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UN TED STATES NUCLEAR REGULATORY COMMISSION

WASHINGTON D'C 2055-401

January 3, 1995

U.S. NUCLEAR REGULATORY COMMISSION
In the Matter of Louisiana Energy Services L.
Docket No. 10-3/03-M1 Official Exhibit No. 247
DOCKET NO. TO STOS PTE DINGUI EXHIBIT TO STORY
OFFERED by: Applicant/Licensee Intervenor NTRS/PC
ADD Chaff Other
IDENTIFIED on 10/03/05 Witness/Rane) Disposal.
Action Taken: (ADMITTED) REJECTED WITHDRAWN
Reporter/Clerk BITTONIL ENGLY :

Hr. Charles E. Bradley, Jr. Office of Uranium Programs Office of Nuclear Energy U.S. Department of Energy NE-33, Germantown Bldg Washington, DC 20585

Dear Hr. Bradley:

This is in response to the U.S. Department of Energy's (DOE's) recent request for recommendations on the potential uses for, and technologies that could facilitate the long-term management of, depleted uranium hexafluoride (DUF₆) stored at the Paducah, Kentucky, and Portsmouth, Ohio, gaseous diffusion plants and at the Oak Ridge Reservation in Tennessee (59 FR 56324, Hovember 10, 1994). The U.S. Huclear Regulatory Commission's interest in the disposition of DUF₆ at these facilities stems from our responsibilities under the Energy Policy Act of 1992 and the effect that the long-term management strategy(s) chosen by DOE for DUF₆ at DOE's Facilities may have on the management of DUF₆ at facilities under the Commission's jurisdiction, that is, the U.S. Enrichment Corporation plants and the proposed Louisiana Energy Services, Claiborne Enrichment Center.

NRC staff has not developed recommendations on the potential uses for DOE's DUF6. Although beneficial uses of the DUF6 may be forthcoming in the future. NRC staff believes that, because of the current, excess world-wide inventory of DUF6, DOE should assume that a significant portion of both DOE and commercial DUF6 will require disposal as waste. Disposal of the DUF6 will likely require conversion of the material to a more stable physiochemical form, such as U₁O₁. HRC staff has recommended in the past that U₂O₂, which is thermodynamically stable and relatively insoluble, is a likely form for disposal. Although DU₂O₂ could be disposed of in limited quantities in conventional near-surface disposal facilities, the very large quantities derived from a significant fraction of the notion's enrichment tailings indicate a need for a unique disposal facility. We have assumed that such a large quantity, in proper form, might well be disposed of in a mined cavity, perhaps an exhausted uranium mine, providing better containment for such a large quantity of depleted uranium.

The current storage methods employed at the Paducah, Portsmouth, and Oak Ridge facilities have demonstrated insignificant impacts from a health and safety standpoint for about fifty years. However, conversion of the DUF6 to U₂O₈ could provide even tafer intermediate storage for possible future use or eventual disposal.

Although we recognize that, currently, there is limited capacity for converting the DUF, to U_3O_4 , and that a mined cavity for its disposal has yet

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to be developed, we believe that conversion of DUF, to V_3O_8 , and placement of the material in a mined cavity, is one long-term management option that should be included in any evaluation of options by DOE for its DUF,.

If you have any quastions concerning the staff's recommendation, please contact Hichael F. Weber, of my staff, at (301) 415-7298.

Sincerely.

151 Cobut F. Burnett for

Robert H. Bernero, Director Office of Huclear Material Safety and Safeguards

ABJUDICATIONS STAFF

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If you have any questions concerning the staff's recommendation, please contact Michael F. Weber, of my staff, at (301) 415-7298.

·Sincerely.

Robert H. Bernero, Director Office of Nuclear Material Safety and Safeguards

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