CENTER FOR NUCLEAR WASTE REGULATORY ANALYSES

CORRECTIVE ACTION REQUEST

CAR No: 2005-3

Associated AR, SR, NCR No: AR 2005-1

PART A: DESCRIPTION OF CONDITION ADVERSE TO QUALITY

Contrary to the requirements of the CQAM, Section 12, Rev.4, Chg. 3, inadequate labeling or other non-compliance(s) to calibration requirements were noted:

1. Fourteen thermocouples used in the CLST laboratory were calibrated, but found without labels. (Section 12.4 (4))

The following apply to the slow-rate strain testing performed in Division 18, Building 90 Annex:

- 2. Load cell and linear displacement transducers (LVDT) calibrated by Division 18 but not labeled as to status. (Section 12.4 (4))
- 3. Thermocouple calibration by Division 18 not documented. (Section 12.4 (6))
- 4. Identification of the digital voltmeter used to take voltage potential across specimen not noted in the test documentation. Traceability not available to evaluate subsequent out of tolerance conditions. (Section 12.7)

Initiated by: R. Weber	Responsible Individual: D. Dunn
Date: May 18, 2005	Response Due: June 15, 2005

PART B: PROPOSED ACTION

1) Extent of Condition:

In addition to the 14 thermocouples, other equipment used in testing identified to be without mandatory calibration information include: (i) temperature controllers used to both control and measure test cell temperature, (ii) a metallurgical microscope used to measure localized corrosion penetration depth and (iii) two colorimeters used to conduct microbial influenced corrosion investigations
 The load cell, LVDT, thermocouple, and electrometer used in slow strain rate tests in Building 90 annex are the only are instruments owned by

2. The load cell, LVDT, thermocouple, and electrometer used in slow strain rate tests in Building 90 annex are the only are instruments owned by Division 18 that are presently used in CNWRA slow strain rate tests. Additional load cells and LVDTs owned by Division 18 have been used in previous stress corrosion cracking tests.

2) Root Cause:

1. The labeling of individual thermocouples was not considered to be practical by CNWRA staff, instead, separate calibration status information was relied upon.

2. The Division 20 staff are not clear on how to control work using equipment and personnel from other SwRI Divisions and comply with the applicable QA requirements.

3) Remedial Action:

1. Individual calibration labels were added to the group of 14 thermocouples.

2. A calibrated thermocouple meter was added to record the temperature of test cells.

Per QAP-019, the LVDT and the load cell require calibration before use. The details of the calibration procedures for these instruments will be documented in the appropriate scientific notebook. Labels will be added to the equipment to indicate calibration status (CALIBRATE BEFORE USE).
 The fine focus knob of the metallurgical microscope used to measure localized corrosion penetration depth was calibrated (i.e., verified to have 1 micrometer graduations) using a calibrated caliper. The verification of the performance was recorded in scientific notebook 308.

5. CALIBRATE BEFORE USE stickers will be added to the colorimeters used in the microbial Influenced corrosion studies.

4) Corrective Action to Preclude Recurrence:

1. Staff shall be retrained on QAP-001 and QAP-019. The requirements of these procedures has been summarized in an training guide with specific

suggestions for initial and in process entries for scientific notebooks. The guidance document provides specific instructions on the use of equipment from other Divisions.

2. Initial scientific notebook entries for ongoing tests will be revised to include additional information on test equipment calibration as required by QAP-001. The revised initial entries will be reviewed and approved by the element manager.

3. Test equipment and instrumentation in Division 18 laboratories will be classified as calibrate before use. Scientific notebook initial entries will specifically identify equipment not owned by Division 20 and the calibrations applied to that equipment. Per QAP-019, details of the calibrations or performance verifications will be documented in the scientific notebooks.

4. Staff will discontinue using the provision listed in CQAM Section 12.4.1(5) allowing calibration labels to be placed on other locations such as equipment containers when it is impractical to attach the calibration labels to the calibrated item. Instead, each calibrated item will be labeled with status information.

Manager/Director:

Date: 6/1/2005

PART C: APPROVAL Comments/Instructions

Proposed Completion Date: July 30, 2005

Proposed Completion Date: July 1, 2005

CENTER FOR NUCLEAR WASTE REGULATORY ANALYSES

CORRECTIVE ACTION REQUEST

CAR No: 2005-3 Director of QA: Do Jul	Associated AR, SR, NCR No: <u>AR 2005-1</u> Date: 6 2005
PART D: VERIFICATION OF CORRECTIVE ACTION IMPLEMENTATION See kelow	Distribution: Original-CNWRA/QA DIRECTOR QA Records ORIGINATOR PRINCIPAL INVESTIGATOR D. DVNM MANAGER J. JA: ASSISTANT DIRECTOR S. Moderaty CNWRA PRESIDENT
Verified by: What Date: 7/27/05	GEDV.P.

Remedice arben Individual calibration to as verified on remaining thermocouples.
4 Thermocouple meters are now calibrated and also to measure the temperature of test cells.
3 Calibration Before lies labels added.
4) Five focus adjustment performance verification is documented in Acientific Notebook
5. "Calibrate Before Use' stickus have been placed on the colorimeters." 1. Training on @4/00/2015 was performed on 46/05 (altader) 2. Training was reformed on OAROOI (su t1). In the Eatures were reveal (attacked) Howstrom este controlion gage blocks an colibrations locker of colipus tin was verified . 4. Calibrate lobeles when und for ease thermorene (etame attack) * Remedia actions verified by: Me SwAI QA 25.27 July 05

CAR 2005-2 Corrective Action Item / CAR 2005-3 Corrective Action Item /

GEOSCIENCES AND ENGINEERING DIVISION MEETING ATTENDANCE

SUBJECT OF MEETING: Training on Procedures QAP-001, QAP-009, QAP-016, QAP-017, QAP-018, QAP-019, and TOP-012

DATE: June 6, 2005	LOCATION: Bidg	g. 189	
PERSON	ORGANIZATION	TITLE/FUNCTION	TELEPHONE NUMBER
Kuang. Tsan Chiang	GED	Sr. Res. Scientist	210.522-2308
Lietaj Yang	GED	Sr. Res. Engineer	210 522 2483
Xihna He	OFD	Res. Scientist	210-522-5194
PAVAN SNUKLA	GED	Rep. Eugineer	210-522-6534
Yi-Ming Pan	FED	Rep. Engineer Principal Engineer	210-522-6640
VIJAY JAIN	11		210-522-5439
DARRELL DUNN	GED	Principal Engineer	210 522 6090
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Form AP-5

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CAR 2005-3 Convective Action Item 1
CAR 2005-3 Corrective Action Item 2
CAR 2005-3 Corrective Action Item 3
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Scientific Notebook Initial and In Process Entries for Laboratory Activities

From QAP-001

<u>Initial Entries</u>. Scientific notebook initial entries shall be made in the beginning of the notebook and whenever substantive changes to the objectives, approach, or methods are made. Initial Entries shall include:

- Title of the experiment, field investigation, computer simulation, or other technical task.
- Names of the individuals performing the activity.
- Description of the objectives of the task and the proposed approach or procedure for achieving the objectives.
- Special personnel training or qualification requirements

Laboratory and Field Activity initial entries shall also include, as appropriate:

- Equipment and materials to be employed during the experiment, including any necessary design or fabrication of experimental equipment and any material/chemical characterizations.
- Measurement parameters and test equipment calibration, accuracy, and precision requirements.
- As applicable, description of suitable and controlled environmental conditions.
- Potential sources of uncertainty and error.

<u>In process entries</u>. Scientific notebook in-process entries shall be made at the time work is performed. Laboratory and field activity in-process entries shall include, as appropriate:

- Evidence that experiment/test prerequisites have been met (i.e., special environmental conditions, equipment calibrations, etc).
- Description of the experiment or field investigation, including detailed description of the step-by-step process followed, either by reference to a Technical Operating Procedure (TOP) or industry Standard Method, or by description in the scientific notebook.
- Variance from Standard Methods shall be clearly identified as such and documented.
- Description of any conditions which may adversely affect the results.
- Identification of samples used and any additional equipment and materials not included as initial entries. Measuring and test equipment shall be identified by item and its calibration status documented.
- Significant data taken and a brief description of the results, to include notation of any unaccepted results.
- Any interim conclusions reached, as appropriate.

Suggested Format for Initial Entries

Title of the experiment, investigation, or other technical task

- This can be short if it is an entry for a specific type of testing. Examples:
- 1. Crevice Chemistry Experiments for Type 316L Stainless Steel
- 2. Oxidation Rate Tests for Alloy 22

Names of the individuals performing the activity

As indicated, put the names of the individuals. All staff and consultants making entries should have completed COI evaluation. Also all persons making entries should sign the front of the notebook

Description of the task objectives, proposed approach or procedure to achieve objectives

- List the objectives of the testing. Examples:
- 1. Determine the chemistry inside crevices during active crevice corrosion
- 2. Determine the oxidation kinetics and characterize intergranular oxidation of Alloy 22

The proposed approach must include specific information on how the tests will be conducted. A test plan and a test matrix should be considered.

The procedure for conducting the tests should be listed or described. If a standard test procedure is used (e.g. ASTM) the procedure number should be listed along with any deviations from the procedure. In many cases a combination of standard ASTM procedures are used along with procedures developed specifically for conducting the types of tests required. All of this information should be included. A reference to a paper or report describing the test procedure may be considered.

Special personnel training or qualification requirements

For many initial entries, this may not be necessary. This is training above and beyond the QA and standard laboratory training. Nevertheless make sure all persons making entries into the notebook have received training on QAP-001. Examples of specialty training such as radiation safety or laser safety training. Consider what type of objective evidence will be required to demonstrate training was obtained.

Equipment

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List the equipment that will be used in the tests. This must include any necessary design or fabrication of experimental equipment. Consider the measurement parameters that are required. All parameters recorded should be obtained from instruments that are either calibrated or the performance of the equipment has been verified. If no calibration or performance verification sticker is present, the instrument is not calibrated and cannot be used for the measurement of critical parameters. Status of equipment with the following labels:

CALIBRATION NOT REQUIRED: This means that the instrument is <u>not calibrated</u>. Unless a performance verification has been conducted and documented, these instruments cannot be used for measuring critical parameters.

CALIBRATE BEFORE USE: The instrument must either be calibrated (SwRI Calibration Laboratory) or the performance must be verified and documented prior to use.

CALIBRATION OVERDUE: The calibration is no longer valid and the instrument needs to be returned to the SwRI Calibration Laboratory for calibration.

Materials to be employed during the experiment

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This must include any necessary design or fabrication of test specimens and any material/chemical characterizations. The following must be included:

- Material specifications (e.g. Type 316L Stainless Steel)
- Heat Number (e.g. P80746)
- Test Specimen Description (e.g. CNWRA Drawing or ASTM standard)

Measurement parameters, test equipment calibration, accuracy, and precision requirements

List parameters that will be measured (e.g. Current density, Potential, pH, Temperature) and identify the instruments that will be used. Specific serial numbers may not be required. If specific serial numbers are included in the initial entry, then every time a different instrument is used, a explanation of the deviation from the initial entry will likely be necessary. The instruments used should either be calibrated or the performance of the instruments will need to be verified. Verification of instrument performance requires documentation with data and should use standard test methods (ASTM), accepted procedures from the manufacturer, or control tests using calibrated instruments or components. The procedure for verifying potentiostat performance (TOP-022) is a good example. A calibrated resistor is used to verify the performance of the entire system.

The equipment used must be compatible with the requirements. When in doubt, check the instrument specifications. Avoid over specification (e.g. specifying ± 0.5 °C when ± 2 °C is acceptable for the tests to be conducted) as several nonconformance reports have been written because parameters were over specified.

Equipment that is not owned by Division 20 should specifically be identified. This equipment should either be calibrated by the SwRI calibration lab or be calibrate before use. In either case, the calibration or performance verification information must be placed in the appropriate scientific notebook. At a minimum the calibration or performance verification must be conducted prior to and after the test data are acquired. Routine scheduled calibration or performance verification must be conducted for equipment that is used for extended periods. In general a 6 month interval should be used but an interval of up to 1 year may be necessary for equipment used in long term tests. The calibration status of the instrument must be clearly visible and must reference the scientific notebook where the calibration or performance verification information is documented.

As applicable, description of suitable and controlled environmental conditions

Include relevant conditions that must be controlled (e.g. test temperature) and specifications of the control. Consider how compliance with the controlled conditions will be determined and documented.

Potential sources of uncertainty and error

As applicable. This will vary significantly from one tests to the next.

Suggestions for In Process Entries

A test specific form that can be completed for each tests has many advantages and is recommended. Test forms can reduce the number of omission mistakes such as missing heat numbers and missing equipment calibration. There are many examples of test forms that have been used successfully. These may serve as examples.

While each type of test has different requirements the following information is typical of routine corrosion tests.

Test Name

1 11

- Material and Heat Number
- Significant Material Preparation (provide a notebook number and page number if applicable)
- Test Solution Preparation including:
 - Concentrations,
 - Measured reagent weights,
 - Reagent lot numbers,
 - Volume of solution prepared,
 - Any stock solutions used (provide notebook number and page number),
 - Starting and ending pH.
- Other environmental conditions such as any gasses used
- Test temperature
- Initial conditions such as
 - corrosion potentials
 - potential of Pt counter electrodes
- Instruments and calibration status
 - model number
 - serial number
 - Calibration or Performance Verification Date
 - Calibration or Performance Verification Due Date
- Values of parameters that were measured in the test
- Electronic data files collected in the test

Include any additional information necessary. If the test results are invalid, the scientific notebook entry should clearly indicate this and why the tests are invalid.

CAR 2005-3 Corrective Action Item 3

5N#695 45

Ins	titute Calibration Lab	oratory Instrument	s Processed:		
L	Stitute Calibration Lab				
A	tion Performed: CHECKIN	Date: 5/19/2005			s . n
<u>Asset</u> 0116		Workorder Model 303064334 TYPE K	Manufacturer Description DURO-SENSE THERMOC	OUPLE 0003	<u> </u>
То	tal Number of Instruments Processed:	1			
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K.J. Ching

SOUTHWEST RESEARCH INSTITUTE CALIBRATION LABORATORY MEMORANDUM

To:	DARRYL WAGAR DIV18 B128
From:	Walt Hill, Metrology Group Leader Institute Calibration Laboratory
Subject:	Status of Calibration Supplier
Ma	nufacturer/Model: MOREHOUSE 10,000 LBF
	Description: PROVING RING
	Serial Number: 3668C-709
	Asset Number: 000748
Wor	k Order Number: 444061883
	Date Calibrated: December 30, 2004
	Supplier: MOREHOUSE, YORK PA - A2LA - 717 843-0081
	Remarks: Morehouse Report # 3668CL3004.
	() Supplier is on the Approved Suppliers List (ASL).
	() Supplier is not on the Approved Suppliers List.
	(* Calibration is ISO 17025 accredited.
	() Calibration is not ISO 17025 accredited.
	() There is no known supplier to meet ISO 17025 accreditation at this time.
Please not if doc umentat	y the Institute Calibration Laboratory, extension 5215, of any discrepancies with the item or calibration ion.
Attachment	(s)

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K. T. Chiaf 6/15/06

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"As Returned"

Certificate of Calibration

and Traceability to the

United States National Institute of Standards & Technology

MOREHOUSE INSTRUMENT CO. SERIES 100: PROVING RING, SERIAL NO. 3668C 10,000 LBF CAPACITY, TENSION & COMPRESSION

The above identified instrument was calibrated in accordance with Section 7 of the American Society for Testing and Materials (ASTM) specification E74-02 entitled, "Standard Practice of Calibration of Force-Measuring Instruments...". We could not provide an "As Received" calibration due to the condition of the Proving Ring when it was returned to us. This calibration is in conformance with the requirements of Morehouse QAM Rev.7, dated 12/04/00, ISO/IEC 17025, and ANSI/NCSL Z540.1.

The result of this calibration as determined by statistical analysis according to Section 8 of ASTM E74-02, is as follows:

ι	J ncertainty	Resolution
Tension:	1.57 Lbf	0.60 Lbf
Compression:	1.61 Lbf	0.58 Lbf

This is within the Manufacturer's tolerance of 5.00 Lbf.

Calibration data is correct for a temperature of 23 degrees C. For a temperature of t degrees C, the values of the calibration factor should be corrected for temperature using the following formula.

 $d_{23} = d_t - .00027 x ((t-23) x d_t)$ Where: $d_{23} = deflection \text{ at a temperature of 23 degrees C.}$ $d_t = deflection \text{ at a temperature of t degrees C.}$ t = temperature, degrees Celsius.

This calibration is certified traceable to the United States National Institute of Standards & Technology according to the following documentation and calibration apparatus used:

Dead Weight Force Machine S/N M-4644 NIST Lab No. 822/255038-95

Uncertainty of Force Standard used to perform this calibration did not exceed +/- 0.002% of applied load

Calibrated By:

. .

MOREHOUSE INSTRUMENT COMPANY, INC. FORCE CALIBRATION LABORATORY 1742 SIXTH AVENUE YORK, PA 17403-2675 U.S.A. PHONE: 717 / 8430081 FAX: 717 / 846-4193 WEB : www.morehouseinst.com

This Certificate shall not be reproduced except in full, without written approval from Morehouse Instrument Company, Inc.

K. J. Chiaf 6/15/06

Date Calibrated:

February 16, 2004 Report No: 3668CB1604 48

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K-7 Chiaf 6/15/06

## Load Cell Calibration

Frame No. 1

Load Cell S/N:

Morohow

Morehouse Proving Ring S/N: 3668C-709

732329

Date:

4/4/2005

Last Cal.: 12/30/2004	Next Cal:	12/30/2006
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	Moorehouse	Adj. Moorehouse			Chart
Load	Ring	Ring Wts.	Meter	Computer	Table Wts.
0	4.0	0.0	0.0	-3.7	0
600	54.2	50.2	599.7	594.7	593
1200	105.7	101.7	1198.0	1194.6	1201.7
1800	155.7	151.7	1791.4	1789.2	1794.2
2400	204.3	200.3	2376.1	2376.2	2371.6
3000	252.3	248.3	2951.4	2952.4	2943.2
3600	300.5	296.5	3522.6	3524.1	3518.5
4200	348.0	344.0	4091.0	4093.0	4086.9
4800	395.0	391.0	4652.7	4655.7	4650.7
5000	425.0	421.0	5010.7	4995.0	5011.3
4800	400.3	396.3	4716.0	4719.0	4714.4
4200	361.0	357.0	4237.3	4239.2	4242.7
3600	313.7	309.7	3697.4	3699.2	3676.3
3000	264.0	260.0	3081.6	3083.1	3082.7
2400	212.5	208.5	2465.4	2465.4	2469.1
1800	160.6	156.6	1849.4	1847.5	1852.3
1200	108.0	104.0	1232.5	1229.5	1228.9
600	56.0	52.0	615.1	610.1	614.2
0	4.0	0.0	2.4	1.5	0.0

### Load Cell Calibration

Frame No2	Date:	4/4/2005
Load Cell S/N:	732328%	
Morehouse Proving Ring	S/N: <u>3660C-70</u>	9

Last Cal.: 12/30/2004

Next Cal: _____12/30/2006

	Moorehouse	Adj. Moorehouse		ſ	Chart
Load	Ring	Ring Wts.	Meter	Computer	Table Wts.
0	4.0	4.0	0.7	2.1	0
600	51.3	51.3	563.2	562.1	558.6
1200	100.6	100.6	1137.1	1136.5	1141.3
1800	148.4	148.4	1711.4	1710.3	1707.6
2400	197.0	197.0	2283.4	2282.4	2284.7
3000	245.0	245.0	2856.1	2855.1	2856.1
3600	293.3	293.3	3429	3427.2	3432.5
4200	340.0	340.0	4001.2	3999.1	3991.1
4800	387.0	387.0	4575.2	4572.3	4554.7
5000	420.0	420.0	4958.8	4955.8	4951.1
4800	385.0	385.0	4578.3	4575.3	4500.6
4200	339.9	339.9	4007.3	4004.8	3870.2
3600	292.0	292.0	3435.3	3433.5	3417
3000	243.3	243.3	2862.6	2861.5	2835.9
2400	195.0	195.0	2286	2288.6	2260.9
1800	148.0	148.0	1718.5	1716	1702.8
1200	100.6	100.6	1142.2	1140.7	1141.3
600	52.0	52.0	567.3	561.6	566.9
0	4.0	4.0	0	2	0

K.J. chiaf 6/15/06

Frame No. 3

Load Cell S/N:

Morehouse Proving Ring S/N:

Last Cal.: 12/30/2004

789126

Next Cal: 12/30/2006

4/4/2005

Load	Moorehouse Ring	Adj. Moorehouse Ring Wts.	Meter	Computer	Chart Table Wts.
0	4.0	0	0.0	-2.0	0.0
600	53.3	49.3	575.3	574.5	582.5
1200	103.6	99.6	1157.8	1156.7	1176.8
1800	153.5	149.5	1740.7	1739.3	1768.1
2400	200.1	196.1	2322.7	2319.0	2321.6
3000	247.3	243.3	2840.1	2839.4	2883.6
3600	287.3	283.3	3304.5	3304.2	3360.8
4200	332.4	328.4	3830.5	3830.4	3900.1
4800	355.6	351.6	4102.2	4102.2	4178.0
5000	374.8	370.8	4328.4	4328.9	4408.2
4800	367.6	363.6	4244.2	4243.1	4321.9
4200	342.7	338.7	3949.7	3949.9	4023.4
3600	302.0	298	3481.8	3481.7	3536.5
3000	250.3	246.3	2910.0	2909.1	2919.3
2400	203.8	199.8	2327.2	2325.6	2365.6
1800	153.3	149.3	1743.7	1742.8	1765.7
1200	103.4	99.4	1161.6	1160.6	1174.4
600	53.5	49.5	577.9	577.0	584.6
0	4.0	0	-0.9	-1.5	0.0

Date:

3668C-709

## Load Cell Calibration

Frame No. _____4____

#### Date: 4/4/2005

Load Cell S/N: 789127

Morehouse Proving Ring S/N: 3668C-709

Last Cal.: 12/30/2004

Next Cal: 12/30/2006

	Moorehouse	Adj. Moorehouse			Chart
Load	Ring	Ring Wts.	Meter	Computer	Table Wts.
0	4	0.0	0.4	0.2	0.0
600	48.3	44.3	532.6	531.6	523.2
1200	96.5	92.5	1091.6	1090.6	1092.8
1800	143.0	139.0	1640.7	1639.7	1643.5
2400	188.3	184.3	2184.9	2182.4	2181.3
3000	234.5	230.5	2727.5	2726.4	2731.0
3600	279.5	275.5	3265.1	3268.1	3267.7
4200	324.0	320.0	3803.3	3802.6	3799.6
4800	367.0	363.0	4332.4	4331.6	4314.7
5000	396.2	392.2	4672.2	4671.3	4665.1
4800	370.5	366.5	4389.4	4389.3	4356.7
4200	336.4	332.4	3944.0	3943.0	3948.0
3600	298.4	294.4	3480.5	3479.3	3493.4
3000	252.6	248.6	2945.2	2944.7	2946.7
2400	202.4	198.4	2349.0	2347.0	2349.0
1800	149.0	145.0	1758.0	1761.1	1714.7
1200	102.0	98.0	1158.0	1156.5	1157.9
600	51.5	47.5	561.3	560.4	561.0
0	4.0	0.0	2.3	1.6	0.0

K.J. Chiaf 6/15/06

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## Initial Scientific Notebook Entry for Corrosion Resistant Material Potentiostatic Tests, Potentiodynamic Tests, Slow Strain Rate Tests

Title: Potentiaostatic tests, polarization tests, slow strain rate tests (SSRT).

**Test Performed by:** Kuang-Tsan Kenneth Chiang, Walter Machowski, Brian Derby

**Objective:** Study the effect of Bicarbonate  $[HCO_3^-]$ , Carbonate  $[CO_3^{2^-}]$ Nitrate  $[NO_3^-]$  to Chloride  $[CI^-]$  concentration ratio in simulated groundwater on stress corrosion cracking susceptibility of mill-annealed Alloy 22 and Alloy 22 weldments using slow strain rate tests.

**Equipment:** EG&G Versastat Serial Number 20104. EG&G Model 352 corrosion software. NEC 586 computer. Keithley Electrometer Model 614 SN 55538 or equivalent. ASTM G-5 Polarization Cell. The SSRT equipment and instrumentation in Div. 18 laboratory is classified as calibrate before use. The SSRT system includes B&B Motor Control Type MSH-54RL Motor, Ohio Gear Model D3 Gear Box, Sensotec LVDT Model VL7A, Bodine DC Motor Speed Control (SN L4904600), 440 multichannel potentiostats with National Instruments Labview data acquisition software 4.1. CNWRA Computer Property No. 2330. Calibration and due dates are provided in data sheets for each test.

**Materials:** Mill-annealed Alloy 22, gas metal-arc welded Alloy 22, gas tungsten-arc welded Alloy 22.

**Specimen Specifications:** Cylindrical CPP specimens 1.195"x.250" and slow strain rate specimens drawings 20-06002-01-321-001 and 20-06002-01-322-002. SSRT per ASTM G129 procedure.

Measurement Parameters: Current and Potential as described in TOP-008. Temperature of solution  $\pm 2^{\circ}$ C.

**Required Level of Accuracy:** Potentials ±5mV. Current ±0.1 microamps.

Uncertainty and Source of Error: Current density calculated as current divided by sample area. Actual current density of corroding areas is not determined. Resolution limit of data acquisition systems may limit accuracy of passive current density measurements.

R. T. Climp 7/20/25

## CAR 2005-3

Corrective Action Item 4

thermocomple #16

 
 SwRi Cai Lab
 By:bit

 Cai: Apr 06, 06
 Due: Apr 06, 06

 An:000201
 Sn:203376W

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## **GEOSCIENCES AND ENGINEERING DIVISION**

## MEMORANDUM

TO: QA Files

FROM: Vijay Jain, Manager - CSPE

SUBJECT: Review of revised initial entries in active lab notebooks

I have reviewed the revised initial entries in the following active scientific laboratory notebooks and found them to have adequately incorporated revised format. This completes the actions required by Item #2 of the Corrective Action to Preclude Reoccurrence for CAR 2005-3.

- SN 670
- SN 706
- SN 718
- SN 715
- SN 695
- SN 713
- SN 617
- SN 641 (closed)
- SN 659
- SN 675
- SN 533

## Initial Scientific Notebook Entry for Corrosion Resistant Material Potentiostatic Tests, Potentiodynamic Tests, Slow Strain Rate Tests

Title: Potentiaostatic tests, polarization tests, slow strain rate tests (SSRT).

**Test Performed by:** Kuang-Tsan Kenneth Chiang, Walter Machowski, Brian Derby

**Objective:** Study the effect of Bicarbonate  $[HCO_3^-]$ , Carbonate  $[CO_3^{2^-}]$ Nitrate  $[NO_3^-]$  to Chloride  $[CI^-]$  concentration ratio in simulated groundwater on stress corrosion cracking susceptibility of mill-annealed Alloy 22 and Alloy 22 weldments using slow strain rate tests.

**Equipment:** EG&G Versastat Serial Number 20104. EG&G Model 352 corrosion software. NEC 586 computer. Keithley Electrometer Model 614 SN 55538 or equivalent. ASTM G-5 Polarization Cell. The SSRT equipment and instrumentation in Div. 18 laboratory is classified as calibrate before use. The SSRT system includes B&B Motor Control Type MSH-54RL Motor, Ohio Gear Model D3 Gear Box, Sensotec LVDT Model VL7A, Bodine DC Motor Speed Control (SN L4904600), 440 multichannel potentiostats with National Instruments Labview data acquisition software 4.1. CNWRA Computer Property No. 2330. Calibration and due dates are provided in data sheets for each test.

**Materials:** Mill-annealed Alloy 22, gas metal-arc welded Alloy 22, gas tungsten-arc welded Alloy 22.

**Specimen Specifications:** Cylindrical CPP specimens 1.195"x.250" and slow strain rate specimens drawings 20-06002-01-321-001 and 20-06002-01-322-002. SSRT per ASTM G129 procedure.

Measurement Parameters: Current and Potential as described in TOP-008. Temperature of solution  $\pm 2^{\circ}$ C.

**Required Level of Accuracy:** Potentials ±5mV. Current ±0.1 microamps.

**Uncertainty and Source of Error:** Current density calculated as current divided by sample area. Actual current density of corroding areas is not determined. Resolution limit of data acquisition systems may limit accuracy of passive current density measurements.

R.7. Chicang 7/20/25

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Initial Scientific Notebook Entry for Dry-Air Oxidation, Humid Air Oxidation, Passive Film.

**Title:** Isothermal oxidation tests, cyclic oxidation tests, dry-air oxidation with deposited salt, oxygen 16/oxygen 18 experiments, microstructure analysis.

## Test Performed by: Kuang-Tsan Kenneth Chiang

**Objective:** Study the oxidation kinetics, oxide morphology, and mechanism of passive film formation in dry air, humid air and thermal cycling conditions. Study the effect of fabrication processes such as welding and post-weld heat treatment on passive film formation.

**Equipment:** High Temperature Microbalance TherMax 700, Lindberg Furnace, Scanning Electron Microscopy (SEM), Energy Dispersive X-Ray Analysis (EDX), X-ray Diffraction (XRD), Secondary Ion Mass Spectroscopy (SIMS) of oxygen 16/oxygen 18 oxidized specimens will be performed by a qualified supplier through a purchase order.

Materials: Alloy C-22, Alloy 825

TITLE_

From Page No.

**Specimen Specifications:** ASTM G54-77 for determining total depth of attack for static oxidation testing.

Measurement Parameters: Temperature of oxidation treatment, weight change after oxidation and thermal cycling.

**Required Level of Accuracy:** Temperature  $\pm 3^{\circ}$ C for oxidation studies, weight measurement  $\pm .005$  mg.

Uncertainty and Source of Error: Duplicated oxidation specimens will be tested for each oxidation conditions.

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Witnessed & Understood by me,	Date	Invented by	Date
		Recorded by K. J. Chianf	7/21/05

Initial Scientific Notebook Entry for Corrosion Resistant Material Potentiostatic and Potentiodynamic Tests.

**Title:** Potentiaostatic tests, polarization tests, crevice repassivation tests, passive current density tests, critical pitting temperature tests, critical repassivation temperature tests.

**Test Performed by:** Kuang-Tsan Kenneth Chiang, Darrell S. Dunn, Brian K. Derby.

**Objective:** Study the effect of Bicarbonate  $[HCO_3^-]$  and Carbonate  $[CO_3^{2^-}]$  to Chloride  $[Cl^-]$  concentration ratios on corrosion resistance of candidate nuclear waste package materials.

**Equipment:** EG&G Versastat Serial Number 20104. EG&G Model 352 corrosion software. NEC 586 computer. Keithley Electrometer Model 614 SN 55538 or equivalent. ASTM G-5 Polarization Cell. Large 2 L glass cells with Teflon tops. Electrochemical Impedance Spectroscopy system including Solaritron 1260 FRA and Solaritron 1287 Potentiostat. ESC 440 multichannel potentiostats with National Instruments Labview data acquisition software or Strawberry Tree data acquisition software. Calibration and due dates are provided in data sheets for each test.

Materials: Alloy C-22

**Specimen Specifications:** Cylindrical CPP specimens 1.195"x.250" and Crevice repassivation specimens with Teflon crevice washers attached to surface.

**Measurement Parameters:** Current and Potential as described in TOP-008. Temperature of solution ±2°C.

**Required Level of Accuracy:** Potentials ±5mV. Current ±0.1 microamp.

Uncertainty and Source of Error: Current density calculated as current divided by sample area. Actual current density of corroding areas is not determined. Resolution limit of data acquisition systems may limit accuracy of passive current density measurements.

K.J. Chianf 7/21/05

Booh 675 142**Revised Initial Entries:** Title: Microbially Influenced Corrosion of Container Materials Names of the individuals performing the activity: Lietai Yang, and Brian Derby, Roger Dykstra, Geri Becker and Stuart Birnbaum (All signatures are in the back of cover page of this book). Objectives and the proposed approach or procedure: Microbially influenced corrosion of alloys in solutions containing different microbes. Two tests will be conducted, one is in electrochemical cell and the other is the total immersion test in vials. In the electrochemical cell, both coupons and coupled multielectrode array sensor (MAS) probes will be used. In the total immersion test, only coupons will be used. The coupons will be taken out at different time intervals for examination of corrosion evidence. Equipments and measurement parameters: Coupling Currents from MAS probes will be measured with the Keithly 2182 nanovolt meter and Keithley 7001 Mainframe Switch. The potentials from electrodes will be measured with a Keithley 617 electrometer SN0679628. The pH will be measured with Orion EA940, SN 2330, Temperature will be measured with a non-contact thermometer (SN 2332580201-0007). Two colorimeters (SN: 46580 (on loan from UTSA) and DR/890 SN: 031090028183) will be used as a quick means to determine rough growth of bacteria. Calibration must be performed before use. Results from these meters will not be used in any report or papers. Only the results from the cultivation method will be used in the reports. Software: In-house developed visual basic code was used to control the two Keithley meters and to store the data in a computer hard drive. The visual basic code was verified (see page 45, Scientific Note Book # 423). Test equipment calibration, accuracy, and precision requirements: The Keithley meters (7001 and 2182) and the visual program were verified on a regular basis. The latest verification was performed on 8/26/2004. See page 294, Scientific Note Book # 604 for details. Keithley 617 electrometer: SN0679628 Cal 09/13/04, Due 09/13/05; Orion EA940 pH meter, SN 2330, Cal 721/04, Due 7/21/05. Calibration with proper standards is needed before use; Non-contact thermometer SN 2332580201-0007 Cal 5/13/05, Due 5/13/06; Two colorimeters (SN: 46580 (on loan from UTSA) and DR/890 SN: 031090028183): Calibration before use is required. **Required level of accuracy:** Temperature: ±3 oC, pH: ±0.5, Potential ±5 mV; Currents: ±10 nA. **Potential Source of Uncertainty:** Temperature in room may change with climate. Heaters may be required in room during the winter season. Special personnel training or qualification requirements: None Material/chemical: Varies (see in-progress entries).

## **Revised Initial Entries for Under-salt Corrosion Tests:**

Title: Corrosion of Alloy 22 in Salt Environment at Elevated Temperatures

## Names of the individuals performing the activity:

Lietai Yang, Steve Young

## Objectives and the proposed approach or procedure:

Measurement of non-uniform corrosion rate of Alloy 22 in the liquid and vapor phases of a system containing mixed salts (NaCl-NaNO3-KNO3) at elevated temperatures. Coupling multielectrode array sensor (MAS) probes will be used in the test.

## Equipments and measurement parameters:

The signal from the MAS probes will be measured with a Keithly 2182 nanovolt meter and Keithley 7001 Mainframe Switch. The salts will be placed in an electrochemical cell which will be heated by a temperature controller. The temperature will be measured with a calibrated thermocouple (SN 293376W) and a calibrated temperature meter (SN:T-94140)

#### Software:

In-house developed visual basic code was used to control the two Keithley meters and to store the data in a computer hard drive. The visual basic code was verified (see page 45, Scientific Note Book # 423).

## Test equipment calibration, accuracy, and precision requirements:

The two Keithley meters and the visual program were verified on a regular basis. The latest verification was performed on 8/26/2004. See page 294, Scientific Note Book # 604 for details. This instrument has been used continuously for the microbially influenced corrosion test. The verification will be conducted as soon as the present test is completed and by 8/26/2005. Thermocouple: SN293376W, Cal'd: Apr 05, 2005, Due: Apr 05, 2006

Temperature Meter: SN: T-94140, Cal'd: May 12, 2005, Due: Nov 11, 2005

### **Required level of accuracy:**

Temperature: ±3 oC, Currents: ±10 nA.

### **Potential Source of Uncertainty:**

Noise from power line or nearby equipment may affect readings from the probes. Pumps and heaters should not be close to the test cell.

## Special personnel training or qualification requirements:

None

#### Material/chemical:

Probe will be made of Alloy 22 wire, solution annealed at 1121 oC for 10 minute followed by water quenching.

## Special personnel training or qualification requirements:

None

## Material/chemical:

Varies (see in-progress entries).