

DUKE POWER COMPANY

POWER BUILDING

422 SOUTH CHURCH STREET, CHARLOTTE, N. C. 28242

WILLIAM O. PARKER, JR.  
VICE PRESIDENT  
STEAM PRODUCTION

June 19, 1980

TELEPHONE AREA 704  
373-4083

Mr. Harold R. Denton, Director  
Office of Nuclear Reactor Regulation  
U. S. Nuclear Regulatory Commission  
Washington, D. C. 20555

Attention: Mr. B. J. Youngblood, Chief  
Licensing Projects Branch No. 1

Subject: McGuire Nuclear Station  
Secondary Water Chemistry Monitoring  
and Control

dockets  
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Dear Mr. Denton:

Please find attached supplemental information on the McGuire Nuclear Station program to monitor and control secondary water chemistry. This information supplements my letters of October 25, 1979 and May 23, 1980 on this subject. Specifically, the information provides updated pages on Table 1 and Table 2 of my October 25, 1979 letter. In addition, Table 8 thru Table 12 is provided to describe other procedures and administrative controls involved in the secondary water chemistry program.

As stated in my October 25, 1979 letter, due to the continuing evolution of chemistry control technology the procedures and administrative control described in this program are subject to modification. Any modifications will be designed in accordance with the operating experience of McGuire Nuclear Station as well as new developments of technology. It should be emphasized that certain administrative limits specified in the secondary water chemistry program are goals we will be striving to attain. A better understanding of system capability will be obtained during the first year of operation.

Very truly yours,

William O. Parker, Jr.  
William O. Parker, Jr. *By DMS*

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Attachment

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THIS DOCUMENT CONTAINS  
POOR QUALITY PAGES

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TABLE I

SECONDARY CHEMISTRY SAMPLING SCHEDULE  
NORMAL OPERATION

ENCLOSURE 3, 5, 8

Duke Power Company  
McGill Nuclear Station

SAMPLE	pH	COD ppm	C. and Inhal. ppb	O <sub>2</sub> ppb	Na ppb	NaR <sub>10</sub> ppb	Cl <sup>-</sup> ppb	H <sub>2</sub> S ppb	MIL. SUS ppb	SO <sub>2</sub> ppb	Fe ppb	Ca ppb	NH <sub>3</sub> ppb	Pb ppb	CROSS P ppb	SPEC ACT pCi/ml	T <sup>131</sup> pCi/ml	TRITIUM pCi/ml
Steam Generator R. D. (A - D)	C	C	C	C	C	C	C	D	B	C	C	C	2/W	5/W	W	D	W	W
Main Steam (A - D)	D	D	C	C	C	C	C	D	D	D	D	D	2/W	5/W	W	W	W	W
Hotwell Pump Discharge	C	C	C*	C	C	C	C	D	D	C	C	C	2/W	5/W	W	W	W	W
Polish Resin Effluent						C	C	D	D	D	D	D	2/W	5/W	W	W	W	W
Feedwater	C	C	C	D	C	D	D	D	D	C	C	C	2/W	5/W	W	W	W	W
C. Heater Draining	3/W																	
G. Heater Draining	3/W																	
Condensate Storage Tank	W	W																
Hotwell Separators (IA1 - IC2)	D																	
Condenser Hotwell (IA1 - IC2)																		
Aux Boiler Feedwater (A or B)																		
Aux Boiler Blowdown (A or B)	D							D										
Polish Resin Cell Effluent															D			
Main Steam Crossover	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D

KEY:

C = Continuous Monitor  
D = Daily  
3/W = 3 Times per Week  
W = Weekly

(T) = Technical Specifications  
\* = Continuously Sampled as Polished Demin. Influuent  
\*\* = Conducted on Resins removed from Spent Gels

TABLE 2

Babcock Power Company  
McGinnis Nuclear Station  
Secondary Chemistry Specifications  
Normal Operation

	COMP	C. COND mmhos	O <sub>2</sub> ppb	N <sub>2</sub> H <sub>4</sub> ppb	Cl ppb	F ppb	Na ppb	Ca ppb	Fe ppb	SPFC
	pH	pphos	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ACT μCi/m <sup>3</sup>
Steam Generator R. I. (A or B)	8.5- 9.0	<2.0	<100	<150	<500	<1000	<1000	<1000	<200	<10
Main Steam (A or D)	9.2	5-12	<0.5	<3						
Boiler Pump Discharge	9.2	1-12	<2.0	<5						
Pooleh Demin Effluent	8.8-	<0.5	<5	<5	<150	<150	<10	<20	<10	
Feedwater	9.2	<0.5	<5	<5	<25	<150	<10	<20	<10	
C. Heater Drains	8.8-	<0.5	<5	<5						
G. Heater Drains	8.8-	9.2	<0.5	<5						
Condensate Storage Tank Mixture Separators	6.0- 8.8-	9.2	5-12							
Condenser Hotwell (IA1 - IC2)	8.8-	9.2	<5	<30						
Potassium Demineralizer				<5	<150					
Cell Effluent				<0.5						
Main Steam Crossover	8.8-	9.2	5-12	<0.5	<1					
Aux Boiler	8.8-	9.2								
Feedwater (A or B)	8.8-	9.2								
Aux Boiler										
Blowdown (A or B)										

(a) N<sub>2</sub>H<sub>4</sub> = to hotwell if feeding from hotwell.

(b) &lt; 250 ppb not to exceed 1 hour, Transient Condition.

(c) Administrative Limit for Chloride or Fluoride is &lt;50 ppb. If exceeded, take corrective action immediately; do not wait for conc. to reach 150 ppb.

\* - Conducted on routine removed from spent cells.

Note: It should be emphasized that these are goals we will be striving to attain.  
A better understanding of actual specific system capabilities will be obtained  
during the first year of operation.

TABLE 8  
ADMINISTRATIVE GUIDELINES FOR LIMITING OPERATION  
FEEDWATER SYSTEM

ENCLOSURE 3.B.4  
Page 2 of 3

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CHEMICAL PARAMETER	PREOPERATIONAL CLEANUP		NORMAL OPERATION (Note 2)	TRANSIENT CONDITIONS LIMITING OPERATION TO:		
	NC SYSTEM < 100 °F	NC SYSTEM > 100 °F (Note 1)		< 100 hrs	< 24 hrs	< 2 hrs
pH	8.8-9.6	8.8-9.2	8.8-9.2	8.7-8.8 9.2-9.3	8.6-8.7 9.3-9.4	8.5-8.6 9.4-9.5
Cat. Cond	< 0.5 $\mu$ mo	< 0.5 $\mu$ mo	< 0.2 $\mu$ mo	0.2-0.5 $\mu$ mo	0.5-1.0 $\mu$ mo	> 1.0 $\mu$ mo
Dissolved O <sub>2</sub> (Note 4)	< 100 ppb	< 5 ppb	< 5 ppb	5-25 ppb	25-100 ppb	> 100 ppb
SiO <sub>2</sub>	< 20 ppb	< 20 ppb	< 10 ppb	20-40 ppb	40-60 ppb	> 60 ppb
Na <sup>+</sup>	< 5 ppb	< 5 ppb	< 1.5 ppb (2.5 Note 3)	2.5-3.5 ppb	3.5-6.0 ppb	> 6.0 ppb
N <sub>2</sub> H <sub>4</sub>	1-2 ppm	75-150 ppb	3 x DO	3 x DO	3 x DO	
SS	< 10 ppb	< 10 ppb	< 10 ppb	10-25 ppb	25-50 ppb	> 50 ppb
Fe	< 10 ppb	< 10 ppb	< 10 ppb	10-25 ppb	25-50 ppb	> 50 ppb
Cu	< 5 ppb	< 5 ppb	< 3 ppb	3-5 ppb	5-10 ppb	> 10 ppb
Pb	< 1 ppb	< 1 ppb	< 1 ppb	1-2 ppb	2-3 ppb	> 3 ppb

Note 1: These parameters apply as long as the ratio of Na<sup>+</sup> concentration in the S/G Blowdown is < 2 times the feed concentration.

Note 2: These parameters apply any time the Na<sup>+</sup> concentration in the S/G Blowdown exceeds 2 times the feedwater concentration.

Note 3: If Na<sup>+</sup> concentration exceeds 1.5 ppb, all available corrective actions should be used to prevent exceeding 2.5 ppb, and restore Na<sup>+</sup> to < 1.5 ppb.

Note 4: These limits apply to the Rotwell. If Dissolved Oxygen > 5 ppb, increase N<sub>2</sub>H<sub>4</sub> concentration to 3 times the DO concentration.

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Note: It should be emphasized that these are goals we will be striving to attain.  
 A better understanding of actual specific system capabilities will be obtained during the first year of operation.

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TABLE 9  
ADMINISTRATIVE GUIDELINES FOR LIMITING OPERATION  
STEAM GENERATOR BLOWDOWN

CHEMICAL PARAMETER	FREQUENTIAL HG SYSTEM < 100 F	CLEARUP HG SYSTEM > 100 F (Note 1)	NORMAL OPERATION (Note 2)	TRANSIENT CONDITIONS LIMITING OPERATION TO:	
				< 100 hrs	< 24 hrs
pH	8.5-9.4	8.5-9.2	8.8-9.2	9.2-9.6 8.8-8.6	9.4-9.5 8.6-8.4
Cat. Cond.	< 2 $\mu$ mos	< 2 $\mu$ mos	< 2 $\mu$ mos	2-4 $\mu$ mos	4-6 $\mu$ mos
$\text{Na}^+$	< 40 ppb	< 40 ppb	< 100 ppb	100-150 ppb	150-250 ppb
$\text{Cl}^-$	< 50 ppb	< 50 ppb	< 50 ppb	50-200 ppb	200-250 ppb
$\text{F}^-$	< 50 ppb	< 50 ppb	< 50 ppb	50-200 ppb	200-250 ppb
SS	< 250 ppb	< 250 ppb	< 1000 ppb	1000-1500 ppb	1500-2500 ppb
$\text{SiO}_2$	< 250 ppb	< 250 ppb	< 1500 ppb	1600-1500 ppb	1500-2000 ppb
Fe	< 250 ppb	< 250 ppb	< 1000 ppb	1000-1500 ppb	1500-2000 ppb
Cr	< 10 ppb	< 10 ppb	< 100 ppb	100-500 ppb	500-1000 ppb

Note 1: These parameters apply as long as the ratio of  $\text{Na}^+$  concentration in the S/G Blowdown to < 2 times the feed concentration.

Note 2: These parameters apply any time the  $\text{Na}^+$  concentration in the S/G Blowdown exceeds 2 times the feedwater concentration.

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TABLE I O

FIGURE 3, A, S

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Duke Power Company  
McGallure Nuclear Station

TABLE II

**DUKE POWER COMPANY**  
**SECONDARY SYSTEM**  
**WET LAYUP SAMPLING SCHEDULE**

TABLE 12

CALIBRATION OF MCGUIRE NUCLEAR STATION'S  
PROCESS ANALYTICAL INSTRUMENTATION  
SECONDARY CHEMISTRY

pH Meters:	Calibrated daily using a minimum of two different purchased buffers, ACS Reagent Grade. Normally pH's of 7 and 9, 10, or 11 are checked.
Conductivity Meters:	Calibrated monthly by cross verifying the reading to a Balabaugh meter. The Balaburgh meter is checked using selected potassium chloride solutions to verify probe response, and checked electronically by our instrument group.
Dissolved Oxygen Analyzers:	Calibrated fully weekly (membrane changed, electrolyte changed, and air calibration), checked for accuracy daily using the indigo carmine method (in addition, chemets are sometimes used).
Hydrazine Analyzers:	Currently calibrated daily by cross verifying the reading using a manual determination. This may be relaxed to weekly in the future if results indicate this is feasible. Also see Note 1.
Silica Analyzer:	Standardized daily using prepared standards of known composition. ACS Reagent Grade chemicals are used.
Sodium Analyzers:	Standardized daily using standards prepared and verified using a flameless Atomic Absorption technique. See Note 1.

If, in the future, the current calibration frequencies of these instruments are found to be excessive, the frequency may be changed. However, it is highly doubtful of the frequency would be less than weekly (except conductivity as these units are extremely stable). No instrument will hold calibration indefinitely and we feel that accurate data is an absolute necessity, regardless of how many hours are required to calibrate these instruments.

NOTE 1: The Q Sum technique is used to insure accuracy of these manual determinations. This technique requires that at least 20 duplicate sets of samples be analyzed under optimum conditions. This data is then used to determine the standard deviation for both accuracy and precision of the method with a 95% confidence level. Graphs are then prepared and each day the method is used, the deviation of both accuracy and precision are plotted. If a positive slope change occurs outside of the 95% confidence limits, sample data is rejected and all samples are reanalyzed. In addition, at least one standard must be analyzed with each subsequent sample that day and that one standard must fall within the standard deviation of the method. This insures that not only all reagents used react as required, but also verifies instrumentation operation and the analytical technique of the technician.