INSTITUTO DE ENERGIA ATÔMICA

I. E. A.

CIDADE UNIVERSITARIA

SIE.0549/07/79.

FEG/MCB.

Cx Postal: 11.049 - Pinheiros End. Telegráfico: IEATÓMICA Telefone (PABX): 211-6011 Telex: (011) 23592 - IENA-BR CGC: 43778448/0001 SÃO PAULO

São Paulo, July 05, 1979.

UNITED STATES NUCLEAR REGULATORY COMMISSION Gerald G. Oplinger, Assistant Director. Export/Import and International Safeguars Office of International Programs.

Washington, D.C. 20555

XSNM01475

Dear Mr. Oplinger

We were very much surprised with the terms of your letter of May 22, adressed to Globe Shipping Co- N.Y. New York.

Your point of view regarding our order to import some NBS standards is for us too difficult to understand.

Our purpose is to get uranium NBS standards of varying isotopic concentrations and not enriched uranium in the form of NBS standards. There is quite a big difference in these two statements.

Anyone familiar with uranium thermionic -/ mass spectrometric analysis knows about the need of reference samples. Just for your knowledge we are sending, im annexe, a couple of papers (all from-U.S.A) dealing with this matter. In these papers you could realise the -/ importance of these materials including uranium over 20% of enrichment.

application are for checking discriminator factors, like is made in any / laboratory over the world. We do not need this material for manufacture / nuclear fuels. In this field our program is quite modest, all of the -/ activities are of laboratory scale and for do that we have imported the / necessary material inclusive from USA. Always under safeguards. It looks / for us rather impossible to make fuels with five gramms of uranium. As / regard an other point of letter you should know our country signed an / agreement with Germany which cover all for our activities in the field of uranium enrichment.

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cont.



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(cont.)

We hope to have clarified your questions and doubts but if you still have further questions regarding this matter / do not hesitate to write me.

Sincerely

HERNANI AUGUSTO LOPES DE AMORIM

Superintendent

CC. Mrs. Alvine R. Spies. Globe Shipping Co. Written: February 1971 Distributed: Merch 1971 APE

LA 4622 UC-80, REACTOR TECHNOLOGY TID-4500

# LOS ALAMOS SCIENTIFIC LABORATORY of the University of California

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Methods of Chemical Analysis for FBR Uranium-Plutonium Oxide Fuel and Source Materials

Edited by

James E. Rein George M. Matlack Glenn R. Waterbury Robert T. Phelps Charles F. Metz

This work was aponeored by the Fuels and Materials Branch of the Division of Reactor Development and Tachnology of the U. S. Atomic Energy Commission.

DETRIBUTION OF THIS DOCL YEST IS INLIGHTE

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POOR I

- Set the amplifier on the 10-mV scale said apply 10 mV from the potentiomstar.
   herord the deflection.
- Repeat step 2 for each ampailier sensilivity setting and the corresponding potentiometer voltage.
- Determine exact scale factors from the ratios of applied voltage/recorder dedection for each sensitivity setting. Use these factors in subsequent determinations.
- b. Recorder linearity
  - Connect a precision potentiometer as is step a-1 above.
  - Set the amplifier on the 1-V scale and apply voltage in increments of 0.1 V up to 1 V. Record the recorder deflection for each increment.
  - Find the cause for and correct any nonlinearity greater than 0.25" of half scale.
- e. Orid resistor linearity and ratio determination. If a standard current source is evaluable, the test described in TID-7029 is advisable.

Calibration of Complete System

Calibrate the complete mass spectrometer at startup, at least every three morehs, and following any siteration, as follows:

- a. Determine the mass discrimination bias both for the electron multiplier and the Faraday cup by measuring the 235 U/238 U ratio on at least three exparate filament loadings of the NRS U-500 standard for each detector. Average the results and calculate the respective bias correction factors by
- [2] R. J. Joses, Ed., Nethod 2.500, Selected Meaeurement Methods for Urantum and Platonium in the Nuclear Fuel Cycle, "USA EC Report TID-1029 (1983).



where B - muse discrimination bias factor per arms in the U-Ps mass range,

 $\overline{R}_a$  = average of observed atom ratio  $^{235}U/^{236}U$  ,  $\overline{R}_a$  = NES stated atom ratio  $^{235}U/^{238}U$  = 1.0003.

Apply these bias corrections to subsequent determinations.

> b. Establish the linear range of both detection systems by measuring the <sup>234</sup>(1),<sup>235</sup>U ratio on separate filament loadings of the NBS U-\$30 standard for each system over a wide range of ion currents at increments corresponding to the amplifier scale factors. To sid interpretation, plot the observed ratio vs the <sup>235</sup>U lun current on semilog paper. Asalyze samples only within the linear range.

Daily Verification of Mass Spectrometer Stabulky

Verify the stability of each mass spectrometer used in any s-hr working shift with the NRS U-010 uranium isotopic standard when uranium samples are analyzed during that shift. Malatain a control that for each instrument for the ratio of  $^{2.5}$ U,  $^{2.5}$ U. When the value of this ratio changes at the .05 significance level, do not analyze samples until the cause is corrected. Possible sources of instability are a deterioration of the electron multiplier, low gain is the electrometer amplifier, electrical or mechanical maifunctions is the recorder, and landequate regulation is the filament power supply.

### Sample Accignia

 Dissolve i mg of the wranium dioxide rample is 8,8 mi of 5M HNO<sub>3</sub> is a 10-mi beaker with the sid of moderate heating, dilute to 5 ml with water, and min

The sample weight and reagent rolemers need not be quantitative because only inotingle abundances are to be measured. If a whole insulator pell of has been dissolved, dilute as aliquot of the nobulion to a uranium concentration of 0.2  $\mu g/mi$  with 1 M HNO  $_{\chi}$ .

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TID 7029 (2 MEN) MASTER

This report was prepared as an account of work sponsored by the United States Government, Newhole United States and the United States Atomic Energy Commission, nor any of their employees, nor any other contractors, or their employees makes any prepared to their contractors, or their employees makes any prepared to their contractors, property makes any prepared to the state of their contractors, property or the process of their contractors, property or the process dischard, no represents that its use their contractors and their contractors of their c

Selected Measurement Methods for Plutonium and Uranium in the Nuclear Fuel Cycle

Second Edition

Edited by
CLEMENT J. RODDEN

Office of Information Services
U.S. Atomic Energy Commission
1972

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POOR ORIGINAL used. Also, for instruments with expanded-scale strip-chart recording, see Ref. 3.

### D. Calibration

1. Mass Discrimination The mass-discrimination bias factor for both uranium and piutonium is determined by analyzing a suitable NBS isotopic reference standard. The general procedure is to obtain several sets of isotopic ratios from several separate loadings of the reference sample and to calculate an average mass-discrimination bias factor, B,

$$B = \frac{1}{C} \left[ \left( \frac{\overline{R}}{R_4} \right) - 1 \right]$$

where B = mass-discrimination bias factor

 $C = \Delta M/M = (238 - 230)/238$  for U'

h = known value of 135 U/138 U for the reference sample

i. = grand-average measured value of 235 U/238 U for all sets of data

2. Linearity The linearity of the ion-current measuring system, including electron multiplier, amplifier, and data-recording system, can be checked using the uranium isotopic reference standard NBS U-930 and measuring the <sup>234</sup>U/<sup>235</sup>U ratio over the range of ion intensities normally used.

## E. Sample Preparation

1. Uranium Prepare uranium samples as a solution of uranyl nitrate of approximately 0.1 to 1 mg of uranium per milliliter in 0.5N to 2N HNO<sub>3</sub>. The sample may contain other elements as impurities up to 10 times the amount of uranium, except for alkali elements, which should be not greater than 10% of the uranium. For purification of samples see Methods 1.1.2.1 and 1.2.2.1.

2. Plutonium Prepare plutonium samples as a solution of approximately 0.01 to 0.1 mg of plutonium per milliliter in 0.5N HNO<sub>3</sub>. The sample may contain other elements, including uranium, as impurities up to 100 to 1000 times the plutonium concentration, except for alkali elements, which should be not greater than 100% of the plutonium. For purification of samples, see Method 2.1.2.1.

# F. Procedures

1. Filament Preparation In preparing the filaments of a multiplefilament assembly, make certain that the filaments are not con-

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