rule.

2 The status of the long-term rulemaking will be
3 discussed by Mr. Madeiros.

4 MR. LAWROSKI: Are there any criteria except for
5 the recombiners?

6 MR. FLEISHMAN: The recombiners have to meet the
7 criteria of Section 50.44. So in other words, they are the
8 same recombiners that have been installed in all of the
9 plants that meet 50.44 criteria.

10 MR. KERR: It seems to me the TMI situation with
11 which this is dealing is a recognition that people would not
12 want venting to occur under any conditions, so they are
13 saying that although we are willing to consider venting in
14 extreme situations before TMI, we are no longer willing to
15 consider it.

16 Now, I have a little bit of a problem when I see
17 that filtered vented containment is one of the alternatives
18 for dealing with hydrogen, but that is a separate problem
19 and I guess we need to keep problems separated.

20 MR. ETHERINGTON: Isn't the capacity of the
21 recombiner so small that it would have no material effect on
22 the Three Mile Island accident?

23 MR. KERR: I think that is true. That is what I
24 am saying. We are dealing with accidents that we believed
25 in before Three Mile Island, namely, the design basis LOCA.

1 MR. ETHERINGTON: The design basis LOCA. Also, the
2 recombining is too slow to be effective, isn't it?

3 MR. KERR: The recombiner deals, I thought, with
4 radiolytic decomposition, primarily?

5 MR. ETHERINGTON: It will get the other hydrogen
6 down in time.

7 MR. KERR: But it will prevent a pressure buildup
8 that might occur due to radiolytic decomposition. In other
9 words, you might have to vent.

10 MR. ETHERINGTON: Are we talking about more than
11 this at that time?

12 MR. FLEISHMAN: Not right now, no; but there are
13 people who believe that these recombiners could have been
14 beneficial during Three Mile Island also; that during a
15 long-term buildup of hydrogen, that the recombiners could
16 function and could mitigate to some extent the effects of
17 the accident.

18 MR. ETHERINGTON: Are you talking days or hours?

19 MR. FLEISHMAN: Weeks.

20 MR. ETHERINGTON: Weeks, yes.

21 MR. FLEISHMAN: The fact that you have this
22 recombiner capability could still help in something like
23 TMI-2.

24 MR. ETHERINGTON: They had already had the
25 explosion.

1 MR. FLEISHMAN: It would not have prevented the
2 explosion.

3 MR. ETHERINGTON: Pardon?

4 MR. FLEISHMAN: It would not have prevented the
5 explosion.

6 MR. ETHERINGTON: It would not have prevented the
7 explosion, no.

8 MR. KERR: Are there other questions or comments?
9 Is Mr. Madeiros up next?

10 MR. FLEISHMAN: Yes.

11 MR. KERR: Mr. Madeiros.

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1 MR. MADEIROS: Good morning. My name is Manny
2 Madeiros from the Office of Standards Development.

3 On May 9th I described and advance notice of
4 proposed rulemaking being prepared by the staff concerning
5 degraded cooling, and I will not repeat that presentation.
6 But I did bring two slides with me from that presentation to
7 kind of bring people up to speed because I see an awful lot
8 more people here in the audience today and on the
9 Subcommittee.

10 The first slide.

11 (Slide)

12 This briefly describes the problem. First of all,
13 the degraded cooling and resultant core damage is treated
14 unevenly in the regulations. You can go to various sections
15 in the regulations and pull out numerous examples of where
16 one place will discuss a 5 percent metal/water reaction,
17 another place 1 percent hydrogen.

18 You get into Part 100, of course, and you are
19 talking a substantial melting. The second point being the
20 safety analysis stops short of the Class 9 accidents and
21 therefore is inadequate and suggests that the designs are
22 not adequate either and that currently --

23 MR. KERR: Excuse me, Mr. Madeiros. I don't know
24 whether that microphone is at all directional, but if it is,
25 I think you have it pointed away from your voice.

1 MR. MADEIROS: Okay. Lastly, I have on here an
2 example of a related problem. I will not dwell in much
3 detail on this because we have discussed that all on May
4 9th, unless there are questions.

5 (Slide)

6 Here I thought from some of the questions being
7 asked this morning that it might be wise to discuss what an
8 advance notice of rulemaking is. It defines the area of
9 concern, explains the problem to the public. It provides
10 the public an opportunity to advise. Normally we allow a
11 60-day comment period.

12 In this particular case for this advance notice of
13 rulemaking, we have allowed 90 days. We elicit advice by
14 asking questions. In this particular case we have asked 18
15 questions. Two of them have to do with hydrogen, and that
16 is what I am going to get to here in a minute.

17 Then we use the public recommendations to shape a
18 proposed rule. We do this at an early stage of development,
19 and this answers somewhat, Dr. Kerr, your question a little
20 bit earlier to these other fellows of why we did not have
21 some specific numbers in mind.

22 The Commission has made the decision to start with
23 an advance notice of rulemaking in this very important and
24 complex area rather than start with a proposed rule. When
25 you start with an advance notice of rulemaking, you are

1 truly looking for the advice and recommendations of the
2 public, of the regulated industry. And if you already have
3 fixed numbers in your mind, I think we somewhat make a
4 charade of the --

5 MR. KERR: Mr. Madeiros, I insist that one can go
6 in with recommendations that are not necessarily fixed. It
7 has been my experience over the years that you get better
8 advice from people if you ask them about something specific
9 than if you just go out and sort of say what should I do?

10 MR. MADEIROS: That is what we intend to do at at
11 the proposed rulemaking stage. We have just started with
12 this one step earlier and have truly left the issue open for
13 the best advice that we can get. We feel there is much room
14 for imagination out there for the industry, and we are going
15 to let the industry exercise its imagination in giving us
16 good advice on the ANR, so that when we prepare a proposed
17 rule and then perhaps have hearings, we will have fixed
18 positions that we can defend logically technically.

19 (Slide)

20 Now, the features of the ANR have not changed.
21 ANR from the first line -- that is advance notice of
22 rulemaking. The feature of the advance notice of rulemaking
23 have not changed materially since we discussed them in
24 Chicago on May 6th. Briefly they are here.

25 We will require a coherent consideration of core

1 damage in design and review. We will analyze a broad range
2 of accidents within and outside traditional design
3 envelopes. And I might take on the controversial point you
4 raised in your opening remarks, I guess, about whether the
5 staff is leaning towards the end of spectrum only kind of
6 accident, and clearly the answer is no.

7 This advance notice of rulemaking covers the whole
8 range of accidents, as I discussed with you earlier, from
9 clad perforation to small amounts of hydrogen up to release
10 of gap activity, up through large amounts of hydrogen
11 release, and then on to melting. But we are asking that the
12 -- we will consider accidents lesser than design basis
13 accidents, historically consider, and accidents within the
14 envelope.

15 So it is not just the end of spectrum problem that
16 we are interested in, and that is kind of what my third
17 point is there so I will not repeat it. . . sically the
18 reason is my fourth point, because we will be considering
19 multiple failures and operator errors.

20 This brings me to the item of particular interest.

21 MR. KERR: Can you tell me whether the staff at
22 this point plans to take a mechanistic as contrasted with a
23 probabilistic approach?

24 MR. MADEIROS: That has not been decided. My
25 guidance on it would be, as much as I have to do with it, to

1 take a mechanistic approach because I do not feel that the
2 probabilistic approach is developed enough to be very
3 practical. The probabilistic approach deals primarily with
4 failure rates.

5 MR. KERR: I think I understand the probabilistic
6 approach. I just wanted to know which one the staff was
7 likely to adopt.

8 MR. MADEIROS: I thought I would explain a little
9 bit because this is not the trendy view in the Commission or
10 the ACRS. Anyway, I will skip that.

11 MR. ETHERINGTON: There seems to me a problem if
12 you take a mechanistic view, with the statement, I believe
13 by one of the commissioners, that containments ought to be
14 able to handle as much hydrogen as was developed at Three
15 Mile Island because that has already happened. That puts
16 you on a real spot, doesn't it?

17 MR. MADEIROS: Yes, it does on that one, perhaps.
18 What I had in mind more, Mr. Etherington, was that the
19 probabilistic approach just does not deal very well with
20 stupidity, with operator errors, with design deficiencies,
21 that sort of thing. It deals primarily with component
22 failure rates. That kind of information is so skimpy in
23 this business as to almost make that worthless from that
24 standpoint.

25 It is not a procedure or a method or a methodology

1 or whatever buzz words you like to use these days. It
2 cannot deal with the operator. It cannot deal with the
3 mistakes an operator would make.

4 MR. ETHERINGTON: I sympathize with your position
5 entirely. I merely point out that this criterion puts you
6 on the spot.

7 MR. MADEIROS: So again, then, it will require
8 imagination is what you are saying, and I think the
9 Commission can rise to the occasion, or the staff.

10 I will now go to the item of particular interest
11 here, and that is the question concerning hydrogen.

12 (Slide)

13 There are 18 questions in the advance notice of
14 rulemaking. Most of them have to do with analysis and
15 design improvements. Two have to do with hydrogen, and
16 these are the two that are in the advance notice of
17 rulemaking that I expect to be published soon. I will give
18 you a schedule on that in a moment.

19 "Are you in favor of requirements to incorporate
20 into containment design systems for controlling combustion
21 of hydrogen?"

22 "Do you favor methods of control that suppress
23 combustion, or do you favor controlled burning?" If you
24 favor suppression of combustion --" here we are talking
25 about the inerting on demand idea, the halogen suppression

1 schemes, the fogging schemes, the steam injections steams
2 that Mort spoke of and others earlier -- let's see -- "or if
3 you favor controlled burning, do you recommend open flames,
4 spark plugs, catalytic combustors, igniters, these glow
5 plugs we have been talking about?"

6 The idea is with this piece of the question that
7 ignition sources in containment are unavoidable so you might
8 as well control them as leave ignition to chance. That is
9 kind of the idea behind of, I think, all of our work when I
10 speak of igniters. And then what percent of the core's
11 zirconium being oxidized, and at what rate would you design
12 for?

13 We would be looking for some good advice to come
14 up with some of the numbers, Mr. Seale, that you were asking
15 about a little bit ago.

16 And then would you respond differently for
17 different reactor or containment types; and if so, what
18 differences do you recommend?

19 And then we would get into the question of
20 inerting. Can everybody see that? I will push it up a
21 little bit.

22 Would you recommend that all nuclear plants
23 operate with a nitrogen-enriched containment atmosphere as
24 some BWR plants currently do? Why or why not; and if not,
25 which types of containment, if any, would you limit nitrogen

1 enrichment to?

2 I think one of the ideas throughout this advance
3 notice of rulemaking, and particularly here, is that you
4 don't want to do things that take away from safety. We have
5 heard arguments -- I am not sure how completely valid they
6 are, but they seem to have some validity -- that some things
7 like inerting could decrease safety rather than increase
8 it. So we would be interested in those kinds of thoughts in
9 response to this question.

10 Now, as of today two commissioners have approved
11 the advance notice of rulemaking. I expect the other two
12 to approve it shortly. As soon as the full Commission
13 approval is received, we will publish the advance notice of
14 rulemaking in the Federal Register for a 90-day comment
15 period. Assuming we were able to do that by the end of
16 September, we would still then by the end of December this
17 year have the advice and response from the regulated
18 industry, from the public that would allow us immediately to
19 start on a proposed rule, and which I expect will take
20 several months to not only prepare but to process through
21 the staff and on its way to the Commission.

22 Now, I believe the Subcommittee has a copy of SECY
23 80-357, which Mort Fleishman mentioned earlier, that
24 contains all of the questions. One of them, Dr. Kerr, I am
25 sure you will be interested in particularly. In Chicago you

1 suggested that and advised that a question be added weighing
 2 the costs of this work against improvements, so I added this
 3 question: Would you consider useful or appropriate
 4 comparisons between nuclear power plant risks and other
 5 risks to which people are exposed? That was the suggestion
 6 of this subcommittee in Chicago, and it has been added to
 7 the rule.

8 One of the things you raised in your opening
 9 remarks was whether the staff was leaning towards a
 10 conservative approach in the design of mitigating systems or
 11 design improvements or a realistic approach or best estimate
 12 approach, I think were your words.

13 This question is part of the question, so again, I
 14 don't come with an answer today but I come with a statement
 15 that that is in the advance notice of rulemaking. We ask
 16 that specific question of the public, and within a short
 17 time expect to have the advice that with our own thoughts
 18 would let us go forward with a firm position at the proposed
 19 rulemaking stage.

20 Let me see. Well, there was another controversial
 21 point you raised and I don't know whether I should take it
 22 on or not, but I guess I will. You wondered whether it was
 23 the staff's view that the training changes and some of the
 24 instructions to operators and that sort of thing would lead
 25 us -- I cannot remember your exact words -- but perhaps to

1 go slower or take a less stringent view of some of the
2 mitigating --

3 MR. KERR: What I wanted to find out is whether
4 the staff thought the probability of producing a certain
5 amount of hydrogen had been significantly decreased as a
6 result of the changes that have occurred and will occur as a
7 result of TMI-2 .

8 MR. MADEIROS: Okay. That is right.

9 MR. KERR: And if this is being taken into account
10 in deciding what one should finally do about mitigation or
11 prevention.

12 MR. MADEIROS: And that is being said quite
13 often. Many publications, a lot of pronouncements have that
14 view that what has been done in these areas as a result of
15 Three Mile Island should result in less hydrogen
16 production. My point of view is I don't think that is so.
17 I think that is more wishful thinking than anything else.

18 MR. KERR: Well, Mr. Madeiros, if all the money
19 and effort that has been spent since Three Mile Island has
20 had no influence on the probability of hydrogen production,
21 it seems to me that we are certainly wasting a lot of
22 resources.

23 MR. MADEIROS: I did not say it had none.

24 MR. KERR: Well, I will say no significant.

25 MR. MADEIROS: No significant.

1 MR. RUBINSTEIN: Mr. Chairman, I believe that
2 represents his personal opinion.

3 MR. MADEIROS: That is correct.

4 MR. RUBINSTEIN: It is the staff's viewpoint, and
5 we have slides prepared in our summary to take that into
6 account, that there have been significant changes which
7 contribute to the reduction of risk.

8 These have been partly the basis for our
9 recommendations, our general recommendations before we got
10 into mitigation devices or distributive ignition systems and
11 halon, to go forward with the ice condensers without
12 modification. That was a very important part of our
13 considerations and we will address this in a moment.

14 MR. MADEIROS: I think we should wait for his
15 detailed comments.

16 With that, I do not think the advance notice of
17 rulemaking work to date answers any of the other points you
18 raised at the beginning. If you have further questions, I
19 would be glad to try to answer them.

20 MR. KERR: Mr. Lee.

21 MR. LEE: The way you envision the schedules for
22 the proposed rulemaking hearings, potential hearings that
23 might take place, by that time do you think you may be able
24 to draw upon some of the results that staff might have
25 performed in their risk assessment analysis, probabilistic

1 assessment and so on?

2 MR. MADEIROS: Perhaps. It depends on what
3 direction this takes after we get the advice. There is a
4 predominant view in the Commission that mitigation is the
5 nature of the game today, that we have reached the point of
6 diminishing returns with regard to prevention, and so
7 mitigation is where we ought to put our work.

8 I personally do not believe that, but it may be
9 so. We may get advice from the public, from the regulated
10 industry that convinces us otherwise, that there is much to
11 be done in the area of prevention. If that turns out to be
12 the case, then I do not see the need perhaps as much for
13 these probabilistic analyses and the research and the kinds
14 of things I think you are suggesting as would be if you were
15 to lean more heavily toward the mitigating things like core
16 catchers and controlled filtered vents and those kind of
17 way-out --

18 MR. LEE: I do not understand why you think
19 prevention would require or prevention could use the
20 probabilistic assessment results while mitigation does not.

21 MR. MADEIROS: Maybe I can give you an example,
22 maybe. Let's suppose that the public advice is that we
23 could get a step increase in improvement in safety by
24 requiring strict literal requirements with operating
25 procedures, for example, and that there should be an effort

1 to do that.

2 There does not need to be a lot of probabilistic
3 risk assessment done or research or study to require that
4 operators operate these plants in strict compliance with
5 operating procedures. That could well be suggested by the
6 regulated industry in lieu of something like a controlled
7 filtered vented scheme.

8 I am not saying that they will, but it could be.
9 And from my point of view that is a very strong need in this
10 business, but that is another subject.

11 MR. LEE: I thought a certain amount of operator
12 actions could perhaps be factored into probabilistic risk
13 assessment analyses.

14 MR. MADEIROS: I don't think so. I don't think
15 probabilistic risk assessment will ever come to grips with
16 the operator.

17 MR. LEE: Even on a relative comparative risk
18 analysis basis.

19 MR. KERR: I don't think you are going to convince
20 him. He does not believe in it.

21 (Laughter.)

22 MR. LEE: I was not trying to convince him about
23 that, but what I was interested in was really whether you
24 would be able to at least utilize, or do you plan on
25 utilizing some of the results that might be available?

1 MR. RUBINSTEIN: May I help in this regard? The
2 staff looks to probabilistic risk assessment to be a very
3 complementary and powerful tool to our deterministic
4 analysis in safety. In the particular instance where Dr.
5 Kerr was talking about in his introduction, for example, if
6 we identified a scenario either through failure modes and
7 effects analysis or very high risk scenario to event tree or
8 fault tree, we would want to go ahead and fix that in the
9 plant, and in this way we would couple deterministic and
10 probabilistic.

11 This is why when we talk about scenarios which
12 would lead to severe core damage -- and we try to identify
13 them -- if we identify those with significant contributions
14 to risk, our first step would be to fix them and reduce the
15 risks.

16 MR. KERR: I think it is fair to say that the
17 non-PAS part of the staff looks on risk assessment somewhat
18 like one should look upon perfume: it is okay to smell it
19 but you should not swallow it.

20 (Laughter.)

21 MR. SIEGEL: In one of your slides you emphasized
22 that you are providing the public an opportunity to advise.
23 I am really concerned about how the public is identified in
24 your mind. It certainly needs to be larger than the
25 industry and the Union of Concerned Scientists. I would be

1 very disappointed if those were the only responses you
2 received.

3 MR. MADEIROS: I guess I would, too. What we do
4 to make sure we don't have that kind of disappointment is
5 first of all you publish it in the Federal Register and hope
6 people see it there.

7 MR. SIEGEL: That is a rather vague and really
8 hopeless hope.

9 MR. MADEIROS: In a lot of respects it is. But
10 what we do in addition, then, and the advance notice of
11 rulemaking states this, is we prepare a long list for
12 selected mailings. We try to get in everybody that is a
13 licensee, potential licensee. We try to get the architect
14 engineer. We try to get all public interest groups. We try
15 to get universities.

16 We try to get on this list of hundreds of people
17 anyone that has expressed an interest in the past in areas
18 to do with nuclear power. They are recorded in a special
19 group in the Commission that keeps track of those kinds of
20 things.

21 MR. SIEGEL: I really hope it would be something
22 more than an invitation to comment, but in some way appoint
23 a delegation of responsibility that organizations, for
24 example, like certain identified professional engineering
25 societies --

1 MR. MADEIROS: We cannot demand the American
2 Nuclear Society will comment.

3 MR. SIEGEL: You cannot demand, but you can make a
4 pointed request to the current president that he has a
5 responsibility to get on the ball and do something about it,
6 or other engineering organizations, different groups than
7 are the convention active responders, so you really involve
8 the public.

9 MR. MADEIROS: We will try. That is the best I
10 can tell you. We will accept any suggestions for
11 improvement there as well. We personally suggest that these
12 technical society meetings, that they get copies of these
13 things and read them and comment to us. We announce to
14 maybe scores of people at some of these meetings that have
15 not seen the Federal Register notice. We have a special
16 list that I mentioned.

17 We use every means that we have been able to think
18 about so far, and we will accept any additional help we can
19 get in that area.

20 If there are no more questions, that is all I
21 have, gentlemen.

22 MR. KERR: Thank you, Mr. Madeiros.

23 Although the schedule does not call for it at this
24 point, I am going to declare a ten-minute break.

25 (Recess.)

1 MR. KERR: We will reconvene and continue with the
2 subject of licensing efforts, and the name I have associated
3 with that is Mr. Shapaker.

4 MR. SHAPAKER: Yes.

5 MR. KERR: How close did I come in my
6 pronunciation?

7 MR. SHAPAKER: Very good. My name is Jim
8 Shapaker. I am with the Containment Systems Branch. I want
9 to discuss briefly the status of the design and testing of
10 the distributed ignition system at the Sequoyah Plant.

11 (Slide)

12 It is felt that after studying various concepts,
13 that TVA feels the distributed ignition system is a very
14 promising concept for improving hydrogen control capability
15 in the event of a degraded core accident.

16 The purpose of the system is to mitigate the
17 consequences of the hydrogen release to the containment
18 under degraded core accident conditions by inducing
19 controlled burns, preferably in the lower compartment, to
20 permit active and passive heat removal systems to dissipate
21 the combustion energy and thereby maintain the containment
22 response within the containment structural design capability.

23 Now, associated with this effort is design,
24 testing and analytical efforts, and I will talk briefly
25 about the design of the system and the testing that is going

1 on, and the analytical efforts will be discussed
2 subsequently.

3 (Slide)

4 TVA has a short-term program and a long-term
5 program, and their short-term program is concerned with the
6 distributed ignition system. There are essentially three
7 developmental phases that they have identified. There is an
8 interim phase which deals with the system installation and
9 operation and testing, and I understand the system has been
10 installed.

11 However, before the system is made operable, prior
12 Commission approval is needed. With respect to the system
13 design itself, the system as it now exists is not a safety
14 grade system, and it consists of 30 glow plugs that are
15 distributed throughout the containment. There are eighteen
16 in the lower compartment, five in the lower plenum of the
17 ice condenser, four in the upper plenum of the ice
18 condenser, and three in the upper compartment.

19 The glow plug that is currently being tested is
20 the GMAC 7-G diesel engine glow plug. The system will
21 utilize standby lighting circuits, and these circuits are of
22 seismic design. They are powered from the emergency buses,
23 and as a result, emergency diesel generator power is
24 available to them and the system will be remote manually
25 controlled from the auxiliary building.

1 The standby lighting cabinet is about 150 feet
2 from the main control room, and they have proposed emergency
3 operating instructions which will instruct an operator to
4 initiate the system in response to the automatic actuation
5 of any safeguards equipment.

6 As a result, the decision to activate the system
7 will not be left up to the operator.

8 (Slide)

9 I have a picture of the Sequoyah containment. I
10 just wanted to give you a feel for the relative locations of
11 these igniters. On the right are given the elevations at
12 which they are located, and the number beneath is the number
13 of glow plugs at each elevation. As you can see, there are
14 14 in the lower compartment region. At the 731-foot
15 elevation, there are four more in the lower compartment.

16 The remaining five are located in the lower plenum
17 of the ice condenser. There are four in the upper plenum of
18 the ice condenser and three in the dome area.

19 (Slide)

20 This just briefly is how the electrical circuit
21 looks. The glow plug device is represented by this here
22 (indicating), and there is a transformer which sets down the
23 voltage from 120 volts to 14 volts, and the glow plug is
24 shown there.

25 MR. LEE: Excuse me. How large are the glow plugs

1 physically?

2 MR. SHAPAKER: The glow plug itself, probably
3 about that long (indicating).

4 MR. LEE: The size does not matter.

5 MR. SHAPAKER: This is what the box looks like
6 that contains the transformer.

7 (Slide).

8 It is about a foot long by six inches wide, eight
9 inches high.

10 MR STREHLOW: Is that designed to be explosion
11 proof, the box itself?

12 MR. SHAPAKER: I don't know.

13 MR STREHLOW: Because is you want the glow plugs
14 to work twice --

15 (Laughter.)

16 MR. STREHLOW: -- the box better be explosion
17 proof or maybe it will crush.

18 MR. BUTLER: In response to that, I think the
19 objective of the glow plug, of course, is to ignite a
20 combustion rather than any severe detonation. Now, the
21 boxes, of course, are designed in a rather rugged fashion.
22 It is a strong box. We hope to have a sample of that box
23 available next Thursday to show to the committee.

24 MR STREHLOW: If the box is crushed by the
25 external explosion, you are in trouble for the next

1 ignition. It should be tested to be strong enough to
2 withstand that sort of pressure.

3 MR. KERR: I think that if that box is crushed by
4 the initial explosion, then we are in trouble, period.

5 MR. RUBINSTEIN: The point of the glow plugs is to
6 initiate a controlled combustion as opposed to detonation.

7 MR. STREHLOW: I am talking about a pressurizer,
8 25 psi, which might crush a weak box.

9 MR. SIEGEL: Do you have any indication where the
10 glow plugs are located with respect to the specific places
11 where one might expect the gas to be released? I mean you
12 have them shown in a rather general volume. They must be
13 much more specifically located than that.

14 MR. SHAPAKER: Yes. TVA does have some more
15 detailed slides where they are specifically located at each
16 elevation. They are located around the reactor vessel within
17 the lower compartments here.

18 (Slide)

19 And they are also in the annular regions outside
20 the shield wall at each elevation, and the ones at the
21 731-foot elevation are just right at the operating deck but
22 in the openings as they go up into the steam generator dog
23 houses. And, of course, the ones in the lower plenum would
24 see any hydrogen sweeping through as it was forced out
25 through the ice condenser.

1 MR. SIEGEL: Does the reactor primarily boundary
2 itself have any new vent on it, say in the upper regions,
3 near which a vent could possibly be opened and near which
4 several igniters specifically could be located?

5 MR. SHAPAKER: There will be a reactor coolant
6 system vent. I don't know exactly --

7 MR. RUBINSTEIN: That will vent into the
8 containment. But TVA will go into some greater detail this
9 afternoon and has Vu-graphs prepared on the distribution of
10 igniters and some of these other design details.

11 MR. SIEGEL: Good.

12 MR. ETHERINGTON: It is true, I suppose, that the
13 igniters will not become effective until you have 5 percent
14 of hydrogen, and therefore you will have an explosion, is
15 that right?

16 MR. SHAPAKER: I am not sure I understand. If
17 there is an indication of an accident as a result of some
18 automatic actuation of various equipment, the operator is
19 instructed to initiate the system. Therefore, it will be
20 ready. And if there is then a hydrogen release --

21 MR. ETHERINGTON: There will be an explosion. The
22 nitrogen will not just burn because you have these things
23 on. It has to get up to the combustion --

24 MR. RUBINSTEIN: The probability --

25 MR. SHAPAKER: There is a concentration at which

1 they will become effective.

2 MR. RUBINSTEIN: We have a graph of the
3 flammability and detonation limits in terms of composition
4 of the air, steam, and hydrogen, and 5 percent --

5 MR. ETHERINGTON: We are all familiar with that.
6 I wanted to make sure there was not any misconception.
7 There is going to be an explosion. That is right, isn't it?

8 MR. KERR: I think there is at least one
9 misconception. I don't know on which side at this point,
10 but I don't think Mr. Rubinstein thinks there will
11 necessarily be an explosion.

12 MR. SIEGEL: I would like to hear the point
13 discussed a little more. I hear words like control burn,
14 detonation, explosion. What are we talking about?

15 MR. STREHLOW: If I could comment, you would expect
16 the glow plug to oxidize hydrogen near its surface at all
17 times no matter what the concentration of hydrogen is, but
18 that would be so slow that you would never rid of the
19 hydrogen by that process. You would expect that when you
20 get to about the 4 percent limit, you would start getting
21 upward propagating flames.

22 You might end up just having a continuous 4
23 percent burn with small pressure rises associated with it at
24 that point. Hydrogen burn is very funny at the lean limit.
25 When it burns upward, it does not burn everything

1 completely.

2 It only burns downward at about 9 percent, then it
3 burns everything completely, both upward or downward. But
4 at 4 percent hydrogen, 5 percent, 6 percent, you might only
5 burn 10 percent or 20 percent of the hydrogen in your
6 apparatus in one burn. The flame will propagate to the
7 ceiling and go out. It won't propagate laterally outward.
8 It does not do that.

9 So that if you have the glow plugs on
10 continuously, you have to expect a slow burn to occur at 4
11 to 5 percent hydrogen wherever the concentration was at that
12 level. And when the concentration gets back to that level,
13 it would do the same thing again.

14 MR. SIEGEL: Meanwhile, you could have other
15 volumes where the hydrogen concentration was rising into the
16 detonability limit.

17 MR STREHLOW: That is a possibility. You might
18 have a very high hydrogen concentration at the top of the
19 reactor in the dome and have very low concentrations down
20 below; but if you have glow plugs up in the dome, you are
21 taking care of that, too, because the hydrogen concentration
22 will go up to 5 percent and you will start getting burning.

23 MR. LEE: But I thought you said you will not see
24 much of a lateral propagation.

25 MR STREHLOW: Not between 4 or 6 or 7 percent.

1 MR. LEE: So you would see what you term
2 controlled burn around the --

3 MR STREHLOW: They are propagating up.

4 MR. LEE: Vertically around the location.

5 MR STREHLOW: It might grow a little bit in size.

6 MR. LEE: That is why I was interested in the size
7 of the glow plugs and so on.

8 MR. RUBINSTEIN: We have some specific insight to
9 the Sequoyah Plant and mixing and combustion. I think Mr.
10 Tinkler can address some of these questions.

11 MR. TINKLER: I would like to say that we will
12 address burning at various concentrations a little later on
13 when we talk about the analytical work. There is some
14 concern over the definition of explosion here. That seems to
15 be one of the problems. If you could define relative flame
16 velocities that you are talking about, then we could
17 probably confirm whether we agree with your statement or not.

18 MR STREHLOW: If you are talking about just flame
19 propagation through the system, then the rate of pressure
20 rises as a function of a normal burning velocity in the
21 system. The large systems have very slow pressurizers like
22 Three Mile Island, only 10 seconds or so, and that was with
23 a moderate flame velocity, I imagine.

24 MR. TINKLER: Do you consider that an explosion?

25 MR STREHLOW: I call that an explosion, yes.

1 MR. TINKLER: In that event, then, we are in
2 effect calculating similar types of transients, and the term
3 "explosion" can be used for a burn where it takes 10
4 seconds, for example.

5 MR STREHLOW: In my opinion, that is true.

6 MR. TINKLER: And we will provide results of
7 calculations where burns occur in 10 seconds.

8 MR STREHLOW: You have to be careful between the
9 difference between an explosion and a detonation.

10 MR. TINKLER: You don't mean detonation when you
11 say explosion.

12 MR STREHLOW: No, not necessarily.

13 MR. TINKLER: You mean a relatively quick pressure
14 rise.

15 MR STREHLOW: Yes.

16 MR. TINKLER: And I think the igniter will be
17 tested within a vessel that will be -- where the hydrogen
18 will be burned at varying concentrations and some of the
19 concentrations will be high enough so that you will see
20 rapid pressure rises.

21 MR STREHLOW: If you do this in a small vessel,
22 you will be able to tell what pressure you see in a large
23 vessel for that same concentration of hydrogen.

24 MR. TINKLER: Like I say, I can only respond that
25 we will be testing hydrogen, various concentrations.

1 MR STREHLOW: What size vessel?

2 MR. TINKLER: Jim will talk about this in a little
3 more detail. It is on the order of 100 cubic feet, one of
4 the vessels.

5 MR STREHLOW: Okay.

6 MR. TINKLER: But we are testing at various
7 concentrations and we would expect that we would get flame
8 propagation in some cases upward and sideways and spherical.

9 MR STREHLOW: Yes.

10 MR. KERR: Are we now all of one mind? Good.
11 Please continue.

12 MR. SHAPAKER: With respect to the glow plug
13 testing, they have conducted some initial tests at the TVA
14 Singleton Laboratory.

15 (Slide)

16 The purpose of these initial tests were to screen
17 alternative igniters and demonstrate that igniters can be
18 used to detonate hydrogen. Some of the characteristics of
19 the plugs that they found were that they determined the glow
20 plug's temperature was a function of applied voltage, and at
21 14 volts they got a temperature of about 1720 degrees
22 Fahrenheit.

23 MR. KERR: You say that they did experiments to
24 determine that glow plugs could ignite hydrogen? Was there
25 some doubt about that?

1 MR. SHAPAKER: Well, since they were not used for
2 this purpose before, they wanted to run some initial tests
3 and confirm in their own mind and give themselves some
4 confidence that they are on the right path.

5 MR. KERR: Sounds reasonable.

6 MR. SHAPAKER: Okay. They also wanted to
7 determine the durability of the glow plugs, and they have
8 operated a specimen successfully at 1720 degrees Fahrenheit
9 for six days, and this particular glow plug was used in the
10 hydrogen burn test. They also wanted to determine the
11 reliability of the glow plug as an ignition source, and they
12 introduced different concentrations of hydrogen and they did
13 achieve ignition in dry air at a 12 volume percent and 7
14 volume percent hydrogen mixture.

15 They wanted to determine the completion of
16 hydrogen burn, and at the 12 volume percent mixture they
17 essentially got 100 percent combustion. The volume of this
18 vessel was about .04 cubic feet. It was a very small vessel.

19 (Slide)

20 MR. KERR: .04 cubic feet?

21 MR. SHAPAKER: Yes.

22 MR. KERR: How did they get the glow plug inside?

23 (Laughter.)

24 MR. SHAPAKER: Well --

25 MR. KERR: Inyway, they did. Okay.

1 MR. SHAPAKER: They have now a subsequent testing
2 program that will continue to be scoping in nature and
3 verify that the glow plugs are a useful concept and to
4 better understand the combustion phenomena using these glow
5 plugs in different environments.

6 The purpose of the test will be to demonstrate
7 that the igniter will initiate a volumetric burn for
8 different environmental conditions, and the different
9 environmental conditions that they are going to select are
10 including a steam environment, and they want to define the
11 hydrogen concentration range over which a volumetric burn
12 will be initiated with these different environmental
13 conditions.

14 In general, the acceptance criteria for the tests
15 are that they are looking for consistency in the data,
16 confirmation of theoretical predictions, and reliable
17 ignition at the high concentrations, around 12 volume
18 percent with essentially complete combustion.

19 In other words, the purpose of these tests will be
20 such that there is no surprise, that they fully understand
21 what phenomena are going on. The test facility that will be
22 used is a spherical vessel. It is about six feet in
23 diameter. It will be heated electrically. It will be
24 equipped with an internal fan to promote mixing and create a
25 draft at the igniter surface.

1 The purpose of creating this draft is to determine
2 what effect the flow of a mixture past the igniter would
3 have on its reliability and effectiveness.

4 MR. LEE: I have a question. I do not understand
5 your rationale for suggesting 12 percent by volume of the
6 hydrogen concentration in your acceptance criteria. I
7 thought you were talking about the control burn which
8 produced low pressure so you would like, perhaps, to somehow
9 initiate ignition at lower concentrations of hydrogen, I
10 thought.

11 MR. SHAPAKER: With the glow plugs on, it will
12 begin to ignite the mixture or whatever will support a flame.

13 MR. GREGORY: Did you test that? Have you tested
14 the glow plug action as the hydrogen concentration rises
15 from zero up through the lower flammable limit? It seems to
16 me that you tested getting it up to 12 percent and then
17 putting on the glow plug. That is not representative of
18 what will happen inside the containment.

19 MR. KERR: Is TVA going to discuss the tests?

20 VOICE: Yes, sir, we are.

21 MR. KERR: Okay.

22 MR. LEE: Could you also perhaps comment, if you
23 would, at this stage on the status of the theoretical models
24 for predicting this type of phenomenon and how it compared,
25 for example, with the often quoted triangular diagram and so

1 on?

2 MR. SHAPAKER: Well, this is what they are -- by
3 conducting these tests, they are fully expecting that their
4 burns will occur. In keeping with that turnery diagram --

5 MR. BUTLER: I think we will defer to a later
6 presentation by Mr. Bowman when he describes the test
7 program that he planned at the Livermore test facility.

8 MR. KERR: Please continue.

9 MR. SHAPAKER: Okay. They will be monitoring the
10 atmospheric temperature and surface temperature of the
11 vessel, and they will have their vessel pressure also, and
12 it will have hydrogen and oxygen analyzers to measure the
13 pre and post-burn concentrations.

14 (Slide)

15 Just briefly, the first series of tests TVA is
16 planning looks like this. You can see the various
17 temperature, pressure and hydrogen concentrations that they
18 want to achieve in the vessel initially, and the degree of
19 mixing that they want to induce inside the vessel.

20 (Slide)

21 Based on the outcome of these tests, they are
22 contemplating the need for possible further testing, and
23 also what they have in mind is to measure the temperature
24 response of the different components and steel or concrete
25 objects.

1 Now, with respect to measuring the temperature
2 response of components, they are planning to submit to the
3 staff in early September plans for their testing to
4 determine the effects of a hydrogen burn on essential
5 components, and the Environmental Qualification Branch is
6 taking an active part in this effort in scoping out this
7 program.

8 MR. ETHERINGTON: What is meant by saturation
9 temperature on the previous slide? What temperature, what
10 pressure is that?

11 MR. SHAPAKER: I guess that would be the
12 saturation temperature corresponding to the total pressure.

13 MR. ETHERINGTON: What pressure?

14 MR. SHAPAKER: The total pressure in the vessel.

15 MR. ETHERINGTON: The pressure that develops in
16 the burn?

17 MR. SHAPAKER: No, for the initial conditions.

18 MR. ETHERINGTON: I see. All right.

19 MR. SIEGEL: The pressure is in the next column.

20 MR. ETHERINGTON: Oh, yes, I see. I did not see
21 that.

22 MR. SHAPAKER: They also want to investigate ways
23 to simulate a spray droplet entrainment in the atmosphere so
24 they can determine to what extent a spray system would have
25 on the effectiveness of the igniters. The schedule for the

1 testing is not quite as shown on this slide.

2 (Slide)

3 We just got additional information last night.
4 The facility is being prepared and they do have to install
5 some heaters and instrumentation yet; and as a result, the
6 test series number 1 schedules is from 9/8 to 9/12, and
7 further tests being planned for 9/15 to 9/28, with a test
8 evaluation from 9/15 to 9/19, and a test report on 10/1.

9 Presumably this test report would not include the
10 results of further tests. It would probably only include the
11 results of their series number 1.

12 (Slide)

13 There is also a Phase II to this program, and this
14 is sort of an improvement phase. These improvements will
15 proceed along with their long-term degraded core task force
16 program, and some of the improvements that will be
17 implemented include providing individual control from the
18 main control room for each igniter, installing additional
19 hydrogen and oxygen monitors to guide the operators so they
20 know more precisely what is going on in the containment
21 within the various compartments.

22 They want to install a plant computer to warn of
23 hydrogen concentrations reaching the detonation limit. The
24 backup diesel power supply will continue to be included in
25 any design improvements. And they want to determine the

1 environmental qualification of the distributed ignition
2 system components. And also they want to determine the
3 effects of the hydrogen burn environment on other components.

4 These two items will be covered in part in this
5 next series of testing in the larger vessel for this
6 vessel. It is large enough to physically contain the glow
7 plug rig that was described earlier.

8 They are also going to be looking into alternate
9 and/or additional igniter locations based on a better
10 understanding of the characteristics of combustion and
11 whatever they find out from these tests.

12 They are also contemplating installing hydrite
13 converters near the reactor vessel vent, PORV discharge and
14 air return fans.

15 MR. LAWROSKI: What is a hydride converter?

16 MR. SHAPAKER: I believe it consists of certain
17 metals that have an affinity for forming a hydride at an
18 elevated temperature as hydrogen would pass over it. So it
19 would serve as an initial getter for any hydrogen being
20 released from the reactor coolant system. They are also
21 contemplating --

22 MR. SIEGEL: Will we hear more about those later?

23 MR. SHAPAKER: Possibly from TVA when they discuss
24 their longer-term efforts.

25 MR. SIEGEL: In that same paragraph, what is the

1 reactor vessel vent?

2 MR. SHAPAKER: This would be the vent prescribed
3 by the TMI Action Plan, I presume.

4 MR. SIEGEL: Doesn't that exist now?

5 MR. SHAPAKER: I don't know if that one is
6 installed or not.

7 MR. KERR: Is there a TVA representative who can
8 respond to the question of whether the vent now exists?

9 VOICE: Dr. Kerr, our engineering design people
10 just came in. If you could repeat the question we could
11 probably get an answer.

12 MR. KERR: I should give them a chance at least to
13 sit down, I suppose.

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1 The question is whether the reactor vessel vent as
2 prescribed by the Lessons Learned at TMI now exists and is
3 ready to be operable on Sequoyah.

4 VOICE: It has not been -- the system has not been
5 installed.

6 MR. KERR: It has not been installed, but what --

7 VOICE: It has not been installed, but the drawing
8 has been issued. The hardware is not in the plant itself.

9 MR. KERR: It is in process. Thank you.

10 Do you want to add something, Mr. Rubinstein?

11 MR. RUBINSTEIN: This was originally slated for
12 January 1, 1981, but it is a requirement which is coupled
13 with the inadequate core cooling and reactor vessel level
14 requirement, and both of these items have been slipping,
15 partially because of material unavailability, for the
16 pressure drop systems, for the inadequate core cooling
17 reactor vessel, and partially because the staff has only
18 started to undertake a review with the Westinghouse owners'
19 group on the reactor coolant system vent itself.

20 So I visualize this requirement being, in effect,
21 late in the year, perhaps December of 1981.

22 MR. KERR: Thank you.

23 Mr. Lawroski.

24 MR. LAWROSKI: Are we going to hear more about
25 hydride converters this afternoon?

1 MR. KERR: Yes.

2 Please continue.

3 MR. SHAPAKER: As a result of trying to improve
4 the diagnostic capability within the containment, additional
5 containment penetrations may have to be installed, so that
6 is another item in their improvement phase.

7 (Slide).

8 The third phase consists of final modifications
9 that will be implemented at the completion of TVA's
10 long-term degraded core task force program. Of course,
11 these remain to be defined as their study continues. The
12 degraded core task force program is a two-year effort, and
13 they have identified some major tasks, and they plan to
14 continue with their study of controlled ignition.

15 They also plan to study ultimate concepts, and
16 they mentioned the use of halon suppressants or possibly
17 fogging systems, and they plan to following along the
18 rulemaking proceeding with their own efforts in risk
19 assessment and core behavior, with hydrogen generation and
20 transport, and with their studies of hydrogen burning and
21 analytical efforts and containment response, and also the
22 structural response of the containment.

23 MR. LEE: A question. Are they going to have both
24 controlled ignition system and some suppressant system, or
25 choose one or the other?

1 MR. SHAPAKER: They will probably have to choose
2 one or the other. They are continuing to study alternate
3 concepts in the event that their current study of the
4 distributed ignition system does not prove to be that
5 desirable.

6 MR. BUTLER: Let me add to that, Jim. They have
7 an interim distributed ignition system in place now. They
8 hope to have it operational with staff approval shortly.
9 However, there are continuing studies under way not only to
10 study further the glow plug efficacy, but also to study
11 other alternative approaches.

12 If the alternative approaches prove better, they
13 will be used in lieu of or even in addition to. It is
14 subject to future determination. Right now they are just
15 going forward with the burning system, the igniters. They
16 might later add or substitute another system.

17 MR. LEE: Those are two conflicting systems. One
18 is to suppress burning; the other is to initiate burning.
19 That is why I am somewhat curious at this stage.

20 MR. BUTLER: You certainly would not use the
21 igniter with the halon system, that is true.

22 MR. LEE: Okay.

23 (Slide)

24 MR. SHAPAKER: This concludes my discussion of the
25 distributed ignition system. I have two other slides on

1 their hydrogen sampling system and mixing systems. I don't
2 know if you want to hear that right now or not.

3 MR. KERR: I think we probably do not unless you
4 think it contributes.

5 Let me ask you if you would be willing to comment
6 on Mr. Gilinsky's question, which is, if I may paraphrase:
7 Is the staff persuaded of the effectiveness of distributed
8 igniters in ice condenser containments?

9 MR. SHAPAKER: Well, I guess I can say it looks
10 promising; but this next series of testing will really
11 provide the most useful information.

12 MR. KERR: You are not yet persuaded but you are
13 willing to be persuaded if the evidence is persuasive.

14 (Laughter.)

15 MR. SHAPAKER: We are keeping an open mind.

16 MR. RUBINSTEIN: The preliminary evidence is quite
17 persuasive, particularly from the very limited Singleton
18 Laboratory work. And we look forward to reaching our full
19 conclusion in December, after the submittal.

20 MR. KERR: Is the staff going to try to determine
21 if the igniters can keep pressures below design pressures if
22 one is faced with TMI quantities of hydrogen?

23 MR. RUBINSTEIN: The staff will discuss analyses
24 this afternoon which deal with the initial hydrogen content
25 and the final hydrogen content as determined experimentally

1 from our work as it is placed in both the MARCUM code, and
2 TVA will address it as it is in the CLASIX code, and
3 sequential burns, which deal with both pressure and
4 temperature.

5 MR. KERR: Thank you.

6 MR. GREGORY: Is the concept you have in mind to
7 keep the igniters running continuously at all times, or only
8 to activate them in the event of an accident?

9 MR. SHAPAKER: They will not be powered at all
10 times. They will only be actuated when certain events occur
11 which cause the automatic initiation of different safety
12 equipment. Then the operating instructions will be for an
13 operator to then immediately go and actuate this system.

14 MR. GREGORY: I would be interested to learn what
15 criteria, what triggers the decision to turn on the igniters
16 and what safeguards are being put in to make sure that they
17 get turned on before you possibly have built up to the
18 detonable limit.

19 MR. KERR: Can the TVA representative hear the
20 question? What I really want to know is whether you plan to
21 address this question in your presentation.

22 MR. LAI: I can address it now. TVA has an
23 emergency procedure written, but of course we are not
24 activating it, pending the NRC staff approval to use the
25 system. The procedure called for in the emergency operating

1 instruction is that you would activate this system
2 immediately after the ECCS system is activated and the
3 emergency core cooling system is initiated.

4 The next step is to make sure the system is on and
5 functional.

6 MR. KERR: Does that respond to your question, Mr.
7 Gregory?

8 MR. GREGORY: Yes. I would have some concern that
9 the controlled ignition system could in certain events cause
10 a detonation if it was not activated soon enough after a
11 hydrogen release. I think some attention ought to be paid
12 to that.

13 MR. KERR: Thank you.

14 MR. SEALE: I assume you are also going to
15 determine that it is really on, and not that the switch has
16 just been flipped.

17 MR. KERR: I will accept that as a gratuitous
18 remark.

19 (Laughter.)

20 MR. KERR: But an appropriate one.

21 Mr. Etherington.

22 MR. ETHERINGTON: The first Vu-graph referred to
23 this as the most promising interim concept. To me it looks
24 like the most promising concept that we have seen. If this
25 is approved and installed, is TVA looking at the possibility

1 of devising an interim -- a periodic testing of individual
2 plugs?

3 MR. LAI: The answer is yes.

4 MR. ETHERINGTON: Good.

5 MR. KERR: With hydrogen, I assume.

6 MR. ETHERINGTON: Yes.

7 MR. LAI: We did not plan to test the igniter in
8 place with hydrogen.

9 (Laughter.)

10 MR. ETHERINGTON: I realize you are not going to
11 fill the containment with hydrogen. Surely you can put a
12 box around it or something.

13 MR. LAI: No, sir. We do not plan to test the
14 igniter inside the plant with any hydrogen. It would be
15 off-site testing.

16 MR. KERR: Are there other questions?

17 (No response.)

18 MR. KERR: Thank you, Mr. Shapaker.

19 I show a presentation by Mr. Tinkler at this point.

20 MR. TINKLER: I am Charles Tinkler. I work in the
21 Containment Systems Branch.

22 This portion of the presentation will deal with
23 the overall NRC efforts to address those technical questions
24 which are related to the issue of hydrogen control.

25 (Slide)

1 Those efforts are primarily in three areas:
2 short-term efforts principally directed towards an
3 evaluation of igniters for degraded core accidents. These
4 efforts are principally being monitored by NRR. The
5 long-term efforts on hydrogen control are being conducted
6 through the Office of Research, and their research will deal
7 with both degraded and melted core accidents.

8 In addition, earlier this year the staff began its
9 studies on the Zion/Indian Point plants in which hydrogen
10 control was also a topic.

11 (Slide)

12 MR. KERR: Mr. Schott.

13 MR. SCHOTT: For general information or
14 orientation, are the Zion and Indian Point things MARK III's
15 or are they ice houses or what are they?

16 MR. TINKLER: They are large drive PWRs.

17 Under the short-term efforts we have a rather
18 large program to evaluate the igniters. We have obtained
19 technical assistance from a number of labs in order to
20 complete our evaluation in a timely manner. Livermore is
21 participating in the igniter evaluation by performing tests
22 on the igniters that have been proposed by TVA for their
23 interim distributed ignition system.

24 Livermore will also conduct a survey of hydrogen
25 detection in plants currently in operation. The igniter

1 tests, as has been talked about previously -- and the tests
2 will be described in much more detail by Mr. Bowman of
3 Livermore later on, but principally the idea is to determine
4 the effectiveness of the igniters in varying atmosphere
5 mixtures, the ability to ignite the hydrogen in varying
6 concentrations of steam.

7 Columbus-Battelle is participating in this
8 program, and they are performing analysis using the MARCH
9 code for the Sequoyah containment in an attempt to
10 analytically determine the effect of use of igniters.
11 Sandia will perform an evaluation of igniters and
12 principally focus on alternative systems in the case that
13 igniters do not eventually end up being very promising.

14 MR. LEE: Mr. Tinkler, what kind of result do you
15 expect to get out of MARCH analysis for the controlled
16 ignition systems and so on?

17 MR. TINKLER: We think the MARCH code represents a
18 substantial improvement over calculations which have been
19 provided, for example, in SECY papers. It provides the
20 capability to do a transient calculation of hydrogen release
21 to the containment and the transient burning of hydrogen.

22 MR. LEE: How about the controlled burning
23 igniters? Are you trying to model that?

24 MR. TINKLER: The data that we learn from tests
25 will be used as the basis for parameter studies and analysis

1 -- the ignition set points, quantity of hydrogen that is
2 burned, for example.

3 Now, we feel that we need a data base from which
4 to do our analysis.

5 MR. LEE: So it is from your experimental tests
6 you expect to get --

7 MR. TINKLER: How much of the hydrogen will burn
8 for a given concentration, and those parameters are input
9 options into the codes that we are talking about. We are
10 able to initiate hydrogen combustion and the burndown limit
11 in the codes so that we can model the heat addition to the
12 atmosphere.

13 MR. LEE: I am still not sure the information that
14 you have has to really come from your experimental efforts.

15 MR. TINKLER: The experimental efforts will
16 verify. It will also provide the basis for future analyses
17 to determine how effective they are in mitigating the
18 consequences of various accidents.

19 MR. LEE: I am particularly interested in what
20 kind of volume a particular igniter could perhaps control,
21 what kind of a volume of hydrogen and what rate and so on.
22 Unless you have this information to put into the MARCH code
23 or any other code, you cannot expect to get any meaningful
24 information.

25 MR. TINKLER: I think the information we will get

1 from the test program will be sufficient to model the
2 calculation in the codes that we have available. Like I
3 say, the tests are designed to determine whether the
4 hydrogen will ignite at various concentrations. That will
5 be bounded for analytical purposes.

6 MR. KERR: The MARCH code does not itself model
7 the microscopic behavior of the burn. If I understood Mr.
8 Lee's questions, they had to do with flame propagation that
9 might occur at low hydrogen concentration. Suppose you get
10 an ignition. How much burn do you get? Where does it go?
11 Don't you need that sort of information to have a good
12 understanding of the energy input from a burn, for example,
13 and how extensive the burn is; or is that too detailed?
14 Maybe you only need a broad brush.

15 MR. TINKLER: I don't think that we need to know
16 the precise definition of the flame front in order to
17 perform the bulk of the containment calculations. We are
18 principally concerned with how much hydrogen is burned and
19 how much energy addition to the containment atmosphere
20 results.

21 MR. KERR: But the MARCH code does not tell you
22 how much hydrogen is burned, does it? You have to input
23 that.

24 MR. TINKLER: Yes.

25 MR. KERR: If you input it, how do you know what

1 to input unless you know something about the behavior of the
2 igniters in a lean mixture of hydrogen?

3 MR. TINKLER: We intend to test the igniters in a
4 lean hydrogen mixture. You have seen data on that at 8 and
5 12 percent, okay? But the igniters will be tested in leaner
6 atmospheres.

7 MR. LEE: The data base is still quite limited.

8 MR. TINKLER: Yes. That is why we are doing the
9 testing.

10 MR. LEE: The six foot diameter test volume still
11 is quite limited, in my opinion, compared to the large
12 containment.

13 MR. TINKLER: One of the test vessels will be
14 approximately ten cubic feet. One of the test vessels will
15 be approximately slightly more than 100 cubic feet.

16 MR. LEE: That is a six foot diameter vessel you
17 are talking about, and I don't know how many glow plugs you
18 have for that 100 cubic foot volume. I don't know whether
19 it makes sense. I have no idea whatsoever at this stage.
20 And to rely on -- oh, I am not saying that you should not
21 try to exercise the MARCH code or anything like that, but I
22 am just curious what kind of information you envision
23 getting out of such an exercise of the computational program.

24 MR. TINKLER: One more try here. I think the
25 numbers from the test data -- you know, we cannot guarantee

1 them to be precise for the plant applications; but with a
2 little judgment we can bound those numbers in order to
3 determine the effects of hydrogen burning at various
4 concentrations. That is the best we could hope to do in any
5 situation where we have conflicts between test data and
6 actual configurations.

7 MR. LEE: That might well be perhaps the best you
8 can do, but I am still interested in what general
9 information you can get out of running the MARCH code that
10 you don't already have from your test program.

11 MR. TINKLER: The calculations you have seen in
12 the SECY papers are hand calculations for an adiabatic heat
13 addition. The MARCH code is a substantial improvement over
14 that type of calculation. Unless we do a transient
15 calculation which takes credit for heat removal mechanisms
16 in the containment, we won't see benefit from any mitigation
17 devices.

18 MR. LEE: I agree with your assessment to a
19 certain extent. As far as hydrogen generators are
20 concerned, perhaps what the MARCH code predicts might be
21 somewhat more accurate or more reliable. As far as the
22 efficacy of the distributed igniters is concerned, I am not
23 sure what additional information a code like the MARCH
24 program can provide.

25 MR. TINKLER: I gave it my best shot.

1 MR. BUTLER: It might be helpful. I think the
2 thrust of your prior question, Mr. Lee, was more in the area
3 of distribution of hydrogen concentrations before you turn
4 on the glow plug, and the concern presumably is you might
5 have appropriate distributions away from the glow plug but
6 no hydrogen near the glow plug. Is that the thrust of your
7 question?

8 MR. LEE: Yes, certainly. That would be my
9 concern.

10 MR. BUTLER: It might be helpful here to indicate
11 that there are return air fans that assist quite a bit to
12 mixing the air in the upper and lower compartments. These
13 are 40,000 CFM fans that give you one complete air change in
14 the lower compartment every five minutes.

15 Now, it is true that the hydrogen addition could
16 conceivably be a point source, and we are trying to engage a
17 consultant to examine the heterogeneity question of hydrogen
18 distributions by postulating point sources and determine how
19 that hydrogen would provide various concentrations as a
20 function of space away from the point source.

21 MR. LEE: That is just one part of my concern,
22 perhaps. The second part was related to what Dr. Strehlow
23 mentioned earlier about the lateral versus vertical
24 propagation of flame, which I am willing to wait and learn a
25 little bit more about a little bit later. But if that type

1 of phenomenon cannot be simulated in a code like the MARCH
2 code and the code is not certified to be a design-type of
3 code -- and I understand the model is not really detailed at
4 this stage -- what additional information can we get out of
5 the code?

6 I still have that concern.

7 VOICE: I agree with most of your comments, Mr.
8 Lee. The MARCH code certainly does not do the microscopic
9 studies. But I think maybe the point should be made that
10 given certain amounts of information on the efficacy or
11 effectiveness of the igniters under different conditions,
12 things that the MARCH code can tell you are what is the
13 effects of that type of ignition in a large system as
14 opposed to the idealized experiments.

15 MR. LEE: How can you tell? I am curious, if you
16 could educate me a little.

17 VOICE: I would just as soon not take too much time
18 at the moment since I am scheduled to be on in a little
19 while. Perhaps I can address that point a little later.
20 But basically what the MARCH code will tell you is, given
21 ignition under these conditions, what is the effect in the
22 full-scale system.

23 MR. STREHLOW: Mr. Butler, your comment raised
24 another question in my mind. With those return fans, they
25 are ducted? The air is ducted by those fans through ducts

1 to circulate?

2 MR. TINKLER: Yes, it is.

3 MR STREHLOW: They can be very dangerous as far as
4 transition is concerned. If you do a burn, a long duct with
5 a fan that is making nice, turbulent flow can be extremely
6 dangerous in causing transition to detonation. This may not
7 happen at 4 percent. It probably would not. But if you got
8 up to 8 percent and those fans are running, you might have a
9 real good time with a detonation going down those ducts.

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1 MR. KERR: Please continue.

2 MR. TINKLER: I don't believe I mentioned I&D
3 Associates, but they are involved, and they will participate
4 with Livermore in an evaluation of data. Another lab not
5 listed here is Brookhaven, which has participated in the
6 Zion/Indian Point studies and has done work on hydrogen.

7 MR. SEALE: I have not heard anything yet, and I
8 was just interested in this one point; and that is, you are
9 putting some igniters in or at least you are examining the
10 possibilities. There are a lot of igniters that are
11 apparently already in there, switches, and maybe even those
12 fan motors you were talking about. And I wonder if there is
13 any evaluation of other ignition sources in such systems?

14 It seems to me that there is very likely going to
15 be ignition whether you put those igniters in there or not.
16 Maybe not as often or as regularly, if you will, or
17 continuously, but there will be some ignition.

18 MR. TINKLER: The idea of adding the ignition
19 system is to more or less improve the reliability which you
20 feel will occur, okay? It is recognized there are random
21 ignition sources inside containment.

22 MR. SEALE: The reason I brought it up, I guess,
23 is you speak of other things besides ignition as an
24 approach, and the idea that anything else might be
25 incompatible with the ignition; and yet, if there are

1 ignition sources there anyway, you really are going to have
2 to accommodate that problem.

3 MR. TINKLER: That would have to be factored into
4 an evaluation of the other mitigation systems.

5 MR. SEALE: Okay.

6 MR. STREHLOW: I might comment on that point. The
7 glow plugs are probably a very strong source of ignition.
8 There is a large surface area. They are very high
9 temperature. If I remember correctly, Three Mile Island
10 went off at 8 percent or something like that hydrogen.

11 That indicates to me that the ignition sources
12 present are not strong sources, and it turns out when you go
13 towards the limit, the strength of the ignition source has
14 to increase markedly, otherwise you will not get flame
15 propagation. So the installation of glow plugs might be a
16 very good thing to do, because you might induce relatively
17 slow burns over a long period of time which will not cause
18 any appreciable pressure rise in a system.

19 MR. SEALE: If they are not there you still --

20 MR. STREHLOW: We don't know until the experiments
21 are done.

22 MR. KERR: It seems to me we may be converging on
23 the TVA position.

24 (Laughter.)

25 MR. SIEGEL: I have kind of a basic difficulty

1 with this approach. I don't know if this is the right time
2 to ask the question, but let me pose it now.

3 It seems to me the objective of this distributed
4 igniter effort and the test program is to demonstrate that
5 you can safely ignite several hundred kilograms or several
6 tens of kilograms of hydrogen. But it seems to me a better
7 objective would be to say that you can safely handle so many
8 kilograms per second of hydrogen coming from some hopefully
9 plausibly identified sources so you'd never tens or hundreds
10 of kilograms in the containment.

11 I don't see any signs that either the test program
12 or the concept is addressed toward this latter question.

13 MR. TINKLER: In the event that we have to provide
14 some sort of system to mitigate the consequences of a
15 degraded core accident, we cannot avoid the introduction of
16 hydrogen in those quantities.

17 MR. SIEGEL: I agree. You need the other as a
18 backup system. But I would like to see some evidence that
19 you have a pilot light near the place where the flame is
20 likely to start rather than have a light bulb in the kitchen
21 which will blow it up when you have it there.

22 MR. TINKLER: What I would like to point out is
23 this is the interim distributive ignition system. Further
24 review of this system, you know, as yet to take place, and
25 the location and number of igniters is one of the subjects

1 that we intend to review in more detail.

2 MR. SIFGEL: Well, frankly, I guess I have
3 expressed my reservations about this being the preferred
4 interim system.

5 MR. KERR: That is not a question.

6 (Laughter.)

7 MR. TINKLER: Okay. The Sandia short-term
8 program, which is being monitored by the Office of Research,
9 as I said previously, will concentrate on alternative
10 systems. They will look at deliberate ignition systems
11 also; and to take the things they will look at under
12 deliberate ignition, they will look at ignition strategies,
13 whether you want to igniters on and leave them on, or
14 whether you want to have intermittent operation. And they
15 will look into analytical verification of igniter
16 effectiveness to see if they could determine from existing
17 test data at what concentrations you burn hydrogen and so
18 forth.

19 Halon systems, some of the issues to consider are
20 concentrations necessary to prevent deflagration or
21 detonation, the effects of recombiner operation in this
22 halon atmosphere, post-accident handling of the halon
23 atmosphere -- the halon only suppresses the burn; it does
24 not remove the hydrogen -- what would a halon system look
25 like, the design concept. And they would evaluate it as a

1 short-term solutio. to the issue of hydrogen control.

2 Water fog, similar type issues will be
3 investigated -- the design concept, what would a system look
4 like, how much water would have to be supplied to the
5 containment, what droplet size in order to suppress, what is
6 its effectiveness in -- to prevent hydrogen burning, what
7 effects would it have by immediately reducing steam
8 concentrations perhaps and thereby changing the flammability
9 characteristics, what would a long-term recovery look like
10 for a containment with a water fog system, since again
11 presumably this only works as long as it is operating, and
12 what aspects of this would lead you to believe that this is
13 a short-term solution to solving the hydrogen control
14 problem.

15 (Slide.)

16 MR. KERR: What is the significance of the term
17 that appears in two of those short-term feasibility --

18 MR. TINKLER: We believe the issue of hydrogen
19 control will be studied for some time in conjunction with
20 the rulemaking, in conjunction with ongoing studies that the
21 staff --

22 MR. KERR: I am not making my question specific
23 enough. Does that mean you are trying to determine whether
24 it would be feasible to implement those very soon?

25 MR. TINKLER: At Sequoyah.

1 MR. KERR: That is what short-term feasibility
2 means.

3 MR. TINKLER: Yes. Could they reasonably hope to
4 install a water fog system at Sequoyah in this time frame.

5 MR. LEE: Are hydrogen -- are there any systems
6 under study by anybody at the present time?

7 MR. TINKLER: I would say that hydride converters
8 are not -- they are not being strongly considered by the
9 staff at this time. TVA has indicated they will look into
10 this area, but it is a technology about which we do not know
11 a great deal.

12 The ultimate capabilities to observe hydrogen are
13 not well understood.

14 MR. GREGORY: I would suggest that you might
15 consult the people at Brookhaven who have a tremendous
16 background on various hydrides.

17 MR. TINKLER: Okay. Brookhaven has suggested
18 hydride converters in connection with the Zion/Indian Point
19 study. Brookhaven continues to work in that area.

20 MR. GREGORY: I would think, just off the top of
21 my head, that it would be difficult to find a hydriding
22 metal that is tolerant to air, because if you have an
23 inerted containment, it would be relatively easy to pull the
24 hydrogen out with a hydride, but with an air-filled
25 containment, I think the hydriding metals are likely to

1 oxidize.

2 MR. TINKLER: Long-term efforts principally being
3 directed by the Office of Research, the last time, I
4 suppose, the staff reported on this, we indicated that a
5 user's request was being prepared or had been prepared for
6 additional research on control or mitigation of melted core
7 accidents. Since that time a user's request has been
8 issued, and to repeat the objective of it, was to provide
9 technical support for conducting a rulemaking on
10 degraded/coremelted accidents.

11 We had hoped to investigate the filter vented
12 containment system designs and hydrogen control systems for
13 generic types of lightwater reactors, various containment
14 types. We had suggested that the research initially focus on
15 hydrogen controls for the ice condenser in the Mark III
16 containment, since they seem to be more sensitive to
17 hydrogen generation.

18 MR. KERR: Have you given the RES any
19 specifications? I mean, for example, on a filter vented
20 containment design. Have you specified any rough criteria
21 for behavior that these have, or is that part of the
22 research program to establish criteria?

23 MR. TINKLER: Part of the research program is to
24 establish the criteria under which these systems must, you
25 know, operate and what their function will be. We have not

1 specified detailed criteria that we think a filter vented
2 containment system must have.

3 MR. KERR: How about non-detailed criteria? Have
4 you given any --

5 MR. TINKLER: We identify criteria against which
6 we think all the systems should be judged -- cost-benefit
7 sort of items, reliability.

8 MR. KERR: For example, have you specified flow
9 rates that you expect that these may have?

10 MR. TINKLER: No. If we had done that, then we
11 would have identified the accident scenarios.

12 MR. KERR: That is so logical that I must agree
13 with you.

14 (Laughter.)

15 MR. TINKLER: And we have not identified the
16 accident scenarios for which they must operate. Okay. And
17 we discussed this before, so --

18 MR. BUTLER: Let me add to that, Charlie.

19 The matter of filter vented containments has been
20 under study for a couple of years at least, and in
21 conjunction with that work sponsored by research, the NER
22 staff has met with the research people and the Sandia people
23 and expressed its comments with respect to containment
24 systems' needs in regard to the filter vented containment.

25 MR. KERR: One of the reasons I ask is because any

1 good research organization is not dedicated to solving
2 problems. They are dedicat,d to finding new problems. I
3 mean, you know, that is what a researcher does. So unless
4 you give him some idea of the problem you want him to solve,
5 he will not try to solve problems; he will try to find
6 problems.

7 At some point it seems to me somebody is going to
8 have to define what the problem is in order so someone can
9 look for a solution. Maybe it is too early to do that.

10 MR. BUTLER: It is our expectation that we will
11 have frequent meetings with the research people as well as
12 with the laboratory support people. It is indeed a bit too
13 early to do that. We still have to meet with the research
14 people in their response to our user's request.

15 MR. TINKLER: Since that time a draft FSR program
16 plan for severe accident phenomenology and mitigation
17 research has been drafted. The logic of the program is
18 similar to the WASH-1400 reactor safety study and the
19 Zion/Indian Point study.

20 About the most I can say is the program is
21 currently under review, and this draft program also deals
22 with much more than hydrogen control. It deals with broader
23 aspects of the severe accident mitigation.

24 (Slide.)

25 The relationship to Zion/Indian Point studies, as

1 previously mentioned, the Zion/Indian Point studies did
2 consider generation and release of hydrogen, and the severe
3 accident to the containment, and the ultimate combustion of
4 that hydrogen.

5 As a result of the Zion/Indian Point studies,
6 Sandia compared a compendium of hydrogen behavior in a
7 post-accident environment, and there were various studies of
8 mitigation features. These studies were done for the
9 Zion/Indian Point plants, so application of all these
10 studies to the Sequoyah plant are not easily made.

11 The general conclusion from the technology
12 exchange meetings and the bulk of the work was that the
13 large dry containments may survive a hydrogen deflagration
14 for degraded core accidents.

15 You might argue what is large, but it does appear
16 for the dry containments they would be able to survive a
17 hydrogen deflagration simply due to the high volume.

18 MR. KERR: In fact, you could almost say large
19 containments would, couldn't you, whether they were dry or
20 wet?

21 MR. TINKLER: Yes.

22 MR. GREGORY: May I go back to the previous slide
23 with your issuing requests for research, and ask you to --
24 ask what mechanisms are you using to solicit new concepts,
25 new ideas from the research community on methods of either

1 preventing hydrogen release or observing hydrogen or
2 combustion techniques?

3 Are your requests going out in such a form that
4 the universities, the industrial contracting labs may
5 respond, or are you addressing these only to the potential
6 users and the national labs?

7 MR. RUBINSTEIN: We have a representative from
8 research here who can address that question.

9 Mr. Hotzen.

10 MR. HOTZEN: The current work, we are planning to
11 use Sandia Laboratories as our center for the hydrogen
12 work. They in turn may funnel work out to universities or
13 other research laboratories, but they are going to be our
14 center for the hydrogen research work.

15 MR. GREGORY: It occurs to me that there are
16 surely a number of research operations around the country
17 that might have ideas to offer in this respect, but they may
18 not even be aware of this kind of problem.

19 MR. HOTZEN: Yes. We are currently talking about
20 possibly holding a conference at Sandia in December and
21 inviting the broad community to attend that for two
22 reasons: one, to inform them of what Sandia is doing, and
23 second, to find out what each of the organizations are doing.

24 As someone mentioned earlier, the hydrogen problem
25 has come up pretty quick. There are a lot of people running

1 in a lot of different directions, and we need to find out
2 what each other are doing.

3 MR. GREGORY: As a member of the industrial
4 contract research community, we scan the Commerce Business
5 daily for requests for proposals in areas that we think we
6 have technology to offer, and I do not recall having seen
7 anything relating to this advertised to the research
8 community.

9 I would just like to make the comment that you do
10 have a mechanism of getting out to the hungry research
11 public, and it might give you fast response on new ideas if
12 funding would be available.

13 MR. HOTZEN: At the moment the main job -- we have
14 been through the compendium at Sandia which raises a lot of
15 interesting areas where research could be done from the
16 scientific point of view. The next step we have to do is
17 find out where that research would be of interest in the
18 nuclear safety area; and that is the next part of the work
19 we are going to do.

20 After that we will be in a position to get a
21 better handle on the specific experiments that we might
22 advertise for assistance on.

23 MR. KERR: Please proceed, Mr. Tinkler.

24 MR. TINKLER: This concludes my presentation at
25 this time.

1 I would like to introduce Mr. Bowman.

2 MR. KERR: Before you conclude completely, back to
3 the MARCH code, does the staff feel fairly confident of the
4 results or the applicability of the results you are getting
5 from that code, or are you using it because it is the best
6 you have, or what is your attitude toward the results
7 insofar as they apply to, say, Sequoyah and other operating
8 plants?

9 MR. TINKLER: It's the best we have, okay. But we
10 feel that any of the results calculated to date are
11 preliminary in nature, okay. The codes that have not been
12 used for this application --

13 MR. KERR: Is "preliminary" a euphemism for
14 "semi-lousy," or do you mean "early?"

15 (Laughter.)

16 I don't know.

17 MR. TINKLER: If anything, it is a euphemism for a
18 lot better than anything else we can do, and we think they
19 are reasonable calculations. Whether we think they can
20 predict pressures within one psi, for example, we cannot say
21 that with any confidence.

22 MR. KERR: What about 10 psi?

23 MR. TINKLER: I think that is the ballpark we are
24 talking about, how much lower we can get it. On MARCH -- we
25 need to learn a little more about it.

1 MR. KERR: I'm not trying to be critical. I'm
2 just trying to understand how you view the results of the
3 code, because I do not -- I do not know.

4 MR. TINKLER: I can only say that based on my
5 limited exposure to the code, I have to say -- repeat it is
6 a substantial improvement over what we had before. I think
7 the code does a reasonable job of modeling the transient.
8 Whether it would do a better job is debatable at this time;
9 but we think it is valuable in determining the effectiveness
10 of the igniters.

11 We surely need to use something like it, or the
12 CLASIX code which is being used by TVA.

13 MR. RUBINSTEIN: May I add to this? We recognize
14 MARCH and CLASIX as essentially unverified in terms of
15 experimental bases, and we will be doing at Battelle with
16 the MARCH code verification through other codes where
17 possible. And I believe that TVA will discuss to some
18 degree their verification program and their code later this
19 afternoon.

20 MR. KERR: Thank you. Thank you, Mr. Tinkler.

21 MR. TINKLER: As I said, Mr. Bowman will discuss
22 the igniter program at Livermore.

23 MR. BOWMAN: My name is Barry Bowman. I am from
24 the Lawrence Livermore National Laboratory. I am going to
25 give you a brief overview of the hydrogen igniter program

1 that we have going on right now at the laboratory. We are
2 doing this under contract to the NRC, and the overview will
3 cover the following items.

4 (Slide.)

5 Very briefly I will go through the program
6 objective, the status of the hydrogen monitoring device
7 survey, a description of the experimental design and
8 instrumentation, and experimental conditions and brief
9 account of the procedures we intend to use to conduct the
10 experiment, and finally, a program schedule to give you an
11 idea of how soon we intend to get everything done.

12 (Slide.)

13 The objectives of the program are fairly
14 explicit. There are three major objectives. The first of
15 these is to test the thermal igniters proposed for use at
16 Sequoyah by TVA in various mixtures of hydrogen, air and
17 steam.

18 Bill Lowry is the project engineer. He is here
19 today.

20 The second objective is to survey and report on
21 the state of the art methods and devices for hydrogen
22 protection, being conducted by Bill Lye, and then to conduct
23 a literature survey of the hydrogen combustion, particularly
24 that conducted by the U.S. Bureau of Mines.

25 To date there are no publications or data

1 available for hydrogen combustion in substantial amounts of
2 steam, so this represents a new work from that standpoint.

3 The igniter itself -- I brought one along to give
4 you an idea of the size of what we are dealing with.

5 MR. SIEGEL: When you say test the thermal
6 igniters, what does that mean -- to see if they light the
7 gas, or to see how the phenomenon of gas combustion goes on
8 in some chamber which has been initiated by the igniters?
9 What does to test thermal igniters mean?

10 MR. BOWMAN: It implies that we will confirm
11 whether igniters will or will not function for various
12 combinations of air, hydrogen and steam. How we detect the
13 burn will be from a detection of pressure rise in the
14 vessel. I will go into that in a little while.

15 MR. SIEGEL: It is literally a test of the
16 igniters, as you say, not a test of the phenomenon of
17 hydrogen combustion as initiated by these igniters.

18 MR. BOWMAN: That is right. Before I cover that,
19 here is where we are in the hydrogen monitoring systems.

20 (Slide.)

21 This is not an all-inclusive list. There are four
22 basic types that we have uncovered -- catalytic device
23 manufactured by these companies: a semi-conductor device; a
24 volume expansion device; and electrochemical oxidation type
25 device.

1 The catalytic device essentially detects the
2 hydrogen by increasing temperature of the catalyst. The
3 semi-conductor device detects it by a change in the current
4 of the conductor. The volume expansion device is a
5 mechanical device; as the hydrogen adsorbs it actually
6 increases in size. And the electrochemical oxidation is
7 basically an electrolyte absorption process where the
8 hydrogen increases or decreases the output of a little
9 battery.

10 (Slide.)

11 The general characteristic of the available
12 systems, they depend on diffusion of the hydrogen. They are
13 fairly slow in response. "Slow" here means 3 to 30 seconds
14 for the first two devices that I mentioned, catalytic and
15 semi-conductor, and from 7 to -- let's see -- 30 to 7
16 minutes for the volume expansion device. Electrochemical
17 oxidation is also the --

18 MR. KERR: What sort of time response does the NRC
19 think should be available?

20 MR. BOWMAN: Why don't you ask the NRC?

21 MR. KERR: They have not told you?

22 MR. BOWMAN: No.

23 MR. KERR: All right. I will ask them later.

24 MR. BOWMAN: The indications were that was
25 adequate, 30 to 30 seconds, but I will --

1 MR. GREGORY: What is the response time of the
2 electrochemical?

3 MR. BOWMAN: Three to 30 seconds, somewhere in
4 that time frame. It's a matter of seconds, not minutes.
5 The slowest is the volume expansion device. Some are
6 subject to contamination if you do have other hydrocarbons
7 in the system, or if you have a fair amount of water in this
8 system, then they also pick up that. The pores on a
9 catalytic detector can actually be plugged by condensation
10 of water.

11 Some are limited in the range of hydrogen
12 detection. Of the devices I mentioned, the catalytic and
13 semi-conductor devices have a range of about .02 percent by
14 volume of hydrogen up to about what is called the lower
15 explosion limit or sometimes referred to as the lower
16 flammability limit of 4 percent or so.

17 The volume expansion device and electrochemical
18 oxidation devices don't have any apparent upper limit. The
19 lower limit for the volume expansion device is on the order
20 of a tenth of a percent, and the electrochemical oxidation
21 is roughly half that, .05 percent.

22 MR. SIEGEL: Are any of those devices based on the
23 thermal conductivity of the gas?

24 MR. BOWMAN: Of the gas? I am not sure.

25 MR. SIEGEL: That is sort of the classical way of

1 measuring hydrogen.

2 MR. BOWMAN: I suspect the semi-conductor device
3 refers to that. I don't know the details. I know the least
4 about that one, so you probably know more than I do.

5 MR. RUBINSTEIN: I believe that type of conductor
6 is in Sequoyah. TVA can answer that.

7 MR. KERR: Is it the view of Livermore that
8 existing devices are adequate for the purpose for which they
9 may be needed in a nuclear plant?

10 MR. BOWMAN: We don't know. It depends on what
11 you mean by "detection" in a nuclear plant.

12 Okay. Right at this point what we are trying to
13 determine, first of all, what is available commercially,
14 what range of --

15 MR. KERR: I guess I don't see what difference it
16 makes what is available commercially unless you know what
17 you want. I mean, you could go to a catalog and find out
18 what is available commercially. I assumed that Livermore
19 was looking for something that would do a job, and it had
20 some idea what the job was.

21 MR. BOWMAN: Let me describe the job a little
22 bit. If you are going to detect hydrogen concentrations
23 that are burnable by these thermal igniters, you need
24 something that will detect normally 4 percent. If you want
25 an opinion, the catalytic converter, if that is the upper

1 bound, if you get concentrations that exceed that, you are
2 going to pick that device almost every time.

3 So in the selection of a device we are not
4 involved in the selection, but we could be involved in
5 making recommendations as to which devices are or are not
6 usable within a plant, and we will do that.

7 MR. KERR: But you are not yet prepared to do that.

8 MR. BOWMAN: Not yet, no. We started this whole
9 process August 1. This is sort of a status report as to
10 where we are.

11 MR. KERR: Thank you.

12 MR. BOWMAN: A description of the experimental
13 apparatus, I have some pictures that will show you better
14 what the system is like. This is a schematic, but the
15 orientation and basic configuration are correct.

16 (Slide.)

17 This is the hydrogen supply, air supply and steam
18 generator that we are going to use to provide various
19 concentrations within this vessel. This is a 10 cubic foot
20 diameter -- test chamber. It is an ASME steel pressure
21 vessel. It is normally 20 inches in diameter and about 60
22 inches in length. We have access points at the side and the
23 top which are about eight inches, and at this end through
24 which protrudes a mixing fan which is about four inches in
25 diameter.

1 Okay. The instrumentation and the glow plug
2 itself are in the system at these approximate locations.

3 (Slide.)

4 We have several thermocouples installed throughout
5 the vessel at the bottom, top, and sides. We have them
6 sticking into the vessel, normally an inch or so, and
7 attached to the internal wall and the outside wall to
8 measure the heat transfer through it.

9 We are also measuring the ambient conditions. We
10 have static pressure transducers located on the end flange
11 and right now on the top flange, and dynamic quartz-type
12 pressure transducers which are flush-mounted to the sidewall
13 of the containment.

14 These two will detect any substantial strength
15 pressure wave that is propagated through the system. We
16 initiate combustion at a higher concentration. We have
17 taken gas samples periodically on the ends, both above and
18 below the center line, to determine the homogeneity of the
19 mixture within the containment vessel at the top, and then
20 periodic samples that are mounted to this flange plate on
21 the side.

22 The glow plug itself is on an extendable rod that
23 sits down inside the vessel through this upper flange. It
24 can be moved up and down so that we can locate it anywhere
25 vertically within the pressure vessel that we like.

1 There is also a differential pressure transducer
2 which just measures the hydrostatic head of the condensate
3 that we are going to get through the process of conducting
4 the experiment.

5 MR. MARK: In the slide before and what you have
6 said, you are prepared to mix hydrogen, air and steam. As
7 soon as one of these igniters has burned a little and the
8 flame is out and the hydrogen keeps coming, do you no longer
9 have normal air to look at?

10 You have nitrogen contaminant probably more than
11 steam. Is there any plan on the part of research or
12 Livermore or NER to add nitrogen fraction contaminate to the
13 system, because it is certainly going to affect the ignition
14 the second or third time that you call on it.

15 MR. BOWMAN: Not specifically, no. We are going
16 to rely on that to tell us what is in the volume at any
17 particular time prior to ignition, immediately after
18 ignition, and any subsequent firings of the glow plug. So
19 you are right. It will be --

20 MR. MARK: If you are going to be subsequently
21 igniting the used air, then you will get these data points
22 free.

23 MR. BOWMAN: Right. We are getting a lot of data
24 points free by this experiment which I think you will see.

25 MR. STREHLOW: Are you going to change the

1 elevation of the glow plug?

2 MR. BOWMAN: We had not planned on changing that.
3 Right now it is sticking down, okay. We can change the
4 orientation of both the glow plug and the entire system if
5 we want.

6 MR. STREHLOW: And are you planning to put the
7 gases in an then turn the glow plug on?

8 MR. BOWMAN: We plan on precharging the vessel,
9 all right, and I will go through that in a little bit.

10 MR. STREHLOW: Okay, sure.

11 MR. BOWMAN: Any other questions on that? I have
12 some photographs that we took just before we left. This
13 will give you an indication of a couple of things -- what it
14 looks like and where we are in assembly.

15 (Slide.)

16 This is the vessel itself. You can see the
17 various penetrations for the pressure transducers at the top
18 and thermocouples. The dynamic pressure transducers come
19 out here and here (Indicating). This is the side eight-inch
20 flange, and then the flange on the end (Indicating).

21 (Slide.)

22 The igniter itself is mounted on this rod that we
23 can control its vertical position (Indicating). The igniter
24 that I passed around is sticking right up here on the end.
25 Everything that goes into the vessel itself is from this

1 type thread forward to the left (Indicating).

2 (Slide.)

3 The mixer itself is an air-driven fan that pokes
4 in the end wall, and the gas sampling bottles we are just
5 starting to assemble. They are normally 70 cubic centimeter
6 bottles with an isolation valve here. They remotely operate
7 a solenoid valve. We have to do all of the experimentation
8 remotely in a bunker just in case we make a mistake.

9 (Slide.)

10 Some of the characteristics of the glow plug that
11 we have arrived at are shown on the vu-graph. We applied
12 various voltages. Incidentally, I did not mention it, but
13 the glow plug itself is instrumented for current voltage and
14 surface temperature.

15 (Slide.)

16 The first four tests are normally with 13 to
17 almost 16 volts pressed across the glow plug itself.
18 Surface temperatures are on the order of 1770 to 1800
19 degrees. This is the RMS voltage. In rise times it was the
20 order to 20 to 30 seconds to come up to that peak
21 temperature.

22 In the subsequent tests below this blue line you
23 will see there is some inconsistency in the data, and this
24 may reflect a breakdown in the plugs, since the voltage was
25 impressed across it. That was well above the design value

1 of 12 volts.

2 (Slide.)

3 An experimental region of interest that we are
4 going to be looking at, here is the famous turnery diagram.
5 I'd like to point out a couple of things. These are not
6 really very strict lines. They are more or less broad bands
7 across here, and this lower detonation line is subject to
8 how the tests were actually conducted, because it is
9 dependent upon the --

10 MR. MARK: Is that a typo on the slide about the
11 hydrogen fractions?

12 MR. BOWMAN: Let me look here. No.

13 MR. MARK: Four-tenths to nine-tenths sounds
14 pretty high.

15 MR. BOWMAN: That is not percent by volume.

16 VOICE: That is a typo.

17 MR. MARK: Under the experimental region by your
18 right hand, maybe a zero would have cleared it up.

19 MR. BOWMAN: I am losing you somewhere. Yes, I am
20 sorry. It is a typo. It is .04 to .09. Yes, in this
21 region here. So 4 to 9 percent by volume.

22 MR. MARK: Do you know what your tank will stand?
23 I mean, can you fill it with two hydrogens and one oxygen
24 and light it and still have the tank there?

25 MR. BOWMAN: We think so, but we are conducting

1 tests in a bunker.

2 (Laughter.)

3 The design pressure on the tank is 200 psi. I
4 think detonation calculations show that we could generate a
5 pressure wave when the pressure is about 5 to 6 atmospheres,
6 if I recall, overpressure.

7 MR. SCHOTT: It is more than that.

8 MR. BOWMAN: I will check that for sure. That was
9 in the presence of a substantial amount of water vapor.

10 MR. STREHLOW: It sounds awfully low.

11 MR. BOWMAN: A relatively low concentration of
12 hydrogen.

13 MR. KERR: Well, they do have the bunker.

14 MR. STREHLOW: Yes..

15 MR. BOWMAN: That is right.

16 MR. SEALE: It may be a 50 cubic foot volume.

17 (Laughter.)

18 MR. BOWMAN: Experimental procedures, we are
19 relying on some of our pretest thermodynamic calculations.

20 (Slide.)

21 We are going to meter the air, precharge the tank
22 with air, and then put in the hydrogen by allowing it to
23 reach equilibrium with a precharged bottle, precharged
24 volume of hydrogen, so we do not inadvertently get more than
25 we want and create this situation.

1 We then inject saturated steam from the steam
2 generator to a precalculated pressure and temperature that
3 is outside of these predicted combustion limits, allow the
4 steam to condense on the side walls, and periodically fire
5 the glow plug, sampling gases just before and just after
6 each firing. This gives us a measure of what the
7 condensation rate on the walls is by post-determination of
8 the water vapor in the samples.

9 After we get an ignition which will detect the
10 pressure in the system, then we will grab a final sample and
11 decide to go on to subsequent firings.

12 (Slide.)

13 MR. LEE: There is no way of measuring the
14 pressure around the glow plug during the -- during the
15 firing process or right after on continuous basis?

16 MR. BOWMAN: No.

17 MR. LEE: So you have to get the pressure
18 difference before and after.

19 MR. BOWMAN: That is right. We are looking at the
20 equilibrium pressure for the tank before and after, and then
21 the quartz impact gauges essentially may give us a peak
22 pressure.

23 MR. LEE: How much time do you feel that you have
24 to give after the firing -- burn was initiated before you
25 reach that pressure in the tank?

1 MR. BOWMAN: I am not certain, okay. That is one
2 of the things that we will find out. The complexity that
3 contributes to that is continuing condensation within the
4 tank of revaporized water vapor; so we don't know.

5 MR. LEE: Presumably from small tests set up like
6 this you could go to a larger containment system, or is that
7 too much to ask?

8 MR. BOWMAN: I don't know, and that is a question
9 I think that comes up: what is the applicability of these
10 tests to an extrapolation to a larger test system or to a
11 containment itself. And Roger --

12 MR. STREHLOW: If you have a small vessel, in
13 other words, almost a round vessel, the extrapolation is
14 straightforward. The maximum rate of pressure rise times
15 the volume to the third power is a constant, so if you
16 determine it in this vessel -- and I think this vessel is
17 large enough -- you should be able to get a realistic
18 pressure time curve. Assuming that there is no flame
19 acceleration due to structural interactions and things like
20 that, you will get the lowest value for pressure rise with
21 time.

22 MR. LEE: How about the fact of the condensation
23 on the walls and so on? Is that going to affect anything?

24 MR. STREHLOW: That is what this big matrix code
25 takes care of. What was the name of the code, the one we

1 were arguing about it?

2 MR. KERR: The MARCH code.

3 MR. STREHLOW: That takes care of the condensation
4 effects in the main vessel. It handles all the losses to
5 the load itself, which you would not do in an adiabatic
6 calculation.

7 MR. LEE: But I have -- I have a concern about
8 what a code can do in lieu of experimental data.

9 MR. KERR: But your answer is that the experiment
10 does not handle that problem.

11 MR. STREHLOW: The experiment will handle the
12 problem. He will get condensation.

13 MR. KERR: It does not handle the extrapolation
14 problem.

15 MR. STREHLOW: The code would -- well, I don't
16 know how to answer that. The main thing that happens, I
17 think, if you have losses is that the maximum pressure is
18 different, that the rate of pressure rise will still be
19 about the same probably, and that can be done by the
20 extrapolation of these experiments here by the maximum
21 pressure. If you do -- you are probably going to do an
22 adiabatic calculation for maximum pressure, aren't you, on
23 this system for complete combustion?

24 MR. BOWMAN: That is right.

25 MR. STREHLOW: You will see a volume that is less

1 than that during the experiment, and you can attribute that
2 to partial combustion, and you can get samples so you can
3 tell how much did burn. In that case you will mix the gas
4 thoroughly before you draw that final grab sample, won't you?

5 MR. BOWMAN: We will turn the fan on prior to
6 combustion.

7 MR. STREHLOW: But I mean after -- you should
8 recirculate it afterwards because you might draw a false
9 sample if you have only partial combustion.

10 MR. SIEGEL: On your chart about experimental
11 procedures, you show that you inject saturated steam, and
12 then you allow the steam to condense slowly, and then
13 periodically the glow plug is fired. Do you know the steam
14 conditions at the time that combustion is occurring? Are
15 those steam conditions varying during the combustion process?

16 MR. BOWMAN: During the combustion process you are
17 going to generate water as a combustion product, and you
18 will also revaporize some of the condensate; so we will not
19 know precisely until we post-process our gas samples what we
20 had at the time just prior to combustion.

21 MR. SIEGEL: Will you know the steam conditions at
22 the time of initiation of combustion?

23 MR. BOWMAN: We will know from the pressure and
24 temperature measurements that we made, so we can infer about
25 where we are on the diagram, but not exactly.

1 MR. LEE: Will you also be able to somehow infer
2 the pressure pulse that might occur based on the pressure
3 that you measure later on?

4 MR. BOWMAN: I'm not sure I understand your
5 question.

6 MR. LEE: The pressure on the glow plug when the
7 maximum temperature is reached, it may be different from the
8 particular pressure that --

9 MR. STREHLOW: There are no waves in this system.
10 The pressure is spatially constant. It is temporally
11 changing, but it is spatially constant. The velocity of
12 sound is 1,000 feet per second, and that thing is only 10
13 feet long. It takes a hundredth of a second, so there is no
14 pressure gradient in the system at all, not with this kind
15 of condition.

16 MR. LEE: Not even with the condensation and so on
17 taking place on the wall?

18 MR. STREHLOW: No way. You will change the
19 pressure with time, but you will not have spatial gradient.
20 So if you measure pressure at the wall, and you measure
21 accurately, you are measuring the pressure at the glow plug.

22 MR. LEE: No, but I am interested in the temporal
23 variation of pressure.

24 MR. STREHLOW: That you will get with those gauges
25 and --

1 MR. LEE: The special --

2 MR. STREHLOW: A megacycle of pressure
3 fluctuation, but you won't see that in this system because
4 the pressure rise is slow.

5 MR. SIEGEL: What is the time scale for the
6 pressure rise? Is it short compared to condensation or
7 evaporation times? I am still concerned about the steam
8 presence.

9 MR. BOWMAN: I think it is short compared to
10 condensation rate. The sort of temperatures we see
11 externally at the test site, the condensation rates are such
12 that we could condense from 70 percent steam down to 30
13 percent steam in the course of about an hour. Okay.
14 Pressure increase within the vessel itself occurred much
15 more rapidly, a matter of seconds, seconds or subseconds.

16 MR. SCHOTT: Can we expand a bit on the
17 temperature control that you do or do not have on the test
18 vessel? It is unheated, uninsulated, but the injection of
19 steam will allow its temperature to go up after?

20 MR. BOWMAN: That is right. The injection of
21 steam will preheat the vessel, all right, and then we just
22 let it cool down. And the control of the experiment really
23 is, you might say, almost uncontrolled, but our control
24 takes place as the steam condenses and the volume of
25 hydrogen comes up naturally.

1 We have some calculations that predict what that
2 volume fraction decrease in steam should be. Some of the
3 tests we are going to conduct in the shakedown period, of
4 course, and we can verify that. There is a mechanism to get
5 the water vapor concentration up to some appreciable value
6 by injecting enough steam to heat the metal and still permit
7 an appreciable water vapor in event ambient condition.

8 MR. BOWMAN: Absolutely.

9 MR. SCHOTT: Seventy degrees or something like
10 that.

11 MR. BOWMAN: Sure.

12 MR. SCHOTT: Another clarification on the tests
13 that have already been done and are being reported here in
14 this extensive graph. On the glow plug temperature I can
15 see how it is easy enough to monitor current voltage
16 continuously to these glow plugs in operation.

17 What instrumentation is laid to these righthand
18 column peak temperature figures?

19 MR. BOWMAN: We had a thermocouple welded to the
20 surface of the glow plug itself.

21 MR. GREGORY: Your experimental procedure is
22 obviously designed to get a matrix of data on combustion
23 under these conditions, but I am concerned it does not seem
24 to include a set of conditions that represent what would
25 happen in the real case of hydrogen in a reactor containment.

1 My concept here is that the reactor containment is
2 filled with a mixture of air and steam, and then you have a
3 steady injection of hydrogen coming from the core. And the
4 glow plugs will be on before you reach the flammable limit,
5 and you are hoping that you will get a lightoff of the
6 hydrogen, a partial burn, but then the hydrogen keeps on
7 coming, and so you get another one and another. And you
8 want to make sure that the glow plug is going to ignite
9 every time before you reach detonation.

10 You are not running that kind of test in your
11 experiment. You are working with a constant ratio of oxygen
12 to hydrogen, and just letting the steam partial pressure
13 drop.

14 MR. BOWMAN: There is a more fundamental question
15 to be answered, all right, and which this whole program
16 really addresses: will the glow plug function if the
17 concentration around the glow plug itself reaches such and
18 such combination of concentrations, all right?

19 So the answer to the more extensive question that
20 you ask is almost implicit in what you get out of this in
21 terms of the glow plug will function under these
22 concentrations locally. All right. If it will function
23 subsequently at different concentrations, if we measure
24 those, then we do have confirmation yes, that it will or
25 not. If the concentration, you know, changes into some

1 range where we have not conducted an experiment, then the
2 question is again open.

3 MR. GREGORY: I think I would be happy if I could
4 see experimental results that show the -- in effect,
5 determine the lower flammability limit of hydrogen in the
6 presence of a glow plug and in the presence of a lot of
7 steam. You are really not determining the flammability
8 limit. You are just determining whether the glow plug will
9 ignite the hydrogen when it is about that limit.

10 MR. BOWMAN: It really is a test of the
11 functionality of this particular device rather than a
12 reproduction of the flammability limit curves.

13 MR. KERR: Mr. Gregory has a point in that if it
14 functions the way you hope it will, it will first light off
15 when one has reached something that turns out experimentally
16 to be the lower flammability of hydrogen in that situation.
17 Maybe you don't have to determine what that is to make
18 certain it will operate. That is precisely the way it will
19 operate, it seems to me.

20 MR. BOWMAN: That is exactly what we will
21 determine.

22 MR. STREHLOW: If this glow plug had a rapid rise
23 time for temperature, I would be suspicious of doing the
24 experiment this way, but it takes 30 seconds for the thing
25 to get hot, and under those circumstances you have a

1 convective field around that glow plug which actually
2 reproduces the same convective field that you would expect
3 to have with a hot plug being contacted by a gas which has
4 the proper composition to burn.

5 So you have a quasi-steady experiment here, and it
6 should model what is happening in a reactor as the
7 concentration is changing. If this thing had a rise time of
8 a tenth of a second, then I would say no, you have to watch
9 your step because you are not duplicating conditions; but
10 this thing is going to have a nicely defined convective flow
11 field around it at all times in this experiment.

12 Now, one thing it should look at it is when it
13 goes off relative to where this 30 second rise is. If it
14 goes off after ten seconds, then you are igniting under
15 worse conditions than the steady state conditions.

16 MR. BOWMAN: We will be monitoring that during the
17 course of the experiment. We know previous experience of
18 data available indicates what the temperature threshold is
19 or even the energy threshold and --

20 MR. SCHOTT: Recognizing that presently planned
21 and financially provided for and manpower provided for
22 experiments at Lawrence Livermore are of the preset mixture
23 switch on the glow plug rather than the continuous injection
24 of fuel type, it still makes -- there is an important aspect
25 to be investigated by the present mode of experimentation,

1 to work with not only accumulated steam but also accumulated
2 nitrogen in place of air.

3 Carson Mark has made this point earlier, but I
4 want to hammer on it. One will not have conditions with a
5 lot of steam. One could, but the function of the deliberate
6 ignition system and hardware for burns subsequent to a first
7 burn is important, and so the reduced oxygen situation can
8 and should be investigated by this mode of investigation.

9 MR. BOWMAN: You are suggesting we fire the glow
10 plug after initial indications of ignition, correct?

11 MR. SCHOTT: That is one way to go about it. The
12 more deliberate way is to start from scratch with extra
13 nitrogen.

14 MR. KERR: Other questions or comments? Please
15 continue.

16 MR. BOWMAN: The experimental schedule, to give
17 you an idea of what we're doing and where we are, is we
18 started this schedule in August. We expect to be complete
19 with at least this series of tests by October 30.

20 The design and assembly, you have already seen
21 some of the pictures of where we are there. We are right on
22 schedule or even a little bit ahead. We anticipate we will
23 start checkout about the middle of September, as indicated.

24 We provide a quick report on the data acquired by
25 October 30 and then a final report some three months

1 subsequent to that, and also the conceptual design for a
2 spray system in this particular configuration, if it is
3 deemed necessary and important.

4 We will also report on a review of the hydrogen
5 detection systems and the literature survey by about the
6 October 30 deadline date.

7 Some of the problems we have already considered
8 that have already come up is what is the influence of glow
9 plug location on combustion -- this refers to the tendency
10 for the stuff to propagate upwards in the presence of
11 concentrations below eight percent -- whether or not a
12 turbulent atmosphere in the tank is desirable; whether you
13 want to leave the fan on or shut it off; the revaporization
14 of the condensed water adversely affecting sampling -- we
15 have already addressed that -- the homogeneity of the
16 mixture, we get an indication of that by sampling at various
17 points within the containment vessel itself. And then the
18 last question, of course, which is one of prime importance,
19 how this can be extrapolated for use in representing a real
20 containment.

21 What we intend to deliver to the NRC is a report
22 of the hydrogen detection monitoring methods, the results of
23 our combustion literature survey, and for the various
24 concentrations of hydrogen, air and steam which should be
25 able to tell you, okay, whether or not the glow plug will

1 function -- function meaning will it light the mixture up.

2 We do that by providing the pre- and post-ignition
3 temperatures and the concentration of the major
4 constituents, and we will make an attempt that if some high
5 pressure wave does propagate through the system -- finally,
6 the energy input to the igniter and its temperature
7 (Inaudible).

8 That is all I have to say.

9 MR. SIEGEL: Will there be any thermal
10 instrumentation on the glow plug during the course of the
11 test?

12 MR. BOWMAN: Just a thermocouple, yes.

13 MR. SIEGEL: You ought to be able to tell from the
14 thermocouple whether ignition is occurring. That will tell
15 you instantaneously whether ignition is occurring.

16 MR. BOWMAN: With the response time of the
17 thermocouple. We probably pick it up on the pressure
18 indicators sooner.

19 MR. KERR: As I look at the schedule and after
20 conferring with Mr. Tinkler, who assures me that the next
21 presentation will require at least an hour, it seems to me
22 this is a sensible time to break for lunch. So I am going
23 to break and reconvene at 1:30.

24 (Whereupon, at 12:30 p.m., the meeting was
25 recessed for lunch, to be reconvened at 1:30 p.m., the same

1 day, Thursday, August 28, 1980.)

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AFTERNOON SESSION

(1:30 p.m.)

1
2
3 MR. KERR: It has been called to my attention that
4 there are people who think it is rather warm in this room.
5 And the conclusion, after careful investigation, is that you
6 people are making too much heat. So I want to urge that
7 everybody think cool for the rest of the day, and we will
8 therefore not overload the air conditioning system quite as
9 much.

10 If air conditioning only means cooling to you, you
11 have a limited perspective. It also can mean heating. And
12 I think, I think that's what's happening to ours.

13 Mr. Tinkler and Mr. Jenz, perhaps --

14 MR. TINKLER: No, Mr. Cybulskis.

15 MR. KERR: Mr. Cybulskis, okay. You lead off? Or
16 does he lead?

17 MR. TINKLER: I'll lead off.

18 MR. KERR: Okay. And he, he comes in as a relief
19 pitcher.

20 (Pause)

21 MR. TINKLER: This portion of the presentation
22 will deal with the analytical efforts which have taken place
23 to date in order to evaluate the role of the igniter system
24 and mitigating the consequences of a degraded core accident,
25 specifically, for the Sequoyah plant.

1 As a point of reference, we're showing a slide
2 which outlines the capability of the Sequoyah containment,
3 as calculated by various organizations. TVA calculated a
4 yield of 33 and an ultimate strength for the containment of
5 approximately 43 psig. Ames (?) calculated a slightly
6 higher yield pressure. R&D Associates calculated a slightly
7 lower yield pressure, using minimum code values for the
8 yield strength of steel. And the Office of Research
9 calculated a similar number for the yield pressure as
10 compared to TVA.

11 As I say, this, this slide, is presented as a
12 point of comparison for the presentations that will proceed
13 after this.

14 To date, the bulk of the analysis has been done by
15 Battelle/Columbus, using the MARCM code, and by TVA through
16 OES, using the CLASIX code.

17 At this time, Mr. Cybulskis, from
18 Battelle/Columbus, will provide a discussion of the results
19 obtained for the Sequoyah plant using the MARCM code.

20 After that, I'll return and discuss the CLASIX
21 results.

22 (Pause)

23 MR. CYBULSKIS: Good afternoon. My name is Peter
24 Cybulskis. I am with Battelle's Columbus Laboratories. And
25 this afternoon I would like to describe briefly the very

1 short study on the behavior of the Sequoyah containment
2 assuming intentional ignition, using the MARCH code.

3 I had described the MARCH code to the Committee,
4 July 2nd, in Los Angeles. So I really hadn't intended to
5 get back to that point. But since there were a number of
6 questions raised, let me very briefly say at least a few
7 words about what MARCH does as intended to do.

8 The initials "MARCH" stand for Meltdown Accident
9 Response Characteristics. MARCH is an outgrowth of the work
10 that we did for the reactor safety study, in which there
11 were a limited number of computer pools developed to treat
12 the meltdown accident. Since that time we have done
13 substantial additional work, which has led to the
14 development of MARCH.

15 What MARCH tries to do is the thermal hydraulics
16 of a core meltdown accident, starting from the initiation of
17 the event, through core heat-up, core meltdown, pressure
18 vessel melt-through, and interaction of the core debris with
19 the concrete ultimately. Throughout the course of the core
20 heat-up history, MARCH is continuously coupled to the
21 containment, where it keeps track of the containment
22 pressure, temperature, composition of the atmosphere, what
23 have you, taking into account static heat sinks and those
24 engineered safety features that might be applicable, which
25 may include spallies (?), containment coolers, ice

1 condenser, et cetera.

2 It is a systems code more so than a detailed code
3 in any one area. As I said before, it attempts to treat the
4 entire meltdown accident scenario. It does have a lot of
5 approximations in it, for various reasons, sometimes because
6 the knowledge is not adequate to describe things exactly; or
7 in some cases we have included bounding matters, in other
8 cases we have included approximations for convenience; and
9 at other times there are phenomena that we feel are not
10 important, so we have not chosen to treat them in great
11 detail.

12 In the limited time here, I don't want to get into
13 great detail on MARCH, unless there are some specific
14 questions.

15 In terms of the treatment of the containment,
16 MARCH is a -- has a capability of treating the containment
17 as a multi-volume system. However, and the inherent
18 assumption in MARCH is that the -- most of the pressure
19 rises within the containment take place slowly and that the
20 pressures can equilibrate three compartments. There are
21 some exceptions to that approximation, particularly when we
22 get into the areas of containment plumbing and the
23 containment failure. But, in general, the assumption is
24 made that the inputs into the containment and the pressure
25 rises are slow enough that the pressure within the entire

1 containment structure goes up and down essentially uniformly.

2 With those brief words on MARCH, let me get on to
3 the work related to Sequoyah.

4 MR. KERR: Excuse me, Mr. Cybulskis, would you be
5 willing to comment briefly, at least, on where those
6 assumptions make sense for the ice condenser containment and
7 the problem being addressed? That is, how good an
8 approximation do you feel that is, in light of this
9 structure and the problem?

10 MR. CYBULSKIS: The -- I don't have any great
11 difficulty with applying MARCH to the type of structure that
12 we have here. With regard to the specific problem that we
13 are discussing today, the hydrogen burning issue, you do run
14 into limitations. - And the limitations have to do with the
15 duration of the burn, or, if you will, the flame velocity.
16 If the flame velocity is relatively slow and the burn takes
17 10, 20, 30 seconds, then that is more than adequate time for
18 the pressures to be essentially equal in all compartments.
19 When you start talking about flame speeds corresponding to
20 burn times of one second or less, or of that order, then I
21 would have to question the effective equilibrium
22 approximation.

23 MR. KERR: Thank you.

24 MR. LEE: I may come back to that point once more,
25 you did indicate earlier in the morning that, indeed, using

1 MARCH code you perhaps differentiate or extrapolate, like
2 you say, from a small test environment to a large
3 containment as to the special effect whether the burn front
4 could be assumed to propagate heat, especially, and so on.

5 MR. CYBULSKIS: No, I'm sorry, that is a
6 misunderstanding. We do not treat the details of the
7 propagation of the burn front in the MARCH code. In the
8 MARCH code the composition of each compartment is assumed to
9 be homogeneous.

10 Now, the thing that we can do in MARCH, and I will
11 get into that in the things that we have done, is, for any
12 given composition of, say, hydrogen or steam, we can make
13 assumptions about the point of hydrogen ignition and the
14 completeness of burning. And this is where the small-scale
15 experiments, hopefully, will come into play. Right now we
16 have taken basically a parametric study, using perhaps a
17 certain amount of judgment and using the information that is
18 available in the literature as to the burning
19 characteristics.

20 Doing the small-scale experiments we discussed
21 earlier give us some data points. For example, at 6-percent
22 mixtures only half the hydrogen would burn. Then we would
23 input those parameters into MARCH in specific places and
24 track the effect of that type of burning throughout the
25 entire containment volume.

1 The thing that you -- that MARCH does that the
2 small-scale experiments don't tell you is what is the effect
3 on the entire system of burning, say, in one compartment and
4 then expanding through the ice condenser into the other
5 compartments. If you -- we can go ahead and make the
6 assumption that all compartments have the same composition
7 and everything burns at once, then the MARCH results, say,
8 in terms of peak pressure, should be very similar to the
9 experiments in small scale.

10 But that, I don't think that would really be
11 particularly interesting. I think the point of interest is,
12 given ignition, say, in the lower compartment, what happens
13 in the system as a whole. That is the type of thing that we
14 need a code like MARCH or CLASIX or some other code.

15 Did I answer the question?

16 MR. LEE: To a certain extent. But one of the
17 concerns I raised was: if the burn was propagated more or
18 less in a vertical column of some kind, without a lot of
19 lateral propagation --

20 MR. CYBULSKIS: Yeah. Let me --

21 MR. LEE: -- then what does a code like MARCH tell
22 in such a scenario?

23 MR. CYBULSKIS: Well, let me -- this is somewhat
24 of an aside, but let me throw up a transparency, for
25 example, that perhaps might shed some light on it.

1 This is data paperback, Furno et al., U.S. Bureau
2 of Mines data, that I'm sure many of you have seen. And as
3 far as I know, there are not too many people that argue with
4 this, this particular data. But what it indicates is volume
5 percent hydrogen versus the pressure increase. This
6 particular curve is a calculated constant volume burning
7 curve. It's easy to reproduce. It's a fairly
8 straightforward calculation. You can even do it with a
9 slide rule if you're so inclined. A home calculator works
10 better.

11 And this, this is the experimental data under a
12 particular set of conditions and using a particular ignition
13 source. And without going into great detail, what this
14 curve tells you is that the next region, here, if you get a
15 composition in this region, given this ignition source, you
16 will get complete combustion. The flame will propagate
17 upward, downward, sideways; there is no problem, the mixture
18 is rich enough.

19 In this mixture here, between 4 and approximately
20 8 percent, you will only get upward propagation and
21 incomplete combustion. And this is the point that was made
22 earlier this morning.

23 And right in the steep part of the curve, the --
24 it's the tricky part, where you may or may not get complete
25 combustion, depending on a lot of factors, including the

1 ignition source, and, in fact, it may not be as reproducible
2 as you would like to have.

3 In the -- as I understand the program, in the
4 small-scale experiments, what it will be attempted to do, if
5 you will, is definition of a similar curve, but it wasn't
6 presented in this context, using the particular glow points
7 then as igniters. And if it takes a 12 percent composition
8 to ignite reliably, then I think we would feel fairly
9 confident that the ignition will go to completion, whether
10 the compartment is large or small.

11 MR. LAWROSKI: Is that a sub 1 atmosphere at the
12 start?

13 MR. CYBULSKIS: This particular atmosphere --

14 MR. LAWROSKI: A little pressure?

15 MR. CYBULSKIS: Yes.

16 MR. LAWROSKI: What could you say about it if you
17 started out with, say, three or four atmospheres?

18 MR. CYBULSKIS: I think, based on what has been
19 published in the literature over the years, the pressure
20 itself has only a small effect on the flammability limits.

21 Of course, the final pressure would be higher,
22 depending on what initial pressure is. But that's again a
23 very straightforward calculation. There's nothing
24 particularly secretive about that.

25 MR. STREHLOW: If I can answer that, any flame

1 system that has a low flame temperature, and this limit
2 flames are low-flame-temperature systems, there's hardly any
3 dissociation. And that means that if you doubled the
4 initial pressure, you'd double the final pressure,
5 essentially; I mean, you'd raise the final pressure by the
6 same amount proportionally. It's almost one-to-one; there's
7 hardly any deviation from that.

8 MR. SEALE: So these delta- P s would just sit on
9 top of any residual?

10 MR. STRELOW: Well, if you have two atmospheres
11 exactly, you end up with twice the delta- P s.

12 MR. MARK: Could you just give a simple, one-word
13 answer: Does MARCH calculate the hydrogen generation or take
14 it as input?

15 MR. CYBULSKIS: MARCH calculates the hydrogen
16 generation. And --

17 MR. MARK: Taking account of the amount of steam?

18 MR. CYBULSKIS: Taking into account the amount of
19 steam, the temperature of the core, the conditions in the
20 core at the time, it generates the hydrogen in the primary
21 system, releases it from the primary system to the --

22 MR. MARK: How does it handle the fact that the
23 steam might not be -- I mean, your reaction rate at 4000
24 degrees -- it'll calculate the hydrogen-steam mixture coming
25 out of the top of the tube, for instance.

1 MR. CYBULSKIS: Yes.

2 MR. MARK: And one could get the input state of
3 the gas, and you do get that when you go into the lower
4 compartment?

5 MR. CYBULSKIS: Yes. The -- what MARCH attempts
6 is, in terms of the hydrogen or steam or whatever is coming
7 off the top of the core, it would come out at a particular
8 temperature, and if the core is very hot, these gases are
9 very hot, they would mix with the gases that are already in
10 the system and transfer heat to structures, et cetera, and
11 then, depending on the accident sequence, in a large break
12 you would have an easier path to the containment than you
13 would in a small break, but it tries into account the path
14 that it has to take to the containment and introduces the
15 gas at the appropriate conditions -- at the calculated
16 conditions.

17 MR. MARK: So if there's only little steam coming,
18 then there'll be very little hydrogen generated?

19 MR. CYBULSKIS: That is correct. And, in fact,
20 that is what we find in most of our calculations, that the
21 extent of reaction, or really the rate of reaction, is
22 controlled by the availability of steam, except for the
23 initial phases of the accident.

24 Now, I must footnote that, or asterisk that, in
25 the sense that most of the calculations we do, of course,

1 are meltdown accidents, where we define an accident sequence
2 and let it take its natural course, if there is a natural
3 course; we generally do not try to arrest the melt.

4 MR. LAWROSKI: Could you put that curve back on
5 again, please?

6 If I increased the starting pressure, would not
7 the -- what you show there as taking off rapidly, at about 8
8 percent, wouldn't that drift to the left more?

9 MR. CYBULSKIS: Not -- not necessarily. It
10 depends -- if you raised the pressure uniformly and you did
11 not change the composition of the gases, it would not drift
12 very much.

13 And let me see if I can lay my hands on -- on
14 another slide, which you have seen many times, and this is
15 the, basically the Shapiro-offette curve. And this, this,
16 the dark line, is at one atmosphere. And the other lines
17 are, one is increased temperature only, and the second line
18 is increased temperature and pressure. And as you can see,
19 the flammability limits -- or the -- built very rapidly with
20 -- with increasing pressure, the temperature does somewhat.

21 Any other questions or comments?

22 MR. LEE: Can you end up with a very hot hydrogen
23 gas, depending on the accident sequence?

24 MR. CYBULSKIS: Yes, you can, depending on the
25 accident sequence. And if the hydrogen enters the

1 containment at a very high temperature and there's oxygen
2 present at the point of entry, you could expect ignition to
3 take place without any other events.

4 MR. LEE: (WORDS UNINTELLIGIBLE) number?

5 MR. CYBULSKIS: I'm sorry?

6 MR. LEE: Can you venture to give a typical number
7 for that?

8 MR. CYBULSKIS: I believe the spontaneous ignition
9 temperature of hydrogen in air is a number that is fairly
10 widely available, and if my memory serves me correctly, it's
11 around 1100 degrees Fahrenheit.

12 MR. LEE: And hydrogen temperature, typically, you
13 get in the --

14 MR. CYBULSKIS: That would have to be the hydrogen
15 temperature to get spontaneous ignition upon contact with
16 air.

17 MR. LEE: Right. But I mean the temperature of
18 hydrogen that you get typically out of MARCH code?

19 MR. CYBULSKIS: They can be -- in our -- it would
20 depend on how you defined the problem. If you defined a
21 large LOCA hot leg break, then you would, essentially, come
22 out into the containment maybe with 3000 degree Fahrenheit.
23 If you define a small break or transient where the gas goes
24 through a long length of pipe and through the steam
25 generator, then the temperature would be down, controlled by

1 the structural heat sense; it may be 500 or even less.

2 MR. LEE: TMI-2? Can you venture --

3 MR. CYBULSKIS: In TMI-2, as we analyzed the
4 accident, all the releases took place through the
5 pressurizer water, so they would have been at the
6 pressurizer water temperature, which was relatively low.

7 MR. STREHLOW: If I could interject an answer to
8 Mr. Lawroski's question, Gary and I were just talking about
9 this, and if you're talking about an ordinary flame, like
10 propane or something like that, the lean limit is unaffected
11 by pressure, essentially. The problem is that the -- this
12 steep change in curve behavior there is due to a number of
13 different phenomena than are usually apply -- or, usually
14 occur in an ordinary flame, so we don't know which way it
15 would go.

16 MR. LAWROSKI: Okay.

17 MR. STREHLOW: The answer is: you'd have to do
18 experiments at high pressure. There's buoyancy forces
19 involved; there's preferential diffusion involved. Both of
20 these are pressure-dependent. And it's toss-up in the air
21 now and ask which way the wind is blowing.

22 MR. KERR: You're saying a very thorough
23 long-range research program should be carried out in this
24 area?

25 MR. STREHLOW: Yes.

1 MR. LAWROSKI: Preferably in Illinois. But thank
2 you.

3 MR. CYBULSKIS: The work that I'm about to
4 describe is a relatively short effort. And I understand,
5 from discussions with the NRC representatives, that we will
6 -- they will be asking us to do further work. So, in that
7 sense, we have run a finite number of computer runs; we have
8 not run perhaps as many as we would like, or as many as
9 we're going to; in that sense, the results are preliminary.

10 Just for reference, this is a cross-section of the
11 Sequoyah containment. In our MARCH modeling we represented
12 it as a two-volume system, the lower compartment and the
13 upper compartment. The volume of the ice condenser in our,
14 the way we modeled it was included with the lower
15 compartment. And the ice condenser -- or the heat removal
16 flux associated with the ice condenser we modeled at the
17 junction between the two. And just for reference, the
18 dividing line between the two compartments is roughly like
19 this.

20 I don't recall the numbers exactly, but I believe
21 it's something like 390,000 cubic foot in the lower
22 compartment and about 900,000 cubic feet in the upper
23 compartment, give or take a little.

24 The accident sequence that we used as the -- for
25 the basis of analysis was SCD sequence, which is a small

1 break, the failure of the emergency core cooling injection
2 system. As I indicated before, the sequence is that it ran,
3 we let the sequence proceed, to meltdown, failure of the
4 vessel head, and a dropping of the core into the reactor
5 cavity. This is really not a point of interest, as I
6 understand it, to the discussion here; however, that, we
7 made no attempt to arrest the accident, at least, not in the
8 short-term effort.

9 And just to give the perspective of what we're
10 talking about, we have the core, given this type of
11 accident, we have core uncover at 43, top of the core
12 beginning to uncover at 43 minutes into the accident, start
13 of melt at 62 minutes, and slumping of the core
14 approximately a half hour later. The time required for the
15 entire core to slump into the bottom of the reactor vessel
16 is a function of modeling assumptions. This represents a
17 particular set of assumptions. It could be different. ♦
18 That's one area where we don't know very much now.

19 Going on, this is just the pressure transient for
20 this accident, without any hydrogen burns, just again to put
21 things in perspective. And we're primarily interested in
22 this region here where the pressure is staying constant. At
23 this point the reactor vessel fails and you're in a
24 different sort of situation.

25 There a number of slides for this accident

1 sequence in the handout. I won't bother to go through those
2 unless there's particular interest in any of them.

3 Having done the no-burning case, then we looked at
4 the -- what would happen under different burning
5 assumptions, different ignition limits, et cetera. And let
6 me just give a series of slides that are more or less
7 typical, before I get into the overall results.

8 All right. This particular -- here again we have
9 pressure time history in the containment for a particular
10 sequence: ignition assumed at 10 volume percent; hydrogen
11 burning to completion -- a burn time of five seconds, I
12 believe, burning only in the lower compartment. And on the
13 time scale that the computer spits out, you can barely see
14 the little blips, these are the 10 percent burns that you
15 see, you get very small pressure releases under these
16 assumptions. The large pressure spike is associated with
17 the head failure and release of large amounts of containment
18 to -- hydrogen to the containment.

19 Corresponding temperature courses, this is again
20 in the lower compartment, you see whenever you get the burn,
21 you get something like 2500 degree temperatures in the
22 atmosphere of the lower compartment; they are rapidly
23 reduced by the action of the recirculating fans blowing the
24 air around through the ice condenser, et cetera.

25 Just for interest, if we look at the partial

1 pressure of hydrogen, which just repeats what we've seen in
2 the other slides, it builds up to our assumed ignition
3 limit, burn to completion, builds up again, burn to
4 completion.

5 The particular case that I'm talking about here,
6 as I indicated, was ignition at 10 volume percent hydrogen.
7 And let me go on to the mass of ice in the containment for
8 this particular case. During the boil-off phase of the
9 accident, the ice is depleted at a relatively slow rate,
10 because the input into the ice bed is controlled by the hole
11 size, effectively. When you get into the hydrogen burn
12 regime, you see small stuff releases in the amount of iceage
13 down by the burn; and of course when the head fails and
14 there's a large burn, you deplete the ice bed completely.

15 MR. LEE: Do I understand correctly from the
16 series of slides that a hydrogen burn in this particular
17 sequence of action doesn't make whole lot of difference?

18 MR. CYBULSKIS: Under this particular set of
19 assumptions, you get very small pressure increases in the
20 containment if you burn at 10 volume percent in the lower
21 compartment.

22 MR. LAWROSKI: Did you say that this is the result
23 when the head fails?

24 MR. CYBULSKIS: When the head fails and releases
25 the rest of the hydrogen into the compartment, it's well

1 above 10 percent, it burns, and you get a very high pressure
2 spike. But that assumes complete core meltdown, which is
3 basically what the MARCH calculation is intended to do, as I
4 indicated before. We -- for the purposes of this brief
5 study, we made no attempt to arrest the accident in any way.

6 MR. LEE: Short of complete core meltdown, what
7 kind of impact could we expect from hydrogen burn?

8 MR. CYBULSKIS: Well, that -- let me get on to my
9 summary of results and that perhaps will answer your
10 question.

11 Let me just throw up one other slide. This is for
12 a slightly different -- well, for a case where we assumed
13 ignition at 8 volume percent and only burned down to 4
14 volume percent, the burn limit. And you would get more
15 spikes, obviously, since you're burning less hydrogen in
16 each start and with that particular set of assumptions.
17 You're starting to approach the -- more or less a continuous
18 combustion of the hydrogen.

19 Let me throw on this very busy slide here. And
20 those of you that don't have handouts may not be able to
21 read it. Apologize for that. That's basically a summary of
22 the runs that we conducted for this initial study.

23 What we looked at were the assumed ignition point
24 in terms of volume percent hydrogen -- we looked at 10, 12,
25 8, 4 percent -- and the assumed burn limit in terms of how

1 far down in hydrogen concentration. We burned -- in most of
2 the cases, as you can see, we assumed complete combustion of
3 the hydrogen. In a number of the cases we assumed that it
4 burned down to the nominal flammability limit.

5 We also looked at the duration of the burn, or the
6 burn time, which corresponds to some assumptions about flame
7 velocity, and within the context of what we're doing here,
8 it didn't seem to make an awful lot of difference (WORDS
9 UNINTELLIGIBLE). But in the presence of the ice, the ice
10 tends to overwhelm the energy release and you don't see much
11 of an effect.

12 Next factor that we looked at was the flame
13 propagation. Since the hydrogen is released -- from one
14 compartment to the other, so flame propagation from
15 compartment one to two or vice versa -- since the hydrogen
16 is released into the lower compartment, you would always
17 reach your set point, or your ignition point, in the lower
18 compartment first before you would reach it in the upper
19 compartment. So what happens is, you burn the lower
20 compartment down to zero, if that's the case, and then some
21 of the hydrogen in the upper compartment, which is not at 10
22 percent, blows back down, you generate some more hydrogen,
23 so the lower compartment keeps building up to 10, while the
24 upper compartment tends to lag behind, though its level of
25 hydrogen increases in a period of time.

1 So in most of the cases we assumed ignition only
2 in the lower compartment. In some of the cases we assumed
3 that the flame would propagate into the upper compartment.
4 That was an assumption, that, in fact, it would propagate;
5 there's no mechanistic calculation to indicate that it would
6 or would not, though, in fact, if, say, you're talking about
7 10 volume percent hydrogen, in the upper compartment you
8 would probably be talking about significant levels -- I'm
9 sorry, if you have 10 volume percent in the lower
10 compartment, you would have significant levels in the upper
11 compartment also.

12 And looking at the results we get, I think they're
13 fairly straightforward. What we see is, if we burn only in
14 the lower compartment and let the hot gases expand through
15 the ice bed into the upper compartment, we get very nominal
16 pressure rises. And that's not too surprising, since the
17 lower compartment is a smaller volume, you're burning a
18 relatively small amount of hydrogen there, you're taking
19 advantage of the ice heat sink as well as expansion volume
20 in the upper compartment. And the pressures that we're
21 talking about here are just a few psi pressure rises.

22 If you make the assumption that the flame
23 propagates into the upper compartment, then you get
24 significantly higher pressure rises.

25 The -- in the last three cases in the table, we

1 arbitrarily reduced the amount of ice in the containment, to
2 see what we would predict in the absence of ice. So, in
3 these three cases, the ice was gone long before the hydrogen
4 came along. In the case 17, we still get a modest pressure
5 rise compared to what it would do if it were, say, an
6 adiabatic combustion. Case 18, we modified the treatment of
7 our suspended water droplets in the atmosphere, basically,
8 reduced the amount of water in the atmosphere, and we get
9 some increase in predicted pressure but still a relatively
10 modest increase. And in the last case there, we let the
11 flame propagate into the -- both compartments, once it
12 started, and then you get some relatively high pressure
13 generated again.

14 But I think the conclusion, based on this
15 relatively limited set of runs, is: given ignition, and if
16 the burn takes place only in the one compartment, your
17 pressure, overall pressure releases are quite modest. And
18 if you compare the column marked "PI," which is, if you
19 will, the actual pressure, against the "PNEW," which is the
20 adiabatic pressure that you would conduct, you'd see quite a
21 significant effect of the containment sprays and the ice on
22 the pressures that we calculated.

23 That is -- I will stop there.

24 Yes, sir?

25 MR. SIEGEL: Siegel. Some of your latter curves,

1 I think it's, well, like the fourth from the last, which
2 shows the temperature in compartment number two, and there's
3 a period of, oh, like, five minutes after that precipitous
4 rise where the atmosphere is up around 1500 degrees
5 Fahrenheit, as I read what I see there. What else is up at
6 this temperature? And how much of the model that's used in
7 calculating this is disappearing during those five minutes?

8 MR. CYBULSKIS: I'm sorry.

9 MR. SIEGEL: I mean, are there fans, sprays, and
10 so forth?

11 MR. CYBULSKIS: This, the ice condenser, if the
12 ice is there, is available to remove heat, basically, in
13 proportion to the flow through it; the sprays are on in this
14 particular accident sequence, so they would be available
15 except for those times when the -- there is ice available.
16 Basically, we can't -- the way the code is set up, we can't
17 have the ice and the spray both.

18 MR. KERR: I think Mr. Siegel has some concern
19 about the fans operating at 5000 degrees Fahrenheit.

20 MR. CYBULSKIS: Well, the -- the -- the numbers
21 that you see there, the 1500 or 2000, whatever, is the
22 temperature of the gases.

23 MR. SIEGEL: Yeah, I know. But in five minutes
24 some recalibration will occur.

25 MR. CYBULSKIS: Let --

1 MR. SIEGEL: If it's just a narrow spike, I can
2 understand; it's just the atmosphere.

3 MR. CYBULSKIS: Let me put on a slide that might
4 perhaps shed some light on that question.

5 I'm not -- I don't believe this in your handout;
6 and perhaps it should have been. But this is for, I think,
7 the previous case 17; where we had the reduced amount of
8 ice. These are the surface temperatures of the various
9 structures in the compartment. The shell is the upper
10 compartment and the iron and concrete structures in the
11 upper compartment -- lower compartment. And basically, what
12 we're seeing is, in this region, very modest temperature
13 rises, and then when you get the head failure it increases.

14 But the heat capacity of the atmosphere is very
15 low compared to the heat capacity available. So, basically,
16 the temperatures that you see are of very short duration in
17 the atmosphere and they're rapidly dissipated by the
18 available heat sinks.

19 So this is perhaps indicative of what you might
20 expect given this set of assumptions.

21 MR. SIEGEL: The other question I have is related
22 to this. On one of your charts dealing with the mass of
23 ice, I'm really quite impressed with the disappearance of a
24 million pounds of ice in a very brief period. That, too,
25 looks like a glacier falling off or something.

1 MR. CYBULSKIS: It's a direct result of the energy
2 input into the ice. Now, there I would suspect that there
3 may be room to question some of the assumptions of our
4 treatment of the ice condenser, whether, in fact, you
5 conduct that much heat to the ice. But the rapid
6 disappearance of that ice, if, in fact, it is a direct
7 result of the large energy input associated, one, with the
8 release of the high-pressure and -temperature steam that's
9 in the primary system, release of the remaining hydrogen and
10 the combustion of that hydrogen, in a short period of time,
11 those represent rather substantial amounts of energy.

12 MR. SIEGEL: Well, how was the ice subdivided in
13 its initial state?

14 MR. CYBULSKIS: The -- the -- our treatment of the
15 ice condenser is based on the experiments that were done
16 with Westinghouse -- or, by Westinghouse in the development
17 of the concept. And in the tests -- and there may be
18 Westinghouse people that could probably describe it better
19 than I can -- but in the tests that were conducted, they
20 measured -- they put steam through the ice and measured the
21 effluent coming out of the top. And basically, what we have
22 done is represented the temperatures inlet and outlet, the
23 ice condenser is almost like a heat exchanger with fixed
24 outlet conditions, based on the experiments.

25 Now, I -- as I say, there are probably people

1 better qualified here to answer about the details of the ice
2 condenser itself (WORDS UNINTELLIGIBLE).

3 MR. MARK: Could I ask what is just meant by your
4 use of the term "head failure": is that melt-through of the
5 bottom of the vessel or something?

6 MR. CYBULSKIS: Yes. That is the --

7 MR. MARK: Okay.

8 MR. CYBULSKIS: -- failure of the reactor vessel
9 bottom.

10 MR. SCHOTT: And that is, that triggers release of
11 -- there's a considerable mass of hydrogen that's been,
12 already been -- represents the oxidation of zirconium --

13 MR. CYBULSKIS: Right. That's --

14 MR. SCHOTT: -- which is withheld from the system.

15 MR. CYBULSKIS: That is correct.

16 MR. SCHOTT: But it's -- what? In solution in the

17 --

18 MR. CYBULSKIS: Right. It's not in solution.

19 It's just that it's released to the primary system, and as
20 your water level in the core drops down, your boil-off rate
21 of steam drops down, and, effectively, there is no, little
22 or no motive force to pump that --

23 MR. SCHOTT: It's just the volume that's --

24 MR. CYBULSKIS: That's --

25 MR. SCHOTT: -- that's about that time been

1 shielded off by --

2 MR. CYBULSKIS: Right.

3 MR. SCHOTT: -- (WORD UNINTELLIGIBLE) that was
4 attacked.

5 MR. CYBULSKIS: And when you're -- now, remember,
6 this is a small break case, the particular sequence we're
7 talking about; it's a two-inch break. You now open up the
8 entire bottom head and you empty the system.

9 MR. RUBENSTEIN: Peter, maybe you ought to place
10 in perspective severe core damage in relationship to the
11 time scale of whether the head fails. Severe core damage
12 primarily deals with scenarios which are terminated. And
13 S2D-2 is a scenario which is unterminated.

14 MR. CYBULSKIS: That is the point I tried to make
15 in my opening remarks, that basically we made no attempt, as
16 a matter of convenience, in this thing in pointing to some
17 results of, say, terminating it; we just ran the code as we
18 normally do it, which is a unterminated, or, if you will,
19 just let -- follow the sequence and whatever, wherever it
20 leads without terminating it short of meltdown.

21 And, in fact, the results following head failure
22 represent complete core meltdown, complete metal-water
23 reaction.

24 MR. SIEGEL: Do they include a core-concrete
25 reaction?

1 MR. CYBULSKIS: In the particular calculations
2 that I presented today, there is no core-concrete
3 reactions. They would take place, well, basically, at about
4 the time that the calculation ended they would start, to
5 compound it.

6 MR. KERR: Mr. Schott?

7 MR. SCHOTT: Just two specifics here. The sets of
8 Vu-graphs in the handout deal with two cases, S2D-B and
9 S2D-2. What's the --

10 MR. CYBULSKIS: Okay.

11 MR. SCHOTT: -- major distinction between these
12 two?

13 MR. CYBULSKIS: The S2D-B, in this case the "B"
14 stands for "base case": no hydrogen burning whatsoever; the
15 hydrogen is generated, it sparked through the system, but we
16 did not burn it.

17 In the case of -- "2" is the one where we burned
18 it. Case 2 in that case corresponds to case 2 in the table
19 of the kinds of conditions under which it was burning.

20 MR. SCHOTT: And the other general matter: in
21 these calculations there is considerable pressure excursion,
22 in the lower chamber where the burn takes place, that's
23 relieved on some time scale. Is the lower chamber really
24 stronger than the total containment by some distinct amount?

25 MR. CYBULSKIS: I would defer that question to

1 representatives of Sequoyah. Perhaps they would care to
2 comment on that.

3 MR. TINKLER: Are you referring to -- can you tell
4 us what curve you're referring to?

5 MR. SCHOTT: I'll try.

6 MR. TINKLER: Is it case two?

7 MR. SCHOTT: Well, it is -- it says --

8 MR. CYBULSKIS: What are the labels? I --

9 MR. SCHOTT: Maybe I can't -- maybe I'm -- was
10 misinterpreting something --

11 MR. CYBULSKIS: What are the labels in the graph?

12 MR. SCHOTT: Let's see. I may have missed, just
13 missed, missed the significance of something as it went by.
14 The temperature in-compartment one, that gets high, but,
15 presumably, the -- is the supposition there that the density
16 is low?

17 MR. CYBULSKIS: No, there is --

18 MR. SCHOTT: That the material has vented?

19 MR. CYBULSKIS: There is no supposition. The
20 temperature does, in fact, get high, but it goes on --

21 MR. SCHOTT: But is the pressure not high when
22 that is so?

23 MR. CYBULSKIS: No, the -- the pressure is not
24 correspondingly high, because the pressure is venting into
25 the ice condenser and the gases are being cooled.

1 MR. SCHOTT: Okay.

2 MR. KERR: Mr. Lee, you have a question?

3 MR. SCHOTT: Probably it's --

4 MR. CYBULSKIS: What I tried to indicate, if you
5 will look at that PNEW column, that represents the adiabatic
6 burn pressure.

7 MR. SCHOTT: So are the --

8 MR. CYBULSKIS: In that particular compartment.
9 So if you want to look at it as being a limiting pressure
10 condition for that compartment, that will give you some
11 indication as to how high the pressure might be if it were
12 not venting into the other compartment.

13 MR. SCHOTT: Probably what I've done here is just
14 -- is followed something too late to the head rupture.
15 There is something here called "pressure in containment
16 volume number one," which is this lower probably, going up
17 to almost 140 psi at 90 (WORDS UNINTELLIGIBLE) --

18 MR. CYBULSKIS: That is following head failure.

19 MR. SCHOTT: That is. Okay, thank you.

20 MR. KERR: Mr. Lee?

21 MR. LEE: I guess I still don't understand the
22 distinction between hydrogen burn pressure and
23 partial-pressure hydrogen in compartment number two and so
24 on?

25 MR. CYBULSKIS: The partial pressure of hydrogen

1 is just that: the partial pressure that the hydrogen would
 2 have at that particular point in time. That's just the --

3 MR. LFE: With or without the burn?

4 MR. CYBULSKIS: Without the burn. That is just
 5 the hydrogen occupies some --

6 MR. LEE: All right.

7 MR. CYBULSKIS: -- some space; therefore, it has
 8 some partial pressure associated with it.

9 MR. LEE: But for the same case when you say
 10 "hydrogen burn pressure," what does it mean then?

11 MR. CYBULSKIS: The hydrogen burn -- or the -- let
 12 -- let me make sure -- the hydrogen burn pressure in -- as
 13 it's labeled in -- let me see if I can find a appropriate
 14 graph perhaps -- is really the adiabatic burn pressure that
 15 you would get if there were no heat removal mechanisms.

16 MR. LEE: And if you had hydrogen burn?

17 MR. CYBULSKIS: Yes.

18 MR. LFE: But you just said there is no hydrogen
 19 burn, because the partial pressure indicates it's zero
 20 practically.

21 MR. CYBULSKIS: Okay. Which graph are you looking
 22 at?

23 MR. LFE: Okay. I'll refer to it. Here's one
 24 that shows Sequoyah S2D-B.

25 MR. CYBULSKIS: Right.

1 MR. LEE: And the ordinate is partial-pressure
2 hydrogen in compartment number two.

3 MR. CYBULSKIS: Okay, that's -- it's not this
4 graph that you're looking at?

5 MR. LEE: No.

6 MR. CYBULSKIS: No.

7 MR. LEE: No, that gets at what I'm most
8 interested, that's my second one.

9 MR. CYBULSKIS: Well, the one that says "partial
10 pressure," there is no burn, but there is, obviously,
11 hydrogen.

12 MR. LEE: But, then, if there is no burn, what is
13 it -- where do we get the burn pressure?

14 MR. CYBULSKIS: The -- in case B, the adiabatic
15 pressure is calculated if it burns: if it burns what would
16 be the pressure.

17 MR. LEE: Is that for lower compartment, upper
18 compartment?

19 MR. CYBULSKIS: For both compartments, if it were
20 to burn at any instant in time, this is what the adiabatic
21 pressure would be.

22 MR. LEE: But when the --

23 MR. CYBULSKIS: It gives you some perspective as
24 to how significant the hydrogen might be, so that when you
25 run the next MAPCH case you know whether to run a case with

1 burning or not.

2 MR. LEE: Are we looking at the results of two
3 different cases, then, although they are all, both labeled
4 case B?

5 MR. CYBULSKIS: Case B is the same case -- as I
6 said, this is a "what if" pressure; we did not actually burn
7 it, but we asked the computer what would it be if it burned.

8 I don't think you -- you still look puzzled.

9 MR. LEE: So the -- let me try to go at it once
10 more -- so the effect of the -- or, the increase in pressure
11 due to hydrogen burn over the initial part of the assumed
12 accident is something like 20 psi? Is that what?

13 MR. CYBULSKIS: I'm -- I'm still --

14 MR. LEE: From this. From this transparency you
15 had.

16 MR. CYBULSKIS: Just looking at this graph --

17 MR. LEE: The first 60 minutes --

18 MR. CYBULSKIS: Right.

19 MR. LEE: -- the increase in pressure due to
20 possible hydrogen burn is 20 psi or thereabouts?

21 MR. CYBULSKIS: It's -- it's -- no, in this
22 particular case it's zero, so that this really represents,
23 is the pressure. There is no burning.

24 MR. LEE: But if you had burning?

25 MR. CYBULSKIS: It can't burn in this point of

1 time. So there's no contribution of burn up to this point.
2 It becomes flammable at this point here.

3 Now, if it were to burn at this point,
4 adiabatically this is what it would be.

5 MR. TINKLER: I think it ought to stay there.

6 MR. CYBULSKIS: Okay.

7 MR. KERR: Are there other questions?

8 Please continue.

9 MR. CYBULSKIS: I -- that concludes my discussion,
10 except perhaps to say that we are planning some further
11 studies, perhaps a little more parametrically than we have
12 done, to try to better define the limiting conditions, if
13 you will, under which the containment may or may not be able
14 to accommodate hydrogen burns. The details of those
15 particular parametric studies are yet to be defined other
16 than they're in the works.

17 I don't know whether you want to add anything to
18 that, Les?

19 MR. RUBENSTEIN: Well, we have the capability of
20 assuming certain hydrogen generation rates and burns, and
21 what we're going to try and do is anticipate a matrix which
22 we can plug our data from the Livermore and the Fenwall
23 experiments into, to get some insight into what the
24 pressures and temperatures would be as calculated by MAPCH.

25 MR. KERR: At various points, you and others have

1 pointed to the fact that this is an unverified code and that
2 it has some approximations and assumptions in it and that it
3 is a systems code which attempts to handle a good many
4 phenomena simultaneously.

5 If you were going to assign confidence, would you
6 say that that part of the code that is treating the hydrogen
7 burn and the consequences thereof is more, or less, or about
8 the same as -- here, in a sense, one is treating hydrogen
9 almost as a human input; although the code does calculate
10 it, it calculates it under a rather special set of
11 circumstances. But given that you have hydrogen, is it your
12 feeling that the burn and the pressure calculations in, say,
13 a Sequoyah-type containment are fairly accurate? Or do you
14 have any comment on that?

15 MR. CYBULSKIS: I -- I feel fairly comfortable
16 about the way we treat the hydrogen burn itself and the --
17 say, the pressure associated with a burn in the specific
18 volume given, the specific composition. That part of it, I
19 think, is fairly straightforward.

20 Now, we do include heat sinks in our subroutine.
21 The slab heat sinks, the spray heat sinks, again I feel
22 reasonably comfortable; I think those are fairly
23 straightforward.

24 The one heat sink -- and which, unfortunately, may
25 be the most important one -- I have some misgiving is, and

1 that's the ice bed. While the ice bed has been proven to be
2 quite effective in handling, say, steam blow-down, and I
3 have no reason to question that, the way we have modeled the
4 ice bed, for also these hydrogen calculations, is the same
5 way as we did for the blow-down, and whether, in fact, the
6 ice bed is equally effective in creating, say, the energy
7 removal from the hydrogen flame, I don't know and I don't
8 have any basis for assessing it right at this moment.

9 MR. KERR: Well, it appears to me that that is a
10 fairly crucial part of the calculation. And if there's a
11 large uncertainty in that, I would think that one would want
12 to investigate that. How does the staff look at this part
13 of the calculation?

14 MR. RUBENSTEIN: We feel much the same way.

15 I think, Peter, you want to talk about the second
16 half of the work that you'll be doing for us, which is
17 towards the verification. And I don't want to use the word
18 in the literal sense that Dr. Tong uses it experimentally,
19 because I think we'll do code comparison, but we will be
20 looking at that.

21 MR. KERR: You're going to compare two unverified
22 codes with the assumption that if they both give the same
23 answer something or other has then been demonstrated -- and
24 I'm not sure what.

25 MR. RUBENSTEIN: We'll try.

1 MR. CIBULSKIS: Let me just make a couple of
2 comments, if I may, in this verification area. There are
3 facets of the MARCH code that are fairly easy to verify, and
4 some of this has been done. There are facets that perhaps
5 never will be verified. Let me just point out, for example,
6 that the structural heat sink models that we use are the,
7 essentially the same as they are in the CONTEMPT code,
8 which, I believe, is an approved code. The heat transfer
9 coefficients from the atmosphere to the structural heat
10 sinks, I forget whose correlation we use, my memory fails me
11 for the moment, but this particular correlation has been
12 shown by a number of people to be conservative insofar as it
13 does not transfer as much heat to the containment -- to the
14 structure as has been experimentally observed.

15 In terms of, say, the blow-down pressures that we
16 might predict, we have some highly simplified blow-down
17 models in our code that would predict the containment
18 response if we put steam in among other things. And the
19 results that we get are consistent with what we see for
20 sample problems for CONTEMPT, for example.

21 So a number of these things are really, either
22 have been or, in fact, are, easy to verify. The areas that
23 are difficult to verify and may never be verified, I think,
24 are some of the areas I alluded to in some of my earlier
25 presentations to the committee, namely, the core slumping

1 models and that type of things. In those areas the only
2 hope that you have of any kind of verification is a
3 comparison with somebody else's code. Hopefully, we won't
4 have too many data points.

5 MR. RUBENSTEIN: But they're beyond the range of
6 interest, core slumps.

7 MR. CYBULSKIS: In the particular discussion at
8 hand --

9 MR. RUBENSTEIN: This time --

10 MR. CYBULSKIS: -- that is correct.

11 Obviously, the -- in the work that I presented
12 here, we just made assumptions that the hydrogen ignites at
13 certain levels and burns down to certain levels. These were
14 assumptions of convenience, for the purposes of the study.
15 And the experimental program that was described earlier
16 will, hopefully, shed some light into these areas. We can
17 then go back and feed that experimental data and repeat the
18 calculations and see what it would make.

19 So, I guess, in summary, as far as the
20 verification is, there are parts of the code that are, in
21 fact, verified or easily verifiable, there are parts of the
22 code that are not as easily verified.

23 MR. KERR: Well, as I have grasped the mass of
24 material that I have heard today, I get the feeling that the
25 MARCH code, given an energy input which would come from

1 burning uniform distributions of hydrogen and air, can
2 probably calculate the pressure fairly accurately.

3 MR. CYBULSKIS: Yes.

4 MR. KERR: That there's going to be an
5 experimental program to determine if igniters will ignite.
6 But at some point, it seems to me, one also has to determine
7 whether given ignition, one gets an amount of burning
8 necessary to produce the energy input which goes into the
9 MARCH code, that is, given a burn in the vicinity of an
10 igniter, how does it propagate. And the code -- and you
11 certainly have not claimed that the code will tell you
12 anything about that. It isn't clear to me that the
13 experiments are going to tell one very much about that.
14 Where does that information come from?

15 MR. CYBULSKIS: I think the --

16 MR. KERR: Or I'm missing something.

17 MR. CYBULSKIS: The point that I tried to make, by
18 showing my first slide, which was really not in my planned
19 presentation, the Furno data, was that above certain levels
20 of hydrogen in air, in particular, perhaps not in steam
21 mixtures, there has been data showing that the combustion
22 will go to completion. And I'm not sure -- to the best of
23 my knowledge, that is an accepted conclusion -- I'm not sure
24 that we have to prove that again.

25 MR. KERR: So that you're saying that at some

1 point one will accumulate a sufficient concentration of
2 hydrogen so that given an ignition at that point one has a
3 burn?

4 MP. CYBULSKIS: Yes. And the burn will go to
5 completion. And I think the experiments that Livermore is
6 doing will shed some light on this for the particular set of
7 igniters.

8 If I might make a passing comment with respect to
9 the -- with respect to the Furno slide that I showed, I
10 believe, if I remember the source correctly, those data were
11 shown for a single spark in a particular mixture. And what
12 they found: that given a single spark in those very lean
13 mixtures, it did not burn at all, and you got these little
14 flamelets that were mentioned earlier, little balls of fire
15 that sort of just rose up.

16 Another part of that particular paper, which isn't
17 perhaps as well known, is that they did do some experiments
18 with repeated sparking. And when they did the experiments
19 with repeated sparking, they, in fact, found that they could
20 get more complete burning than they did with just a single
21 spark, though the number of experiments in that area is very
22 limited.

23 So, I guess, the only thing I'm trying to point
24 out there is that we're not completely in the dark on
25 hydrogen behavior. There is a base of knowledge available.

1 MR. KERR: Thank you.

2 Mr. Lee?

3 MR. LEE: Based on your MARCH calculations, can
4 you somehow give me a rough idea what kind of benefit, or
5 reduction in pressure in containment, we can expect if we
6 have functional distributed igniters?

7 MR. CYBULSKIS: Well, I come back, if I may, to
8 the, what perhaps is, to me, basically much of a basic
9 conclusion, though I'm not sure that everybody would agree
10 with me, but if you let hydrogen accumulate to significant
11 quantities in a containment of this type and then get
12 ignition, for whatever reason, a la Three Mile Island, or
13 whatever, at a significant concentration, then you have a
14 very high probability of failing the containment. And the
15 whole point of the intentional ignition effort, as I
16 understand it, is to make sure that you never reach those
17 very high concentrations of hydrogen that can, in fact, fail
18 containment. What you're trying to do is to keep the
19 hydrogen concentration down to manageable levels. And the
20 manageable levels may not be well defined, but basically
21 what you're trying to do is keep the pressure within some
22 envelope. And a good way to do that is to make sure that
23 the hydrogen doesn't accumulate to a high level.

24 MR. RUBENSTEIN: One of the next presentations
25 will address some specifics in terms of pressure and

1 temperature with the CLASIX code for Sequoyah.

2 MR. CYBULSKIS: Well, I won't keep you waiting.

3 MR. KERR: Thank you, Mr. Cybulskis.

4 That brings us, I believe, to a presentation by --
5 or you, you're next up. Okay.

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1 MR. TINKLER: This portion of the presentation
2 will deal with the analysis that has been provided by TVA,
3 principally performed by OPS.

4 This slide says some of the same. It indicates
5 that TVA believes that the study of igniters will continue
6 for about a year and they will determine sensitivity to
7 critical parameters for various accident scenarios, which
8 they are waiting for us to identify, in order to determine
9 the containment response for distributed ignition system.

10 The bulk of the calculations have been done using
11 the CLASIX code. The code is under development. Put to
12 summarize, we think the code contains the models that are
13 necessary to perform a calculation of the containment
14 response. It contains features typical for containments and
15 features unique to ice condenser containments.

16 This is a list of some of the capabilities that
17 CLASIX has. CLASIX is a multivolume containment code. It
18 is not restricted to equal pressures. It can calculate
19 different pressures in different rooms. It has the
20 capability to model a vent from the upper compartment into
21 either the environs or can model a vent path directly from
22 the upper compartment to the lower compartment, which would
23 bypass the ice condenser.

24 It has an ice condenser model slightly more
25 refined than the MARCH code presently uses. The ice

1 condenser model is similar to that used in the LCTIC 3 code,
2 which is a licensing code used by Westinghouse CPS to
3 perform design analysis on ice condensers. And the ice
4 condenser heat transfer model is taken principally from the
5 Walt's Mill (?) test data.

6 It has the capability to model the recirculation
7 fans which draw suction in this case from the upper
8 compartment and discharge to the lower compartment in the
9 dead-ended regions in the containment. It has the
10 capability to model both the lower inlet and intermediate
11 deck doors. This is one of the reasons why you may get
12 different pressures, because they act as check valves in
13 reverse direction.

14 It has the capability to model individual
15 atmosphere constituents -- oxygen, hydrogen, nitrogen and
16 steam. It uses ASME values for saturated steam. It has the
17 capability to model sprays. It will model sprays whether
18 the ice is there or not.

19 Input to the code can be in the form of hydrogen,
20 nitrogen or pure heat additions, and input would also be
21 break flow, steam water. It has several burning control
22 options in order to determine the sensitivity to the
23 ignition criteria or setpoints that you choose to
24 investigate.

25 As I stated earlier, the TVA has provided

1 preliminary analytical results. These results are for the
2 small break LOCA with a failure of ECCS, the S D
3 sequence. The rate of hydrogen release to the containment
4 was based on a MARCH calculation provided to Westinghouse.

5 Onset of hydrogen release was approximately one
6 hour after accident initiation, and the hydrogen continued
7 to be released for approximately 3000 seconds. They
8 released approximately 1550 pounds of hydrogen, representing
9 about 70 percent of the zirc reaction.

10 The base case, which we will be looking at a
11 little bit later, assumed hydrogen combustion when 10 volume
12 percent hydrogen was reached in the various compartments.
13 They have done a number of sensitivity studies regarding the
14 air return fan, burning parameters and ice availability.

15 This is a list of the base case parameters that we
16 used. As I said, the CLASIX code calculation begins when
17 hydrogen is released to containment, when hydrogen is
18 generated. Up until that time, in order to initialize the
19 CLASIX code, results from the LOTIC code, which is a review
20 and approve code developed by Westinghouse, and the initial
21 conditions, volumes, temperatures, pressures, the ice mass
22 at the start of the CLASIX calculation, heat transfer
23 properties, are taken from the LOTIC analysis.

24 Burn parameters, and we have several of them
25 listed, are user options. They are hydrogen for ignition

1 within a compartment, hydrogen concentrations necessary for
2 propagation of the hydrogen burn into adjacent compartments,
3 and the oxygen levels necessary to sustain the ignition.
4 Those are the values that we used in the base case.

5 The air return fans were modeled, both of them, in
6 the base case, assuming a constant capacity of 40,000 CFM.
7 The spray system was modeled with a flow rate of 6,000 GPM
8 at a temperature of 125 degrees. Sprays are modeled as
9 droplets and modeled, a film coefficient to the droplets, a
10 value of 20.

11 The ice condenser drain temperature was
12 conservatively assumed to be 32 degrees. As I have said,
13 break release data was obtained from results using the MARCP
14 code to calculate the primary system response. The CLASIX
15 code does not model the primary system; it is simply a model
16 of the containment.

17 This is a figure of a ice condenser model as seen
18 by CLASIX. The Sequoyah model was also a four-volume
19 model. The fan was modeled, took suction from the upper
20 compartment and discharged into both the dead-ended volumes
21 and the lower compartment volume, and they also modeled the
22 flow path from the upper compartment to the lower
23 compartment through the operating deck drain holes, which is
24 a bypass.

25 The ultimate capacity of the code is seven

1 volumes.

2 This shows the hydrogen release that was used in
3 the CLASIX calculation as calculated from a MAFCH analysis
4 for the S D transient. As we said, the onset of hydrogen
5 production occurs in roughly an hour after accident
6 initiation and proceeds for approximately 3000 seconds.
7 This represents -- at the top of the curve -- this is
8 integrated mass release and it represents about 70 percent
9 of the core reaction.

10 At this time the core slumped and they terminated
11 the calculation because they consider this evaluation of
12 degraded core accidents.

13 This is a plot of some of the results obtained
14 using the CLASIX code for the base case. This is a plot of
15 the temperature in the lower compartment. What we see here
16 is a series of nine burns in the lower compartment. The
17 igniters are turned on. Hydrogen concentration reaches 10
18 percent and the hydrogen is burned off.

19 Hydrogen concentration then builds up until the 10
20 percent setpoint is reached, and again it burns. The
21 temperature hovers around 2200 degrees for each of the nine
22 burning cycles. During each cycle, approximately 100 pounds
23 of hydrogen are burned, resulting in about 6 million Btus
24 released to the containment.

25 MR. KERR: Mr. Tinkler, you mentioned sprays

1 earlier. Are these calculations made assuming that the
2 sprays are operating?

3 MR. TINKLER: Yes.

4 MR. KERR: What turns them on?

5 MR. TINKLER: Containment high pressure setpoint.

6 MR. KERR: That setpoint, then, is not reached
7 until the pressure --

8 MR. TINKLER: It would have been reached long
9 before the onset of hydrogen generation. It is reached at
10 about, I think it is, 2.5 to 3 psig.

11 MR. KERR: I thought in TMI-2 it was -- one of the
12 evidences for detonation was that sprays aren't turned on
13 until one reached a fairly high pressure.

14 MR. TINKLER: Well, TMI-2 has a considerably
15 higher spray setpoint, too.

16 MR. KERR: The ice containment is a very low --

17 MR. TINKLER: Ice condensers have lower
18 containment spray setpoints.

19 MR. KERR: Thank you.

20 MR. TINKLER: Because of lower design pressures.

21 This is another temperature plot. From this plot
22 we see that the first burn in the lower compartment does
23 propagate into the ice condenser volume and it results in a
24 temperature excursion in the ice condenser of around 1200
25 degrees. The remaining burns did not propagate into the ice

1 condenser volume simply because the concentration of
2 hydrogen was sufficient to meet the criteria of the analysis.

3 This is a plot of the temperature in the upper
4 compartment volume. Again, you see one departure from the
5 normal transient in that the temperature rises briefly to
6 about 150 degrees. This is due to the fact that one of the
7 burns did occur and did propagate into the ice condenser and
8 results in a higher exit temperature from the ice condenser,
9 so you discharge hotter gases into the upper compartment and
10 it raises the upper compartment temperature.

11 I don't think you could see that phenomenon
12 necessarily from MARCHE, depending on how the calculation was
13 done.

14 This is a series of pressure plots. The total
15 pressure in the lower compartment. Pressure prior to
16 hydrogen burning is approximately 22.5 psia, and it arises
17 to approximately 26.5 for the first burn. The pressure
18 transient closely resembles the temperature transient, and
19 the peaks are obviously occurring when the hydrogen is
20 burning.

21 The pressure is decreased from the peak pressure
22 calculated each burn in approximately two minutes. The
23 first pressure peak is slightly higher than the others
24 because the concentration of hydrogen, while it is at 10
25 percent for each of the burns, the absolute magnitude of

1 hydrogen present in the lower compartment can vary simply
2 because the mass of various constituents in the lower
3 compartment varies during the transient.

4 So you may have had slightly more actual pounds of
5 hydrogen in the lower compartment for the first burn.

6 This is a plot of the pressure in the upper
7 compartment. It is very similar to the lower compartment.
8 There is a slightly different pressure because the burn did
9 propagate into the ice condenser.

10 This is a plot of the lower compartment oxygen,
11 partial pressure. From this figure it looks like the oxygen
12 concentration in the lower compartment at the start of the
13 transient is on the order of 10 percent and decreases during
14 the burn as oxygen is depleted, and oxygen is returned to
15 the lower compartment via fans and the cycle is allowed to
16 start over again.

17 This is a plot of the hydrogen partial pressure in
18 the lower compartment. First burn occurs when the hydrogen
19 concentration is 10 percent and the partial pressure is
20 correspondingly around 2.3 psi.

21 This is a plot of the steam partial pressure in
22 the lower compartment.

23 This is a plot of the ice mass. As can be seen,
24 during each burning cycle we get a very rapid change in the
25 ice mass, although the overall transients are rather

1 smooth. As a matter of fact, the shape doesn't really
2 change that drastically from the beginning of hydrogen
3 burning to the end, but that may vary due to steam release
4 rates from the primary system also.

5 One can roughly estimate the efficiency of the ice
6 in removing energy due to hydrogen burning by looking at the
7 slope of this transient during a burn cycle. You could
8 extrapolate the slope at the higher points in order to
9 estimate roughly how much energy of the burn is being
10 removed by the ice.

11 After the transient is over, we have about 3000
12 pounds of ice which should represent at least 40 million
13 Btus of energy removal capability. As I say, this was done
14 conservatively, assuming the drain temperature to be 32
15 degrees. The hotter the drain temperature, the more ice
16 that would be left following the transient.

17 TVA has provided a number of sensitivity studies
18 to indicate trends that they see. The first case, as I
19 said, was the base case. There was a total of 900 pounds of
20 hydrogen burned. There was a series of nine burns, 100
21 pounds per burn. The peak temperature in the lower
22 compartment was approximately 2200 degrees. The icebed saw a
23 peak temperature of 1200 degrees. The upper compartment saw
24 a peak temperature of only 150 degrees, with peak pressures
25 of 26.5 and 28.5.

1 They performed a case assuming that the hydrogen
2 was ignited at 8 percent and that the burn would propagate
3 into other compartments at 8 percent. This case resulted in
4 slightly more hydrogen being burned, with less hydrogen
5 being burned at a time.

6 This case also saw approximately 36 total burns
7 within the containment, the bulk of them occurring in the
8 lower compartment, but there were also burns in the ice
9 condenser and a burn in the upper compartment. You see, the
10 upper compartment temperature is substantially higher than
11 the base case. It rises to about 260 degrees.

12 But the sprays were effective in removing heat
13 addition to the upper compartment in this case.

14 TVA has performed a case assuming one air fan in
15 operation. The base case assumed both air fans, and there
16 was a negligible change in the results.

17 TVA has also performed a case assuming no ice in
18 the ice condenser exists after the first two of seven
19 burning cycles. This is a nonmechanistic analysis in that
20 if they do the S D transient, they have ice in there. But
21 this case was done in order to indicate what type of
22 sensitivity they may have to varying accident scenarios,
23 although you would have to precisely look at the accident
24 scenario to determine at what time you ran out of ice.

25 There are obviously scenarios where ice is

1 depleted, but you would have to balance that against the
2 energy that would be remaining in the core along with the
3 releases to the lower compartment.

4 This case burned 350 pounds of hydrogen, not
5 significantly different. They burned slightly more hydrogen
6 each time. They had only seven burning cycles, all of which
7 were determined to have occurred in the lower compartment.
8 The peak pressure for this case was 41 psia.

9 The final case shown here is a case with no air
10 fans, where mixing of the atmosphere constituents is simply
11 based on pressure differentials between compartments. That
12 is how flows are calculated. Without the air fans, there is
13 a great deal of difficulty in predicting beforehand where
14 the hydrogen will end up.

15 In this case a great deal of hydrogen accumulated
16 in the upper compartment, resulting in a burn in the upper
17 compartment with a very high pressure of 92 psia.

18 MR. ETHERINGTON: Is that kind of pressure
19 difference on the ducts effective without the fans?

20 MR. TINKLER: Yes. If you have a delta-p between
21 the upper and lower compartment, it will drive it through
22 that to the ducts and to the fans.

23 MR. ETHERINGTON: So you don't really -- that
24 means if you have no fan system.

25 MR. TINKLER: Well, this burn may occur, say, on

1 the order of 10 to 20 seconds. I can't say it is
2 guaranteed, but it is doubtful that the pressure could be
3 equalized between those two compartments through the fan
4 system in 10 seconds.

5 MR. ETHERINGTON: Isn't that equally true if the
6 fans were running?

7 MR. TINKLER: Yes.

8 MR. ETHERINGTON: So the fans don't really make
9 much difference one way or the other.

10 MR. TINKLER: As far as differential pressures
11 there?

12 MR. ETHERINGTON: Yes.

13 MR. TINKLER: No, they don't make much difference,
14 I wouldn't guess. What they do make a difference in is
15 mixing the hydrogen.

16 MR. ETHERINGTON: Yes, yes.

17 MR. TINKLER: The no fans case also -- I think at
18 one point the lower compartment was oxygen depleted or there
19 was not sufficient oxygen to sustain the ignition.

20 MR. STREHLCK: I am curious on one point. You say
21 when one chamber burns, and the other doesn't burn. Is that
22 what you actually do in the code, you allow combustion only
23 in one chamber and nothing happens in the other chamber?

24 MR. TINKLER: You specify conditions under which
25 the burn will propagate. You specify the condition at which

1 the burn in one compartment will propagate to another, and
2 you can specify a propagation delay time.

3 MR STREHLOW: But it always propagates, in other
4 words.

5 MR. TINKLER: Excuse me?

6 MR STREHLOW: It always propagates?

7 MR. TINKLER: No.

8 MR STREHLOW: I have a problem with that, because
9 if the pressure rises from 15 psi initial pressure to 20 psi
10 --

11 MR. TINKLER: It rises from approximately 22.5 to
12 about --

13 MR STREHLOW: About 4 psi, right?

14 MR. TINKLER: 4 psi.

15 MR. STREHLOW: The point is you are going to
16 displace some of the gas from the chamber that contains
17 adequate hydrogen to the chamber that doesn't. It is not
18 going to mix with that gas; it is going to burn in that
19 chamber. The flame will propagate after it. I mean if you
20 have a vented vessel, for example, that has got combustion
21 going on in it, you get a tremendous fireball outside that
22 vessel.

23 So that when you pressurize this vessel and gas is
24 transferred to the second vessel to pressurize it, some of
25 that gas contains the hydrogen. And you don't have any

1 flame arresters in there, so the flame will get up in
2 there. Now, it won't burn all of it because there is not
3 that much hydrogen up there, but there will be combustion in
4 the second chamber if there is combustion in the first
5 chamber, at all times.

6 MR. TINKLER: I understand.

7 MR. STREHLOW: But that is not what the code does,
8 right?

9 MR. TINKLER: The code can do that by specifying a
10 -- you could ultimately model that by specifying a low
11 setpoint for propagation.

12 MR. STREHLOW: But it doesn't at the present time.

13 MR. TINKLER: Well, these runs did not. It could.

14 MR. STREHLOW: Okay, that is the answer.

15 MR. TINKLER: Like I say, it is anticipated that
16 sensitivity studies into the ignition criteria that are used
17 in the analysis will continue for some time. There is no
18 claim that we have searched out and found the worst case for
19 ignition criteria. That wasn't the intent in these cases.

20 The intent was to pick parameters they thought
21 were reasonable. I believe that would be the case. I believe
22 at 10 volume percent, which was the base case, they believed
23 that the combustion will be fairly complete, and therefore
24 it is a reasonable calculation.

25 MR. MARK: Did I catch correctly that this Case 5

1 is the one in which you have still nine cycles but --

2 MR. TINKLER: No. The case with no fans. It is
3 not quite the same because the burning occurs in different
4 places at slightly different times. This case was not run
5 until completion.

6 MR. MARK: It is the no ice that had nine cycles,
7 two with ice and seven without.

8 MR. TINKLER: The no ice case had seven cycles,
9 two with ice, five without. I will provide you some slides
10 on that a little later. As a matter of fact, right now.

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This is the case with no ice. Again, it's a non-mechanistic analysis, which is conservative. This is the total pressure in the upper compartment; pressure in the lower compartment. Very similar.

MR. KERR: Mr. Tinkler, I'm beginning to run into a problem in that I have about three more hours of presentation and only about two more hours of time.

MR. TINKLER: I only have a few more slides.

MR. KERR: Okay. I was just going to suggest that if you could talk 30% faster --.

MR. TINKLER: Okay. Like I said, the transient looks very similar to the other transients you've seen. You see the two pressure peaks for the cases where ice is remaining, and the two cases where ice has been removed.

You would expect the pressure in the containment to rise after the ice has melted out, irregardless of hydrogen burning, because you've lost the heat sink, and you see that in a normal design basis LOCA analysis. But the pressure is approximately 41 psi, which is still far below the yield pressure containment.

This is a plot of the temperature in the lower compartment; about 23, 2400°F. We were burning slightly more hydrogen in the cycle.

MR. ETHERINGTON: Do you consider 41 psi below the yield?

1 MR. TINKLER: The yield was estimated between 27 psig
2 using a conservative value for the yield strength, and between
3 27 and 36 I believe psig. We calculated 41 psia.

4 MR. LAWROSKI: Mr. Tinkler, in view of what happened
5 in the case 5 with no air fans to the upper compartment peak
6 pressure, how comfortable would you be with a system that only
7 has two air fans?

8 MR. TINKLER: You could argue that anything that
9 increases reliability of the air fans would be an improvement.
10 Since they are sensitive to it.

11 As we tried to emphasize, this is all work that has
12 been done on a rather short timeframe, and we have many things
13 that we're going to continue to review. Some of these topics
14 include the location and number of igniters. It may be that we
15 decide we will want additional igniters located in some strategic
16 places where high point vents are, the pressurizer release valve
17 or whatever. Hydrogen monitoring and mixing, those systems were
18 designed for what is on the docket as a design basis accident.
19 Are they adequate or suitable for this kind of accident mitigation?

20 We need to know more about the ability of igniters
21 to function in a turbulent flowing atmosphere. Procedures and
22 strategy for igniter operations; we need to learn more about
23 CLASIX, the solutions scheme, models in the containment, specific
24 details, that is, and the input parameters. Analysis results
25 need more review. Mixing of constituents, sensitivity studies,

1 burning criteria and the effects of different hydrogen generation
2 rates.

3 We do think the no-ice case does demonstrate some
4 insensitivity to the accident scenario.

5 The effects of local detonations need more work done,
6 they need more review.

7 MR. KERR: Excuse me, when you say the effects of, are
8 you able now to predict the possibility of local detonations?

9 MR. TINKLER: We don't calculate concentrations that
10 are detonable in the analysis we've done.

11 MR. KERR: Is your model capable of doing that?

12 MR. TINKLER: Of doing a detonation calculation?

13 MR. KERR: Of doing a calculation which would find
14 pockets of detonable mixtures, if they existed, in this structure?

15 MR. TINKLER: If we picked a small, a very small,
16 volume, we have 7-node capacity, and if we picked a very small
17 node and we introduced hydrogen in a node adjacent to it, we
18 could force the code possibly to calculate it.

19 MR. KERR: Do you plan to make an effort to explore
20 for this possibility? Do you consider it a serious problem or
21 unlikely?

22 MR. TINKLER: We consider the problem of local detona-
23 tions to require more work by both the applicant and the staff.
24 Preliminary calculations and judgments are that they can withstand
25 local detonations. Presumably, concentrations may be higher in

1 the lower compartment, but it's a smaller volume. If you could
2 withstand a detonation of a local pocket, you could most surely
3 withstand equilibrium pressure after the detonation, because
4 we're talking about smaller quantities of hydrogen. But in any
5 event, it does need more work.

6 Protective measures for essential equipment; we need
7 to learn a little more about the response of equipment to the
8 temperature transients and what may be necessary to insure that
9 they'll be operable. Whatever equipment is needed, that is.

10 Conclusions. The preliminary conclusion is that the
11 distributing ignition system has significant potential as a
12 short-term solution to the problem of hydrogen control. It should
13 be stressed that the system is most useful when operating in
14 conjunction with existing heat removal systems. If you don't
15 have other heat removal mechanisms, burning it with igniters is
16 not a great advantage.

17 And we intend to explore various aspects of both
18 igniters and possibly additional heat removal methods.

19 MR. KERR: Thank you sir. Are there questions?

20 MR. MARK: When you say the very short term, do you
21 mean something that can be implemented quickly or something
22 you'll try to get away from quickly?

23 MR. TINKLER: Something that could be implemented
24 quickly. TVA has indicated that this is an an interim distribu-
25 ting ignition system and they will continue to study other

1 measures for hydrogen control.

2 MR. MARK: But there may not be any very good reason
3 to change if you learn a little more about it.

4 MR. TINKLER: That is correct. That's not to say it
5 couldn't be the final system.

6 MR. GREGORY: I have a concern about the curve which
7 shows depletion of oxygen in the lower compartment. If you drop
8 the oxygen pressure by about half for a short period of time, it,
9 in effect, is putting the hydrogen-oxygen ratio up and your
10 conditions for initiation of ignition will be quite different
11 during that period while oxygen is coming back into the system.
12 You, in effect, conceivably increased or you've certainly altered
13 the hydrogen-oxygen ratio in the gas.

14 MR. KERR: How do you remove oxygen while at the same
15 time removing hydrogen?

16 MR. GREGORY: You remove both, but they'll come back
17 in at different rates because they're coming back from different
18 sources. The oxygen, I assume, is coming in from the upper
19 compartment and the hydrogen is coming from a break in the lower
20 compartment. And they're coming in at different rates and it
21 will be pretty complicated to model but it ought to be looked at.

22 MR. TINKLER: It does look like, though, that the
23 oxygen is returned to their ignition setpoint much more quickly
24 than the hydrogen is built up, just based on the next slide.

25 MR. KERR: If the igniters work, as one might predict

1 that they will, is that a problem? One would simply get a
2 flammability limit earlier than otherwise might occur.

3 MR. SCHOTT: In the regime that's being deal with
4 here in this series of calculations, it appears that depletion
5 of oxygen -- rather, depletion of hydrogen at the immediate time
6 of the burn is quite complete; that of oxygen, while it looks
7 serious, like down to half, is not for any useful period of time
8 or a dangerous period of time so complete as to prevent burning
9 of any hydrogen that might accumulate. That is, one is always
10 in the regime where there is more than enough oxygen to handle
11 however much hydrogen may be present. And as long as that is
12 so, the 10% criterion on hydrogen is the limiting one, and even
13 that is just an assumption.

14 It appears to me that the series of calculations that
15 have been done both with the CLASIX and the MARCH codes are
16 severe cases of the pressure and temperature transients in that
17 they have not -- they have worked with hydrogen accumulation up
18 to this 10% level, or 8% --

19 MR. TINKLER: That was a point that we battered around
20 this morning.

21 MR. SCHOTT: Which one is sure will burn rather
22 completely, whatever is in short supply, which in every instance
23 is the hydrogen. So it's a conservative test of the ability of
24 a repetitive burn system to accommodate a rather small number of
25 burns with rather substantial, 100 pound, accumulations of

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1 hydrogen each. And the real situation that may result from
2 having --. It's as though the igniters were triggered at will
3 at this 10%, which is not what's actually being proposed to be
4 done.

5 MR. TINKLER: But you're saying then that they hydrogen
6 may be burned at even a slower rate, smaller quantities at a
7 time.

8 MR. SCHOTT: More continuously in time. And probably
9 mixing -- to the point where mixing becomes the limitation
10 instead of what's now -- the set of parameters that now are
11 governing the calculations.

12 MR. KERR: Other questions or comments?

13 MR. LEE: I'm still somewhat puzzled. Is there any
14 accident scenario that could get us to the upper flammability
15 limit instead of the lower flammability limit?

16 MR. TINKLER: Are there any?

17 MR. LEE: Yes. Or should they consider those
18 sequences?

19 MR. TINKLER: It's difficult for me to say whether
20 there's --. Seventy percent of the core reaction is what we
21 showed as the endpoint for our calculation in some definition of
22 a degraded core accident. That represents near 17% hydrogen,
23 17%-18% hydrogen concentration in the containment at dry atmos-
24 phere. So --

25 MR. LEE: Seventeen percent of --?

1 MR. TINKLER: No. Seventy percent of the core
2 corresponds to 17% molar concentration of hydrogen in the contain-
3 ment uniformly.

4 MR. LEE: In a mixture of air and hydrogen?

5 MR. TINKLER: Yes, it's primarily a dry atmosphere.
6 More steam reduces the concentration.

7 MR. MARK: But you will go through the upper limit.
8 You have hydrogen and steam coming out the end of a pipe, and
9 it's a low percent hydrogen as far as the air is concerned. Now
10 you'll add air.

11 MR. TINKLER: In a local region we could. That's
12 based on bulk containment.

13 MR. KERR: Are there any other questions?

14 MR. SCHOTT: That 70% based on zirconium, is that
15 hydrogen that is all released through the vent, or is some
16 hunk of that 70% still harbored in the unburst reactor?

17 MR. TINKLER: That 70% is in the containment.

18 MR. KERR: Thank you, Mr. Tinkler. Mr. Butler?
19 I show you for 10 minutes now.

20 MR. BUTLER: I'd like to make some conclusory remarks
21 for the staff's portion of today's presentation. I certainly
22 appreciate the comments and also some of the answers and questions
23 raised by the members of the committee and its consultants.
24 We covered a lot of ground today, covering different programs,
25 each with different objectives. We've dealt with long-term

1 programs and short-term programs; generic issues, generic
2 criteria, as well as specific issues and specific criteria for
3 Sequoyah for the near term. I recognize it's somewhat difficult
4 for those new to the issue and the problem to keep these different
5 programs and issues separated, but I think it's important that
6 an effort be made.

7 We did cover a lot of the generic and overall kinds of
8 issues. But for us, that kind of matter can await further develop-
9 ments in further meetings. For us, the urgent thing is the near-
10 term issue, Sequoyah; the interim distributed ignition system
11 and whether it is good enough for an interim period pending
12 further understanding of the characteristics of the IDIS.

13 The program has moved very fast. We had a meeting with
14 the ACRS July 15th. Subsequent to that meeting, all this stuff
15 we discussed today was prepared. We wrote work statements,
16 contracted with a number of labs and got them well underway with
17 their programs. I think you can understand that the staff has
18 worked very vigorously, and certainly the TVA applicant has
19 worked harder and faster. They have limited time to present
20 their story, additional to those things that we presented based
21 on their input to us.

22 Now, let me go through some of the slides, just to
23 refocus on the stuff I showed you earlier today and the urgency
24 of our schedule. Starting with the September 2 date --

25 MR. KERR: If you read those things one by one, I'm

1 going to --.

2 MR. BUTLER: I won't. You've seen it this morning.
3 Starting with the September 2 date we have the mini-PSAR, and
4 we expect to conclude all our activities by December and meet
5 again with the ACRS in January and with the Commission also in
6 January. We need a tentative decision next week with respect to
7 whether Sequoyah can reasonably be authorized to run at full
8 power for this interim period pending further understanding of
9 the issue.

10 This slide simply summarizes the status for all the
11 classes of containments, and you'll note in here for the ice
12 condensers that the specific criteria and measures for hydrogen
13 control are under consideration. We hope to conclude that matter
14 within the next few days.

15 I have one slide on staff position and then the staff
16 conclusions. The staff position is that the existing provisions
17 at Sequoyah which satisfy Section 50.44 are sufficient near-term
18 requirements to warrant full power licensing. We have accelerated
19 programs, both staff and applicant, to qualify and implement
20 measures additional to those of 50.44. The timeframe is during
21 the next four months. These additional measures, if found
22 effective and appropriate for Sequoyah, will be considered as
23 requirements also for other ice condenser plants.

24 In conclusion now, since the TMI short-term Lessons
25 Learned items have been implemented at Sequoyah, and placing the

1 Sequoyah plant in the same risk space as Surry and the Peach
2 Bottom station, both of which are operating; since aggressive
3 applicant and staff programs are in place to improve the hydrogen
4 management, capability at Sequoyah in the timeframe of the next
5 four months; since preliminary work shows the integrated distribut
6 ignition system to be a very promising approach, at least for a
7 substantial fraction of scenarios that lead to degraded cores;
8 furthermore, since backup programs are in place, should the
9 IDIS prove unacceptable, the staff concludes that full power
10 licensing of Sequoyah need not await completion of these near
11 term studies and experiments.

12 MR. KERR: Are there questions?

13 Thank you very much, Mr. Butler. The agenda now has
14 scheduled a presentation by RDA Associations. Mr. Hubbard, are
15 you going to start?

16 MR. GREGORY: May I ask a question before we start.
17 is it my understanding that the Commission is recommending
18 allowing Sequoyah to run at full power without the ignition
19 system we've been talking about? Is that fair?

20 MR. KERR: I think it has reached the position that
21 they are going to recommend this to the Commission, I believe.

22 MR. RUBENSTEIN: That's true. Between now and December
23 the startup program is such that I wouldn't even want to hitch
24 on December. We feel confident that the changes made by the
25 action plan have improved the risks to Sequoyah and placed it

1 equal with other plants. But in any case, until the resolution
2 of the interim distributed ignition system occurs in December,
3 potentially 25 effective full power days of operation will occur.

4 MR. KERR: Mr. Hubbard?

5 MR. HUBBARD: My talk is going to be very short, which
6 the Chair will be glad to hear, I'm sure, as much to catch an
7 airplane as any other reason, however. But I made the talk length
8 proportional to the length of time RDA has been involved in this
9 subject of containment safety, which would be about a minute.

10 Our involvement came through a request from Commissioner
11 Gilinsky to respond to a set of fairly specific questions about
12 Sequoyah containment. And although we have a group of people at
13 RDA who are intensely interested in the subject, we have spent a
14 very short time on it. The total effort was about four man weeks
15 and split between a look at the structure and hydrogen problems.
16 So I think it's probably a little unlikely that we're going to be
17 able to say anything to you that you haven't heard before,
18 other than our opinions on the matter.

19 The first thing that struck us was, of course, that one
20 is constrained by rates, and that at some rate of zirconium
21 reaction and burning of hydrogen, igniters don't do anything for
22 you. And we took the base case of no ice present and 1% per
23 minute zirconium reacting, and that saturates the heat removal
24 capacity of the plant.

25 So whatever else we have to say about it, we're talking

1 about what can one do if the rate of production is lower than
2 that, and I don't know of any data that limit the rate to that
3 value. In case the ice is present, I understand that Donald C.
4 Cook's study indicated about twice that value to saturate the
5 heat removal capacity.

6 So, one might ask then what next, assuming that the
7 rate is lower than that. What one would like to do is use the
8 pilot light approach rather than the light in the kitchen, and
9 that is to surround all the possible sources of hydrogen produc-
10 tion with a little flame so that you get it immediately, before
11 it can build up in concentration. And, of course, probably that's
12 how the igniters will be distributed somewhere near. But as a
13 practical matter, you can't accomplish that goal, so gradually
14 the concentration of hydrogen will build up in the containment,
15 and there's probably no way to avoid that.

16 Therefore, there seemed to be two ways to go with
17 igniters, that being the case. One would be to try to keep the
18 concentration below 9% or 10%, whatever is the value, wherein
19 propagation is rapid. The other would be --. And in that case,
20 I haven't heard any data that exist that would give you informa-
21 tion that tells you how many you need in that case. I'm sure
22 it's not 30. I mean I'm sure, but after all, if you spread them
23 uniformly, that's 30 in 30,000 cubic meters. It stretches the
24 imagination that one per 1000 cubic meters is going to suck in
25 all the mixture and burn it.

1 The other approach is to say well, we'll keep track of
2 the concentration and let the igniters go when you reach some
3 limit at which burning is more rapid, 9% or 10%. And in which
4 case, then, you are stuck with burning whatever is in the compart-
5 ment that's isolated. You know, the structure is compartmentalized
6 and the problem with that, as I see it, is the upper compartment
7 is very large and 10% of it is like 2000 cubic meters. A 10%
8 concentration is like 200 kilograms of hydrogen; that will
9 create -- you know, burning that will create a pressure too
10 rapidly for the heat removal equipment to handle. It looks like
11 two-thirds of the case that we examined at 300 kilograms of the
12 whole containment, which would be close to four atmospheres.

13 So it seems to us that because of the uncertainty in
14 the production rate and the uncertainties in the mixing in the
15 lower containment; after all, it's produced rich and becomes lean
16 as mixing takes place, probably largely in the ice compartment,
17 the uncertainty of where you ought to put the igniters --
18 until you know those facts and the lack of information about
19 igniters, we concluded that it's probably not a reasonable thing
20 to expect that igniters could play a role of pressure control
21 until those data were available, at least on the slow burning.

22 There's another thing that has to be added to that,
23 some of the judgment that went into that has to do with kind of,
24 I would say, a pathological task that RDA, against accepting
25 the output of large systems' computer codes, without the

1 adjunct of a large experimental program to go with it, and I
2 haven't any experimental data that would be needed for this use.
3 So that was the reason -- I mean, that in a nutshell summarizes
4 our feelings about the subject.

5 In addition, I think one should point out that if the
6 ice is there, and if the sprays are working, they do remove
7 steam from the hydrogen-steam mixture that's produced. And it
8 seems to me any fluctuations in the containment have a chance,
9 therefore, for leaving you with a dried out mixture that's rich
10 enough to possibly detonate. And it was just mentioned -- I'd
11 be very leery of detonating 10 kilograms of hydrogen next to
12 the containment wall.

13 MR. SCHOTT: As a gas?

14 MR. HUEBARD: You don't think so? The energy content
15 of 10 kilograms is several hundred of high explosive, and just
16 on an energy basis alone --. And if you have a thick enough
17 layer, I mean, the speed that counts is not the detonation speed,
18 you know it's a very sharp spike, but it's just the sound speed.
19 So you'll have a pressure there for a few milliseconds and
20 that's something that really ought to be looked at.

21 So the other part of the question that we were asked
22 to speak on was what about alternatives. And it's very clear
23 that nobody likes inerting a containment, which would solve this
24 problem and create a lot of others, I guess. And I must say,
25 as was pointed out before, it's an awfu' lot easier to find the

1 problems than it is to come up with solutions. And I guess our
2 total time on this specific problem was maybe a couple weeks, so
3 I think we really can't add anything to the programs that are
4 ongoing here already.

5 MR. KERR: Are there questions?

6 MR. GREGORY: You indicate, I think, in your written
7 material, and you were alluding verbally, that you might prefer
8 to go the inerting route rather than the combustion route.

9 MR. HUBBARD: I mean, it just avoids the problem.
10 That was the point. It raises a lot of other problems.

11 MR. GREGORY: May I ask if the containment will fail
12 or not with the pressure rise that's brought about simply by
13 the release of the extra gas? If we get something like 70% of
14 the cladding material reacted, and all that hydrogen comes out
15 into the containment, just the pressurized use of the extra
16 gas that's in there, will that cause the containment to fail or
17 not?

18 MR. HUBBARD: I don't think so, no. It's just the
19 energy released in producing the hydrogen that's split evenly
20 between hydrogen production and burning. It's the added
21 energy, it's not the gas.

22 MR. KERR: I think the answer is no, but I expect it
23 depends on the scenario that one postulates.

24 MR. BUTLER: That particular problem was considered
25 and reported in SECY 80-107, wherein we made the determination

1 that pressure, although above design pressure, is well below
2 the failure pressure of the ice condenser.

3 MR. KERR: Thank you very much, Mr. Hubbard. That
4 brings me to TVA, and I expect having listened this long, you
5 would like to be heard. In fact, the parenthesis says "new
6 information".

7 My name is Wang Lau with TVA, Nuclear Engineering
8 Branch. We also have a few people from Westinghouse, Offshore
9 Power and also from our home office here today. They can answer
10 questions later.

11 We were asked to give a 15 minute talk about new
12 information. Since we have not talked with this particular sub-
13 committee we weren't sure exactly what new information was or
14 old information. But I heard from this morning and this afternoon,
15 a lot of questions were asked, a lot of information was presented.
16 Of course, I will not duplicate that information. I do intend
17 to cover some of the things that were not covered this morning.
18 Not with the intention of presenting a complete story in 15
19 minutes because that's not possible to do, but I will try to use
20 my presentation to kind of stimulate your questions so that some
21 of the questions that were asked this morning if they were not
22 answered properly, you can try me again and see if I can do a
23 little bit better.

24 The first thing I want to cover is the igniter system
25 a little. Let me tell you some of the major topics I will cover

1 and I want to cover the installation and operation of thermal
2 igniters, locations and things like that. And I will briefly
3 mention spark igniters that we considered at one time, and
4 about the Singleton Lab tests, the Fenwall tests, a little bit
5 about the halon study and some information about degraded core.

6 The igniter system looks something like this. It has
7 a General Motors type diesel igniter with a transformer in there
8 to drop it to the proper voltage; a little ring shield at the
9 top to stop the containment spray from getting to it directly,
10 and the assembly is pretty rugged. We have sent samples of
11 these igniters with the whole assembly to Livermore and to NRC
12 and to Fenwall and, of course, our own Singleton Lab for testing.

13 This is just a simple electrical circuit.

14 MR. LAWROSKI: That will do it for our Chairman,
15 anyway.

16 MR. LAU: I beg your pardon?

17 MR. KERR: He is alluding to the fact that at one
18 point I was known as an electrical engineer, and he thinks that
19 will be the one thing that I understand about today's presenta-
20 tion.

21 (General laughter.)

22 MR. LAU: We have some electrical engineers in our
23 task force and I do have to keep them busy, you know.

24 (Laughter.)

25 This is a rough outline of our Sequoyah plant, and

1 you notice there are six different elevations. These elevations
2 are sealevel elevations. This is a very busy diagram. I'll
3 show a little bit better one.

4 This gives you a little better view. I also intend
5 to show you this one later. Now, 669 is the floor level, basement
6 level, and you can come up. This elevation is directly under
7 the air return -- I mean, the ice condenser. These numbers are
8 the number of igniters at these elevations, in those particular
9 planes. There's four in the ice condenser in the upper plenum;
10 five in the lower plenum; six directly under the ice condenser;
11 and those are in what we call the deadend volume, but they are
12 pretty wide open really. Notice the colors, because going to
13 the next picture, it will show you a little about the detonation
14 thing you will consider.

15 The four yellow ones are the steam generators. These
16 four are, of course, air coolant pumps; this is the pressurizer,
17 this is the reactor cavity. The ice condenser goes 300° around
18 this -- may even go down to 60° for refueling and other things.
19 These two locations are the air return fan locations, and they
20 are not under the ice condenser, they are just within the 60°
21 opening and they suck directly from the upper compartment into
22 there.

23 This is the pressurizer joint tank. This is where
24 the reactor vessel vent discharge point is. Those are the
25 locations of interest.

1 MR. SIEGEL: What is the reactor vessel vent?

2 MR. LAU: Yes, we have a reactor vessel vent since
3 TMI, and that is just an opening in case you have that bubble
4 that we talk about like at TMI inside the reactor vessel. We can
5 vent it out. It doesn't get trapped inside. That's a new
6 feature.

7 MR. SEIGEL: That can be controlled from the control
8 room?

9 MR. LAU: Yes, sir.

10 MR. SCHOTT: Is that up or down in this picture here?
11 Downward into here?

12 MR. LAU: It's vented out to the lower compartment
13 just outside the ice condenser doors, near the ice condenser
14 doors. In this picture, it would be around elevation 700, plus
15 or minus. Okay?

16 MR. SCHOTT: That's a duct from someplace.

17 MR. LAU: There is the reactor vessel vent.

18 MR. SIEGEL: Where is the nearest igniter to that?

19 MR. LAU: Okay. Let's see if I can remember all those
20 figures. Same color as the igniter, orange. This is one
21 elevation, that's 689, the basement. To give you a rough idea,
22 the orange dots are the igniters. Those numbers don't really
23 mean a whole lot. The A, B, C means the reclose(?) of the
24 diesel generator because it is backed up by emergency power and
25 seismic and all that. The Glow represents the glow plug. L is

1 the emergency lighting fixture that we are not using.

2 MR. SEALE: All of the G's on here are glow plug
3 locations, right?

4 MR. LAU: Yes, sir. And this is at the basement.
5 The (?) was scattered, and this is a little high elevation,
6 700 under the ice condenser. That's 731. You'll recall that
7 731 is around there, inside the ice condenser in the lower
8 plenum. It's kind of like a manifold as far as the ice
9 condenser is concerned.

10 MR. SIEGEL: Are you saying that the nearest plug is
11 inside the ice -- the nearest plug to the reactor vent is --

12 MR. LAU: No, sir. We've got -- The nearest plug to the
13 reactor vessel vent? Nearest plug to what?

14 MR. SIEGEL: To the exit from the reactor vessel vent.

15 MR. LAU: No, there are a few in the lower compartment.
16 The reactor vessel vent is venting to the lower compartment,
17 and there are --.

18 MR. KERR: What is the nearest glow plus? One foot,
19 10 feet? 20 feet?

20 MR. LAU: Let me see. In this drawing, this is 90°,
21 so this is around 45° at around 731 elevation. So let's take
22 that 731 elevation at 45°; I would say the igniters are located
23 in this area. This area is about 65 feet, so this is about, oh,
24 say about 20 feet.

25 MR. KERR: Okay. Is that a good enough estimate? Good.

1 MR. SCHOTT: It's a good enough estimate but it's not
2 close enough.

3 MR. LAU: This elevation, 792, has something interesting
4 in one corner, and that is the location of the control panel in
5 the auxiliary building. And this control panel is approximately
6 100 feet from the main control room. You can run out there and
7 throw the switch and run back. But these igniters are manually
8 controlled, not from the main control room. You do have to get
9 out there, but there's no radioactive stuff at that elevation;
10 there are electrical things.

11 MR. KERR: I should point out that it's very unlikely
12 that any hydrogen would ever come out of that vent. So if
13 you're going to put the glow plugs there to catch the most likely
14 exit spot of hydrogen, that's probably not it.

15 MR. GREGORY: I thought I heard somebody say that the
16 purpose of the vent was to let out the " bubble."

17 MR. KERR: Well, that was the Three Mile Island bubble
18 which maybe was there and maybe was not. But remember, at
19 Three Mile Island there was a lot of hydrogen released and it
20 didn't come out through a vent. And it may not come out through
21 it --. In fact, I think it somewhat unlikely that it will come
22 out the vent the next time.

23 MR. SIEGEL: It came out of a vent which happened to
24 be the exist from the pressurizer, but if you provide another
25 more suitable vent, why shouldn't it come out there?

1 MR. KERR: But this suitable vent is only the vent in
2 case one develops a bubble in the top of the vessel and can't
3 get rid of it.

4 MR. GREGORY: So if you ever use the vent, you will
5 get hydrogen coming out.

6 MR. KERR: That's true.

7 MR. GREGORY: And I would not like to operate a
8 kitchen range, for example, where the pilot light is 20 feet
9 away from the burner. That doesn't seem to be --

10 MR. KERR: I defer to your wisdom.

11 MR. SEALE: I think Dr. Kerr's point might be, though,
12 that it's equally important to ask whether or not there's a glow
13 plug near the relief valves for the pressurizer.

14 MR. KERR: I'll take credit for that viewpoint.

15 (Laughter.)

16 MR. LAU: Of course, there's a pressurizer relief in
17 Sequoyah that relieves to the reactor coolant drain tank, the
18 tank that I showed earlier, and from there, there is a
19 rupture. Just like in TMI. I mean, that rupture -- the
20 hydrogen will be dispersed in the lower compartment.

21 MR. SIEGEL: Do I understand that the location of the
22 glow plugs is really constrained by the location of your emer-
23 gency lighting system?

24 MR. LAU: I'm not so sure that constrained is the
25 proper word, but this is --

1 MR. SEIGEL: Determined by?

2 MR. LAU: In the interim system, the first phase, we
3 do want to go in there and use whatever is available. Our reason
4 is that the emergency lighting circuit is usually located in a
5 place, by design, that is wide open to give good illumination
6 of what is going on. So the coupling is pretty good because of
7 that. And the conduit is there, it is seismic and backed up by
8 diesel and the location (?) is good, and that's what we
9 used.

10 Later on, in our future study if we do find out that
11 there are better locations such as -- we might reroute the
12 reactor vessel vent, we might put the igniters further away and
13 we might put them closer.

14 MR. KERR: Mr. Lau, I think if we get answers this
15 long to every question that we ask; I think he has the answer
16 to his question now.

17 MR. LAU: Thank you. Three more in the very top
18 dome.

19 I have here a drawing showing the approximate -- it's
20 a hydrogen scheme, designed for that purpose -- to collect hydro-
21 gens at various locations especially at the deadend volume and
22 the steam generator enclosures and things like that. So we can
23 use the air return fans to kind of circulate the hydrogen burner.
24 Get better mixing.

25 As Mr. Butler said earlier, the air return fan has

1 something like 40,000 cfm speed and you would have a complete
2 air change in the lower compartment in about four or five
3 minutes. So it is a highly turbulent, well mixed case.

4 I heard a lot of questions this morning and this
5 afternoon about pocketing, detonation and so on and so forth.
6 I think we should keep in mind that the flammability lower limit
7 is around 4%, and you'll get a lot better chance of igniting at,
8 say, 8% or 10%. The detonation limit we're talking about, 18%
9 and higher.

10 Now, there is a margin of, say, about 10%. We are
11 therefore talking about concentration gradient in space of
12 around 10%. With this kind of mixing and with the wide open
13 areas and with all the turbulence inside and in the upper compart-
14 ment we have the sprays, of course people can always argue with
15 me, but I really think that we have pretty good mixing, such
16 that that kind of concentration gradient is not exceeded,
17 considering we have 30 igniters and they are not that far away.
18 Okay? But right now I do not have any numbers to back up what
19 I've mentioned about this concentration gradient.

20 Any questions so far? Okay. We'll talk a little
21 bit about some of the Singleton tests we have. We have endurance
22 tests and it looks like -- you can always fail a plug. If you
23 intentionally put overvoltage on it, we have successfully
24 ruined a few of them at at 16 volts or so. But if the reactor
25 coolant fails, it usually fails within about the first 30 minutes

1 or so. At a bout 14 volts we are pretty successful in keeping
2 it alive. Now, some of the old plugs we bought they fail. I
3 think it's because of the shelf life or maybe there's some
4 moisture in them or something like that. But all the 1979 plugs
5 are pretty successful, and I guess the Lawrence Livermore people
6 use the plugs and they're okay, too.

7 The endurance tests -- we are planning more tests. We
8 are buying large samples, a kind of quality assurance sampling
9 test to establish some kind of confidence level on an igniter.
10 We use, of course, a future installation of an igniter or a
11 replacement igniter and we prove it would work.

12 We feel very comfortable with the temperature require-
13 ments we've put on the igniters. That's important because we
14 want to prove that the igniter works.

15 The staff mentioned this afternoon that the spontaneous
16 ignition temperature in the ideal situation, dry air, is around
17 1100^oF. Now, we are using about 1700^o. Something above 1500^o
18 is our criterion, so there's plenty of margin there. The steam
19 concentration does not affect the spontaneous ignition tempera-
20 ture that much; according to some of the published literature
21 from the British research, the steam concentration in the range
22 we are talking about might add about 50^oC or maybe 100^oF to
23 the spontaneous ignition temperature. So we have plenty of
24 margin, we feel very comfortable with the fact that you will
25 ignite. Certainly at the 8% and 10% range. And, of course,

1 we'll confirm those in our future testing.

2 MR. STREHLOW: I've got a comment here. Remember we
3 said that if you have a 2% hydrogen mixture you'll probably get
4 low localized chemical reaction in the neighborhood of the
5 glow plug; they hydrogen would not burn but would be decomposed,
6 it would oxidize; it wouldn't really be a flame. There's a
7 possibility that this might shorten the life of the plug
8 considerably, and if you're considering putting these plugs on
9 and leaving them run for weeks to make sure that there's no
10 hydrogen buildup that's going to go to explosion, and you soak
11 them in a 2% hydrogen mixture, they may burn out in a few days.
12 And that's one --

13 MR. KERR: You do not intend to leave them on --?

14 MR. LAU: We have some of the igniters in the lab and
15 we've been burning them for over a week.

16 MR. STREHLOW: In a hydrogen atmosphere?

17 MR. LAU: Not in a hydrogen atmosphere.

18 MR. STREHLOW: That's the point I'm making. I'm
19 making the point that if you put them in a hydrogen atmosphere
20 of 2% or 3%, the plug's temperature will be higher than the
21 temperatures that you measure because of the oxidation reaction
22 occurring at the surface. And this will cause the plug tempera-
23 ture to rise above the temperatures. And the thing that burns
24 those things out is to get them too hot.

25 MR. KERR: This could be a problem in the long term

1 evolution of hydrogen, but I would guess that it's more likely
2 that that will be slow and would be thought of as being taken
3 care of by recombiners, if recombiners are still in existence.

4 MR. SIEGEL: My impression was that this could be a
5 serious point. If you're doing this for a couple of days, you're
6 in a time that's short compared to where recombiners are in
7 place but long compared to a deterioration time of the plug.

8 MR. KERR: You may be right.

9 MR. DILWORTH: This is George Dilworth, TVA.
10 Recombiners in our containments are in place. They're inside the
11 containment, two of them, part of the design basis for this plant.
12 And we do not intend to run the igniters for extended periods of
13 time. When the hydrogen is produced through the S2D situation
14 that we've been talking about today, after that's over with, those
15 9 burns, the igniters will not be left on. The recombiners are
16 sufficiently capable of taking care of any small residual hydrogen
17 that might be released.

18 MR. KERR: However, one should not ignore the fact
19 that nature does not necessarily recognize all of the S2D as
20 a possible method of evolving hydrogen, and it seems to me it
21 is possible that one might get something slow compared to that
22 and fast compared to recombiners.

23 MR. LAU: At one time, we considered spark igniters.
24 They are definitely a very efficient ignition source. We kind
25 of dropped it for our Phase 1 program, interim program, because

1 we were not sure of the effect due to electromagnetic interfer-
2 ence, and we are contracting people to make a study and go ahead
3 with a design of a (?) or something of that nature and
4 maybe if it's successful we'll pick it up again. But, of course,
5 we have to worry about electromagnetic shield. We might also
6 have a magnetic effect on the flame burning. So that's being
7 studied.

8 MR. LEE: Are these glow plugs used in any major
9 ways other than in diesel engines?

10 MR. LAU: I do not know of any.

11 MR. LEE: Just diesel engines? That's where they
12 are used?

13 MR. LAU: Yes.

14 MR. LEE: Are any controlled ignition devices in use
15 for controlling hydrogen in any industry?

16 MR. LAU: Of course, the Bureau of Mines has a lot
17 of -- something of this nature with igniters, but they're not
18 for hydrogen. They're for various types of gases. And there
19 are other industries, the chemical industry uses igniters, but
20 not for the purpose of hydrogen that I know of.

21 The Fenwall test will have a seven foot diameter test
22 vessel. We intend to push droplets in it later on, and we
23 intend to have a fan circulating the atmosphere inside to make
24 sure that the burning is still there.

25 By the way, a couple days ago we had a kind of

1 shakedown demonstration at Fenwall. I hesitate to call it a test
2 because it was not one of the scheduled tests. And they put
3 about 12% hydrogen in there in dry air, and they got ignition.
4 They measured pressure pounds, and don't quote me on the final
5 pressure we got because I don't know anything about calibration
6 or pressure sensors or anything like that. But they received
7 about a 67 psi gauge, which is pretty much what you'd expect
8 in an adiabatic burn in a closed volume with no venting.
9 12% hydrogen. The pressure rise was about .4 second. Since
10 the vessel was 7 feet in diameter --.

11 MR. LAWROSKI: Where was the decimal on the second?

12 MR. LAU: Zero point four (0.4). Since it was
13 a seven-foot diameter, the igniters were put in the center, so
14 the distance from the center to the sides is about three or four
15 feet, so that would give you a flame speed of approximately 10
16 feet per second, plus or minus. That's exactly the kind of
17 thing we're talking about.

18 MR. DILWORTH: Wang, just a minute. This is George
19 Dilworth again. To save time, we're going to have a more
20 detailed presentation on the 5th to the full Committee of the
21 Fenwall test. I think it would be better to wait until that
22 time rather than receive a lot of questions on those tests today.

23 MR. LAU: Okay. Our schedule for those tests is in
24 the handout, and we intend to have a first phase of the testing
25 done by approximately the last part of September.

1 The halon study. We have contracted a contractor to
2 study halon with Duke and ADP joining in that study. The major
3 thing to study, of course, is the biological effects if it's
4 virtually activated, and also, whatever the decomposition product
5 may do to your system.

6 MR. KERR: That doesn't require a lot of study, does it?

7 MR. LAU: It's a very funny thing. This decomposition
8 product I understand will reach some kind of equilibrium.
9 At least, that's what the chemical engineer told me.

10 I want to conclude by saying that we are using the
11 S2D or MARCH code to study. We do not know that certainty
12 evolved in the MARCH code because TVA does not have the capability
13 to generate the MARCH data. We are using the CLASIX code to
14 make the containment study. We are testing to make sure that
15 igniters work.

16 Our approach is that we want to make sure that a
17 hydrogen source be used that will find a fair range of accidents.
18 We want to make sure that the containment analysis using the
19 CLASIX code predicts situations that we can stand in our contain-
20 ment, and we want to make sure by testing and whatever, the
21 literature and available data, to prove that our assumptions
22 and things like that used in the computer model are accurate.
23 And we believe that that's a logical approach to the solution.

24 MR. KERR: Thank you, Mr. Lau. Questions? This brings
25 us to a presentation from NSA by Mr. Miller.

1 MR. MILLER: My name is Al Miller. I am
2 summarizing the workshop that we held in Palo Alto back in
3 March, where we gathered a bunch of experts to talk about
4 containment stability.

5 There is a summary of that workshop available to
6 anyone who wants it. I will leave this one here for the
7 committee if they would like.

8 I will race through these first slides. I think
9 that we will see the purpose. This is the hydrogen burn.
10 We have a little more detail on that showing the apparent
11 pressure rate followed by a generalized increase in
12 temperature all the way around the containment, somewhat
13 implying a generalized burn throughout the containment, and
14 the tree balance showing somewhere around 50 percent metal
15 water reaction taking place there.

16 These are all in the document, if you would like
17 more detail. I will not dwell on them. The conclusions
18 from our workshop -- these are conclusions from our
19 workshop, and not necessarily NSAC conclusions. The first
20 conclusion was that the Three Mile containment was not
21 challenged by the hydrogen burn. It would not have been
22 challenged if there had been 75 percent or even 100 percent
23 metal water reaction.

24 The second conclusion was, there is some
25 insecurity in being able to predict the pressure rises in

1 large containments. There are scale effects, and these
2 should be studied further. I will go into what Avery is
3 doing with respect to that a little bit later. We will talk
4 about that a little bit later.

5 The most important conclusion was the third that
6 the highest probability for successful mitigation of
7 hydrogen accumulation problems was controlled ignition,
8 perhaps coupled with the use of water sprays.

9 The filler venting concept received almost no
10 support. Spark ignition of hydrogen in concentrations less
11 than 9 percent was unreliable. There is some data by
12 Rockwell that show that they had some difficulty igniting
13 hydrogen concentrations less than about 8 percent in a
14 reliable manner.

15 The inerting of containment atmosphere is a
16 possible option but there is very great concern because of
17 the occupational hazards that it poses. The experience in
18 India was mentioned where they lost a few people.

19 The use of halon to suppress detonations was
20 talked about, but again the questions that were raised by
21 TVA were again raised here.

22 MR. LAWROSKI: The regular halon.

23 MR. MILLEP: Halon 1301 is the one that the data
24 is on.

25 MR. LAWROSKI: They may to date not have been very

1 serious.

2 MR. MILLER: Halon is used quite a bit. I know
3 Three Mile has a halon system.

4 A few final words on the work that will be done at
5 Avery, or what we are proposing to do at Avery.

6 We are proposing some large scale testing,
7 probably at the Nevada test site, probably in a 50-foot
8 diameter sphere, 100 psi steel sphere, looking at the scale
9 effects. There will be room to put in fans, obstructions,
10 promoting turbulence, testing different kinds of ignitions,
11 ignitions of various sources, different percentages, perhaps
12 going up to detonable mixtures by using subsatmospheric
13 conditions.

14 That in a nutshell is about what I have to say.

15 MR. KERR: Thank you, Mr. Miller.

16 Referring to one of your conclusions that ignition
17 is unreliable in concentrations of hydrogen less than 8
18 percent for spark ignition, did you consider only spark, or
19 did you talk about glow-plugs, for example?

20 MR. MILLER: Yes. Some of that data is in the
21 blue bound volume. Spark ignition via the AI work, the
22 Atomic International work was not reliable, whereas using a
23 detonable mixture to ignite was reliable, and using a flame
24 to ignite was reliable at those regions, even down to 2 or 4
25 percent.

1 MR. KERR: Did the glow plug get any
2 consideration.

3 MR. MILLER: The work there was to look for
4 pressure response, and not ignition types.

5 MR. KERR: Did you workshop talk any about what
6 was deemed to be an appropriate way of considering the
7 amount of hydrogen involved? Is there a consensus that one
8 ought to consider, 100 percent metal water reaction, or 50
9 or 70, or 20? Is there an approach to how one might decide
10 on an appropriate amount to consider, or do you have any
11 suggestion for the committee?

12 MR. MILLER: That was a subject of the workshop.
13 The workshop was, you have got it, now what do you do with
14 it, what are the responses.

15 MR. KERR: Your conclusion is that burning looks
16 like a promising approach, but more questions need to be
17 answered before one is certain of that?

18 MR. MILLER: Exactly, yes.

19 MR. KERR: Are there other questions?

20 MR. LEE: This unreliability of ignition at around
21 8 percent or below that, in light of one of these slides
22 that we saw earlier in the afternoon which indicated a steep
23 increase in pressure somewhere around 8 percent
24 concentration, does that mean that we have a very narrow
25 window in which we should operate our glow plugs?

1 MR. MILLER: The work at AI dealt with sparkplugs.

2 MR. LEE: I understand that.

3 MR. MILLER: Small sparkplugs, 120 sparks per
4 second sparkplugs, and the correlation between that and the
5 glow plugs is questionable.

6 MR. LAWROSKI: May we not also forget getters,
7 too, hydrogen getters?

8 MR. MILLER: We talked about getters at the
9 workshop.

10 MR. LAWROSKI: If you get one that is truly good,
11 you are going to have problems with oxygen.

12 MR. MILLER: We talked about them at the workshop,
13 and they did not receive too much support. The R&D types
14 are excited about it because it is a fantastic R&D project.

15 (General laughter.)

16 MR. LAWROSKI: They have their place at the
17 battery. You have an inventory of hydrogen. If you have a
18 getter for this purpose, wouldn't you agree that you are
19 kind of fighting the battle between something that is going
20 to take the hydrogen out without getting goofed up.

21 MR. GREGORY: I think that if you are not going to
22 inert the containment, then the best thing to do with
23 hydrogen is to burn it in a controlled way. If you are
24 going to inert the containment, then you still have to get
25 rid of the hydrogen somehow, and then you would use a

1 getter.

2 MR. LEE: In order to get this type of benefit out
3 of our sprays, what kind of spray system do we need compared
4 to what we have right now?

5 MR. MILLER: These nozzles were standard. The
6 numbers are in here, but they were standard BWR nozzles in
7 existence. I have forgotten the number. They were standard
8 nozzles giving you a 500 micron droplet.

9 MR. LEE: So we don't have anything other than
10 guaranteeing that the sprays will work.

11 MR. MILLER: I am sure that you could optimize
12 that. But there was no attempt made in this study to
13 optimize.

14 There were questions concerning the actual heat
15 sink supply is not really big enough to give you that
16 pressure suppression. So there were questions concerning,
17 was there a mechanical energy transfer; was a lot of the
18 energy of that deflagration going into actually breaking the
19 droplet apart as opposed to actually heating up and
20 vaporizing the droplet. You really could not account for
21 that pressure decrease just in a heat transfer heat sink
22 type mode.

23 There is certainly a lot of room for
24 optimization. This was just using standard BWR spray
25 nozzles.

1 MR. LAWROSKI: My acquaintance with the
2 suppressants -- were there any really good ones that were
3 talked about at the workshop?

4 MR. MILLER: Von Alby was there, and the one
5 mentioned by him was the Halon 1301, the CF3BR. I think
6 that is the one most commonly in use today.

7 MR. SEALE: I would assume your big Nevada site
8 test with the big container, and so on, will include some
9 glow plug work?

10 MR. MILLER: Yes. A variety of ignition sources
11 are planned to be evaluated. I can virtually guarantee you
12 that the TVA ones will be used as one of the options. It is
13 not my project, but I can almost guarantee that they will be
14 used.

15 MR. SEALE: Okay.

16 MR. MILLER: The schedule on that is that we are
17 hoping to be starting initial testing next summer, next
18 September, somewhere in that area. Money has been allocated
19 by our task forces. It hasn't been allocated by our Board
20 of Directors. We are looking for co-funding. There are a
21 bunch of options in the mill.

22 MR. KERR: Is your work coordinated with, in the
23 sense that you are going to take full account of the work
24 being done by Livermore?

25 MR. MILLER: Yes. We are trying to stay closely

1 involved with all of the work, the Livermore, the Sandia,
2 and Fettel-Frankfurt.

3 MR. KERR: Are there further questions.

4 (No response.)

5 MR. KERR: Thank you, Mr. Miller.

6 (Whereupon, at 4:15 p.m., the Committee went into
7 executive session.)

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EXECUTIVE SESSION

1
2 MR. KERR: The executive session is now in
3 session. It is an open executive session.

4 I would like your comments on the two questions
5 raised by Mr. Gilinsky, if you have any at this point, since
6 we must deal with these as a committee at our next meeting.
7 Do you have any suggestions for me, or the subcommittee?

8 Do those members of the subcommittee have any
9 suggestions as to appropriate comments in response to Mr.
10 Gilinsky's questions, the first one being, "Does the
11 committee believe additional hydrogen control measures are
12 necessary for ice containments?"

13 MR. SIESS: Additional to what?

14 MR. KERR: I don't know. I assume that it is
15 additional to what now exists.

16 MR. SIESS: By what now exists, do you mean what
17 was required of Sequoyah?

18 MR. KERR: I am not sure.

19 MR. SIESS: We wrote two letters in whatever month
20 it was, July. We wrote one letter on Sequoyah. We wrote a
21 letter to Mr. Gilinsky in response to a question he had
22 raised separately. Did we say anything in those letters
23 about hydrogen control measures for ice condensers?

24 MR. KERR: We commented, I think, about the
25 approach that TVA was taking, and said that it seemed

1 reasonable.

2 MR. SIESS: We had a letter from Mr. Gilinsky and
3 a response to him at the July meeting. I don't have that in
4 my collection here, so I am not quite sure about that. I
5 thought the committee had a position that we did think
6 something additional was required for ice condenser
7 containments over what was already there before Sequoyah
8 started talking about igniters.

9 MR. KERR: My interpretation of the letter, which
10 is probably not as good as yours --

11 MR. SIESS: I am not trying to interpret the
12 letter. I said, a position, but whether we have written it
13 down anywhere, I don't know.

14 MR. KERR: We don't have a position until it
15 becomes written down.

16 MR. SIESS: I know.

17 I think that in some of the letters we have
18 written on other things, we have talked about hydrogen
19 control for ice condensers. But specifically, I guess, the
20 question from Mr. Gilinsky is whether either of the letters
21 we wrote on July 15, or whatever it was, made a clear
22 position, or whether we can cite something else, or whether
23 we should write another letter.

24 MR. KERR: He refers to the letter of July 15th to
25 him on hydrogen control measures at Sequoyah.

1 MR. SIESS: That is the letter I don't have in
2 front of me, or the Sequoyah letter, for that matter, right
3 at the moment.

4 What do you think the committee's position is on
5 hydrogen control measures for ice condenser containments?
6 Didn't we write a letter? Have we said anything about Mark I
7 or Mark II containments on hydrogen.

8 MR. KERR: We have written a letter on TMI lessons
9 learned in which we agreed with the staff's recommendation
10 on inerting Mark I and Mark II containments. I don't
11 remember our having mentioned ice condensers in that same
12 letter.

13 MR. SIESS: My recollection is that we did.

14 MR. KERR: We did not recommend inerting them, did
15 we?

16 MR. SIESS: No. We recommended that they be
17 looked at.

18 MR. KERR: Yes.

19 MR. SIESS: There was a strong indication that at
20 least we thought that there might be something needed to be
21 done about ice condensers.

22 MR. KERR: I think that that is the case, yes.

23 I don't know whether Mr. Gilinsky is talking about
24 the general case or about Sequoyah. I guess we can find out
25 by asking, maybe.

1 MR. LAWROSKI: When we talk about Sequoyah, that
2 is the immediate model?

3 MR. KERR: It seems to me that that would make
4 some sense.

5 MR. LAWROSKI: We must not forget that there will
6 be at least 10 of these things. Sooner or later we are
7 going to have to bite the bullet.

8 MR. SIESS: In the letter to Mr. Gilinsky, we
9 simply said, regarding the control of large amounts of
10 hydrogen "is discussed to some extent in the committee's
11 Sequoyah letter." So that takes care of that. We can now
12 look at the Sequoyah letter.

13 I think that somebody needs to look back further
14 because I thought I sensed a position by the committee that
15 ice condensers, at least, had a problem.

16 MR. KERR: There certainly has been a concern
17 about ice condensers.

18 MR. SIESS: The Committee in its March 11, 1980,
19 report on the VTOL items recommended the licensees develop
20 reliability assessments for their plants, and that design
21 studies of possible hydrogen control in filter vented
22 containments be required. This was mentioned in the
23 Sequoyah letter.

24 I don't see any position in here. I see an awful
25 lot of words, but no position.

1 MR. KERR: Do any of our consultants have advice
2 for us at this point that they would be willing to
3 volunteer?

4 MR. SEALE: I don't know whether it is advice, but
5 it is some comments.

6 It seems to me that the ignition approach is the
7 right way to go. I really shudder to think of all of the as
8 yet unresolved engineering problems you would get into if
9 you tried to do inerting, and so on, at this stage of the
10 game in a system that was designed the other way. It does
11 not seem like a good idea.

12 I am a little concerned about a problem that was
13 mentioned by, I believe, Dr. Strehlow, regarding the
14 possibility of the ducts and the turbulent mixtures in those
15 ducts serving as a sort of a quasi-containment, I guess, for
16 directing some flame fronts, or something like, from one
17 region to another. I am not sure if I am saying it quite
18 right, but I think that that the is the substance of what he
19 said.

20 To me it is kind of interesting that the location
21 of the plugs at this stage seems to be a question of
22 convenience rather than maybe the most strategic place in
23 which they might be located. I think that that can be a
24 dangerous convenience to accept. I would hope that the TVA
25 people would look at it in more detail and if there wasn't a

1 more systematic way of going about locating those things.

2 Certainly, the glow plugs look like the right
3 approach. I think sparkplugs would cause all kinds of
4 problems in control instrumentation and things like that
5 that we have not even thought about yet just because of the
6 electro-magnetic interference effects.

7 In general, I think the idea of going ahead as the
8 staff recommends, recognizing that at some point fairly soon
9 down the road some decisions are going to be available with
10 the glow plug approach, and hopefully putting the glow plug
11 in a more propitious set of locations is the reasonable and
12 prudent thing to do.

13 MR. SIEGEL: I more or less second the comments
14 that Seale has made. I think the glow plugs are an
15 appropriate interim solution to be adopted. I feel rather
16 strongly that the vent system -- you cannot move the glow
17 plugs, the exists from the vent system ought to be somehow
18 adapted to be located closer to one or more glow plugs. I
19 personally would emphasize the more or so, to give some
20 redundancy.

21 As you said, Dr. Kerr, I don't think we can fully
22 identify where the hydrogen is going to come from in the
23 next event, but at least there are some plausible places
24 where it might be planned for. I think that those plans
25 should be coordinated with the locations of the glow plugs.

1 If it is too hard to move the plugs, then let's adapt the
2 vent system, but combine them in some reasonably logical
3 fashion.

4 If this can be done in a timely way so that the
5 proposed operation at one percent power can occur more or
6 less simultaneously with the installation of this combined
7 vent and glow plug system, then I have no reservations about
8 the recommendations of the staff.

9 On the other hand, I would dislike seeing the
10 thing being an interim-interim measure where 100 percent
11 operation continues for some extended period before even the
12 glow plugs come into availability.

13 MR. KERR: If I understand the staff's
14 recommendation, they are willing to recommend that Sequoyah
15 be permitted to go to 100 percent power with the idea that
16 since they are in a testing phase the total amount of 100
17 percent power operation between now and the glow plug
18 approval time is likely to be something like not more than a
19 month full power days.

20 MR. SIEGEL: What concerns me about that
21 particular recommendation is whether operation at 100
22 percent power would impede any additional, hopefully, minor
23 modifications to the glow plug array or the installation of
24 some additional vent system. I would rather have those
25 changes added to the interim plan, rather than to have those

1 changes impeded by the 100 percent operation even for a
2 brief period.

3 DR. FERR: I think the points you are raising are
4 very good. I don't know enough about the ignition system to
5 know where the plugs should be put. It occurs to me that if
6 you put them too close to the source of hydrogen, at least
7 it seems to me in principle, to be possible that you will be
8 igniting the hydrogen in a situation in which the
9 concentration is considerably greater than if it were
10 permitted to diffuse somewhat, which might be an argument
11 against putting it near a vent. On the other hand, if you
12 put it near the vent, then you are more likely to ignite it
13 earlier.

14 What I am saying is that there are possible
15 trade-offs here which I would not know how to answer on the
16 basis of what I have heard so far.

17 MR. SIEGEL: I fully agree with you. There is
18 some optimum place, and that optimum place would be -- It
19 would be extremely fortunate if that optimum was where the
20 emergency lightbulb happened to be.

21 MR. LEE: May I make some comments?

22 MR. KWAN: Please.

23 MR. LEE: I feel somewhat less encouraged,
24 perhaps, about the ignition system than the two gentlemen
25 who have commented on the system, for a couple of reasons.

1 One is, even with the addition of the ignition system, one
2 may not be able to disregard the possibility of local
3 detonations, and with all the compartments, and so on, one
4 might have to contend with.

5 The second point, perhaps, Brookhaven National Lab
6 people reported that one might have to worry about some of
7 the foam padding or cover that exists around the ice
8 condenser, and so on, and not based on my expertise at all,
9 which is practically nothing in this area, but my gut
10 feeling is to rely on flame, even at low propagation speed,
11 I feel a little bit uneasy about it. I would like to
12 explore the possibility of using some suppressant, halon
13 1301 or some other mechanism, or a combination of several
14 possibilities.

15 Those things, perhaps, should be explored a little
16 bit more, perhaps as actively, although I feel the issue of
17 ignition system perhaps has gone far along so that we should
18 certainly try out more either in the test facilities or also
19 in the actual plant in some way.

20 I feel somewhat uneasy about the ignition system
21 partly because in response to one of my earlier questions,
22 if this type of system had ever been used with hydrogen, the
23 answer was no. To try it out on a large containment for the
24 first time, I feel somewhat uneasy.

25 MR. SIEGEL: I would like to add a comment, or a

1 question that Mr. Lee's comments reminded me of. In reading
2 the description, I, too, was concerned about this one word
3 description of the foam insulation around the ice
4 condenser. In view of the things that we have heard about
5 in airplanes and in the BART system, that can be a very
6 suspect material, particularly if you have an atmosphere
7 that is up at 1500 degrees for a few minutes.

8 We did not hear anything about that today in
9 response to my concerns about what else happens when the
10 atmosphere gets up to 1500 degrees for several minutes.

11 MR. KUBER: Mr. Gregory?

12 MR. GREGORY: I would like to agree with the
13 general comments that were made over here that an ignition
14 system does seem to be the way to go, and more consideration
15 should be given to the design placement of the igniters.

16 In general, there are two standard ways of dealing
17 with flammable gases leaking into spaces. One is to
18 deliberately eliminate the ignition sources, and the other
19 is to deliberately ignition sources in a place where you
20 ignite the flammable gas before it gets out of control.

21 The obvious example that we are all familiar with
22 is in our own gas domestic appliances, the pilot flame, or
23 the spark igniter is always located very, very close to the
24 burner. I think we have to look at the same philosophy in
25 the system.

1 It should be possible to make some pretty educated
2 guess as to where the most likely place is that hydrogen
3 will come out into the containment in the event of various
4 types of accidents, and to make sure that there is an
5 ignition source right there.

6 I would like to make one observation that would
7 bother me a little bit about glow plugs. Glow plugs are not
8 normally used in hydrogen ignition or flammable gas ignition
9 systems, whereas standing pilots and spark ignition system
10 have got a record of being used for this kind of thing.

11 We are embarking on an ignition technique which
12 perhaps is not well established in the industry, it just
13 occurs to me that a sparkplug in an engine operates many,
14 many millions of time in its life, and a glow plug only
15 operates for a few thousand times in its life. I just
16 wonder if the spark system should not be, perhaps,
17 reconsidered.

18 MR. KERR: Any further comments on that?

19 MR. SEALE: I will just observe that IEEE 279
20 control rod circuits are a beautiful antenna, and more than
21 once reactors have been scrambled because of high frequency
22 electro-magnetic disturbances.

23 MR. SIEGEL: But hopefully the reactors have been
24 scrambled when these things go on.

25 MR. SEALE: That is right.

1 MR. KERR: Are there any other comments?

2 (No response.)

3 MR. KERR: Let me thank all of you for your
4 attendance and participation. The meeting is adjourned.

5 (Whereupon, at 4:30 p.m., the meeting was
6 adjourned.)

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NUCLEAR REGULATORY COMMISSION

This is to certify that the attached proceedings before the
ACRS Subcommittee on Class 9 Accidents

in the matter of:

Date of Proceeding: August 28, 1980

Docket Number: _____

Place of Proceeding: Washington, D. C.

were held as herein appears, and that this is the original transcript thereof for the file of the Commission.

David R. Parker

Official Reporter (Typed)



(SIGNATURE OF REPORTER)

NUCLEAR REGULATORY COMMISSION

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Suzanne Babineau

Official Reporter (Typed)

Suzanne Babineau

Official Reporter (Signature)

*Page 1/5RB
+ 2*

HYDROGEN BURNING IN
ICE CONDENSER CONTAINMENT

by

Peter Cybulskis

Presentation to:

ACRS Class 9 Accident Subcommittee
Washington, D.C.

August 28, 1980

BATTELLE
Columbus Laboratories
505 King Avenue
Columbus, Ohio 43201

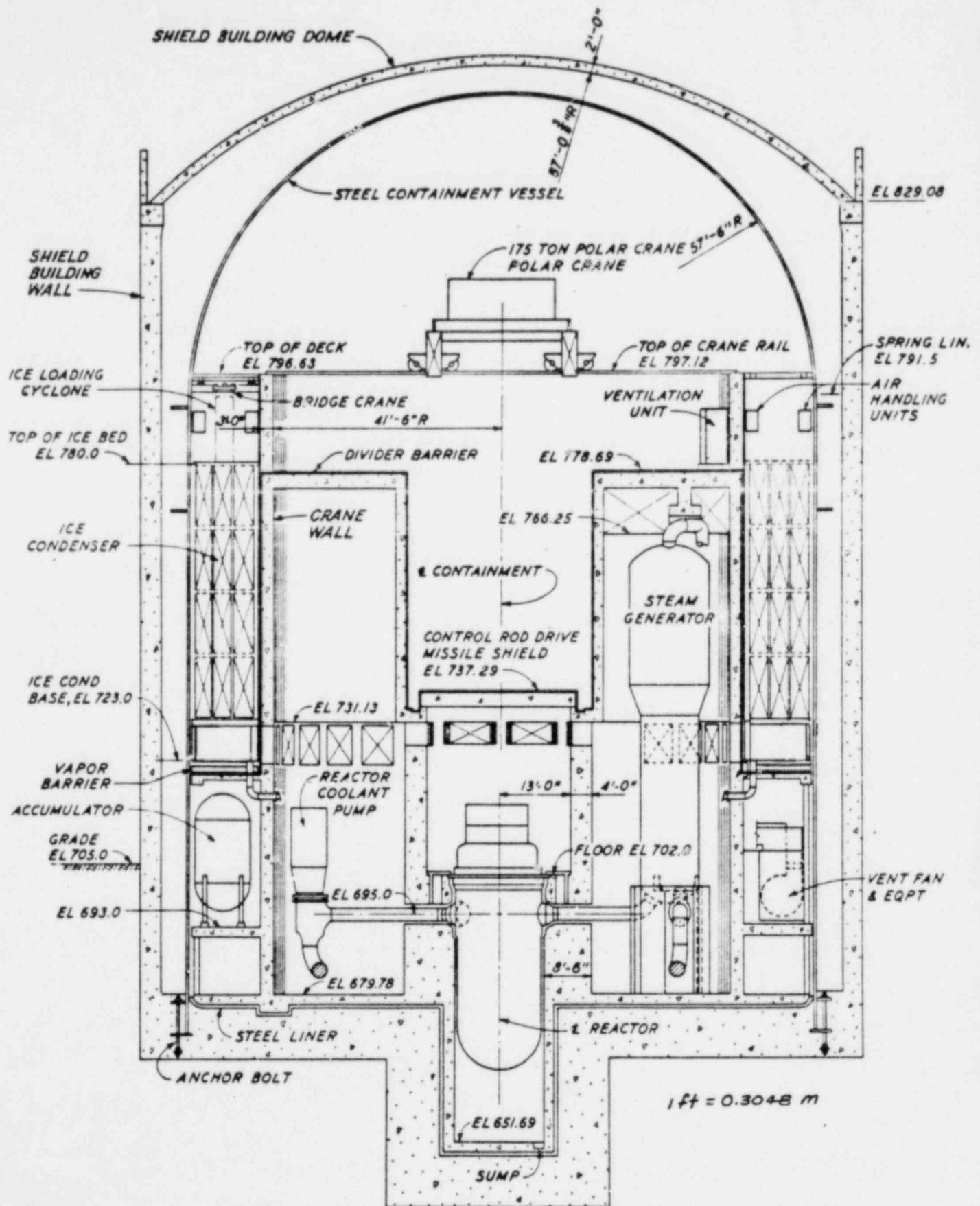
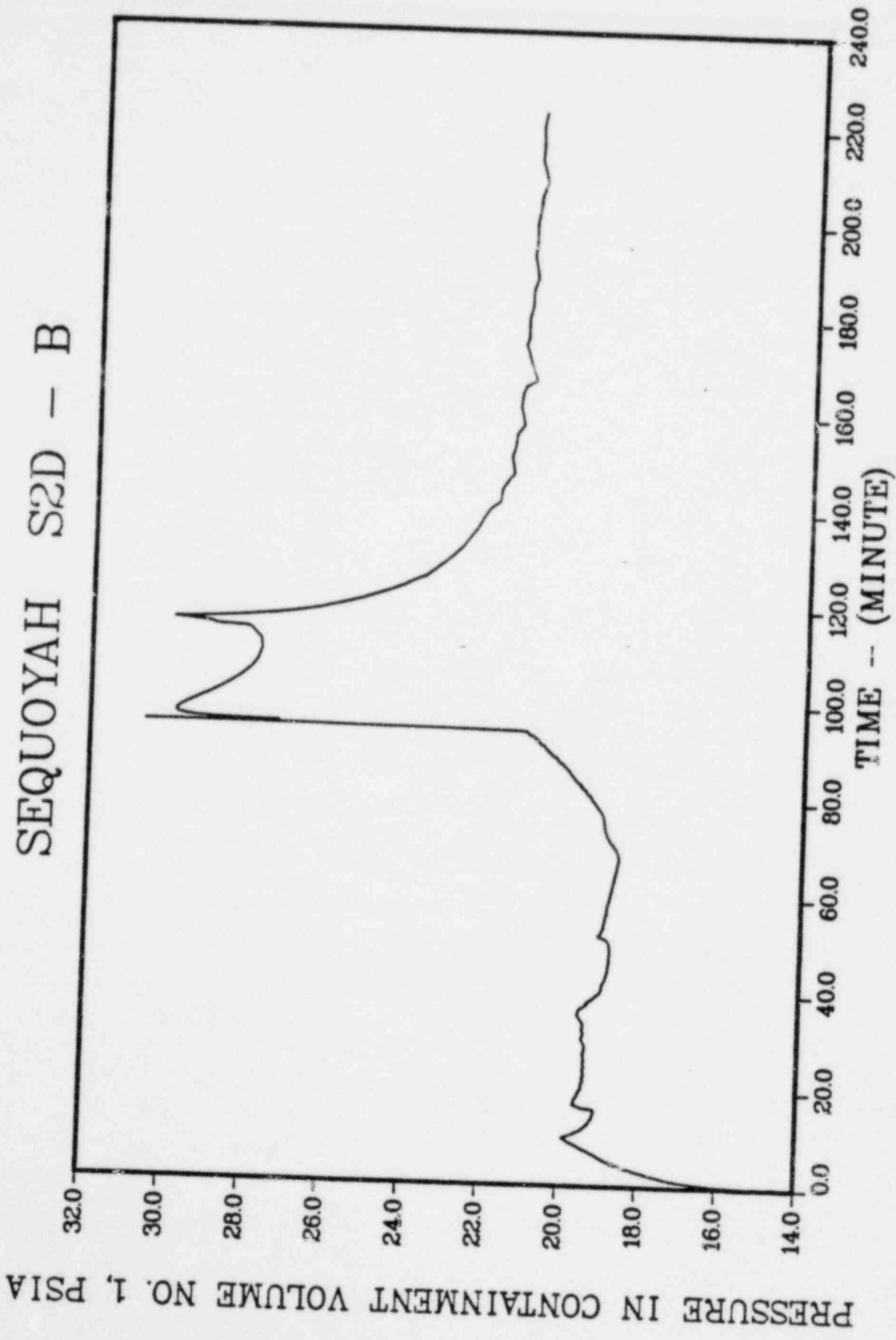
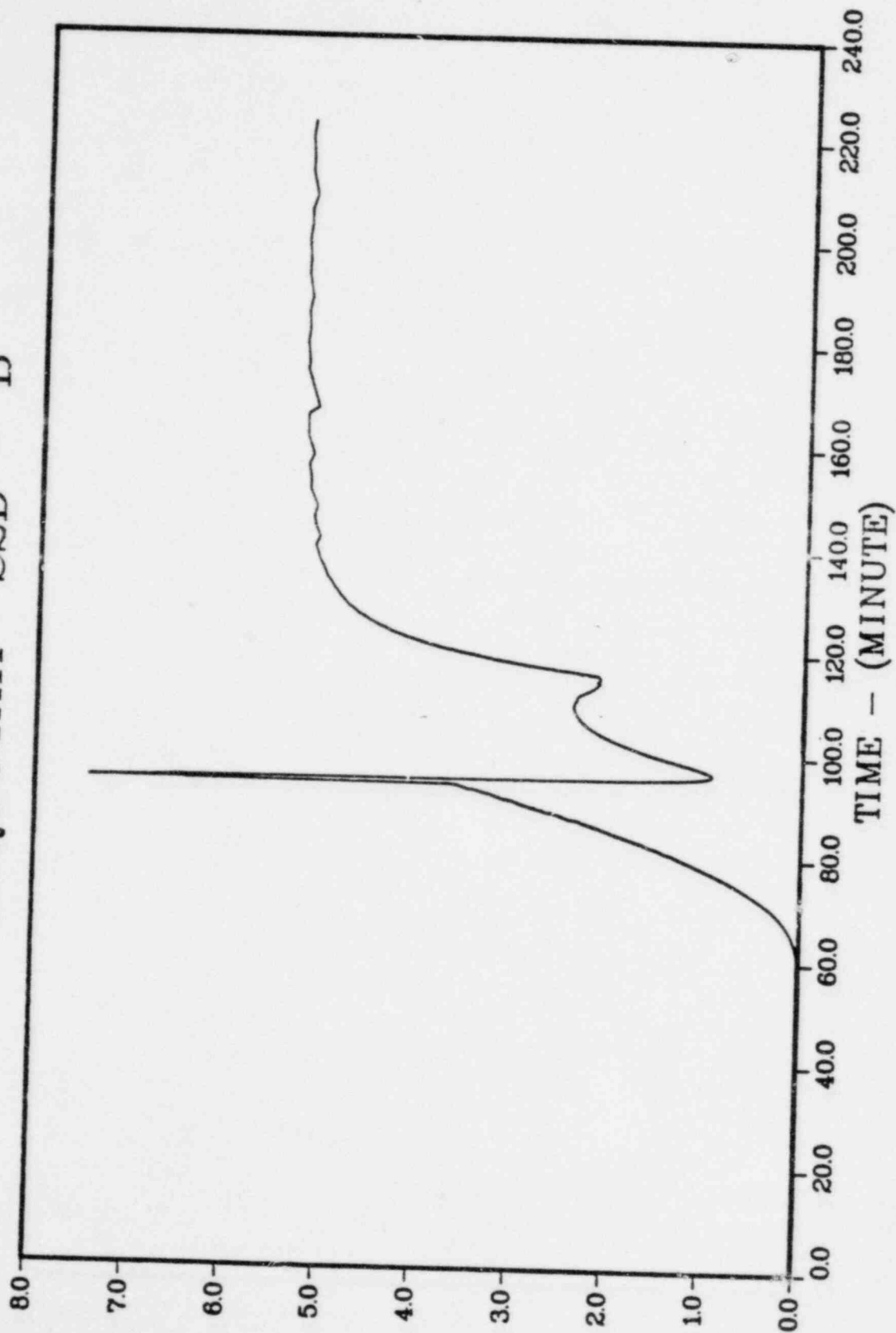


Figure 3.8.1-1 Reactor Building Elevation

SEQUOYAH S2D - B

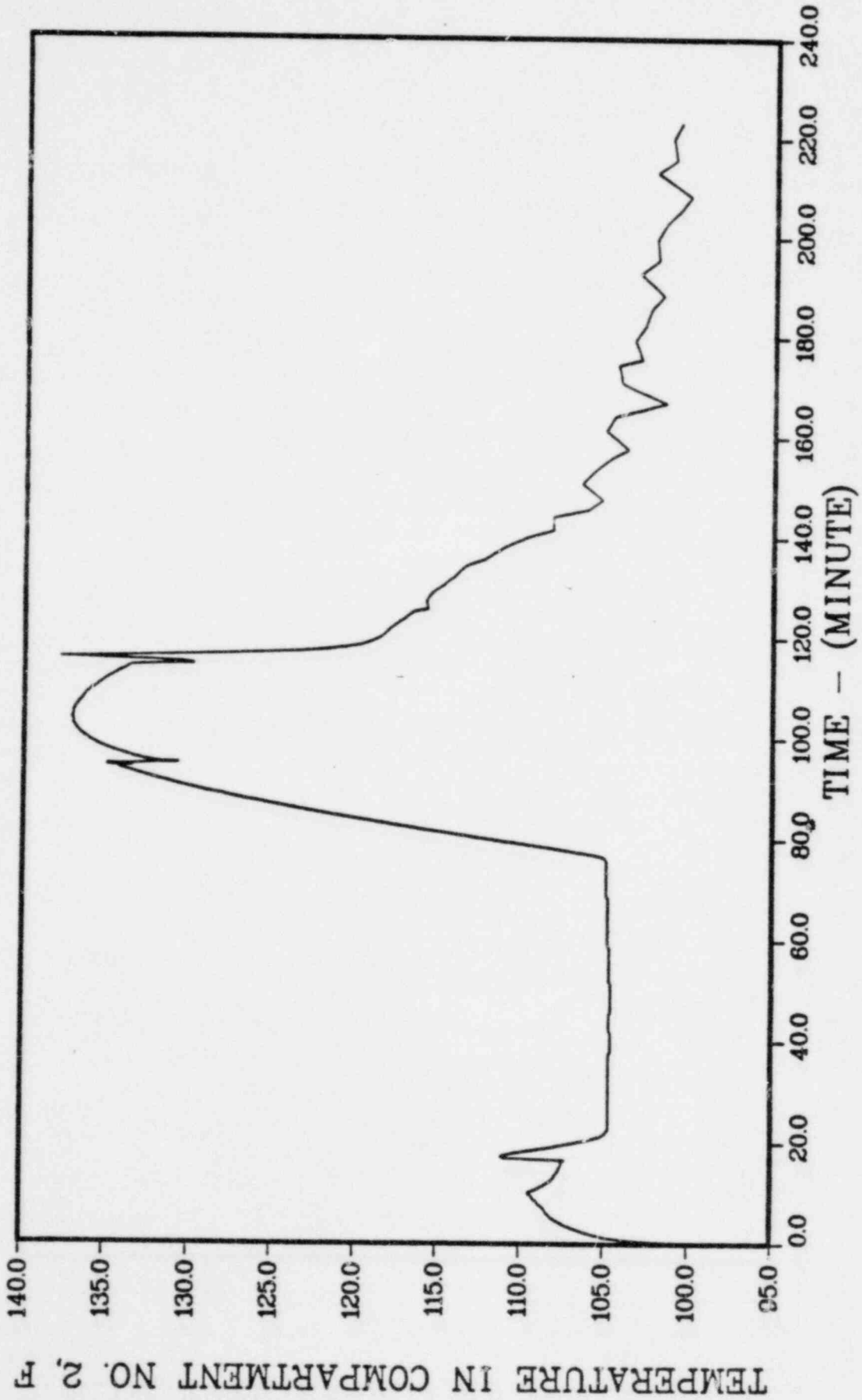


SEQUOYAH S2D - B

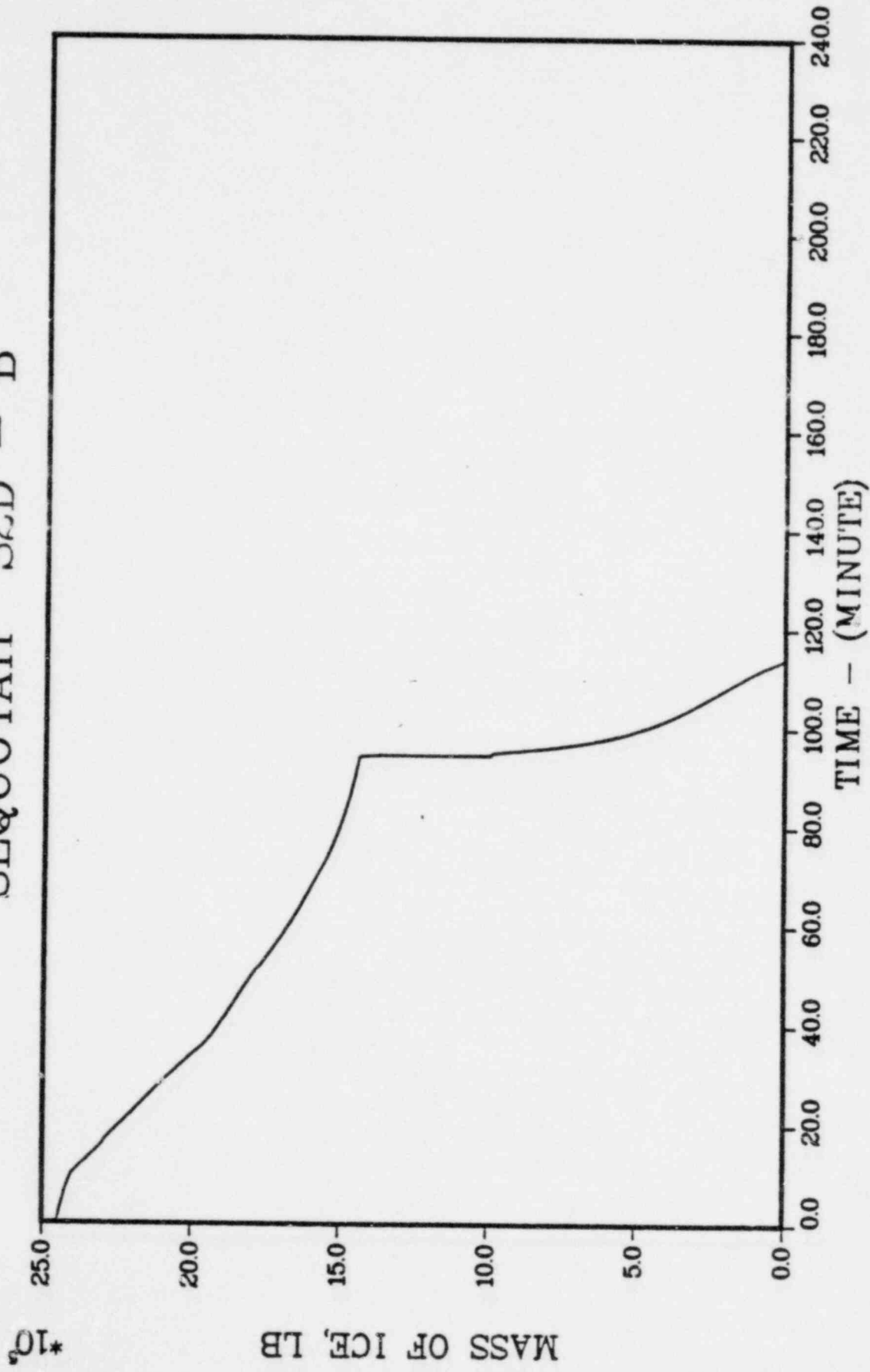


PARTIAL PRESSURE OF H2 IN COMPARTMENT NO.1, PSIA

SEQUOYAH S2D - B



SEQUOYAH S2D - B

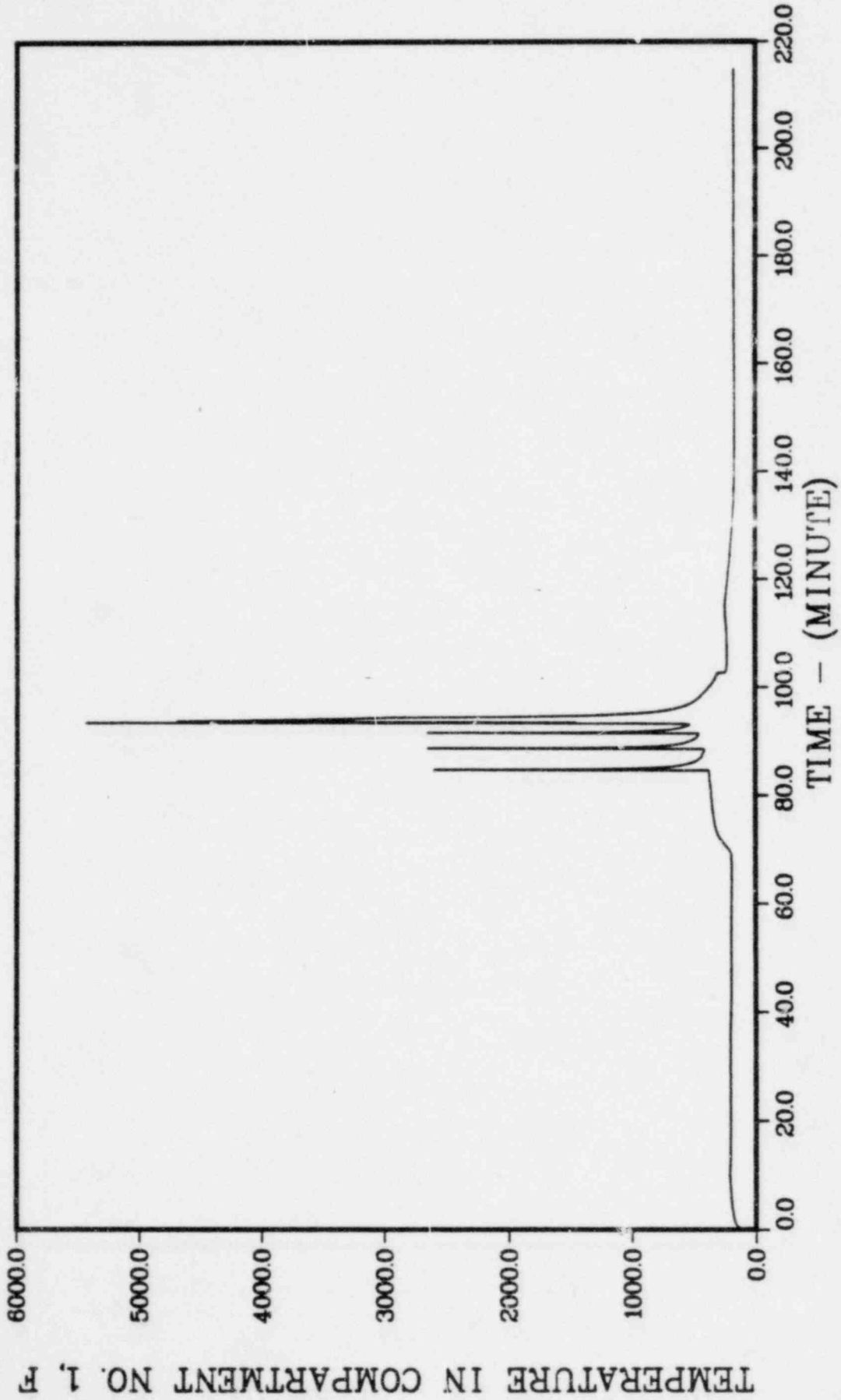


SUMMARY OF RESULTS

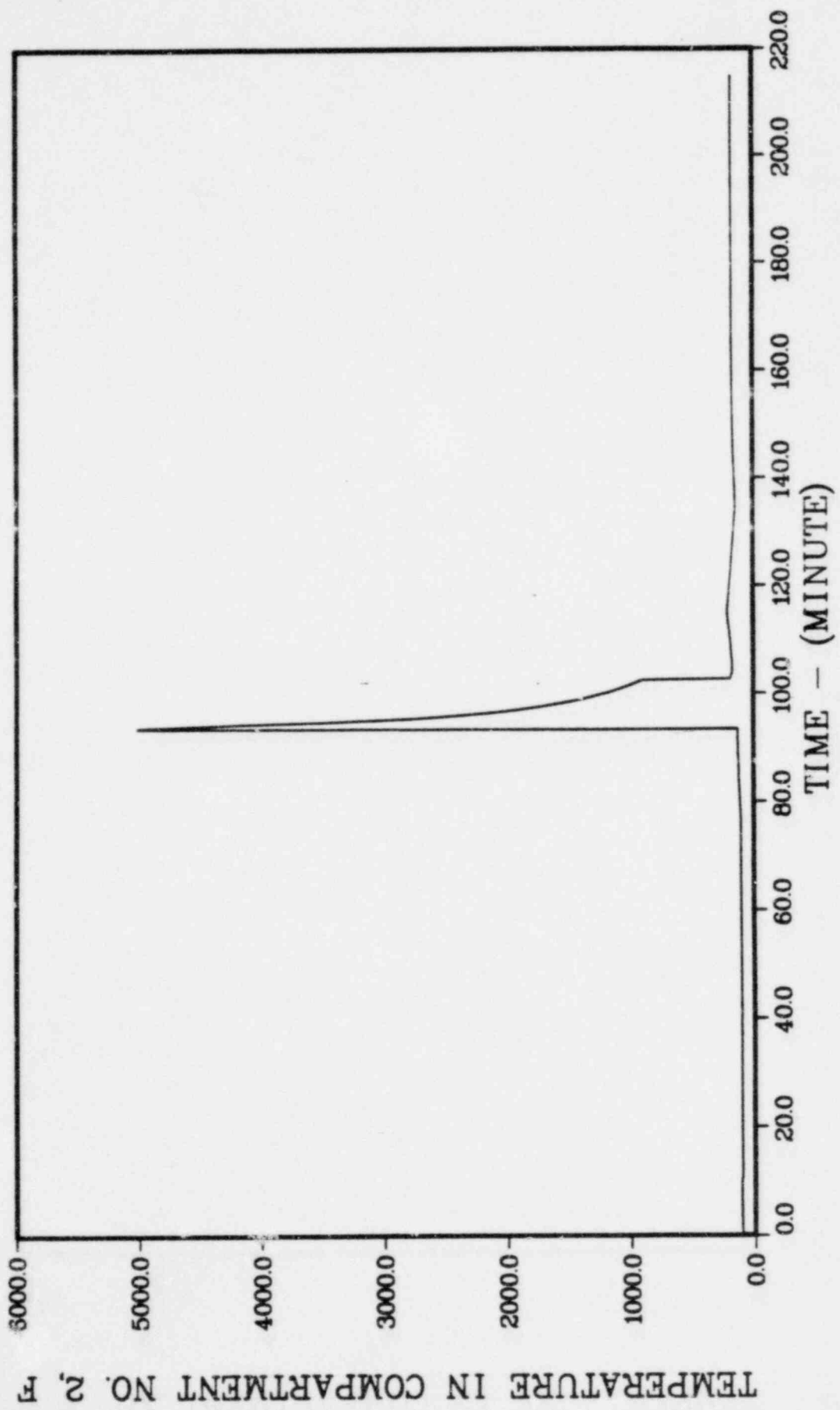
Case	Ignition Point v/o H ₂	Burn Limit v/o H ₂	Burn Time, sec	Flame Propagation 1 † 2	PT psia ⁽¹⁾	PNEW psia ⁽²⁾	PMAX psia ⁽³⁾
1	10	0	1	no	23	58	141
2	10	0	5	no	23	58	136
3	10	0	25	no	22	58	131
1X	10	0	1	yes	44	58	126
1X ⁽⁵⁾	10	0	1	yes	53	66	150
4	10	4	1	no	22	44	122
5	10	4	25	no	22	44	114
6	12	0	1	no	24	64	141
7	12	0	25	no	23	64	137
6X ⁽⁵⁾	12	0	1	yes	60	71	181
8	8	0	1	no	22	51	132
9	8	0	25	no	22	51	127
10	8	4	1	no	22	36	120
11	8	4	25	no	21	36	110
10X	8	4	1	yes	27	36	112
1B	4	0	1	yes	24	41	111
17 ⁽⁴⁾	10	0	1	no	31	79	146
18 ^(4,5)	10	0	1	no	35	68	223
19 ^(4,5)	10	0	1	yes	50	66	149

- (1) Peak pressure prior to core slump
- (2) Peak adiabatic H₂ burn pressure prior to core slump
- (3) Peak containment pressure after head failure
- (4) Ice melt complete at 21 minutes
- (5) Modified treatment of suspended water droplets

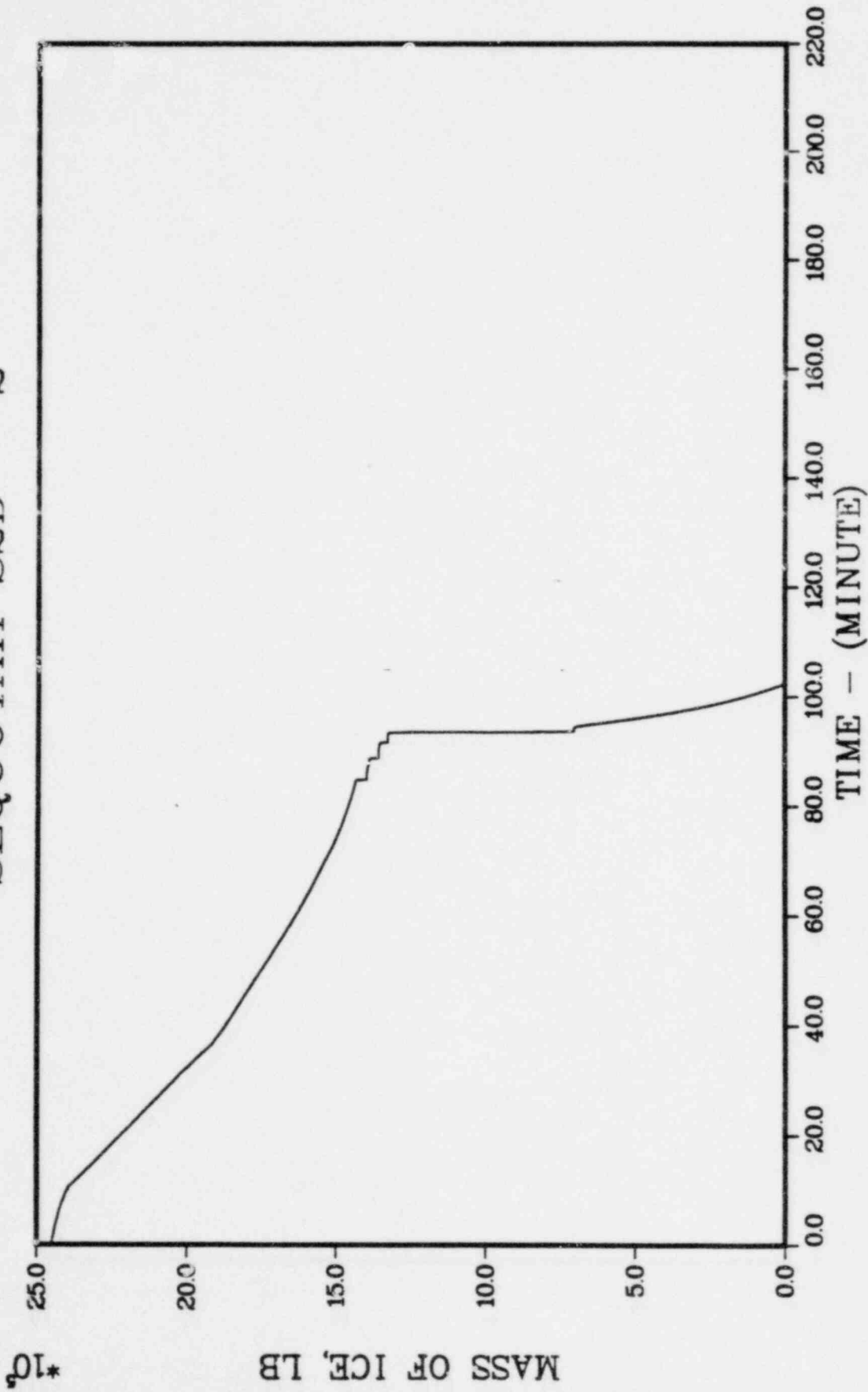
SEQUOYAH S2D - 2



SEQUOYAH S2D - 2



SEQUOYAH S2D - 2



TVA ANALYSES

. ANALYTICAL EFFORT

- WESTINGHOUSE/OFFSHORE POWER SYSTEMS

- ABOUT/YEAR STUDY OF CRITICAL PARAMETERS FOR VARIOUS ACCIDENT
SCENARIOS TO DETERMINE CONTAINMENT RESPONSE

- USING CLASIX CODE (UNDER DEVELOPMENT)

CLASIX CAPABILITIES

1. VENT FROM UPPER COMPARTMENT
2. ICE CONDENSER
3. RECIRCULATION FAN
4. DOORS - LOWER INLET AND INTERMEDIATE
5. INDIVIDUAL REPRESENTATION OF O_2 , H_2 , N_2 AND H_2O
6. SATURATED AND SUPER-HEATED STEAM
7. SPRAYS
8. H_2 , N_2 AND HEAT ADDITIONS
9. BREAK FLOW
10. BURN CONTROL

• PRELIMINARY ANALYTICAL RESULTS

- SELECTED SMALL BREAK LOCA RESULTING IN DEGRADED CORE COOLING (S₂D SEQUENCE OF WASH-1400)

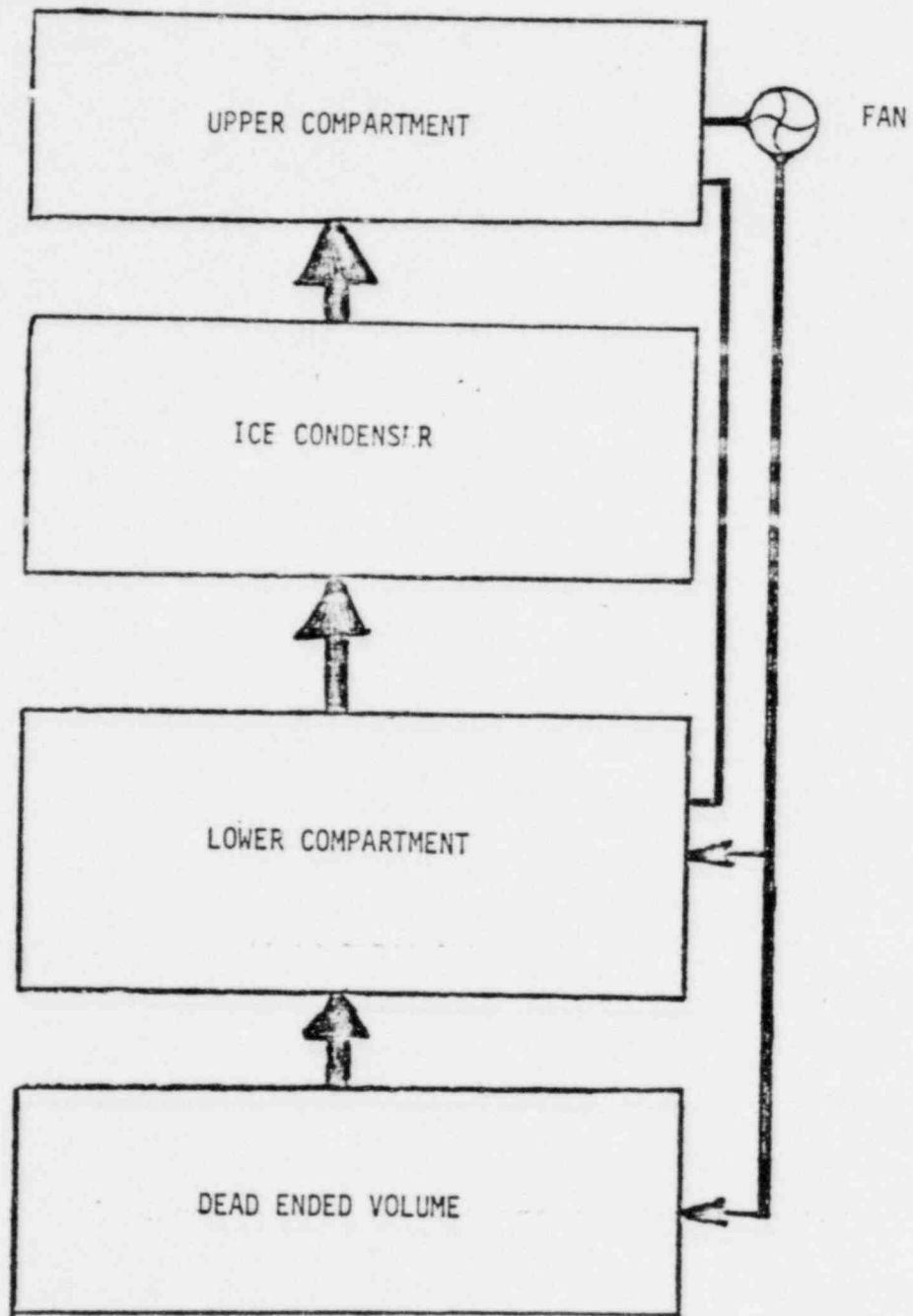
- RATE OF HYDROGEN RELEASE BASED ON MARCH CODE CALCULATION (ONSET OF HYDROGEN RELEASE 3500 SEC AFTER ACCIDENT INITIATION AND ASSUMED TO CONTINUE UNIMPEDED FOR 3000 SEC, RESULTING IN REACTION OF ABOUT 70% OF TOTAL ZIRCONIUM IN CORE)

- HYDROGEN COMBUSTION ASSUMED WHEN 10 VOLUME PERCENT HYDROGEN REACHED

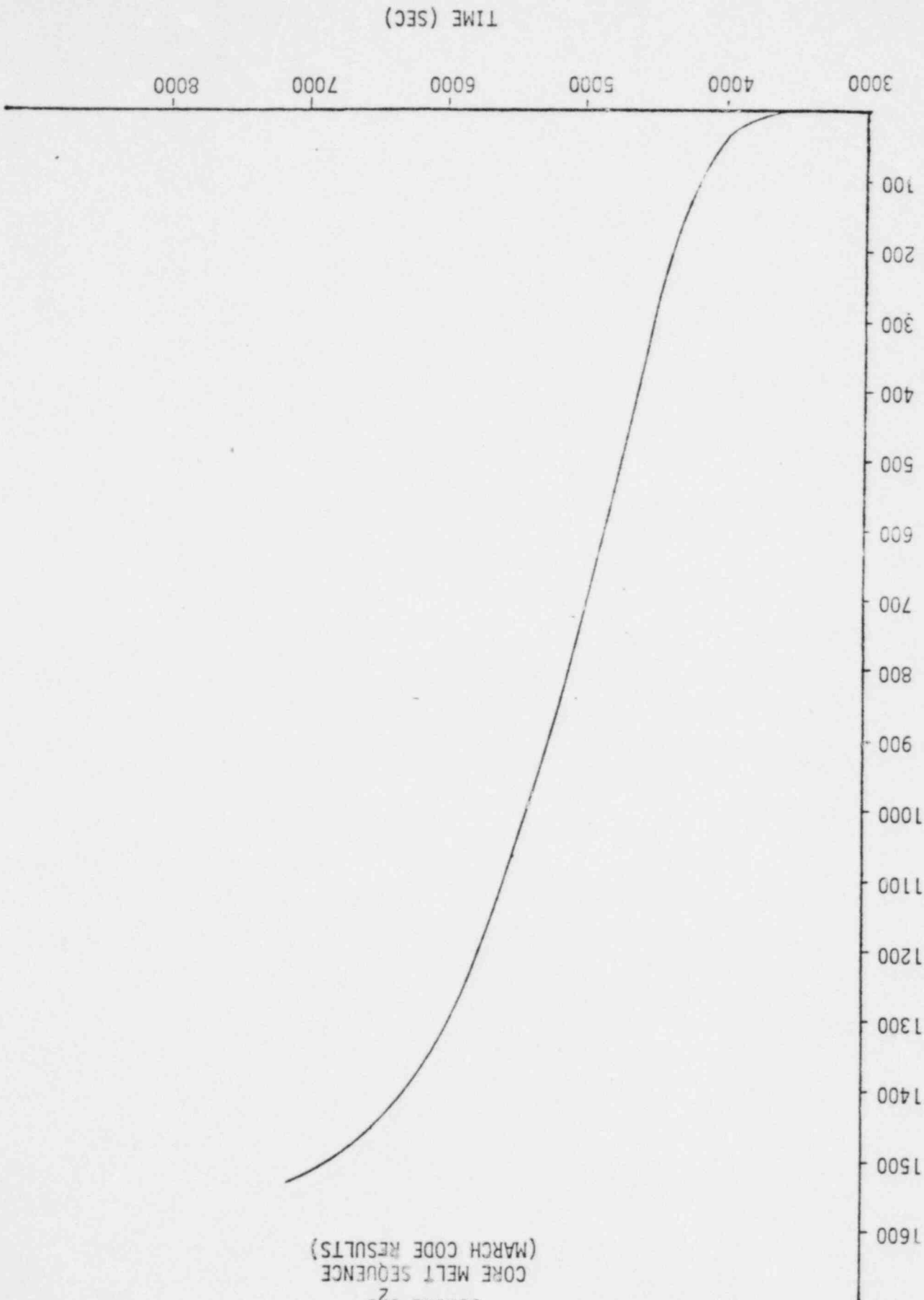
- VARIED ASSUMPTIONS REGARDING AIR RETURN FAN AND UPPER COMPARTMENT SPRAY PERFORMANCE, AND ICE AVAILABILITY.

BASE CASE PARAMETERS

- | | | |
|------------------------------------|--------------------------------|-----------------------------|
| 1. INITIAL CONDITIONS: | VOLUMES | |
| | TEMPERATURES | |
| | PRESSURES | LOTIC |
| | ICE MASS | CODE |
| | ICE HEAT TRANSFER AREA | |
| 2. BURN PARAMETERS: | H ₂ FOR IGNITION | 10 V/O |
| | H ₂ FOR PROPAGATION | 10 V/O |
| | O ₂ FOR IGNITION | 5 V/O |
| 3. AIR RETURN FANS: | NUMBER OF FANS | 2 |
| | CAPACITY OF EACH FAN | 40000 CFM |
| 4. SPRAY SYSTEM: | FLOW RATE | 6000 GPM |
| | TEMPERATURE | 125 F |
| | HEAT TRANSFER COEFFICIENT | 20 BTU/HR FT ² F |
| 5. ICE CONDENSER DRAIN TEMPERATURE | | 32 F |
| 6. BREAK RELEASE DATA | | MARCH CODE |



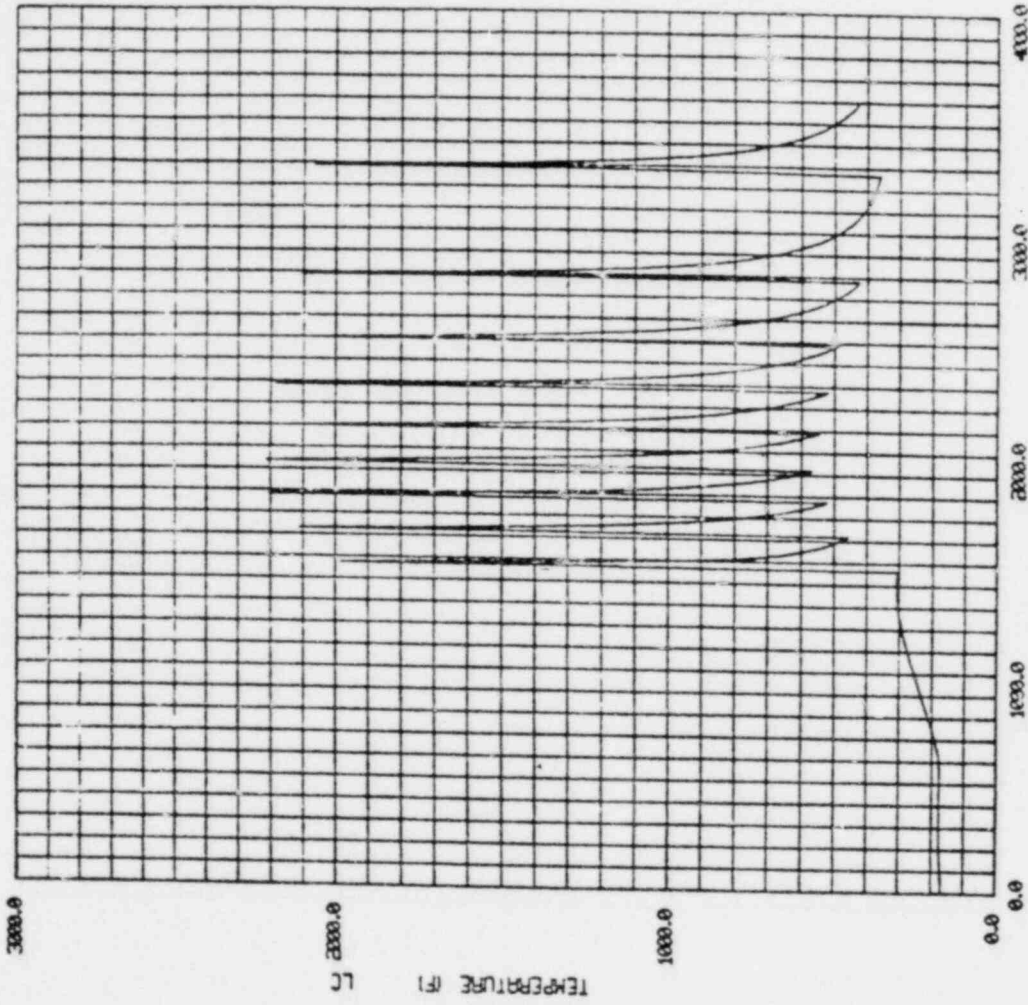
CLASIX MODEL OF ICE CONDENSER CONTAINMENT



HYDROGEN PRODUCTION
 DURING S₂D
 CORE MELT SEQUENCE
 (MARCH CODE RESULTS)

READY-

FRAME 01 1

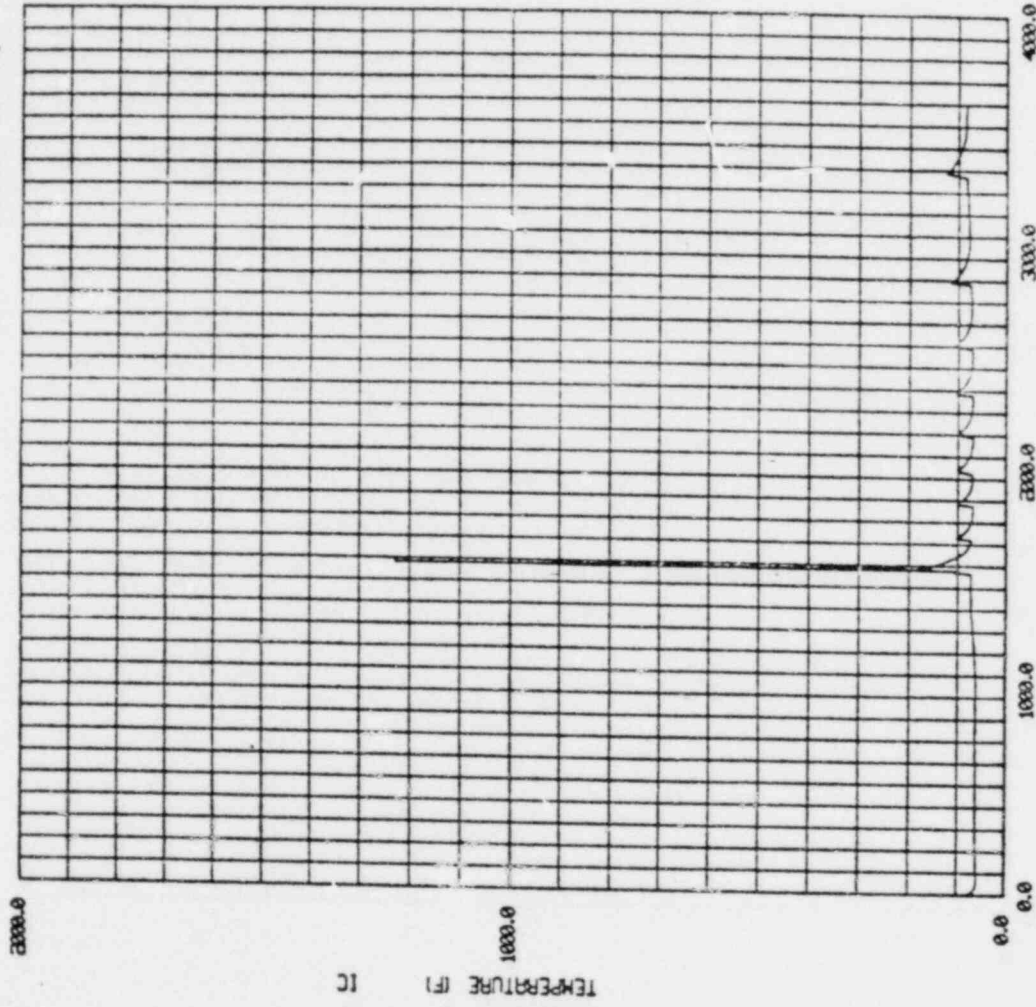


TUA S2D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 6FPS T+3480 BASE1

LOWER COMPARTMENT TEMPERATURE

READY-

FRAME #2

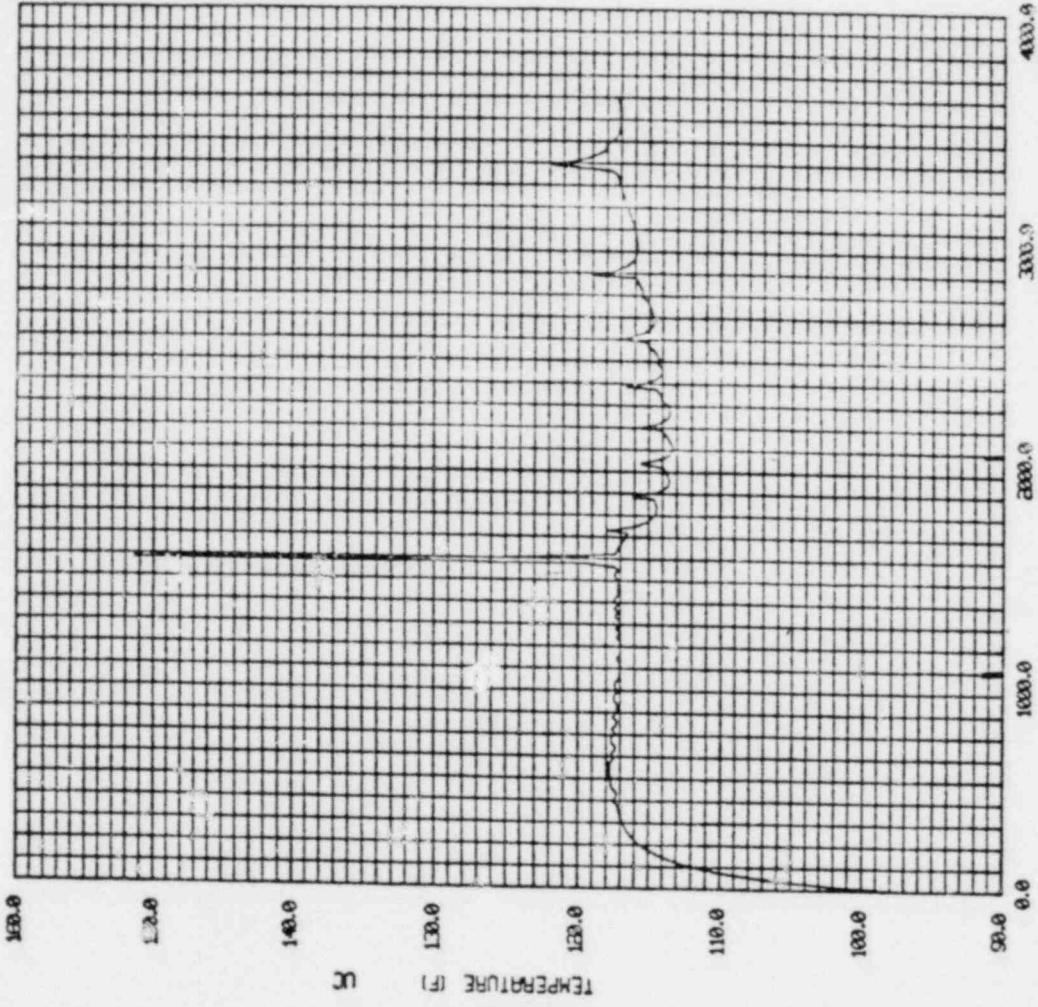


TIME (SECONDS)
TWA S2D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U O G/PS T+3480 BASE1

ICE CONDENSER TEMPERATURE

READY-

FRAME 03 1

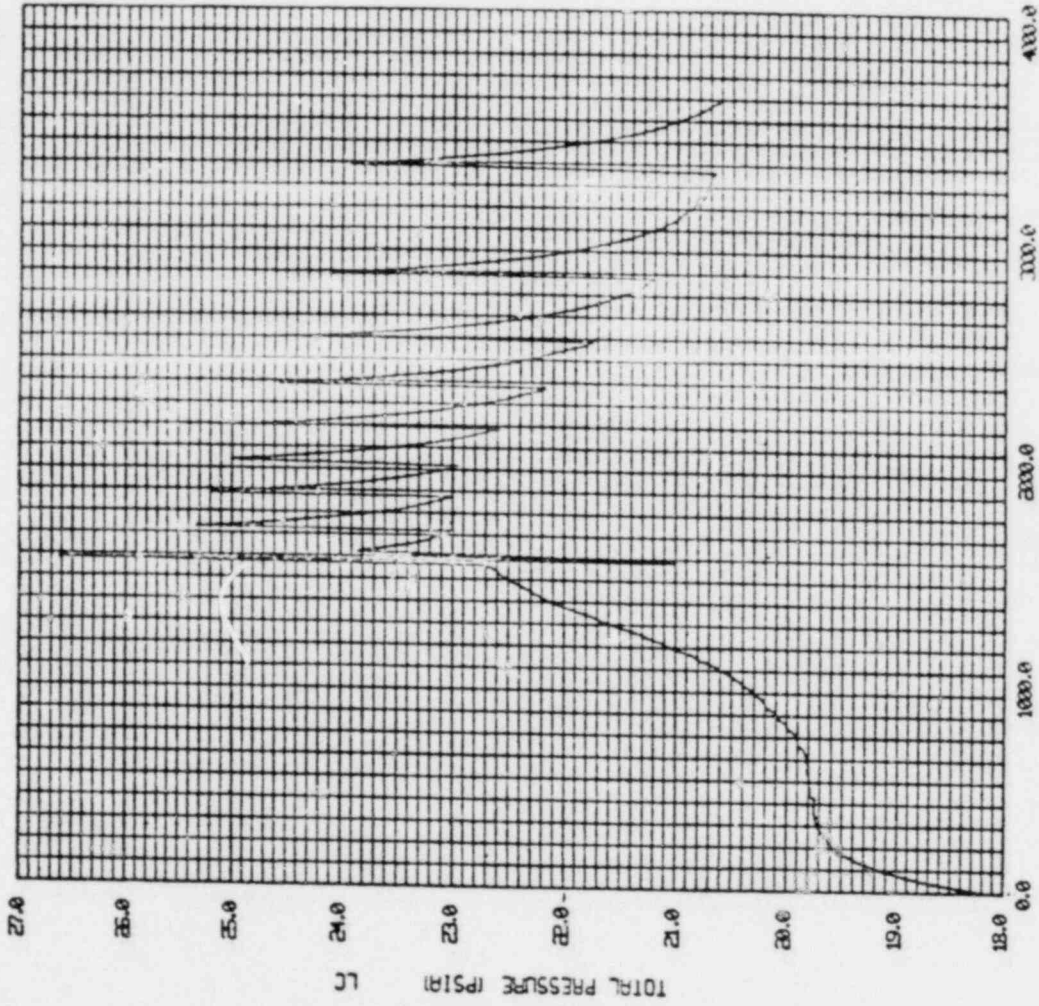


TWA S2D CASE1 2 FAN 1 SPRAY BURST 100 PCT AT 10 U @ 6FPS 1+3430 BASE1

UPPER COMPARTMENT TEMPERATURE

READY-

FRAME 05 F.5

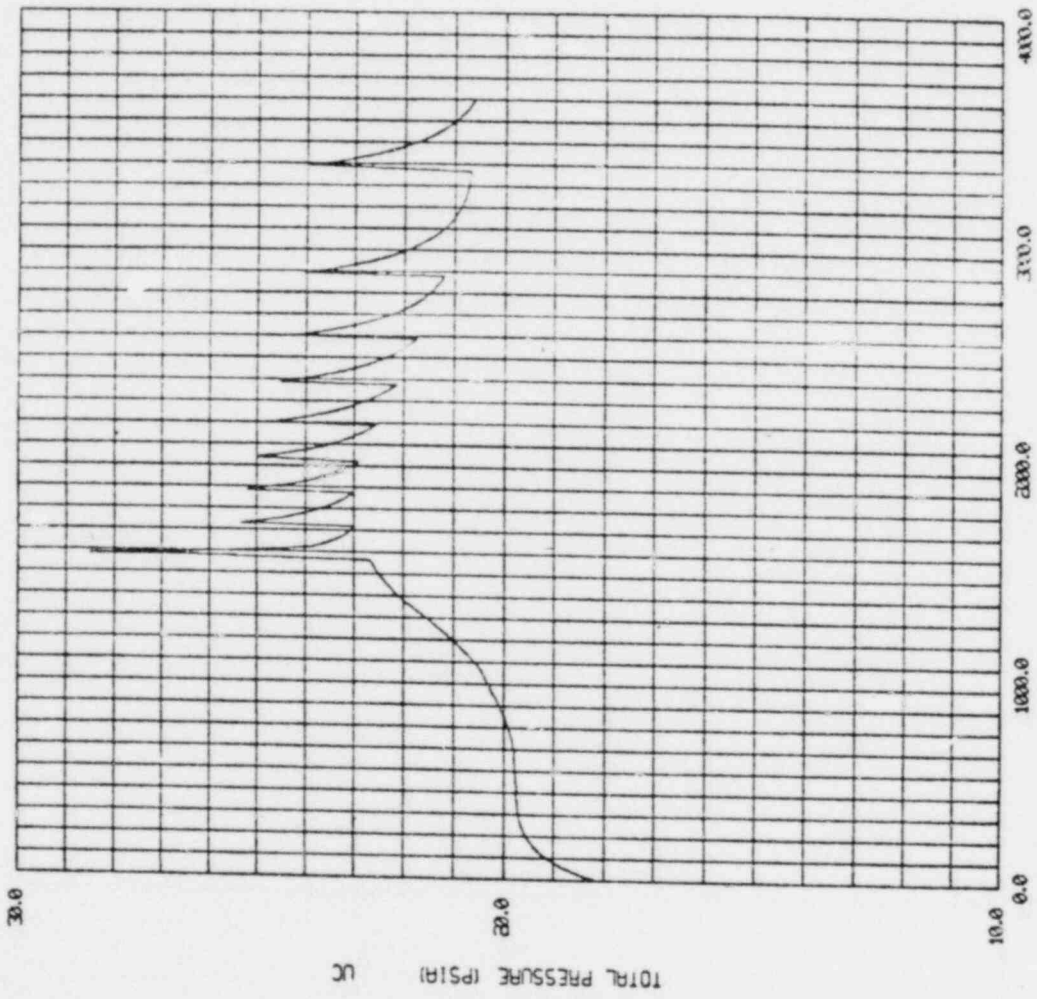


TUA S2D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 JFFS T-3480 BASE1

LOWER COMPARTMENT PRESSURE

READY-

FRAME 07 1

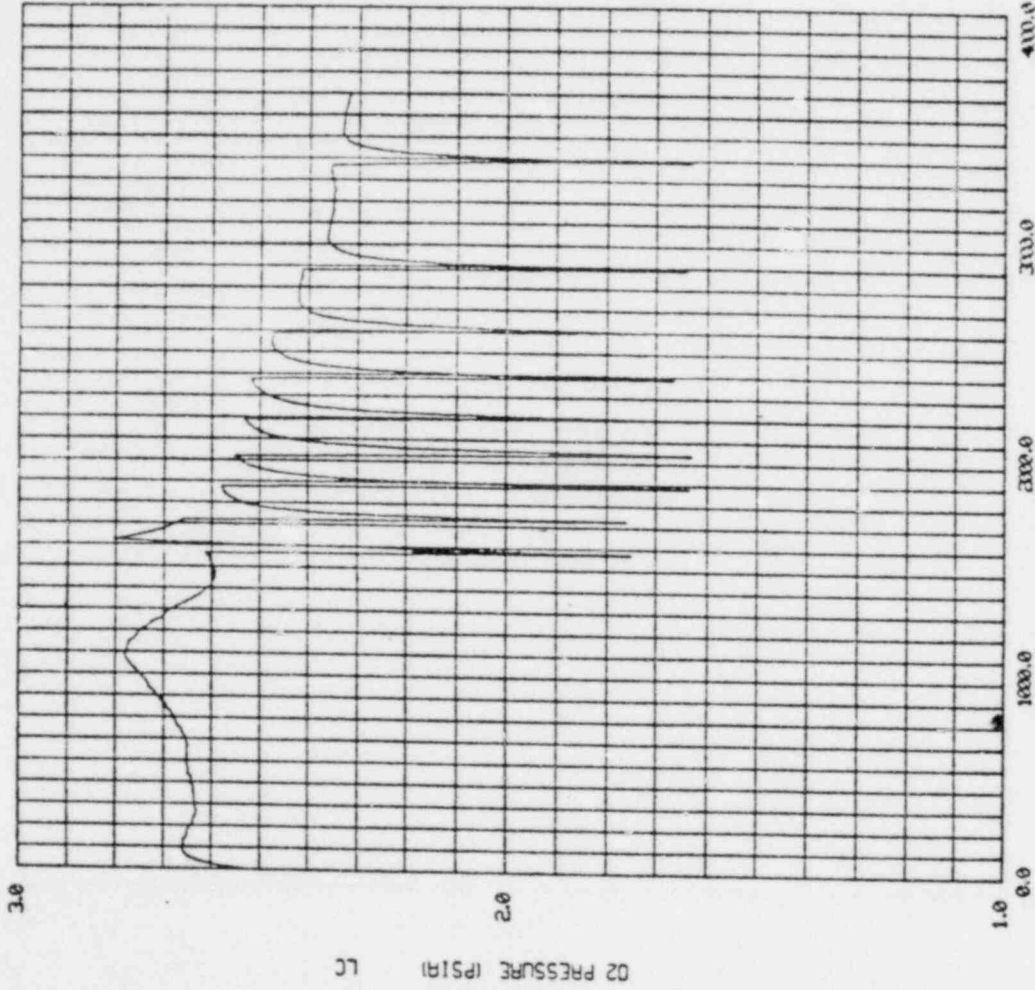


TUA 52D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 6FPS T+3480 BASE1

UPPER COMPARTMENT PRESSURE

READY-

FRAME 09 1

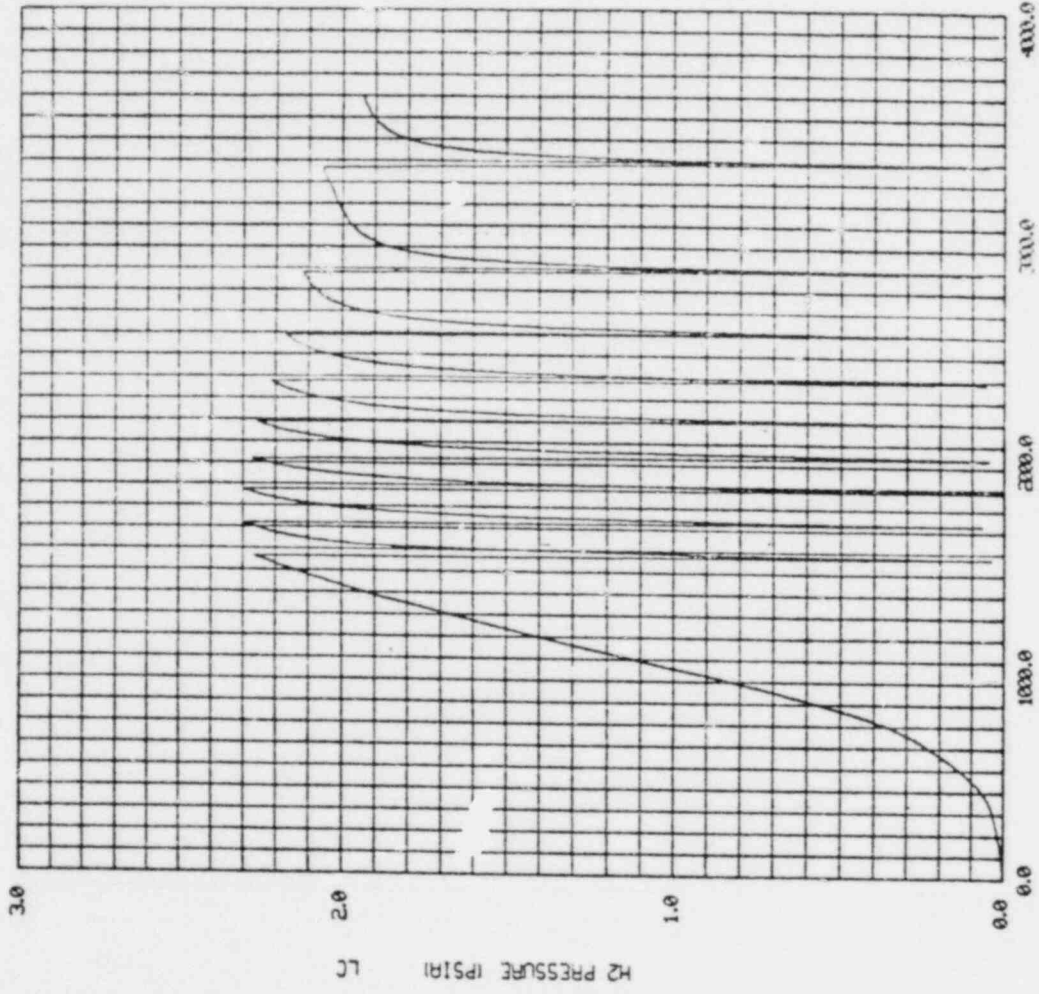


TVA S2D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 0 6FPS T+3480 BASE1

LOWER COMPARTMENT OXYGEN

READY-

FRAME 17 F.17

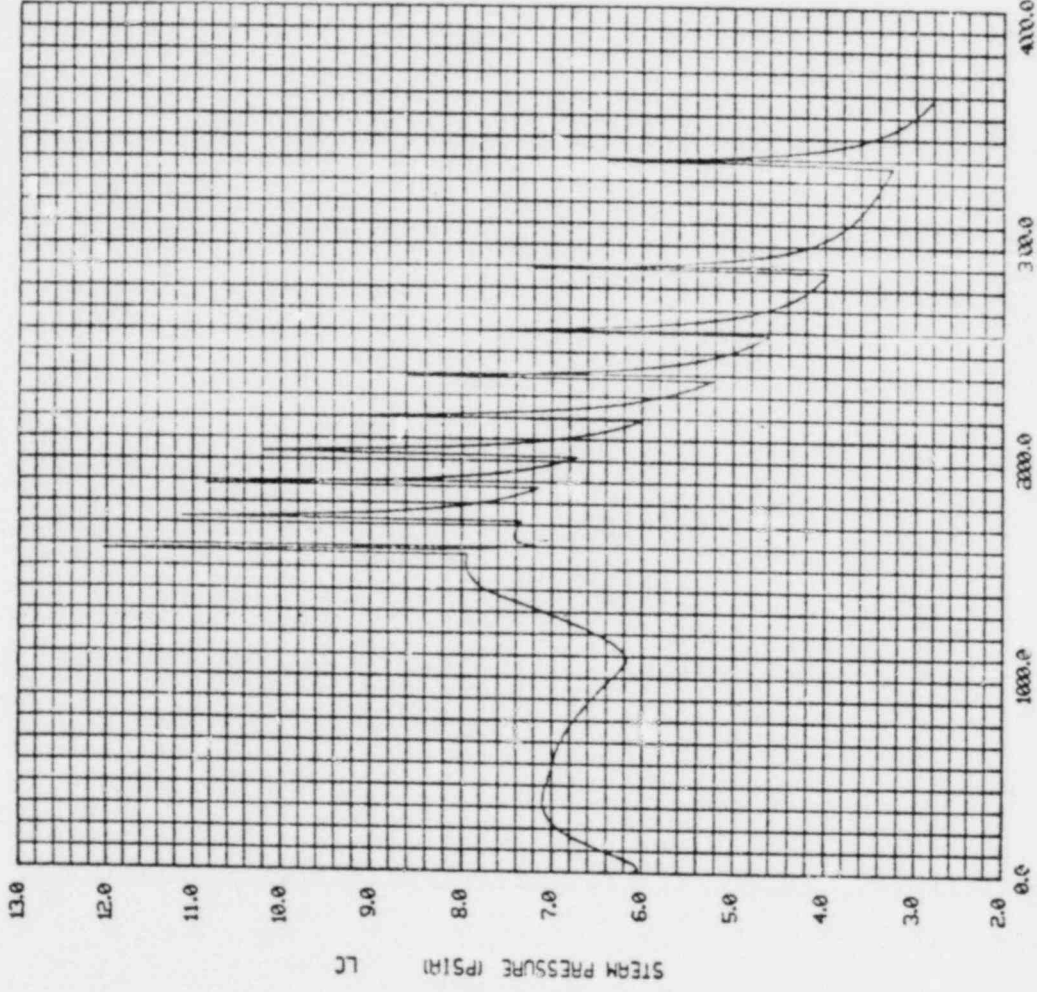


TIME (SECONDS)
TWA 52D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 EFPS T+3480 ERSE1

LOWER COMPARTMENT HYDROGEN

READY-

FRAME 21 F, 21

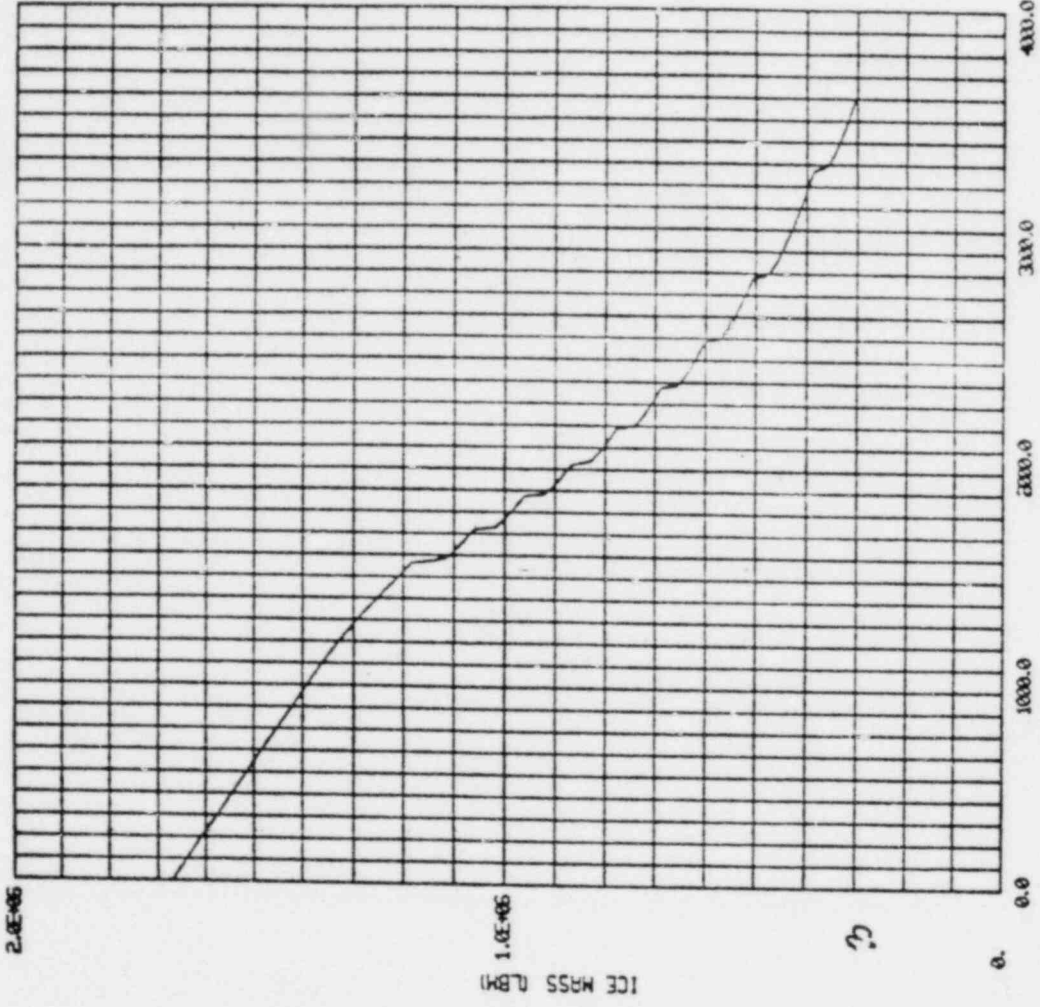


TWA 52D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 6FPS T+3480 EASE1

LOWER COMPARTMENT STEAM

READY-

FRAME 41 F, 41



TUA 52D CASE1 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 6FPS T+J480 BASE1

ICE MASS

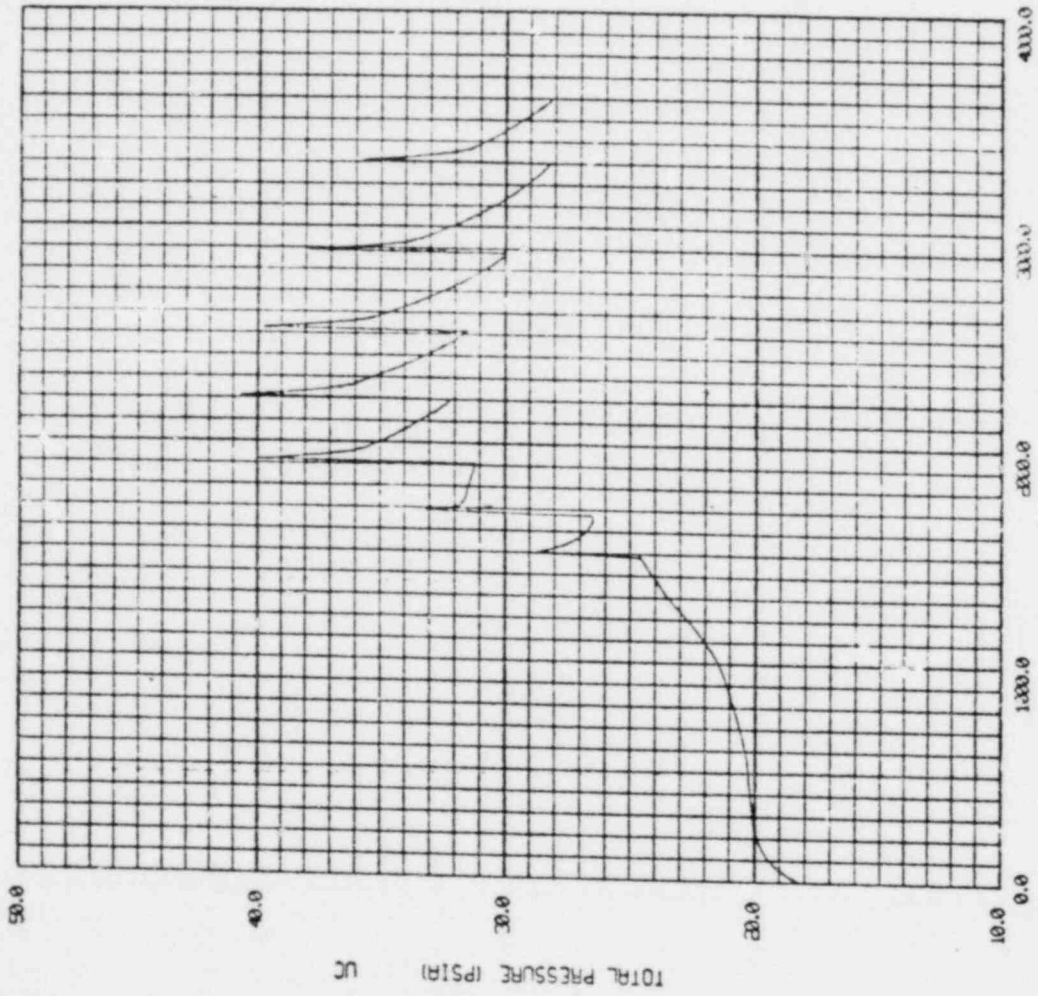
TABLE 1. PRELIMINARY CONTAINMENT ANALYSIS SENSITIVITY STUDIES

	TOTAL H ₂ BURNED (LB)	PEAK TEMP. (°F)			PEAK PRESS (PSIA)	
		LOWER COMPARTMENT	ICE BED	UPPER COMP.	LOWER COMP.	UPPER COMP.
1. BASE CASE	900	2200	1200	150	26.5	28.5
2. H ₂ IGNITION AND PROPAGA- TION @ 8%	1050	1200	700	260	28.5	30.5
3. 1 AIR FAN	900	2200	1350	160	26.5	29.5
4. NO ICE*	850	2400	2000	270	41	41
5. NO AIR FANS	1200	2370	2580	1090	46.4	92.4

* ICE EXISTS ONLY FOR THE FIRST TWO OF 7 BURNING CYCLES.

READY-

FRAME 07 1

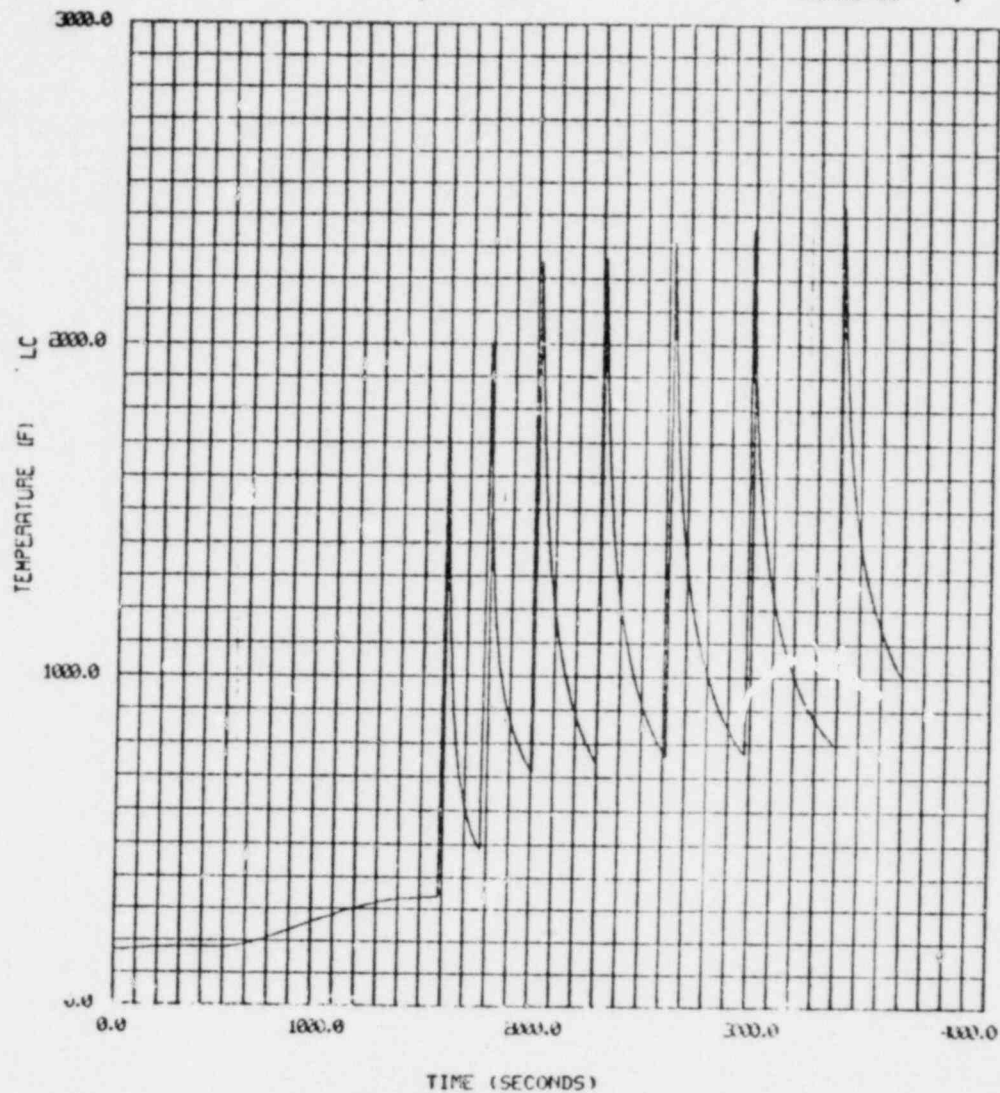


TUN S2D CASES 2 FAN 1 SPRAY BURN 100 PCT AT 10 U 0 12 FPS T+3480 -ICE

UPPER COMPARTMENT PRESSURE

READY-

FRAME 01



TWA S20 CASES 2 FAN 1 SPRAY BUR 1 100 PCT AT 10 U 8 GFPS T+3480 -ICE

LOWER COMPARTMENT TEMPERATURE

TOPICS FOR FURTHER REVIEW

1. LOCATION AND NUMBER OF IGNITERS
2. H₂ MONITORING AND MIXING
3. ABILITY OF IGNITERS TO FUNCTION IN TURBULENT, FLOWING ATMOSPHERE
4. PROCEDURES AND STRATEGY FOR IGNITER OPERATION
5. ANALYSIS MODELS
CLASIX
 - A. SOLUTION SCHEME
 - B. CONTAINMENT MODELING
 - C. INPUT PARAMETERS
6. ANALYSIS RESULTS
 - A. MIXING OF CONSTITUENTS
 - B. SENSITIVITY STUDIES
 - I. BURN CRITERIA
 - II. DIFFERENT H₂ GENERATION RATES
7. EFFECTS OF LOCAL DETONATIONS OR RAPID BURNS USING CLASIX. INPUT AS A BOUNDARY CONDITION TO CLASIX, START FROM DETONATION.
8. PROTECTIVE MEASURES FOR ESSENTIAL EQUIPMENT

DELIBERATE IGNITION SYSTEM (DIS)

PRELIMINARY CONCLUSION: SIGNIFICANT POTENTIAL FOR
MITIGATING THE CONSEQUENCES OF DEGRADED CORE ACCIDENT
WHEN OPERATING IN CONJUNCTION WITH EXISTING CONTAINMENT
HEAT REMOVAL MECHANISMS.

HYDROGEN PROBLEMS IN SEQUOYAH CONTAINMENT

INTRODUCTION

This letter report completes the RDA response to a request from the Nuclear Regulatory Commission to critique the ultimate strength analysis of the Sequoyah containment. This second report deals with the last two tasks of the work statement.

1. How would the analyses and results be altered if the stresses are caused by ignition/detonation of 300-600 Kg of hydrogen distributed uniformly and nonuniformly in the containment?
2. To what extent can distributed ignition sources mitigate the effects of hydrogen?

A preliminary discussion of these topics was attended by Commissioner Gilinsky and Dr. John Austin at RDA on 18 July 1980.

RESULTS

1. a) 300 kg of H₂ gas mixed uniformly with the air and steam (if less than 40 percent steam) in the Sequoyah containment volume following an accident would be completely combustible if ignited (see Figure 1). This complete combustion could occur so rapidly as to exceed the capacity of the available heat removal processes, and could produce a pressure as high as 5.5 atmospheres, thus rupturing the containment (see Table 2). The combustion of 600 kg of H₂ would of course have more severe consequences.

- b) A nonuniform distribution of 300 kg of H₂ present in the containment would consist of parcels of gas richer in H₂ than the uniform distribution. If these separated parcels formed while the blowers were operating, they would probably be mixed, combustible and perhaps detonable. If they were all detonable and all ignited, the damage to the containment would be worse than that due to ignition of a uniform mixture. If the gas parcels were not detonable, the pressure upon combustion would probably be at least as high as the uniform distribution. Under some circumstances, it would be possible to collect pockets of gas too rich in H₂ to burn. As the outer edges of such pockets mix with air, partially combustible mixtures would form. The results of igniting such a distribution would clearly depend on the sizes of the parcels and the timing.

It should be noted that harmless mixtures of H₂, air and steam may become highly combustible or detonable as steam is condensed out (see Appendix B). Thus one mechanism employed for removing heat from the containment also removes the combustion inhibitor from the containment.

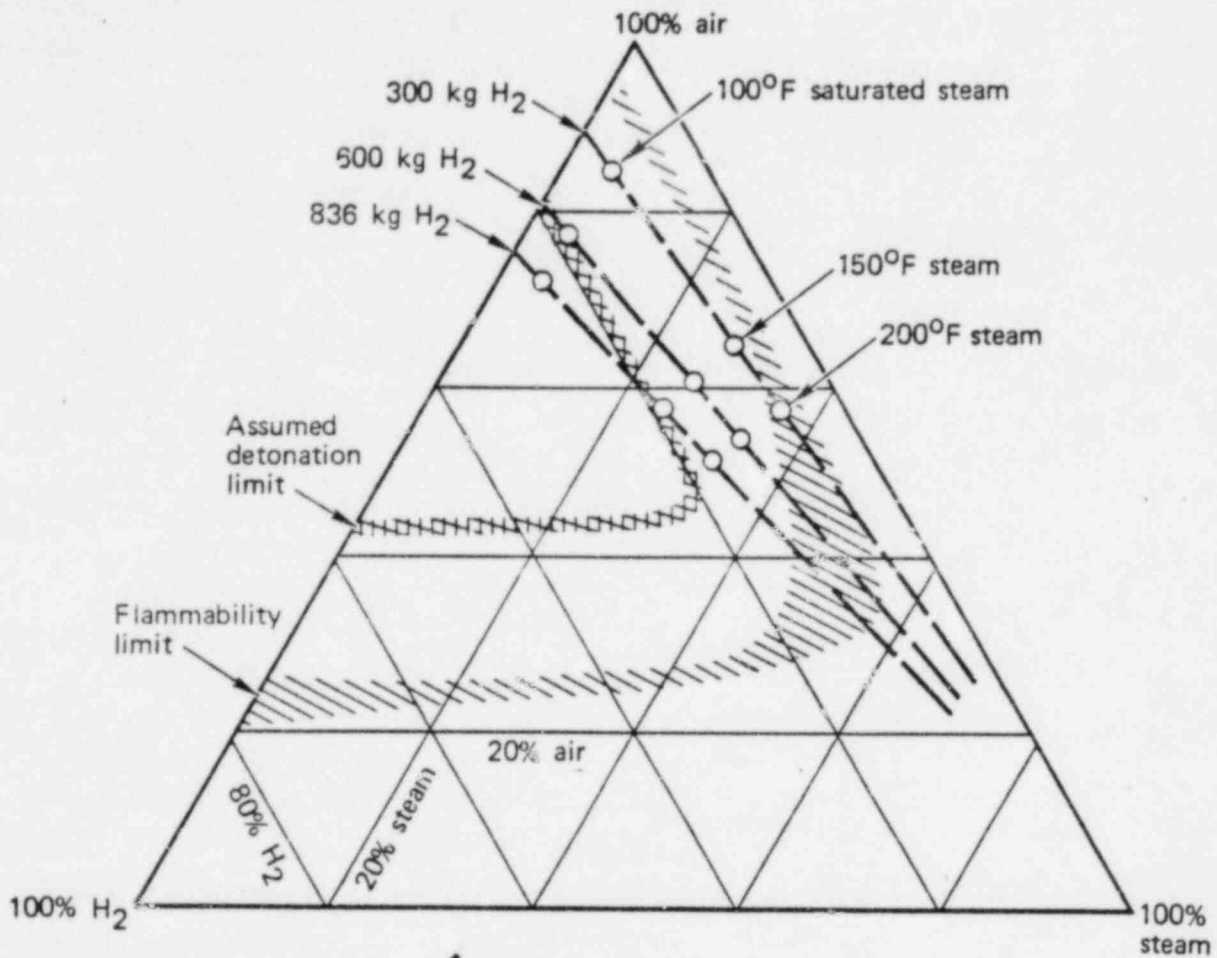
2. If the rate of hydrogen formation is sufficiently low, and the mixing of H_2 is complete and rapid so that all the gas in the containment gradually increases in H_2 concentration, then the presence of enough igniters could prevent overpressurizing the containment. This would be accomplished by releasing the heat of combustion at low concentrations over a long enough period of time to be handled by the heat removal equipment. However, if the Zr reaction rate is high relative to heat removal processes, then igniters might only delay containment failure. Table 3 shows that a 1 percent per minute Zr reaction rate, accompanied by the burning of hydrogen at its rate of formation, would match the steady-state heat removal capacity of the RHR equipment.

If the H_2 is not thoroughly mixed, then there is a possibility of igniting a detonable pocket of gas with an igniter. If left to its natural end, such an H_2 -rich pocket could disperse below the detonation limit (~ 20 percent H_2) when its ignition would cause less of a problem.

Since the possible rates of generation of H_2 following an accident and the rate, place, and degree of mixing with air are highly uncertain, the use of igniters can only be an uncertain means of pressure control. Improper use might be detrimental rather than helpful. On the other hand, if it is assumed that there are many unavoidable ignition sources in the containment, it is certainly true that control of the time and place of ignition is preferable to chance. In this sense the use of igniters seems beneficial.

COMMENT

It is our opinion that the uncertainties in H_2 generation and mixing are so dependent on hardware details and scenarios that they are unlikely to be greatly reduced by further work. For this reason we believe it may be a better use of resources to explore thoroughly the feasibility of using an inert atmosphere in the containment, so as to avoid the hydrogen burning problem.



Limits of flammability and detonation based on Shapiro and Moffette WAPD-SC-545, as reproduced in WASH 1400.

Figure 1. Uniform mixtures in the Sequoyah containment vessel.

TABLE 1. INPUT DATA FOR SEQUOYAH PLANT

1. Free volume of containment vessel ^(a)	$3.2 \times 10^4 \text{ m}^3$
Weight of contained air at 27°C, 1 atm.	$3.7 \times 10^4 \text{ kg}$
Gram moles of air	1.3×10^6
Gram moles of oxygen	2.7×10^5
2. Weight of zirconium in core ^(b)	$1.9 \times 10^4 \text{ kg}$
Gram moles of zirconium	2.1×10^5
3. Yield of 100% Zirconium-water reaction	
Weight of hydrogen	836 kg
Gram moles of hydrogen	4.2×10^5
Heat of reaction ^(c) , $\text{Zr} + \text{H}_2\text{O}$	$1.1 \times 10^{11} \text{ joules}$
Heat of H_2 burn ^(d) (to form liquid H_2O)	$1.2 \times 10^{11} \text{ joules}$
Total heat of reaction + burn	$2.3 \times 10^{11} \text{ joules}$
4. Molar quantities and partial air pressure of saturated steam in containment	
At 100°F (38°C) vapor = 8.1×10^4 moles = 0.06 atm.	
150°F (66°C) = 5.9×10^5 moles = 0.25 atm.	
200°F (93°C) = 8.4×10^5 moles = 0.78 atm.	

NOTES:

- (a) Sequoyah Nuclear Plant, Preliminary Safety Analysis Report (PSAR), Table 5.2-1 gives the total containment active volume as 1,142,000 ft³, comprised of 730,000 in the upper compartment, 125,000 in the ice compartment, and 287,000 in the lower compartment.
- (b) Sequoyah PSAR, Tabel 1.3-1, gives the clad weight as 41,993 lb.
- (c) G. W. Keilholtz, ORNL-NSIC-120, Annotated Bibliography of Hydrogen Considerations in Light-Water Power Reactors, Feb. 1976, Table 1, Heat of Reaction = 122 to 137 kcal/mole Zr.
- (d) Lewis and Von Elbe,
p. 685, 68.3 kcal/mole H_2O .

TABLE 2

	H ₂ Quantity		
	300 kg	600 kg	836 kg
1. Percent Zr Reaction	36%	72%	100%
2. Moles H ₂	1.5x10 ⁵	3.10 ⁵	4.2x10 ⁵
3. Partial Pressure @ 300°k (atmospheres)	0.12	0.23	0.32
4. Molar Ratio $\frac{H_2}{Air}$, Uniform Distribution	0.11	0.23	0.32
5. Detonatable (D) or Combustible (C) ^a Mixture, no steam present	C	D	D
6. H ₂ Concentration Multiplier Required relative to uniform mixture ^a			
a) to reach detonation regime	2.0	1.0	1.0
b) to reach stoichiometric ratio of 0.42:1 for H ₂ :air	3.8	1.8	1.3
7. Steam Vapor Pressure Required: ^b			
a) to prevent detonation of uniform mixture	0	0.1 atm	0.4 atm
b) to prevent combustion of uniform mixture	0.9 atm	2.0 atm	2.3 atm
8. Energy Release in 100% Combustion, Joules (liquid water product)	4.3x10 ¹⁰	8.6x10 ¹⁰	1.2x10 ¹¹
9. Final Absolute Pressure in Adiabatic Combustion (Initial Air Partial Pressure 1 atm, Initial Temperature 300°k) ^c			
a) No steam, 100% combustion	5.5 atm	10.0	13.3 atm
b) No steam, 50% combustion	3.3	5.8	7.3
c) Steam @ 190°F, 50% combustion	4.1	6.5	8.3

NOTES:

- (a) Approximate, based on regimes outlined in Figure 1.
- (b) Approximate, based on regimes outlined in Figure 1, plus molar concentrations of saturated steam as a function of pressure.
- (c) Assuming products of combustion behave as ideal gases, and assuming a constant-volume reaction.

TABLE 3. HEATING AND COOLING RATES IN SEQUOYAH CONTAINMENT

Time when Fission Product Heat (Cumulative) Equals Total Heat of Reaction	3000 sec
Rate of Heating at the 1% per min Zr Reaction Rate	
Zr Reaction	18.0 MW
H ₂ Burning	20.0
Total	38.0 MW
Rate of Fission Product Heating at 2 hours (when ice has been melted in DBA)	27 MW
Steady-state ^a Cooling Capacity of the 2 RHR Heat Exchangers	67 MW
Net Margin of Cooling Capacity (Beyond Chemical Reactions @ 1%/min and Fission Product Heating)	2 MW

NOTES:

- (a) Sequoyah PSAR, Table 6.3-2 cites 2 heat exchangers, each having a capacity of 1.15×10^8 BTU/h at specified conditions.

APPENDIX A

LITERATURE SEARCH ON EXTENT OF HYDROGEN BURNING AND FLAMMABILITY LIMITS FOR MIXTURES OF H₂, AIR, AND STEAM

In considering the effects of 300 kg to 600 kg H₂ in the Sequoyah containment vessel, questions of lean mixture flammability limits and the extent of combustion are important. The 1976 literature survey by Keilholtz (1) provided citations for most of the sources used in this brief study, and provided much of the available data on flammability and extent of combustion.

EXTENT OF COMBUSTION

Keilholtz states that combustion of 100 percent of the hydrogen will not occur until the hydrogen comprises about 10 vol percent of the H₂-air mixture. A partial combustion data point of 50 percent combustion is quoted for a 5.6 vol percent H₂ mixture in air. This point is attributed to Shapiro and Moffette (2), a reference that we were unable to obtain in the available time. However, Furno, et al. (8) indicate about 90 percent combustion for an initial mixture of 8.5 percent H₂ as compared with 5-10 percent combustion for mixtures of 6.9-7.4 percent H₂. If 300 kg H₂ were uniformly distributed throughout the active volume of the Sequoyah Unit 1 containment vessel, it would constitute a 10 vol percent mixture with air (neglecting steam), and hence could burn completely.

FLAMMABILITY LIMIT

The lean mixture threshold of flammability is given by Keilholtz as 4.1 vol percent H₂ in air but at this concentration, Egerton (3) as well as Keilholtz point out that the flame front is not coherent, and flame propagation is upward only.

Downward propagation begins with a hydrogen concentration of about 9 vol percent (1), (3). Drell and Belles (4) state that a 9 percent mixture will burn completely (a point to be compared with the Keilholtz 10 percent mixture for 100 percent combustion). Even the lean mixture non-coherent flames are postulated to burn a mixture that is richer than the original mixture, because the high diffusion rate of H_2 permits access of additional H_2 to the flame (4). The diffusion rate of H_2 is also important to the dispersal of segregated pockets of hydrogen, and will be discussed later.

STEAM DILUTION

The effects of dilution by steam are potentially important. Drell and Belles (4) state that inert diluents have scarcely any effect on the lean-mixture limit of flammability, where 300-600 kg of H_2 in Sequoyah would be, if uniformly distributed. They claim water vapor has effects similar to CO_2 , and they show data of Coward and Jones (5) (which we were unable to obtain) such that only after more than half the mixture is CO_2 does the fraction of H_2 required for flammability begin to increase. These findings are consistent with the ternary mixture chart of Shapiro and Moffette for H_2 , air, and steam, wherein the lean mixture flammability limit is at a nearly constant H_2 fraction as the steam content increases from zero to about 50 vol percent.

DETONATION

Shapiro and Moffette indicate a triangular shaped detonation regime in their ternary mixture chart, a regime bounded approximately by a 19 vol percent H_2 line at the lean mixture boundary and a 45 vol percent air line at the rich mixture boundary. Although the original reference was not available to us, it appears that the authors constructed the detonation regime by extrapolating from data on dry mixtures of H_2 and

air. We note that Drell and Belles show the range of detonability of H_2 in air from 18.3 vol percent to 50 vol percent H_2 . We could find no information on the effects of inert diluents on the detonability of hydrogen-air mixtures, and we note the caption on the Shapiro-Moffette ternary mixture chart: "Assumed Detonation Limits." We conclude that the effects of steam on detonability of H_2 -air mixtures are essentially unknown. The nearest information we could find was cited by Keilholtz, and this pertains to detonations in Knallgas-team mixtures (6). Knallgas is a stoichiometric mixture of H_2 and O_2 . In reference (6), experiments indicated that a minimum of about 65 vol percent Knallgas in saturated steam at $100^\circ C$ was required for detonation. This would correspond to about 44 percent H_2 .

The occurrence of detonation is also influenced by the size and configuration of the vessel, and the nature of the walls (4,7), which further complicates efforts to predict detonation precisely.

REFERENCES

1. G. W. Keilholtz, "Annotated Bibliography of Hydrogen Considerations in Light-Water Power Reactors, ORNL-NSIC-120, Feb. 1976.
2. Z. M. Shapiro and T. R. Moffette, "Hydrogen Flammability Data and Application to PWR Loss-of-Coolant Accident," WAPD-SC-545, 1957.
3. Sir Alfred C. Egerton, "Limits of Inflammability," Fourth Symposium (International) on Combustion, the Williams and Wilkins Co., Baltimore, 1953, pp. 4-13.
4. I. L. Drell and F. E. Belles, "Survey of Hydrogen Combustion Properties," National Advisory Committee on Aeronautics Report 1383, 1958.
5. H. F. Coward and G. W. Jones, "Limits of Flammability of Gases and Vapors," U. S. Bureau of Mines, Bulletin 503, 1952.
6. J. A. Luker and E. C. Hobaica, "Effect of the Initial Mixture Density on the Formation of Detonation in Knallgas Saturated with Water Vapor," Journal of Chemical and Engineering Data, 6, 2 April 1961, pp. 253-256.
7. L. B. Adler, E. C. Hobaica, and J. A. Luker, "The Effect of External Factors on the Formation of Detonation in Saturated Knallgas-Steam Mixtures," Combustion and Flame, 3, 481, 1959.
8. A. L. Furno, E. E. Cook, J. M. Kuchta, and D. S. Burgess, "Some Observations on Near-Limit Flames," Thirteenth Symposium (International) on Combustion, 1971.

APPENDIX B

HYDROGEN-AIR MIXING BY FAN

Air recirculation fans are provided in the Sequoyah containment for returning air to the lower compartment after a postulated blowdown. Two such fans are provided, each having a rated capacity of about 40,000 cfm. The purpose of the fan-induced recirculation is to convey steam produced by residual heating to the ice condenser, if the emergency core cooling system should fail (failure of the ECCS is also a situation that could permit a zirconium-water reaction and hydrogen generation). The design basis for the recirculation system is an air flow rate of 40,000 cfm, corresponding to the operation of one fan. Some parameters related to mixing and burning of hydrogen in an air flow of 40,000 cfm have been calculated, and are presented in Table 4.

The air velocities in the ice condenser and upper plenum are low. Nevertheless, the flow would be turbulent in the upper plenum of the ice compartment, so the flow entering the upper compartment should be well mixed. If hydrogen were being generated by a 1% per minute reaction of zirconium (as an example), the rate of hydrogen flow would be about 10% of the air flow, giving a mixture containing about 9% H₂. This would be combustible, according to the literature cited elsewhere in this report.

A reference calculation is illustrated in Figure 2, where mixtures of 40,000 cfm air and the hydrogen yields of various rates of zirconium reaction are plotted on the ternary mixture chart. Each reaction rate corresponds to a straight-line locus, with steam rate determining the position on any line. The one point plotted on each line is for a steam rate that corresponds to the heat release rate of the Zr-H₂O reaction and the latent

heat of vaporization of water. It can be seen in Figure 2 that the yield of Zr-H₂O reaction rates in excess of 2% per minute can produce detonable mixtures with 40,000 cfm of air if the steam content is sufficiently low. Rates of several percent per minute were calculated for some accident scenarios in WASH 1400.

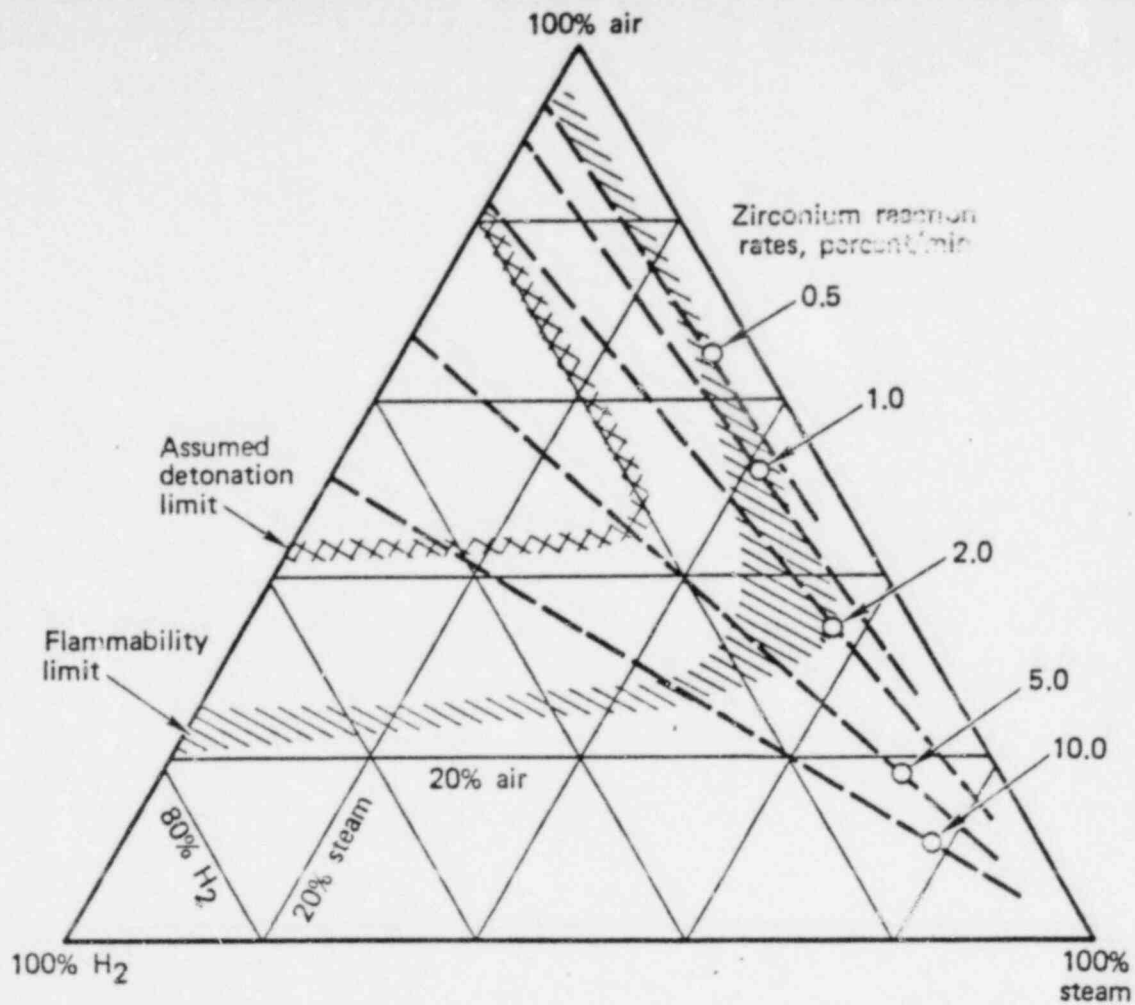
Table 4. Air Circulation Parameters

Design Data From Sequoyah PSAR

Number of Blowers	2
Capacity of Each Blower	40,000 cfm
Ice Condenser: Flow Area (net)	1,326 ft ²
Height	48 ft
Annular Thickness	11 ft
Effective Circumferential Length	267 ft
Lower Compartment Active Volume	2.87x10 ⁵ ft ³
Total Containment Active Volume	1.24x10 ⁶ ft ³

Derived Parameters, for One Blower Operating

Air Velocity: a) In Ice Bed	30 ft/min
b) In Upper Plenum of Ice Compartment	14 ft/min
Air Reynolds Number in Upper Plenum (kinematic viscosity of air @ 50°C = 1.15x10 ⁻² ft ² /min)	2.6x10 ⁴
Air Residence Time in: Ice Compartment	1.6 min
: Lower Compartment	7.2 min
: Total Active Volume of Containment	31 min



Limits of flammability and detonation based on Shapiro and Moffette WAPD-SC-545, as reproduced in WASH 1400.

Figure 2. Locus of state points for mixtures of 40,000 cfm air with the hydrogen yield of various Zr reaction rates.

Law
Page 4

1. Controlled Ignition Phase I Program Status

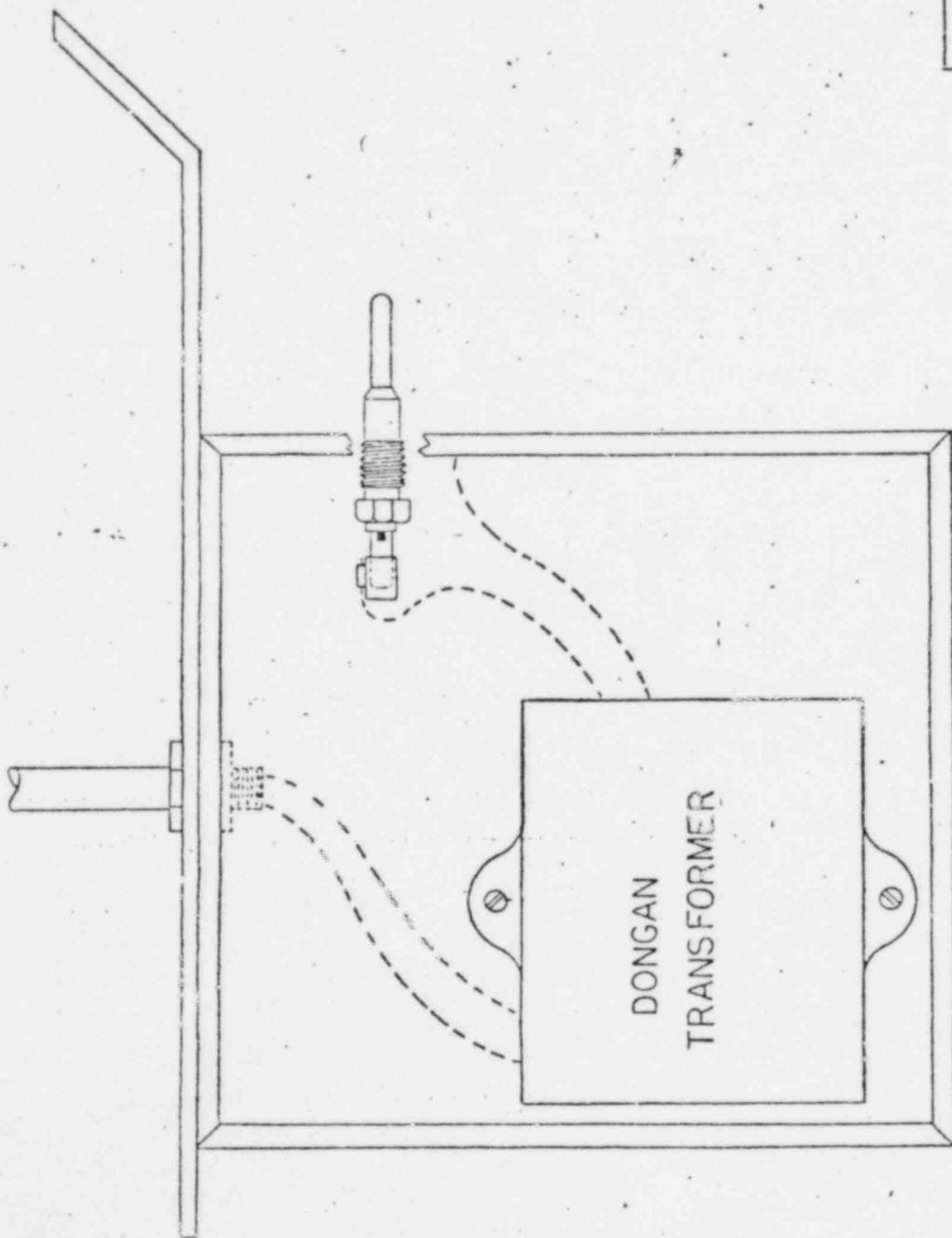
- o Description, installation, and operation of thermal igniters.
- o Spark igniters and electromagnetic interferences

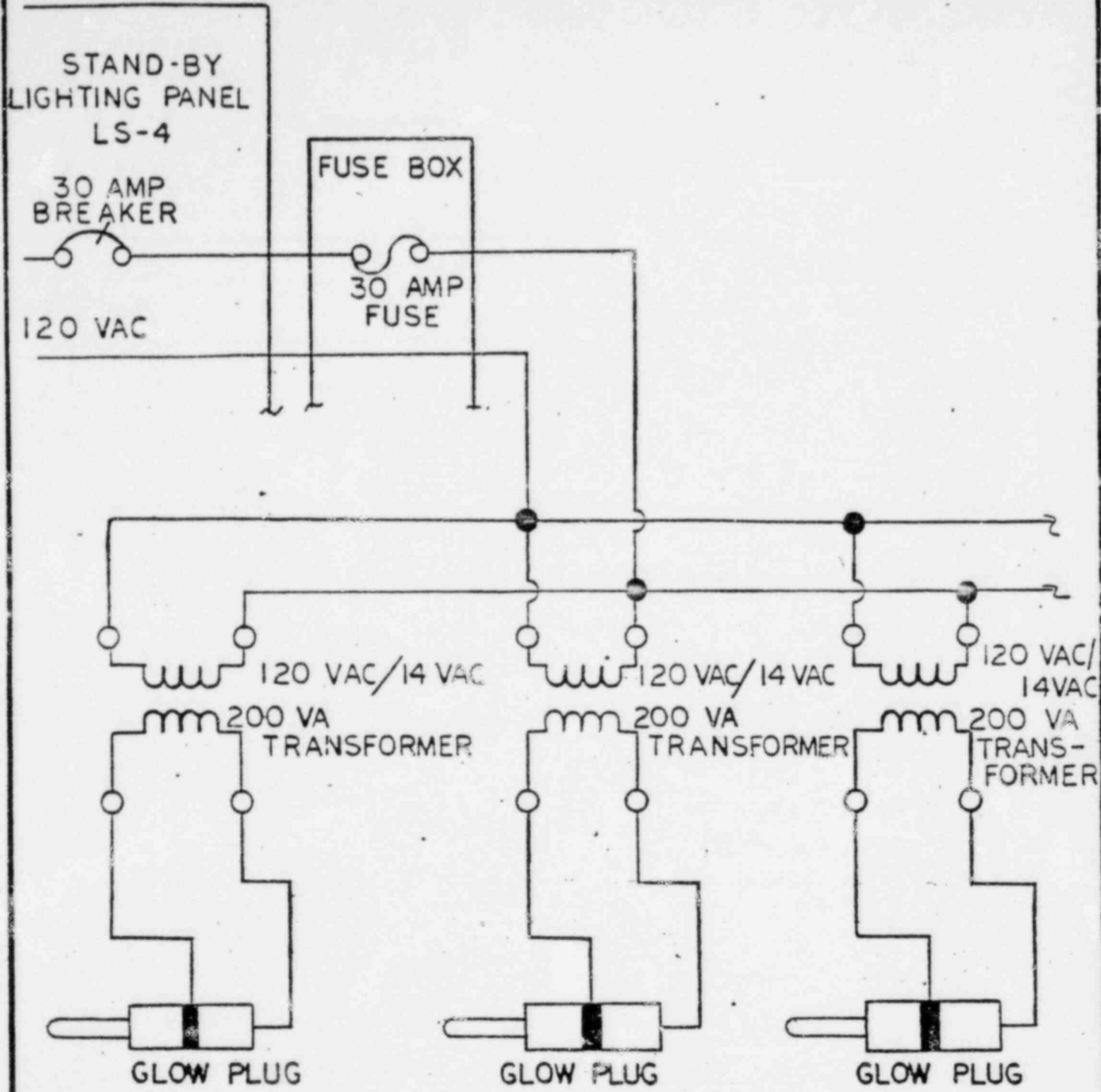
2. Singleton Lab Tests

3. Fenwall Tests

4. Halon Study

5. Degraded Core





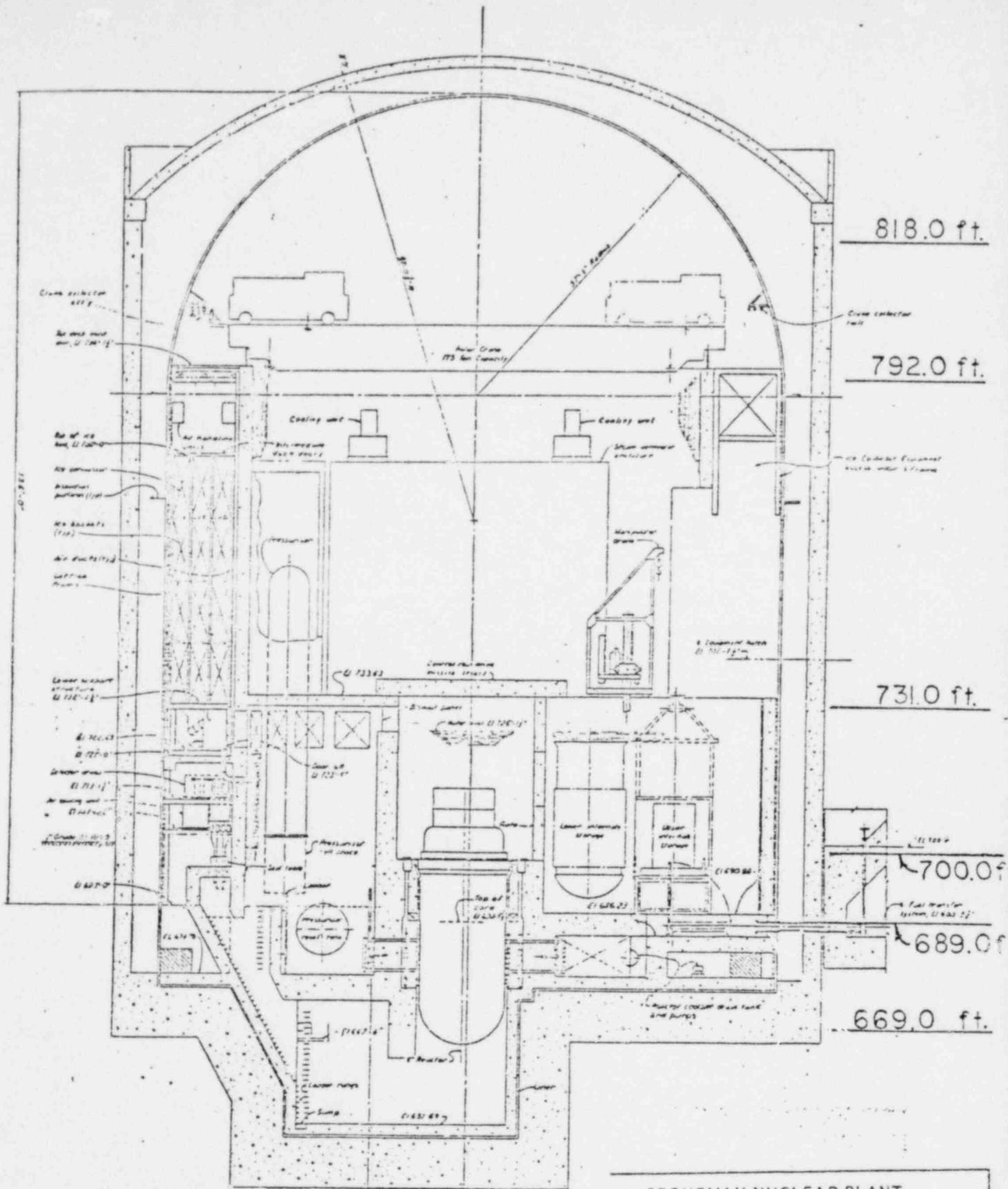
TYPICAL OF ONE CIRCUIT

(NUMBER OF GLOW PLUGS PER CIRCUIT VARIES FROM ONE TO EIGHT)

ELECTRICAL SCHEMATIC FIGURE 15									
TENNESSEE VALLEY AUTHORITY DIVISION OF ENGINEERING DESIGN									
SUBMITTED			RECOMMENDED			APPROVED			
KNOXVILLE									

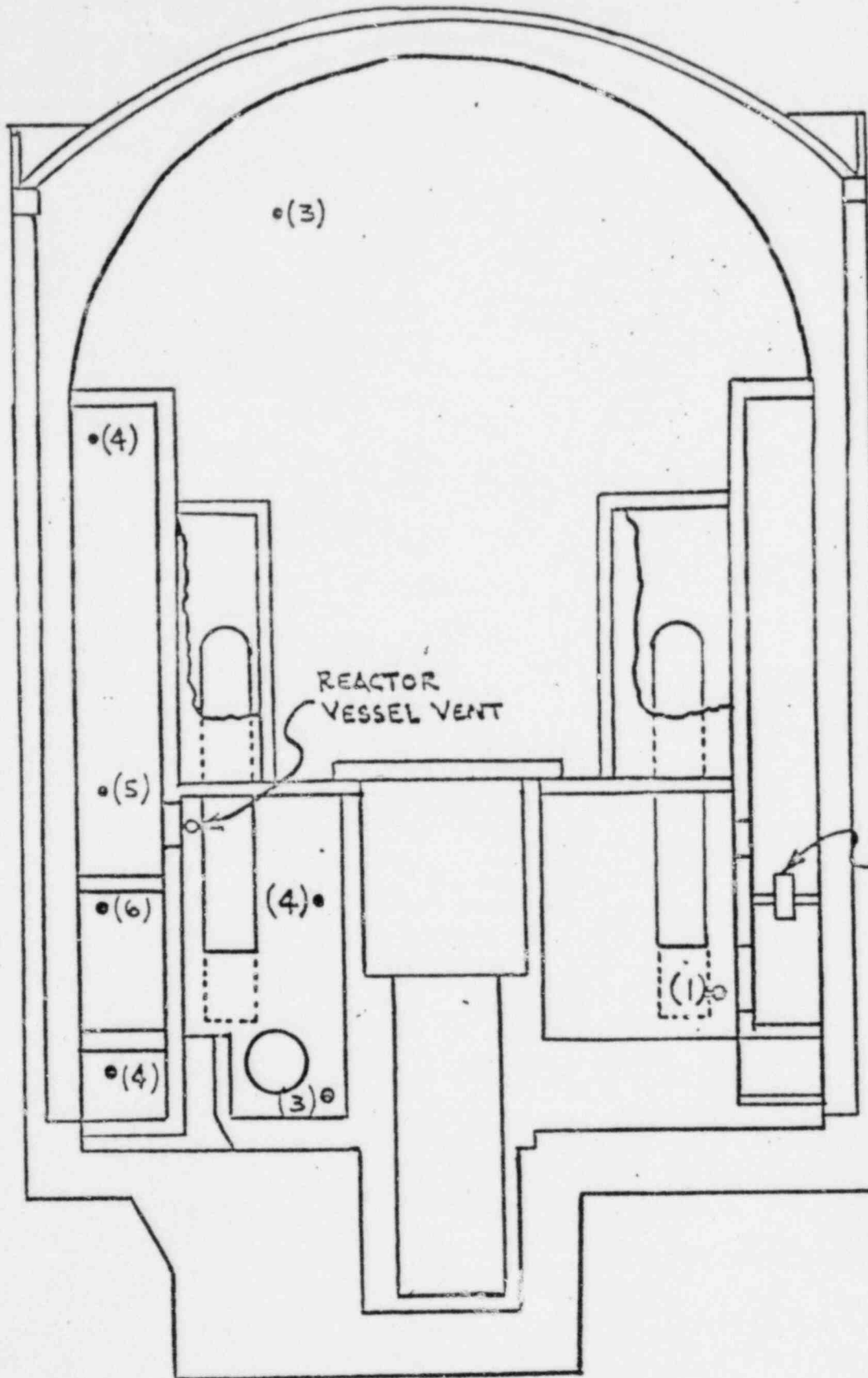
REV. NO.	ECN. NO.	DATE	DESIGN	OPENED	CHECKED	SUPV.	ENGR.	WSP.	SUBMITTER	RECH.	APPR.

P3MM



SEQUOYAH NUCLEAR PLANT
 FINAL SAFETY
 ANALYSIS REPORT

CONTAINMENT LIGHTING
 FIXTURE ELEVATION
 FIGURE 6.2-141



818.0 ft.

792.0 ft.

731.0 ft.

AIR RETURN FAN

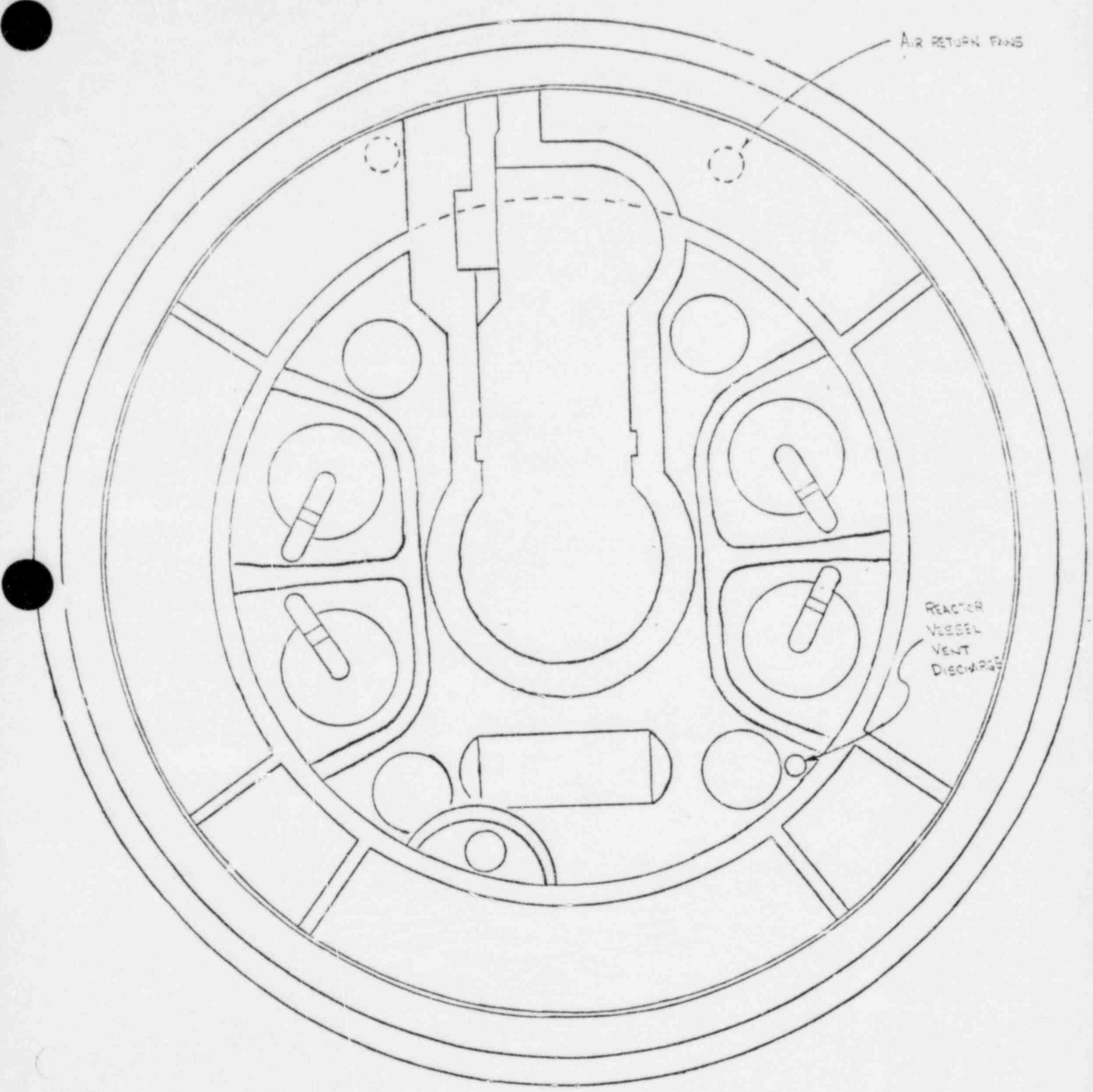
700.0 ft.

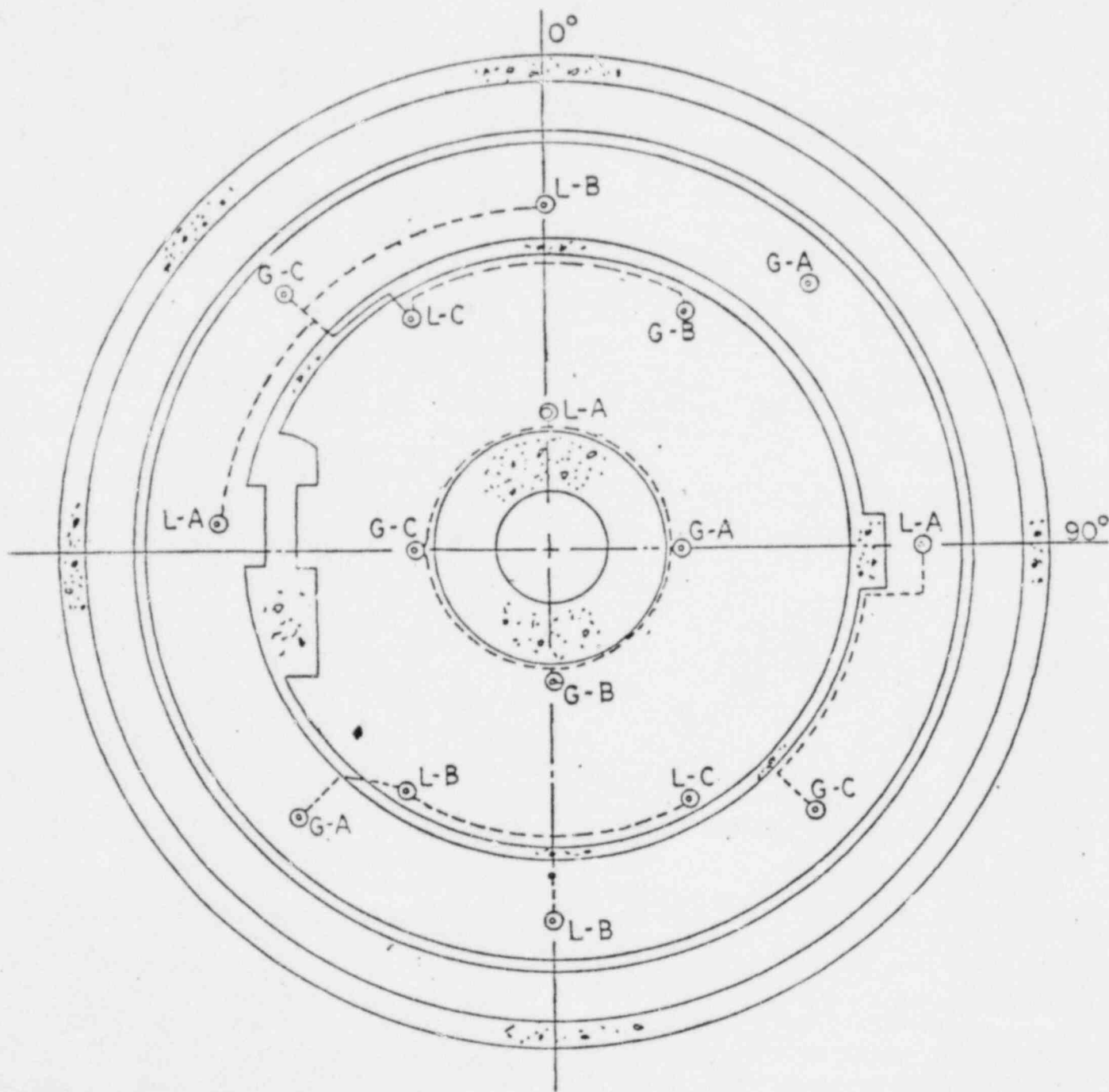
689.0 ft.

669.0 f

AIR RETURN FANS

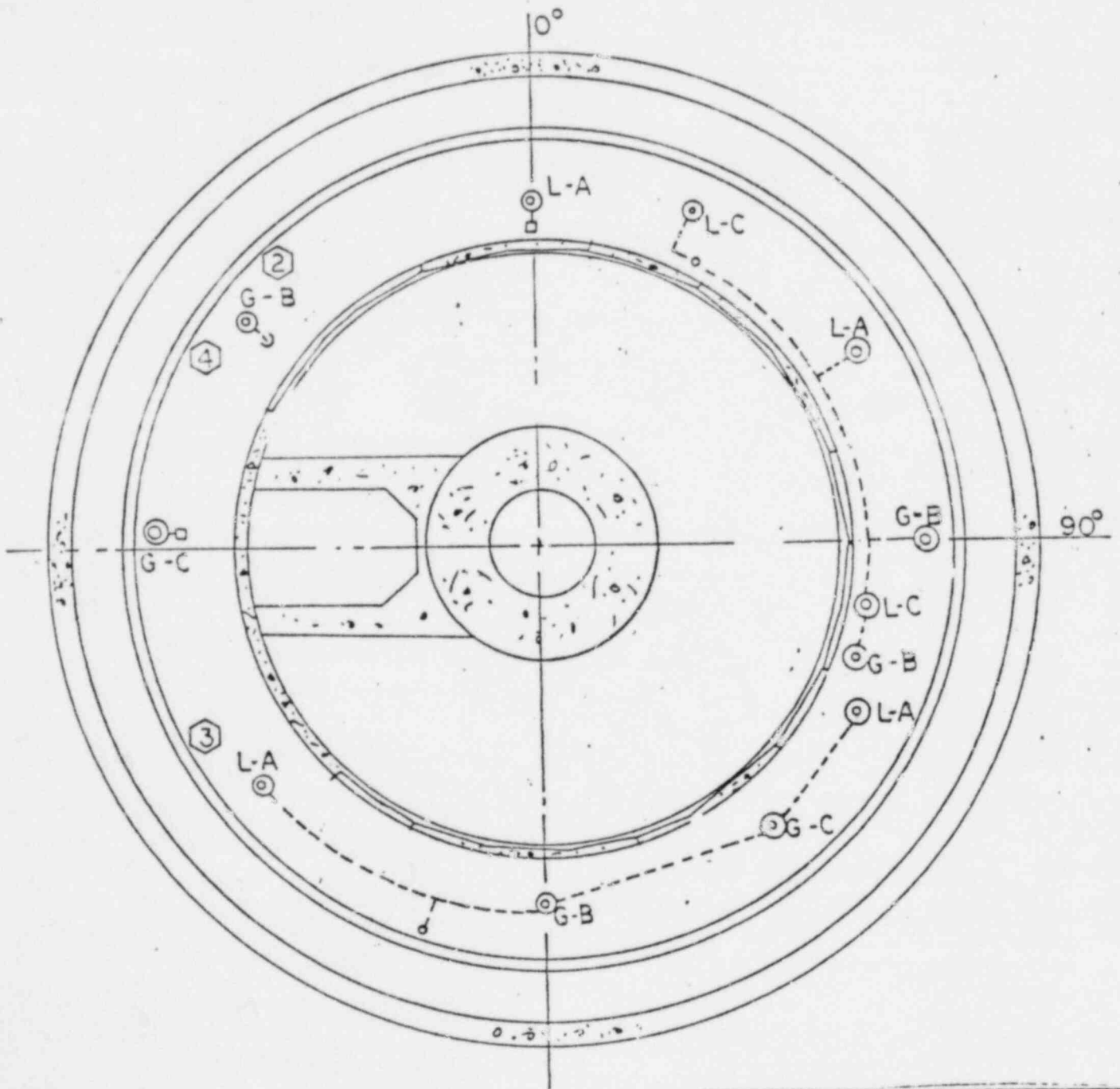
REACTOR
VESSEL
VENT
DISCHARGE





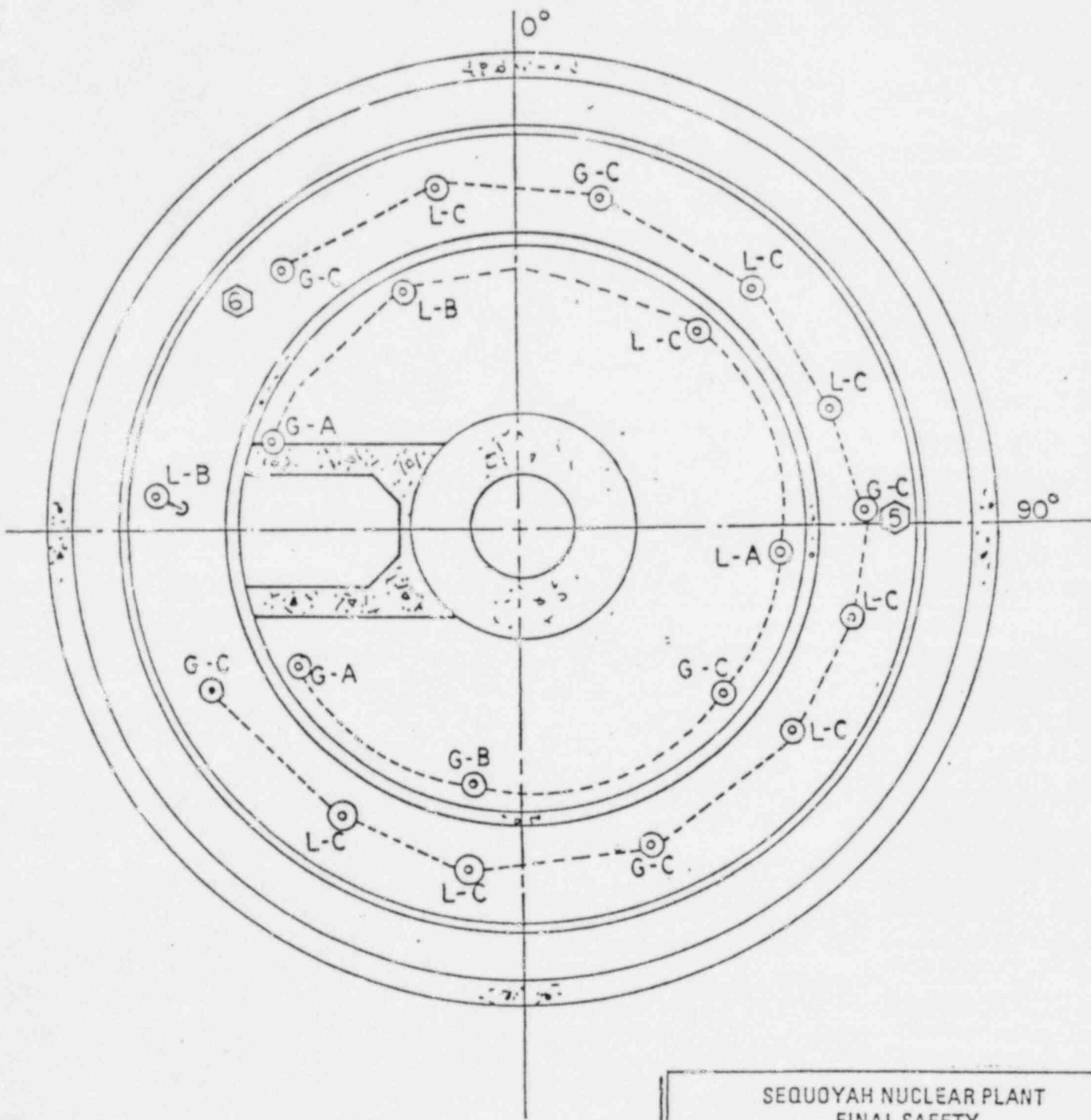
SEQUOYAH NUCLEAR PLANT
 FINAL SAFETY
 ANALYSIS REPORT

CONTAINMENT LIGHTING FIXTURES
 EL. 689.0'
 FIGURE 6.2-142



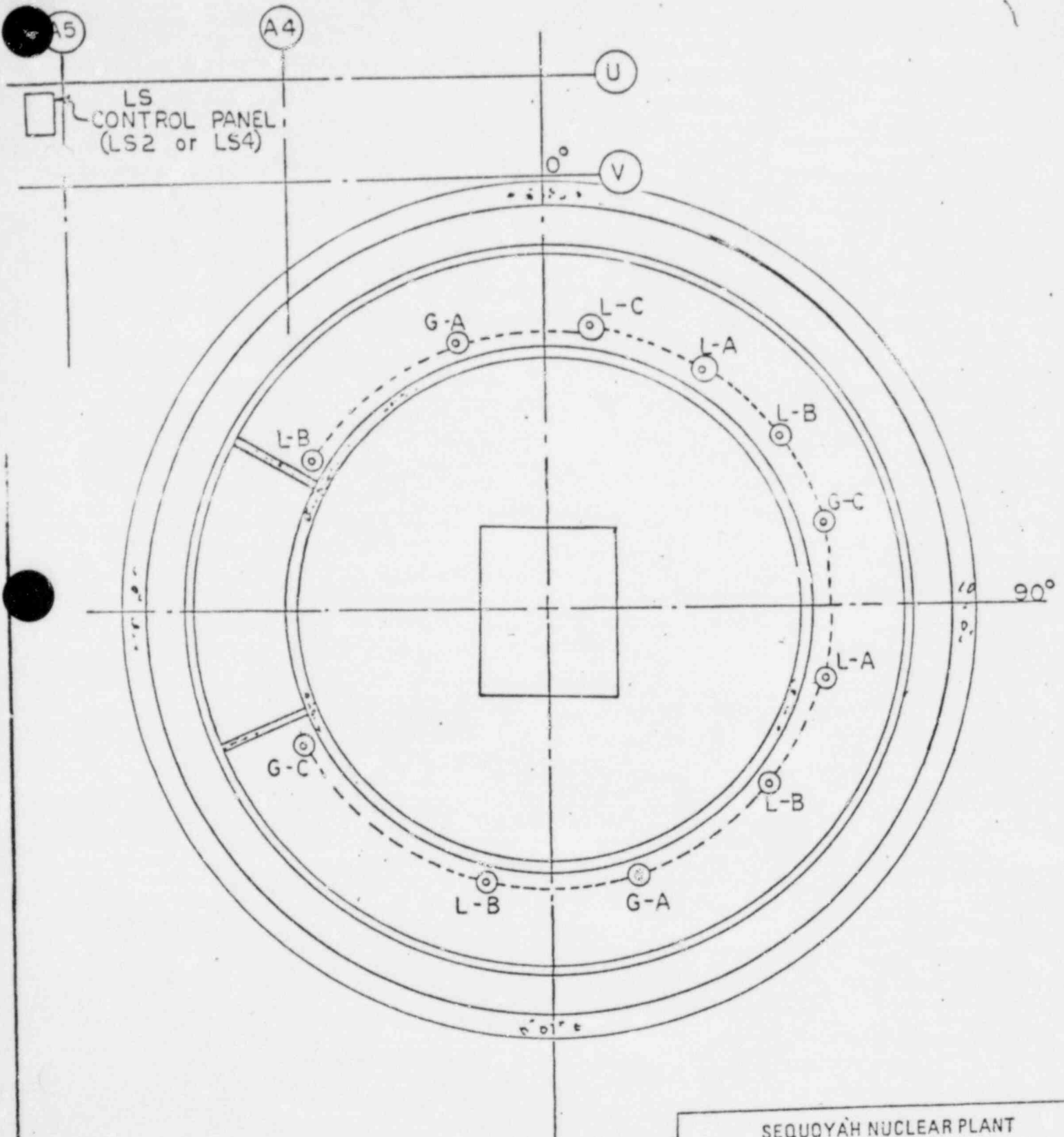
SEQUOYAH NUCLEAR PLANT
 FINAL SAFETY
 ANALYSIS REPORT

CONTAINMENT LIGHTING FIXTURES
 EL. 700.3
 FIGURE 6.2-143



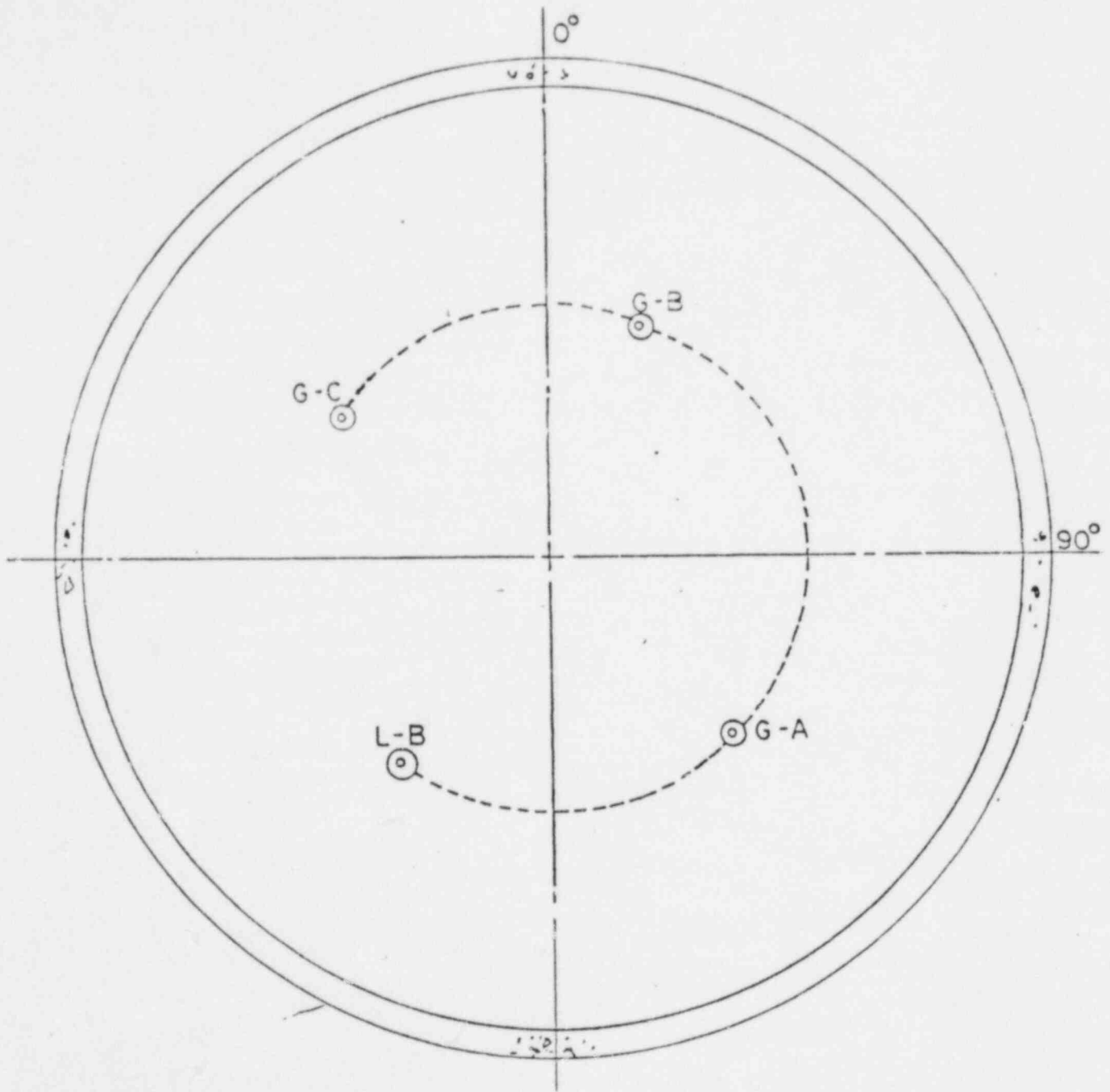
SEQUOYAH NUCLEAR PLANT
FINAL SAFETY
ANALYSIS REPORT

CONTAINMENT LIGHTING FIXTURES
EL 731.0'
FIGURE 6.2-144



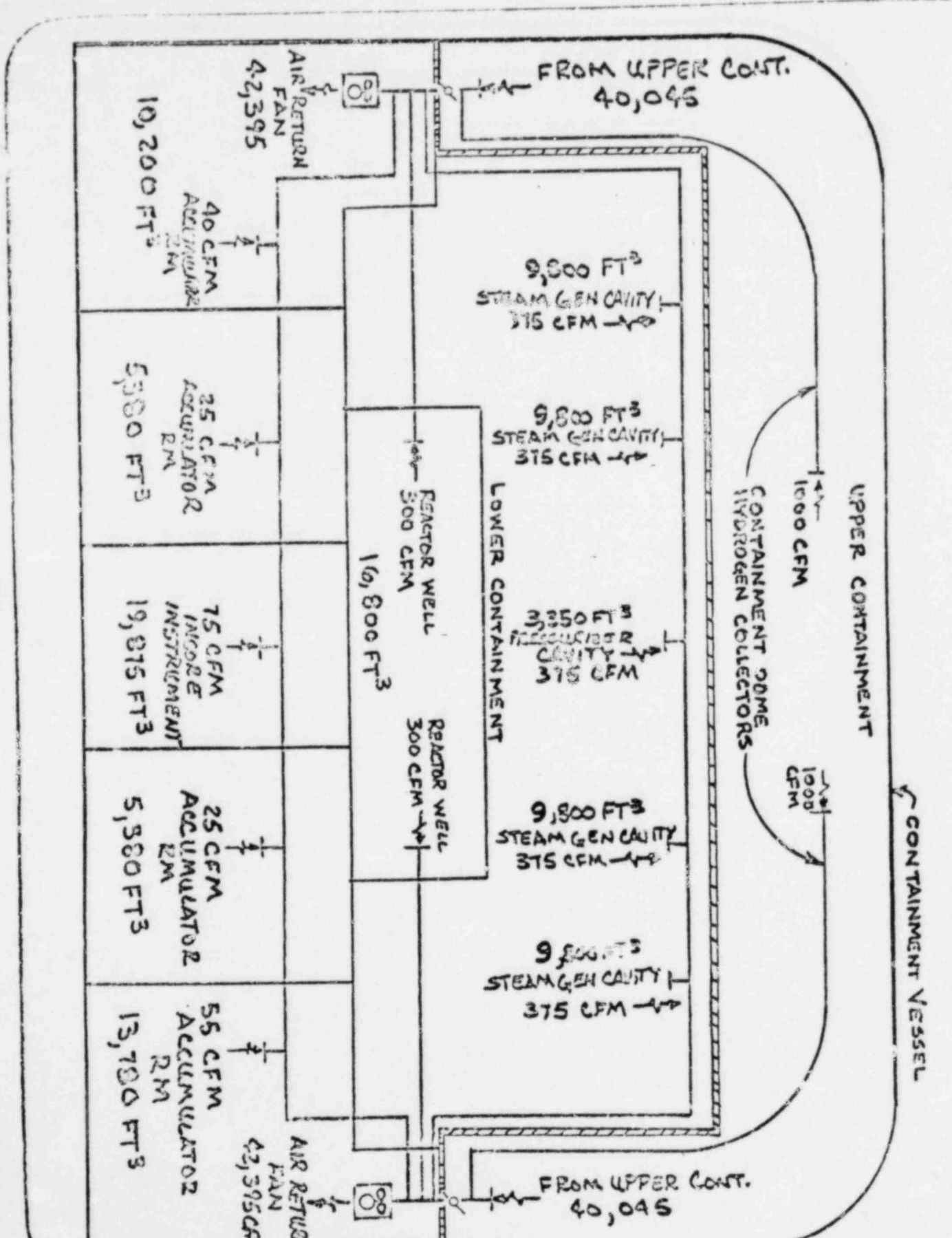
SEQUOYAH NUCLEAR PLANT
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 ANALYSIS REPORT

CONTAINMENT LIGHTING FIXTURES
 EL 792.0'
 FIGURE 6.2-145



SEQUOYAH NUCLEAR PLANT
FINAL SAFETY
ANALYSIS REPORT

CONTAINMENT LIGHTING FIXTURES
EL 818.0'
FIGURE 6.2-146



10,200 FT³
40 CFM
ACCUMULATOR
RM

5,300 FT³
25 CFM
ACCUMULATOR
RM

19,815 FT³
75 CFM
INCORE
INSTRUMENT

5,530 FT³
25 CFM
ACCUMULATOR
RM

13,700 FT³
55 CFM
ACCUMULATOR
RM

9,800 FT³
STEAM GEN CAVITY
375 CFM

9,800 FT³
STEAM GEN CAVITY
375 CFM

9,800 FT³
STEAM GEN CAVITY
375 CFM

9,800 FT³
STEAM GEN CAVITY
375 CFM

9,800 FT³
STEAM GEN CAVITY
375 CFM

REACTOR WELL
300 CFM

16,800 FT³

REACTOR WELL
300 CFM

FROM UPPER CONT.
40,045

PRELIMINARY TESTING TO IDENTIFY

COMMERCIALY AVAILABLE IGNITERS

TESTING CONDUCTED AT TVA'S

SINGLETON LABORATORIES

1.0 Introduction

TVA has a testing program which is being conducted at TVA's Singleton Laboratory to obtain preliminary information about the performance of commercially available igniters. The purpose of these tests was to screen alternative igniters and to gain a degree of confidence that the igniters could ignite hydrogen. The tests were not run under normal laboratory test conditions since the objective was to identify which igniters, if any, were most promising as subjects for more detailed testing and evaluation. Nonetheless, TVA gained considerable information and assurance that commercially available igniters could ignite hydrogen.

2.0 Preliminary Screening

A number of igniter types were evaluated, ranging from high energy spark igniters to large diameter (1-1/2" I.D.) heater coils. Although the spark plug type igniter was considered an excellent candidate for this application, it was rejected prior to preliminary testing due to potential problems with electromagnetic interference (EMI) with critical instrumentation. TVA's Electrical Engineering Branch is researching the problems associated with EMI generators, and spark type igniters may be considered at a later date for use in Sequoyah unit 2 or Watts Bar.

Two other potential candidates, both coil heaters, were rejected

after the first one, a large diameter (1-1/2" I.D.) coil, could not reach sufficient surface temperature, and the second one failed at the connector in less than five minutes. Therefore, testing was restricted to diesel engine glow plugs, since they were known to be capable of achieving the 1500^oF minimum surface temperature desired by TVA and because of their rugged design.

TVA determined that at 12 volts ac, acceptable surface temperatures could be achieved but that considering line losses, variances in system voltages, possible plug cooling due to high humidity, and other effects, TVA would need to operate the plugs at 13 volts ac \pm 1 volt.

Since the possibility existed that TVA could overstress the plugs by overvoltage, TVA consulted glow plug manufacturers and identified two types of failure modes which could be expected. The first type of failure caused by overstressing would be the failure of the heater wire within the glow plug sheath. This type of failure due to the breaking of the circuit would outwardly cause the plug to discontinue glowing. The second type of failure caused by overstressing would involve offgassing of the glow plug tip. Unlike the first type of failure after offgassing, the glow plug may continue to glow; however, the surface temperature would drop significantly.

3.0 Description of Glow Plugs

Glow plugs manufactured by three different companies have been

levels both on the primary and secondary side and at the plug were measured by a digital voltmeter (Fluke model number 8024A), and the current levels were measured by an amp meter (Triplett model number 10 type 2). The surface temperature of each of the glow plugs was measured by either a thermocouple (type S) connected to a potentiometer (Leeds and Northrop model number 8690-2) in contact with the surface of the plug or by an optical pyrometer (Pyro model number 85). A total of 12 plugs have been tested to date.

4.2 Surface Temperature

A GMAC model 7G plug was operated at 12, 14, and 16 volts ac. Surface temperatures as measured by the thermocouple were 1480, 1550, and 1650^oF, respectively. Since the thermocouple would be expected to increase local heat loss and hence reduce the measured local surface temperature of the thin-walled plug sheath, these values were probably somewhat lower than actual surface temperatures. This conclusion was supported by later readings with the pyrometer while testing another GMAC model 7G at 14 volts ac and getting 1720 \pm 15^oF.

A Bosch plug has been tested at 13 volts ac. It produced a surface temperature of 1700^oF as measured by an optical pyrometer. Based on these results, TVA concluded that the diesel glow plugs could produce the desired surface temperatures.

4.3 Voltage Tests

Voltage tests have been completed on only the GMAC model 7G plugs. Based on tests on 5 GMAC 7G plugs, reliable operation at 14 volts was confirmed by two other 7G plugs failed at 16 volts ac after a few minutes.

Inconclusive testing on 2 Bosch plugs resulted in failure when operated at 14 volts ac; however, one Bosch plug operated satisfactorily at 13 volts ac. One Isusi plug was tested at 14 volts ac but lasted for only 30 minutes.

4.4 Extended Operation

Endurance tests have been performed on only two plugs for extended periods of time. A GMAC model 7G plug was operated continuously for 148 hours without failure and was later used in the hydrogen burning tests. A Bosch 10.5 volt plug was operated at 13 volts for 90 hours, then cooled down for two hours and turned back on. It has been running continuously after being reenergized since August 20, 1980, at 10 a.m.

5.0 Hydrogen Testing

One igniter (AC 7G) was installed in a "PARR" (229HC6-T316-031579-

5142) pressure vessel in order to determine feasibility of igniting hydrogen in a sealed container. The vessel lid has a silicone rubber sealed gas injection sampling port. Hydrogen concentrations in the vapor phase were determined before and after ignition intervals. An ignition interval is the time current flows through the igniter circuit. The hydrogen was measured by a Perkin-Elmer gas chromatograph equipped with 3920 thermal conductivity detector and an M-2 integrator. The chromatograph was standardized with hydrogen and air mixtures prepared from research grade hydrogen and laboratory air.

Temperature measurements were made with a mercury and glass (484635, ASTM 9C) thermometer. Temperatures reported are ambient for tests 1 through 3. Prior to tests 4-10, 100 grams of water was added to the vessel. The vessel was heated by a temperature adjustable hot plate to saturation temperature of the water and maintained throughout the test. The reported temperature is the water temperature after completion of the test. Results of the 10 ignition tests are given in table 1.

6.0 Future Tests at Fenwall Laboratories

TVA and Westinghouse have contracted with Fenwall Laboratories to perform hydrogen burn testing on the AC igniter and its mounting enclosure in an enclosed vessel. Attachment 1 is the proposed Test Plan for the testing. The final test plan is being prepared by Fenwall and should be available in the near future. These tests are designed to prove the effectiveness of this

igniter assembly to burn a volumetric quantity of hydrogen in environmental conditions which approximate postulated accident conditions inside containment.

7.0 Conclusions and Summary

The purpose of these tests at Singleton was to select a commercially available igniter that was capable of igniting hydrogen. From the results obtained, the GMAC model 7G glow plug produces more than adequate temperatures at a range of voltages that can be provided inside the Sequoyah containment.

In addition, the plug seems capable of extended operation at high temperatures and has been shown in small tests to be able to ignite 12 percent and lower volumetric quantities of hydrogen. Although it has not been tested as thoroughly, the Bosch plug appears like it may also be an optional igniter.

E50239.06

TABLE 1
HYROGEN IGNITION TESTS

<u>Test No.</u>	<u>Vessel Contents</u>	<u>Temp. (°F)</u>	<u>Initial Hyd. Conc. (% Hyd.)</u>	<u>Final Hyd. Conc. (% Hyd.)</u>	<u>Ignition Intervals (Min.)</u>
1	Hyd., Air	90	12.5	0.1	5
2		80	7	0.1	5
3		80	3.5	0.1	5
4	Hyd. Air, Water	120	12	0.1	3
5		180	14	0.5	3
6		180	4	2.5	1
7		180	2.5	1.5	1
8		180	1.5	1.3	1
9		180	11	5	1
10		180	5	2	1.3

Vessel Volume 1.1 dm³ (0.039 ft³)

Operating Voltage 12V dc

E50239.07

ATTACHMENT 1

SUMMARY

SEQUOYAH PLANT

HYDROGEN IGNITER TEST PLAN

1. Introduction

The following describes tests to be conducted on a type of hydrogen igniter to be installed in the Sequoyah Nuclear Plant. The igniter consists of a "glow plug" as used in diesel engines, the surface of which exceeds 1500° F and serves as a hot surface to initiate hydrogen burning, and a power transformer and an enclosure for the unit. The function of the igniters in the nuclear power plant containment is to burn hydrogen, in accidents where it could be released, when it reaches a burnable concentration thereby precluding its buildup to high concentration levels. The tests will be conducted by Fenwall, Incorporated, at their facilities in Ashland, Massachusetts. The unit, consisting of glow plug and enclosed transformer, will be placed in a test vessel and subjected to a range of environmental conditions (including hydrogen concentration, temperature, pressure, and steam), and its hydrogen ignition performance monitored.

1.1 Purpose of Tests

The primary purpose of the tests is to demonstrate that the igniter will initiate a volumetric burn of the hydrogen for the specified environmental conditions (pressure, temperature, water vapor). A secondary objective of the tests is to narrow down the hydrogen concentration range for which a volumetric burn of hydrogen will be initiated.

1.2 Acceptance Criteria

For the initial set of tests, the following acceptance criteria will be used:

1. Data generated are internally consistent (i.e., ignition at 8% consistently produces low pressure rise).
2. Data gathered confirm theoretical predictions.
3. Igniters reliably ignite mixtures at high (12%) concentration and provide relatively complete combustion.

2. Description of Igniter

The igniter is a General Motors Ac Division Model 7G glow plug

(thermal resistive heating element) requiring 14V ac supply at a maximum of 8-1/2 amps. The surface temperature of the plug as measured by an optical pyrometer should be a minimum of 1500° F. TVA has measured 1720° F surface temperature on one of the glow plugs at their facilities. The igniter is powered by 120V ac stepped down to 14V ac. The power transformer is a Dongan Electric, Incorporated, Model 52-20-187 specially wound transformer having the following characteristics:

120V	RMS	AC	on primary side
14V	RMS	AC	on secondary side
200V	A	Min.	

Class H (High temperature insulation)

Open style with 18" flexible leads

Certified capability that transformer will operate at 220° C.

The igniter and transformer are mounted as a unit as shown in Figure 1 with the glow plug extending from the side. The unit is encased in a 1/8-inch steel plate box type casing and sealed with a rubber seal for water tightness.

3. Description of Test Facility

The tests will be conducted by Fenwall, Incorporated, at their facilities in Ashland, Massachusetts.

3.1 Test Vessel

The igniter unit will be tested in a spherical vessel in excess of six feet in diameter. The internal volume of the test vessel is 1000 gallons (134 ft³). The vessel is constructed of carbon steel (exterior) and is lined with stainless steel. The vessel is designed for a working pressure of 500 lb/ft². The vessel is equipped with five diameter access ports (four on circumference, 90° apart,

and fifth at the top), one of which is drilled to attach to a manifold with valves and connecting lines to air, steam, and hydrogen makeup sources.

The vessel is heated externally via electrical heaters. The vessel will be equipped internally with a fan to promote mixing and also to create a draft at the igniter heating surface during testing when desired.

3.2 Instrumentation and Measurements

The vessel is instrumented with two pressure transducers to monitor the pressure including the pressure transient during the hydrogen burn. The output is carrier amplified and feeds to an oscillograph device. Thermocouples are provided which will monitor vessel atmosphere temperature prior to and after a burn. In addition, a thermocouple will be used to measure the temperature of its heated

surface. Gas mixtures will be formed using pressure instrumentation and a partial pressure method in which a given gas is added until the appropriate partial pressure is indicated. Sampling capability exists via a 1-inch by 1-foot lecture bottle. Hydrogen and oxygen analyzers will be provided to measure pre- and post-burn concentrations of these gases:

	<u>O₂ Analyzer</u>	<u>H₂ Analyzer</u>
Manufacturer	Hays Republic	Hays Republic
Model	A 00632	SH-A-00643D
Range	0-5%/0-20%	0-5%/0-20%
Accuracy	± 1% F.S.	± 2% F.S.

4. Test Plan

4.1 Identification of Tests

The unit consisting of the glow plug and encased transformer will be positioned in the test vessel (via 18-inch port) with the glow plug heating surface located near the center of the test vessel. Various mixtures of H₂, steam, and air will be adjusted with pressure and temperature as specified and then the igniter turned on. The pressure transient will be recorded and the mixture analyzed for H₂ and O₂ content prior to and after the burn. The test matrix for the first series of 12 tests is shown in Table 1. Initial total pressures of 15, 21, and 27 lb/ft² will be covered at hydrogen concentrations of 8 and 12-volume percent. Initial temperature will vary from 180° F (dry case) to 350° F (superheated steam) with most of the tests being conducted at saturation temperature corresponding to the pressure to be tested. In addition, a fan will be located in the test vessel to provide drafts of 5 and 10 FPS in the vicinity of the glow plug to simulate turbulence which may be developed in the vicinity of the igniters.

Further testing will be developed based on the outcome of test series #1, and may include addition of an instrumented transmitter and steel or concrete surfaces with thermocouples attached to measure temperature response on hydrogen burn. In addition, means to simulate spray droplet entrainment in the atmosphere are under investigation.

4.2 Test Procedure

The basic procedure is to adjust mixture concentration temperature and pressure, then energize glow plug and record the pressure and temperature transient. Hydrogen concentration after the burn will be measured to assess completeness of burn. The steps for the different tests are as indicated in Table 2. In one of the tests with a steam environment, the glow plug will be energized after the steam, pressure, and temperature environment conditions

are reached, but before hydrogen is added, and allowed to stand for two hours. Then the glow plug will be deenergized, hydrogen adjusted, and then the glow plug energized. The purpose of this is to allow for preburn exposure to the environment.

4.3 Test Schedule

The test schedule is tentatively planned as follows:

Facility Preparation	8/18 through 8/29
Test Series No. 1	9/1 through 9/5
Subsequent Tests	9/8 through 9/12
Test Evaluation	9/15 through 9/19

DE01;SQNHVD.AA

TABLE 1

TEST SERIES NO. 1

<u>Test</u>	<u>Temp (°F)</u>	<u>Total Pressure*</u> <u>(Gauge)</u>	<u>Hydrogen Concentration</u> <u>(Volume Percentage)</u>	<u>Fan Induced Flow Speed (fps)</u>
1	180	0	12	0
2	180	0	8	0
3	Sat temp	6	12	0
4	Sat temp	6	8	0
5	Sat temp	12	12	0
6	Sat temp	12	8	0
7	Sat temp	6	12	5
8	Sat temp	6	8	5
9	Sat temp	6	8	10
10	Sat temp	6	12	10
11	350	12	12	0
12	350	12	12	10

*This is the total pressure due to air, hydrogen, and steam. For tests 1 and 2, the pressure will be higher than 0 due to the added hydrogen partial pressure and the evaluated temperature.

PROPERTIES OF HALON 1301

- o LOW BOILING POINT (-72°F)
- o LOW TOXICITY (UL GROUP 6)
- o INSOLUBLE IN WATER (0.0095 W/O)
- o INERT
- o LOW RADIOLYTIC DECOMPOSITION
(0.00023 g/d/R/h)
- o NO LONG TERM ACTIVE MIXING REQUIRED
AFTER INJECTION

SUITABILITY OF HALON 1301

- o PREVENTS HYDROGEN IGNITION AT SUFFICIENT CONCENTRATIONS.

- o ATLANTIC RESEARCH CORPORATION REPORT SHOWED HALON 1301 SUITABLE FOR USE IN A MARITIME REACTOR CONTAINMENT.

- o INITIAL STUDY BY ARC FOR AEP/DUKE/TVA INDICATES HALON 1301 SUITABLE FOR USE IN ICE CONDENSER CONTAINMENT.

AREAS OF FURTHER STUDY BY ARC FOR AEP/DUKE/TVA
ON HALON 1301

- o EFFECT OF HALON 1301 AND ITS DECOMPOSITION PRODUCTS ON CONTAINMENT MATERIALS
- o TEMPERATURE AND PRESSURE EFFECTS ON CONTAINMENT DUE TO INADVERTENT ACTUATION
- o POTENTIAL PROBLEMS ON LONG TERM ACCIDENT RECOVERY
- o EFFECT OF HIGH CORE TEMPERATURES ON HALON 1301 DECOMPOSITION
- o POTENTIAL FOR NON-INERTED HYDROGEN POCKET DETONATION TO INITIATE COMBUSTION IN INERTED MIXTURES
- o PERSONNEL HAZARD DUE TO INADVERTENT OPERATION
- o SYSTEM DESIGN AND INCORPORATION

Depleted Core

- o TVA is following the state-of-the-art developments at national laboratories (Battelle, Columbus, Brookhaven, Oak Ridge), AIF, EPRI, etc.
- o TVA is building the capability to use MARCH as a starting point.
- o MARCH is not intended for design.
- o TVA has set a goal to obtain a hydrogen generation rate curve (into the containment) for a fair range of core damage accidents.

STATEMENT IN RESPONSE TO QUESTION -

DO ICE CONDENSERS NEED ADDITIONAL HYDROGEN MITIGATION SYSTEMS?

The nuclear power industry and NRC have identified many lessons in the TMI-2 event. As a result of studies by the staff, the Kemeny Commission, consultants, ACRS, and others, including TVA's own Nuclear Program Review, a large number of changes have been identified. Some were implemented almost immediately, some are in various stages of implementation, and others are the subject of intensive study or planned rulemaking. The issue of the effects of hydrogen generation from degraded cores was considered by many, including TVA, as one of the more important raised.

We are addressing all of our containment designs; while the lower design pressure is a disadvantage for this issue, the ice condenser containment also has definite advantages, including a large, passive heat removal capability.

As a result of its Nuclear Program Review, TVA committed to:

Study ways to contain larger amounts of hydrogen and to backfit feasible features into the Sequoyah design. (TVA Nuclear Program

Review: Sequoyah Nuclear Plant and the report of the President's Commission on the accident at Three Mile Island, November 1979.)

TVA moved immediately to fulfill that commitment by committing significant resources to the issue. That effort continues and at this point has resulted in a significant study of degraded core accidents and their mitigation, in a long range plan to further study and act on the recommendations of that study, and in installation of an interim distributed ignition system. We feel that the steps taken and planned to reduce the likelihood and minimize the extent of core damage events, when coupled with the plants' inherent capability to withstand substantial core damage (about 25% metal-water reaction), would provide a sufficient degree of safety for the short term until TVA's and others' studies could be completed. However, since TVA is committed to make feasible improvements in the safety of our plants, we proceeded to install the interim distributed ignition system once we were convinced that it would not reduce plant safety and had the promise of increasing the amount of metal-water reduction that the plant could withstand. Our efforts are being placed on determining how much increase in capability the interim system affords and on our long term program which addresses other alternative measures in addition to controlled ignition.

TVA believes: that Sequoyah can be safely operated at least in the short term until our studies can be completed; that the plant already has significant capabilities to withstand a range of core damage

events; and that the interim distributed ignition system increases this range of capability. We are firmly committed by policy, by staff opinion, and by actual work to take the lead on this safety issue.

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