

03/03/01

9:23 sensor in MgCl₂ bottle, from lot # 00036

RM = 30.97% , 22.92°

9:31 in NaNO₃ bottle. lot # 961772A

9:41 RM = 70.61% at 23.12°

9:50 RM = 70.0% 23.1°

9:54 ~~9:44~~ ^{9:54} in MgCl₂ - NaCl bottle. from lot # 00036 and #006924

10:01 RM = 34.54 at 23.2°

10:01 sensor to KCl bottle from lot # 005573

10:25 RM = 77.57% at 22.6°

10:26 to NaNO₃ - NaCl bottle. from lot # 961772A and #006924

10:35 RM = 69.18% at 22.96°

10:35 to NaNO₃ - KNO₃ - NaCl lot # 7028kDPD

10:49 RM = 65.53 at 22.67°

10:50 to MgCl₂ bottle

10:57 RM = 35.88% 22.9°

12:29 RM = ~~32.61%~~ ^{32.61%} 22.7° (should 32.8°) -

12:29 to KCl bottle lot # 005573 ✓

12:46 RM = 80.1% , 23.2°

12:50 RM = 78.85% 23.1°

Jingmei 4/2/02

3/04/01

9:00 to MgCl₂ + NaCl bottle.

16:37 30.55% 22.86°

8/10/01

17:07 to NaCl + NaNO₃ + KNO₃ (3-salt) bottle.

8/13/01

8:53 67.34% 21.55°

9:13 66.17% ^{63.5%} 21.67°

9:13 to NaCl - NaNO₃ bottle (2-salt bottle).

9:40 70.57% 21.9°

9:50 69.8% 22.5°

10:10 raise oven Temperature. to 50°

11:55 46.45% , RM = 66.46%

15:09 46.78% 66.62% NaCl - NaNO₃

15:09 changed to 3-salt bottle.

17:05 46.72% 62.63%

3/14/01

8:41 46.91 59.61% }
11:39 46.59 60.56% } $\frac{60.56 + 59.61}{2} = 60.09$

to KCl bottle

12:46

13:52 45.66% 70.24%

14:06 46.22% 73.94%

Jingmei 4/1/02

16:06 to NaNO₃ Agarn.
 14:41 46.38², 69.46% Glan T=48²
 14:42 to KNO₃ bottle
 14:45 45.54² 82.14%
 15:11 46.43² 87.94% Glan T=49²
 15:17 to NaCl
 15:46 46.34², 77.41%
 17:00 46.37 77.27%
 to 3-salts bottle.
 raise Taren to 20².

8/17/01
 8:16 67.34², 52.83%
 9:22 65.21², 52.6% bottle too dry, water added.
 9:26 to Two-salt
 11:38 64.25² 54.07% Glan T=64²
 11:41 to MgCl₂
 11:56 63.64², 32.3%
 14:01 64.11² 31.37% o.k.

J. Young
 4/3/02

8/16/01

From now use a
 11:35 to KCl bottle. Teflon sleeve on
 12:50 68.8² 74.83% sensor tip to avoid
 14:29 68.98² 74.57% salt contamination and
 14:30 to MgCl₂ bottle. to reduce sensitivity
 16:41 68.75² 25.63% -increase stability.
 to 3-salt bottle.
 8/17/01 J. Young 8/16/01

8:09 68.96² 46.68% 3-salt
 to NaNO₃ bottle
 10:14 68.75², 56.68%
 10:27 68.76² 56.71
 to NaCl
 10:24 67.85 67.60%
 11:18 68.48 67.71
 to Two-salt
 13:15 68.88², 53.83%
 to KCl ^{oven} to 85².

14:04 84.74² 71.39% ↓ 85².
 14:05 to Two salt
 14:10 80.42², 52%
 15:44 85.02² 51.40%
 15:56 To 3-salt bottle.
 J. Young
 4/3/02

16:39 35.21% 40.24%
 16:41 to MgCl₂·6H₂O bottle
 17:32 35.32% 24.09
 17:35 Change oven Temp to 40°C.
 08/10/01 (Monday)
 8:34 38.16% 26.24%
 8:40 to KCl bottle
 9:34 38.15% 71.81
 9:40 to Two salts
 10:17 38.13% 57.84%
 10:18 to 3-salts
 11:05 38.17% 62.70%
 11:05 to NaCl bottle
 11:17 38.25% 66.37%
 Test terminated.

J. Young
 4/3/02

Data processing.

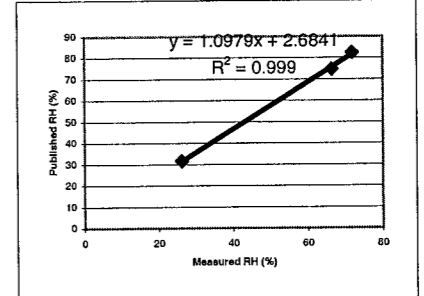
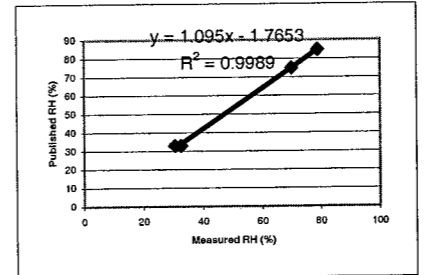
DRH_test_Hygrometer_summary.xls

Table. Comparison of Deliquescence points of single salt with salt mixtures

Salts	Deliquescence Point (Percent)						
	16.5oC	23oC**			38.13oC		
		8/8-8/13, 2001, morning			8/20/2001, morning		
Dates of data	Published*	Measured**	Corrected†	Published*	Measured**	Corrected†	Published*
Pure NaCl	76.5				66.4	75.78466	74.68
Pure NaNO3	76	70	74.8882	74.8			
Pure KNO3	95						
NaCl+NaNO3 +KNO3	30.5†	63.5	67.7707		52.7	60.74343	
NaCl+NaNO3		69.8	74.6692		57.84	66.386636	
KCl		78.85	84.57895	84.7	71.81	81.724299	82.32
MgCl2.6H2O		32.61	33.94615	32.8	26.24	31.692996	31.6
MgCl2.6H2O+ NaCl		30.55	31.69045	32.8			
			1.095	-1.7618		1.0979	2.8841

* ANL-EBS-MD-000001 REV 00 ICN 01 (2001)
 † Handbook value (Wreast, 1981)
 ** Hygrometer measurement

(also green span, 1977) J. Young 8/20/01



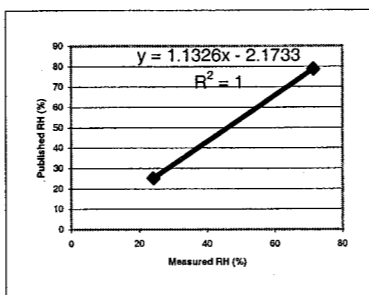
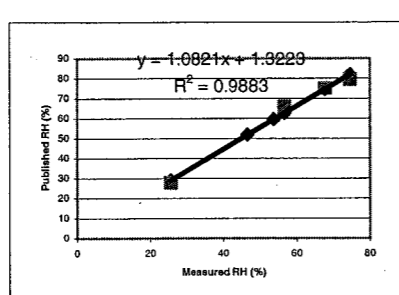
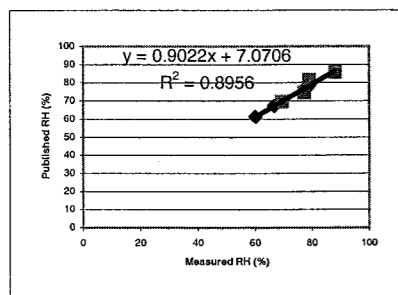
† — Corrections are based on the linear regression for pure salts.

* CRWMS M&O, Environment on the Surfaces of the Drip Shield and Waste Package Outer Barrier, ANL-EBS-MD-000001, REV 00 ICN 01, Las Vegas NV, Office of Civilian Radioactive Waste Management System, Management and Operating Contractor, 2000.

J. Young
 4/3/02

DRH_test_Hygrometer_summary.xls

Deliquescence Point (Percent)									
Dates	48oC			69oC			84.8		
	Measured**	Corrected	Published*	Measured**	Corrected	Published*	Measured**	Corrected	Published*
8/13/2001 -8/14/01 Afternoon	77.27	76.783594	74.45	67.71	74.591291	75.06			
Pure NaCl	77.27	76.783594	74.45	67.71	74.591291	75.06			
Pure NaNO3	69.46	69.737412	69.5	56.71	62.688191	66.04			
Pure KNO3	88	86.4642	85.8						
NaCl+NaNO3+KNO3	60.085	61.279287		46.68	51.834728		40.24	43.40252	
NaCl+NaNO3	66.62	67.175164		53.83	59.571743		51.4	56.04234	
KCl	78.9	78.25418	81.5	74.57	82.014497	79.49	71.39	78.68301	78.68
MgCl2.6H2O				25.63	29.056523	27.77	24.09	25.11103	25.11
MgCl2.6H2O+NaCl									
		0.9022	7.0706		1.0821	1.3223		1.1326	-2.1733

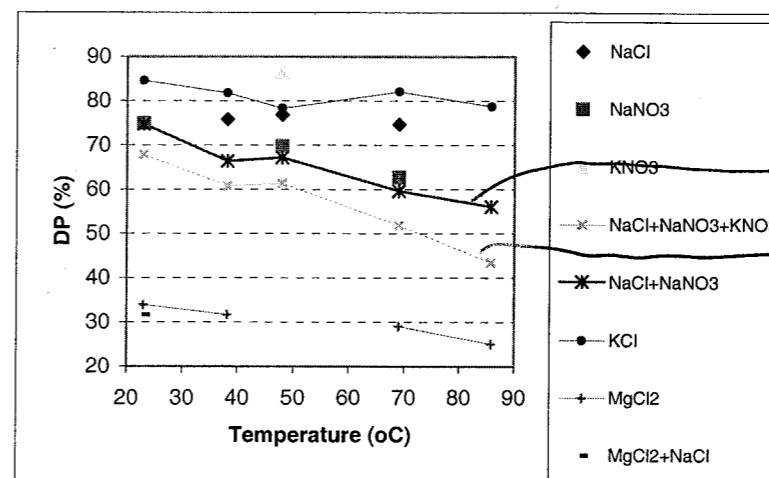


J. yams
4/3/02

DRH_test_Hygrometer_summary.xls

Delequescence Data

Salts	Temperature (oC)				
	23	38.13	48	69	85.8
NaCl		75.78466	76.783594	74.591291	
NaNO3	74.8882		69.737412	62.688191	
KNO3			86.4642		
NaCl+NaNO3	67.7707	60.74343	61.279287	51.834728	43.402524
NaCl+NaNO3	74.6692	66.386636	67.175164	59.571743	56.04234
KCl	84.57895	81.724299	78.25418	82.014497	78.683014
MgCl2	33.94615	31.692996		29.056523	25.111034
MgCl2+NaCl	31.69045				



$y = -0.2835x + 76.794$
 $r^2 = 0.9674$

$y = -0.3749x + 76.79$
 $r^2 = 0.9674$

J. yams
4/3/02

Continued on

10/22/01

Experimental Determination of the Deliquescence Points of Multi-component Salts and the Conductivity Behavior of Salt Deposits As a Function of Relative Humidity

Lietai Yang, Roberto T. Pabalan, Lauren Browning
Center for Nuclear Waste Regulatory Analyses
Southwest Research Institute
San Antonio, TX.

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4/3/02

Table 1 Comparison between measured and published deliquescence points for pure salts

Temperature (°C)	Measured (%)	Published (%)	Error (%)

DRM 2-2-10/22/01

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DRM - Test - Hygrometers
Summary - xls
Comparison - Error.

Table 2 Measured deliquescence points for salt mixtures

Temperature (°C)	23.0	38.1	48.0	69.0	85.8

94% Confid. interval ± 1.51%
Jug.
10/22/01

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file: same as Table 1.
Tab: Measured - DP - Mixture

File: DP-30.C.xls
Tab: Data

Table 3. Measured values of impedance and relative conductances of salts at different relative humidities at 30°C.

RH (%)	Impedance	Reciprocal of	Rel	RH DP (%)

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File: DP-5000.xls

Tab: Data

Table 4. Measured values of impedance and relative conductances of salts at different relative humidities at 50°C.

RH (%)	Impedance	Reciprocal of	Rel.	RH-DP (%)
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10/2

7

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Table 4. Measured values of impedance and relative conductances of salts at different relative humidities at 50°C. Cont'd.

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10/2

9

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File: DP_Hyfrometer

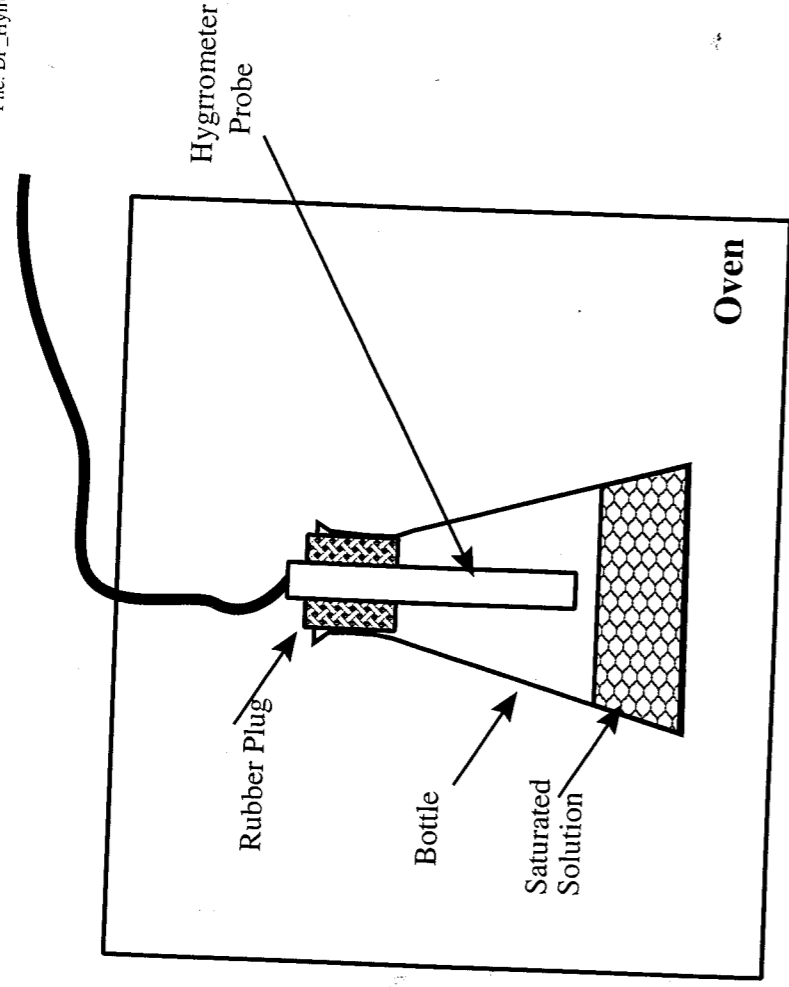


Figure 1

J. Jans 4/3/02

File:Teflon_Cell3

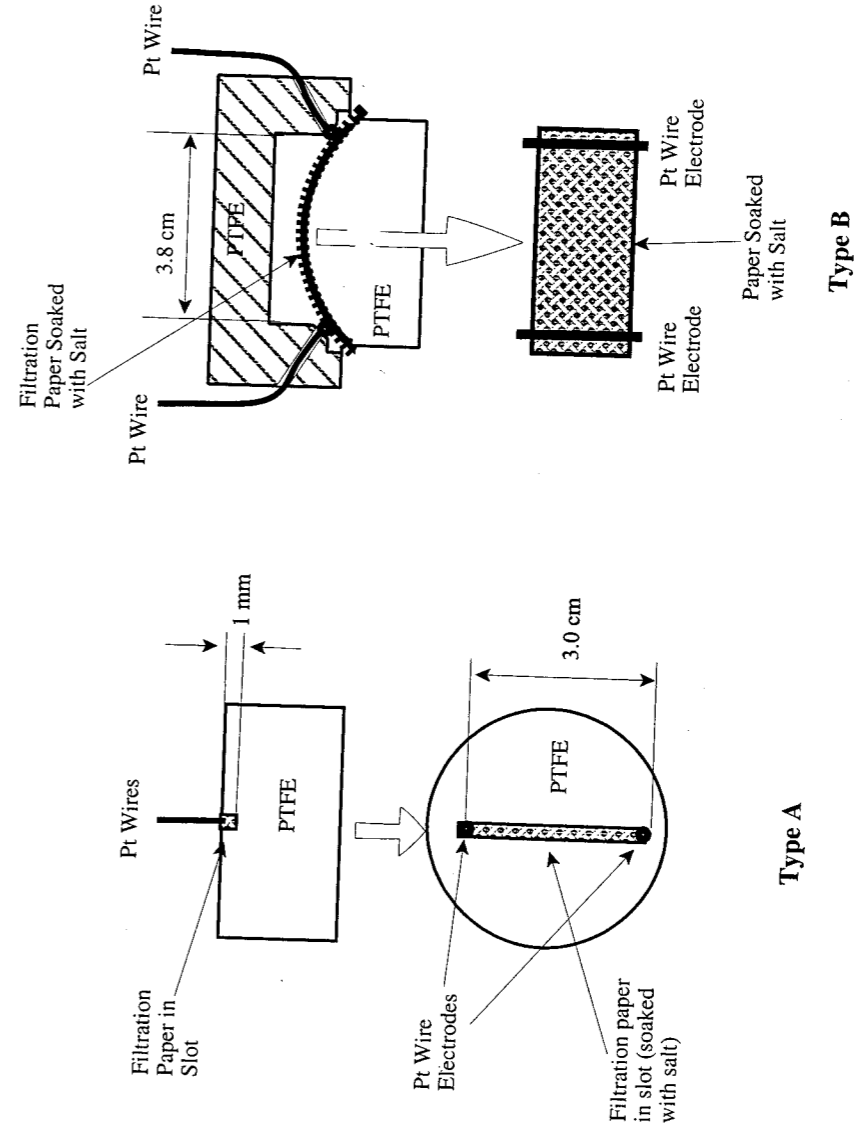


Figure 2

J. Jans 4/3/02

File: DRH_test_Hygrometer_summary.xls / TabChart_Compare_Published

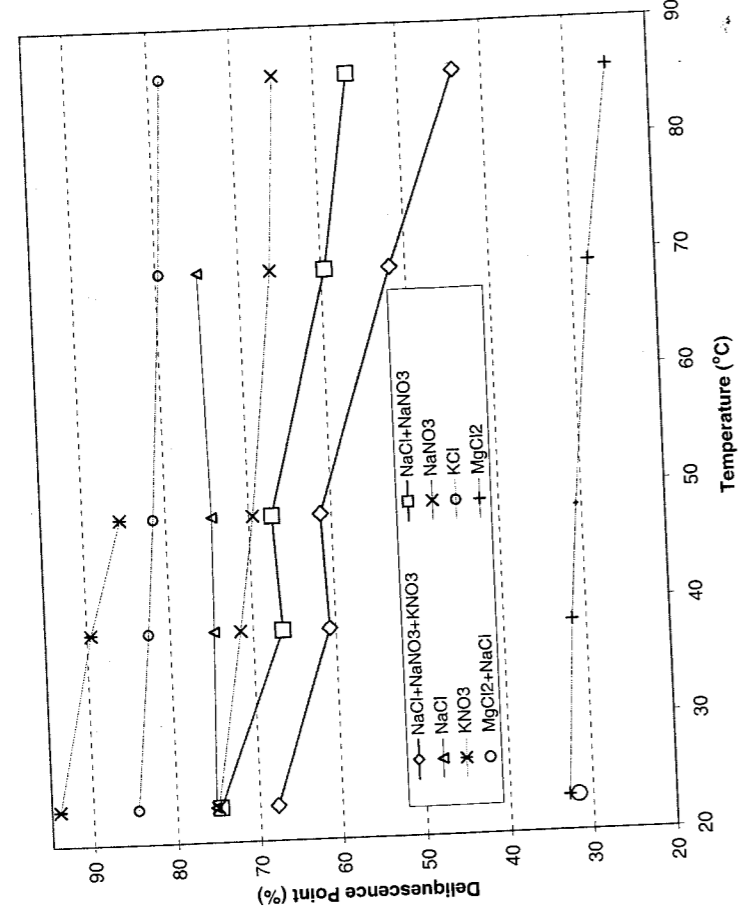


Figure 3 Comparison between the deliquescence points of pure and mixed salts. Note: Values for salt mixtures were measured; values for pure salts are from literature (Greenspan, 1977).

J. Jansz
4/3/02

File: 0612b_00_14a_01a.xls / Tab: Chart1

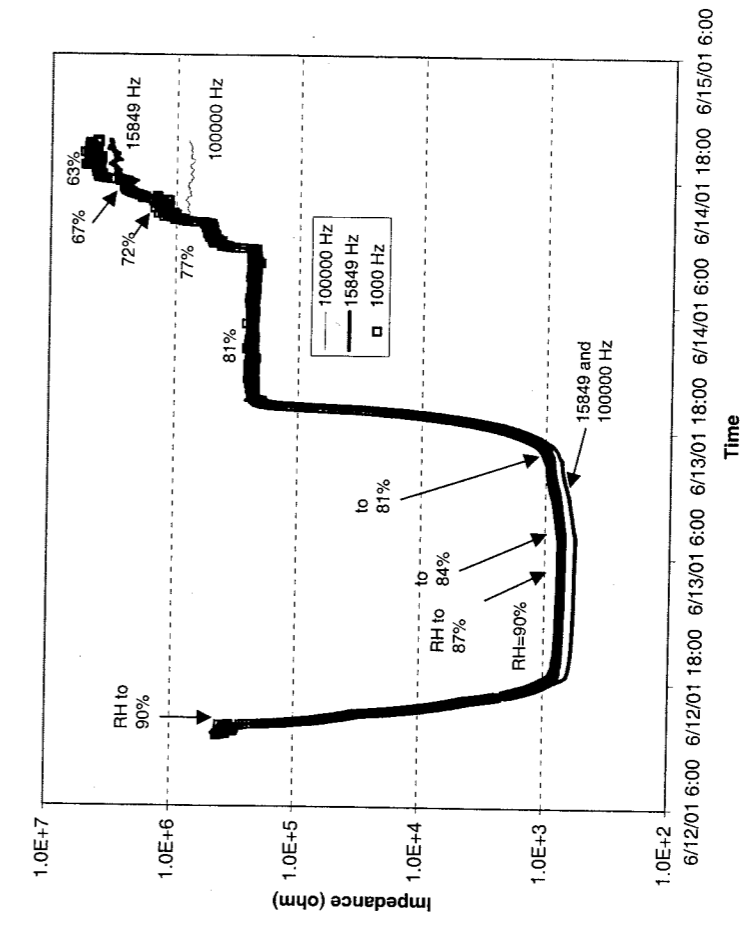


Figure 4 Typical responses of impedances measured at three frequencies to the changes in relative humidity. Note: Filtration Paper dimension: 1x3x30 mm.

J. Jansz
4/3/02

File: 0612b_DP.xls / Tab: DP_Chart

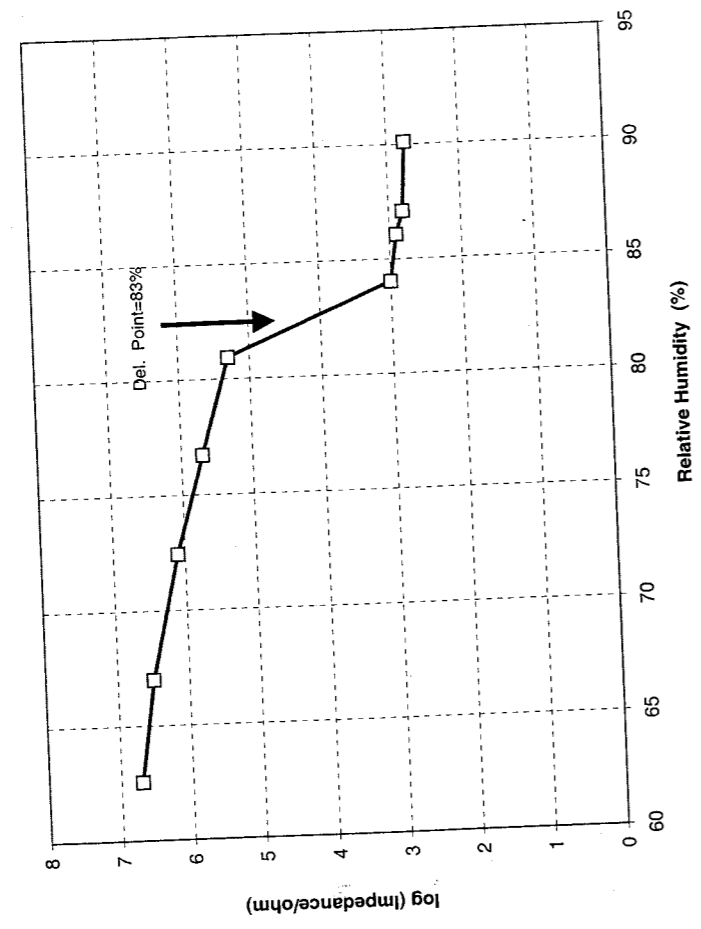


Figure 5 Relationship between the impedance and the relative humidity
Note: Filtration Paper dimension: 1x3x31 mm.

J. Yang
4/3/02

File: DP_300C.xls / Tab: Chart

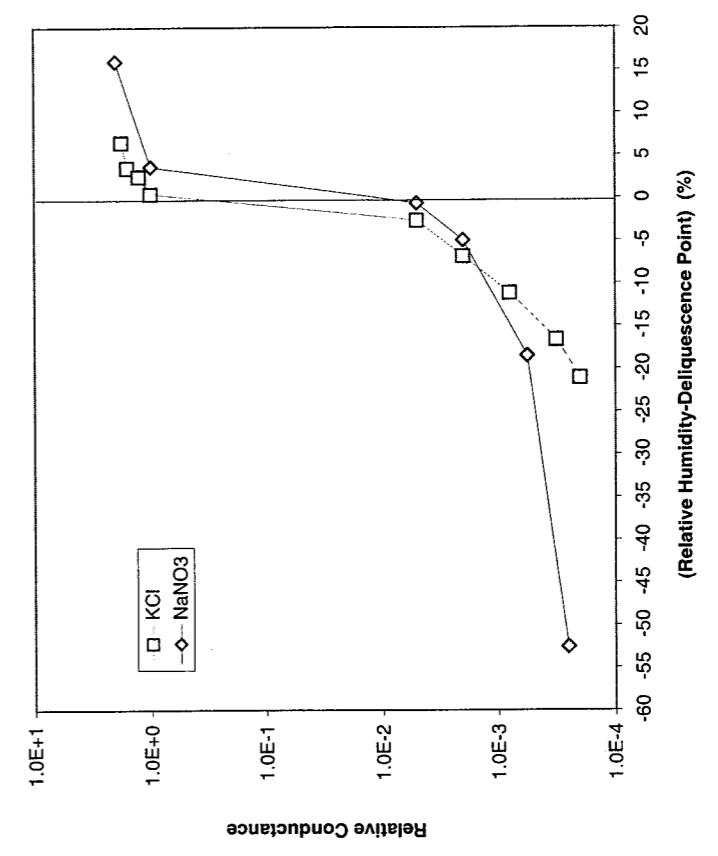


Figure 6. Measured relative conductances of salts as a function of the difference between the relative humidity and the deliquescence points at 30°C.
Note: Cell: Type A; Filtration Paper dimension: 1x3x30 mm.

J. Yang
4/3/02

File: DP_500C.xls / Tab: Chart

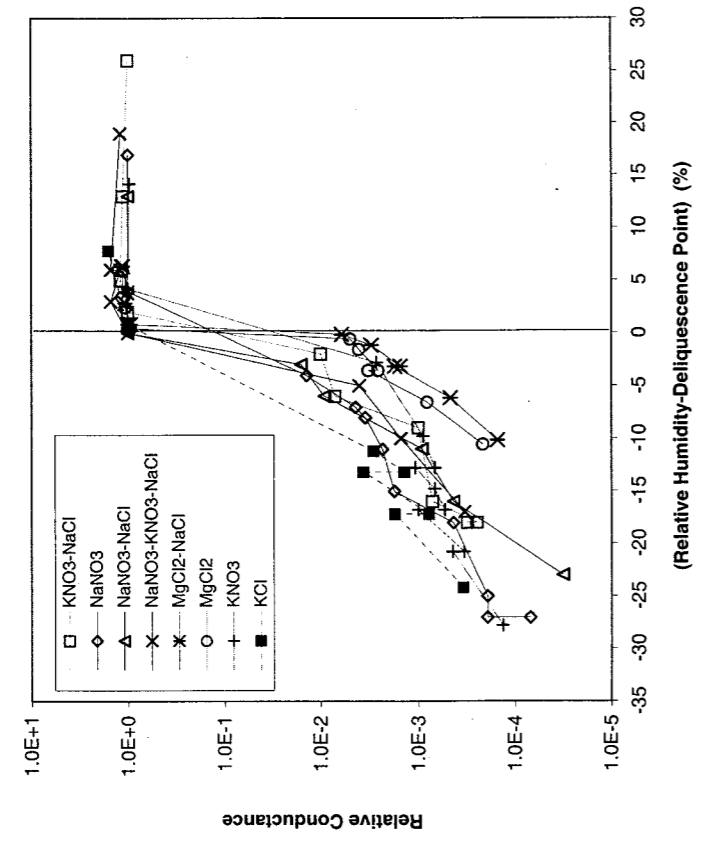


Figure 7 Measured relative conductances of salts as a function of the difference between the relative humidity and the deliquescence points of the salts at 50°C.

Note: Cell: Type B; Filtration Paper dimension: 0.23x26x46 mm.

J. Yang
4/3/02

REFERENCES

ASMT Designation: E104-85 (Reapproved 1996), "Standard practice for maintaining constant relative humidity by means of aqueous solutions Standard", 1996

Brossia, C.S., L. Browning, D.S. Dunn, O.C. Moghissi, O. Pensado and L. Yang. Effect of Environment on the Corrosion of Waste Package and Drip Shield Materials. CNWRA 2001-003. San Antonio, TX: Center for Nuclear Waste Regulatory Analyses. September 2001.

CRWMS M&O, Environment on the Surfaces of the Drip Shield and Waste Package Outer Barrier, ANL-EBS-MD-000001, REV 00 ICN 01, Las Vegas NV, Office of Civilian Radioactive Waste Management System, Management and Operating Contractor, 2000.

CRWMS M&O, In-Drift Precipitate/Salts Analysis, ANL-EBS-MD-000045, Revision 00 ICN02, Las Vegas NV, Office of Civilian Radioactive Waste Management System, Management and Operating Contractor, 2001.

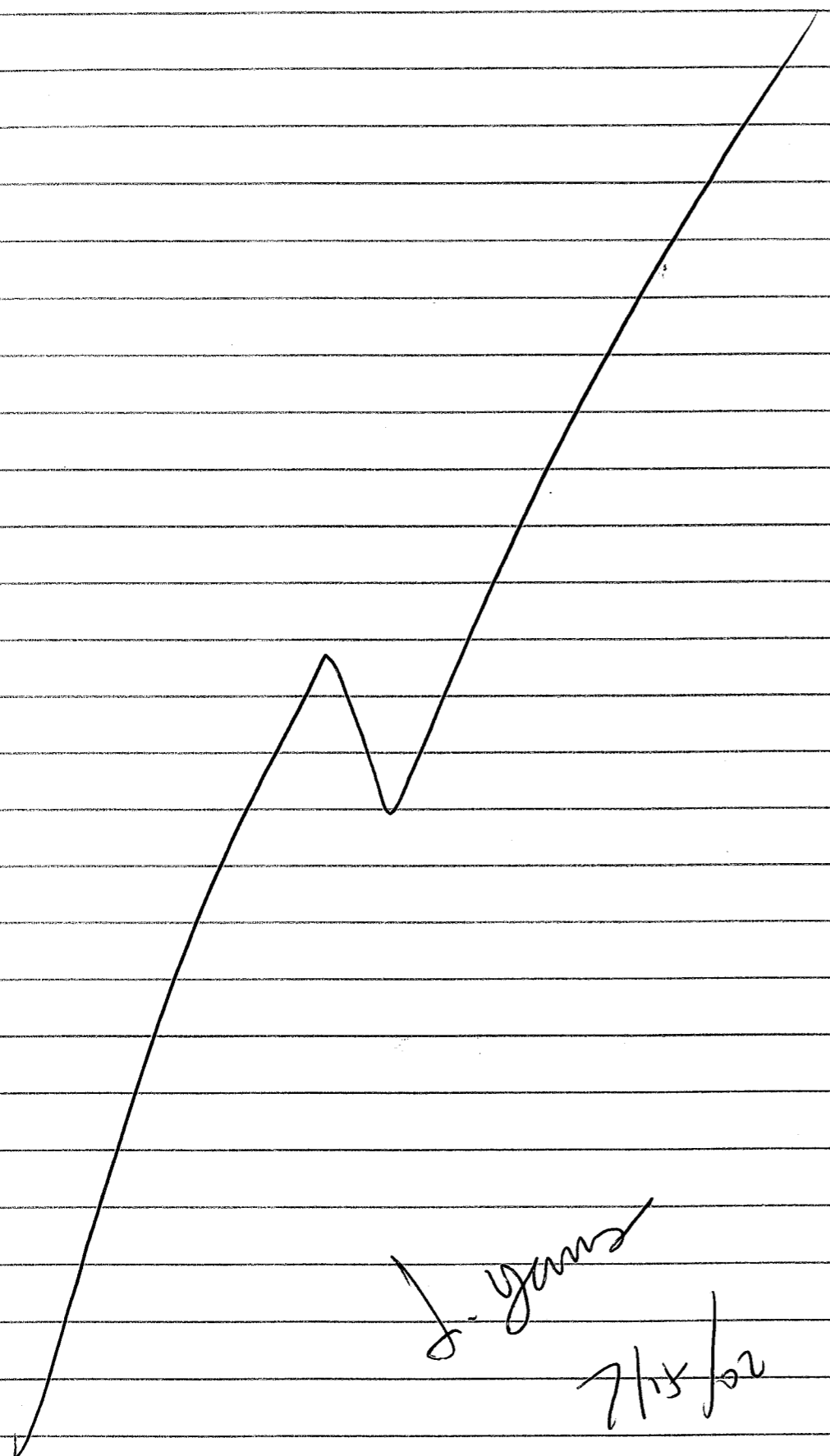
Greenspan, Lewis, "Humidity fixed points of binary saturated aqueous solutions", J. of Research of the National Bureau of Standards- A. Physics and Chemistry Vol. 81A, 89-96, 1977

Leygrafer C., and T. E. Graedel, "Atmospheric Corrosion", Chapter, 2, Wiley Interscience, New York, 2000,

Pabalan et al., 2001, MRS Paper

Weast, R.C. and Astle, M.J., eds. 1981. CRC Handbook of Chemistry and Physics: A Ready Reference Book of Chemical and Physical Data. 62nd Edition. Boca Raton, Florida: CRC Press.

J. Yang
2/3/02



J. Lyons
7/15/02

DRH and Corrosion Test.

Corrosion test: Sensors ^{chan. 1-8} ① PTFE coated 1010 CS, 8 elect.
 ② PTFE coated SS304, 8 elect. ^{chan. 9-16.}

Sensors placed upward, filtration paper soaked with NaCl soln (saturated) were placed on top of sensors. (Next page) lot # 006924

Sensor 1010 CS connected to ch. 1-8; SS304 to ch. 9-16 of DRH - conductivity test. Multi-cell instrument

Cell I - NaCl + KCl ^{lot # 006924} paper ^{lot # 005573} 1.5 cm wider than base
 Cell II - NaCl ^{lot # 006924} paper 1.0 mm wider than base

(DRH = 74.43% - Greenspan)

8/20/01

13:43 Z plot file started, 0820a on Cell I

13:45 $(1/T_a/T_w) = 19/21$ $Z_I = 10^{21} \Omega$

13:51 18/21 to Cell II.

13:56 18²/21² Sensor program started, 0820a

Raise temp. 1 min interval.

14:37 40²/51² $Z_{II} = 310 \Omega$ to Cell I.

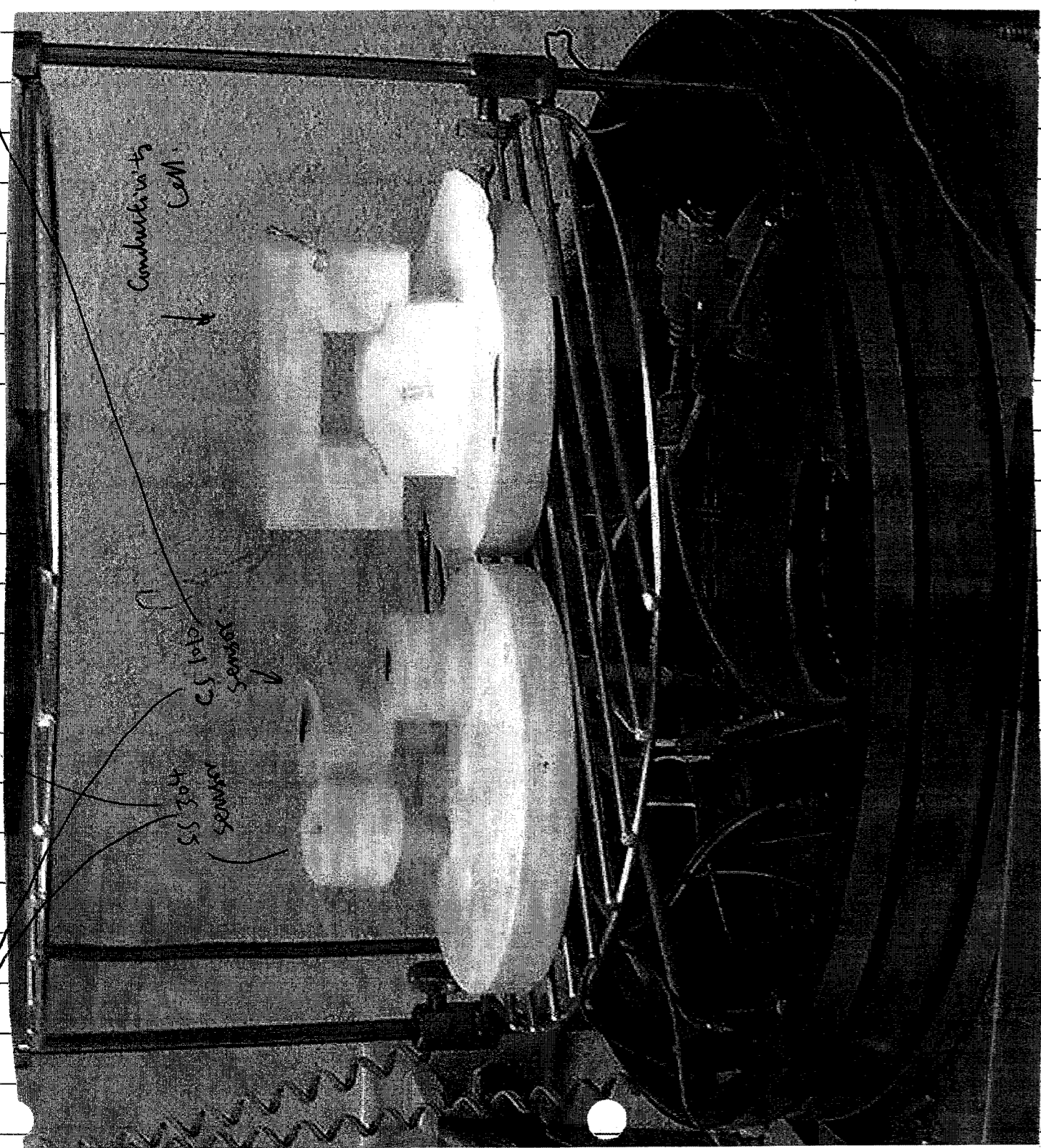
14:52 40²/51².

15:02 change humidity to Cell I

J. Lyons
4/13/02

EDS Analysis see page 151 30pk #497. (Do not have enough material to do solun. Analysis) J.Y. 07/20/03

Cont. Analysis see page 4, Book 522



made on 6/19/00

see page 34 BWA 424 J.Y. 05/01/03

J.Y. 09/09/02

15:14 41/50²
 15:33 41/50. H. chamber opened, put put move Hall soln. sat'd at 22².
 15:37. H. chamber closed.
 15:39 40/50².
 15:30 41/50²
 17:16 41/50² (53%) to cell I.
 17:25 restart sensor program. 0820b.txt.
 17:30 41/50² to cell II.
 8/21/01
 7:58 41/50². Changed Humidity. to cell I
 8:38 40/50².
 8:
 13:
 11:59 39/50 to cell II
 15:17 39/50 to cell I
 17:13 39/50 to II at 17:14
 8/22/01
 8:00 Sensor program restarted 0822a to cell I
 Filter: 40; dt = 2 min, settle ch.: 17 J.Y. 8/22/01
 8:10 39/50 changed Humidity (~~38/50~~ 22
 C 38/50², Tanker = 22², Knob: 1 turn)
 12:39 37/50 (44%)
 8/23/01
 7:54 Z_{II} = 10^{6.2} v. to cell II.
 J.Y. 4/6/02

8:05 37/50², Z_I = 10^{5.5} change RH.
(36/50, 17², 1 turn)

12:51 35/50 to cell II

8/24/01 8
8:14 Z_{II} = 10^{6.5} to cell I

8:21 35/50 change RH
(15/50.5, 0², full turns)

12:37 26/50² (14%) to (36/50, 17², 1 turn) to cell II

16:06 Z_{II} = 10⁶ 33 d.g. 8/24/01 to cell I

17:17 34/51 to (38/50, 22², 1 turn).

8/25/01
9:24 35/50 to (38/50, 22², 1 turn).

10:49 37/50

11:43 37/50

08/27/01 (Monday).

8:04 37/50 to (39.5/50, 27.5², 1 turn) to cell II

8:21

12:36 39/50 to (41/50, 32.5², 1 turn).

18:14 41/50 (88%) to cell I

8/28/01

8:04 41/50 to (42.5/50, 32.5, 1 turn) to cell II

12:29 #5179 41.5/50².

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4/3/02

8/29/01
8:14 41.5/50 to cell I
14:24 42/50 to (43.6/50, 32.5², 1/2 turn) to cell II

14:26 43/51
17:17 42/50

8/30/01
8:07 42/50 (64%) to cell I

12:30 to cell II

17:15 42/50 to (43.7/50, 32.5², 1/4 turn)

17:19 44/50 (72%)

8/31/01

8:19 43/50 Sensor program stopped.

8:28 44/50 to cell I

8:36 44/50

13:19 44/44.5² to cell II

17:06 #7461 44/50 to cell I

17:12 to cell II

9/4/01 after Labour Day.

8:06 #9993, Sensor program stopped at 5:39 a.m. 9/2/01.

8:12 43/50, Sensor prog started, to cell I.
0904A

8:14 to cell II

13:32 43/50 to (43.6/50, 32.5, 1/4 turn).

J. Young
4/3/02

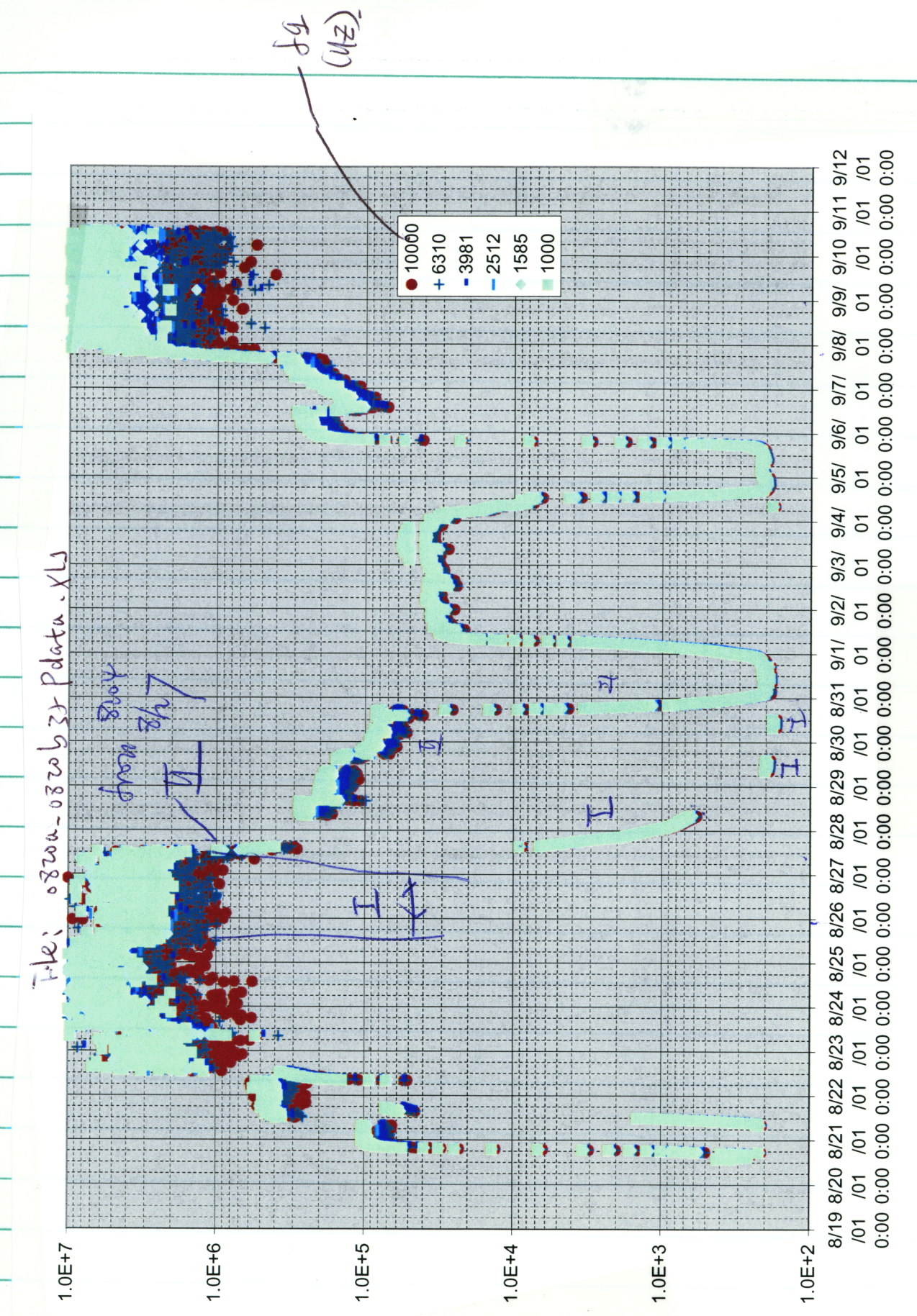
9/5/01
 8:16 48/50 (76%) to cell I
 8:18 to cell II
 17:20 43/50

9/6/01
 11:29 41/51 (56%) to (43, 32.4, 1/2 turn).
 17:18 42/51

9/9/01
 13:09 Reservoir dry. 9/9/01
 Water in humidifier gone.
 paper on SI sensor blown out.

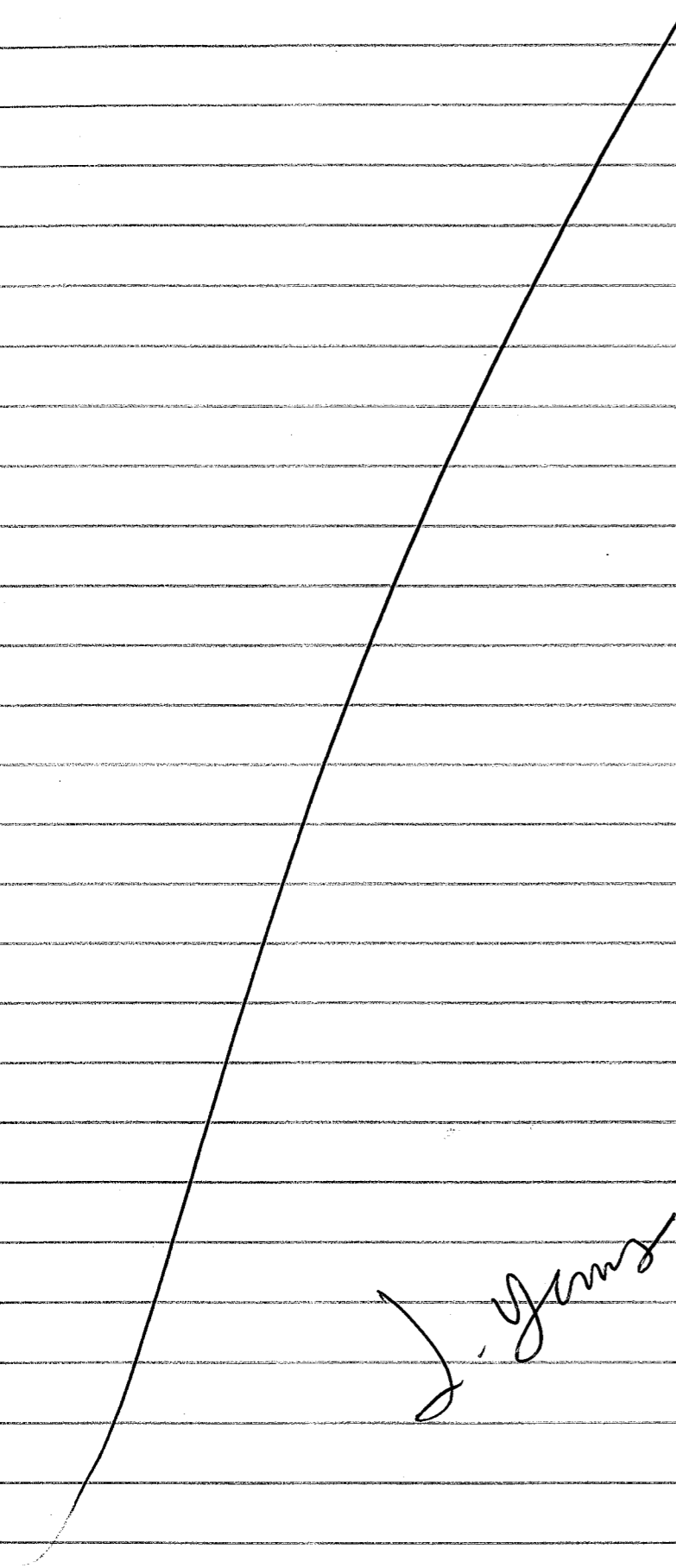
13:11 30/51
 13:13 30/50. test terminated.

J. Young
 4/3/02



J. Young
 4/3/02

Smith
20/11/02

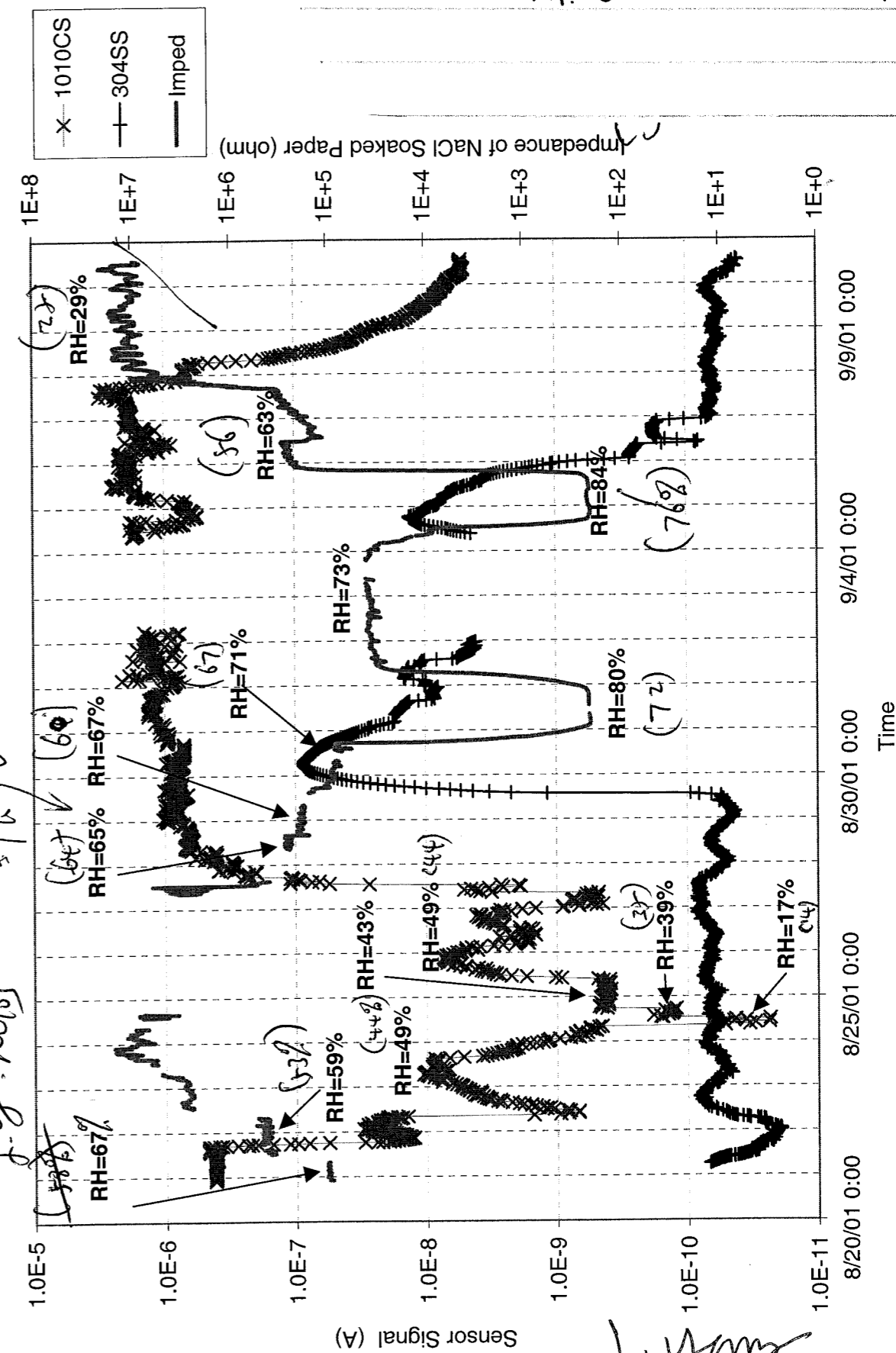


0820_1314_0910_Sensor_extracted_CS_SS.xls

Chart1

J.Y. 8/26/03
7/6 See Next Page.

J.Y. 8/6/03



Smith
20/3/02

RH correction

0820a_0820b35PData.xls

directly measured value

Correction 9% at about 84% (Notebook 70) and 4% at about 30% (Notebook 12)

	Factor	Corrected	RH=	Factor
58	6.592593	64.59259	84	9
53	6.12963	59.12963	30	4
44	5.296296	49.2963		
35	4.462963	39.46296		
14	2.518519	16.51852		
38	4.740741	42.74074		
44	5.296296	49.2963		
58	6.592593	64.59259		
60	6.777778	66.77778		
64	7.148148	71.14815		
72	7.888889	79.88889		
66	7.333333	73.33333		
76	8.259259	84.25926		
56	6.407407	62.40741		
25	3.537037	28.53704		

DH of NaCl=74%, according to Greenspan

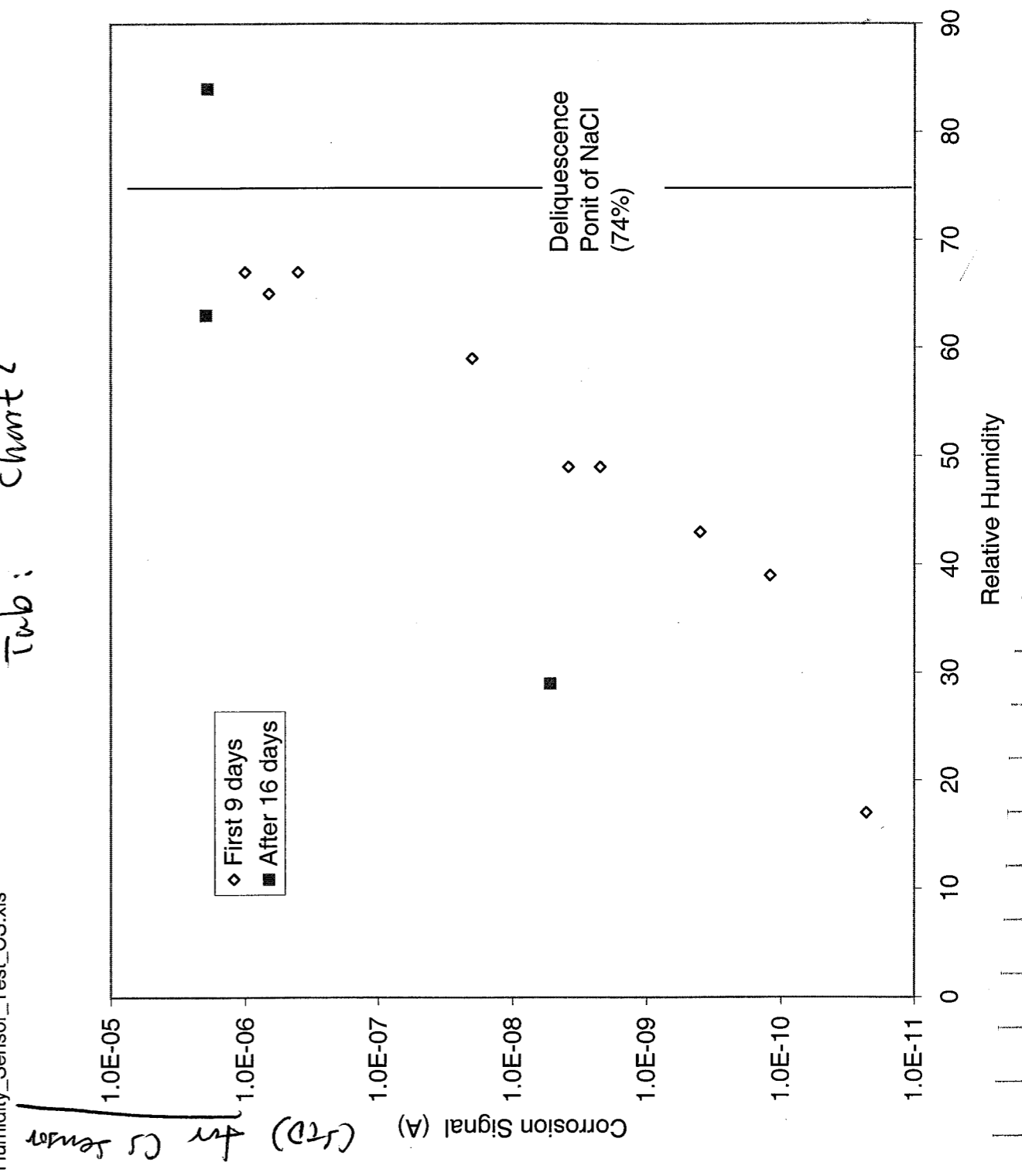
J. Young

4/3/02

Corrections

Humidity_Sensor_Test_CS.xls

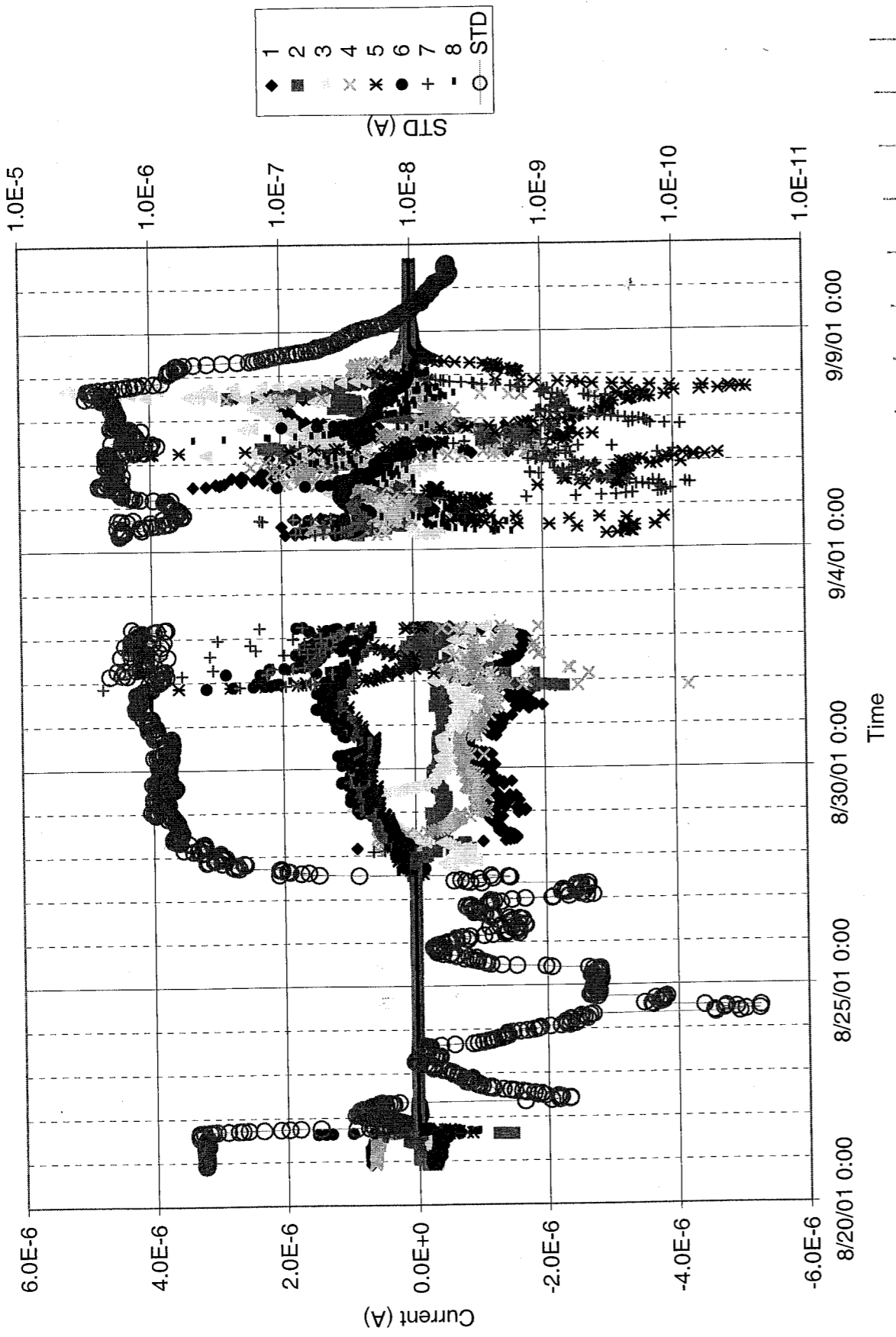
Tub: Chart 2



J. Young

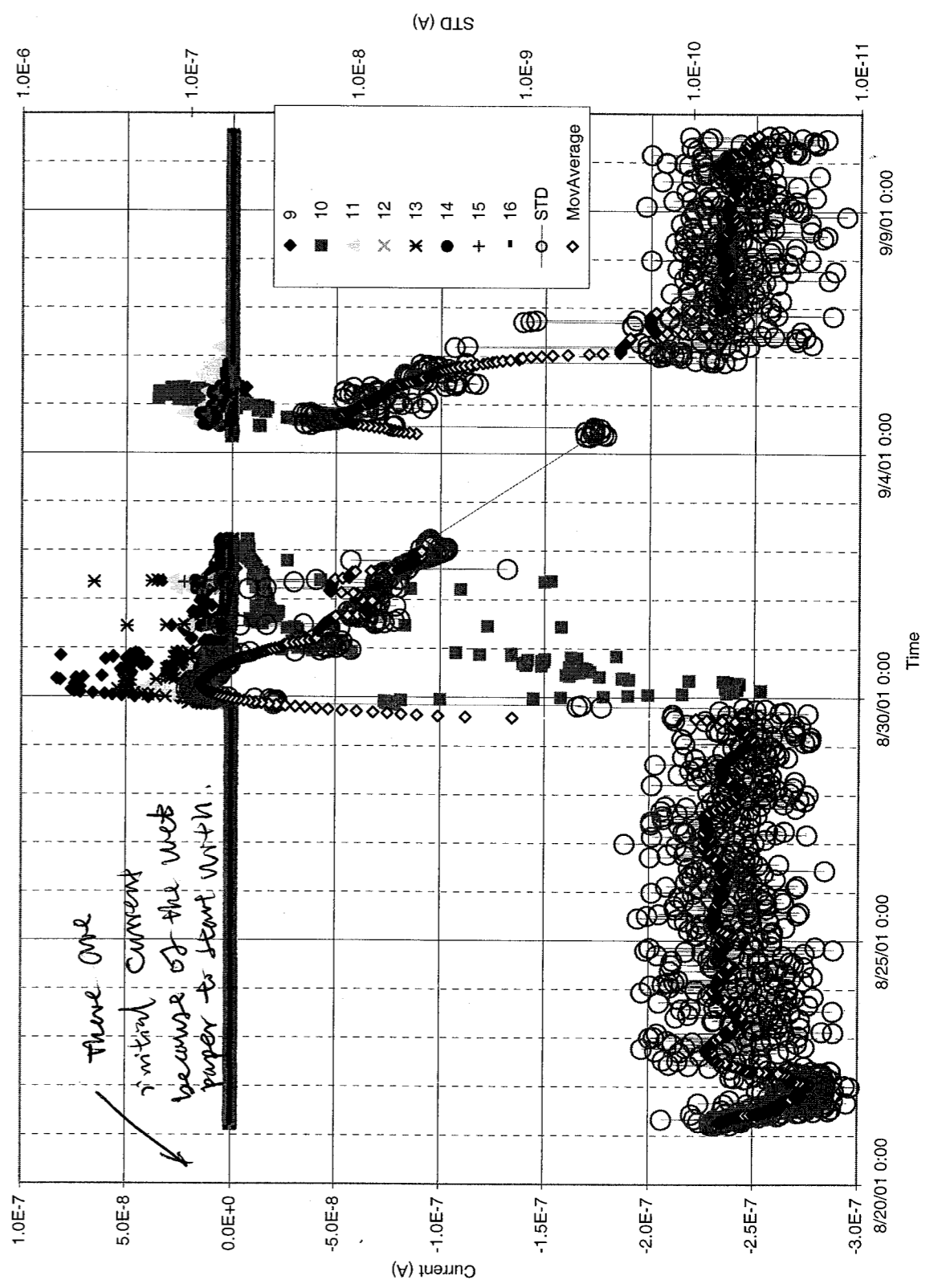
4/3/02

Carbon steel sensor.



20/3/02
L. Young

S.S. sensor



20/3/02
L. Young

6/20/02

Corrosion of Carbon steel and S. stainless steel materials under salt deposits in Humidity controlled environment.

Test #1 using KCl

Sensors Carbon steel: D-CS1010-2, 16 working Electrodes
S. steel: D-SS316-1, 16 working electrodes
See note book #497 For details (pages 110, 111)
Confirmatory Analysis of CS1010 and SS316
see page 153, Book 497 For SS316
page 151, Book 497 For CS1010 (EDS only)

14:00 KCl salts, lot # 005173.
dried in oven at 100°C, for 2 hours
Humidity chamber set to 50% RH and 50°C.

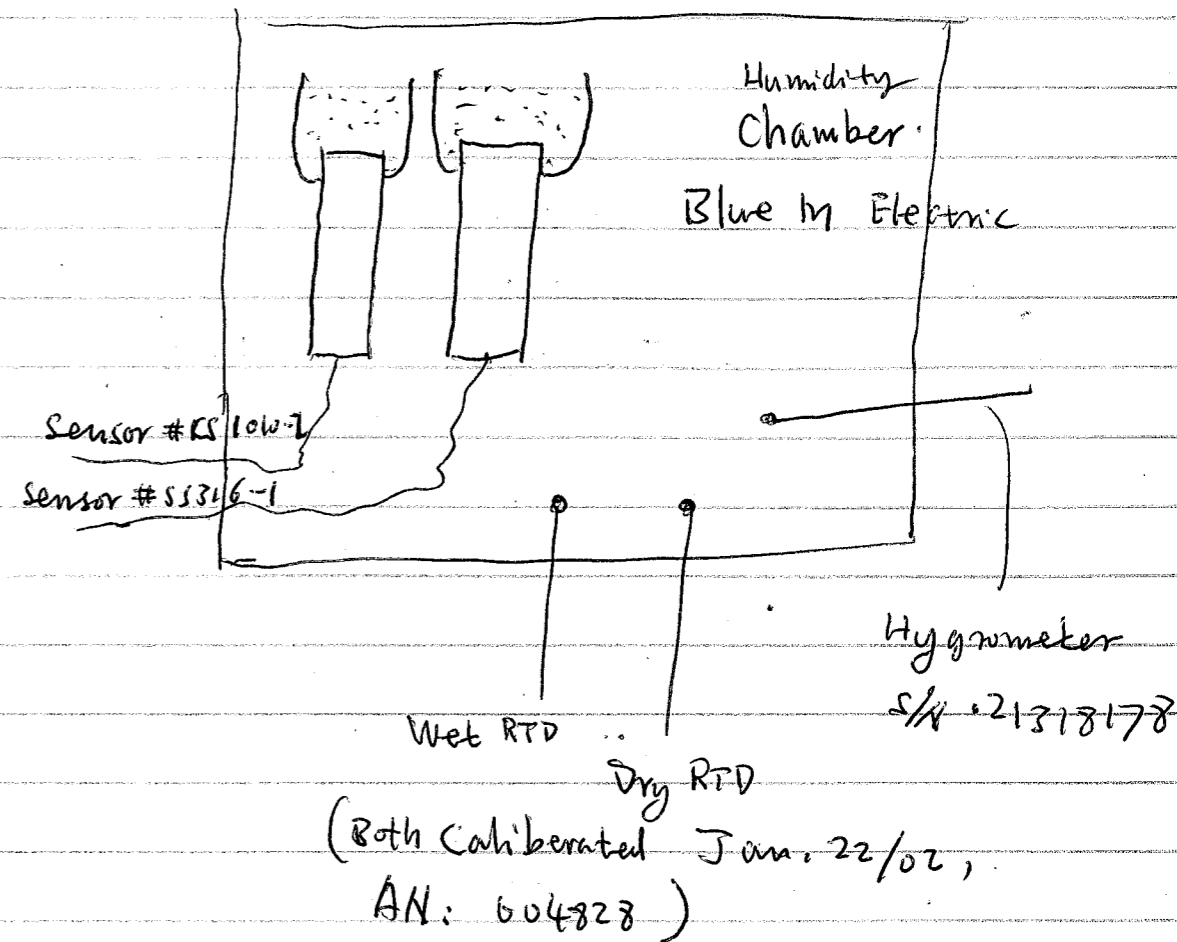
16:25 J. Yarns 6/21/02
RH(w) =
RH(meter) = 38.5%

16:30 dried salt transferred to two of sensors
see page 173 for set up

16:45 Td = 49°C, Tw = 33°C, RH(meter) = 39.07%

6/21/02 7:49 Wet water gone refilled.
RH(meter) = 36.72%

8:10 Td = 49°C, Tw = 34°C, RH(meter) = 37.6%, T(meter) = 47.41°C
J. Yarns 7/12/02



8:25, Dpoint = 1330, Connectors #1-4 to Sensors in Humidity Chamber.

Connections:

connector: #1 #2 #3 #4
Sensor: D-SS316-1(S) D-CS1010-2(S) D-CS1010-2(L)
D-SS316-1(L)

Sensor program has been used for crevice test.
Data program name: 02-0618a.

12:21 Tw/Td = 34°C/49°C, RH(meter) = 37.32, T(meter) = 47.32°C.
P channel #1-32 = -6.8 -3.1 mV Except #23
Found return of Ch. 23 broken.
Meter connectors returned to the crevice sensor
J. Yarns 7/12/02

13:41. New connection.

Connectors	II-2	II-3	II-4	II-5
Resistors	100k-1	100k-2	100k-3	100k-4
Sensor cable	D-SS316-1(S)	D-SS316-1(L) ^{2.9.} 6/21/02	D-CS1010-2(S)	D-CS1010-2(L)

14:27 sensor program started. 02-0621a. Results see pages 183-184 2.9. 7/10/02

16:47 $T_d/T_w = 49^{\circ} / 33.5^{\circ}$, RH (meter) = 37.76%, T (meter) = 47.30°C
Set $(T_d/T_w) = 52 / 38.5^{\circ}$, cooler = 14°C.

6/24/02 $T_d/T_w = 51 / 35^{\circ} \Rightarrow [35\% RH]$ RH (m) = 34.79%, T (m) = 48.64°C

8:25 Signal of sensors: -2 ~ -8.5 μV . with 100k \sim Resist.

8:32. Set to 65% RH. as Target. ($51^{\circ} / 42^{\circ} = T_d/T_w$) cooler = 14°C.

9:03 still $T_d/T_w = 51 / 35$ change cooler = 25°C.

9:05 $T_d/T_w = 51 / 38$ change cooler = 30°C

9:15 $T_d/T_w = 51 / 40$ change cooler = 34°C

9:25 $T_d/T_w = 51 / 41$
10:36 $T_d/T_w = 51 / 41$ RH (m) = 57.2%, T (m) = 48.86°C
Sensor signals -2.6 ~ -9.7 μV .

$T_d/T_w = 51 / 42$ RH (m) = 56.64% T (m) = 48.82
2:26 sensor signal: -2.0 ~ -20.2 μV (56.57 - 59.35%)

J. Yang
7/15/02

6/25/02

9:35. $T_d/T_w = 51 / 41$. RH (m) = 59.86 T (m) = 48.82°C
 $\rightarrow 56\% RH$
18.6 ~ 18 μV sensor signal.

6/26/02

8:43 $T_d/T_w = 51 / 42 \sim 52 / 42$ RH (m) = $\frac{62.2\%}{58.05} T (m) = 48.83^{\circ} C$
sensor signal SS: -0.05 ~ -15.87 μV !
CS: +790 ~ -870 μV !
Comoding!

8:53 sensor signal SS: +4 ~ -6 μV
CS: +913 ~ -890 μV

9:06 Set $T_d/T_w = 51 / 33.5^{\circ}$. Target ($51 / 33.5^{\circ}$) (37.7% RH, JPL 6-26-02)

9:16 $T_d/T_w = 51 / 38$ change cooler = 30°C

9:22 $T_d/T_w = 51 / 38$ change cooler = 20°C

9:41 $T_d/T_w = 51 / 34$ change cooler = 19°C

9:55 $T_d/T_w = 51 / 34$ ~~change cooler~~ change cooler = 18°C (JPL 6-26-02)

10:07 $T_d/T_w = 51 / 33$ 34.5% RH

13:27 $T_d/T_w = 51 / 33$ 31.6% RH
Sensor signal SS: -5 ~ -3 μV
CS: +1.6 ~ -10 μV

J. Yang
7/15/02

14:00 Sensors were removed and discoloration due to corrosion of carbon steel was noted on the carbon steel sensor. The salt directly above the carbon steel sensor was also discolored.
 No discoloration noted on stainless steel sensor.

14:16 Sensors were polished ^{John Sady} and ⁶⁻²⁶⁻⁰² at 1000-grit and placed back on humidity chamber

14:54 Sensors were connected same as pg 174.
 15:20. Program interrupted; and restarted. Results see pages 181-186

new file: 02-0626a. Chamber close to dry the system.
 4:45 pm. $T_d/T_w = 51/33^{\circ}$, $RH(M) = 36.78$.

~~Chamber opened J.Y. 6/27/02~~
 Signals: -13 - -0.8 μV (SS)
 -4 - +22 μV (CS) J.Y. 6/27/02
 6/27/02 Chamber closed, after sensors repolished to 1000 grit. ~~Pre dry it before salt put in.~~
 8:27 $T_d/T_w = 51/33^{\circ}$, $RH(M) = 37.41$

8:41 signals - 5.6 - -0.59 μV (SS)
 - 4.8 - -0.6 μV (CS).
 $T_d/T_w = 51/33^{\circ}$, $RH(M) = 37\%$.

8:52. Kell Salt (dried over night at 50 $^{\circ}$, lot # 00573) placed on top of sensor, with a beaker over
 8:53 Set $T_d/T_w = 51/33^{\circ}$. the sensors and the hygrometer.

J. Yano 7/15/02

8:56 $T_d/T_w = 50/33$ $RH(M) = 56.9\%$, $T(M) = 36.7^{\circ}$

9:03 $T_d/T_w = 51/33$ $RH(M) = 42.2\%$ $T(M) = 45.3^{\circ}$

9:16 $T_d/T_w = 51/34$ $RH(M) = 38.2\%$ $T(M) = 47.2^{\circ}$

10:19 $T_d/T_w = 51/33$ $RH(M) = 35.9\%$ $T(M) = 47.69^{\circ}$.

11:27, Data# 247. Connector II-3 disconnected for size test.

13:43 $T_d/T_w = 51/3$ $RH(M) 35.4\%$ $T(M) = 47.7^{\circ}$
 Range: 35.48% - 36.16%

13:54 Connector II-3 Returned
 Connector II-5 Disconnected
 Data pt. 277

6-28-02

9:27 Connector II-5 Returned JPL 6-28-02
~~Connector II-3 Disconnected~~
 $T_d/T_w = 51/34$ $RH(M) = 36.8\%$ $T(M) = 47.75^{\circ}$

9:30
 Signals Data pt. 510
 SS: -6.546 - -3.435 μV
 CS: -7.248 - -3.867 μV

9:32 Connector II-3 Disconnected

13:00 Data 554 $T_d/T_w = 51/34$ $RH(M) = 36.14\%$ $T(M) = 47.77^{\circ}$
 SS -6.029 - -4.5 μV only one connected
 CS -6.89 - -3.661 μV

J. Yano 7/15/02

6/28/02

16:42 Data #602

Connector II-5 disconnected for size effect test
connector II-3 returned

$T_d/T_w = 51/34$, $RH(m) = 35.34\%$, $T(m) = 47.79^\circ C$
 $\hookrightarrow 34\% RH$ $\hookrightarrow 36.00\%$ $\hookrightarrow 47.80^\circ C$

17:26 Set: $52/38.5^\circ C$, cooler = $34^\circ C$.

17:27 set: $52/38.5^\circ C$, set cooler: $34^\circ C$.

17:38 $T_d/T_w = 51/39^\circ C$, $RH(m) = 50.4\%$, $T(m) = 47.93^\circ C$.

Connector II-5 returned.

6/31/02

JPL 6-31-02

7/2/02 J.Y.

7:23

SS: $-9.859 - -3.416 \mu V$

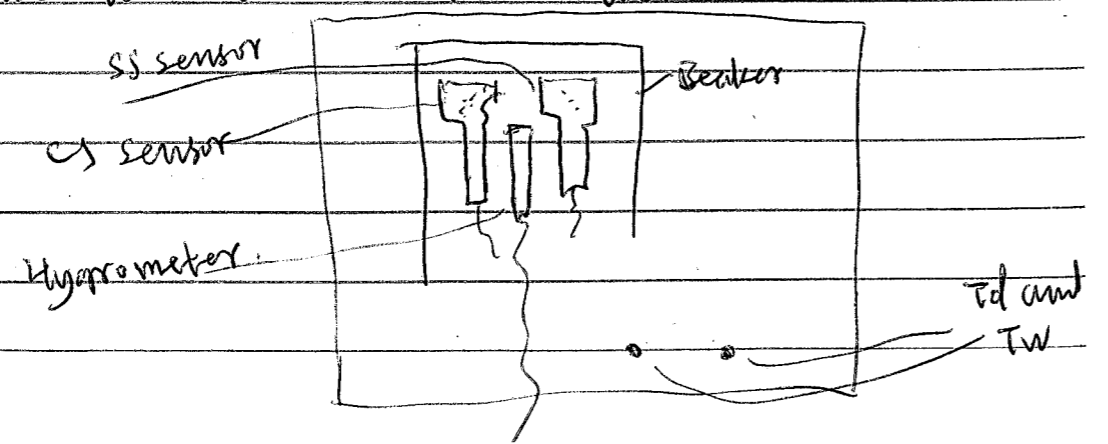
CS: $-2642136.03 - -3.934 \mu V$

J.Y. 7/10/02
this level of signal
one only seen on connector
II-5, not II-4
must be due to bad
connection of return.

$T_d/T_w = 51/40$, $RH(m) = 61.31\%$, $T(m) = 47.73^\circ C$

9:11 $T_d/T_w = 51/40$, $RH(m) = 60.70 - 62.36\%$, $T(m) = 47.77^\circ C$
 $\hookrightarrow 53\% RH$ $\hookrightarrow 47.78^\circ C$

The RH from T_d/T_w is much less than $RH(m)$.
probably due to the location difference.



J. Yarns 7/15/02

The temp. of $T(m)$ is only $47.7^\circ C$.
The temp of T_d is $51^\circ C$.

The $RH(T_d/T_w)$ was closer to $RH(m)$ previously
(see page 175). That time the $T(m)$ was $48.8^\circ C$, because
it was closer to the bottom.

9:29 II-5 Removed

10:32 signals from II-4.
-3 - $-7.3 \mu V$

13:10 Data 1357

SS: $-11.632 - -3.776 \mu V$

CS (II-4) only: $-8.809 - -5.973 \mu V$

14:19 II-5 Returned

16:11 II-5 Removed

17:23 $T_d/T_w = 51/40$, $RH(m) = 61.16\%$, $T(m) = 47.77^\circ C$

Data 1407 SS: $-7.774 - -4.163 \mu V$

CS (II-4 only): $-6.497 - ~~-5.747~~ -5.747 \mu V$
JPL 7-1-02

J. Yarns
7/15/02

7/2/02 7131

$T_d/T_w = 51/39$ $T(m) = 47.72^\circ C$ $RH(m) = 51.65\%$

~~stir Magnet was not stirring~~ ^{JPL} 7-8-02

Data 1582

SS: -13.139 - -3.891 mV

CS: (II-4 only) -7.124 - -5.594 mV

8:37 II-5 Returned

8:46 II-3 Removed

9:24 $T_d/T_w = 51/40$ $T(m) = 47.77^\circ C$ $RH(m) = 61.77 - 63.27\%$

SS: (II-2 only) -4.91 - -3.731 mV

CS: -10.51 - -3.045 mV

12:07 $T_d/T_w = 51/40$

$RH(m) = 62.52 - 63.61\%$

$T(m) = 47.82^\circ C$

Data 1639 SS: (II-2 only) -4.944 - -1.355 mV

CS: -5.532 - -3.769 mV

16:42 II-3 Returned

16:53 $T_d/T_w = 51/40$ $T(m) = 47.79^\circ C$ $RH(m) = 61.73 - 62.85\%$

Data 1695 SS: -7.506 - -3.773 mV

CS: -7.56 - -4.83 mV

Jyams
7/15/02

7/3/02

$T_d/T_w = 51/40$ $RH(m) = 61.63 - 62.94\%$ $T(m) = 47.7^\circ C$

Data 1868

~~SS: -7.142 - -1.623 mV~~ ^{JPL}

~~CS: -7.742 -~~ ⁷⁻³⁻⁰²

SS: -7.742 - -1.623 mV

CS: -6.203 - -3.504 mV

9:14 II-5 Removed

11:43 $T_d/T_w = 51/40$ $RH(m) = 62.03 - 63.31\%$ $T(m) = 47.78^\circ C$

SS ^{JPL 7-3-02} ~~-4.558 - +8.098 mV~~ -5.723 - -3.877 mV

CS (II-4 only) -7.472 - -4.368 mV

7/5/02

8:31 Data # 2452

$T_d/T_w =$ $RH(m) = 58.73 - 60.28$, $T(m) = 47.76 - 47.27^\circ C$

signals
SS (II-2, II-3): -3.25 - -5.70

CS (II-4 only): -4.20 - +0.47

8:51 Data # 2456 II-5 returned

10:51 signals SS -8.7 - -4.13 mV; CS: -4.2 - -9.8 mV
II-5 removed.

14:00 program restarted. 0705a

Jyams

7/15/02

Results see pages 187-188

7-8-02

8:19 Data 821

SS: -5.039 - -3.63 μ V

CS:(II-4 only) -4.913 - -3.553 μ V

$T_d/T_w = 51/35$ RH(m) 34.48 - 35.03% $T(m) = 47.89^\circ$ C

8:54 II-5 Returned to CS

9:13 Found reason for low RH — Water in reserve tank empty, — refilled.

9:14. $T_d/T_w = 51/40$

16:46 $T_d/T_w = 51/40$ RH(m) = 54.72 - 56.26% $T(m) = 47.97^\circ$ C

Data 910

SS -4.509 - +1.627 μ V

CS -5.562 - -3.484 μ V

7-9-02 $51/39$

8:10 $T_d/T_w = 51/40$ RH(m) 57.01 - 58.25% $T(m) = 47.83^\circ$ C

Data 1083

SS: -5.867 - -3.296 μ V

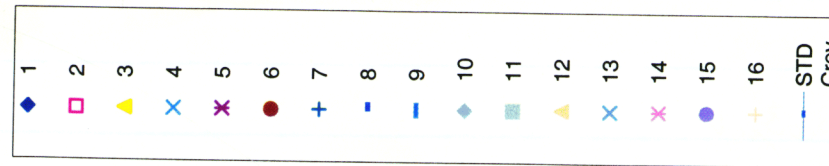
CS: -6.989 - -4.546 μ V

11:25 II-3 Removed

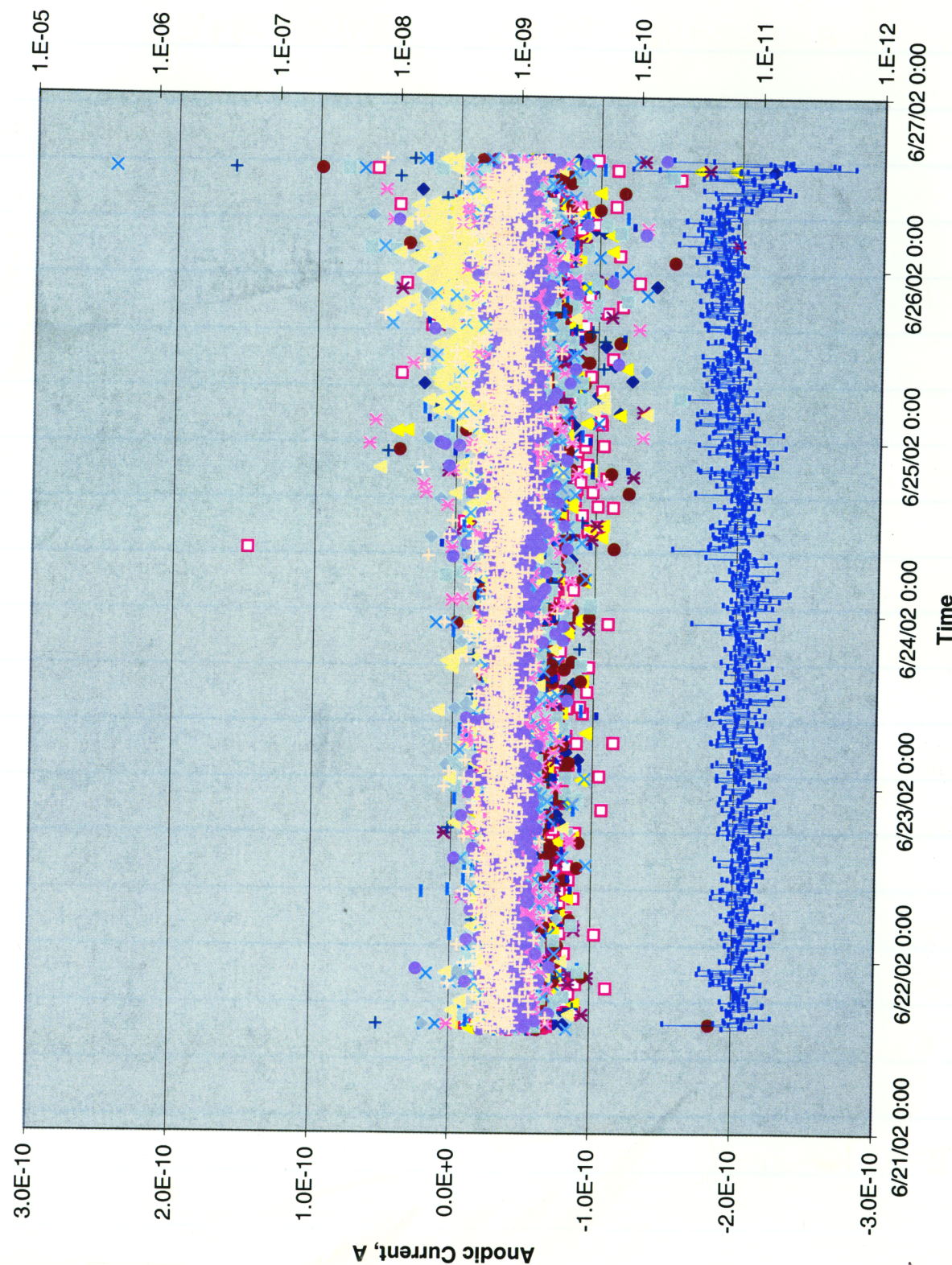
New Data File: 02-0709a see pages 191-192.

Handwritten signature
7/15/02

element



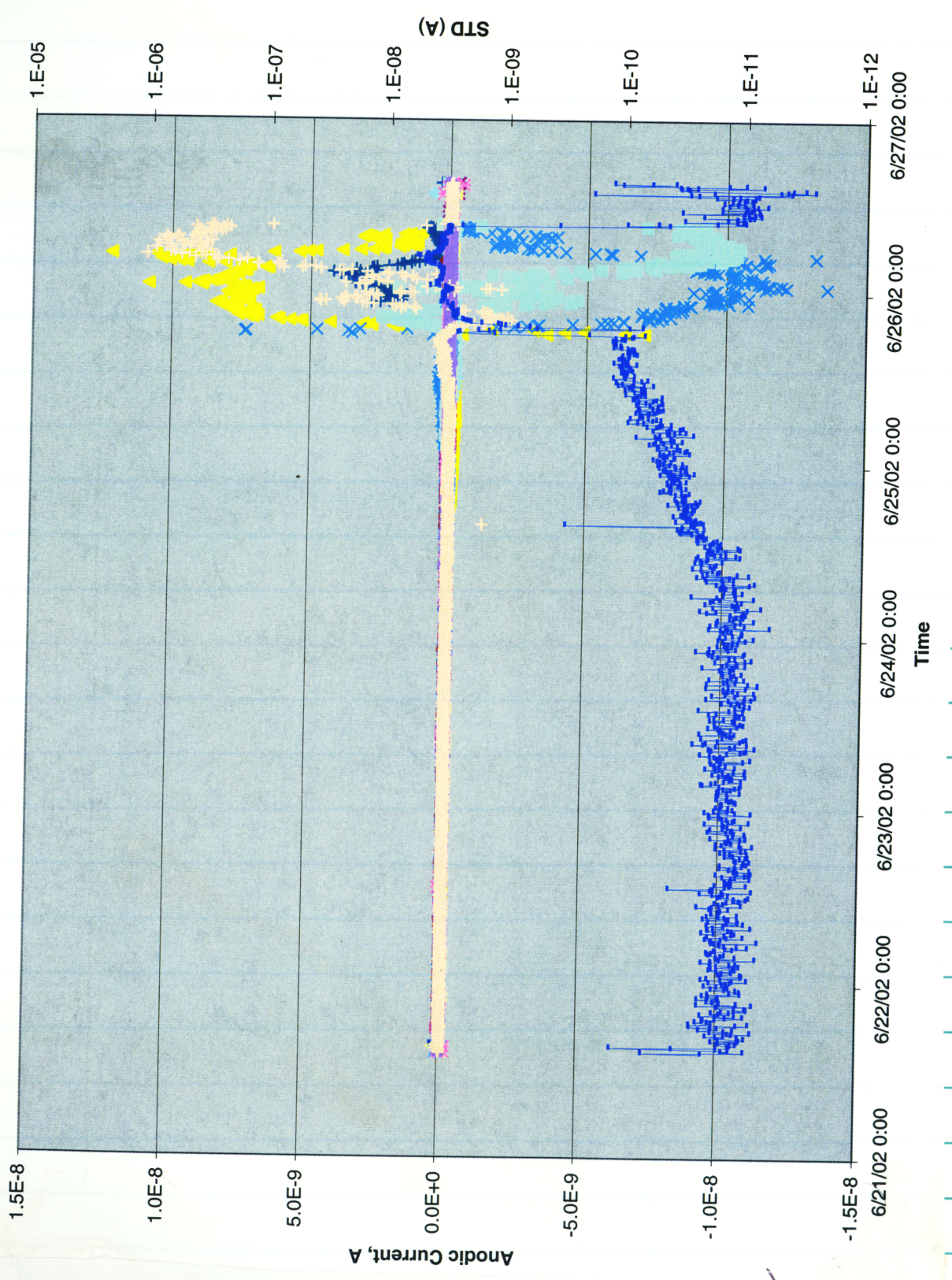
02_0621a00_02_Salt_Sensor.xls SS316_Chart



Handwritten signature
7/14/02

02_0621a00_02_Salt_Sensor.xls / Tab: CS_1010_Chart

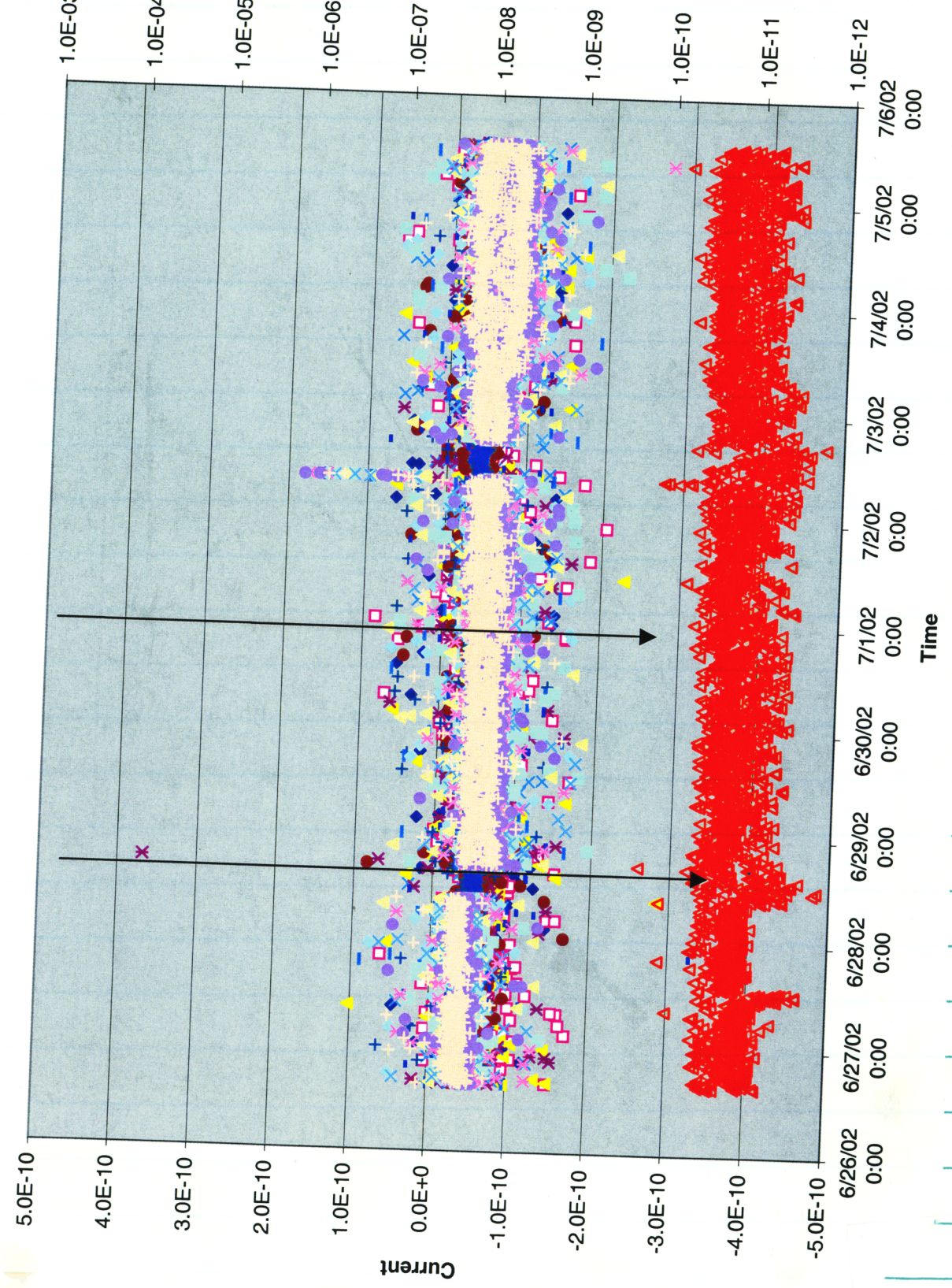
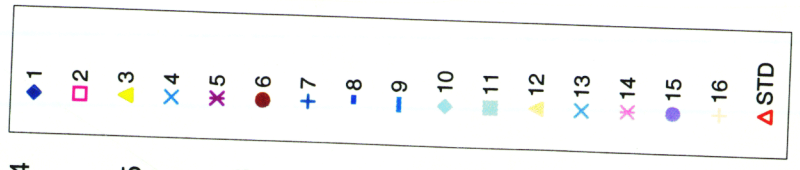
electrode



20/11/2 sample

02_0626a00_a25_salt.xls Chart_SS316-1

electrode

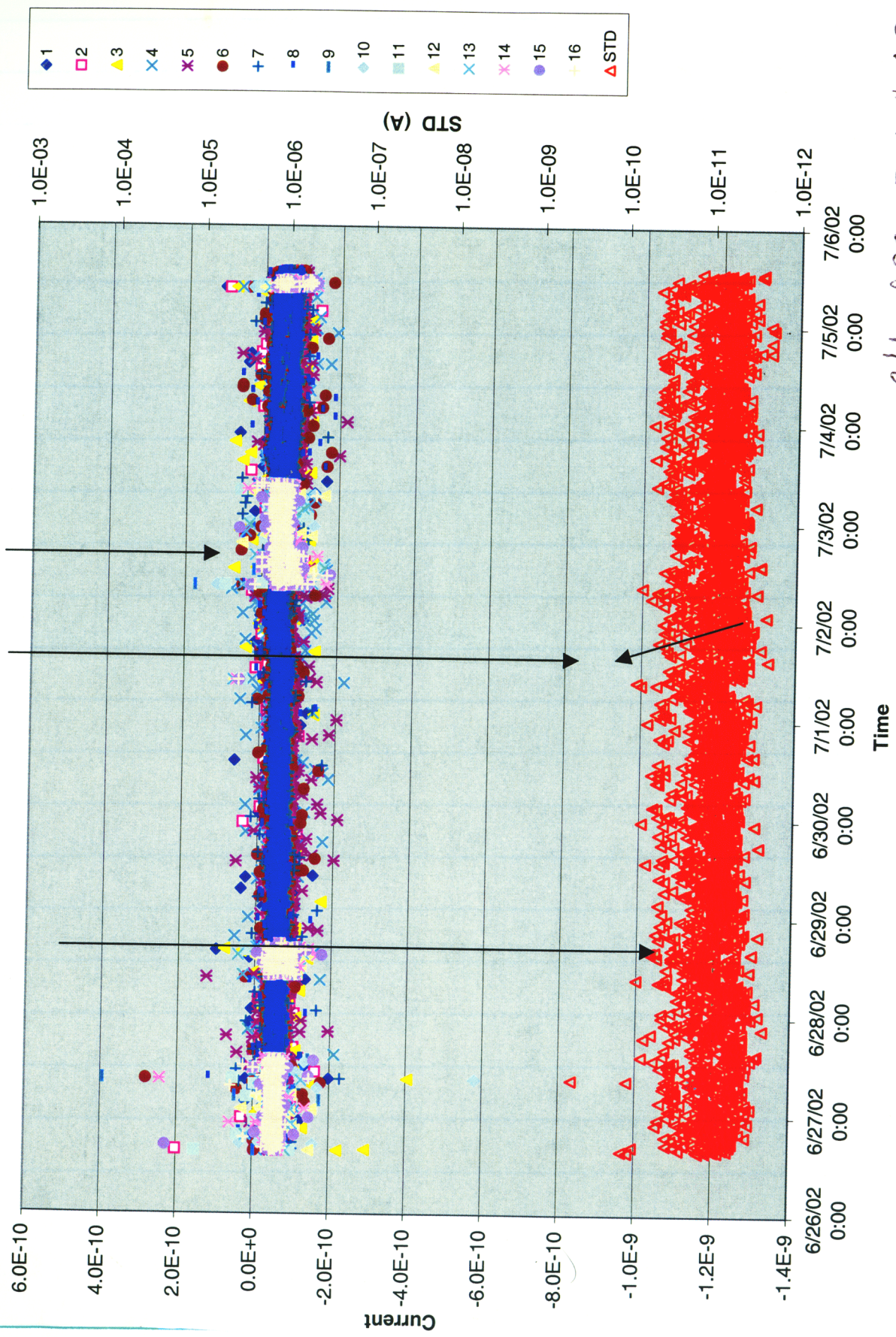


20/11/2 sample

show see page 34 of Book # 533 Fry. 8/6/02

02_0626a00_a25_salt.xls Chart_CS1010-1

electrode

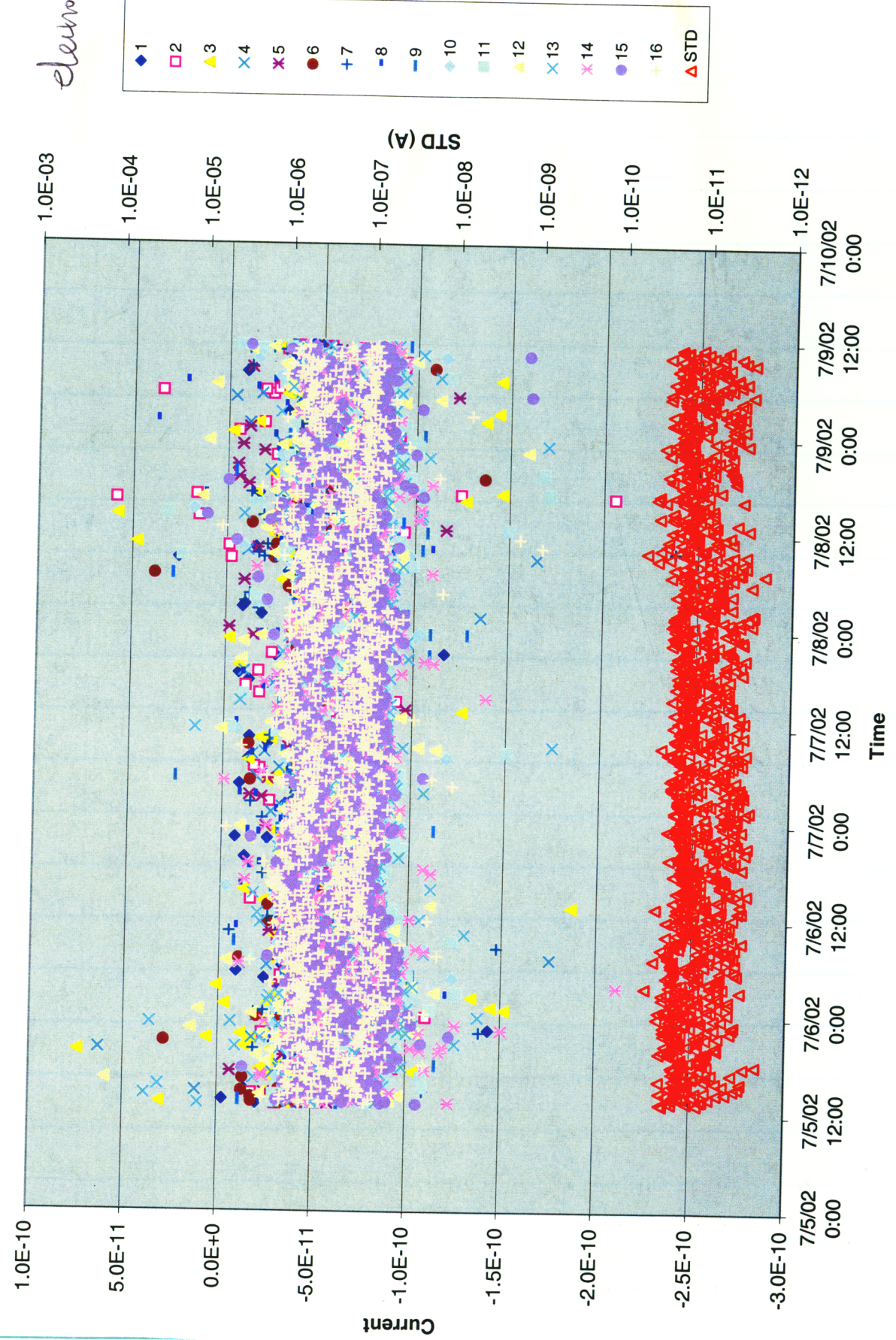


also see Book # 533
Page 39
Aug, 26/6/02

20/5/12 smh/f

02_0705a00_a02_salt.xls Chart_SS316-1

electrode

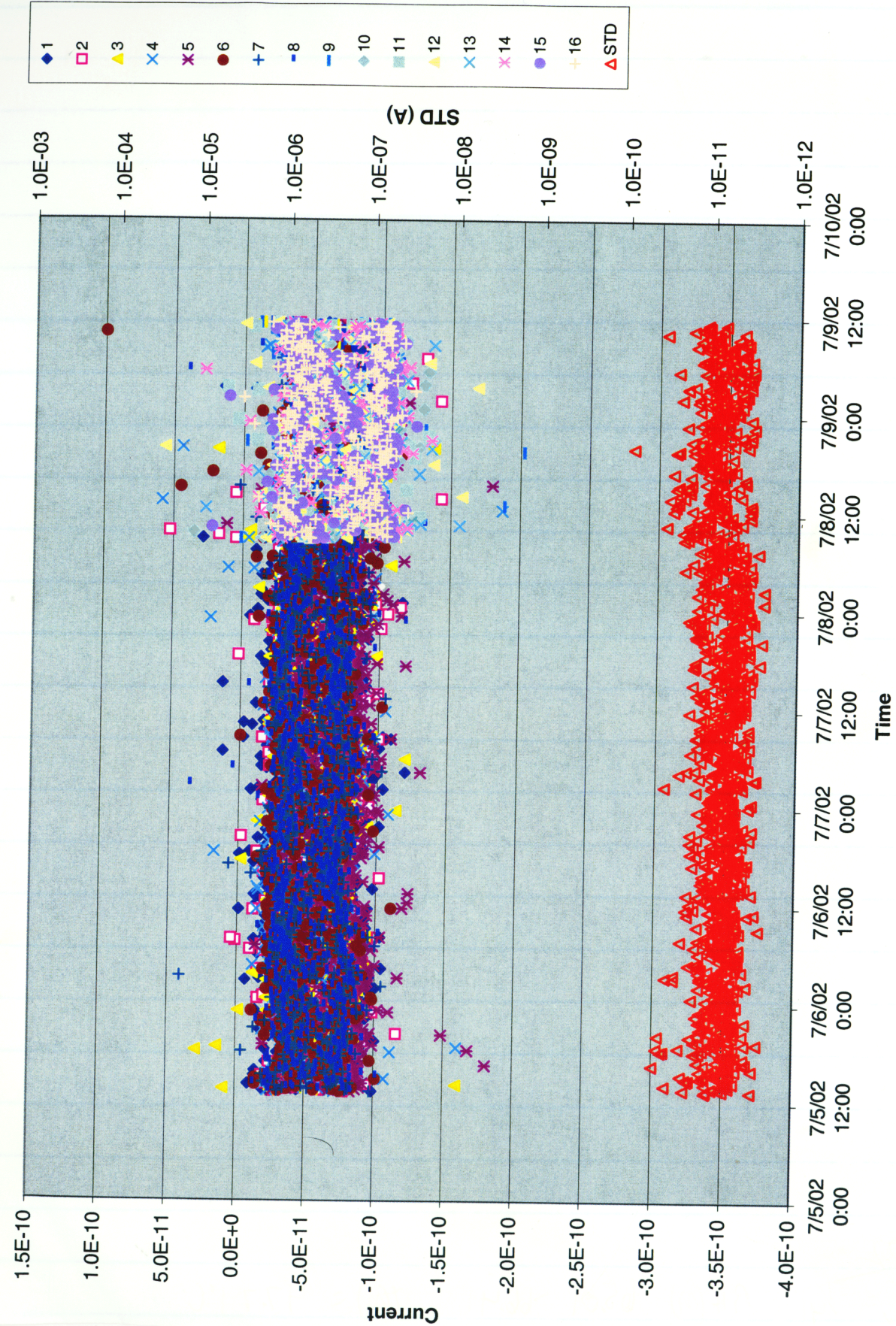


Also see page 39
of Book # 533
Aug, 26/6/02

20/5/12 smh/f

02_0705a00_a02_salt.xls Chart_CS1010-1

electrode



J. Yang
7/15/02

7-10-02

T_d/T_w JAL
7-10-02

13:44

$T_d/T_w = 51/40 \sim 39$, RH(m) = 58.30 - 59.87% $T(m) = 47.90^\circ C \pm 0.00$

13:46

Set Temp $T_d/T_w = 52/40$ Target Humidity 65%

15:11

~~$T_d/T_w = 52$~~ J.J. 7/10/02

$T_d/T_w = 51/40$, RH(m) = 59.91 - 61.26% $T(m) = 47.90$

15:14

Set Temp $T_d/T_w = 52/41$

15:12

Data 309, program terminated. II-3 returned.

15:22

program restarted. 02-0710a. see pages 191-192

15:34

$T_d/T_w = (51-52)/(40-41)$

Set $T_d/T_w = 51.5/41$

16:11

$T_d/T_w = (50-51)/40$ RH(m) = 61.98 - 63.19% $T(m) = 47.63^\circ C$

16:17

Set $T_d/T_w = 51.3/41.32$, II-3 removed.

7-11-02

8:05

$T_d/T_w = 51/41$ RH(m) = 64.78 - 65.87% $T(m) = 47.69^\circ C$

17:00

$T_d/T_w = 51/41$ RH(m) 65.18 - 66.64% $T(m) = 47.80^\circ C$

7-12-02

9:01 $T_d/T_w = 51/41$ RH(m) 65.01 - 66.41% $T(m) = 47.74^\circ C$

J. Yang
7/15/02

7/12
 17:32 program restarted. 02-07126 RH(m) = 65-66.5%
 7/15/02 program restarted 02-0715a

7/15/02
 8:52 $T_d/T_w = 51/41 - 51/42$ RH(m) = 68.23 - 69.54% T(m) = 47.79°C

$T_d/T_w = 51/41 - 51/42$ RH(m) = 68.32 - 69.56% T(m) = 47.80°C

SS (#2 only): -5.453 - -3.973 μV
 CS: -647.249 - +389.21 μV

18:10
 7/15/02 program restarted II-3 returned.
 7-9-7/16/02 02-0715b. Results see page 39, Book #533

7/16/02
 8:16 $T_d/T_w = 51/40 - 51/41$ RH(m) = 66.43 - 67.83% T(m) = 47.69°C

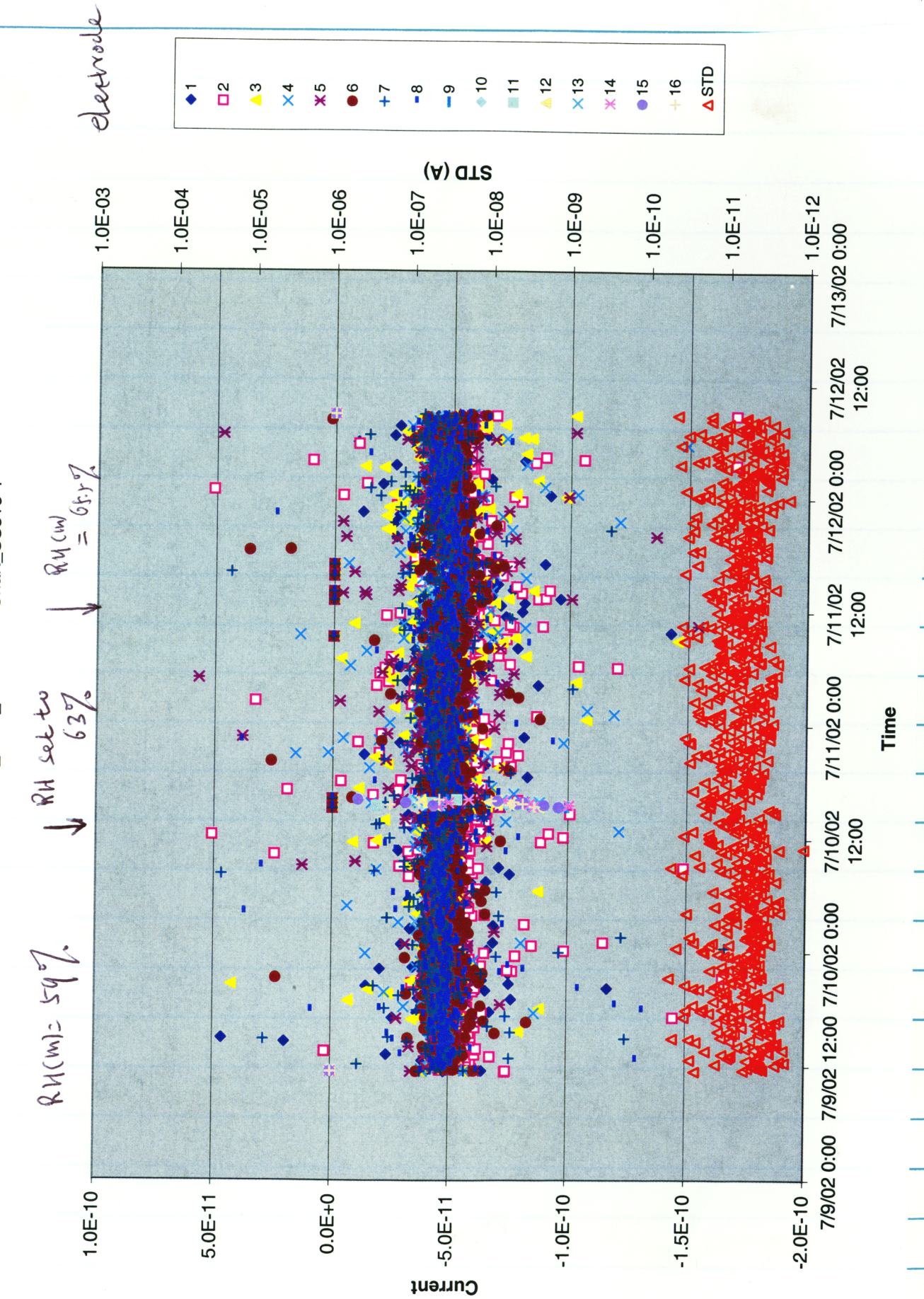
8:08 Data# 46
 SS: -6.166 - -2.86 μV
 CS: -464.005 - +268.734 μV

8:20 program change, new file: 02-0716a.
 8:20 set $T_d/T_w = 51.8/42$ Target RH = 70% Results see page 39, Book #533

9:12 II-3 Removed
 9:23 RH(m) = 68.70 - 90.07%
 9:24 set $T_d/T_w = 51.8/43$ SPL 7-16-02
 11:21 II-3 Returned - continued on page 197
 J. Young 7/16/02

Results see pages 145-146

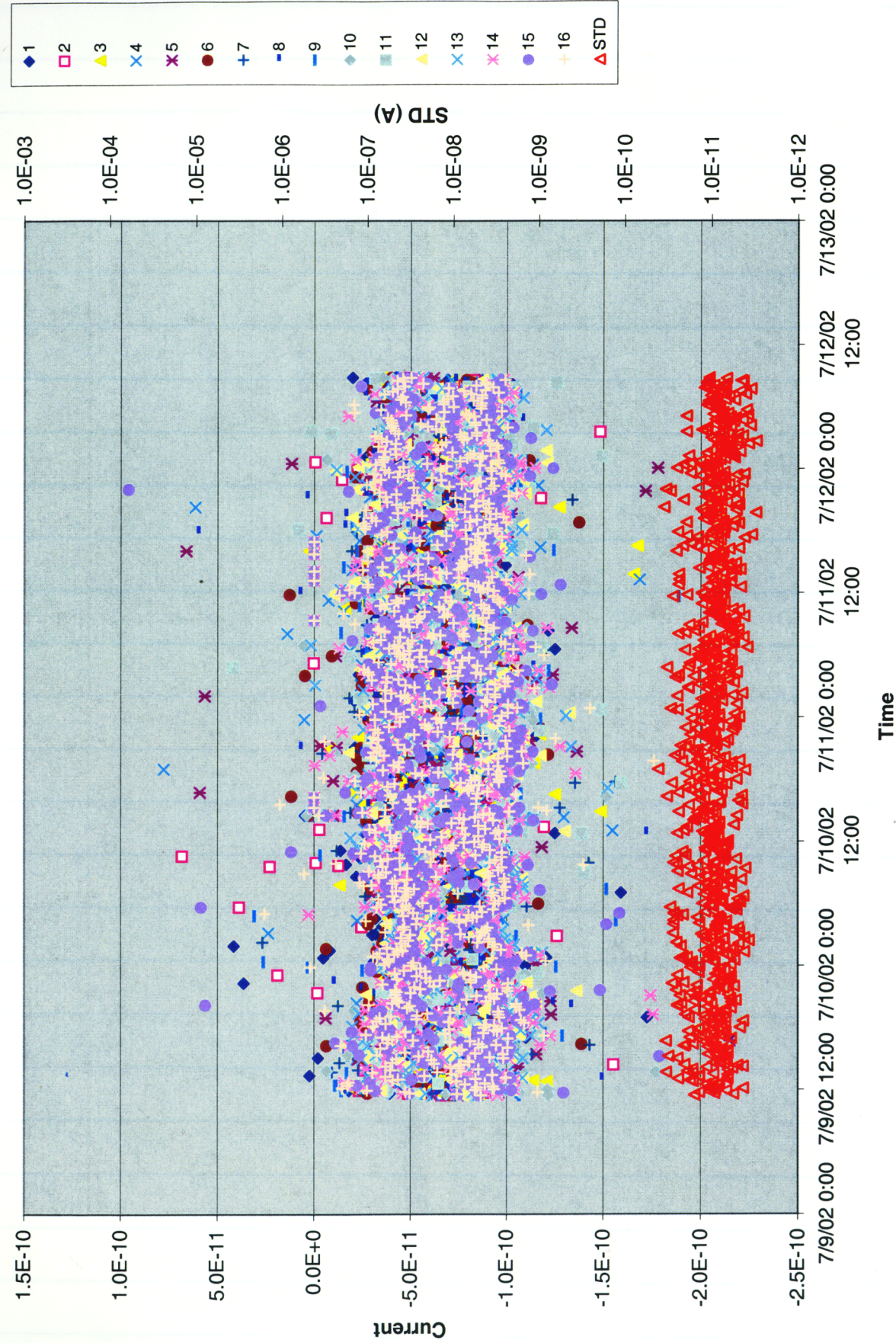
02_0709a00_10a0_salt.xls Chart_SS316-1



J. Young 7/18/02

electrode

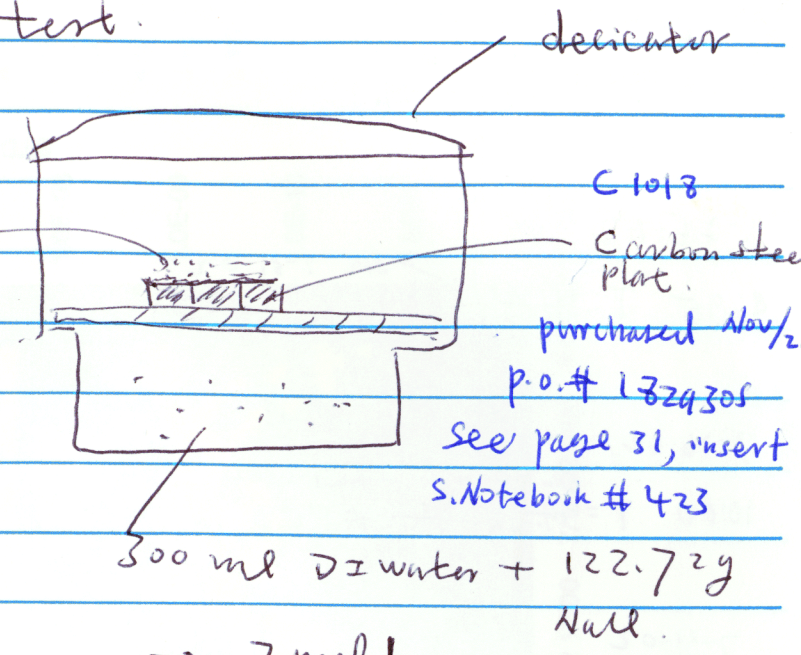
02_0709a00_10a0_salt.xls Chart_CS1010-1



2/24/02

07/15/02

Corrosion product test.



NaCl: lot # 020814

→ 7 molal
(solubility 6.1 → 6.25 molal @ 40°C)

Carbon steel polished to 320 grit.
washed with Acetone

11:14. Decicator placed in oven, 50°C.

17:00. T (indicator) = 51°C ; calibrated thermometer placed in oven.

7-16-02

1:50 T = 51.1°C

7-17-02

10:30 T = 49.8°C

15:07 T = 49.1°C

7-19-02

9:36 T = 49.1°C

7-22-02

8:51 T = 49.1°C

Justin Sandley 7-25-02