ATTACHMENT 4 TO NL-04-162

 $\ddot{}$

 $\ddot{}$

POLESTAR NON-PROPRIETARY CALCULATION

(42 PAGES)

ENTERGY NUCLEAR OPERATIONS, INC INDIAN POINT NUCLEAR GENERATING UNIT 3 DOCKET 50-286

PSAT 3056CT.QA.04 Non-Proprietary

L.

 \sim .

Page 1 of 11
Rev.0 $\begin{bmatrix} 1 & 2 & 3 & 4 \end{bmatrix}$

CALCULATION TITLE PAGE

CALCULATION NUMBER: PSAT 3056CT.QA.04

CALCULATION TITLE: Component Leakage Iodine Release with Boundary Layer Effect Calculation of Indian Point 3 Engineered Safety Feature

3

4

Table **of Contents**

Attachment 1: XL Spreadsheet Calculation of Pool Evaporation Rates (2 pages)

Attachment 2: XL Spreadsheet Calculation of Iodine Concentration in Bulk Gas Phase with BL Effect, Hallway Outside (4 pages)

Purpose

In accordance with the requirements of Section 4.2, Reference [1], the purpose of this calculation is to determine the Indian Point 3 (IP3) engineered safety feature (ESF) component leakage iodine release to the environment with and without consideration of the effect of the gas boundary layer that exists at the interface between the ESF liquid leakage and the bulk gas in the room into which the ESF leakage occurs (i.e., liquid-gas interface).

Methodology

The approach is to apply the Reference [2] Standard Review Plan (SRP) guidance that if the calculated flash fraction is less than 10% or if the water is less than 212 F, then an amount of iodine smaller than 10% of the iodine in the leakage may be used if justified by the actual sump pH history and ventilation rates. In addition, the effect of the gas boundary layer existing at the liquid-gas interface on the bulk gas iodine concentration has been included. The steps in the calculation are as follows:

Evaluate the elemental iodine concentration in the ESF liquid. This is a function of the \bullet core inventory of iodine, the iodine release from the core during the DBA LOCA (this iodine is assumed to go into solution in the RCS and containment sump liquid which is in turn recirculated through certain primary auxiliary building (PAB) rooms as ESF liquid), the total ESF liquid mass, and the ESF liquid pH.

PSAT 3056CT.QA.04 Page 3 of 11

- Assume that ESF liquid leaks onto the floors of certain PAB rooms during the recirculation and cumulates to form a pool. [I
	- 11

Evaluate the gaseous iodine concentration at the pool surface. II

11

Evaluate the resistance to the mass transport of iodine across the boundary layer, which is referred to as decontamination factor (DF) in this report. If

11

• Using the volumetric flow of gas from the ESF gas space to the environment and gaseous iodine concentration in the ESF gas space (with or without consideration of boundary layer DF), calculate iodine release to the environment.

The ESF liquid is circulated outside containment beginning at 6.5 hours after the initiating event per Reference [3], items 4.2-4.4. Per Assumption 4, it is assumed that leakage can occur in the RHR cells, the Sump Tank Room, the Hallway to the Sump Room, the Hallway Outside (which includes the Sump Room), the Piping Penetration Area, and the High Head Pump Room.

Assumptions

- Assumption 1: Flashing need not be considered.
- Justification: The peak post-recirculation ESF liquid temperature is about 208 F at 6.5 hours (Reference [3]). This is less than 212 F (saturation temperature at atmospheric pressure). Thus, flashing will not occur.
- Assumption 2: Iodine partitioning to the gas phase is assumed to take place instantaneously under equilibrium conditions at the ESF leakage pool surface. The partition coefficient depends only on the temperature of the liquid.
- Justification: The assumption of equilibrium conditions is made to simplify the calculation, and in addition is conservative since the gaseous iodine concentration is actually lower before reaching equilibrium condition.
- Assumption 3: The liquid is assumed to always be wvell-mixed and homogeneous.

PSAT 3056CT.QA.04 PSAT 3056CT.QA.04

- Justification: This assumption is conservative, since it over-estimates aqueous phase iodine concentration near the pool surface and therefore over-estimates the release to the gas phase.
- Assumption 4: The continuous ESF leakage (i.e., 4 gph) will be assumed to be distributed in all rooms in which ESF components are located.
- Justification: Distributing the ESF leakage involves a greater gas phase volume and thus effectively increases the partitioning of I_2 to the gas phase, since per Assumption 2, the gaseous iodine concentration at the pool surface is in equilibrium with the aqueous phase iodine, and so the larger the gas volume, the greater the gaseous I_2 content that is necessary to maintain the equilibrium gaseous concentration. In addition to resulting in a larger gaseous I_2 content, distributing the ESF leakage will increase the volumetric exchange rate to the environment. The combination of these effects will therefore lead to a conservative iodine release to the environment.
- Assumption 5: The volumetric exchange with the environment will be assumed to be the sum of the exchange rates of all ESF rooms into which leakage can occur with the rooms being well-mixed.
- Justification: As mentioned above it is conservative to assume that the flow from all ESF rooms into thhich leakage can occur is available since maximizing the exchange with the environment will maximize the iodine release. It is also conservative to assume well-mixed rooms since time for mixing would slow the exchange with the environment relative to the well-mixed case.
- Assumption 6: Per Assumption 8 below, the ESF liquid leakage is expected to pool on the floor of the area into which the leakage occurs. However, for purposes of quantifying the effect of the gas-liquid boundary layer, the ESF leakage is assumed to be in the form of a liquid **film** on the floor, walls, and structural surfaces of the ESF area into which the leakage occurs.
- Justification: In considering the effect of the boundary layer, it is conservative to maximize the area of the liquid surface which will in turn increase the mass of iodine transported across the boundary layer. Thus, while it is expected that the liquid will be pooled on the floor, this assumption provides significant conservatism.
- Assumption 7: Transient effects will be neglected in calculating the iodine concentration in the bulk gas region.
- Justification: In the boundary layer calculation, both the steady state and the transient equation for bulk gas region iodine concentration are solved and it is shown that equilibrium is reached quickly compared to the time intervals used to characterize temperature.

Page 5 of 11
Rev.0 $\begin{bmatrix} 2 & 3 & 4 \end{bmatrix}$

Assumption 8: The ESF leakage is assumed to pool on floor surfaces. [[

Page 6 of 11
Rev.0 $\begin{bmatrix} 0 & 2 & 3 & 4 \end{bmatrix}$

it

References

Reference 1: PSAT 3056CT.QA.02, "Implementing Procedure for Design Control for Calculation of Indian Point 3 Engineered Safety Feature Component Leakage Iodine Release," Rev. 0.

Page 7 of 11 $Rev.0 \, 1234$

- Reference 2: NUREG-0800, NRC Standard Review Plan, Section 15.6.5, Appendix B, "Radiological Consequences of a Design Basis Loss of Coolant Accident: Leakage from Engineered Safety Features Components Outside Containment".
- Reference 3: PSAT 3056CT.QA.03,"Plant-Specific Design Input for Calculation of Indian Point 3 ESF Component Leakage Iodine Release", Rev. 0.
- Reference 4: II

1]

- Reference 5: D. Powers and S. Burson, "A Simplified Model of Aerosol Removal by Containment Sprays," NUREG/CR 5966, June, 1993.
- Reference 6: L. Soffer et al., "Accident Source Terms for Light-Water Nuclear Power Plants," NUREG-1465, February, 1995.
- Reference 7: R. Sher and J. Jokiniemi, "NAUAHYGROS 1.0: A Code for Calculating the Behavior of Aerosols in Nuclear Plant Containments Following a Severe Accident," EPRI Report TR-102775, July, 1993.
- Reference 8: Reference 9: ll and the second state of the second state \mathbf{I} \mathbf{I} 1I
- Reference 10: "Calculation of Indian Point 2 Engineered Safety Feature Component Leakage Iodine Release with Boundary Layer Effect", PSAT 126.02CT.QA.05, Rev.1, Polestar Applied Technology, Inc.

Calculation

The basic calculational approach is described in detail in Reference [10], and only requires minor numerical changes to account for differences in some of the input data for IP3, e.g. core inventory of iodine, assumed pH, peak average pool temperature, and geometry and flow data. [1]

Н

As previously noted, external recirculation begins at 6.5 hours after the initiating event. There are 7 rooms into which ESF leakage can occur and through which there is bulk gas flow. The geometry and flow characteristics of these rooms, taken from Reference [3], are shown in Table 1. They are the volume (V) , the volumetric flow rate to environment (V) , the total surface area $(A, \text{ taken to be the sum of all wall surfaces, except for the ends since the ends are open, plus }$ additional area equal to twice the floor area to conservatively account for equipment surfaces), the length of the room (L) , the cross-section area of the room (A_c) , and flow velocity *(U, derived* from \dot{V} divided by A_c). The volumetric exchange rate with the environment, \dot{V}/V , is $(4000+4000+200+12000+7500+11700+4000) = 33400$ cfm or 2.004E6 cfh divided by the total volume of 1.02857E5 ft3 = 19.48 vol/hr.

Page 8 of 11
Rev.0 $\boxed{1}$ 2 3 4

]1

 $\begin{array}{c} \hline \end{array}$

Table 1. Geometry & flow characteristics of rooms into **which** ESF leakage can occur

a The supplied flow rate includes the flow in the RHR cells, which has been subtracted to obtain V-dot.

b Irregular shape; values chosen for length and width are arbitrary and preserve the floor area.

Table 2. **1[**

Page 9 of 11
Rev.0 $[1]$ 234

11

]] The details of the iodine release calculation are presented in Appendix B and the results are given in Table 3, which is the same as the table in Appendix B.

Table **3. Fraction of incoming iodine released to the environment during external**

recirculation beginning at 6.5 hours.

The boundary layer DFs are calculated in Attachment 2, following the method used in ref. [10]. The results are shown in Table 4 below and the minimum DF is 10.

Room	V (m ³)	V-dot (m ³ /s)	V -dot/ V (hr^{-1})	A (m ²)	L (m)	A_c (m ²)	U (m/s)	DF
RHR Room 31	75	1.89	90.7	118	5.79	13	0.146	H
RHR Room 32	113	1.89	90.7	155	5.79	19.5	0.097	
Sump Tank Room	55.6	0.094	6.1	77.7	3.35	16.6	0.0057	
Hallway to Sump Room	34	5.66	599	85.5	6.1	5.57	1.016	
Hallway Outside	617	3.54 ^a	20.7	619	13.7 ^b	45	0.079	
Penetration Area	1389	1.89	4.9	858	13.0 ^b	107	0.018	
High Head Pump Room	629	5.52	31.6	689	17.9	35	0.157	11

Table 4. **The** boundary layer DF for all PAB rooms in question

Results

In Table 5 below, the fractional releases with boundary layer effects are shown. For $pH =$ 7.27 and 7.61, a DF of IO was applied to the releases shown in Table 3 above. For pH = 7.9, the result for $DF = 1$ is shown. The results are in units of fraction of the incoming iodine in the ESF leakage into the ESF component room(s) (i.e., fraction of the total dissolved iodine in the 4 gph leakage water). In the first 6.5 hours (23400 seconds) the release is zero since recirculation has not yet started. The fractional releases depend strongly on the pH, and for the same DF, differ by an order of magnitude between $pH = 7.9$ and $pH = 7.27$. If

HI

For pH = 7.27, the average release rate (with DF = 10) is about 2.7% of the total incoming rate of iodine in the leakage water. For $pH = 7.61$, the average release rate (with $DF = 10$) is about 0.9%, while for $pH = 7.9$, the limiting value with $DF = 1$ is about 2.7%.

It is noted that the results reported here are quite conservative with respect to several effects in addition to the cooling due to heat transfer, as has been discussed previously: II

11

Table 5. Fraction of incoming iodine released to the environment during external recirculation beginning at 6.5 hours **with** boundary layer effects.

Time period	$pH = 7.9$ $DF = 1$	$pH = 7.61$ $DF = 10$	$pH = 7.27$ $DF = 10$
0 to 6.5 hours	N/A	N/A	N/A
6.5 to 8 hours	0.0268	0.00881	0.02652
8 to 24 hours	0.0278	0.00921	0.02787
1 day to 4 days	0.0274	0.00907	0.02746
4 to 30 days	0.0265	0.00881	0.02697

PSAT 3056CT.QA.04 Page 11 of 11

 Rev.0 [1] 2 3 4

Conclusions

The conclusions from this calculation are as follows:

• The iodine release from ESF leakage for IP3 is very sensitive to the assumed pH and the boundary layer DF. The fractional release for $pH = 7.27$ is approximately 2.7%. For $pH =$ 7.61 it is about 0.9%, while for $pH = 7.9$ it is about 2.7% (DF = 1). [[

]1

Page 1 of 12
Revision 1

Appendix A. [[

Revision 1

IL

Page 3 of 12
Revision 1

—...

 \mathbf{I}

Revision 1

 \mathbf{r}

Page 5 of 12
Revision 1

1[

Page 6 of 12
Revision 1

 \hfill

 $\ddot{}$

Page 7 of 12
Revision 1

 \hfill II

 \bar{z}

Page 8 of 12
Revision 1

 $\rm II$

Revision 1

 \mathbf{r}

 \mathbf{I}

Page 10 of 12
Revision 1

 \mathbf{I}

Page 11 of 12
Revision 1

 $\overline{}$

 $\hat{\boldsymbol{\theta}}$

 \mathbf{I}

 \bar{z}

Page 12 of 12
Revision 1

 $\Delta \phi$

 \mathbf{I}

PSAT 3056CT.QA.04 **PSAT 3056CT.QA.04** PSAT 3056CT.QA.04

Appendix B. Calculation of 1P3 ESF Component Leakage Iodine Release Definition of variables

 $\label{eq:1} \mathbf{f}_{\text{l}}(\mathbf{pH}) := \left[\frac{\left(\mathbf{H}(\mathbf{pH})\cdot\mathbf{mpl}^{-1}\right)^2}{\mathbf{d} + \mathbf{c}\cdot\mathbf{H}(\mathbf{pH})}\right]$ $PC(T) := 10^{(6.29 - 0.0149 \cdot T \cdot K^{-1})}$ Eq. Ia is rewritten as: **²** $I_{2L} = I_{ion}^{2} \cdot f_1(pH)$ Partition coefficient as a function of temperature (K). Eq. 2 is rewritten as:

$$
I_{2G} = I_{2L}PCT^{-1} = I_{ion}^2 \cdot f_1(pH) \cdot PCT^{-1}
$$

Other variables and given conditions are

Page 2 of 13 Revision 1

 $\rm H$

Page 3 of 13
Revision 1

---- -

[1

Page 4 of 13
Revision 1

From 23400 to 28800 sec

 $\tau_i := 23400 \text{ sec}$ $\tau_f := 28800 \text{ sec}$ $x := 0$ -mpl $Frac:= RF(\tau_i,\tau_f,\varepsilon,\lambda,T,pH,V_G,x)$ $Frac{dV}{dr} = (2.88 \times 10^4 \text{ } 5.553035 \times 10^{-5} \text{ } 7.9 \text{ } 315.2 \text{ } 0.026789)$ RFC_{H₀ Frac₄} 11

Page **5 of 13** Revision **1**

From 28800 to 30000 sec

PSAT 3056CT.QA.04 PSAT 3056CT.QA.04

Revision 1

 $\tau_i := 80000 \text{ sec}$ $\tau_f := 86400 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_j, \tau_f, \epsilon, \lambda, T, pH, V_G, x)$ $Frac{t}{s} = \begin{pmatrix} 8.64 \times 10^4 & 5.549809 \times 10^{-5} & 7.9 & 316 & 0.027761 \end{pmatrix}$ From 86400 to 90000 sec $T := 316K$ $RFC_{H_7} := Frac_4$ $\tau_j := 86400 \text{ sec}$ $\tau_f := 90000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = (9 \times 10^4 \text{ } 5.549832 \times 10^{-5} \text{ } 7.9 \text{ } 315.9 \text{ } 0.027661)$ From 90000 to 100000 sec $T := 315.9 K$ $RFC_{H_2} := Frac_4$ $\tau_i := 90000 \sec \qquad \qquad \tau_f := 100000 \sec$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $x := Frac_1 \cdot mp$: T := 315.9 K $Frac{a}{b} = \left(1 \times 10^5 \times 5.549936 \times 10^{-5} \times 7.9 \times 315.9 \times 0.027664\right)$ From 100000 to 172800 sec $\epsilon := 0.05$ $\tau_i := 100000$ sec $\tau_f := 172800$ sec $x := \text{Frac}_1 \cdot \text{mp}_2$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{t}{s} = (1.728 \times 10^5 \text{ 5.548301} \times 10^{-5} \text{ 7.9} \text{ 316.3} 0.028044)$ From 172800 to 200000 sec $\tau_i := 172800 \text{ sec}$ $\tau_f := 200000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = \left(2 \times 10^5 \times 5.548714 \times 10^{-5} \times 7.9 \times 315.9 \times 0.027651\right)$ From 200000 to 300000 sec $RFC_{H_Q} := Frac_4$ $T := 316.3K$ $RFC_{H_{10}} := Frac_4$ $T := 315.9K$ $RFC_{H_{11}} := Frac_{4}$ $\tau_i := 200000 \sec \qquad \qquad \tau_f := 300000 \sec \qquad \qquad x := Frac \cdot r$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = \left(3 \times 10^5 \text{ } 5.551902 \times 10^{-5} \text{ } 7.9 \text{ } 315.2 \text{ } 0.026998\right)$ From 300000 to 345600 sec $\tau_i := 300000$ sec $\tau_f := 345600$ sec $x := \text{Frac}_1 \cdot \text{mp}$ $\text{Frac} := \text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, \text{pH}, V_G, x)$ $Frac{T}{s} = (3.456 \times 10^5 \text{ 5.552916} \times 10^{-5} \text{ 7.9} \text{ 315} \text{ 0.026827})$ $T := 315.2K$ $RFC_{H_{12}} := Frac_4$ $T := 315 K$ $RFC_{H_{13}} := Frac_4$

PSAT 3056CT.QA.04 **PSAT 3056CT.QA.04**

Revision 1

From 345600 to 400000 sec

 $\tau_i := 345600 \text{sec}$ $\tau_f := 400000 \text{sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = (4 \times 10^5 \text{ } 5.554165 \times 10^{-5} \text{ } 7.9 \text{ } 314.8 \text{ } 0.026652)$ From 400000 to 500000 sec $\tau_i := 400000 \sec \qquad \qquad \tau_f := 500000 \sec$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (5 \times 10^5 \text{ } 5.555847 \times 10^{-5} \text{ } 7.9 \text{ } 314.7 \text{ } 0.026572)$ From 500000 to 2590000 sec $T := 314.8K$ $\text{RFC}_{\text{H}_{14}} := \text{Frac}_4$ $x := Frac_1 \cdot mp$. $T := 314.7K$ $RFC_{H_{15}} := Frac_{4}$ $\varepsilon := 0.5$ $\tau_i := 500000 \text{sec}$ $\tau_f := 2590000 \text{sec}$ $x := \text{Frac }$, mp. Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = \left(2.59 \times 10^6 \right. 5.562135 \times 10^{-5} \left. 7.9 \right. 314.5 \left. 0.026437 \right)$ Assuming that pH is 7.61, the release fraction of iodine atoms are: $\varepsilon := 2.10^{-3}$ $\lambda := 19.48 \text{ hr}^{-1}$ pH := 7.61 From 23400 to 28800 sec $\tau_i := 23400 \text{ sec}$ $\tau_f := 28800 \text{ sec}$ $x := 0$ -mpl Frac:= $RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (2.88 \times 10^4 \text{ } 5.190837 \times 10^{-5} \text{ } 7.61 \text{ } 315.2 \text{ } 0.08806)$ From 28800 to 30000 sec $\tau_i := 28800 \text{ sec}$ $\tau_f := 30000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = \left(3 \times 10^4 \right.$ 5.190364 $\times 10^{-5}$ 7.61 315.7 0.091329) From 30000 to 40000 sec $E := 0.005$ $\tau_f := 30000 \text{ sec}$ $\tau_f := 40000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = (4 \times 10^4 \cdot 5.186616 \times 10^{-5} \cdot 7.61 \cdot 315.9 \cdot 0.09187)$ $T := 314.5K$ $RFC_{H_{16}} := Frac_{4}$ $T := 315.2K$ $RFC_{M_0} := Frac_4$ $T := 315.7K$ $RFC_{M_1} := Frac_4$ $T := 315.9 K$ $RFC_{M_2} := Frac_4$

From 40000 to 50000 sec

 $\tau_i := 40000 \sec$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ τ_f := 50000 sec x := Frac₁·mp T := 316K Frac^T = $(5 \times 10^4 \text{ 5.184997} \times 10^{-5} \text{ 7.61 316 0.092117})$ RFC_{M₃ := Frac₄} From 50000 to 60000 sec $\tau_i := 50000 \sec \tau_f := 60000 \sec \times \pi = \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{dV}{dr} = (6 \times 10^4 \text{ } 5.184319 \times 10^{-5} \text{ } 7.61 \text{ } 316 \text{ } 0.092081)$ *T :=* 316K $RFC_{M_{d}} := Frac_{4}$ From 60000 to 70000 sec $\tau_i := 60000 \text{ sec}$ $\tau_f := 70000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $T := 316.2 \text{ km}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = \left(7 \times 10^4 \right.$ 5.183004 $\times 10^{-5}$ 7.61 316.2 0.092694 From 70000 to 80000 sec $\tau_i := 70000 \sec \qquad \qquad \tau_f := 80000 \sec \qquad \qquad x := \text{Frac}_i \cdot \text{mp}.$ $T := 316 \text{K}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{t}{s} = \left(8 \times 10^4 \right.$ 5.182975 $\times 10^{-5}$ 7.61 316 0.092023) From 80000 to 86400 sec RFC_{M_5} := $Frac_4$ RFC_{M_6} := $Frac_4$ $\tau_i := 80000 \sec \qquad \qquad \tau_f := 86400 \sec \qquad \qquad x := \text{Frac}, \text{mp}$ $\text{Frac} := \text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = \left(8.64 \times 10^4 \cdot 5.182982 \times 10^{-5} \cdot 7.61 \cdot 316 \cdot 0.092027\right)$ From 86400 to 90000 sec $\tau_i := 86400 \text{ sec}$ $\tau_f := 90000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (9 \times 10^4 \text{ } 5.183075 \times 10^{-5} \text{ } 7.61 \text{ } 315.9 \text{ } 0.091697)$ From 90000 to 100000 sec $T := 316K$ RFC_{M_7} := $Frac_{4}$ $T := 315.9K$ $RFC_{M_8} := Frac_4$

PSAT 3056CT.QA.04 **PSAT 3056CT.QA.04**

Revision 1

 $\tau_{\textbf{i}} := 90000 \text{sec}$ $\tau_{\textbf{f}} := 100000 \text{sec}$ $\textbf{x} := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{t}{s} = \left(1 \times 10^5 \text{ 5.183332} \times 10^{-5} \text{ 7.61 315.9 0.091709}\right)$ From 100000 to 172800 sec $\varepsilon := 0.05$ $\tau_i := 100000 \sec \qquad \qquad \tau_f := 172800 \sec$ Frac:= $RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $\text{Frac}^T = (1.728 \times 10^5 \cdot 5.179914 \times 10^{-5} \cdot 7.61 \cdot 316.3 \cdot 0.092931)$ From 172800 to 200000 sec $\tau_i := 172800 \text{ sec}$ $\tau_f := 200000 \text{ sec}$ $Frac:= RF(\tau_i,\tau_f,\epsilon,\lambda,T,pH,V_G,x)$ $Frac{T}{S}=\left(2\times 10^5 \text{ } 5.180831\times 10^{-5} \text{ } 7.61 \text{ } 315.9 \text{ } 0.091613\right)$ From 200000 to 300000 sec $x := Frac_1 \cdot mp'$ $T := 315.9K$ $RFC_{M_9} := Frac_4$ $T := 316.3K$ $RFC_{M_{10}} := Frac_{4}$ $T := 315.9K$ $x := Frac_1 \cdot mp$ $RFC_{M_{11}} := Frac_{4}$ $\tau_i := 200000 \text{sec}$ $\tau_f := 300000 \text{sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dx} = (3 \times 10^5 \cdot 5.18788 \times 10^{-5} \cdot 7.61 \cdot 315.2 \cdot 0.08954)$ From 300000 to 345600 sec $\tau_i := 300000 \text{sec}$ $\tau_f := 345600 \text{sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (3.456 \times 10^5 \cdot 5.190106 \times 10^{-5} \cdot 7.61 \cdot 315 \cdot 0.089053)$ From 345600 to 400000 sec $T := 315.2K$ $RFC_{M_{12}} := Frac_4$ $T := 315K$ $RFC_{M_{13}} := Frac_4$ $\tau_i := 345600 \text{sec}$ $\tau_f := 400000 \text{sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac^T = (4 \times 10^5 \text{ 5.192704} \times 10^{-5} \text{ 7.61 314.8 0.08852})$ From 400000 to 500000 sec $\tau_i := 400000$ sec $\tau_f := 500000$ sec $x := \text{Frac}_1 \cdot \text{mp}$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{t}{s} = (5 \times 10^5 \text{ } 5.196211 \times 10^{-5} \text{ } 7.61 \text{ } 314.7 \text{ } 0.088313)$ $T := 314.8K$ $RFC_{M_{14}} := Frac_4$ $T := 314.7K$ $RFC_{M_{15}} := Frac_4$

From 500000 to 2590000 sec $\tau_i := 500000$ sec $\tau_f := 2590000$ sec Frac:= $RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $\epsilon := 0.5$ $x := Frac_1 \cdot mp!$ $T := 314.5K$ $Frac{d}{dt} = \left(2.59 \times 10^6 \right.$ 5.208562× 10⁻⁵ 7.61 314.5 0.088067) Assuming that pH is 7.27, the release fraction of iodine atoms are: $\varepsilon := 2.10^{-3}$ $\lambda := 19.48 \text{ hr}^{-1}$ pH := 7.27 From 23400 to 28800 sec $\tau_i := 23400 \text{ sec}$ $\tau_f := 28800 \text{ sec}$ $x := 0$ -mp! Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{dV}{dr} = \left(2.88 \times 10^{4} \quad 4.143422 \times 10^{-5} \quad 7.27 \quad 315.2 \quad 0.265162\right)$ From 28800 to 30000 sec $\tau_i := 28800 \text{ sec}$ $\tau_f := 30000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $RFC_{M_{16}} := Frac_{4}$ $T := 315.2K$ $RFC_{L_0} := Frac_4$ $T := 315.7K$ $\text{Frac}^* = (3 \times 10^7 \cdot 4.139628 \times 10^7 \cdot 7.27 \cdot 315.7 \cdot 0.278069)$ From 30000 to 40000 sec $\tau_i := 30000 \text{ sec}$ $\tau_f := 40000 \text{ sec}$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $\varepsilon := 0.005$ $x := Frac_1 \cdot mp'$ $RFC_{L_1}:= \text{Frac}_{4}$ T:=315.9 *K* $Frac{a}{b} = (4 \times 10^4 \quad 4.12693 \times 10^{-5} \quad 7.27 \quad 315.9 \quad 0.278577)$ From 40000 to 50000 sec $RFC_{L_2} := Frac_4$ $\tau_j := 40000 \sec \qquad \qquad \tau_f := 50000 \sec \qquad \qquad x := \text{Frac}_1 \cdot \text{mp} \qquad \qquad T := 316 \text{ K}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = \left(5 \times 10^4 \quad 4.123248 \times 10^{-5} \quad 7.27 \quad 316 \quad 0.278745\right)$ From 50000 to 60000 sec RFC_{L_3} := $Frac_{4}$ $\tau_i := 50000 \sec \qquad \qquad \tau_f := 60000 \sec \qquad \qquad x := \text{Frac}_i \cdot \text{mp}$ T:= 316 K $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$

Frac^T = $(6 \times 10^4 \text{ } 4.122018 \times 10^{-5} \text{ } 7.27 \text{ } 316 \text{ } 0.278457)$ RFC_{L₄} := Frac₄ From 60000 to 70000 sec $\tau_i := 60000 \text{ sec}$ $\tau_f := 70000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $T := 316.2 \text{ kg}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{1}{2} = (7 \times 10^{4} \text{ } 4.11913 \times 10^{-5} \text{ } 7.27 \text{ } 316.2 \text{ } 0.280131)$ RFC_{L₅ := Frac₄} From 70000 to 80000 sec $\tau_{\text{i}} := 70000 \text{ sec}$ $\tau_{\text{f}} := 80000 \text{ sec}$ $x := \text{Frac}$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (8 \times 10^4 \text{ } 4.119493 \times 10^{-5} \text{ } 7.27 \text{ } 316 \text{ } 0.278025)$ From 80000 to 86400 sec $T := 316K$ $RFC_{L₆} := Frac₄$ τ_i:= 80000sec **τ_f:= 86400sec x:= Frac_i·mp** T:= 316 K Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ Frac¹ = (8.64× 10^{*} 4.119662× 10⁻³ 7.27 316 0.278071) RFC_{L₇ := Frac₄} From 86400 to 90000 sec $\tau_i := 86400 \text{ sec}$ $\tau_f := 90000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{dV}{dr} = (9 \times 10^4 \text{ } 4.120024 \times 10^{-5} \text{ } 7.27 \text{ } 315.9 \text{ } 0.277104)$ From 90000 to 100000 sec $T := 315.9K$ $RFC_{L_R} := Frac_4$ $\tau_i := 90000 \text{ sec}$ $\tau_f := 100000 \text{ sec}$ $x := \text{Frac}_i \cdot \text{mp}$ $T := 315.9 \text{ K}$ Frac:= $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = (1 \times 10^5 \text{ } 4.120831 \times 10^{-5} \text{ } 7.27 \text{ } 315.9 \text{ } 0.277203)$ From 100000 to 172800 sec $\epsilon = 0.05$ $\tau_i := 100000$ sec $\tau_f := 172800$ sec $x := \text{Frac}_1 \cdot \text{mp}$ $Frac:= RF(\tau_i,\tau_f,\epsilon,\lambda,T,pH,V_G,x)$ $Frac{d}{dt} = (1.728 \times 10^5 \text{ } 4.11355 \times 10^{-5} \text{ } 7.27 \text{ } 316.3 \text{ } 0.280518)$ From 172800 to 200000 sec $RFC_{L_0} := Frac_4$ $T := 316.3K$ $RFC_{L_{10}} := Frac_{4}$

PSAT 3056CT.QA.04 **PSAT 3056CT.QA.04**

Revision **1**

 $\tau_i := 172800 \text{ sec}$ $\tau_f := 200000 \text{ sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ Frac¹ = $(2 \times 10^3 \text{ } 4.116181 \times 10^{-3} \text{ } 7.27 \text{ } 315.9 \text{ } 0.276476)$ RFC_{L₁₁} := Frac_{*L*} From 200000 to 300000 sec $T := 315.9K$ $\tau_f := 200000$ sec $\tau_f := 300000$ sec $x := Frac_1 \cdot mp$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = (3 \times 10^5 \quad 4.132961 \times 10^{-5} \quad 7.27 \quad 315.2 \quad 0.271233)$ From 300000 to 345600 sec $\tau_i := 300000 \sec \qquad \qquad \tau_f := 345600 \sec \qquad \qquad x := Frac_i \cdot mp!$ $Frac := RF(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dx} = \left(3.456 \times 10^5 \quad 4.137765 \times 10^{-5} \quad 7.27 \quad 315 \quad 0.270537\right)$ From 345600 to 400000 sec $T := 315.2K$ $RFC_{L_{12}} := Frac_4$ $T := 315K$ $RFC_{L_{13}} := Frac_4$ $\tau_i := 345600 \text{sec}$ $\tau_f := 400000 \text{sec}$ $x := \text{Frac}_1 \cdot \text{mp}$ $Frac:= RF(\tau_i, \tau_f, \epsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = \left(4 \times 10^5 \quad 4.143152 \times 10^{-5} \quad 7.27 \quad 314.8 \quad 0.269334\right)$ From 400000 to 500000 sec $\tau_i := 400000 \sec \qquad \qquad \tau_f := 500000 \sec \qquad \qquad x := Frac_i \cdot mp'$ Frac := $\text{RF}(\tau_i, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{d}{dt} = (5 \times 10^5 \text{ } 4.150024 \times 10^{-5} \text{ } 7.27 \text{ } 314.7 \text{ } 0.269196)$ From 500000 to 2590000 sec $\epsilon := 0.5$ $T := 314.8K$ $RFC_{L_{14}} := Frac_{4}$ $T := 314.7K$ $RFC_{L_{15}} := Frac_4$ $\tau_i := 500000$ sec $\tau_f := 2590000$ sec $x := \text{Frac}_i \cdot \text{mp}$ $Frac := RF(\tau_1, \tau_f, \varepsilon, \lambda, T, pH, V_G, x)$ $Frac{a}{b} = \left(2.59 \times 10^6 \quad 4.169013 \times 10^{-5} \quad 7.27 \quad 314.5 \quad 0.269715\right)$ $T := 314.5K$ $RFC_{L_{16}} := Frac_4$

Iodine release fractions during **X/Q** time Intervals:

Page 13 of 13 Revision 1

$$
t_{i+2} := (i - 8) \cdot 10000(
$$

\n
$$
t_{13} := 34560(
$$

\n
$$
t_{16} := 2.5910^{6}
$$

\n
$$
Ave(i_{0}, i_{1}, RFC, t) := \begin{cases} \text{dummy} \leftarrow 0 \\ \text{for } j \in i_{0} \dots i_{1} \\ \text{dummy} \leftarrow \text{dummy} + RFC_{j}(t_{j} - t_{j-1}) \end{cases}
$$

\n
$$
\frac{\text{dummy}}{t_{i_{1}} - t_{i_{0}-1}}
$$

From 6.5 to 8 hour:
\n
$$
RFC_{H_0} = 0.0268
$$
 $RFC_{M_0} = 0.0881$ $RFC_{L_0} = 0.26516$
\nFrom 8 to 24 hour:
\n $Ave(1,7,RFC_H,t) = 0.02778$ $Ave(1,7,RFC_M,t) = 0.09213$ $Ave(1,7,RFC_L,t) = 0.27869$
\nFrom 1 to 4 days:
\n $Ave(8,13,RFC_H,t) = 0.02737$ $Ave(8,13,RFC_M,t) = 0.09074$ $Ave(8,13,RFC_L,t) = 0.27458$
\nFrom 4 to 30 days:
\n $Ave(14,16,RFC_H,t) = 0.02645$ $Ave(14,16,RFC_M,t) = 0.08809$ $Ave(14,16,RFC_L,t) = 0.26968$

Summary of the results for iodine release fractions

References:

1. PSAT 3056CT.QA.03, "Plant Specific Design Input for Calculation of Indian Point 3 ESF Component Leakage Iodine Release" Rev. 0.

2. L. Soffer et al., "Accident Source Terms for Light-Water Nuclear Power Plants, " NUREG-1465, February, 1995.

PSAT 3056CT.QA.04
Attachment 1 Attachment 1 Rev.

2 1

 \hfill [[

PSAT 3056CT.QA.04 Page 2 of 2
Attachment 1 Attachment 1 Rev. 1

[[

Attachment 2 XL Spreadsheet Calculation of Iodine Concentration In Bulk Gas with BL Effect, Hallway Outside

It

PSAT 3056CT.QA.04
Attachment 2 Attachment 2 Rev.

fl

 \mathbf{I}

PSAT 3056CT.QA.04 PSAT 3056CT.QA.04 Page 3 of 4 Attachment 2

 $\overline{}$

Attachment 2 XL Spreadsheet Calculation of Iodine Concentration In Bulk Gas with BL Effect

 $\,$ [[

 $\mathbf{l}% =\mathbf{l}^{T}\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf{v}^{T}\mathbf{v}^{T}+\mathbf$

PSAT 3056CT.QA.04 Attachment 2

Page 4 of 4 Rev. 1

Attachment 2 XL Spreadsheet Calculation of Iodine Concentration In Bulk Gas with BL Effect

Calculation of elemental iodine concentration at liquid edae of gas bounday laver

 \mathbf{I}

ATTACHMENT 5 TO NL-04-162

COMMITMENTS

ENTERGY NUCLEAR OPERATIONS, INC INDIAN POINT NUCLEAR GENERATING UNIT 3 DOCKET 50-286

NL-04-162 Docket 50-286 Attachment 5 Page 1 of 1

 $\overline{}$

 \sim 100 \pm 100 \pm

COMMITMENTS

