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Q200506060007

Scientific Notebook No. 578: Department of  
Energy (DOE) Alloy 22 Repassivation Tests -  
Continuation of Scientific Notebook No. 540  
(08/08/2002 through 04/03/2003)

# LABORATORY NOTEBOOK

CNWRA/SwRI

CNWRA  
CONTROLLED  
COPY 578

NOTEBOOK NO. \_\_\_\_\_  
ISSUED TO DARRELL DUNN *Darrell Dunn DD*  
ON \_\_\_\_\_ 20\_\_\_\_  
DEPARTMENT \_\_\_\_\_  
RETURNED \_\_\_\_\_ 20\_\_\_\_  
*Birby* - Brian K. Deeby - BKD



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TITLE \_\_\_\_\_

From Page No. \_\_\_\_\_

Continued Testing from Notebook # 505 + # 540

**Initial Scientific Notebook Entry for Department of Energy Alloy 22 Localized Corrosion Resistance Tests**

**Title:** DOE Alloy 22 Repassivation Tests.

**Tests Performed by:** Darrell S. Dunn, Div 20; Brian Derby, Div. 18

**Objectives:** Determine the effect of fabrication processes on the localized corrosion susceptibility of Alloy 22.

**Equipment:** Laboratory oven for exposure of test specimens at 600 to 1125 °C, Thermocouple and thermocouple meter. Keithley 614/617. Solartron 1287 Potentiostat and CorrView Software or equivalent, Electrochemical test cell.

**Materials:** Base alloy: Alloy N06022 Alleghany Ludlum heat 059902LL2 welded (GTAW) using Inco Alloys 622 heat XX2048BG filler. Other materials and heats to be added and identified prior to testing.

**Specimen specifications:** Specimens will be equivalent to 20.01402.571.006 rev. 1 unless otherwise specified. Location of specimens with respect to weld will be identified.

**Measurement Parameters:** Temperature and time of exposure, Potential and Current of specimen during test.

**Required level of accuracy:** Temperature of thermal exposure  $\pm 10$  °C Temperature during corrosion tests  $\pm 2$  °C, Time of exposure  $\pm 1$  minute, Potentials  $\pm 1$  mV, Current  $\pm 0.1$  microamp.

**Uncertainty and Sources of Error:** Current measurement error can occur for localized corrosion processes because the actively corroding area is not the same as the surface area of the test specimen.

Copies from NA # 505 + # 540

To Page No. \_\_\_\_\_

Witnessed & Understood by me, \_\_\_\_\_

Date \_\_\_\_\_

Invented by \_\_\_\_\_

Date \_\_\_\_\_

Recorded by \_\_\_\_\_

B. K. O. J.

3/12/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: Same As pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059502LL2  
Taco Alloy 622 heat XX 2048BG filler - Specimen Type 3 Row 4 Bot  
Contains Weld Material - 600 Grit polished Finish - with 2PTFE Crevice  
Washers Attached At 50 In-Oz Using Pasta # 6104 cal 3/6/03 due 9/6/03

Start wt: 34.31025g Sartorius Genius SN# 12509099 cal 11/15/02 due 5/15/03  
End wt: 34.30972g

Solution: 0.001 M NaCl  
0.119g NaCl lot# 027678  
+ DI water to 200mls

pH start: 5.832 Fisher Accumet 950 meter SN# 3340 cal 8/7/02 due 8/7/03  
pH End: 6.709 pH probe # 13-620-296 SN# 2251257 PL

Potentiostat: EG & G model # 273 SN# 41108

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 SN# 0249092

Temperature: 95°C Hg Thermometer SN# 498-170 cal 5/10/02 due 5/10/03

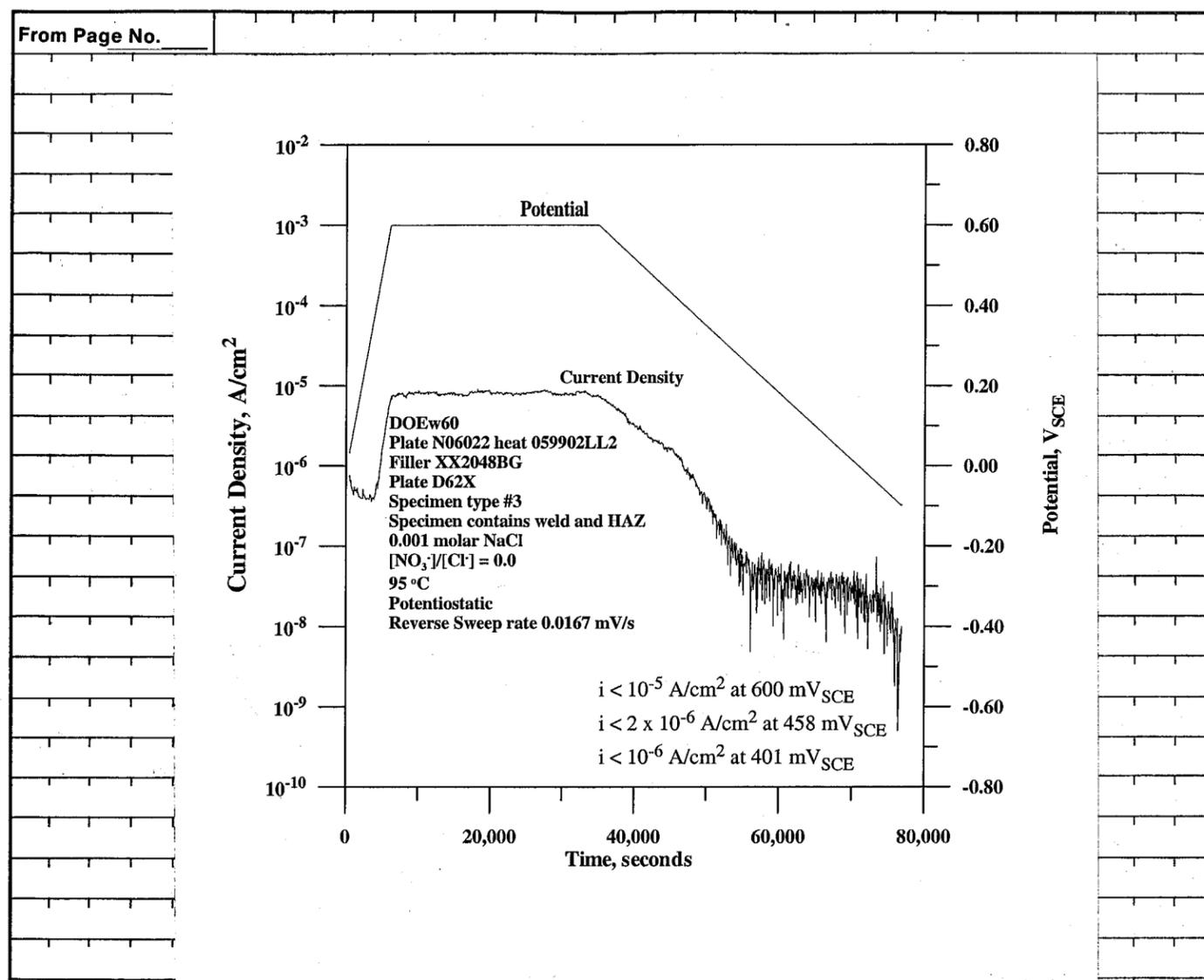
Ecorr = -337mv Keithley 614 SN# 0704934 cal 5/26/02  
Ept = +340mv due 5/26/03

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: No Crevice Corrosion 1/24 feet of crevice washer  
mild staining on facial surfaces of specimen

Date DOE-W60 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	3/27/03



Witnessed & Understood by me, \_\_\_\_\_

Date \_\_\_\_\_

Invented by \_\_\_\_\_

Date \_\_\_\_\_

Recorded by \_\_\_\_\_

3/28/03

Date \_\_\_\_\_ To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	3/28/03

From Page No. \_\_\_\_\_ Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum heat 059902LL2  
Inco Alloy 622 Heat XX2048BG filler - Specimen Type 3 Row 1 Top  
Contains weld Material - 600 Grit polished finish - with 2 PTFE  
Crevice Washers Attached At 50 In-Oz Using Proto # 6104  
SN# 139072 cal 3/6/02 due 9/6/03

Start wt: 33.96390g Santorionic Genius SN# 12509099 cal 11/15/02  
End wt: 33.96217g due 5/15/03

Solution: 0.01 M NaCl  
1.170g NaCl lot # 027578  
+ DI water to 2000ml

pH start: 5.715 Fisher Accumet 950 meter SN# 3340 cal 8/7/02 due 8/7/03  
pH End: 7.366 pH probe Fisher # 13-620-296 SN# 2291257PB

Potentiostat: EG & G Model # 273 SN# 10120

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 SN# 0251439

Temperature: 95°C H<sub>2</sub> Thermometer SN# 00-387 cal 5/10/02  
due 5/16/03

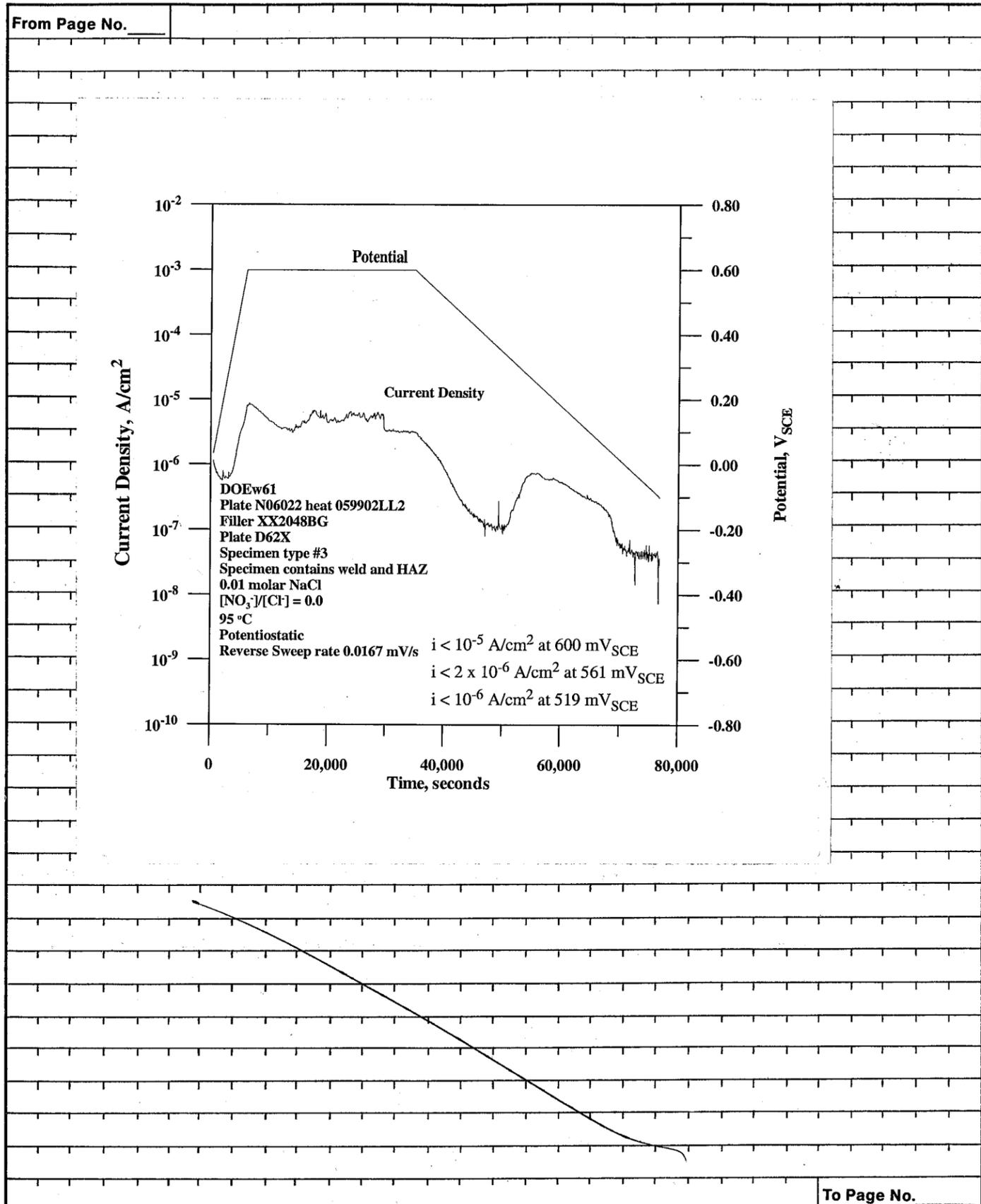
E<sub>corr</sub> = -311 mV Keithley 614 SN# 0704934 cal 5/26/02  
E<sub>pT</sub> = +87 mV due 5/26/03

Solution Deaerated with 99.999% O<sub>2</sub>

Specimen Examination: crevice corrosion on 2/24 feet of crevice washer  
n/o surface staining

DATA DOE-w61 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>Bi [Signature]</i>	3/07/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	3/28/03

From Page No. \_\_\_\_\_ Repassivation Potential of Alloy C-22

Objective: Same as pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 heat XX2048BG Filler - Specimen Type 3 Row 1 Bottom  
Contains weld material - 600 grit polished finish - with 2 PTFE  
Crevice washers Attached AT 50 In-02 Using Proto #6104 SN#139672  
cal 3/6/02 due 9/6/03

Start wt = 34.03725g Sartorius Genius SN#12809099 cal 11/15/02 due 5/13/03  
End wt = 34.03295g

Solution: 0.1 M NaCl  
11.691g NaCl lot # 027878  
+ DI water To 2000mls

pH start = 5.983 Fisher Accumet 950 meter SN#3340 cal 8/7/02 due 8/2/03  
pH End = 7.286 pH probe # 13-620-296 SN#2291257 PG

Potentiostat: EG & G model # 273 SN#41108

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 SN# 0249092

Temperature: 95°C Hg Thermometer SN#H98470 cal 5/10/02 due 5/10/03

Epot2 = -512mV Keithley 614 SN# 0704934 cal 5/24/02 due 5/24/03  
Ept = +95mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 1/24 feet of crevice washer  
staining color tint on all surfaces of specimen

Date DOE-W62

To Page No. \_\_\_\_\_

Witnessed & Understood by me,

Date

Invented by

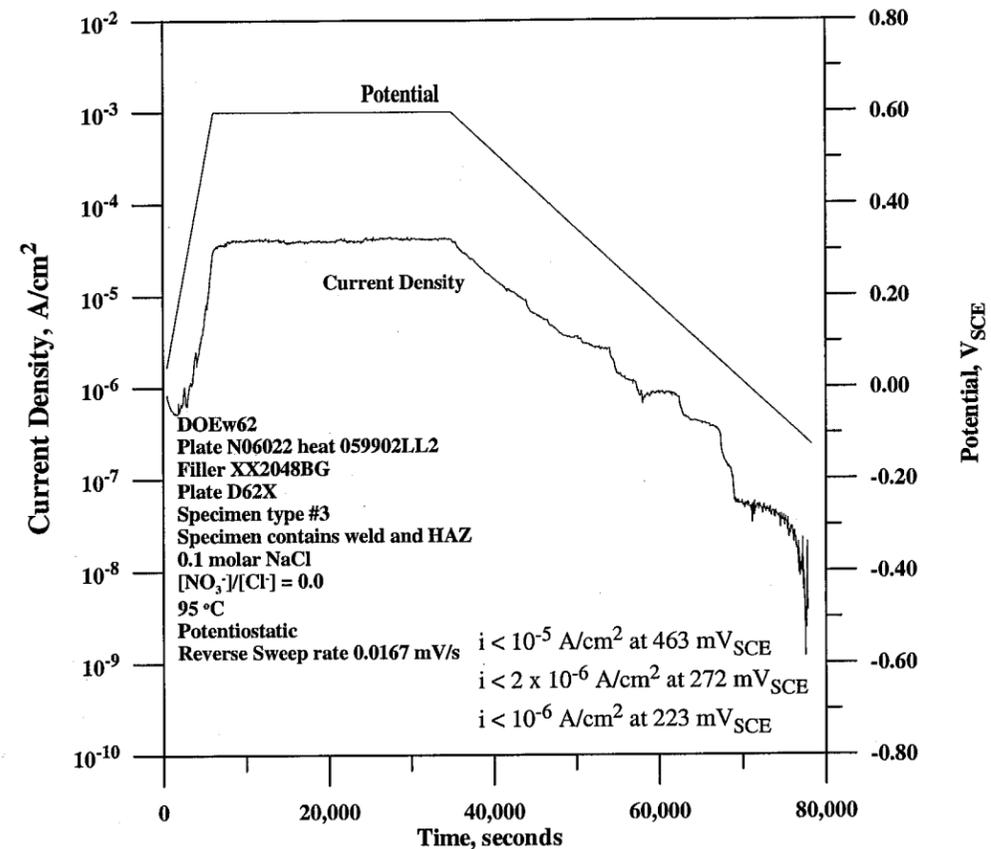
Date

Recorded by

*Bick*

4/1/03

From Page No. \_\_\_\_\_



Witnessed & Understood by me,

Date

Invented by

Date

Recorded by

*Daniel Du*

4/2/03

To Page No. \_\_\_\_\_

From Page No. \_\_\_\_\_ Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022- Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XZP4856 filler - Specimen Type 1 Panel M10  
Doesn't contain weld Material - 600 grit polished Finish - with 2 PTFE Ceramic  
washers Attached At 50 In-Oz Using Petro 6104 SN# 139072 cal 3/6/03 pre 9/6/03

Start wt = 34.1798g Satorious Genius SN# 12809099 cal 11/15/02 pre 5/15/03  
End wt = 34.1797g

Solution 0.001 M NaCl  
0.119g NaCl lot # 027878  
+ DE To 2000 ml

pH start = 5.947 Fisher Accumet 950 meter SN# 3340 cal 8/7/02 pre 8/7/03  
pH End = 7.506 pH probe Fisher #13-620-296 SN# 2291257 PL

Potentiostat = EG & G model #273 SN# 10120

Counter Electrode = Pt Flg

Reference: Fisher 13-620-52 SN# 0251429

Temperature: 95°C Hg Thermometer SN# 00-387 cal 5/10/02 pre 5/10/03

Ecorr = -474 mV Keithley 614 SN# 0704934 cal 5/26/02 pre 5/26/03  
Ept = -93 mV

Solution Degassed with 99.999% N<sub>2</sub>

Specimen Examination: No Ceramic Corrosion 1/24 feet of Ceramic Washer  
w/o surface staining

DATA DOE-W63

To Page No. \_\_\_\_\_

Witnessed & Understood by me, \_\_\_\_\_

Date \_\_\_\_\_

Invented by \_\_\_\_\_

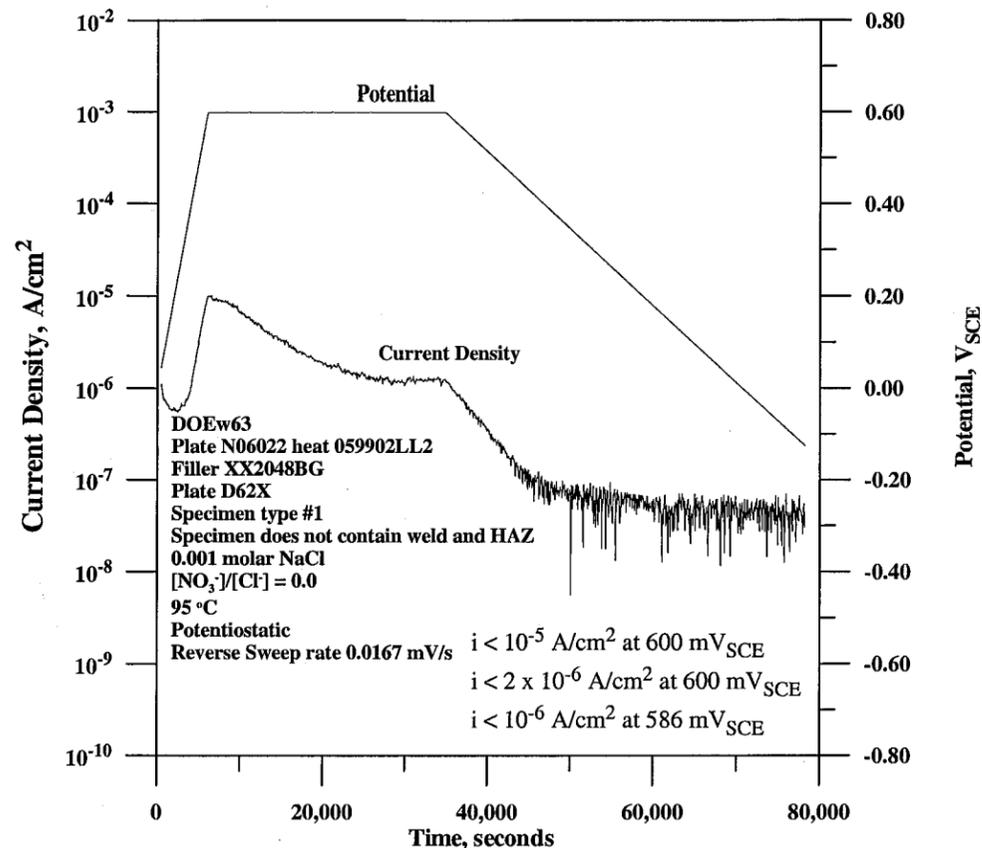
Date \_\_\_\_\_

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4/1/03

*Bill [Signature]*

From Page No. \_\_\_\_\_



Witnessed & Understood by me, \_\_\_\_\_

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Invented by \_\_\_\_\_

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Recorded by \_\_\_\_\_

4/2/03

*David [Signature]*

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: Same As #1

Specimen: DOE Alloy N06022- Allegheny Ludlum Heat 059902442  
Inco Alloy 622 heat XX2048BG filler - Specimen Type 1 Row 2 Top  
Doesn't contain weld material - 600 Grit polished finish - with 2 PTFE  
Crevice washers Attached At Sp In-Oz Using Photo #6104 SN# 139077  
cal 3/6/03 Due 9/6/03

Start wt = 34.23934g SARTORIUS Genius SN# 12809099 cal 11/15/02 due 5/15/03  
End wt = 34.23847g

Solution: 0.01M NaCl  
1.171g NaCl lot # 02787R  
+ DI water to 2000ml

pH start = 6.168 Fisher Asset 950 meter SN# 3340 cal 8/7/02 due 8/7/03  
pH End = ~~6.820~~ 6.820 pH probe # 13-620-296 SN# 2291257 P6  
4/3/03

Potentiostat: EG & G model # 273 SN# 41108

Counter Electrode: Pt Flay

Reference: Fisher 13-670-52 SN# 0249897

Temperature: 95°C Hg thermometer SN# H98-170 cal 5/10/02 due 5/10/03

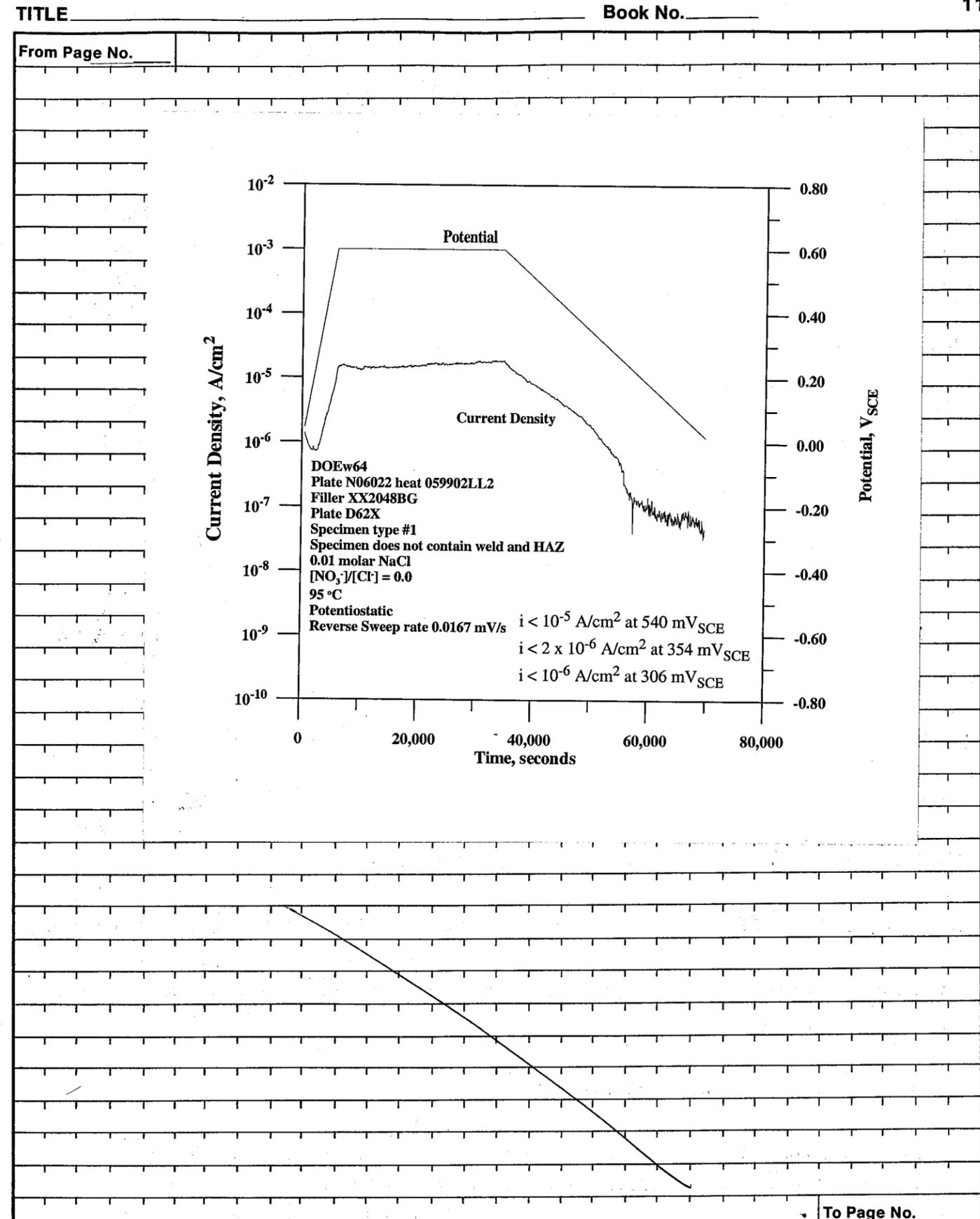
Ecorr = -415mV Keithley 614 SN# 0704934 cal 5/26/02 due 5/26/03  
Ept = +92mV

Solution Deaeration with 99.999% N<sub>2</sub>

Specimen Examination: No crevice corrosion 1/24 feet of crevice washer  
mild grey tint staining on all surfaces of specimen

Data DOE-464 To Page No. \_\_\_\_\_

Witnessed & Understood by me, \_\_\_\_\_ Date \_\_\_\_\_  
Invented by \_\_\_\_\_ Date 4/3/03  
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Invented by \_\_\_\_\_ Date 4/3/03  
Recorded by *[Signature]*

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Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XX 2048BG filler - Specimen Type 1 Row 2 min  
Doesn't Contain Weld Material - 600 grit polished Finish with 2 PTFE  
Crevice washers Attached At 50 In-O<sub>2</sub> Using Photo #6104 sn#139072  
Cal 3/6/03 Due 9/6/03

Start wt = 34.28693g Sartorius Genius sn#12909099 Cal 11/15/02  
End wt = 34.27994g Due 5/15/03

Solution 0.1 M NaCl  
11.689g NaCl 027878  
+ DI water To 200mls

pH start = 5.894 Fisher Accumet 950 meter sn#3340 Cal 8/7/02 Due 8/7/03  
pH End = 6.489 pH probe Fisher #13-620-296 sn#2291257 PL

Potentiostat: EG&G Model # 273 sn#10120

Counter Electrode: PT Flag

Reference: Fisher 13-620-52 sn#0251439

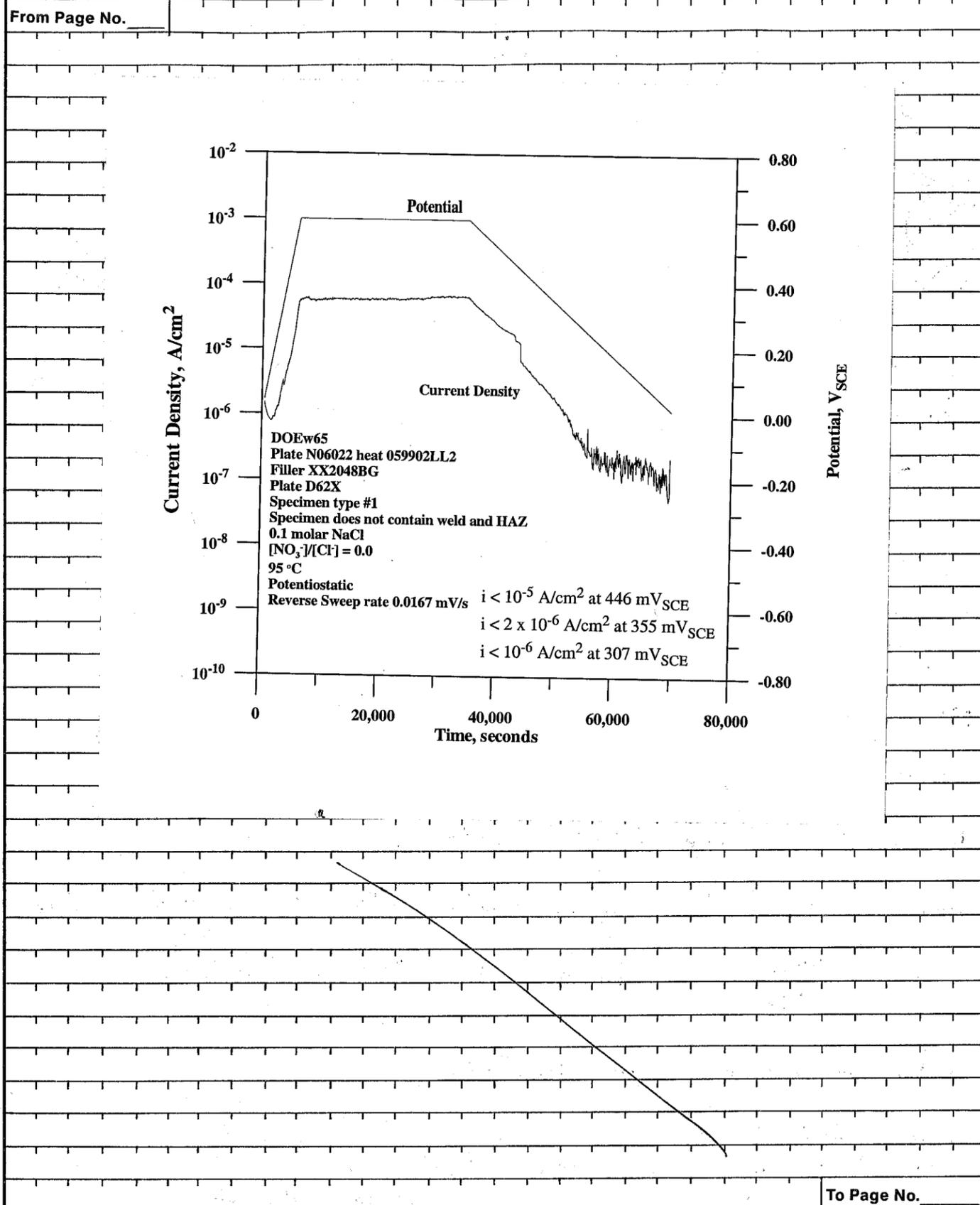
Temperature: 95°C H<sub>2</sub> Thermometer sn#00-387 Cal 5/10/02 Due 5/10/03

Evam = -494mV Keithley 614 sn#0704924 Cal 5/26/02 Due 5/26/03  
Ept = -125mV

Solution Deaerates with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 1/24 feet of crevice  
washer. Color that staining on All surfaces of Specimen  
Data DOE-w65

Witnessed & Understood by me, \_\_\_\_\_ Date \_\_\_\_\_  
Invented by \_\_\_\_\_ Date 4/3/03  
Recorded by *[Signature]*



Witnessed & Understood by me, \_\_\_\_\_ Date \_\_\_\_\_  
Invented by \_\_\_\_\_ Date 4/3/03  
Recorded by *[Signature]*

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Repassivation Potential of Alloy C-22

Objective: Same As pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 heat XX2048BG filler - Specimen Type 1 Row 2 Bot  
Do not contain weld material - 600 Grit polished finish - with 2 PTFE  
Crevice washers Attached At 50 In-Oz Using Proto sn# 139072 #6104  
cal 3/6/03 due 9/6/03

Start wt: 34.21146g Sartorius Genius SN# 12809099 cal 11/15/02 due 5/15/03  
End wt: 34.21058

Solution: 0.25 m NaCl  
29.236g NaCl lot# 627828  
+ DI water To 200mls

pH Start = 6.219 Fisher Accumet 950 meter SN# 3340 cal 8/7/02 due 8/7/05  
pH End = 7.567 pH probe # 13-620-296 SN# 2291257 P6

Potentiostat: EG & G model # 273 SN# 41108

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 SN# 0245052

Temperature: 95°C Hg Thermometer SN# H94-170 cal 5/10/02 due 5/10/03

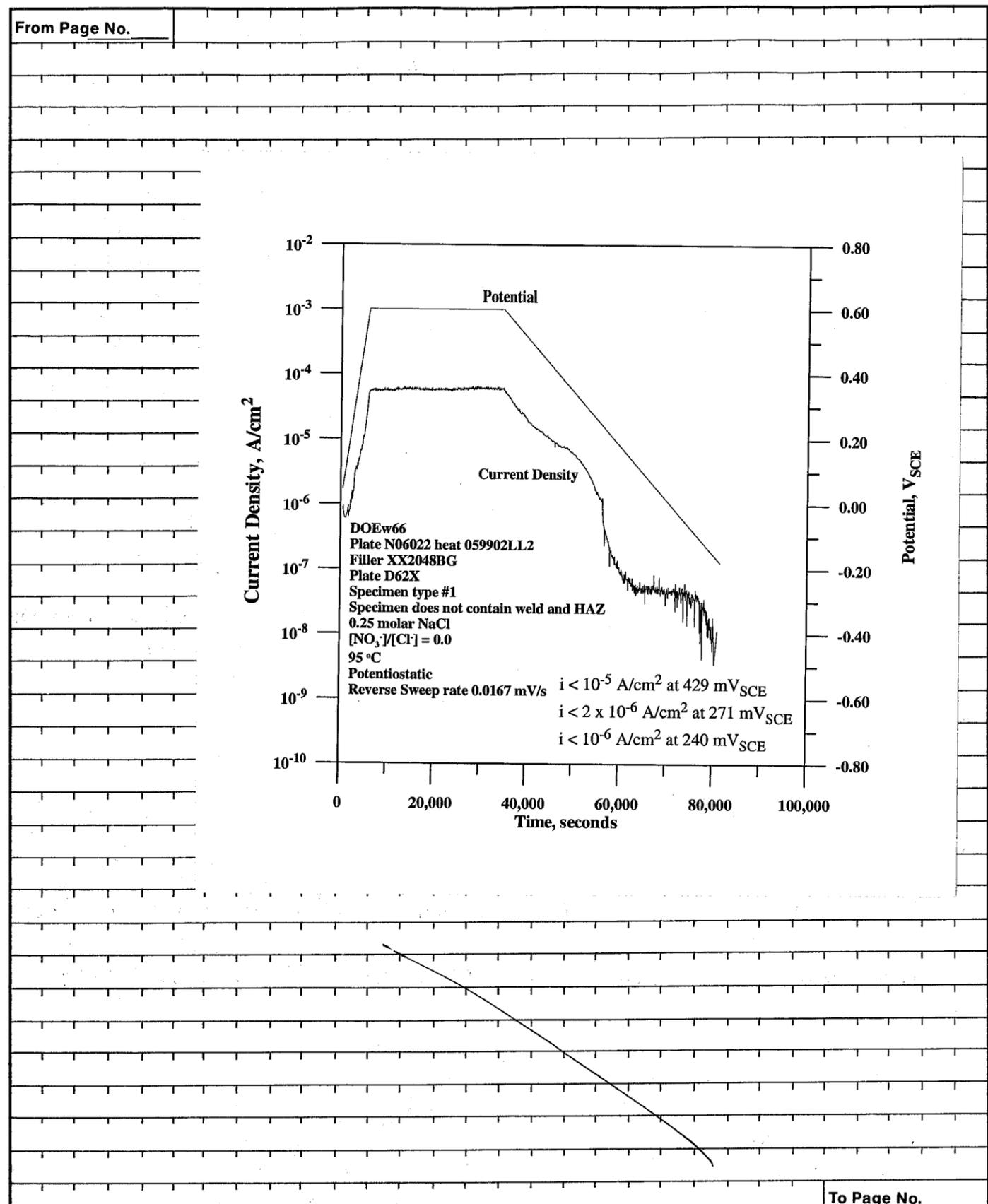
E<sub>corr</sub> = -443 mV Keithley 614 SN# 6704934 cal 5/26/02  
E<sub>pt</sub> = +39 mV due 5/26/03

Solution bubbled/aerated with 99.999% N<sub>2</sub>

Specimen Examination No Crevice Corrosion 0/24 feet of crevice washer  
Gold tint staining on all surfaces of specimen

Date DOE\_w66 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/4/03
		<i>B. [Signature]</i>	



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		Recorded by	4/7/03
		<i>[Signature]</i>	

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: Same As #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 heat XX2048BG filler - Specimen Type Row 3 Top  
Doesn't contain weld material - 600 Grit polished Finish - with 2 PTFE  
Crevice washers Attached At 50 In-Oz Using Probe #6104 SN#139072  
cal 3/6/03 due 9/6/03

start wt: 34.28055g Satorius Genius SN#12809099 cal 11/15/02 due 5/15/03  
End wt: 34.23851g

Solution: 0.5 M NaCl  
58.49g NaCl 027878  
+ DI water to 200ml

pH Start: 6.079 Fisher Accumat 250 meter SN#3340 cal 8/7/02 due 8/7/03  
pH End: 7.783 pH probe #13-620-296 SN#2291257 P6

potentiostat: EG & G Model #273 SN#10120

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 SN#0251439

Temperature: 95°C H<sub>2</sub> Thermometer SN#00-387 cal 5/9/02 due 5/10/03

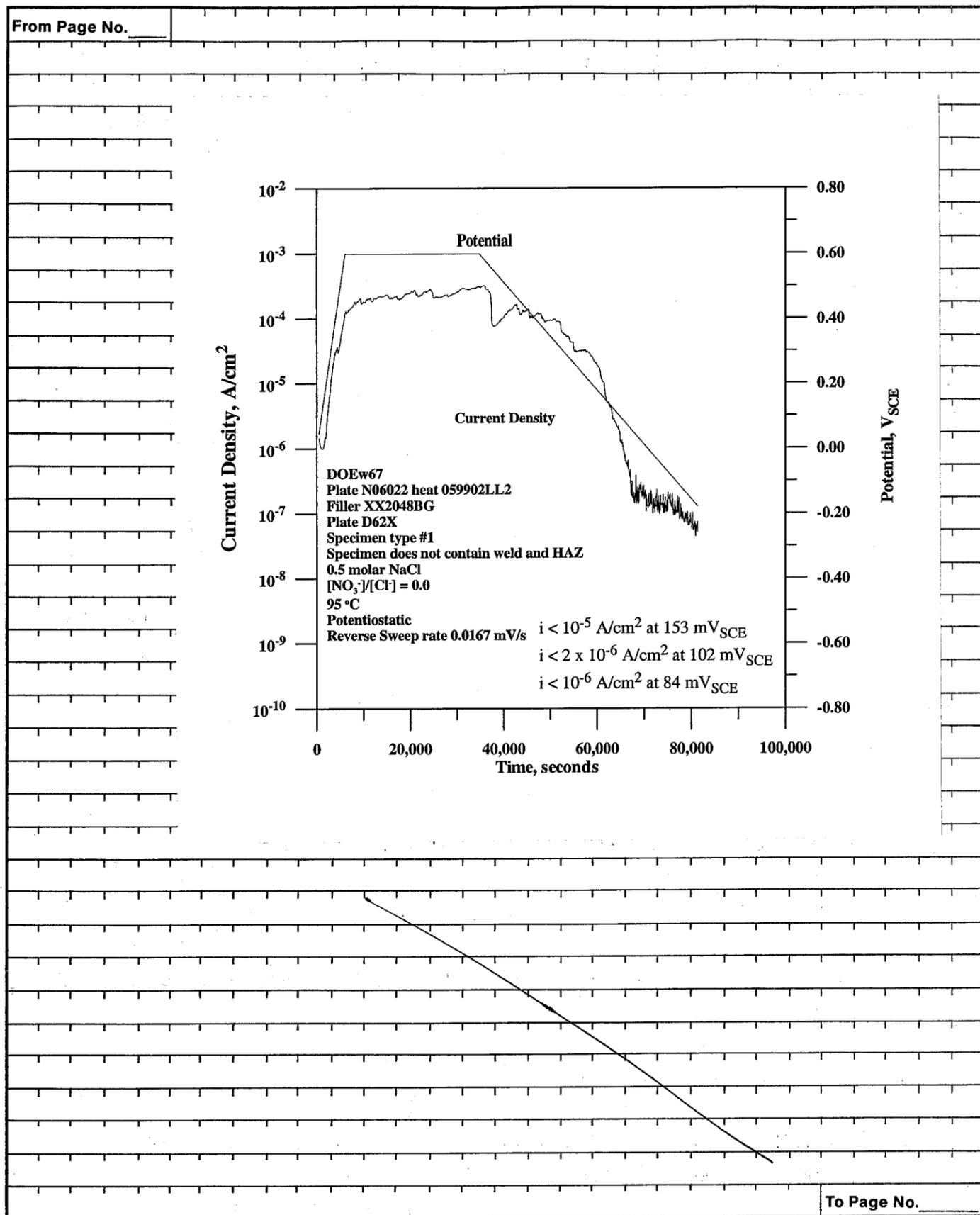
E<sub>corr</sub> = -466 mV Keithley 614 SN#0704934 cal 5/26/02 due 5/26/03  
E<sub>pt</sub> = -200 mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 8/24 feet of crevice washer  
Cold tint staining on All Surfaces

Data DOE W67

Witnessed & Understood by me,	Date	Invented by	Date	To Page No. _____
		Recorded by <i>[Signature]</i>	4/4/03	



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		Recorded by <i>[Signature]</i>	4/7/03	

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### Repassivation Potential of Alloy C-22

Objective: Same As pg #1

Specimen: DOE Alloy N06022 - Alleghany Ludlum heat 059902LL2  
Incp Alloy 622 heat XX 2048BG filler - Specimen Type 2 Row 1 Top  
Down: Carbon web Material - 600 Grit polishes Finish - with 2 PTFE  
Crevice washers Attached At 50 In-Oz Using Paper #6104 SN# 139672  
Cal 3/6/03 Due 9/6/03

Start wt: 34.15618g      Sartorius Corus SN# 1289059 cal 11/15/02 due 5/15/03  
End wt: 34.15512g

Solution: 0.01 M NaCl  
11.695g NaCl Lot # 027878  
+ DI Water To 2000ml

pH start: 5.819      Fisher Accumet 950 meter SN# 3340 cal 8/7/02 due 8/7/08  
pH End: 7.578      pH probe # 13-620-296 SN# 2291257 PL

Potentiostat: EG & G model #273 SN# 41108

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 SN# 0249092

Temperature: 95°C      Hg Thermometer SN# 498-170 cal 5/10/02 due 5/10/03

Ecorr -371 mV      Keithley 614 SN# 0704934 cal 5/26/02 due 5/26/03  
E<sub>pt</sub> +181 mV

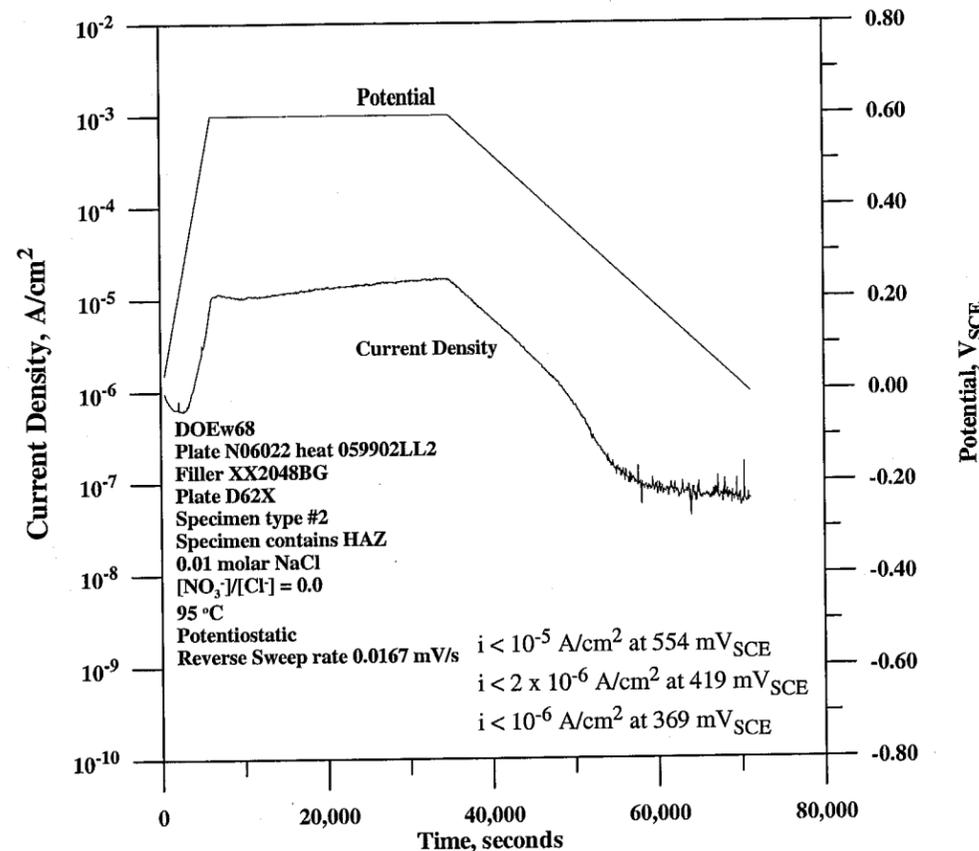
Solution Deaerates with 99.999% N<sub>2</sub>

Specimen Examination: No crevice corrosion 1/24 feet of crevice washer  
Staining on All surfaces of specimen

Date DOE-68 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. J. [Signature]</i>	4/7/03

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Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/10/03

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### Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Alleghany Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XX2048BG filler - Specimen Type 2 Row 4 Top  
Doesn't contain weld material - 600 grit polished finish - with 2 PTFE  
Crevice washer Attaches At 50 In-Oz Using Puro #6104 SN#135072  
cal 3/6/03 Due 9/6/03

Start wt: 34.24414g Santoripus Genis SN#12809099 cal 11/15/03 due 5/15/03  
End wt: 34.24399g

Solution: 0.001 M NaCl  
0.125g NaCl lot # 027578  
+ DI water to 200mls

pH start: 5.796 Fisher Accuret 950 meter SN#3340 cal 8/7/02 due 7/03  
pH End: 7.542 pH probe #13-620-296 SN#2291257P6

Potentiostat: EG & G Model #273 SN#16120

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 SN#025143

Temperature: 95°C Hg Thermometer SN#00-387 cal 3/10/02 due 5/10/03

Equip: -492 mV Keithley 614 SN#0704934 cal 5/26/02 due 5/26/03  
E<sub>pt</sub>: -137 mV

Solution Deaerated with 99.999% N<sub>2</sub>

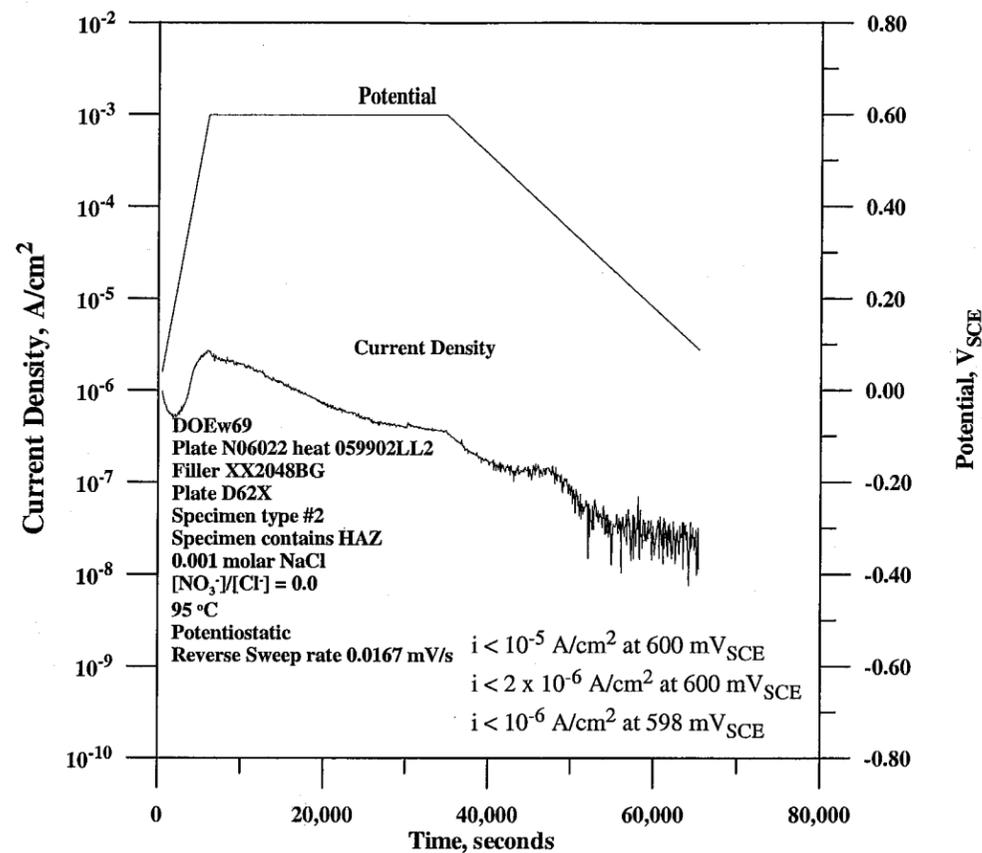
Specimen Examination: No crevice corrosion - 1/24 feet of crevice washer  
mild surface staining

Data DOE-W69

To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. J. D.</i>	4/1/03

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Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/10/03

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Repasivation Potential of Alloy C-22

Objective: Same As pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 heat XX2048BG filler - Specimen Type 2 Row 3 Bottom  
Donor: Contain weld Material - 600 Grit polished Finish - with 2 PTFE  
Crevice Washers Attached At 50 In-Oz Using Proto 6104 SN# 139072  
Cal 3/6/03 Due 9/6/03

Start wt = 34.06341 g Sartorius Genius SN# 12809099 cal 11/15/02 Due 5/15/03  
End wt = 34.63785 g

Solution: 0.25 M NaCl  
29.228 g NaCl lot # 027878  
+ DI water To 2000 ml

pH Start = 5.754 Fisher Accuret 950 meter SN# 3340 cal 9/7/02 Due 8/7/03  
pH End = 8.075 pH probe # 13.620.296 SN# 2291257 PL

Potentiostat: EG & G Model # 273 SN# 41108

Counter Electrode: Pt Flag

Reference: Fisher 13-62052 SN# 0249092

Temperature: 95°C Hg Thermometer SN# M98-170 cal 5/10/02 Due 5/10/03

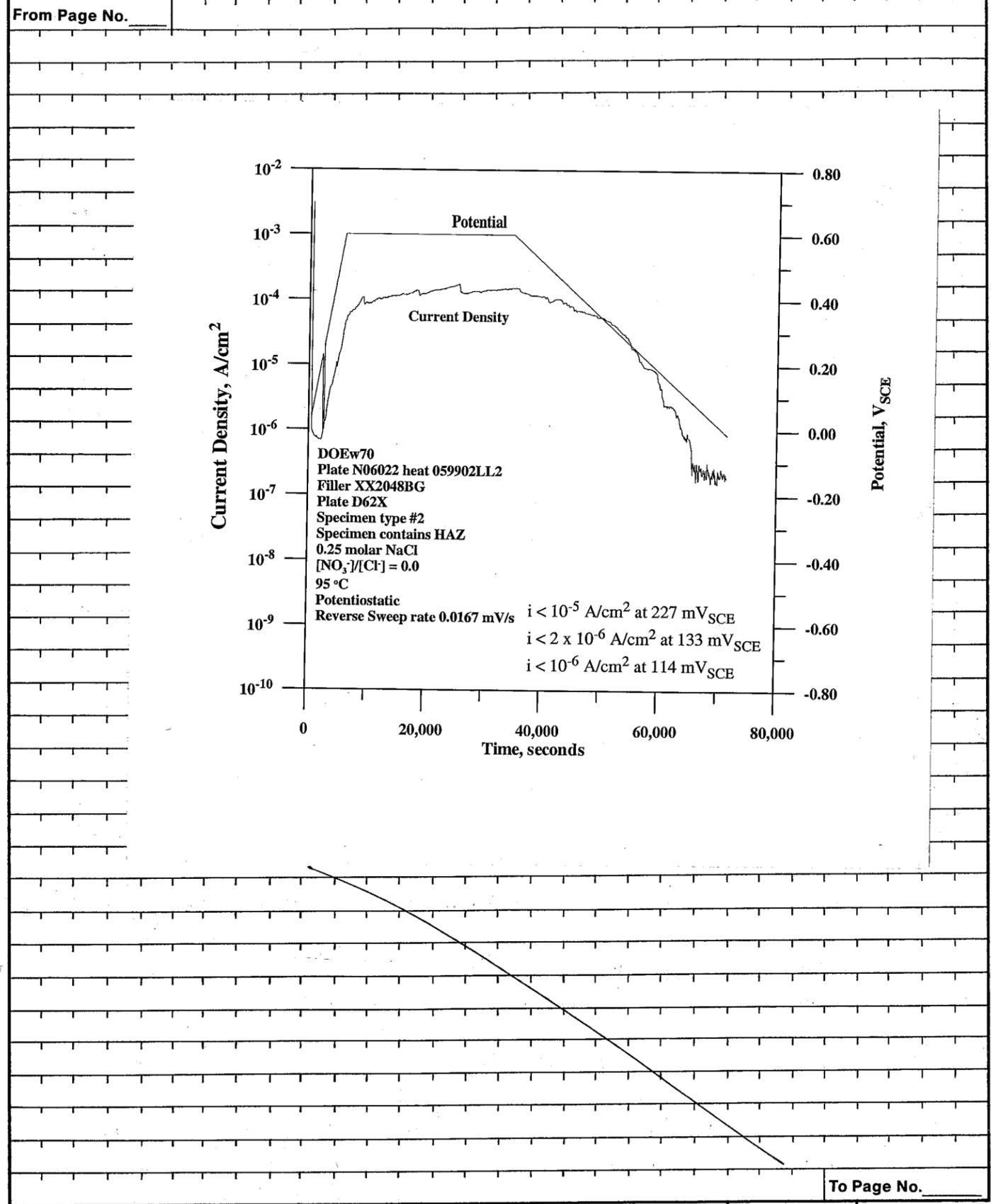
Ecorr = -475 mV Keithley 614 SN# 0704934 cal 5/26/02 Due 5/26/03  
Ept = -123 mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 7/24 feet of crevice washer  
Cable tint staining on All surfaces of Specimen

Data DOE-W70 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/9/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/14/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XX2048BG filler - specimen Type 2 Row 4 Bottom  
Doesn't contain Lab Material - 600 Grit polishes Finish - with 2 PTFE  
Crevice Washer Attaches At 50 In.-Oz Using Proto #6104 sn#139072  
cal 3/6/03 due 9/6/03

Start wt = 34.3098g      Sanjourious Genius sn#12809099 cal 11/15/02  
End wt = 34.2809g      due 5/15/03

Solution      0.1 m NaCl  
11.690g NaCl lot#027878  
+ DI water to 2000mls

pH Start = 6.079      Fisher Accumat 950 Meter #3340 cal 8/7/02 due 8/7/03  
pH End = 7.249      pH probe #13-620-296 sn#2291257P6

potentiostat: EG+G model #273 sn#10120

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 sn#0251439

Temperature: 95°C      Hg thermometer sn#00-387 cal 5/10/02  
due 5/10/03

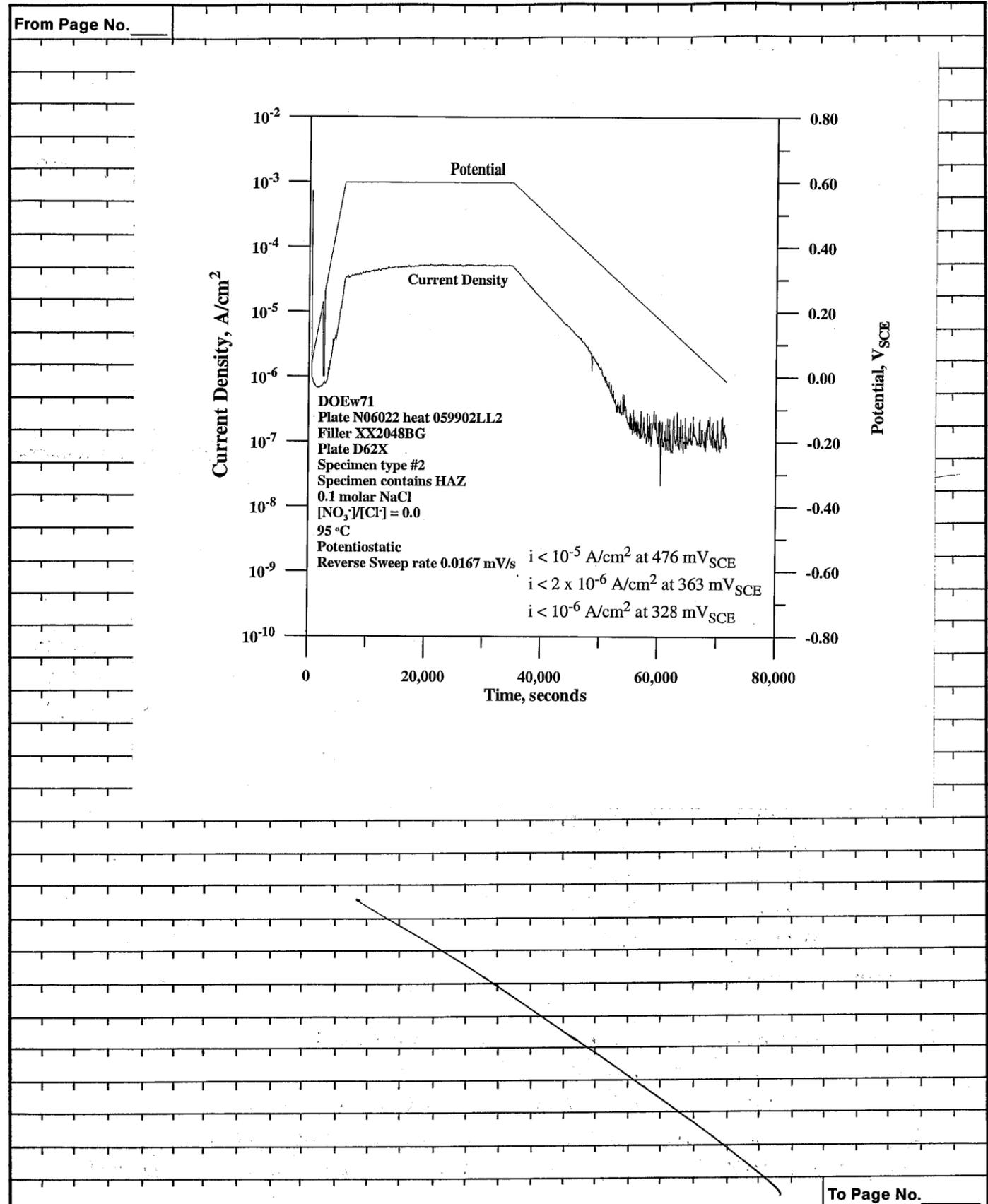
E<sub>corr</sub> = -503mv      Keithley 614 sn#0704934  
E<sub>pt</sub> = -193mv      cal 5/26/02 due 5/26/03

Solution Deaerates with 99.999% N<sub>2</sub>

Specimen Examination: No Crevice Corrosion 1/24 feet of Crevice Washer  
Color that staining on All Surfaces

Data DOE-W71      To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/9/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/14/03

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Thermal Treatment of Specimens

specimens Are Type 2 Near weld Area Haz (3 specimens)

Oven: Linberg Model # S1333 SN# 909172

Oven set point: 875°C

Temperature check: Omega HM22 Microprocessor thermometer  
SN# T 94140  
cal 10/29/02 Due 4/29/02

Used Type K Thermocouple SN# 322 cal 1/14/03 due 6/14/03  
7/14/03  
DD 4/11/03

Oven Temp Reading: 880.5°C

Procedure Thermally Anneal @ 875°C for 5 min

Start: 874°C	Oven Temp:
1:00 min	876°C
2:00 min	876°C
3:00 min	875°C
4:00 min	875°C

End: Remove

\* After procedure then Solution Anneal:

Oven Set point = 1125°C

Omega HM22 SN# T 94140 cal 10/29/02 due 4/29/02

Type K Thermocouple SN# 322 cal 1/14/03 due 6/14/03

Temp = 1134.8°C Total Time 15:00 min @ 1125°C

To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. [Signature]</i>	4/10/03

TITLE \_\_\_\_\_

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Thermal Treatment of Specimens

Specimens Are Type 3 that contain weld Material  
Total of 4 specimens - Solution Annealed @ 1125°C  
for a total of 15 min.

Oven: Linberg Model # S1333 SN# 909172

Oven set point: 1125°C

Temperature check: Omega H22 Microprocessor thermometer  
SN# T 94140 cal 10/29/02 due 4/29/03

Used Type K Thermocouple SN# 322 cal 1/14/03 due 6/14/03  
7/14/03  
DD 4/11/03

Oven temp Reading = 1134.8°C

All Specimens from pg 26 and 27 were  
polished to a 600 Grit prior to thermal treatment  
and will be repolished to 600 Grit finish

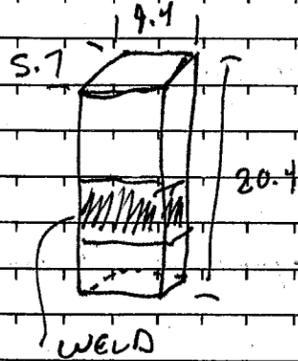
To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. [Signature]</i>	4/10/03

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### Thermal Treatment of Specimens

SPECIMENS: CUT FROM PLATE D62X AND CONTAIN  
WELD METAL SPECIMENS ARE APPROXIMATELY  
20.4 x 9.1 x 5.7 mm STARRETT 734M SN 024371  
CAL 2/7/03 DUG 3 DD 4/11/03 DUG 2/7/04



DETAILS OF WELD LOCATED  
IN NOTEBOOK 505

BASE METAL HEAT 059902LL2  
FILLER METAL XX2048 BG

OVEN SETPOINT 1300°C  
TEMP CHECK 1299°C OMEGA HH22 SN T54140  
CAL 10/29/02 DUG 4/29/03  
WITH TYPE K THERMOCOUPLE SN 332  
CAL 1/14/03 DUG 6/4/03 7/14/03  
DD 4/11/03

OVEN SET POINT 1250°C  
TEMP CHECK 1249°C OMEGA HH22 SN T54140  
TYPE K SN 332

OVEN SET POINT 1200°C DD 4/11/03  
TEMP CHECK 1200°C OMEGA HH22 SN T54140  
TYPE K SN 332

### SPECIMEN SOLUTION ANNEAL TEMPS AND TEMP

- 1300°C FOR 15 min WATER QUENCH
- 1250°C FOR 15 min WATER QUENCH
- 1200°C FOR 15 min WATER QUENCH

To Page No. \_\_\_\_\_

Witnessed & Understood by me,

Date

Invented by

Date

Recorded by

4/11/03

From Page No. \_\_\_\_\_

Witnessed & Understood by me,

Date

Invented by

Date

Recorded by

4/10/05

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From Page No. \_\_\_\_\_

Repasivation Potential of Alloy C-22

Objective: Same As pg #1

Specimen: DOE Alloy N06022-Allegheny Ludlum Heat 059902LL2  
Jaco Alloy 622 heat xx 2048BG Filler - Specimen Type 3 Row 2 Top  
Contains web Material - 600 Grit Polished finish - with 2 PTFE Crevice Washers  
Attached At SA In-Oz Using Probe #6104 SN#139072 cal 3/6/03 due 9/6/03  
\* Specimen Solution Annealed see pg #27

Start wt = 33.84807g Sartorius Genius SN#12809099 cal 11/15/02 due 5/15/03  
End wt = 33.84719g

Solution: 0.001 M NaCl  
0.117g NaCl Lot # 02787R  
+ DI water to 200mls

pH Start = 5.787 Fisher Accumet 950 meter SN#3340  
pH End = 8.649

Potentiostat = EG&G Model #273 SN#41108

Counter Electrode = Pt Flay

Reference: Fisher 13-620-52 SN#0249092

Temperature: 95°C Hg Thermometer SN#498-170 cal 5/10/02 due 5/10/03

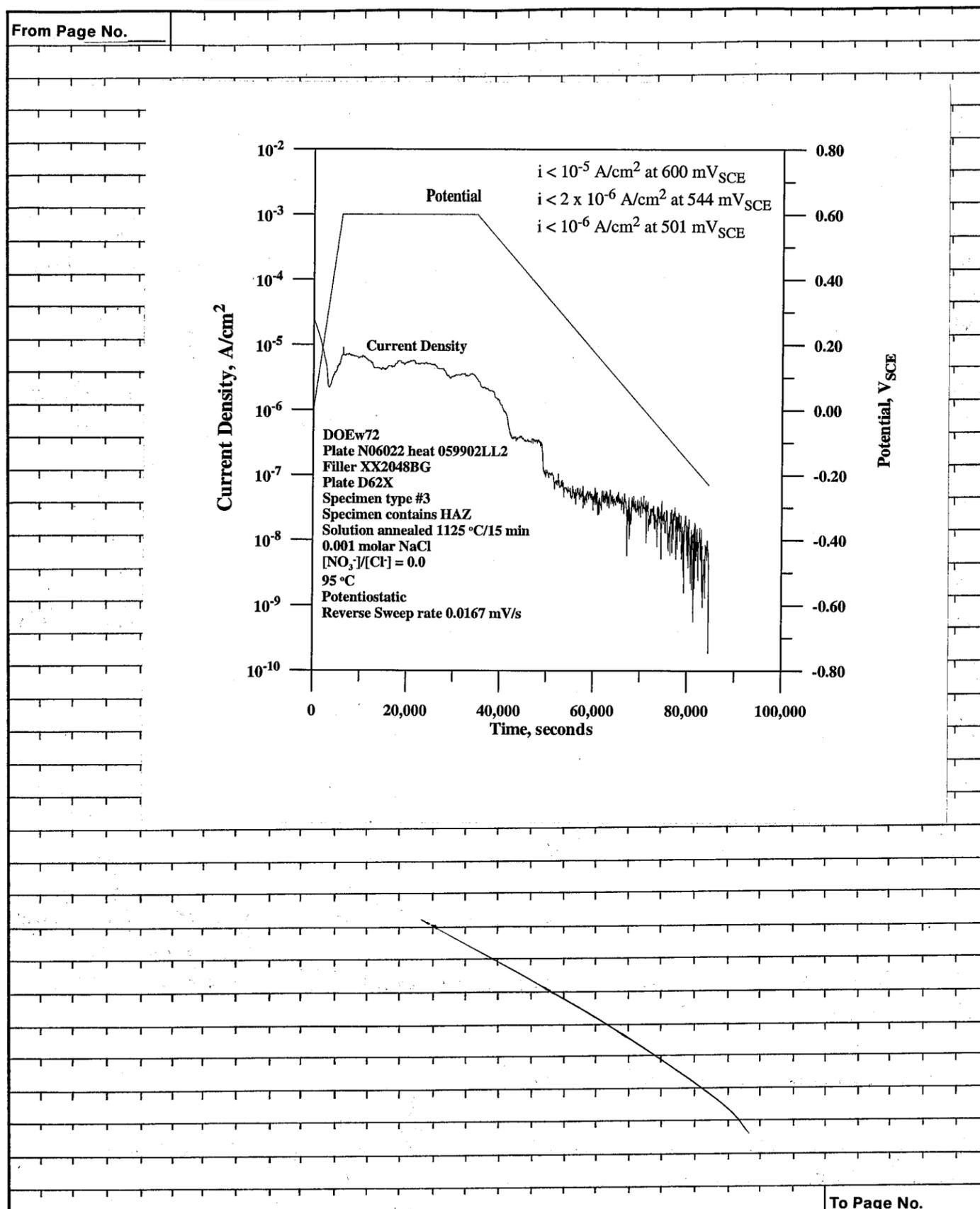
Epot = -507mV Keithley 614 SN#0704934 cal 5/26/02 due 5/26/03  
Ept = +165mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 7/24 feet of Crevice Washer  
Mild Staining on All Surfaces of Specimen

Data DOE\_W72 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. [Signature]</i>	4/14/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/17/03

From Page No. \_\_\_\_\_

Repasivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XX 204566 Filler - Specimen Type 3 Row 2 Bottom  
Contains weld material - 600 grit polished finish - with 2 PTFE crevice washers  
Attached At 50 To-02 Using Photo 6104 SN# 135072 cal 3/6/03 due 9/6/03  
\* Specimen Solution Annealed see pg #27

Start wt: 33.99578g Sartorius Genius SN# 12509099 cal 11/15/02 due 5/15/07  
End wt: 33.98930g

Solution: 0.01 M NaCl  
1.172g NaCl lot # 027878  
+ DI water to 2000ml

pH Start = 5.728 Fisher Accumet 950 Meter SN# 3340 cal 8/7/02 due 8/7/03  
pH End = 8.409 pH probe # 13-620-296 SN# 2291257 PL

Potentiostat: EG+G model # 273 SN# 10120

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 SN# 0251435

Temperature: 95°C Hg Thermometer SN# 00-387 cal 5/10/02 due 5/10/03

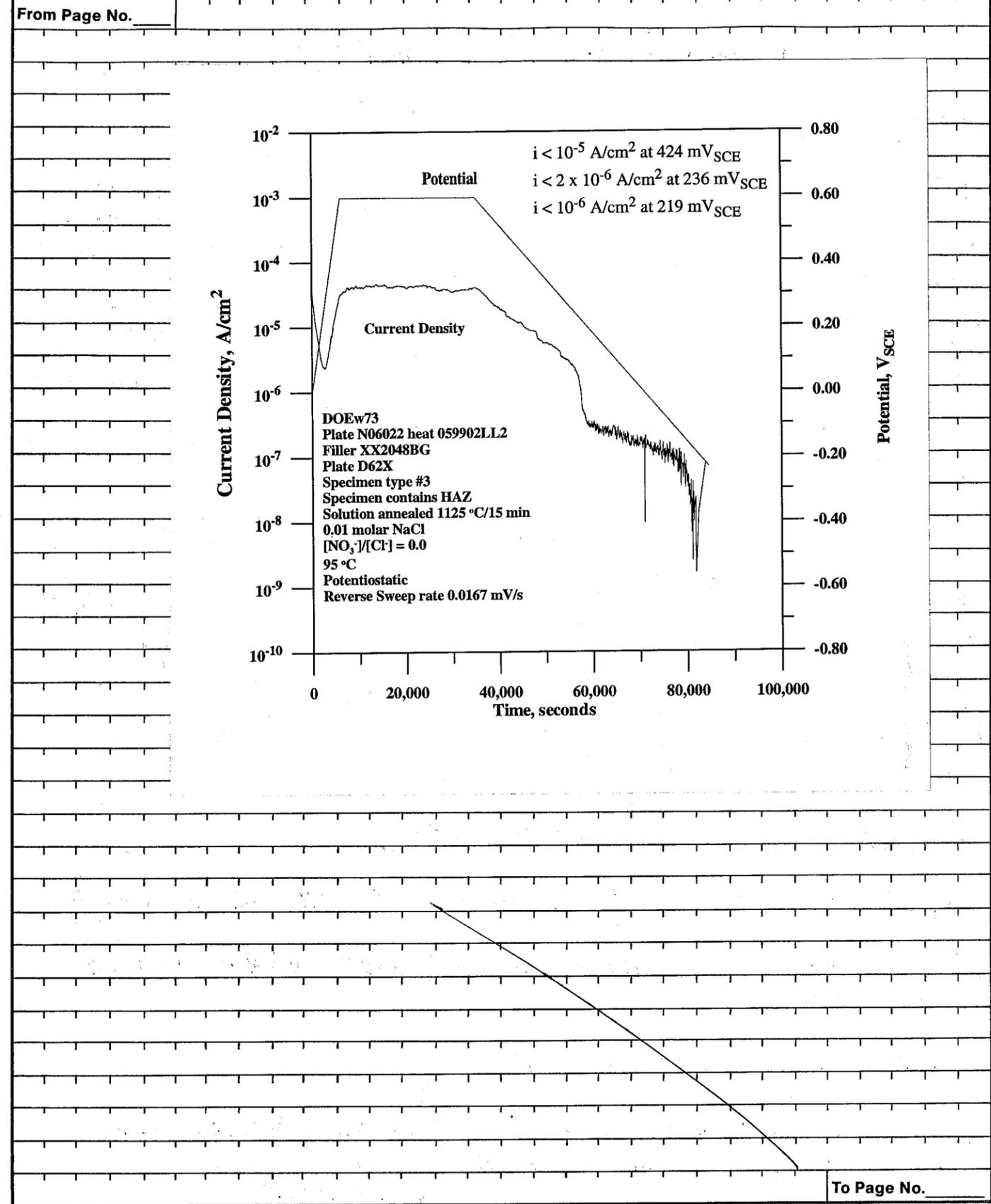
Ecorr = -508 mV Keithley 614 SN# 0704934 cal 5/26/02 due 5/26/03  
Ept = -65 mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 1/24 feet of crevice washer  
Gold lot staining on All surfaces of specimen

Date DOE-W73 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/14/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/17/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: same As pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
InCO Alloy 622 heat XX2048BG filler - Specimen Type 3 Row 4 Bottom Top  
Contains weld Material - 600 grit polished finish - with 2 PTFE crevice washers  
Attached At 50 In-Oz Using Parts # 6104 sn# 139072 cal 3/6/03 pvc 9/6/03  
\* Specimen Solution Annealed See pg #27

Start wt = 33.9564g     Sartorius Genius sn# 12809099 cal 11/15/02 pvc 5/15/03  
End wt = 33.90280g

Solution: 0.1 M NaCl  
11.694g NaCl lot # 027578  
+ DI water to 2000ml

pH start = 6.348     Fisher Accumet 950 meter sn# 3340 cal 8/7/02 pvc 8/7/02  
pH End = 8.098     pH probe # 13-620-296 sn# 2291257 Pb

Potentiostat: EG&G Model #273 sn# 41108

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 sn# 0216492

Temperature: 95°C     Hg Thermometer sn# 1198-170 cal 5/10/02 pvc 5/10/02

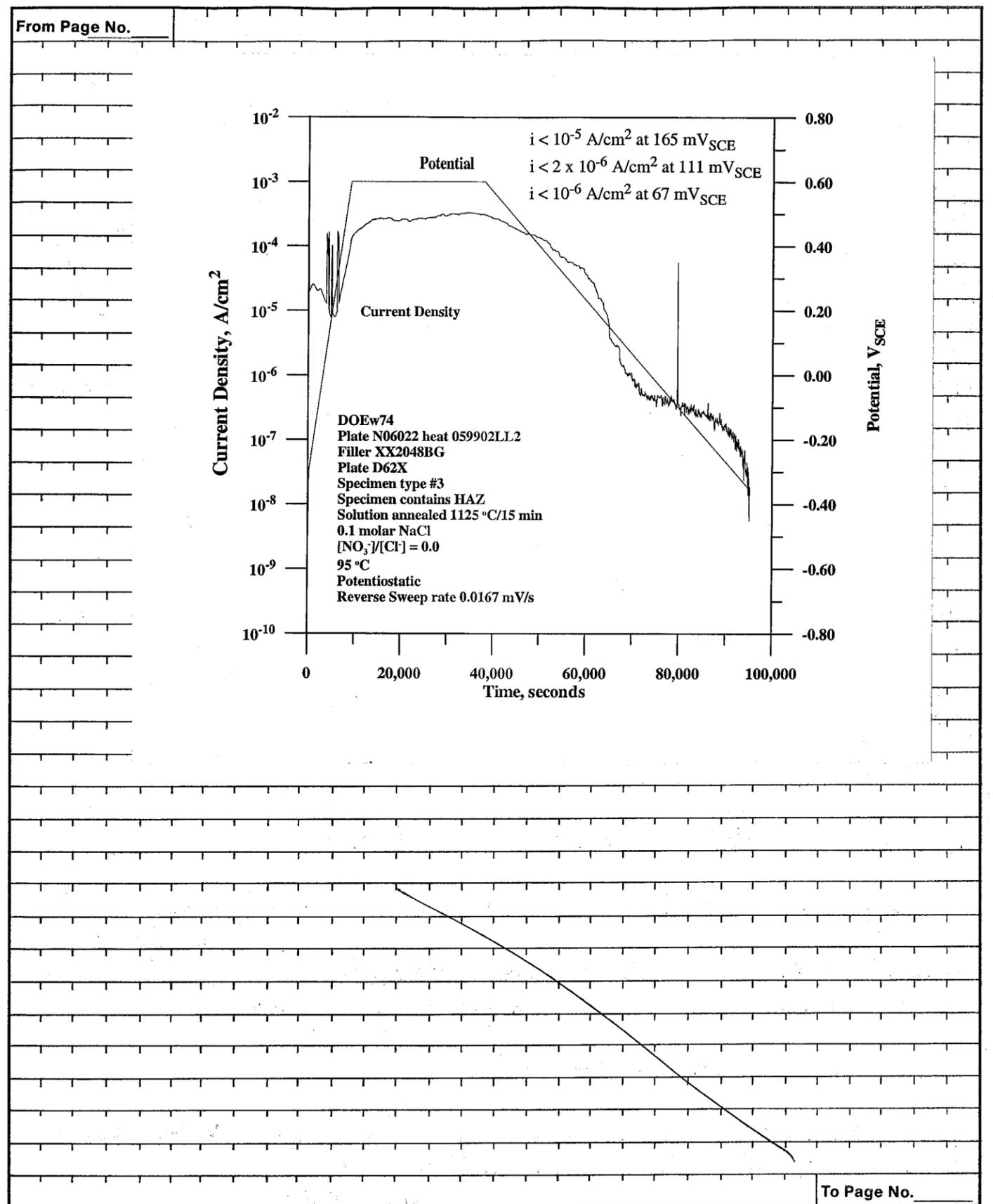
Epot = -552mV     Keithley 614 sn# 0704934 cal 5/26/02 pvc 5/26/02  
Ept = +160mV

Deaerated Solution with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 20/24 Feet of Crevice washer  
Staining, Galvanic on All Surfaces of Specimen

Data DOE-W74     To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/18/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/21/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: same as pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2  
Inco Alloy 622 Heat XX2048BG filler - Specimen Type 3 Row 4 bottom  
Contains weld Material - 600 Grit polished Finish - with 2 PTFE Coresic Washers  
Attached At 50 In-Oz using Probe #6104 SN#139077 cal 3/6/03 due 9/6/03  
\* Specimen Solution Annealed See pg #27

Start wt: 34.01910g Sartorius Genius SN#12809099 cal 11/15/02 due 5/15/03  
End wt: 33.89036g

Solution: 0.25 M NaCl  
29.26g NaCl Lot# 027478  
+ DI water to 2000mls

pH Start: 6.328 Fisher Accumet 950 meter SN#3340 cal 8/7/02 due 4/7/03  
pH End: 8.126 pH probe #13-620-296 SN#2291257 PL

Potentiostat: EG + G model #273 SN#10120

Counter Electrode: Pt Flag

Reference: Fisher 13-020-52 SN#0251439

Temperature: 95°C H<sub>2</sub> thermometer SN#00-387 cal 5/14/02 due 5/14/03

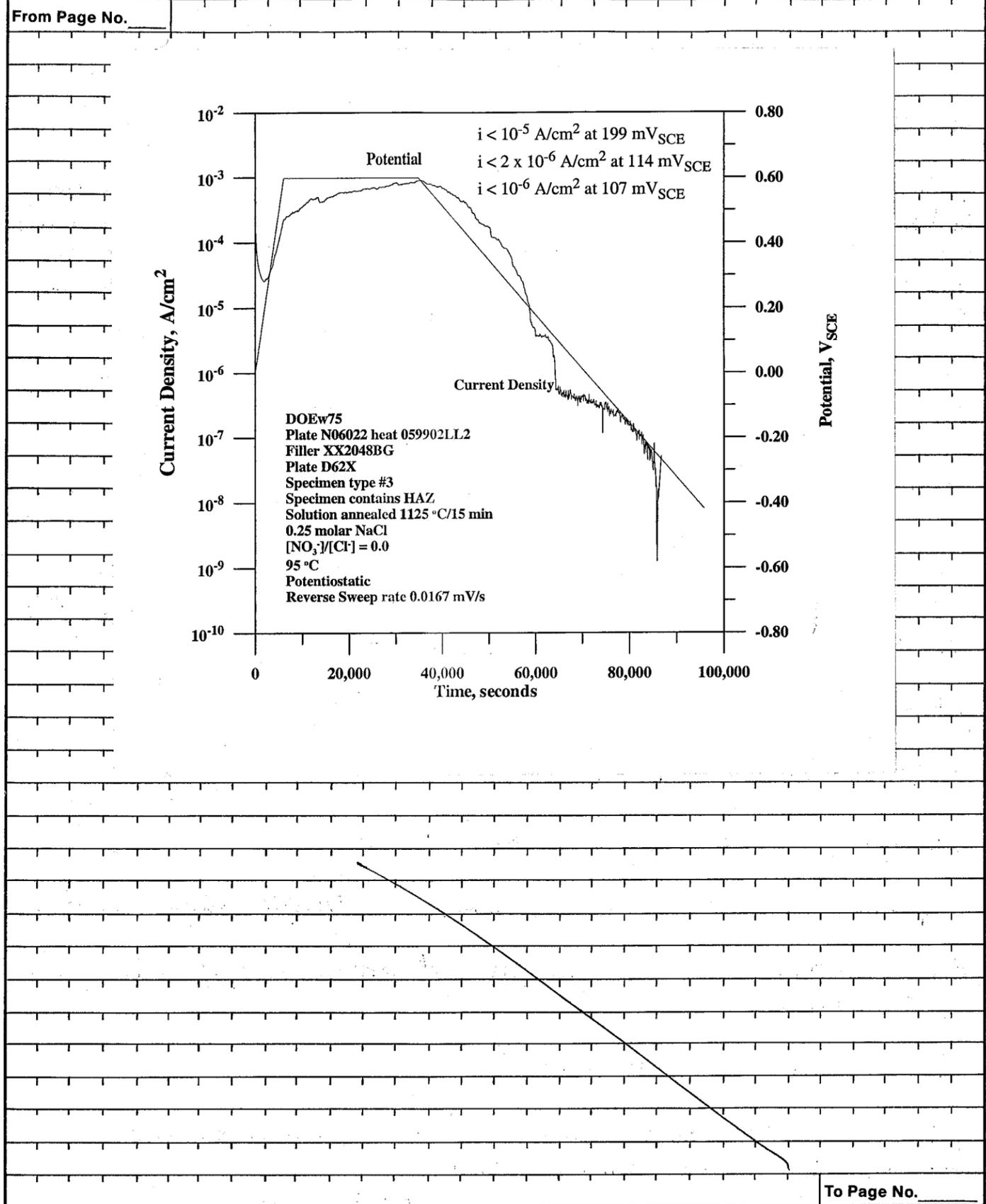
Equip: -528 mV Keithley 614 SN#0704934  
Ept: +21 mV cal 5/26/02 due 5/26/03

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Coresic corrosion on 20/24 feet of Coresic Washer  
Staining on All Surfaces of specimen Gold tint.

Date DOE-wts To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/18/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/21/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: Same As #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2 - Inco Alloy 622 heat # 204566 Filler - Specimen Type 2 Row 2 Top - Doesn't contain weld material 600 grit polished finish - with 2 PTFE Crevice Washers Attached At 50 In-Oz Using Pasta 6104 SN#139072 cal 3/4/03 due 9/6/03 \* Specimen Thermally Aged 5 min @ 870°C then Solution Annealed see pg # 26

Start wt = 34.00892g Satorious Genius SN#12809099 cal 11/15/02 due 5/15/03  
End wt = 34.00728g

Solution: 0.001 M NaCl  
0.124g NaCl lot # 027878  
+ DI To 2000mls

pH start = 6.112 Fisher Accumet 950 meter SN#3340 cal 8/7/02 due 8/7/03  
pH End = 6.773 pH probe # 13-620-296 SN#2291257 Pb

Potentiostat = EG & G model # 273 SN#41108

Counter Electrode: Pt Flay

Reference: Fisher 13-620-52 SN#0244092

Temperature: 95°C Hg Thermometer SN#H98-170 cal 5/16/02 due 5/16/03

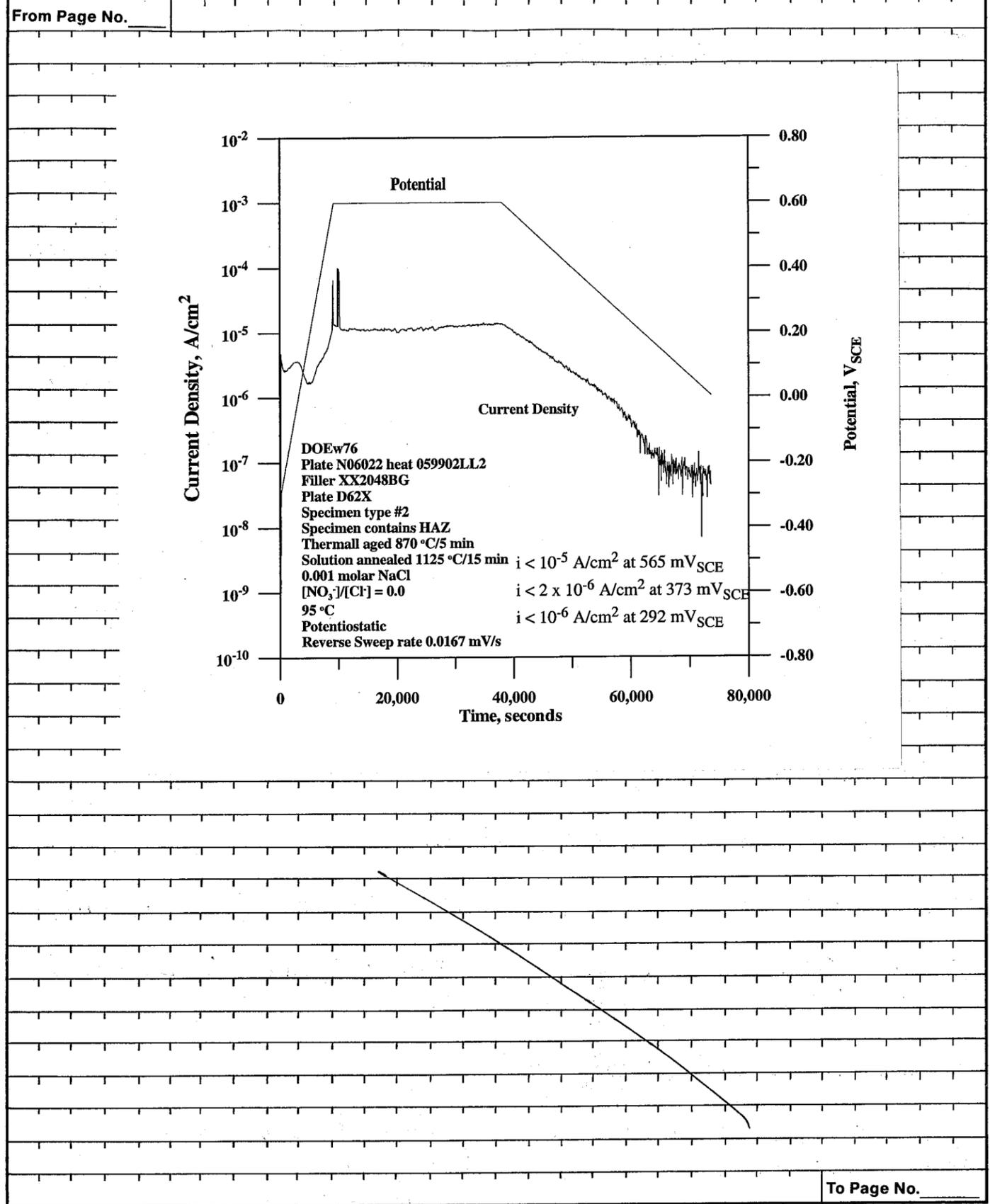
E<sub>corr</sub> = -541mV Keithley 614 SN#0204934 cal 5/26/02 due 5/26/03  
E<sub>pt</sub> = +118mV

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: No crevice corrosion 1/24 Feet of crevice washer Gold tint staining on All surfaces of Specimen

Date DOE\_w76 To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/21/03



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/22/03

From Page No. \_\_\_\_\_

Repassivation Potential of Alloy C-22

Objective: See pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2 - Inco Alloy 622  
Heat XX2048BG Filler - Specimen Type 2 Row1 Center - Down? Contain web Material  
600 grit Polished Finish - with 2 PTFE crevice washers Attached At 50 In. Oz. Using  
Pasta # 6104 SN# 139072 Cal 3/6/03 due 9/6/03 \* Specimen Thermally Aged  
for 5 min @ 870°C then solution Annealed see pg # 26

Start wt = 33.87151g Sartorius Genius SN# 1209099 cal 11/15/02 due 5/15/03  
End wt = 33.86965g

Solution: 0.01 M NaCl  
1.176g NaCl Lot# 027878  
+ DI water to 2000mls

pH start = 5.981 Fisher Accumet 950 meter SN# 3340 cal 8/7/07 due 8/7/07  
pH End = 7.665 pH probe Fisher #13-620-296 SN# 2291257 PL

Potentiostat: EG & G model # 273 SN# 10120

Counter Electrode: Pt Flag

Reference: Fisher 13-620-52 SN# 0251439

Temperature: 95°C Hg Thermometer SN# 00-387 cal 5/10/02 due 5/10/03

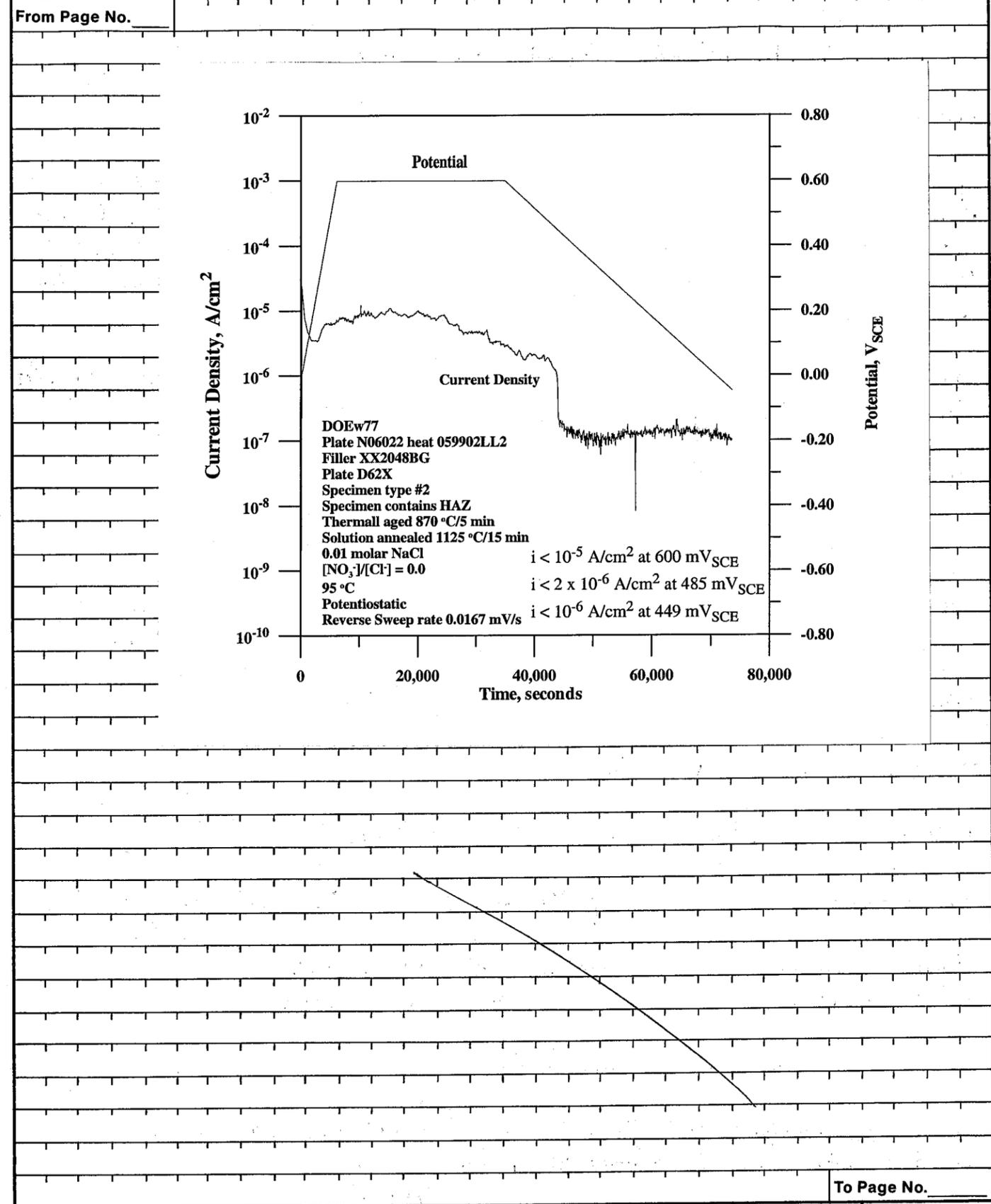
Equip: -539 mV Keithley 614 SN# 0704534 cal 5/26/02 due 5/26/03  
Ept: +18 mV

Solution Degenerates with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 3/24 feet of crevice washer  
Color tint staining mls on All surfaces of Specimen

Data DOE-W77

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>B. F. J.</i>	4/21/05



Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by <i>[Signature]</i>	4/22/05

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Repassivation Potential of Alloy C-22

Objective: same as pg #1

Specimen: DOE Alloy N06022 - Allegheny Ludlum Heat 059902LL2 - Inco Alloy 622  
Heat XX 2048BG filler - Specimen Type 2 Row 1 Bottom - Doesn't contain Welo Material  
600 Grit polished Finish - with 2 PTFE crevice washers Attached At 50 In-Oz  
Using Probe #6104 SN#139077 cal 3/6/03 due 9/6/03 \* Specimen Thermally Aged  
@ 870°C for 5 min then Solution Annealed See pg #26.

Start wt: 33.78510g Sartorius Genius SN#12209099 cal 11/15/02 due 5/15/03  
End wt: 33.76565g

Solution: 0.1 M NaCl  
11.691g NaCl lot# 027878  
+ DI water - to 200ml

pH Start = 5.893 Fisher Accumat 950 Meter SN#3340 cal 8/7/02 due 8/7/03  
pH End = 8.073 pH probe SN#13-620-296 SN#229125796

Potentiostat = EG + G Model #273 SN#41108

Counter Electrode = Pt Flag

Reference = Fisher 13-620-52 SN#024902

Temperature: 95°C Hg Thermometer SN#H98-170 cal 5/10/02 due 5/10/03

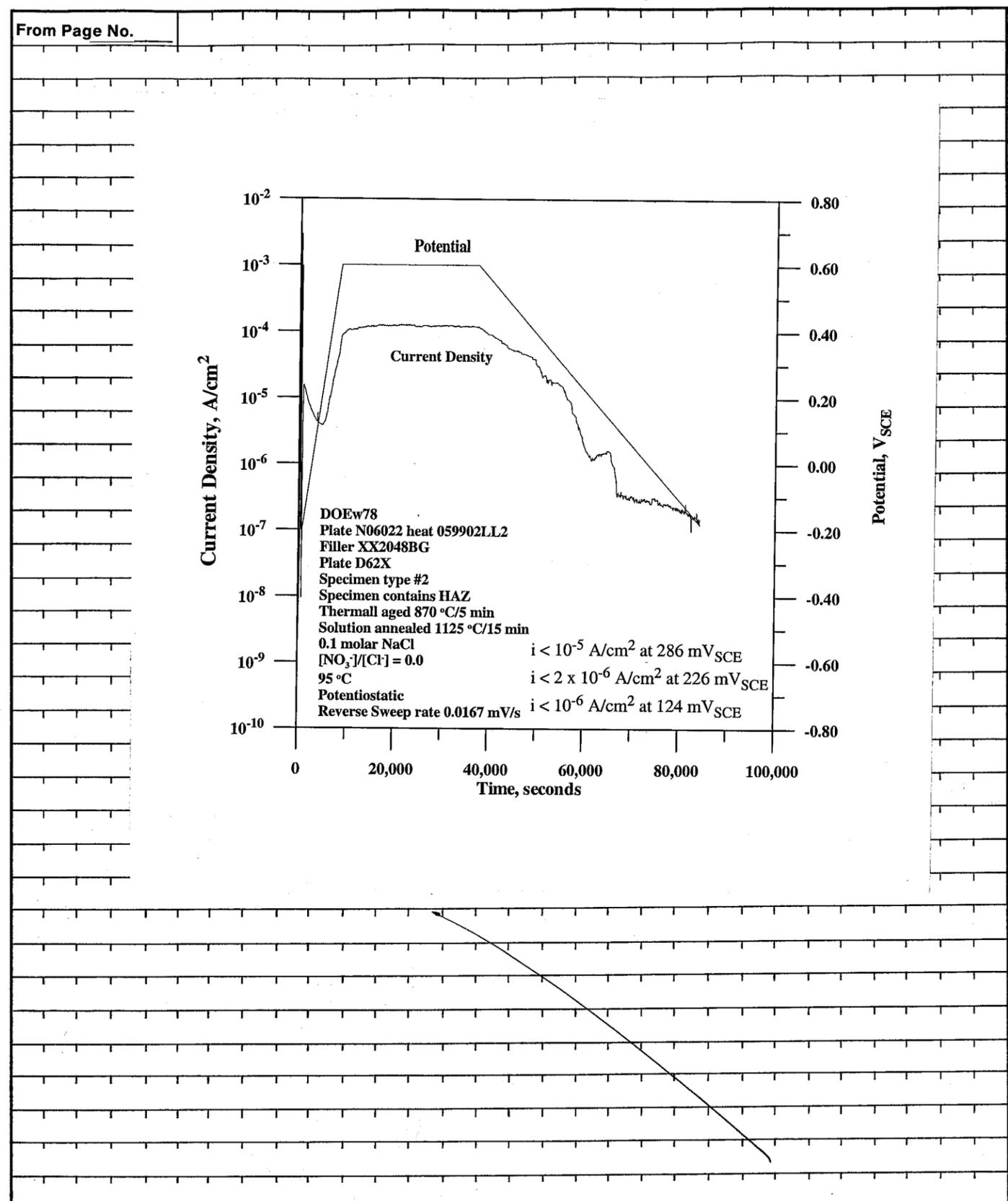
Equip = -430mv Keithley 614 SN#0704934  
Ept = -110mv cal 5/26/02 due 5/26/03

Solution Deaerated with 99.999% N<sub>2</sub>

Specimen Examination: Crevice Corrosion on 1/24 feet of Crevice Washer.  
Capit that staining on All surfaces of Specimen

Done DOE WTS To Page No. \_\_\_\_\_

Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/23/03 4/16/04



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Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by	4/29/03



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I have reviewed this scientific notebook and find it in compliance with QAP-001. There is sufficient information regarding procedures used for conducting tests, acquiring and analyzing data so that another qualified individual could repeat the activity.

*[Handwritten Signature]*

6/6/05

To Page No. \_\_\_\_\_

Witnessed & Understood by me,

Date

Invented by

Date

Recorded by

### ADDITIONAL INFORMATION FOR SCIENTIFIC NOTEBOOK NO. 578

<b>Document Date:</b>	08/08/2002
<b>Availability:</b>	Southwest Research Institute® Center for Nuclear Waste Regulatory Analyses 6220 Culebra Road San Antonio, Texas 78228
<b>Contact:</b>	Southwest Research Institute® Center for Nuclear Waste Regulatory Analyses 6220 Culebra Road San Antonio, TX 78228-5166 Attn.: Director of Administration 210.522.5054
<b>Data Sensitivity:</b>	<input checked="" type="checkbox"/> "Non-Sensitive" <input type="checkbox"/> Sensitive <input type="checkbox"/> "Non-Sensitive - Copyright" <input type="checkbox"/> Sensitive - Copyright
<b>Date Generated:</b>	06/04/2005
<b>Operating System:</b> (including version number)	Windows
<b>Application Used:</b> (including version number)	Excel and others unknown
<b>Media Type:</b> (CDs, 3 1/2, 5 1/4 disks, etc.)	1 - CD
<b>File Types:</b> (.exe, .bat, .zip, etc.)	xls, dat
<b>Remarks:</b> (computer runs, etc.)	Media contains one folder named DOE welded Alloy 22 containing data files.