

CENTER FOR NUCLEAR WASTE REGULATORY ANALYSES

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Issued to Kuang-Tsan Kenneth Chiang K. T. Chiang

KTC

On October 1, 2003

Department CNWRA CC 20

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WALTER J. MACHOWSKI Walter Machowski WJM

Returned April 14, 2009

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**Initial Scientific Notebook Entry for Corrosion Resistant Material
Potentiostatic and Potentiodynamic Tests.**

Title: Potentiostatic tests, polarization tests, crevice repassivation tests, passive current density tests, critical pitting temperature tests, critical repassivation temperature tests.

Test Performed by: Kuang-Tsan Kenneth Chiang, Darrell Dunn, Brian K. Deby

Objective: Study the effect of Nitrate [NO₃⁻] to Chloride [Cl⁻] concentration ratio on corrosion resistance of candidate materials.

Equipment: EG&G Versastat Serial Number 20104. EG&G Model 352 corrosion software. NEC 586 computer. Keithley Electrometer Model 614 SN 55538 or equivalent. ASTM G-5 Polarization Cell. Large 2 L glass cells with Teflon tops. Electrochemical Impedance Spectroscopy system including Solartron 1260 FRA and Solartron 1287 Potentiostat. ESC 440 multichannel potentiostats with National Instruments Labview data acquisition software or Strawberry Tree data acquisition software.

Materials: Alloy C-22

Specimen Specifications: Cylindrical CPP specimens 1.195" x .250" and Crevice repassivation specimens with Teflon crevice washers attached to surface.

Measurement Parameters: Current and Potential as described in TOP-008. Temperature of solution $\pm 2^{\circ}\text{C}$.

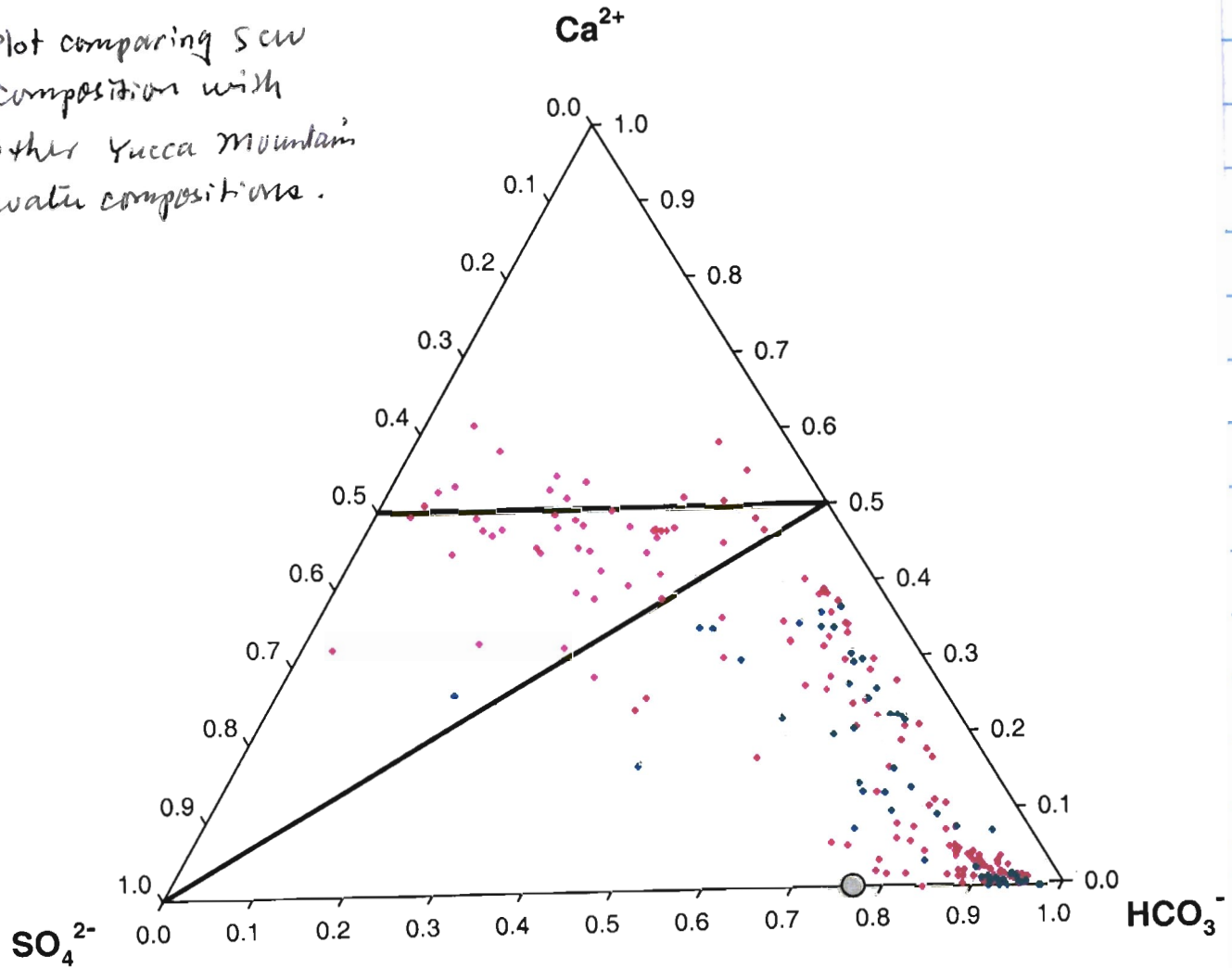
Required Level of Accuracy: Potentials $\pm 5\text{mV}$. Current less than 0.1 microamp.

Uncertainty and Source of Error: Current density calculated as current divided by sample area. Actual current density of corroding areas is not determined. Resolution limit of data acquisition systems may limit accuracy of passive current density measurements.

K.T. Chiang 10/2/03

Yucca Mountain
pore, perched, & ground waters

Plot comparing SCW
composition with
other Yucca Mountain
water compositions.



Pink - Yucca Mountain unsaturated zone porewater compositions
Green - Yucca Mountain perched and saturated zone water compositions
Grey circle - Simulated Concentrated Water (DOE corrosion experiments)

Source: Robert Pabalan

Reference UCRL-ID-132286

Formulation and Make-up of Simulated Concentrated Water,
High Ionic Content Aqueous Solution

Greg Odowski, April 4, 1997

K.T. Clancy
11-2-03

1.0 PURPOSE

This procedure describes the formulation and make-up of Simulated Concentrated Water (SCW), a high-ionic-content water to be used for Activity E-20-50 "Long-Term Corrosion Studies." This water has an ionic content which is nominally a factor of a thousand higher than that of "representative" waters at or near Yucca Mountain. "Representative" waters were chosen as J-13 well water [Harrar, 1990] and "perched" water at Yucca Mountain [Glassley, 1996] (see Table 1). J-13 well water is obtained from ground water that is in contact with the Topopah Spring tuff, which is the repository horizon rock. The "perched" water is located in the Topopah Spring tuff, but below the repository horizon and above the water table. A nominal thousand times higher ionic content was chosen to simulate the water that would result from the wetting of salts which have been previously deposited on a container surface.

The expected composition of the SCW is given in Table 1. It is anticipated that the actual composition of test solutions will be within $\pm 20\%$ of these values. The changes in the corrosive properties of the test solutions will be acceptable within these values. In addition similar type materials are tested in the same test vessel, so minor vessel to vessel variation of solution composition is of limited significance.

Both of the "representative" waters have similar corrosive characteristics. The solution pH's and the concentrations of the aggressive anions (Cl^- , F^- , and SO_4^{2-}) are essentially equivalent from a corrosion stand point.

This aqueous solution is one of the four aqueous test solutions to be used in the activity. The other aqueous solutions included a Simulated Dilute Water (SDW), a simulated acidic concentrated water (SAW), and a simulated basic concentrated water (SBW).

This TIP documents the chemical reagents, reactant air, and the procedures used to make-up the aqueous solution for Activity E-20-50. More than 12,000 liters (3,170 gallons) of Simulated Concentrated Water solution are required for the test vessels for implementation of the full test matrix of the activity plan.

2.0 SCOPE

This procedure applies to the Simulated Concentrated Water solution, one of the aqueous solutions that are to be used in the test vessels for Activity E-20-50 "Long-Term Corrosion Studies."

3.0 RESPONSIBILITIES

The Principal Investigator (PI) or designee is responsible for:

- the conduct of the activities and methods described in this procedure, and
- maintaining laboratory scientific notebooks.

The Task Area Leader (TAL) is responsible for:

- ensuring that the requirements of this procedure are implemented,
- ensuring that personnel conducting the work are qualified and are trained to this procedure,
- verifying that this procedure meets the objectives of the Scientific Investigation Plan (SIP) "Metal Barrier Selection and Testing" (SIP-CM-01, Rev.3, WBS # 1.2.2.5.1) and Activity E-20-50 "Long-Term Corrosion Studies", and
- ensuring approval of this procedure.

The YMP Quality Assurance Manager (QA Manager) is responsible for:

- monitoring the work to assure proper implementation of this procedure, and
- assuring its continued effectiveness.

Simulated Concentrated Water (SCW)

Simulated Dilute Water (SDW)

K. J. Clancy
11-2-03

SCW Composition

4.0 COMPOSITION OF SCW AND REACTANT AIR

4.1 Aqueous Solution Composition

The Simulated Concentrated Water (SCW) has a ionic composition that is nominally a factor of a thousand higher than that of "representative" water of Yucca Mountain. "Representative" waters were chosen J-13 well water [Harrar, 1990] and "perched" water at Yucca Mountain [Glassley, 1996]. J-13 well water is obtained from ground water that is in contact with the Topopah Spring tuff, which is the repository horizon rock. The "perched" water is located in the Topopah Spring tuff, but below the repository horizon and above the water table. The thousand times higher ionic content was chosen to simulate the water that may result from wetting of salts and minerals that have been deposited on the container surfaces.

The composition of J-13 well water and the "perched" water are given in Table 1. Only ions with concentration greater than 0.5 ppm are included in this table. Minor constituents have been detected in J-13 well water; these include Li, B, Al, Mn, Fe, Sr, and PO_4^{3-} ions. These constituents have been reported in the 10-100 $\mu\text{g/liter}$ concentration. The most consistently determined minor constituents are Li and B at mean (several studies) concentrations of 48 and 134 $\mu\text{g/liter}$, respectively. The minor constituents are not explicitly included in the SCW. However, the reagent chemicals have some impurities, which may include the above noted impurities. The minor constituents at the reported concentrations are not expected to significantly effect the corrosion of the test specimens.

It is worth noting the differences in the "representative" waters. In terms of the calcium and bicarbonate concentrations, the "perched" water is higher in both of these constituents, and has a higher pH; these both are probably due to contact with calcium carbonate (CaCO_3) minerals. The concentrations of Na^+ , K^+ , F^- , and NO_3^- are slightly higher in J-13 well water.

Table 1. Compositions of "representative" Yucca Mountain waters J-13 well water and "perched" water, and the estimated composition of the simulated concentrated water.

Constituent	J-13 (mg/l) (mg/l)	"Perched" (mg/l)	Simulated Concentrated Water (estimated)
Na	45.80	36	40,900
Si	28.5	37	27 (60C); 49 (90C)
Ca	13.0	25	< 1
K	5.04	1.7	3,400
Mg	2.01	2.2	< 1
F^-	2.18	0.7	1,400
Cl^-	7.14	6.3	6,700
NO_3^-	8.78	4	6,400
SO_4^{2-}	18.4	15	16,700
HCO_3^-	128.9	147	70,000
CaCO_3	—	—	47,500 (precipitate)
MgCO_3	—	—	7,300 (precipitate)
pH	7.41	8.1	

K.J. Ching
11-02-03

The following paragraphs explain the reasoning used to arrive at the composition and formulation of the SCW. The silica content is based on solubility of α - cristobalite which is believed to be the dominant soluble silica phase of Yucca Mountain rock at 60 and 90°C [Wolery, 1983; Knauss, 1987]. Silica may be added to the solution by dissolution of sodium silicate. The salts, in general, are concentrated by a factor of a thousand over an "average" of the "representative" Yucca Mountain waters. The exceptions are calcium (Ca^{2+}), magnesium (Mg^{2+}), and bicarbonate (HCO_3^-), which are all at lower concentration.

Previous studies have indicated that extensive concentrating of water with relative ionic concentrations like those of the "representative" water (high bicarbonate) results in the precipitation of calcium and magnesium carbonates and silica-base minerals [Drever, 1982]. Calcite (CaCO_3) will precipitate first, with some magnesite (MgCO_3) co-precipitating in the calcite. The remaining magnesium will precipitate in silica-based minerals.

Simulations of concentrating the "representative" waters were run using the computer program "Geochemist's Workbench, Release 2.2" [Bethke, 1994]. The results generated were in agreement with the qualitative predictions based on the general solution composition. All of the calcium and magnesium was effectively precipitated during concentrating of the solution; a few ppm of each remained in solution.

The estimated composition of the SCW is given in Table 1. It is expected that the actual composition of test solutions will be within $\pm 20\%$ of these values. The changes in the corrosive properties of the test solutions within these values will not be significant. In addition similar type materials are tested in the same test vessel, so minor vessel-to-vessel variation of solution composition is of limited significance.

4.2 Reactant Air

Reactant air is compressed building air which has been purified to remove hydrocarbons and water. Air will be purified by flowing through a Whatman Zero Air Generator (see Section 6.0). Nominal flow rates through each test vessel will be 200 ml/min. Air will exit through a condenser to remove water; this greatly reduces the amount of water loss from the test vessels.

Reactant air serves two purposes: 1) it keeps the oxygen content of the vessels constant, and 2) the slightly pressurized test vessel will keep the potentially contaminated room air out of the test vessels.

5.0 REAGENTS AND FORMULATION

5.1 Reagent Chemicals

In order to obtain the solution composition given in Table 1, various combinations of chemicals can be used. A spreadsheet has been developed which calculates the composition of a solution based on the added chemicals. **Copies of typical outputs of the spreadsheets are shown in Appendix A for 60 and 90°C solutions; the amount of silica changes with temperature.** Many of the chemicals listed in the spreadsheet are not used in this particular example. The inclusion of numerous chemicals in the spreadsheet allows the user the freedom to choose the needed chemicals based on availability, cost, and personal preference.

The algorithm to arrive at reagent concentrations was a trial and error method. The quantities of reagents required was estimated, and the spreadsheet calculated the total ionic content of the theoretical solution. Iteration was continued until an acceptable match was achieved.

A few guidelines were used in choosing the reagents. The choices for bicarbonate ions were NaHCO_3 or KHCO_3 , since these are the common commercial source of bicarbonate. The use of potentially hazardous materials such as HF, MgF_2 , and CaF_2 was avoided. The more soluble salts (minerals) were chosen, for example, magnesium sulfate was chosen over the less soluble carbonate and nitrate salts.

K. J. Chuaif
11/02-03

Also solution silica will be obtained by the addition of sodium silicate. (Calculations showed that dissolution of solid silica phases would take extended periods of time (>1000 days) in order for sufficient amounts of silica to dissolve.) Note using sodium silicate will result in slightly elevated sodium concentrations.

Using sodium silicate will result in the formation of hydroxyls equal to the number of moles of sodium atoms added. **In order to neutralize the hydroxyls, an equal number of moles of acid (hydrochloric, nitric, or sulfuric) will be added.**

Since a large percentage of both the calcium and magnesium will form carbonate precipitates, it was not necessary to add soluble salts of these ions to the level of the concentrating. However excess of these ions will be added such that precipitates of calcite and magnesite will form.

A word of caution in using the spreadsheet: the calculations assume that the chemicals dissolve completely and may therefore over estimate the composition of some species. The user must therefore be aware of potential solubility problems. A listing of the solubilities of various chemical is shown in Appendix B.

A typical example of chemicals used to make-up of the aqueous solution are listed in Table 2 along with the quantities required per 1000 l of solution.

Table 2. An example of the reagents and quantities required per 1000 liters of simulated aqueous solution.

Reagent	Quantity @ 60°C (gms / 1000 liters)	Quantity @ 90°C (gms / 1000 liters)
NaHCO ₃	128,450.0	128,297.0
NaF	3,182.6	3,182.6
Na ₂ SO ₄	12,236.4	12,254.5
Na ₂ SiO ₃ •5H ₂ O	204.0	370.0
MgSO ₄ •7H ₂ O	21,392.0	21,392.0
CaCl ₂ •2H ₂ O	7,598.0	7,598.0
Ca(NO ₃) ₂ •4H ₂ O	12,168.5	12,168.5
HCl	3.73	70.1
KCl	6,417.8	6,282.0
KHCO ₃	10.4	192.5
CaCO ₃	37,117.3	37,117.3
H ₂ SO ₄	89.3	76.79

K. J. Ching

11-2-03

The chemicals and the quantities used in making up the test solutions will be listed in the Scientific Notebook or electronic media.

5.2 Purified Water

The make up of the test solutions requires large quantities of low ionic content water is required. The use of LLNL de-ionized water is acceptable. This water has an ionic content typically less than 1 ppm. This is less than 0.001% of the ionic content due to the added chemicals. The source of the water used in testing will be recorded in the scientific notebook.

5.3 Reactant Gas

The reactant air will be purified before entering the test vessels.

6.0 EQUIPMENT

A balance that can measure to 0.1 grams is acceptable for make-up of the test solutions. An acceptable balance is:

Mettler Balance Model # AT200
Serial Number 1114463500

Mettler Balance Model # PC16
Serial Number A51361

An air purifier for cleaning the building compressed air is required. The following unit or equivalent is acceptable:

Whatmann Type 76-818NA Zero Air Generator
Unit Serial Number 768180065B
Tower Module Serial Number 76811-10116B

This air purifier removes hydrocarbon to 0.1 ppm.

7.0 PROCEDURE

The following procedure will be followed in making-up of the Simulated Concentrated Water solutions for the test vessels:

- 1) Purified water is emplaced in the cleaned vessel; the liquid level is slightly less than the required depth for testing. (Need to account for rise in water level due to the specimens and racks, and the density decrease due to raising the water temperature to the test temperature.)
- 2) The amount of purified water added to the test vessels is estimated.
- 3) The required amounts of reagent chemicals is determined and entered in the scientific notebook or electronic media.
- 4) The purified water is heated to a nominal temperature of 40°C. This will accelerate reactions that occur in solution.
- 5) The water will be stirred. The stirrer mounted on the vessel is sufficient.

K. J. Chung

11-2-03

- 6) **Add chemicals to water. No particular order is required for chemical additions except that sulfuric acid will be the last chemical added to the test vessel.**
- 7) **Concentrated sulfuric acid shall be diluted 500-1000 times the required volume using deionized water and then added to the test vessel.**
- 8) **The vessel is sealed and brought to testing temperature for at least 24 hours.**
- 9) **The specimen racks are inserted into test vessel.**
- 10) **A sample of the test solution is withdrawn for analysis approximately a day after the level of water reaches the normal operation set point.**

Note: The liquid level in the test vessels will self-adjust to the required level. If the liquid level is low, the liquid level control system will add purified water. If the liquid level is high, water removal by the air purge will occur; this may be slow but it will occur. It is preferred to add water rather than to remove water, since the control system shuts down the heaters when the liquid level is above a certain height.

8.0 QA RECORDS

Any data that is pertinent to this TIP shall be entered into the Scientific Notebook or electronic media for Activity E-20-50. This shall include, but is not be limited to the chemical used lot # manufacturer supplied analysis, and actual reagent chemical amounts used for make-up.

9.0 REFERENCES

C.M. Bethke, The Geochemist's Workbench, Version 2.0; A Users Guide to Rxn, Tact, React, and Gtplot, Hydrogeology Program, University of Illinois, 1994.

J.I. Drever, The Geochemistry of Natural Waters, Prentice-Hall, Inc., Englewood Cliffs, NJ, 1982.

W. Glassley, private communication, 1996.

J.E. Harrar, J.F. Carley, W.F. Isherwood, and E. Raber, "Report of the Committee to review the Use of J-13 Well Water in Nevada Nuclear Waste Storage Investigations," Lawrence Livermore National Laboratory report UCID-21867, Livermore California, January 1990.

K.G. Knauss, W.J. Beiriger, D.W. Peifer, "Hydrothermal Interaction of Sodic Wafers of Topopah Spring Tuff with J-13 Water at 90 and 150°C Using Dickson-Type, Gold-Bag Rocking Autoclaves: Long-Term Experiments," Lawrence Livermore National Laboratory Report UCRL-53722, May 1987.

T.J. Wolery, Memo GCC-83-3/1773w, "Summary of Silica Solubility Data for Acid-to-Neutral pH Conditions," 16 Nov. 1983.

K. J. Ching 11-02-03

Appendix B. Solubilities in Water

Compound	Formula		Mol Wt.	Solubility gms/100cc	Solubility mg / l	T(°C)	Solubility gms/100cc	Solubility mg/l	T(°C)
Calcium sulfate	CaSO4	nat anhydrite	136.14	0.2090	2090	30	0.1619	1619	100
Calcium sulfate dihydrate	CaSO4•H2O	nat gypsum	172.17	0.2410	2410		0.222	2220	100
Calcium nitrate	Ca(NO3)2		164.09	121.2000	1E+06	18	376	3760000	100
Calcium Chloride	CaCl2		110.99	74.5000	745000	20	159	1590000	100
Calcium Fluoride	CaF2	nat flouride	78.08	0.0016	16	18	0.0017	17	26
Calcium Carbonate	CaCO3	calcite	100.09	0.0014	14	25	0.0018	18	75
Calcium hydroxide	Ca(OH)2		74.09	0.1850	1850	0	0.077	770	100
Sodium sulfate	Na2SO4	nat thenardite	142.04	4.7600	47600	0	42.7	427000	100
Sodium nitrate	NaNO3	soda niter	84.99	92.1000	921000	25	180	1800000	100
Sodium Chloride	NaCl	halite	58.44	35.7000	357000	0	39.12	391200	100
Sodium Fluoride	NaF	nat villiumite	41.99	4.2200	42200	18		0	
Sodium carbonate	Na2CO3		105.99	7.1000	71000	0	45.5	455000	100
Sodium Bicarbonate	NaHCO3		84	6.9000	69000	0	16.4	164000	60
Sodium hydroxide	NaOH		40	42.0000	420000	0	347	3470000	100
Sodium Silicate	Na2SiO3	metasilicate	122.06	soluble		-	soluble; dissolves		-
Magnesium sulfate	MgSO4		120.37	26.0000	260000	0	73.8	738000	100
Magnesium nitrate	Mg(NO3)2•6H2O		256.41	124.0000	-1E+06		vs		
Magnesium chloride	MgCl2		95.22	54.2500	542500	20	72.7	727000	100
Magnesium fluoride	MgF2	nat sellaite	62.31	0.0076	76	18	i		
Magnesium carbonate	MgCO3	nat magnesite	84.32	0.0106	106			0	
Magnesium carbonate trihydrate	3MgCO3•Mg(OH)	nat hydromag	365.34	0.0400	400		0.011	110	
Magnesium carbonate basic	MgCO3•3H2O	nat nesqueho	138.37	0.179	1790	16	d		
Magnesium hydroxide	Mg(OH)2	nat brucite	58.33	0.0009	9	18	0.004	40	100
Potassium sulfate	K2SO4	nat arcanite	174.27	12	120000	25	24.1	241000	100
Potassium sulfate, hydrogen	KHSO4	nat mercallite	136.17	36.3	363000	0	121.6	1216000	100
Potassium nitrate	KNO3	saltpeter	101.11	13.3	133000	0	247	2470000	100
Potassium chloride	KCl	nat sylvite	74.56	23.8	238000	20	56.7	567000	100
Potassium fluoride	KF		58.1	92.3	923000	18	vs		
Potassium carbonate	K2CO3		138.21	112	1E+06	20	156	1560000	100
Potassium Carbonate, hydrogen	KHCO3		100.12	22.4	224000		60	600000	60
Potassium Hydroxide	KOH		56.11	107	1E+06	15	178	1780000	100

Source: CRC Handbook of Chemistry and Physics, R.C. Weast, M.J. Astle, eds., CRC Press, Inc., Boca Raton, FL, 1981.

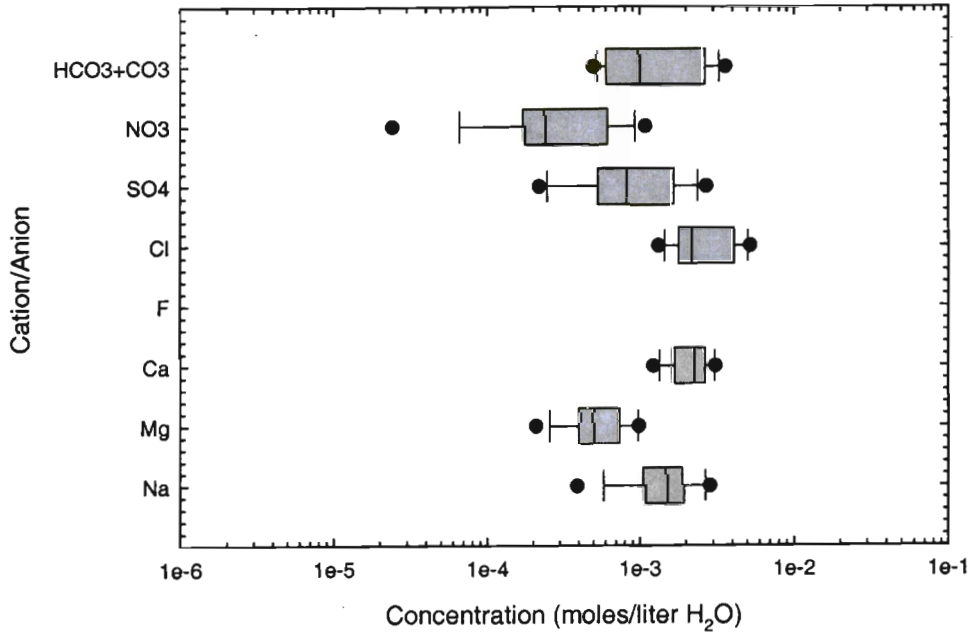
Page 1 of 1

Appendix B, TIP-CM-07,
Rev. 0, CN TIP-CM-07-0-1

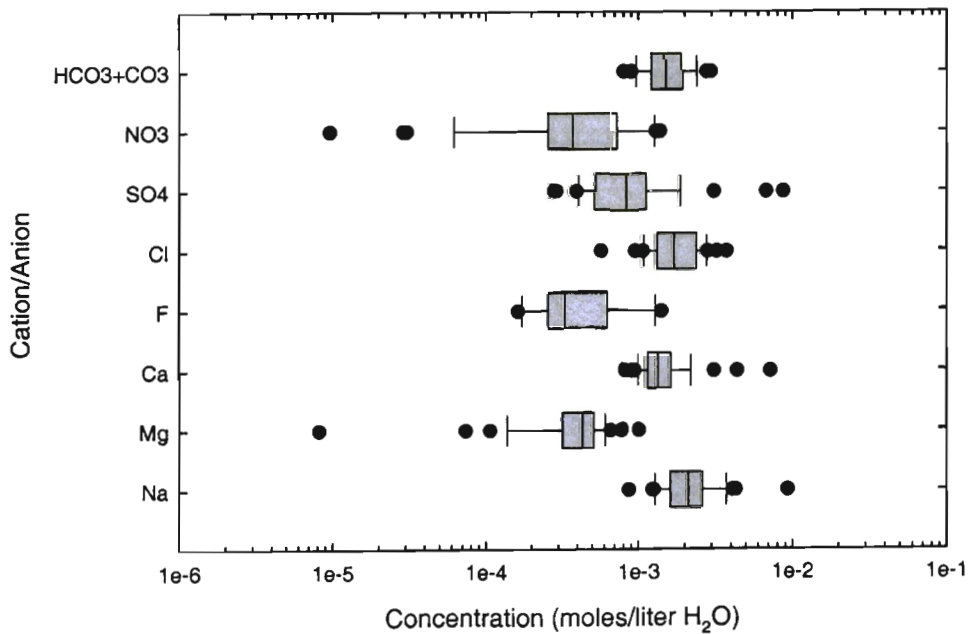
K.T. Ching 11-02-03

Discussions: Darrel Dunn, Bobby Pabalon, Ken Chiang Tue 10/28/03 1pm

Yucca Mountain Ca-Cl-Type Unsaturated Zone Waters
(Yang et al., 1996, 1998, 2003)



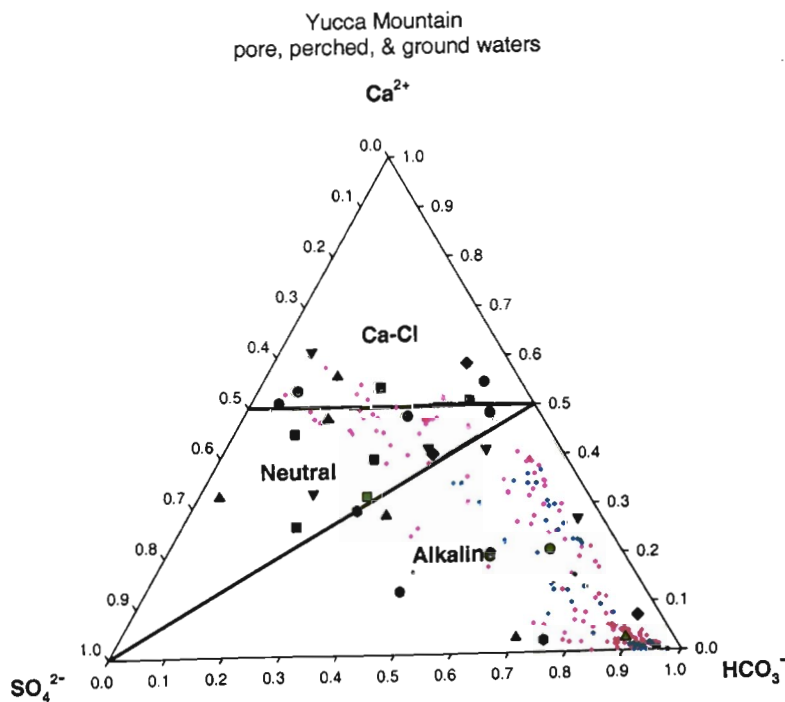
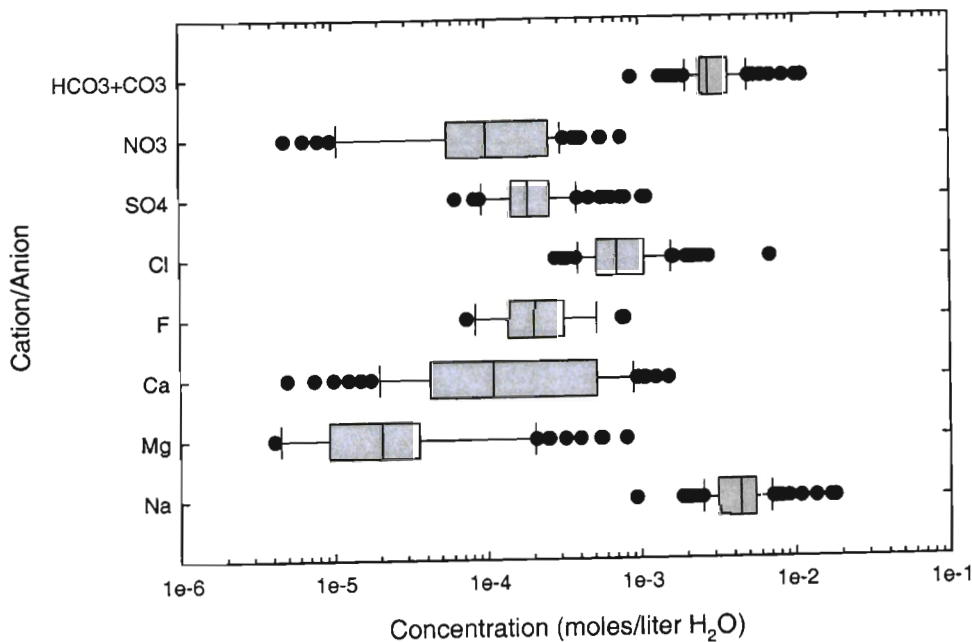
Yucca Mountain Neutral-Type Unsaturated Zone Waters
(Yang et al., 1996, 1998, 2003)



Ken Chiang
11/2/03

Discussion D. Dunn, R. Pabalon, Ken Chiang 10/28/03 1pm-3pm

Yucca Mountain Alkaline-Type Unsaturated Zone Waters
(Yang et al., 1996, 1998, 2003)



unsaturated zone porewater (pink)
perched and saturated zone water (dark green)
compositions used in evaporation simulation (light green)

Ken Chiang
11-2-03

24 Cylindrical Specimens machined to the following drawing dimensions

78668

Darrell S. Dunn
SwRI-CNWRA
Phone: (210) 522-6090
Fax: (210) 522-5184

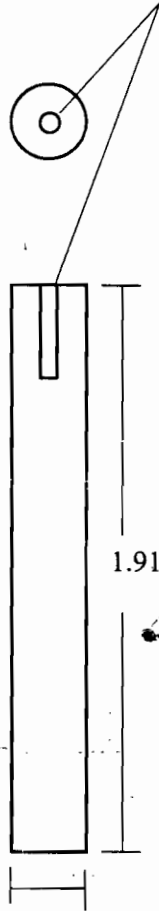
N.T.C

Cylindrical Test Specimen

92 R-B

CNWRA Drawing 20.01402.571.019

#5-40 thread centered minimum 0.250" deep



This information is to be completed at time of fabrication

Material: C-22 Alloy

Heat #: 2277-3-3266

Specimen Orientation: Specimens Need To Be perpendicular To Roll

Other: Quantity = 24 Specimens

1.915" ± 0.005"


0.250" ± 0.003"

Procedure: IPS-wt-827

Project # _____

J.C. # 78668 LOCATION CC 31

TOTAL PCS. INSPECTED	<u>6 of 24</u>	EQUIPMENT	<u>Mic 2191790</u>
TOTAL PCS. ACCEPTED	<u>6</u>		<u>Cal 30-6C-3</u>
TOTAL PCS. REJECTED	<u>0</u>		<u>Thd Base</u>
"NR #" IF REJECTS	<u>NA</u>		

INSPECTOR 

DATE NOV -3 2003

Darrell S. Dunn 10/15/01
Initiated by D. Dunn Date

B. Mabrito 10/15/01
QA Approval B. Mabrito Date

V. Jain 10/15/01
Reviewed by V. Jain Date

K.T. Chiang 11/10/03

Kuang-Tsan Ken Chiang

From: Walter Smithson [walter.smithson@swri.org]
Sent: Monday, November 03, 2003 1:26 PM
To: Kuang-Tsan K Chiang; Darrell S Dunn
Cc: Brian K Derby
Subject: CYL. TEST SPECIMENS C-22 ALLOY

6PCS. OF 24PCS. COMPLETE

Kuang-Tsan Ken Chiang

From: Walter Smithson [walter.smithson@swri.org]
Sent: Thursday, November 06, 2003 11:27 AM
To: Kuang-Tsan K Chiang; Darrell S Dunn
Cc: Brian K Derby
Subject: cyl. test specimen -

final 18pcs. of 24 complete at ms

18000

N.T.C

92 R-B

Cylindrical Test Specimen

CNwRA Drawing 20.01402.571.019

#5-40 thread centered minimum 0.250" deep

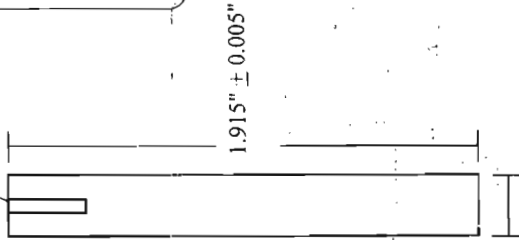
This information is to be completed at time of fabrication

Material: C-22 Alloy

Heat #: 2277-3-3266

Specimen Orientation: Specimens Neck To Be perpendicular To Roll

Other: Quantity = 24 Specimens



Walter Smithson 10/15/01
 Date: 10/15/01
 Q&A Approval B: Mabritto

6 pcs ok
 18 pcs OK NOV - 6 2003
 Cal 30-6C-3
 Mic 5056-1

Darrell S. Dunn 10/15/01
 Date: 10/15/01

Initiated by: D. Dunn
 Reviewed by: V. Jain

Darrell S. Dunn
SwRI-CNwRA
Phone: (210) 522-6090
Fax: (210) 522-5184

K.T. Chiang 11/10/03 KTC 11/10/03 03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish2277-3-3266
810 4/27/04Initial Weight: 12.54034g Model: Sartorius Genius SN: 12809099
Final Weight: 12.53842g Cal: 5/15/03 Due: 11/15/03SOLUTION: Simulated Concentrated H₂O12.966g KCl Lot# 006242 Na₂SO₄ 41.41g Lot# 025157
10.870g NaCl Lot# 034103 NaHCO₃ 192.79g Lot# 025478
17.503g NaNO₃ Lot# 020809 NaF 6.153g Lot# 990555
+ DI water To 2000mlsReagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 04Initial pH: 7.43 Model: Fisher Accumet 950 Meter SN: 3340
Final pH: 8.89 Cal: 8/11/03 Due: 8/11/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/09

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -468 mv Model: Keithley 614 SN: 0704934

Ept: -221 mv Cal: 6/09/03 Due: 6/09/04

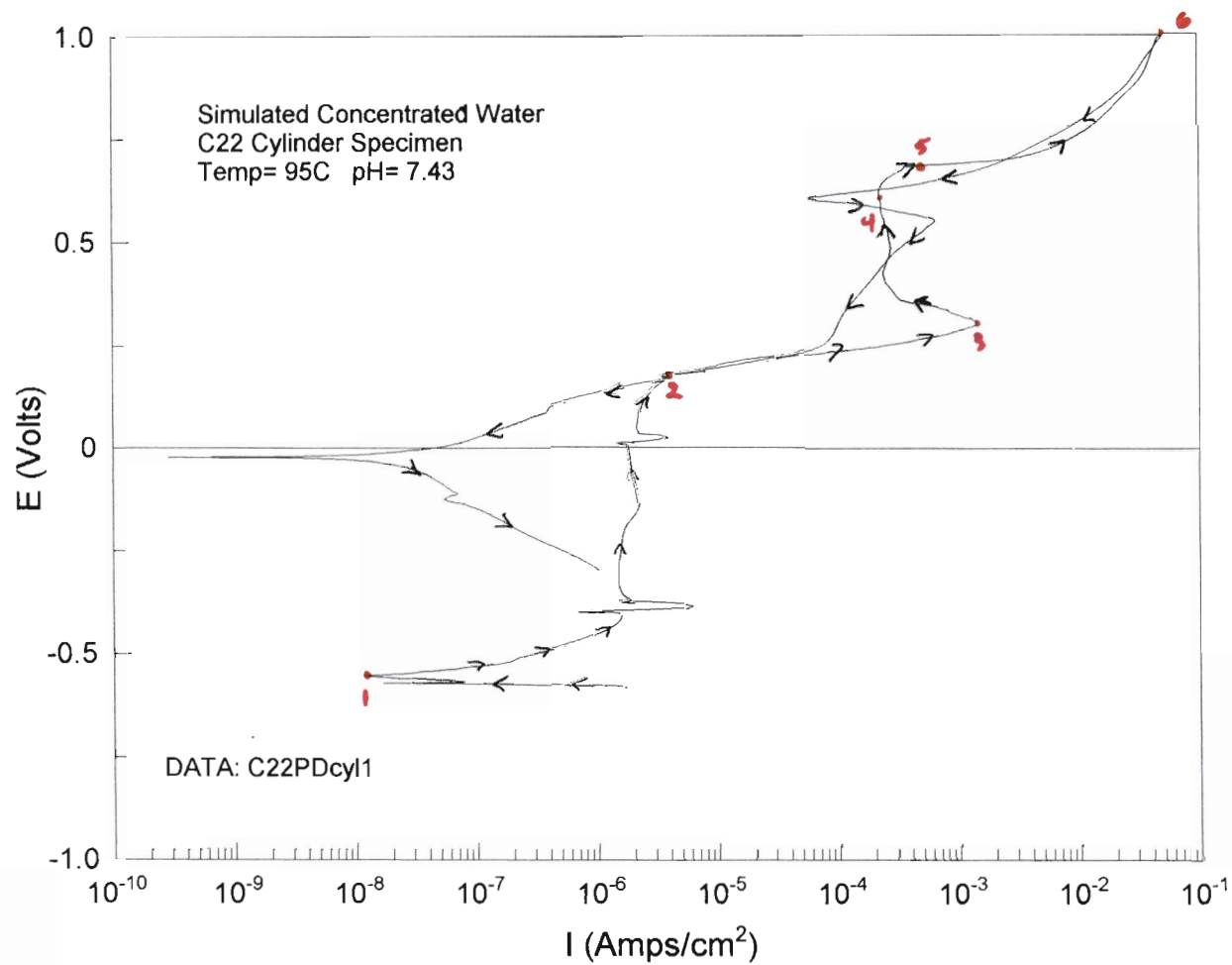
Applied (vs SCE): none

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No Visual Sign of Corrosion or Pitting
Surface Has Dull tint staining - Except Area In
Vapor phase of cell

Data: C22 PD cyl 1

B. F. J.
11/11/03



	<u>Potential</u>	<u>Current</u>
1	- 0.551	1.2187×10^{-8}
2	0.159	3.3373×10^{-6}
3	0.297	0.0013905
4	0.626	0.00022667
5	0.683	0.0047025
6	0.995	0.0487

Di. E. J. 11/12/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish2277-3-3266 5/10
6/27/04Initial Weight: 12.1797g Model: Sartorius Genius SN: 12809099
Final Weight: 12.17927g Cal: 5/15/03 Due: 11/15/03SOLUTION: Simulated Concentrated H₂O minus NO₃
12.970g KCl Lot# 006242 192.83g NaHCO₃ Lot# 025478
10.894g NaCl Lot# 034103 6.196g NaF Lot# 991559
41.40g Na₂SO₄ Lot# 025157Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 04 8/12/03Initial pH: 7.78 Model: Fisher Accumet 950 Meter SN: 3340
Final pH: 9.01 Cal: 8/11/03 Due: 8/11/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -673mV Model: Keithley 614 SN: 0704934

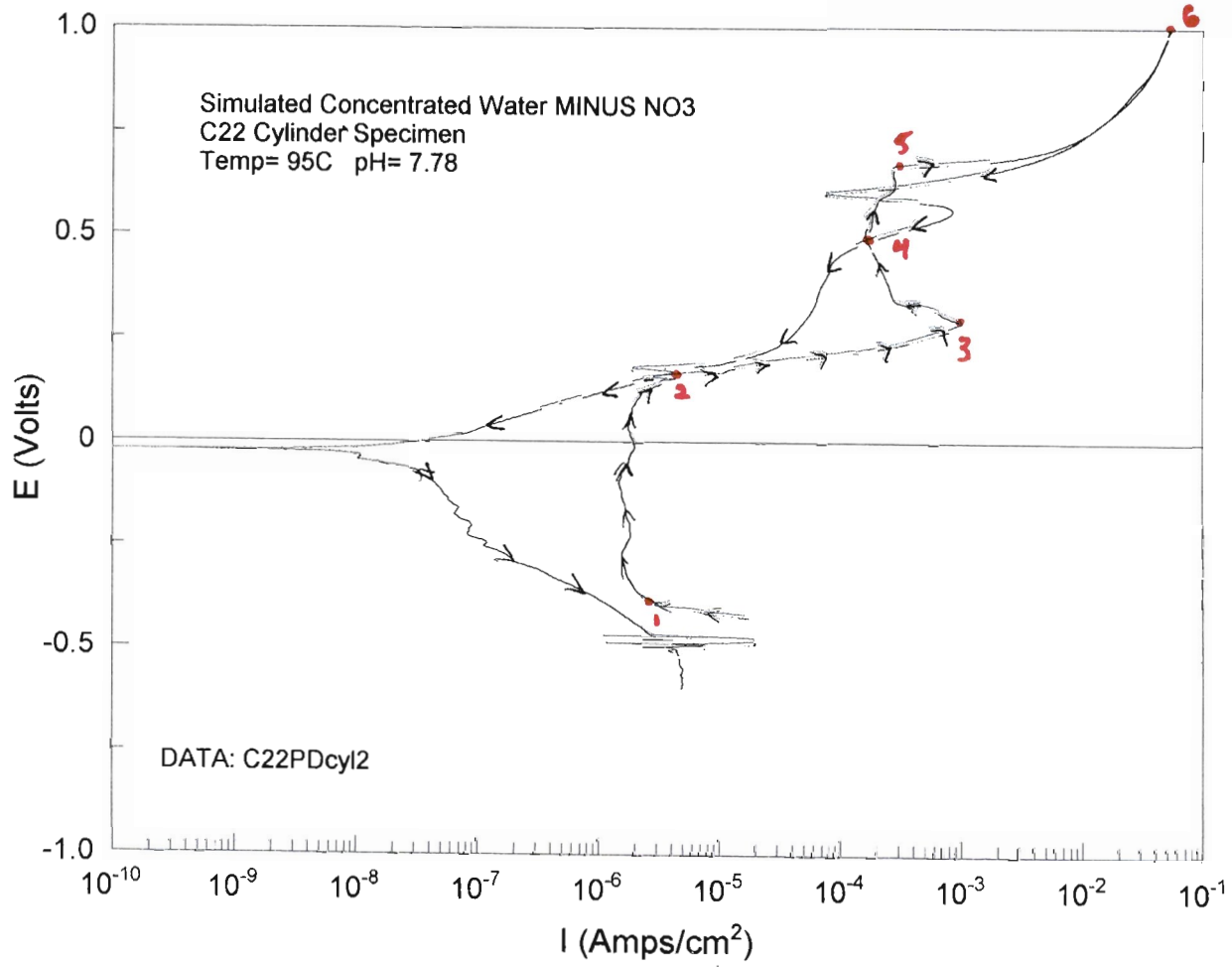
Ept: -227mV Cal: 6/09/03 Due: 6/09/04

Applied (vs SCE): none

Potentiostat: Solartron 1480 SN# 00240551

specimen Examination: No visual signs of Corrosion/Pitting
Surface Has A Dull Tint staining - Area In Vapor
phase of Cell look great No staining* Note: Simulated Concentrated H₂O was minus NO₃

Data C22PDex/2 Bi-RJ 11/12/03



	<u>Potential</u>	<u>Current</u>
1	-0.400	3.1727×10^{-6}
2	0.166	8.971×10^{-6}
3	0.280	0.00098319
4	0.488	0.0001651
5	0.662	0.00030346
6	0.995	0.052418

R. D. J. 11/15/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
2277-3-3266 810
6/27/01

Initial Weight: 12.29263g Model: Sartorius Genius SN: 12809099
Final Weight: 12.29102g Cal: 5/15/03 Due: 11/15/03

SOLUTION: Simulated Concentrated Water minus Na₂SO₄
12.962g KCl Lot # 006242 192.73g NaHCO₃ Lot # 024924
10.872g NaCl Lot # 034103 6.20g NaF Lot # 951559
17.587g NaNO₃ Lot # 020809
+ DI water to 2000mls

Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 04 11/14/03

Initial pH: 7.76 Model: Fisher Accumet 950 Meter SN: 3340
Final pH: 9.14 Cal: 8/11/03 Due: 8/11/04
pH Probe: #13-620-296 SN: 2291257P6

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

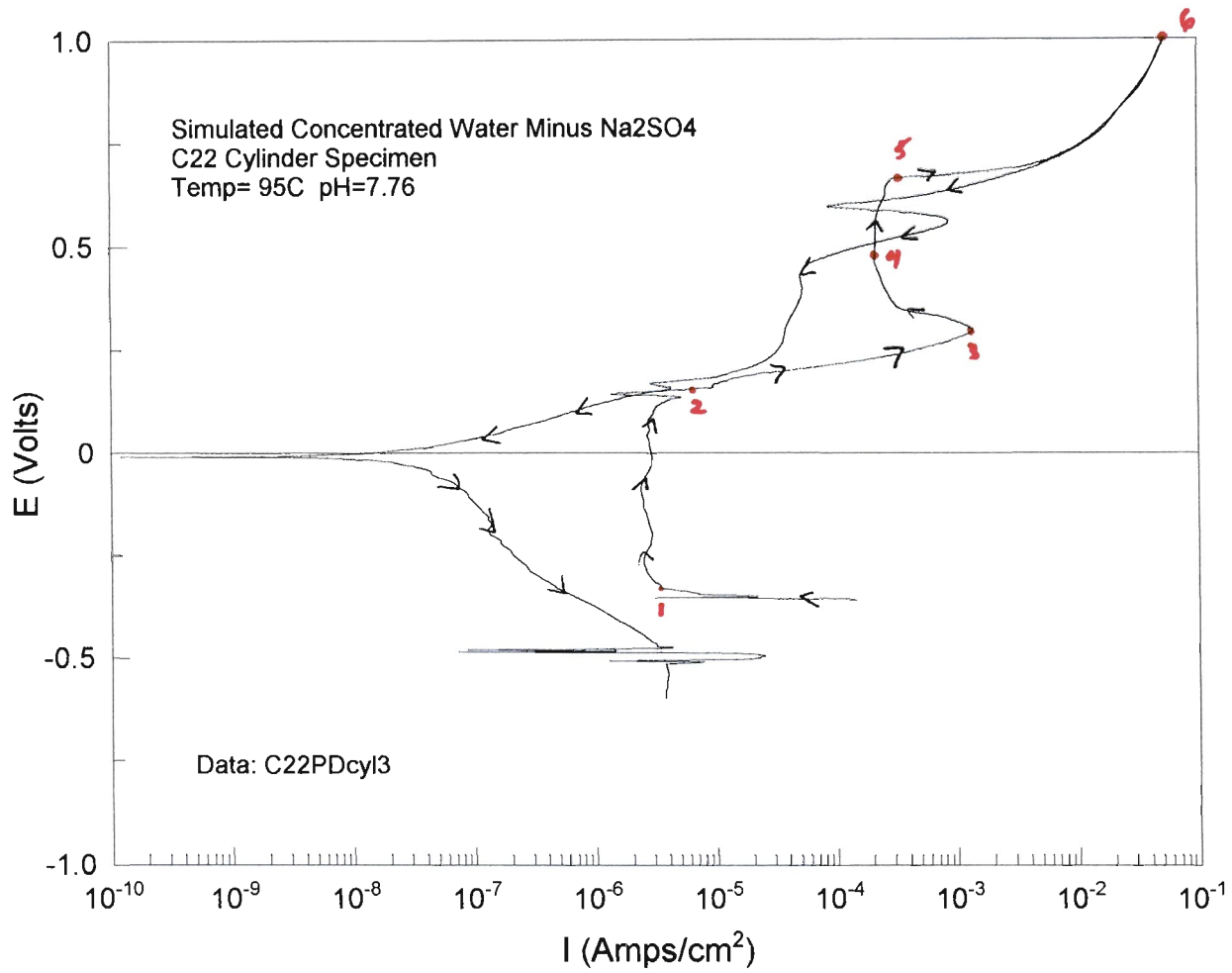
Counter Electrode: Platinum Flag
Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen
Ecorr: -620 mV Model: Keithley 614 SN: 0704934
Ept: -244 mV Cal: 6/09/03 Due: 6/09/04
Eapplied (vs SCE): none

Potentiostat: Solartron 1480 SN# 00240557

Specimen Examination: No Sign of Corrosion/Pitting on Any Surface of Specimen; Surface has Dull tint staining

Date: C22PD 6/13 B. R. J. 11/14/03



	<u>Potential</u>	<u>Current</u>
1	-0.333	3.5079×10^{-6}
2	0.159	9.3344×10^{-6}
3	0.292	0.0013224
4	0.453	0.00020772
5	0.654	0.00028103
6	0.996	0.051672

B. E. S. J. 11/17/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
2277-3-3266 8/03
6/27/04Initial Weight: 12.4036g Model: Sartorius Genius SN: 12809099
Final Weight: 12.4004g Cal: 11/14/03 Due: 5/14/04SOLUTION: Simulated Concentrated water minus NaF
12.96g KCl Lot# 006242 142.7g NaHCO₃ Lot# 028924
16.86g NaCl Lot# 034103 + DI water To 2000ml
17.50g NaNO₃ Lot# 020809
41.39g Na₂SO₄ Lot# 025157Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/04Initial pH: 7.84 Model: Fisher Accumet 950 Meter SN: 3340
Final pH: 9.16 Cal: 8/11/03 Due: 8/11/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -561mV Model: Keithley 614 SN: 0704934
Ept: -307mV Cal: 6/09/03 Due: 6/09/04

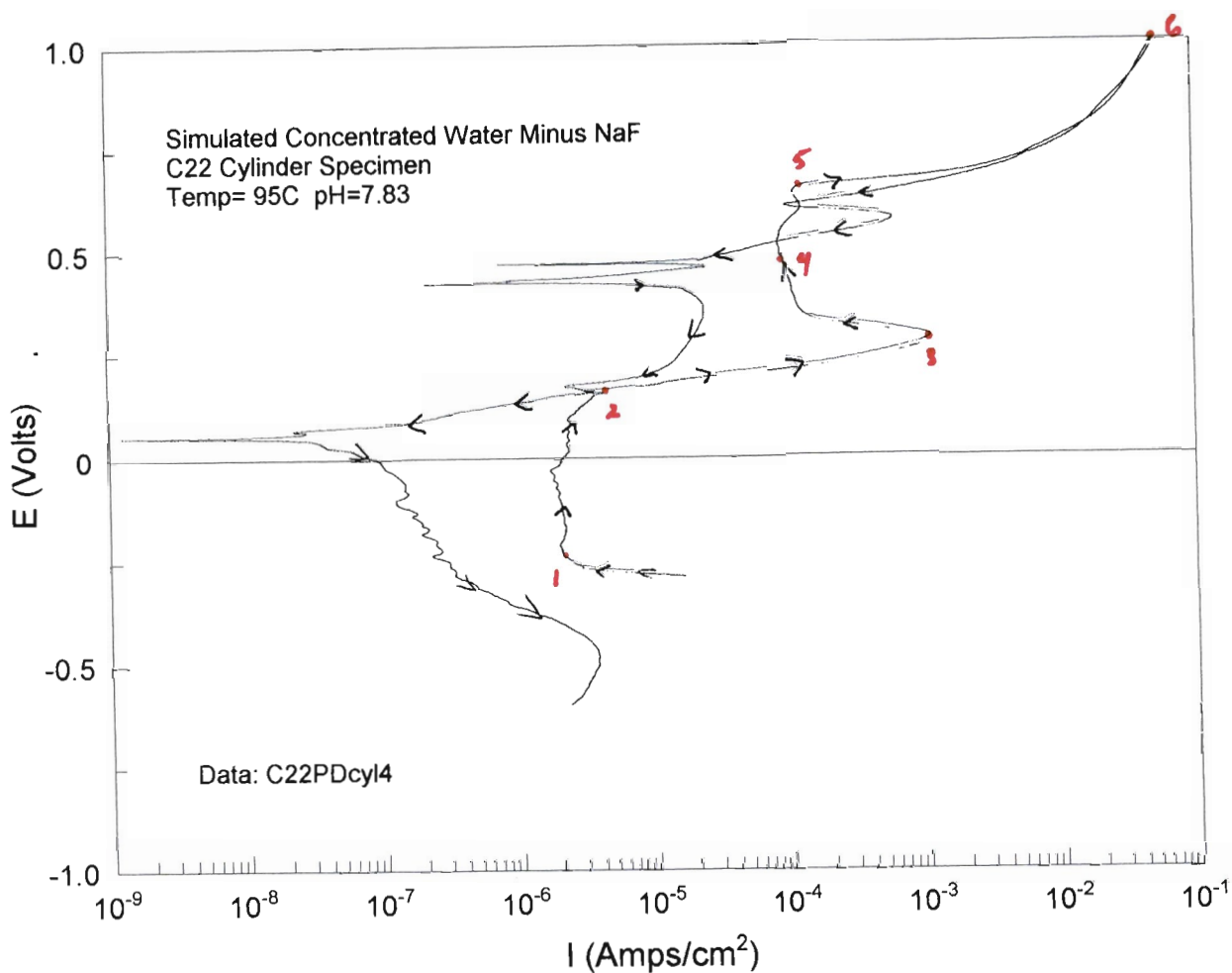
Applied (vs SCE): none

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No Visual Signs of Corrosion/Pitting
on surface. Dull tint staining on All surfaces
of specimen

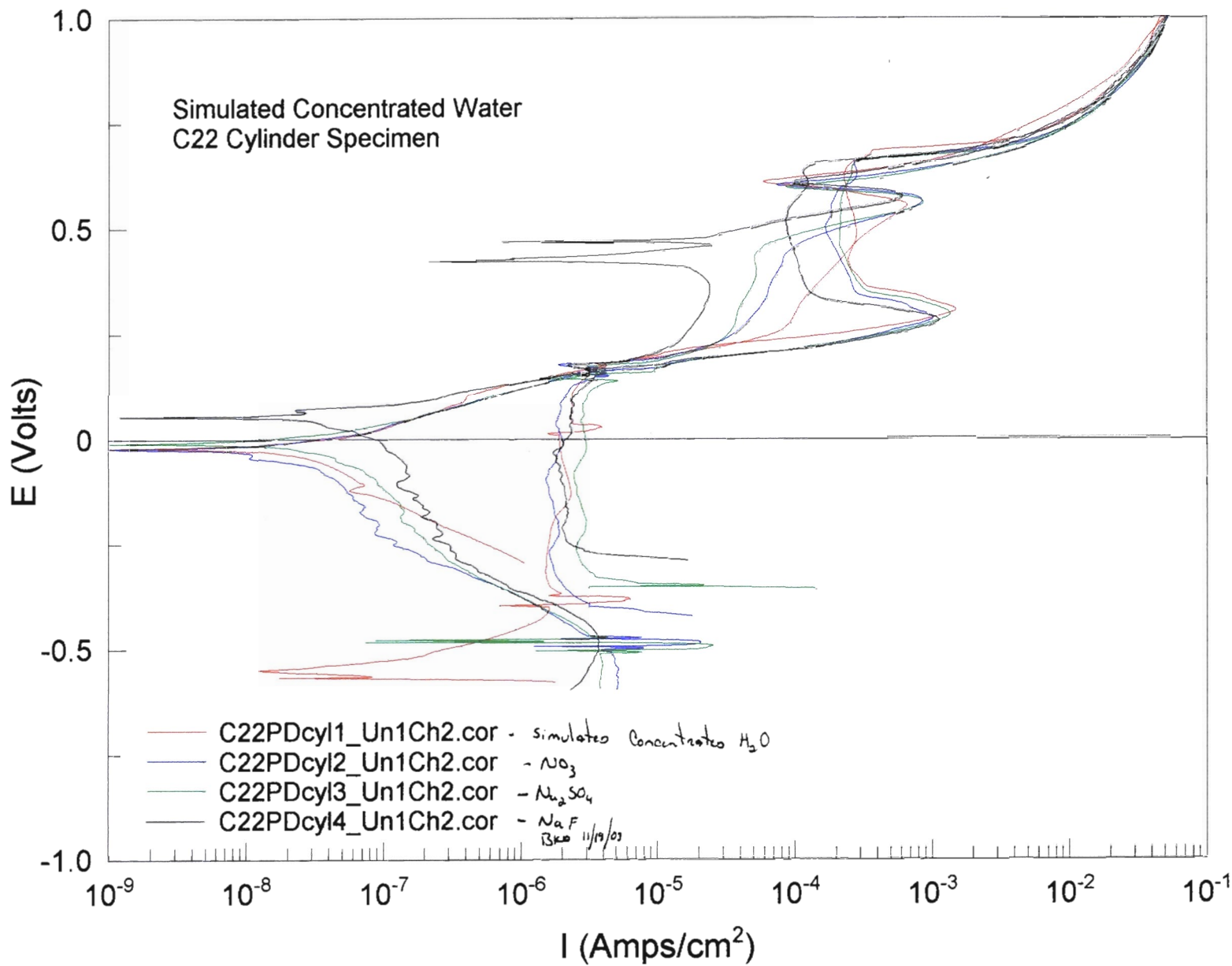
Date C22PO04/4

B. K. J. 11/12/03



	<u>Potential</u>	<u>Current</u>
1	- 0.255	2.4917×10^{-6}
2	0.171	6.7891×10^{-6}
3	0.276	0.6011077
4	0.526	8.7174×10^{-5}
5	0.656	0.00016785
6	0.997	0.052362

B. F. J. 11/18/03



Bi E
11/19/03

New Test Matrix:

Continue with C-22 Cylinders HT # 2277-3:3266

Solution will be Simulated Concentrated Water

But Instead of Removal of $\text{NaF} - \text{Na}_2\text{SO}_4$

And NaNO_3 will Decrease Molarity of Solution

* Note: will Reduce Molarity of Each Reagent $\text{NaF} - \text{Na}_2\text{SO}_4$

And NaNO_3 In Simulated Concentrated Water

See Each Individual Test for Concentrations

Temp: will Remain 95°C

All other test parameters will Remain the Same

All other Equipment will Remain the Same through out Testing In Simulated Concentrated Water

Bi Rdf 11/20/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ 2277-3-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
819
6/27/06Initial Weight: 12.45528g Model: Sartorius Genius SN: 12809099 610/12/03
Final Weight: 12.45489g Cal: 11/14/03 Due: 5/14/03 04SOLUTION: Simulated Concentrated water Decarbox NaF
12.964g KCl Lot# 006242 41.40g Na₂SO₄ Lot# 025157
10.869g NaCl Lot# 034103 192.71g NaHCO₃ Lot# 028924
17.051g NaNO₃ Lot# 020809 3.093g NaF Lot# 99559
+ DI water to 2000mlReagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 04 810/11/03Initial pH: 7.76 Model: orion SN: S001A
Final pH: 9.21 CAL: 1/9/03 DUE: 1/9/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/03

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -649mV Model: Keithley 614 SN: 0704934

Ept: -305mV Cal: 6/09/03 Due: 6/09/04

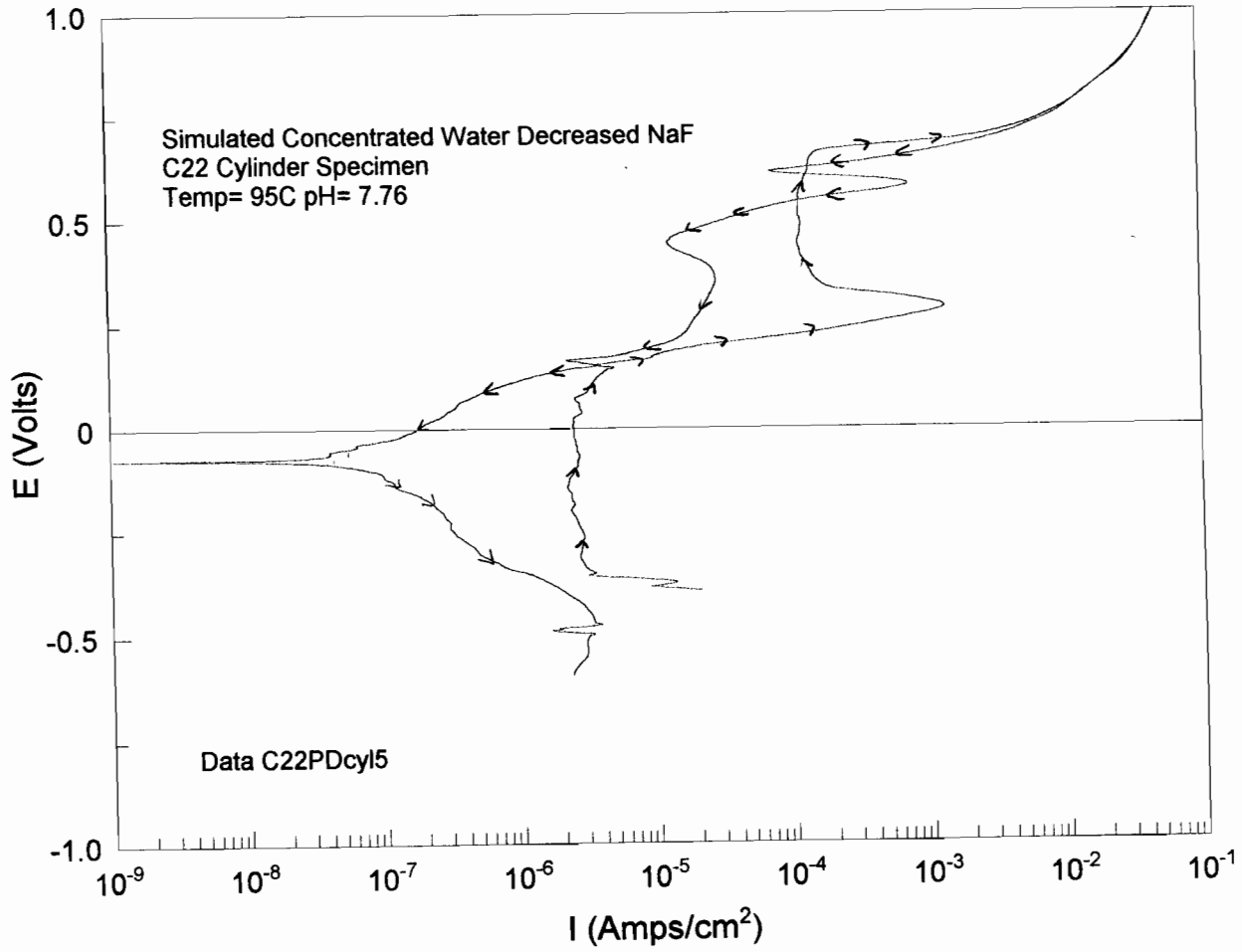
Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No Visual Sign of Corrosion/Pitting
on Specimen. Dull tint staining on All Surfaces

data C22PO g15

11/21/02 B-PJ



Handwritten signature and date: 11/2/02

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3206~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
 2277-3-3266 ^{6/27/06}

Initial Weight: 12.3349g Model: Sartorius Genius SN: 12809099 ^{6/14/03}
 Final Weight: 12.33465g Cal: 11/14/03 Due: 5/14/08 ⁰⁴

SOLUTION: Simulated Concentrated water Decreases NaNO_3
 12.961g KCl Lot # 066242 41.39g Na_2SO_4 Lot # 035451
 10.861g NaCl Lot # 034103 192.70g NaHCO_3 Lot # 028924
 8.773g NaNO_3 Lot # 020809 6.190g NaF Lot # 991555
 + DI water to 2000mls

Reagents measured with Model: OHAUS SN: 2883
 Cal: 7/29/03 Due: 1/29/08 ⁰⁴

Initial pH: 7.71 Model: orion SN: S001A
 Final pH: 9.34 CAL: 1/9/03 DUE: 1/9/04
 pH Probe: #13-620-296 SN: 2291257P6

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
 Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag
 Reference Electrode: Fisher SCE

13-620-52 SN: 0251439

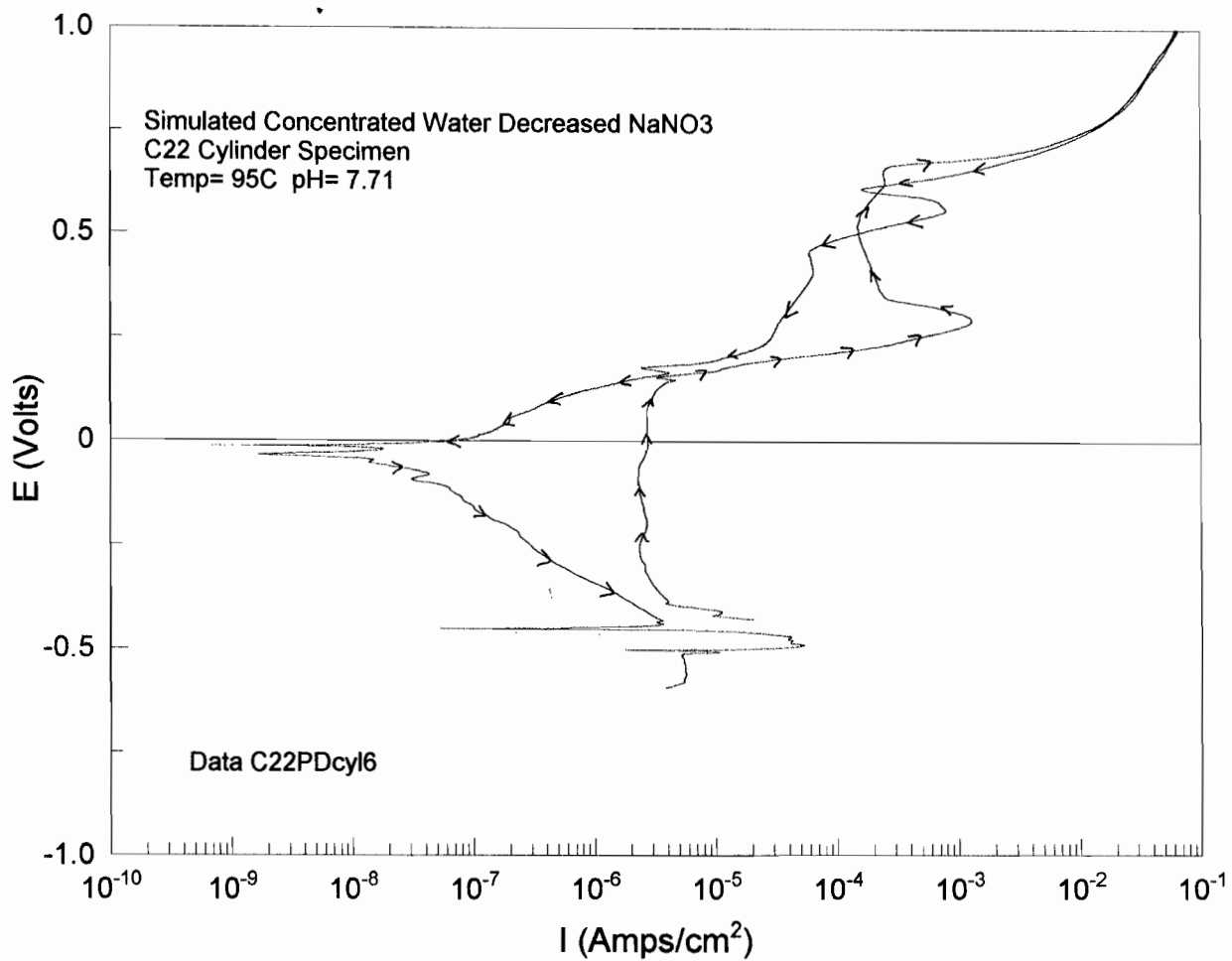
Gas: 99.999% Nitrogen
 Ecorr: - 636 mV Model: Keithley 614 SN: 0704934
 Ept: - 249 mV Cal: 6/09/03 Due: 6/09/04
 Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No visual signs of corrosion / P.H.ig on
 Specimen - dull tint staining on surface

Data C22 PD Cyl 6

B. [Signature] 11/24/03



B-K 11/24/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ 2277-3-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
016
6/27/06Initial Weight: 12.43485g Model: Sartorius Genius SN: 12809099
Final Weight: 12.43395g Cal: 11/14/03 Due: 5/14/08 04 12/10/08SOLUTION: simulated concentrated water Decreased Na_2SO_4
12.966g KCl lot# 006242 20.69g Na_2SO_4 lot# 035451
16.864g NaCl lot# 034103 192.75g NaHCO_3 lot# 028924
17.521g NaNO_3 lot# 020809 6.192g NaCl lot# 991559
+ O₂ water to 2000 ml'sReagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 04 10/10/03Initial pH: 7.74 Model: orion SN: S001A
Final pH: 9.21 CAL: 1/9/03 DUE: 1/9/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -445mV Model: Keithley 614 SN: 0704934

Ept: -233mV Cal: 6/09/03 Due: 6/09/04

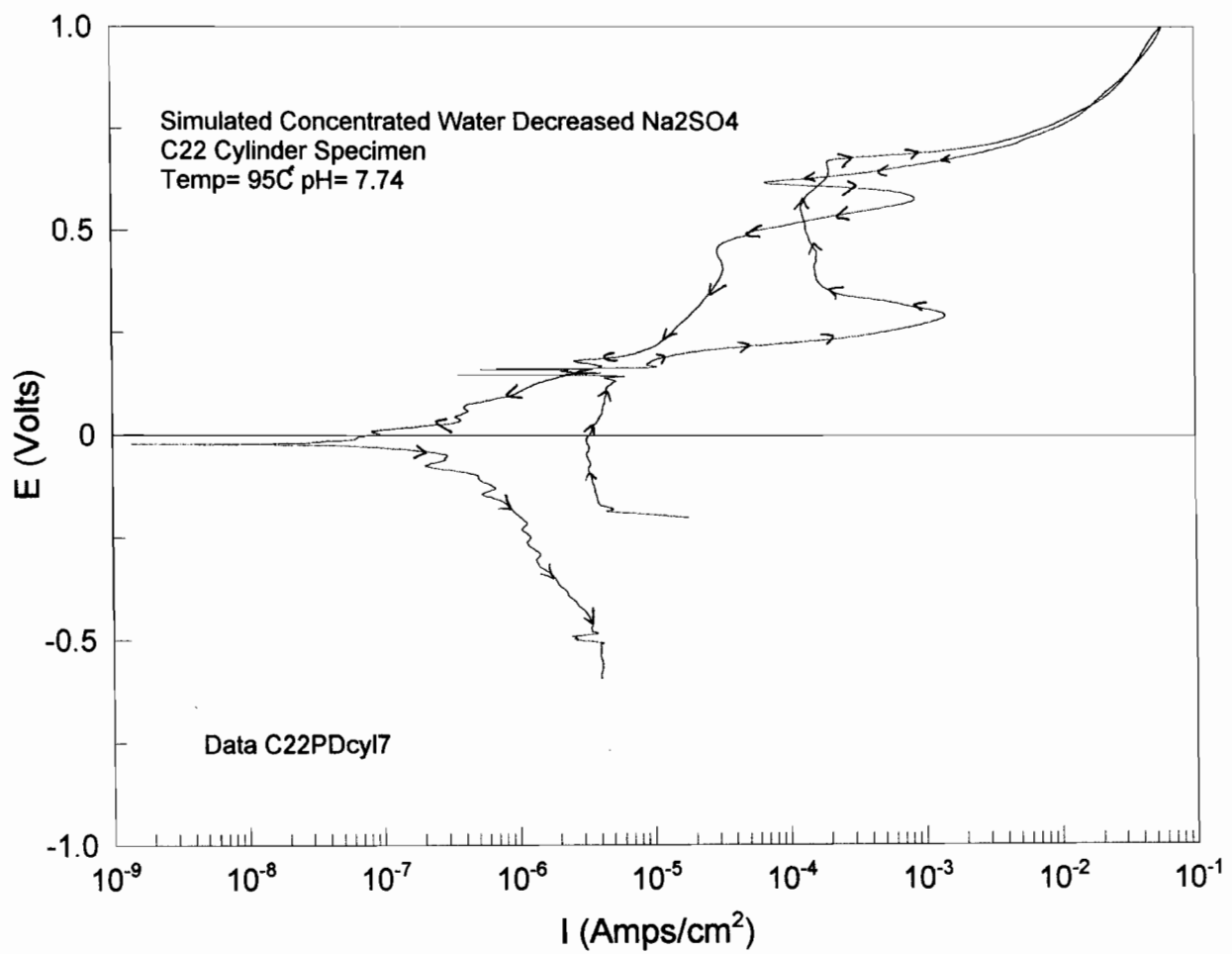
Eapplied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No sign of corrosion/pitting
staining on all surfaces of specimen (Dull Tint)

data C22POC/17

B. E. J. 11/24/03



See pg 41 for Graph of Test Data
C22PD cyl #5 - #7 - Comparison Graph

[Signature] 11/25/03

^{pore-}
112 water calculation from Bobby Pabalan

Kuang-Tsan Ken Chiang

From: R Pabalan [rpabalan@cnwra.swri.edu]
Sent: Wednesday, December 03, 2003 1:33 PM
To: Darrell Dunn; Ken Chiang
Subject: UZ porewater evaporation results



Synth UZ
rewater Evap at 95-

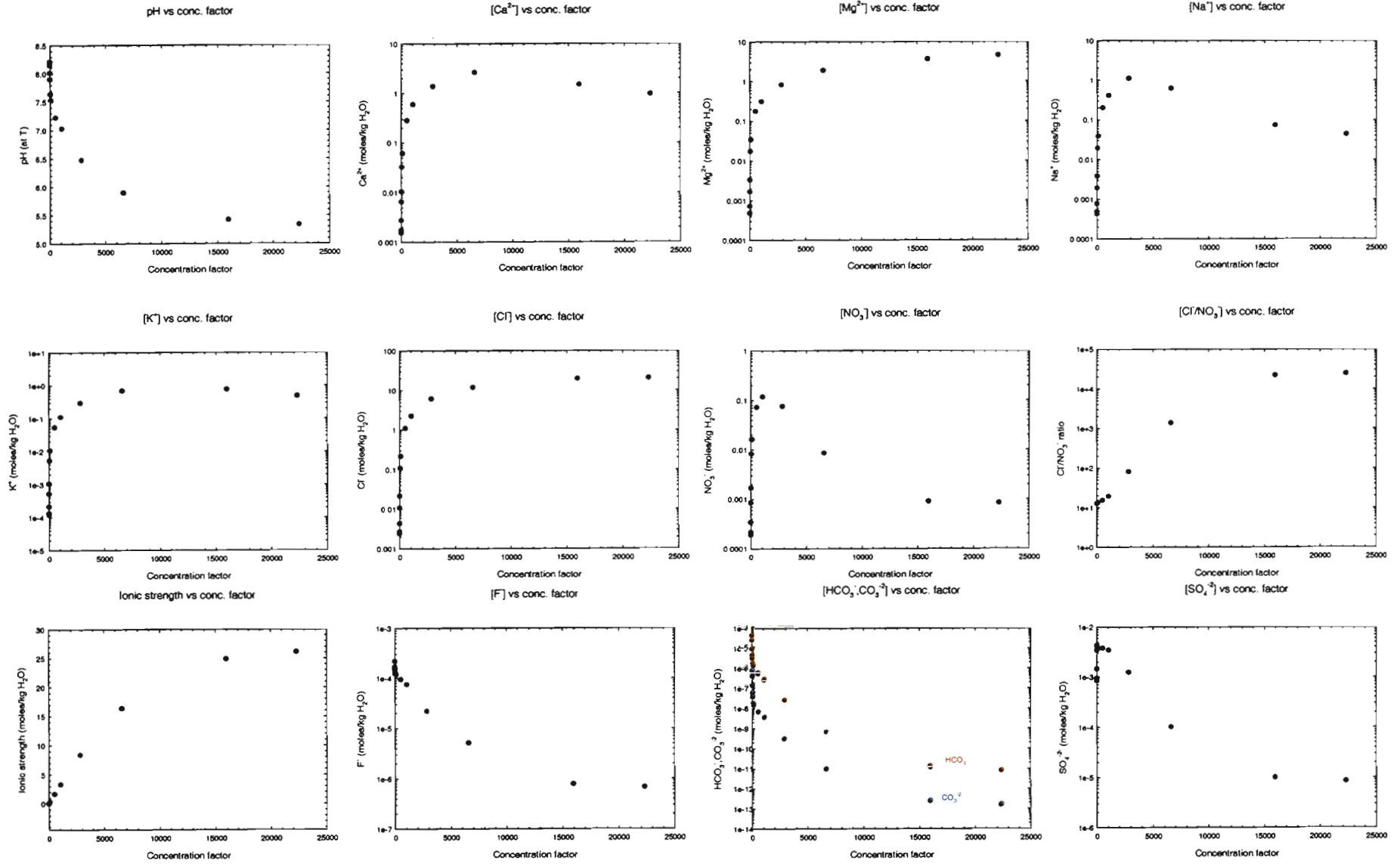
Darell, Ken,

Attached is a Word file with plots of pH, ionic strength, and aqueous species concentration as a function of concentration factor and temperature (95, 110, 125, and 140 C) for UZ porewater. The porewater composition was taken from Rosenberg et al. (2001).

bobby

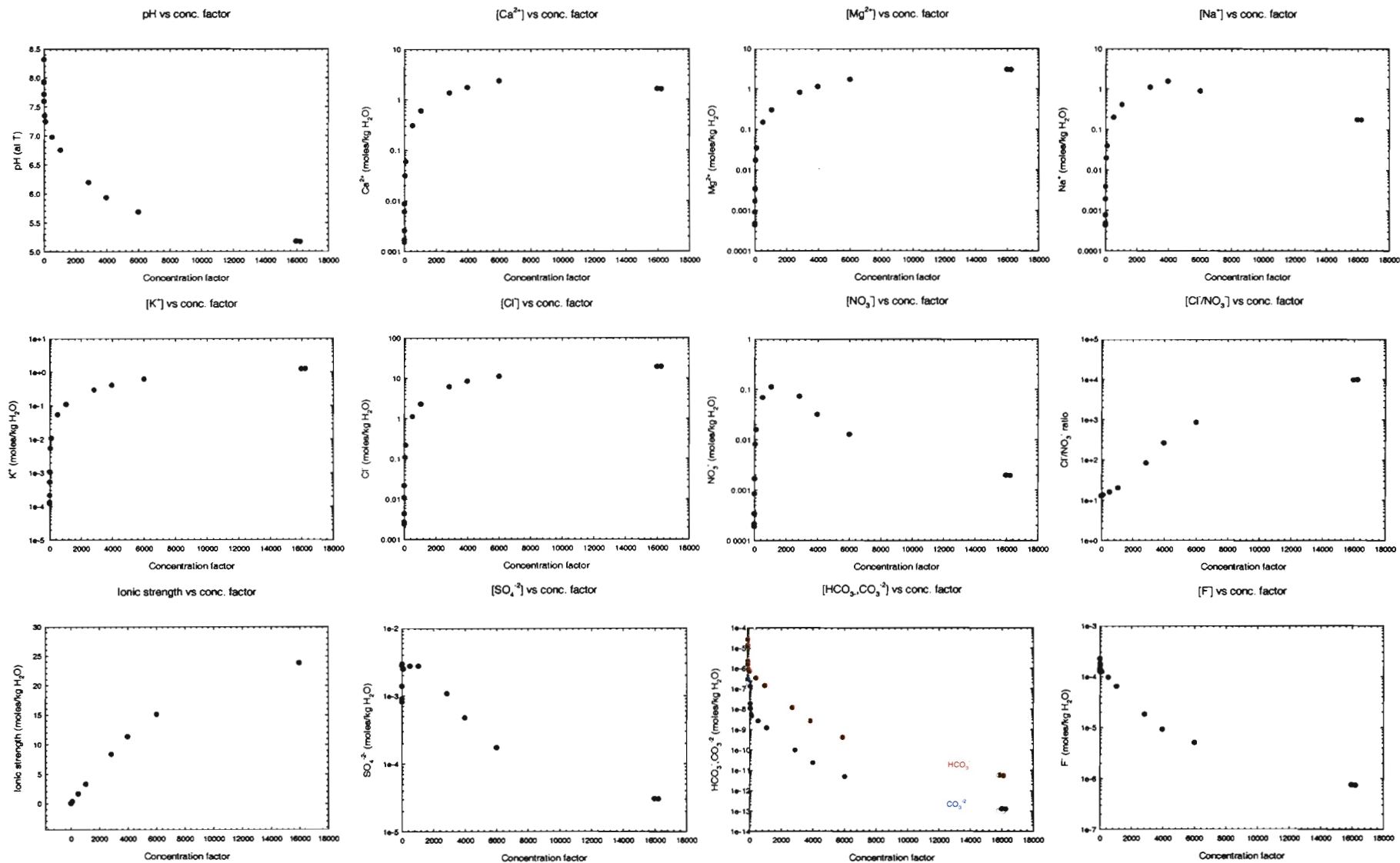
K. T. Chiang 12/4/03

Synthetic UZ porewater evaporation at T=95 °C



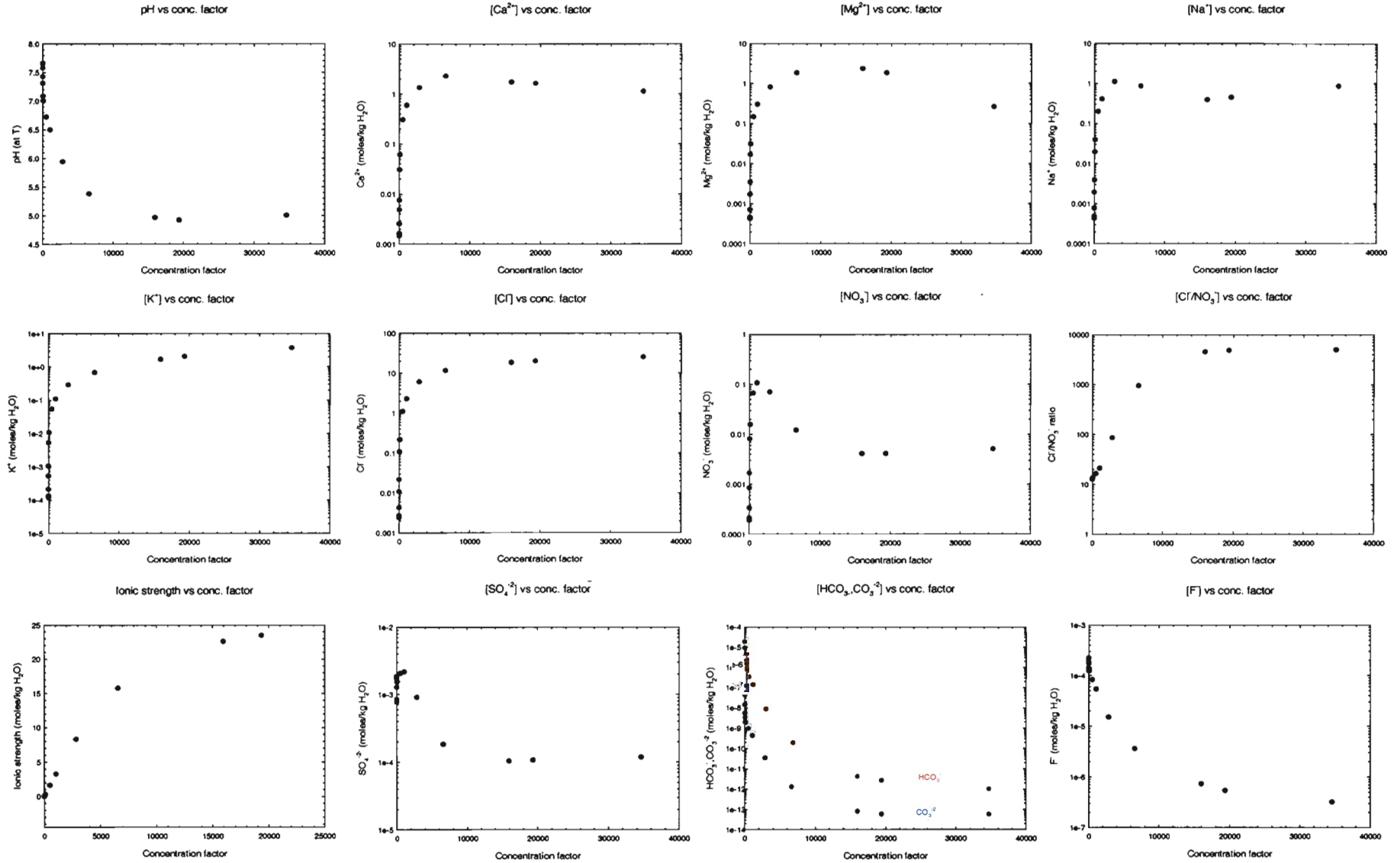
R. J. Chisep 12/14/03

Synthetic UZ porewater evaporation at T=110 °C



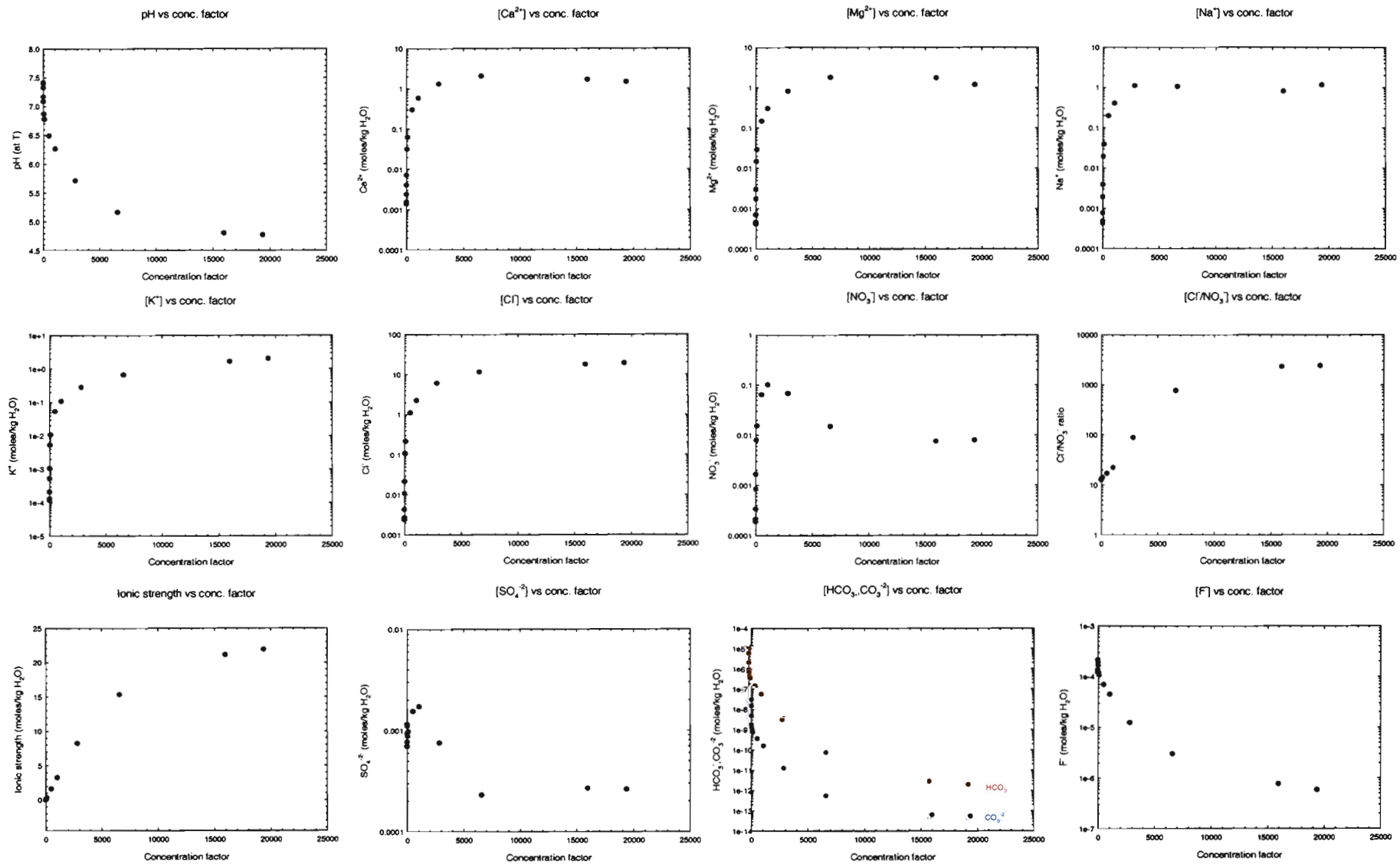
K. S. Ulrich 12/4/05

Synthetic UZ porewater evaporation at T=125 °C

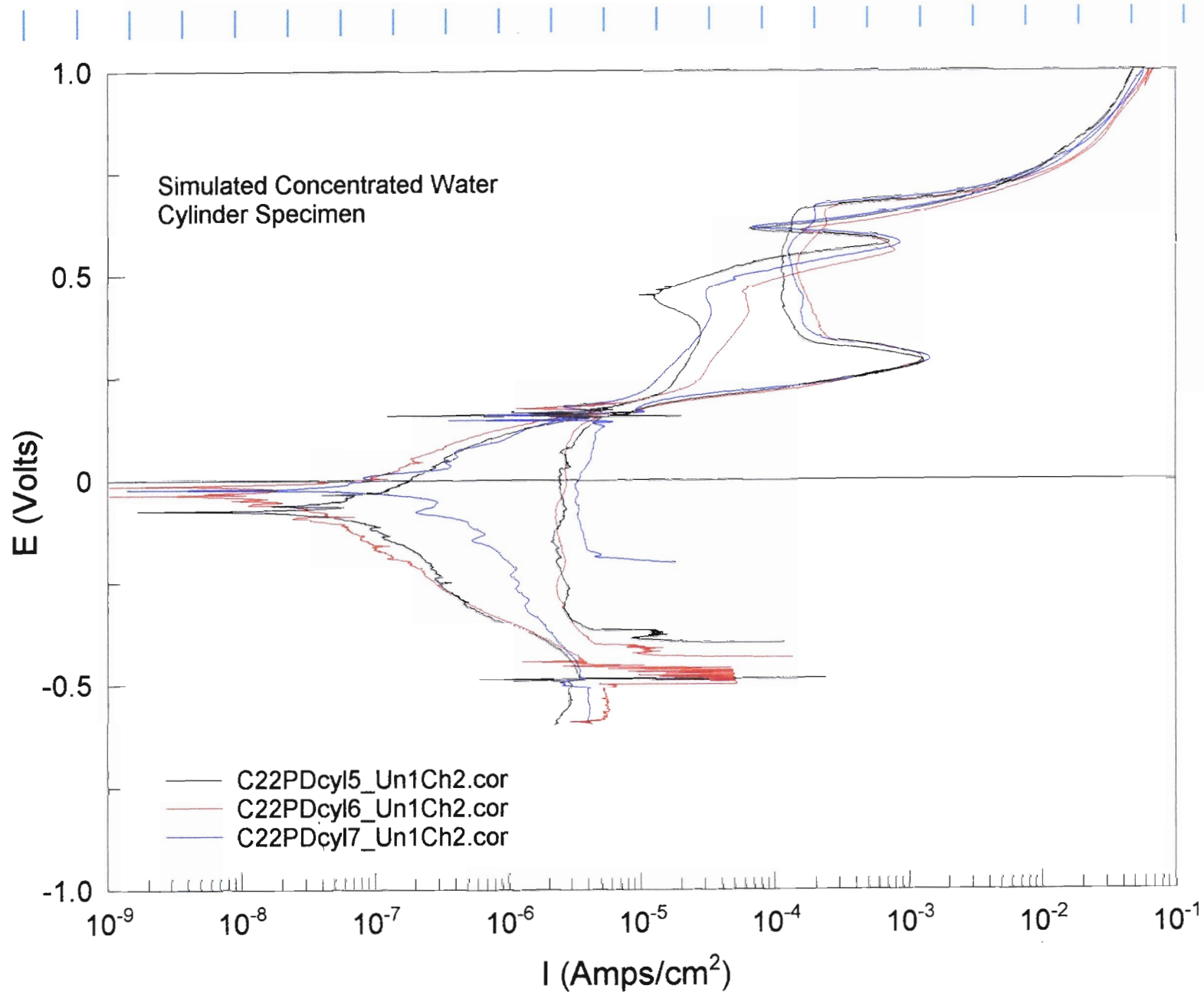


K.T. C. Leap 12/14/03

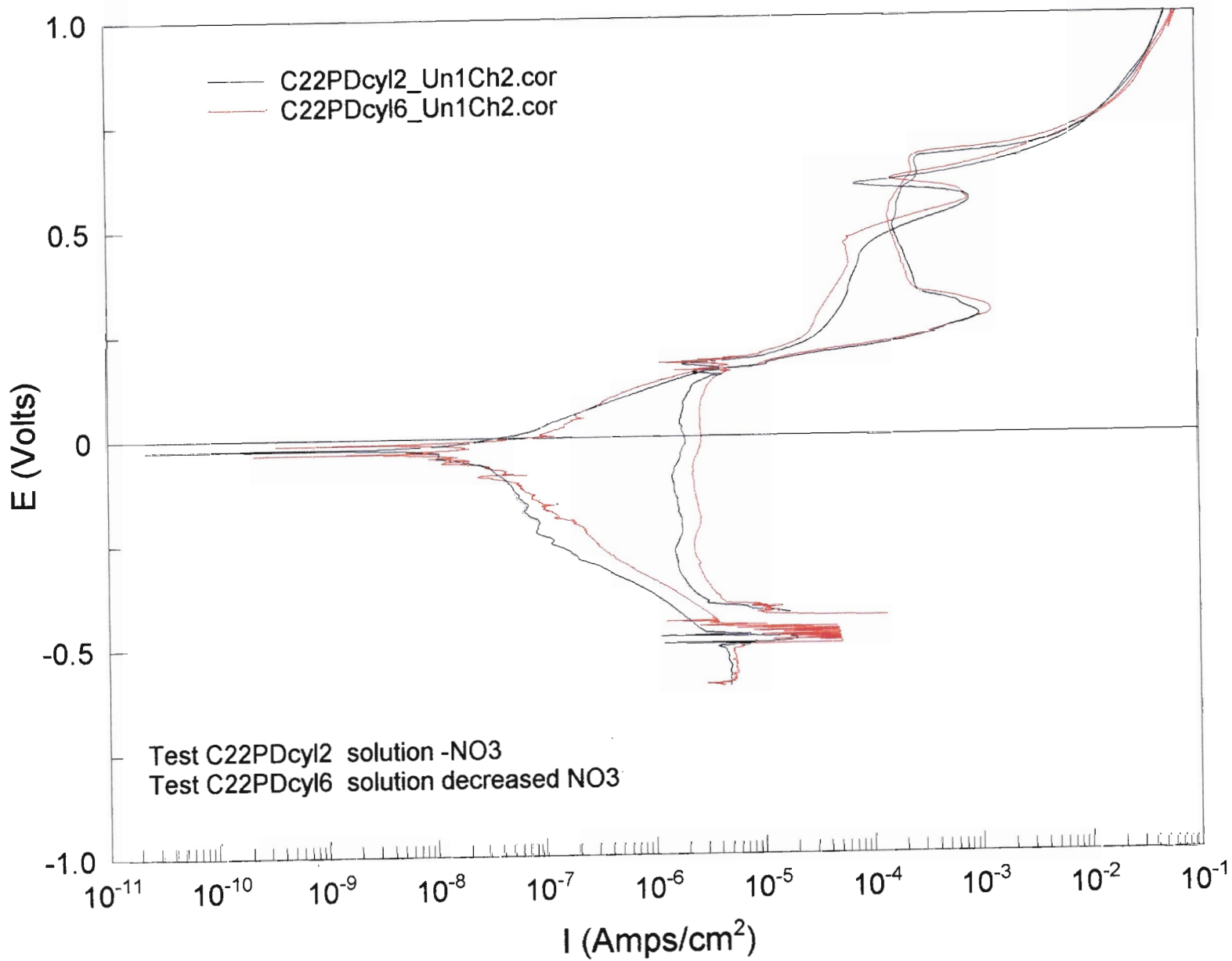
Synthetic UZ porewater evaporation at T=140 °C



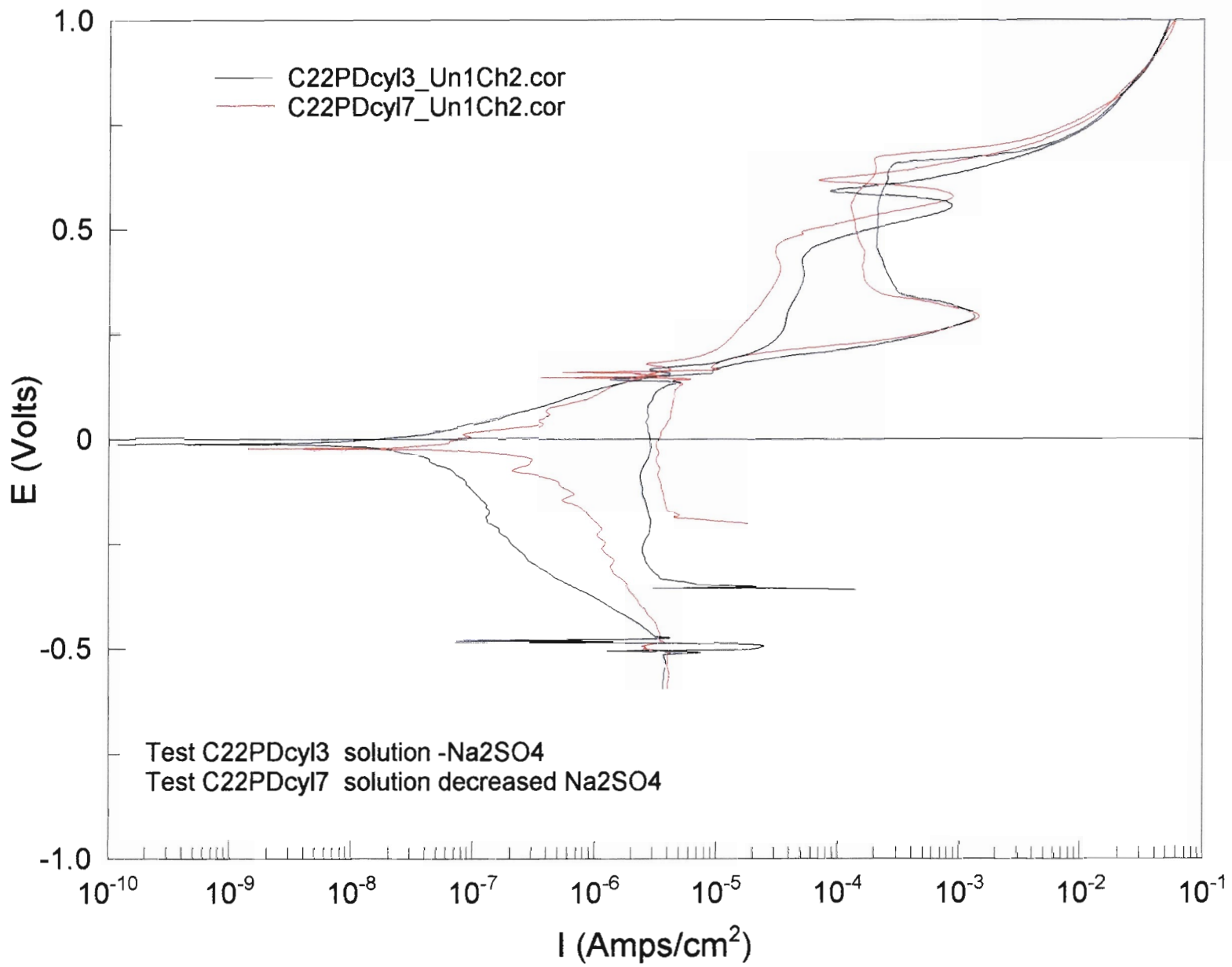
K. J. (Lionel) 12/14/03



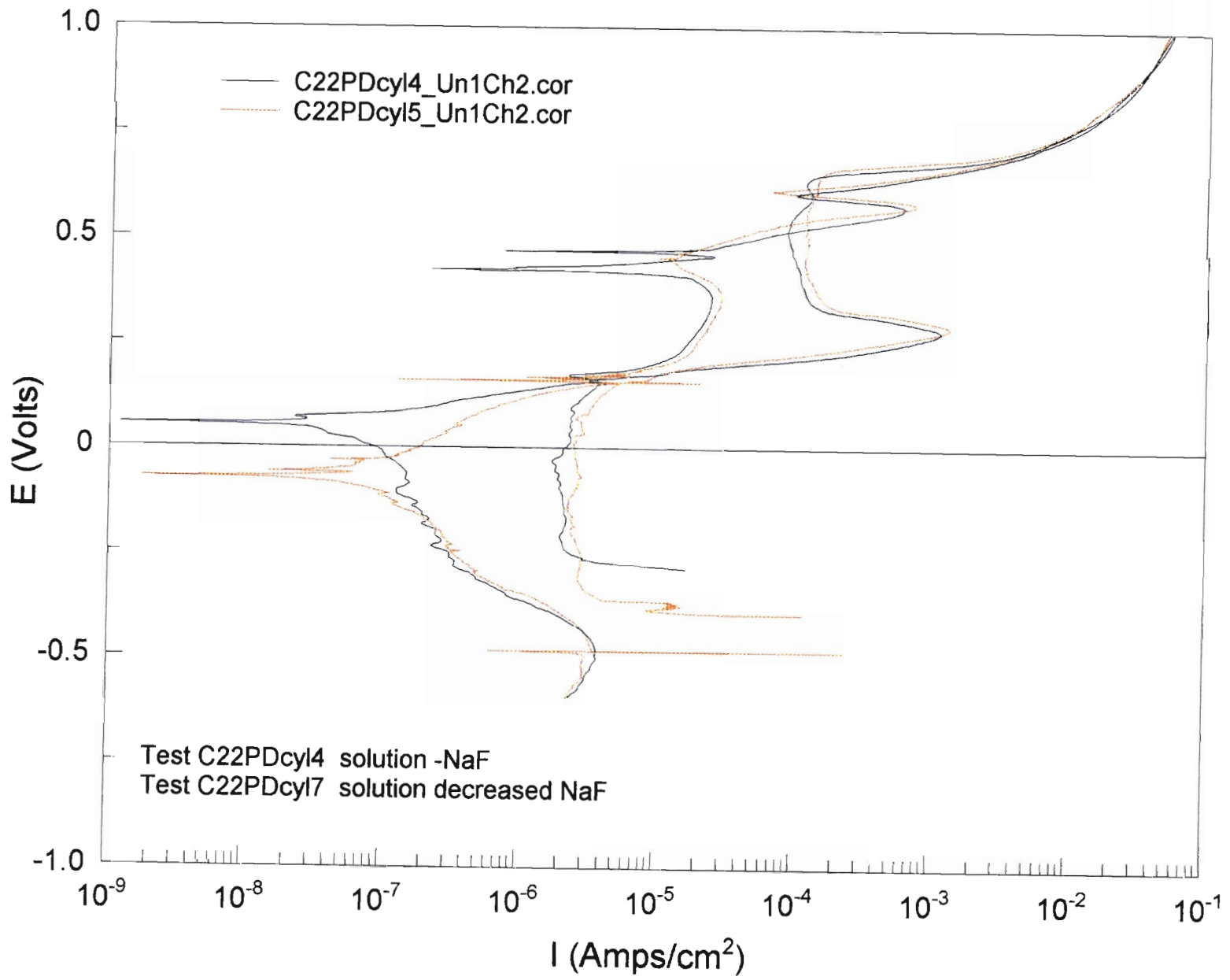
[Handwritten signature]
12/4/03



S. J. D.
12/5/03

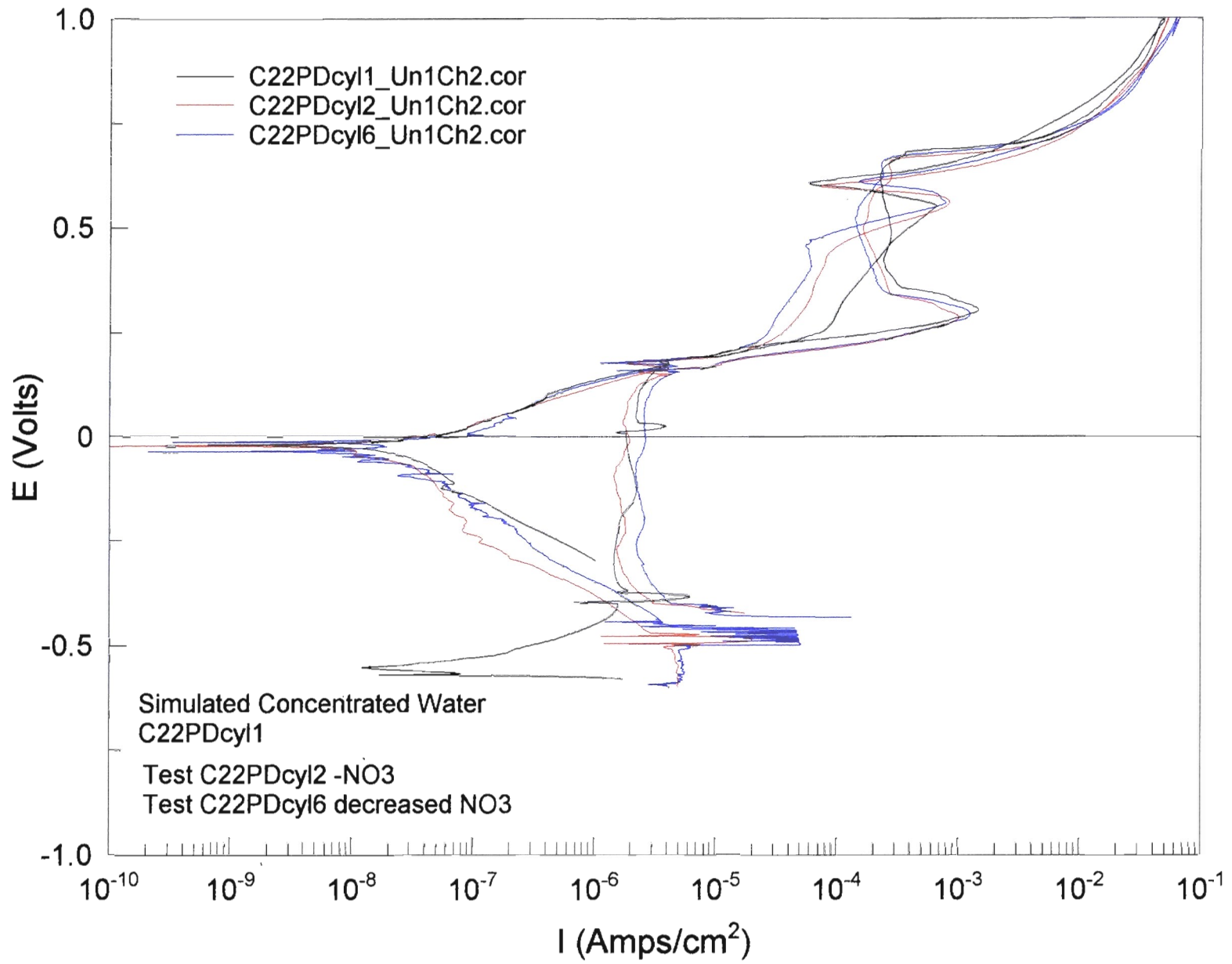


50/5/21
AC

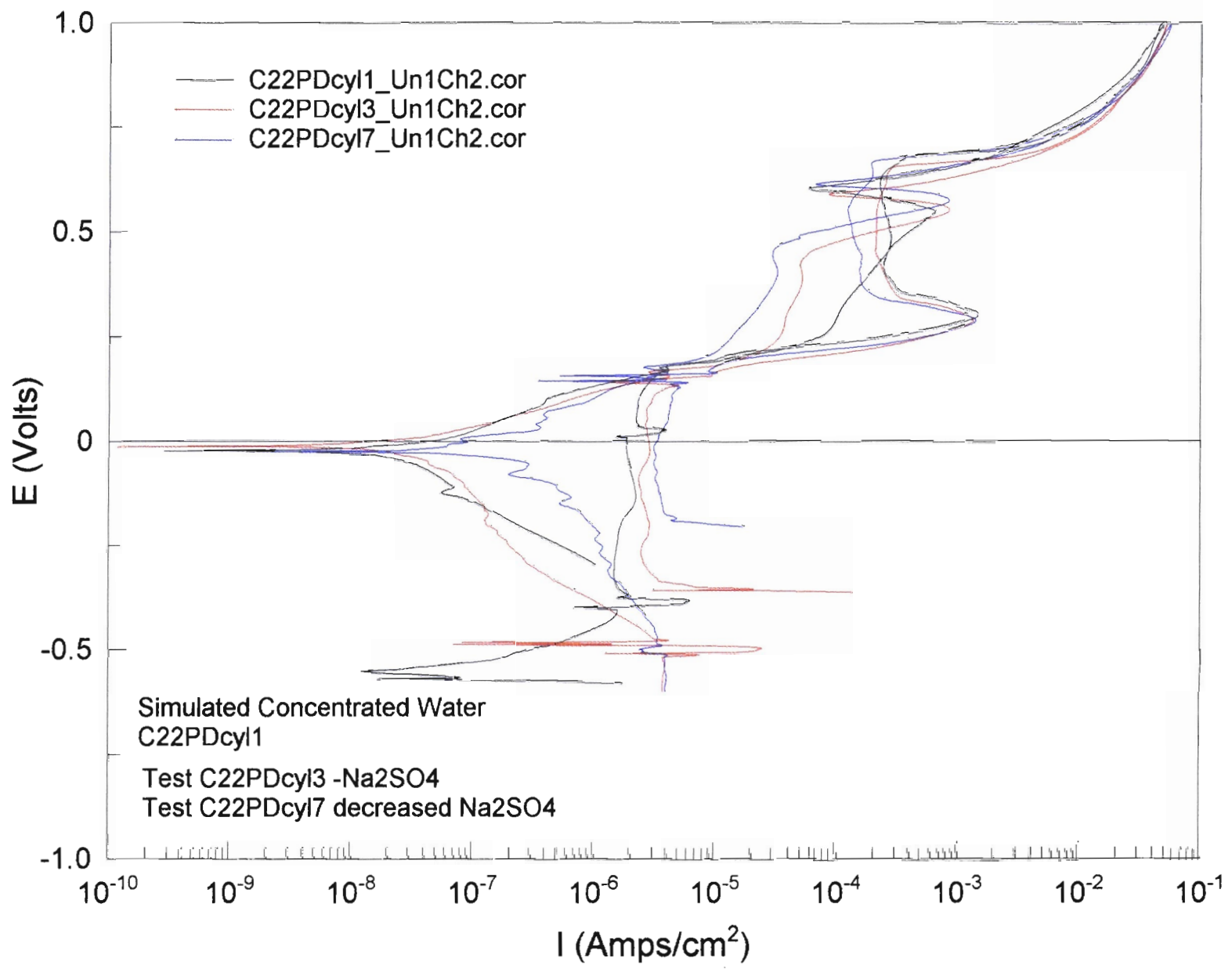


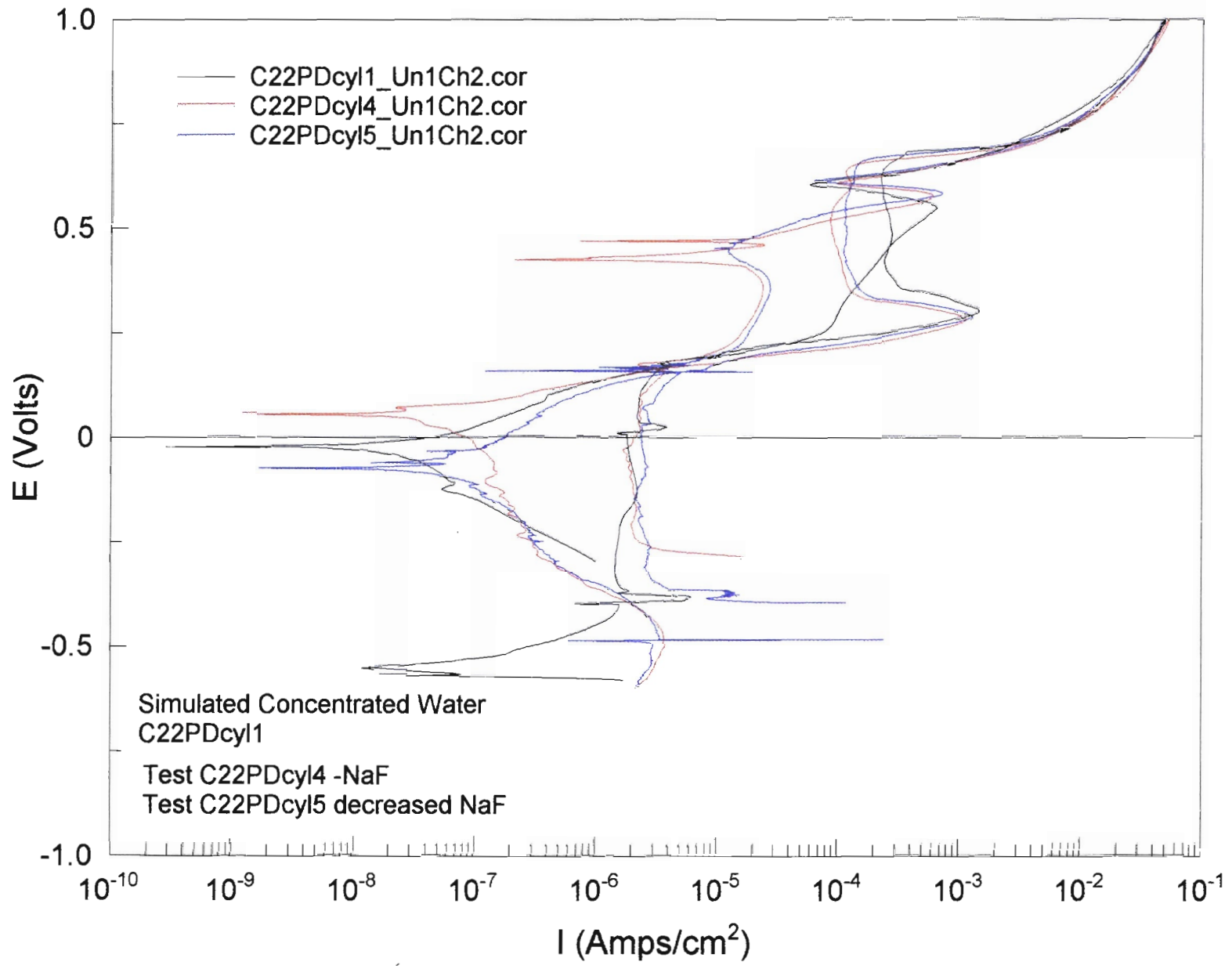
S. K. S.
12/5/03

12/1/02
B. J. [Signature]

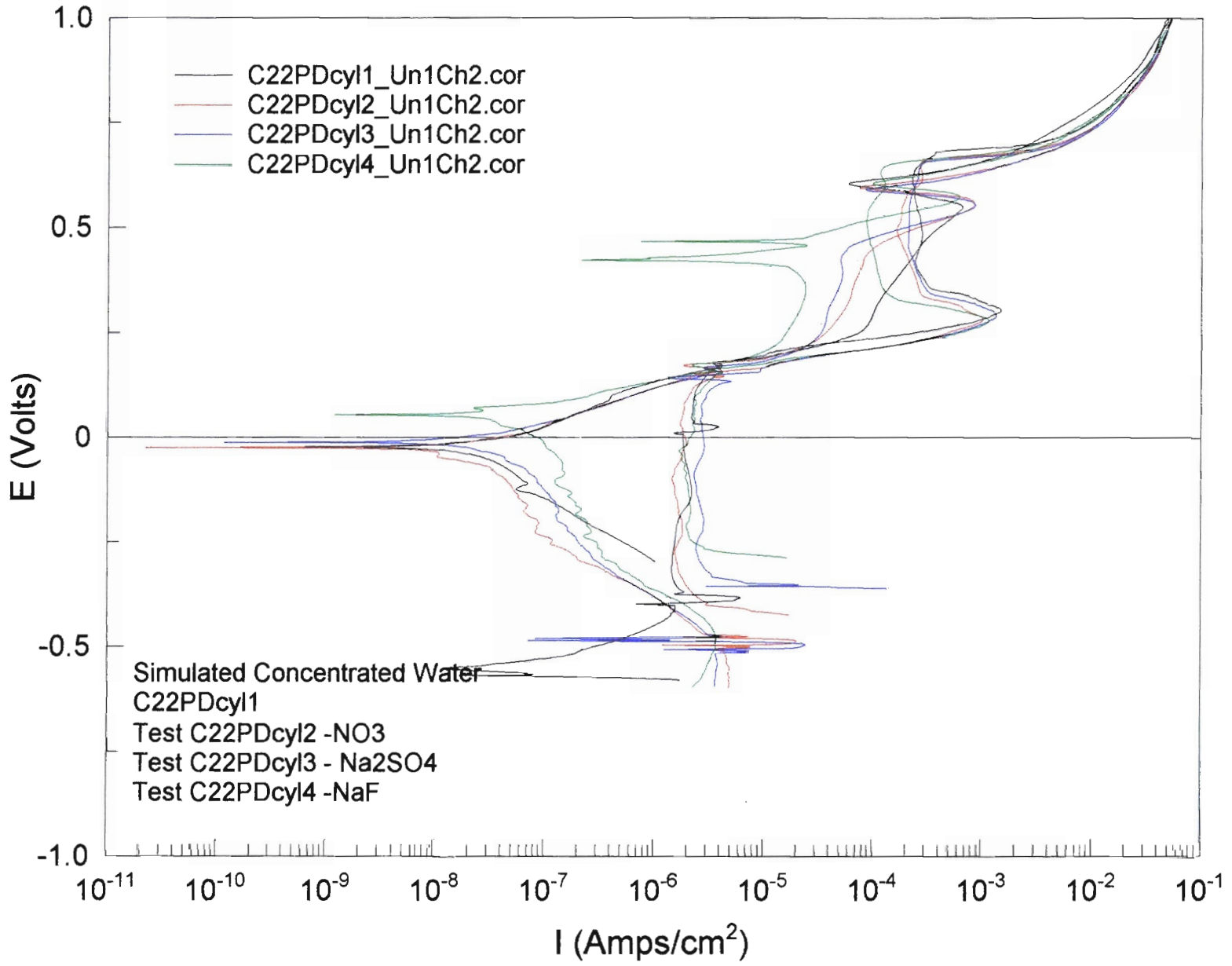


12/11/11
B. S.





6/18/21
S. A.

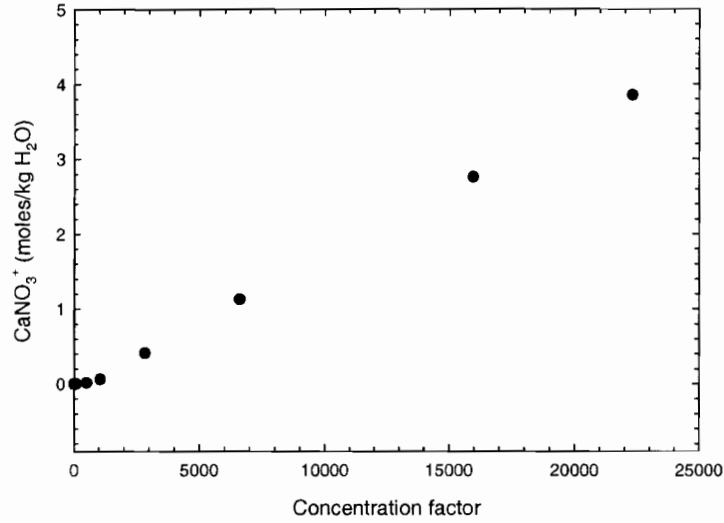


Note: Same graph as on pg #8

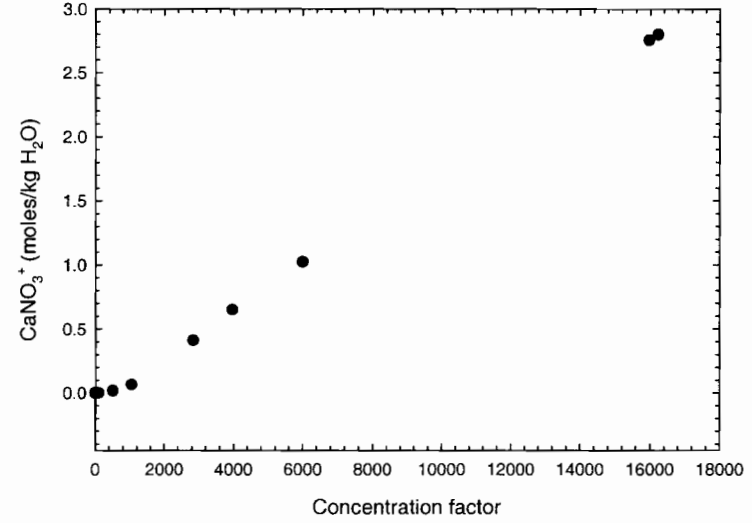
S. P. 12/8/05

Synthetic UZ Porewater

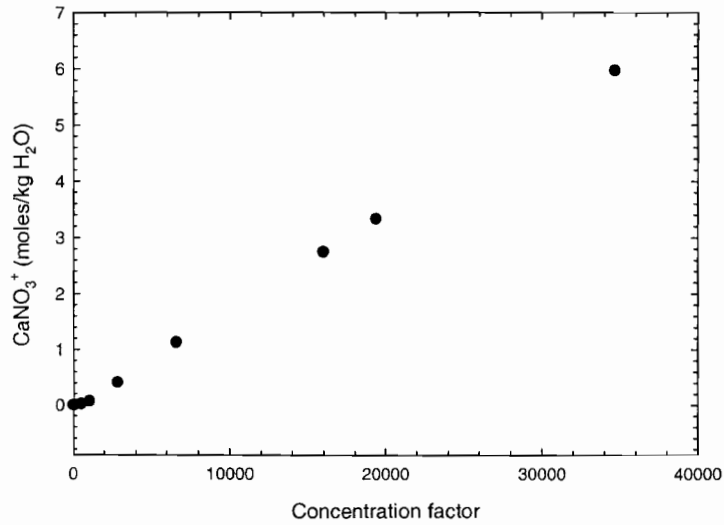
[CaNO₃⁺] vs conc. factor
Temp = 95 °C



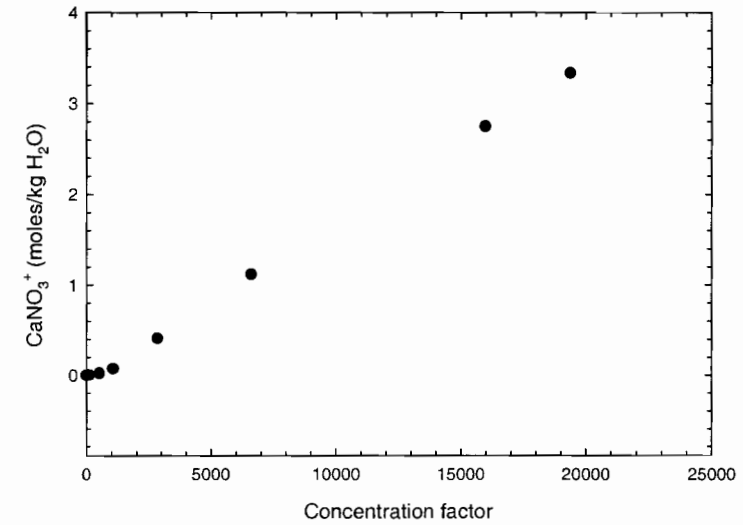
[CaNO₃⁺] vs conc. factor
Temp = 110 °C



[CaNO₃⁺] vs conc. factor
Temp = 125 °C



[CaNO₃⁺] vs conc. factor
Temp = 140 °C



Bill
12/1/05

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-6-3266~~ CNWRA Drawing 20.01402.571.019 polished to a
600 grit finish 2277-3-3266 ^{8/19} 4/27/04Initial Weight: 12.41790g Model: Sartorius Genius SN: 12809099
Final Weight: 12.41468g Cal: 11/14/03 Due: 5/14/04

SOLUTION: Simulated Concentrated water - KCl

10.873g NaCl lot# 034103 192.7g NaHCO₃ lot# 028424
17.540g NaNO₃ lot# 020909 6.194g NaF lot# 991559
41.61g Na₂SO₄ lot# 033451

+ DI water to 2000mls

Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/04 ^{8/19} 7/15/03Initial pH: 8.17 Model: orion SN: S001A
Final pH: 9.24 CAL: 1/9/03 DUE: 1/9/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: C251439

Gas: 99.999% Nitrogen

Ecorr: -624 mV Model: Keithley 614 SN: 0704934
Ept: -464 mV Cal: 6/09/03 Due: 6/09/04

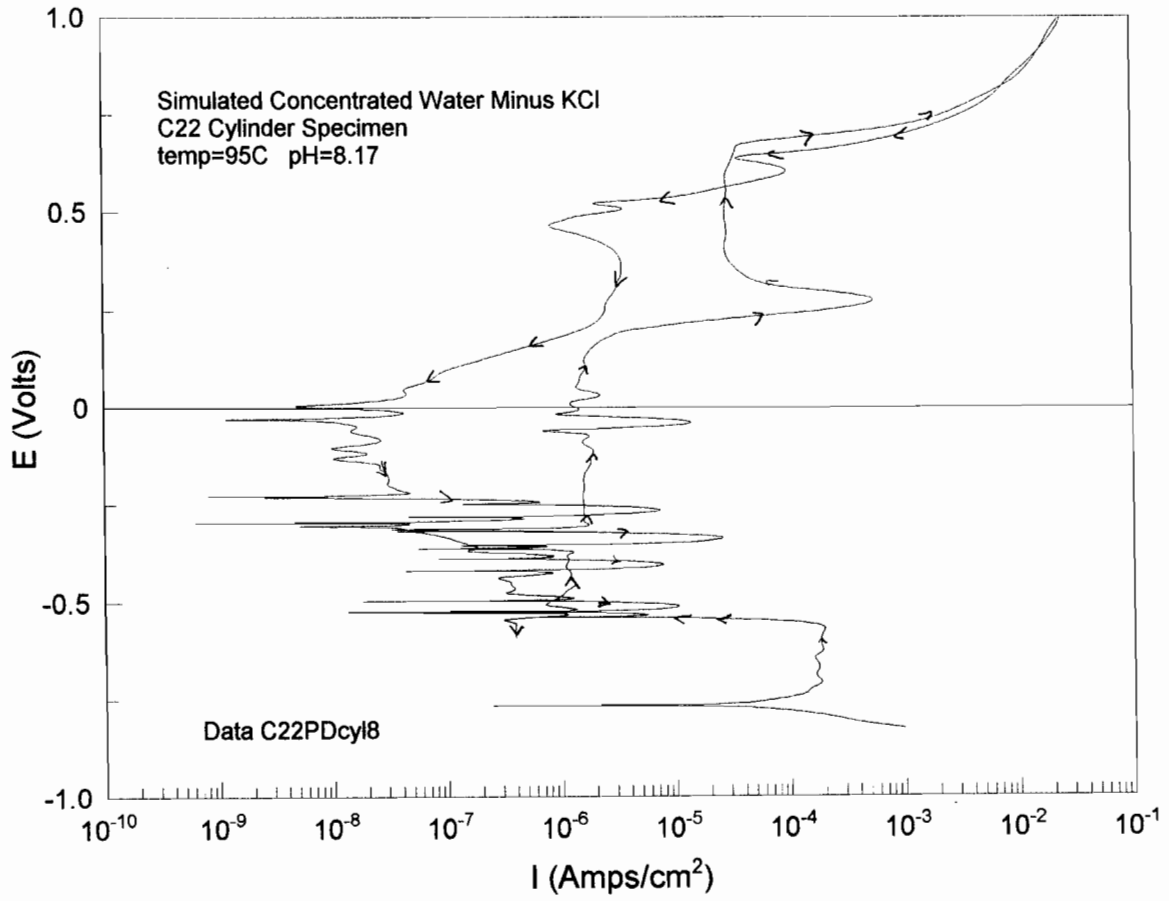
Eapplied (vs SCE): —

Potentiostat: Solartron 1480 SN# 06240551

Specimen Examination: No Visual Signs of Corrosion/Pitting
Surface Has Dull Tint Staining

Date C22ED cyl 8

B. J. 12/1/03



B. P. J.
12/16/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-6-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

2277-3-2266 plus 6/27/06

Initial Weight: 12.41745 Model: Sartorius Genius SN: 12809099
Final Weight: 12.41745 Cal: 11/14/03 Due: 5/14/04

SOLUTION: Simulated Concentrated Water Minus NaCl
12.983g KCl Lot # 006247 192.76g Na₂CO₃ Lot # 028924
17.581g NaNO₃ Lot # 020809 6.201g NaF Lot # 991559
41.63g Na₂SO₄ Lot # 025451
+ DI water to 2000ml

Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/03 ⁸¹⁰ 12/14/03
Initial pH: 7.87 Model: orion SN: S001A
Final pH: 9.37 CAL: 1/9/03 DUE: 1/9/04
pH Probe: #13-620-296 SN: 2291257P6

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -241 mV Model: Keithley 614 SN: 0704934
Ept: -7 mV Cal: 6/09/03 Due: 6/09/04

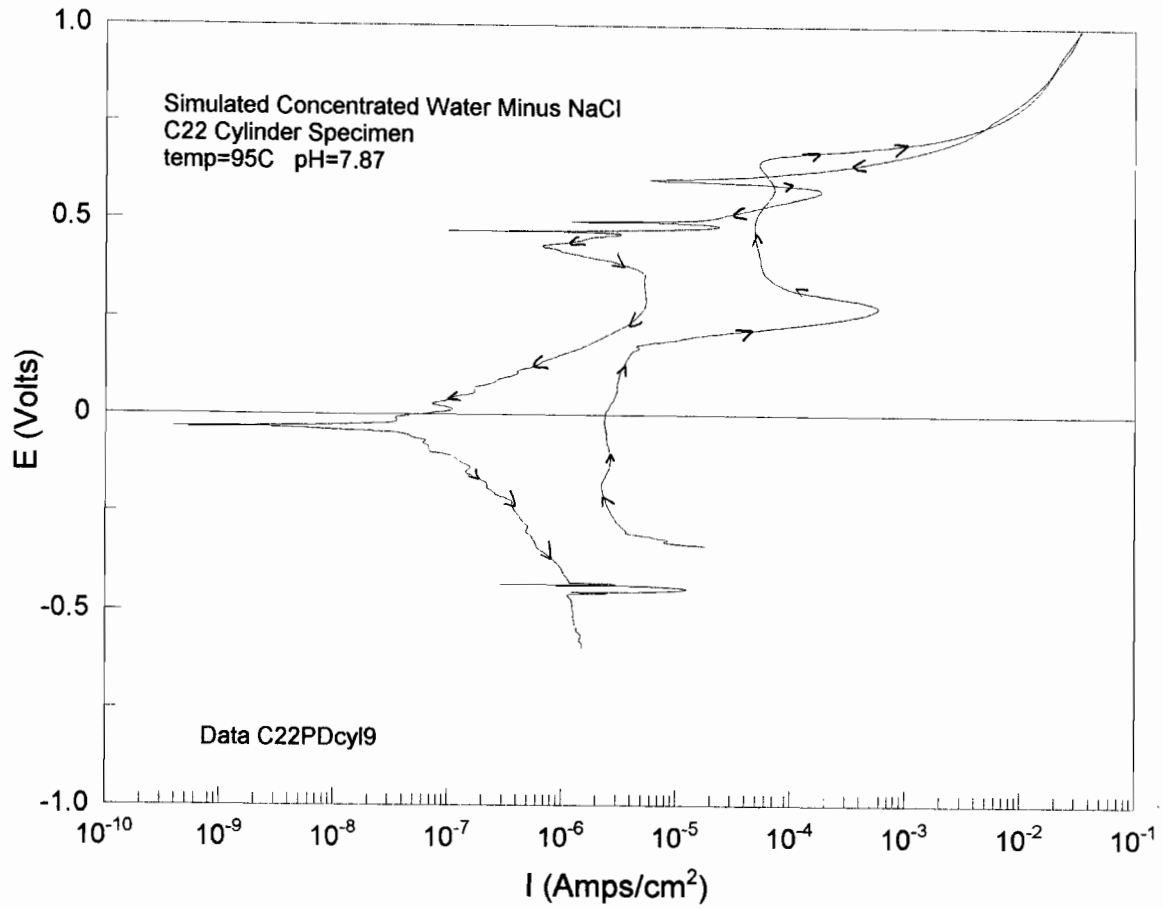
Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No Visual signs of Corrosion/Pitting
Surface Has Dull tint staining

Data C22PD cyl 9

B. [Signature] 12/16/03



B. F. J.
12/17/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# 2277-8-3266 CNWRA Drawing 20.01402.571.019 polished to a
600 grit finish 2277-3-3266 DJJ
6/27/06Initial Weight: 12.26250g Model: Sartorius Genius SN: 12809099
Final Weight: 12.25725g Cal: 11/14/03 Due: 5/14/04SOLUTION: Simulates concentrated water minus Cl^-
17.502g KNO_3 Lot# 026809 192.71g $NaHCO_3$ Lot# 028924
41.39g Na_2SO_4 Lot# 035451 6.193g NaF Lot# 991559
+ DI Water To 2000 mlsReagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/04 8/10/03Initial pH: 7.78 Model: orion SN: S001A
Final pH: 9.26 CAL: 1/9/03 DUE: 1/9/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -224 mV Model: Keithley 614 SN: 0704934

Ept: -118 mV Cal: 6/09/03 Due: 6/09/04

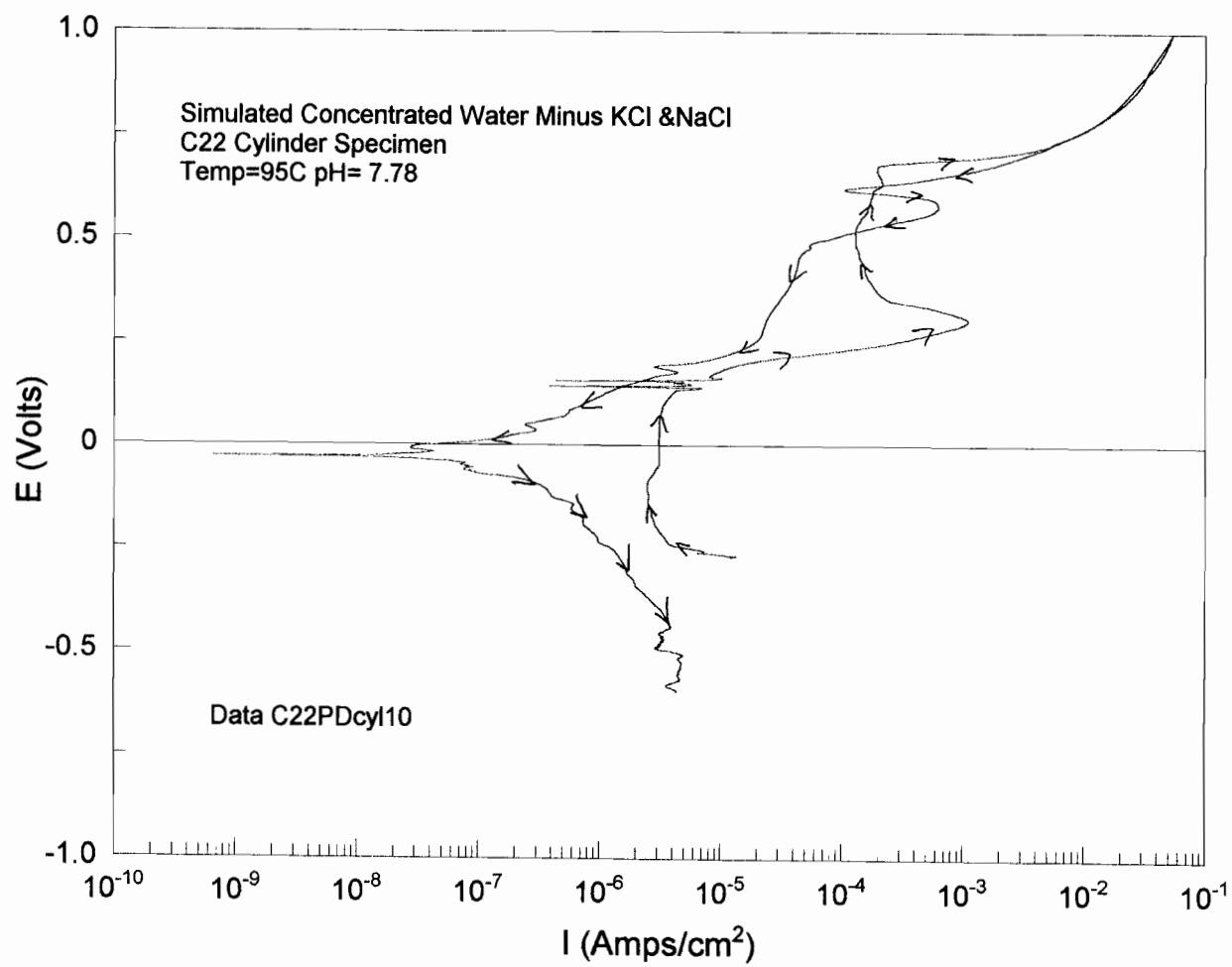
Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No visual signs of corrosion/pitting
Surface has Dull Tint Staining - slight
Build up of Material

Data C22PCyl 10

B. F. J.
12/12/03



B. E. J.
12/18/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish2277-3-3266 ⁸⁶⁰
6/27/04Initial Weight: 12.42197g Model: Sartorius Genius SN: 12809099
Final Weight: 12.42240g Cal: 11/14/03 Due: 5/14/04

SOLUTION: Simulated concentrated water - KCl

10.877g NaCl Lot# 034103 192.69g NaHCO₃ Lot# 028966
17.522g NaNO₃ Lot# 020809 6.192g NaF Lot# 991559
41.39g Na₂SO₄ Lot# 035451Reagents measured with + DI water to 2000mls
Model: OHAUS
Cal: 7/29/03SN: 2883
Due: 1/29/03 04 ^{12/14/03}Initial pH: 7.73
Final pH: 9.18Model: orion
CAL: 1/9/03
pH Probe: #13-620-296SN: S001A
DUE: 1/9/04
SN: 2291257P6

TEST TEMPERATURE:

Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE

13-620-52 SN: 0251435

Gas: 99.999% Nitrogen

Ecorr: -510mV

Model: Keithley 614

SN: 0704934

Ept: -277mV

Cal: 6/09/03

Due: 6/09/04

Eapplied (vs SCE): -

Potentiostat: Solartron 1480

SN# 00240551

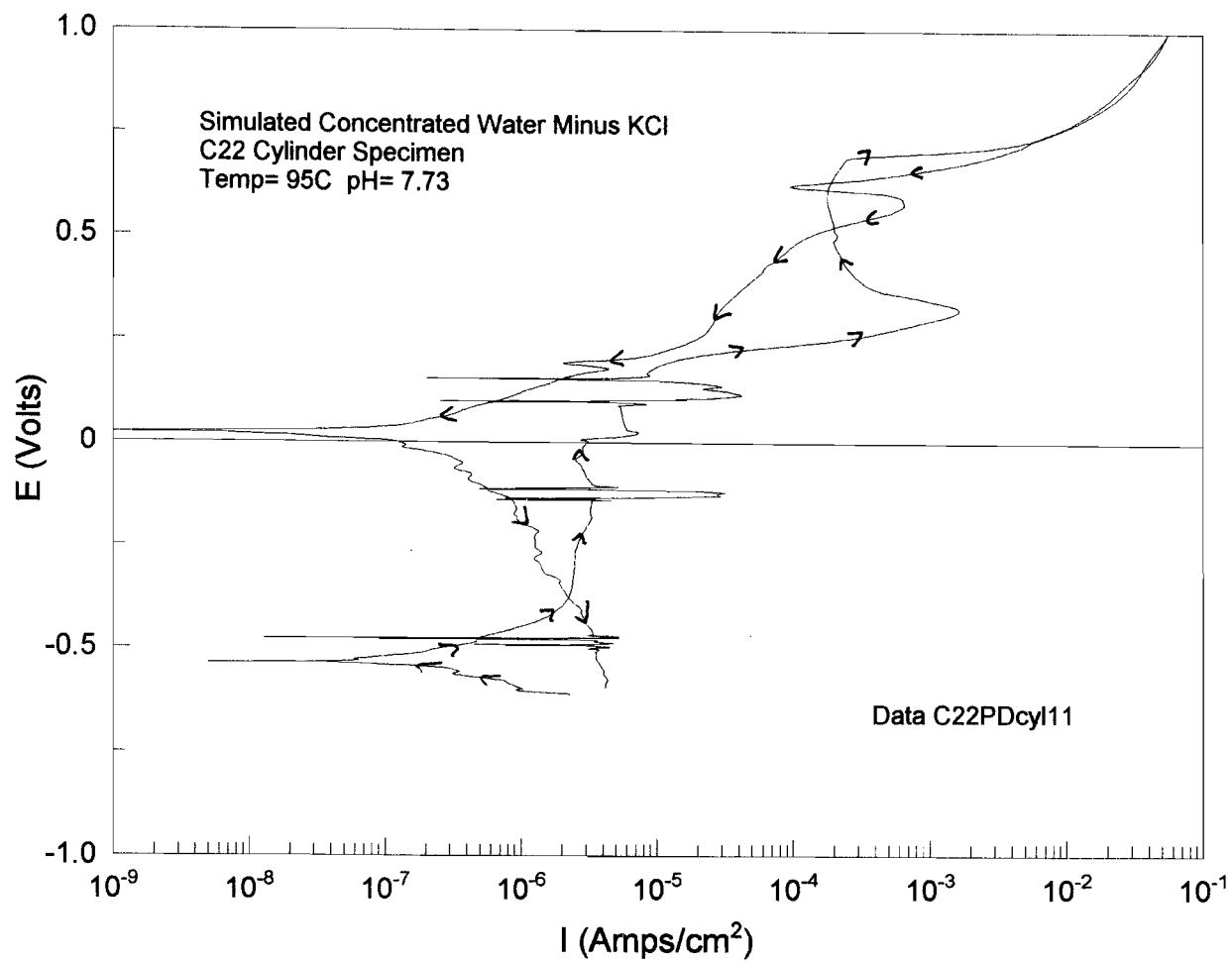
Specimen Examination:

No Visual Signs of Corrosion/pitting
surface has dull tint staining

* Note Rerun of Test C220Cyl18

Data C220Cyl11

B. J. J.
12/19/03



B. K. J.
12/14/03

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# 2277-8-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

2277-3-3266 315
6/27/04Initial Weight: 12.4061g Model: Sartorius Genius SN: 12809099
Final Weight: 12.4095g Cal: 11/14/03 Due: 5/14/04SOLUTION: simulated concentrated water x2 NaF
12.974g KCl lot# 006242 41.42g Na₂SO₄ lot# 035451
10.861g NaCl lot# 035421 192.69g NaHCO₃ lot# 028966
17.562g NaNO₃ lot# 020809 * 12.43g NaF lot# 991559
+ DI water To 2000mlReagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/04Initial pH: 7.69 Model: orion SN: 2230
Final pH: 8.91 Cal: 7/15/03 DUE: 7/15/04
pH Probe: #13-620-296 SN: 2291257P6TEST TEMPERATURE: Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251439

Gas: 99.999% Nitrogen

Ecorr: -550mV Model: Keithley 614 SN: 0704934

Ept: -255mV Cal: 6/09/03 Due: 6/09/04

Applied (vs SCE): -

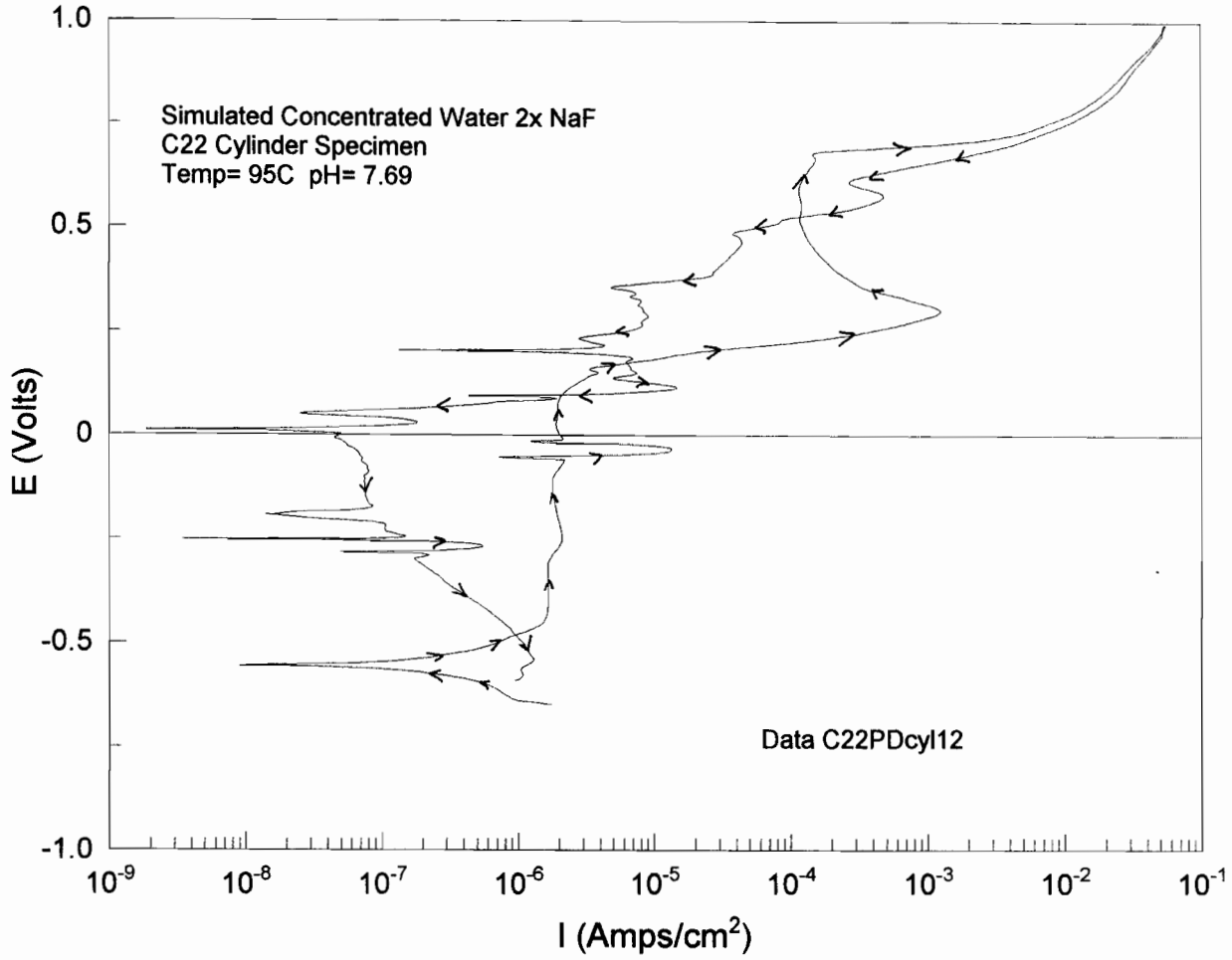
Potentiostat: Solartron 1480 SN# 00240551

Specimen Examination: No Visual signs of Corrosion/Pitting

All surface Area has Dull tint staining

date 02/09/12

B. P. J.
1/13/04 KTC
1/13/04



[Handwritten signature]
1/14/03 1/14/04
KTC 1/14/04

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-6-3266~~ 2277-3-3266 ⁵⁰⁰ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

6/27/04

Initial Weight: 12.3809g Model: Sartorius Genius SN: 12809099
Final Weight: 12.31700g Cal: 11/14/03 Due: 5/14/04

SOLUTION: Simulated Concentrated Water minus $NiNO_3$ X 2 NaF
12.974g KCl Lot# 006242 192.71g $NaHCO_3$ Lot# 028966
10.860g NaCl Lot# 035421 12.44g NaF Lot# 991559
41.41g Na_2SO_4 Lot# 038451
+ DI water To 2000mls

Reagents measured with Model: OHAUS SN: 2883
Cal: 7/29/03 Due: 1/29/04

Initial pH: 7.69 Model: orion SN: 2230
Final pH: 8.93 Cal: 7/15/03 DUE: 7/15/04
pH Probe: #13-620-296 SN: 2291257P6

TEST TEMPERATURE: Measured with Hg Thermometer SN: C96-377
Cal: 7/15/03 Due: 1/15/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 0251459

Gas: 99.999% Nitrogen

Ecorr: -698mV Model: Keithley 614 SN: 0704934
Ept: -383mV Cal: 6/09/03 Due: 6/09/04

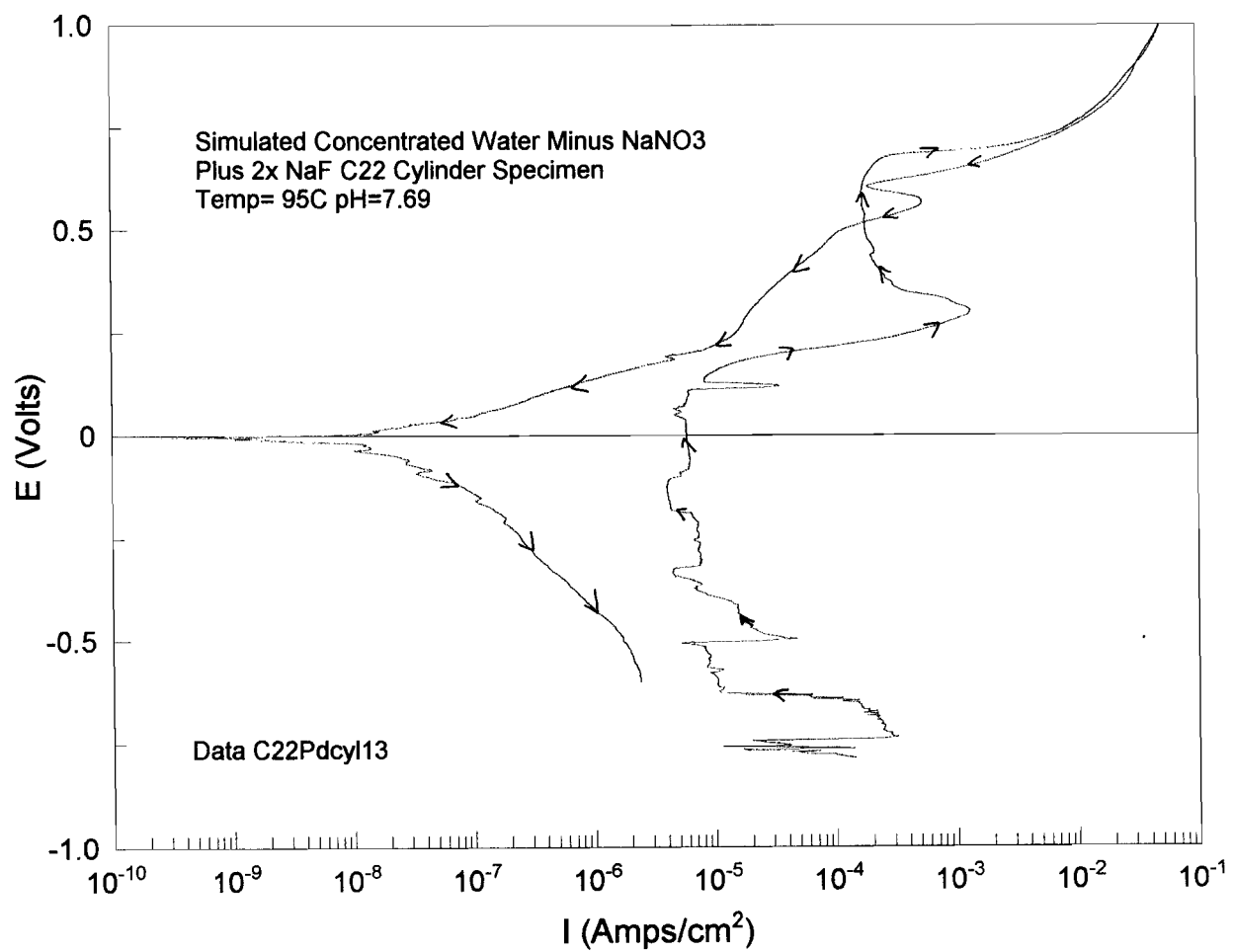
Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00240557

Specimen Examination: No Visual Signs of Corrosion / Pitting
dull tint staining on All Surfaces of Specimen

data C22POC/13

[Signature]
4/14/03 1/14/04 1/14/04 KTC

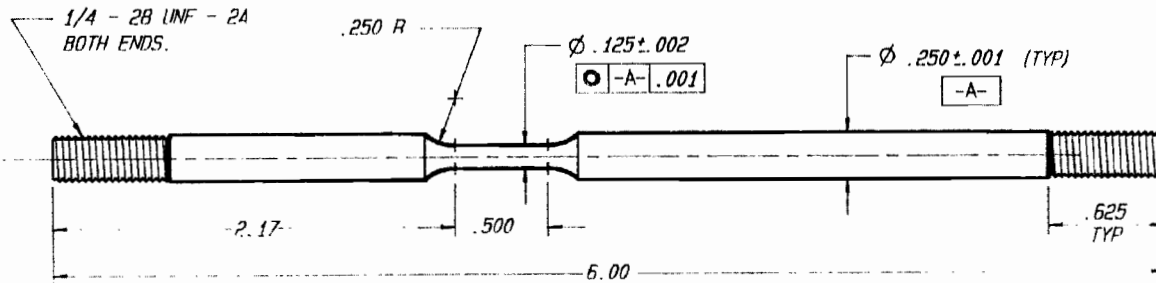


K.J. Chan 1/15/04

Ken Chiang
 SwRI-CNWRA
 Phone: (210) 522-2308
 Fax: (210) 522-5184
 e-mail: Kchiang@swri.org

SwRI DRAWING # 20-03704-042-001

LTH	DESCRIPTION	DATE	APPROVED
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NOTE: 1. DO NOT UNDERCUT RADII
 2. USE LOW STRESS MACHINING PROCEDURE

K.J. Chiang
4/8/04

FIND NO.	QTY REQ.	MATERIAL	CODE IDENT NO.	PART OR IDENTIFYING NO.	NOMENCLATURE OR DESCRIPTION	PARTS LIST		
						DECIMALS	FRAC-TIONS	CONTRACT
		0.5 THICK PLATE SUPPLIED				4-10-1992	A. NAGY	
		16 RMS						
SOUTHWEST RESEARCH INSTITUTE						SLOW STRAIN RATE SPECIMEN		
DRAWING NO. 20-3704-042-1						SCALE 2 = 1		
SHEET						DRAWING NO. 20-3704-042-1		

K.J. Chiang 1-13-04
 Initiator: K. Chiang Date

Praveen Jain for V. Jain
 Reviewer: V. Jain Date 1/13/04

Mark R. Shurtown 1/13/04
 QA Approval: R. Briant Date

**Initial Entry for Corrosion Resistant Material:
Slow Strain Rate Tests:**

Title: Slow Strain Rate Tests

Tests performed by: *Walter Machowski*

Objective: see page 5 of *Notebook #695*

Equipment: *ESC 440 multichannel Potentiostat
w/ National Instruments Labview data acquisition
software. Glass test cell (1500ml) w/ Teflon
top & bottom. Load cell #1 on Frame #5
SWRI Test Frame #5*

Materials: Alloy C-22 mill annealed; *PMAW and BTAW
alloy 22 welded*

Specimens: Cylindrical SSR Specimens; see drawing on page 62
HT # 2277-3-3292 Silica WNA813/XY1977.8611 8/27/06

*see revised initial entry on
p. 70 of Notebook #695*

W. I. Chris
4/26/04

Load Cell Calibration Data

Load Cell 1

<u>Volts</u>	<u>1</u>	<u>2</u>	<u>3</u>	<u>Load Cell 4</u>
0.463	500	496	495	507
0.893	998	996	994	998
1.325	1500	1497	1500	1508
1.755	2000	1995	1999	2005
2.185	2499	2499	2500	2505
2.614	2999	2999	2996	2997
3.041	3501	3495	3497	3492
3.410	3995	3992	3993	3983

LVDT Calibration

<u>Gage ID</u>	<u>Value</u>	<u>1</u>	<u>2</u>	<u>3</u>
-----	0	1.000	1.000	1.000
X369A	0.0625	0.938	0.938	0.938
Y242A	0.1251	0.875	0.875	0.875
Y455A	0.2500	0.749	0.749	0.749
Y649A	0.5001	0.500	0.500	0.500
Y189B	1.000	0.000	0.000	0.000

*All readings in inches

K. J. Chis
4/26/04

SLOW STRAIN RATE TESTObjective: see page #5 *NOTEBOOK #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

*Heat # ~~2277-8-3266~~ 2277-3-3266 ⁸⁷ 4/27/04*Solution: *N/A specimen run in air*

Reagents measured with

Model:

SN:

Cal:

Due:

Counter Electrode:

Reference Electrode:

Gas:

Ecorr:

Eapplied:

Potentiostat:

SN:

Specimen Visual:

*metallic sheen - ~~no~~ discoloration
looks like normal ductile failure*

$$e^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

data file: SSRMA22 - air

Dante J. Macfarlane 4/26/2004

SLOW STRAIN RATE TEST

SSRMA_SCW01

Objective: see page #5

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 2277-3-3266 ⁰⁹ 6/27/06
Heat # ~~2275-8-3266~~Solution: Simulated Concentrated Water ^{w/DF} x2 liters

12.956	KCl	Lot # 006242	NaHCO ₃	192.684	# 028966
10.862	NaCl	# 035421	NaF	6.185	# 006679
17.503	NaNO ₃	# 897674	pH	7.76	adj to 8.90 w/ NaOH
41.384	Na ₂ SO ₄	# 035451			

Reagents measured with

Model: OHAUS
Cal: 2 FEB 04SN: 2883
Due: 2 AUG 04

Counter Electrode: Pt flag

Reference Electrode: Ag/AgCl

Gas: 99.999 N₂

Ecorr: -196 mV in house w/ 3M KCl

Eapplied: +415 mV

Potentiostat: FSC 440-2 SN: 9209138

Specimen Visual:

no discoloration - many crack initiation sites; looks like brittle failure

post test pH 9.70 $E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$ * Temp offset between RT and 95°C $\Delta 13 \text{ mV}$ data file: SSRMA_SCW01
SSRMA22_SCW01

Natalie J. Markowski 4/26/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Heat # ~~2277-8-5266~~ 2277-3-3266 *SK2/27/06*

Solution:

*same solution as p.66
SCW
(used remaining 1 liter)*

Reagents measured with

Model: *OHAUS*
Cal: *2 FEB 04*SN: *2883*
Due: *2 AUG 04*Counter Electrode: *Pt Flag*

Reference Electrode:

*As/AsCl
in house w/ 3M KCl*Gas: *99.999 N₂*Ecorr: *-230 mV*Eapplied: ** +415 mV*Potentiostat: *ESC 440-2* SN: *9209138*

Specimen Visual:

*very slight "gold-bronze"
discoloration; many crack initiation sites;
looks like brittle fracture*

$$i^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

final pH *9.86*** Temp offset between RT + 95°C Δ 13 mV*data file: *SSRMA22-SCW02**Walter J. MacKroschi 4/
5/11/04*

SLOW STRAIN RATE TEST

Objective: see page #5

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Heat # ~~2297-8-3266~~ 22-77-3-3266 810 6/27/01

Solution: X 1 liter w/DF
 96.339 NaHCO₃ # 991559
 6.479 KCl Lot # 005573
 5.438 NaCl Na₂SO₄ # 035421
 20.699 ~~NaHCO₃~~ w/10M # 028964
 3.088 NaF w/10M # 991559

SCW mins NO₃
 7.34
 pH_{ini} 8.73
 pH_{final} 9.02

Reagents measured with Model: OHAUS SN: 2883
 Cal: 2 FEB 04 Due: 2 AUG 04

Counter Electrode: Pt Flag

Reference Electrode: Ag/AgCl
 in home w/3M KClGas 99.995 N₂

Ecorr: -403 mV

* Applied: +415 mV

Potentiostat: ESC 440-2 SN: 9209138

Specimen Visual:

slightly dull color; still
 metallic grey-silver; many crack initiation
 sites; looks like brittle fracture

$$\epsilon' = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* Temp offset between RT and 95°C $\Delta 13 \text{ mV}$ pH adj from 7.34 \rightarrow 8.73 w/10M NaOH

data fill: SSRMA22-SCW03

Walter J. Meehan 5/12/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook # 695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

SCW *minim say* = *Heat # 2277-8-3266* *810*
Solution: *x liter w/DE* *2277-3-3266* *6/27/06*6.480g KCl *Lot # 105573* 96.341g NaHCO₃ #028966
5.433g NaCl #035421 3.109g NaF #006679
8.749g NaNO₃ #020809 pH_{ini} 7.45 *adj* → 8.73
pH_{final} 9.99

Reagents measured with

Model: OHAUS
Cal: 2 FEB 04SN: 2883
Due: 2 AUG 04Counter Electrode: *PT flag*Reference Electrode: *Ag/AgCl*
*in buret w/ 3M KCl*Gas: *99.999 N₂*Ecorr: *-142 mV** Applied: *+415 mV*Potentiostat: *KSC 440-2* SN: *920 913 8*

Specimen Visual:

slightly dull color; still metallic
grey-silver; many crack initiation sites;
look like brittle fracture

$$\epsilon^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* Temp offset between RT and 95°C $\Delta 13 \text{ mV}$
*pH adjusted w/ 10M NaOH*data file: *SSRMA22-SCW04**Dan J. Mackowski* 5/18/04

SLOW STRAIN RATE TEST

Objective: see page #5

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: x 1 liter DI

Heat # ~~2277-8-3266~~
2277-3-3266 810
6/27/04

SCW minus NaCl

6.46 g KCl

Lot # 005673

96.349 g NaHCO₃8.758 NaNO₃

020809

028966

20.697 Na₂SO₄

035451

3.098 NaF

006679

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 2 FEB 04

Due: 2 AUG 04

Counter Electrode: Pt flag

Reference Electrode: Ag/AgCl

Gas: N₂ (99.999%)

Ecorr: -167 mV

in house w/ 3M KCl

* Applied: +415 mV

Potentiostat: ESC 440-2

SN: 9209138

Specimen Visual:

metallic-grey silver; crack
initiation sites; looks like brittle fracture

$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

$$\text{pH} = 7.38 \rightarrow 8.81$$

final 9.98

* temp difference between RT and 95 °C Δ13 mV

data file: SSRMA22-SCW06

Walter J. Markowski
5/24/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *x 1 liter DI SCW minus F⁻*
Heat # *2277-8-3266* 22.77-3-3266 BMD 6/27/06

6.460 g KCl #005573	20.693 Na ₂ SO ₄ 035451
5.340 NaCl 035421	96.352 NaHCO ₃ 028966
8.754 NaNO ₃ 020809	

Reagents measured with

Model: *DTAUS*SN: *2883*Cal: *2 FEB 04*Due: *2 AUG 04*Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl*Gas: *N₂ 99.999*Ecorr: *-335mV**in house w/3M KCl** Applied: *+415mV*Potentiostat: *FSC 440-2*SN: *9209138*

Specimen Visual:

metallic grey color; many crack initiation sites; look like brittle fracture

$$e^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

pH *7.37* *adi* \rightarrow *8.72* 10.00 Serial* Temp difference between RT and 95°C Δ *13mV*data file: *SSRMA22-SCW05**K.T. Ching 5/28/04*

SLOW STRAIN RATE TEST

Objective: see page #5

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: * 1 liter DI

Heat # ~~2277-8-3266~~ ^{8/20/06}
2277-3-3266 6/27/06
3.8M NaCl + 0.38M NaNO₃44.243 g NaNO₃ lot # 020809
222.27 g NaCl # 035421

Reagents measured with

Model: OHAUS
Cal: 2 FEB 04SN: 2883
Due: 2 Aug 04

Counter Electrode: Pt flag

Reference Electrode: Ag/AgCl
in house w/ 3M KClGas: N₂ (99.999) Ecorr: -311 mV

* Applied: +415 mV Potentiostat: ESC 440-2 SN: 9209138

Specimen Visual:

metallic grey-silver; very slight
staining; looks like ductile fracture

$$i^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

pH = 4.10 → 9.80
final 10.00

* temp difference between RT and 95°C Δ13 mV

data file: SSRMA22-SCW 07

Walter J. MacKowski
6/1/2004

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: \times 1 liter DI *Heat # ~~2277-8-3266~~ 2277-3-3266 8/20/06 6/27/06*
3.8M NaCl + 0.38M NaNO₃*same solution as p. 72
(used on SSRMA22 - SCW07)
(used remaining 500 ml)*

Reagents measured with

Model: *ORAU5*SN: *2883*Cal: *2 FEB 04*Due: *2 AUG 04*

Counter Electrode:

Reference Electrode:

Gas:

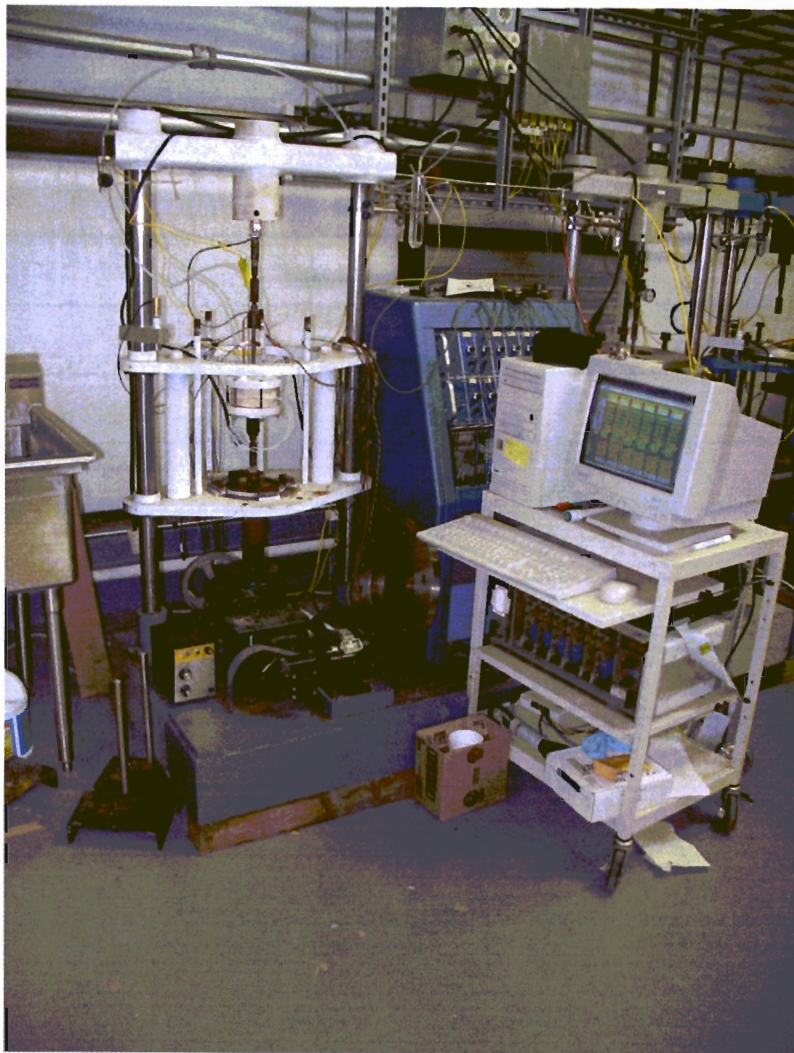
Ecorr: *-283 mV** Applied: *+215 mV* Potentiostat: *ESC440-2* SN: *9209138*

Specimen Visual:

metallic grey-silver; very slight staining; looks like ductile fracture

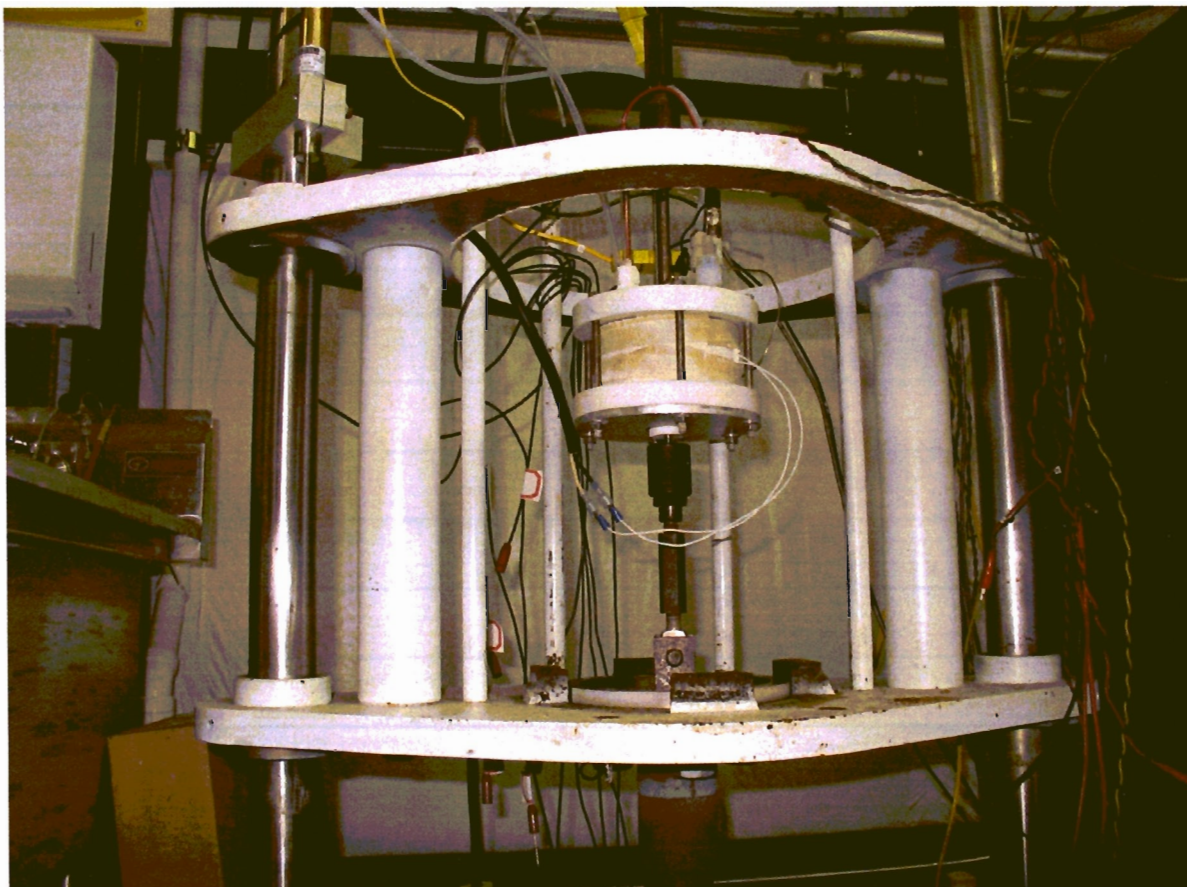
$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

*pH = 4.10 \rightarrow 4.80
final 10.09** temp difference between RT and 95°C Δ 13 mVdata file: *SSRMA22 - SCW08**Walter J. Macroski 6/8/2004*



Test frame with test cell and peripheral
data acquisition equipment

Walter J. Macfarlane
6/11/04



Close-up of test cell mounted in
test frame.

Walter J. Marchinski
6/11/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *Heat # ~~2277-8-3266~~ 2277-3-3266 8/27/04*
x 1 liter DI 3.8M NaCl + 0.38M NaNO₃44.290 g NaNO₃ Lot # 020809

222.300 g NaCl # 035421

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 2 FEB 04

Due: 6 AUG 04

Counter Electrode: *PE flag*Reference Electrode: *Ag/AgCl*Gas: *N₂ (99.999%)*Ecorr: *-283 mV**in hood w/ 3M KCl** Applied: *+415 mV*Potentiostat: *KSC 440-2* SN: *9209138*

Specimen Visual:

metallic grey-silver; looks like ductile fracture

$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

*pH 8.18
not adjusted
final pH 6.77*** temp difference between RT and 95°C $\Delta 13 \text{ mV}$* *data file: SSRMA22-SCW09**Walter J. Macchiarini
6/14/04*

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Neat # ~~2277-8-3266~~ 2277-3-3266 6/27/06

Solution:

*same solution mixed for SCW09
on p. 76*

Reagents measured with

Model: *OTDAUS*SN: *2883*Cal: *2 FEB 04*Due: *6 Aug 04*Counter Electrode: *PT flay*Reference Electrode: *Ag/AgCl*Gas: *N₂ (99.999)*Ecorr: *-297**413m KCl*Eapplied: *+215mV*Potentiostat: *ESC 440-2*SN: *9209138*

Specimen Visual:

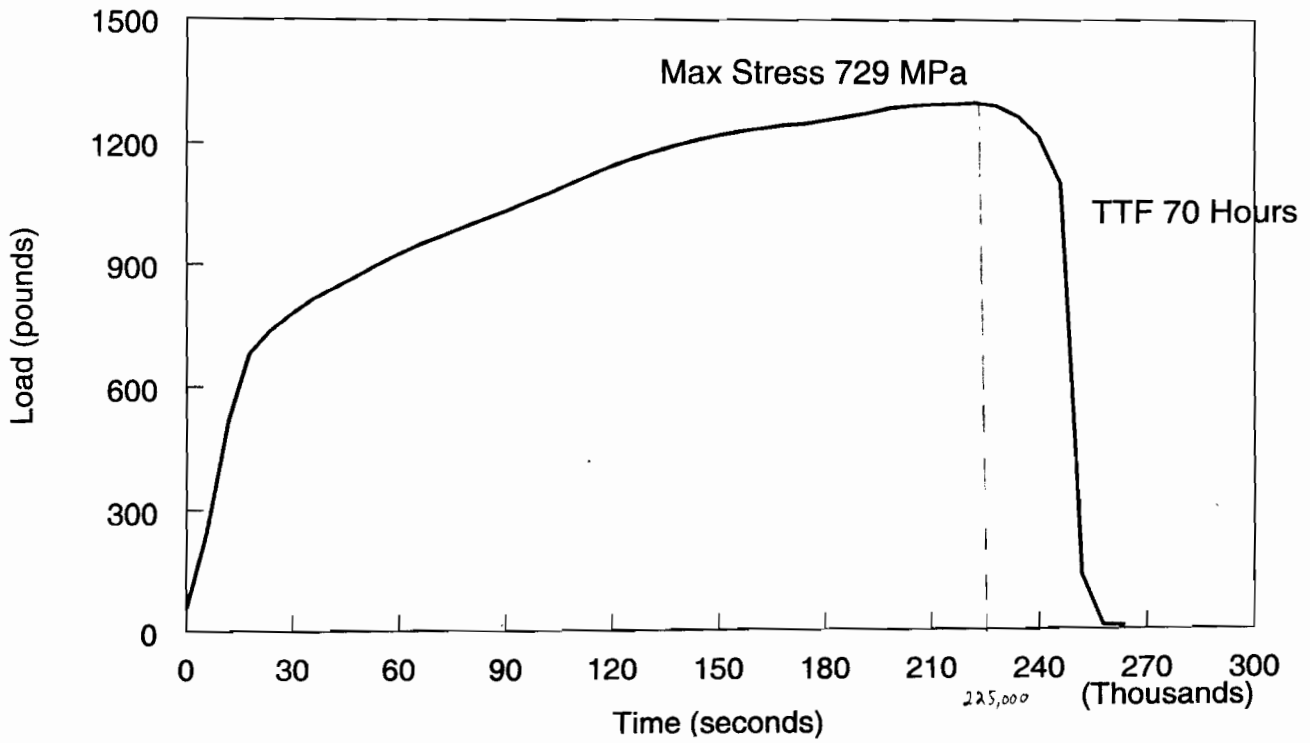
*metallic grey-silver; looks like
ductile fracture*

$$E^{\circ} = 3.2 \times 10^{-6} \text{ S}^{-1}$$

** temp difference RT and 95°C $\Delta 13\text{mV}$* *pH 8.15
not adjusted**final pH 7.49**data file: SSRMA22 - SCW10**Walter J. Macherhi
6/21/04*

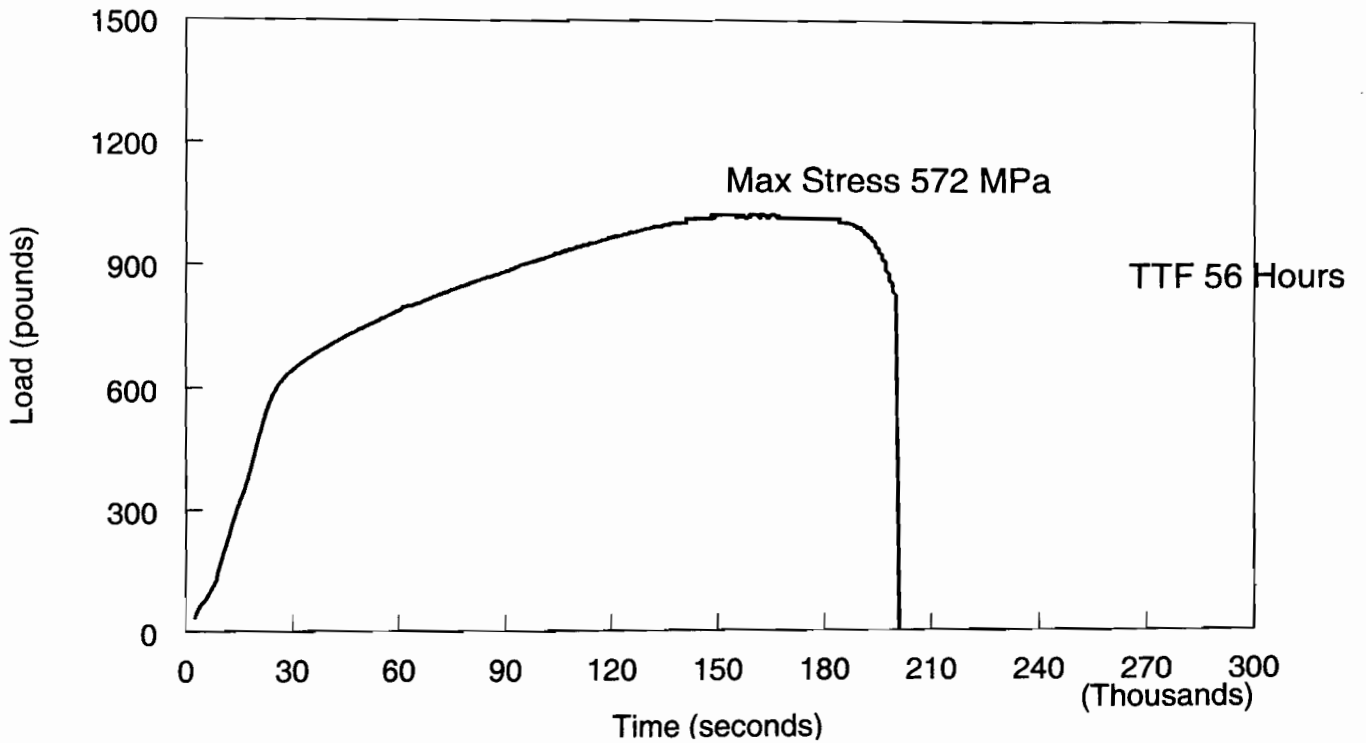
Slow Strain Rate test Ambient Air

SSRMA22_air



Simulated Concentrated Water

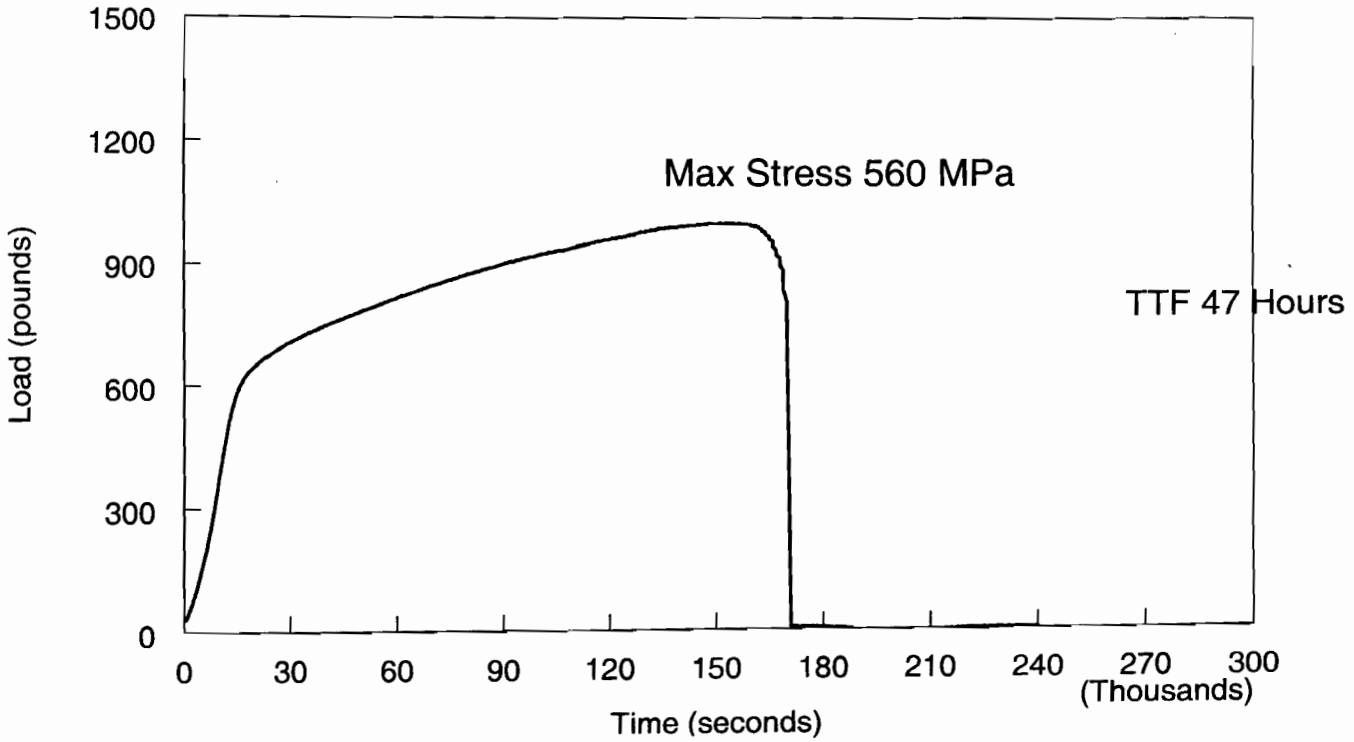
Test SSRMA22_SCW01



Walter J. Macchiarini 6/21/04

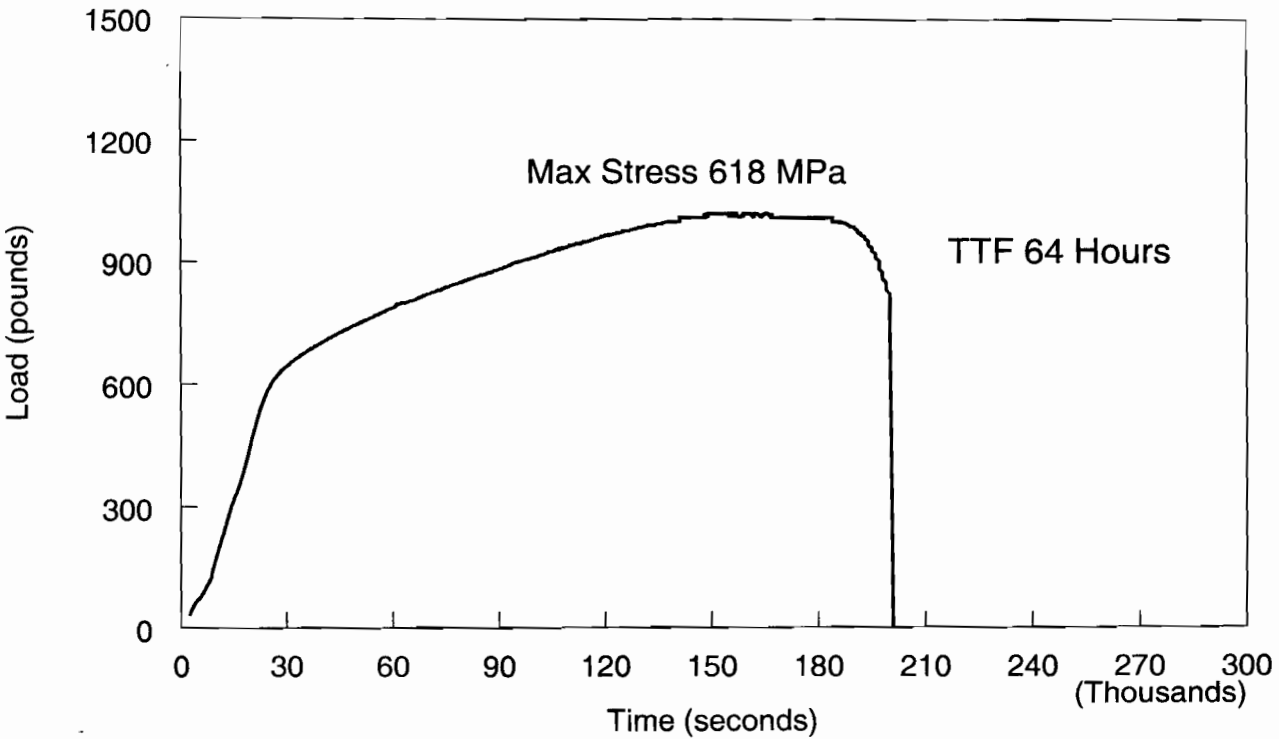
Slow Strain Rate test Simulated Concentrated Water

Test SSRMA22_SCW02



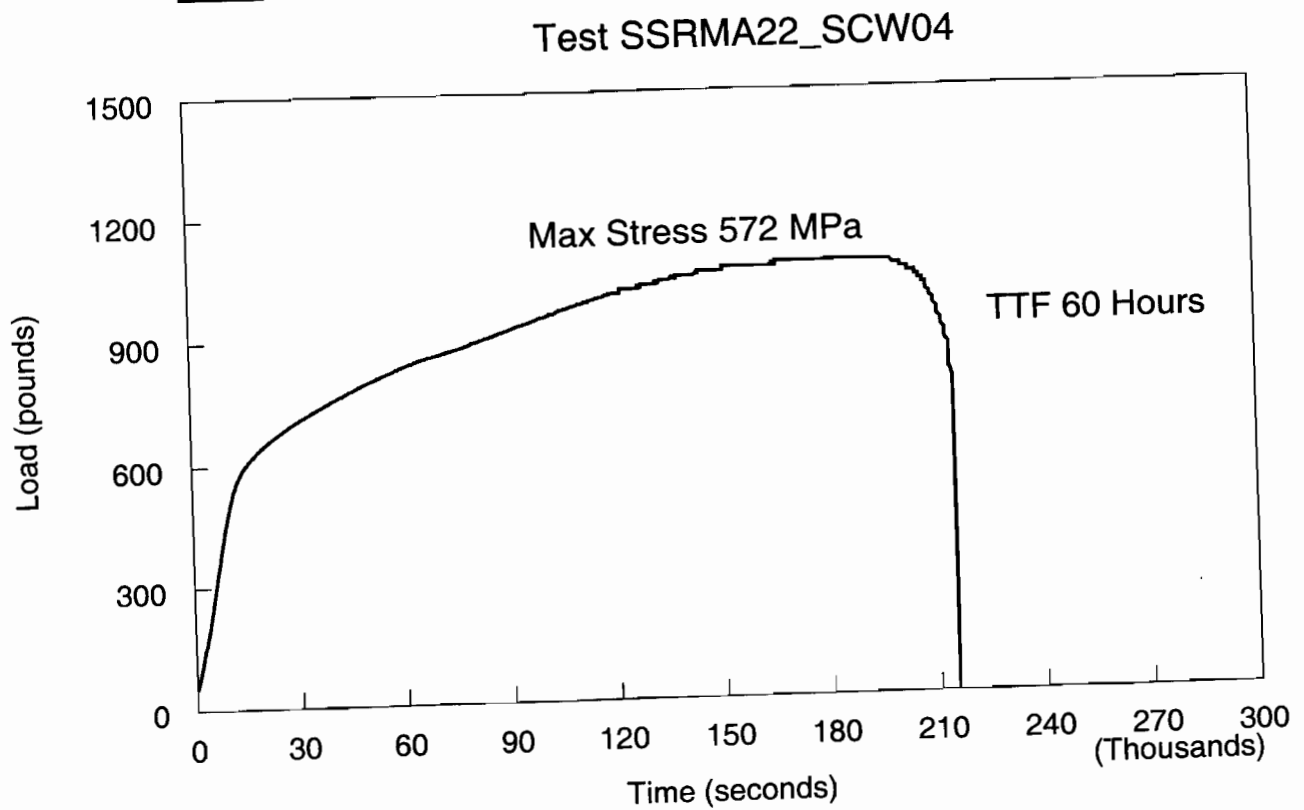
Simulated Concentrated Water - NO3

Test SSRMA22_SCW03

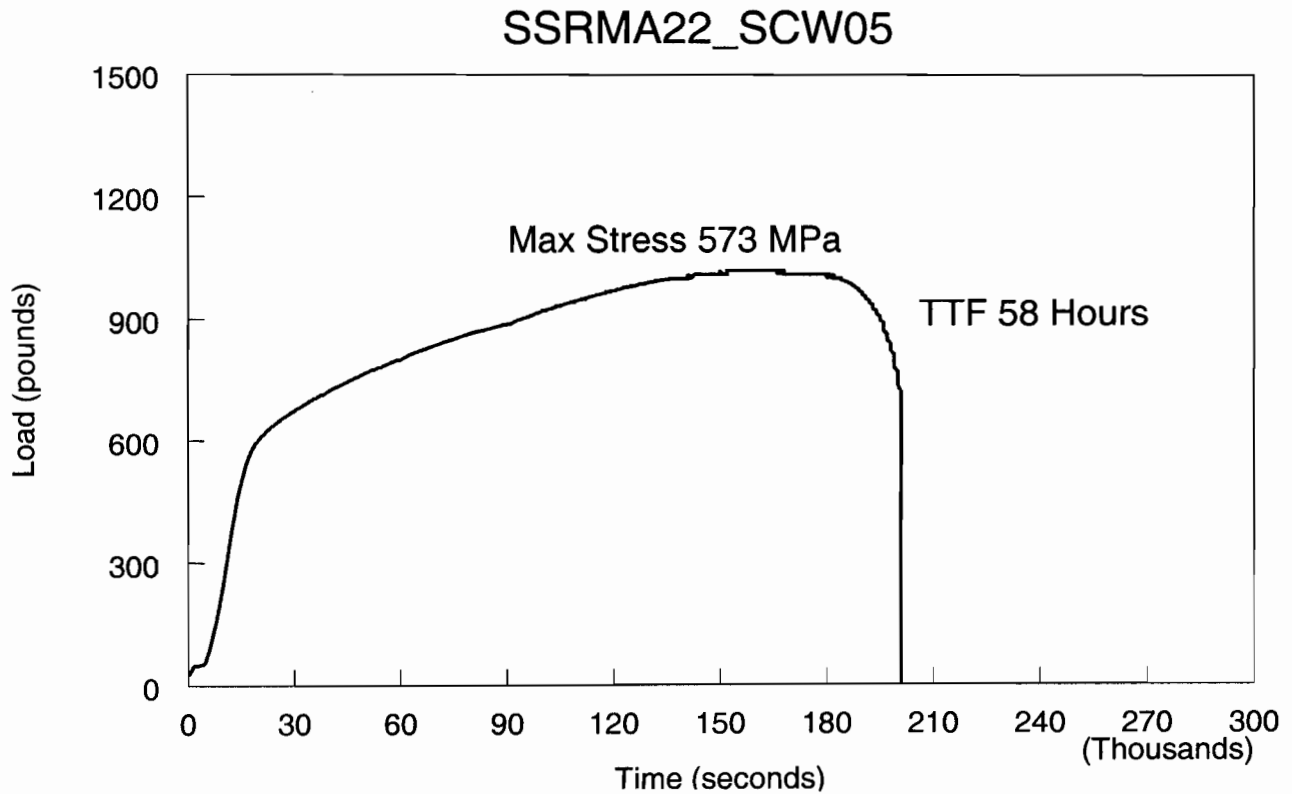


Walter J. MacKooli 6/21/04

Slow Strain Rate test Simulated Concentrated Water - SO4



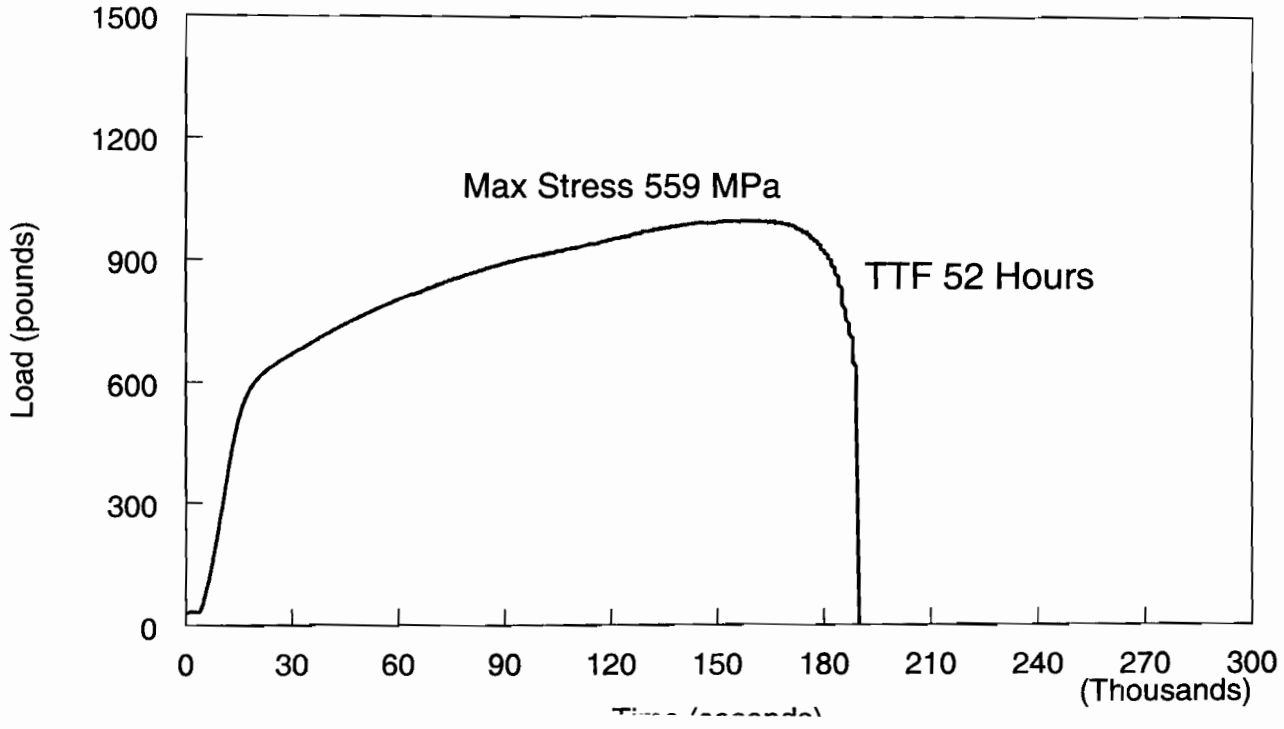
Simulated Concentrated Water - F



Walter J. Macintosh 6/21/04

Slow Strain Rate Test Simulated Concentrated Water - NaCl

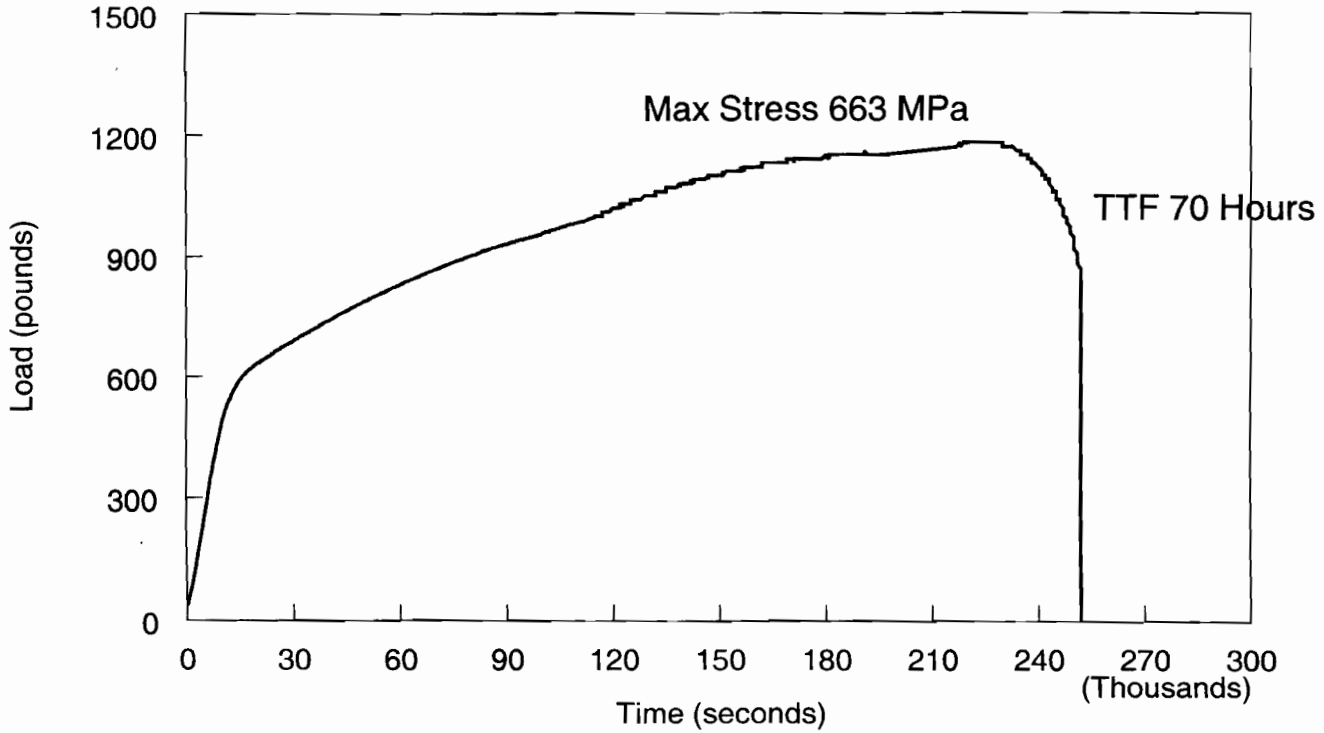
SSRMA22_SCW06



3.8M NaCl+0.38M NaNO3

+400mV v. Ag/AgCl

SSRMA22_07



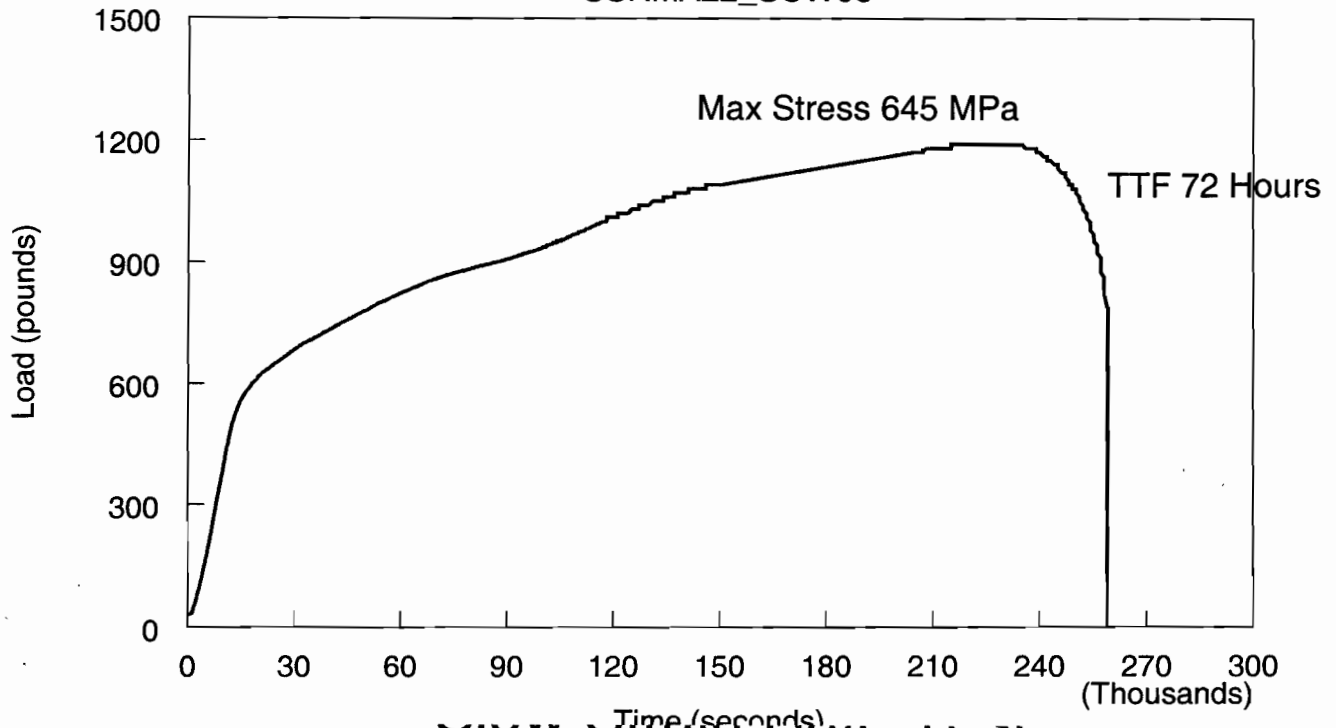
Walter J. Macfarlane 6/21/04

Slow Strain Rate Test

3.8M NaCl+0.38M NaNO3

+200 mV v. Ag/AgCl

SSRMA22_SCW08

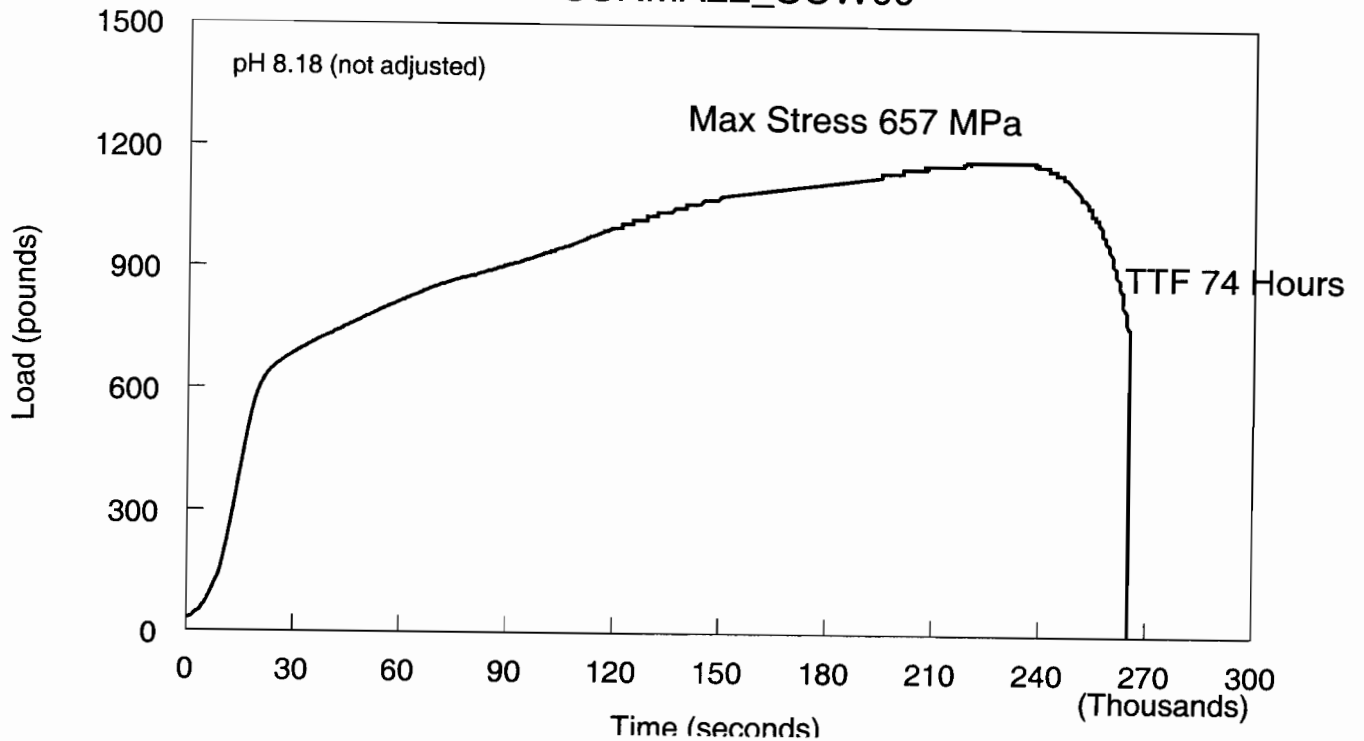


Slow Strain Rate Test

3.8M NaCl+0.38M NaNO3

+400mV v. Ag/AgCl

SSRMA22_SCW09



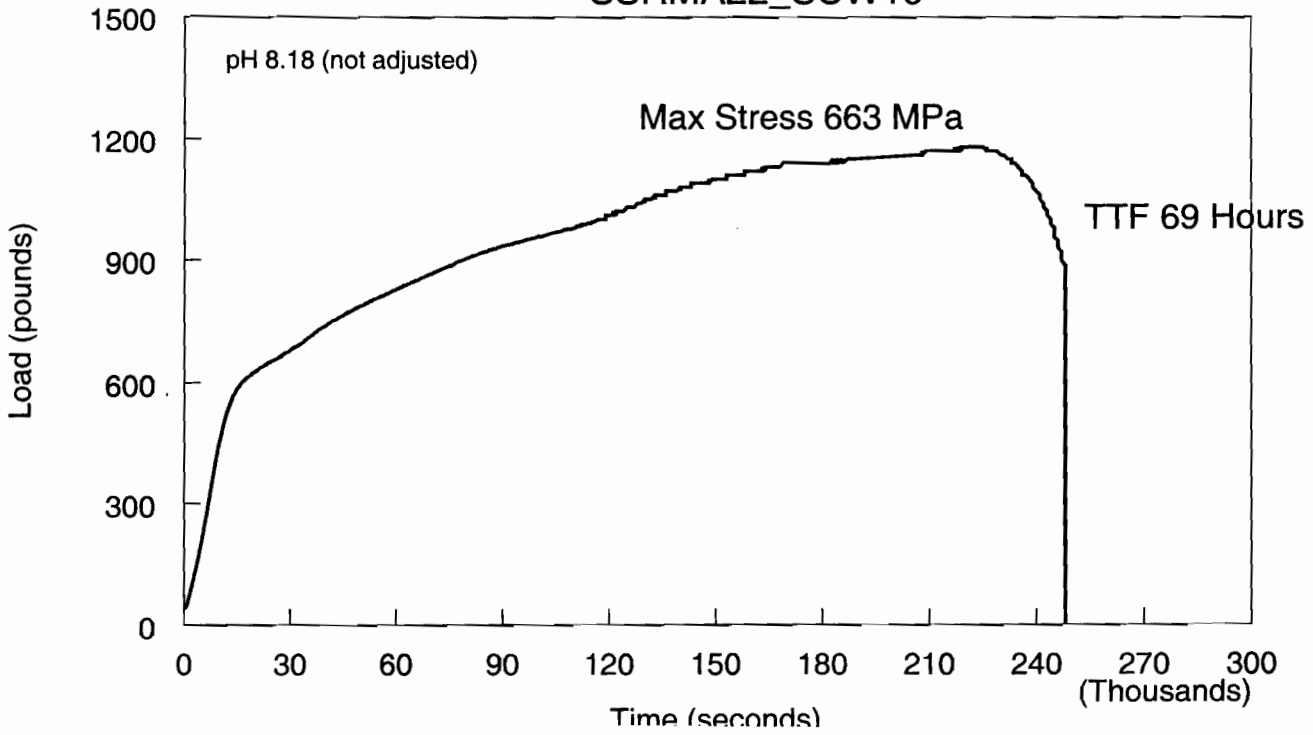
Walter J. Mackowski 6/21/04

Slow Strain Rate Test

3.8M NaCl+0.38M NaNO3

+200 mV v. Ag/AgCl

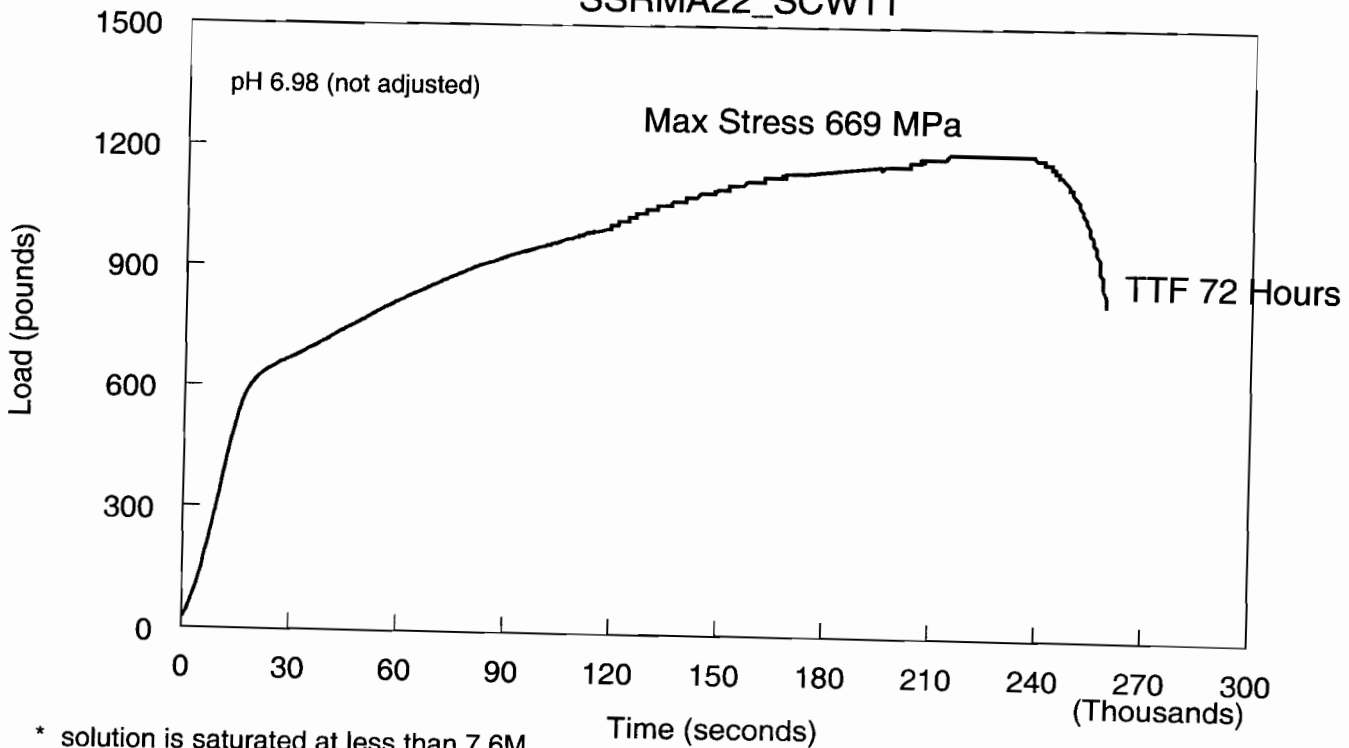
SSRMA22_SCW10



7.6M* NaCl+0.38M NaNO3

+400 mV v. Ag/AgCl

SSRMA22_SCW11



* solution is saturated at less than 7.6M

Walter J. Macchowski 6/30/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

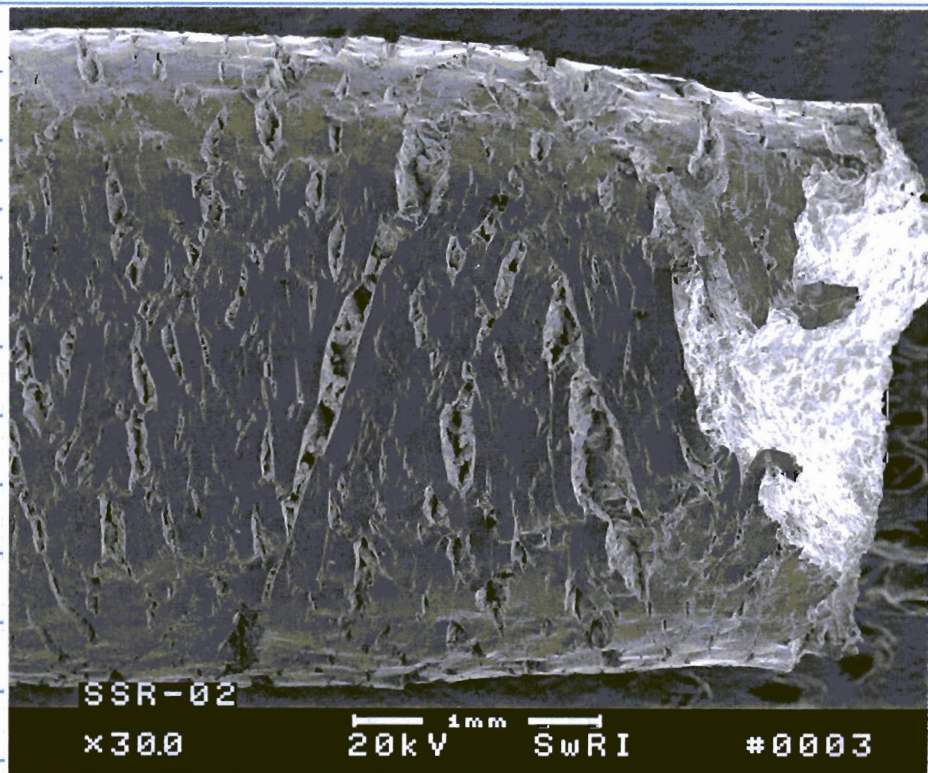
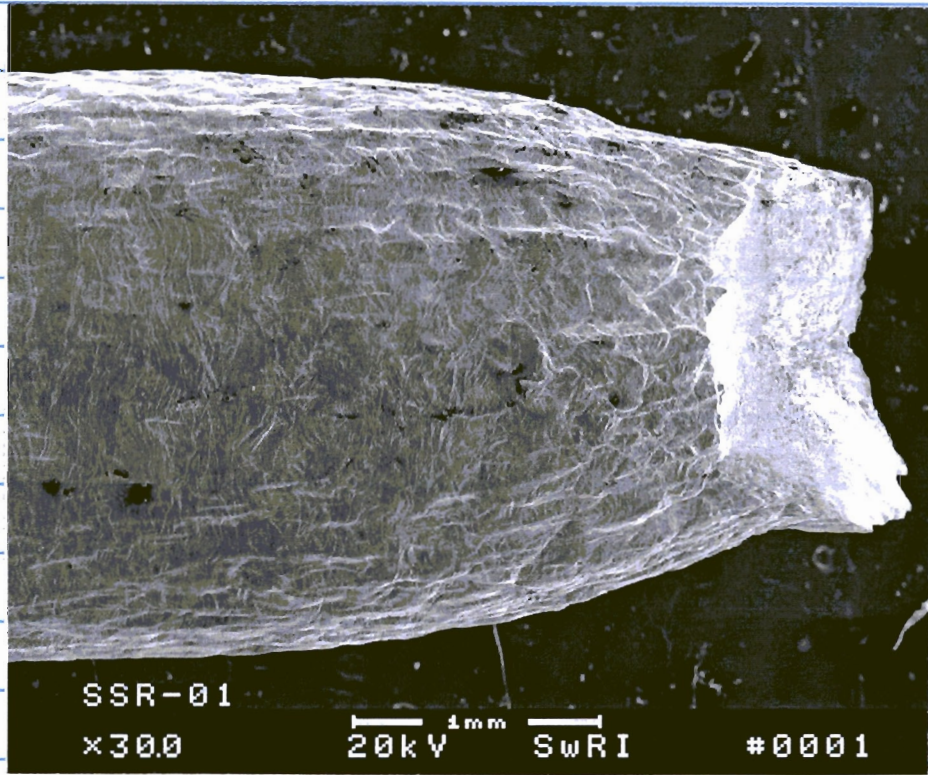
Solution: $\times \frac{1}{2}$ liter *Neat #2277-8-3266 6/22/06*
± 2277-3-3266
7.6 M NaCl + 0.38 M NaNO₃222.39 g NaCl lot # 035421 pH 6.98
32.680 NaNO₃ # 020809 pH 7.22Reagents measured with Model: *ONARA 5* SN: *2883*
Cal: *2 FEB 04* Due: *6 AUG 04*Counter Electrode: *PT flag* Reference Electrode: *H₂/PtCl*
in home w/ 3M HCl
Gas: *N₂ (99.99%)* Ecorr:Eapplied: *** +415 mV* Potentiostat: *GSC 440-2* SN: *9209138*

Specimen Visual:

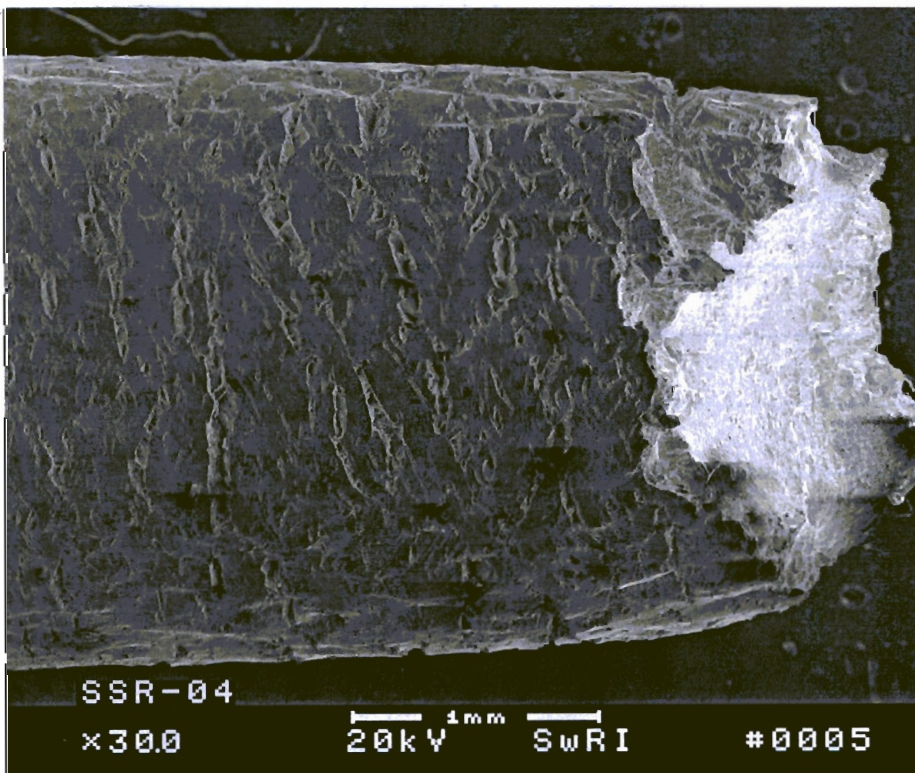
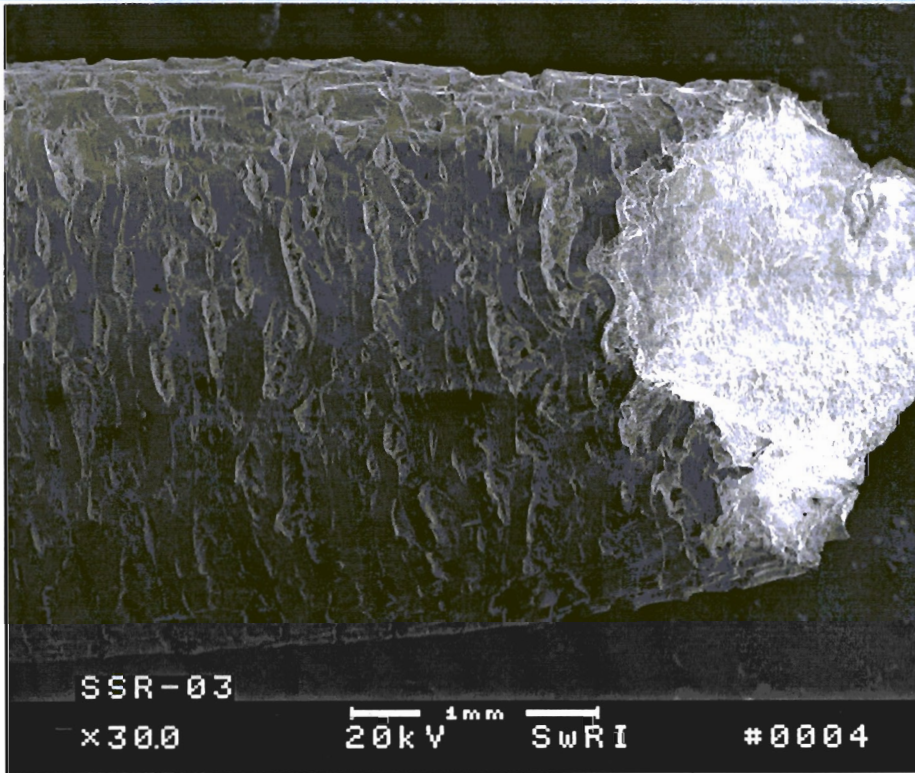
*grey-silver mottled; some
slight staining (darker grey); looks
like ductile fracture*

$$e^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

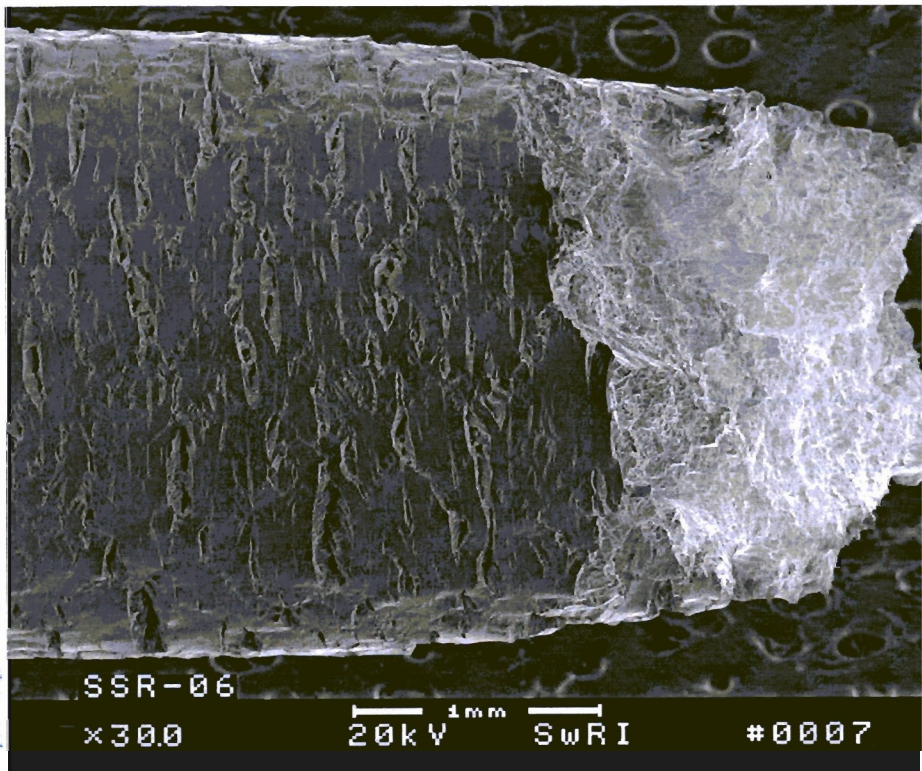
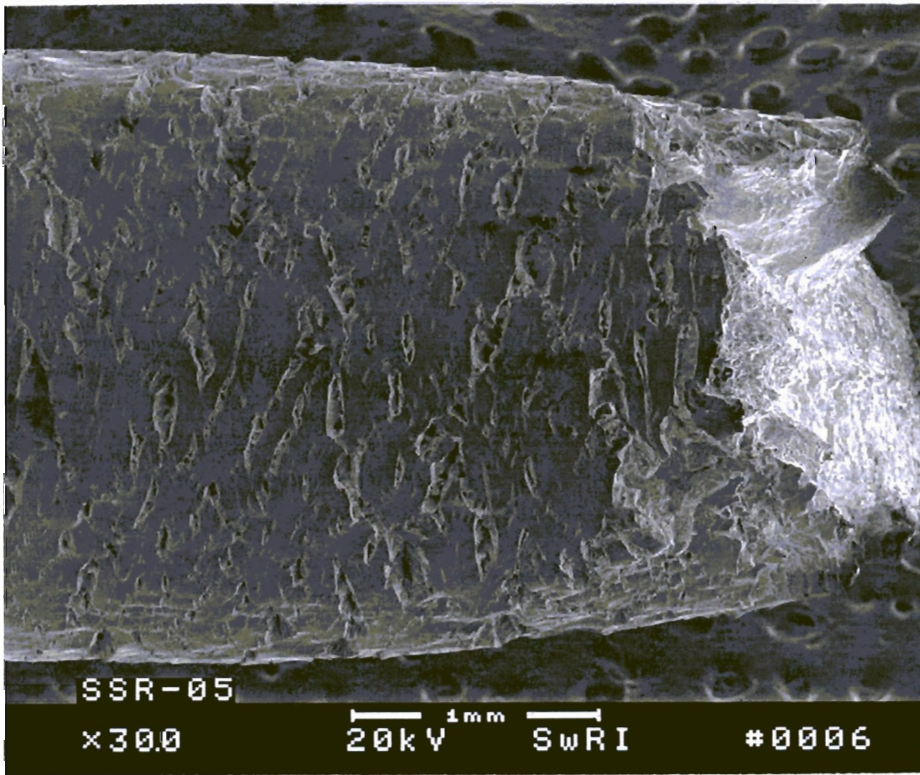
*** Temp difference between RT + 95°C = Δ13 mV*** this concentration is above saturation; solution plus
residual salt were put into test cell**data file: SSRMA22-SCW11**Walter J. Macdonald*
6/28/04



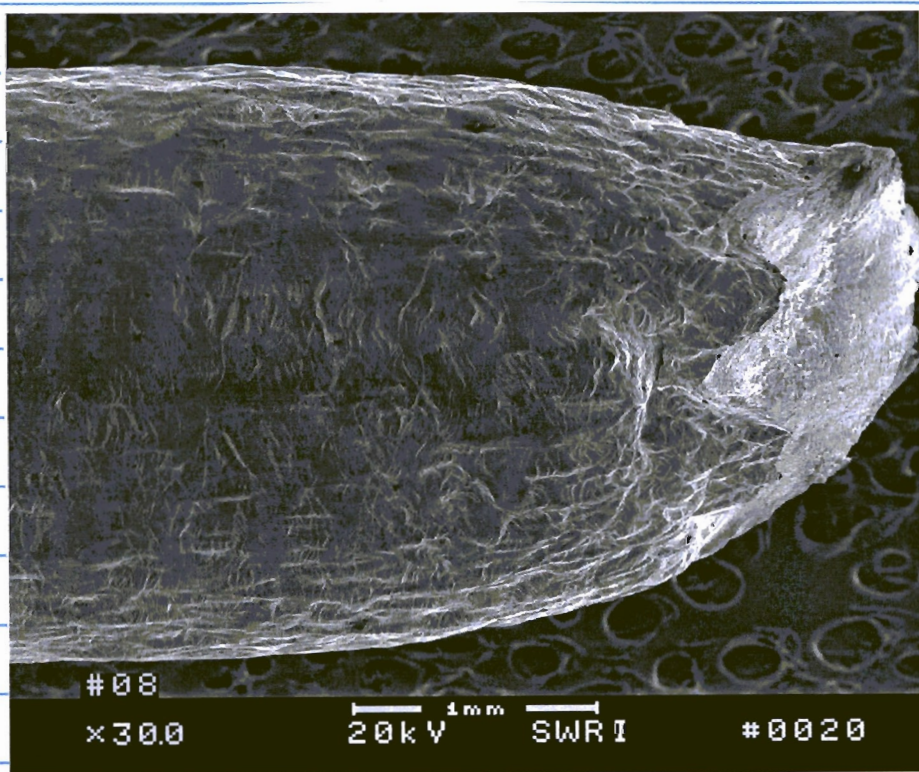
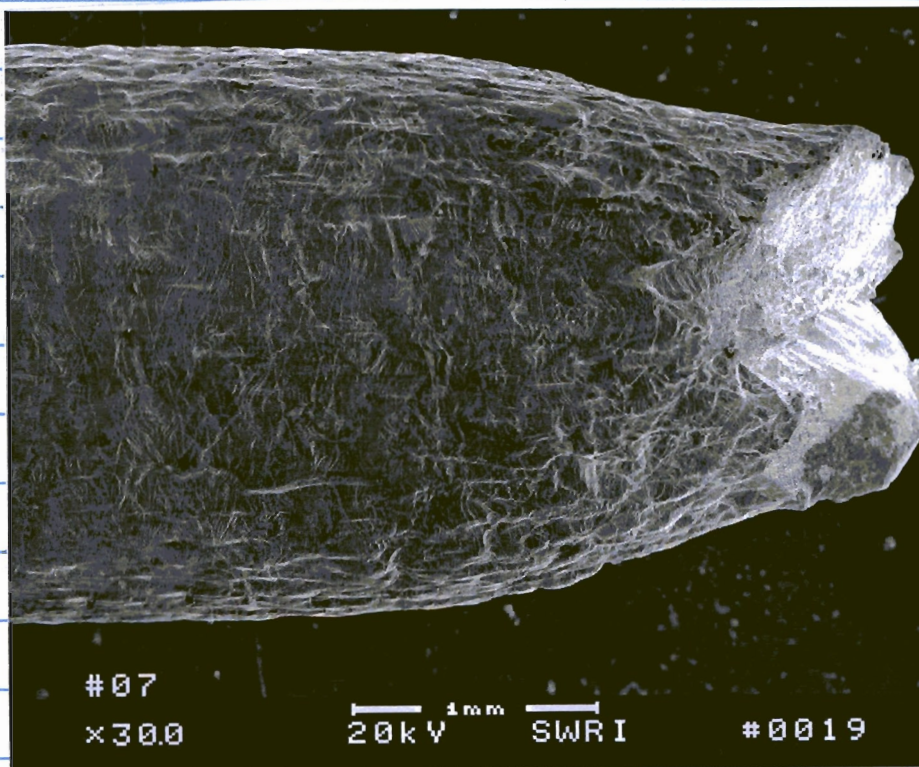
K.T. Chiang 6/23/04



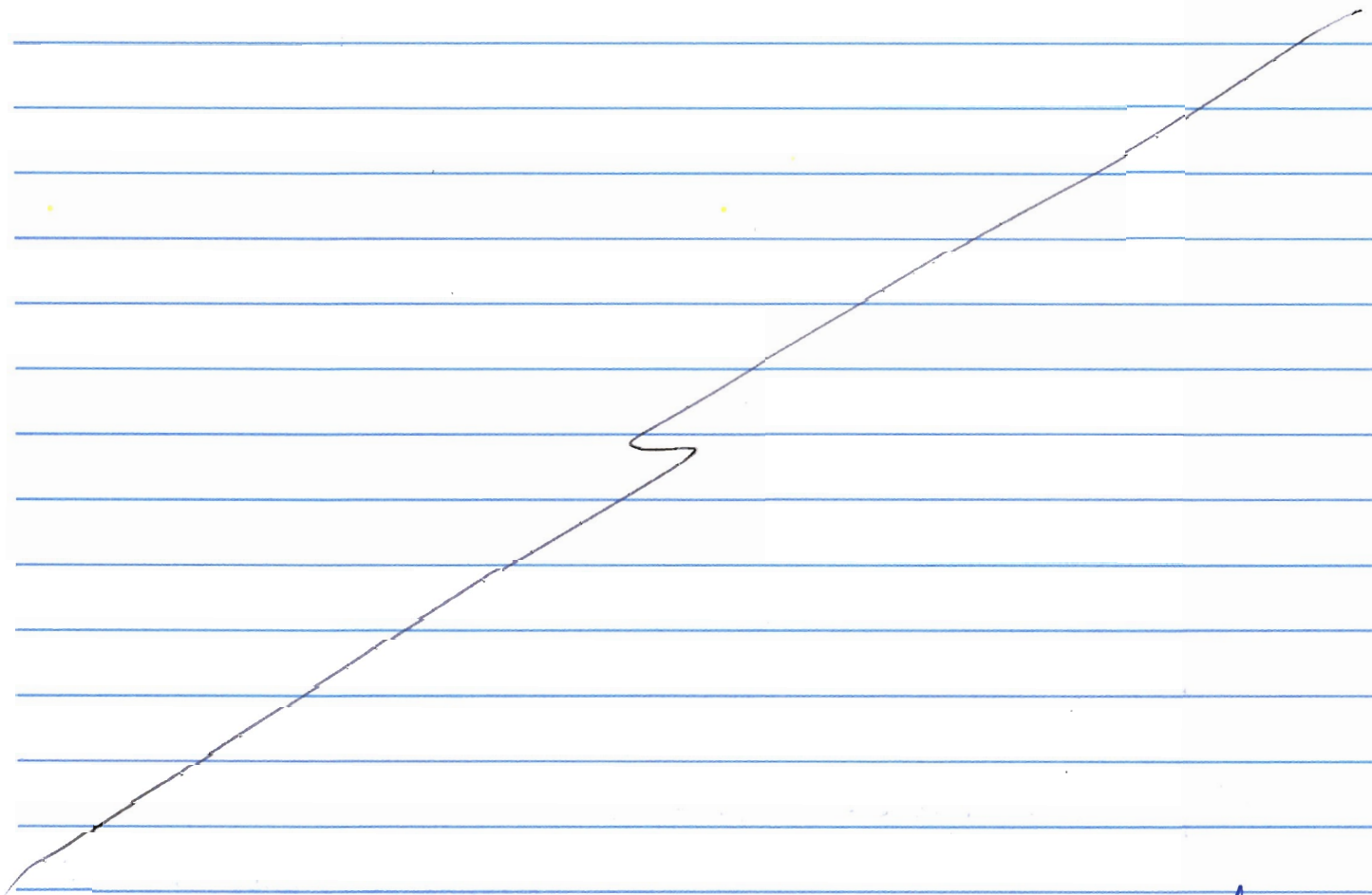
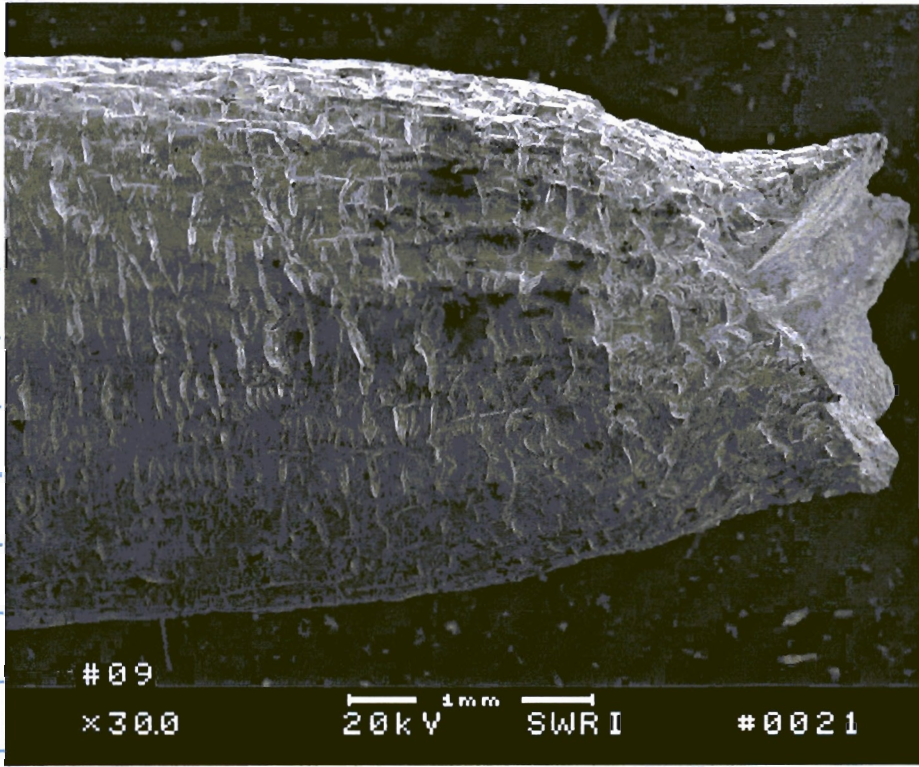
K. S. Ching 6/29/04



K.T. Chiu
6/23/04



K. J. Chiang
6/23/04



K.-T. Ching
6/23/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\times \frac{1}{2}$ liter

2.720 g NaCl lot # 035421 } amount in
 4.342 g NaNO₃ ~~# 0280~~ ^{B10} ₀₂₀₈₀₉ } SCW sol'n
 # 2277-8-3266 ^{B10} _{6/27/06}

pH 5.44 adj to 9.43 post pH 7.38

Reagents measured with

Model: OHAUS
Cal: 2 FEB 04SN: 2883
Due: 6 AUG 04

Counter Electrode: Pt flag

Reference Electrode: Ag/AgCl
in house w/ 3M KClGas: N₂ (99.999%) Ecorr: -288 mV

* Applied: +415 mV Potentiostat: LSC 440-2 SN: 9209138

Specimen Visual:

grey-silver metallic; looks like
ductile fracture

$$E^{\circ} = 3.2 \times 10^{-6} \text{ V}$$

* temp diff between RT and 95°C $\Delta 13 \text{ mV}$

data file: SSRMA22-SCW12

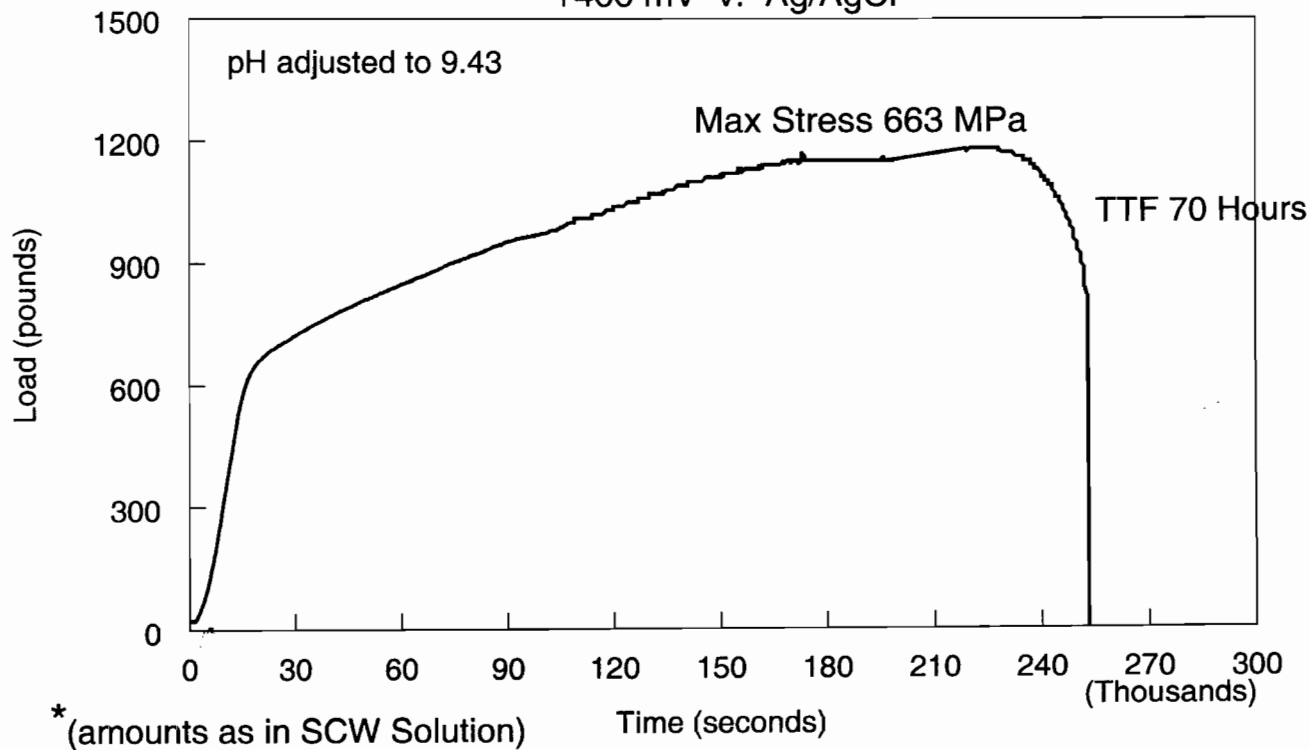
Walter J. MacKowski
7/2/04

Slow Strain Rate Test

NaCl + NaNO₃*

SSRMA22_SCW12

+400 mV v. Ag/AgCl



Walter J. Neerach
7/2/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #665*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: x 1 liter

Heat # 2277-8-3266
2277-3-3266 *1/27/04*
SCW minus NaHCO_3

KCl 6.4725 #005573

 Na_2SO_4 20.691 #035451

NaCl 5.427 #035421

NaF 3.113 #991559

 NaNO_3 8.759 #020809pH 6.87 $\xrightarrow{\text{adj}}$ 9.79

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 2 FEB 04

Due: 2 Aug 04

Counter Electrode: *PT flag*Reference Electrode: *Ag/AgCl*
*in home w/ 3M KCl*Gas: N_2 (99.999)

Ecorr: -305 mV

* Applied: +415 mV

Potentiostat: ESC440-2

SN: 9209138

Specimen Visual:

grey silver metallic color
looks like ductile fracture

$$e^{\circ} = 3.2 \times 10^{-4} \text{ s}^{-1}$$

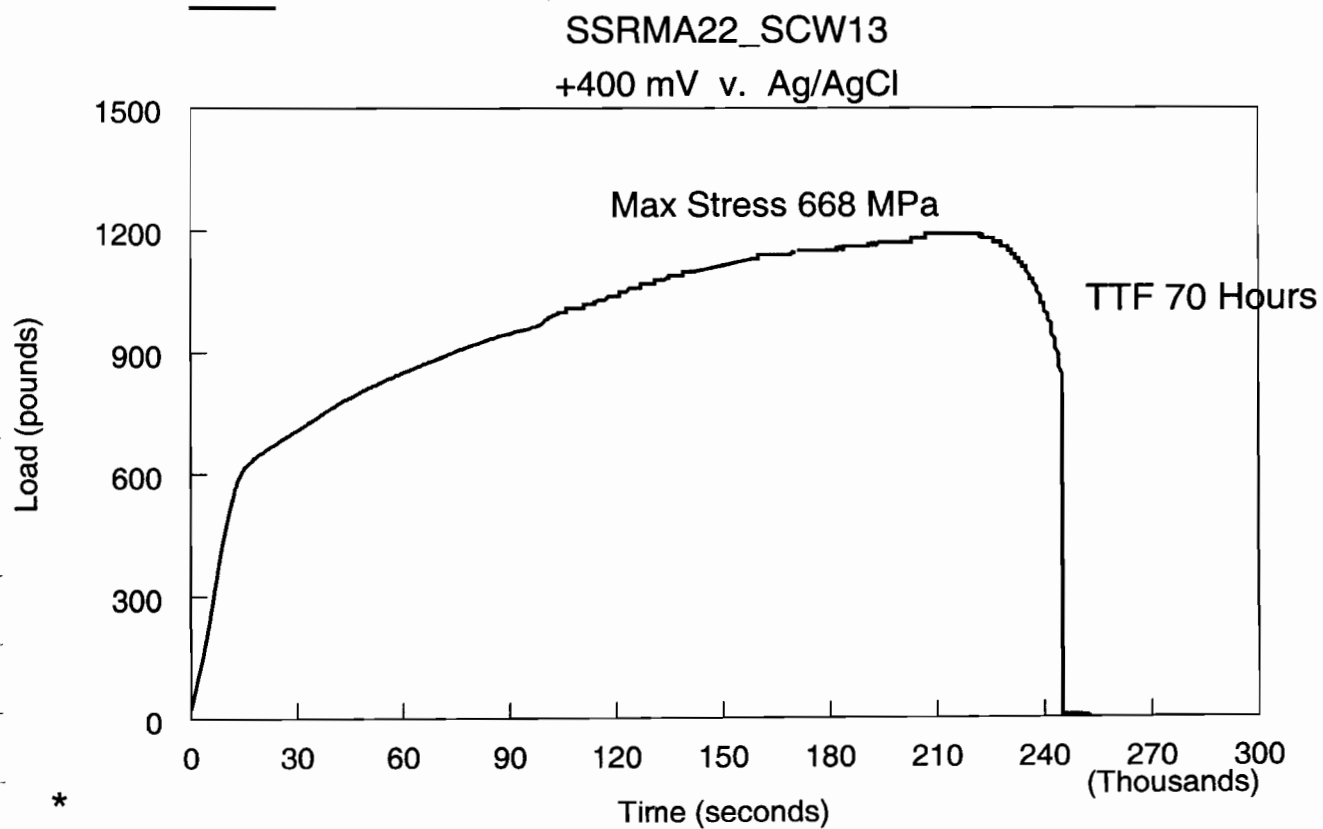
post pH 7.53

* Temp offset between RT and 95°C $\Delta 13 \text{ mV}$

file name: SSR MA 22 - SCW 13

Walter J. Markon
7/19/04

Slow Strain Rate Test SCW minus NaHCO₃



Walter J. MacKowski
7/19/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #685*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

*Heat # ~~2277-8-3266~~
2277-3-3266* ³⁴⁰_{6/27/04}

Solution:

N/A specimen run in air

Reagents measured with

Model:

Cal:

N/A

SN:

Due:

Counter Electrode:

Reference Electrode:

Gas:

N/A

Ecorr:

N/A

Applied:

Potentiostat:

SN:

Specimen Visual:

*metallic sheen; looks like
ductile failure*

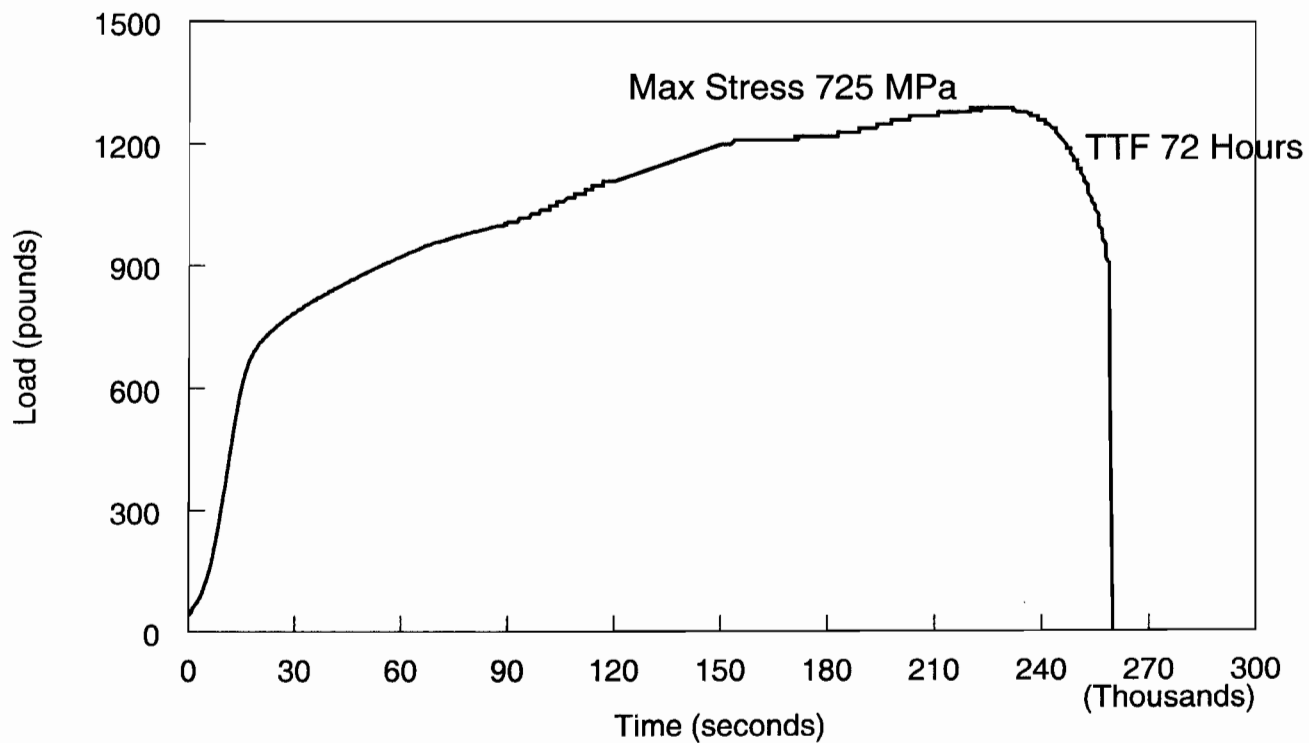
$$E^0 = 3.2 \times 10^{-6} \text{ S}^{-1}$$

*data file: SSRMA22 - air 2**Dattaj Machwani
7/20/04*

Slow Strain Rate Test

Ambient Air

SSRMA22_air2



Walter J. Machinski
1/20/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook # 695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\times \frac{1}{2} l$ *Heat # 2277-8-5266*
2277-3-3266 ⁸¹⁰ _{6/27/06}48.171g NaHCO_3 #028924pH 6.50 $\xrightarrow{\text{adj}}$ 8.58 $\xrightarrow{\text{final}}$ 10.38

Reagents measured with

Model: *OHUUS*SN: *2883*Cal: *2 FEB 04*Due: *2 MAR 04*Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl*
*in house w/3M KCl*Gas: *N₂ (99.999)*Ecorr: *-294 mV** Applied: *+415 mV*Potentiostat: *ESC 440-2* SN: *9209138*

Specimen Visual:

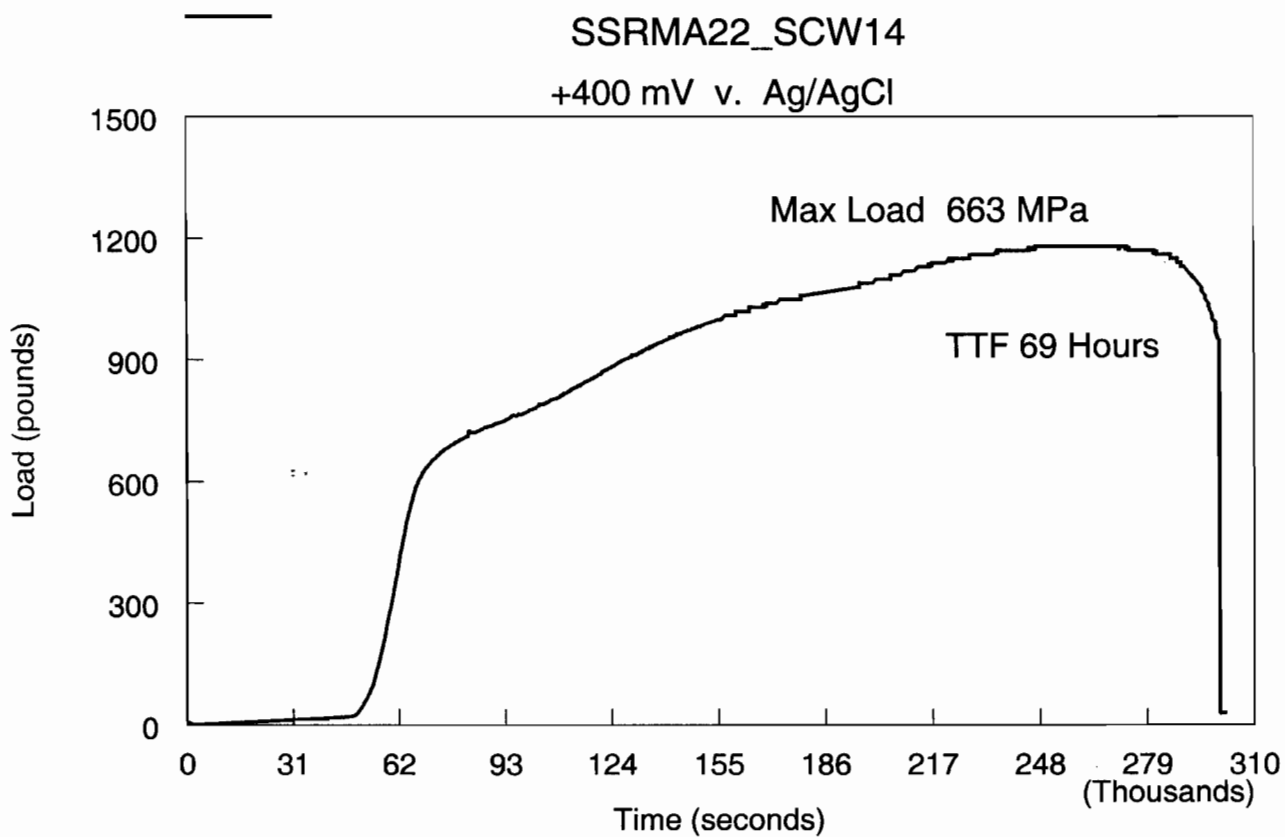
looks like ductile fracture
still silver-grey metallic, slight
stain

$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* temp difference between RT & 95°C $\Delta 13 \text{ mV}$ data file: *SSRMA22_S0014**Walter J. MacKowski*
7/23/04

Slow Strain Rate Test

NaHCO₃ as in SCW



Walter J. Mochowski
7/23/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

~~Heat #2277-8-3266~~
2277-3-3266Solution: $\times \frac{1}{2}$ liter

KCl	3.241 g	#005573	NaHCO ₃	48.21 #028924
NaCl	2.720 g	035421	NaF	1.560 #991559
NaNO ₃	4.375	020809		
Na ₂ SO ₄	10.348	035451		

pH 7.71 $\xrightarrow{\text{adj}}$ 8.66
pot test 10.18

Reagents measured with

Model: OHAUS
Cal: 2 FEB 04SN: 2883
Due: 2 AUG 04

Counter Electrode: Pt flag

Reference Electrode: Ag/AgCl

Gas: N₂ (99.999)E_{corr}: -186 mV

in home -13m KCl

*E_{applied}: +415 mV

Potentiostat: FSC440-2 SN: 9209138

Specimen Visual:

*silver-grey metallic, looks like
brittle fracture*

$$i^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

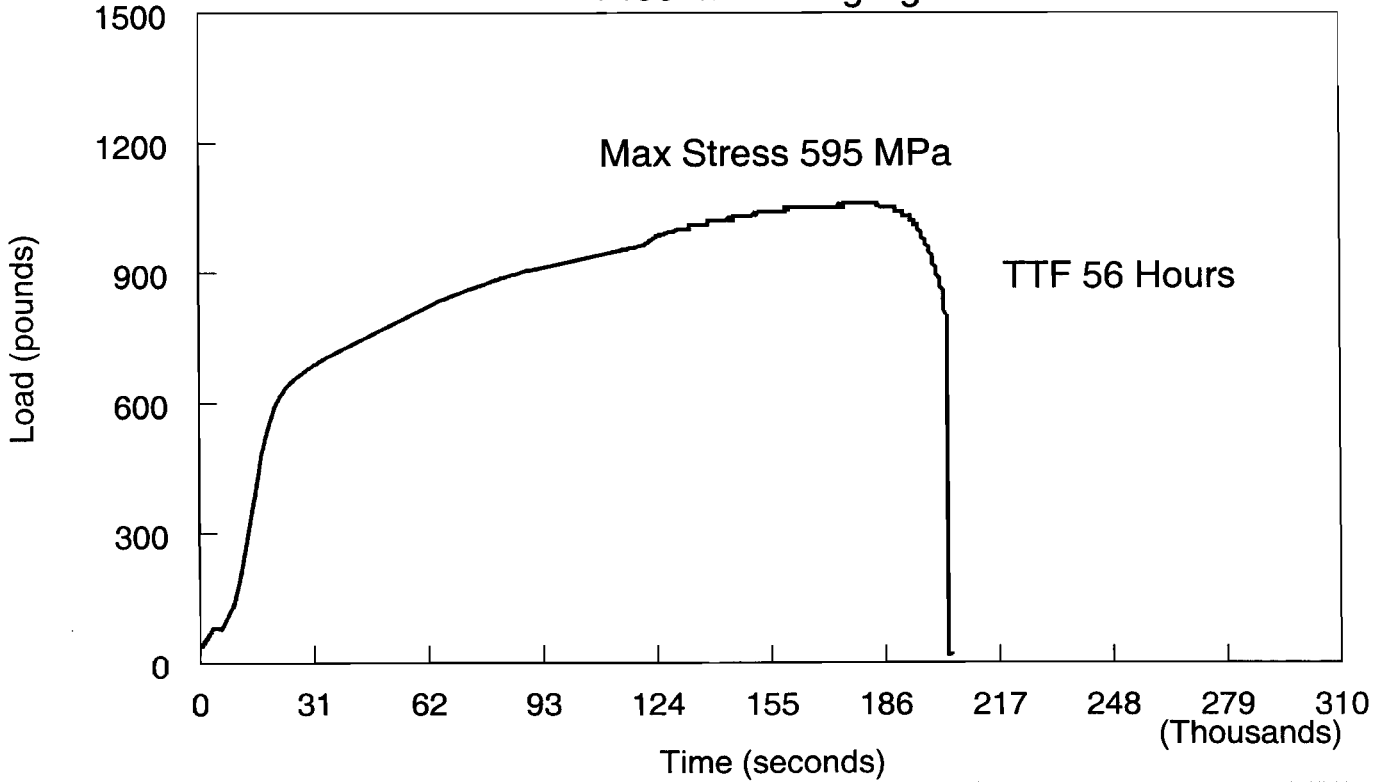
 \rightarrow temp difference between RT and 95°C $\Delta 13 \text{ mV}$

data file: SSRMA22-SCW15

Walter J. Michowski
8/3/04

Slow Strain Rate Test Simulated Concentrated Water

SSRMA22_SCW15
+400 mV v. Ag/AgCl



Walter J. Macroski
8/3/04

SLOW STRAIN RATE TEST

Objective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *x 1/2 liter*

Heat # ~~2277-8-5266~~ 2277-3-3266 6/27/06

KCl	3.252g	#005573	pH 7.76
NaCl	2.726	035509	adi → 8.68
NaNO ₃	4.372	020809	
NaHCO ₃	48.18	028294	post test 9.78

Reagents measured with

Model: *ONTAS*
Cal: *2 FEB 04*

SN: *2883*
Due: *4 AUG 04*

Counter Electrode: *PT flag*

Reference Electrode: *Ag/AgCl*
in house 1M KCl

Gas: *N₂ (99.999)* Ecorr: *-68mV*

* Applied: *+415mV* Potentiostat: *LSC440-2* SN: *9209138*

Specimen Visual: *brittle fracture*

note: *Luggin probe failed; i.e. some solution was lost and there is a question as to the actual potential applied since reference solution was lost too.*

$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$

* *temp effect RT → 95°C Δ13mV*

data file: SSRMA22_SCW16

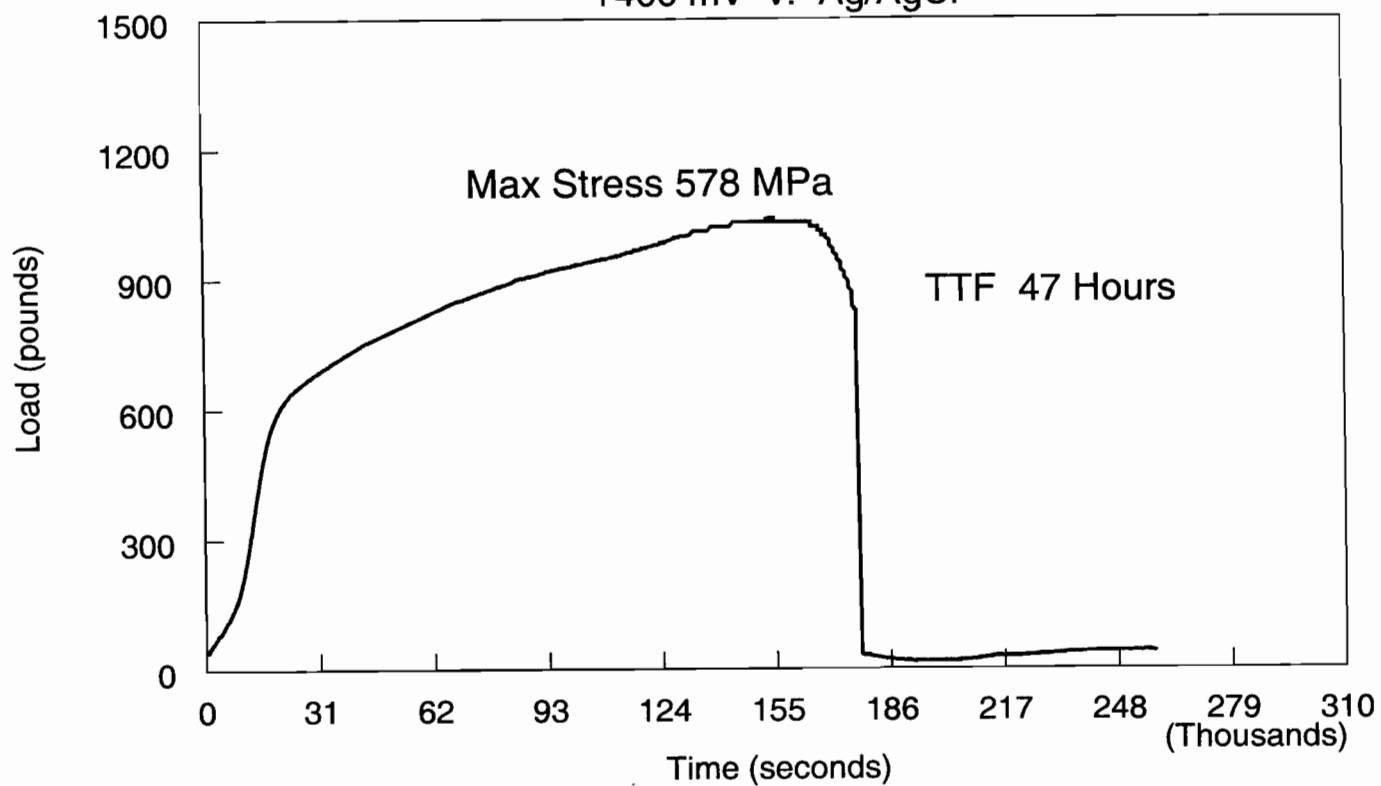
Walter J. MacKowski
8/10/04

Slow Strain Rate Test

SCW minus F and ~~NO3~~

SSRMA22_SCW16

+400 mV v. Ag/AgCl

SO₄ 8/10/04
WJMWalter J. Machowski
8/10/04

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *~ 1/2 liter**Heat # ~~2277-8-3266~~ 2277-3-3266* $\frac{8/12/06}{6/22/06}$

KCl 3.275g #005573

pH 8.45

NaCl 2.738 035509

adi → 8.70NaNO₃ 4.383 020809NaHCO₃ 48.18 028294

final 10.14

Reagents measured with

Model: *ONTANS*

SN: 2883

Cal: 2 FEB 04

Due: 4 Aug 04

Counter Electrode: *PT flag*Reference Electrode: *Ag/AgCl*Gas: *N₂ (99.999)*Ecorr: *-228mV**in house 1/3M KCl** Applied: *+415mV*Potentiostat: *ECS440-2* SN: *9209138*

Specimen Visual:

metallic silver-grey; looks like brittle fracture

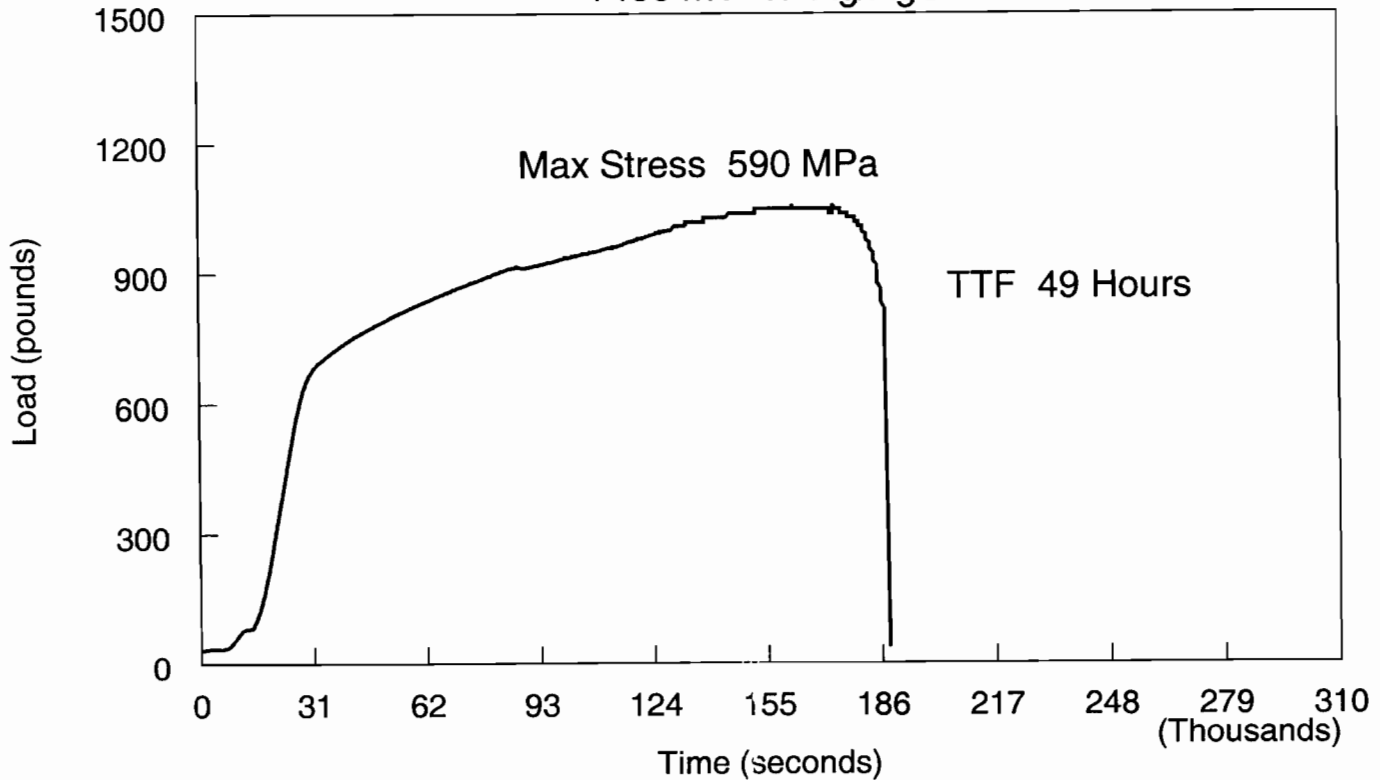
$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp effect from RT → 95°C Δ13mV*data file: *SSRMA22-SCW17**Walter J. MacKowski*
8/12/04

Slow Strain Rate Test SCW minus F and SO4

SSRMA22_SCW17

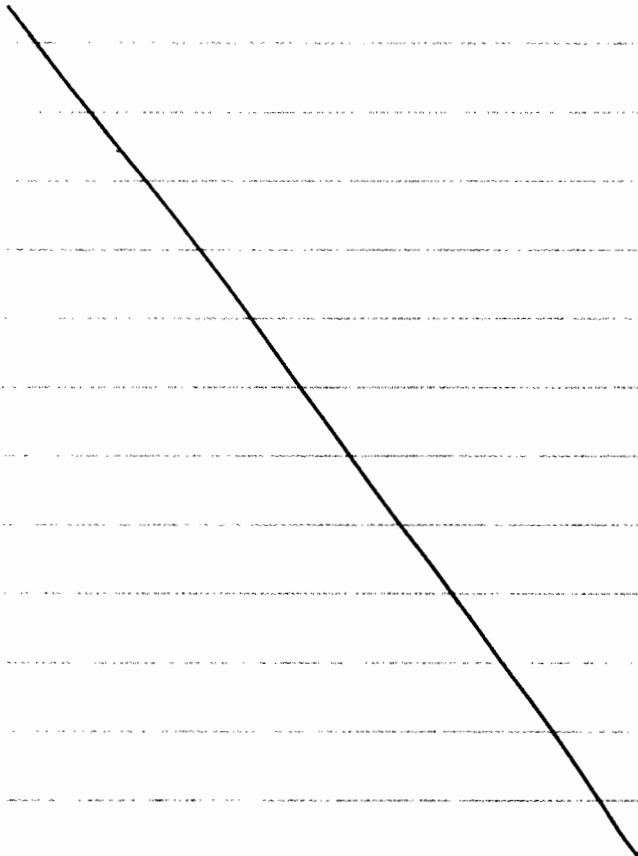
+400 mV v. Ag/AgCl



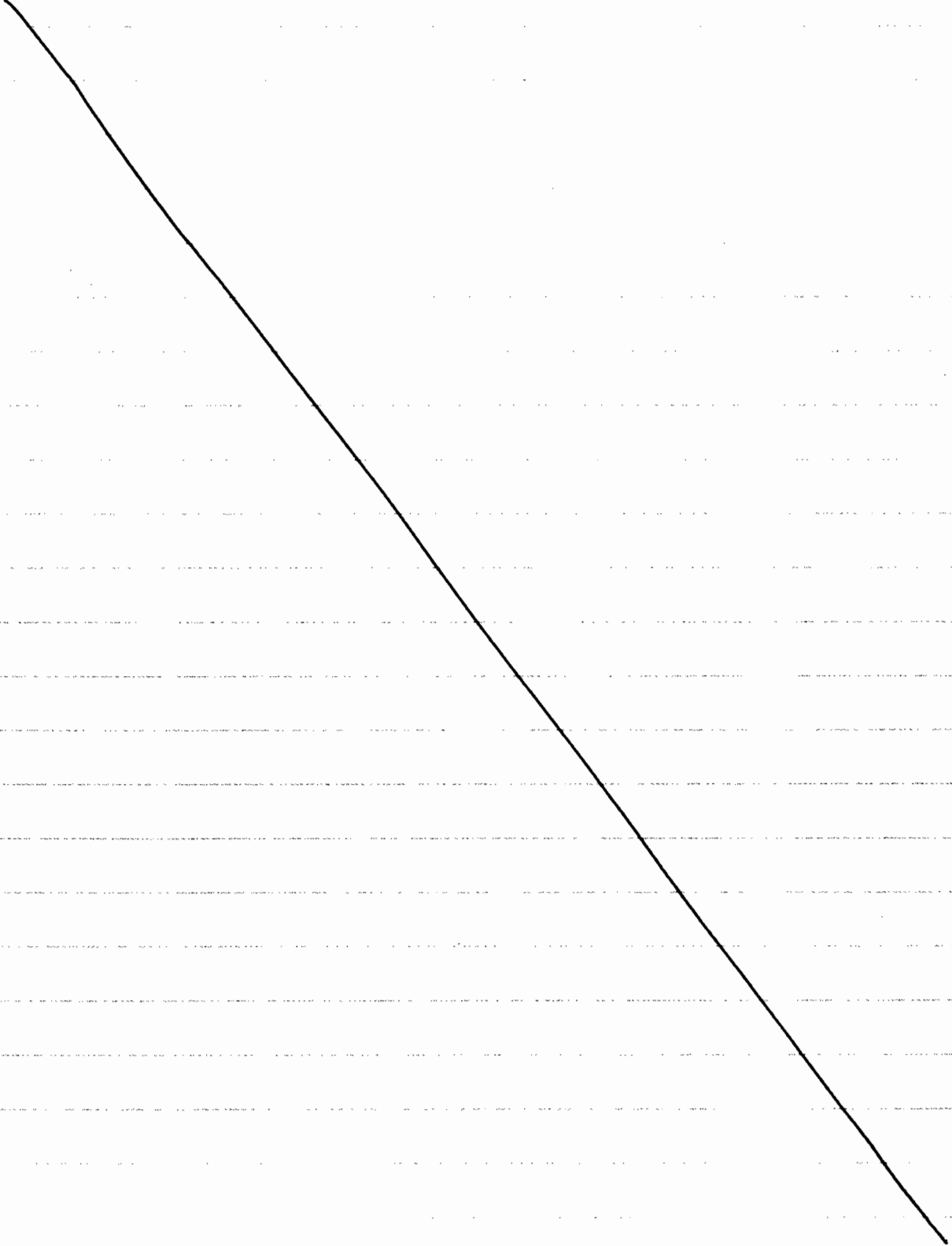
This is a duplicate of SSRMA22-SCW16
which was run due to the failure of the probe
in *-SCW16.

Walter J. MacKowski
5/12/04

It should be noted that in solutions containing HCO_3 , the amount of OH required to raise the pH is relatively considerable. There is an equilibrium that exist between CO_3 , HCO_3 , and OH such that in any solution only two of these can exist in appreciable quantities. The effect here is that the addition of OH reduces the HCO_3 concentration by converting it to CO_3 . Therefore adjusting the pH of these solutions with OH is, in effect, altering the HCO_3 concentration of the original SCW solution.



Walter J. MacLachlan
8/12/04



Walter J. Machowski
8/12/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: x 1 liter

Heat # ~~2277-8-3266~~ ⁸¹⁰
2277-3-3266 6/27/06

KCl	6.487 g	#005573	pH 7.806
NaCl	5.444	035509	↓ adj
NaHCO ₃	96.38	028954	8.726
			9.6/01 ^{10.14}
			Final 10.36

Reagents measured with

Model: ONAUS

SN: 2883

Cal: 15 JUL 04

Due: 15 JAN 05

Counter Electrode: *PT flag*Reference Electrode: *As/AsCl*
*in house w/3M KCl*Gas: *N₂ (99.999%)* Ecorr: *-203 mV** Applied: *+415 mV* Potentiostat: *ESC440-2* SN: *9209138*

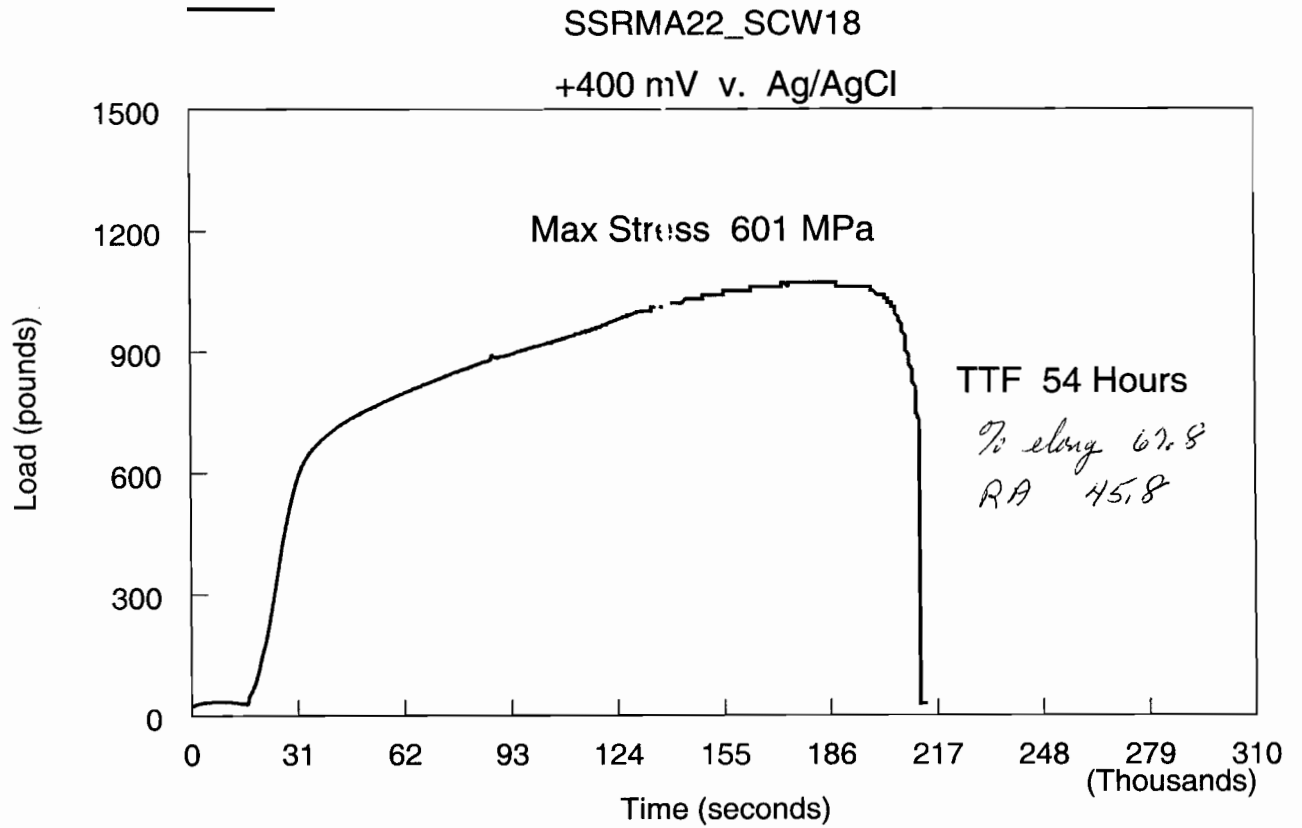
Specimen Visual:

*metallic silver-gray; looks
like brittle fracture*

$$i^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp effect from RT to 95°C Δ13 mV**data file: SSRMA22 - SOW18**Walter J. Marchowski
9/2/04*

Slow Strain Rate Test Cl and HCO₃ as in SCW



Walter J. Macchiaroli
9/2/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\times \frac{1}{2}$ liter*NaHCO₃ 96.341 g #028924**NaF 3.077 g #028924**pH 7.856 $\xrightarrow{\text{adj}}$ 8.645 final pH 10.28**Next #2277-8-3266
2277-3-3266 6/27/04*

Reagents measured with

Model: *ORAU5*SN: *2883*Cal: *15 JUL 04*Due: *15 JAN 05*Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl**in 0.1M KCl*Gas: *N₂ (99.999%)*Ecorr: *-238 mV** Applied: *+415 mV*Potentiostat: *KSC 440-2*SN: *9209138*

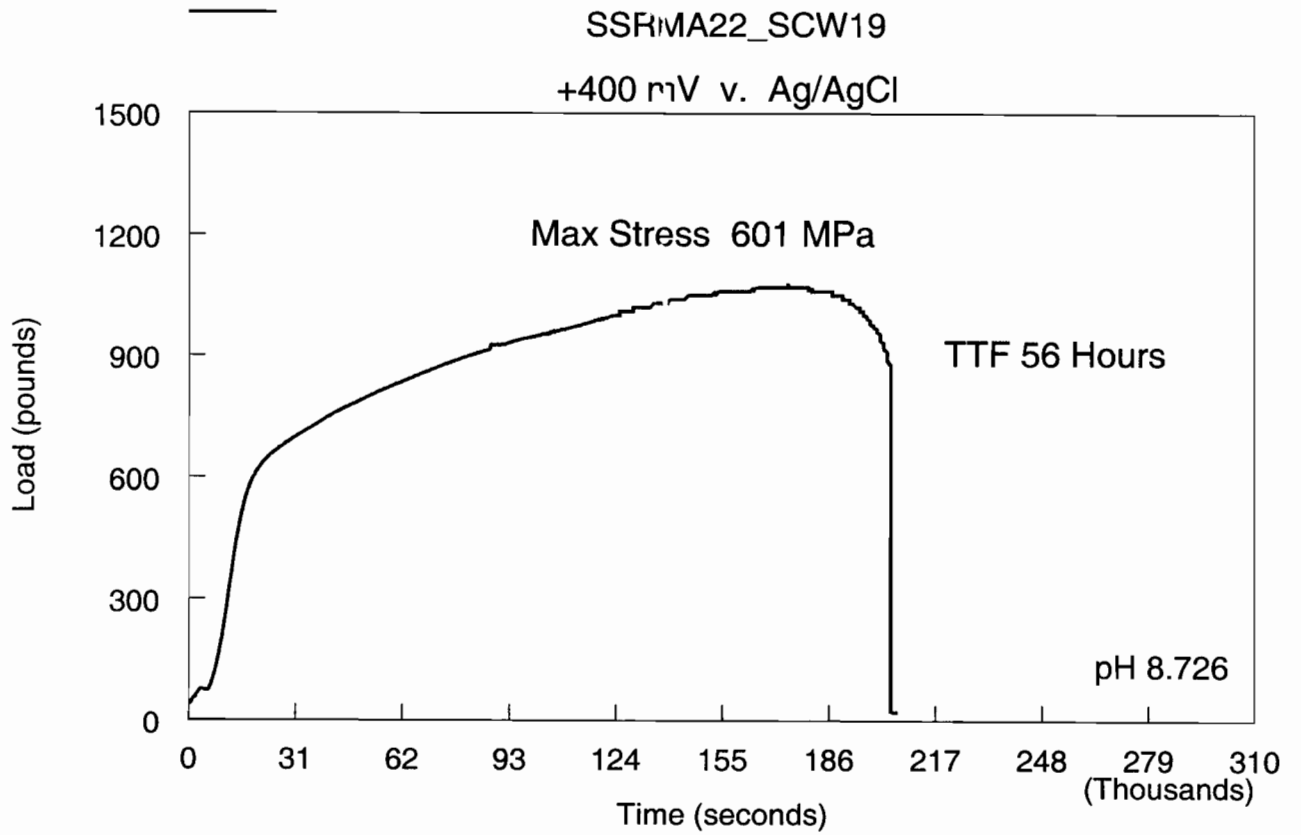
Specimen Visual:

*silver metallic gray; could
be brittle failure*

$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp effect from RT to 95°C $\Delta 13 \text{ mV}$* *data file: SSRMA22-SCW19**Walter J. MacLachlan
9/3/04*

Slow Strain Rate Test F and HCO₃ as in SCW



Walter J. Mochowski
9/3/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *1 liter*~~Heat # 2277-8-5266~~
2277-3-3266 ⁸¹⁰ 6/27/06

NaHCO ₃	88.240 g	# 028924	pH 8.784
NaCl	29.252 g	# 035509	final
NaOH	7.662	# 033972	pH 10.42

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 15 JUL 04

Due: 15 JAN 05

Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl*
*in house w/3M NaCl*Gas: *N₂ (99.999%)* Ecorr: *-268 mV** Applied: *+415 mV*Potentiostat: *65C 440-2* SN: *9209138*

Specimen Visual:

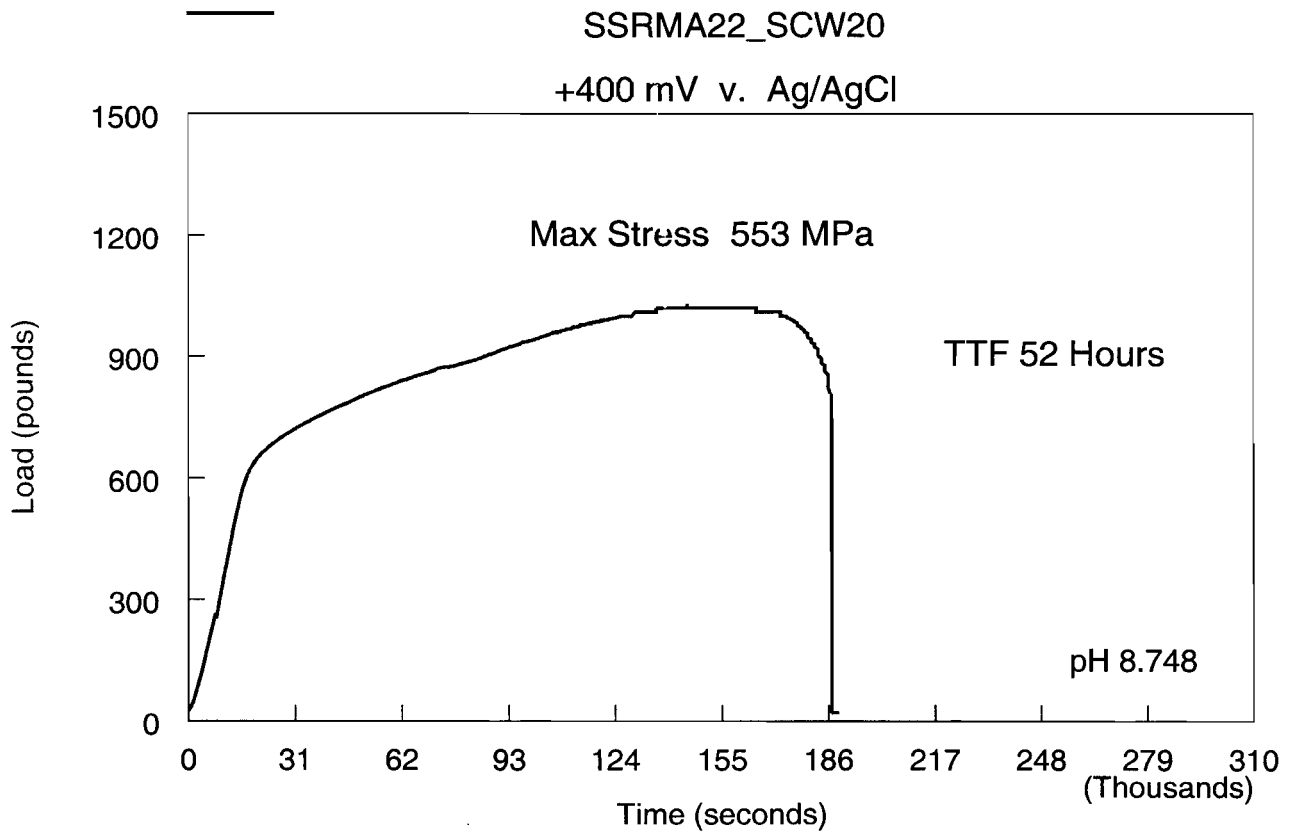
metallic silver-gray; looks like brittle fracture

$$\epsilon^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* *tens effect from RT to 95°C* $\Delta 15 \text{ mV}$ data file: *SSRMA22-SCW20**Walter J. MacKowski*
9/2/04

Slow Strain Rate Test

1.05M NaHCO₃; 0.5M NaCl; 0.19M NaOH



Walter J. MacLowski
9/7/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: *x 2 liters**Heat #2277-8-3266
2277-3-3266 ⁹⁰
6/2/06*

NaHCO ₃	176.57 g	#028966	pH 8.683
NaCl	58.50	#035509	not adj.
NaOH	15.35	#033972	
KCl	223.65	#005573	final pH 10.1

Reagents measured with

Model: *OHAUS*SN: *2883*Cal: *15 JUL 04*Due: *15 JAN 05*Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl*Gas: *N₂ (99.999%)*Ecorr: *-234mV**in house ~13M NaCl
w/ 0.1M KCl**Applied: *+415mV*Potentiostat: *ESC 440-2* SN: *9209138*

Specimen Visual:

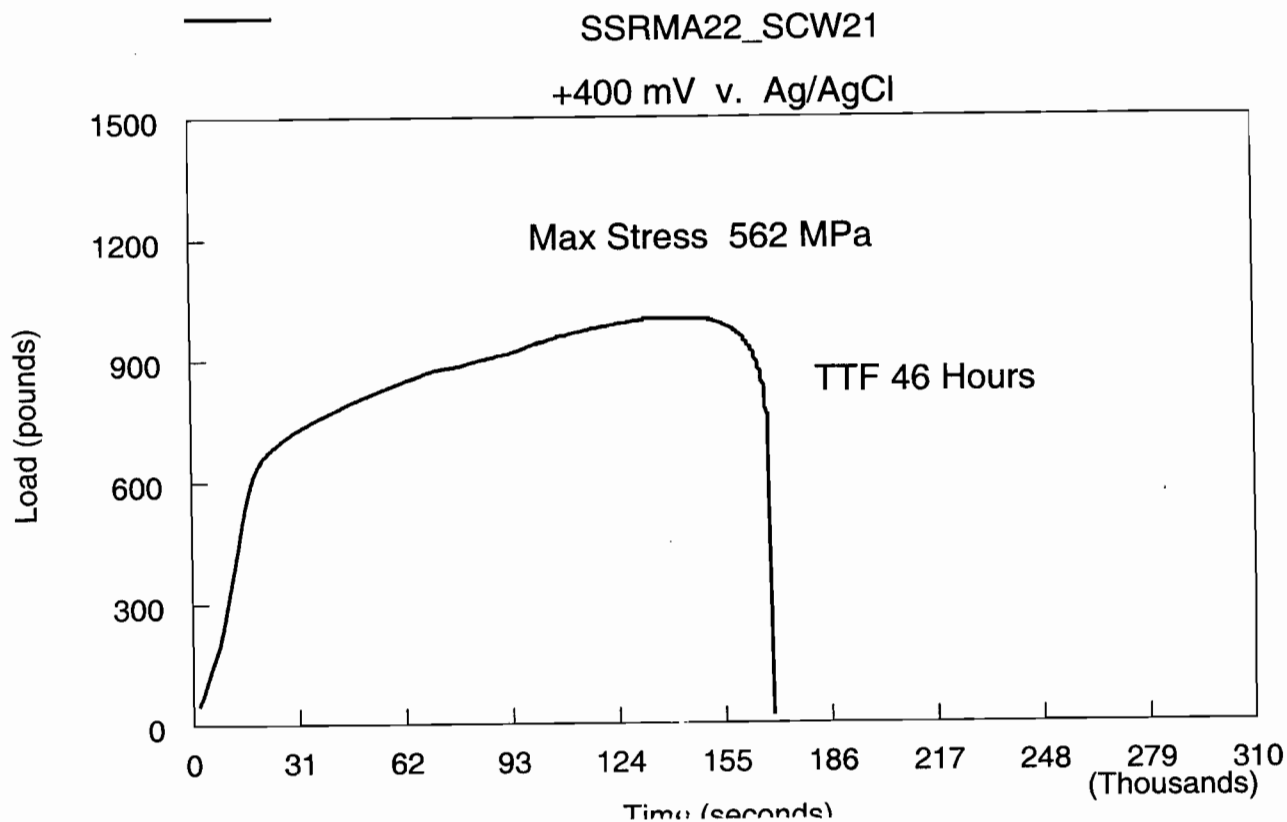
*metallic silver-gray;
looks like brittle fracture*

$$\epsilon^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp effect from RT to 95°C $\Delta 13mV$* *data file: SSRMA22-SCW21**Walter J. Macdonald
9/17/04*

Slow Strain Rate Test

1.05M NaHCO₃; 0.5M NaCl; 0.19M NaOH; 1.5M KCl



Walter J. MacKowski
9/17/04

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# 2277-8-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

2277-3-2266
4/27/04

Initial Weight: 12.46190g Model: Sartorius Genius SN: 12809099
Final Weight: 12.33808g Cal: 5/14/04 Due: 11/14/04

SOLUTION: SCW - NaHCO₃

KCl = 12.974g Lot# 005573 Na₂SO₄ = 41.39g Lot# 035451
NaCl = 10.874g Lot# 035509 NaF = 6.198g Lot# 991559
NaNO₃ = 17.869g Lot# 020809 + DI water 2000mls

Reagents measured with Model: OHAUS SN: 2883
Cal: 7/15/04 Due: 7/15/05

Initial pH: 7.012 Model: orion SN: 2330
Final pH: 7.836 CAL: 7/21/04 DUE: 7/21/05
pH Probe: #13-620-296 SN: 4079126

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: 4198-149
Cal: 6/8/04 Due: 12/4/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 3289206

Gas: 99.999% Nitrogen

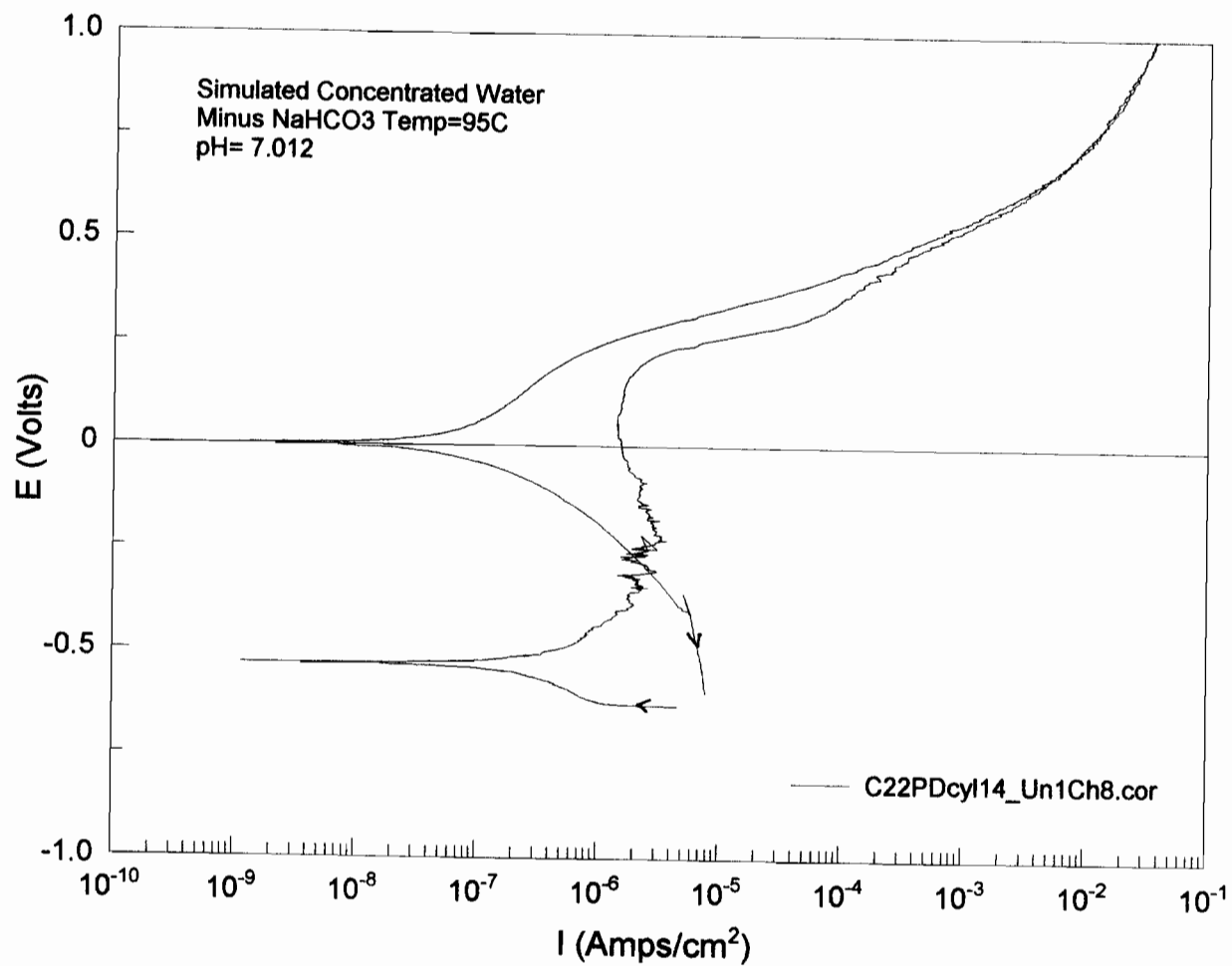
Ecorr: -545 mV Model: Keithley 614 SN: 6704936
Ept: -43 mV Cal: 6/7/04 Due: 6/7/05
Eapplied (vs SCE): ~

Potentiostat: Solartech 1480 SN# 00238265

Specimen Examination: No pitting - Mild Surface Etching on Specimen
Staining on All Surfaces of Specimen

Data C22 PD cyl 14

B. E. J. 9/1/04



B. J. 9/8/04

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish2277-3-3266
By
4/27/04Initial Weight: 12.37657g Model: Sartorius Genius SN: 12809099
Final Weight: 12.37598g Cal: 5/14/04 Due: 11/14/04SOLUTION: .5M NaCl + 1.05M NaHCO₃ + 1.5M KCl + 0.19M NaOH
58.48g NaCl Lot # 035509 15.241g NaOH # 033972
176.46g NaHCO₃ Lot # 628924
223.68g KCl Lot # 005573Reagents measured with Model: OHAUS SN: 2883
Cal: 7/15/04 Due: 7/15/05Initial pH: 8.790 Model: Orion SN: 2330
Final pH: 8.826 CAL: 7/21/04 DUE: 7/21/05
pH Probe: #13-620-296 SN: 4079126TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: 458-149
Cal: 6/8/04 Due: 12/8/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 3289206

Gas: 99.999% Nitrogen

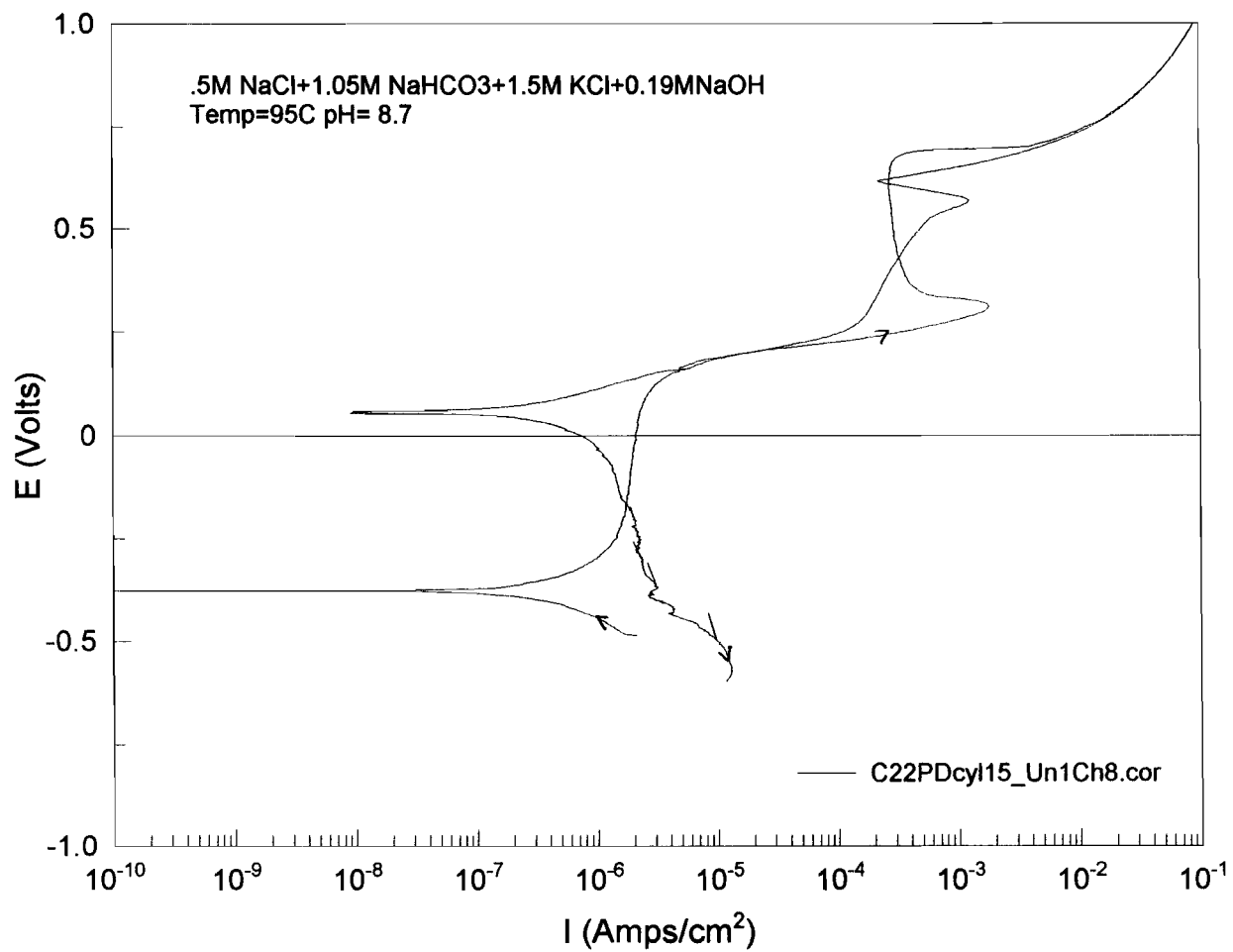
Ecorr: -389mV Model: Keithley 614 SN: 0704936
Ept: +70mV Cal: 6/7/04 Due: 6/7/05
Applied (vs SCE): -

Potentiostat: Solentech 1450 SN# 06238265

Specimen Examination: No pitting on specimen very m.h. surface
staining will Repolish Specimen for further testing

Data C2200 cyl 15

B. K. J. 9/8/04



[Handwritten signature] 9/8/05

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish
2277-3-3266 87
4/27/04

Initial Weight: 12.3584g Model: Sartorius Genius SN: 12809099
 Final Weight: 12.35550g Cal: 5/14/04 Due: 11/14/04

SOLUTION: .5 m NaCl + 1.05 m NaHCO_3 + 0.19 m NaOH

58.46g NaCl lot # 035509
 176.53g NaHCO_3 lot # 028924
 15.245g NaOH lot # 033972

Reagents measured with Model: OHAUS SN: 2883
 Cal: 7/15/04 Due: 7/15/05

Initial pH: 8.910 Model: orion SN: 2330
 Final pH: 9.123 CAL: 7/21/04 DUE: ~~7/21/05~~ 7/21/05 8/20/06
 pH Probe: #13-620-296 SN: 4079126

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: M 98-149
 Cal: 6/8/04 Due: 12/8/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 3289206

Gas: 99.999% Nitrogen

Ecorr: -383mV Model: Keithley 614 SN: 0704936
 Ept: -287mV Cal: 6/7/04 Due: 6/7/05

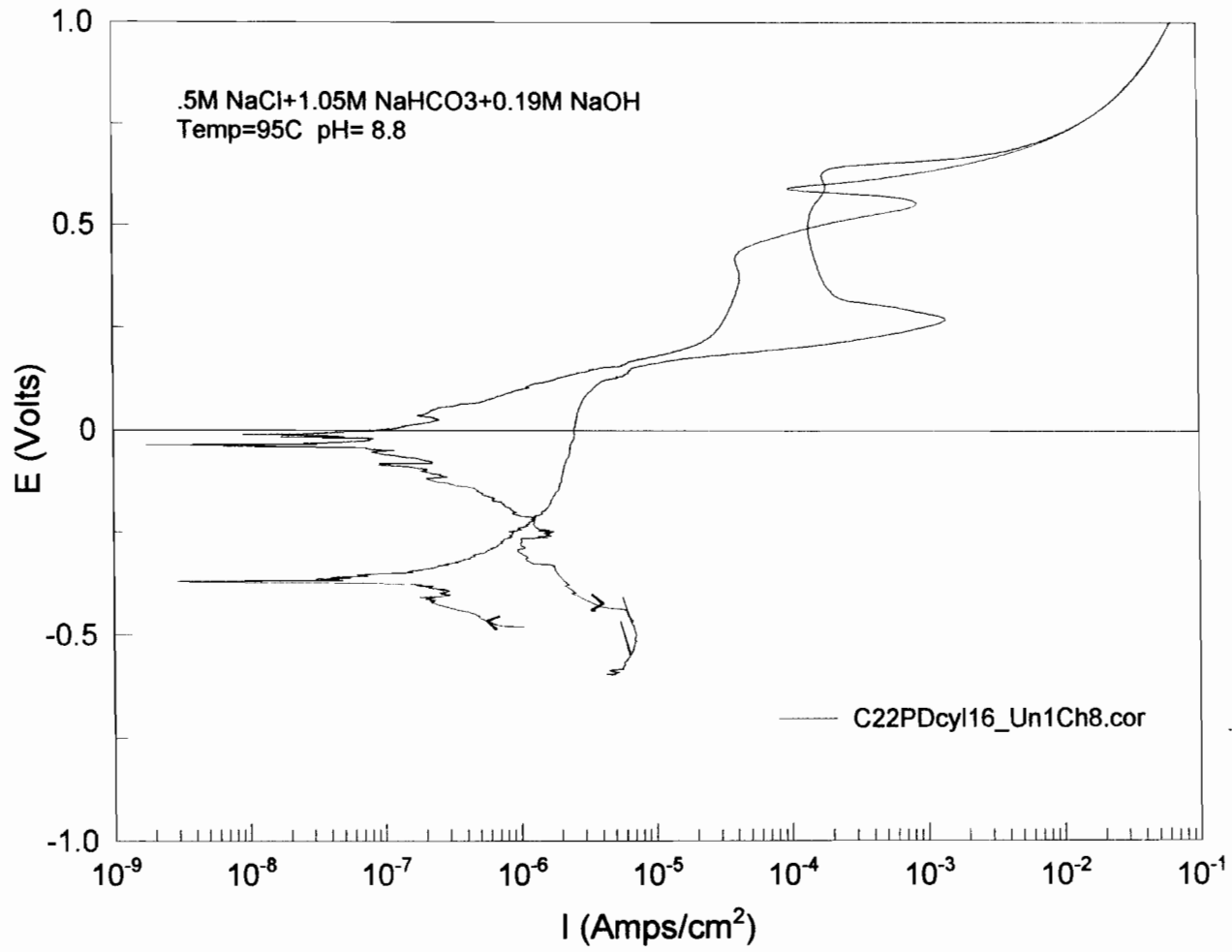
Applied (vs SCE): -

Potentiostat: Solartron 1480 SN# 00238265

Specimen Examination: No Corrosion or pitting - Surface staining
 on specimen

* will Repolish Specimen for further testing
 Data C22 POCyl 1b

S. H. / 9/8/04



B. K. D. 9/9/04

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3286~~ 2277-3-3266 ⁸¹⁰ CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

6/27/06

Initial Weight: 12.3429g Model: Sartorius Genius SN: 12809099
 Final Weight: 12.3372g Cal: 5/14/04 Due: 11/14/04

SOLUTION: SCW with NaCl + NaNO₃ Only
 10.870g NaCl Lot # 035509
 17.511g NaNO₃ Lot # 020809
 + DI water to 2000mls

Reagents measured with Model: OHAUS SN: 2883
 Cal: 7/15/04 Due: 7/15/05

Initial pH: 5.324 Model: orion SN: 2330
 Final pH: 8.023 CAL: 7/21/04 DUE: 7/21/05
 pH Probe: #13-620-296 SN: 4079126

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: H 98-149
 Cal: 6/8/04 Due: 12/8/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-52 SN: 3289206

Gas: 99.999% Nitrogen

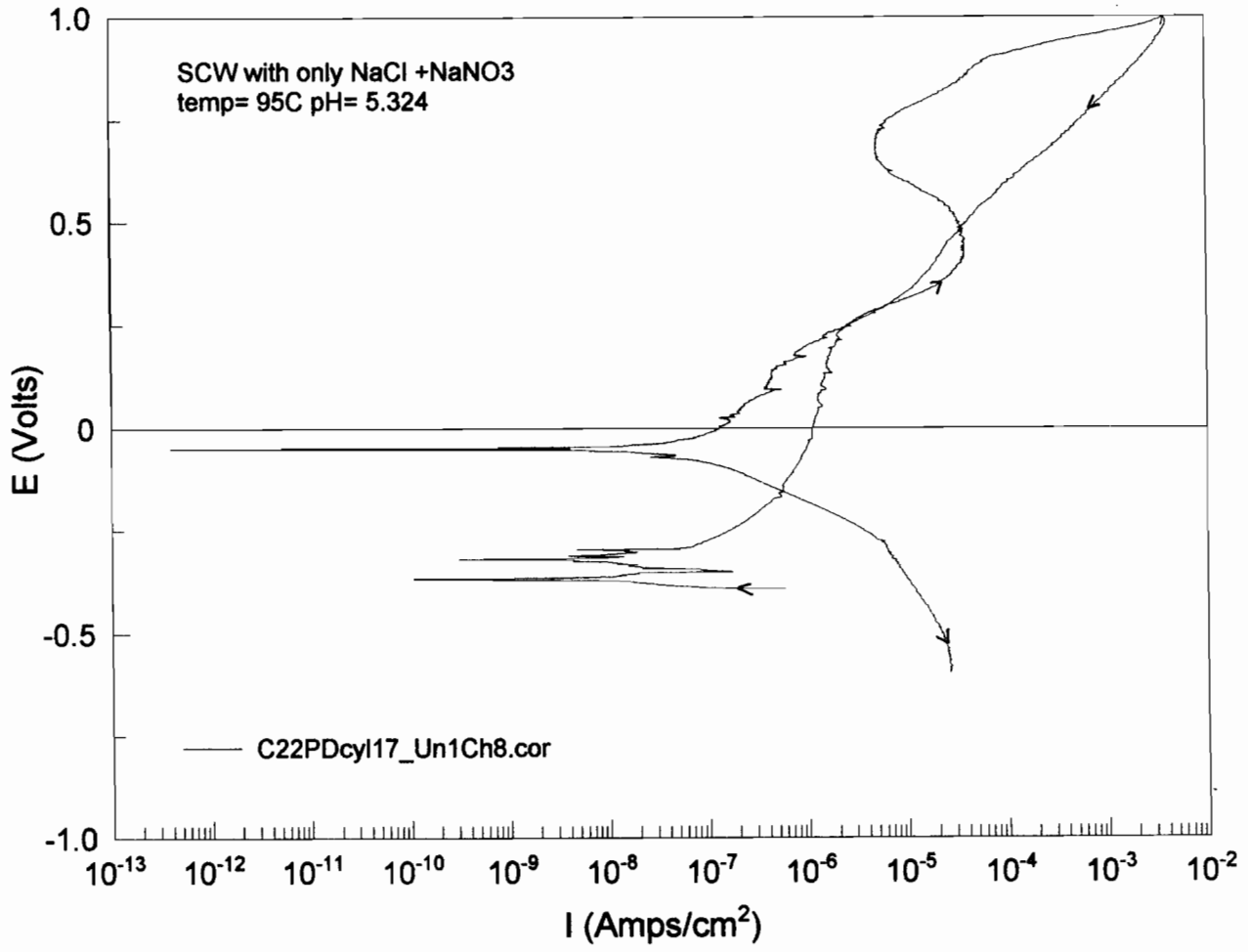
Ecorr: -264mV Model: Keithley 614 SN: 6704936
 Ept: -98mV Cal: 6/7/04 Due: 6/7/05
 Applied (vs SCE): -

Potentiostat: Solatron 1480 SN# 06238265

Specimen Examination: No sign of corrosion or pitting
 Gold color tint staining on All surfaces of Specimen

Data C22-POCY117

[Signature] 9/13/06



B. F. J.
9/14/01

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# ~~2277-8-3266~~ 2277-3-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish8/22
6/27/06

Initial Weight: 12.32176g Model: Sartorius Genius SN: 12809099
 Final Weight: Cal: 5/14/04 Due: 11/14/04

SOLUTION: 3.8 m NaCl + 0.38 m NaNO₃
 444.16g NaCl lot # 035509
 64.62g NaNO₃ lot # 020809
 + DI water to 2000 ml

Reagents measured with Model: OHAUS SN: 2883
 Cal: 7/15/04 Due: 7/15/05

Initial pH: 5.507 Model: orion SN: 2330
 Final pH: 6.718 CAL: 7/21/04 DUE: 7/21/05
 pH Probe: #13-620-296 SN: 4079126

TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: H 98-149
 Cal: 6/8/04 Due: 12/8/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 17-620-52 SN: 3289206

Gas: 99.999% Nitrogen

Ecorr: -460 mV Model: Keithley 614 SN: 6704936
 Ept: -51 mV Cal: 6/7/04 Due: 6/7/05

Applied (vs SCE): -

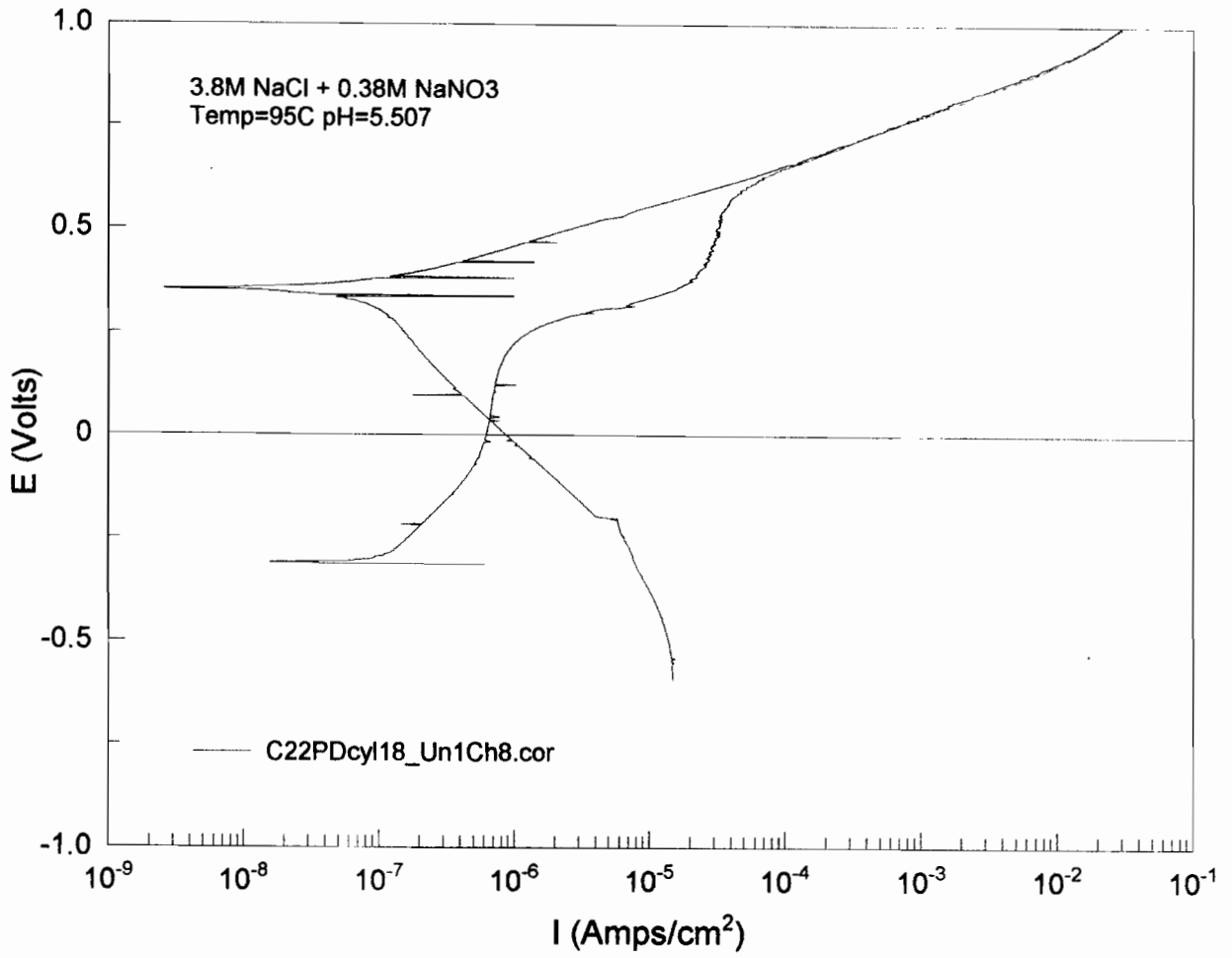
Potentiostat: Solartron 1480 SN# 60238265

Specimen Examination: No corrosion or pitting - Gold tint
 staining on All surfaces of Specimen

* Note will Repolish Specimen for further Testing

date C22POCyl

De F... 9/14/04



B. D. J. 9/15/04

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C22 Cylinder Heat# 2277-8-3266 CNWRA Drawing 20.01402.571.019 polished to a 600 grit finish

2277-3-3266 8kg
6/27/04Initial Weight: 12.27016g Model: Sartorius Genius SN: 12809099
Final Weight: 12.25543g Cal: 5/14/04 Due: 11/14/04SOLUTION: 3.8 M NaCl + 0.38 M NaNO₃
444.18g NaCl Lot # 035509
64.60g NaNO₃ Lot # 020809
+ DI water To 2000 mLReagents measured with Model: OHAUS SN: 2883
Cal: 7/15/04 Due: 7/15/05
Initial pH: 6.231 Model: orion SN: 2370
Final pH: 7.609 CAL: 7/21/04 DUE: 7/21/05
pH Probe: #13-620-296 SN:TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: H 98-149
Cal: 6/9/04 Due: 12/8/04

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 17-620-52 SN: 3289206

Gas: 99.999% Nitrogen

Ecorr: -459 mV Model: Keithley 614 SN: 0704936
Ept: -63 mV Cal: 6/7/04 Due: 6/7/05
Applied (vs SCE): -

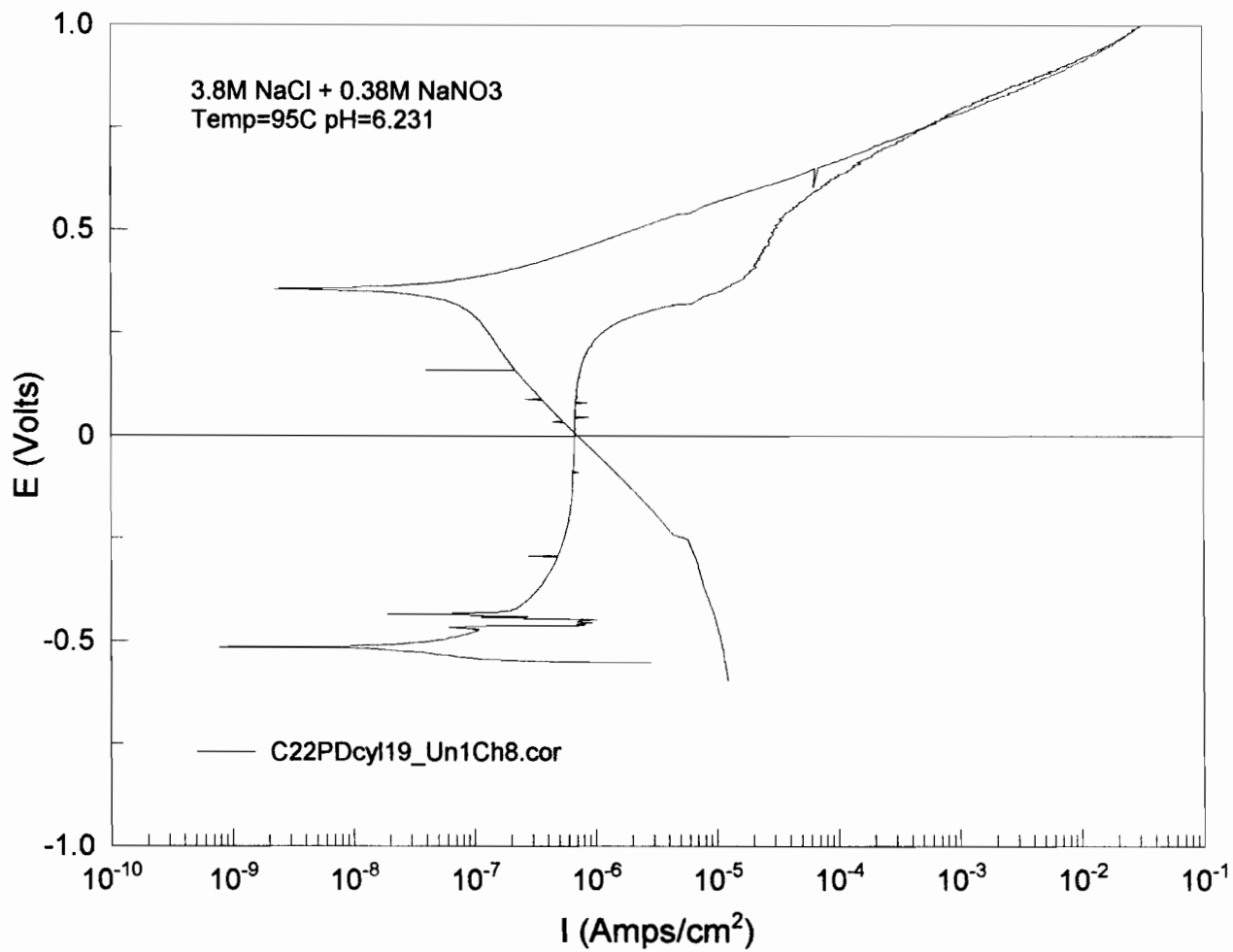
Potentiostat: Solatron 1480 SN# 00238265

Specimen Examination: No corrosion or pitting - Gold tint

Staining on All Surfaces of Specimen

Data C22 POCyl 1g * Re Run of Test #18

A. E. J. 9/15/04



B. K. S.
9/16/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\approx \frac{1}{2}$ liter
*Neat # ~~2277-8-3266~~
 2277-3-3266 ⁸⁷ 6/27/06
 Lot*

NaHCO ₃	48.156 g	# 028924
NaCl	14.997 "	# 035509
KCl	123.09 g	# 105573 # 006242

Reagents measured with

Model: *ORAU5*SN: *2883*Cal: *15 JUL 04*Due: *15 JAN 05*Counter Electrode: *Pt flag* Reference Electrode: *Ag/AgCl in house*Gas: *N₂ (99.999%)* Ecorr: *-267 mV* *w/ 3M KCl** Applied: *+415 mV* Potentiostat: *ESC 440-2* SN: *9209138*

Specimen Visual:

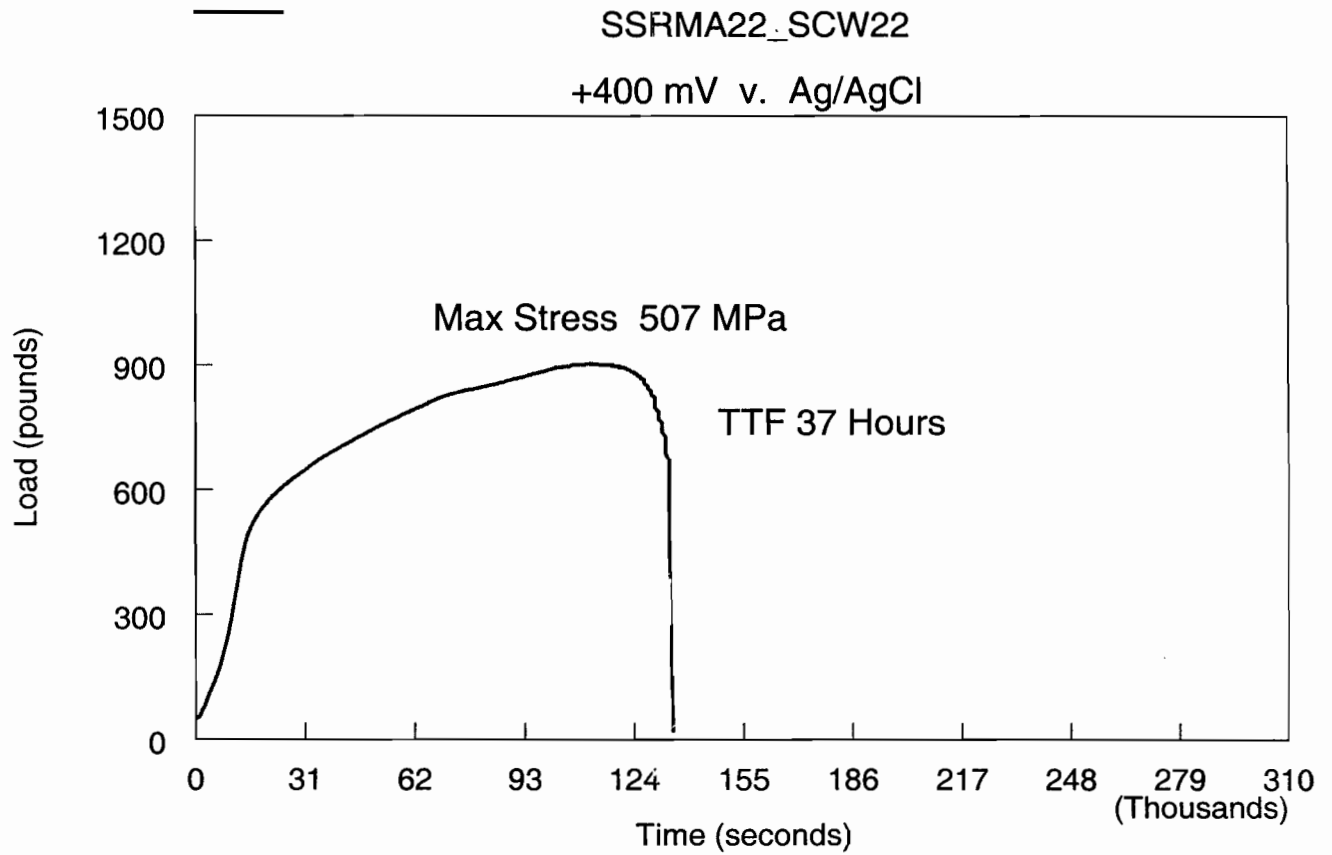
*silver-gray metallic;
 much secondary cracking; brittle
 failure*

$$E^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp effect from RT to 95°C $\Delta 13 \text{ mV}$* data file: *SSRMA22-SCW22**Walter J. Mackowski
 10/24/04*

Slow Strain Rate Test

0.5M NaCl; 3.3M KCl; NaHCO₃ as in SCW



Walter J. Mockroski
10/21/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: * $\frac{1}{2}$ liter~~Heat # 2277-8-3266~~
2277-3-3266 ^{3kg} 6/27/04
pH 8.69

NaCl 14.926 g #35309

NaHCO₃ 48.10 g #02894

KCl 264.83 g #006242

post pH 9.98

(pH not adjusted)

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 15 JAN 04

Due: 15 JAN 05

Counter Electrode: Pt flag Reference Electrode: Ag/AgCl in home

Gas: N₂ (99.999) Ecorr: -290mV

* 13M KCl

Applied: * 415mV

Potentiostat: ESL440-2 SN: 9209138

Specimen Visual:

*silver-gray metallic;
severe secondary cracking; brittle failure*

$$E^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

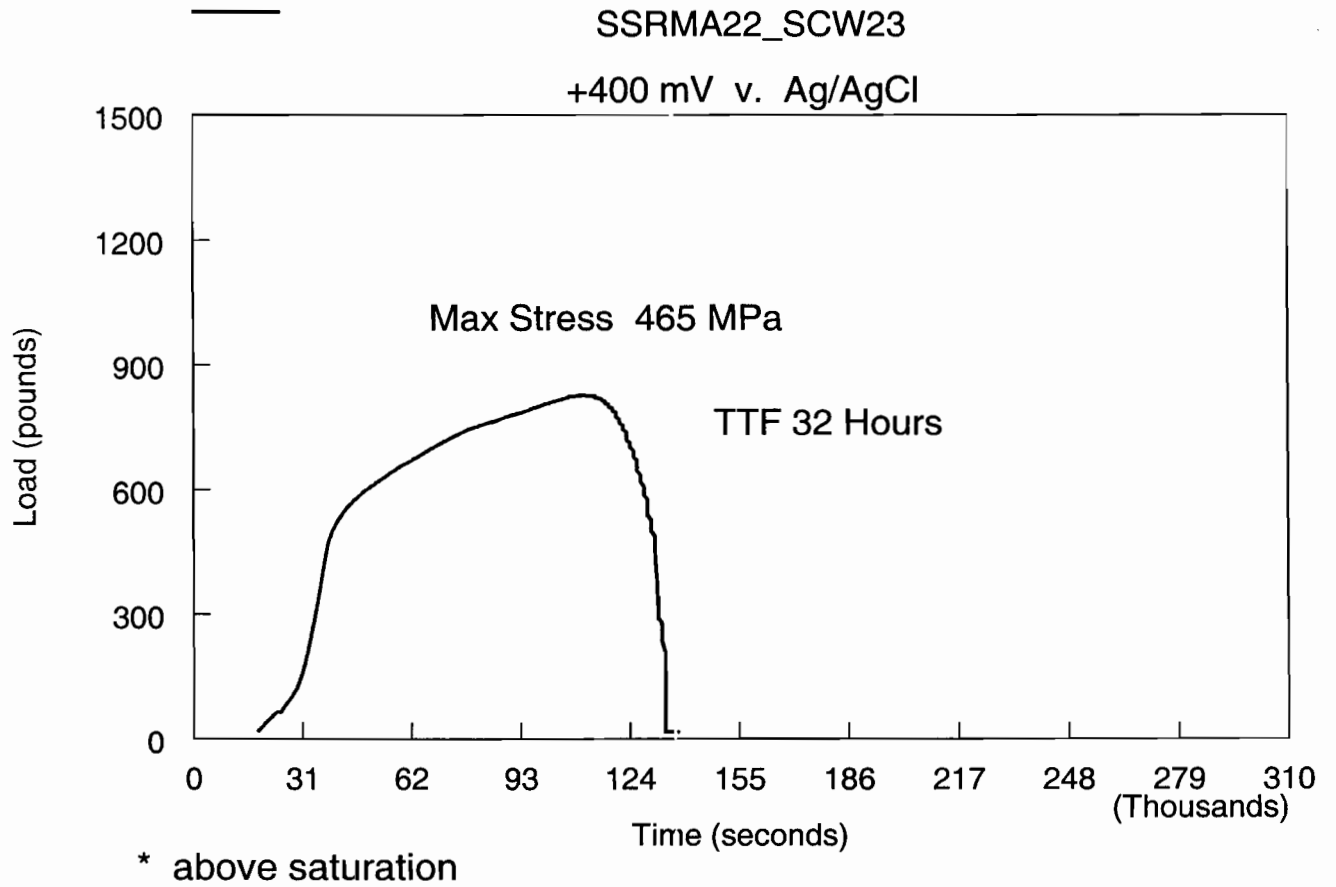
** Temp effect from RT to 95°C 415mV*

data file: SSRMA22-SCW23

*Walter J. Macchiarri
10/25/04*

Slow Strain Rate Test

0.5M NaCl; 7.1M* KCl; NaHCO₃ as in SCW



Walter J. Macroschi
10/25/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 2277-3-3266 ⁸¹⁹ 6/27/06~~Heat # 2277-8-3266~~

Solution: x 1/2 liter

NaCl 14.903g # 035509

NaHCO₃ 48.20g # 028924

KCl 264.83g # 006242

pH 8.62

not adj

final pH 10.24

Reagents measured with

Model: *ORION*

SN: 2883

Cal: 15 JUL 04

Due: 15 JAN 05

Counter Electrode: *Pt flag* Reference Electrode: *Ag/AgCl in h₂O*Gas: *N₂ (99.999%)* Ecorr: -280mV*w/3M KCl*Applied: +200mV Potentiostat: *FSC440-2* SN: 9209138

Specimen Visual:

*silver metallic; looks
like ductile failure*

$$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

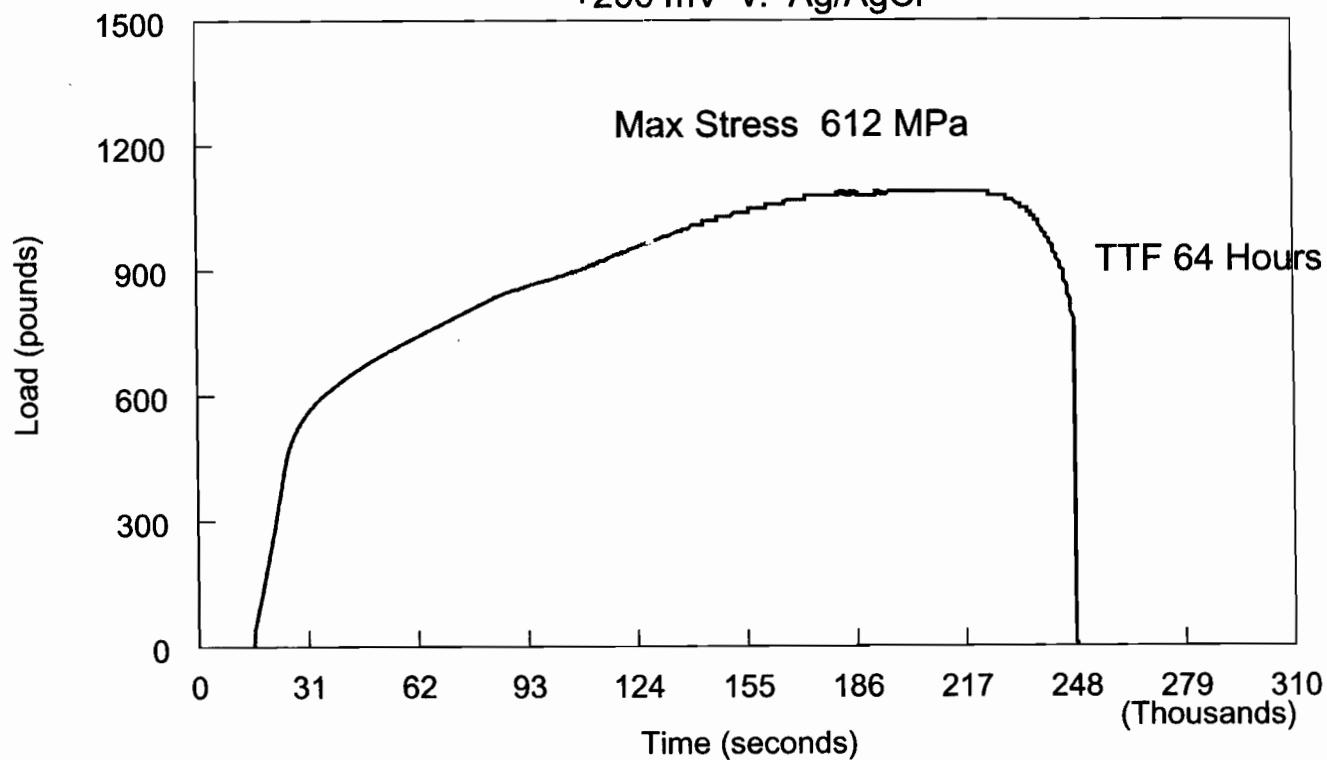
*data file: SSRMA22-SCW24**Walter J. Macfarlane**11/5/04**10/5/04 wjm*

Slow Strain Rate Test

0.5M NaCl; 7.1M* KCl; NaHCO₃ as in SCW

SSRMA22_SCW24

+200 mV v. Ag/AgCl



* above saturation

*Walter J. Michrowski**11/5/04*

SLOW STRAIN RATE TESTObjective: see page #5 *notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\approx \frac{1}{2}$ liter~~Heat # 2277-8-3266~~
2277-3-3266 ^{SP} 6/27/01

NaCl	14.924 g	# 035509	pH 8.62
NaHCO ₃	48.340	# 028924	not adj
KCl	264.75	006242	final pH 10.20

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 1554204

Due: 1554205

Counter Electrode: Pt flag Reference Electrode: Ag/AgCl in brine

Gas: N₂ (99.999%) Ecorr: -296 mV w/3M KCl

Applied: +100 mV Potentiostat: ECS 440-2 SN: 9209138

Specimen Visual:

*silver-metallic; looks like
ductile fracture*

$$\epsilon^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

data file: SSRMA22-SCW25

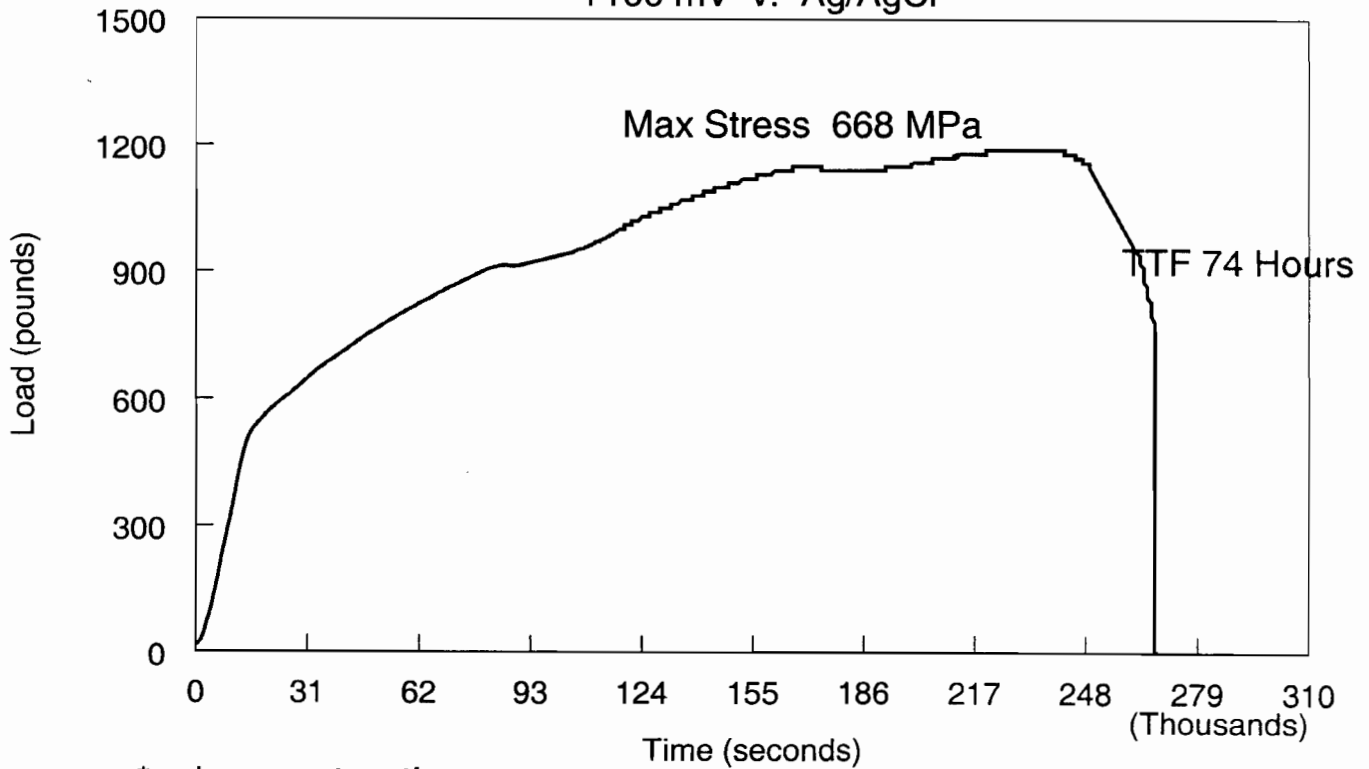
Walter J. MacKowski
11/12/04

Slow Strain Rate Test

0.5M NaCl; 7.1M* KCl; NaHCO3 as in SCW

SSRMA22_SCW25

+100 mV v. Ag/AgCl



* above saturation

Walter J. Macfarlane
11/12/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\approx \frac{1}{2}$ liter *Heat # ~~2277-8-3266~~ ^{BDO} 2277-3-3266 6/27/06*

NaCl	14.934 g	# 041475	pH 8.69
NaHCO ₃	48.18	# 028924	wt adj.
KCl	264.80	{ 006242 005573	final pH 10.83

Reagents measured with	Model: <i>OXTR45</i>	SN: 2883
	Cal: 15 JUN 04	Due: 15 JUN 05

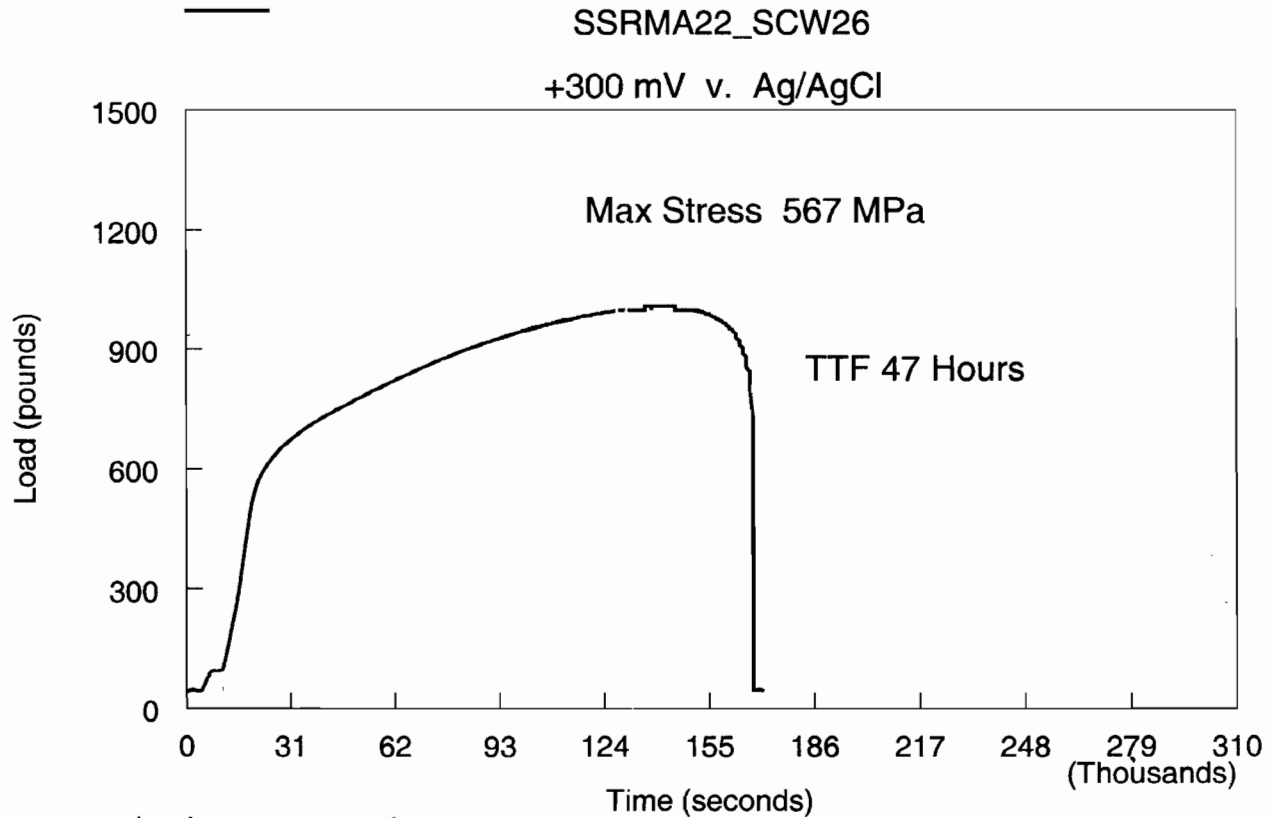
Counter Electrode: *Pt flag* Reference Electrode: *Ag/AgCl in house*Gas: *N₂ (99.999%)* Ecorr: *-355 mV* *4/3M KCl*Eapplied: *+300 mV* Potentiostat: *KESC440-2* SN: *9209138*Specimen Visual: *silver-gray metallic;
secondary cracking; looks like
brittle fracture*

$$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

data file: *SSRMA22-SOW26**Walter J. MacKowski*
11/15/04

Slow Strain Rate Test

0.5M NaCl; 7.1M* KCl; NaHCO₃ as in SCW



* above saturation

Walter J. Mochowski
11/15/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution:

*Glycerin**Heat # ~~2277-8-3206~~
2277-3-3266 ⁷⁰ 6/27/06*

Reagents measured with

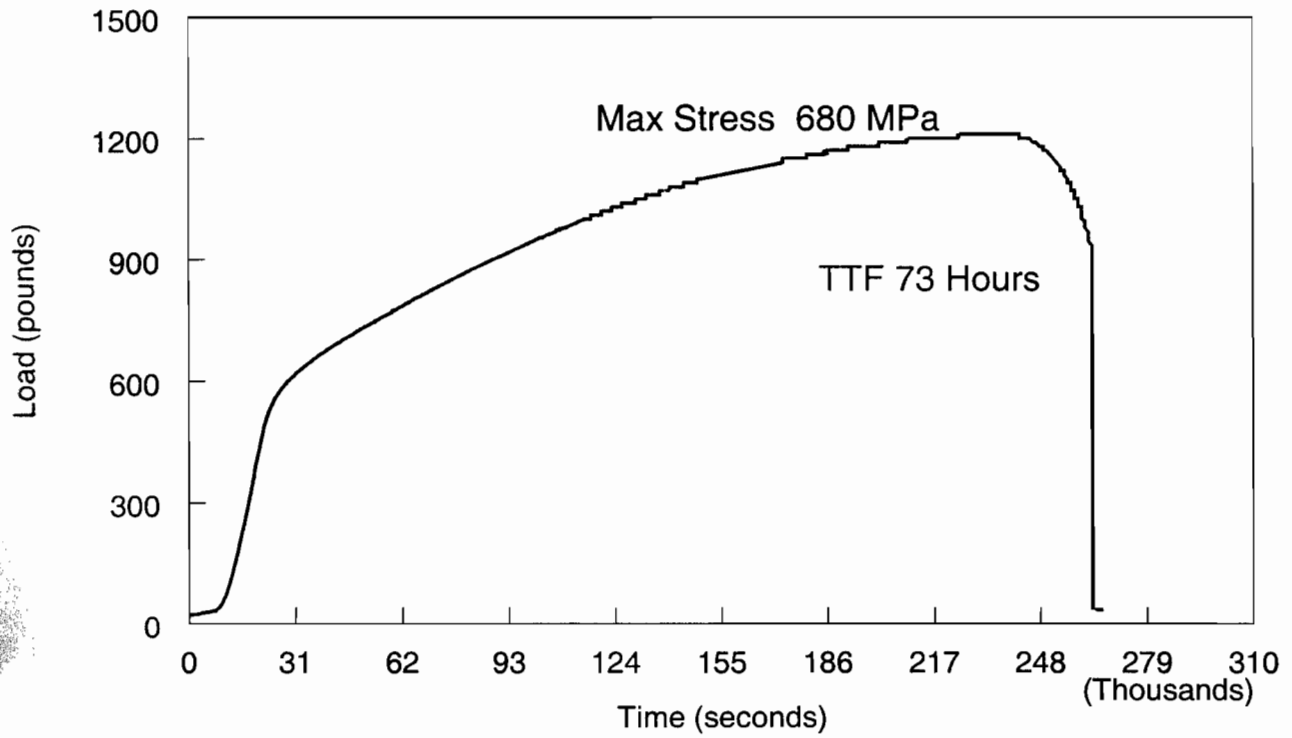
Model: *NA*
Cal: *NA*SN: *NA*
Due: *NA*Counter Electrode: *NA*Reference Electrode: *NA*Gas: *N₂ (99.999)* Ecorr: *NA*Eapplied: *NA*Potentiostat: *NA*SN: *NA*

Specimen Visual:

*silver metallic, ductile failure**file: SSRMA22-SCW27**Walter J. Mackowski
11/23/04*

Slow Strain Rate Test Glycerin @ 95 C

SSRMA22_SCW27



Walter J. Macintosh

11/23/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\approx \frac{1}{2}$ literHeat # ~~2277-8-3266~~ ^{BLW}
2277-3-3266 6/27/04

NaCl	14.92 g	# 041475
NaHCO ₃	48.13	# 028924
KCl	264.83	# 005573
NaNO ₃	44.29	020809

pH_{in} 7.41
↓ no adjpH_{final} 9.56

Reagents measured with

Model: OHAUS
Cal: 15 JUL 04SN: 2883
Due: 15 JAN 05Counter Electrode: Pt flag Reference Electrode: Ag/AgCl in house
w/3M KClGas: N₂ (99.999) Ecorr: -326 mV* Applied: +400 mV Potentiostat: ~~ESC 400-2~~ SN: 9209138

Specimen Visual:

w/duy
6/27/04 ESC 440-2

metallic silver-gray; secondary cracking
brittle failure

$$\epsilon^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* did not apply extra 15 mV as previously done

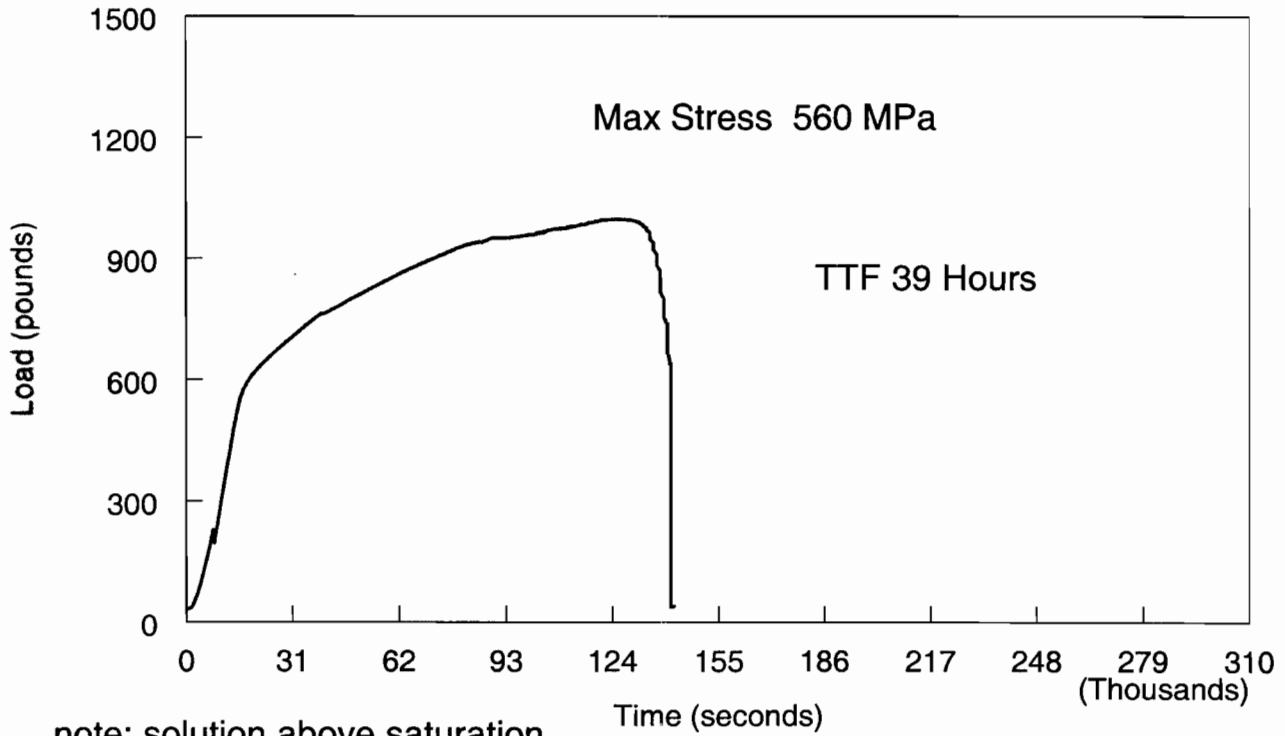
data file: SSRMA22-SCW28

Walter J. Macherush
12/13/04

Slow Strain Rate Test

0.5M NaCl; ~~0.6M~~ KCl; 0.38M NaNO₃; NaCO₃ as in SCW

— **7.1M** SSRMA22_SCW28
wjm 12/13/04



Walter J. Nuckowski
12/13/04

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\approx \frac{1}{2}$ literHeat # ~~2277-8-3266~~ ²²⁷⁷⁻³⁻³²⁶⁶ 6/27/06

NaCl	14.72g	# 041475	pH _{ini} 7.87
NaHCO ₃	48.07	028924	↓ w/adj
KCl	123.10	005573	pH _{final} 9.44
NaNO ₃	44.24	020809	

Reagents measured with

Model: *ORION S*SN: *2883*Cal: *15 JUL 04*Due: *15 JAN 05*Counter Electrode: *Pt flag*Reference Electrode: *Ag/AgCl in home*Gas: *N₂ (99.999)*Ecorr: *-350 mV**w/3M KCl** Applied: *+415*
*+400 mV*Potentiostat: *ECS 440-2* SN: *9209138*

Specimen Visual:

*grayish; secondary cracking;
little failure*

$$\epsilon^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* difference in T from RT to 95°C $\Delta +15 \text{ mV}$

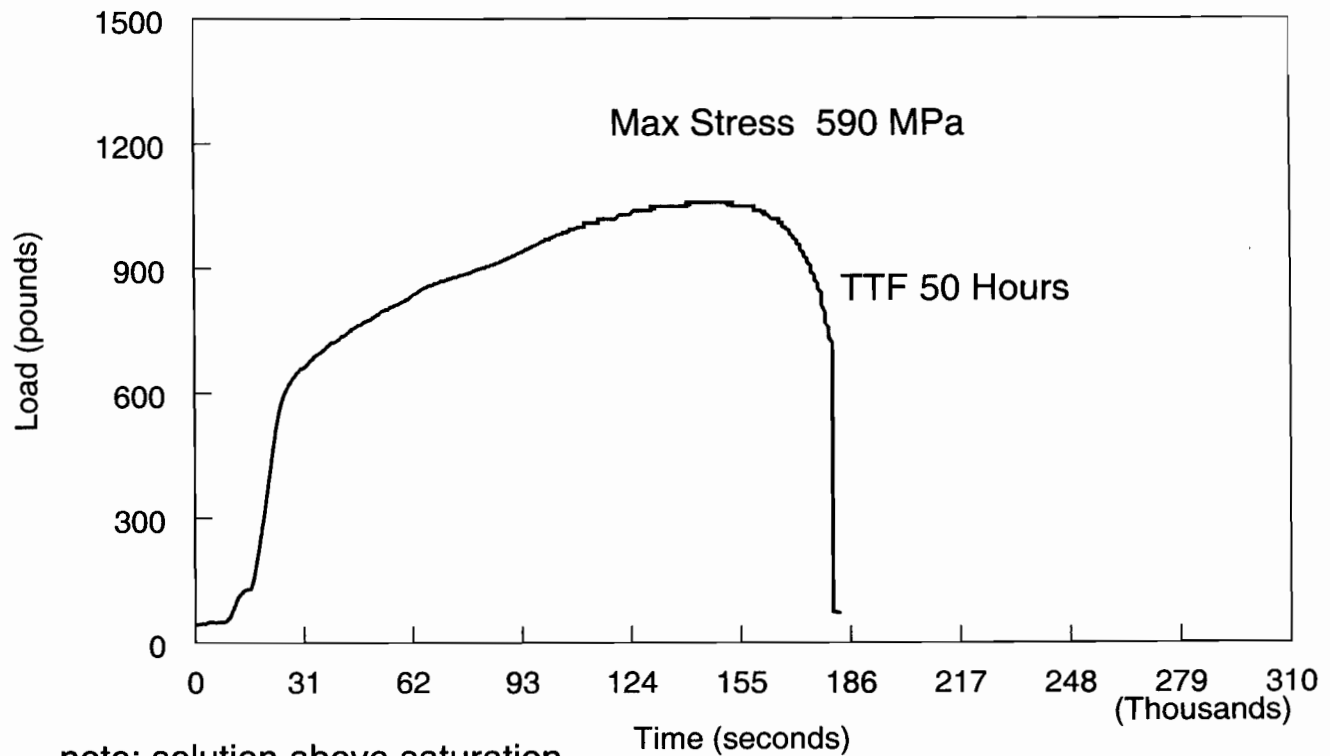
data file: *SSRMA22_SW29*

Walter J. MacLuski
12/17/04

Slow Strain Rate Test

0.5M NaCl; 3.3M KCl; 0.38M NaNO₃; NaCO₃ as in SCW

SSRMA22_SCW29



note: solution above saturation

Walter J. Macchowski
12/17/04

Calculation Results
Calculation was conducted using OLI StreamAnalyzer 1.2

SCW01					
Calculation Summary		Calculation Summary		Calculation Summary	
SinglePoint1 Calculation for Work 1		SinglePoint1_NaOH Calculation for Work 1		SinglePoint1_NaOH_850C Calculation for Work 1	
Automatic Chemistry Model Databank: Public		Automatic Chemistry Model Databank: Public		Automatic Chemistry Model Databank: Public	
Unit Set: Default		Unit Set: Default		Unit Set: Default	
Isothermal Calculation Temperature: 25 °C Pressure: 1 atm		Isothermal Calculation Temperature: 25 °C Pressure: 1 atm		Isothermal Calculation Temperature: 95 °C Pressure: 1 atm	
Stream Inflows		Stream Inflows		Stream Inflows	
Water: 2000 g Sodium chloride: 10.862 g Potassium chloride: 12.956 g Sodium nitrate: 17.503 g Sodium sulfate: 41.384 g Sodium bicarbonate: 192.684 g Sodium fluoride: 8.185 g Sodium hydroxide: 0 g		Water: 2000 g Sodium chl: 10.862 g Potassium: 12.956 g Sodium nit: 17.503 g Sodium sul: 41.384 g Sodium bic: 192.684 g Sodium flu: 8.185 g Sodium hy: 25 g		Water: 2000 g Sodium chl: 10.862 g Potassium: 12.956 g Sodium nit: 17.503 g Sodium sul: 41.384 g Sodium bic: 192.684 g Sodium flu: 8.185 g Sodium hy: 25 g	
Speciation Summary		Speciation Summary		Speciation Summary	
Total number of species: 113		Total number of species: 113		Total number of species: 113	
User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Species Liquid Sp H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 HCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KNHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl CH-1 HN03 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - V KF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2O KH2O K2CO3 HF - Aq KOH 2H2O K2CO3.1.H+1 KOH.1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HN03 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH.2H2O KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF.Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2SR SO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaHCO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4		User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Species Liquid Sp H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KNHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl CH-1 HN03 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - V KF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2O KH2O K2CO3 HF - Aq KOH 2H2O K2CO3.1.H+1 KOH.1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HN03 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH.2H2O KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF.Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2SR SO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaHCO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4		User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Species Liquid Sp H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KNHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl CH-1 HN03 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - V KF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2O KH2O K2CO3 HF - Aq KOH 2H2O K2CO3.1.H+1 KOH.1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HN03 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH.2H2O KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF.Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2SR SO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaHCO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4	
Stream Parameters		Stream Parameters		Stream Parameters	
Stream Am: 2281.6 g Temperature: 25 °C Pressure: 1 atm pH: 7.6633 pH Ionic Strength: 1.5687 mol/kg H2O Osmotic Pressure: 70.757 atm Water Activity: 0.94658 Activity Electrical Cond. specific: 0.080133 1/ohm Electrical Cond. molar: 50.942 cm/ohm-mol Viscosity, absolute: 1.3987 cP Viscosity, relative: 1.5703 cP/cP H2O		Stream Am: 2306.6 g Temperature: 25 °C Pressure: 1 atm pH: 8.0496 pH Ionic Stre: 2.0967 mol/kg H2O Osmotic Pr: 73.217 atm Water Actv: 0.94844 Activity Electrical C: 0.092006 1/ohm Electrical C: 49.19 cm/ohm-mol Viscosity, : 1.4813 cP Viscosity, : 1.6531 cP/cP H2O		Stream Am: 2306.6 g Temperature: 95 °C Pressure: 1 atm pH: 8.0607 pH Ionic Stre: 2.0783 mol/kg H2O Osmotic Pr: 83.095 atm Water Actv: 0.96083 Activity Electrical C: 0.22901 1/ohm Electrical C: 127.43 cm/ohm-mol Viscosity, : 0.52864 cP Viscosity, : 1.7804 cP/cP H2O	
Density		Density		Density	
Total Aqueous Vapor Solid 2nd Liquid g/ml g/ml g/ml g/ml		Total Aqueous Vapor Solid 2nd Liquid g/ml g/ml g/ml g/ml		Total Aqueous Vapor Solid 2nd Liquid g/ml g/ml g/ml g/ml	
Enthalpy		Enthalpy		Enthalpy	
Total and Phase Flows (Amounts)		Total and Phase Flows (Amounts)		Total and Phase Flows (Amounts)	
Mole		Mole		Mole	
Mass		Mass		Mass	
Volume		Volume		Volume	
Scaling Tendencies		Scaling Tendencies		Scaling Tendencies	
solids within temperature n Temperature Range		solids within Temperature Range		solids within Temperature Range	
Sodium bicarbonate 1 0 200.00 °C inside range		Sodium bic 0.33407 0 200.00 °C inside range		Sodium bic 1 0 200.00 °C inside range	
Sodium fluoride 0.10306 0 100.00 °C inside range		Sodium flu 0.11544 0 100.00 °C inside range		Sodium flu 0.096034 0 100.00 °C inside range	
Sodium sulfate decahydrate 0.050002 0 32.400 °C inside range		Sodium sul 0.078199 0 32.400 °C inside range		Sodium sul 0.0225147 19 241.00 °C inside range	
Sodium fluoride anhydrous 0.033332 20 40.000 °C inside range		Sodium sul 0.056571 0 32.400 °C inside range		Sodium sul 0.011471 35 37 109.00 °C inside range	
Potassium bicarbonate 0.011273 0 70.000 °C inside range		Sodium flu 0.042782 20 40.000 °C inside range		Hexaoxide 8.51E-03 30 150.00 °C inside range	
Sodium sulfate 8.74E-03 19 241.00 °C inside range		Sodium sul 0.92E-03 19 241.00 °C inside range		Sodium zn 3.28E-03 0 350.00 °C inside range	
Sodium carbonate decahyd 4.07E-03 0 32.000 °C inside range		Potassium 8.94E-03 0 70.000 °C inside range		Sodium nit 7.44E-04 0 300.00 °C inside range	
Potassium nitrate(VI) 3.12E-03 0 110.00 °C inside range		Potassium 3.17E-03 0 110.00 °C inside range		Potassium 8.89E-04 0 110.00 °C inside range	
Sodium thiosulfate 2.49E-03 0 350.00 °C inside range		Sodium chl 2.30E-03 0 350.00 °C inside range		Potassium 3.51E-04 0 200.00 °C inside range	
Sodium nitrate 1.87E-03 0 300.00 °C inside range		Sodium nit 1.90E-03 0 300.00 °C inside range		Potassium 3.91E-04 9 7 292.00 °C inside range	
Potassium sulfite(VI) 9.00E-04 9 7 292.00 °C inside range		Potassium 8.08E-04 9 7 292.00 °C inside range		Potassium 6.61E-04 0 200.00 °C inside range	
Potassium chloride 6.48E-04 0 200.00 °C inside range		Potassium 8.74E-04 0 200.00 °C inside range		Potassium 3.29E-06 17 7 40.00 °C inside range	
		Potassium hy 5.86E-06 data valid 8 inside range		Sodium hy 9.56E-07 data valid 8 inside range	

R. J. Ching
12/19/04

Table with 3 columns: Input/Output, Value, Units/Conditions. Includes Potassium fluoride dihydrate, Sodium hydrogen difluoride, Potassium carbonate 1.5 hr, Potassium bisulfate(VI), Sodium bisulfate, Potassium hydroxide dihydrate, Sodium hydroxide monohydrate, Sodium sulfate bisulfate, Potassium hydroxide.

Table with 3 columns: Input/Output, Value, Units/Conditions. Includes Potassium, Sodium hydr, Potassium, Potassium, Sodium hydr, Potassium, Sodium bis, Sodium sulf, Potassium.

Table with 3 columns: Input/Output, Value, Units/Conditions. Includes Potassium, Sodium, Potassium, Potassium.

Table with 3 columns: Input/Output, Value, Units/Conditions. Includes Potassium, Sodium, Potassium, Sodium, Potassium, Sodium.

Species Output (True Species)

Table with 5 columns: Species, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists various chemical species like H2O, HCl, NaNO3, etc.

Species Output (True Species)

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Molecular Output (Apparent Species)

Table with 5 columns: Species, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists various chemical species like H2O, NaOH, CO2, etc.

Molecular Output (Apparent Species)

Table with 5 columns: Species, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists various chemical species like H2O, NaOH, CO2, etc.

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Molecular Output (Apparent Species)

Table with 5 columns: Species, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists various chemical species like H2O, NaOH, CO2, etc.

Element Balance

Table with 5 columns: Element, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists elements like C, Cl, F, H, N, Na, O, S.

Element Balance

Table with 5 columns: Element, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists elements like C, Cl, F, H, N, Na, O, S.

Element Balance

Table with 5 columns: Element, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists elements like C, Cl, F, H, N, Na, O, S.

Element Balance

Table with 5 columns: Element, Total, Aqueous, Vapor, Solid, 2nd Liquid. Lists elements like C, Cl, F, H, N, Na, O, S.

Species Activity Coefficients

Table with 5 columns: Species, Value. Lists activity coefficients for various species like H2O, HCO3-1, etc.

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Species Activity Coefficients

Table with 5 columns: Species, Value. Lists activity coefficients for various species like H2O, HCO3-1, etc.

Species K(eq)-Values

Table with 2 columns: Species, Value. Lists K values for various species like H2O, HCO3-1, etc.

Species K(eq)-Values

Table with 2 columns: Species, Value. Lists K values for various species like H2O, HCO3-1, etc.

Species K(eq)-Values

Table with 2 columns: Species, Value. Lists K values for various species like H2O, HCO3-1, etc.

Species K(eq)-Values

Table with 2 columns: Species, Value. Lists K values for various species like H2O, HCO3-1, etc.

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Table with 2 columns: Species Name and Value. Includes species like (HF)2-Aq, HSO4-1, HSO4-2, CO3-2, Cl-1, Na2F+1, F-1, HF2-1, H+1, OH-1, NO3-1, K+1, KSO4-1, NaHCO3-1, Na+1, NaSO4-1, SO4-2.

Species Mobilities

Table with 2 columns: Species Name and Value. Includes species like HCO3-1, HSO4-1, CO3-2, Cl-1, Na2F+1, F-1, HF2-1, H+1, OH-1, NO3-1, K+1, KSO4-1, NaHCO3-1, Na+1, NaSO4-1, SO4-2.

Species Self Diffusivities

Table with 2 columns: Species Name and Value. Includes species like H2O, HCO3-1, HSO4-1, CO2-Aq, CO3-2, Cl-1, Na2F+1, F-1, (HF)2-Aq, HCl-Aq, HF-Aq, H+1, OH-1, NO3-1, HNO3-Aq, KHSO4-Aq, KCl-Aq, K+1, KSO4-1, NaHCO3-Aq, NaCO3-1, NaF-Aq, Na+1, NaNO3-Aq, NaSO4-1, SO4-2, SO3-Aq, H2SO4-Aq.

Table with 2 columns: Species Name and Value. Includes species like (HF)2-Vaq, HCl-Aq, HCl-Vap, HF2-1, HF-Aq, HNO3-Aq, HNO3-Va, HNO3, KHSO4-A, KHSO4, K2CO3, K2CO3.1E, KCl-Aq, KCl, KF, KF.2H2O, KF.4H2O, KOH, KOH.2H2C, KOH.1H2C, KNO3, K2SO4, KSO4-1, K2SO4.1H, NaHCO3, NaHCO3-Aq, NaHCO3, NaHSO4, Na2CO3, Na2CO3.1, Na2CO3.7, Na2CO3.7, Na2CO3.11, NaCl, NaF-Aq, NaF, NaF.Na2Si, NaHF2, NaOH, NaOH.1H2, NaNO3-A, NaNO3, Na2SO4, Na2SO4.1, Na2SO4.N, Na2SO4.1, NaSO4-1, SO3-Aq, SO3-Vap, H2SO4-A, H2SO4-V.

Species Mobilities

Table with 2 columns: Species Name and Value. Includes species like HCO3-1, HSO4-1, CO3-2, Cl-1, Na2F+1, F-1, HF2-1, H+1, OH-1, NO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Species Self Diffusivities

Table with 2 columns: Species Name and Value. Includes species like H2O, HCO3-1, HSO4-1, CO2-Aq, CO3-2, Cl-1, Na2F+1, F-1, (HF)2-Aq, HCl-Aq, HF-Aq, H+1, OH-1, NO3-1, HNO3-Aq, KHSO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3-A, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A.

Table with 2 columns: Species Name and Value. Includes species like HF-Vap, HNO3-Aq, HNO3-Va, HNO3, KHSO4-A, KHSO4, K2CO3, K2CO3.1E, KCl-Aq, KCl, KF, KF.2H2O, KF.4H2O, KOH, KOH.2H2C, KOH.1H2C, KNO3, K2SO4, KSO4-1, K2SO4.1H, NaHCO3, NaHCO3, NaHSO4, Na2CO3, Na2CO3.1, Na2CO3.7, Na2CO3.7, Na2CO3.11, NaCl, NaF-Aq, NaF, NaF.Na2Si, NaHF2, NaOH, NaOH.1H2, NaNO3-A, NaNO3, Na2SO4, Na2SO4.1, Na2SO4.N, Na2SO4.1, NaSO4-1, SO3-Aq, SO3-Vap, H2SO4-A, H2SO4-V.

Species Mobilities

Table with 2 columns: Species Name and Value. Includes species like HCO3-1, HSO4-1, CO3-2, Cl-1, Na2F+1, F-1, HF2-1, H+1, OH-1, NO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Species Self Diffusivities

Table with 2 columns: Species Name and Value. Includes species like H2O, HCO3-1, HSO4-1, CO2-Aq, CO3-2, Cl-1, Na2F+1, F-1, (HF)2-Aq, HCl-Aq, HF-Aq, H+1, OH-1, NO3-1, HNO3-Aq, KHSO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3-A, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A.

Table with 2 columns: Species Name and Value. Includes species like HF-Vap, HNO3-Aq, HNO3-Va, HNO3, KHSO4-A, KHSO4, K2CO3, K2CO3.1E, KCl-Aq, KCl, KF, KF.2H2O, KF.4H2O, KOH, KOH.2H2C, KOH.1H2C, KNO3, K2SO4, KSO4-1, K2SO4.1H, NaHCO3, NaHCO3, NaHSO4, Na2CO3, Na2CO3.1, Na2CO3.7, Na2CO3.7, Na2CO3.11, NaCl, NaF-Aq, NaF, NaF.Na2Si, NaHF2, NaOH, NaOH.1H2, NaNO3-A, NaNO3, Na2SO4, Na2SO4.1, Na2SO4.N, Na2SO4.1, NaSO4-1, SO3-Aq, SO3-Vap, H2SO4-A, H2SO4-V.

Species Mobilities

Table with 2 columns: Species Name and Value. Includes species like HCO3-1, HSO4-1, CO3-2, Cl-1, Na2F+1, F-1, HF2-1, H+1, OH-1, NO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Species Self Diffusivities

Table with 2 columns: Species Name and Value. Includes species like H2O, HCO3-1, HSO4-1, CO2-Aq, CO3-2, Cl-1, Na2F+1, F-1, (HF)2-Aq, HCl-Aq, HF-Aq, H+1, OH-1, NO3-1, HNO3-Aq, KHSO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3-A, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A.

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SCW03										SCW04														
Calculation Summary SinglePoint_NaOH Calculation for Work1 Automatic Chemistry Model Database: Public Unit Set: Default Isothermal Calculation Temperature: 25 °C Pressure: 1 atm					Calculation Summary SinglePoint_NaOH_95oC Calculation for Work1 Automatic Chemistry Model Database: Public Unit Set: Default Isothermal Calculation Temperature: 95 °C Pressure: 1 atm					Calculation Summary SinglePoint1 Calculation for Work1 Automatic Chemistry Model Database: Public Unit Set: Default Isothermal Calculation Temperature: 25 °C Pressure: 1 atm					Calculation Summary SinglePoint_NaOH Calculation for Work1 Automatic Chemistry Model Database: Public Unit Set: Default Isothermal Calculation Temperature: 25 °C Pressure: 1 atm					Calculation Summary SinglePoint Automatic Chemistry Model Database: Public Unit Set: Default Isothermal Calculation Temperature: 25 °C Pressure: 1 atm				
Stream Inflows Water: 1000 g Sodium chl: 5.438 g Potassium: 6.479 g Sodium ntr: 0 g Sodium sul: 20.699 g Sodium bic: 96.339 g Sodium flu: 3.068 g Sodium hy: 9 g					Stream Inflows Water: 1000 g Sodium chl: 5.438 g Potassium: 6.479 g Sodium ntr: 0 g Sodium sul: 20.699 g Sodium bic: 96.339 g Sodium flu: 3.068 g Sodium hy: 9 g					Stream Inflows Water: 1000 g Sodium chl: 5.43 g Potassium: 6.48 g Sodium ntr: 8.749 g Sodium sul: 0 g Sodium bic: 96.341 g Sodium flu: 3.109 g Sodium hy: 0 g					Stream Inflows Water: 1000 g Sodium chl: 5.43 g Potassium: 6.48 g Sodium ntr: 8.749 g Sodium sul: 0 g Sodium bic: 96.341 g Sodium flu: 3.109 g Sodium hy: 9 g					Stream Inflows Water: 1000 g Sodium chl: 5.43 g Potassium: 6.48 g Sodium ntr: 8.749 g Sodium sul: 0 g Sodium bic: 96.341 g Sodium flu: 3.109 g Sodium hy: 9 g				
Species Summary Total number of species: 113					Species Summary Total number of species: 113					Species Summary Total number of species: 113					Species Summary Total number of species: 113					Species Summary Total number of species: 113				
User Inflows Related Inflows Aqueous Species Vapor Species Solid Species Second Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KNaHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.IC02 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF.2H2O HClO4 HCl - Aq KHCO3 KHSO4 HF-1 KOH K2CO3 HF - Aq KOH.2H2O K2CO3.1.EH+1 KOH.1H2O KF OH-1 KNO3 KF.2H2O NO3-1 K2SO4 KF.4H2O HNO3 - Aq K2SO4.1H2O KOH KHSO4 - Aq NaHCO3 KOH.2H2C KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1NaNO3 - Aq NaF.Na2SO4 Na2CO3.7NaSO4-1 NaHF2 Na2CO3.1ISO4-2 NaOH NaF.Na2SiSO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4					User Inflows Related Inflows Aqueous Species Vapor Species Solid Species Second Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KNaHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.IC02 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF.2H2O HClO4 HCl - Aq KHCO3 KHSO4 HF-1 KOH K2CO3 HF - Aq KOH.2H2O K2CO3.1.EH+1 KOH.1H2O KF OH-1 KNO3 KF.2H2O NO3-1 K2SO4 KF.4H2O HNO3 - Aq K2SO4.1H2O KOH KHSO4 - Aq NaHCO3 KOH.2H2C KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1NaNO3 - Aq NaF.Na2SO4 Na2CO3.7NaSO4-1 NaHF2 Na2CO3.1ISO4-2 NaOH NaF.Na2SiSO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4					User Inflows Related Inflows Aqueous Species Vapor Species Solid Species Second Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KNaHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.IC02 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF.2H2O HClO4 HCl - Aq KHCO3 KHSO4 HF-1 KOH K2CO3 HF - Aq KOH.2H2O K2CO3.1.EH+1 KOH.1H2O KF OH-1 KNO3 KF.2H2O NO3-1 K2SO4 KF.4H2O HNO3 - Aq K2SO4.1H2O KOH KHSO4 - Aq NaHCO3 KOH.2H2C KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1NaNO3 - Aq NaF.Na2SO4 Na2CO3.7NaSO4-1 NaHF2 Na2CO3.1ISO4-2 NaOH NaF.Na2SiSO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4					User Inflows Related Inflows Aqueous Species Vapor Species Solid Species Second Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KNaHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.IC02 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF.2H2O HClO4 HCl - Aq KHCO3 KHSO4 HF-1 KOH K2CO3 HF - Aq KOH.2H2O K2CO3.1.EH+1 KOH.1H2O KF OH-1 KNO3 KF.2H2O NO3-1 K2SO4 KF.4H2O HNO3 - Aq K2SO4.1H2O KOH KHSO4 - Aq NaHCO3 KOH.2H2C KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1NaNO3 - Aq NaF.Na2SO4 Na2CO3.7NaSO4-1 NaHF2 Na2CO3.1ISO4-2 NaOH NaF.Na2SiSO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4					User Inflows Related Inflows Aqueous Species Vapor Species Solid Species Second Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KNaHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.IC02 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Va K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)6 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF.2H2O HClO4 HCl - Aq KHCO3 KHSO4 HF-1 KOH K2CO3 HF - Aq KOH.2H2O K2CO3.1.EH+1 KOH.1H2O KF OH-1 KNO3 KF.2H2O NO3-1 K2SO4 KF.4H2O HNO3 - Aq K2SO4.1H2O KOH KHSO4 - Aq NaHCO3 KOH.2H2C KCl - Aq NaHSO4 KOH.1H2C K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H NaCO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1NaNO3 - Aq NaF.Na2SO4 Na2CO3.7NaSO4-1 NaHF2 Na2CO3.1ISO4-2 NaOH NaF.Na2SiSO3 - Aq NaOH.1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH.1H2O Na2SO4 Na2SO4 Na2SO4.NaHSO4 Na2SO4.10H2O Na2SO4.10H2O SO3 H2SO4				
Stream Parameters Stream Am: 1141 g Temperature: 25 °C Pressure: 1 atm pH: 8.7918 Ionic Stron: 1.8686 mol/kg H2O Osmotic Pi: 68.794 atm WaterActiv: 0.95132 Activity Electrical C: 0.084842 1/ohm Electrical C: 50.594 cm2/ohm-mol Viscosity, r: 1.4549 cP Viscosity, r: 1.6334 cP/cP H2O					Stream Parameters Stream Am: 1141 g Temperature: 95 °C Pressure: 1 atm pH: 8.7029 Ionic Stron: 1.8158 mol/kg H2O Osmotic Pi: 77.557 atm WaterActiv: 0.95365 Activity Electrical C: 0.20673 1/ohm Electrical C: 128.09 cm2/ohm-mol Viscosity, r: 0.51849 cP Viscosity, r: 1.7452 cP/cP H2O					Stream Parameters Stream Am: 1120.1 g Temperature: 25 °C Pressure: 1 atm pH: 7.7199 Ionic Stron: 1.3556 mol/kg H2O Osmotic Pi: 67.753 atm WaterActiv: 0.95154 Activity Electrical C: 0.074641 1/ohm Electrical C: 51.954 cm2/ohm-mol Viscosity, r: 1.3695 cP Viscosity, r: 1.5364 cP/cP H2O					Stream Parameters Stream Am: 1129.1 g Temperature: 25 °C Pressure: 1 atm pH: 8.8146 Ionic Stron: 1.6915 mol/kg H2O Osmotic Pi: 66.471 atm WaterActiv: 0.95283 Activity Electrical C: 0.082646 1/ohm Electrical C: 48.969 cm2/ohm-mol Viscosity, r: 1.4047 cP Viscosity, r: 1.5771 cP/cP H2O					Stream Parameters Stream Am: 1129.1 g Temperature: 25 °C Pressure: 1 atm pH: 8.8146 Ionic Stron: 1.6915 mol/kg H2O Osmotic Pi: 66.471 atm WaterActiv: 0.95283 Activity Electrical C: 0.082646 1/ohm Electrical C: 48.969 cm2/ohm-mol Viscosity, r: 1.4047 cP Viscosity, r: 1.5771 cP/cP H2O				
Total and Phase Flows (Amounts) Total Mole: 58.923 Total Mass: 1141 g Total Volume: 1.0457 L					Total and Phase Flows (Amounts) Total Mole: 58.79 Total Mass: 1141 g Total Volume: 1.0669 L					Total and Phase Flows (Amounts) Total Mole: 58.373 Total Mass: 1120.1 g Total Volume: 1.0466 L					Total and Phase Flows (Amounts) Total Mole: 58.769 Total Mass: 1129.1 g Total Volume: 1.0452 L					Total and Phase Flows (Amounts) Total Mole: 58.769 Total Mass: 1129.1 g Total Volume: 1.0452 L				
Scaling Tendencies solids with Temperature Range Sodium bic: 0.91633 0 200.00 °C inside range Sodium flu: 0.10781 0 100.00 °C inside range Sodium sul: 0.05269 0 32.400 °C inside range Sodium cai: 0.051344 0 32.000 °C inside range Sodium flu: 0.036404 20 40.000 °C inside range Potassium: 0.010188 0 70.000 °C inside range Sodium sul: 9.05E-03 19 241.00 °C inside range Sodium chl: 2.64E-03 0 350.00 °C inside range Potassium: 9.02E-04 9.7 292.00 °C inside range Potassium: 6.78E-04 0 200.00 °C inside range Potassium: 3.2E-06 17.7 40.200 °C inside range Sodium hy: 8.07E-08 data valid 0 inside range					Scaling Tendencies solids with Temperature Range Sodium bic: 0.34409 0 200.00 °C inside range Sodium flu: 0.073725 0 100.00 °C inside range Sodium sul: 0.022515 19 241.00 °C inside range Sodium cai: 7.31E-03 35.37 109.00 °C inside range Hexaacidur: 4.33E-03 30 150.00 °C inside range Sodium chl: 2.98E-03 0 350.00 °C inside range Potassium: 3.75E-04 9.7 292.00 °C inside range Potassium: 3.5E-04 0 200.00 °C inside range Glaserite: 3.31E-05 35 150.00 °C inside range Sodium hy: 6.07E-08 data valid 0 inside range Potassium: 3.7E-10 33 143.00 °C inside range Potassium: 2.92E-11 data valid 0 inside range					Scaling Tendencies solids with Temperature Range Sodium bic: 0.98994 0 200.00 °C inside range Sodium flu: 0.097127 0 100.00 °C inside range Potassium: 0.013043 0 70.000 °C inside range Sodium cai: 3.99E-03 0 32.000 °C inside range Potassium: 3.37E-03 0 110.00 °C inside range Sodium chl: 2.18E-03 0 350.00 °C inside range Sodium ntr: 1.54E-03 0 300.00 °C inside range Potassium: 6.54E-04 0 200.00 °C inside range Potassium: 3.61E-06 17.7 40.200 °C inside range Sodium hy: 8.0E-07 data valid 0 inside range Potassium: 3.8E-09 80.000 °C inside range Potassium: 1.29E-11 0 33.000 °C inside range					Scaling Tendencies solids with Temperature Range Sodium bic: 0.88195 0 200.00 °C inside range Sodium flu: 0.10425 0 100.00 °C inside range Sodium cai: 0.050008 0 32.000 °C inside range Potassium: 0.010779 0 70.000 °C inside range Potassium: 3.40E-03 0 110.00 °C inside range Sodium chl: 2.41E-03 0 350.00 °C inside range Sodium ntr: 1.67E-03 0 300.00 °C inside range Potassium: 6.80E-04 0 200.00 °C inside range Potassium: 3.0E-06 17.7 40.200 °C inside range Sodium hy: 7.5E-08 data valid 0 inside range Potassium: 3.02E-06 80.000 °C inside range Potassium: 1.67E-10 0 33.000 °C inside range					Scaling Tendencies solids with Temperature Range Sodium bic: 0.88195 0 200.00 °C inside range Sodium flu: 0.10425 0 100.00 °C inside range Sodium cai: 0.050008 0 32.000 °C inside range Potassium: 0.010779 0 70.000 °C inside range Potassium: 3.40E-03 0 110.00 °C inside range Sodium chl: 2.41E-03 0 350.00 °C inside range Sodium ntr: 1.67E-03 0 300.00 °C inside range Potassium: 6.80E-04 0 200.00 °C inside range Potassium: 3.0E-06 17.7 40.200 °C inside range Sodium hy: 7.5E-08 data valid 0 inside range Potassium: 3.02E-06 80.000 °C inside range Potassium: 1.67E-10 0 33.000 °C inside range				

K.T. Chiong
12/19/04

Summary		Calculation Summary		Calculation Summary		Calculation Summary		Calculation Summary	
L_NaOH_95oC Calculation for Work1		SinglePoint1 Calculation for Work1		SinglePoint_NaOH Calculation for Work1		SinglePoint_NaOH_95oC Calculation for Work1		SinglePoint1 Calculation	
Chemistry Model Databanks Public		Automatic Chemistry Model Databanks Public		Automatic Chemistry Model Databanks Public		Automatic Chemistry Model Databanks Public		Automatic Chemistry Model Databanks Public	
Unit Set: Default		Unit Set: Default		Unit Set: Default		Unit Set: Default		Unit Set: Default	
Calculation Temperature: 95 °C Pressure: 1 atm		Isothermal Calculation Temperature: 25 °C Pressure: 1 atm		Isothermal Calculation Temperature: 25 °C Pressure: 1 atm		Isothermal Calculation Temperature: 95 °C Pressure: 1 atm		Isothermal Calculation Temperature: 95 °C Pressure: 1 atm	
Components		Stream Inflows		Stream Inflows		Stream Inflows		Stream Inflows	
1000 g Water 5.43 g Sodium chl 6.46 g Potassium 8.749 g Sodium nit 0 g Sodium sul 96.341 g Sodium bic 3.198 g Sodium flu 9 g Sodium hy		Water 1000 g Sodium chl 5.34 g Potassium 6.46 g Sodium nit 8.754 g Sodium sul 30.693 g Sodium bic 96.352 g Sodium flu 0 g Sodium hy 0 g		Water 1000 g Sodium chl 5.34 g Potassium 6.46 g Sodium nit 8.754 g Sodium sul 30.693 g Sodium bic 96.352 g Sodium flu 0 g Sodium hy 0 g		Water 1000 g Sodium chl 5.34 g Potassium 6.46 g Sodium nit 8.754 g Sodium sul 30.693 g Sodium bic 96.352 g Sodium flu 0 g Sodium hy 0 g		Water 1000 g Sodium chl 5.34 g Potassium 6.46 g Sodium nit 8.758 g Sodium sul 20.697 g Sodium bic 96.349 g Sodium flu 3.098 g Sodium hy 0 g	
Summary		Speciation Summary		Speciation Summary		Speciation Summary		Speciation Summary	
Number of species: 113		Total number of species: 113		Total number of species: 113		Total number of species: 113		Total number of species: 113	
Related Inflows Aqueous Vapor Solid Second Liquid Species Species Liquid Species		User Inflows Related Aqueous Vapor Solid Second Liquid Species Species Liquid Species		User Inflows Related Aqueous Vapor Solid Second Liquid Species Species Liquid Species		User Inflows Related Aqueous Vapor Solid Second Liquid Species Species Liquid Species		User Inflows Related Aqueous Vapor Solid Second Liquid Species Species Liquid Species	
CO2 H2O H2O - Vap K2SO4 KNaSO4 H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 K2SO4 KHSO4-1 (HF)2 - Vap KHCO3 2Na2SO4 CO2 - Aq HCl - Vap KHSO4 (HF)2 CO3-2 HF - Vap K2CO3 HCl CH-1 HNO3 - Vap K2CO3 1.5H2O NaF Na2F+1 SO3 - Vap KCl (HF)6 F-1 H2SO4 - V/F HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KHSO4 HF-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3 1.E+1 KOH 1H2O KF CH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4 1H2O KOH KHSO4 - Aq NaHSO4 KOH 2H2KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3 10H2O K2SO4 NaHSO4 - Aq Na2CO3 7H2O K2SO4 1H NaCO3-1 Na2CO3 1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3 1H NaNO3 - Aq NaF Na2SO4 Na2CO3 3H NaSO4-1 NaHF2 Na2CO3 1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O SO3 H2SO4		H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4 CO2 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHSO4 HCl CH-1 HNO3 - Vap K2CO3 1.5H2O NaF HF Na2F+1 SO3 - Vap KCl (HF)6 F-1 H2SO4 - V/F HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KHSO4 HF-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3 1.E+1 KOH 1H2O KF CH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4 1H2O KOH KHSO4 - Aq NaHSO4 KOH 2H2KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3 10H2O K2SO4 NaHSO4 - Aq Na2CO3 7H2O K2SO4 1H NaCO3-1 Na2CO3 1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3 1H NaNO3 - Aq NaF Na2SO4 Na2CO3 3H NaSO4-1 NaHF2 Na2CO3 1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O SO3 H2SO4		H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4 CO2 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHSO4 HCl CH-1 HNO3 - Vap K2CO3 1.5H2O NaF HF Na2F+1 SO3 - Vap KCl (HF)6 F-1 H2SO4 - V/F HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KHSO4 HF-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3 1.E+1 KOH 1H2O KF CH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4 1H2O KOH KHSO4 - Aq NaHSO4 KOH 2H2KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3 10H2O K2SO4 NaHSO4 - Aq Na2CO3 7H2O K2SO4 1H NaCO3-1 Na2CO3 1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3 1H NaNO3 - Aq NaF Na2SO4 Na2CO3 3H NaSO4-1 NaHF2 Na2CO3 1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O SO3 H2SO4		H2O CO2 H2O H2O - Vap K2SO4 KNaSO4 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4 Na2CO3 KCl K2SO4 KHSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4 CO2 - Aq HCl - Vap KHSO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHSO4 HCl CH-1 HNO3 - Vap K2CO3 1.5H2O NaF HF Na2F+1 SO3 - Vap KCl (HF)6 F-1 H2SO4 - V/F HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KHSO4 HF-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3 1.E+1 KOH 1H2O KF CH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4 1H2O KOH KHSO4 - Aq NaHSO4 KOH 2H2KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3 10H2O K2SO4 NaHSO4 - Aq Na2CO3 7H2O K2SO4 1H NaCO3-1 Na2CO3 1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3 1H NaNO3 - Aq NaF Na2SO4 Na2CO3 3H NaSO4-1 NaHF2 Na2CO3 1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O SO3 H2SO4			
Parameters		Stream Parameters		Stream Parameters		Stream Parameters		Stream Parameters	
1129.1 g 96 °C 1 atm 8.736 pH 1.5641 mol/kg H2O 73.528 atm 0.95592 Activity 0.19702 1/ohm 123.7 cm2/ohm-mol 0.48913 cP 1.6796 cP/cP H2O		Stream Am 1147.6 g Temperature 25 °C Pressure 1 atm pH 7.6877 Ionic Stre 1.5325 mol/kg H2O Osmotic Pi 70.241 atm WaterActiv 0.94921 Activity Electrical C 0.008223 1/ohm Electrical C 51.599 cm2/ohm-mol Viscosity, i 1.3937 cP Viscosity, r 1.5847 cP/cP H2O		Stream Am 1156.6 g Temperature 25 °C Pressure 1 atm pH 8.7726 Ionic Stre 2.0332 mol/kg H2O Osmotic Pi 73.214 atm WaterActiv 0.94824 Activity Electrical C 0.009062 1/ohm Electrical C 50.827 cm2/ohm-mol Viscosity, i 1.4558 cP Viscosity, r 1.5344 cP/cP H2O		Stream Am 1156.6 g Temperature 95 °C Pressure 1 atm pH 9.689 Ionic Stre 2.0298 mol/kg H2O Osmotic Pi 82.853 atm WaterActiv 0.95045 Activity Electrical C 0.22258 1/ohm Electrical C 130.26 cm2/ohm-mol Viscosity, i 0.50444 cP Viscosity, r 1.8979 cP/cP H2O		Stream Am 1136.4 g Temperature 25 °C Pressure 1 atm pH 7.7015 Ionic Stre 1.5114 Osmotic Pi 65.356 WaterActiv 0.95329 Electrical C 0.075481 Electrical C 50.796 Viscosity, i 1.4049 Viscosity, r 1.5773	
Total Aqueous Vapor Solid 2nd Liquid g/g/g/g/g 1.0403 0 0 0 0 Density		Total Aqueous Vapor Solid 2nd Liquid g/g/g/g/g 1.0678 0 2.1907 0 0 Density		Total Aqueous Vapor Solid 2nd Liquid g/g/g/g/g 1.1008 0 0 0 0 Density		Total Aqueous Vapor Solid 2nd Liquid g/g/g/g/g 1.0574 0 0 0 0 Density			
Total Aqueous Vapor Solid 2nd Liquid cal cal cal cal cal -4.04E+06 -4.04E+06 0 0 0 Enthalpy		Total Aqueous Vapor Solid 2nd Liquid cal cal cal cal cal -4.15E+06 -4.13E+06 -1.6443 0 0 Enthalpy		Total Aqueous Vapor Solid 2nd Liquid cal cal cal cal cal -4.18E+06 -4.18E+06 0 0 0 Enthalpy		Total Aqueous Vapor Solid 2nd Liquid cal cal cal cal cal -4.10E+06 -4.10E+06 0 0 0 Enthalpy			
Phase Flows (Amounts)		Total and Phase Flows (Amounts)		Total and Phase Flows (Amounts)		Total and Phase Flows (Amounts)		Total and Phase Flows (Amounts)	
Total Aqueous Vapor Solid 2nd Liquid mol mol mol mol mol 58.594 58.594 0 0 0 Mole		Total Aqueous Vapor Solid 2nd Liquid mol mol mol mol mol 58.695 58.622 0 0.072489 0 Mole		Total Aqueous Vapor Solid 2nd Liquid mol mol mol mol mol 59.158 59.158 0 0 0 Mole		Total Aqueous Vapor Solid 2nd Liquid mol mol mol mol mol 59.029 59.029 0 0 0 Mole			
Total Aqueous Vapor Solid 2nd Liquid g g g g g 1129.1 1129.1 0 0 0 Mass		Total Aqueous Vapor Solid 2nd Liquid g g g g g 1147.6 1141.5 0 6.0896 0 Mass		Total Aqueous Vapor Solid 2nd Liquid g g g g g 1156.6 1156.6 0 0 0 Mass		Total Aqueous Vapor Solid 2nd Liquid g g g g g 1156.6 1156.6 0 0 0 Mass			
Total Aqueous Vapor Solid 2nd Liquid L L L L L 1.0654 1.0654 0 0 0 Volume		Total Aqueous Vapor Solid 2nd Liquid L L L L L 1.0622 1.0494 0 2.77E-03 0 Volume		Total Aqueous Vapor Solid 2nd Liquid L L L L L 1.0607 1.0507 0 0 0 Volume		Total Aqueous Vapor Solid 2nd Liquid L L L L L 1.0638 1.0638 0 0 0 Volume			
Scaling Tendencies		Scaling Tendencies		Scaling Tendencies		Scaling Tendencies		Scaling Tendencies	
Temperature Range 0.32549 0.200.00 °C inside range 0.099785 0.100.00 °C inside range 6.86E-03 35.37 109.00 °C inside range 2.22E-03 0.350.00 °C inside range 7.58E-04 0.110.00 °C inside range 6.83E-04 0.300.00 °C inside range 3.21E-04 0.200.00 °C inside range 7.31E-08 data valid 8 inside range 4.87E-10 33.143.00 °C inside range 2.87E-14 data valid 8 inside range		Solids with Temperature Range Sodium bic 1 0.200.00 °C inside range Sodium sul 0.075015 0.32.400 °C inside range Sodium sul 0.013072 19.241.00 °C inside range Sodium sul 0.010681 0.70.000 °C inside range Sodium sul 4.10E-03 0.32.000 °C inside range Potassium 3.00E-03 0.110.00 °C inside range Sodium chl 2.54E-03 0.350.00 °C inside range Sodium nit 1.89E-03 0.300.00 °C inside range Potassium 1.21E-03 9.7.292.00 °C inside range Potassium 6.29E-04 0.200.00 °C inside range Potassium 1.94E-09 80.000.00 °C inside range Potassium 2.09E-10 data valid 8 inside range		Solids with Temperature Range Sodium bic 0.07983 0.200.00 °C inside range Sodium sul 0.082144 0.32.400 °C inside range Sodium sul 0.064338 0.32.000 °C inside range Sodium sul 0.014568 19.241.00 °C inside range Potassium 9.41E-03 0.70.000 °C inside range Potassium 3.02E-03 0.110.00 °C inside range Sodium chl 2.88E-03 0.350.00 °C inside range Sodium nit 1.89E-03 0.300.00 °C inside range Potassium 1.10E-03 9.7.292.00 °C inside range Potassium 8.44E-04 0.200.00 °C inside range Potassium 2.14E-08 80.000.00 °C inside range Potassium 1.34E-10 0.33.000 °C inside range		Solids with Temperature Range Sodium bic 0.08048 0.200.00 °C inside range Sodium sul 0.096258 19.241.00 °C inside range Hexaacidul 0.012342 30.150.00 °C inside range Sodium sul 8.00E-03 35.37 109.00 °C inside range Sodium chl 3.58E-03 0.350.00 °C inside range Sodium nit 7.27E-04 0.300.00 °C inside range Potassium 6.12E-04 0.110.00 °C inside range Potassium 4.48E-04 9.7.292.00 °C inside range Potassium 3.64E-04 0.200.00 °C inside range Glauberite 5.44E-06 35.150.00 °C inside range Potassium 3.39E-10 33.143.00 °C inside range Potassium 3.65E-11 data valid 8 inside range		Solids with Temperature Range Sodium bic 1 0.090442 Sodium sul 0.047832 Sodium sul 0.029861 Potassium 0.01176 Sodium chl 8.04E-03 Sodium chl 4.12E-03 Potassium 3.12E-03 Sodium nit 1.59E-03 Sodium chl 1.14E-03 Potassium 9.01E-04 Potassium 3.10E-04	

K. T. Chiao
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Table with 2 columns: chemical species and numerical values. Includes species like KH2O, K2CO3, KCl, etc.

obilities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, CO2, HCl, etc.

Diffusivities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, CO2, HCl, etc.

Table with 2 columns: chemical species and numerical values. Includes species like KH2O, K2CO3, KCl, etc.

Species Mobilities

Table with 2 columns: chemical species and numerical values. Includes species like HCO3-, HSO4-, CO3-2, etc.

Species Self Diffusivities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, HCO3-, HSO4-, etc.

Table with 2 columns: chemical species and numerical values. Includes species like K2CO3, KCl, KF, etc.

Species Mobilities

Table with 2 columns: chemical species and numerical values. Includes species like HCO3-, HSO4-, CO3-2, etc.

Species Self Diffusivities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, HCO3-, HSO4-, etc.

Table with 2 columns: chemical species and numerical values. Includes species like KCl, KF, K2H2O, etc.

Species Mobilities

Table with 2 columns: chemical species and numerical values. Includes species like HCO3-, HSO4-, CO3-2, etc.

Species Self Diffusivities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, HCO3-, HSO4-, etc.

Table with 2 columns: chemical species and numerical values. Includes species like (HF)2, HCl, HCl-Aq, etc.

Species Mobilities

Table with 2 columns: chemical species and numerical values. Includes species like HCO3-, HSO4-, CO3-2, etc.

Species Self Diffusivities

Table with 2 columns: chemical species and numerical values. Includes species like H2O, HCO3-, HSO4-, etc.

Handwritten signature 'K. J. Chiang' and date '12/19/04'.

		SCW08				SCW08	
Calculation Summary		Calculation Summary		Calculation Summary		Calculation Summary	
SinglePoint_NaOH Calculation for Work1		SinglePoint_NaOH_95C Calculation for Work1		SinglePoint1 Calculation for Work1		SinglePoint1 Calculation for Work1	
Automatic Chemistry Model		Automatic Chemistry Model		Automatic Chemistry Model		Automatic Chemistry Model	
Databases: Public		Databases: Public		Databases: Public		Databases: Public	
Unit Set: Default		Unit Set: Default		Unit Set: Default		Unit Set: Default	
Isothermal Calculation		Isothermal Calculation		Isothermal Calculation		Isothermal Calculation	
Temperature: 25 °C		Temperature: 95 °C		Temperature: 25 °C		Temperature: 25 °C	
Pressure: 1 atm		Pressure: 1 atm		Pressure: 1 atm		Pressure: 1 atm	
Stream Inflows		Stream Inflows		Stream Inflows		Stream Inflows	
Water	1000 g	Water	1000 g	Water	1000 g	Water	1000 g
Sodium chl	0 g	Sodium chl	222.27 g	Sodium chl	222.27 g	Sodium chl	222.27 g
Potassium	8.48 g	Potassium	8.48 g	Potassium	0 g	Potassium	0 g
Sodium nit	8.758 g	Sodium nit	8.758 g	Sodium nit	32.86 g	Sodium nit	32.86 g
Sodium sul	20.697 g	Sodium sul	20.697 g	Sodium sul	0 g	Sodium sul	0 g
Sodium bic	96.349 g	Sodium bic	96.349 g	Sodium bic	0 g	Sodium bic	0 g
Sodium flu	3.098 g	Sodium flu	3.098 g	Sodium flu	0 g	Sodium flu	0 g
Sodium hy	9 g	Sodium hy	9 g	Sodium hy	0 g	Sodium hy	0 g
Speciation Summary		Speciation Summary		Speciation Summary		Speciation Summary	
Total number of species: 113		Total number of species: 113		Total number of species: 113		Total number of species: 113	
Aqueous Vapor Solid Second Species Species Species Liquid Spe H2O H2O - Vap K2SO4.KNaSO4 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 HSO4-1 (HF)2 - Vap KHCO3 CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 Cl-1 HNO3 - Vap K2CO3.1.5H2O Na2F+1 SO3 - Vap KCl F-1 H2SO4 - VKF (HF)2 - Aq KF 2H2O HCl - Aq KF 4H2O KOH KH2SO4 HF - Aq KOH 2H2O H+1 KOH 1H2O OH-1 KOH NO3-1 KF 2H2O NO3-1 HNO3 - Aq KF 4H2O HNO3 - Aq KH2SO4 - Aq KOH KH2SO4 - Aq KCl - Aq KOH 2H2K KCl - Aq K+1 KOH 1H2K K+1 KSO4-1 KNO3 KSO4-1 NaHCO3 - Aq K2SO4 NaHCO3 - Aq Na2CO3 -1 K2SO4.1H Na2CO3-1 NaF - Aq NaHSO4 NaF - Aq Na+1 Na2CO3 Na+1 NaNO3 - Aq Na2CO3.1.NaNO3 - Aq NaSO4-1 NaHF2 Na2CO3.7 NaSO4-1 SO3 - Aq NaOH Na2CO3.1 SO4-2 H2SO4 - Aq NaF Na2S9 SO3 - Aq NaOH NaOH 1H2O Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 OH2O Na2SO4 10H2O	User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 HCO3-1 HCO3-1 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 NaCl K2SO4.KH HSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1 CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Vap K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)2 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2SO4 HF2-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3.1.5H+1 KOH 1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH 2H2K KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H Na2CO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O H2SO4	User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Liquid Spe H2O CO2 H2O H2O - Vap K2SO4.KNaSO4 HCO3-1 HCO3-1 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 NaCl K2SO4.KH HSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1 CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Vap K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)2 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2SO4 HF2-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3.1.5H+1 KOH 1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH 2H2K KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H Na2CO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O H2SO4	User Related Aqueous Vapor Solid Second Inflows Inflows Species Species Liquid Spe H2O CO2 H2O H2O - Vap 2Na2SO4.Na2CO3 NaCl H2CO3 HCO3-1 CO2 - Vap 2Na2SO4.Na2CO3 KCl K2SO4.KH HSO4-1 (HF)2 - Vap KHCO3 NaNO3 2Na2SO4.1 CO2 - Aq HCl - Vap KH2SO4 Na2SO4 (HF)2 CO3-2 HF - Vap K2CO3 NaHCO3 HCl Cl-1 HNO3 - Vap K2CO3.1.5H2O NaF HF Na2F+1 SO3 - Vap KCl NaOH (HF)2 F-1 H2SO4 - VKF HNO3 (HF)2 - Aq KF 2H2O KHCO3 HCl - Aq KF 4H2O KH2SO4 HF2-1 KOH K2CO3 HF - Aq KOH 2H2O K2CO3.1.5H+1 KOH 1H2O KF OH-1 KNO3 KF 2H2O NO3-1 K2SO4 KF 4H2O HNO3 - Aq K2SO4.1H2O KOH KH2SO4 - Aq NaHCO3 KOH 2H2K KCl - Aq NaHSO4 KOH 1H2K K+1 Na2CO3 KNO3 KSO4-1 Na2CO3.10H2O K2SO4 NaHCO3 - Aq Na2CO3.7H2O K2SO4.1H Na2CO3-1 Na2CO3.1H2O NaHSO4 NaF - Aq NaCl Na2CO3 Na+1 NaF Na2CO3.1.NaNO3 - Aq NaF Na2SO4 Na2CO3.7 NaSO4-1 NaHF2 Na2CO3.1 SO4-2 NaOH NaF Na2S9 SO3 - Aq NaOH 1H2O NaHF2 H2SO4 - Aq NaNO3 NaOH 1H2O Na2SO4 Na2SO4 Na2SO4 Na2SO4 NaHSO4 Na2SO4 NaHSO4 Na2SO4 10H2O Na2SO4 10H2O H2SO4				
Stream Parameters		Stream Parameters		Stream Parameters		Stream Parameters	
Stream Am	1144.4 g	Stream Am	1144.4 g	Stream Am	1256 g	Stream Am	1256 g
Temperature	25 °C	Temperature	95 °C	Temperature	25 °C	Temperature	25 °C
Pressure	1 atm	Pressure	1 atm	Pressure	1 atm	Pressure	1 atm
pH	8.7905 pH	pH	8.712 pH	pH	6.9781 pH	pH	6.9781 pH
Ionic Stron	1.8711 mol/kg H2O	Ionic Stron	1.7814 mol/kg H2O	Ionic Stron	4.1096 mol/kg H2O	Ionic Stron	4.1096 mol/kg H2O
Osmotic Pi	66.896 atm	Osmotic Pi	74.889 atm	Osmotic Pi	236.74 atm	Osmotic Pi	236.74 atm
WaterActiv	0.99272 Activity	WaterActiv	0.98105 Activity	WaterActiv	0.94702 Activity	WaterActiv	0.94702 Activity
Electrical C	0.26365 1/ohm	Electrical C	0.22414 1/ohm	Electrical C	0.21771 1/ohm	Electrical C	0.21771 1/ohm
Electrical C	49.313 cm2/ohm-mol	Electrical C	124.7 cm2/ohm-mol	Electrical C	56.917 cm2/ohm-mol	Electrical C	56.917 cm2/ohm-mol
Viscosity, r	1.4634 cP	Viscosity, r	0.51744 cP	Viscosity, r	1.3684 cP	Viscosity, r	1.3684 cP
Viscosity, r	1.6318 cP/cP H2O	Viscosity, r	1.7416 cP/cP H2O	Viscosity, r	1.5362 cP/cP H2O	Viscosity, r	1.5362 cP/cP H2O
Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid
g/ml 0 1.0819	g/ml 0 1.0819	g/ml 0 1.0616	g/ml 0 1.0616	g/ml 0 1.1463	g/ml 0 1.1463	g/ml 0 1.1463	g/ml 0 1.1463
Density	Density	Density	Density	Density	Density	Density	Density
Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid
cal 0 -4.12E+06	cal 0 -4.12E+06	cal 0 -4.08E+06	cal 0 -4.08E+06	cal 0 -4.20E+06	cal 0 -4.20E+06	cal 0 -4.20E+06	cal 0 -4.20E+06
Enthalpy	Enthalpy	Enthalpy	Enthalpy	Enthalpy	Enthalpy	Enthalpy	Enthalpy
(Amounts)		(Amounts)		(Amounts)		(Amounts)	
Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid
mol 58.493	mol 58.937	mol 58.784	mol 58.784	mol 63.805	mol 63.805	mol 63.805	mol 63.805
Mole	Mole	Mole	Mole	Mole	Mole	Mole	Mole
Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid
g 1133.3	g 1144.4	g 1144.4	g 1144.4	g 1256	g 1256	g 1256	g 1256
Mass	Mass	Mass	Mass	Mass	Mass	Mass	Mass
Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid	Aqueous Vapor Solid 2nd Liquid	Total Aqueous Vapor Solid 2nd Liquid
L 1.0475	L 1.0471	L 1.068	L 1.068	L 1.0948	L 1.0948	L 1.0948	L 1.0948
Volume	Volume	Volume	Volume	Volume	Volume	Volume	Volume
Scaling Tendencies		Scaling Tendencies		Scaling Tendencies		Scaling Tendencies	
solids with Temperature Range		solids with Temperature Range		solids with Temperature Range		solids with Temperature Range	
0 200.00 °C inside range	Sodium bic 0.91798 0 200.00 °C inside range	0 200.00 °C inside range	Sodium bic 0.33639 0 200.00 °C inside range	0 350.00 °C inside range	Sodium chl 0.26333 0 350.00 °C inside range	0 350.00 °C inside range	Sodium chl 0.26333 0
0 100.00 °C inside range	Sodium flu 0.10757 0 100.00 °C inside range	0 100.00 °C inside range	Sodium flu 0.07298 0 100.00 °C inside range	0 300.00 °C inside range	Sodium nit 0.023261 0 300.00 °C inside range	0 300.00 °C inside range	Sodium nit 0.023261 0
0 32.405 °C inside range	Sodium sul 0.061844 0 32.405 °C inside range	0 32.405 °C inside range	Sodium sul 0.018467 19 241.00 °C inside range	0 80.00 °C inside range	Sodium hy 3.47E-12 12 80.00 °C inside range	0 80.00 °C inside range	Sodium hy 3.47E-12 12
20 40.000 °C inside range	Sodium cal 0.05173 0 32.000 °C inside range	0 32.000 °C inside range	Sodium cal 7.19E-03 35.37 108.00 °C inside range				
0 70.000 °C inside range	Sodium flu 3.52E-02 20 40.000 °C inside range	20 40.000 °C inside range	Hexaoxid 2.86E-03 30 150.00 °C inside range				
19 241.00 °C inside range	Potassium 9.98E-03 0 70.000 °C inside range	0 70.000 °C inside range	Sodium chl 1.63E-03 0 350.00 °C inside range				
0 32.000 °C inside range	Sodium sul 8.77E-03 19 241.00 °C inside range	19 241.00 °C inside range	Sodium chl 6.72E-04 0 300.00 °C inside range				
0 110.00 °C inside range	Potassium 3.15E-03 0 110.00 °C inside range	0 110.00 °C inside range	Potassium 6.96E-04 0 110.00 °C inside range				
0 350.00 °C inside range	Sodium nit 1.75E-03 0 300.00 °C inside range	0 300.00 °C inside range	Potassium 3.15E-04 97 292.00 °C inside range				
97 292.00 °C inside range	Sodium chl 1.27E-03 0 350.00 °C inside range	0 350.00 °C inside range	Potassium 1.98E-04 0 200.00 °C inside range				
0 200.00 °C inside range	Potassium 6.96E-04 97 292.00 °C inside range	97 292.00 °C inside range	Glauberite 2.31E-05 35 150.00 °C inside range				
	Potassium 3.18E-04 0 200.00 °C inside range	0 200.00 °C inside range	Sodium hy 7.87E-06 data valid if inside range				
Species Output (True Species)		Species Output (True Species)		Species Output (True Species)		Species Output (True Species)	
Total Aqueous Vapor Solid 2nd Liquid		Total Aqueous Vapor Solid 2nd Liquid		Total Aqueous Vapor Solid 2nd Liquid		Total Aqueous Vapor Solid 2nd Liquid	
g/mol	mol/kg H2O	g/mol	mol/kg H2O	g/mol	mol/kg H2O	g/mol	mol/kg H2O
H2O	1000	H2O	55.508	H2O	1000	H2O	55.508
NaNO3	6.6396	NaNO3	0.07812	NaNO3	6.6396	NaNO3	0.07812
HCl	3.31E-12	HCl	9.08E-14	HCl	3.31E-12	HCl	9.08E-14

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17.7 40.200 °C inside range
data valid if inside range
0 80.000 °C inside range
data valid if inside range
0 33.000 °C inside range
12 80.000 °C inside range
0 82.500 °C inside range
data valid if inside range

Potassium 3.21E-06 17.7 40.200 °C inside range
Sodium hy 8.09E-06 data valid if inside range
Potassium 2.46E-06 0 80.000 °C inside range
Potassium 1.31E-10 0 33.000 °C inside range
Sodium hy 7.82E-11 12 80.000 °C inside range
Potassium 1.04E-11 data valid if inside range
Sodium bis 1.97E-12 data valid if inside range
Sodium sul 1.71E-14 0 82.500 °C inside range
data valid if inside range
Potassium 5.28E-16 data valid if inside range

Potassium 3.86E-10 33 143.00 °C inside range
Potassium 2.41E-11 data valid if inside range
Sodium bis 1.80E-11 data valid if inside range
Potassium 2.42E-14 data valid if inside range

HNO3 2.37E-08 3.77E-10 0 0
Cl-1 134.84 3.8032 0 0
H+1 6.30E-08 6.25E-08 0 0
OH-1 1.07E-06 6.25E-08 0 0
NO3-1 18.907 0.30358 0 0
Na+1 94.48 4.1098 0 0

HNO3 2.37E-08 3.77E-10
Cl-1 134.84 3.8032
H+1 6.30E-08 6.25E-08
OH-1 1.07E-06 6.25E-08
NO3-1 18.907 0.30358
Na+1 94.48 4.1098

Species Output (True Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, KCl, NaNO3, NaF, CO2, HF2, HCl, HF, HNO3, KH2SO4, SO3, Na2F+1, F-1, HF2-1, OH-1, HNO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Molecular Output (Apparent Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, NaCl, NaNO3, NaOH, HCl, Element Balance, Total, Aqueous, Vapor, Solid, 2nd Liquid, Species Activity Coefficients, H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Molecular Output (Apparent Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, NaCl, NaNO3, NaOH, HCl, Element Balance, Total, Aqueous, Vapor, Solid, 2nd Liquid, Species Activity Coefficients, H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species Output (True Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, KCl, NaNO3, NaF, CO2, HF2, HCl, HF, HNO3, KH2SO4, SO3, Na2F+1, F-1, HF2-1, OH-1, HNO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Molecular Output (Apparent Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, NaCl, NaNO3, NaOH, HCl, Element Balance, Total, Aqueous, Vapor, Solid, 2nd Liquid, Species Activity Coefficients, H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species Output (True Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, KCl, NaNO3, NaF, CO2, HF2, HCl, HF, HNO3, KH2SO4, SO3, Na2F+1, F-1, HF2-1, OH-1, HNO3-1, K+1, KSO4-1, NaCO3-1, Na+1, NaSO4-1, SO4-2.

Molecular Output (Apparent Species)

Table with columns: Total, Aqueous, Vapor, Solid, 2nd Liquid. Rows include H2O, NaCl, NaNO3, NaOH, HCl, Element Balance, Total, Aqueous, Vapor, Solid, 2nd Liquid, Species Activity Coefficients, H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species K(eq)-Values

Table with columns: Species, Value. Rows include H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species K(eq)-Values

Table with columns: Species, Value. Rows include H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species K(eq)-Values

Table with columns: Species, Value. Rows include H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

Species K(eq)-Values

Table with columns: Species, Value. Rows include H2O, HNO3-1, HNO3-1, CO2-Aq, CO3-2, Na2F+1, Cl-1, Na2F+1, F-1, HF2-1, HCl-Aq, HF2-1, HF-Aq, H+1, OH-1, HNO3-Ac, KH2SO4-A, KCl-Aq, K+1, KSO4-1, NaHCO3, NaCO3-1, NaF-Aq, Na+1, NaNO3-A, NaSO4-1, SO4-2, SO3-Aq, H2SO4-A, H2O-Vap, CO2-Vap, HF2-Vap, HCl-Vap, HF-Vap, HNO3-Va, SO3-Vap, H2SO4-V.

K.T. Chao
12/18/04

N07		SCW11		SCW12	
Calculation Summary					
SinglePoint_NaOH_950C Calculation for Work1					
Automatic Chemistry Model					
Database: Public					
Unit Set: Default					
Isothermal Calculation					
Temperature: 25 °C					
Pressure: 1 atm					
Stream Inflows					
Water	1000 g	Water	500 g	Water	500 g
Sodium chl	222.27 g	Sodium chl	222.39 g	Sodium chl	2.72 g
Potassium	0 g	Potassium	0 g	Potassium	0 g
Sodium nit	32.66 g	Sodium nit	32.66 g	Sodium nit	4.342 g
Sodium sul	0 g	Sodium sul	0 g	Sodium sul	0 g
Sodium bic	0 g	Sodium bic	0 g	Sodium bic	0 g
Sodium flu	0 g	Sodium flu	0 g	Sodium flu	0 g
Sodium hy	1.60E-03 g	Sodium hy	0 g	Sodium hy	7.00E-04 g
Specification Summary					
Total number of species: 113					
User	Related	Aqueous	Vapor	Solid	Second
Species	Inflows	Species	Species	Species	Liquid Spec
H2O	Vap K2SO4 KNaSO4	H2O	H2O	Vap K2SO4 KNaSO4	
CO2	Vap 2Na2SO4 Na2CO3	NaCl	H2CO3	HCO3-1	CO2 - Vap
(HF)2	Vap KHCO3	KCl	K2SO4 KNHSO4-1	(HF)2 - Vap KHCO3	
HCl	Vap KHSO4	NaNO3	2Na2SO4.CO2 - Aq	HCl - Vap KHSO4	
HF	Vap K2CO3	Na2SO4	(HF)2 CO3-2	HF - Vap K2CO3	
HN03	Vap K2CO3.1.5H2O	NaHCO3	HCl	Cl-1	HN03 - Vap K2CO3.1.5H2O
SO3	Vap KCl	NaF	HF	Na2F+1	SO3 - Vap KCl
H2SO4	V KF	NaOH	(HF)6	F-1	H2SO4 - V KF
	KF.4H2O		HN03	(HF)2 - Aq	
	KF.4H2O		KHCO3	HCl - Aq	
	KOH		KHSO4	HF-2 - Aq	
	KOH.2H2O		K2CO3	HF - Aq	
	KOH.1H2O		K2CO3.1.EtH+1	KOH.1H2O	
	KNO3		KF	OH-1	
	K2SO4		KF.2H2O	NO3-1	
	K2SO4.1H2O		KF.4H2O	HN03 - Aq	
	NaHCO3		KOH	KHSO4 - Aq	
	NaHSO4		KOH.2H2O	KOH.2H2O KCl - Aq	
	Na2CO3		KOH.1H2O K+1	Na2CO3	
	Na2CO3.10H2O		KNO3	KS04-1	
	Na2CO3.7H2O		K2SO4	NaHCO3 - Aq	
	Na2CO3.1H2O		K2SO4.1H NaCO3-1	Na2CO3.1H2O	
	NaCl		NaHSO4	NaF - Aq	
	NaF		NaHSO4	NaF	
	NaF.Na2SO4		Na2CO3.1.NaNO3 - Aq	NaF.Na2SO4	
	NaHF2		Na2CO3.7 NaSO4-1	NaHF2	
	NaOH		Na2CO3.1 SO4-2	NaOH	
	NaOH.1H2O		NaF.Na2Si SO3 - Aq	NaOH.1H2O	
	NaHSO4		NaHF2	H2SO4 - Aq	
	Na2SO4		Na2SO4	NaOH.1H2O	
	Na2SO4		Na2SO4	Na2SO4	
	Na2SO4.NaHSO4		Na2SO4.NaHSO4	Na2SO4.NaHSO4	
	Na2SO4.10H2O		Na2SO4.10H2O	Na2SO4.10H2O	
	SO3		SO3	SO3	
	H2SO4		H2SO4	H2SO4	
Stream Parameters					
Stream Arr: 1255 g					
Temperature: 25 °C					
Pressure: 1 atm					
pH: 9.7817 pH					
Ionic Stron: 4.1066 mol/kg H2O					
Osmotic Ph: 236.74 atm					
WaterActiv: 0.84702 Activity					
Electrical C: 0.21771 l/ohm					
Electrical C: 59.816 cm2/ohm-mol					
Viscosity: 1.3654 cP					
Viscosity: 1.5363 cP/cP H2O					
Scaling Tendencies					
acids w/ln Temperature Range					
350.00 °C	inside range	Sodium chl	0.26334	0.350.00 °C	inside range
300.00 °C	inside range	Sodium nit	0.023261	0.300.00 °C	inside range
60.000 °C	inside range	Sodium hy	2.21E-09	12.60 000 °C	inside range
Species Output (True Species)					
Vapor	Solid	2nd Liquid	Total	Aqueous	Vapor
mol	mol	n/a	mol	mol/kg H2K mol	mol
0	0	0	H2O	1000	55.508
0	0	0	OH	8.8399	0.78121
0	0	0	OH	5.20E-15	1.43E-16

R.T. Clow
12/19/04

Table with 5 columns: Species, Total mol, Aqueous mol/kg H2O, Vapor mol, Solid mol, 2nd Liquid n/a. Rows include HNO3, Cl-, OH-, NO3-, Na+, H2O, HCl, NaOH, NaHSO4, Na2SO4, Na2CO3, Na2CO3.1f, NaCl, NaF, NaF.Na2Si, NaHF2, NaOH.1H2, NaOH.1H2, NaNO3.A. Includes Molecular Output and Element Balance sections.

Table with 5 columns: Species, Total mol, Aqueous mol/kg H2O, Vapor mol, Solid mol, 2nd Liquid n/a. Rows include HCl, HNO3, Cl-, OH-, NO3-, Na+, H2O, HCl, NaOH, NaHSO4, Na2SO4, Na2CO3, Na2CO3.1f, NaCl, NaF, NaF.Na2Si, NaHF2, NaOH.1H2, NaOH.1H2, NaNO3.A. Includes Molecular Output and Element Balance sections.

Table with 5 columns: Species, Total mol, Aqueous mol/kg H2O, Vapor mol, Solid mol, 2nd Liquid n/a. Rows include HNO3, Cl-, OH-, NO3-, Na+, H2O, HCl, NaOH, NaHSO4, Na2SO4, Na2CO3, Na2CO3.1f, NaCl, NaF, NaF.Na2Si, NaHF2, NaOH.1H2, NaOH.1H2, NaNO3.A. Includes Molecular Output and Element Balance sections.

K.J. Ching 12/19/04

NaNO3	28.748 mol%
Na2SO4	0.44942 mol%
Na2SO4	1.90E-07 mol%
Na2SO4.N	162.44 mol%
Na2SO4.H	0.046844 mol%
NaSO4-1	0.12008 mol%
SO3 - Aq	7728 mol%
SO3 - Vap	1.64E+09 mol%
H2SO4 - A	2.57E+10 mol%
H2SO4 - V	3.86E+06 mol%

Species Mobilities

Cl-1	3.64E-04 m2/ohm-mol
H+1	1.87E-03 m2/ohm-mol
OH-1	9.38E-04 m2/ohm-mol
NO3-1	3.40E-04 m2/ohm-mol
Na+1	2.39E-04 m2/ohm-mol

Species Self Diffusivities

H2O	1.67E-09 m2/s
HCO3-1	6.06E-10 m2/s
HSO4-1	9.06E-10 m2/s
CO2 - Aq	1.40E-09 m2/s
CO3-2	6.71E-10 m2/s
Cl-1	1.34E-09 m2/s
Na2F+1	1.06E-09 m2/s
F-1	1.00E-09 m2/s
(HF)2 - Aq	6.28E-10 m2/s
HCl - Aq	1.45E-09 m2/s
HF2-1	1.32E-09 m2/s
HF - Aq	9.12E-11 m2/s
H+1	5.68E-09 m2/s
OH-1	3.30E-09 m2/s
NO3-1	1.26E-09 m2/s
HNO3 - Aq	1.35E-09 m2/s
KHSO4 - A	1.06E-09 m2/s
KCl - Aq	1.23E-09 m2/s
K+1	1.31E-09 m2/s
KSO4-1	6.82E-10 m2/s
NaHCO3-1	7.07E-10 m2/s
NaCO3-1	5.90E-10 m2/s
NaF - Aq	7.81E-10 m2/s
Na+1	9.30E-10 m2/s
NaNO3 - A	6.58E-10 m2/s
NaSO4-1	6.28E-10 m2/s
SO4-2	7.12E-10 m2/s
SO3 - Aq	7.66E-11 m2/s
H2SO4 - A	7.51E-10 m2/s

NaNO3 - A	17.412 mol%
NaNO3	28.748 mol%
Na2SO4	0.44942 mol%
Na2SO4	1.90E-07 mol%
Na2SO4.N	162.44 mol%
Na2SO4.H	0.046844 mol%
NaSO4-1	0.12008 mol%
SO3 - Aq	7728 mol%
SO3 - Vap	1.64E+09 mol%
H2SO4 - A	2.57E+10 mol%
H2SO4 - V	3.86E+06 mol%

Species Mobilities

Cl-1	2.85E-04 m2/ohm-mol
H+1	1.31E-03 m2/ohm-mol
OH-1	7.35E-04 m2/ohm-mol
NO3-1	2.57E-04 m2/ohm-mol
Na+1	1.87E-04 m2/ohm-mol

Species Self Diffusivities

H2O	1.17E-09 m2/s
HCO3-1	5.66E-10 m2/s
HSO4-1	6.25E-10 m2/s
CO2 - Aq	9.97E-10 m2/s
CO3-2	4.66E-10 m2/s
Cl-1	9.36E-10 m2/s
Na2F+1	7.39E-10 m2/s
F-1	7.17E-10 m2/s
(HF)2 - Aq	5.74E-10 m2/s
HCl - Aq	1.01E-09 m2/s
HF2-1	9.22E-10 m2/s
HF - Aq	6.35E-11 m2/s
H+1	4.17E-09 m2/s
OH-1	2.57E-09 m2/s
NO3-1	8.76E-10 m2/s
HNO3 - Aq	9.36E-10 m2/s
KHSO4 - A	7.26E-10 m2/s
KCl - Aq	9.07E-10 m2/s
K+1	9.36E-10 m2/s
KSO4-1	4.67E-10 m2/s
NaHCO3-1	4.91E-10 m2/s
NaCO3-1	4.18E-10 m2/s
NaF - Aq	5.36E-10 m2/s
Na+1	6.81E-10 m2/s
NaNO3 - A	5.80E-10 m2/s
NaSO4-1	4.31E-10 m2/s
SO4-2	4.86E-10 m2/s
SO3 - Aq	5.28E-11 m2/s
H2SO4 - A	5.13E-10 m2/s

NaNO3	28.748 mol%
Na2SO4	0.44942 mol%
Na2SO4	1.90E-07 mol%
Na2SO4.N	162.44 mol%
Na2SO4.H	0.046844 mol%
NaSO4-1	0.12008 mol%
SO3 - Aq	7728 mol%
SO3 - Vap	1.64E+09 mol%
H2SO4 - A	2.57E+10 mol%
H2SO4 - V	3.86E+06 mol%

Species Mobilities

Cl-1	6.29E-04 m2/ohm-mol
H+1	2.88E-03 m2/ohm-mol
OH-1	1.62E-03 m2/ohm-mol
NO3-1	5.88E-04 m2/ohm-mol
Na+1	4.12E-04 m2/ohm-mol

Species Self Diffusivities

H2O	2.28E-09 m2/s
HCO3-1	1.13E-09 m2/s
HSO4-1	1.30E-09 m2/s
CO2 - Aq	1.90E-09 m2/s
CO3-2	9.44E-10 m2/s
Cl-1	1.91E-09 m2/s
Na2F+1	1.51E-09 m2/s
F-1	1.39E-09 m2/s
(HF)2 - Aq	1.15E-09 m2/s
HCl - Aq	2.01E-09 m2/s
HF2-1	1.88E-09 m2/s
HF - Aq	1.26E-10 m2/s
H+1	8.32E-09 m2/s
OH-1	4.78E-09 m2/s
NO3-1	1.79E-09 m2/s
HNO3 - Aq	1.88E-09 m2/s
KHSO4 - A	1.48E-09 m2/s
KCl - Aq	1.57E-09 m2/s
K+1	1.93E-09 m2/s
KSO4-1	9.73E-10 m2/s
NaHCO3-1	9.82E-10 m2/s
NaCO3-1	8.46E-10 m2/s
NaF - Aq	1.09E-09 m2/s
Na+1	1.27E-09 m2/s
NaNO3 - A	1.20E-09 m2/s
NaSO4-1	8.94E-10 m2/s
SO4-2	1.02E-09 m2/s
SO3 - Aq	1.07E-10 m2/s
H2SO4 - A	1.08E-09 m2/s

NaNO3	28.748 mol%
Na2SO4	0.44942 mol%
Na2SO4	1.90E-07 mol%
Na2SO4.N	162.44 mol%
Na2SO4.H	0.046844 mol%
NaSO4-1	0.12008 mol%
SO3 - Aq	7728 mol%
SO3 - Vap	1.64E+09 mol%
H2SO4 - A	2.57E+10 mol%
H2SO4 - V	3.86E+06 mol%

Species Mobilities

Cl-1	6.29E-04 m2/ohm-mol
H+1	2.88E-03 m2/ohm-mol
OH-1	1.62E-03 m2/ohm-mol
NO3-1	5.88E-04 m2/ohm-mol
Na+1	4.12E-04 m2/ohm-mol

Species Self Diffusivities

H2O	2.28E-09 m2/s
HCO3-1	1.13E-09 m2/s
HSO4-1	1.30E-09 m2/s
CO2 - Aq	1.90E-09 m2/s
CO3-2	9.44E-10 m2/s
Cl-1	1.91E-09 m2/s
Na2F+1	1.51E-09 m2/s
F-1	1.39E-09 m2/s
(HF)2 - Aq	1.15E-09 m2/s
HCl - Aq	2.01E-09 m2/s
HF2-1	1.88E-09 m2/s
HF - Aq	1.26E-10 m2/s
H+1	8.32E-09 m2/s
OH-1	4.78E-09 m2/s
NO3-1	1.79E-09 m2/s
HNO3 - Aq	1.88E-09 m2/s
KHSO4 - A	1.48E-09 m2/s
KCl - Aq	1.57E-09 m2/s
K+1	1.93E-09 m2/s
KSO4-1	9.73E-10 m2/s
NaHCO3-1	9.82E-10 m2/s
NaCO3-1	8.46E-10 m2/s
NaF - Aq	1.09E-09 m2/s
Na+1	1.27E-09 m2/s
NaNO3 - A	1.20E-09 m2/s
NaSO4-1	8.94E-10 m2/s
SO4-2	1.02E-09 m2/s
SO3 - Aq	1.07E-10 m2/s
H2SO4 - A	1.08E-09 m2/s

K.J. Ching
12/19/04

Na2SO4 0.14445 mol%
 Na2SO4 2.13E-03 mol%
 Na2SO4 N 155.99 mol%
 Na2SO4 II 26.308 mol%
 NaSO4-I 0.060006 mol%
 SO3 - Aq 1922.2 mol%
 SO3 - Vap 7879.0 mol%
 H2SO4 - A 3.19E+10 mol%
 H2SO4 - V 248.07 mol%

Species Mobilities

Cl-1 1.62E-03 m2/ohm-mol
 H+1 5.02E-03 m2/ohm-mol
 OH-1 3.50E-03 m2/ohm-mol
 NOS-1 1.45E-03 m2/ohm-mol
 Na+1 1.18E-03 m2/ohm-mol

Species Self Diffusivities

H2O 7.73E-09 m2/s
 HCO3-1 9.92E-09 m2/s
 HSO4-1 4.18E-09 m2/s
 CO2 - Aq 6.73E-09 m2/s
 CO3-2 3.49E-09 m2/s
 Cl-1 6.18E-09 m2/s
 Na2F+1 5.57E-09 m2/s
 F-1 4.90E-09 m2/s
 (HF)2 - Aq 4.09E-09 m2/s
 HCl - Aq 6.48E-09 m2/s
 HF2-1 6.90E-09 m2/s
 HF - Aq 4.87E-10 m2/s
 H+1 1.83E-09 m2/s
 OH-1 1.29E-09 m2/s
 NOS-1 5.55E-09 m2/s
 HNO3 - Aq 5.81E-09 m2/s
 KH2SO4 - A 5.51E-09 m2/s
 KCl - Aq 5.01E-09 m2/s
 K+1 5.77E-09 m2/s
 KSO4-1 3.32E-09 m2/s
 NaHCO3-1 3.45E-09 m2/s
 NaCO3-1 3.08E-09 m2/s
 NaF - Aq 3.88E-09 m2/s
 Na+1 4.48E-09 m2/s
 NaNO3 - A 4.07E-09 m2/s
 NaSO4-1 3.12E-09 m2/s
 SO4-2 3.58E-09 m2/s
 SO3 - Aq 3.98E-10 m2/s
 H2SO4 - A 3.98E-09 m2/s

KHSO4 - A 51.933 mol%
 KH2SO4 3.3683 mol%
 K2CO3 54473 mol%
 K2CO3 1 E 859.86 mol%
 KCl - Aq 120.41 mol%
 KCl 7.5444 mol%
 KF 7251.8 mol%
 KF 2H2O 365.98 mol%
 KF 4H2O 63.719 mol%
 KOH 4.83E+10 mol%
 KOH 2H2C 1572.6 mol%
 KOH 1H2C 17545 mol%
 KNO3 0.78972 mol%
 K2SO4 0.011589 mol%
 KSO4-1 0.13228 mol%
 K2SO4 1H 0.010377 mol%
 NaHCO3-1 2.359 mol%
 NaHCO3 0.40374 mol%
 NaHSO4 356.65 mol%
 Na2CO3 7.3005 mol%
 Na2CO3 1 0.11036 mol%
 Na2CO3 7 0.22354 mol%
 NaCO3-1 0.28348 mol%
 NaCO3 1 1.0111 mol%
 NaCl 38.197 mol%
 NaF - Aq 1.2908 mol%
 NaF 0.2625 mol%
 NaF Na2S 3.18E-03 mol%
 NaHF2 7.28E-03 mol%
 NaOH 8.05E+06 mol%
 NaOH 1H2 84322 mol%
 NaNO3 - A 17.412 mol%
 NaNO3 28.746 mol%
 Na2SO4 0.44942 mol%
 Na2SO4 1.90E-07 mol%
 Na2SO4 N 162.44 mol%
 Na2SO4 II 0.046844 mol%
 NaSO4-1 0.12009 mol%
 SO3 - Aq 7726 mol%
 SO3 - Vap 1.84E+09 mol%
 H2SO4 - A 2.57E+10 mol%
 H2SO4 - V 3.86E+05 mol%

Species Mobilities

HSO4-1 3.52E-04 m2/ohm-mol
 Cl-1 5.25E-04 m2/ohm-mol
 Na2F+1 4.12E-04 m2/ohm-mol
 F-1 3.78E-04 m2/ohm-mol
 HF2-1 5.15E-04 m2/ohm-mol
 H+1 2.40E-03 m2/ohm-mol
 OH-1 1.35E-03 m2/ohm-mol
 NOS-1 4.90E-04 m2/ohm-mol
 K+1 5.03E-04 m2/ohm-mol
 KSO4-1 2.62E-04 m2/ohm-mol
 Na+1 3.44E-04 m2/ohm-mol
 NaSO4-1 2.40E-04 m2/ohm-mol
 SO4-2 5.52E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.18E-09 m2/s
 HCO3-1 1.08E-09 m2/s
 HSO4-1 1.23E-09 m2/s
 CO2 - Aq 1.82E-09 m2/s
 CO3-2 9.00E-10 m2/s
 Cl-1 1.83E-09 m2/s
 Na2F+1 1.44E-09 m2/s
 F-1 1.33E-09 m2/s
 (HF)2 - Aq 1.10E-09 m2/s
 HCl - Aq 1.91E-09 m2/s
 HF2-1 1.79E-09 m2/s
 HF - Aq 1.20E-10 m2/s
 H+1 8.03E-09 m2/s
 OH-1 4.59E-09 m2/s
 NOS-1 1.71E-09 m2/s
 HNO3 - Aq 1.79E-09 m2/s
 KH2SO4 - A 1.41E-09 m2/s
 KCl - Aq 1.52E-09 m2/s
 K+1 1.78E-09 m2/s
 KSO4-1 9.25E-10 m2/s
 NaHCO3-1 9.39E-10 m2/s
 NaCO3-1 8.05E-10 m2/s
 NaF - Aq 1.04E-09 m2/s
 Na+1 1.22E-09 m2/s
 NaNO3 - A 1.14E-09 m2/s
 NaSO4-1 8.50E-10 m2/s
 SO4-2 9.73E-10 m2/s
 SO3 - Aq 1.02E-10 m2/s
 H2SO4 - A 1.01E-09 m2/s

KHSO4 - A 51.933 mol%
 KH2SO4 3.3683 mol%
 K2CO3 54473 mol%
 K2CO3 1 E 859.86 mol%
 KCl - Aq 120.41 mol%
 KCl 7.5444 mol%
 KF 7251.8 mol%
 KF 2H2O 365.98 mol%
 KF 4H2O 63.719 mol%
 KOH 4.83E+10 mol%
 KOH 2H2C 1572.6 mol%
 KOH 1H2C 17545 mol%
 KNO3 0.78972 mol%
 K2SO4 0.011589 mol%
 KSO4-1 0.13228 mol%
 K2SO4 1H 0.010377 mol%
 NaHCO3-1 2.359 mol%
 NaHCO3 0.40374 mol%
 NaHSO4 356.65 mol%
 Na2CO3 7.3005 mol%
 Na2CO3 1 0.11036 mol%
 Na2CO3 7 0.22354 mol%
 NaCO3-1 0.28348 mol%
 NaCO3 1 1.0111 mol%
 NaCl 38.197 mol%
 NaF - Aq 1.2908 mol%
 NaF 0.2625 mol%
 NaF Na2S 3.18E-03 mol%
 NaHF2 7.28E-03 mol%
 NaOH 8.05E+06 mol%
 NaOH 1H2 84322 mol%
 NaNO3 - A 17.412 mol%
 NaNO3 28.746 mol%
 Na2SO4 0.44942 mol%
 Na2SO4 1.90E-07 mol%
 Na2SO4 N 162.44 mol%
 Na2SO4 II 0.046844 mol%
 NaSO4-1 0.12009 mol%
 SO3 - Aq 7726 mol%
 SO3 - Vap 1.84E+09 mol%
 H2SO4 - A 2.57E+10 mol%
 H2SO4 - V 3.86E+05 mol%

Species Mobilities

HSO4-1 3.52E-04 m2/ohm-mol
 Cl-1 5.25E-04 m2/ohm-mol
 Na2F+1 4.12E-04 m2/ohm-mol
 F-1 3.78E-04 m2/ohm-mol
 HF2-1 5.15E-04 m2/ohm-mol
 H+1 2.40E-03 m2/ohm-mol
 OH-1 1.35E-03 m2/ohm-mol
 NOS-1 4.90E-04 m2/ohm-mol
 K+1 5.03E-04 m2/ohm-mol
 KSO4-1 2.62E-04 m2/ohm-mol
 Na+1 3.44E-04 m2/ohm-mol
 NaSO4-1 2.40E-04 m2/ohm-mol
 SO4-2 5.52E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.18E-09 m2/s
 HCO3-1 1.08E-09 m2/s
 HSO4-1 1.23E-09 m2/s
 CO2 - Aq 1.82E-09 m2/s
 CO3-2 9.00E-10 m2/s
 Cl-1 1.83E-09 m2/s
 Na2F+1 1.44E-09 m2/s
 F-1 1.33E-09 m2/s
 (HF)2 - Aq 1.10E-09 m2/s
 HCl - Aq 1.91E-09 m2/s
 HF2-1 1.79E-09 m2/s
 HF - Aq 1.20E-10 m2/s
 H+1 8.03E-09 m2/s
 OH-1 4.59E-09 m2/s
 NOS-1 1.71E-09 m2/s
 HNO3 - Aq 1.79E-09 m2/s
 KH2SO4 - A 1.41E-09 m2/s
 KCl - Aq 1.52E-09 m2/s
 K+1 1.78E-09 m2/s
 KSO4-1 9.25E-10 m2/s
 NaHCO3-1 9.39E-10 m2/s
 NaCO3-1 8.05E-10 m2/s
 NaF - Aq 1.04E-09 m2/s
 Na+1 1.22E-09 m2/s
 NaNO3 - A 1.14E-09 m2/s
 NaSO4-1 8.50E-10 m2/s
 SO4-2 9.73E-10 m2/s
 SO3 - Aq 1.02E-10 m2/s
 H2SO4 - A 1.01E-09 m2/s

KCl - Aq 26.986 mol%
 KCl 16.596 mol%
 KF 54867 mol%
 KF 2H2O 1752.1 mol%
 KF 4H2O 6.69E+06 mol%
 KOH 3.86E+08 mol%
 KOH 2H2C 7614.8 mol%
 KOH 1H2C 23065 mol%
 KNO3 3.2519 mol%
 K2SO4 0.02886 mol%
 KSO4-1 0.087738 mol%
 K2SO4 1H 0.067899 mol%
 NaHCO3-1 0.52509 mol%
 NaHCO3 0.65701 mol%
 NaHSO4 376.35 mol%
 Na2CO3 1.316 mol%
 Na2CO3 II 0.23977 mol%
 Na2CO3 7 1.3745 mol%
 NaCO3-1 0.7632 mol%
 NaCO3 1 0.91296 mol%
 NaCl 35.516 mol%
 NaF - Aq 1.197 mol%
 NaF 0.31449 mol%
 NaF Na2S 3.16E-03 mol%
 NaHF2 0.051038 mol%
 NaOH 1.94E+06 mol%
 NaOH 1H2 16297 mol%
 NaOH 1H2C 4.5857 mol%
 NaNO3 58.88 mol%
 Na2SO4 0.14445 mol%
 Na2SO4 2.13E-03 mol%
 Na2SO4 N 155.99 mol%
 Na2SO4 II 26.308 mol%
 NaSO4-1 0.060006 mol%
 SO3 - Aq 1922.2 mol%
 SO3 - Vap 7879.0 mol%
 H2SO4 - A 3.19E+10 mol%
 H2SO4 - V 248.07 mol%

Species Mobilities

HSO4-1 8.60E-04 m2/ohm-mol
 Cl-1 1.28E-03 m2/ohm-mol
 Na2F+1 1.18E-03 m2/ohm-mol
 F-1 1.01E-03 m2/ohm-mol
 HF2-1 1.44E-03 m2/ohm-mol
 H+1 3.98E-03 m2/ohm-mol
 OH-1 2.77E-03 m2/ohm-mol
 NOS-1 1.15E-03 m2/ohm-mol
 K+1 1.20E-03 m2/ohm-mol
 KSO4-1 6.81E-04 m2/ohm-mol
 Na+1 9.19E-04 m2/ohm-mol
 NaSO4-1 6.37E-04 m2/ohm-mol
 SO4-2 1.46E-03 m2/ohm-mol

Species Self Diffusivities

H2O 7.43E-09 m2/s
 HCO3-1 3.76E-09 m2/s
 HSO4-1 3.99E-09 m2/s
 CO2 - Aq 6.47E-09 m2/s
 CO3-2 3.33E-09 m2/s
 Cl-1 5.90E-09 m2/s
 Na2F+1 5.32E-09 m2/s
 F-1 4.70E-09 m2/s
 (HF)2 - Aq 3.89E-09 m2/s
 HCl - Aq 6.20E-09 m2/s
 HF2-1 6.61E-09 m2/s
 HF - Aq 4.48E-10 m2/s
 H+1 1.77E-09 m2/s
 OH-1 1.24E-09 m2/s
 NOS-1 5.31E-09 m2/s
 HNO3 - Aq 5.57E-09 m2/s
 KH2SO4 - A 5.27E-09 m2/s
 KCl - Aq 4.88E-09 m2/s
 K+1 5.53E-09 m2/s
 KSO4-1 3.18E-09 m2/s
 NaHCO3-1 3.30E-09 m2/s
 NaCO3-1 2.94E-09 m2/s
 NaF - Aq 3.71E-09 m2/s
 Na+1 4.29E-09 m2/s
 NaNO3 - A 3.90E-09 m2/s
 NaSO4-1 2.98E-09 m2/s
 SO4-2 3.40E-09 m2/s
 SO3 - Aq 3.81E-10 m2/s
 H2SO4 - A 3.53E-09 m2/s

K. T. Chiang
 12/19/04

Na2SO4 1.90E-07 mol/P
 Na2SO4.N 1.62E-04 mol/P
 Na2SO4.H 0.046544 mol/P
 NaSO4-I 0.12009 mol/P
 SO3 - Aq 7726 mol/P
 SO3 - Vap 1.64E+09 mol/P
 H2SO4 - A 2.57E+10 mol/P
 H2SO4 - V 3.86E+06 mol/P

Species Mobilities

HCO3-1 2.64E-04 m2/ohm-mol
 CO3-2 4.40E-04 m2/ohm-mol
 H+1 2.08E-03 m2/ohm-mol
 OH-1 1.17E-03 m2/ohm-mol
 NaCO3-1 1.96E-04 m2/ohm-mol
 Na+1 2.98E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.39E-09 m2/s
 HCO3-1 1.15E-09 m2/s
 HSO4-1 1.31E-09 m2/s
 CO2 - Aq 1.98E-09 m2/s
 CO3-2 9.63E-10 m2/s
 Cl-1 1.89E-09 m2/s
 Na2F+1 1.53E-09 m2/s
 F-1 1.41E-09 m2/s
 (HF)2 - Aq 1.20E-09 m2/s
 HCl - Aq 2.10E-09 m2/s
 HF2-1 1.89E-09 m2/s
 HF - Aq 1.32E-10 m2/s
 H+1 7.97E-09 m2/s
 OH-1 4.68E-09 m2/s
 NO3-1 1.80E-09 m2/s
 HNO3 - Aq 1.95E-09 m2/s
 KHSO4 - A 1.55E-09 m2/s
 KCl - Aq 1.80E-09 m2/s
 K+1 1.84E-09 m2/s
 KSO4-1 9.91E-10 m2/s
 NaHCO3 1.02E-09 m2/s
 NaCO3-1 8.63E-10 m2/s
 NaF - Aq 1.15E-09 m2/s
 Na+1 1.28E-09 m2/s
 NaNO3 - A 1.25E-09 m2/s
 NaSO4-1 9.11E-10 m2/s
 SO4-2 1.04E-09 m2/s
 SO3 - Aq 1.12E-10 m2/s
 H2SO4 - A 1.10E-09 m2/s

Na2SO4 1.90E-07 mol/P
 Na2SO4.N 1.62E-04 mol/P
 Na2SO4.H 0.046544 mol/P
 NaSO4-I 0.12009 mol/P
 SO3 - Aq 7726 mol/P
 SO3 - Vap 1.64E+09 mol/P
 H2SO4 - A 2.57E+10 mol/P
 H2SO4 - V 3.86E+06 mol/P

Species Mobilities

HCO3-1 2.52E-04 m2/ohm-mol
 CO3-2 4.19E-04 m2/ohm-mol
 H+1 1.99E-03 m2/ohm-mol
 OH-1 1.12E-03 m2/ohm-mol
 NaCO3-1 1.87E-04 m2/ohm-mol
 Na+1 2.84E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.34E-09 m2/s
 HCO3-1 1.13E-09 m2/s
 HSO4-1 1.29E-09 m2/s
 CO2 - Aq 1.84E-09 m2/s
 CO3-2 9.44E-10 m2/s
 Cl-1 1.89E-09 m2/s
 Na2F+1 1.50E-09 m2/s
 F-1 1.38E-09 m2/s
 (HF)2 - Aq 1.18E-09 m2/s
 HCl - Aq 2.08E-09 m2/s
 HF2-1 1.85E-09 m2/s
 HF - Aq 1.29E-10 m2/s
 H+1 7.92E-09 m2/s
 OH-1 4.60E-09 m2/s
 NO3-1 1.77E-09 m2/s
 HNO3 - Aq 1.92E-09 m2/s
 KHSO4 - A 1.51E-09 m2/s
 KCl - Aq 1.58E-09 m2/s
 K+1 1.81E-09 m2/s
 KSO4-1 9.71E-10 m2/s
 NaHCO3 1.02E-09 m2/s
 NaCO3-1 8.46E-10 m2/s
 NaF - Aq 1.12E-09 m2/s
 Na+1 1.28E-09 m2/s
 NaNO3 - A 1.22E-09 m2/s
 NaSO4-1 8.93E-10 m2/s
 SO4-2 1.02E-09 m2/s
 SO3 - Aq 1.09E-10 m2/s
 H2SO4 - A 1.09E-09 m2/s

Na2SO4.N 155.99 mol/P
 Na2SO4.H 26.306 mol/P
 NaSO4-I 0.080006 mol/P
 SO3 - Aq 1922.2 mol/P
 SO3 - Vap 70679 mol/P
 H2SO4 - A 3.19E+10 mol/P
 H2SO4 - V 248.07 mol/P

Species Mobilities

HCO3-1 6.77E-04 m2/ohm-mol
 CO3-2 1.20E-03 m2/ohm-mol
 H+1 3.35E-03 m2/ohm-mol
 OH-1 2.33E-03 m2/ohm-mol
 NaCO3-1 5.28E-04 m2/ohm-mol
 Na+1 7.73E-04 m2/ohm-mol

Species Self Diffusivities

H2O 7.80E-09 m2/s
 HCO3-1 3.92E-09 m2/s
 HSO4-1 4.17E-09 m2/s
 CO2 - Aq 3.84E-09 m2/s
 CO3-2 3.40E-09 m2/s
 Cl-1 6.15E-09 m2/s
 Na2F+1 5.54E-09 m2/s
 F-1 4.91E-09 m2/s
 (HF)2 - Aq 4.14E-09 m2/s
 HCl - Aq 6.58E-09 m2/s
 HF2-1 6.87E-09 m2/s
 HF - Aq 4.77E-10 m2/s
 H+1 1.90E-08 m2/s
 OH-1 1.28E-08 m2/s
 NO3-1 5.53E-09 m2/s
 HNO3 - Aq 5.91E-09 m2/s
 KHSO4 - A 5.59E-09 m2/s
 KCl - Aq 5.04E-09 m2/s
 K+1 5.75E-09 m2/s
 KSO4-1 3.33E-09 m2/s
 NaHCO3 3.50E-09 m2/s
 NaCO3-1 3.09E-09 m2/s
 NaF - Aq 3.95E-09 m2/s
 Na+1 4.47E-09 m2/s
 NaNO3 - A 4.14E-09 m2/s
 NaSO4-1 3.13E-09 m2/s
 SO4-2 3.95E-09 m2/s
 SO3 - Aq 4.04E-10 m2/s
 H2SO4 - A 3.74E-09 m2/s

(HF)2 - Aq 51387 mol/P
 (HF)2 - Va 19124 mol/P
 HCl - Aq 1.09E+06 mol/P
 HCl - Vap 9.92796 mol/P
 HF2-1 4.1247 mol/P
 HF - Aq 6.17E-04 mol/P
 HF - Vap 1.9019 mol/P
 HNO3 - Aq 20.085 mol/P
 HNO3 - Va 1.40E+05 mol/P
 KHCO3 1.8452 mol/P
 KHSO4 - A 51.903 mol/P
 KHSO4 3.3653 mol/P
 K2CO3 54473 mol/P
 K2CO3.1E 859.86 mol/P
 KCl - Aq 120.41 mol/P
 KCl 7.5444 mol/P
 KF 7251.8 mol/P
 KF.2H2O 395.95 mol/P
 KF.4H2O 83.719 mol/P
 KOH 4.83E+10 mol/P
 KOH.2H2O 1572.5 mol/P
 KOH.1H2O 17545 mol/P
 KNO3 0.78972 mol/P
 K2SO4 0.011589 mol/P
 KSO4-1 0.13226 mol/P
 K2SO4.1H 0.010577 mol/P
 NaHCO3 - 2.399 mol/P
 NaHCO3 0.40374 mol/P
 NaHSO4 356.65 mol/P
 Na2CO3 7.3005 mol/P
 Na2CO3.H 0.11036 mol/P
 Na2CO3.7 0.22354 mol/P
 NaCO3-1 0.26346 mol/P
 Na2CO3.1 1.0111 mol/P
 NaCl 38.197 mol/P
 NaF - Aq 1.2905 mol/P
 NaF 0.2625 mol/P
 NaF.Na2SO4 3.18E-03 mol/P
 NaHF2 7.29E-03 mol/P
 NaOH 8.06E+06 mol/P
 NaOH.1H2O 64322 mol/P
 NaOH3 - A 17.412 mol/P
 NaNO3 26.745 mol/P
 Na2SO4 0.44842 mol/P
 Na2SO4 1.90E-07 mol/P
 Na2SO4.N 162.44 mol/P
 Na2SO4.H 0.046544 mol/P
 NaSO4-1 0.12009 mol/P
 SO3 - Aq 7726 mol/P
 SO3 - Vap 1.64E+09 mol/P
 H2SO4 - A 2.57E+10 mol/P
 H2SO4 - V 3.86E+06 mol/P

Species Mobilities

HCO3-1 2.48E-04 m2/ohm-mol
 HSO4-1 2.89E-04 m2/ohm-mol
 CO3-2 4.12E-04 m2/ohm-mol
 Cl-1 4.25E-04 m2/ohm-mol
 Na2F+1 3.34E-04 m2/ohm-mol
 F-1 3.08E-04 m2/ohm-mol
 HF2-1 4.17E-04 m2/ohm-mol
 H+1 1.95E-03 m2/ohm-mol
 OH-1 1.10E-03 m2/ohm-mol
 NO3-1 3.97E-04 m2/ohm-mol
 K+1 4.07E-04 m2/ohm-mol
 KSO4-1 2.15E-04 m2/ohm-mol
 NaHCO3 1.84E-04 m2/ohm-mol
 Na+1 2.70E-04 m2/ohm-mol
 NaSO4-1 1.95E-04 m2/ohm-mol
 SO4-2 4.48E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.24E-09 m2/s
 HCO3-1 1.09E-09 m2/s
 HSO4-1 1.24E-09 m2/s
 CO2 - Aq 1.89E-09 m2/s
 CO3-2 9.10E-10 m2/s
 Cl-1 1.83E-09 m2/s
 Na2F+1 1.45E-09 m2/s
 F-1 1.34E-09 m2/s
 (HF)2 - Aq 1.12E-09 m2/s
 HCl - Aq 1.98E-09 m2/s
 HF2-1 1.80E-09 m2/s
 HF - Aq 1.23E-10 m2/s
 H+1 7.98E-09 m2/s
 OH-1 4.54E-09 m2/s
 NO3-1 1.71E-09 m2/s
 HNO3 - Aq 1.83E-09 m2/s
 KHSO4 - A 1.44E-09 m2/s
 KCl - Aq 1.54E-09 m2/s
 K+1 1.78E-09 m2/s
 KSO4-1 9.33E-10 m2/s
 NaHCO3 9.95E-10 m2/s
 NaCO3-1 8.14E-10 m2/s
 NaF - Aq 1.07E-09 m2/s
 Na+1 1.23E-09 m2/s
 NaNO3 - A 1.18E-09 m2/s
 NaSO4-1 8.57E-10 m2/s
 SO4-2 9.82E-10 m2/s
 SO3 - Aq 1.04E-10 m2/s
 H2SO4 - A 1.03E-09 m2/s

K.Y. Ching
 12/19/04

SCW15

Calculation Summary

SinglePoint_NaOH Calculation for Work1

Automatic Chemistry Model Databanks: Public

Unit Set: Default

Isothermal Calculation Temperature: 25 °C Pressure: 1 atm

Stream Inflows table with columns: Water, Sodium chl, Potassium, Sodium nit, Sodium sul, Sodium bic, Sodium flu, Sodium hydr.

Speciation Summary

Total number of species: 113

Species list table with columns: User, Related, Inflows, Species, Aqueous, Vapor, Solid, Second Liquid, Species

Stream Parameters

Stream Am: 573.95 g, Temperature: 25 °C, Pressure: 1 atm, pH: 8.6601, Ionic Stron, Osmotic Pi, WaterActiv, Electrical C, Electrical C, Viscosity, r

Denisty and Enthalpy table with columns: Denisty, Enthalpy

Total and Phase Flows (Amounts)

Total and phase flows table with columns: Mole, Mass, Volume

Scaling Tendencies

Scaling tendencies table with columns: Solids with Temperature Range

SCW16

Calculation Summary

SinglePoint_NaOH_95C Calculation for Work1

Automatic Chemistry Model Databanks: Public

Unit Set: Default

Isothermal Calculation Temperature: 95 °C Pressure: 1 atm

Stream Inflows table with columns: Water, Sodium chl, Potassium, Sodium nit, Sodium sul, Sodium bic, Sodium flu, Sodium hydr.

Speciation Summary

Total number of species: 113

Species list table with columns: User, Related, Inflows, Species, Aqueous, Vapor, Solid, Second Liquid, Species

Stream Parameters

Stream Am: 573.95 g, Temperature: 95 °C, Pressure: 1 atm, pH: 8.5819, Ionic Stron, Osmotic Pi, WaterActiv, Electrical C, Electrical C, Viscosity, r

Denisty and Enthalpy table with columns: Denisty, Enthalpy

Total and Phase Flows (Amounts)

Total and phase flows table with columns: Mole, Mass, Volume

Scaling Tendencies

Scaling tendencies table with columns: Solids with Temperature Range

SCW17

Calculation Summary

SinglePoint1 Calculation for Work1

Automatic Chemistry Model Databanks: Public

Unit Set: Default

Isothermal Calculation Temperature: 25 °C Pressure: 1 atm

Stream Inflows table with columns: Water, Sodium chl, Potassium, Sodium nit, Sodium sul, Sodium bic, Sodium flu, Sodium hydr.

Speciation Summary

Total number of species: 113

Species list table with columns: User, Related, Inflows, Species, Aqueous, Vapor, Solid, Second Liquid, Species

Stream Parameters

Stream Am: 558.53 g, Temperature: 25 °C, Pressure: 1 atm, pH: 7.7294, Ionic Stron, Osmotic Pi, WaterActiv, Electrical C, Electrical C, Viscosity, r

Denisty and Enthalpy table with columns: Denisty, Enthalpy

Total and Phase Flows (Amounts)

Total and phase flows table with columns: Mole, Mass, Volume

Scaling Tendencies

Scaling tendencies table with columns: Solids with Temperature Range

SCW18

Calculation Summary

SinglePoint_NaOH Calculation for Work1

Automatic Chemistry Model Databanks: Public

Unit Set: Default

Isothermal Calculation Temperature: 25 °C Pressure: 1 atm

Stream Inflows table with columns: Water, Sodium chl, Potassium, Sodium nit, Sodium sul, Sodium bic, Sodium flu, Sodium hydr.

Speciation Summary

Total number of species: 113

Species list table with columns: User, Related, Inflows, Species, Aqueous, Vapor, Solid, Second Liquid, Species

Stream Parameters

Stream Am: 561.63 g, Temperature: 25 °C, Pressure: 1 atm, pH: 8.6481, Ionic Stron, Osmotic Pi, WaterActiv, Electrical C, Electrical C, Viscosity, r

Denisty and Enthalpy table with columns: Denisty, Enthalpy

Total and Phase Flows (Amounts)

Total and phase flows table with columns: Mole, Mass, Volume

Scaling Tendencies

Scaling tendencies table with columns: Solids with Temperature Range

Handwritten signature: K.T. Ching 12/19/04

Potassium 3.16E-06 17.7 40.200 °C inside range
Sodium hy 1.14E-07 data valid if inside range
Potassium 1.87E-05 3 80.000 °C inside range
Potassium 1.07E-10 0 33.000 °C inside range
Sodium hy 5.82E-11 12 60.000 °C inside range
Potassium 1.40E-11 data valid if inside range
Sodium bic 2.79E-12 data valid if inside range
Sodium sul 2.63E-14 0 82.500 °C inside range
Potassium 3.96E-18 data valid if inside range

Potassium 2.77E-10 33 143.00 °C inside range
Potassium 3.80E-11 data valid if inside range
Sodium bic 2.67E-11 data valid if inside range
Potassium 1.75E-14 data valid if inside range

Species Output (True Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, KCl, NaNO3, NaHCO3, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, H+2, NaOH, NaF, NaCl, Na2SO4, SO3, Na2S2O8, HF, HSO4-1, HSO4-2, HSO4-3, H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3, H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Output (True Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, KCl, NaNO3, NaHCO3, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, H+2, NaOH, NaF, NaCl, Na2SO4, SO3, Na2S2O8, HF, HSO4-1, HSO4-2, HSO4-3, H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Output (True Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, KCl, NaNO3, NaHCO3, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, H+2, NaOH, NaF, NaCl, Na2SO4, SO3, Na2S2O8, HF, HSO4-1, HSO4-2, HSO4-3, H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Output (True Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, KCl, NaNO3, NaHCO3, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, H+2, NaOH, NaF, NaCl, Na2SO4, SO3, Na2S2O8, HF, HSO4-1, HSO4-2, HSO4-3, H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Molecular Output (Apparent Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Molecular Output (Apparent Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Molecular Output (Apparent Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Molecular Output (Apparent Species)

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Element Balance

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include C(+4), Cl(-1), F(-1), H(+1), N(+5), Na(+1), O(-2), S(+6).

Element Balance

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include C(+4), Cl(-1), F(-1), H(+1), N(+5), Na(+1), O(-2), S(+6).

Element Balance

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include C(+4), Cl(-1), F(-1), H(+1), N(+5), Na(+1), O(-2), S(+6).

Element Balance

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include C(+4), Cl(-1), F(-1), H(+1), N(+5), Na(+1), O(-2), S(+6).

Species Activity Coefficients

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Activity Coefficients

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Activity Coefficients

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species Activity Coefficients

Table with 6 columns: Total, Aqueous, Vapor, Solid, 2nd Liquid, n/a. Rows include H2O, NaOH, CO2, HCl, HNO3, KOH, F-1, HF-1, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species K(eq)-Values

Table with 2 columns: Species, Value. Rows include H2O, HCO3-1, HCO3-2, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species K(eq)-Values

Table with 2 columns: Species, Value. Rows include H2O, HCO3-1, HCO3-2, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species K(eq)-Values

Table with 2 columns: Species, Value. Rows include H2O, HCO3-1, HCO3-2, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Species K(eq)-Values

Table with 2 columns: Species, Value. Rows include H2O, HCO3-1, HCO3-2, CO2, HCl, HNO3, HCO3-1, HCO3-2, OH-1, OH-2, H+1, Cl-1, Na2F+1, NaF, HSO4-1, HSO4-2, HSO4-3, SO3.

Key: Chief
12/19/04

(HF)2 - Aq	51387 mol%O
(HF)2 - Vap	19124 mol%O
HCl - Aq	1.695E-06 mol%O
HCl - Vap	0.52796 mol%O
HF2 - 1	4.127 mol%O
HF - Aq	6.17E-04 mol%O
HF - Vap	19019 mol%O
HNO3 - Aq	20.065 mol%O
HNO3 - Va	1.402E+05 mol%O
KHSO4	1.8452 mol%O
KHSO4 - A	51.933 mol%O
KHSO4	3.9883 mol%O
K2CO3	54473 mol%O
K2CO3.1.E	859.86 mol%O
KCl - Aq	120.41 mol%O
KCl	7.5444 mol%O
KF	7251.8 mol%O
KF.2H2O	395.96 mol%O
KF.4H2O	63.719 mol%O
KOH	4.83E+10 mol%O
KOH.2H2C	1572.6 mol%O
KOH.1H2C	17545 mol%O
KN03	0.78972 mol%O
K2SO4	0.011589 mol%O
KSO4-1	0.13228 mol%O
K2SO4.1H	0.010377 mol%O
NaHCO3	2.389 mol%O
NaHCO3	0.40374 mol%O
NaHSO4	356.65 mol%O
Na2CO3	7.3005 mol%O
Na2CO3.II	1.0111 mol%O
NaCl	38.197 mol%O
NaF - Aq	1.2906 mol%O
NaF	0.2625 mol%O
NaF.Na2Si	3.16E-03 mol%O
NaOH	8.05E+06 mol%O
NaOH.1H2	64322 mol%O
NaNO3 - A	17.412 mol%O
NaNO3	28.746 mol%O
Na2SO4	0.44942 mol%O
Na2SO4.N	162.44 mol%O
Na2SO4.II	0.046844 mol%O
Na2SO4.N	162.44 mol%O
Na2SO4.II	0.046844 mol%O
NaSO4-1	0.12009 mol%O
SO3 - Aq	7726 mol%O
SO3 - Vap	1.64E+09 mol%O
H2SO4 - A	2.57E+10 mol%O
H2SO4 - V	3.86E+05 mol%O

Species Mobilities	
HCO3-1	2.62E-04 m2/ohm-mol
CO3-2	4.35E-04 m2/ohm-mol
Cl-1	4.49E-04 m2/ohm-mol
H+1	2.05E-03 m2/ohm-mol
OH-1	1.16E-03 m2/ohm-mol
NO3-1	4.20E-04 m2/ohm-mol
K+1	4.30E-04 m2/ohm-mol
NaCO3-1	1.94E-04 m2/ohm-mol
Na+1	2.94E-04 m2/ohm-mol

Species Self Diffusivities	
H2O	2.36E-09 m2/s
HCO3-1	1.14E-09 m2/s
HSO4-1	1.31E-09 m2/s
CO2 - Aq	1.96E-09 m2/s
CO3-2	9.59E-10 m2/s
Cl-1	1.92E-09 m2/s
Na2F+1	1.52E-09 m2/s
F-1	1.41E-09 m2/s
(HF)2 - Aq	1.19E-09 m2/s
HCl - Aq	2.08E-09 m2/s
HF2-1	1.89E-09 m2/s
HF - Aq	1.30E-10 m2/s
H+1	8.12E-09 m2/s
OH-1	4.71E-09 m2/s
NO3-1	1.80E-09 m2/s
HNO3 - Aq	1.94E-09 m2/s
KHSO4 - A	1.53E-09 m2/s
KCl - Aq	1.58E-09 m2/s
K+1	1.84E-09 m2/s
KSO4-1	9.88E-10 m2/s
NaHCO3	1.01E-09 m2/s
NaCO3-1	6.89E-10 m2/s
NaF - Aq	1.19E-09 m2/s
Na+1	1.29E-09 m2/s
NaNO3 - A	1.23E-09 m2/s
NaSO4-1	9.05E-10 m2/s
SO4-2	1.04E-09 m2/s
SO3 - Aq	1.10E-10 m2/s
H2SO4 - A	1.09E-09 m2/s

Species Self Diffusivities	
H2O	2.17E-09 m2/s
HCO3-1	1.05E-09 m2/s
HSO4-1	1.20E-09 m2/s
CO2 - Aq	1.81E-09 m2/s
CO3-2	8.81E-10 m2/s
Cl-1	1.78E-09 m2/s
Na2F+1	1.40E-09 m2/s
F-1	1.30E-09 m2/s
(HF)2 - Aq	1.09E-09 m2/s
HCl - Aq	1.90E-09 m2/s
HF2-1	1.74E-09 m2/s
HF - Aq	1.20E-10 m2/s
H+1	7.89E-09 m2/s
OH-1	4.41E-09 m2/s
NO3-1	1.86E-09 m2/s
HNO3 - Aq	1.77E-09 m2/s
KHSO4 - A	1.39E-09 m2/s
KCl - Aq	1.51E-09 m2/s
K+1	1.71E-09 m2/s
KSO4-1	9.02E-10 m2/s
NaHCO3	9.26E-10 m2/s
NaCO3-1	7.96E-10 m2/s
NaF - Aq	1.03E-09 m2/s
Na+1	1.19E-09 m2/s
NaNO3 - A	1.13E-09 m2/s
NaSO4-1	8.25E-10 m2/s
SO4-2	9.49E-10 m2/s
SO3 - Aq	1.01E-10 m2/s
H2SO4 - A	9.94E-10 m2/s

HF - Vap	379.93 mol%O
HNO3 - Aq	4.002 mol%O
HNO3 - Va	929.44 mol%O
H2CO3	7.7419 mol%O
KHSO4 - A	15.597 mol%O
KHSO4	14.067 mol%O
K2CO3	2670 mol%O
K2CO3.1.E	2055.7 mol%O
KCl - Aq	26.988 mol%O
KCl	16.596 mol%O
KF	54667 mol%O
KF.2H2O	1732.1 mol%O
KF.4H2O	6.89E+09 mol%O
KOH	3.89E+08 mol%O
KOH.2H2C	7814.8 mol%O
KOH.1H2C	23065 mol%O
KN03	3.2319 mol%O
K2SO4	0.02685 mol%O
KSO4-1	0.067733 mol%O
K2SO4.1H	0.087969 mol%O
NaHCO3	0.52509 mol%O
NaHCO3	0.65701 mol%O
NaHSO4	376.35 mol%O
Na2CO3	1.316 mol%O
Na2CO3.II	0.23607 mol%O
Na2CO3.7I	1.3745 mol%O
NaCO3-1	0.7832 mol%O
Na2CO3.II	0.91296 mol%O
NaCl	35.518 mol%O
NaF - Aq	1.197 mol%O
NaF	0.31448 mol%O
NaF.Na2Si	3.16E-03 mol%O
NaOH	0.051038 mol%O
NaOH	1.94E+05 mol%O
NaOH.1H2	16297 mol%O
NaNO3 - A	4.5867 mol%O
NaNO3	56.88 mol%O
Na2SO4	0.14484 mol%O
Na2SO4	2.13E-03 mol%O
Na2SO4.N	155.99 mol%O
Na2SO4.II	26.308 mol%O
NaSO4-1	0.090006 mol%O
SO3 - Aq	1922.2 mol%O
SO3 - Vap	75979 mol%O
H2SO4 - A	3.18E+10 mol%O
H2SO4 - V	248.07 mol%O

Species Mobilities	
HCO3-1	6.19E-04 m2/ohm-mol
HSO4-1	6.62E-04 m2/ohm-mol
CO3-2	1.10E-03 m2/ohm-mol
Cl-1	9.88E-04 m2/ohm-mol
Na2F+1	8.89E-04 m2/ohm-mol
F-1	7.79E-04 m2/ohm-mol
HF2-1	1.11E-03 m2/ohm-mol
H+1	3.07E-03 m2/ohm-mol
OH-1	2.13E-03 m2/ohm-mol
NO3-1	8.87E-04 m2/ohm-mol
K+1	9.23E-04 m2/ohm-mol
KSO4-1	5.24E-04 m2/ohm-mol
NaCO3-1	4.83E-04 m2/ohm-mol
Na+1	7.07E-04 m2/ohm-mol
NaSO4-1	4.90E-04 m2/ohm-mol
SO4-2	1.12E-03 m2/ohm-mol

Species Self Diffusivities	
H2O	7.40E-09 m2/s
HCO3-1	3.70E-09 m2/s
HSO4-1	3.92E-09 m2/s
CO2 - Aq	6.42E-09 m2/s
CO3-2	3.29E-09 m2/s
Cl-1	5.62E-09 m2/s
Na2F+1	5.23E-09 m2/s
F-1	4.66E-09 m2/s
(HF)2 - Aq	3.86E-09 m2/s
HCl - Aq	6.15E-09 m2/s
HF2-1	6.51E-09 m2/s
HF - Aq	4.47E-10 m2/s
H+1	1.74E-08 m2/s
OH-1	1.22E-08 m2/s
NO3-1	5.22E-09 m2/s
HNO3 - Aq	5.52E-09 m2/s
KHSO4 - A	5.20E-09 m2/s
KCl - Aq	4.83E-09 m2/s
K+1	5.46E-09 m2/s
KSO4-1	3.12E-09 m2/s
NaHCO3	3.27E-09 m2/s
NaCO3-1	2.90E-09 m2/s
NaF - Aq	3.68E-09 m2/s
Na+1	4.25E-09 m2/s
NaNO3 - A	3.95E-09 m2/s
NaSO4-1	2.93E-09 m2/s
SO4-2	3.34E-09 m2/s
SO3 - Aq	3.76E-10 m2/s
H2SO4 - A	3.48E-09 m2/s

NaHCO3	2.389 mol%O
NaHCO3	0.40374 mol%O
NaHSO4	356.65 mol%O
Na2CO3	7.3005 mol%O
Na2CO3.II	0.11036 mol%O
Na2CO3.7I	0.22354 mol%O
NaCO3-1	0.28348 mol%O
Na2CO3.II	1.0111 mol%O
NaCl	38.197 mol%O
NaF - Aq	1.2906 mol%O
NaF	0.2625 mol%O
NaF.Na2Si	3.16E-03 mol%O
NaOH	7.28E-03 mol%O
NaOH	8.05E+06 mol%O
NaOH.1H2	64322 mol%O
NaNO3 - A	17.412 mol%O
NaNO3	28.746 mol%O
Na2SO4	0.44942 mol%O
Na2SO4	1.90E-07 mol%O
Na2SO4.N	162.44 mol%O
Na2SO4.II	0.046844 mol%O
NaSO4-1	0.12009 mol%O
SO3 - Aq	7726 mol%O
SO3 - Vap	1.64E+09 mol%O
H2SO4 - A	2.57E+10 mol%O
H2SO4 - V	3.86E+05 mol%O

Species Mobilities	
HCO3-1	2.62E-04 m2/ohm-mol
CO3-2	4.35E-04 m2/ohm-mol
Cl-1	4.49E-04 m2/ohm-mol
H+1	2.05E-03 m2/ohm-mol
OH-1	1.16E-03 m2/ohm-mol
NO3-1	4.20E-04 m2/ohm-mol
K+1	4.30E-04 m2/ohm-mol
NaCO3-1	1.94E-04 m2/ohm-mol
Na+1	2.94E-04 m2/ohm-mol

Species Self Diffusivities	
H2O	2.36E-09 m2/s
HCO3-1	1.14E-09 m2/s
HSO4-1	1.31E-09 m2/s
CO2 - Aq	1.96E-09 m2/s
CO3-2	9.59E-10 m2/s
Cl-1	1.92E-09 m2/s
Na2F+1	1.52E-09 m2/s
F-1	1.41E-09 m2/s
(HF)2 - Aq	1.19E-09 m2/s
HCl - Aq	2.08E-09 m2/s
HF2-1	1.89E-09 m2/s
HF - Aq	1.30E-10 m2/s
H+1	8.12E-09 m2/s
OH-1	4.71E-09 m2/s
NO3-1	1.80E-09 m2/s
HNO3 - Aq	1.94E-09 m2/s
KHSO4 - A	1.53E-09 m2/s
KCl - Aq	1.58E-09 m2/s
K+1	1.84E-09 m2/s
KSO4-1	9.88E-10 m2/s
NaHCO3	1.01E-09 m2/s
NaCO3-1	6.89E-10 m2/s
NaF - Aq	1.19E-09 m2/s
Na+1	1.29E-09 m2/s
NaNO3 - A	1.23E-09 m2/s
NaSO4-1	9.05E-10 m2/s
SO4-2	1.04E-09 m2/s
SO3 - Aq	1.10E-10 m2/s
H2SO4 - A	1.09E-09 m2/s

NaHCO3	2.389 mol%O
NaHCO3	0.40374 mol%O
NaHSO4	356.65 mol%O
Na2CO3	7.3005 mol%O
Na2CO3.II	0.11036 mol%O
Na2CO3.7I	0.22354 mol%O
NaCO3-1	0.28348 mol%O
Na2CO3.II	1.0111 mol%O
NaCl	38.197 mol%O
NaF - Aq	1.2906 mol%O
NaF	0.2625 mol%O
NaF.Na2Si	3.16E-03 mol%O
NaOH	7.28E-03 mol%O
NaOH	8.05E+06 mol%O
NaOH.1H2	64322 mol%O
NaNO3 - A	17.412 mol%O
NaNO3	28.746 mol%O
Na2SO4	0.44942 mol%O
Na2SO4	1.90E-07 mol%O
Na2SO4.N	162.44 mol%O
Na2SO4.II	0.046844 mol%O
NaSO4-1	0.12009 mol%O
SO3 - Aq	7726 mol%O
SO3 - Vap	1.64E+09 mol%O
H2SO4 - A	2.57E+10 mol%O
H2SO4 - V	3.86E+05 mol%O

Species Mobilities	
HCO3-1	2.62E-04 m2/ohm-mol
CO3-2	4.35E-04 m2/ohm-mol
Cl-1	4.49E-04 m2/ohm-mol
H+1	2.05E-03 m2/ohm-mol
OH-1	1.16E-03 m2/ohm-mol
NO3-1	4.20E-04 m2/ohm-mol
K+1	4.30E-04 m2/ohm-mol
NaCO3-1	1.94E-04 m2/ohm-mol
Na+1	2.94E-04 m2/ohm-mol

Species Self Diffusivities	
H2O	2.36E-09 m2/s
HCO3-1	1.14E-09 m2/s
HSO4-1	1.31E-09 m2/s
CO2 - Aq	1.96E-09 m2/s
CO3-2	9.59E-10 m2/s
Cl-1	1.92E-09 m2/s
Na2F+1	1.52E-09 m2/s
F-1	1.41E-09 m2/s
(HF)2 - Aq	1.19E-09 m2/s
HCl - Aq	2.08E-09 m2/s
HF2-1	1.89E-09 m2/s
HF - Aq	1.30E-10 m2/s
H+1	8.12E-09 m2/s
OH-1	4.71E-09 m2/s
NO3-1	1.80E-09 m2/s
HNO3 - Aq	1.94E-09 m2/s
KHSO4 - A	1.53E-09 m2/s
KCl - Aq	1.58E-09 m2/s
K+1	1.84E-09 m2/s
KSO4-1	9.88E-10 m2/s
NaHCO3	1.01E-09 m2/s
NaCO3-1	6.89E-10 m2/s
NaF - Aq	1.19E-09 m2/s
Na+1	1.29E-09 m2/s
NaNO3 - A	1.23E-09 m2/s
NaSO4-1	9.05E-10 m2/s
SO4-2	1.04E-09 m2/s
SO3 - Aq	1.10E-10 m2/s
H2SO4 - A	1.09E-09 m2/s

Na2CO3
Na2CO3.II
Na2CO3.7I
NaCO3-1
Na2CO3.II
NaCl
NaF - Aq
NaF
NaF.Na2Si
NaOH
NaOH.1H2
NaNO3 - A
NaNO3
Na2SO4
Na2SO4
Na2SO4.N
Na2SO4.II
NaSO4-1
SO3 - Aq
SO3 - Vap
H2SO4 - A
H2SO4 - V

Species Mc	
HCO3-1
CO3-2
Cl-1
H+1
OH-1
NO3-1
K+1
NaCO3-1
Na+1

Species Sa	
H2O
HCO3-1
HSO4-1
CO2 - Aq
CO3-2
Cl-1
Na2F+1
F-1
(HF)2 - Aq
HCl - Aq
HF2-1
HF - Aq
H+1
OH-1
NO3-1
HNO3 - Aq
KHSO4 - A
KCl - Aq
K+1
KSO4-1
NaHCO3
NaCO3-1
NaF - Aq
Na+1
NaNO3 - A
NaSO4-1
SO4-2
SO3 - Aq
H2SO4 - A

K.T. Chinn
12/19/04

Summary					SCW18														
<p>Summary</p> <p>1_NaOH_950C Calculation for Work1</p> <p>Chemistry Model</p> <p>Databanks: Public</p> <p>Unit Set: Default</p> <p>Calculation</p> <p>Temperature: 95 °C</p> <p>Pressure: 1 atm</p> <p>ows</p> <p>500 g</p> <p>2.726 g</p> <p>3.252 g</p> <p>4.372 g</p> <p>0 g</p> <p>48.16 g</p> <p>0 g</p> <p>3.1 g</p> <p>Summary</p> <p>ser of species: 113</p>					<p>Calculation Summary</p> <p>SinglePoint1 Calculation for Work1</p> <p>Automatic Chemistry Model</p> <p>Databanks: Public</p> <p>Unit Set: Default</p> <p>Isothermal Calculation</p> <p>Temperature: 25 °C</p> <p>Pressure: 1 atm</p> <p>Stream Inflows</p> <p>Water: 1000 g</p> <p>Sodium chl: 5.444 g</p> <p>Potassium: 6.487 g</p> <p>Sodium nit: 0 g</p> <p>Sodium sul: 0 g</p> <p>Sodium bic: 96.38 g</p> <p>Sodium flu: 0 g</p> <p>Sodium hy: 0 g</p> <p>Speciation Summary</p> <p>Total number of species: 113</p>					<p>Calculation Summary</p> <p>SinglePoint_NaOH Calculation for Work1</p> <p>Automatic Chemistry Model</p> <p>Databanks: Public</p> <p>Unit Set: Default</p> <p>Isothermal Calculation</p> <p>Temperature: 25 °C</p> <p>Pressure: 1 atm</p> <p>Stream Inflows</p> <p>Water: 1000 g</p> <p>Sodium chl: 5.444 g</p> <p>Potassium: 6.487 g</p> <p>Sodium nit: 0 g</p> <p>Sodium sul: 0 g</p> <p>Sodium bic: 96.38 g</p> <p>Sodium flu: 0 g</p> <p>Sodium hy: 8 g</p> <p>Speciation Summary</p> <p>Total number of species: 113</p>					<p>Calculation Summary</p> <p>SinglePoint_NaOH_950C Calculation for Work1</p> <p>Automatic Chemistry Model</p> <p>Databanks: Public</p> <p>Unit Set: Default</p> <p>Isothermal Calculation</p> <p>Temperature: 95 °C</p> <p>Pressure: 1 atm</p> <p>Stream Inflows</p> <p>Water: 1000 g</p> <p>Sodium chl: 5.444 g</p> <p>Potassium: 6.487 g</p> <p>Sodium nit: 0 g</p> <p>Sodium sul: 0 g</p> <p>Sodium bic: 96.38 g</p> <p>Sodium flu: 0 g</p> <p>Sodium hy: 8 g</p> <p>Speciation Summary</p> <p>Total number of species: 113</p>				
<p>Related Inflows</p> <p>Aqueous Species</p> <p>Vapor Species</p> <p>Solid Species</p> <p>Second Liquid Species</p> <p>CO2 H2O</p> <p>H2CO3 HCO3-1</p> <p>K2SO4.KNHSO4-1</p> <p>(HF)2 - Vap KHCO3</p> <p>2Na2SO4.CO2-Aq</p> <p>(HF)2 CO3-2</p> <p>HCl Cl-1</p> <p>HF Na2F+1</p> <p>(HF)6 F-1</p> <p>HNO3 (HF)2-Aq</p> <p>KHCO3 HCl-Aq</p> <p>KHSO4 HF-1</p> <p>K2CO3 HF-Aq</p> <p>K2CO3.1.5H2O</p> <p>KF OH-1</p> <p>KF.2H2O NO3-1</p> <p>KF.4H2O HNO3-Aq</p> <p>KOH KHCO3-Aq</p> <p>KOH.2H2O KOH-Aq</p> <p>KOH.1H2C.K+1</p> <p>KNOS KSO4-1</p> <p>K2SO4 NaHCO3-Aq</p> <p>NaHSO4 NaF-Aq</p> <p>Na2CO3 Na+1</p> <p>Na2CO3.1.NaNO3-Aq</p> <p>Na2CO3.7.NaSO4-1</p> <p>Na2CO3.1.SO4-2</p> <p>NaF.Na2S.SO3-Aq</p> <p>NaHF2 H2SO4-Aq</p> <p>NaOH H2O</p> <p>Na2SO4</p> <p>Na2SO4.NaHSO4</p> <p>Na2SO4.10H2O</p> <p>SO3</p> <p>H2SO4</p>					<p>User Related Inflows</p> <p>Aqueous Species</p> <p>Vapor Species</p> <p>Solid Species</p> <p>Second Liquid Species</p> <p>H2O CO2</p> <p>H2CO3 HCO3-1</p> <p>K2SO4.KNHSO4-1</p> <p>(HF)2 - Vap KHCO3</p> <p>2Na2SO4.CO2-Aq</p> <p>(HF)2 CO3-2</p> <p>HCl Cl-1</p> <p>HF Na2F+1</p> <p>(HF)6 F-1</p> <p>HNO3 (HF)2-Aq</p> <p>KHCO3 HCl-Aq</p> <p>KHSO4 HF-1</p> <p>K2CO3 HF-Aq</p> <p>K2CO3.1.5H2O</p> <p>KF OH-1</p> <p>KF.2H2O NO3-1</p> <p>KF.4H2O HNO3-Aq</p> <p>KOH KHCO3-Aq</p> <p>KOH.2H2O KOH-Aq</p> <p>KOH.1H2C.K+1</p> <p>KNOS KSO4-1</p> <p>K2SO4 NaHCO3-Aq</p> <p>NaHSO4 NaF-Aq</p> <p>Na2CO3 Na+1</p> <p>Na2CO3.1.NaNO3-Aq</p> <p>Na2CO3.7.NaSO4-1</p> <p>Na2CO3.1.SO4-2</p> <p>NaF.Na2S.SO3-Aq</p> <p>NaHF2 H2SO4-Aq</p> <p>NaOH H2O</p> <p>Na2SO4</p> <p>Na2SO4.NaHSO4</p> <p>Na2SO4.10H2O</p> <p>SO3</p> <p>H2SO4</p>					<p>User Related Inflows</p> <p>Aqueous Species</p> <p>Vapor Species</p> <p>Solid Species</p> <p>Second Liquid Species</p> <p>H2O CO2</p> <p>H2CO3 HCO3-1</p> <p>K2SO4.KNHSO4-1</p> <p>(HF)2 - Vap KHCO3</p> <p>2Na2SO4.CO2-Aq</p> <p>(HF)2 CO3-2</p> <p>HCl Cl-1</p> <p>HF Na2F+1</p> <p>(HF)6 F-1</p> <p>HNO3 (HF)2-Aq</p> <p>KHCO3 HCl-Aq</p> <p>KHSO4 HF-1</p> <p>K2CO3 HF-Aq</p> <p>K2CO3.1.5H2O</p> <p>KF OH-1</p> <p>KF.2H2O NO3-1</p> <p>KF.4H2O HNO3-Aq</p> <p>KOH KHCO3-Aq</p> <p>KOH.2H2O KOH-Aq</p> <p>KOH.1H2C.K+1</p> <p>KNOS KSO4-1</p> <p>K2SO4 NaHCO3-Aq</p> <p>NaHSO4 NaF-Aq</p> <p>Na2CO3 Na+1</p> <p>Na2CO3.1.NaNO3-Aq</p> <p>Na2CO3.7.NaSO4-1</p> <p>Na2CO3.1.SO4-2</p> <p>NaF.Na2S.SO3-Aq</p> <p>NaHF2 H2SO4-Aq</p> <p>NaOH H2O</p> <p>Na2SO4</p> <p>Na2SO4.NaHSO4</p> <p>Na2SO4.10H2O</p> <p>SO3</p> <p>H2SO4</p>					<p>User Related Inflows</p> <p>Aqueous Species</p> <p>Vapor Species</p> <p>Solid Species</p> <p>Second Liquid Species</p> <p>H2O CO2</p> <p>H2CO3 HCO3-1</p> <p>K2SO4.KNHSO4-1</p> <p>(HF)2 - Vap KHCO3</p> <p>2Na2SO4.CO2-Aq</p> <p>(HF)2 CO3-2</p> <p>HCl Cl-1</p> <p>HF Na2F+1</p> <p>(HF)6 F-1</p> <p>HNO3 (HF)2-Aq</p> <p>KHCO3 HCl-Aq</p> <p>KHSO4 HF-1</p> <p>K2CO3 HF-Aq</p> <p>K2CO3.1.5H2O</p> <p>KF OH-1</p> <p>KF.2H2O NO3-1</p> <p>KF.4H2O HNO3-Aq</p> <p>KOH KHCO3-Aq</p> <p>KOH.2H2O KOH-Aq</p> <p>KOH.1H2C.K+1</p> <p>KNOS KSO4-1</p> <p>K2SO4 NaHCO3-Aq</p> <p>NaHSO4 NaF-Aq</p> <p>Na2CO3 Na+1</p> <p>Na2CO3.1.NaNO3-Aq</p> <p>Na2CO3.7.NaSO4-1</p> <p>Na2CO3.1.SO4-2</p> <p>NaF.Na2S.SO3-Aq</p> <p>NaHF2 H2SO4-Aq</p> <p>NaOH H2O</p> <p>Na2SO4</p> <p>Na2SO4.NaHSO4</p> <p>Na2SO4.10H2O</p> <p>SO3</p> <p>H2SO4</p>				
<p>rameters</p> <p>561.63 g</p> <p>95 °C</p> <p>1 atm</p> <p>8.5751 pH</p> <p>1.3618 mol/kg H2K SCW16, 95C</p> <p>70.042 atm</p> <p>0.95778 Activity</p> <p>0.18117 1/ohm</p> <p>123.97 cm2/ohm-mol</p> <p>0.47262 cP</p> <p>1.5908 cP/cP H2O</p>					<p>Stream Parameters</p> <p>Stream Am: 1108.3 g</p> <p>Temperature: 25 °C</p> <p>Pressure: 1 atm</p> <p>pH: 7.7419</p> <p>Ionic Stre: 1.2091 mol/kg H2O</p> <p>Osmotic Pi: 61.057 atm</p> <p>WaterActiv: 0.95616 Activity</p> <p>Electrical C: 0.06618 1/ohm</p> <p>Electrical C: 52.46 cm2/ohm-mol</p> <p>Viscosity, ε: 1.3419 cP</p> <p>Viscosity, r: 1.5055 cP/cP H2O</p>					<p>Stream Parameters</p> <p>Stream Am: 1116.3 g</p> <p>Temperature: 25 °C</p> <p>Pressure: 1 atm</p> <p>pH: 8.7798</p> <p>Ionic Stre: 1.5081 mol/kg H2O</p> <p>Osmotic Pi: 59.515 atm</p> <p>WaterActiv: 0.95754 Activity</p> <p>Electrical C: 0.074587 1/ohm</p> <p>Electrical C: 50.84 cm2/ohm-mol</p> <p>Viscosity, ε: 1.3649 cP</p> <p>Viscosity, r: 1.5323 cP/cP H2O</p>					<p>Stream Parameters</p> <p>Stream Am: 1116.3 g</p> <p>Temperature: 95 °C</p> <p>Pressure: 1 atm</p> <p>pH: 8.8695</p> <p>Ionic Stre: 1.3716 mol/kg H2K SCW1795C</p> <p>Osmotic Pi: 65.891 atm</p> <p>WaterActiv: 0.96045 Activity</p> <p>Electrical C: 0.1762 1/ohm</p> <p>Electrical C: 124.66 cm2/ohm-mol</p> <p>Viscosity, ε: 0.47109 cP</p> <p>Viscosity, r: 1.5856 cP/cP H2O</p>				
<p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>g/mol g/mol g/mol g/mol g/mol</p> <p>1.0352 7.75E-04</p> <p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>cal cal cal cal cal</p> <p>-2.01E+06 -2.01E+06 -197.09 0 0</p>					<p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>g/mol g/mol g/mol g/mol g/mol</p> <p>1.0627 0 0 0 0</p> <p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>cal cal cal cal cal</p> <p>-4.06E+06 -4.06E+06 0 0 0</p>					<p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>g/mol g/mol g/mol g/mol g/mol</p> <p>1.0718 0 0 0 0</p> <p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>cal cal cal cal cal</p> <p>-4.09E+06 -4.09E+06 0 0 0</p>					<p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>g/mol g/mol g/mol g/mol g/mol</p> <p>1.033 0 0 0 0</p> <p>Total Aqueous Vapor Solid 2nd Liquid</p> <p>cal cal cal cal cal</p> <p>-4.02E+06 -4.02E+06 0 0 0</p>				
<p>Phase Flows (Amounts)</p> <p>Total mol Aqueous Vapor Solid 2nd Liquid</p> <p>mol mol mol mol mol</p> <p>29.16 29.157 3.05E-03 0 0</p> <p>Total g Aqueous Vapor Solid 2nd Liquid</p> <p>g g g g g</p> <p>561.63 561.56 0.071011 0 0</p> <p>Total L Aqueous Vapor Solid 2nd Liquid</p> <p>L L L L L</p> <p>0.63407 0.54247 0.091803 0 0</p>					<p>Total and Phase Flows (Amounts)</p> <p>Total mol Aqueous Vapor Solid 2nd Liquid</p> <p>mol mol mol mol mol</p> <p>58.05 58.05 0 0 0</p> <p>Total g Aqueous Vapor Solid 2nd Liquid</p> <p>g g g g g</p> <p>1108.3 1108.3 0 0 0</p> <p>Total L Aqueous Vapor Solid 2nd Liquid</p> <p>L L L L L</p> <p>1.0429 1.0429 0 0 0</p>					<p>Total and Phase Flows (Amounts)</p> <p>Total mol Aqueous Vapor Solid 2nd Liquid</p> <p>mol mol mol mol mol</p> <p>58.401 58.401 0 0 0</p> <p>Total g Aqueous Vapor Solid 2nd Liquid</p> <p>g g g g g</p> <p>1116.3 1116.3 0 0 0</p> <p>Total L Aqueous Vapor Solid 2nd Liquid</p> <p>L L L L L</p> <p>1.0416 1.0416 0 0 0</p>					<p>Total and Phase Flows (Amounts)</p> <p>Total mol Aqueous Vapor Solid 2nd Liquid</p> <p>mol mol mol mol mol</p> <p>58.231 58.231 0 0 0</p> <p>Total g Aqueous Vapor Solid 2nd Liquid</p> <p>g g g g g</p> <p>1116.3 1116.3 0 0 0</p> <p>Total L Aqueous Vapor Solid 2nd Liquid</p> <p>L L L L L</p> <p>1.0806 1.0806 0 0 0</p>				
<p>ndencies</p> <p>Temperature Range</p> <p>0.32756 0 200.00 °C inside range</p> <p>4.32E-03 35.37 109.00 °C inside range</p> <p>2.06E-03 0 350.00 °C inside range</p> <p>7.50E-04 0 110.00 °C inside range</p> <p>5.85E-04 0 300.00 °C inside range</p> <p>3.24E-04 0 200.00 °C inside range</p> <p>3.13E-10 33 143.00 °C inside range</p> <p>1.96E-14 data valid if inside range</p> <p>1.3618 mol/kg H2K SCW16, 95C</p> <p>input (True Species)</p>					<p>Scaling Tendencies</p> <p>acids with Temperature Range</p> <p>Sodium bic: 0.90033 0 200.00 °C inside range</p> <p>Potassium: 0.013745 0 70.000 °C inside range</p> <p>Sodium cai: 3.58E-03 0 32.000 °C inside range</p> <p>Sodium chl: 1.34E-03 0 350.00 °C inside range</p> <p>Potassium: 6.84E-04 0 200.00 °C inside range</p> <p>Potassium: 3.26E-09 0 80.000 °C inside range</p> <p>Potassium: 1.42E-11 0 33.000 °C inside range</p> <p>Sodium hy: 5.20E-12 12 60.000 °C inside range</p> <p>Potassium: 5.06E-19 data valid if inside range</p> <p>Species Output (True Species)</p>					<p>Scaling Tendencies</p> <p>acids with Temperature Range</p> <p>Sodium bic: 0.82424 0 200.00 °C inside range</p> <p>Sodium cai: 0.04034 0 32.000 °C inside range</p> <p>Potassium: 0.011592 0 70.000 °C inside range</p> <p>Sodium chl: 2.13E-03 0 350.00 °C inside range</p> <p>Potassium: 6.94E-04 0 200.00 °C inside range</p> <p>Potassium: 3.09E-08 0 80.000 °C inside range</p> <p>Potassium: 1.60E-10 0 33.000 °C inside range</p> <p>Sodium hy: 6.95E-11 12 60.000 °C inside range</p> <p>Potassium: 5.89E-18 data valid if inside range</p> <p>Species Output (True Species)</p>					<p>Scaling Tendencies</p> <p>acids with Temperature Range</p> <p>Sodium bic: 0.30742 0 200.00 °C inside range</p> <p>Sodium cai: 5.14E-03 35.37 109.00 °C inside range</p> <p>Sodium chl: 1.87E-03 0 350.00 °C inside range</p> <p>Potassium: 3.30E-04 0 200.00 °C inside range</p> <p>Potassium: 4.22E-10 33 143.00 °C inside range</p> <p>Potassium: 2.63E-14 data valid if inside range</p> <p>Ionic Stre: 1.3716 mol/kg H2K SCW1795C</p> <p>Species Output (True Species)</p> <p>Total g Aqueous Vapor Solid 2nd Liquid</p> <p>mol mol mol mol mol</p>				

K.T. Chua
12/19/04

1.316 mol%
 0.23607 mol%
 1.3745 mol%
 0.7632 mol%
 0.91296 mol%
 35.518 mol%
 1.197 mol%
 0.31449 mol%
 3.18E-03 mol%
 0.251036 mol%
 1.94E+05 mol%
 16297 mol%
 4.5667 mol%
 56.88 mol%
 0.14446 mol%
 2.13E-03 mol%
 155.99 mol%
 26.308 mol%
 0.090006 mol%
 79679 mol%
 1922.2 mol%
 3.19E+10 mol%
 248.07 mol%

obilities
 6.78E-04 m2/ohm-mol
 1.20E-03 m2/ohm-mol
 1.08E-03 m2/ohm-mol
 3.35E-03 m2/ohm-mol
 2.33E-03 m2/ohm-mol
 9.68E-04 m2/ohm-mol
 1.01E-03 m2/ohm-mol
 5.27E-04 m2/ohm-mol
 7.72E-04 m2/ohm-mol

Diffusivities
 7.82E-09 m2/s
 3.91E-09 m2/s
 4.18E-09 m2/s
 6.78E-09 m2/s
 3.48E-09 m2/s
 6.15E-09 m2/s
 5.54E-09 m2/s
 4.91E-09 m2/s
 4.10E-09 m2/s
 6.52E-09 m2/s
 6.88E-09 m2/s
 4.73E-10 m2/s
 1.82E-08 m2/s
 1.29E-08 m2/s
 5.53E-09 m2/s
 8.85E-09 m2/s
 5.53E-09 m2/s
 5.01E-09 m2/s
 5.76E-09 m2/s
 3.32E-09 m2/s
 3.46E-09 m2/s
 3.07E-09 m2/s
 3.91E-09 m2/s
 4.47E-09 m2/s
 4.10E-09 m2/s
 3.11E-09 m2/s
 3.55E-09 m2/s
 4.00E-10 m2/s
 3.71E-09 m2/s

NaCO3-1 0.28348 mol%
 Na2CO3.11 1.0111 mol%
 NaCl 38.197 mol%
 NaF - Aq 1.2906 mol%
 NaF 0.2625 mol%
 NaF.Na2Si 3.16E-03 mol%
 NaHF2 7.28E-03 mol%
 NaOH 8.05E+06 mol%
 NaOH.1H2 64322 mol%
 NaNO3 - A 17.412 mol%
 NaNO3 28.746 mol%
 Na2SO4 0.44942 mol%
 Na2SO4 1.90E-07 mol%
 Na2SO4.N 162.44 mol%
 Na2SO4.11 0.046844 mol%
 NaSO4-1 0.12009 mol%
 SO3 - Aq 7726 mol%
 SO3 - Vap 1.64E+09 mol%
 H2SO4 - A 2.57E+10 mol%
 H2SO4 - V 3.86E+05 mol%

Species Mobilities

HCO3-1 2.64E-04 m2/ohm-mol
 CO3-2 4.39E-04 m2/ohm-mol
 Cl-1 4.54E-04 m2/ohm-mol
 H+1 2.08E-03 m2/ohm-mol
 OH-1 1.17E-03 m2/ohm-mol
 K+1 4.35E-04 m2/ohm-mol
 NaCO3-1 1.96E-04 m2/ohm-mol
 Na+1 2.98E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.37E-09 m2/s
 HCO3-1 1.15E-09 m2/s
 HSO4-1 1.31E-09 m2/s
 CO2 - Aq 1.97E-09 m2/s
 CO3-2 9.60E-10 m2/s
 Cl-1 1.92E-09 m2/s
 Na2F+1 1.52E-09 m2/s
 F-1 1.41E-09 m2/s
 (HF)2 - Aq 1.19E-09 m2/s
 HCl - Aq 2.08E-09 m2/s
 HF2-1 1.89E-09 m2/s
 HF - Aq 1.31E-10 m2/s
 H+1 6.08E-09 m2/s
 OH-1 4.70E-09 m2/s
 NO3-1 1.80E-09 m2/s
 HNO3 - Aq 1.94E-09 m2/s
 KHSO4 - A 1.53E-09 m2/s
 KCl - Aq 1.60E-09 m2/s
 K+1 1.84E-09 m2/s
 KSO4-1 9.88E-10 m2/s
 NaHCO3-1 1.02E-09 m2/s
 NaCO3-1 8.80E-10 m2/s
 NaF - Aq 1.14E-09 m2/s
 Na+1 1.29E-09 m2/s
 NaNO3 - A 1.24E-09 m2/s
 NaSO4-1 9.08E-10 m2/s
 SO4-2 1.04E-09 m2/s
 SO3 - Aq 1.11E-10 m2/s
 H2SO4 - A 1.09E-09 m2/s

NaCO3-1 0.28348 mol%
 Na2CO3.11 1.0111 mol%
 NaCl 38.197 mol%
 NaF - Aq 1.2906 mol%
 NaF 0.2625 mol%
 NaF.Na2Si 3.16E-03 mol%
 NaHF2 7.28E-03 mol%
 NaOH 8.05E+06 mol%
 NaOH.1H2 64322 mol%
 NaNO3 - A 17.412 mol%
 NaNO3 28.746 mol%
 Na2SO4 0.44942 mol%
 Na2SO4 1.90E-07 mol%
 Na2SO4.N 162.44 mol%
 Na2SO4.11 0.046844 mol%
 NaSO4-1 0.12009 mol%
 SO3 - Aq 7726 mol%
 SO3 - Vap 1.64E+09 mol%
 H2SO4 - A 2.57E+10 mol%
 H2SO4 - V 3.86E+05 mol%

Species Mobilities

HCO3-1 2.47E-04 m2/ohm-mol
 CO3-2 4.11E-04 m2/ohm-mol
 Cl-1 4.24E-04 m2/ohm-mol
 H+1 1.94E-03 m2/ohm-mol
 OH-1 1.09E-03 m2/ohm-mol
 K+1 4.06E-04 m2/ohm-mol
 NaCO3-1 1.53E-04 m2/ohm-mol
 Na+1 2.78E-04 m2/ohm-mol

Species Self Diffusivities

H2O 2.29E-09 m2/s
 HCO3-1 1.11E-09 m2/s
 HSO4-1 1.26E-09 m2/s
 CO2 - Aq 1.90E-09 m2/s
 CO3-2 9.27E-10 m2/s
 Cl-1 1.89E-09 m2/s
 Na2F+1 1.47E-09 m2/s
 F-1 1.36E-09 m2/s
 (HF)2 - Aq 1.15E-09 m2/s
 HCl - Aq 2.01E-09 m2/s
 HF2-1 1.83E-09 m2/s
 HF - Aq 1.25E-10 m2/s
 H+1 7.91E-09 m2/s
 OH-1 4.58E-09 m2/s
 NO3-1 1.74E-09 m2/s
 HNO3 - Aq 1.87E-09 m2/s
 KHSO4 - A 1.48E-09 m2/s
 KCl - Aq 1.56E-09 m2/s
 K+1 1.79E-09 m2/s
 NaHCO3-1 9.80E-10 m2/s
 NaCO3-1 8.30E-10 m2/s
 NaF - Aq 1.10E-09 m2/s
 Na+1 1.25E-09 m2/s
 NaNO3 - A 1.19E-09 m2/s
 NaSO4-1 8.78E-10 m2/s
 SO4-2 1.00E-09 m2/s
 SO3 - Aq 1.07E-10 m2/s
 H2SO4 - A 1.06E-09 m2/s

NaF - Aq 1.197 mol%
 NaF 0.31449 mol%
 NaF.Na2Si 3.16E-03 mol%
 NaHF2 0.051038 mol%
 NaOH 1.94E+05 mol%
 NaOH.1H2 16297 mol%
 NaNO3 - A 4.5667 mol%
 NaNO3 56.88 mol%
 Na2SO4 0.14445 mol%
 Na2SO4 2.13E-03 mol%
 Na2SO4.N 155.99 mol%
 Na2SO4.11 26.308 mol%
 NaSO4-1 0.090006 mol%
 SO3 - Aq 1922.2 mol%
 SO3 - Vap 75679 mol%
 H2SO4 - A 3.19E+10 mol%
 H2SO4 - V 248.07 mol%

Species Mobilities

HCO3-1 6.84E-04 m2/ohm-mol
 CO3-2 1.18E-03 m2/ohm-mol
 Cl-1 1.06E-03 m2/ohm-mol
 H+1 3.29E-03 m2/ohm-mol
 OH-1 2.29E-03 m2/ohm-mol
 K+1 9.90E-04 m2/ohm-mol
 NaCO3-1 5.18E-04 m2/ohm-mol
 Na+1 7.58E-04 m2/ohm-mol

Species Self Diffusivities

H2O 7.77E-09 m2/s
 HCO3-1 3.98E-09 m2/s
 HSO4-1 4.12E-09 m2/s
 CO2 - Aq 6.74E-09 m2/s
 CO3-2 3.45E-09 m2/s
 Cl-1 6.08E-09 m2/s
 Na2F+1 5.48E-09 m2/s
 F-1 4.86E-09 m2/s
 (HF)2 - Aq 4.07E-09 m2/s
 HCl - Aq 6.48E-09 m2/s
 HF2-1 6.80E-09 m2/s
 HF - Aq 4.70E-10 m2/s
 H+1 1.80E-08 m2/s
 OH-1 1.27E-08 m2/s
 NO3-1 5.47E-09 m2/s
 HNO3 - Aq 5.81E-09 m2/s
 KHSO4 - A 5.49E-09 m2/s
 KCl - Aq 4.99E-09 m2/s
 K+1 5.69E-09 m2/s
 KSO4-1 3.29E-09 m2/s
 NaHCO3-1 3.44E-09 m2/s
 NaCO3-1 3.04E-09 m2/s
 NaF - Aq 3.83E-09 m2/s
 Na+1 4.42E-09 m2/s
 NaNO3 - A 4.07E-09 m2/s
 NaSO4-1 3.08E-09 m2/s
 SO4-2 3.52E-09 m2/s
 SO3 - Aq 3.97E-10 m2/s
 H2SO4 - A 3.68E-09 m2/s

K. J. Ching
 12/19/04

Slow Strain Rate Testing of MA Alloy 22													
Test ID	Solution Composition	pH Initial	pH Adj'd	pH Final	Eapp(mV)	Max Stress	TTF (hrs)	RA (%)	% Elong	Adj TTF (h)	Tf/Tair	SCC	
air	ambient air	NA	NA	NA	NA	729	70	77.7	81	70.21	70.73		
air2	ambient air	NA	NA	NA	NA	725	72	76.9	77.8	71.25			
air3	ambient air	NA	NA	NA	NA								
SSRMA22_SCW01	SCW	7.76	8.9	9.7	400	572	56	49.9	65.8	54.79	0.77	Y	
SSRMA22_SCW02	SCW	7.76	8.9	9.7	400	560	47	38.5	52	46.67	0.66	Y	
SSRMA22_SCW03	SCW minus NO3	7.34	8.73	9.02	400	618	64		70.2	54.58	0.77	Y	
SSRMA22_SCW04	SCW minus SO4	7.45	8.73	9.99	400	572	60	52.2	68.8	59.58	0.84	Y	
SSRMA22_SCW05	SCW minus F	7.37	8.72	10	400	573	58	50.4	62	54.58	0.77	Y	
SSRMA22_SCW06	SCW minus NaCl	7.38	8.81	9.98	400	559	52	44.6	58.8	51.54	0.73	Y	
SSRMA22_SCW07	3.8M NaCl + 0.38M NaNO3	4.1	9.8	10	400	663	70	76.2	79.8	69.92	0.99	N	
SSRMA22_SCW08	3.8M NaCl + 0.38M NaNO3	4.1	9.8	10.09	200	645	72	68.6	78.8	71.25	1.01	N	
SSRMA22_SCW09	3.8M NaCl + 0.38M NaNO3	8.18	not adj		400	657	74	69.5	79	71.67	1.01	N	
SSRMA22_SCW10	3.8M NaCl + 0.38M NaNO3	8.18	not adj		200	663	69	73.7	78	68.86	0.97	N	
SSRMA22_SCW11	7.6M NaCl + 0.38M NaNO3	6.98	not adj		400	669	72	73	80.2	71.67	1.01	N	
SSRMA22_SCW12	NaCl+NaNO3 armts as in SCW	5.44	9.43	7.38	400	663	70	71.2	79.2	69.46	0.98	N	
SSRMA22_SCW13	SCW minus NaHCO3	6.87	9.79	7.53	400	668	70	69.5	77	68.13	0.96	N	
SSRMA22_SCW14	NaHCO3 as in SCW	6.5	8.58	10.38	400	663	69	61.6	77	68.89	0.97	N	
SSRMA22_SCW15	SCW	7.71	8.66	10.18	400	595	56	49.3	59.8	54.59	0.77	Y	
SSRMA22_SCW16	SCW minus F and SO4	7.76	8.68	9.78	400	578	47	52.7	53.6	47.79	0.68	Y	
SSRMA22_SCW17	SCW minus F and SO4	8.45	8.7		400	590	49	42.2	56.8	48.52	0.69	Y	
SSRMA22_SCW18	Cl and HCO3 as in SCW	7.806	8.726		400	601	54	45.8	67.8	53.89	0.76	Y	
SSRMA22_SCW19	F and HCO3 as in SCW				400	601	56			54.98	0.78	Y	

Slow Strain Rate Testing of MA Alloy 22													
Test ID	Solution Composition	pH Adj'd	mol/kg H2O				SO4-2	HCO3-	F-	I (Ionic Strength)			
air	ambient air	NA											
air2	ambient air	NA											
air3	ambient air	NA											
SSRMA22_SCW01	SCW	8.9	0.083148	1.7168	0.17878	0.097914	0.065232	0.71667	0.055389	2.0957			
SSRMA22_SCW02	SCW	8.9	0.083148	1.7168	0.17878	0.097914	0.065232	0.71667	0.055389	2.0957			
SSRMA22_SCW03	SCW minus NO3	8.73	0.083016	1.5528	0.17919		0.065678	0.79793	0.056213	1.8686			
SSRMA22_SCW04	SCW minus SO4	8.73	0.086535	1.4386	0.17907	0.098792		0.79948	0.05685	1.6915			
SSRMA22_SCW05	SCW minus F	8.72	0.081648	1.6888	0.17727	0.098118		0.096518	0.79169	2.0332			
SSRMA22_SCW06	SCW minus NaCl	8.81	0.082948	1.5564	0.086285	0.098616	0.064409	0.79674	0.056304	1.8711			
SSRMA22_SCW07	3.8M NaCl + 0.38M NaNO3	9.8		4.1096	3.8032	0.30638				4.1096			
SSRMA22_SCW08	3.8M NaCl + 0.38M NaNO3	9.8											
SSRMA22_SCW09	3.8M NaCl + 0.38M NaNO3	not adj											
SSRMA22_SCW10	3.8M NaCl + 0.38M NaNO3	not adj											
SSRMA22_SCW11	7.6M NaCl + 0.38M NaNO3	not adj		75.397	6.0912	0.46797				6.5592			
SSRMA22_SCW12	NaCl+NaNO3 armts as in SCW	9.43		0.19466	0.093083	0.10154				0.19466			
SSRMA22_SCW13	SCW minus NaHCO3	9.79	0.079153	0.49637	0.17963	0.10178	0.083606			0.65919			
SSRMA22_SCW14	NaHCO3 as in SCW	8.58		1.1202				0.89747		1.2156			
SSRMA22_SCW15	SCW	8.66	0.08331	1.6021	0.17943	0.098393	0.06438	0.083654	0.05624	1.8796			
SSRMA22_SCW16	SCW minus F and SO4	8.68	0.086963	1.3272	0.17999	0.099122		0.86313		1.5296			
SSRMA22_SCW17	SCW minus F and SO4	8.7											
SSRMA22_SCW18	Cl and HCO3 as in SCW	8.726	0.086665	1.2722	0.17948			0.82948		1.5081			
SSRMA22_SCW19													

Slow Strain Rate Testing of MA Alloy 22													
Test ID	Solution Composition	pH Final	mol/kg H2O at 95C				SO4-2	HCO3-	F-	I (Ionic Strength)			
air	ambient air	NA											
air2	ambient air	NA											
air3	ambient air	NA											
SSRMA22_SCW01	SCW	9.7	0.079147	1.6068	0.17866	0.088945	0.10743	0.51845	0.056501	2.0753			
SSRMA22_SCW02	SCW	9.7	0.079147	1.6068	0.17866	0.088945	0.10743	0.51845	0.056501	2.0753			
SSRMA22_SCW03	SCW minus NO3	9.02	0.07868	1.4249	0.17906		0.1077	0.58635	0.057501	1.8158			
SSRMA22_SCW04	SCW minus SO4	9.99	0.086412	1.2638	0.17895	0.091904		0.59322	0.058436	1.5541			
SSRMA22_SCW05	SCW minus F	10	0.076119	1.5676	0.17713	0.089361	0.1598	0.57367		2.0068			
SSRMA22_SCW06	SCW minus NaCl	9.98	0.079732	1.4074	0.086208	0.090939	0.089888	0.589	0.057705	1.7814			
SSRMA22_SCW07	3.8M NaCl + 0.38M NaNO3	10		4.1096	3.8032	0.30638				4.1096			
SSRMA22_SCW08	3.8M NaCl + 0.38M NaNO3	10.09											
SSRMA22_SCW09	3.8M NaCl + 0.38M NaNO3												
SSRMA22_SCW10	3.8M NaCl + 0.38M NaNO3												
SSRMA22_SCW11	7.6M NaCl + 0.38M NaNO3			75.397	6.0912	0.46797				6.5592			
SSRMA22_SCW12	NaCl+NaNO3 armts as in SCW	7.38		0.19311	0.093083	0.099994				0.19311			
SSRMA22_SCW13	SCW minus NaHCO3	7.53	0.073321	0.52866	0.17949	0.098534	0.11346			0.71561			
SSRMA22_SCW14	NaHCO3 as in SCW	10.38		0.95501				0.65907		1.0961			
SSRMA22_SCW15	SCW	10.18	0.078945	1.446	0.17943	0.090391	0.10715	0.6114	0.057717	1.8796			
SSRMA22_SCW16	SCW minus F and SO4	9.78	0.086839	1.1325	0.17987	0.092976		0.64634		1.3618			
SSRMA22_SCW17	SCW minus F and SO4												
SSRMA22_SCW18	Cl and HCO3 as in SCW		0.086535	1.1031	0.17935			0.6277		1.3716			
SSRMA22_SCW19													

K. T. Chuang
1/3/05

SLOW STRAIN RATE TEST

Objective: see page #5

Notebook #695

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

* Solution: α $\frac{1}{2}$ L
 Weight # ~~2277-8-3266~~
 2277-3-3266 ^{B10} 6/27/06

14.928 g NaCl #041475

264.58 g KCl #006242

Reagents measured with

Model: OHAUS

SN: 2883

Cal: ~~15 JUL 04~~

15 JAN 05

Due: ~~15 JAN 05~~14 JUL 05 ^{B10} 6/27/06

Counter Electrode: Pt flag

Reference Electrode: As/AsCl w/3MKCl

in house

Gas: N_2 (99.999)

Ecorr: -263 mV

* Applied: +415 mV

Potentiostat: E5E440-2 SN: 9209138

Specimen Visual:

metallic silver-gray
 ductile fracture

$$E^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* above saturation

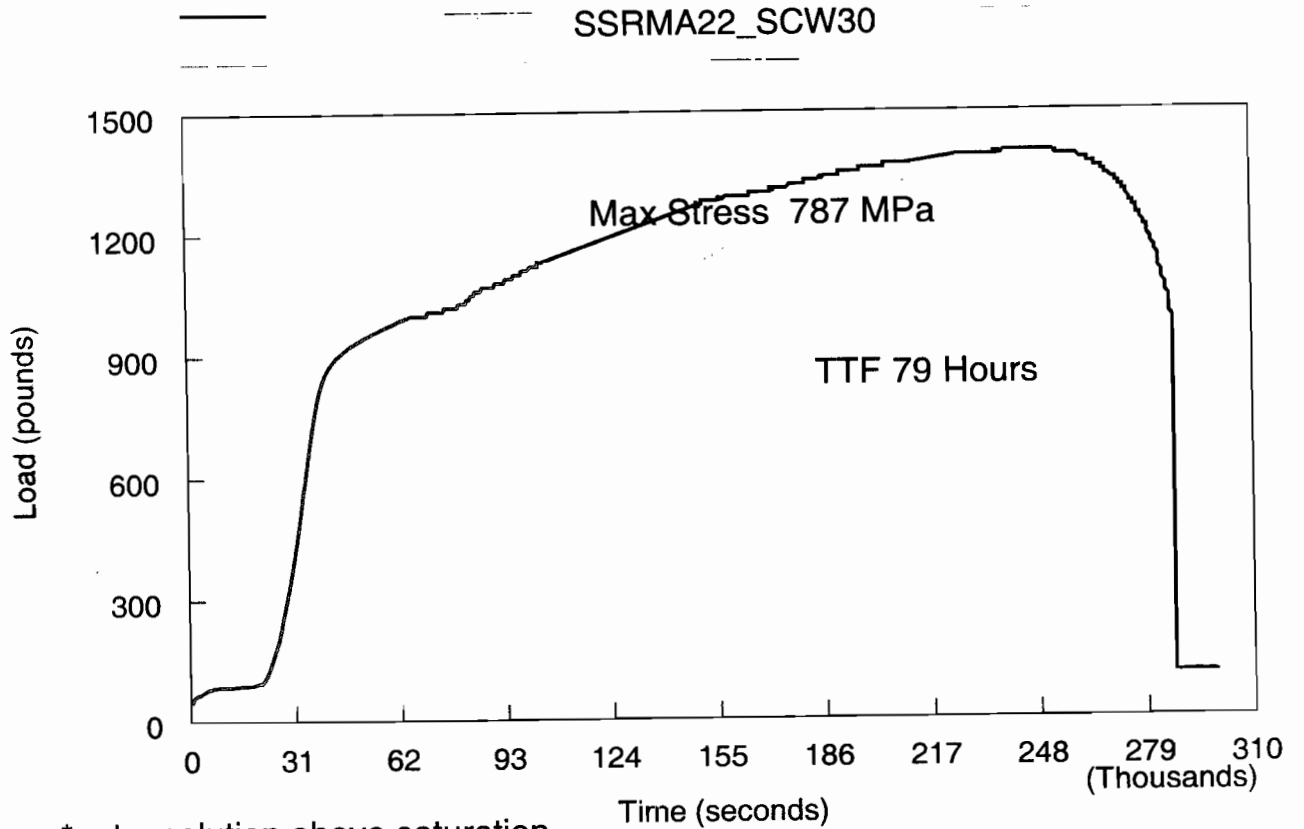
* problem w/ probes, E_{app} not accurate (see data)

data file: SSRMA22 - SCW30

Walter J. Macdonald
 2/1/2005

Slow Strain Rate Test

0.5 M NaCl; 7.1* M KCl



*note: solution above saturation

Walter J. Mackowski
2/1/2005

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C-22 Cylinder Heat # 2277-3-3266 polishes
To A 600 Grit FinishInitial Weight: 12.70363g Model: Sartorius Genius SN: 12809099
Final Weight: 12.69287g Cal 11/10/04 Due: 5/10/05SOLUTION: 0.5 m NaCl + 7.1 m KCl
58.45g NaCl Lot # 041475
1058.79g KCl Lot # 035662
+ DI To 2000 mls
* Not SolubleReagents measured with Model: OHAUS SN: 2883
Cal: 7/15/04 Due: 1/15/05Initial pH: 7.013 Model: orion SN: 2330
Final pH: 7.642 CAL: 7/21/04 DUE: 7/21/05
pH Probe: #13-620-296 SN: 4079126TEST TEMPERATURE: 95°C Measured with Hg Thermometer SN: 188304
Cal: 1/6/05 Due: 7/6/05

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE 13-620-51 SN: 9214081

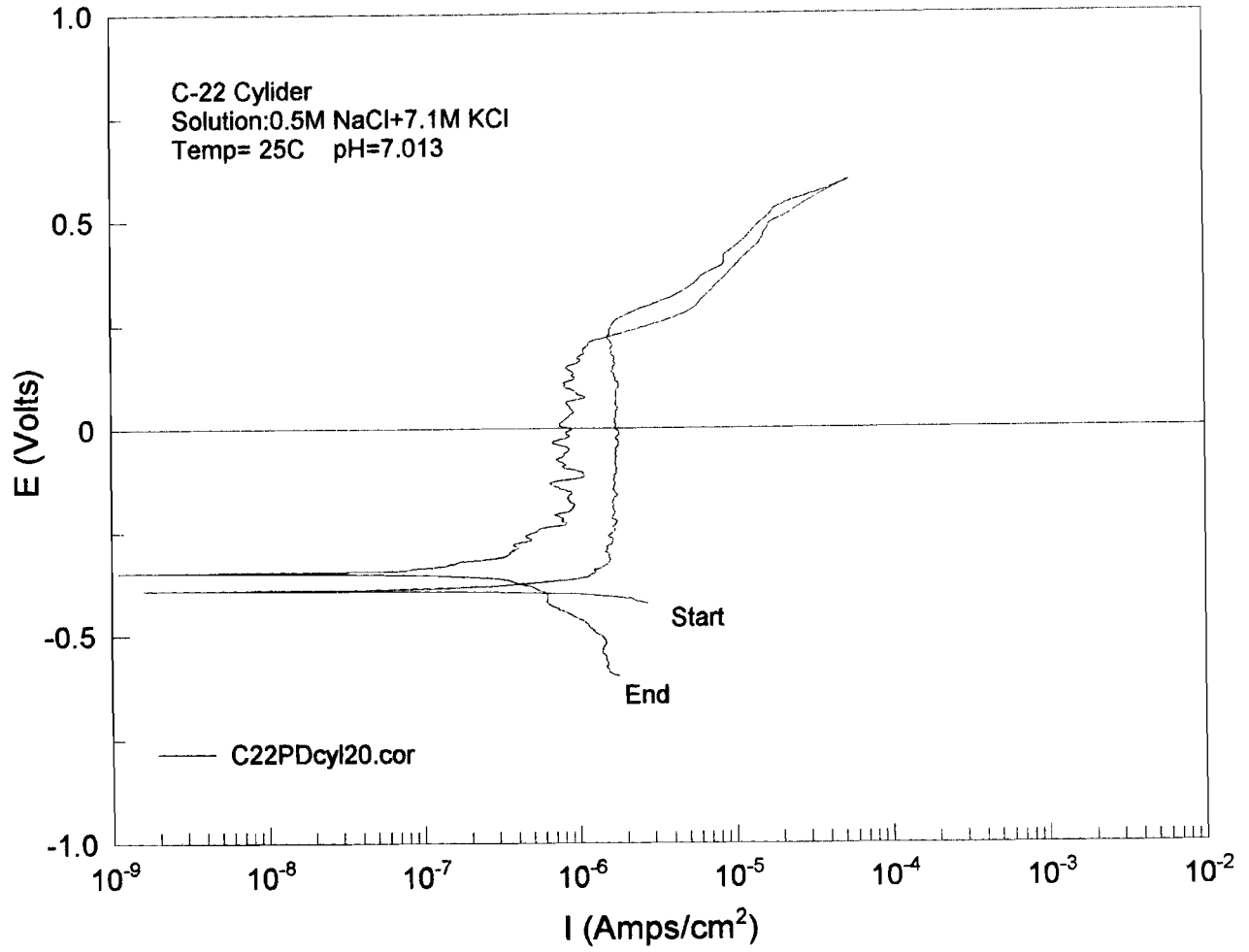
Gas: 99.999% Nitrogen

Ecorr: -322 mV Model: Keithley 614 SN: 0704926
Ept: -56 mV Cal: 6/7/04 Due: 6/7/05

Potentiostat: Solution 1287. SN# 00148500

Specimen Examination: No Visual Sign of Corrosion or Pitting
Very Mild Surface Staining mainly At
Solution Vapor Interface* Solution Above Saturation - Solids In Bottom of Test
Cell Data C22 PD cyl 20

S. P. J. 2/3/05



B. F. J. 2/3/05

POTENTIODYNAMIC TEST

Objective: see page #5

Specimen: C-22 Cylinder Heat # 2277-3-3266 polished To A
600 Grit Finish.Initial Weight: 12.57937g Model: Sartorius Genius SN: 12809099
Final Weight: 12.58224g Cal 11/10/04 Due: 5/10/05

SOLUTION:

0.5 M NaCl + 7.1 M KCl + 1.147M NaHCO₃
58.46g NaCl lot # 041475
1058.82g KCl lot # 035662
192.82g NaHCO₃ lot # 028924
+ DI To 2000 ml * Not Soluble

Reagents measured with

Model: OHAUS SN: 2883
Cal: 7/15/04 Due: 1/15/05

Initial pH: 7.962

Model: orion SN: 2330

Final pH: 8.907

CAL: 7/21/04 DUE: 7/21/05

pH Probe: #13-620-296 SN: 4079126

TEST TEMPERATURE: 95°C

Measured with Hg Thermometer SN: 188304
Cal: 1/6/05 Due: 7/6/05

Counter Electrode: Platinum Flag

Reference Electrode: Fisher SCE

13-620-51

SN: 9214081

Gas: 99.999% Nitrogen

Ecorr: -655 mV

Model: Keithley 614

SN: 6704936

Ept: -5 mV

Cal: 6/7/04

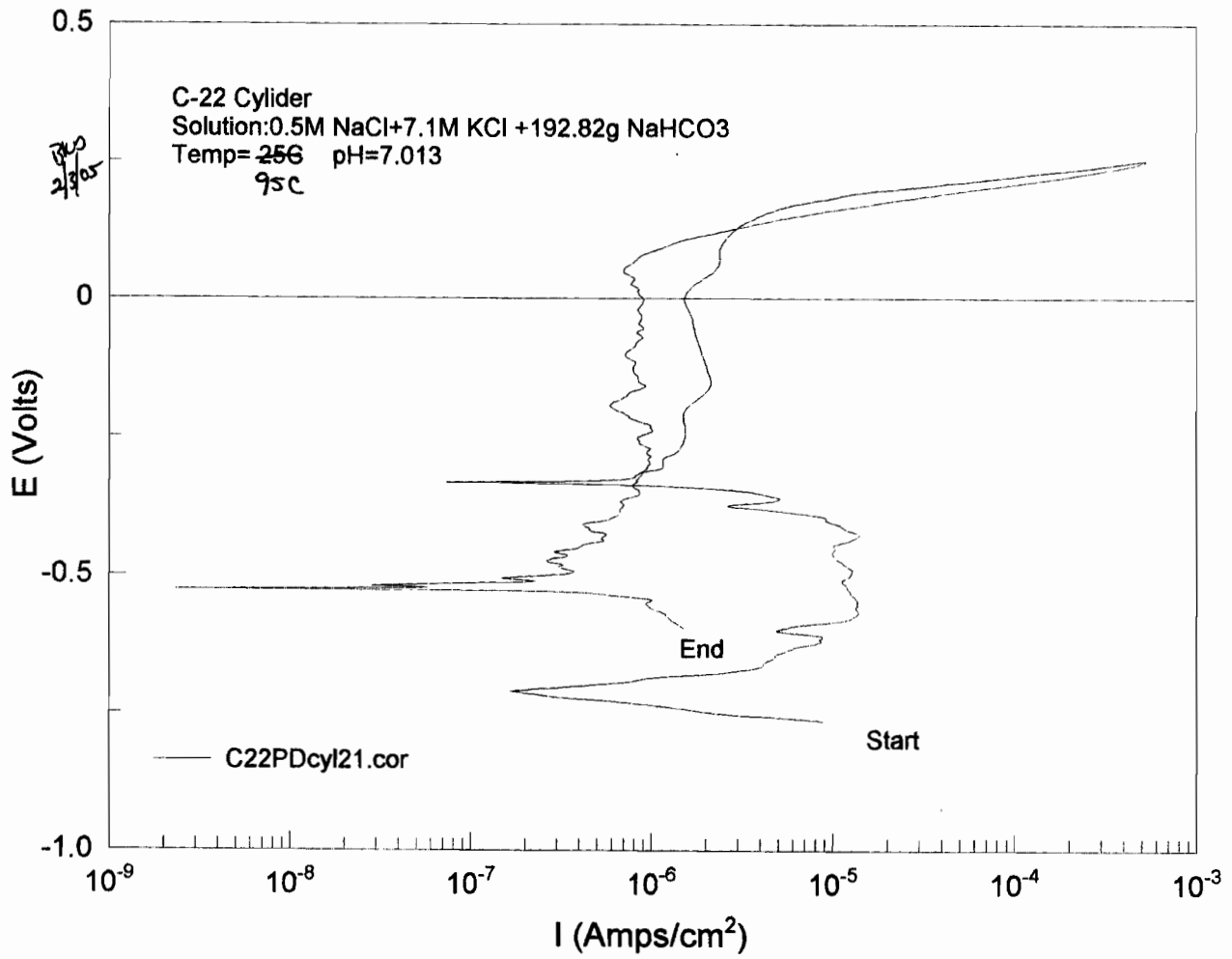
Due: 6/7/05

Potentiostat: Solatron 1287

SN# 60148500

Specimen Examination: No Visual Sign of Corrosion / Pitting
Very mild staining on Specimen At Solution Vapor
Interface* Solution Above Saturation - Solis In Bottom of
Test CellData ~~C22 PD 01/21~~ C22 PD 01/21
2/3/05

B. P. J. 2/3/05



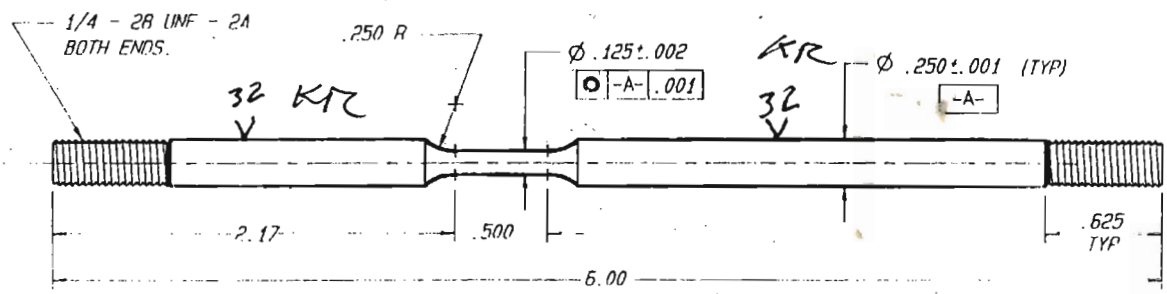
B. R. J. 2/3/05

Ken Chiang
 SwRI-CNWRA
 Phone: (210) 522-2308
 Fax: (210) 522-5184
 e-mail: Kchiang@swri.org

ALLOY 22
 NICKEL
 CHROMIUM
 MOLY
 IRON

176

SwRI DRAWING # 20-03704-042-001



NOTE: 1. DO NOT UNDERCUT RADII
 2. USE LOW STRESS MACHINING PROCEDURE

PERPENDICULAR
 TO
 ROLLING

Procedure: **IOS-WI-821**
 Project # _____
 TOTAL PCS. INSPECTED _____
 TOTAL PCS. ACCEPTED _____
 TOTAL PCS. REJECTED _____
 "NR #": IF REJECTS _____

INSPECTOR



J.C. # _____
 II
 NA

LOCATION: **QC31 MS**

EQUIPMENT: **Optical Comp. 7188**
Phil Gary

DATE: **JAN 10 2005**

K.J. Chiang
 2/4/05

ITEM NO.	QTY REQ.	SIZE	CODE IDENT NO.	PART OR IDENTIFYING NO.	NOMENCLATURE OR DESCRIPTION
PARTS LIST					
BASIC DIMENSION	DECIMALS		FRAC-TIONS	CONTRACT	SOUTHWEST RESEARCH INSTITUTE SLOW STRAIN RATE SPECIMEN DRAWING NO. 20-3704-042-1 SCALE 2 = 1 SHEET
UNDER 6	5 PLACE	3 PLACE		DRAW 4-10-1992 A. NAGY	
6-64 THRU	2 .01	2 .003	2 1/32	OKD	
OVER 64	2 .03	2 .015	2 1/16	MECH	
ANGLES	2 .00	2 .015	2 1/8	ELECT	
MATERIAL 0.5 THICK PLATE SUPPLIED					
FINISH 16 RMS					

K.J. Chiang 1-13-04
 Initiator: K. Chiang Date

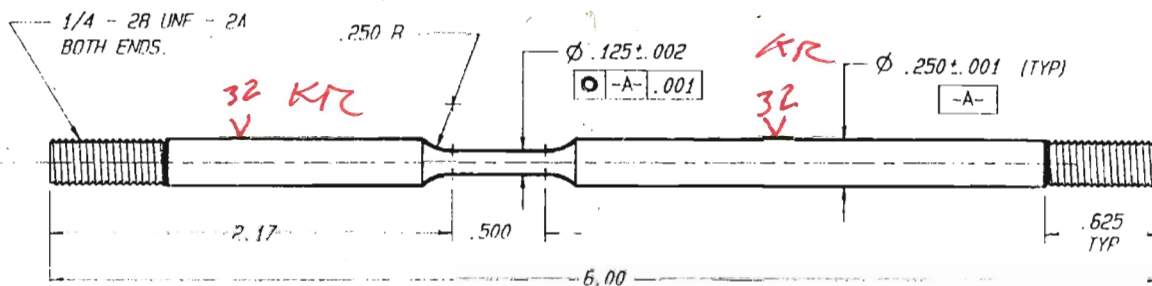
Praveen Kumar for V. Jain
 Reviewer: V. Jain Date 1/13/04

Mark R. Shusterman 1/13/04
 QA Approval: R. Brient Date for

177

Ken Chiang
 SwRI-CNWRA
 Phone: (210) 522-2308
 Fax: (210) 522-5184
 e-mail: Kchiang@swri.org

SwRI DRAWING # 20-03704-042-001



- NOTE:
- DO NOT UNDERCUT RADII
 - USE LOW STRESS MACHINING PROCEDURE

ITEM NO.	QTY	SIZE	CODE IDENT NO.	PART OR IDENTIFYING NO.	NOMENCLATURE OR DESCRIPTION
					0.5 THICK PLATE SUPPLIED
					16 RMS
PARTS LIST					
BASIC DIMENSIONS		DECIMALS		CONTRACT	
	2 PLACE	3 PLACE	FRACTIONS	DRAWN	4-10-1992 A. NAGY
UNDER 8	± .01	± .005	± 1/32	CHKD	
8-24 INCL	± .03	± .010	± 1/16	MECH	
OVER 24	± .08	± .015	± 1/8	ELECT	
ANGLES	± 0° 30'		± 1° 0'		
DRAWING NO. 20-3704-042-1					
SCALE 2 = 1 SHEET					

K.J. Chiang 1-13-04
 Initiator: K. Chiang Date

Parul Jain for V. Jain
 Reviewer: V. Jain Date 1/13/04

Mark R. Shusterman 1/13/04
 QA Approval: R. Brient Date

LOCATION CC 30 M.S.
 EQUIPMENT 0-1 Me 5056
 J.C. # 80498
 Cal 30-6C-3
 Surface Area 7188
 Procedure: L.P.C.
 Project # 60
 TOTAL PCS. INSPECTED
 TOTAL PCS. ACCEPTED
 TOTAL PCS. REJECTED
 "NR #" IF REJECTS
 INSPECTOR [Signature]
 DATE NOV 10 2004
 K.J. Chiang 2/4/05

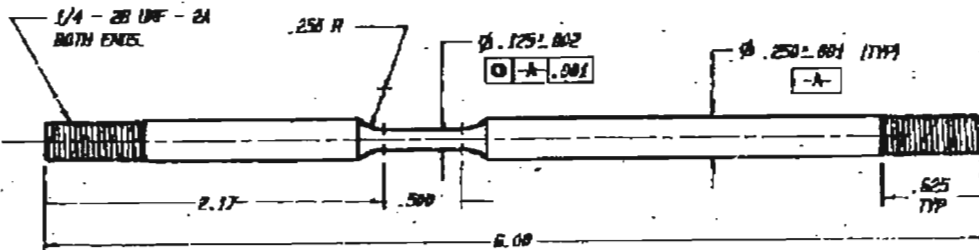
From: S0/H/2 K.J. Chiang 1/4/04

01/20/2004 11:02 #026 P.002/002

Ken Chiang
SwRI-CNWFIA
Phone: (210) 522-2308
Fax: (210) 522-5184
e-mail: Kchiang@swri.org

Post-it® Fax Note	7571	Date	1-9-04	# of pages	1
To	Larry	From	Gilbert Rodriguez		
Co./Dept.		Co.	SWRI		
Picture #		Phone #	210-522-5704		
Fax #	713 928-6004	Fax #	210-522-5084		

SwRI DRAWING # 20-03704-042-001



NOTE:
 1. DO NOT UNDERCUT ANGLE
 2. USE LOW STRESS MACHINING PROCEDURE

ITEM NO.	QTY	UNIT	DESC	UNIT NO.	PART OR IDENTIFYING NO.	DESCRIPTION OR IDENTIFICATION	MATERIAL		REMARKS
							GRADE	STANDARD	
							0.5 THICK PLATE		SUPPLIED
							16 IN5		

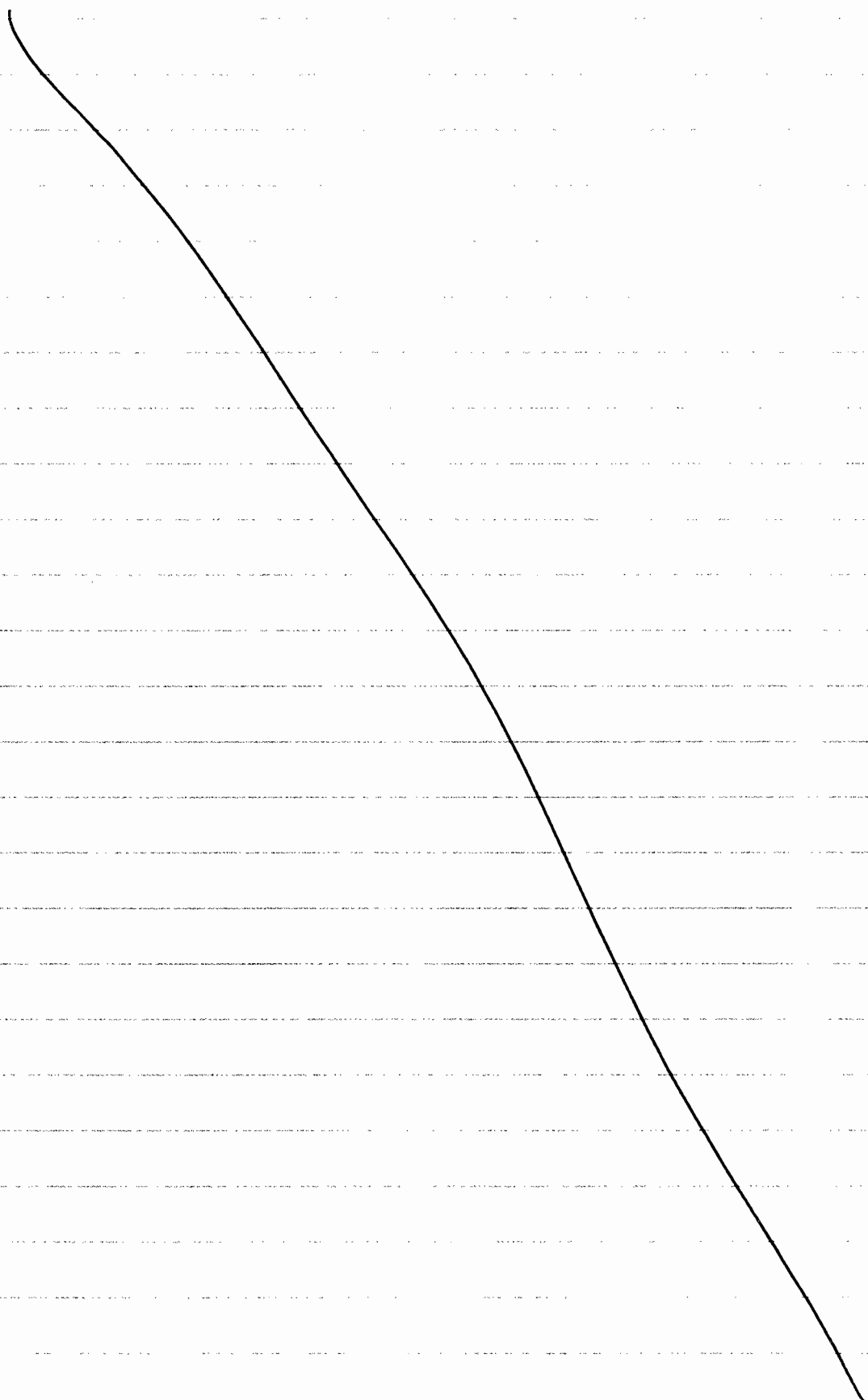
MATERIAL		GRADE		STANDARD		REMARKS	
GRADE	STANDARD	GRADE	STANDARD	GRADE	STANDARD	GRADE	STANDARD

K.J. Chiang 1-13-04
 Initiator: K. Chiang Date

Samuel D... for V. Jain
 Reviewer: V. Jain Date 1/13/04

Mark R. Shurtz 1/13/04
 QA Approval: R. Brient Date

Procedure: LOCATION **0031 M.S.**
 Project # **90337**
 J.C. # **66**
 TOTAL PCS. INSPECTED
 TOTAL PCS. ACCEPTED
 TOTAL PCS. REJECTED
 "NR" # IF REJECTS **NA**
 SENT BY: XEROX TETRA... DATE: 01/17/04 12:18:41 PM 11:33AM
 5225084
 INSPECTOR



2/8/05
Walter J. Machowski

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: $\frac{1}{2}$ liter*Heat # ~~2277-8-3266~~
2277-3-3266 ⁸⁰ 6/21/06*

KCl	3.276	# 035662	NaHCO ₃	48.22	028924
NaCl	2.745	041475	NaF	1.548	006679
NaNO ₃	4.371	020809			
Na ₂ SO ₄	10.355	035461	pH	7.74 $\xrightarrow{\text{adj}}$	8.31
					final 9.03

Reagents measured with

Model: OHAUS

SN: 2883

Cal: 14 JAN 05

Due: 14 JUL 05

Counter Electrode: Pt flag Reference Electrode: Ag/AgCl w/3M KCl

Gas: N₂ (99.999) Ecorr: -198 mV*in house*

Eapplied: +200 mV

Potentiostat: E5C 440-2 SN: 9209138

Specimen Visual:

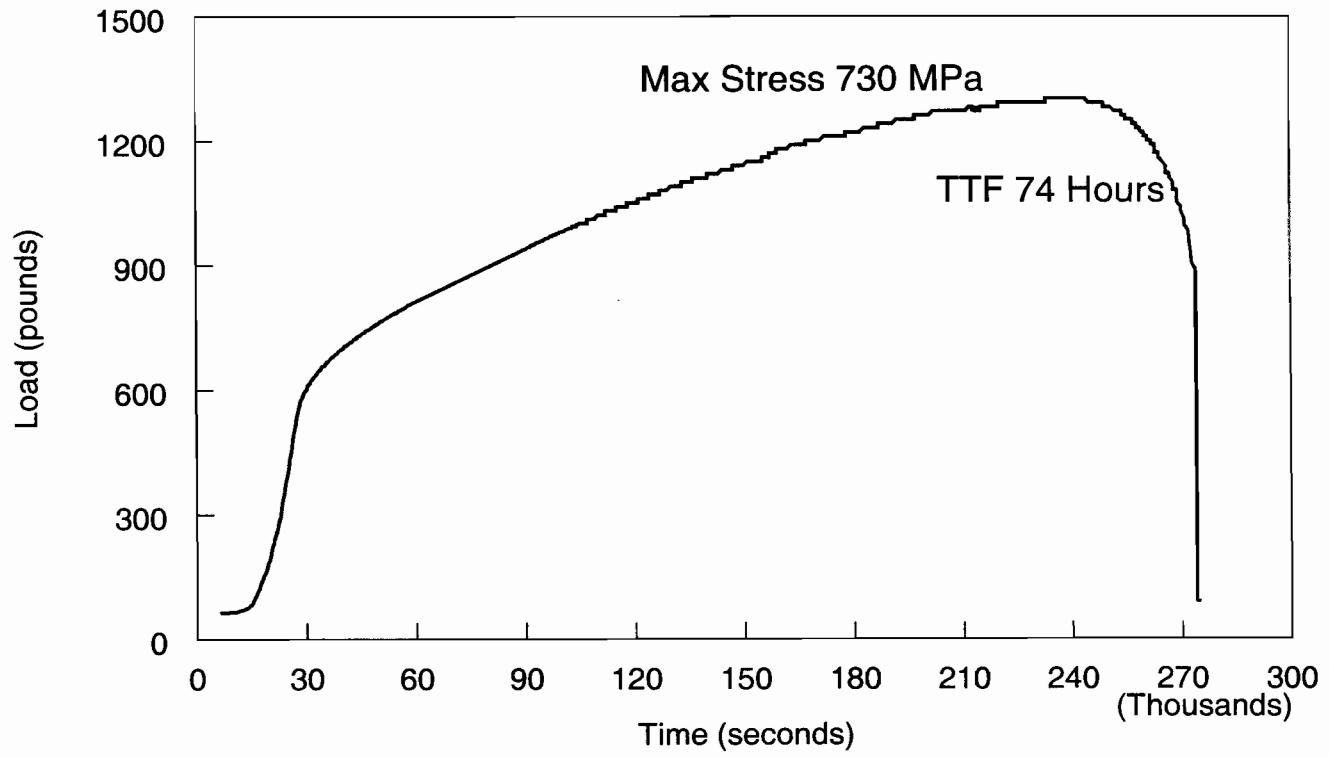
*gray-metallic; no evidence
of cracking, ductile failure*

$$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

*data file: SSRMA22 - SCW 31**2/8/05**Walter J. MacKowski*

Slow Strain Rate Test Simulated Concentrated Water

Test SSRMA22_SCW31
+200 mV



2/8/05
Walter J. Markowski

SLOW STRAIN RATE TEST

Objective: see page #5 ^{Notebook #695} → GMAW "A"
 Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001

Solution: ^{Welded specimen heat #2277-8-} ~~3266~~ ^{BP 6/27/06}
 $\frac{1}{2}$ liter ^{GMAW 2277-3-3292 5.11u WWS13/XY19775611} ~~2277-3-3266~~
 used remainder of *SCW 31 solution

final pH 8.85

Reagents measured with Model: ~~OT045~~ SN: 2883
 Cal: 14 JAN 05 Due: 14 JUL 05

Counter Electrode: Pt flag Reference Electrode: Ag/AgCl 43M KCl
 Gas: N_2 (99.999) Ecorr: -190mV in house

* Applied: +415mV Potentiostat: FSC 440-2 SN: 9209138

Specimen Visual:

bright, shiny, no cracks
 evident, however short TTF

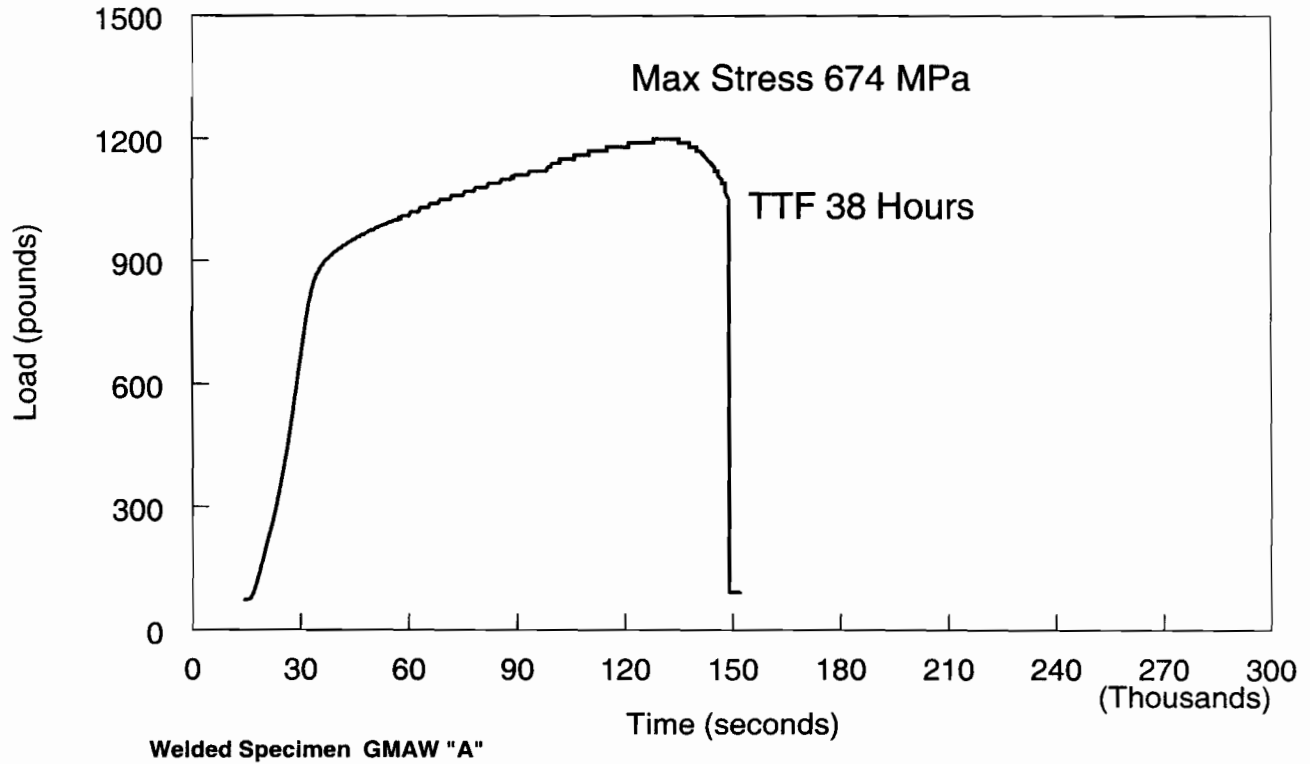
$\epsilon^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$
 * temp correction at 98°C

data file: SSRMA22-SCW32

Walter J. Machowski
 2/14/06

Slow Strain Rate Test Simulated Concentrated Water

Test SSRMA22_SCW32
+400 mV



Walter J. Machowski
2/14/05

SLOW STRAIN RATE TESTObjective: see page #5 → ^{notebook} #695 welded pec # 11A

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 Weloco

Heat # ~~2277-8-3266~~ 2277-5-3292

Solution: K / L

~~2277-5-3266~~ 6/27/06 Filler W813/XX1778G11

88.71 g NaHCO₃ # 028924
 29.25 " NaCl # 042966
 7.68 g NaOH # 897895
 111.79 g KCl # 03566 ±

Reagents measured with

Model: OHAUS
Cal: 14 JAN 05

SN: 2883

Due: 14 JUL 05

Counter Electrode: Pt flg

Reference Electrode: Ag/AgCl w/3M KCl
in buretGas: N₂ (99.999)

Ecorr: -188 mV

* Applied: +415 mV

Potentiostat: LSC 440-2 SN: 9209138

Specimen Visual:

$$\epsilon^{\circ} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

* temp correction @ 95°C

data file: SSRMA22-SCW33

2/25/05

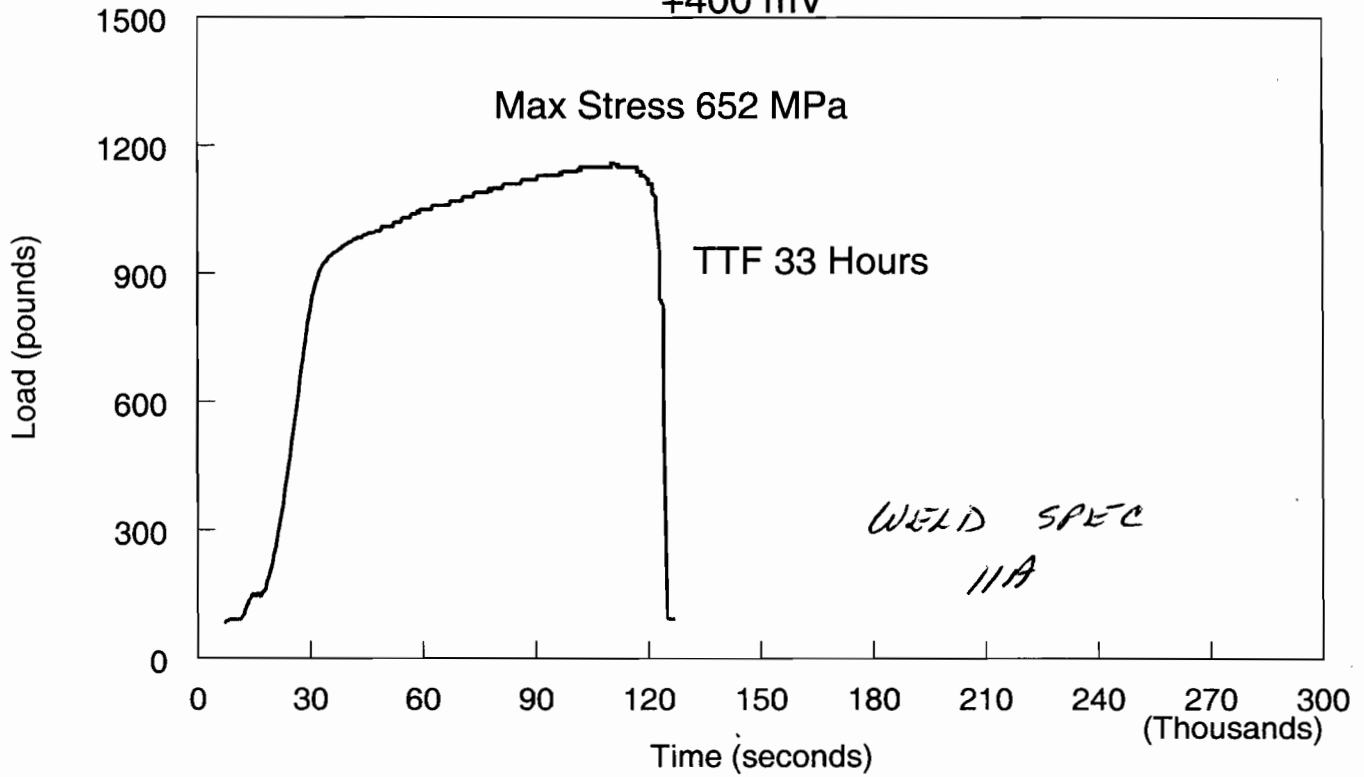
Walter J. Kucharski

Slow Strain Rate Test

1.05M NaHCO₃; 0.5M NaCl; 0.19M NaOH; 1.5M KCl

Test SSRMA22_SCW33

+400 mV



2/25/05
Walter J. Macintosh

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695*
Q GMAW weld spec #12A

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *welco* *1363*
Heat ~~2278-8-3266~~ *4/27/04*
 Solution: *ambient air* *2277-3-3292 filler WN813*
XX19778611

Reagents measured with Model: *N/A* SN:
 Cal: Due:

Counter Electrode: Reference Electrode:

Gas: *N/A* Ecorr:

Applied: Potentiostat: *N/A* SN:

Specimen Visual:

looks like ductile fracture

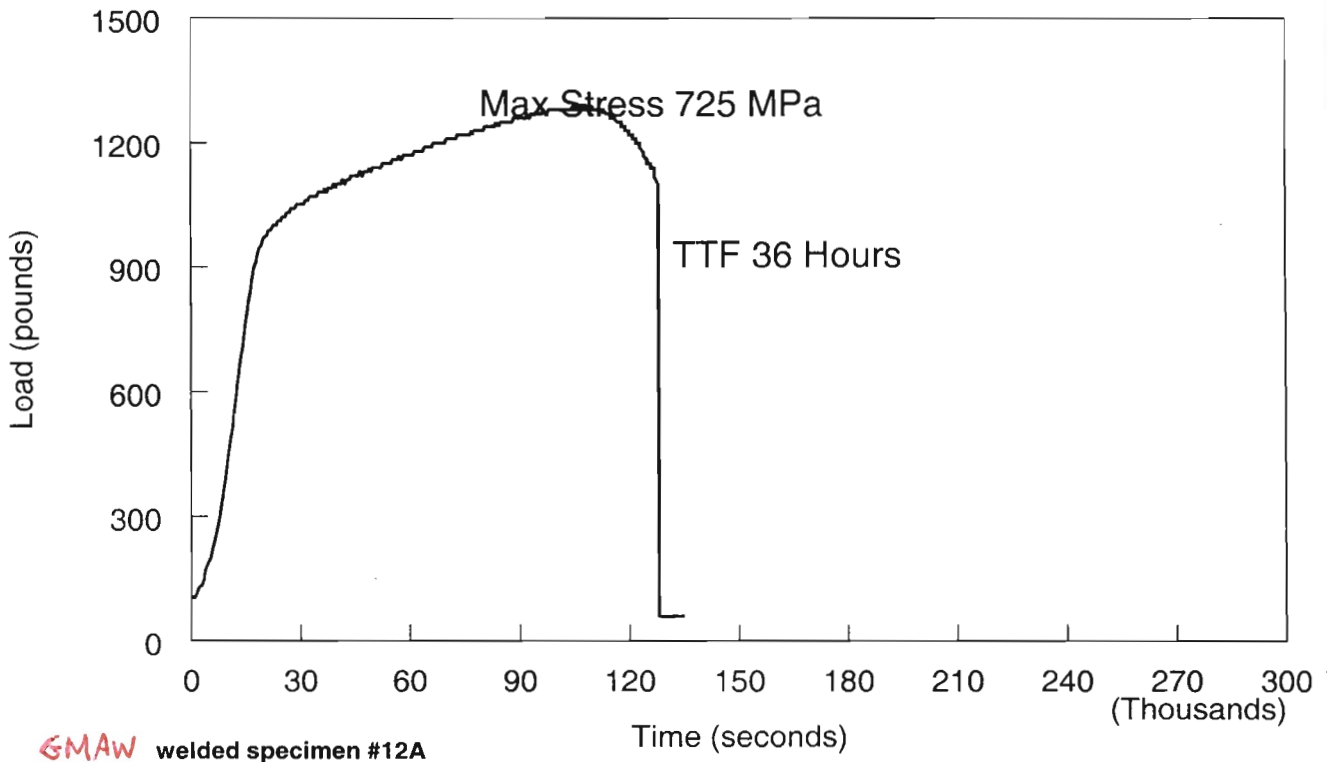
$$E^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

file: *SSRMA22-SCW34*

Walter J. Machowski
3/9/05

Slow Strain Rate Test Air @ Ambient Temperature

Test SSRMA22_SCW34



Walter J. Macintosh
3/9/05

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695 GMAW #12 B*
 Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *welbes Heat 2278-8-3266* *8/27/04*
 Solution: *2277-3-3292 WJ813/XY19778611*

ambient air

Reagents measured with Model: *N/A* SN:
 Cal: *N/A* Due:

Counter Electrode: Reference Electrode:

Gas: *N/A* Ecorr:
 Applied: Potentiostat: *N/A* SN:

Specimen Visual:

looks like ductile fracture

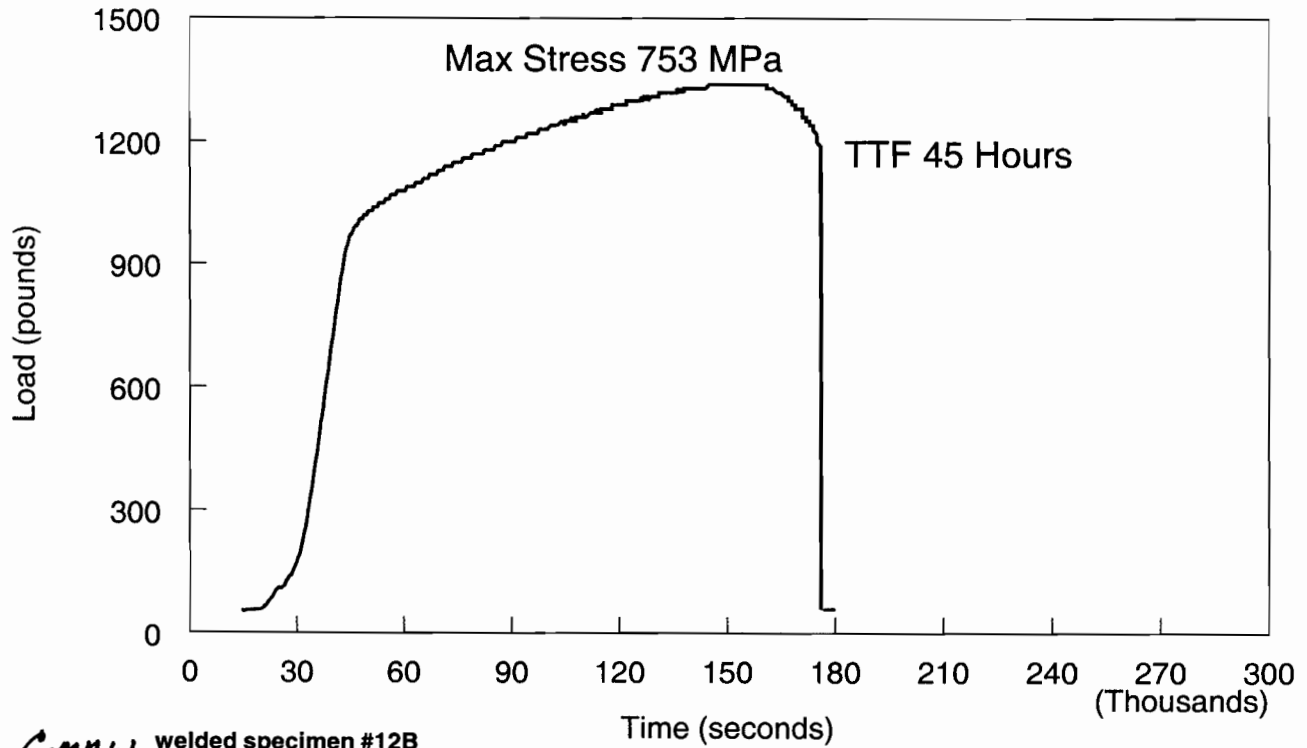
$$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

data file - SSR MA22 - SCW35

Walter J. MacKowski
3/11/05

Slow Strain Rate Test Air @ Ambient Temperature

Test SSRMA22_SCW35



GMAW welded specimen #12B

Walter J. Machowski
3/11/05

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695* *GTAW #10A*
 Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *weldeo* *8/27/04*
Heat ~~2278-8-3264~~
 Solution: *2277-3-3292 WN 813/XY19770611*

ambient air

Reagents measured with Model: SN:
 Cal: *N/A* Due:
 Counter Electrode: *N/A* Reference Electrode:
 Gas: Ecorr:
 Applied: Potentiostat: *N/A* SN:
 Specimen Visual:

looks like ductile fracture

$$\dot{\epsilon}^D = 3.2 \times 10^{-6} \text{ s}^{-1}$$

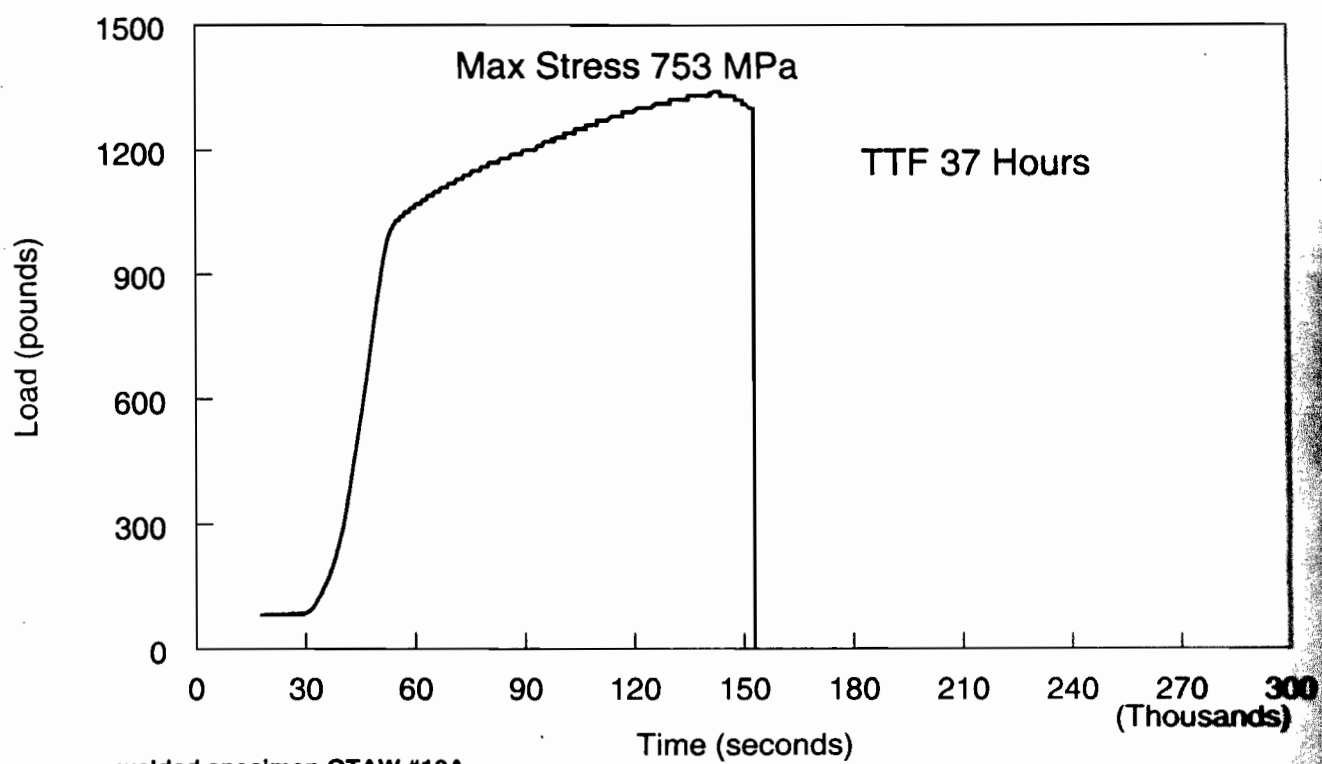
data file: SSRMA22-SCW 36

Walter J. MacKowski
3/21/05

Slow Strain Rate Test

Air @ Ambient Temperature

Test SSRMA22_SCW36



welded specimen GTAW #10A

Walter J. Machowski
3/21/05

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695* GTAW #10B
 Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *welds*
~~2278-8-3266~~
 Solution: 2277-3-3292 *WN/813* *OK 6/27/06*
XX 1977 8611
ambient air

Reagents measured with Model: *N/A* SN:
 Cal: *N/A* Due:

Counter Electrode: Reference Electrode:

Gas: *N/A* Ecorr:
 Applied: Potentiostat: *N/A* SN:

Specimen Visual:

looks like ductile fracture

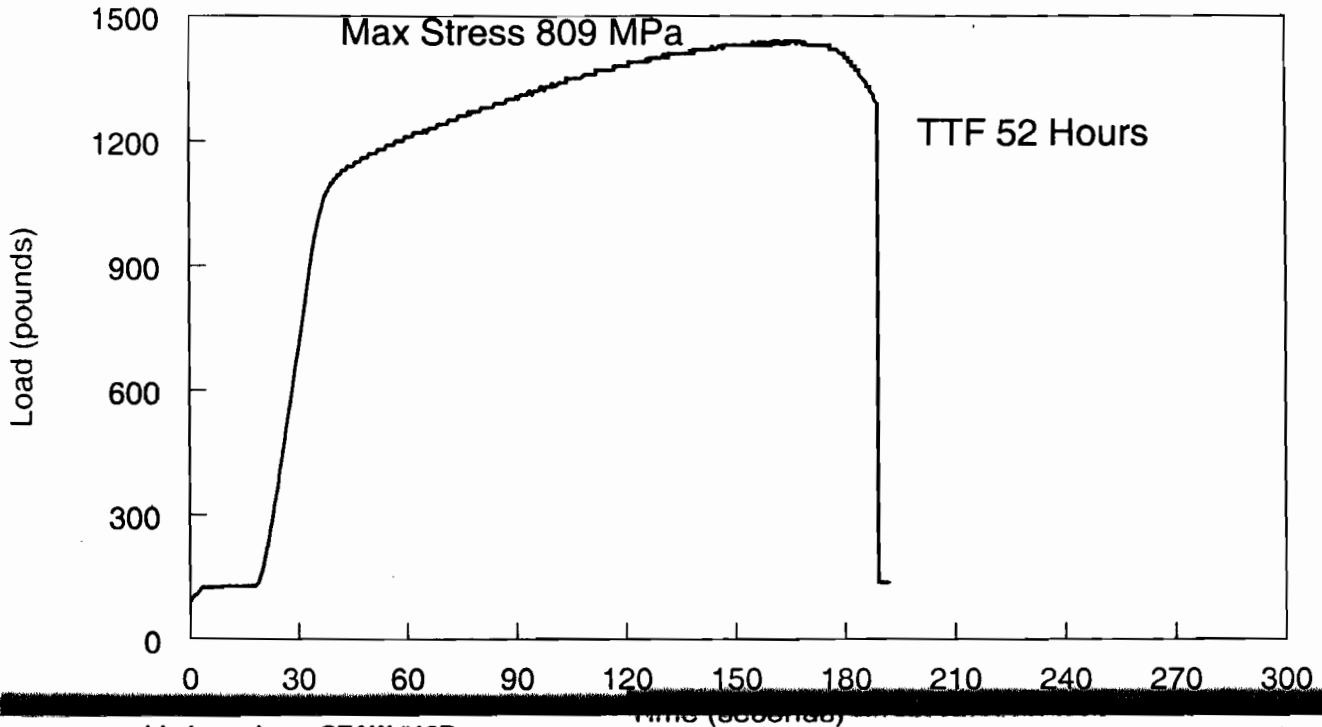
$$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$$

data file: SSRMA22-SCW37

Walter J. Mackowski
3/21/05

Slow Strain Rate Test Air @ Ambient Temperature

Test SSRMA22_SCW37



welded specimen GTAW #10B

Walter J. Macchowski
3/21/05

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695 G M A M #13A*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *welcos*
Heat 2278-8-3266 *3/27/06*
2277-3-3292 *WN 813/XY19770611*

~~Solution:~~ *6/27/06*
thermally aged @ 1125°C
for 20 min
ambient air

Reagents measured with Model: SN:
Cal: Due:

Counter Electrode: Reference Electrode:

Gas: *N/A* ~~Ecorr:~~

Applied: Potentiostat: SN:

Specimen Visual:

looks like ductile fracture

Both 13 A's 0 Specimen ex #194 & #195 ps #196 & #197
Oven: Lindberg sw # 909172 model #51333
Omega AH22 sw # T94140 cal: 11/4/04 due: 5/4/05
thermocouple # 325 cal: 2/7/05 due: 6/05/05
set point: 1125°C oven Temp: 1134°C 20 min H₂O Quenches
6/27/06

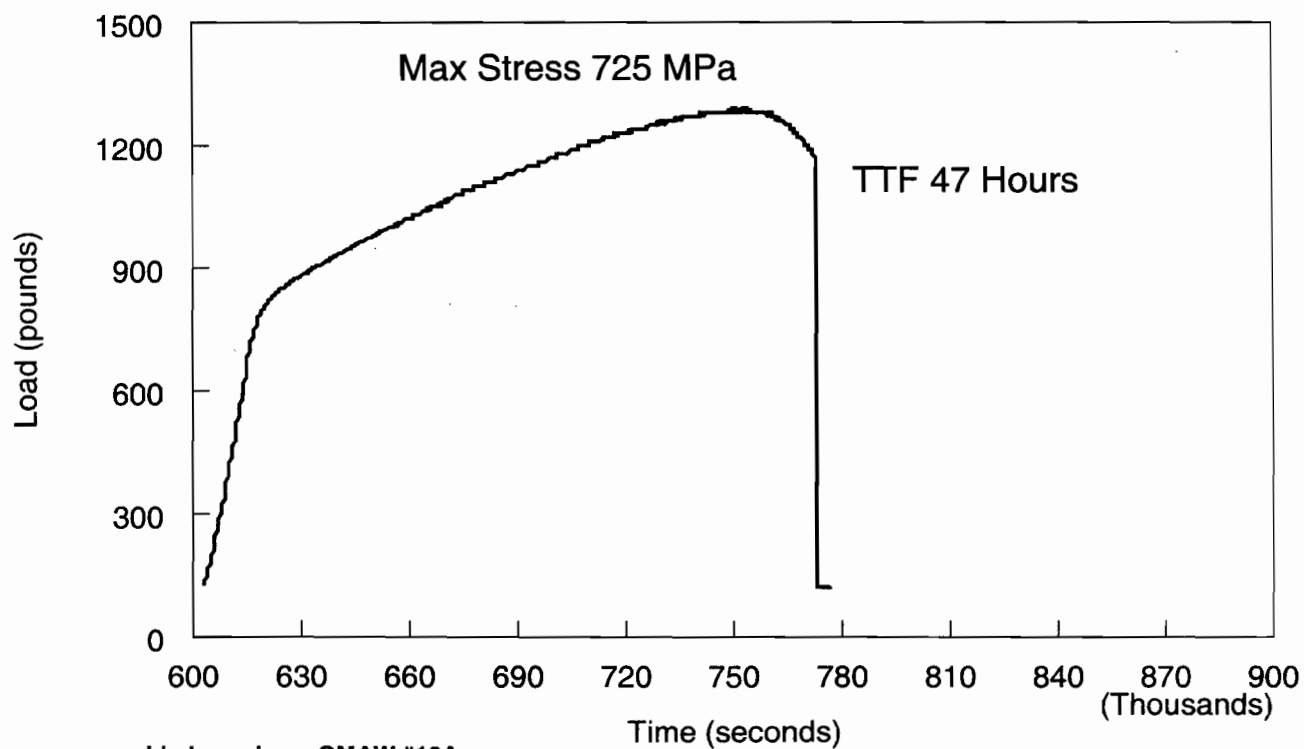
$\epsilon^0 = 3.02 \times 10^{-6} s^{-1}$

data file: SSRMA22 - SCW 38

Walter J. Macfarlane
3/30/05

Slow Strain Rate Test Air @ Ambient Temperature

Test SSRMA22_SCW38



welded specimen GMAW #13A
aged at 1125°C for 20 minutes

Walter J. Machowski
3/30/05

SLOW STRAIN RATE TEST

Objective: see page #5 *Notebook #695 GMAW #13B*

Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *w/loes*
Heat 2278-8-3266
2277-7-3292 *WN813/xx1977 5G11* *skt 6/27/06*

Solution: *thermally aged @ 1125°C / 20 min*
ambient air

Reagents measured with Model: SN:
 Cal: Due:

Counter Electrode: Reference Electrode:

Gas: *N/A*
 Ecorr:

Applied: Potentiostat: SN:

Specimen Visual: *looks like ductile fracture*

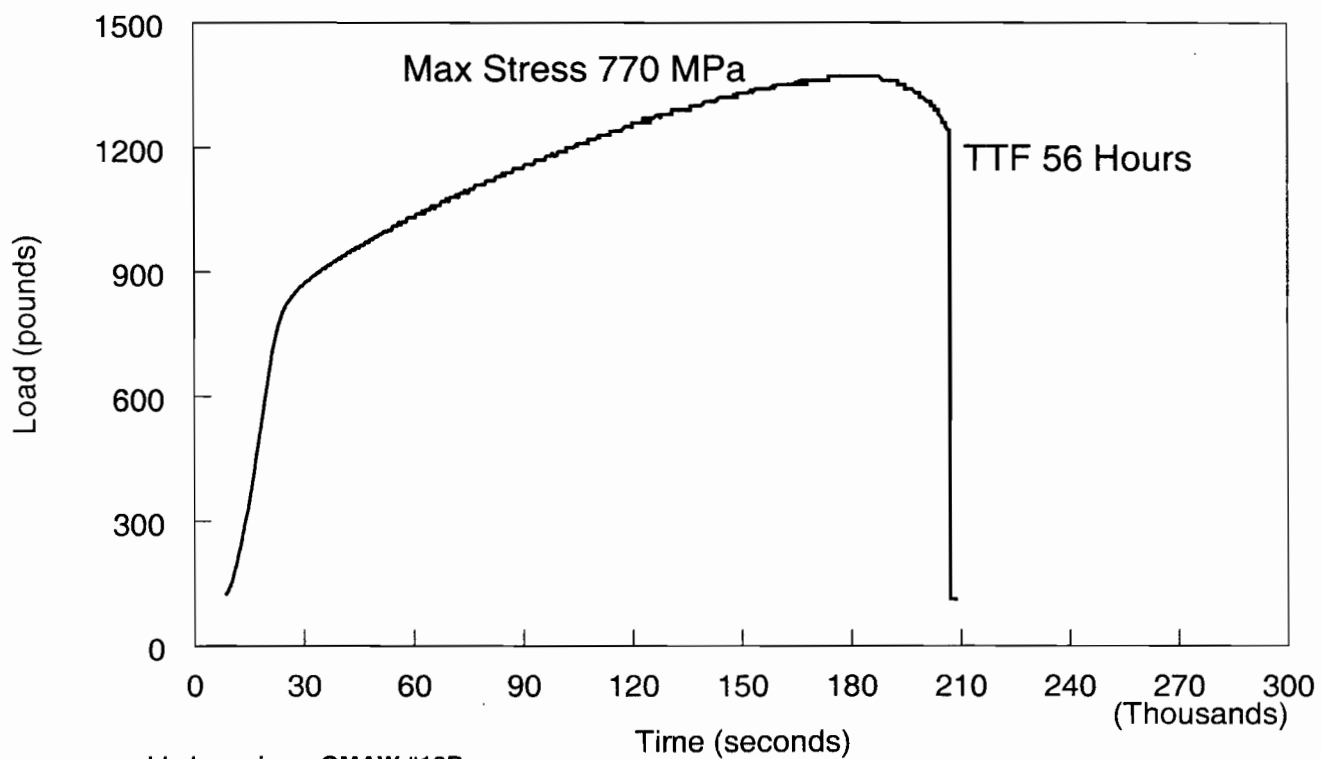
$\dot{\epsilon} = 3.2 \times 10^{-6} \text{ s}^{-1}$

skt 6/27/06
Test SSRMA22 - SCW39

Walter J. MacKorchi
4/5/05

Slow Strain Rate Test Air @ Ambient Temperature

Test SSRMA22_SCW39



welded specimen GMAW #13B
aged at 1125°C for 20 minutes

Walter J. Machinski
4/5/05

SLOW STRAIN RATE TESTObjective: see page #5 *Notebook #695 RMAW #14B*Specimen: MA Alloy 22 SwRI Drawing # 20-03704-042-001 *weldeo**Heat 2278-8-3266**2277-3-3292 WNR3/vx19776611**OKP
6/27/06*

Solution:

*balance of solution - SCW 33
on page 184*

Reagents measured with

Model: *OXFORS*SN: *2883*Cal: *as on p. 184*Due: *as on p. 184*Counter Electrode: *Pt foil*

Reference Electrode:

*Ag/AgCl 0.3M KCl*Gas: *N₂ (99.999)*Ecorr: *-200 mV**in house*** Applied: 415 mV*Potentiostat: *KSC440-2* SN: *9209138*

Specimen Visual:

*some cracks visible
maybe brittle fracture*

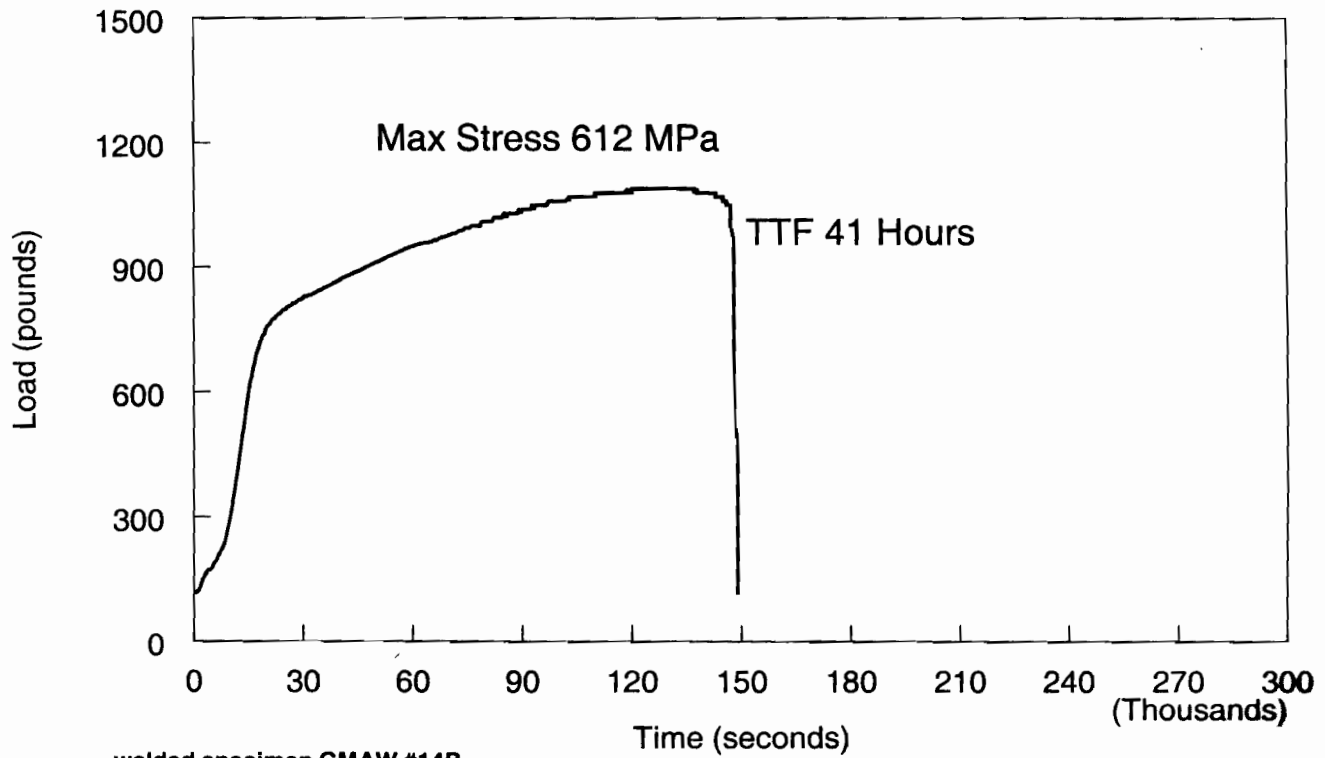
$$\epsilon^0 = 3.2 \times 10^{-6} \text{ s}^{-1}$$

** temp correction (+15 mV) @ 95°C**OKP
6/27/06*Test *SSEMA 22 - SCW 40**Walter J. Markwardt
4/6/05*

Slow Strain Rate Test

1.05M NaHCO₃; 0.5M NaCl; 0.19 NaOH; 1.5M KCl

Test SSRMA22_SCW40



welded specimen GMAW #14B
aged at 1125°C for 20 minutes

Walter J. Macchiarini
4/6/05

Continues In NB # 695.
 R.F. Chiving 4/11/05
 2157-67106

Test ID	Solution Composition	pH Initial	pH Adj'd	pH Final	Eapp(mV)	Max Stress (Mpa)	TTF(hrs)	RA(%)	% Elong	SCC?
SSRMA22_SCW28	0.5M NaCl; 7.1M KCl 0.38M NaNO3; NaCO3 as in SCW	7.87	not adj'd	9.36	400	560	39	55.9	44.4	Y
SSRMA22_SCW29	same as *28 except 3.3M KCl	7.41	not adj'd	9.44	400	590	50	50	55	Y
SSRMA22_SCW30	same as *26 except no HCO3		not adj'd		400*	787	79	76.9	87.6	N
SSRMA22_SCW31	SCW	7.73	8.31	9.03	200	730	74	77	85.2	N
SSRMA22_SCW32	SCW	7.73	8.31	8.85	400	674	38	65.9	41.6	N
SSRMA22_SCW33	1.05M NaHCO3; 0.5M NaCl 0.19M NaOH; 1.5M KCl	8.58	not adj'd	9.81	400	652	33	30	36.6	Y
SSRMA22_SCW34	AIR - GMAW #12A	N/A	N/A	N/A	N/A	725	36	56.9	38.8	N/A
SSRMA22_SCW35	AIR - GMAW #12B	N/A	N/A	N/A	N/A	753	45	48.2	50.4	N/A
SSRMA22_SCW36	AIR - GTAW #10A	N/A	N/A	N/A	N/A	753	37	51.6	40.8	N/A
SSRMA22_SCW37	AIR - GTAW #10B	N/A	N/A	N/A	N/A	809	52	49.3	59.8	N/A
SSRMA22_SCW38	AIR - GMAW #13A Thermally Aged 1125°C / 20 min	N/A	N/A	N/A	N/A	725	47	50.4	53.2	N/A
SSRMA22_SCW39	AIR - GMAW #13B Thermally Aged 1125°C / 20 min	N/A	N/A	N/A	N/A	770	56	54.8	62.4	N
SSRMA22_SCW40 #14B 1125°C/20 min	1.05M NaHCO3; 0.5M NaCl 0.19M NaOH; 1.5M KCl	8.58	not adj'd	9.73	400	612	41	53.7	45.2	Y
SSRMA22_SCW41 #14A 1125°C/20 min	SCW	7.79	8.69	10.5	400	669	50	58	57	N