

EXTENDED DRY STORAGE AND TRANSPORTATION: MODEL FOR EVALUATING VACUUM DRYING ADEQUACY

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DISCLAIMER

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Introduction

• Potential excessive residual moisture following incomplete drying of spent nuclear fuel (SNF)

• A time-dependent integration model, to assess potential radiolysis, producing oxygen and hydrogen, and potential effects on the integrity of SNF, cladding, internals, and flammability

• Interim progress made for the process of model abstraction, model integration, and uncertainty assessments. Quantitative results will be presented in the future.



Outline

- Quantity of Water Remaining after Drying: water reacts with SNF, cladding and internals
- Temperature: affects reactions
- Radiolysis: produces oxygen and hydrogen; affects reactions; poses flammability potential
- Relative Humidity (RH): affects reaction
- Oxidation and Hydration of SNF: affect SNF integrity
- Cladding Oxidation: affects cladding integrity



Outline (continued)

- Cladding Splitting Strain Estimate due to SNF Swelling: affects integrity of SNF and cladding
- Flammability
- Hydrogen-Absorption-Induced Damage
- Aqueous Corrosion: affects reactions
- Integration and Benchmarking



- Quantity of Water Remaining after Drying
 - Unbound residual liquid water, unbound water vapor, and water chemically bound to hydroxide and hydrate species
 - Properly executed vacuum drying procedure result in:
 1 to 5 moles (0.02 to 0.1 L [0.7 to 3.5 oz]) water inventory
 - up to 55 moles (~ 1 L [35 oz]) are conservatively assumed
- Temperature
 - Spatially distributed and decreases over time



Temperature (continued)

• Temperature distribution in the SNF basket assembly (left). The scales for temperature are in K (F = 1.8 K – 459.4); SNF and cladding in the five zones calculated using mean values of low-end SNF and cladding initial temperatures (right)





Radiolysis and Relative Humidity (RH)

• Radiolysis: Global Approximation for Various Intermediate Species

 $2H_2O \leftrightarrow 2H_2 + O_2$

- Decomposition rate of water by radiolysis: exponential time function
- Recombination additional steps for molecular collision needed, compared with decomposition

- **Relative Humidity (RH)**
- The radiolysis removes water molecules by dissociating them into oxygen and hydrogen
- RH is modified by radiolysis



Oxidation and Hydration of SNF

• Upon cladding breach, the UO₂ could be oxidized to form:

- U₄O₉, U₃O₇, and U₃O₈ in dry air (less than 40 percent RH)

- Hydrated uranium oxides, such as schoepite (UO₃•xH₂O, x = 0.5 to 2), in humid air (greater than 40 percent RH) or in an aqueous environment



Grain boundary oxidation of AM-105 spent nuclear SNF to U_4O_9 . Optical ceramographs, as polished: (a) 95 hrs, Bulk O/M = 2.05; (b) 420 hrs, bulk O/M=2.17; (c) 775.5 hrs, bulk O/M = 2.24; and (d) 1,677 hrs, bulk O/M = 2.31 (Einziger, et al., 1992)

- Chemical Arrhenius-type Kinetics
- Actual Rates are Controlled by Radiolysis Kinetics or Chemical Kinetics



Cladding Oxidation, and Cladding Splitting: Strain Estimate due to SNF Swelling

- Cladding Oxidation
 - $\mathbf{Zr} + \mathbf{O}_2 = \mathbf{ZrO}_2$ for dry air and
 - Zr + 2H₂O = ZrO₂ + 2H₂ for water or humid air (e.g., steam)
 - Model by Hillner, et al (1994) and 8 other similar kinetics
 - Large surface area of cladding
 - Actual Rates Controlled by Radiolysis Kinetics or Chemical Kinetics

- Cladding Splitting: Strain Estimate due to SNF Swelling
 - Oxidation of UO_2 to U_3O_8 can generate stress on cladding as U_3O_8 swells (36 percent when there is 100 percent conversion to U_3O_8).
 - Various threshold values of strain are assessed to determine the cladding splitting condition.
 - Observed threshold values:
 6.5% strain (100% conversion);
 5.1% volume expansion correlation (25% conversion);
 2% strain (50% conversion)



Flammability and Hydrogen-Absorption-Induced Damage

- Flammability
 - The cladding will not absorb radiolysis-generated hydrogen.
 - The flammability criterion requires the volume fraction of any flammable gas to be more than 5 percent, with oxygen and ignition.

- Hydrogen-Absorption-Induced Damage
 - Cladding: Zirconium oxide is expected to limit the absorption of molecular hydrogen:

Total possible amount of hydrogen from 55 mole water is assessed.

- Canister Internals: the mechanical properties may or may not be degraded with hydrogen concentration exceeding a threshold value.



- Aqueous Corrosion
 - Aqueous corrosion could occur in vapor when RH is greater than a threshold value: At above the threshold RH
 - (i) SCC of the canister's internal structural components could occur in carbon and stainless steels; no relevant data available
 - (ii) Shadow corrosion (a form of galvanic corrosion) between cladding and spacer-grid material could occur.

Aqueous Corrosion, and Integration and Benchmarking

- Integration and Benchmarking
- In each time step, all reactions and gas generations are assessed
- Bench marking with long-term demonstration data (e.g., hydrogen amount)



Summary

Program is under development that evaluates:

- Time-dependent integrated models for temperature, relative humidity, and radiolysis kinetics to decompose water into oxygen and hydrogen.
- Various abstracted models for temperature, radiolysis, relative humidity, and chemical reactions of SNF, cladding and internals.
- Time integrated model: in each time step, all reactions and gas generations are assessed.
- Assess degradation of SNF, cladding and internals.
- Possibility of flammability conditions.