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In reply, please refer to NISM 72-28 March 30, 1972

Director, Division of Materials Licensing United States Atomic Energy Commission Washington, D.C. 20545

Attention: Mr. Donald A. Nussbaumer, Chief Fuel Fabrication and Transportation Branch

Reference: AEC License No. SNM-33, Docket 70-36

Subject: Revision of Green Room Nuclear Safety Basis

Gentlemen:

On October 29, 1971, we advised you that an application for revision of the nuclear safety of low enriched scrap reprocessing based on favorable geometry and/or poisoning of process vessels and equipment would be submitted by April 1, 1972. This letter, attached revised Subpart 823, Supplemental Safety Analysis Report, and list of effective pages comprise that application.

Nuclear Safety of all of the process vessels and equipment has been assured by favorable geometry, poisoning, and specific computer analyses.

Although the Green Room revisions would be extensive, the basic process would remain the same and would be performed in the same area. As you can imagine, considerable engi-neering effort has gone into the preliminary design of the revision and still more detailed equipment and process design work will be required before the modifications could actually be made. Assuming your early review and approval of the application and no delays in engineering and equipment procurement, the earliest possible time that we could start the modifications would be October, 1972. However, beginning about August, 1972, processing of several contiguous orders for uranium oxide will be It is anticipated that these orders will be under way. completed during the third quarter of 1973. Because it will be necessary to completely cease Green Room operations for a period of at least three months in order to make the revisions, we intend that they be made following completion of the orders.

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As further indicated in the application, air cleaning consisting of high efficiency filtration systems and wet scrubber systems with appropriate provisions for future high efficiency filtration would be installed at the same time that the Green Room is shut down. We would also make provisions for continuous monitoring at that time. Although it might be possible to improve air cleaning on existing process equipment in the Green Room at an earlier time than proposed, it is highly preferable that all of the modifications be made at one time to avoid installing air cleaning that may be unusable with the new process equipment and to avoid another extended shut down.

Please let me know if you have any questions regarding the application.

Sincerely,

Peter Loysen, Manager Nuclear and Industrial Safety Ł

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EFFECTIVE PAGES

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SECTION 800 - CHEMICAL OPERATIONS

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SUBSECTION 820 - PROCESSING

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823. Green Room

823.1 System Description

The Green Room is designed for the following low enrichment processing: (1) Chemical recovery of scrap materials, (2) oxidation-reduction of UO_2 , (3) blending and packaging the product. The maximum enrichment of materials to be processed will not exceed 4.1%.

When not processing low enrichment materials, certain equipment may be used to calcine combustible material of large bulk-low uranium contact with any enrichment.

The Green Room location and equipment arrangement is shown on drawing D-509-2007. Simplified flow diagrams are shown on Figure 823.3-I, II and III.

823.2 Material to be Processed

2.1 Wet Recovery

Wet recovery operations may be performed on all types of scrap materials such as contaminated uranium compounds, rejected pellets, cleanup residues, etc.

2.2 Oxidation - Reduction

Oxidation-reduction may be performed on UO_2 pellets or other material for which wet recovery is not required.

2.3 Blending and Packaging -

The material may be in the form of ADU, UO2, U308, etc.

2.4 Calcining

Combustible material such as paper, absolute filters of large bulk, low uranium content may be processed.

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823.3 Process and Equipment Design

3.1 Muffle Box Loading and Unloading Hood, Muffle Boxes, Furnace & Cooler.

 UO_2 transferred into 240-3 for wet recovery or dry reprocessing will be transferred into trays in the Muffle Box Hood (5-8). Trays will then be loaded into a Muffle Box (5-7-1 through 5-7-6). A maximum of 18 trays will be loaded into one box and the box will be sealed. The boxes will be placed into 1 of 2 Furnaces (5-4) (5-5) for oxidation, oxidation-reduction, calcining or pyrohydrolysis. Upon completion of the furnace cycle, the boxes will be placed in the cooler (5-6) for cooling the boxes and contents with a water spray.

The cooled boxes will be moved to the Muffle Box Hood (5-8) for unloading from individual trays. In the case of clean scrap, the UO₂ will be screened through a screen and a magnetic separator located in hood (5-8). The sized product will be collected in a 5 gallon pail for blending on blender (5-12), sampling in hood (5-9), and weighing on scale (5-11).

UO2 that has been oxidized or calcined to U_3O_8 for wet recovery will also be unloaded in hood (5-8).

3.2 Feed Preparation

The U_3O_8 will be transferred to the feed preparation hood (1-10). All feed material for wet recovery will be processed through a screen and a magnetic separator prior to introduction of the U_3O_8 into the slurry feed hood (1-1-1).

3.3 Dissolution

A pre-determined weight of U_3O_8 from the feed preparation hood (1-10) is introduced into a 6" to 9"\$\$\$\$ vessel (1-1) via the slurry feed hood (1-1-1). Water will also be introduced into(1-1)from a 9"\$\$\$\$ water feed vessel (1-9-2). The U_3O_8 and water will be mixed by an agitator in the vessel. The resulting slurry will be pumped into one of three 9"\$\$\$\$\$ dissolvers (1-2) (1-3) (1-4) with external steam\$\$\$\$ water coils. Nitric acid will be added to the dissolvers from the 9"\$\$\$\$\$\$\$\$\$ HNO_3 feed vessel (1-9-3). The dissolvers will be steam heated and mixed by an agitator. After the U_3O_8 is dissolved, the impure UO_2 (NO_3)_2 solution, with a uranium concentration of 50 to 250 grams

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823.3 Process and Equipment Design (con't)

3.3 Dissolution (con't)

U/1., is allowed to digest to insure complete uranium dissolution. After digestion, steam may be introduced through the hot $UO_2 (NO_3)_2$ solution to remove the excess HNO₃. The excess HNO₃ may also be removed by boiling the $UO_2(NO_3)_2$ solution. After the excess nitrates have been removed, the $UO_2(NO_3)_2$ solution is cooled prior to acid insoluble filtration.

In the case of the low grade dissolver (1-4), the solids are allowed to settle to the bottom of the vessel. After settling, the supernatant $UO_2(NO_3)_2$ solution is removed from the top into dissolvers (1-2) or (1-3) for nitrate removal. After the $UO_2(NO_3)_2$ solution is removed, the solids are removed from the bottom of the vessel (1-4) via the acid insoluble discharge hood (1-4-1) and collected in a 5 gallon pail.

3.4 Acid Insoluble Filtration

The $U_{0_2}(NO_3)_2$ solution containing acid insolubles is pumped through an initial filter press (1-5) to remove the majority of solids. To improve the clarity of the $UO_2(NO_3)_2$ solution, a filter aid may also be introduced into the press or $UO_2(NO_3)_2$ solution from the 9"Ø filter-aid make up vessel (1-8). After the initial filtration, the $UO_2(NO_3)_2$ solution is pumped through a polish filter (1-6) to insure maximum clarity prior to pumping it into the $UO_2(NO_3)_2$ hold tank (1-7). A turbidmeter may be utilized as an overcheck to prevent solids from collecting in the hold tank (1-7). Details of the $UO_2(NO_3)_2$ hold tank (1-7) are shown on sketch 823.3 - IV. The acid insolubles will be removed from the filter presses(1-5) and (1-6)into a 5 gallon pail, for additional processing or disposition to burial.

3.5 <u>UO₂(NO₃)₂ Hold Tank and Dilution</u>

The 500 gallon $UO_2(NO_3)_2$ hold tank (1-7) is filled with Raschig rings.

The $UO_2(NO_3)_2$ solution will be recycled in the tank (1-7) to obtain homogeneity prior to sampling for uranium concentration, etc. After mixing and sampling, the $UO_2(NO_3)_2$ solution will be diluted NUCLEAR FLIELS CORPORATION

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823.3 Process and Equipment Design (con't)

3.5 UO2(NO3)2 Hold Tank and Dilution (con't)

with water from the 9" \emptyset water dilution feed vessel (2-2) to obtain the desired uranium concentration.

3.6 UO2 (NO3)2 Solution pH Adjustment

The diluted $UO_2(NO_3)_2$ solution is pumped into a 9"\$\overline\$ pH adjustment vessel (2-3) where the pH is adjusted with NH₄ OH from the NH₄OH make up (9-2). The NH₄OH make up vessel is a sealed tank with one (1) vent to atmosphere. The NH₄OH addition may be controlled by a continuous pH monitor and flow control system.

3.7 UO4 Precipitation and Aging

The U04 is precipitated in a 9"Ø vessel (2-4) with a hydrogen peroxide solution pumped from the hydrogen peroxide make up tank (9-3). The peroxide make up tank is a sealed tank with a vent to atmosphere. The resulting U04 slurry is recycled through another 9"Ø vessel (2-5) to ensure complete precipitation and to assure particle growth. After aging is complete, the overflow from (2-5)is collected in a 9"Ø surge vessel (2-6) for pumping to the centrifuge.

3.8 UO4 Separation

The UO_4 slurry is pumped from vessel (2-6) into a Sharples centrifuge (3-1) model No. P-600 or P-660. The UO_4 cake is discharged from the centrifuge into a furnace tray and placed into dryer (5-2) via the transfer hood (5-1). Details of the dryer are shown on Sketch 823.3-V. After the UO_4 is dry, it is removed from the oven and transferred to muffle box (5-7-1 through 5-7-6) via transfer hood (5-1). Boxes are sealed and placed in furnaces (5-4) (5-5) for reduction.

The centrifuge supernatant is discharged into a 9"Ø filtrate feed vessel (3-2) for final clarification in filter press (3-3). Solids from the press (3-3) are treated the same as the solids from the centrifuge (3-1). Details of the centrifuge are shown on Sketch .823.3-VI. The filtrate is pumped into filtrate hold tanks (4-1) (4-2) and mixed prior to sampling for uranium and discharge to the

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823.3 Process and Equipment Design (con't)

3.8 UO₄ <u>Separation(Con't)</u>

effluent holding pond. Details of the filtrate hold tanks are shown on Sketch 823.3-IV.

A turbidmeter may be used as an overcheck to insure that no solids are transferred into the hold tanks containing Raschig rings.

The alternate method of UO_4 separation will be utilization of the filter press (3-3) as the primary filter and the impure ADU polish filter (4-6) as the final filtration. Details of the filter press are shown on Sketch 823.3-VII.

3.9 Filtrate Treatment

In the event of filtrates containing recoverable quantities of uranium, the filtrate is pumped into a 9" \emptyset filtrate sparge vessel (4-3) with external steam/water coils. Steam is introduced into the heated filtrate to remove the hydrogen peroxide. After peroxide removal, the sparged filtrate is pumped into a 9" \emptyset filtrate cooling vessel (4-4) with external cooling coils prior to uranium precipitation in a 9" \emptyset ADU precipitation vessel (4-5). Uranium is precipitated with ammonium hydroxide made up from the vessel(9-2). The impure ADU Slur is pumped into press (3-3) for filtration. The impure ADU cake is removed from the press for additional processing or disposition to burial. The filtrate from the press (3-3) is pumped into a 9" \emptyset impure ADU polish filter (4-6) for final clarification before pumping into the filtrate hold tank (4-1) (4-2). The filtrate is mixed and sampled for uranium before discharge to the effluent holding pond.

3.10 UF₆ Cylinder Wash

The uranyl fluoride solution obtained from washing the heel from the low enriched cylinders are transferred into the ADU precipitation vessel (4-5) for uranium precipitation. The treatment of the uranyl fluoride solution is the same as described in Subpart 823.3.9.

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823.3 Process and Equipment Design (con't)

3.11 Analytical Control Hood

The analytical control hood (1-11) will be utilized for analytical control work. Typical work performed in hood will be centrifuging of samples, acid base titrations, preparation of liquid samples, and pH overchecks.

3.12 Utility Hood

The utility hood (1-12) will be used as a general purpose hood. The hood will have a 9'' vessel (1-12-1) for leeching filter media and a 3-1/2" safe slab vessel (1-12-2) for acid washing of empty trays.

3.13 Mill Hood

The mill hood (5-10) will be used to mill coarse powder and dewaxed pellets. Product will be collected in a 5 gallon pail.

3.14 Blending and Packaging

Milled and blended product is blended in a 5 gallon pail on the blender (5-12), weighed on scale (5-11), and sampled in hood (5-9). One pail at a time is handled in each of these locations.

Individual batches are cross-blended into a lot in the hood (10-0). The hood is a specially designed dust control hood. A maximum of 5 pails can be combined in a single operation in the following manner:

A single pail will be placed in the hood. After closing the hood, the lid will be removed and the pail placed on a four wheel dolly in a canted position. The dolly rides on rails along the back of the hood.

In front of the rail carrier will be located five stations, each separated from the next by a concrete slab one foot thick, approximately three feet high and two feet wide to isolate each station from the next from a neutron interaction standpoint. An empty five gallon pail will be positioned in each station.

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Process and Equipment Design (con't)

3.14 Blending and Packaging (con't)

This operation will be repeated until the entire batch is blended. The individual pails having received the incremental parts of the blend will be fitted with a lid, removed from the hood, and placed on a blender (5-12) and tumbled to accomplish blending. At this point, the pails will be weighed and transferred to storage.

3.15 Transfer

823.3

Furnace tray carts (11-1-1 and 11-1-2) will be used to transport filled trays to and from various process operations and for empty tray storage. The top of the cart will hold a maximum of nine (9) filled furnace trays stacked on 3" centers. The bottom of the cart will provide storage area for empty trays in a vertical position. Details of this cart are shown on Sketch 823.3-VIII.

Four (4) muffle box carts and a muffle box lift are provided for the transport of muffle boxes. These carts may be positioned near the furnaces or coolers during loading and unloading operations and placed in the north west portion of the area when awaiting further processing.

3.16 Storage

Inprocess storage racks are placed strategically throughout the storage area. These racks provide either three (3) or four (4) unit horizontal rows and two (2) tiers high. Three (3) foot center to center spacing is provided in both directions.

A storage area is provided for longer term and scrap storage. This area provides space for twenty units with the same separation as for the inprocess storage racks.

3.17 Utility and Support

Several utility or support tanks or vessels are located in the process area. These tanks or vessels are:

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	. 823.3	Proc	ess and Equipment Design (con't)
		3.17	Utility and Support (con't)
		•	 Ammonium Hydroxide Make Up (9-2) Hydrogen Peroxide Make Up (9-3) Deionized Water Storage (9-1), and Nitric Acid Bulk Storage (1-9-1)
	•		All of these vessels will be totally enclosed with one (1) vent to atmosphere.
	823.4	Nucl	ear Criticality Safety
			General
	•		Vessels, tanks and other equipment used to process solutions have nuclear safe geometries or are filled with Borosilicate Glass Raschig Rings. Process equipment used to process wet and dry powders and cakes have been analyzed by the use of computer codes to determine their degree of criticality under normal operating conditions and after one (1) administrative error. Storage containers have nuclear safe geometries and the interaction of arrays have been analyzed using the solid angle method.
	•		Several basic conditions apply throughout the area and are listed below:
			 Vessels which do not normally contain SNM will be separated from pipes or other vessels containing SNM by air breaks or they will have nuclear safe geometries or they will be filled with Borosilicate Glass Raschig Rings.
	· . ·		2) Pipe sizes and pipe intersections will meet the criteria listed on page 20, TID-7016, Rev. 1.
·	•		3) The use of Borosilicate Glass Raschig Rings will meet the standards and criteria set forth in the standard ANSI N 16.4 - 1971
		4.2	Safe_Geometry Vessels and Tanks
		•	Feed, mixing, dissolving and hold tanks or vessels have 9" inner diameters. A 9" inner diameter is nuclearly safe for enrichments

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823.4	Nuclear Criticality Safety (con't)
	4.2 <u>Safe Geometry Vessels and Tanks (con't</u>)
	not exceeding 4.1% as shown on Figure 309-XVI. Tanks and vessels with 9" or less inner diameters are:
	1) Slurry Make-up(1-1)
	2) Water Feed(1-9-2)
	3) Dissolver(1-2)
· ·	4) Dissolver(1-3)
	5) Low Grade Dissolver(1-4)
	6) Nitric Acid Feed(1-9-3)
	7) Filter Aid Make-up(1-8)
	8) Dilution Water Feed(2-2)
	9) PH Adjustment(2-3)
	10) UO ₄ Precipitator (2-4)
	11) $UO_4 Aging(2-5)$
ł.	4
1	13) Filtrate Feed(3-2)
	14) Filtrate Sparge(4-3)
	15) Filtrate Cooling(4-4) 16) ADU Broadsitation (4-5)
·	16) ADU Precipitation(4-5)
	17) Impure ADU Polish Filter(4-6) 18) Filter Leach Versel(1, 12, 1)
	18) Filter Leach Vessel(1-12-1)
•	The interaction effects are analyzed in the Nuclear Safety Evaluati Interaction Calculations.
· ·	
	A centrifuge is used for the separation of solids and supernatant.
	The centrifuge has a void volume not exceeding 24 liters which is
	` nuclearly safe for enrichments not exceeding 4.1% as shown on
	Figure 309-XVIII. The interaction effects are analyzed in the
· ·	Nuclear Safety Evaluation - Interaction Calculations.
	and a second
	Wet solids may be removed directly from the dissolvers and
	collected in 5-gallon pails (18.9 liters). This volume is nuclear
•	safe for enrichments not exceeding 4.1% as shown on Figure 309-XVII
	4.3 Poisoned Vessels and Tanks
	Large, unsafe tanks are filled with Borosilicate Glass Raschig Rings to provide nuclear safety. Poisoned tanks and vessels are:

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823.4	Nucl	ear Criticality Safety (con't)	· · · · · · · · · · · · · · · · · · ·
	4.3	Poisoned Vessels and Tanks (con't)	
		 UO₂(NO₃)₂ Hold Tank(1-7) Filtrate Hold Tank(4-1) Filtrate Hold Tank(4-2) 	
	4.4	Equipment Safe by Analysis	
		Solids may be collected in polish filters and The largest and most reactive filter or filter filter press 3-3. The maximum effective multi this filter press is 0.859 as shown in NED-175 largest and most reactive filter or filter pre- filters or filter presses are less reactive th Nuclear safe filters and filter presses are:	r press is the UO ₄ -ADU plication factor for 59. Since this is the ess, other smaller
	•	 Acid Insoluble Initial Filter(1-5) Acid Insoluble Polish Filter(1-6) UO₂-ADU Filter Press(3-3) Impure ADU Polish Filter(4-6) 	
		These filters and filter presses are separated equipment containing SNM by at least 2½ feet. relatively low effective multiplication factor distance, neutron interaction is considered ne evaluated further.	Thus with the r plus this separation
		Furnace trays have dimensions of 1½"x 16"x21" is nuclearly safe for enrichments not exceedin on Figure 309-XIX.	• The 1½" thickness ng 4.1% as shown
		Furnace trays are placed in a dryer to remove prior to pyrohydrolysis. The effective multip a loaded dryer under normal process conditions conditions where the material might be at opt complete water reflection, the maximum effect factor is 0.960. These results are shown in safe equipment used with these dryers are:	plication factor for s is 0.695. Under imum moderation and ive multiplication
	·.	1) Dryer(5-2) 2) Dryer(5-3)	
· · ·			
	·		
			•

いい日にし · Ć FUELS CORPORATION LICENSE: SNM-33, Docket: 70-36 10 of 11 Page 800 - CHEMICAL OPERATION SECTION: Subsection: 820 - Processing Approved 823 - Green Room 3-30-72 823.4 - Nuclear Criticality Safety Issued Supersedes New 823.4 Nuclear Criticality Safety (con't) 4.4 Equipment Safe by Analysis (con't) 3) Furnace Tray Cart(11-1-1) 4) Furnace Tray Cart(11-1-2) These dryers are separated from each other and other process equipment containing SNM by at least three (3) feet. Thus with the low effective multiplication factor plus this separation distance, neutron interaction is considered negligible and is not evaluated further. Dried furnace trays are next placed in muffle boxes for loading into furnaces for pyrohydrolysis. After pyrohydrolysis, the muffle boxes are transferred to a cooler for cool down. The effective multiplication factor for a loaded muffle box under normal process conditions is 0.650. Under conditions where the material might be at optimum moderation, the maximum effective multiplication factor is 0.958. These results are shown in NED-1759. Nuclear Safe equipment used with these muffle boxes are: 1) Furnace(5-4)2) Furnace(5-5)3) Cooler(5-6)4) Muffle Box Carts (4) 5) Muffle Box Lift Individual muffle boxes are separated from each other and other process equipment c ontaining SNM by at least two (2) feet. Thus, with the low effective multiplication factor plus this separation distance, neutron interaction is considered negligible and is not evaluated further. 4.5 Array Analysis The containers used for storage are 5-gallon pails (18.9 liters). This volume is nuclearly safe for enrichments not exceeding 4.1% as shown on Figure 309-XVIII. Arrays of 5-gallon pails will be used for both inprocess and long term storage. Interaction is analyzed in the Nuclear Safety Evaluation - Interaction Calculations.

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	New
823.4 Nuclear Criticality Safety (con't.)	
4.6 <u>Batch Control Safety</u> Material to be calcined will be low level wastes and scraps. This material contains sma metal or compound form with enrichments up to enrichment. Nuclear safety will be maintained cessing with a limit of 350 grams of U-235 per gram U-235 batch may be handled in each piece time. All other SNM will be removed from inid equipment before such transfer and processing limit of 350 grams of U-235 per batch will be of material balance, chemical analysis or othe techniques. A 350 gram batch limit is nuclear 309-I. Equipment used for this batch processing	11 amounts of SNM in and including full by using batch pro- batch. Only one 350 of equipment at any ividual pieces of may take place. The determined by means r appropriate analytical ly safe as shown on Table
1) Muffle Boxes (5-7-1 through 6)	
2) Muffle Box Loading & Unloading Hood (5-8)	
3) Furnace (5-4)	
4) Furnace (5-5)	
5) Cooler (5-6)	

- 6) Muffle Box Carts (4)
- 7) Muffle Box Lift

When each batch is processed, it will be removed from the Green Room for further processing or storage according to Subsection 810 or 820.

LICENSE:	SNM-33, Docket: 70-36	Page 1 of 1
SECTION:	800 - Chemical Operation	
SUBSECTION:	820 - Processing	Approved
SUBPART:	823 - Green Room	、 、
	823.5 - Health Physics	Issued 3/30/72
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823.5 Health Physics

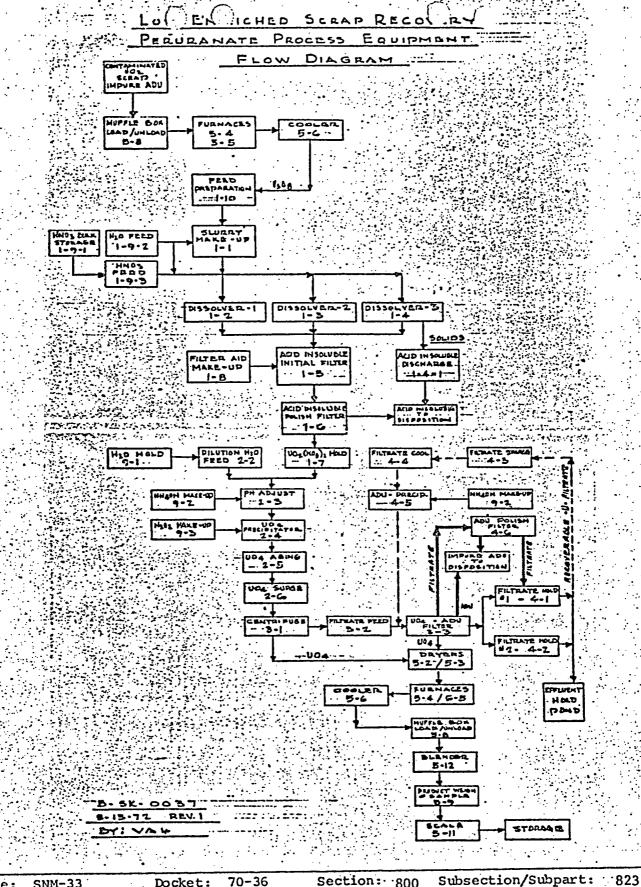
5.1 Ventilation

Ventilation control will be accomplished by installation of equipment that conforms to the criteria of Subpart 404.2. Cleaning of air and gaseous effluents will be achieved in accord with the criteria of Subpart 404.3. Sketch 823.5 - IX diagrams the general ventilation control and air cleaning systems serving this equipment.

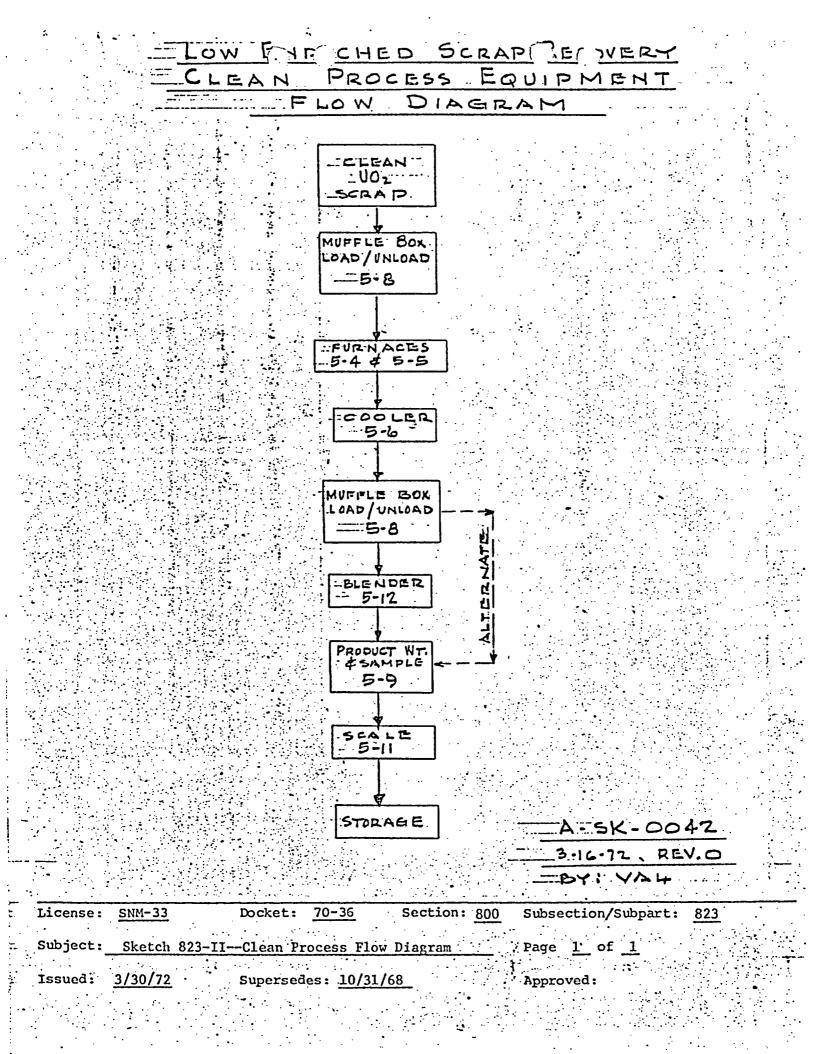
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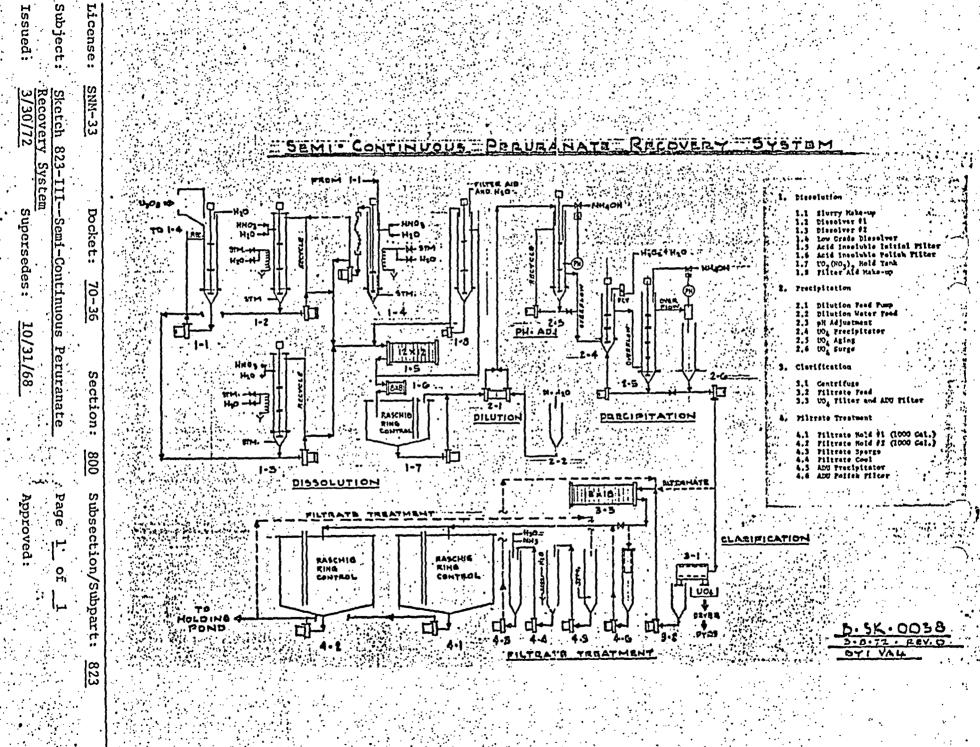
5.2 Liquid Effluents

Process liquid effluents are treated as described in Subpart 823.3.9 prior to discharge to the holding pond. Muffle box cooling water is discharged to the effluent pond, the output of which is monitored in accord with the requirements of Subpart 404.4.

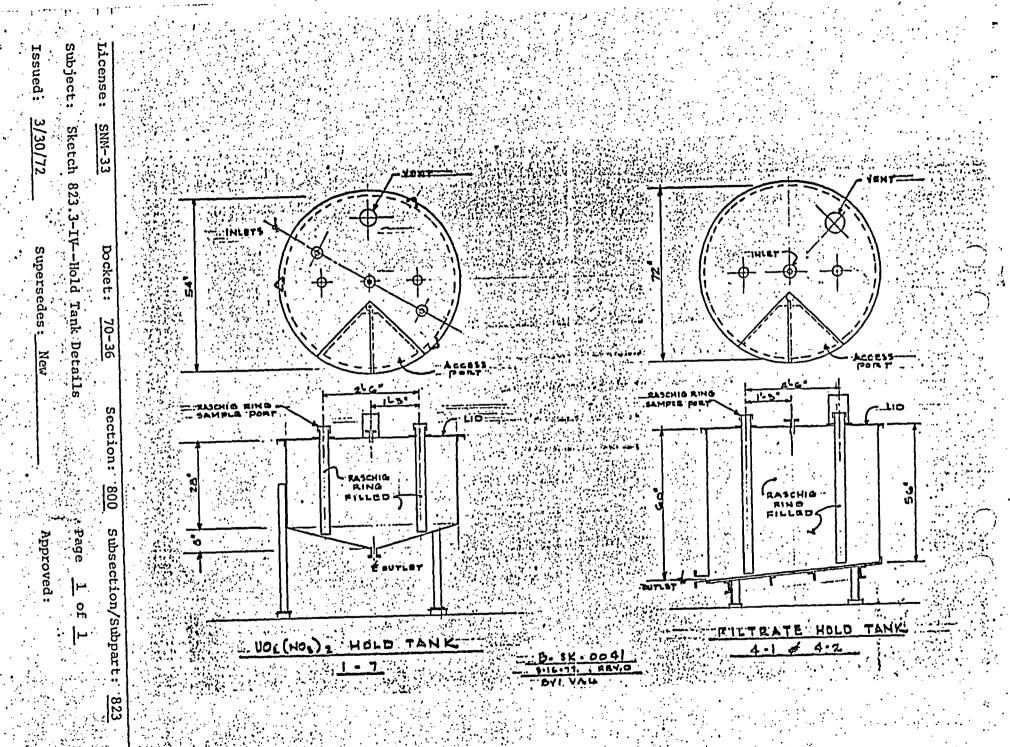


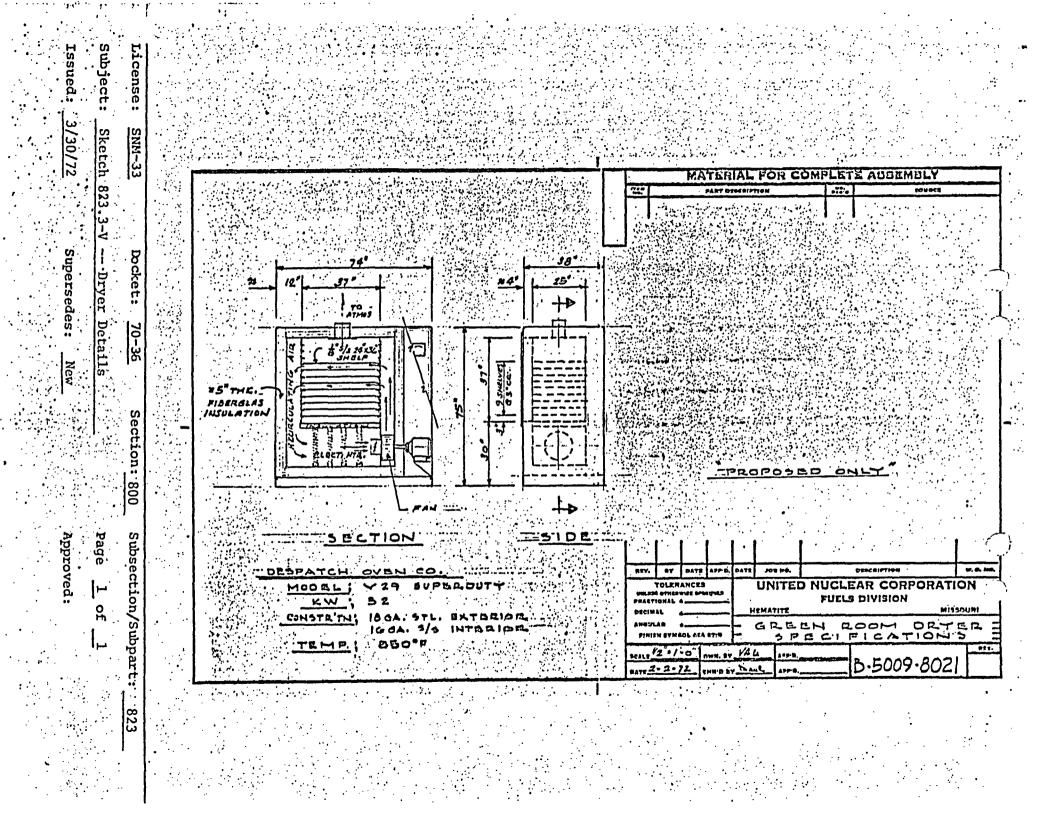
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Subject:	Sketch 823.3-I Perur	anate Process Flow	Page 1	of $\underline{1}$
	Diagram			
Issued	3/30/72 Supersed	les: <u>10/31/68</u>	Approved:	•

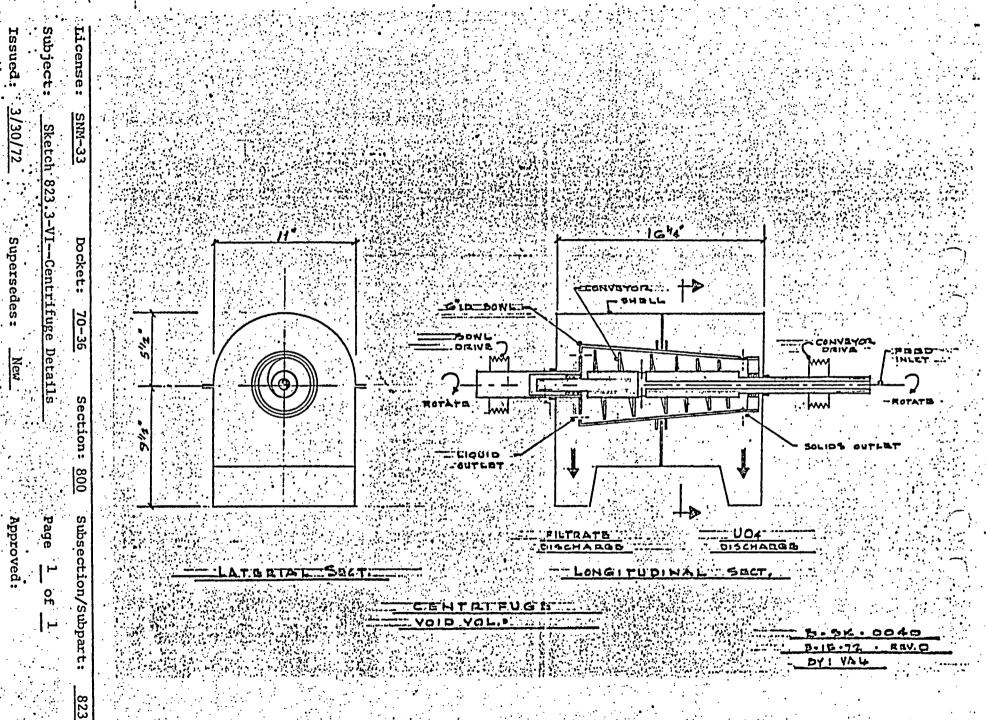


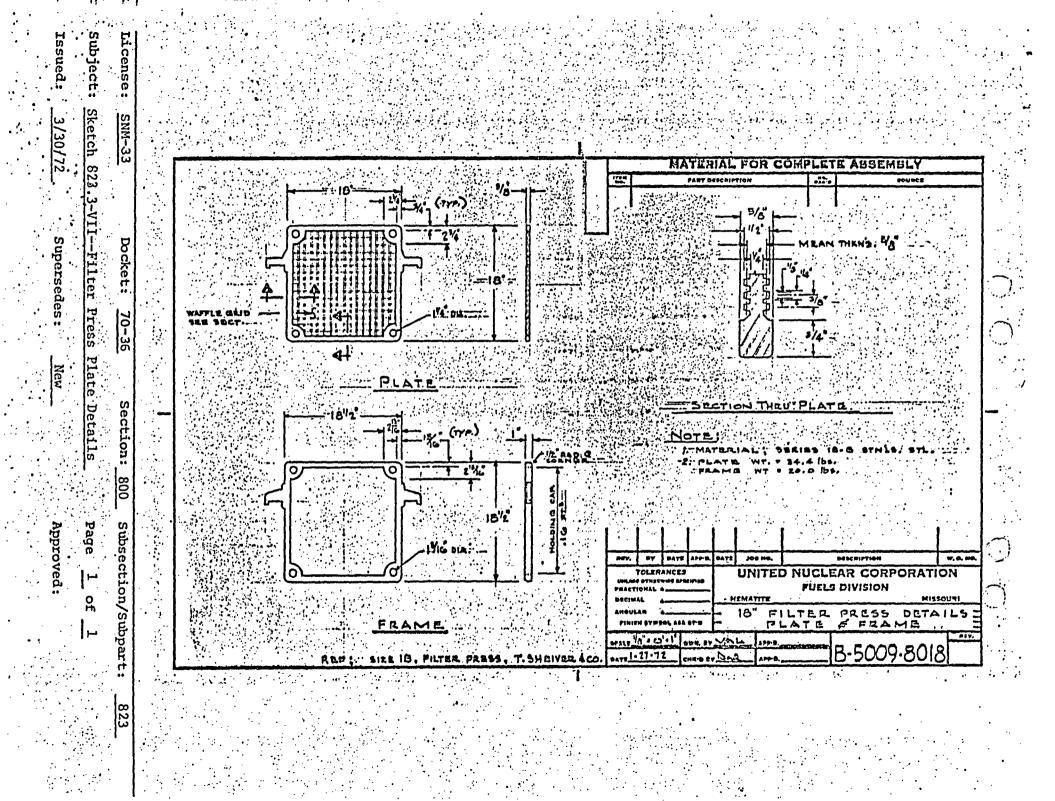


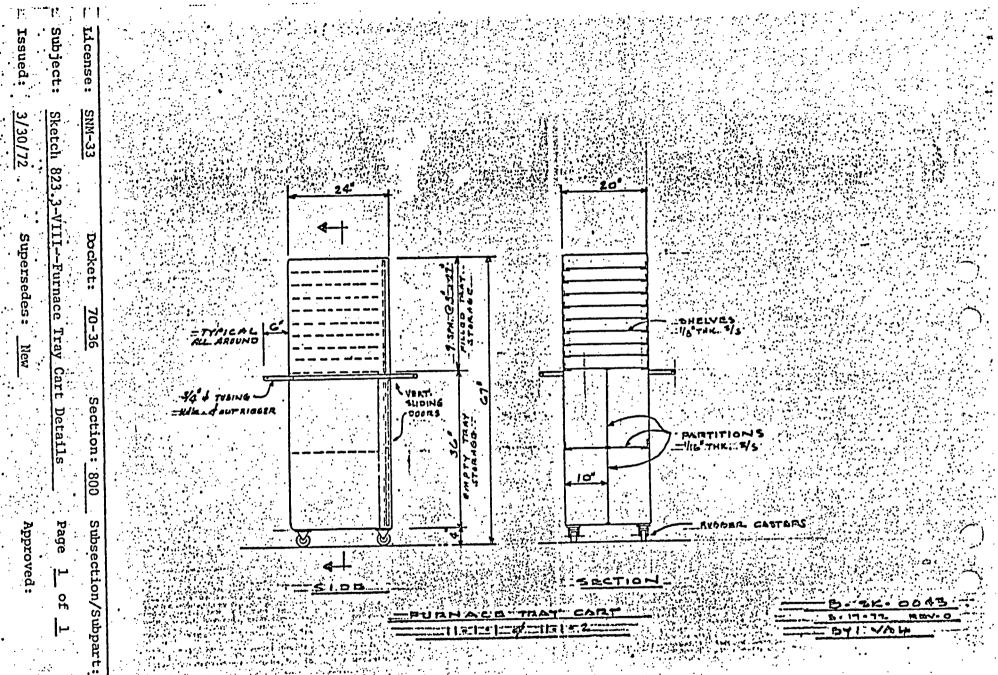
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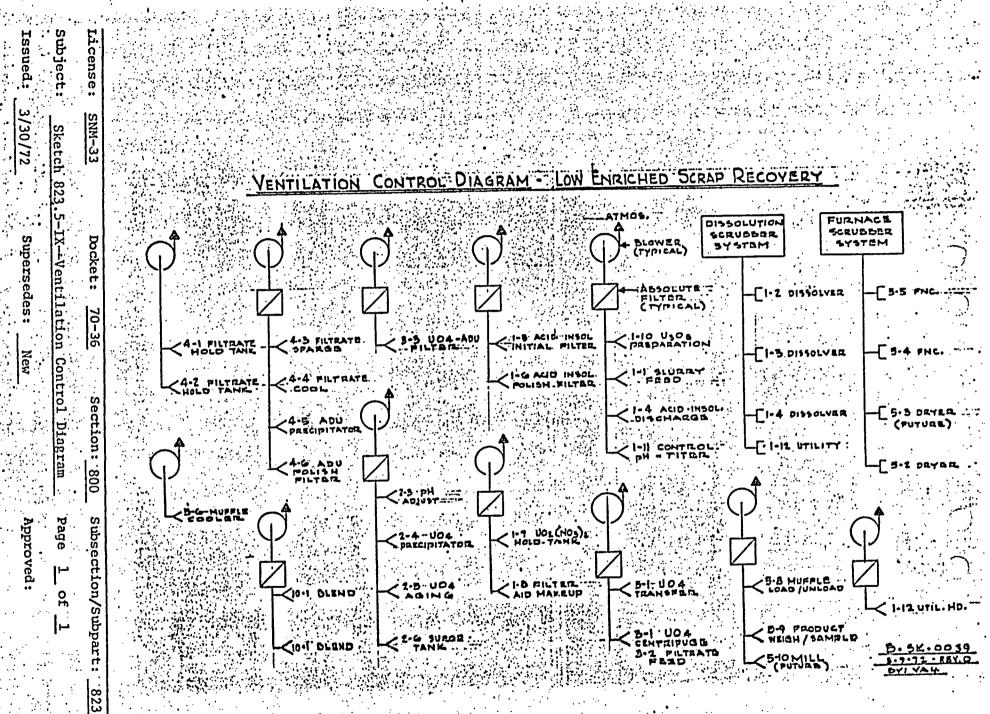








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DELEPTION (D)	•
As can be seen from Dwg. D- 509-2007, The content more	I unit in the dissolution
computer, would be the Low Grade Dissolver 1-4	•
II, INTERACTICAL CALCULATICAS	
A.Contribution from Slummer Make-up Vessel 1-1	
d= q =,75', L= 7', 4/2 = 3,5', h= 4,51	•
-2 A $= \frac{2}{5} \frac{d}{15} \sin \Theta = \frac{1.5}{4.5} (.614) = .157$	where $tw : 0 \cdot \frac{1/2}{n} = \frac{35}{4.5} - 178$ Sin $0 \cdot 1614$
B. Contribution from Discolurn 1-2	5460.1619
d. 9"= 75', LE7'. 42=3.5', H=6'	
$\Omega_{0} = \frac{2}{h} \sin C = \frac{15}{60} (.50) = .096$	Where tun 0 = 4/2 = 3.5 = .58
C. Centribution from Dissolver 1-3	sine=.so
d=9"=175', L=7', 1/2=3.5', h=9.25'	•
$0 = \frac{24}{24} = \frac{7}{2} = \frac{7}{2} = \frac{3}{5} = \frac{5}{14} = \frac{1}{14}$	L/2 3.5 - 770
$\Omega_{\mathcal{L}} = \frac{2c!}{h} \sin \Theta \cdot \frac{1.5}{4.25} (.353) \cdot .044$	where $tim 0 = \frac{L/2}{h} = \frac{3.5}{9.25} = .378$ sing = .353
D. Contribution from Acid Insoluble Initial Filter	· 1-5
d = 18'' = 15', L = 4', L/2 = 7.', h = 8.25'	
$A_{D} = \frac{2}{h} \sin \Theta = \frac{3}{5125} (.236) = .086$	where tem G = 4/2 = 2.
	3m G = ,2%
E. Contribution from Acid Inscluble Polish Filten 1-	6
d = 175', L= 1', 1/25', h= 14.25'	
$A_E = \frac{2c!}{h} \sin \Theta = \frac{1.5}{1425} (.35) \cdot .037$	where tem 6. 4 = 14.25 = , 3;
1425 (1710)	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
F. Contribution from Utility Hovel Vousel 1-12-1	
d=1"=,75", L=7", L/2 = 3.5, h= 20.75	
$A_{F} = \frac{2e!}{h} 5m G = \frac{1.5}{20.15} (.166) = .013$	where time + 12 = 3.5 = 16
	Sm @ = 166

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. _{Дт}	Jense:	<u> 3000-33</u>	Docket: <u>70-36</u>	Section: 800_	Subsection/Subpart: <u>823.4</u>	
Sul	pject:	Nuclear Safety	Evaluation - Interac	tion .	Page 1 of 4	•
		Calculations -	Low Grade Dissolver	1-4		÷
Iss		3/30/72	Supersedes: <u>New</u>	î	Approved:	•
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License: <u>SNM-33</u> Docket: <u>70-36</u> Section: <u>800</u> Subsection/Subpart: <u>823.4</u> Subject: <u>Nuclear Safety Evaluation - Interaction</u> Page <u>2</u> of <u>.4</u> <u>Calculations Low Grade Dissolver 1-4</u> Issued: <u>3/30/72</u> Supersedes: <u>New</u> Approved: II. INFERNCTION CALCULATIONS (continued) N. Contribution from Fir...ke Cool Vessel 4-4 de 175', $L \cong 7'$, 4/2 = 3.5', h = 19.5' $\Omega_N = \frac{2d}{h} \sin \Theta = \frac{1.5}{19.5} (.177) = .014$

0. Contribution from ADU Precipitation 4-5 $d_{2,75}$ ', L'=7', $L'_{2}=3.5$ ', h=23.25' $Q_{10}=\frac{2d}{h}\sin \Theta=\frac{1.5}{23.25}(.147)=.009$

P. Contribution from Filtrate Hold Trink 4-2 d = 7a'' = 6', L = 60'' = 5', L/2 = 2.5', h = 22.5' $\Delta p = \frac{2el}{h} \sin \Theta = \frac{12}{22.5} (.11) = .0.59$

Q. Centribution from Filtrete Hold Tende 4-1 $d = 6', L = 5', L'_2 = 2.5', h = 25'$ $\Delta_q = \frac{24}{h} \sin \theta = \frac{12}{25} (25B) = ,038$

R Contribution from 404-AD4 Filter 3=3 $d = 18^{+} = 1.5^{+}, L = diagonal = 21^{-1}, L/2=1^{+}, h = 15^{+}$ $\Omega_{R} = 2\frac{1}{2} = \frac{3}{18} (.055) = .009$ where ten $\Theta = \frac{1}{18} = \frac{1}{18} = .056$ $Sin \Theta = .055$

where tune = 1/2 = 3.5 = , 18

Whare ten 6 = 1/2 - 35 = ,149

where tun 6 = 4/2 = 2.5 = . 111

Where true + 4/2 = 2.5 = ,081

5m @ = 147

· Si O + 11

5m0 = 1008

sind - , 177

5. Centribution from Muffle Box at Trime for Horsel 5-1. $d = 21'' \cdot 175'$, L. diagonal 2' 5', L/2 = 2.5', h = 19.' $\Omega_s = \frac{2d}{h} \sin \Theta = \frac{3.5}{19} (.13) = 1024$ $\sin \Theta = .13$

T. Contribution from Furshace Tray Cant 1-11 in Aisb $d = 24'' \cdot 2', L = 20'', \frac{1}{2} = 10'' = .83', h = 10'$

 $\Omega_{1} = \frac{2}{h} \sin \Theta = \frac{4}{10} (.013) = \frac{033}{033}$ where $\tan \Theta = \frac{1/2}{h} = \frac{1/2}{10} = .083$

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 Calculations - Low Grade Dissolver 1-4
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II, INTERACTION CALCULATIONS (continued) U. Contribution from Sunnounding Equipment not Specifically Analyzed Din = .5 (assumed)

Vo Total Interaction

- Ay = 2 - DA + 1 + - 2 4 = 1.352 steradiums

III. K. EFF of Low Grade Dissolven 1-4

The Low Grade Dissover 1-4 hos the follow dimensions N= 4.5" = 11.4 cm, L = 84" = 213 cm

Using Eqn. (7), NDEO-1050, the second tric buckling is $B_{q}^{2} = \left(\frac{2.405}{1+6}\right)^{2} + \left(\frac{TT}{L+2\delta}\right)^{2} = .0281 \text{ cm}^{-2}$

Where SEbarel = 3 cm, DP-532

The infinite multiplication factor and migration area at optimum moderation (V420/V402 = 3.75) are.

km = 1.4., Fig.1, NEDO-1137

M² ¥ 35cm Fib 2, Neno - 1137

Using Eqn. (3), NDEO-1050, The effective multiplication factor is

Keff - 1+M3 Bg2 - +706

TV ALLOWABLE INTERACTION

The allowable interaction 15

ΩA = 9-10.K = 1.94 stenaelians

V. CONCLUSIONS

Since the total interaction is less them the allowable interaction; the equipment arnangement is miclearly safe,

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 Calculations - Low Grade Dissolver 1-4
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- I. DESCRIPTION
- As can be seen from Dwg. D-509-2007, the center most unit in the U04-Filtrate complex would be the U04 Aging Vessel 2-5
- I. INTERACTION CALCULATIONS
 - A. Contribution from UO4 Precipitator 2.4 d= 9" = .75', L ≅ 7', 4/2 = 3.5', h= 3.5'
 - $-\Omega_{A} = \frac{2c}{h} \sin \theta = \frac{1.5}{3.5} (.707) = ,303$
 - where $ten \theta = \frac{42}{11} = \frac{3.5}{3.5} = 1$ Sin $\theta = .707$ B. Contribution from 104 Sunge Vessel
 - ΩB = QA = .303
 - C. Contribution from Filtrate Springe Vessel 4-3 $d = .75', L \cong 7', \frac{1}{2} = 3.5', h = 9'$ $\Omega_{c} = \frac{2d}{h} \sin \Theta = \frac{1.5}{9} (.39!) = .065$
 - whone tun (= 3,5 = = . 39 $\sin \Theta = 36$
 - D. Contribution from Filtrite Cool Vossel 4-4 d=175', L=7', 42=3.5', h= 6' when $frightarrow \Theta = \frac{325}{6} = 158$ sin $\Theta = 150$ $\Omega_{D} = \frac{1}{h} \sin G = \frac{1.5}{6} (.50) = .125$
 - E. Contribution from ADA Precipitation 4-5
 - d = . 75', L= 7', 4/2 = 3.5', h= 4.5'
 - Whence $trim \Theta = \frac{3.5}{7.5} = 178$. Sim $\Theta = 161$ $\Omega_E = \frac{2d}{h} \sin \Theta \cdot \frac{15}{45} (.61) = .203$
 - F. Cuntribution from Impure ADA Polish Filter 4-6 d=75', L= 7', 42 . 3.5', h= 6.5' whenc $tz_{11}\Theta = \frac{3.5}{0.5} = 1.54$ $\sin \Theta = 1.48$ $\Omega_{F} = \frac{24}{5} \sin 6 = \frac{15}{65} (148) = 111$
- 61 Contribution from Filtrate Hold Tomk 4-1 d= 72'26', L=60" 5', 4/2= 25', h= 11.5' AG = 2d sine = 12 (1213) = 221-
 - When $c \quad tun \Theta = \frac{1/2}{h} = \frac{2.5}{1k^2} = \frac{1}{2}$ Sin $\Theta = \frac{1}{2}$

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Subject • Nuclear	<u> Safety Evaluation - Inter</u>	raction	Page <u>1</u> of <u>2</u>
Calculat	ions - UO4 Aging Vessel	2-5	
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II. INFERACTION CALCULATIONS (continued):

H. Centinbution From Filtrate Hold Tank 4-2

DH = DG = .271 I. Contribution from Filtrate Feed Vessel 3.2

d=.75',:2=7', 4/2= 3.5', h= 8.25'

 $\Pi_{f} = \frac{2d}{h} \cdot 5m \Theta \cdot \frac{1.5}{5.25} \left(i39 \right) = :072 \qquad \text{where } fm \Theta = \frac{1/2}{h} = \frac{3.5}{8.25} = .424$ Sin $\Theta = .39$

J. Contribution from UO4-AD4 Filter Prices 3-3

d= 18"= 1.5', L= 4.25', 4/2= 2.13', h= 7'

 $\Omega_{3} = \frac{2d}{h} \sin \Theta_{2} = \frac{30}{7} (.290) = .124$ where $\tan \Theta = \frac{1/2}{h} = \frac{2.13}{7} = .304$ sin $\Theta < .240$

K. Contribution from Surrounding Equipment.

All other equipment is greater them 10' Gway and partially shielded, on is completly shielded. Therefore, Their contribution is neglected

"L. Total Interaction

 $\Lambda_{L} = 2 \Omega_{\rho} + \cdots + \Omega_{K} = 1.740$ sterediums

II. K-EFF OF- WOW AGING VESSEL 2-5

- the k-eff is the same as for the Low Grade Dissolver 1-4 since they are the Sume size and type vessels

keff =..707

IV; ALLOWABLE TNTERACTION

The allowable interaction is

QA = 9-10 K = 1.94 steradions

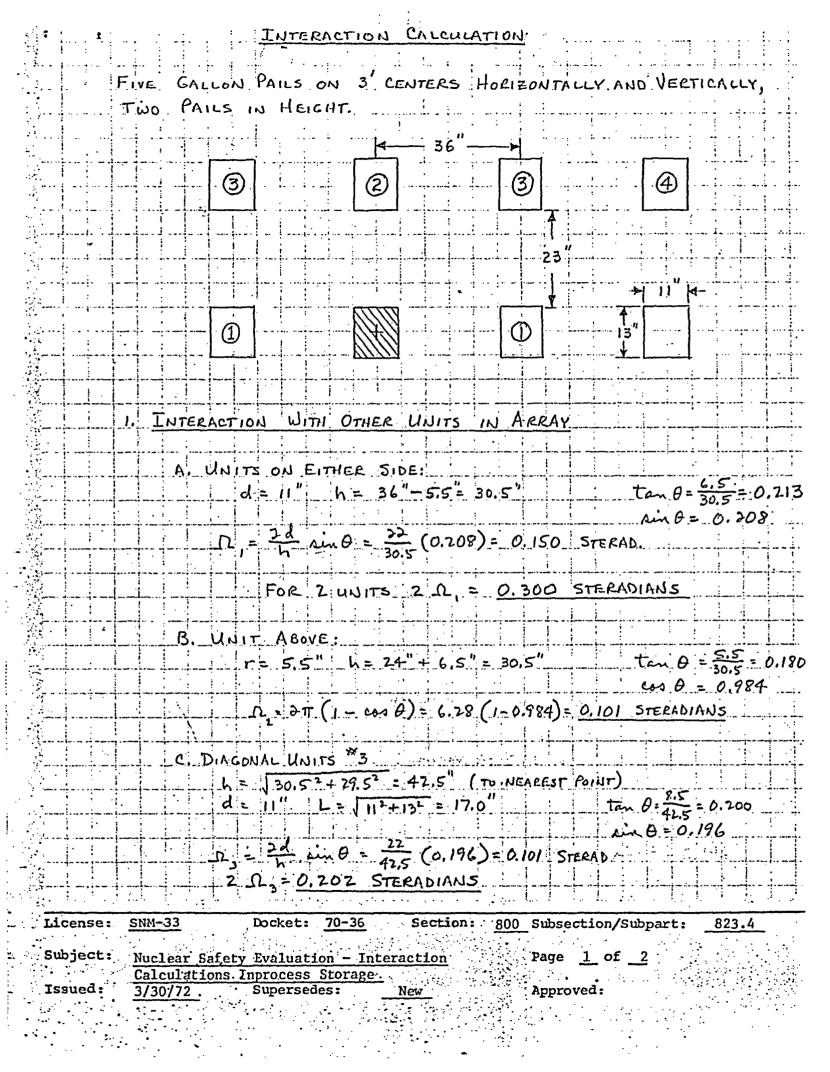
V, CONCLUSIONS

Since the botal interaction is less than the allowable interactions

the equipment arrangement is mucheauly safer

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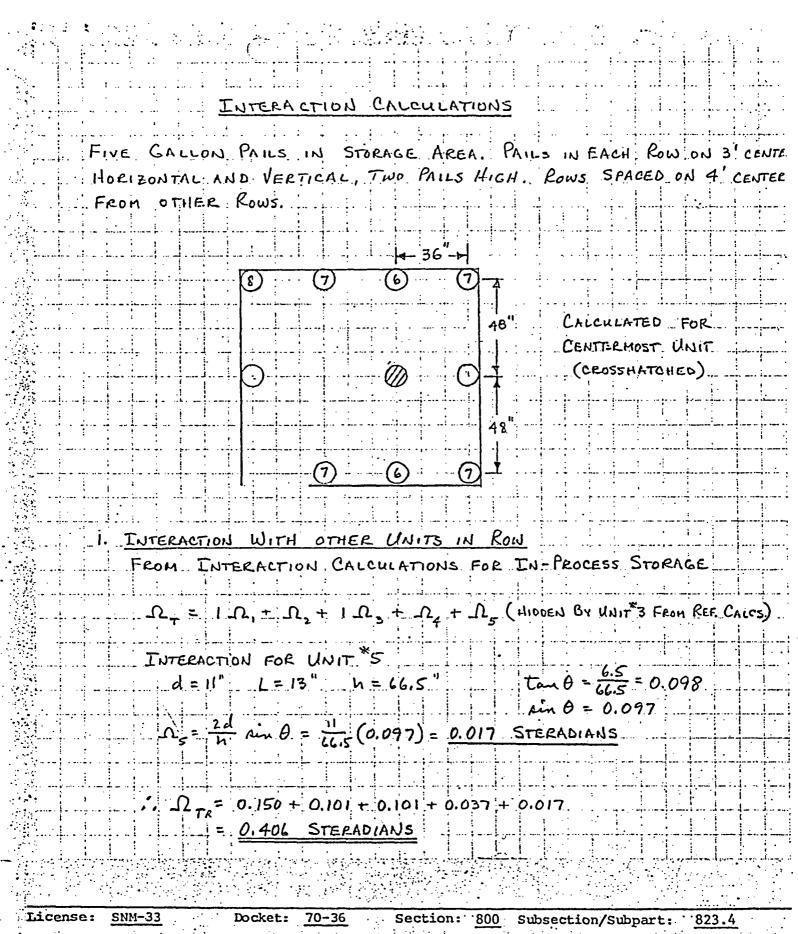


D. DiAconne UNIT #4
h =
$$\sqrt{795^{-4} 66.5^{+}} = 70.5^{-6}$$

d = 11" L = 17.0"
 $a_{1} + \frac{24}{16} a_{101} d = \frac{12}{70.5} (0.117)$
 $a_{1} + \frac{24}{16} a_{101} d = \frac{22}{70.5} (0.117)$
 $a_{2} + \frac{24}{16} a_{2} + 2A_{2} + A_{4} = 0.300 + 0.101 + 0.302 + 0.027$
 $a_{2} + \frac{24}{16} (0.157 + 0.101 + 0.102 + 0.027)$
 $a_{2} + \frac{24}{16} (0.157 + 0.101 + 0.102 + 0.027)$
 $a_{2} + \frac{24}{16} (0.157 + 0.101 + 0.102 + 0.027)$
 $a_{2} + \frac{24}{16} (0.157 + 0.101 + 0.102 + 0.027)$
 $a_{2} + \frac{24}{16} (0.157 + 0.101 + 0.028)$
 $a_{2} + \frac{24}{17} (0.157 + 0.101 + 0.028)$
 $a_{3} + \frac{24}{17} (0.157 + 0.101 + 0.028)$
 $a_{3} + \frac{24}{17} (0.157 + 0.028)$
 $a_{3} + \frac{24}{17} (0.157 + 0.028)$
 $a_{3} + \frac{24}{17} (0.128 + 0.028)$
 $a_{3} + \frac{24}{17} (0.107 + 0.0005 + 0.0005 + 0.0005 + 0.0015$

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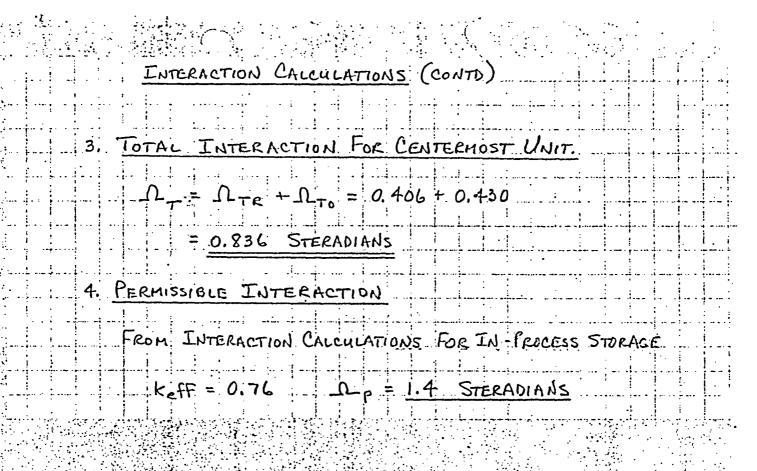
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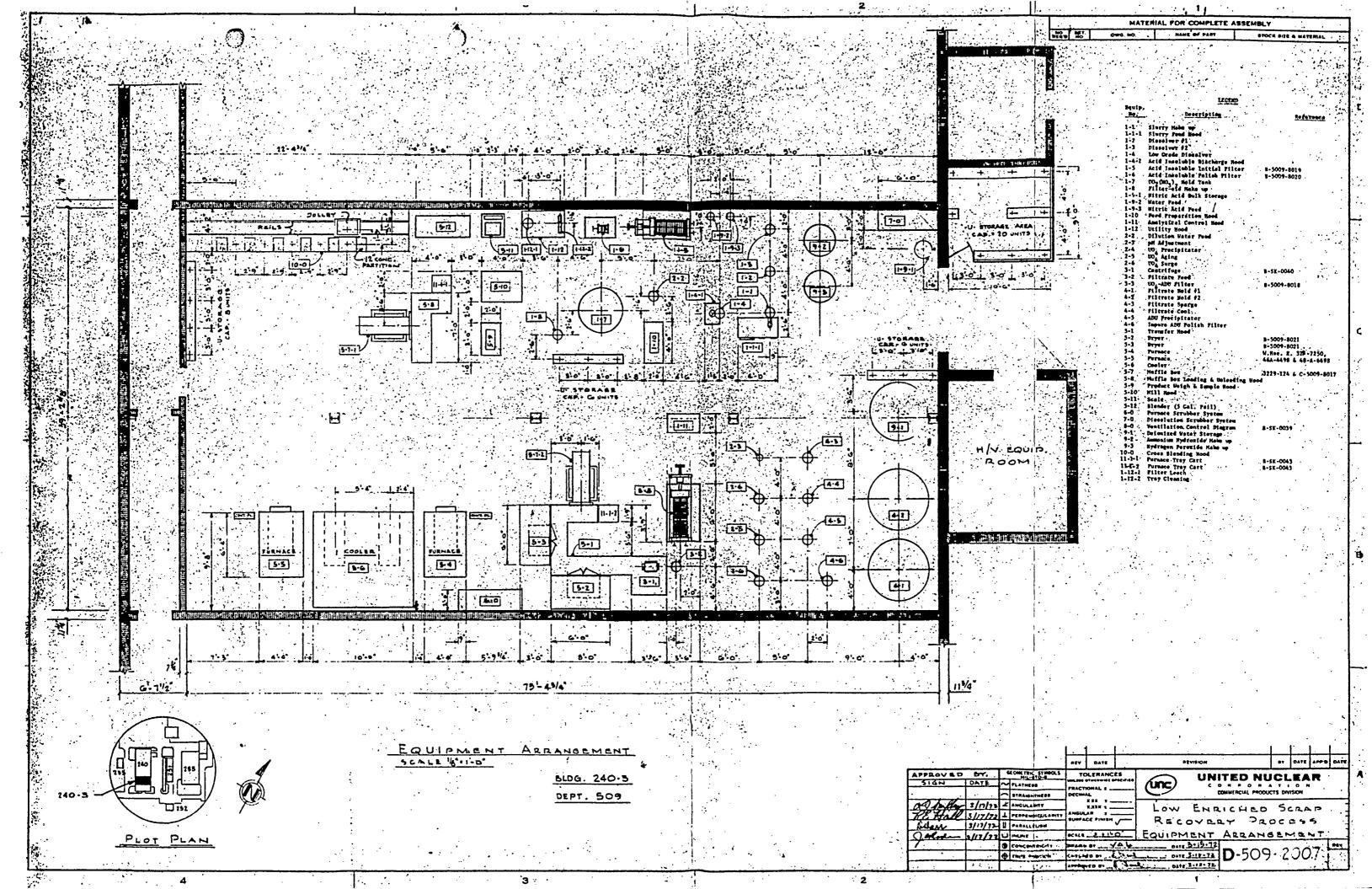
Subject:Nuclear Safety Evaluation - InteractionPage 1 of 3Calculations.Green Room Storage AreaIssued:3/30/72Approved:

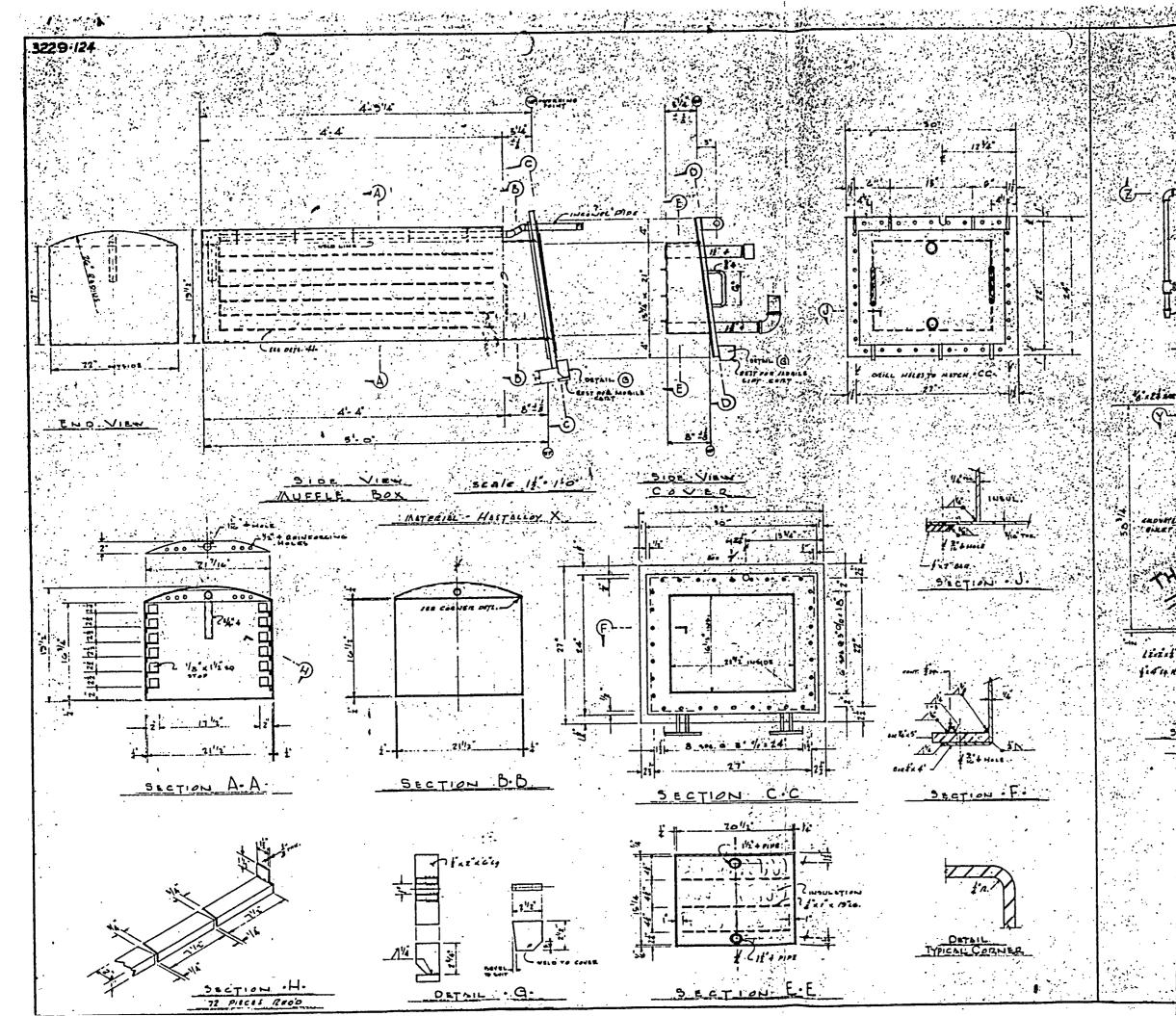
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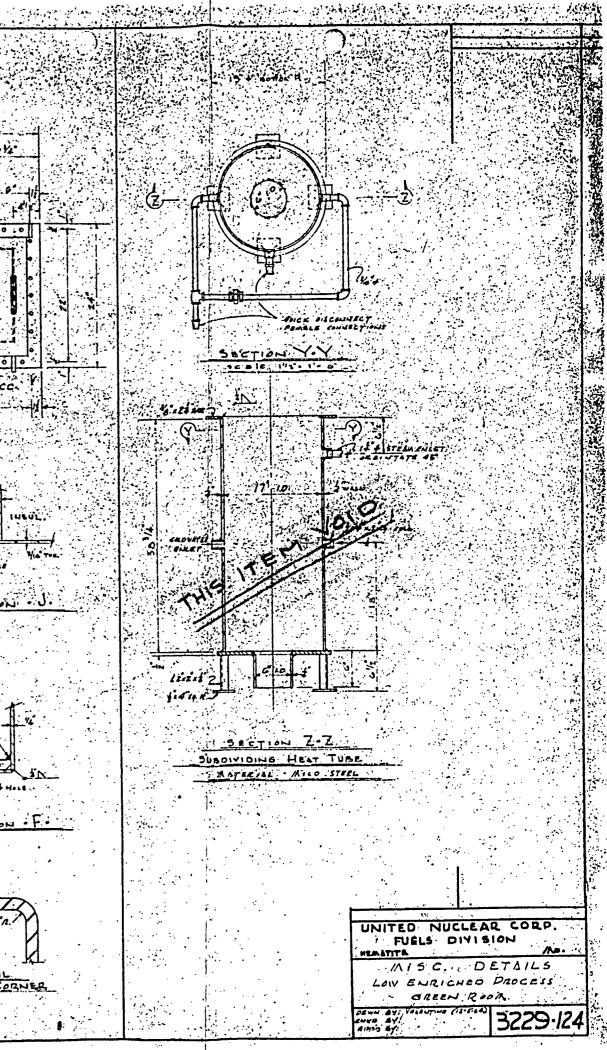
	e and the set of the set
	INTEL TION CALCULATIONS (CON.)
	2. INTERACTION WITH UNITS IN OPPOSING ROWS
	~ 같은 물건을 잘 안 하는 것 같은 것이라는 것 같은 것 같
	FOR SIMPLICITY OF CALCULATION, UNITS IN OPPOSING ROWS ARE
	CONSIDERED AS A SINGLE CYLINDER OF HEIGHT THE SUM OF THE
	HEIGHT OF A DRUM PLUS THE DIAGONAL DIMENSION OF THE SECOND
	DRUM. SEPARATION WILL BE THE CLOSEST STRAIGHT LINE DISTANCE BETWEEN THE UNITS.
	THE UNITS.
	A. UNITS ×6
	$h = (48 - 5.5) = 47.5'' d = 11'' Tan \theta = \frac{15}{42.5} = 0.353$
	$L = 13.0" + 17.0" = 30"$ $\Delta in \theta = 0.333$
	$\Omega_{L} = \frac{2d}{h} \operatorname{Ain} B = \frac{11}{42.5} (0.333) = 0.086 \text{ STERAD}$
	$2 \Omega_6 = 0.172$ STERADIANS
	B. UNITS #7
	$h = \sqrt{42.5^2 + 30.5^2} = 52.3'' d = 11'' L = 30'' tan \theta = \frac{15}{52.3} = 0.287$
	2d
	$\Omega_{\eta} = \frac{2d}{n} \sin \theta = \frac{11}{52.3} (0.276) = 0.058$
	$4-\Omega_{\gamma} = 0.232$ STERADIANS
	C. UNIT * 8
	15
	$h = \sqrt{47.5^2 + 66.5^2} = 79.0^{-1} d = 11^{-1} L = 30^{-1} \tan \theta = -79 = 0.190$ Ain $\theta = 0.187$
	$\Omega_{g} = \frac{2d}{h} \operatorname{Ain} \Theta = \frac{11}{79} (0, 187) = 0.026 \text{ STERADIANS}$
	79 (Crier) - Core Sickedins
	D. TOTAL INTERACTION WITH OPPOSING ROWS
	$\Omega_{+0} = 2 \Omega_{+} + 4 \Omega_{+} + \Omega_{+} = 0.172 + 0.732 + 0.026$
	$\Omega_{\tau_0} = 2 \Omega_{c} + 4 \Omega_{r} + \Omega_{s} = 0.172 + 0.732 + 0.026$ = <u>0.430</u> STERADIANS
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INTER-OFFICE MEMO

ro R. Kropp

GULFUNIT

NUCLEAR FUELS CORPORATION

NED-1759 DATE March 31, 1972

FROM E. Fass

COPY TO

SUBJECT Criticality Safety Analysis of the Gulf United Green Room Process Equipment (7801-012)

1.0 Summary and Conclusions

An analysis has been performed to determine the criticality safety of three pieces of process equipment in the Green Room at the Gulf United Hematite plant. The three pieces of equipment examined were: 1) the 18" filter press, 2) the proposed drying ovens, and 3) the muffle boxes and furnace. The uranium enrichment in each step in this process was fixed at a maximum 5 w/o U^{235} .

The analyses of the filter press and drying ovens considered a full swing of water content in the uranium product. For both these pieces of equipment the calculated k_{eff} was less than 0.900 and are therefore considered safe.

For normal operations with the muffle boxes and furnace, the calculated effective multiplication factors were less than 0.900 and therefore these pieces of equipment are safe. The muffle box process was also examined for varying amounts of excess water in the product as was done in the case of the filter press and drying ovens. The maximum multiplication of the muffle boxes was 0.958 \pm 0.011 and the furnace was 0.955 \pm 0.012, at the 95% confidence interval. Although these calculated values of keff are greater than 0.900 the process is considered safe from a criticality accident because of the confidence in the computing technique used and the high degree of conservatism used in modeling the problems. For example, a more realistic representation of the U²³⁸ resonance integral used in the muffle box analysis at the peak in its reactivity curve reduced the maximum keff from 0.958 \pm 0.011 to 0.914 \pm 0.011.

2.0 Description of the Process Equipment

The filter press consists of a linear array of alternating stainless steel plates and frames. (1) The plates are 18" square in cross section and 5/8" thick. The face on each side of the plate is a recessed "waffle" grid 16 1/2" square with a mean thickness of 3/8". The frame has an outer square dimension of 18 1/2" and a 1" thickness, the inner section of the frame is a void of 16 5/8" square dimension. The plates and frames can be stacked alternately until a maximum linear dimension of 40" is reached which is the limit of the 18" filter press. The array is secured at both ends by steel plates of an estimated mean thickness of 2". The wet uranium product is collected in the frames between opposing faces of the neighboring plates.

When operations with the filter press are complete, the frames and plates are separated and the wet product is put into trays(2) of either 18 gage stainless steel or Inconel. The inside dimensions of the trays are 21"x 16'x 1 1/2".

A maximum of 18 such trays are loaded into a Despatch model V29 drying oven(3) to drive off free water from the product. The trays are loaded onto shelves of 1/8" stainless steel, two to a shelf. It is not physically possible to stack two trays vertically on any shelf other than the top. The vertical spacing of the shelves is 3" and nine shelves are used to provide the space for 18 trays. This inner portion of the oven is constructed of a $37" \times 39" \times 25"$ 16 gage steel shell. This is contained within an outer 18 gage shell with overall dimensions of $74" \times 75" \times 38"$, the inner surface of this outer shell is lined with 5" thick fiberglas insulation.

After the product is dried in the ovens, the trays are removed and placed in the muffle boxes(4). Each box can hold . 3 trays in line on each of 7 shelves giving a maximum of 21 trays. The vertical spacing between shelves is approximately 2 1/2" and the outer shell of the box is made of 1/4" thick Hastelloy X. The overall outer dimensions of the muffle box are 60" x 19 1/2" x 22". The boxes are supported on stands which keep the boxes 42" above the floor.

To reduce the product to its final chemical form, UO2, a single box is loaded into the furnace(5) and fired. In the furnace the box is exposed to heating coils and surrounded by insulating and supporting brick. The neutron reflective properties of the furnace walls were treated in a conservative approach by replacing the walls with tight fitting water reflectors around the muffle box.

3.0 Method of Analysis

Reactivity levels for the filter press were calculated using the DTF(6) code. The 16 group Hansen-Roach cross sections (7) were used and linear anisotropic hydrogen scattering was accounted for in the S4 approximation to the transport equation. The calculational model assumed infinite slab geometry of homogeneous mixtures of materials with the finite square cross section accounted for by an appropriate input buckling.

Reactivity levels for the most reactive loadings of the filter press and for all cases studied for the drying ovens and muffle boxes were calculated using the KENO (8) multi-group Monte Carlo criticality code. Again, the 16 group Hansen-Roach cross sections were used with P_1 linear anisotropic hydrogen scattering.

These codes have been widely used to determine criticality of similar individual unit and array type systems. References 9 and 10 contain thorough discussions on the accuracy of the methods.

3.1 The 18" Filter Press

There are two kinds of product which may be processed in the filter press as well as in the other process equipment. One is $UO_4 \cdot 2H_20$ with a maximum theoretical density of 4.66 gm/cm³() and the other is ammonium diuranate (ADU) whose classic anhydride formula is (NH₄)₂U₂O₇ with a maximum theoretical density estimated at 3.0 gm/cm³(1).

To determine the most reactive condition in the press, a series of DTF calculations were performed assuming that either product may be found in any amount in the filter press with the remaining volume being occupied by water. The results of the DTF calculations are given in Table 1 and Figure 1. Since the UO4.2H₂O has a higher theoretical density than ADU, and hence a greater U^{235} concentration for a given H/U^{235} ratio, it shows a higher reactivity curve than the corresponding one for ADU. For this reason, further calculations of the filter press and other process equipment was restricted to the UO4.2H₂O product. It may be noted on the curves in Figure 1 that some of the calculated data points lie off the smooth curves. This deviation is the result of changing U^{238} cross section sets which cannot be done as a continuous function of $\sum_{s}^{epi}/N_{U}^{238}$

with the Hansan Roach library but only as a step-wise function. The maximum multiplication constant of 0.915 occured for $UO_4 \cdot 2H_2O$ with 30 volume percent water, $H/U^{235} = 150$.

KENO calculations were performed for this loading of the filter press. With the three dimensional geometry, it was possible to treat the finite extent of the press in the KENO input. The calculated k_{eff} after 14,100 neutron histories was 0.845 + 0.010.

A KENO calculation was also performed for this problem with a thick water reflector placed around the outside of the filter press. The KENO calculated k_{eff} was 0.859 \pm 0.010. KENO results for this problem, and for the other pieces of process equipment are presented in Table 2.

3.2 The Drying Ovens

The wet product (assumed to be only $UO_4 \cdot 2H_2O$ because of its higher reactivity for a given volume) is placed into trays and loaded into the ovens. The trays were assumed to be stainless steel since this material has a smaller neutron absorption cross section than Inconel. The following conservative assumptions were made regarding reflective properties (1) The concrete block wall in back of the around the oven: oven and the concrete floor under it were replaced by thick water reflectors. (2) operating personnel may be standing at the face of the oven and so a 15 cm water reflector was also placed at this surface. Since another oven may be placed next to the one being considered, a neutron reflecting boundary con-The top dition was improved adjacent to one side of the oven. and other side were taken to be nonreentrant surfaces. Thus, the calculated keff's are really those for a system of two fully loaded, partially reflected, drying ovens.

The k_{eff} of the drying ovens was calculated as a function of the amount of water in the product as was done for the problem with the filter press. However, since the method used here was Monte Carlo, it was necessary to restrict the number of calculations to the vicinity in which the maximum k_{eff} was expected to occur. Table 2 and Figure 2 give the results of the KENO calculations for this system. The maximum k_{eff} calculated was 0.806 ± 0.012 at 30% water by volume ($H/U^{235} = 150$).

3.3 The Muffle Boxes and Furnace

When the product is removed from the ovens, it has lost most of the free water present before being dried. For the first part of the muffle box analysis, it was assumed that the only water present in the product was the water of hydration of the UO₄ giving an $H/U^{235} = 80$ for the dry product. It was also assumed the bulk product density was 70% of theoretical, i.e. 3.26 gm/cm³.

While loaded with product, a muffle box may either be loaded in the furnace or placed on its stand among other boxes similarly charged.

The first problem examined was that involving the box in the furnace. Since the material and geometry description of the furnace would have been difficult to treat in the KENO input it was decided that a conservative simplification would be to replace the furnace walls by a thick water reflector wrapped around the box. The KENO calculated k_{eff} was 0.695 + 0.012.

Next, the problem of the planar array of boxes was considered. The flange on the face of the boxes provides a minimun side to side separation of 10", it was assumed that the face to back separation of adjacent boxes was zero, that is, the array is packed as tight as the physical dimensions of the boxes will permit. It was further assumed that the concrete floor, 42" below the boxes, was a thick water reflector. In KENO, the boundaries on the sides, face, and back of a single box were assigned meutron reflective boundary conditions to simulate the infinite array in the X-Y dimensions, the top of the box was considered a nonreentrant surface. The KENO calculated k_{∞} for this array was 0.651 + 0.010.

The planar array was evaluated as a function of water content in the $UO_4 \cdot 2H_2O$ product. That is, the product was not dried, but was assumed to take any form possible as in the filter press stage. The KENO calculated k_∞ as a function of array water content is shown in Figure 2. The maximum calculated k_∞ was 0.958 ± 0.011. It may be added that in order to maintain a smooth curve through the peak in the reactivity curve the U^{238} cross section set of $G_p=100$ barns was used for all points, even through this represents a high degree of conservatism in this region. A single box from this array was then surrounded by a thick water reflector, simulating the furnace, giving a calculated Since these values of keff are high, the k_{eff} of 0.955 + 0.012. peak in the reactivity curve for the muffle boxes was examined with a more accurate value of the potential scattering cross section per resonance absorber atom. In particular the Op at this peak is 202 barns. When the U^{238} cross section set corresponding to $O_p = 200b$ was used in KENO, the peak dropped from $k_{\infty} = 0.958 \pm 0.011$ to 0.914 ± 0.011 . This point is plotted on Figure 2 with the reactivity curve for the muffle boxes.

4.0 References

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- 4. Gulf United Nuclear Fuels Corporation Drawing Number 3229-124
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- Carlson, B. G., et.al., DTF Users Manual, United Nuclear Corporation, UNC Physics Math 3321, Vol. 1 (November 1963); Vol. II (May 1964).
- Hansen, G.E., and Roach, W.H., "Six and Sixteen Group Cross Sections for Fast and Intermediate Critical Assemblies," LAMS-2543, Los Alamos Scientific Laboratory (December 1960).
- 8. Whitesides, G.E., and Cross, N.F., "KENO A Multigroup Monte Carlo Criticality Program," CTC-5, Union Carbide Corporation, Nuclear Division, Computing Technology Center, Oak Ridge, Tennessee, (September 1969).
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- 10. Crume, E. C., "Some Considerations in Regard to the Development and Use of the KENO Program", CONF-680909 Proceedings of the Livemore Array Symposium, Lawrence Radiation Laboratory, pp. 18-22 (September, 1968).
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EF/ah Attachments

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DTF Results for the 18" Filter Press as a Function of Product Form and Water Content

Product	Water <u>Volume, %</u>	Corresponding <u>H/U²³⁵</u>	k
004.5H20	$\begin{array}{c} 0.0 \\ 10.0 \\ 20.0 \\ 30.0 \\ 40.0 \\ 50.0 \\ 60.0 \\ 70.0 \\ 80.0 \end{array}$	79 97 119 147 185 238 317 450 714	0.842 0.875 0.900 0.915 0.899 0.879 0.827 0.762 0.658
(NH ₄)2 ^{U207}	$\begin{array}{c} 0.0 \\ 10.0 \\ 20.0 \\ 30.0 \\ 40.0 \\ 50.0 \\ 60.0 \\ 70.0 \end{array}$	79 104 136 177 231 307 421 610	0.600 0.663 0.693 0.711 0.732 0.714 0.679 0.630

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Table 2

Summary of KENO Calculated Results for the Green Room Process Equipment

<u>Filter Press</u> - with 5 w/o U^{235} as $U0_4 \cdot 2H_2^0$ in 30 v/o water

Reflector	$k_{eff} \pm 1$ std.deviation
None	0.845 + 0.005
15 cm. water	0.859 + 0.005

Drying Ovens - with 5 w/o U^{235} as $U0_4 \cdot 2H_20$

v/o of water in Product	$k_{eff} \pm 1$ std. deviation
20.0	0.787 + 0.007
30.0	0.806 + 0.006
40.0	0.786 ± 0.006

Muffle Boxes -

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1) Normal Operating Conditions, 5 w/o U^{235} as $U0_4 \cdot 2H_20$, density = 3.26 gm/cm³, no excess water in the product.

System

 $k \pm 1$ std. deviation

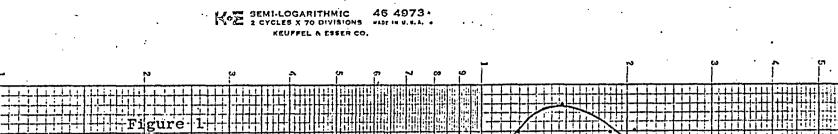
Infinite Planar Array of Boxes Single Box with 15 cm Water Reflector

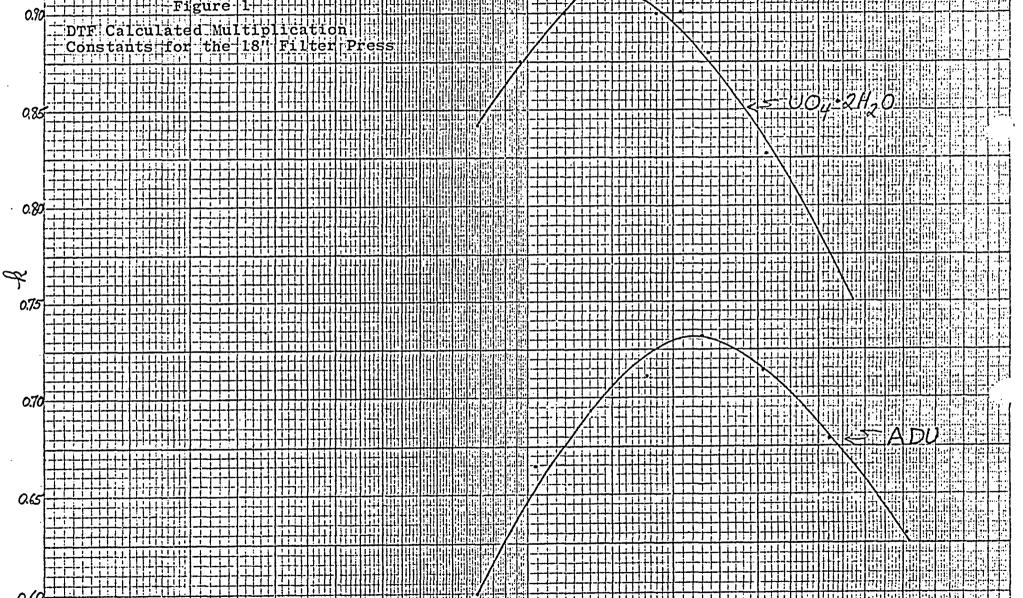
 $k_{\infty} = 0.651 \pm 0.005$ $k_{eff} = 0.695 \pm 0.006$

2) Abnormal Operating Conditions, 5 w/o U^{235} as $U0_4 \cdot 2H_20$ with excess water in the product.

Infinite Planar Array: v/o Wator in Product	$k_{\infty} \pm \text{std. deviation}$
$\begin{array}{c} 0.0 \ (/= 4.66 \ \mathrm{gm/cm}^3) \\ 20.0 \\ 30.0 \\ 40.0 \end{array}$	$\begin{array}{r} 0.875 \pm 0.005 \\ 0.925 \pm 0.006 \\ 0.958 \pm 0.006 \\ 0.933 \pm 0.006 \end{array}$

Single Box with 30 v/o water in product and 15 cm Water reflector $k_{eff} = 0.955 \pm 0.006$





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۱.1 11 Figure 2 KENO Results for the Drying Ovens and Muffle Boxes. Values are given ± 1 standard deviation 0.98 0.96 0.94 53 0.92 σp = 200 barns 0.90 PLANAR ARRAY INFINTE OF MUFFLE BOXES 0.88 0.86 ¥ 0.84 0.82 0.80 SYSTEM OF TWO PARTIALLY REFLECTED 0.78 DRYING OVENS FASS E. Ň .55 70 80 90 100 150 200

H/U²³⁵

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